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1 **In-depth Analysis of Lime-Hemp Concrete and Water Vapor Interactions: Effect of water**
2 **default and prediction of the sorption behavior**

3

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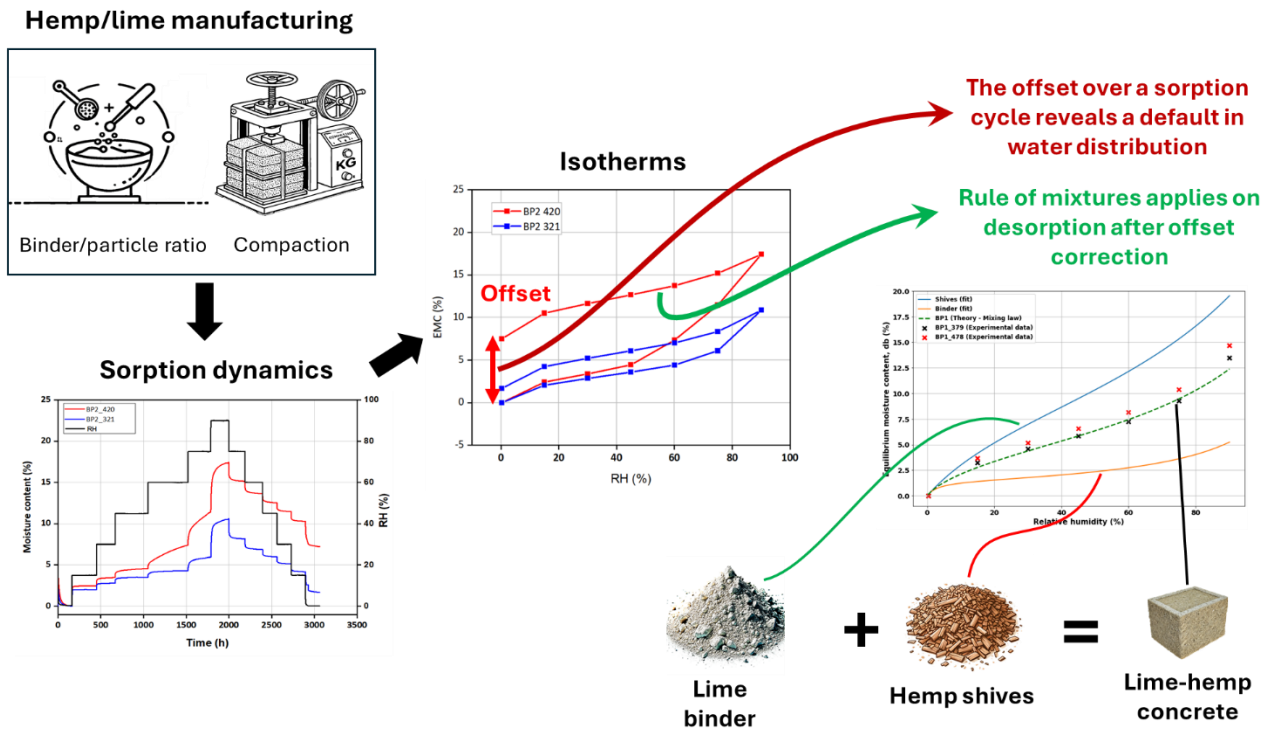
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Graphical abstract



24 **Abstract**

25 Lime-hemp concrete (LHC) emerges as a sustainable building material due to its low embodied
26 energy, carbon storage capabilities, and interesting properties for both winter and summer
27 comfort. However, a comprehensive understanding of its moisture behavior is pivotal for its
28 development and application in construction. This study investigates the moisture sorption
29 behavior and isotherm characteristics of LHC across four formulations varying in density (321–
30 478 kg/m³) and binder/particle weight ratios (BP = 1 and 2). Using a strict equilibrium criterion,
31 over 3000 hours of Dynamic Vapor Sorption (DVS), experiments revealed some formulations
32 failed to reach equilibrium during adsorption at RH levels above 60%, indicating irreversible
33 processes characterized by offsets in equilibrium moisture content (EMC) at 0% RH after a
34 complete cycle. These phenomena were attributed to insufficient water availability during
35 mixing and/or excessive compaction. Formulations with a higher weight ratio (B/P=2) and
36 significant compaction, such as BP2_420, exhibited the highest desorption offset (7.5% EMC),
37 while those with a lower B/P weight ratio (B/P=1), such as BP1_379, showed reduced offsets
38 below 2%, due to better water distribution. The study also showed that reversible sorption
39 behavior, corrected for offsets, could be accurately described using the Guggenheim-Anderson-
40 de Boer (GAB) model. Finally, the rule of mixtures reliably predicted sorption isotherms by
41 combining the GAB parameters of hemp shive particles and binders, with deviations limited to
42 a maximum error of 2.3%.

43 **Keywords:** sorption; lime-hemp concrete; hysteresis; water-availability, rule of mixtures

44

45 **1. Introduction**

46 The construction sector accounts for approximately 40% of global energy use and material
47 resource consumption, contributing significantly to greenhouse gas emissions [1,2]. This is why
48 this sector is under strong pressure to respond to global environmental concerns, in order to
49 reduce the sector's carbon footprint and resource consumption. To mitigate these impacts, there
50 is a growing shift towards developing and adopting eco-friendly building materials, including
51 bio-based alternatives that can offer lower embodied energy compared to traditional
52 construction materials. Among these innovations, vegetable concrete has emerged as a
53 promising solution, combining the thermal insulation and moisture regulation properties of
54 plant-based aggregates, such as hemp or flax, with the structural capabilities of mineral binders
55 [3]. This synergy not only ensures good thermal performances but also enhances indoor comfort
56 through improved hygrothermal properties [4]. These benefits make bio-based concretes a
57 distinctive alternative to conventional building materials, addressing both ecological and
58 economic challenges within the construction sector [5,6].

59 Lime-hemp concrete (LHC), commonly known as hempcrete, is one of the most promising bio-
60 based concretes due to its technical, environmental, and economic benefits [7,8]. While
61 traditional concrete includes coarse aggregates, the term "concrete" in LHC refers to its
62 composite nature and application in construction. The versatility of LHC makes it suitable for
63 a wide range of walls, floors, and roofs in both new construction and renovation [9]. Despite its
64 relatively low compressive strength generally, requiring the integration of auxiliary structural
65 supports to provide load-bearing walls, the LHC's remarkable hygrothermal behavior makes it
66 possible to envisage massive development [10,11]. With thermal conductivity ranging between
67 0.06 and 0.16 $\text{W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$, it serves as an effective insulator, potentially eliminating the need for
68 additional insulation layers [12,13]. Like other plant-based materials, LHC is an efficient
69 humidity regulator thanks to its pronounced adsorption and desorption characteristics [4,14].

