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1 **Rare earth elements as indicators of post-Marinoan**
2 **(~635 Ma) paleoceanographic changes from the Amazon**
3 **Craton**

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21 **ABSTRACT**

22 The extensive deglaciation linked to snowball Earth climate elapsed in terminal
23 Neoproterozoic caused dramatic changes in the ocean's chemistry, exceptionally recorded
24 in the cap carbonate sequence. One of the most spectacular examples is the post-Marinoan
25 Puga cap carbonate (~635-622 Ma), exposed in the Amazon Craton associated with
26 expressive sea level changes related to the glacial-isostatic adjustment (GIA) resulting in
27 continuous mixed-waters, coastal upwelling succeeded by long-term transgression under
28 greenhouse climax. To better understand the complex dynamics during this period, we
29 conducted a comprehensive multiproxy study encompassing sedimentological, rare earth
30 elements, and stable isotopic analysis. The Rare Earth Element + Yttrium (REY) patterns
31 in the Puga cap carbonate reflect the marine conditions aftermath of the aftermath of the
32 snowball Earth. Specifically, the Y/Ho ratios in the cap dolostones consistently fall below
33 modern seawater values interpreted as seawater composition dilution due to the meltwater
34 influx from the continents during deglaciation. However, superchondritic Y/Ho ratios (up
35 to 70) and positive Eu anomalies (Eu/Eu* up to 3.6) are observed in the basal portion of
36 the cap dolostone. These results suggest the upwelling of hypersaline seawater with some
37 contribution from contemporaneous hydrothermal fluids in shallow coastal water. These
38 findings are interpreted as evidence of ocean destratification related to upwelling, ocean
39 circulation, and glacioisostatic adjustment. The cap dolostone occurs with minimal
40 siliciclastic contributions, consistent with a siliciclastic starvation model. It could be
41 attributed to the persistent shallow-water conditions in the coastal zone, significantly
42 influenced by glacial isostatic adjustment (GIA). This process led to continuous uplift,
43 resulting in a shallow shelf isolated from direct oceanic processes. Barrier-inland facies
44 were not observed in the studied succession, suggesting that irregular diamicton
45 morphologies may have created protected areas against waves. This scenario maintained
46 a consistently shallow water column and followed a significant landward shoreline
47 migration depositional pattern. These dramatic changes have been assumed to have
48 resulted from the rapid recovery of primary productivity during deglaciation, with the
49 proliferation of microbial communities and induced dolomite precipitation. The $\delta^{13}\text{C}$
50 isotopic signature is consistent with these interpretations and with the values found for
51 the Ediacaran in other cap carbonates. Mg concentration was constant during restricted
52 paleoenvironmental conditions on dolomitic platforms. The rapid rise in sea level led to
53 the dispersion of Mg ions, leading to the definitive interruption of dolomicrite
54 precipitation, abruptly replaced by seas supersaturated in CaCO_3 with massive aragonite
55 precipitation coinciding with the increase in detrital components under the influence of a
56 long-term transgression in the initial Ediacaran.

57 **Keywords:** cap dolostone; starvation siliciclastic, plumeword; restriction; upwelling

58 **1 INTRODUCTION**

59

60 The transition from the extreme Snowball Earth climate state to more temperate
61 conditions represents a pivotal interval (~720–635 Ma) of Earth's geologic history. These
62 drastic changes were manifested in marine sediments through the dramatic transformation
63 from a globally frozen ocean to an ice-free one (Hoffman et al. 2017, 2021). Significant

64 perturbations triggered large changes to sea level during the pre-, syn-, and post-glacial
65 stages, primarily driven by the effects of glacial eustasy (global mean sea level), glacial-
66 isostatic adjustment (GIA), and the thermal expansion of the oceans under greenhouse
67 conditions (Allen and Hoffman, 2005; Benn et al. 2015; Creveling and Mitrovica, 2014;
68 Hoffman et al. 2017, 2021, Mitchell et al. 2021, Rame and Marotzke, 2022). During the
69 advance of the syn-deglacial transgression, meltwater production and surface warming
70 created stable density stratification (Yang et al. 2017). Salinity below the meltwater layer,
71 which encompassed the majority of the global ocean at this time, has been estimated to
72 be three times higher than the modern ocean, creating a stable stratification that may have
73 lasted thousands to hundreds of thousands of years (Shield, 2005; Liu et al. 2014; 2018;
74 Yang et al. 2017; Hoffman et al. 2017). The GIA and oceanic circulation were crucial
75 factors for the end of stratification during the post-melting stage (Yang et al., 2017; Rame
76 and Marotzke, 2022).

