



Using X-ray induced Auger electron spectroscopy transitions to explore the surface reactivity of CIGS: In and Ga degradation kinetics at different probing depths

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Using X-ray induced Auger electron spectroscopy transitions to explore the surface reactivity of CIGS: In and Ga degradation kinetics at different probing depths

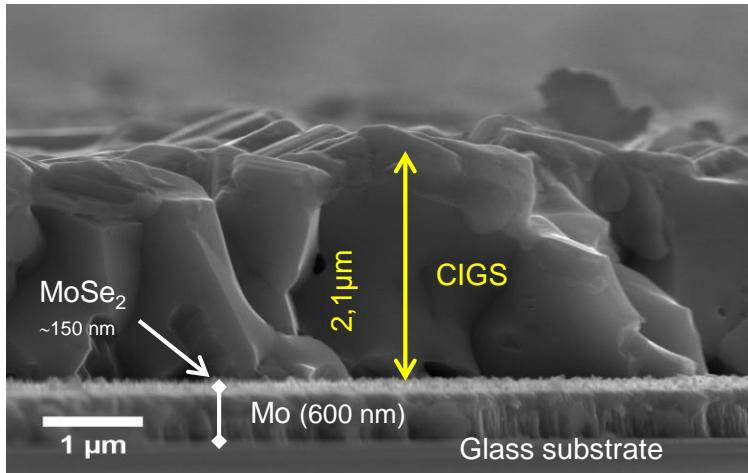
Anna Gagliardi,¹ Neal Fairley,² Solène Béchu¹

¹ Institut Lavoisier de Versailles (ILV), Versailles Saint-Quentin-en-Yvelines University, CNRS, UMR 8180, Versailles, France.

² Casa Software Ltd, Bay House, 5 Grosvenor Terrace, Teignmouth, Devon TQ14 8NE, United Kingdom.



$\text{CuIn}_{0.7}\text{Ga}_{0.3}\text{Se}_2$: a solar device material

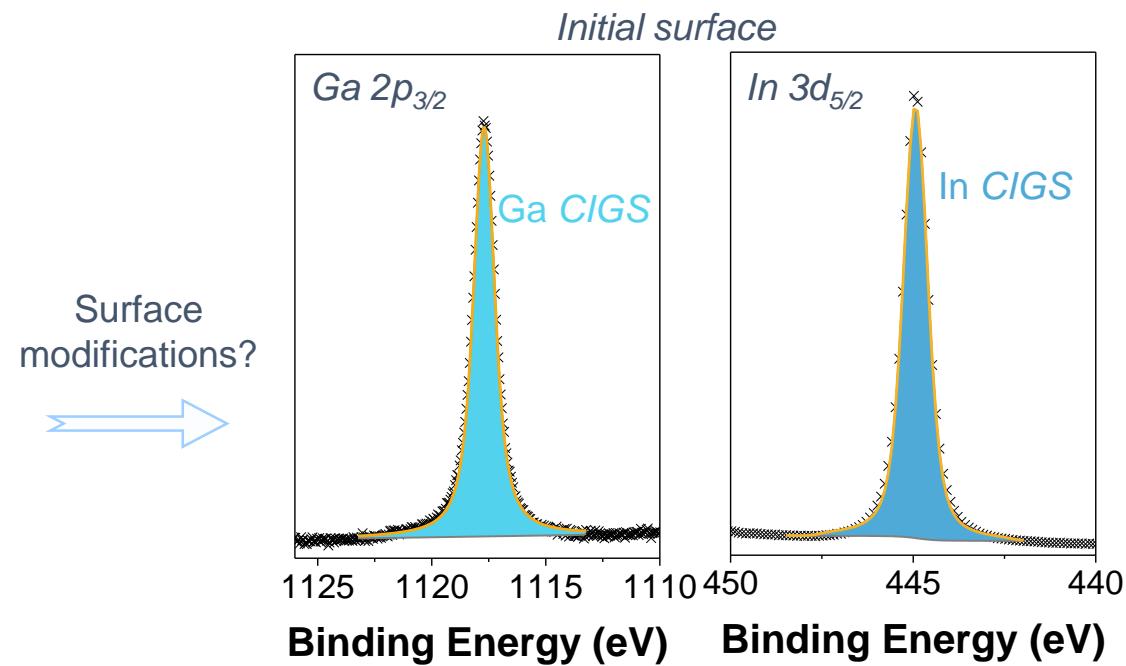
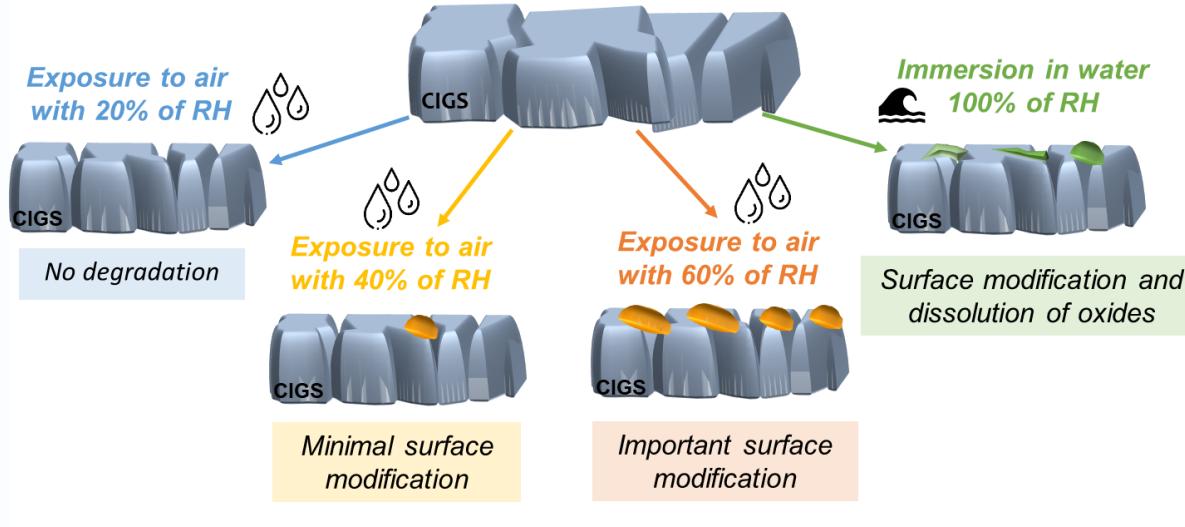


- $\text{Cu}(\text{In}_x\text{Ga}_{1-x})\text{Se}_2$ (CIGS): solar absorber with efficiencies up to 23.6%
- **Long-term stability of the cell:** main factors responsible for degradation

- Relative humidity
- Temperature
- Illumination
- Gazeous environment

M. Theelen, F. Daume, *Sol. Energy.*, 2016, 133, 586–627

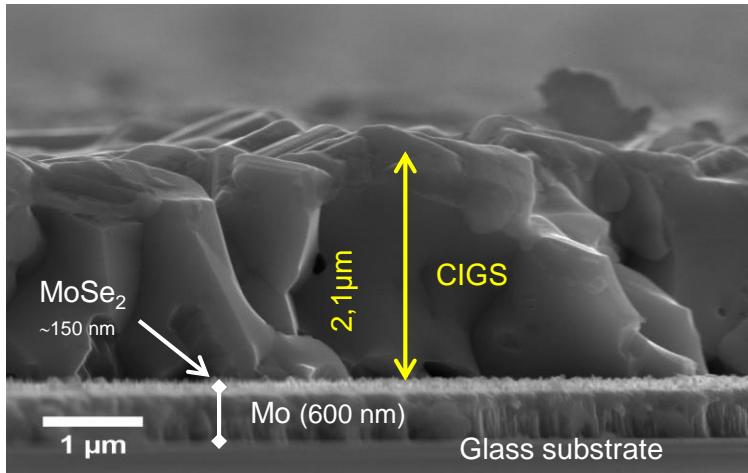
Degradation of
CIGS material itself?



➡ Kinetics of degradation?