70 These properties enable LHC to stabilize indoor humidity levels by absorbing excess moisture
71 when the humidity is high and releasing it back into the environment when conditions are drier.
72 This dynamic moisture regulation capability not only mitigates the risk of condensation within
73 the building structure but also plays a crucial role in preserving air-indoor quality [15].
74 Additionally, LHC's hygroscopic behavior enhances summer comfort by increasing thermal
75 inertia through latent heat exchange over daily variations of external conditions (temperature
76 and relative humidity), further boosting its suitability for modern, sustainable construction
77 [4,16].

78 However, previous studies have highlighted the critical influence of moisture content on the
79 heat and moisture transfer and storage dynamics of bio-based materials like hemp concrete
80 [14,17,18]. Therefore, a thorough understanding of moisture content fluctuations within LHC
81 is essential for accurately characterizing its hygrothermal behavior in response to environmental
82 changes. Experimental data on the hygroscopic properties of hemp concrete remain somewhat
83 limited and show considerable variability across different investigations. Moreover, the
84 methodologies and parameters employed for the hygroscopic characterization of hemp concrete
85 differ and are the subject of ongoing discussion [19]. Traditional methods for determining
86 sorption isotherms (equilibrium moisture content, EMC, as a function of relative humidity of
87 the air, RH), as outlined in standards such as NF EN ISO 12571, often rely on saturated salt
88 solutions or climatic chambers [4,20,15,17]. While these methods are widely accepted, they
89 have faced criticism for their labor-intensive nature and potential inaccuracies. Recent research
90 has shifted towards dynamic vapor sorption (DVS) with enhanced efficiency and precision
91 [17,21], provided the equilibrium criterion is demanding.

92 Numerous studies have been conducted to investigate the hygroscopic behavior of hemp
93 concrete. For instance, one study demonstrated that LHC, with binder-to-hemp and water-to-
94 hemp weight ratios of 1.2 and 2.38, respectively, reached a moisture content of approximately

95 16% at 90% relative humidity (RH) [22]. Sorption hysteresis has also been reported in several
96 works, attributing these findings to mechanisms such as capillary condensation hysteresis,
97 contact angle hysteresis, and the ink-bottle effect, depending on the combination of pore sizes
98 present and RH applied [15,22,23]. Note that the sorption hysteresis, always observed in
99 lignocellulosic materials, is attributed to the effect of sorption history on the macromolecular
100 conformation [24,25] rather than explanations involving liquid water, as this process occurs in
101 the domain of bound water. All previous phenomena, more related to capillary condensation,
102 can only explain the upper part of the sorption isotherm [26]. It is important to remember that
103 sorption hysteresis is defined as the gap between the two envelop curves obtained during
104 sorption cycles. During a cycle, these envelop curves pass through the origin ($EMC = 0$ at $RH = 0$).
105 The situation is more complex when the offset observed at $RH = 0$ after one cycle reveals
106 an irreversible mass uptake. A study examined three types of hemp concrete with different
107 compositions and manufacturing methods: precast, sprayed, and molded [4]. Their results
108 showed variations in sorption capacity, hysteresis, and irreversibility (offset after a full cycle:
109 adsorption until 97 % RH followed by desorption to 0 % RH). In particular, the authors
110 observed a significant irreversible mass in sprayed and molded hemp concrete compared to
111 precast variants. Another study confirms the presence of hysteresis phenomena accompanied
112 by an irreversible portion of desorption [27]. This could be due to the carbonation reaction of
113 the binder at a high RH level [4].

114 Research was also carried out to compare hemp concretes with identical formulations but
115 varying compaction levels [28]. Their findings revealed that the most densely compacted
116 sample demonstrated increased equilibrium moisture content at every RH level. They attributed
117 these results to a higher presence of mesopores for the denser mixture, which is a consequence
118 of the applied compaction stresses. These pore sizes tend to increase adsorption by capillary
119 condensation within the material. In fact, capillary condensation occurs in small pores, and

120 according to the Kelvin equation, only pores with a radius larger than 96 nm will remain empty
121 at 98.9% RH [26]. This suggests that the denser samples contain a significant number of
122 mesopores smaller than this threshold, contributing to the increased EMC observed at higher
123 RH. In addition to empirical investigations, some efforts have been dedicated to developing
124 numerical models to predict hygrothermal properties, mainly focusing on hysteresis effects
125 between adsorption and desorption phases [29,14,30]. These models show that ignoring
126 hysteresis can lead to inaccurate predictions. For example, Derluyn et al. [29] found that
127 neglecting hysteresis in concrete can underestimate durability risks, while Oumeziane et al. [14]
128 and Rémond et al. [31] demonstrated that including hysteresis in biosourced materials is needed
129 to predict the moisture buffering effect accurately.

130 While various results concerning the sorption of vegetable concrete have been reported in the
131 literature, the correlation between the composition of pure components of LHC, its post-
132 processing factors like compaction, and its sorption behavior (notably hysteresis and
133 irreversibility, which are crucial for LHC performance) remains largely unexplored. Moreover,
134 the theoretical approach traditionally used to predict the properties of composite materials from
135 their individual components has not been applied to LHC. By integrating this approach, the
136 present study offers novel insights into the interplay between the constituents of LHC and their
137 collective impact on its hygroscopic behavior.

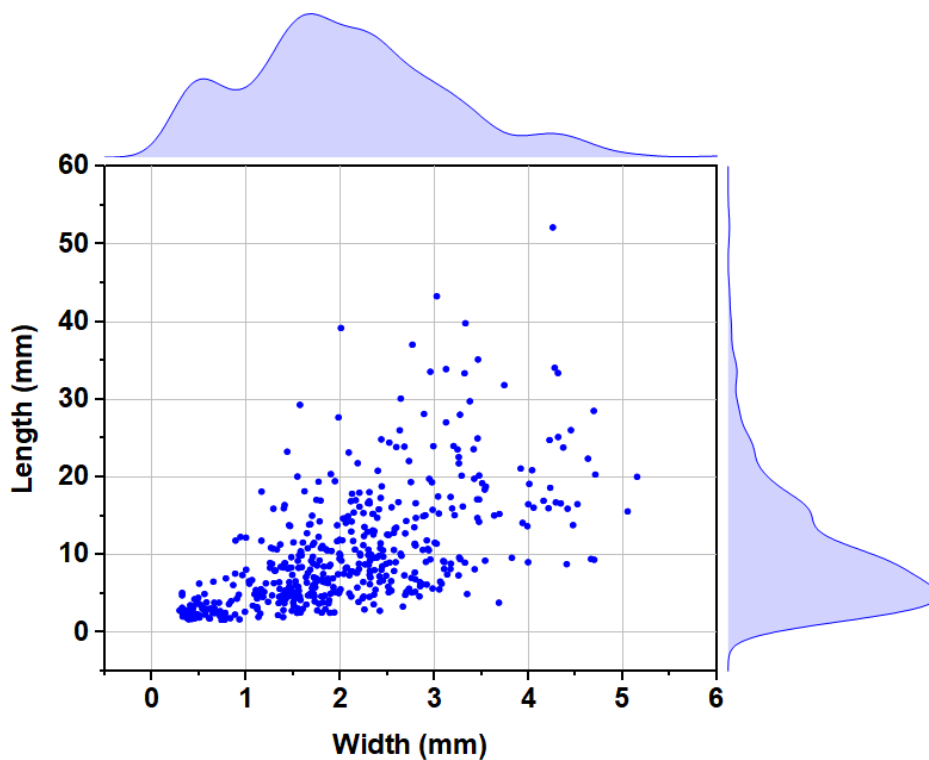
138 Accordingly, this study investigates the sorption phenomena and isotherm characteristics of
139 LHC by examining four contrasted formulations, as well as their individual components (binder
140 and particle). The selected formulations vary in binder/particle weight ratio (B/P), with ratios
141 of 1 (BP1) and 2 (BP2), and in compaction, resulting in different densities. This work relies on
142 DVS measurements using a very strict equilibrium criterion, which turned into a full sorption-
143 desorption cycle of more than 3000 hours. To the best of our knowledge, this represents the
144 first instance of such an extended dynamic sorption experiment being conducted on this type of