77 Globally deposited cap carbonate sequences record a critical interval documenting
78 the transition out of the Marinoan Snowball Earth (Hoffman et al. 2017). These sequences
79 are recognized as the Cryogenian-Ediacaran boundary and consist of basal Ediacaran cap
80 dolostones, which are conformably overlain by cap limestones. These associations exhibit
81 striking similarities across continents and have been globally correlated through
82 geochronological and geochemical means (e.g., Halverson et al., 2005; Corsetti and
83 Lorentz, 2006; Hoffman et al., 2017; Yu et al., 2022).

84 Current cap carbonate formation models have been proposed to explain the
85 genesis of the Marinoan cap dolostones. In one model, it is suggested that increased
86 seawater alkalinity played a crucial role in the precipitation and widespread
87 dolomitization observed during cap dolostone deposition. These factors include the
88 overturning of deep anoxic ocean waters (Grotzinger and Knoll, 1995; Hoffman et al.,

89 1998; James et al., 2001), intense chemical weathering (Hoffman et al. 1998; Higgins and
90 Schrag, 2003), Hydrothermal alkalinity model: alkalinity increases linked to extensive
91 hydrothermal activity along basin-bounding faults (Gernon et al., 2016) gas-hydrate
92 destabilization and massive methane release (Jiang et al. 2003, Jiang et al. 2006; Kennedy
93 et al. 2008; Wang et al. 2008), alkalinity increases associated to microbial activity (e.g.,
94 sulfate reduction) (Nogueira et al. 2003,2007, 2022; Font et al. 2006) and meltwater
95 plumes (Shields, 2005; Yang et al. 2017; Liu et al.2014; Santos et al. 2021).

96 Despite the widespread distribution and intense study of these post-Marinoan
97 sequences and the various proposed models, there has yet to be a consensus regarding
98 their genesis. One aspect that remains to be understood is the microfacies changes from
99 dolostone to limestone. This texture and facies composition transition is key to
100 unraveling the intricate processes during the initial Ediacaran. The precipitation
101 mechanism of cap dolostones remains a subject of intense debate. One reason for these
102 ongoing debates is the variability between units; for example, many units record
103 enigmatic barite horizons (e.g., Bao et al. 2008; Crockford et al. 2016; 2018), while many
104 others do not. Despite this variability, to date, two main hypotheses have been proposed.
105 The first suggests that dolostones formed directly, retaining their primary characteristics
106 and reflecting the chemistry of the original depositional environment (Kennedy, 1996;
107 Shields, 2005; Nogueira et al., 2003; Nedeléc et al., 2007; Font et al., 2006; Santos et al.
108 2021, 2023). On the other hand, the second hypothesis proposes that the dolostones
109 underwent secondary dolomitization, wherein an aragonite or calcite precursor was
110 transformed into dolomite (Fabre et al. 2013; Higgins et al. 2018; Ahm et al. 2019;
111 Gammon et al. 2012; Gammon, 2012).

112 Rare earth elements and yttrium (REE+Y) have been extensively utilized in
113 paleoenvironmental and paleoceanographic studies to provide insights into the aftermath

114 of Snowball Earth events (Font et al. 2006; Nedelec et al. 2007, Huang et al. 2009; Meyer
115 et al. 2012; Wang et al. 2014; Viera et al. 2015; Hohl et al. 2015, 2017; Caxito et al. 2018;
116 Verdel et al. 2018; Wu et al. 2019; Zhao et al. 2021; Cui et al. 2021; Ren and Li, 2023)..
117 These elements, when analyzed in ancient non-skeletal carbonates such as microbialites,
118 serve as proxies for reconstructing paleoenvironmental conditions and offer valuable
119 information about past seawater conditions (Schere and Suits, 1980; Shaw and
120 Wasserburg, 1985; Webb and Kamber, 2000; Kamber and Webb, 2001; Nothdurft et al.
121 2004; Huang et al. 2009; Ling et al. 2013). The incorporation of REE+Y into carbonate
122 minerals is controlled by various physicochemical processes, such as water chemistry,
123 pH, and temperature (Wood, 1990; Nozaki and Alibo, 2003; Chen et al., 2015; Bolhar et
124 al., 2015; Rahlf et al. 2021; Zhao et al. 2022). Consequently, the utility of marine
125 carbonates as recorders of oceanographic conditions hinges upon the preservation quality
126 of these carbonates, well-established diagenetic constraints, and a comprehensive
127 understanding of contamination by noncarbonate phases.