$$\frac{Ga_{ox}}{Ga_{CIGS}} \text{ & } \frac{In_{ox}}{In_{CIGS}}$$

$\text{CuIn}_{0.7}\text{Ga}_{0.3}\text{Se}_2$: a solar device material

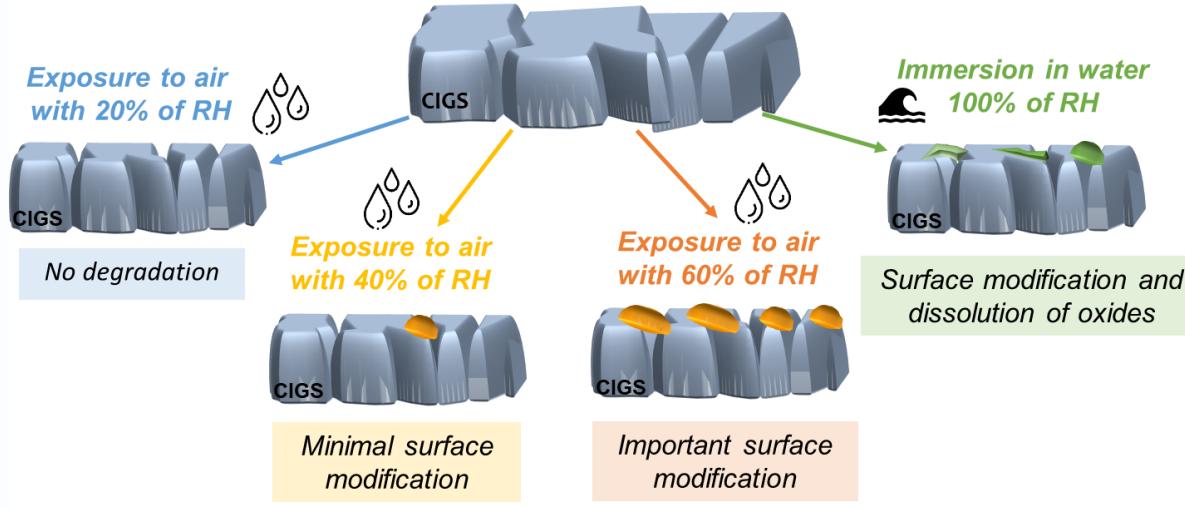


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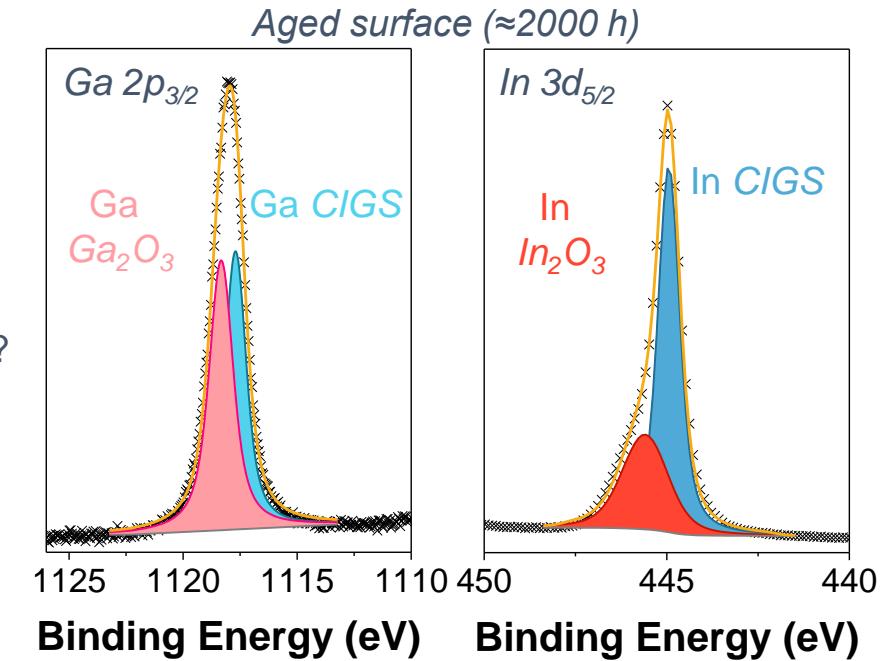
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M. Theelen, F. Daume, *Sol. Energy.*, 2016, 133, 586–627

Degradation of
CIGS material itself?



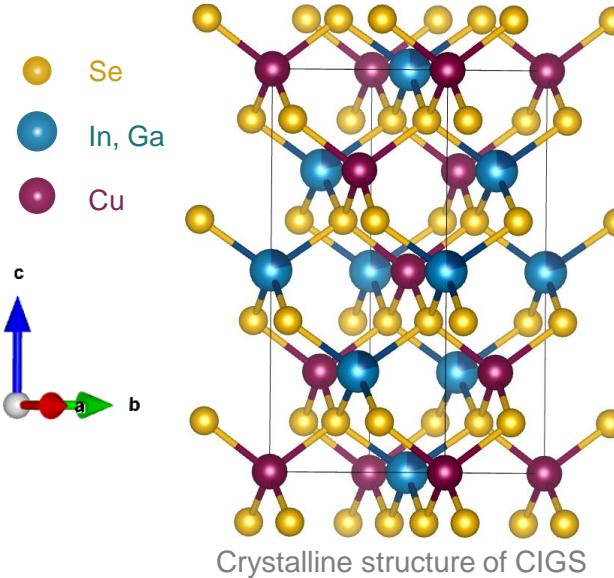
Surface modifications?



Kinetics of degradation?

$$\frac{\text{Ga}_{\text{ox}}}{\text{Ga}_{\text{CIGS}}} \text{ & } \frac{\text{In}_{\text{ox}}}{\text{In}_{\text{CIGS}}}$$

Why using X-AES transitions? The case of CuIn_{0.7}Ga_{0.3}Se₂

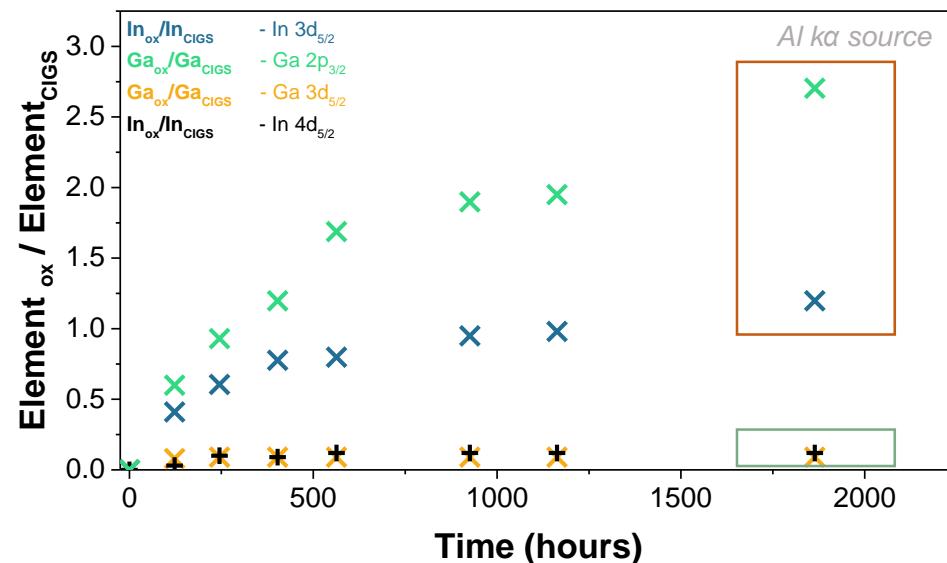


Evolution of In and Ga under aging?



most decomposed photopeaks

photopeaks with the same probing depth



S. Béchu et al., Appl. Surf. Sci., 2022, 576, 151898

Atomic level

Binding Energy
Kinetic Energy
Depth probed

$In M_{4,5}N_{4,5}N_{4,5}$
1105 - 1068
381 - 417
 $\approx 3 \text{ nm}$

$Ga 2s$

1298
188.6
 $\approx 2 \text{ nm}$

$Ga 2p$

1143 - 1116
343.6 – 370.6
 $\approx 3 \text{ nm}$

$In 3s$
826
660.6
 $\approx 4 \text{ nm}$

$In 3p$
702 - 664
822.6
 $\approx 5 \text{ nm}$

$In 3d$
451 - 443
1035.6 – 1043.6
 $\approx 6 \text{ nm}$

$Ga L_3M_{4,5}M_{4,5}$
430 - 410
1058 - 1075
 $\approx 6 \text{ nm}$

$In 4s$
122
1364.6
 $\approx 7 \text{ nm}$

$Ga 3s$
158
1328.6
 $\approx 7 \text{ nm}$

$Ga 3p$
107 - 103
1379.6 - 1383.6
 $\approx 8 \text{ nm}$

$In 4d$
16
1470.6
 $\approx 8 \text{ nm}$

$Ga 3d$
18
1468.6
 $\approx 8 \text{ nm}$

Different kinetics due to the different probing depths

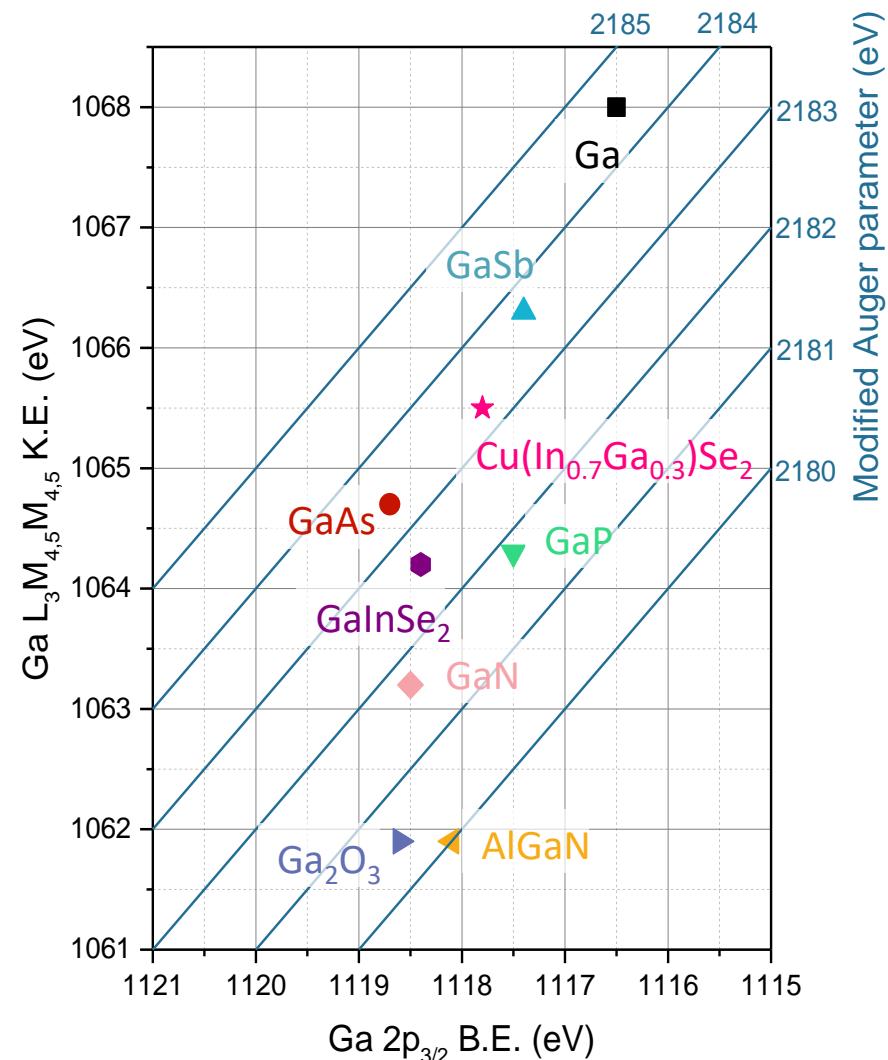
Extreme surface ($\approx 2 - 3 \text{ nm}$)

