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Singly and doubly charged clusters of (S^2) metals. Ab initio and model calculations on Mg_n^+ and Mg_n^{++} $(n \le 5)$

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Résumé. — Des calculs *ab initio* précis incluant la corrélation électronique ont été effectués pour Mg_2^+ et Mg_2^{++} . Le deuxième système présente un minimum local. A l'aide d'un procédé de diabatisation, on a pu définir un hamiltonien modèle simple porté par les produits des configurations atomiques s², s¹ et s⁰. L'hamiltonien modèle inclut les effets de polarisation instantanée (ainsi que les effets à trois corps) et la répulsion coulombienne des charges positives pour les agrégats doublement chargés. Il peut être considéré comme une représentation de un ou deux trous actifs dans une bande s polarisable. Le modèle prédit que la structure la plus stable de Mg_3^+ et Mg_3^{++} . Pour Mg_5^+ et Mg_5^{++} existent également deux minima quasiment dégénérés : le tétraèdre centré et le carré centré. L'existence de telles structures a été confirmée par des calculs Hartree-Fock, par minimisation directe de l'énergie et calcul de la matrice des constantes de force pour la conformation optimale, mais on démontre que dans l'approche du modèle à particules indépendantes les longueurs de liaisons sont trop longues et les structures centrées sont trop stables. L'énergie du minimum de Mg_5^{++} est située à 1,71 eV au-dessus de la somme des énergies des systèmes Mg_3^+ et Mg_2^+ . Les potentiels d'ionisation verticaux ont été calculés et on démontre que Mg_4^+ , Mg_5^+ et Mg_5^{++}

Abstract. — Accurate *ab initio* CI calculations on Mg_2^+ and Mg_2^+ have been performed. The second appears to have a local minimum. Through a diabatization procedure, it was possible to define a simple model Hamiltonian spanned by products of atomic s^2 , s^1 and s^0 configurations. The model Hamiltonian includes instantaneous polarization energies (and three-body effects) and Coulombic repulsion of the holes in doubly charged clusters, and may be considered as treating one or two active holes in the polarizable s band. The model predicts linear Mg_3^+ and Mg_3^{++} structures and nearly degenerate square and centred equilateral triangular structures for Mg_4^+ and Mg_4^{++} ; Mg_5^+ and Mg_5^{++} both present nearly degenerate minima of centred tetrahedron and centred square structures. The existence of such structures has been confirmed by UHF *ab initio* (gradient + force) calculations; it is shown however that single-determinantal approaches overestimate the bond lengths and unduly favour the centred structures; the Mg_5^{++} minimum is 1.71 eV above the $(Mg_3^+ + Mg_2^+)$ dissociation limit and 1.41 eV under the (3 Mg + 2 Mg^+) dissociation limit. The vertical Ionisation Potentials (IP) have been calculated and it is shown that Mg_4^{++} , Mg_5^+ and Mg_5^{++} may evaporate a neutral atom.

1. Introduction.

The atoms of column II of the periodic table have ground state $s^2({}^1S)$ configurations. The atoms may be considered to keep essentially this ground state structure in small clusters, despite some weak hybridization with the $sp({}^{3,1}P)$ configuration. The electronic population of a neutral cluster of such elements may thus be seen as an (s) full band. The positive ion may consequently be treated as a single hole problem, delocalized on the different sites of the cluster through a single-hole operator. A doubly charged cluster may be considered as a two-hole system in the s band, and treated like a two-electron problem, including correlation. The present paper proposes such simplified models for singly and doubly charged clusters of (s^2) atoms. For Mg clusters the diagonal effective energies and interactions are extracted from accurate MO-CI calculations on Mg_2^{n+} (n = 1, 2) and from the experimental Mg_2 potential curve. The problem received an increased interest from the recent observation of an Hg_5^{++} doubly charged cluster [1], from electron impact on a beam of mercury clusters. The appearance at 22 eV of such a peak in the mass spectrometer analysis was quite surprising since the Coulombic repulsion between the two holes should compel such a small cluster to explode into two singly charged fragments. Brechignac *et al.* [1] suggested that the positive charge in this cluster might be concentrated on the central atom of a compact cluster, the Hg^{++} atom (the ionization potential of which is 29 eV) inducing stabilizing polarization forces on the surrounding neutral atoms.

The present work does not concern Hg clusters since the correct treatment of such a heavy atom requires the inclusion of relativistic effects; their treatment is in progress in our laboratory through relativistic pseudopotentials [2, 3], which will be applied further to this precise problem through calculations of Hg₂, Hg₂⁺, Hg₂²⁺ potential curves. The present paper is limited to the isoelectronic problem of Mg clusters, in order to see wether an Mg₅²⁺ cluster may exist at least as a stable local minimum. The question is treated through both model and *ab initio* MO calculations, which compare fairly well.

2. A simplified model Hamiltonian for Mg_n^+ and Mg_n^{++} clusters.

2.1 NEUTRAL INTERACTIONS. — Let us assume that in the neutral ground state, each atom essentially keeps an (s^2) configuration. This means that the cluster is essentially held by dispersion forces. This also means that the cluster should have a compact form, maximising the number of nearest neighbours through a packing of tetrahedra. The dissociation energy of Mg_n into n Mg atoms should be directly related to the number p of nearest neighbours. Assuming an additivity of repulsion energies and dispersion forces (i.e. cancellation between three-body repulsions and three-body dispersive energies) the dissociation energy per atom $D_e^{(n)}$ of the cluster should be

$$D_{\rm e}^{(n)} = p D_{\rm e}/2 \tag{1}$$

where D_e is the dissociation energy of the dimer. This quantity is rather weak for Mg₂ ($D_e = 0.05 \text{ eV}$) and equation (1) certainly underestimates the dissociation energy of the cluster since the cohesive energy of the metal is 1.1 eV, much larger than $(12/2) D_e/2$, which would be the asymptotic limit of equation (1). This may be due to the involvement of the p band, to attractive interaction with second neighbours and/or an attractive balance between many-body forces. It is clear however that the dissociation energy of Mg_n is much smaller than that of the corresponding positive cluster Mg_n⁺, which involves delocalization and polarization energies. The precise research of the geometry and energy of the weak Mg_n neutral cluster is not in the scope of our model, and we simply assume that its wave function is dominated by products of (s^2) atomic configurations

$$\psi_0 = \mathcal{A} \prod_{i=1}^n \varphi_i (2 \, i - 1, 2 \, i) + \cdots$$
(2)

where A is an antisymmetrizer, φ_i is either an $s_i \overline{s_i}$ product of 3s atomic orbitals, or an angularly correlated function

$$\varphi_i = \lambda \cdot |s_i \overline{s_i}| + \mu \cdot (|x_i \overline{x_i}| + |y_i \overline{y_i}| + |z_i \overline{z_i}|).$$