128 Abundant carbonate deposits with well-preserved primary textures occur in the
129 Southern Amazon Craton associated with the Marinoan Puga cap carbonate. Previous
130 studies have predominantly focused on microbial carbonate petrology, sedimentology,
131 and $\delta^{13}\text{C}$ and $\delta^{18}\text{O}$ geochemical values of carbonates (Nogueira et al. 2003; 2022; Font et
132 al. 2006; 2010; Sansjofre et al. 2011; Soares et al. 2020; Romero et al. 2020; Santos et al.
133 2021,2023). However, despite well-established geological constraints, this succession
134 lacks a comprehensive multiproxy study to constrain the post-Marinoan conditions and
135 test previous carbonate precipitation models, thereby refining our understanding of the
136 factors driving the changes in the carbonate factory over this critical time interval. The
137 evaluation of the Puga cap carbonate exposed in the southern part of the Amazon Craton
138 represents a unique opportunity to achieve this end due to its exceptional preservation,

139 largely unaffected by severe late-stage diagenetic modification (Sansjofre et al. 2014;
140 Soares et al. 2020; Santos et al. 2021). Therefore, a comprehensive multiproxy study
141 incorporating sedimentological aspects (microfacies and facies), rare earth elements (e.g.,
142 Y/Ho, Eu/Eu* and Ce/Ce* anomalies), and isotopic composition of $\delta^{13}\text{C}$ and $\delta^{18}\text{O}$ may
143 provide the needed insights into the post-Marinoan conditions in this region. For example,
144 the impact of meltwater and stable stratification on ocean circulation in this region and
145 the potential influence of siliciclastic starvation on precipitation and carbonate factory
146 changes (from dolostone to aragonite) remain to be evaluated.

147

148 **2. GEOLOGICAL BACKGROUND**

149

150 The focus of this study is the Puga cap carbonate, which is in the Southern
151 Amazon Craton and is associated with the basal deposits of the intracratonic Araras-Alto
152 Paraguai Basin installed during the Late Cryogenian (Fig. 1 A and B). (Nogueira et al.
153 2003; 2019; Gaia et al. 2017; Afonso and Nogueira, 2018; Santos et al. 2020; Soares et
154 al. 2020; Romero et al. 2020; Santos et al. 2021,2023). This basin overlies crystalline and
155 metamorphic basement rocks of the Cuiabá Group that characterize the Paraguai Belt
156 (Nogueira et al. 2019; Santos et al. 2020). The basal fill of this basin is composed of
157 Cryogenian rocks of the Puga Formation (pelites, sandstones, dropstones, and glacial
158 diamictites) overlain by earliest Ediacaran carbonate rocks of the Araras Group (Fig. 1C),
159 which consists of four formations: 1) Mirassol d'Oeste (Cap dolostone), composed of
160 dolomitic layers interpreted as a shallow platform environment; 2) Guia (Cap Limestone),
161 composed of carbonates and shales, interpreted as a moderately deep platform; 3) Serra
162 do Quilombo, composed of shallow to moderately deep dolomitic platforms; and 4)
163 Nobres, consisting of tidally influenced dolomites (Nogueira and Riccomini, 2006;

164 Nogueira et al. 2003; 2007; 2019;2022; Rudnitzki et al. 2016; Romero et al. 2020; 2022;
165 Soares et al. 2013; 2020; Santos et al. 2021).