Sub surface ($\approx 8 \text{ nm}$)

B.E and K.E of In and Ga elements for an Al Ka source
Probing depths are calculated with the TPP2M formula for CIGS

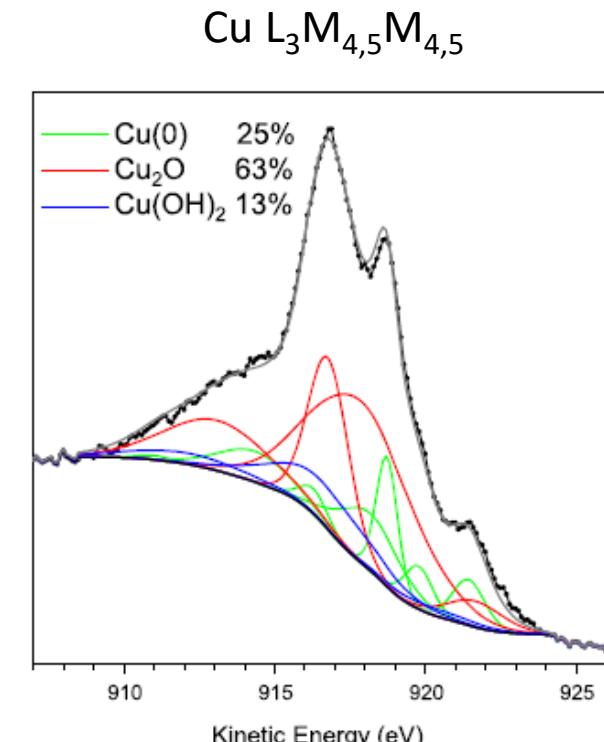
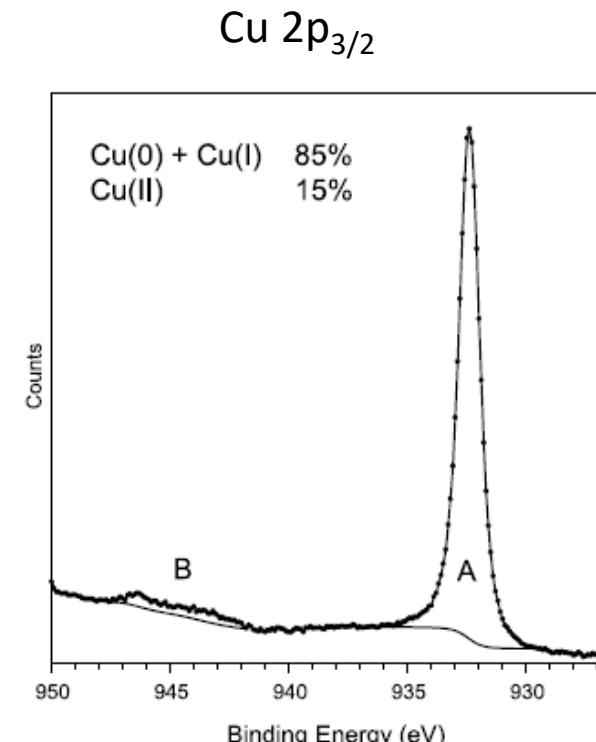
The X-AES transitions: phase identifications and quantification

Wagner plot and modified auger parameter



Ga $2p_{3/2}$ – Ga LMM Wagner plot

Quantifications with X-AES lines



Advanced analysis of copper X-ray photoelectron spectra

Mark C. Biesinger*



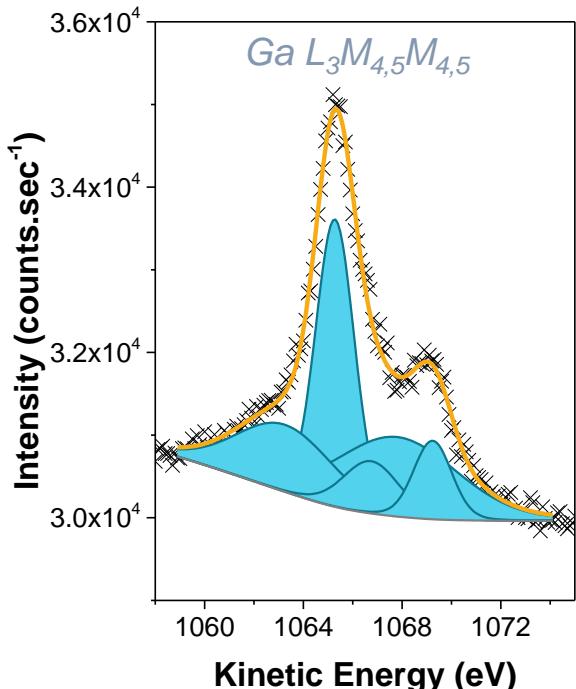
Surf. Interface Anal. 2017, 49, 1325–1334

Non-linear versus linear least square decomposition

2 methods to decompose the X-AES transitions

Non-linear least square decomposition

Arbitrary use of Gaussian Lorentzian curves to simulate one chemical environment



NLLS

$Ga\ L_3M_{4,5}M_{4,5}$ non-linear least square decomposition for CIGS

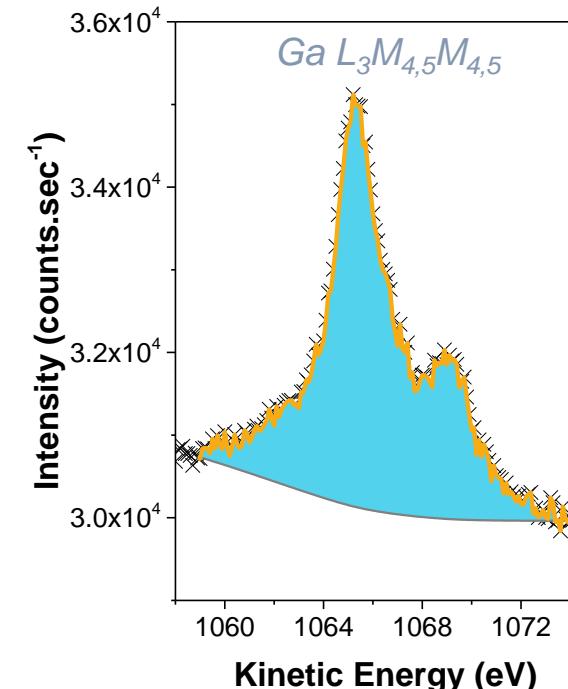
✓ Possibility of slight adjustments for the positions, the FWHM values

✗ For a mix of chemical environments, necessity of many arbitrary peaks → possible error

J. J. Moré, *Numer. Anal.*, **1978**, 630, 105

Linear least square decomposition

Implementation of a spectral shape



LLS

$Ga\ L_3M_{4,5}M_{4,5}$ linear least square decomposition for CIGS

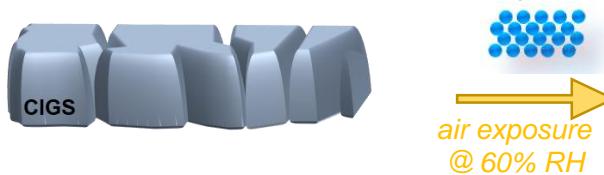
✓ Easy to implement with different chemical environments

✗ No FWHM or position adjustments for a series of fits

G. H. Golub and C. Reinsch, *Numer. Math.*, **1970**, 14, 403

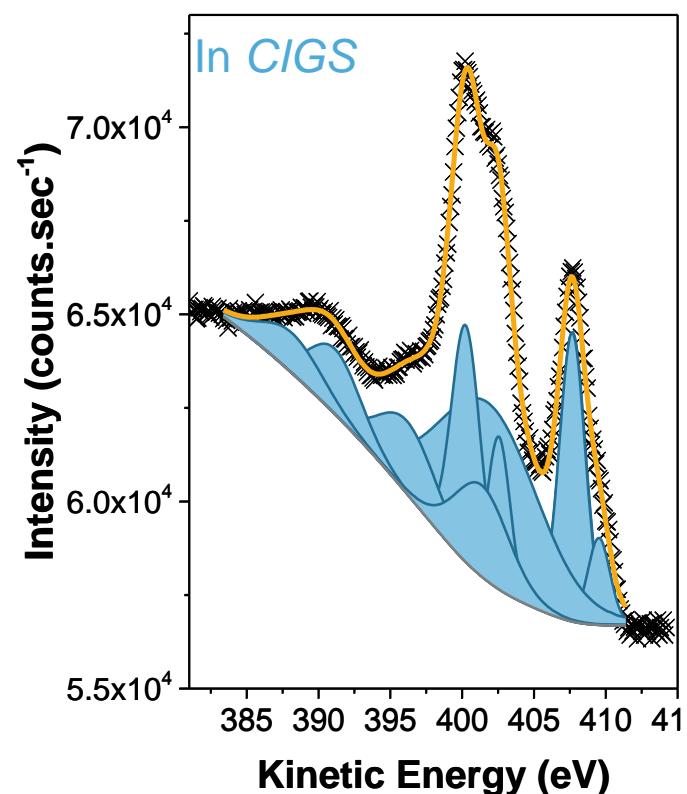
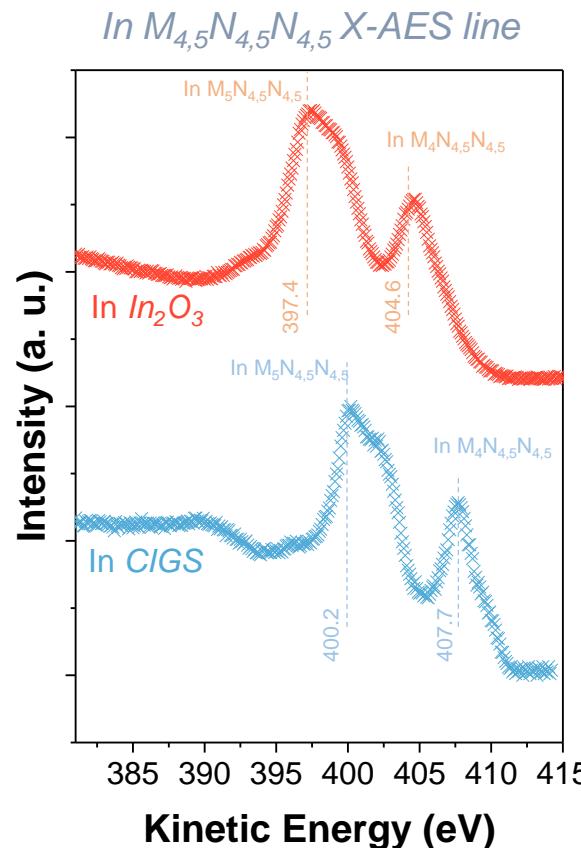
NLLS decomposition of CIGS X-AES transitions