The interaction energy between neutral atoms i, j at distance r_{ij} is supposed to be equal to the interaction energy of the dimer $R(r_{ij}) = R_{ij}$ and the total cohesive energy would be

$$E^{n} = \sum_{i>j} R(r_{ij}) = \sum_{i>j} R_{ij}.$$
 (3)

2.2 MODEL HAMILTONIAN FOR THE SINGLY POSITIVE CLUSTER. — For the positive cluster Mg_n^+ one may consider, according to the Valence Bond (VB) description, a series of *n* localized functions, where the hole concerns the various atoms. If the hole is on the *i*th atom

$$\phi_i^+ = \mathcal{A}\left[\prod_{j < i} \varphi_j (2j - 1, 2j)\right] \times \\ \times s_i (2i - 1) \cdot \left[\prod_{k > i} \varphi_k (2k - 2, 2k - 1)\right] = a_{\overline{i}} \psi_0 \quad (4)$$

where a_i is an annihilation operator of a β spin electron on the site *i*. Then the wave function of the cluster is a linear combination of the ϕ_i^+ 's

$$\psi_{\lambda}^{+} = \sum_{i} c_{i\lambda} \phi_{i}^{+}$$

$$\overline{H} \psi_{\lambda}^{+} = E_{\lambda}^{+} \psi_{\lambda}^{+}.$$
(5)

The matrix elements of the $n \times n$ model Hamiltonian \overline{H} are easily obtained as follows. For the diagonal elements, one may write

$$\langle \phi_i^+ | \overline{H} | \phi_i^+ \rangle = \sum_{(j>k)\neq i} R_{jk} + \sum_{j\neq i} R_{ij}^+ + IP_1.$$
 (6)

 R_{jk} represents the instantaneous interaction between neutral centres, supposed to be equal to that in the corresponding neutral systems. R_{ij}^+ is the interaction between the positive atom *i* and the neutral atom *j*. Its determination will be given below, and IP_1 is the lowest ionization potential of the atom.

The off-diagonal element

$$\langle \phi_i^+ | \overline{H} | \phi_j^+ \rangle = -F_{ij}^+$$
 (7)

represents the hopping integral between atoms *i* and *j*. If φ_i is reduced to $s_i \overline{s_i}$, then

$$\langle \phi_i^+ | \overline{H} | \phi_j^+ \rangle = \langle a_{\overline{i}} \psi_0 | \overline{H} | a_{\overline{j}} \psi_0 \rangle = -F_{ij} \quad (8)$$

where F is the Fock operator of the ground state. In fact (see below) we have used effective F_{ij}^+ integrals which introduce some reorganization effects.

For Mg_2^+ the problem reduces to a 2 × 2 matrix

$$\begin{pmatrix} R_{12}^{+} & -F_{12}^{+} \\ & R_{12}^{+} \end{pmatrix}$$
which generates two roots ${}^{2}\Sigma_{u}^{+}, {}^{2}\Sigma_{g}^{+} \\ E({}^{2}\Sigma_{u}^{+}) = R_{12}^{+} + F_{12}^{+} \\ E({}^{2}\Sigma_{g}^{+}) = R_{12}^{+} - F_{12}^{+}$

and thus the functions R_{12}^+ and F_{12}^+ are immediately extracted from the knowledge of the exact potential curves of the diatomic cation

$$R_{12}^{+} = \left(E(^{2}\Sigma_{u}^{+}) + E(^{2}\Sigma_{g}^{+}) \right) / 2.$$
(9)

$$F_{12}^{+} = \left(E(^{2}\Sigma_{u}^{+}) - E(^{2}\Sigma_{g}^{+}) \right) / 2.$$
 (10)

Two things should be noticed :

i) the method is identical, at this step, to a Diatom in Molecules (DIM) procedure [4];

ii) it includes polarization forces since R_{12}^+ involves both a purely repulsive part and an $\alpha_1/2 r^4$ polarization energy of the adjacent atom by the positive charge. Since the singly ionic cluster only bears one positive charge the polarization energy keeps an additive form.

2.3 MODEL HAMILTONIAN FOR THE DOUBLY CHARGED CLUSTERS. — For a two-hole cluster, one may imagine two types of elementary VB situations.

— The two holes may be on different atoms, generating a function ϕ_{ij}^{++} . For $S_z = 0$ one must introduce two determinants :

$$\phi_{i\bar{j}}^{\pm +} = \mathcal{A}\left(\prod_{k\neq i,j} \varphi_k \times s_i \cdot \overline{s}_j\right) = a_{\bar{i}} a_{j,} \psi_0. \quad (11)$$

$$\phi_{\bar{i}j}^{++} = \mathcal{A}\left(\prod_{k\neq i,j} \varphi_k \times \overline{s_i}, s_j\right) = a_i a_{\bar{j}} \psi_0. \quad (12)$$

Their number is thus n(n-1). Their effective energy

$$\langle \phi_{ij}^{\pm +} | \overline{H} | \phi_{ij}^{\pm +} \rangle = 2 IP_1 + \sum_{(k>l)\neq i,j} R_{kl} + \sum_{k\neq i,j} (R_{ki}^{\pm} + R_{kj}^{\pm}) + R_{ij}^{\pm +} + \delta_{pol}$$
 (13)

involves, besides the energy $2 IP_1$ needed to ionize two atoms, the interactions between the instantaneously neutral atoms, the interactions between the positive and the neutral atoms, and the Coulombic repulsion between the holes, R_{ij}^{++} which weakly deviates from r_{ij}^{-1} . The last term δ_{pol} is a correction to the polarization energy which is not additive, since the field created by the two holes on the neutral polarizable atom k must be combined vectorially. Let us call $\mathbf{E}_{ki} = \mathbf{r}_{ki}/r_{ki}^3$ the field created by the hole *i* on the neutral atom k. Then the polarization energy is

$$E_{\text{pol}} = \sum_{k \neq i,j} \left[-\frac{\alpha_1}{2} (\mathbf{E}_{ki} + \mathbf{E}_{kj})^2 \right].$$
(14)

Since R_{ki}^+ involved a term $-\frac{\alpha_1}{2}E_{ki}^2, \delta_{pol}$ is given by

$$\delta_{\text{pol}} = \sum_{k \neq i,j} \left[-\alpha_1 \, \mathbf{E}_{ki} \cdot \mathbf{E}_{kj} \right]. \tag{15}$$

It gives the three-body correction to the polarization energy.