145 material. These extensive tests provide a unique dataset for detailed analysis of hysteresis and
146 irreversibility effects to clarify the mechanisms of moisture interaction in these materials.
147 Furthermore, the study proves that the rule of mixtures applies to the reversible part of the
148 sorption isotherm, which allows the properties of the composite materials to be predicted by
149 their individual components.

150 **2. Materials and method**

151 *2.1. Raw materials*

152 LHC samples assessed in this research were manufactured using a lime binder, hemp shive
153 particles, and a water-retaining agent (Hydroxypropyl-Methyl Cellulose). The hemp shives,
154 provided by Agrochanvre in Normandy, France, were harvested in September 2021, featuring
155 lengths ranging from 1 to around 40 mm and widths from 0.5 to 5 mm. The particle's sizes were
156 measured by image analysis (ImageJ version 1.53) and are shown in **Fig 1**. Over 1000 particles
157 were analyzed, providing detailed data [11]. These shives possess a bulk density of roughly 100
158 kg.m^{-3} according to the supplier and an apparent density of about 300 kg.m^{-3} . The binder used
159 in the concrete mix was Tradical® Thermo, comprising approximately 80% air lime and 20%
160 cement. HydroxyPropyl Methyl Cellulose (HPMC), sourced from DEFI - Houillères de
161 Cruéjols, was added to the mix as an additive at a 2% weight-to-weight ratio relative to the
162 mixing water, aiming to improve the mixture's stability by reducing mass mobility.



163
 164 Fig. 1. Hemp shive particles geometric analysis: scatter plot and side distributions of width
 165 and length, weighted by particle size.

166

167 ***2.2. Materials processing***

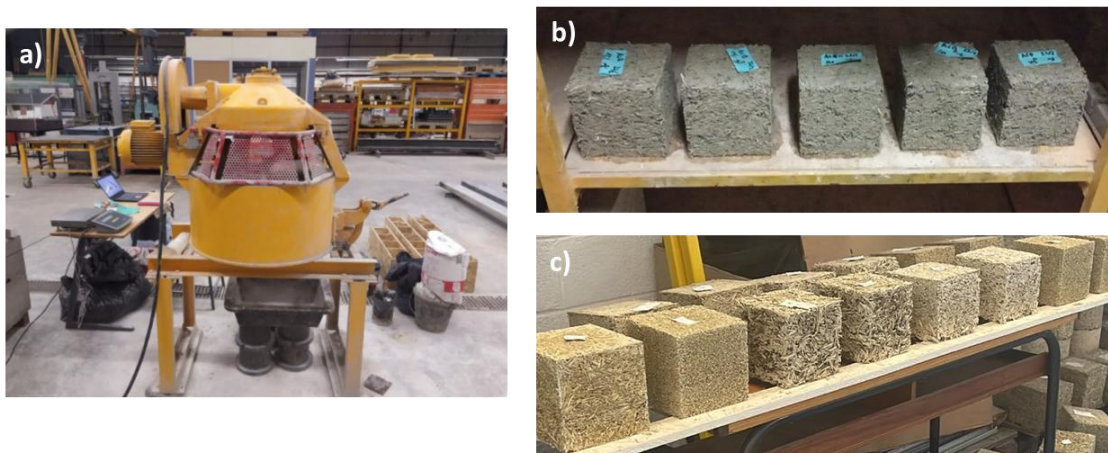
168 Following the experimental design outlined by Frantz et al. [11], this study focused on materials
 169 within the "Raw" category (composed of "building" certified hemp particles, in bulk, directly
 170 from Agrochanvre's processes). First, the hemp particles were pre-wetted with 170% water
 171 relative to their mass (with a moisture content of 12%, dry basis) for 5 minutes in a horizontal
 172 shaft mixer (manufactured by 3R) (**Fig 2a**). In parallel, the binder was prepared by mixing 40%
 173 (w/w) water with an additional 20% water relative to the particle mass to compensate for water
 174 absorption by the hemp particles. The water-retaining agent (HPMC, 2 % w/w) was added to
 175 this mixing water using a standard kitchen mixer. The binder-water mixture was then added to
 176 the pre-wetted hemp particles, and everything was mixed for an additional 8 minutes. The target
 177 densities for the hemp concrete were achieved by precisely measuring all components with a
 178 precision scale (Minebea Intec) and molding the mixture into $15 \times 15 \times 15 \text{ cm}^3$ cubes, in

179 accordance with the NF EN 12 390–3 standard (**Fig 2b-c**). The required quantity of mixture
 180 was manually compacted to fit the molds volume. For the denser sample (BP1), with a target
 181 density of 478 kg/m³, a hydraulic press was needed to reach the required compaction. To ensure
 182 homogeneity, each cube was compacted into three layers of 5 cm, and a metal plate was placed
 183 on top to evenly distribute the pressure. Demolding occurred 30 minutes after compaction,
 184 sufficient to maintain the cubic shape of the specimens and ensure their cohesion and stability.
 185 After demolding, all materials were initially stored at 70% relative humidity and 20°C for 48
 186 hours, then transferred to ambient conditions (20°C and 50% relative humidity) for stabilization
 187 over approximately one year. (**Fig. 2b-c**).

188 Additionally, a control cube, composed solely of hydrated binder (Tradical® Thermo product),
 189 was prepared with 40% (w/w) water and compacted to reach a density of 1,300 kg/m³, according
 190 to the manufacturer's specifications. Four material configurations, named BP1-379, BP1-478,
 191 BP2-321, and BP2-420 (**Table 1**), were selected in the present work to separate the effects of
 192 B/P ratios and density on sorption behavior while retaining configurations suitable for insulated
 193 walls.

194 Table 1. Apparent density, binder/particle ratio (weight/weight), and proportion of mixed
 195 materials.

Samples	Binder (g)	Particles (g)	Water (g)	Binder/Particle (B/P)	Apparent density (kg.m ⁻³)
BP1-379	579	579	347	1	379
BP1-478	723	723	434	1	478
BP2-321	552	276	276	2	321
BP2-420	736	368	368	2	420



196
197 Fig. 2. (a) Shaft Mixer used in mixing stage, (b) conditioning in a climatic chamber (70% RH,
198 20°C), and (c) conditioning at ambient conditions.