Modifications induced by the aging



NLLS

Growth of In_2O_3 and Ga_2O_3 phases in addition to the CIGS one
Identified with the modified Auger parameter



	PP
Position	407.7 ± 0.1
FWHM	2.10 ± 0.05
Area	/

Peak 6	Peak 7	Peak 8	Peak 9
PP-7.4	PP-11.5	PP-16.5	PP-19.5
PP*1.11	PP*2.34	PP*2.04	PP*2.45
PP*0.91	PP*0.59	PP*0.46	PP*0.34

9 arbitrary peaks to decompose the
 $\text{In } \text{M}_{4,5}\text{N}_{4,5}\text{N}_{4,5}$ X-AES transition of CIGS
→ 5 for the $\text{Ga } \text{L}_3\text{M}_{4,5}\text{M}_{4,5}$ X-AES transition of CIGS

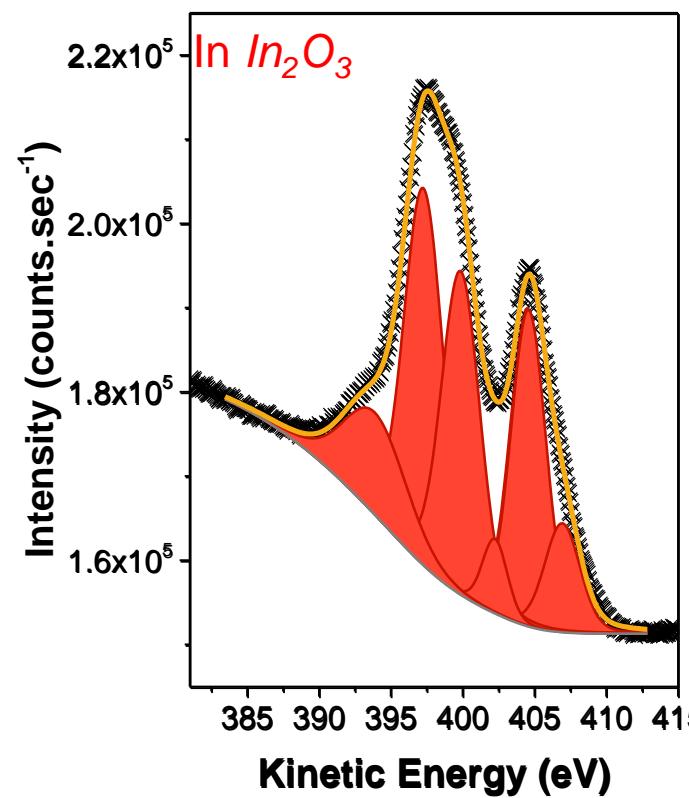
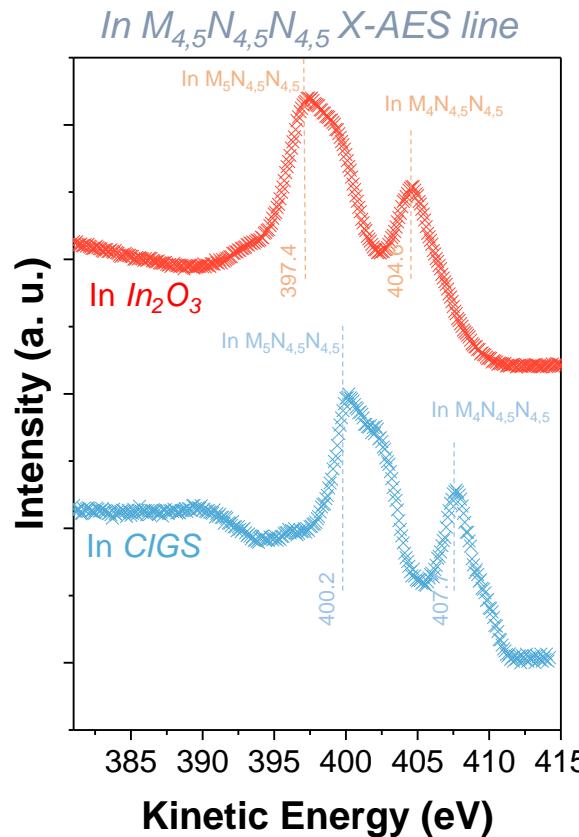
NLLS decomposition of CIGS X-AES transitions

Modifications induced by the aging



NLLS

Growth of In_2O_3 and Ga_2O_3 phases in addition to the CIGS one
Identified with the modified Auger parameter



	PP
Position	404.5 ± 0.1
FWHM	2.85 ± 0.05
Area	/

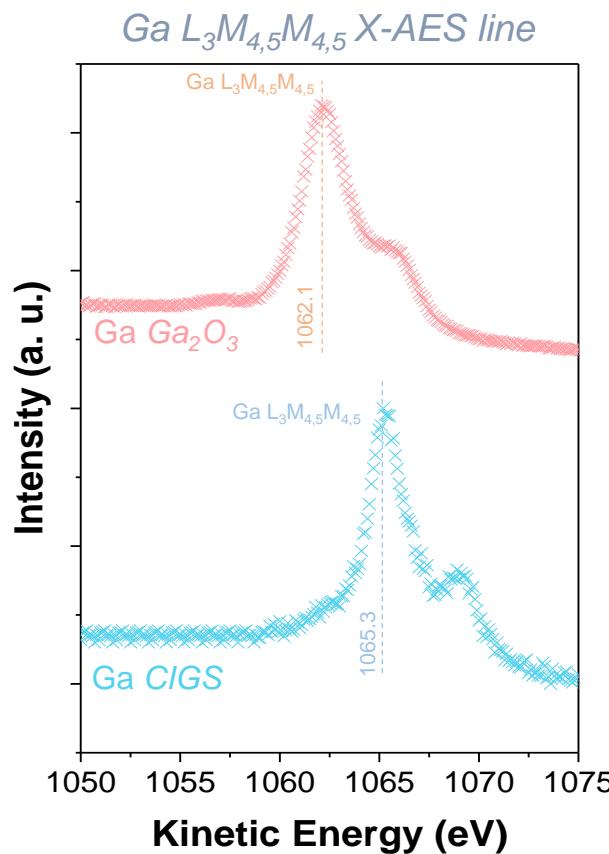
Peak 5	Peak 6
PP-7.3	PP+10.7
PP*1.13	PP*1.81
PP*1.34	PP*0.58

6 arbitrary peaks to decompose the
 $\text{In } \text{M}_{4,5}\text{N}_{4,5}\text{N}_{4,5}$ X-AES transition of In_2O_3
→ 5 for the $\text{Ga } \text{L}_3\text{M}_{4,5}\text{M}_{4,5}$ X-AES transition of Ga_2O_3