— The two holes may be on the same atom, which is now s^0 , while the others are neutral (s^2). The corresponding VB function is

$$\phi_{i\bar{i}}^{++} = \mathcal{A}\prod_{k\neq i} (\varphi_k) = a_{\bar{i}} a_i \psi_0.$$
 (16)

The diagonal energy associated with this function is

$$\langle \phi_{ii}^{\pm +} | \overline{H} | \phi_{ii}^{\pm +} \rangle = IP_2 + \sum_{(k>l)\neq i} R_{kl} + \sum_{k\neq i} R_{ki}^{2+}.$$
(17)

 IP_2 is the double-ionization potential

$$IP_2 = E(Mg^{++}(3s^0)) - E(Mg(3s^2))$$

 R_{ki}^{2+} is an effective interaction between Mg and Mg⁺⁺, and its practical extraction will be given below.

One may conceive two main types of off-diagonal interactions :

i)
$$\langle \phi_{i\bar{j}}^{++} | \overline{H} | \phi_{i\bar{k}}^{++} \rangle = F_{(i)j\bar{k}}^{++} \cong F_{j\bar{k}}^{+}$$
. (18)

The hopping integral corresponding to the jump of a hole from the site j to the site k may be taken equal to its value for the singly positive cluster. Actually

$$F_{(i)ik}^{++} = F_{ik}^{+} - \langle j | J_i | k \rangle$$

where J_i is the Coulombic operator relative to the (3s) atomic orbital of the atom *i*; the three-centre integral should be small compared to the monoelectronic (essentially kinetic) integral.

One may neglect the bielectronic integrals (corresponding to $2e^{-}$ jump) such as

$$\langle \phi_{i\bar{k}}^{++} | H | \phi_{p\bar{q}}^{++} \rangle = 0$$
 if $i \neq p$ and $k \neq q$

(even when i = q and p = k); this approximation is usual in chemistry, in the various Zero Differential Overlap (ZDO) systematics [5, 6], and in Solid State Physics in the Hubbard Hamiltonian [7] which is the crudest version of this approximation; the ZDO approximation consists in neglecting the bielectronic interaction involving overlap distributions.

ii)
$$\langle \phi_{j\bar{j}}^{\pm +} | \overline{H} | \phi_{i\bar{i}}^{\pm +} \rangle = -F_{ij}^{2+},$$
 (19)

represents the hopping integral for the jump of a hole

Nº 8

from a doubly charged atom to an (adjacent) neutral atom, this jump creating a pair of positively charged atoms. In a simple model one may write

$$F_{ij}^{2+} = -\langle a_{\overline{j}} a_{j} \psi_{0} | \overline{H} | a_{\overline{i}} a_{i} \psi_{0} \rangle = -\langle i | F - J_{i} | j \rangle.$$
(20)

In that precise case the integral $\langle i | J_i | j \rangle$ cannot be neglected, since it only involves two centres, and $F_{ij}^{2+} \neq F_{ij}^{+}$.

$$\begin{vmatrix} a\overline{b} & | & b\overline{a} \\ 2 IP_1 + R_{ab}^{++} & K_{ab} \\ & 2 IP_1 + R_{ab}^{++} \end{vmatrix}$$

The two-hole model requires the knowledge of three supplementary functions, R_{ij}^{++} , R_{ij}^{2+} and F_{ij}^{2+} . This information may be obtained from the potential curves of the doubly charged diatom Mg₂⁺⁺. In that case one has two electrons in two orbitals, and the problem is isomorphic to the H₂ problem. There are four determinants $|a\overline{b}|$, $|b\overline{a}|$, $|a\overline{a}|$ and $|b\overline{b}|$ where *a* and *b* are the two 3s AO's of the two atoms A and B. The matrix takes the form

| $ a\overline{a} $ | $ b\overline{b} $ |
|----------------------|----------------------|
| F_{ab}^{2+} | F_{ab}^{2+} |
| F_{ab}^{2+} | F_{ab}^{2+} |
| $IP_2 + R_{ab}^{2+}$ | K _{ab} |
| | $IP_2 + R_{ab}^{2+}$ |

The bicentric exchange integrals K_{ab} may be neglected.

The triplet state $(|a\overline{b}| - |b\overline{a}|)/\sqrt{2}$, of ${}^{3}\Sigma_{u}^{+}$ symmetry has an energy

$$E(^{3}\Sigma_{u}^{+}) = 2 IP_{1} + R_{ab}^{++}$$
(21)

which determines R_{ab}^{++} (or the deviation of the repulsion from its coulombic part). The singlet ${}^{1}\Sigma_{u}^{+}$ state $(|a\overline{a}| - |b\overline{b}|)/\sqrt{2}$ has an energy

$$E(^{1}\Sigma_{u}^{+}) = IP_{2} + R_{ab}^{2+}$$
(22)

which determines the function R^{2+} .

The $({}^{1}\Sigma_{g}^{+})$ ground state energy is given by a second degree equation

$$\frac{|a\overline{b}| + |b\overline{a}|/\sqrt{2} \qquad |a\overline{a}| + |b\overline{b}|/\sqrt{2}}{E(^{3}\Sigma_{u}^{+}) - x \qquad 2F_{ab}^{2+}} \qquad \begin{pmatrix} c_{1} \\ c_{2} \end{pmatrix} = 0$$
$$E(^{1}\Sigma_{u}^{+}) - x \qquad \end{pmatrix}$$

and the knowledge of the ${}^{1}\Sigma_{g}^{+}$ ground state energy $E({}^{1}\Sigma_{g}^{+})$ will determine the value of F_{ab}^{2+} as a function of the interatomic distance

$$F_{ab}^{2+} = -\frac{1}{2} \left[E({}^{3}\Sigma_{u}^{+}) - E({}^{1}\Sigma_{g}^{+}) \right]^{1/2} \times \left[E({}^{1}\Sigma_{u}^{+}) - E({}^{1}\Sigma_{g}^{+}) \right]^{1/2}.$$
 (23)