166 The cap dolostone comprises stratiform stromatolites with gypsum
167 pseudomorphs, giant stromatolites with tubestone, and wave-dominated dolostone facies
168 (peloidal dolomudstone with parallel laminations; peloidal dolopacktone with quasi-
169 planar laminations, and peloidal dolopacktone with megaripples beds). The cap limestone
170 is predominantly composed of limestones with megaripples, wavy laminations, quasi-
171 planar and crystal fans, and subordinate laminated shale (for more details, see Nogueira
172 et al. 2003; 2018; Sansjofre et al. 2014; Soares et al. 2020; Romero et al. 2020; Santos et
173 al. 2021,2023). Post-Marinoan cap carbonate deposition is generally constrained through
174 U-Pb zircon dating and more recent Re-Os dating to between 635.2 ± 0.5 Ma to 632.3 ± 5.9
175 Ma respectively (Condon et al., 2005; Rooney et al., 2015). Consistent with this are ages
176 of 627 ± 32 Ma for cap dolostone and 622 ± 33 Ma for the cap limestone obtained through
177 Pb-Pb dating, indicating a lower Ediacaran age (Babinski et al. 2006; Romero et al. 2013)
178 and an age of 623 ± 3 [9] Ma U–Pb in situ in seafloor crystal fans was obtained recently
179 by Carvalho et al. (2023). This latter age is interpreted as a new constraint on the transition
180 from shallow to intermediate burial diagenesis of the Puga cap carbonate (Fig. 1C). The
181 isotopic values of $\delta^{13}\text{C}$ from -5 ‰ on the hood to -2 ‰ and $^{87}\text{Sr}/^{86}\text{Sr}$ between 0.7074 and
182 0.7089 (Fig. 1C) are compatible with global curves (e.g., Otavi Group, Namibia)
183 (Nogueira et al. 2019).

184 **3. MATERIALS AND METHODS**

185 In total, 97 samples were analyzed from the cap dolostone (Mirassol d'Oeste
186 Formation) and cap limestone (Guia Formation) of the Puga cap carbonate, southern
187 Amazon craton (Fig.1). Of these 97 samples, 37 were sampled via outcrop from the
188 Terconi section in the Terconi quarry and 60 were sampled from the Tangará section with

189 41 samples being taken from drill cores and 19 samples from outcrops (Fig.3 and Fig.4).
190 The samples followed the stratigraphic proposed and facies/microfacies criterion
191 presented by Nogueira et al. 2003; Sansjofre et al. 2014 and Santos et al. 2021,2023.

192 3.1 Trace elements

193 **3.1.1 REE geochemistry - acetic acid leaching**

194 In total, 42 samples of the drillcore JOR, which captures the cap dolostone from
195 the Tangará section, were prepared for major/minor/trace/REE + Y concentrations to
196 avoid REE contributions from non-carbonatic components (e.g., clays and oxides)
197 following the digestion protocol of Rongemaille et al. (2011). Approximately 20 mg of
198 powder was weighed into acid-cleaned 2.5 mL microcentrifuge tubes, and 0.5 mL of
199 trace-metal grade acetic acid at a concentration of 0.85 N (i.e., 5%) was added and left to
200 dissolve the samples overnight at room temperature. Samples were centrifuged, and 0.25
201 mL aliquots of the supernatant were diluted in 4.75 mL PFA-distilled nitric acid prepared
202 to a concentration of 0.45 N HNO₃ (i.e., 2%) with 10 ppb indium as an internal standard
203 to control instrumental drift. Analyses were conducted on a Thermo Scientific
204 ELEMENT XR high-resolution inductively coupled plasma mass spectrometer (HR ICP-
205 MS) of the Pôle Spectrométrie Océan (PSO) at Institut Universitaire Européen de la Mer,
206 Université de Bretagne Occidentale, France. Using low, medium, and high-resolution
207 modes allowed for measuring major/minor/trace/REE + Y concentrations at a constant
208 dilution factor during a single analytical session. Repeat analyses of the standard
209 reference material CAL-S (N = 15) showed 1 σ precision of 7–9%, with results within 10%
210 of published values. Concentrations were determined by external calibration with a
211 synthetic, matrix-matched calibration solution.

212 **3.1.2 REE geochemistry - 2 N nitric acid leaching**

213

214 In addition, 56 outcrop samples were analyzed, leaching 37 from the Terconi
215 section and 19 from the Tangará section using total bulk dissolution. Rock fragments free
216 of apparent fractures or recrystallization were carefully selected and crushed in an agate
217 mortar. Chemical analysis was performed by ICP-AES (major and minor elements) and
218 ICP-MS (trace elements) at the spectrochemical laboratory (SARM) of the Centre de
219 Recherches en Pétrographie et Géochimie in Vandœuvre-les-Nancy. Samples were
220 prepared by fusion with LiBO₂ (980 °C during 60 min), followed by dissolution in an
221 HNO₃ (1 mol L⁻¹)–H₂O₂ (~0.5% v/v)–glycerol (~10% v/v) mixture to obtain a diluted
222 solution (Carignan et al. 2001). Precision and accuracy were better than 1% of the
223 measured value (mean 0.5%) for major–minor elements and 8% for trace metals, as
224 checked by international standards and analysis of replicate samples (Carignan et al.
225 2001).