199
200 To perform the sorption tests, small samples was cut from each LHC cube using a saw. This
201 section was then ground using a hammer to obtain a representative powder sample of
202 approximately 1 g for analysis. Similarly, 1-g samples of the pure components, hemp shive
203 particles and hydrated binder, were also prepared for analysis.

204 ***2.3. Water vapor sorption***

205 The sorption tests were conducted using dynamic vapor sorption (DVS) (ProUmid GmbH &
206 Co. KG, Ulm, Germany) equipped with a 23-slot tray. This instrument is a cutting-edge, fully
207 automated system designed for assessing the water vapor sorption of several samples within a
208 controlled temperature and RH environment. The device offers a gravimetric accuracy of 0.1 μg
209 for samples up to 22 g, which is the case of samples used in this study. An internal mass
210 reference corrects any microbalance drift, ensuring outstanding balance stability, even during
211 prolonged measurements.

212 Sorption isotherms were measured at $25 \pm 0.1^\circ\text{C}$, under nitrogen flow, with RH levels ranging
213 from 0 to 90 %, by steps of 15 %, and then back to 0 %, by steps of -15 %. The measurement
214 procedure starts with obtaining the sample's dry weight (equilibrium mass at 0% RH). Then,
215 the desired humidity level is set, and changes in the sample's weight are recorded every 20
216 minutes. Equilibrium in a dynamic sorption experiment is defined as the time at which the mass

217 derivative is less than the set criterion. In this study, this criterion (dm/dt) was set at 0.0001%
 218 per 200 minutes, which is very demanding. With this stringent criterion, each RH step required
 219 a minimum of 3000 minutes and a maximum of 330 hours for the samples to reach equilibrium.
 220 Upon reaching equilibrium, the subsequent humidity step was initiated.
 221 The sorption tests were performed in duplicate for each sample, ensuring reproducibility and
 222 accuracy of the sorption measurements. Data analysis and figure production were done using
 223 in-house Python scripts. For all samples, equilibrium moisture content values (EMC, dry basis)
 224 were determined using the mass average over the last fifteen data of each RH step.

225 **2.4. GAB model**

226 The GAB model (Eq. 1) was employed to fit the sorption isotherms of LHC due to its well-
 227 established applicability for characterizing moisture sorption isotherms in porous and
 228 hygroscopic materials. The GAB model extends the widely used BET model, enabling the
 229 description of sorption behavior across a broader range of relative humidity (0–90%), making
 230 it ideal for this study. This model was applied for adsorption after calculating the moisture
 231 content at each humidity level using the initial dry mass equilibrated at 0 % RH. For desorption,
 232 to correct for the irreversible residual mass (offset after a complete sorption cycle), the moisture
 233 content was calculated using the mass obtained after desorption at 0 % RH as the dry mass. The
 234 model parameters were adjusted to fit the experimental data using a Python script, based on the
 235 Root Mean Square Error (RMSE) (Eq. 2). Finally, the SEE function was also employed to
 236 evaluate the capability of the GAB model in fitting the experimental data points.

$$EMC = \frac{MC_m \cdot C \cdot k \cdot RH}{(1 - k \cdot RH) \cdot (1 - RH + C \cdot k \cdot RH)} \quad \text{Eq. (1)}$$

237
 238 Where EMC is the equilibrium moisture content, MC_m represents the monolayer moisture
 239 content ($kg_{\text{water}}/kg_{\text{dry mass}}$) or as a percentage (%), RH is the relative humidity of the air (%), C

240 is the GAB constant and related to the heat of sorption of the first layer and k is another GAB
 241 constant associated with the heat of sorption for the multilayer.

$$RMSE = \sqrt{\frac{1}{N} \sum_{i=1}^N (EMC_{exp} - EMC_{pred})^2} \quad \text{Eq. (2)}$$

242

$$SEE = \sqrt{\frac{\sum_{i=1}^N (EMC_{exp} - EMC_{pred})^2}{df}} \quad \text{Eq. (3)}$$

243

244 Where N is the number of data points; EMC_{exp} the experimental equilibrium moisture content
 245 (% d.b) and EMC_{pred} the equilibrium predicted moisture content (% d.b), df is the degree of
 246 freedom, defined as $N - p$, where p is the number of the parameter in the model.

247 **2.5. Additivity of the sorption behavior**

248 The principle of additivity, also known as the rule of mixtures, was applied to predict the
 249 sorption isotherms for mixtures, based on the sorption behaviors of the individual pure
 250 components. The objective was to test the prediction of the LHC sorption isotherm by an
 251 average weighted by the mass fraction of the constituents isotherms. More clearly, the EMC of
 252 the composite at each RH value is computed as the weighted average, by the mass ratio, of the
 253 respective EMC values of lime and hemp obtained by the GAB curves at this RH value.

254 To validate the rule of mixtures, the sorption isotherms for the individual components, shives
 255 and hydrated binder (Tradical® Thermo) were then established in addition to the tests on the
 256 materials.

257

258 **3. Results and discussion**

259 ***3.1. Raw data: sorption dynamics and sorption isotherms***

260 **Fig. 3** depicts the moisture dynamics of the tested samples at varying relative humidity (RH)
261 levels. As can be seen, some samples, including the (hydrated) binder, fail to reach the
262 equilibrium state (defined at $dm/dt < 0.0001 \text{ \%}/200 \text{ min}$) during adsorption at the highest RH
263 levels, even after several hundred hours, especially above 60 % RH (**Fig. 3c**). In contrast,
264 equilibrium is always reached in desorption. This difference in behavior between adsorption
265 and desorption, observed for hydrated binder and some LHC formulations (**Fig. 3b and c**),
266 reveals an irreversible process, confirmed by the offset observed at 0 % RH after a complete
267 cycle.