2.4 DIATOMIC POTENTIAL CURVES OF Mg_2 , Mg_2^+ AND Mg_2^{++} AND *r* DEPENDENCE OF THE 6 PARAMETERS OF THE MODEL HAMILTONIAN. — The model involves 6 parameters *R*, R^+ , F^+ , R^{++} , R^{2+} , F^{2+} . The first is identical to the ground state potential energy of the neutral dimer and rather than calculating it, a function fitted to the experimental spectroscopic parameters [8-9] has been used

$$R = \mathcal{M}(r) \exp[a_4 \exp(a_1 - r)]$$
(24)

where \mathcal{M} is a Morse function involving three parameters $a_1 a_2 a_3$ (see Table I). The second (R^+) and third (F^+) functions were determined from accurate MO-CI calculations on Mg⁺₂, using a [3s 2p 1d] basis set. The ${}^{2}\Sigma_{g}^{+}$ and ${}^{2}\Sigma_{u}^{+}$ potential curves were in agreement with a previous calculation by Stevens and Krauss [10]. The resulting functions have been fitted as follows :

$$F^{+} = -(a_1 + a_2 r) \exp(-a_3 r)$$
 (25)

$$R^{+} = \mathcal{M}^{+}(r) - 0.5 \,\alpha_{1}(r^{-4}) \,f_{1}(r) \qquad (26)$$

where \mathcal{M}^+ is a Morse function, α_1 the polarizability of neutral atoms and $f_1(r)$ a short range screening of the electric field. The three last parameters come from calculations on Mg_2^{++} , which were achieved in the same basis set (and complete CI). The main problem concerned the definition of R^{2+} , the energy of the $(|a\overline{a}| - |b\overline{b}|)/\sqrt{2}$ state, dissociating into $Mg^{2+} + Mg(3s^2) (IP_2 = 22.67 \text{ eV})$, due to a curve crossing with a state dissociating into $Mg^+(3p)$ + $Mg^+(3s)$, at 19.72 eV above $Mg(3s^2) + Mg(3s^2)$, i.e. 2.9 eV below the preceding dissociation limit (IP_2) . An ab initio diabatization procedure [11] has been applied to that problem in order to define nearly diabatic ${}^{1}\Sigma_{u}^{+}$ and ${}^{1}\Sigma_{g}^{+}$ excited states keeping the $Mg(3s^{2}) + Mg^{++}(3s^{0})$ character from infinite distances to the relevant short range region [12]. The resulting parameters have been fitted as follows

$$R^{++} = r^{-1} + \mathcal{M}^{++}(r) - \alpha_2 r^{-4} f_2(r) \qquad (27)$$

$$R^{2+} = \mathcal{M}^{2+}(r) \tag{28}$$

$$F^{++} = -(a'_1 + a'_2 r) \exp(-a'_3 r)$$
 (29)

where R^{++} involves a Coulombic r^{-1} repulsion, a Morse potential and a (screened) polarizability part, α_2 being the polarizability of Mg⁺ atoms. R^{2+} has been fitted by a simple Morse function. 7.34894

7.46170

0.004322

0.050549

Table Ia. — R parameters of the model. M are Morse functions: $a_2[(1 - \exp(-a_3(r - a_1)))^2 - 1]$ and f are screening functions: $1 - a_4 \exp[-a_5(r - a_6)^2]$.

Table Ib. — F parameters of the model.

 $\mathcal{M}^{2+}(r)$

| Parameter | Analytical function | <i>a</i> ₁ (au) | a_2 (hartree au ⁻¹) | $a_3 ({\rm au}^{-1})$ |
|-----------|------------------------------|----------------------------|-----------------------------------|-----------------------|
| F + | $(a_1 + a_2 r) \exp(-a_3 r)$ | 0.646202 | 0.269301 | 0.571140 |
| F^{2+} | | 2.44539 | 0.998238 | 0.722744 |

The explicite representation of the polarization energy in equation (26) is not compulsory for singly charged clusters. The determination of a screening function for the electric field was required for a correct estimate of the three-body term in equation (15). The screening function has been determined by comparing the energies of $Mg_A + Mg_B^+$ (3s frozen) in minimal and extended basis sets for Mg_A . An analogous procedure has been used for $Mg_A^+ + [Mg_B^{(+)} (3s$ frozen)]. The potential curves used for the parametrization have been plotted in figure 1a), c). The ground state potential curve of Mg^{++} is repulsive with r_{ab}^{-1} behaviour, but it presents a local minimum for r = 2.93 Å which is due to the mixing between the « pseudo-neutral » $(|a\overline{b}| + |b\overline{a}|)/\sqrt{2}$ configuration with the doubly ionic $(|a\overline{a}| + |b\overline{b}|)/\sqrt{2}$ configuration (as in H₂), through the hopping integral F^{2+} . The

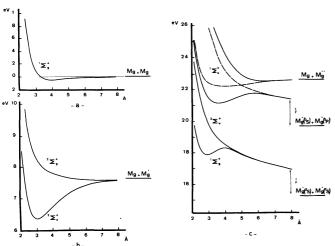


Fig. 1. -a) Experimental Mg₂ potential curve. b) Ab initio Mg_2^+ ($^2\Sigma_u^+$ and $^2\Sigma_g^+$) potential curves. c) Ab initio adiabatic (----) and diabatic (-----) Mg₂⁺⁺ potential curves.

parameters have been plotted as functions of the interatomic distance in figure 2. They have the expected r-dependence. One should notice however that for short r values, the F^{2+} integral has a larger amplitude than F^+ ; this increase may be due either to the effect of the hole (cf. Eq. (20)) or to the indirect role of the Mg(3p) + Mg(3s) state which belongs to the same range of energy. Standard deviation from experimental data for the R parameter of Mg₂ and from *ab initio* calculations for the R^{k+} and F^{k+} parameters of Mg₂⁺ and Mg₂⁺⁺ is less than 10^{-3} hartrees.

0.894654

0.505123

0.317424

0.092967

3. Results.

3.1 VERTICAL IONIZATION POTENTIALS. — Assuming compact structures for the neutral clusters (i.e. equilateral triangle for Mg₃, tetrahedron for Mg₄ and regular trigonal bipyramid for Mg₅ with $r = r_e(Mg_2) = 7.35$ a.u. interatomic distances), the vertical ionization potentials have been calculated and they appear in table II for both single and double ionizations. Notice the rapid decrease of the ionization potentials; for a regular linear chain or cycle of natoms the ionization potential would decrease as n^{-1} (as shown from Huckel type analytic derivations [13]). If, for a given value of n, the cluster becomes more compact, the increasing number of nearest neighbours, and consequently of hopping integrals, results in a lowering of the ionization potential. This increased compactness explains the rapid decrease of the calculated ionization potentials.