LLS decomposition of CIGS X-AES transitions

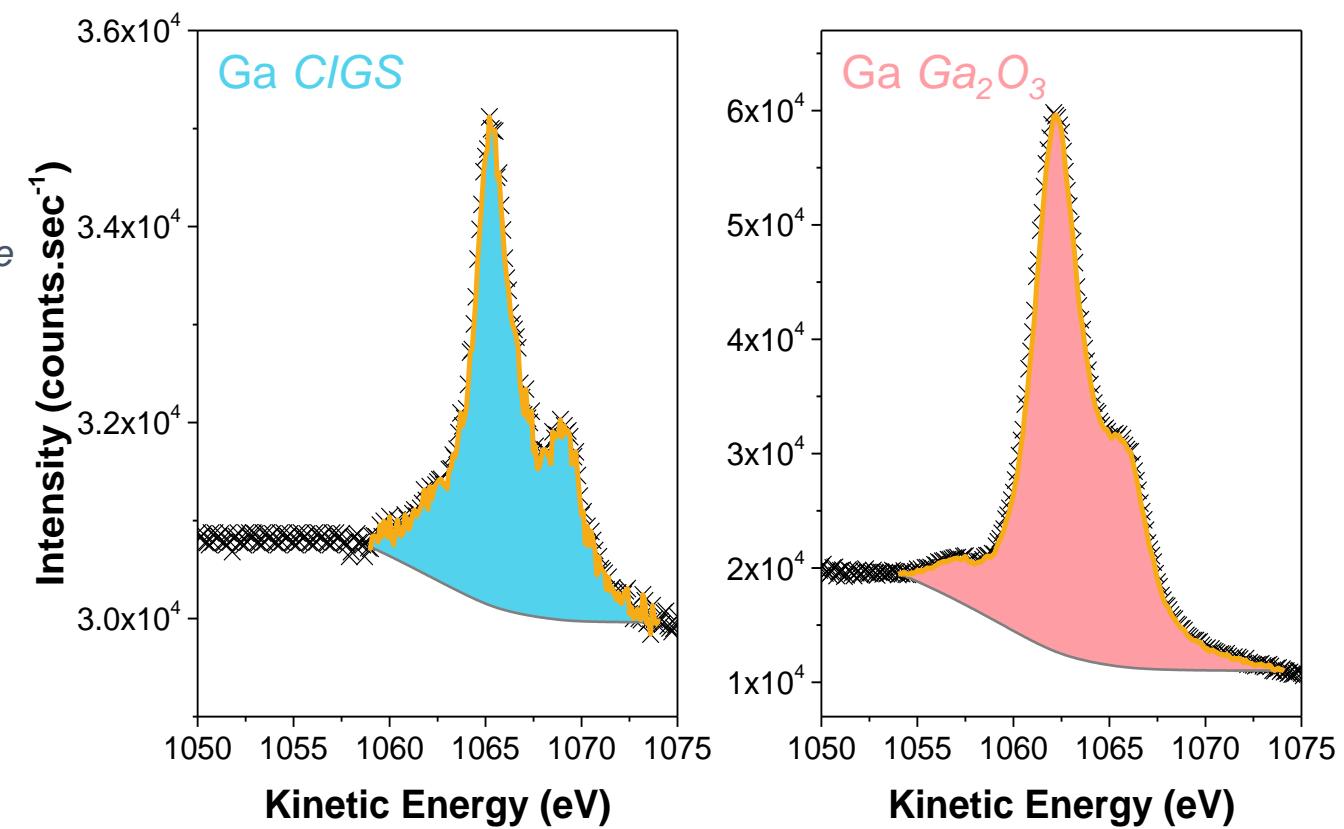
LLS

Modifications induced by the aging

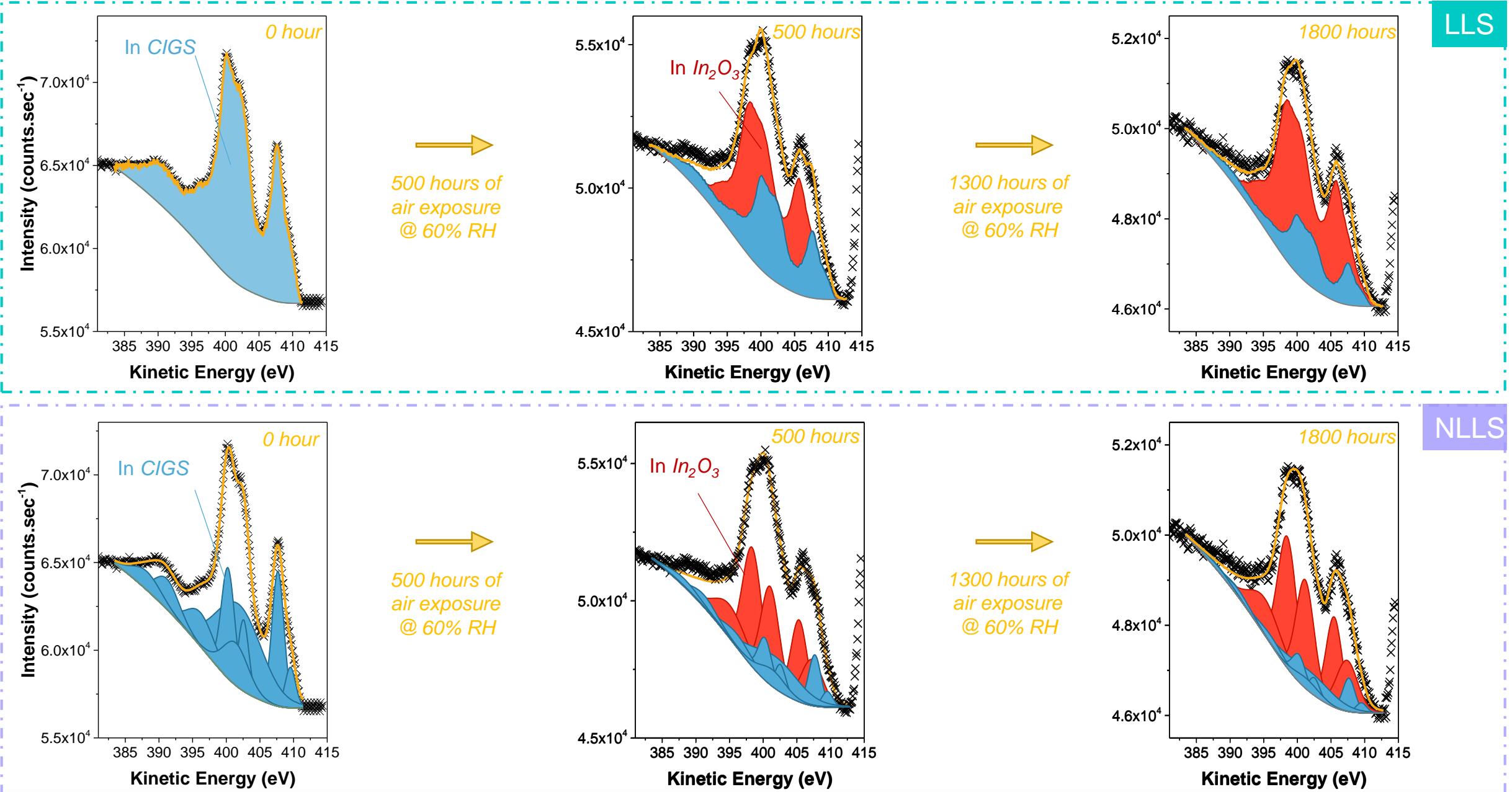


Thermo Nexsa
P.E. 20 eV; step size: 0.1 eV; dwell time: 100 ms
Bckg: U2 Tougaard – resolution 0.52 eV

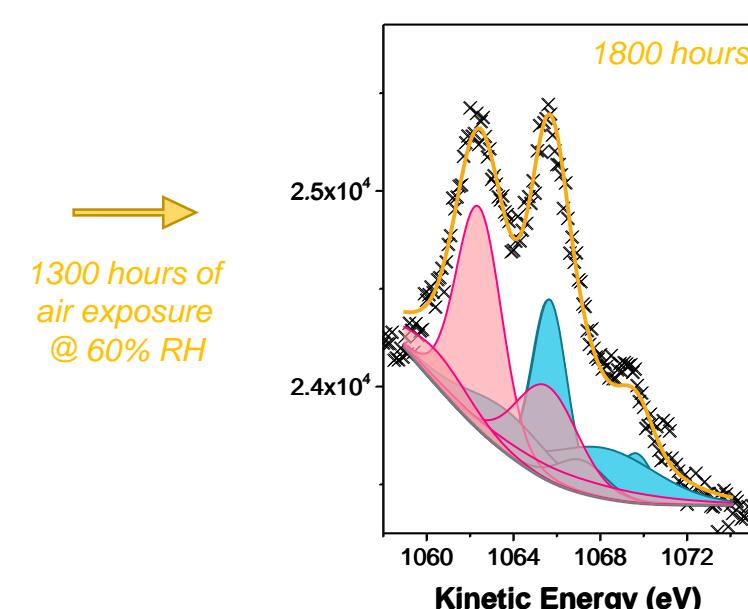
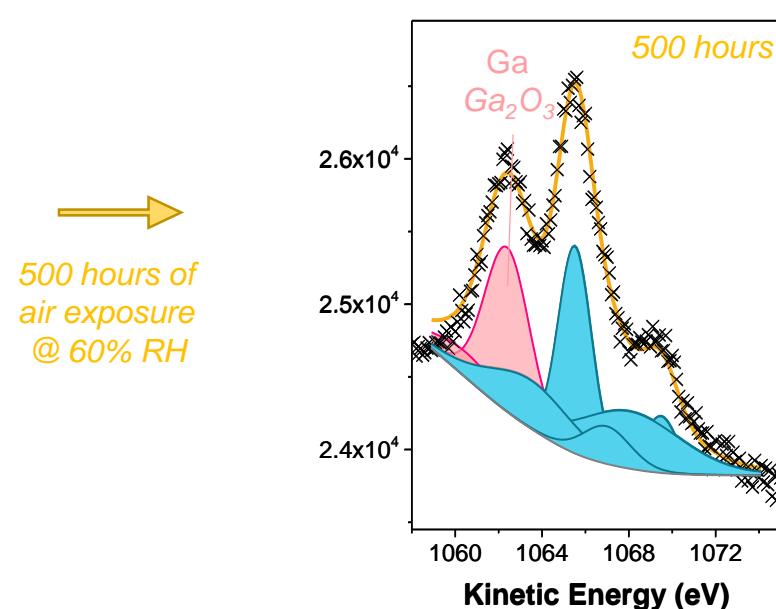
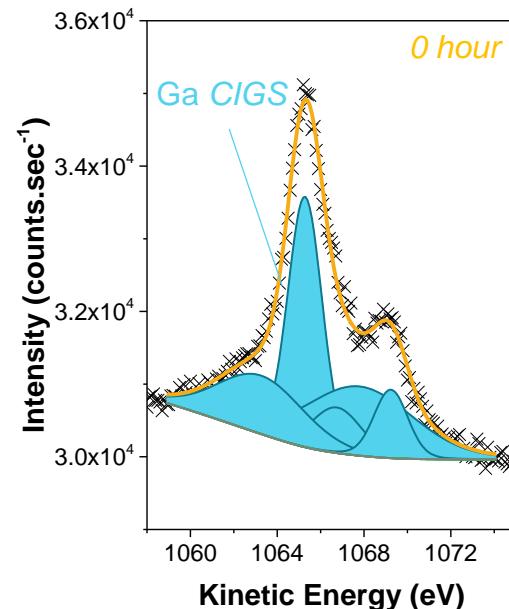
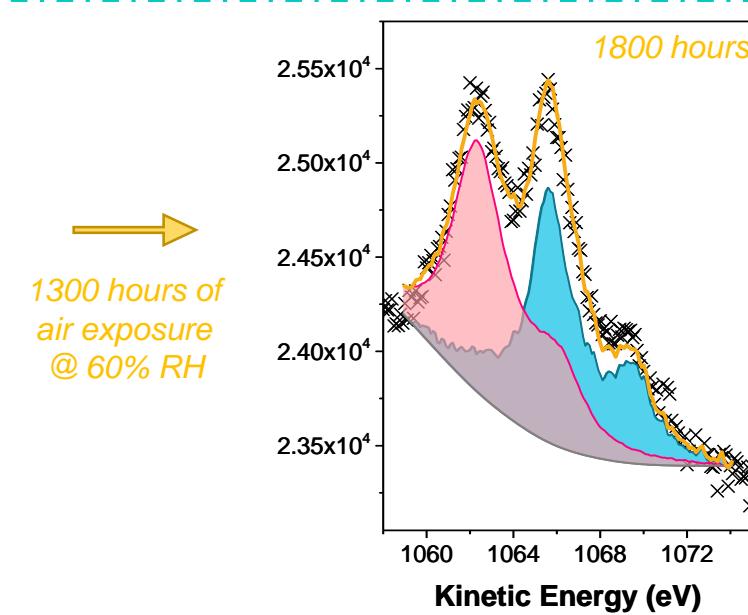
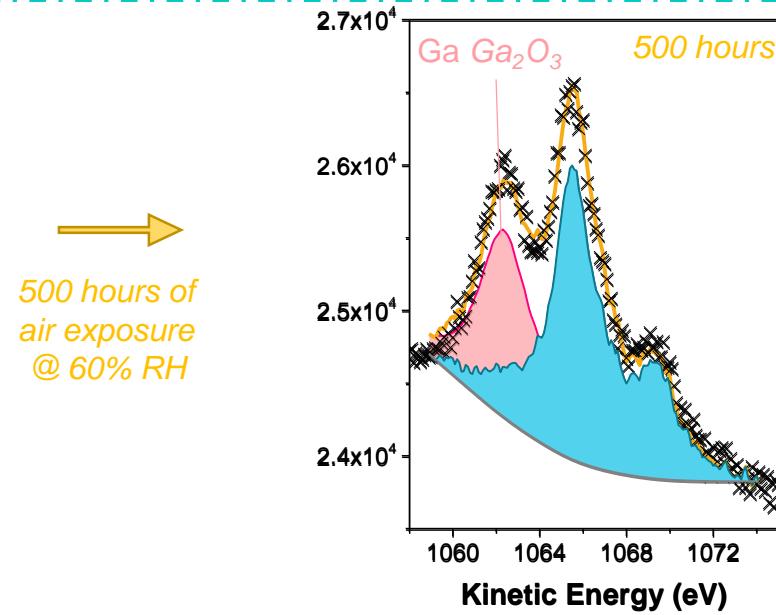
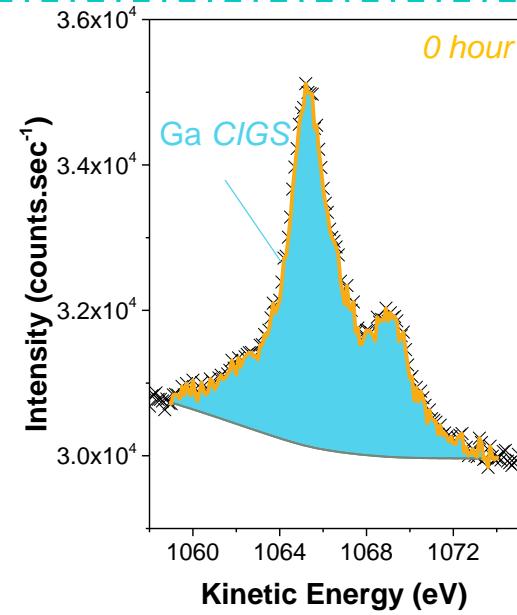
Growth of In_2O_3 and Ga_2O_3 phases in addition to the CIGS one
Identified with the modified Auger parameter



