

Controlled Ru (II) and amp- Rh (III) catalyzed [4+2] and amp- [4+1] carbo-annulation via C–H Activation using Sulfoxonium ylide and amp- Maleimide: Facile access to Fused Benzoisoindole and amp- Spiroindanone

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Raju Vaggu, Akhila Muraharikar, Yenamareddy Bhargav Kumar, Yarasi Soujanya, René Grée, et al.. Controlled Ru (II) andamp- Rh (III) catalyzed [4+2] andamp- [4+1] carbo-annulation via C– H Activation using Sulfoxonium ylide andamp- Maleimide: Facile access to Fused Benzoisoindole andamp- Spiroindanone. European Journal of Organic Chemistry, 2024, 27 (47), pp.e202400885. 10.1002/ejoc.202400885 . hal-04822723

HAL Id: hal-04822723 https://hal.science/hal-04822723v1

Submitted on 17 Feb 2025

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COMMUNICATION

DOI: 10.1002/adsc.201((will be filled in by the editorial staff))

Controlled Ru (II) & Rh (III) catalyzed [4+2] & [4+1] carboannulations via C–H Activation using Sulfoxonium ylide and Maleimide: Facile access to Fused Benzoisoindole & Spiroindanone

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Received: ((will be filled in by the editorial staff))

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/adsc.201#######.((Please delete if not appropriate))

Abstract. A well-ordered strategy of sulfoxonium ylides as a directing group as well as carbene source for the Rucatalysed [4+2] and Rh-catalysed [4+1] carbo-annulations with maleimides has been described. These novel methodologies have been successfully established for the synthesis of complex spiroindanones and benzoisoindoles containing compounds by the transition metal-mediated C-H activation/functionalization reactions. Further, Rhcatalysed [4+1] annulated compounds were obtained in the absence of any oxidant. A possible rationale for the remarkable divergent selectivity between the two catalysts has been proposed.

Keywords: C-H Activation; Carbo-annulation; Transition Metal Catalyst; Benzoisoindole; Spiroindanone

Spiroindanones^[1] and benzoisoindoles^[2] structures are present in bioactive molecules, as well as in drugs, which explains the special attention paid by synthetic chemists^[3,4,8d] to these structures. Molecules of this type show biological activities that include anti-tumor, antibiotic^[5] and α 1A-adrenoceptor antagonist activity^[6,7] (Figure 1).



Figure 1. Representative examples of spiroindanone and fused benzoisoindole bioactive molecules.

In the past decades, many groups have developed elegant strategies towards the synthesis of various types of spiroindanones^[8] and benzoisoindoles^[9] by using transition metal catalysts.^[10]

Representative previous approches:



Present work:



Scheme 1. Previous approaches and present work.

The weakly coordinating group sulfoxonium ylide^[11a] is a recent addition as a directing group for C-H activation and it can used for further transition metal catalysed reactions.[11] The first examples were reported few years back on the reactions of such ylides with various alkynes to afford naphthol derivatives (Scheme 1).^[11f] The Rh(III) catalysts were used first for this kind of [4+2] annulation for C-H activation of alkenes,^[12] and then air-stable Co-catalysts proved to be efficient as well. Reactions with activated alkenes^[13] have been developed more recently affording, under Rh(III) catalysis, the indanones via a [4+1] annulation strategy with acrylates and acrylamides.^[14a] And very recently, similar moiety was reported using stoichiometric amount of metal oxidant^[11d] for the synthesis of spiroindanones or indanolylpyrrolidine-2-5-diones. In parallel to these studies, various types of carbo- and heterocyclic structures have been prepared by similar approaches using sulfoxonium ylides and transition metal catalysts, mostly using Rh-catalysts with oxa/azabicyclic olefins,^[15a] allenolates,^[15c] anthranils,^[15f,15g] then using Ru-catalysts for homocouplings^[15d,15e] and Co(III) complexes with arenes^[15h] and dioxazolones.^[15i]

In extension to our previous studies,^[16] we have decided to explore the reaction of these sulfoxonium ylides with readily available maleimides as coupling partners. Therefore, the goal of this publication is to demonstrate that the less expensive Ru (II) catalyst can be used in [4+2] carbo-annulation reactions for a facile route to complex benzoisoindoles. The protocol also allowed an improvement for efficient synthesis of spiroindanones via an Rh (III) catalysed C-H activation following a [4+1] carbo-annulation under oxidant free conditions. Further, we propose mechanisms for these reactions and an explanation for the selectivity difference between the two catalytic systems.