It may be interesting to analyse the space and spin symmetries of the lowest states of the ion, as predicted by our model, and compared to the monoelectronic picture, which suggests to empty the highest occupied MO (or MO's if a degeneracy occurs).

The lowest state of the triangular Mg_3^+ is of course degenerate, according to the monoelectronic picture (to which our model Hamiltonian is here identical).

3.03449

 R^{++}

 R^{2+}

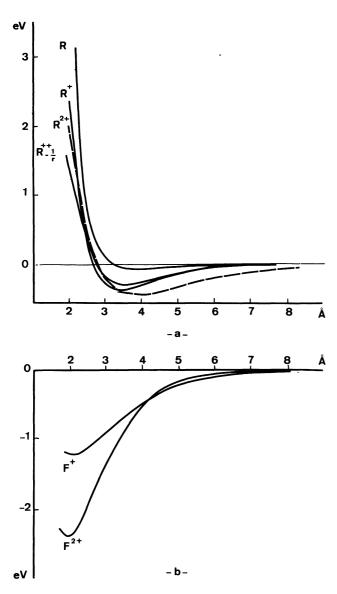
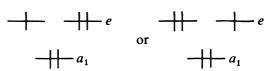


Fig. 2. — *r*-dependence of the various parameters of the model Hamiltonian. a) diagonal terms (for sake of comparison, the r^{-1} repulsion has been subtracted from R^{++}). b) off-diagonal integrals.

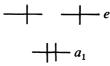
Table II. — Vertical and adiabatic ionization potentials (in eV).

| Single ionization | | | Double ionization | | |
|-------------------|----------|-----------|-------------------|-----------|--|
| n | vertical | adiabatic | vertical | adiabatic | |
| 1 | 7. | 64 | 22.67 | | |
| 2 | 6.80 | 6.40 | 18.32 | 17.87 | |
| 3 | 6.50 | 5.96 | 16.77 | 17.97 | |
| 4 | 6.19 | 5.53 | 15.88 | 14.69 | |
| | | (5.60) | | (15.29) | |
| 5 | 5.73 | 5.20 | 14.89 | 14.32 | |
| | | (5.22) | | (14.41) | |

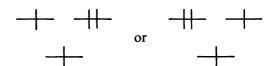
(The numbers in parentheses are secondary minima.)



A Jahn-Teller distortion must necessarily occur. In the same geometry one may expect that the lowest state of Mg_3^{++} will be of A_1 symmetry.



It actually is a triplet ${}^{3}A_{1}$ state. The next upper states are degenerate and would correspond to a different space part E

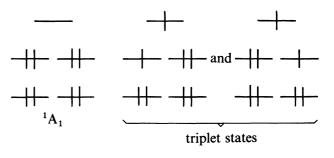


with a singlet spin arrangement. The ${}^{1}A_{1}$ associated with the lower triplet is higher in energy, which is surprising from an MO filling point of view.

The tetrahedral Mg_4^+ lower state is triply degenerate, as expected from the monoelectronic picture. The doubly charged ground state is quintuply degenerate.



For the bipyramid configuration the Mg_5^+ lower state is ${}^{2}A_1$ with larger charges on the corners of the pyramids. The doubly charged lowest state is ${}^{1}A_1$ and corresponds to two holes in the same highest MO; two degenerate triplet states are only 0.016 eV above.



The occurrence of low lying open shell states, sometimes contradicting the MO filling rule, examplifies the importance of the repulsion between the holes, due to the fact that

$$J_{ii} > J_{ij}$$

eV 23|

22

21

18

3.2 RELAXED GEOMETRIES AND ADIABATIC IONIZATION POTENTIALS. — The results appear in table III, figures 3-4.

The Mg_2^+ and Mg_2^{++} structures have already been given, and one may see (Fig. 1c)) that the latter is a local minimum in the repulsive ground state potential curve; this local minimum might accept up to 6 vibrational levels, the barrier being 0.35 eV.

The ground state geometries of Mg_3^+ and Mg_3^{++} are both linear with longer distances (3.1 Å) in Mg_3^+ than in Mg_3^{++} (3.0 Å); this is a general trend in the series. The states are of ${}^2\Sigma_g^+$ and ${}^1\Sigma_g^+$ symmetry in agreement with the orbital model (ionizations from the highest occupied MO).

For both Mg_4^+ and Mg_4^{++} two geometries are in competition, namely the square and the centred triangle. Both are real minima on the potential surfaces, the former being 0.6 eV below for Mg_4^+ , 0.07 eV below for Mg_4^{++} . The nearest neighbour distances are shorter in the centred triangle than in the square, especially for the doubly charged species $(\Delta r = 0.2 \text{ Å})$. The states are of 2A_1 and 1A_1 symmetries for the centred triangle, ${}^2B_{2g}$ and ${}^1A_{1g}$ for the square, in agreement with the MO picture.

For Mg_5^+ and Mg_5^{++} two nearly degenerate minima were found, corresponding to the centred tetrahedron (more stable by 0.02 eV in Mg_5^+) and centred square (preferred in Mg_5^{++} by 0.09 eV). They all are fully symmetric, and correspond to ionizations of the highest valence MO in the delocalized MO picture.

Mg 17 16 15 14 13 5 3 2 8 7 Mg 6 5 0 n Мg -1

Fig. 3. — Intrinsic energies of Mg_n , Mg_n^+ and Mg_n^{++} clusters (model Hamitonian), relative to *n* separated Mg atoms for \bigcirc neutral ground state conformation. \triangle and \blacktriangle , relaxed conformations of the ions.