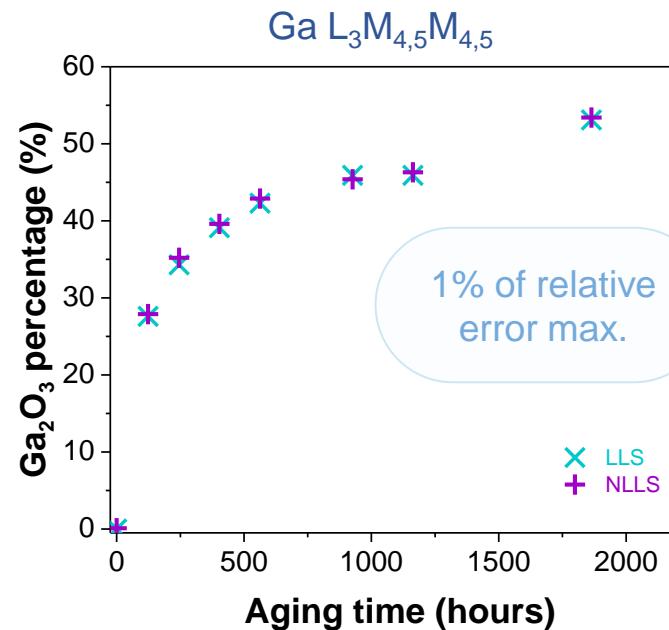
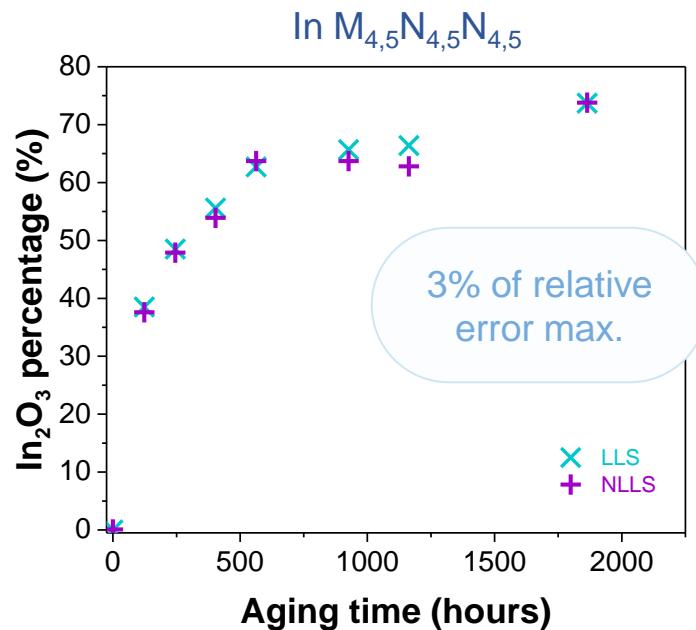
Evolution of the In $M_{4,5}N_{4,5}N_{4,5}$ X-AES transitions over air aging



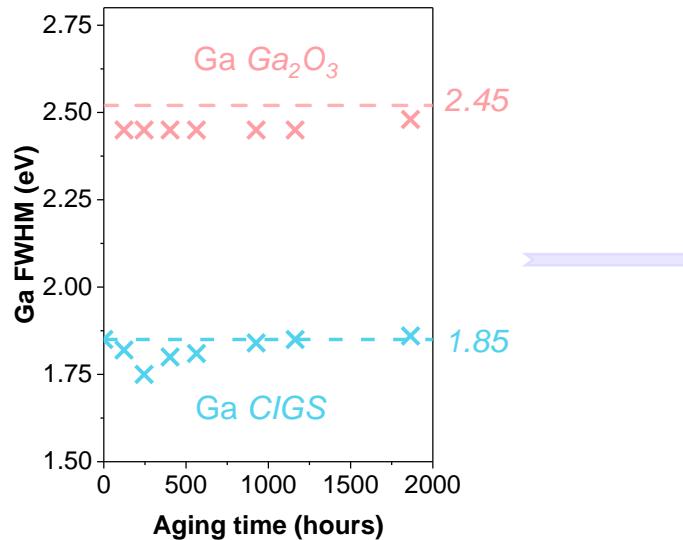
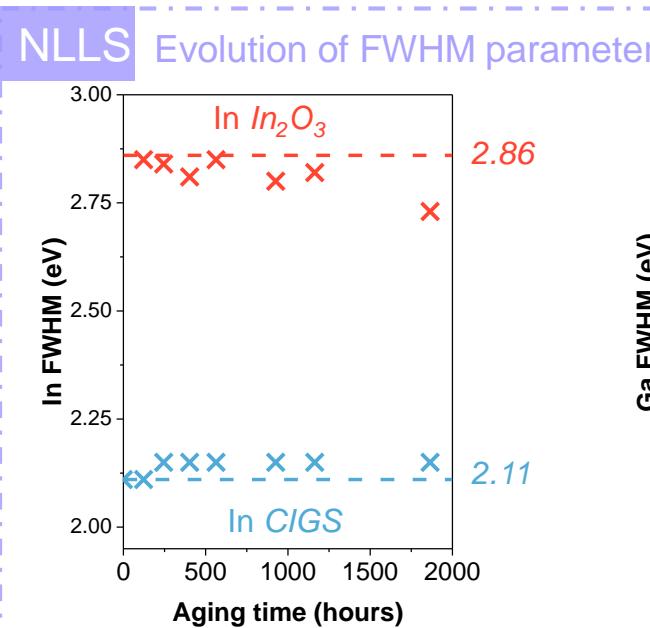
Evolution of the Ga $L_3M_{4,5}M_{4,5}$ X-AES transitions over air aging