Initially, we started to explore our reaction (Table 1) with phenylsulfoxonium ylide 1a and N-phenyl maleimide 2 in presence of 5 mol % of [Cp*RhCl₂]₂ with 0.5 eq. of acetic acid and 10 mol% of AgSbF₆ in TFE as solvent at 90 °C for 10h^[14] and we obtained the desired adduct 3aa in 20% yield (entry 1). Then, we investigated the role of the acetic acid (entries 2-4) and we found the best results with 2 equivalents, affording 3aa with 75% yield (entry 4). Other silver salts, in 10 mol%, proved to be less efficient (entries 5-7). Increasing the quantity of AgSbF₆ (up to 30% mol) did not improve the vield of the reaction (entries 8-9). On the other hand, PivOH and NaOAc proved to have a low impact on the [4+1] carbo-annulation reaction furnishing the product 3aa in 45% and 40% yield respectively (entries 10 and 11). Next, we investigated different solvents in which TFE demonstrated the best yields (entries 12-17), although THF was another good possibility (entry 18). We studied also the role of the temperature and we found the best results at 90°C, affording **3aa** with 75% yield (entries 19-22). The reaction gave only poor yields in the absence of the silver salt (entry 23) or in the absence of acetic acid (entry 24). Finally, the yield was found to decrease (entry 25) in the absence of inert gas.

 Table 1. Optimization of the reaction under Rhcatalysis^a

\bigwedge		٦] - 0 - ر	hCp*Cl ₂] ₂ (5 AgSbF ₆ (10	5 mol %) mol)	
		N I Ph	AcOH(2e	q)	
1a		2	TFE, 90 °	C	3aa / N`Ph O
Entry	Silver salt	Additive	Solvent	Temp	Yield
	(10 mol %)	(2 eq.)	borvent	$(0^{\circ}C)$	$(\%)^{[b]}$
1	AgSbF ₆	AcOH	TFE	90	20
		(0.5 eq.)			
2	AgSbF ₆	AcOH	TFE	90	40
		(1.0 eq.)			
3	AgSbF ₆	AcOH	TFE	90	60
		(1.5 eq.)			
4	$AgSbF_6$	AcOH	TFE	90	75
		(2.0 eq.)			
5	$AgBF_4$	AcOH	TFE	90	50
6	AgOAc	AcOH	TFE	90	20
7	AgOTf	AcOH	TFE	90	30
8	AgSbF ₆	AcOH	TFE	90	74
	(20 mol %)				
9	AgSbF ₆	AcOH	TFE	90	63
	(30 mol %)				
10	AgSbF ₆	PivOH	TFE	90	45
11	AgSbF ₆	NaOAc	TFE	90	40
12	$AgSbF_6$	AcOH	DCE	90	20
13	$AgSbF_6$	AcOH	DMF	90	Trace
14	AgSbF ₆	AcOH	EtOAc	90	20
15	AgSbF ₆	AcOH	DMSO	90	No reaction
16	AgSbF ₆	AcOH	PhCl	90	No reaction
17	AgSbF ₆	AcOH	Toluene	90	50
18	$AgSbF_6$	AcOH	THF	90	75
19	$AgSbF_6$	AcOH	TFE	100	70
20	AgSbF ₆	AcOH	TFE	120	65
21	AgSbF ₆	AcOH	TFE	80	65
22	AgSbF ₆	AcOH	TFE	70	50
23	None	AcOH	TFE	90	25
24	AgSbF ₆	None	TFE	90	20
25°	AgSbF ₆	AcOH	TFE	90	30

Reaction conditions: ^{a)}**1a**(0.25mmol), **2**(0.375 mmol), [Rh(Cp*)Cl₂]₂ (5 mol%), AgSbF₆, (10 mol%), AcOH (2 eq.), N₂(g), TFE (1.5 ml), 90 °C, 12 to 22h. ^{b)}Isolated yield, ^{c)}Absence of nitrogen gas.