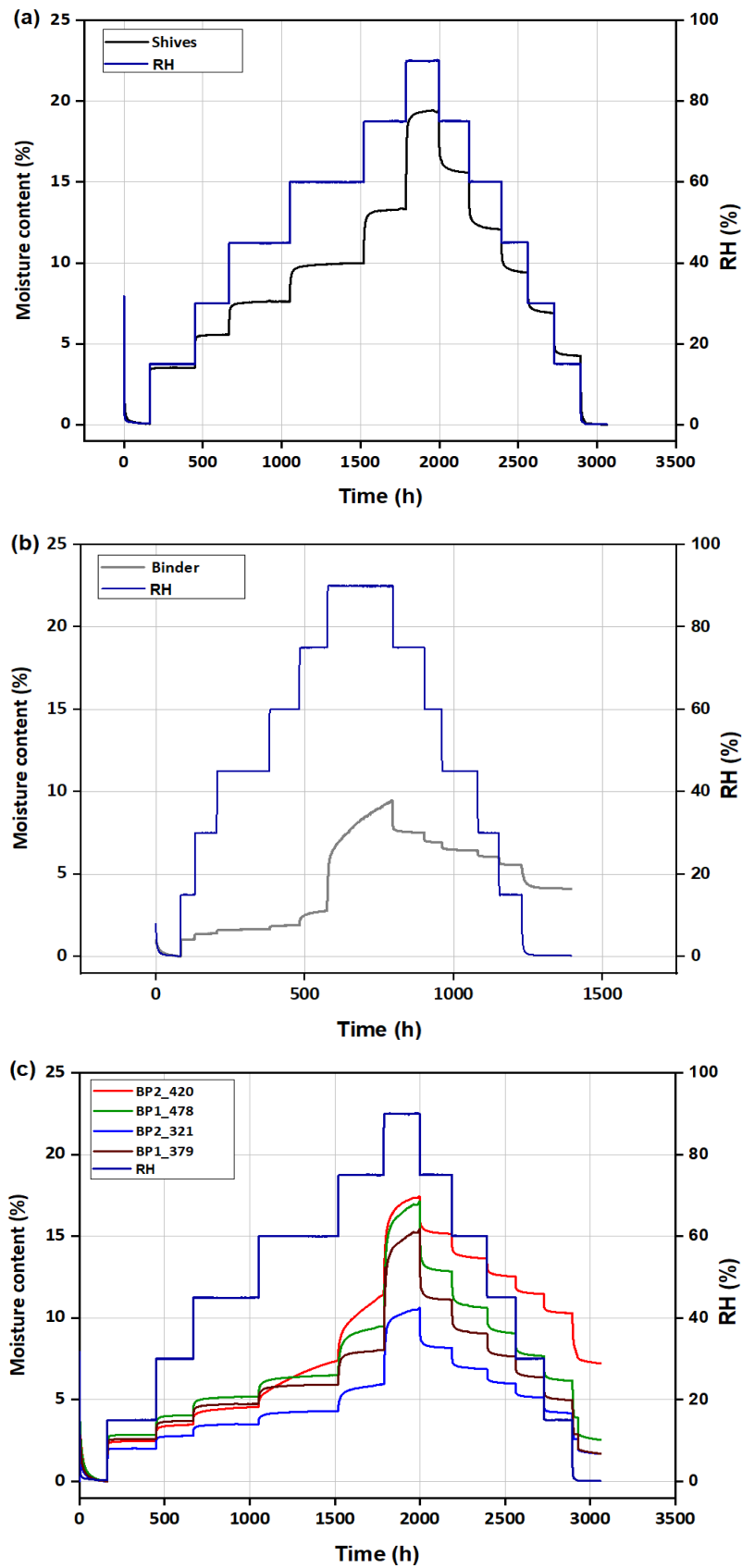
268 A very high offset is observed for formulation BP2_420 (exceeding 7 %), whereas formulations
269 BP1_478, BP1_379, and BP2_321 exhibit lower offsets, ranging between 2 % and 3 %. These
270 observations suggest several complex interactions between humidity and the material's
271 physicochemical composition, potentially influenced by the water-to-binder ratio and
272 compaction effects. This explanation is confirmed by the fact that this irreversible behavior is
273 observed at high RH, at values not seen by the material during its storage after manufacturing.
274 The manufacturing method consisted of adding 40 % water to the binder relative to its mass
275 (the manufacturer's minimum recommendation), to which 20 % extra water (percentage of
276 particle mass) was added to avoid consumption of setting water by particles. Two percent water-
277 retaining agent (HPMC) was introduced into the mixing water before its incorporation with the
278 binder. Following that, the blend was thoroughly mixed with pre-wetted particles. This process
279 was designed to reduce water mobility and thus minimize setting problems. Nevertheless,
280 considering the water-retaining capacity of hemp shive particles, it is probable that a portion of
281 this extra water remains accessible to the binder. Consequently, binders in formulations with a
282 B/P=1 ratio may potentially receive more water compared to those with a B/P=2 ratio.

283 Additionally, compaction, while crucial for achieving the desired material density, can induce
284 a non-uniform moisture distribution within the material [14]. Indeed, high compaction may lead
285 to local drainage of water from highly compacted zones, potentially resulting in incomplete
286 dispersion and interaction of the lime binder in these areas. Furthermore, compaction reduces
287 porosity, which may limit the diffusion of CO₂ required for air-lime carbonation [32,33],
288 particularly at relative humidity levels between 50 % and 70 % [34]. Reduced porosity also
289 restricts water retention and storage within the matrix [14], which is critical for maintaining the
290 hydration and carbonation processes.