LLS and NLLS decompositions comparison

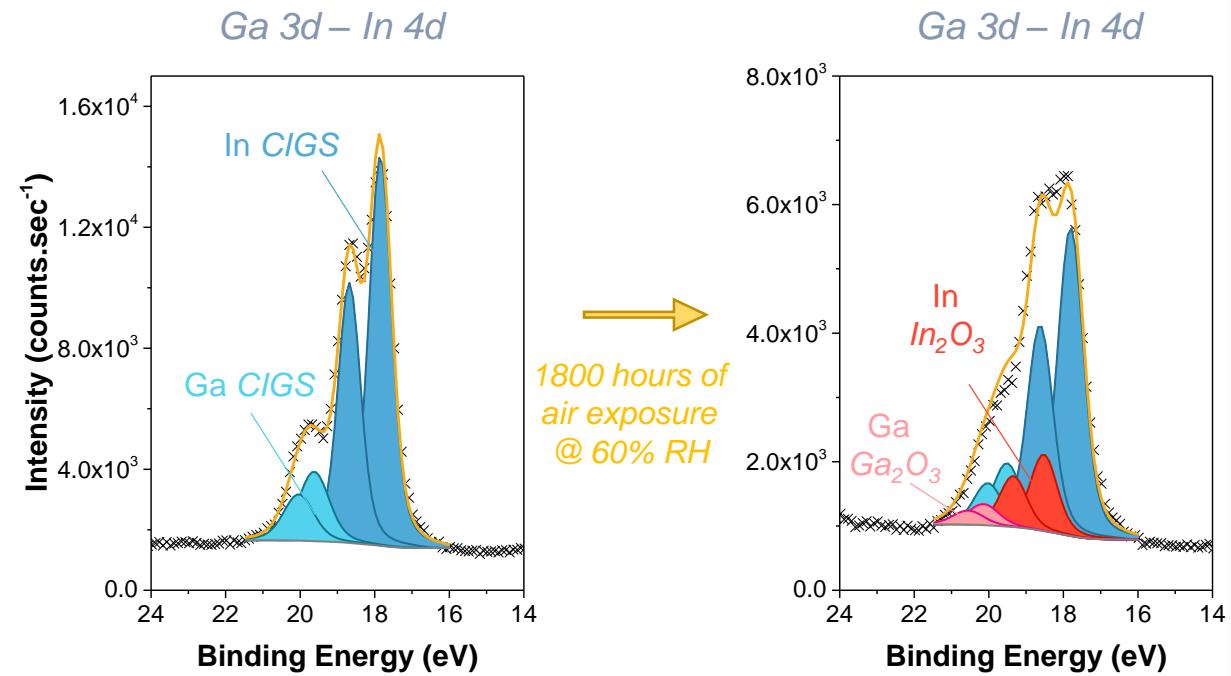
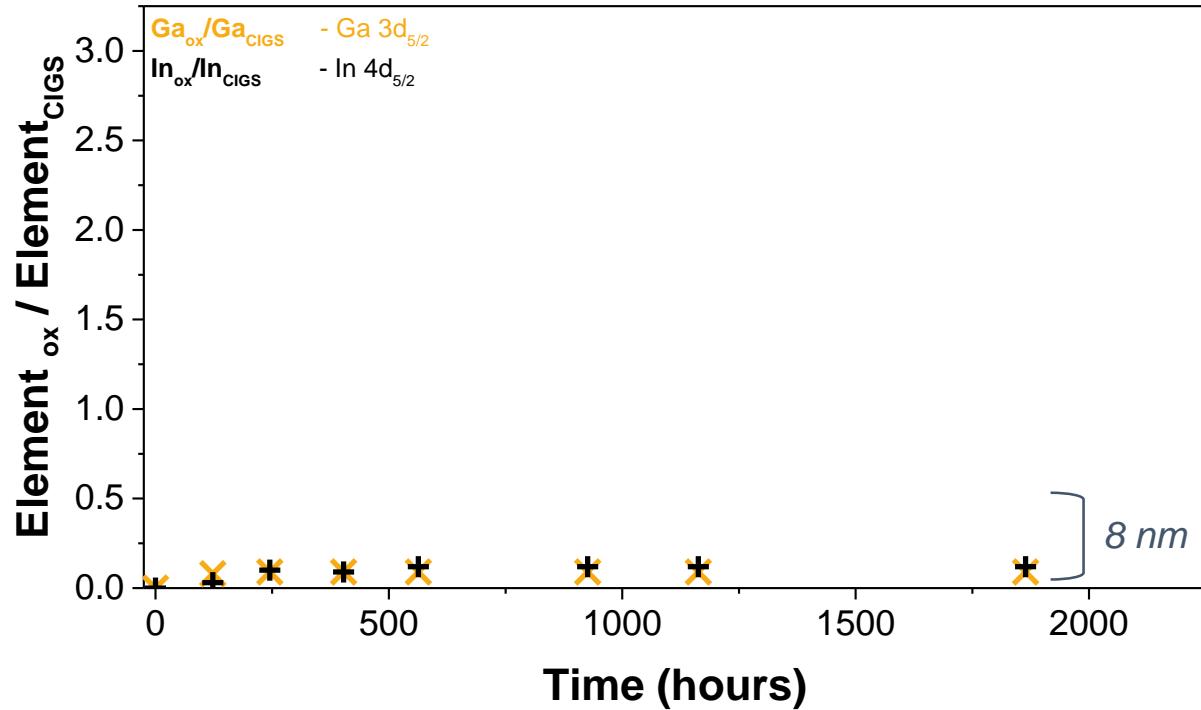


Linear least squares decomposition and non-linear least squares decomposition provide similar results for X-AES transitions.

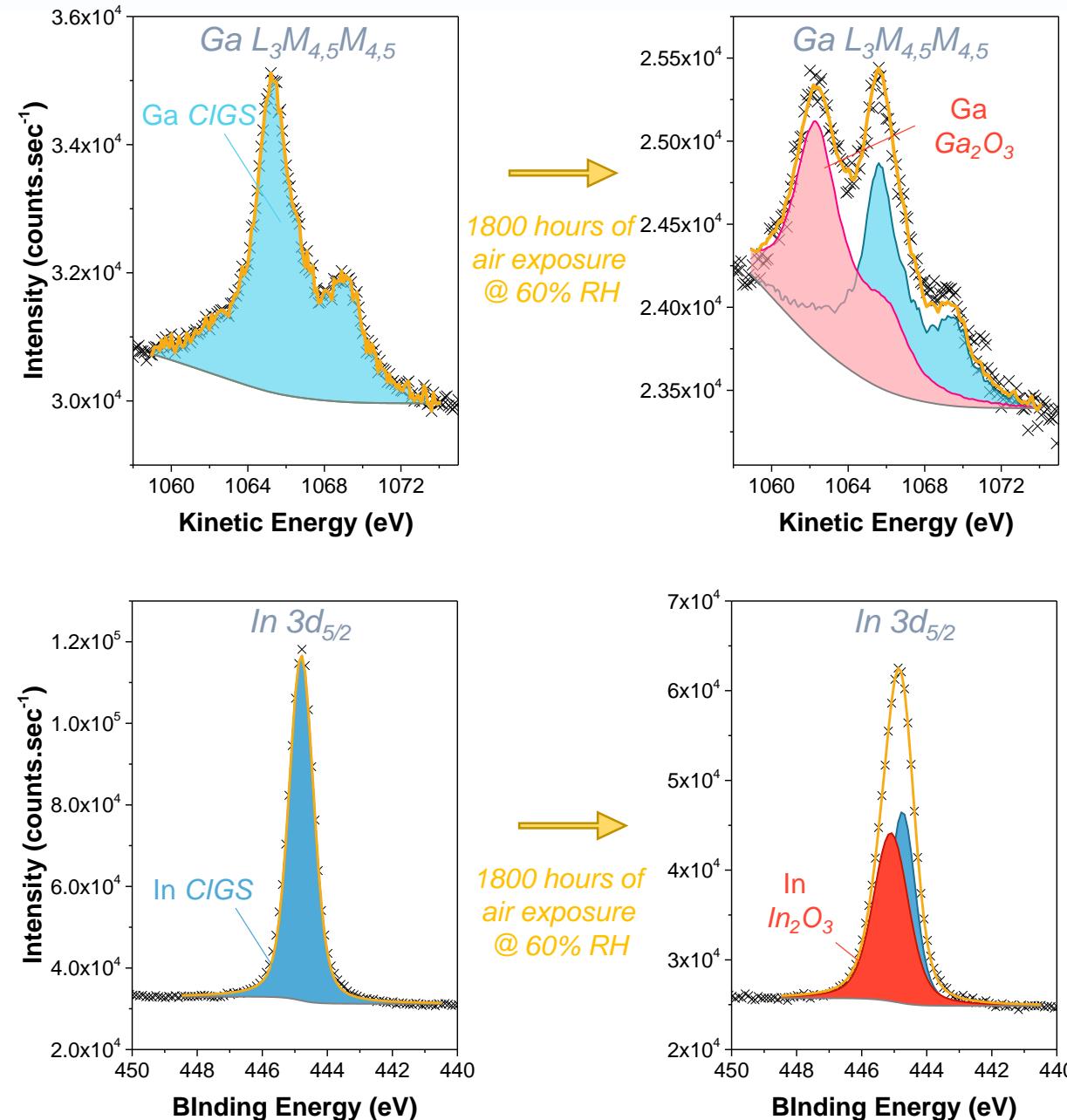
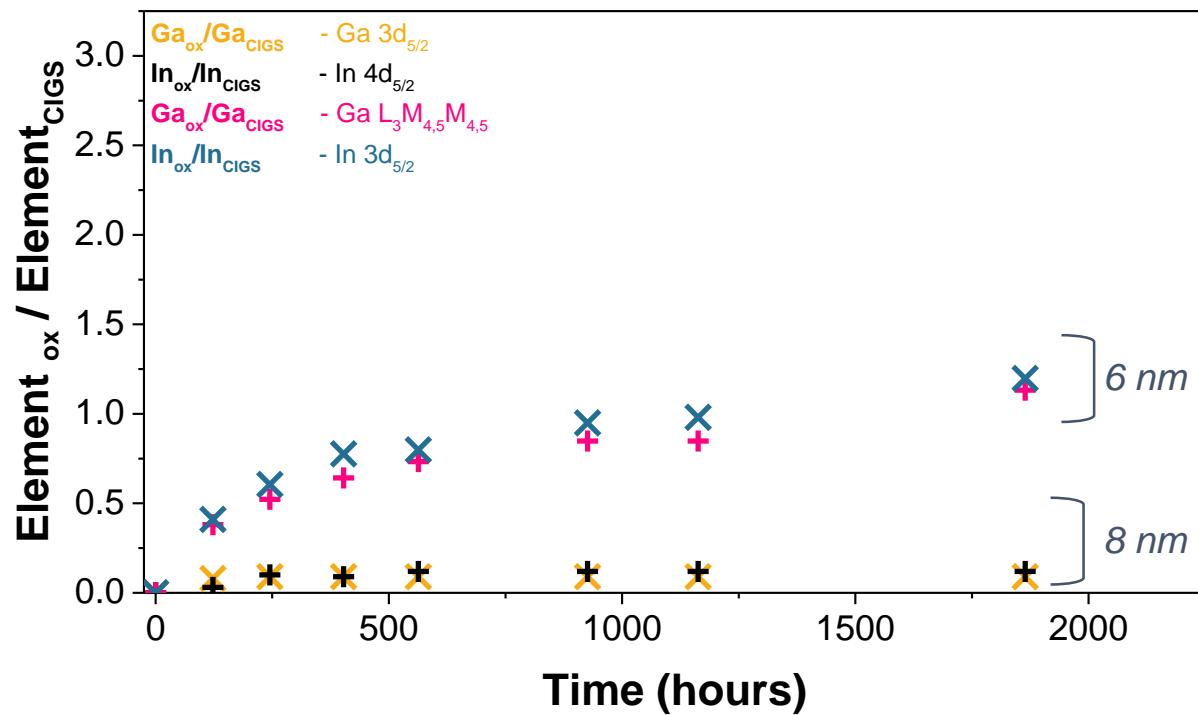