226

227 **3.2 Stable isotopes ($\delta^{13}\text{C}$ and $\delta^{18}\text{O}$)**

228

229 In total, 44 samples of the drill core JOR, which captures the cap dolostone from
230 the Tangará section, were prepared for stable isotope analyses and were performed at the
231 National Museum of Natural History in Paris at the SMIMM Platform. The samples were
232 first ground into a powder in an agate mortar. CO₂ was extracted by dissolution with
233 100% H₃PO₄ at 70 °C under a primary vacuum. Stable carbon and oxygen isotopic
234 compositions of the product CO₂ were measured using a Kiel IV device. Standards CA21
235 – Marble – NBS19 were calibrated to the $\delta^{13}\text{C}$ sample/ref data. These standards have been
236 calibrated relative to V-PDB using two NBS19 and other international standards. The

237 results are presented in δ -notation relative to the international Standards V-PDB for $\delta^{13}\text{C}$
238 and $\delta^{18}\text{O}$. The external reproducibility for $\delta^{13}\text{C}$ and $\delta^{18}\text{O}$ measurements is 0.1‰ and
239 0.15‰, respectively (1σ).

240

241 **4. RATIONALE FOR THE USE OF RARE EARTH ELEMENTS + YTRIUM** 242 **(REE+Y) AS PALEOCEANOGRAPHIC INDICATORS**

243

244 Ancient nonskeletal carbonates, including microbialites, can record ambient
245 seawater rare earth elements and yttrium (REE+Y) concentrations, making them excellent
246 potential archives for paleoenvironment proxies (Schere and Suits, 1980; Shaw and
247 Wasserburg, 1985; Webb and Kamber, 2000; Kamber and Webb 2001; Nothdurft et al.
248 2004; Huang et al. 2009; Ling et al. 2013; Hodskiss et al., 2021). However, their utility
249 is predicated on the ability to control for noncarbonate contamination, as well as the
250 impacts of diagenesis (Bolhar and Van Kranendonk, 2007; Rongemaille et al., 2011; Zang
251 et al. 2015; Van Kranendonk et al. 2003; Bolhar et al. 2004; Nothdurft et al. 2004; Olivier
252 and Boyet, 2006; Bolhar et al. 2015). In this study, we primarily utilize Y/Ho ratios, Ce
253 anomalies, and Eu anomalies to explore the post-Marinoan record of the Amazon craton.

254 The Y/Ho ratio is a geochemical parameter distinguishing between continental
255 and marine influence in marine sediments (Bau et al. 1996; Nozaki et al. 1997; Wang et
256 al. 2014). Yttrium (Y) is more abundant in continental rocks, released during weathering,
257 and transported by rivers. Holmium (Ho) is associated with marine waters (Lawrence et
258 al. 2006). Consequently, the Y/Ho ratio of estuaries shows an increase from crustal values
259 of ~ 27 to typical marine high values of >40 , together with the increasing salinity towards
260 the open ocean (Lawrence and Kamber., 2006). REE + Y is relatively flat in freshwaters
261 with typical Y/Ho values ≈ 26 (Taylor and McLennan, 1985; Kamber et al., 2005;

262 Lawrence et al., 2006; Bolhar and Van Kranendonk, 2007; Babechuk et al., 2015). Values
263 between ≈ 36 and 26 typically relate to inland waters or mixed seawater-freshwater
264 systems (Webb and Kamber, 2000; Van Kranendonk et al., 2003; Bolhar et al., 2004;
265 Shields and Webb, 2004; Lawrence). Similarly, contamination by detrital components
266 typically overestimates carbonate concentrations of elements such as Al, Ti, Th, and Zr
267 and will reduce Y/Ho ratios to values of ≈ 27 , which is similar to the continental crust and
268 much lower than expected seawater Y/Ho (Webb and Kamber, 2000; Nothdurft et al.
269 2004; Uhlein et al. 2021).