Table III. — Energies E(eV) and nearest neighbour distances a(Å) of Mg_n^+ and Mg_n^{++} clusters (the zero of energy is n Mg atoms).

| | | | а | | Ε | | |
|---------------------|---|----------|--|--------------|--------------|----------------------|---------------|
| Structure | n | Charge | Symmetry | UHF | Model | UHF | Model |
| linear | 2 | + ++ | $\frac{{}^{2}\Sigma_{u}^{+}}{{}^{1}\Sigma_{g}^{+}}$ | 3.17 2.93 | 2.99 2.93 | 5.52 16.71 | 6.35 17.82 |
| linear | 3 | + + + | ${}^{2}\Sigma_{g}^{+}$ ${}^{1}\Sigma_{g}^{+}$ | 3.23 3.00 | 3.10 2.99 | 5.01 14.42 | 5.81 16.21 |
| square | 4 | +++ | ${}^{2}B_{2}$ ${}^{1}A_{1}$ | 3.51 3.22 | 3.20 3.12 | 5.18 14.13 | 5.23 14.39 |
| centred triangle | 4 | + + + | ${}^{2}A_{1}$ ${}^{1}A_{1}$ | 3.33 3.06 | 3.15 2.94 | 4.83 13.45 | 5.30 14.99 |
| centred tetrahedron | 5 | +++ | ² A ₁ ¹ A ₁ | 3.39 3.13 | 3.16 2.93 | 4.71 12.84 | 4.77 13.87 |
| centred square | 5 | + + + | ² A ₁ ¹ A ₁ | 3.49 3.23 | 3.15 2.95 | 4.96 13.38 | 4.75 13.96 |

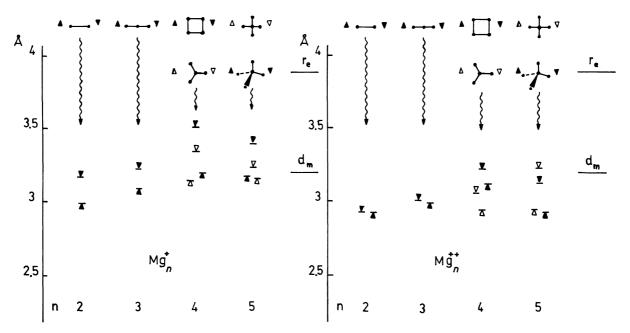


Fig. 4. — Interatomic distances between nearest neighbours in positive ions (relaxed conformations). ∇ , ∇ UHF *ab initio* calculation. \triangle , \triangle model calculation. r_e Mg₂ experimental r_e . d_m metallic nearest neighbour distance.

It seems likely that for larger and larger clusters the number of competing minima will increase and their exhaustive research will become a real problem. In these structures, whose topology is similar, the interatomic distances are almost identical.

For centred structures, the positive charge tends to be concentrated on the central atom; this charge is close to one for doubly charged species; this may be understood by the decrease in electronic repulsions when the central atom has no valence electron, by the polarization of the surrounding neutral electrons and by the role of cross-roads played by the central atom in the hole delocalization.

The relaxation energies (differences between vertical and adiabatic ionization potentials) are quite large :

0.40 eV for Mg_2^+ , 0.54 eV for Mg_3^+ , 0.66 eV for Mg_4^+ and 0.53 eV for Mg_5^+

0.45 eV for Mg₂⁺⁺, 0.41 eV for Mg₃⁺⁺, 1.19 eV for Mg₄⁺⁺ and 0.57 eV for Mg₅⁺⁺.

There is no clear trend in this series.

The adiabatic ionization potentials appear in table II They decrease regularly with n, and it is clear that one is still quite far from the asymptote, which would be the metallic values (one and two extraction potentials of the metal).

3.3 EVAPORATION, AGGREGATION AND FRAGMENTA-TION PROCESSES. — Figure 3 gives the intrinsic energies of Mg_n , Mg_n^+ and Mg_n^{++} taking the energy of *n* Mg separated atoms as the zero of energy. This picture makes easy a direct discussion of the enthalpy variations in the processes

$$Mg_n^+ \stackrel{b}{\rightleftharpoons} Mg_{n-1}^+ + Mg$$

$$Mg_n^{++} \stackrel{b}{\rightleftharpoons} Mg_{n-1}^{++} + Mg$$

(neglecting the zero vibration contributions). The aggregation process (a) occurs from the relaxed structure of Mg_{n-1}^{k+} (k = 1, 2), toward a relaxed geometry of Mg_n^{k+} (k = 1, 2). The decrease of the relaxed geometry energies in figure 3 indicates that the aggregation process is always exoenergetic. We have not analysed the existence of barriers in these processes, but it is quite clear — as summarized in figure 5 - that the aggregation of a neutral atom on the central atom of the centred structure of Mg_{n-1}^+ and Mg_{n-1}^{++} should proceed without difficulty, i.e. in the same symmetry and with small rearrangements. The processes which change the symmetry, i.e. those involving the square geometries, are more problematic and the research of activation barriers should be performed.

The evaporation processes are only possible from the vertical ionizations. They are only excenergetic for $4^+ \rightarrow 3^+ + 1$ and $5^+ \rightarrow 4^+$ (square) + 1 and for $5^{++} \rightarrow 4^{++}$ (square) + 1. The energy excess is very low (see Fig. 3) and the existence of barriers is possible in view of the large geometry (and symmetry) changes.

The possible fragmentation of Mg_5^{++} into $Mg_2^{+} + Mg_3^{+}$ has been analysed. The exoenergeticity is 1.72 eV when starting from the lowest energy tetrahedral geometry toward the relaxed geometry Mg_3^{+} and Mg_2^{+} ions. For a fragmentation pathway keeping C_{2v} symmetry, the intersection between the relevant ${}^{1}A_1(Mg_5^{++})$ and ${}^{1}B_2(Mg_3^{+} + Mg_2^{+})$ potential surfaces has its lowest energy 0.21 eV above the tetrahedron Mg_5^{++} energy minimum. This quantity may be considered as an estimate of the barrier.

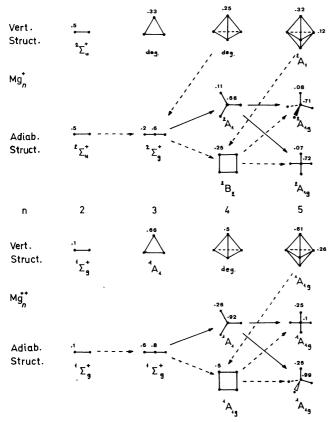
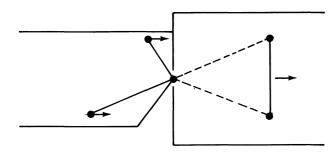
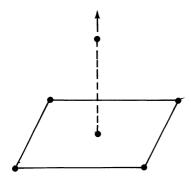


Fig. 5. — Net charges, symmetries of the clusters, and schematic views of possible aggregation and evaporation processes. \longrightarrow allowed processes. \longrightarrow hypothetic processes (exoenergetic but with possible barriers).