Only small FWHM variations observed during the aging

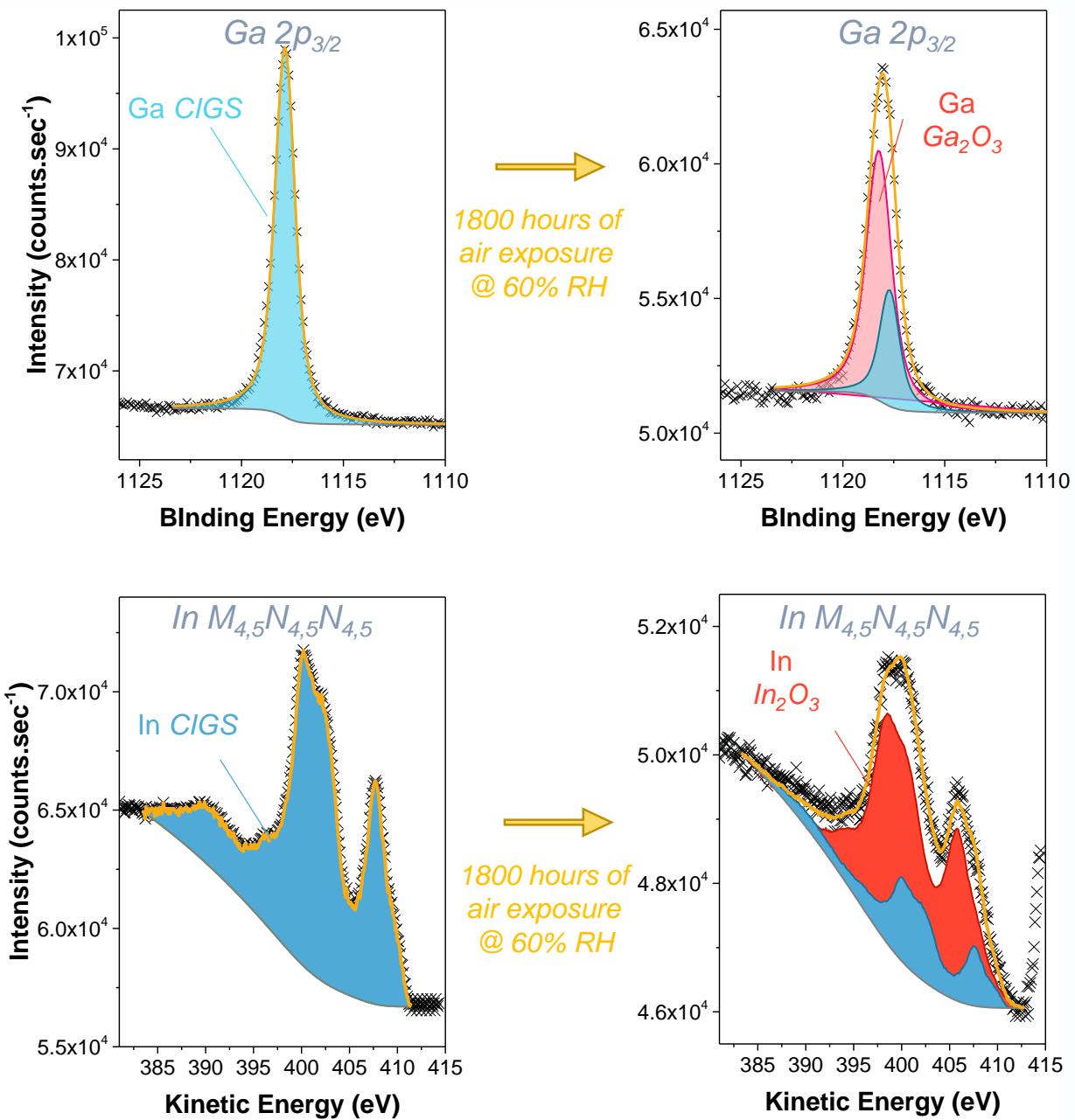
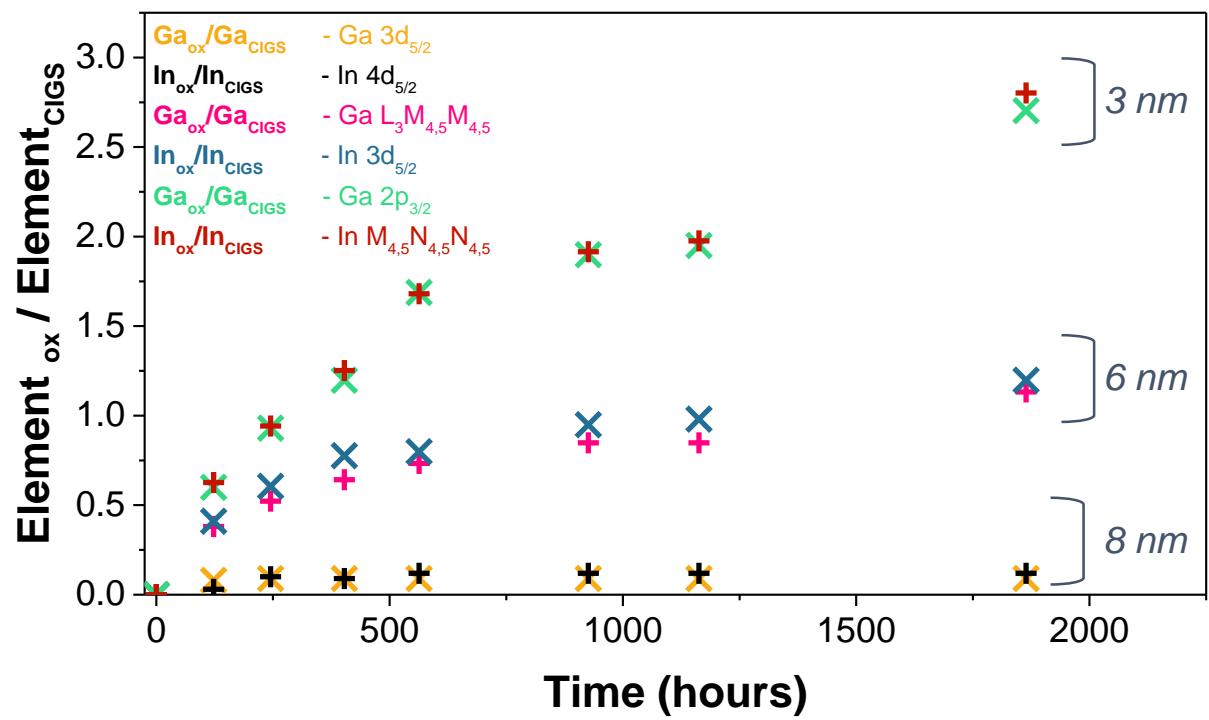
Application to CIGS aging



Application to CIGS aging



Application to CIGS aging



Conclusions

Decomposition of X-AES lines → 2 methods with advantages and disadvantages giving similar results

LLS

- Easy to use, with only one shape per chemical environment
- No possibility to provide any adjustments with aging (FWHM values...)



NLLS

- Possibility to perform evolution along with air aging (FWHM values, positions...)
- Up to 15 arbitrary peaks to simulate two chemical environments



Application to CIGS aging → access to a depth range probed for 2 elements within the same scale

- 3 different depths probed for Ga and In: 3, 6, and 8 nm using XPS photopeaks and X-AES lines
- Progressive oxidation from the extreme surface to the sub-surface, with a higher level of oxidation on the extreme surface





Neal Fairley



Anna Gagliardi



THANK YOU

Want to know more about the degradation of the sub surface of CIGS?
Don't miss Anna Gagliardi's talk
This afternoon @ 2.50 p.m.
11 NRG

S.B. and A.G. thank the French Agence Nationale de la Recherche for funding under the contract ANR-22-CE50-0008-01