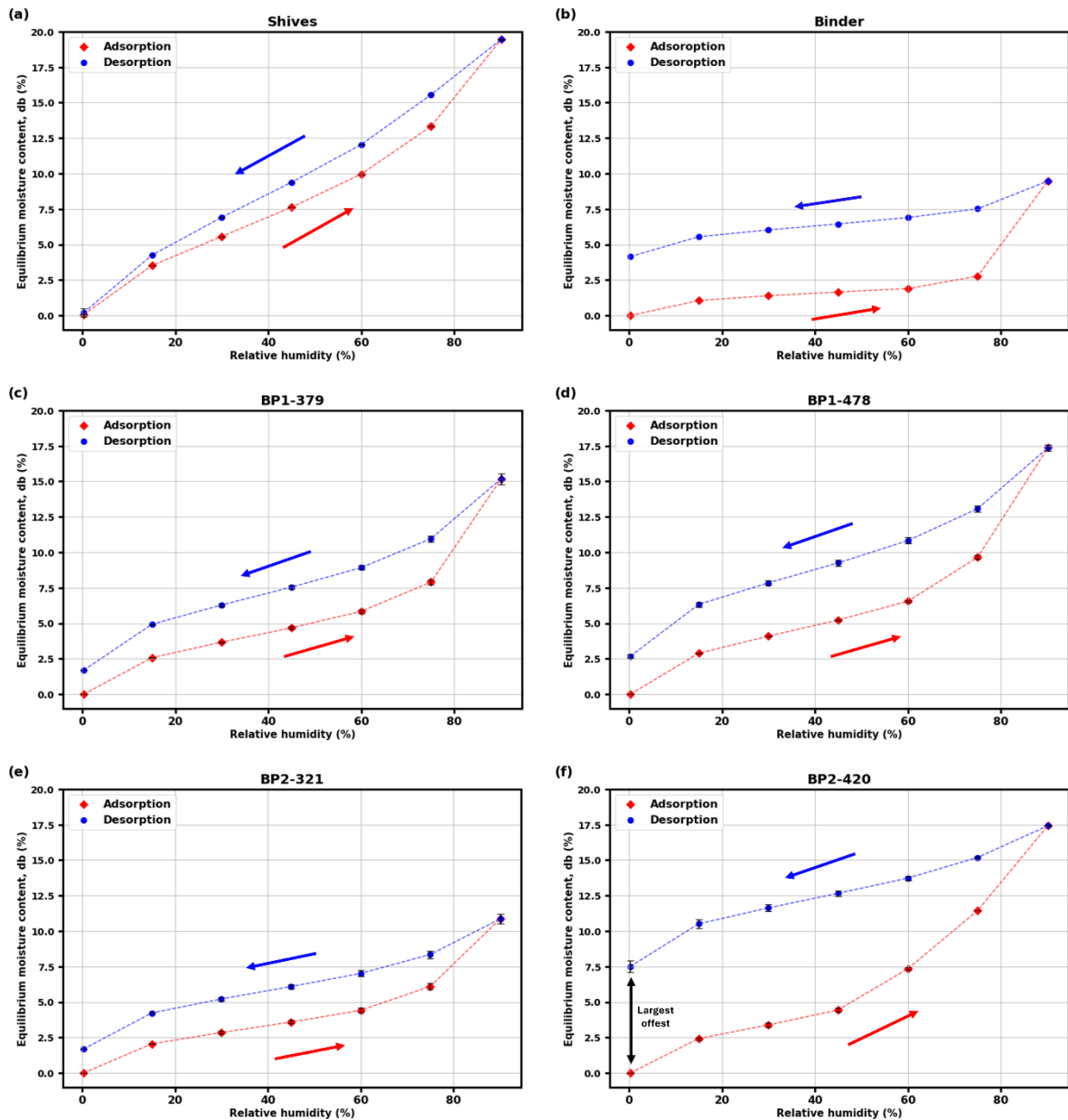
291 These hypotheses corroborate the sorption dynamics (**Fig. 3**) and sorption isotherm curves (**Fig.**
292 **4**). Tests on shive particles alone allow equilibrium to be reached relatively quickly at all RH
293 levels for both adsorption and desorption, with no offset, indicating a reversible cycle.
294 Conversely, the mass setting of the binder without the particles shows a ramp and has not
295 reached equilibrium at 90 % RH (**Fig. 3b**). This seems to confirm that the water quantity (40 %
296 of the binder) is insufficient. Additionally, the compaction required to reach the target density
297 of 1,300 kg.m⁻³ could explain the less water availability and the resulting delay in carbonation.
298 Formulation BP2_420 exhibits the same phenomenon from the 60 % RH level. This is the worst
299 case explained by the combination of the highest proportion of binder (B/P=2) with the
300 significant compaction needed to obtain 420 kg.m⁻³ concrete (**Fig. 3c, red line**). The resulting
301 offset on desorption is the highest with 7.5 % of EMC at 0 % RH (**Fig. 4f**). The proportion of
302 binder in formulation BP1_478 is lower (B/P=1), which is beneficial regarding the amount of
303 available water. However, such a small ratio means a high proportion of low-density particles,
304 which required a very severe compaction likely to induce local drainage that explains an offset
305 of approximately 2.5 % of water content. Consistently, this formulation did not reach the
306 equilibrium from 75 % RH level (**Fig. 3c, green line**). Finally, formulations BP1_379 and
307 BP2_321 are those presenting the lowest offset values, below 2 % (**Fig. 4c,e**). Formulation

308 BP1_379 required significant compaction for its manufacture but has a lower proportion of
309 binder (B/P=1) favorable for water availability, and BP2_321 required almost no compaction.
310 The LHC moisture contents obtained in this study are consistent with existing literature. At
311 lower relative humidity (RH) levels (below 45%), the sorption capacity is less than 5% for all
312 formulation, which aligns with findings from various studies [4,17,35]. However, at higher RH
313 levels ($\geq 70\%$), significant variations in moisture content are observed depending on the LHC
314 formulation, reflecting the variability also reported in the literature due to differences in the
315 formulation tested. For instance, Collet et al. [4] observed such variability when testing three
316 different hemp-lime concrete formulations: sprayed, molded, and precast. They found that at
317 higher RH levels, sprayed and molded hemp concretes exhibited similar mass changes, with
318 water content reaching approximately 20% at 97% RH and irreversible mass changes around
319 5% after desorption. In contrast, precast hemp concrete showed water content peaking at around
320 44% at 97% RH and irreversible mass changes of about 2%. These differences were attributed
321 to the higher hemp/binder ratio in precast concrete and the capillary condensation that occurs
322 in the pore structure at high humidity, as well as possible lime carbonation.

323



324 Fig. 3. Raw data from sorption tests: moisture content and relative humidity of the air over time
 325 for various materials: a) hemp shives, b) hydrated binder, c) Formulations BP1-379, BP1-478,
 326 BP2-321, and BP2-420.



327 Fig. 4. Isotherm curves obtained (adsorption in red and desorption in blue) for : a) hemp shives,
 328 b) lime binder, c) BP1-379, d) BP1-478, e) BP2-321, and f) BP2-420. Standard deviations are
 329 included in the figures, but they are so small that the bars remain inside the markers.

330

331 **3.2. Sorption behavior analyzed from the desorption data**

332 The previous section proved that the adsorption data include an irreversible contribution added
 333 to the true, reversible, sorption behavior. These data are, therefore, unsuitable for analyzing
 334 pure sorption behavior. Conversely, the desorption curves, corrected for the offset, can be
 335 assumed to represent the pure sorption behavior. As explained in the material and methods, the

336 offset correction consists of using the final mass at the end of the complete sorption cycle as a
337 reference dry mass. Consequently, all moisture contents were computed as the difference of
338 actual mass minus the reference dry mass divided by the same reference dry mass.

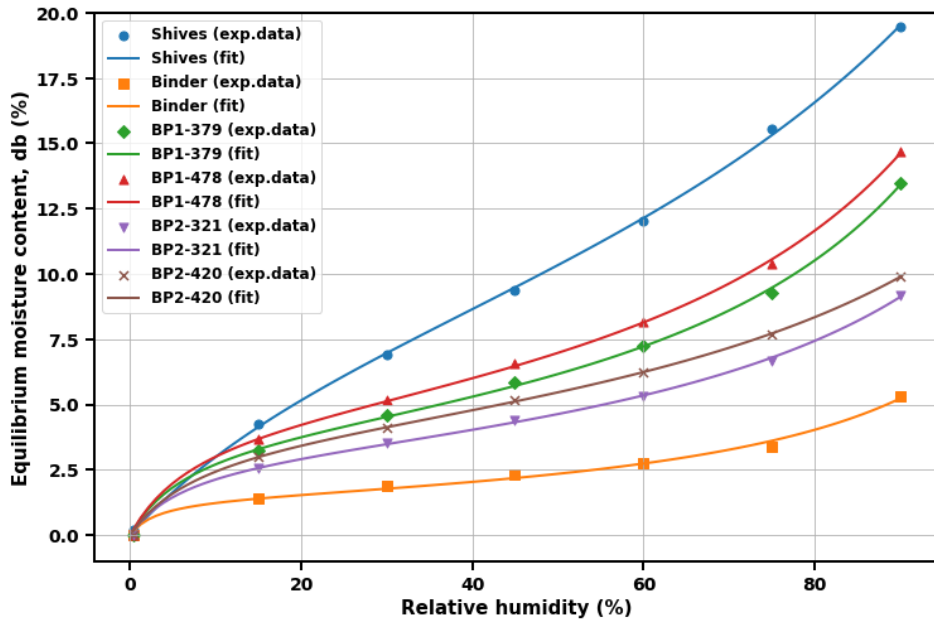
339 The isothermal desorption curves (**Fig. 5**) reveal that sorption characteristics and capacities
340 depend significantly on the sample. The sigmoid curves obtained for the different formulations
341 are characteristic of a monolayer-multilayer sorption material (classification IUPAC type II
342 isotherm). Hemp particles exhibit a high hygroscopicity, with an EMC of 13 % at 75 % RH.
343 This is consistent with the hydrophilic nature of lignocellulosic biomass, namely due to the
344 numerous hydroxyl groups in amorphous macromolecules [24,36,37]. On the contrary, instead
345 of a distinctive “Point B” (inflection point at low RH values), the binder matrix exhibits a
346 gradual curvature, which is an indication of a significant amount of overlap of monolayer
347 coverage and the onset of multilayer adsorption [38]. The material's hygroscopicity is much
348 lower, with an EMC of 2.6 % at 75 % RH. The various formulations tested show hygroscopicity
349 values between these two extremes, reflecting a combination of the hemp's high sorption
350 capacity and the binder's lower contribution.