We carried out the next investigations at optimal standard conditions (entry 8) to explore the scope of this [4+1] annulation reaction as regards to both the aryl sulfoxonium ylides and the maleimides (Scheme 2). The model ylide 1a reacted with *N*-H maleimide to give the desired spiro-lactone 3ba in 71% yield,

similar to the one obtained with the N-phenyl maleimide **3aa** (75%). Subsequently, we tried with Nalkyl substituted maleimide to give the desired product **3ca** in 73% yield. Then, we performed reactions with alkyl containing N-phenyl substituted maleimides and we were pleased to obtain the desired products 3da-**3ea**, as well as the benzyl derivative **3fa** in 72%-73% yields. In the next series, we employed N-aryl substituted maleimides with halogens on the aromatic moiety. They similarly afforded the corresponding annulated spiro products 3ga-3ha-3ia and 3ja in fair yields (69% to 72%). We also explored the reaction of a sulfoxonium ylide with a methyl and t-butyl added in *para* position on the aromatic ring with N-phenyl maleimide, affording the annulated spiro product 3ka in 74% yield and 3ma in 74% yield. Satisfyingly, the sulfoxonium ylide with a chloro in ortho to the ketone reacted as well to give the desired annulated spiroproduct **3la** in 70% yield. In addition, naphthol substitution furnished the anticipated spiro-product 3na in 65% yield. The reactions with meta-substituted sulfoxonium ylide with electron donating and withdrawing groups furnished the desired spiro product 30a and 3pa in 71% and 63% respectively. All these heterocycles have spectral and analytical data in agreement with the proposed structures. In addition, the structure of 3ea has been confirmed by X-Ray crystallographic studies (see Supplementary Information).



AgSbF₆(10mol%), AcOH(2 eq), N₂(g), TFE(1.5ml), 90 °C, 10-20h.

Scheme 2. Synthesis of spiroindanones with sulfoxonium

ylides and maleimides via Rh (III) catalysed [4+1] carboannulation.

In parallel to previous studies, we have also explored the reaction of these two components, in the presence of a Ru (II) catalyst: $[RuCl_2(p-cymene)_2]_2$. After optimization of the reaction conditions (see the Supporting Information for details), we were delighted to find that the reaction of the sulfoxonium ylide **1a** and *N*-phenyl maleimide in presence of 5 mol % Ru (II) catalyst with 2 eq. of acetic acid and 15 mol% of AgSbF₆ in TFE at 80 °C for 14h afforded in 76% yield the [4+2] annulated product **4aa**, fully characterized by its spectral and analytical data. Then, we explored the substrate scope of this [4+2] annulation reaction with several aryl sulfoxonium ylides and various *N*substituted maleimides (Scheme 3).



Reaction conditions⁸:1(0.25mmol), **2**(0.375 mmol), [RuCl₂(p-cymene)]₂ (5 mol%), AgSbF₆ (15 mol%), AcOH(2 eq), N₂,TFE (1.5 ml), 80 °C, 10 to 20h.

Scheme 3. Synthesis of benzoisoindoles from sulfoxonium ylides and maleimides via Ru (II) catalysed [4+2] carbo-annulation.

First, sulfoxonium ylide **1a** reacted well with *N*-H maleimide affording **4ba** in 70% yield. Next, the same ylide reacted smoothly with two *N*-alkyl substituted maleimides to give the desired products **4ca - 4da** in 72%-71% yield, and with the *N*-cyclohexyl derivative

affording 4ea in 76% yield. The reaction of 1a worked accordingly well with two N-benzyl maleimides to afford 4fa and 4ga in 77% and 72% yield respectively. Similarly, the reaction of **1a** went on smoothly with four different N-arylmaleimides to give 4ia, 4ja, 4ka, and 4la in 71%-72% yields. Finally, this annulation reaction was performed also with five other sulfoxonium ylides. For those bearing electron donating in *para* position of the aromatic group, the desired compounds 4ma, 4na, 4qa, 4ra and 4sa were obtained in similar 72-78% yields. The structure of 4ma was confirmed by X-Ray crystallography.^[18] For electron-withdrawing vlides, possessing the substituents in *para* position the awaited compounds 40a and 4pa were obtained in slightly lower yields (60% and 69% respectively).