270 Ce anomalies are a valuable geochemical tool to assess past redox conditions in
271 sedimentary environments. Cerium (Ce) exists in two oxidation states: Ce (III) and Ce
272 (IV). Ce (III) is preferentially incorporated into sediments in reducing conditions, leading
273 to a higher Ce anomaly, $Ce/Ce^* > 1$. Conversely, Ce (IV) is more stable and retained in
274 the sediments under oxic conditions, resulting in a lower Ce anomaly, $Ce/Ce^* < 1$. Ce
275 anomalies in sedimentary records have been widely used to reconstruct marine redox
276 conditions over time (Kamber and Webb, 2001; Bolhar et al., 2004; Bau and Alexander,
277 2006; Bolhar and van Kranendonk, 2007; Pourret et al., 2008; Planavsky et al. 2010; Ling
278 et al. 2013; Hood and Wallace, 2015; Rodler et al. 2016; Sahoo et al. 2016; Wallace et al.
279 2017; Zhang and Shield, 2022).

280 The Eu/Eu* anomaly is a useful indicator for investigating changes in redox and
281 hydrothermal-related parameters (Bau, 1991; Klinkhammer et al., 1994; Barrat et al.,
282 2000; Lawrence et al., 2006; Alexander et al., 2008; Parsapoor et al. 2009). Under
283 reducing conditions, such as hydrothermal systems influence of high-temperature fluids
284 associated with fumaroles, Eu (II) is preferentially incorporated into sediments, resulting
285 in higher Eu/Eu* anomalies. In contrast, Eu (III), under oxidizing conditions, led to lower
286 Eu/Eu* anomalies. High Eu anomalies are absent from the general characteristics of

287 marine carbonates. The Eu/Eu* ratio in sedimentary records allows us to infer past
288 variations in hydrothermal activity or the presence of Eu in detrital sediments, especially
289 within calcic plagioclase. This insight significantly enhances our understanding of its
290 influence on sediment geochemistry and the resulting shifts in redox conditions over time.
291 (Michard, 1989; Klinkhammer et al., 1994; Haas et al., 1995; Bau and Dulski, 1999;
292 Douville et al., 1999; Barrat et al., 2000; Dai et al., 2016; Liu et al., 2016 Cui et al. 2019).

293 **5. RESULTS**

294 5.1 Major and Trace Elements

295 Trace element data from the Calcário Tangará and Terconi Quarry sections (Fig.
296 2 and Fig. 3) are listed in Table 1. They are described below, following the stratigraphic
297 proposed by Nogueira et al. 2003 and Santos et al. 2021,2023. The cap dolostone
298 (Mirassol d'Oeste Formation) was subdivided from base to top into four microfacies:
299 stratiform doloboundstones with gypsum pseudomorph, domical doloboundstones with
300 tubestone, and wave-dominated dolostone facies (peloidal dolograinstone/dolomudstone
301 with quasi-planar laminations and low-angle truncations, and peloidal dolograinstone
302 with megaripples beds) (see Soares et al. 2020, Santos et al. 2021,2023 for more details).

303 The measured REE+Y data of all Puga cap carbonate samples were normalized
304 to those of post-Archean Australian shale (PAAS) by McLennan, 1989. Anomalies of La,
305 Ce, and Eu were calculated using the equations following (1) Lawrence et al. (2006) and
306 (2) Frimel 2009. Overall, REE+ Yttrium (Σ REE+Y) patterns from the Puga cap carbonate
307 are similar in both sections (Fig. 4 and Fig. 5). The Puga cap carbonate shows a
308 progressive enrichment of REEs up section, which is lower than those in the PAAS by
309 less than two orders of magnitude, in the Tangará Section Σ REE+Y: 1.58–60.41 ppm,
310 average = 5.79ppm, n = 61; Fig. 4) and the Terconi Section (Σ REE+Y: 1.4–97.6 ppm,
311 average = 11.4. ppm, n = 37; Fig. 5).