351 To accurately describe the sorption behavior from desorption data, the GAB model was used to
352 fit the desorption data (**Fig. 5**) for all materials. This model is widely recognized for its
353 effectiveness in quantifying the moisture sorption properties of various materials. As already
354 commented, some samples underwent chemical interactions between the vapor and the sample
355 during adsorption, which fundamentally altered the physical state of the sample. Such
356 interactions render the GAB model inapplicable for adsorption since it should apply to pure,
357 reversible, sorption behavior. For this reason, the GAB parameters were not fitted for adsorption
358 data.

359 The GAB model provided an excellent fit to the experimental data, as evidenced by the close
360 agreement between observed data points and the fitted values (**Fig. 5**). The quality of the fit is

361 assessed by the low error values of two objective indicators: the root mean square error (RMSE)
362 and the sum of squared errors (SSE) (**Table 2**).

363 These GAB parameters quantify the interaction between moisture and the material. The
364 monolayer moisture content (MC_m), represents the moisture content at which bound water
365 molecules are the most strongly held (availability of active sites for water sorption by the
366 material), and the constants C and k are related to the heat of sorption for the first and subsequent
367 layers of adsorbed water, respectively. When K approaches one, there is almost no distinction
368 between multilayer molecules and liquid molecules (water molecules beyond the monolayer
369 have the same characteristics as the molecules in the bulk liquid). In the present study, the
370 mechanism of sorption is deduced from the combination of these parameters. The GAB
371 parameters for binder ($C = 39.3$, $K = 0.80$, $MC_m = 1.45\%$) in Table 2 illustrate the presence of
372 a monolayer, with strongly bound water molecules and a multilayer, which has characteristics
373 comparable with bulk liquid. Shives ($C = 6.2$, $K = 0.62$, $MC_m = 9.68\%$) have more available
374 active sites than binder, but water molecules are less strongly bound to the material and a more
375 structured multilayer, where the water molecules are different from bulk liquid molecules. The
376 mixed binder-shives (particles) materials show intermediate parameter values, reflecting the
377 combined characteristics of the pure materials in the LHC formulations.



378
 379 Fig. 5. Desorption isotherms, after offset correction, of the studied materials fitted by the GAB
 380 model (symbols: experimental values, continuous line: predicted values).
 381

382 Table 2. Parameters of the GAB model fitted to the experimental desorption curves together
 383 with the quality criteria: Root mean square error (RMSE) and the sum of squared errors (SSE).

Constants	Desorption data					
	Shives	Binder	BP1-379	BP1-478	BP2-321	BP2-420
MC_m (%db)	9.68	1.45	4.07	4.73	3.18	4.02
k	0.62	0.80	0.78	0.76	0.73	0.68
C	6.16	39.26	18.74	17.20	20.66	17.86
RMSE	0.12	0.13	0.12	0.11	0.09	0.06
SSE	0.11	0.11	0.11	0.08	0.05	0.03

384
 385 ***3.3. Rule of Mixtures applying GAB models***

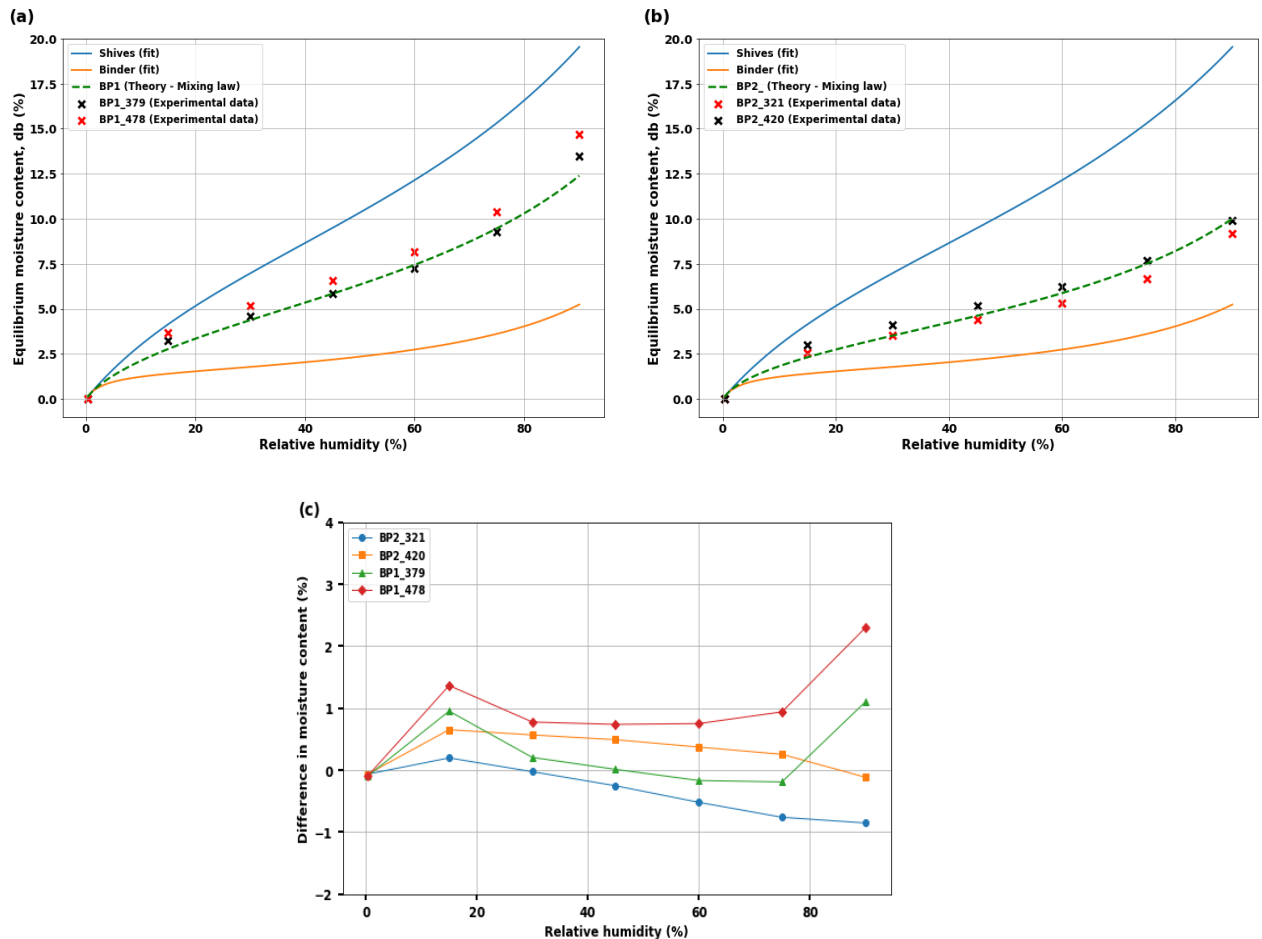
386 The GAB parameters for hemp shive particles and binders were used to predict the sorption
 387 behavior of formulations by the rule of mixtures (RoM). To do this, the EMC of both pure
 388 components are computed using the corresponding GAB models over the RH range and the
 389 EMC of the vegetable concrete is predicted as a weighted average of the mass ratio of the
 390 components. This allows the sorption curve of any vegetable concrete to be plotted. Its
 391 application in predicting the sorption isotherms of LHC mixtures has provided insightful data