In the control experiments (Scheme 4), the sulfoxonium ylide proved to be a key directing group for the reaction with maleimide leading to the spirolactones and indanones. Similar reactions were performed starting with phosphonium derivative^[17] 5 and in that case, no reaction occurred (without any traces of 3aa or 4aa). Interestingly under our standard reaction conditions, phenacyl bromide 6 reacted with *N*-phenyl maleimide furnishing the phenacyl directed Rh (III) and Ru (II) catalyzed arene alkylation product 7a in 60 % yield with no Heck type adduct formation. Again, no traces of 3aa or 4aa were observed during latter reactions. Finally, we demonstrated that the reaction can be performed on gram scale (5 mmol) too under standard conditions furnishing the spirocompound 3aa in 66% yield and the annulated product 4aa in 71% yields.



Scheme 4. Control experiments

Based on the experiments and preceding literature ^[13,14] on Rh(III) catalysis the following plausible

mechanism has been proposed (Scheme 5). The vlide 1 forms a five membered rhodacycle ring A after C-H activation followed by the concerted metalation deprotonation. Then, a seven membered rhodacycle **B** is formed by the insertion of maleimide onto A and the intermediate C is generated by the β -hydride elimination of **B**. At this point, two pathways appear possible to obtain the final product. In pathway-1 the intermediate C leads to the 6-membered rhodacycle D by the 1,3-insertion of hydride. Next, it leads to intermediate E by elimination of DMSO and internal oxidation of Rh (III) to Rh (V). Then, the intermediate **F** is generated by migratory insertion. In pathway-2, the intermediate C generates the hydride D' by elimination of DMSO and internal oxidation. Next, D' leads to the formation of E' by a 1,3-insertion, and an intramolecular 1,4-addition generates the intermediate **F**. In both pathways, the desired product **3** is generated by the final proto-demetalation of **F** by AcOH.



Scheme 5. Plausible mechanism for C-H activation followed by [4+1] carbo-annulation

In the case of Ru (II) we did not obtain the same [4+1] carbo-annulated products as before with the rhodium catalyst, but instead we obtained benzoisoindole-type products through a [4+2] carbo-annulation and we can propose a mechanism for this process, as indicated in Scheme 6. The ylide 1 forms a five membered chelated intermediate **G** by a concerted metalation deprotonation process and then it leads to intermediate **H** by the insertion of maleimide. The internal oxidation of Ru (II) to Ru (IV) and the elimination of DMSO leads to the formation of intermediate **I**. Next, a reductive elimination gives **J** which, after protodemetalation generate the final product **4**.

The preference of Rh (III) catalyst towards spiro product may be attributed to a greater Lewis acidity of Rhodium with smaller ionic radius (0.665 Å) and therefore it is electron deficient compared to Ruthenium. Thus, *int*-**B** can undergo β -hydride elimination and Rh immediately gets attached to the carbon atom by 1,3 insertion leading to the formation of the spiro *int*-**F**. However, looking at the plausible mechanism of Ru catalysed annulation has offered no similar possibility.



Scheme 6. Plausible mechanism for C-H activation followed by [4+2] carbo-annulation

In conclusion, we have developed a new strategy to synthesize both complex spiroindanone-type molecules and benzoisoindole derivatives by taking advantage of the capabilities of sulfoxonium ylide to act as directing group as well as carbene source in C-H activation/functionalization reactions, followed by transition metal-mediated carbo-annulations with maleimide derivatives. In the case of the Ru (II)activation catalysed C-H the benzoisoindole derivatives 4 were obtained in a [4+2] annulation process with N-phenyl maleimide. To the best of our knowledge, it is the first example of such a reaction of ruthenium-catalysed reaction with sulfoxonium ylides. On the other hand, the directed Rh (III)-catalysed [4+1] tandem annulation provided the complex spiro molecules 3. Herein subsequently, rhodium-catalysed [4+1] carbo-annulation reaction with sulfoxonium ylides has been realized in the absence of any additional metal oxidant such Cu(OAc)₂, previously reported.^[11d] The divergent reactivities of Ru(II) and Rh(III) catalyst systems on similar substrates (sulfoxonium ylides and maleimides) furnishing very different scaffolds via C-H activation followed by [4+2] or [4+1] carbo-annulation has been projected establishing a comprehensive mechanistic pathways, highlighting the possible reasons for a fundamentally different behaviour of these catalytic cycles. Both annulation reactions have a broad scope and they are performed under oxidant free and mild reaction conditions. In addition, this work not only expands the application of sulfoxonium ylides in C-H activation, but also provides a new strategy for switchable access to one-pot construction of polyheterocyclic scaffolds.