Another fragmentation process,

 Mg_5^{++} (centred square) $\rightarrow Mg_4^{+}$ (square) $+ Mg^{+}$



has also been explored. Its exoenergeticity is 1.09 eV, lower than the preceding one. It is quite clear that the doubly charged species tend to explode into equal size fragments, due to the *n* dependence of the stability of Mg_n^+ clusters.

$$Mg_n^+ + Mg_n^+ < Mg_{n+1}^+ + Mg_{n-1}^+ < \cdots < \ < Mg_{2n-1}^+ + Mg^+$$

The barrier for the extraction process of the central atom of Mg_5^{++} is only 0.14 eV.

3.4 OBSERVABILITY OF DOUBLY CHARGED CLUSTERS. ---The existence of local minima in the potential energy hyper-surface of Mg_n^{++} is not sufficient to insure the observability of these clusters, since these minima are significantly higher than the result of a fragmentation into $Mg_p^+ + Mg_q^+(p + q = n)$. Another factor which may prevent the observation of the doubly charged clusters is the great difference between the geometry of the neutral cluster and that of this local minimum. A vertical double ionization would not lead to the doubly charged cluster since it brings an important energy excess and the system could hardly achieve the travel from the original neutral cluster conformation to that of the doubly charged minimum and stay in that region which is only surrounded by weak barriers.

This statement is schematically illustrated in figure 6, representing for instance the Mg₃^(k+) (k = 0, 1, 2) potential curves as functions of two parameters only $r = r_{AB} = r_{BC}$ and $\theta = \widehat{ABC}$. It is quite clear that the vertical double ionization gives a small probability to stay in the local minimum of the upper surface.

As a very important fact, one should notice that in all the cases studied here, the stable conformations of Mg_n^+ and Mg_n^{++} are quite similar. This means that an ionization from Mg_n^+ to Mg_n^{++} would be nearly vertical, and would lead close to the bottom of the Mg_n^{++} structure, giving an opportunity for this cluster to have a significant lifetime. One must thus assume that the ejections of the two electrons under the electron impact are not simultaneous. During the time separating the first and second electron emission, the cluster Mg_n^+ should rearrange toward the stable Mg_n^+ configuration, which is an absolute minimum, the second emission proceeding without important geometry changes towards the secondary minimum of the Mg_n^{++} cluster (cf. Fig. 6).

These remarks are crucial for small clusters, and for all cases where the charges are quite concentrated, inducing important rearrangements of the positive cluster with respect to the neutral one. For very large clusters the rearrangements might become small (*vide infra*) and creations of doubly charged clusters might proceed in one step.

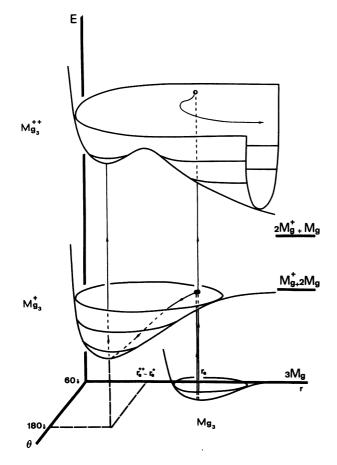


Fig. 6. — Schematic view of the lowest potential surfaces of $Mg_3^{(k+)}$ (k = 0, 1, 2) as functions of $r = r_{AB} = r_{BC}$ and $\theta = \widehat{ABC}$.

4. Comparison with *ab initio* SCF calculations; methodological discussion.

4.1 RESULTS OF SCF CALCULATIONS. — Valence only [14] SCF calculations have been performed on both singly and doubly charged clusters, using (3s, 2p) AO basis sets (i.e. deleting the 3d AO involved in the MO-CI calculation on Mg_2^+ and Mg_2^{++}). A UHF (Unrestricted Hartree Fock) formalism was used with $S_z = 1/2$ for Mg_n^+ and $S_z = 0$ for Mg_n^{++} , but for the latters the solutions were always pure singlets.

These calculations have been performed to check; i) the occurrence of minima close to the structures predicted by our model; ii) the possible importance of p AO's which do not appear explicitly in our scheme.

The energies and geometries appear in table III; they concern real minima obtained from a gradient program and force constants calculations. They confirm that all the predicted minima also appear in a monoelectronic approach. This agreement is not too surprising since, for the studied minima, the model wave-functions were consistent with single and double ionizations from the highest occupied MO. Significant differences appear however :

i) the ionization potentials are always much lower

than in our model. This is already true for n = 1, since the correlation of the s² pair in Mg is very important, increasing the ionization potential by about 1 eV; this defect is still acting in the cluster UHF calculation;

ii) the relative energy ordering of the two minima for Mg_4^+ , Mg_4^{++} and Mg_5^+ is reversed with respect to our model; one may notice that for Mg_4^+ and Mg_4^{++} the monoelectronic calculation favours the centred structure;

iii) the bond interatomic distances are much larger (see Fig. 4).

These two points of disagreement will be discussed later. The electronic populations calculated according to the Mulliken criterion would indicate a more even net charge distribution; they also would indicate a significant p population on the central atom. However a careful examination of the MO's shows that p_c AO's do not appear with coefficients larger than 0.15; the hybridization is therefore rather weak and the apparent p population is essentially due to the overlap of the p_c AO's of the central atom with the s AO's of the surrounding atoms, in a typical artifact of Mulliken's population analysis. This is confirmed by the fact that the p_c diagonal element of the density matrix is negligible (< 0.03 eV).

4.2 COMPARISON BETWEEN MODEL AND *ab initio* SCF CALCULATIONS. — The discrepancies between our model and SCF results may be easily understood. It is clear that *ab initio* SCF calculations introduce many-body repulsive effects which are not taken into account in our model. However these many-body effects are not very large and they cannot explain the differences in bond distances and in the relative energy orderings of the Mg_4^+ and Mg_4^{++} isomers.

i) Regarding the interatomic distances, one should notice that in a cluster the probability for two adjacent atoms to be both neutral is $1 - \frac{1}{n}$ for singly charged clusters and $1 - \frac{2(2 n - 1)}{n(n + 1)}$ for doubly charged clusters. In both cases this probability goes rapidly to one. One may therefore expect that when *n* increases the mean interatomic distances tend toward that of the neutral cluster (i.e. the metal for large *n*). Since the monoelectronic picture does not take into account the dispersion energy between neutral atoms, it will let the cluster explode and the interatomic distances will tend to infinity. This failure will also occur in positive pluster energy of the hole in

clusters since the delocalization energy of the hole is bound by kF^+ (k being a scalar, depending on the compactness of the cluster) as are the upper or lower limits of a band. Since the repulsive interactions ΣR increase as n, the delocalization of the hole, included in the monoelectronic scheme, cannot fight efficiently against the increasing repulsions, and the positive cluster is more and more loosely bound with larger and larger interactomic distances. This defect, which does not appear in our model, is very difficult to overcome in *ab initio* calculation, since it would require the calculation of dispersion forces — a task which is already difficult for Mg_2 [15].

ii) Regarding the relative energies of centred versus square structures in Mg^+ and Mg^{++} , one should notice that the Hartree-Fock approximation only takes into account the *static* field. In homogeneous structures, as is the square cluster, the charge is equally spread, the static field is weak everywhere; on the contrary if the charge is well concentrated on one atom — as occurs for the central atom of the centred triangle — the static field is important on the surrounding atoms, and the HF approximation is able to take this effect into account.