392 **(Fig. 6)**. This figure shows the moisture desorption behavior corrected by the offset value of
393 the mixtures in comparison to their individual constituent hemp shives and hydrated binder.
394 The experimental data were put together with the RoM predictions as markers for each specific
395 mixture formulation, highlighting the interactions between the components.

396 The desorption isotherms for all formulations (BP1_379, BP1_478, BP2_321, and BP2_420)
397 reveal that the RoM works quite well in predicting the equilibrium moisture content on average
398 **(Fig. 6)**. By removing the irreversible adsorption effects (offset), the experimental desorption
399 isotherms align closely with the RoM predictions. In some cases, the experimental data almost
400 perfectly overlay the theoretical curves, indicating that RoM is a reliable method for modeling
401 these formulations. However, closer inspection reveals deviations from the RoM predictions,
402 particularly for BP1_478, which exhibits the largest discrepancies. The experimental data for
403 BP1_478 are substantially higher than the RoM predictions, especially at relative humidity
404 levels above 60%, indicating that the RoM underestimates the moisture absorption capacity of
405 this formulation. Keeping in mind that this configuration required the highest compaction level,
406 leading to a reduction in pore size, consistent with the finding of Gourlay et al.[28], although
407 the deviations remain limited in amplitude. Similarly, the experimental data for BP2_321 are
408 slightly lower than the RoM predictions for a wide range of humidity levels (40-90%),
409 indicating that the RoM overestimates the moisture uptake in this case, but with an error that
410 remains less than -1 % **(Fig. 6c)**. In contrast, BP1_379 shows experimental values very close to
411 the RoM predictions, with deviations becoming noticeable only at higher humidity levels.
412 Finally, BP2_420 demonstrates a strong alignment with the RoM, with minimal deviations,
413 suggesting that the additivity assumption holds well for this formulation.

414 In conclusion, while the RoM provides a good approximation for most formulations, it struggles
415 to fully capture the complexity of moisture absorption behaviors, particularly for BP1_478 at
416 higher humidity levels, where a maximum error of 2.3% was observed between the predicted

417 and experimental values (**Fig. 6c**). Nonetheless, for most cases, the sorption isotherms can be
 418 predicted with sufficient accuracy using the RoM combined with the GAB model, which is
 419 derived from the main constituents of the vegetal concrete, such as lime and hemp.



420
 421 Fig. 6. Application of the rule of mixtures to the desorption isotherms of the studied materials.
 422 a) Formulations with a B/P ratio of 1, b) Formulations with a B/P ratio of 2, and c) The
 423 difference between the predicted data (obtained using the rule of mixtures) and the experimental
 424 data.

425 The long sorption tests applied in the present work, combined with our experimental design,
 426 suggest that the offset observed in some samples can be attributed to insufficient water
 427 availability during manufacture. It seems possible to improve the manufacture of LHC, with
 428 their complex mineral-vegetal interactions, by adding the necessary amount of water, which
 429 should vary according to the proportions of constituents and to the compaction level. Indeed,
 430 achieving the optimum setting is crucial for maximizing the concrete's mechanical strength,

431 which can only be attained when the binder has fully undergone hydration and carbonation
432 reactions.

433 **4. Conclusion**

434 Lime-hemp concrete (LHC) represents a compelling research topic due to its low embodied
435 energy, carbon sequestration potential, and natural thermal regulation. As a bio-based material,
436 it offers promising solutions for sustainable and energy-efficient construction. This study
437 provides a detailed investigation into the sorption behavior and isotherm characteristics of LHC,
438 focusing on four formulations with varying binder/particle weight ratios ($B/P = 1$ and $B/P = 2$)
439 and densities. Using Dynamic Vapor Sorption (DVS) measurements over a 3000-hour cycle,
440 critical interactions between relative humidity (RH) and LHC components were uncovered,
441 shedding light on its moisture behavior and potential for optimization. Key findings revealed
442 that some formulations (e.g., BP2_420) failed to reach equilibrium during adsorption at RH
443 levels above 60 %, leading to irreversible mass gain and offsets in the desorption phase. These
444 issues were linked to insufficient water availability during fabrication, exacerbated by
445 compaction and binder/particle ratios, which caused uneven water distribution due to drainage
446 and reactivation of binder reactions under humid conditions. In contrast, desorption curves
447 consistently reached equilibrium more rapidly across all RH levels, providing reliable data for
448 further analysis and modeling. The Guggenheim-Anderson-de Boer (GAB) model was
449 employed with success to fit with the experimental value. Finally, the rule of mixtures
450 demonstrated effective predictability of hygroscopic behavior for all formulations, with a
451 maximum error of 2.3% observed between the predicted and experimental values, primarily at
452 high RH levels. The theoretical models applied in this study, including the GAB model and rule
453 of mixtures, provide a robust framework for understanding the moisture behavior of LHC, and
454 can thus be used to predict the sorption behavior of new formulations. The findings offer
455 insights that can inform the refinement of fabrication processes, particularly in areas such as

456 binder/particle ratios, compaction methods, and water management, enhancing the material's
457 performance and potential scalability in construction. Further investigations will concern the
458 evaluation of the long-term performance of LHC under varying environmental conditions to
459 better understand its durability and hygrothermal behavior in real-world scenarios.

460 **Author contributions**

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464 Writing - Review & Editing; Supervision.

465 *Nils Frantz*: Conceptualization; Methodology; Investigation.

466 *Patrick Perré*: Conceptualization; Methodology; Formal analysis; Data Curation; Writing -
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474

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