Experimental Section

General procedure for the synthesis of compounds (3aa-3la): An oven-dried Schlenk tube equipped with a magnetic stir bar was charged with $[Cp*Rh(Cl_2)]_2$ (0.0125 mmol, 5 mol %), AgSbF₆ (0.025 mmol, 10 mol %), AcOH (2 equiv) in a small portion TFE (0.5 ml), **1** (0.25 mmol, 1.0 equiv), maleimide **2** (0.375 mmol, 1.5 equiv) followed by TFE (1.5 ml). The tube was then flushed several times with nitrogen gas, to ensure removal of moisture and placed in a preheated silicon oil bath at 90°C and the reaction mixture was stirred at the same temperature for 10-16 h under nitrogen atmosphere. After completion of the reaction (monitored by TLC), the reaction mixture was then allowed to cool down to room temperature and filtered through the short silica bed with (100-200 mesh) and concentrated under reduced pressure. The resulted crude mixture was purified by flash column chromatography ethyl acetate/hexane (30:70) on silica gel afforded **3** as white colour solid in 65-75% isolated yield.

General procedure for the synthesis of compounds (4aa-4pa): An oven-dried Schlenk tube equipped with a magnetic stir bar was charged with $[Ru(p-cymene)Cl_2]_2$ (0.0125 mmol, 5 mol %), AgSbF₆ (0.025 mmol, 15 mol %), AcOH (2 equiv) in a small portion TFE (0.5 ml), 1 (0.25 mmol, 1.0 equiv), maleimide 2 (0.375 mmol, 1.5 equiv) followed by TFE (1.5 ml). The tube was then flushed several times with nitrogen gas, to ensure removal of moisture and placed in a pre-heated silicon oil bath at 80°C and the reaction mixture was stirred at the same temperature for 10-16 h under nitrogen atmosphere. After completion of the reaction (monitored by TLC), the reaction mixture was then allowed to cool down to room temperature and filtered through the short silica bed with (100-200 mesh) and concentrated under reduced pressure. The resulted crude mixture was purified by flash column chromatography, ethyl acetate/hexane (30:70) on silica gel afforded **4** as white colour solid in 60-78% isolated yields.

X-Ray Data: Deposition numbers CCDC 2314310 (for **3ea**), and CCDC 2314306 (for **4ma**) contain the supplementary crystallographic data for this paper. These data are provided free of charge by the joint Cambridge Crystallographic Data Centre and Fachinformationszentrum Karlsruhe Access Structures service.

Acknowledgements

V. R. thank CSIR for fellowship. Y. B. K. thank CSIR-IICT for financial support. All the authors thanks the Indo-French "Joint Laboratory for Natural Products and Synthesis toward Affordable Health" program for the platform to carry out the finding jointly. We sincerely thank the Department of Analytical and Structural Chemistry at CSIR-IICT for all the analytical support and especially for X-ray crystal data. We gratefully acknowledge the overall support by CSIR-IICT to perform the research (IICT/Pubs./2024/124)

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COMMUNICATION

Controlled Ru (II) and Rh (III) catalysed [4+2] & [4+1] carbo-annulations via C–H Activation using Sulfoxonium ylides and Maleimides: Facile access to Fused Benzoisoindoles and Spiroindanones

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● TM switchable reactivity | ● C-H Activation | ● Carbo-annulation | ● Oxidant free

A new strategy has been developed to synthesize both complex spiroindanone-type molecules and benzoisoindole derivatives by taking advantage of the capabilities of sulfoxonium ylide to act as directing group as well as carbene source in the C-H activation/functionalization reactions, followed by transition metal-mediated carbo-annulations in [4+2] or [4+1] with maleimide derivatives