Our model deals with instantaneous repolarizations in all VB local structures, and therefore takes into account an effect which would only appear through double excitations in the MO scheme (change of the hole from the highest occupied MO to lower MO plus one excitation to the outer polarization space).

These comments — which will be developed in detail in a forthcoming paper [16] — explain the preference of the monoelectronic calculation for centred structures, as due to an unbalanced purely static treatment of polarization effects.

5. Conclusion.

A careful extraction of information from the potential curves of the diatoms Mg₂, Mg₂⁺ and Mg₂⁺⁺, corresponding to the states keeping an $s^p + s^q(p + q = 4, 3, 2, p, q \le 2)$ character, permits to define a simple model Hamiltonian of low size for the study of singly and doubly charged clusters. The Hamiltonian involves only 6 parameters and its size is n^2 for doubly charged

clusters. The parameters being analytic functions of the interatomic distances gradient techniques may be easily applied.

Since the extraction of the parameters has been done from calculations involving basis sets and CI, they should provide some information for large cluster studies, which are unattainable from direct MO-CI calculations since they rapidly require reductions of the basis set and/or severe CI truncations. The superiority of our model over *ab initio* SCF calculations appeared clearly in the preceeding discussion on both the internuclear distances (unreliable in SCF calculations) and the energy ordering of various minima (unduly preferred as centred structures in SCF calculations).

The present work has shown stable structures for Mg_n^+ and Mg_n^{++} for $n \leq 5$. It will be interesting to analyse the general topology of the potential energy hypersurface(s), taking benefit of some recent theoretical advances [17]. This would give better insurance regarding the number of minima, a problem which will be of increasing importance for larger clusters. It will also be worthwile to analyse the existence and heights of barriers for aggregation, evaporation and fragmentation processes.

Extension to larger clusters of Mg atoms is in progress. An isoelectronic series may be treated, namely the Hg_n^+ or Hg_n^{++} clusters studied in reference [1]. The extraction of the basic parameters for this heavy atom is in progress. Finally one should mention that a direct extension to rare gas clusters is easy to perform, the number of AO being then increased to 3 *n* (and the size of the matrix to $(3 n)^2$ for doubly positive clusters). In that last case the problem of hybridization, which may lead to difficulties for large values of *n* in s² atoms, would be avoided.

References

- BRECHIGNAC, C., BROYER, M., CAHUZAC, Ph., DELA-CRETAZ, G., LABASTIE, P. and WÖSTE, L., Chem. Phys. Lett. 118 (1985) 174.
- [2] ERMLER, W. C., LEE, Y. S., CHRISTIANSEN, P. A. and PITZER, K. S., *Chem. Phys. Lett.* 81 (1981) 70.
 CHRISTIANSEN, P. A., BALASUBRAMANIAN, K. and PITZER, K. S., *J. Chem. Phys.* 76 (1982) 5087.
- [3] TEICHTEIL, Ch., PÉLISSIER, M. and SPIEGELMANN, F., Chem. Phys. 81 (1983) 273, 283.
- [4] KUNTZ, P. J., Chem. Phys. Lett. 16 (1972) 581; TULLY,
 J. C., Modern Theoretical Chemistry, Vol. 7,
 G. A. Segal ed. (New York, Plenum), 1977, p. 173.
- [5] PARISER, R. and PARR, R. G., J. Chem. Phys. 21 (1953) 466, 767;
 - POPLE, J. A., Trans. Far. Soc. 49 (1953) 1375.
- [6] POPLE, J. A., SANTRY, D. P. and SEGAL, G. A., J. Chem. Phys. 43 (1965) 129;

- BAIRD, N. C. and DEWAR, M. J. S., J. Chem. Phys. 50 (1969) 1262.
- [7] HUBBARD, J., Proc. R. Soc. London A 276 (1963) 238.
- [8] HUBER, K. P. and HERZBERG, G., Constants of Diatomic Molecules (Van Nostrand, N. Y.) 1979.
- [9] SCHEINGRABER, H. and VIDAL, C. R., J. Chem. Phys. 66 (1977) 3694.
- [10] STEVENS, W. J. and KRAUSS, M., J. Chem. Phys. 67 (1977) 1977.
- [11] SPIEGELMANN, F. and MALRIEU, J. P., J. Phys. B 17 (1984) 1259;
 - CIMIRAGLIA, R., MALRIEU, J. P., PERSICO, M. and SPIEGELMANN, F., J. Phys. **B18** (1985) 3073.
- [12] DURAND, G., BERNIER, A. and SPIEGELMANN, F., to be published. The effective coupling between the $[Mg^{++} + Mg(3s^2)]$ and $[Mg^+(3s) + Mg^+(3p)]$ configurations is a $\langle 3s | F | 3p_z \rangle$ hopping integral, which increases at short distances, preventing

journal de physique. — t. 47, nº 8, août 1986

the diabatic potential curves to touch the adiabatic ones in this region.

- [13] See for example : SALEM, L., The Molecular Theory of Conjugated Systems (Benjamin, N. Y.) 1965.
- [14] DURAND, Ph. and BARTHELAT, J. C., Chem. Phys. Lett. 27 (1974) 191.
 - BARTHELAT, J. C. and DURAND, Ph., Gazzetta Chim. Ital. 108 (1978) 225.
- [15] PURVIS, G. D. and BARTLETT, R. J., J. Chem. Phys. 71 (1979) 548.
- [16] MALRIEU, J. P., DURAND, G., DAUDEY, J. P., to be published.
- [17] LIOTARD, D., Symmetries and Properties of non-rigid molecules : A Comprehensive Survey. Proceedings of an International Symposium, Paris, 1-7 July 1982, J. Maruani and J. Serre (eds), p. 323 (reference therein).