312 5.1.1. Calcário Tangará Quarry

313 5.1.1.1 Cap dolostone from Tangará Quarry

314 5.1.1.1.1 Microbialites - Tangará Quarry

315

316 In the Calcário Tangará Quarry section (Fig. 2), the microbialites
317 (doloboundstones stratiform stromatolites with gypsum pseudomorph cement and
318 doloboundstones domical stromatolites with tubestone associated) of the cap dolostone
319 showed Mg/Ca ratios ranging from 0.6 to 1.0 (average = 0.8 ± 0.18 ; n = 42), and low
320 concentration of Sr from 22. to 87 ppm (average = 50 ± 14 ppm) (Fig. 2). The stratiform
321 doloboundstones with gypsum pseudomorphs cement have low concentrations of
322 $\Sigma\text{REE}+\text{Y}$ ranging from 1.6 to 27.04 ppm (average = 4 ± 4.7 ppm; n = 32), showed a flat
323 $(\text{REE} + \text{Y})_{\text{SN}}$ with a depleted LREE pattern (average $(\text{Pr}/\text{Yb})_{\text{SN}} = 1.0 \pm 0.3$; n= 33) (Fig.
324 6A, C and E). Negative La anomalies, $\text{La}/\text{La}^* = 0.8$ to 2.2, average = 0.9 ± 0.3). Ce
325 anomalies are mostly weakly negative, with $\text{Ce}/\text{Ce}^* = 0.4$ to 1.0 (average = 0.9 ± 0.2),
326 and 3) strong positive Eu anomalies, $\text{Eu}/\text{Eu}^* = 1.0$ to 3.2 (average = 1.5 ± 0.5). These
327 facies record slightly super-chondritic Y/Ho ratios (average = 32 ± 10 ; n = 32), with some
328 sample reaching Y/Ho values up to 70 (super-chondritic) (Fig. 7). The Y/Ho ratio of
329 these facies showed an absence of correlation factors ($\text{R}^2 = 0.023$); Th ($\text{R}^2 = 0.0017$); Ni
330 ($\text{R}^2 = 0.0024$), and Pb ($\text{R}^2 = 0.063$).

331 The domical doloboundstones with tubestone (giant stromatolites) exhibit ΣREE
332 +Y concentrations between 3.0 to 34.8 ppm (average = 3.7 ± 0.97 ; n=10) and have a flat
333 $(\text{REE} + \text{Y})_{\text{SN}}$ (Fig. 4 B) with a depleted LREE pattern (average $(\text{Pr}/\text{Yb})_{\text{SN}} = 0.7 \pm 0.97$;
334 n = 10) (Fig. 6A, C, and E), occurring with 1) Weakly negative Ce anomalies, Ce/Ce^*
335 = 0.8 to 1.1, (average = 0.9 ± 0.1); 2) Positive Eu anomalies, $\text{Eu}/\text{Eu}^* = 1.0$ to 1.9, (average
336 = 1.3 ± 0.3); and 3) negative La anomalies, $\text{La}/\text{La}^* = 0.8$ to 1, (average = 0.9 ± 0.1). The

337 Y/Ho ratio in this location was found to be between 27 and 30 (average = 29 ± 1) (Fig. 7)
338 and shows an absence of correlation factor with Zr ($R^2 = 0.2$); Th ($R^2 = 0.3$); Ni ($R^2 =$
339 0.24), and Pb ($R^2 = 0.30$).

340 5.1.1.1.2 Wave-dominated dolostone facies Tangará Quarry

341

342 The wave-dominated dolostone facies (peloidal dolomudstone with parallel
343 lamination, peloidal dolopacktone with quasi-planar lamination, and peloidal
344 dolopacktone with megaripples beds) show Mg/Ca ratios ranging from 0.6 to 0.9,
345 (average = 0.6 ± 0.11 ; $n=9$) and with low Sr concentrations between 62.75 and 173 ppm
346 (average = 86.4 ± 34 ppm) are similar to those of the basal section (Fig. 2). The
347 concentrations of the detrital component towards the top of the cap dolostone, with Al =
348 12.34 to 37.88 ppm (average = 25 ± 9.7), Zr = 7 to 20 ppm (average = 13 ± 5 ppm) and Th
349 = 0.59 to 1.77 (average = 1.18 ± 0.5 ppm) (Fig. 2). The transition from cap dolostone to
350 cap limestone record the decreased purity toward the top of the Puga cap carbonate. The
351 wave-dominated dolostone facies exhibit $\Sigma\text{REE} + \text{Y}$ concentrations between 12.3 to 39.02
352 ppm (average = $26 \text{ ppm} \pm 8.3$; $n=10$), have a flat $(\text{REE} + \text{Y})_{\text{SN}}$ (Fig. 4C), with a depleted
353 LREE pattern (average $(\text{Pr}/\text{Yb})_{\text{SN}} = 1.0 \pm 0.1$) (Fig. 6A, C, and E),. and show the
354 following attributes: 1) Absence of La anomalies $\text{La}/\text{La}^* = 0.8$ to 0.9 (average = $0.84 \pm$
355 0.03). 2) weakly negative Ce anomalies $\text{Ce}/\text{Ce}^* = 0.8$ (average = 0.78 ± 0.02), 2). Eu
356 anomalies are small to absent, $\text{Eu}/\text{Eu}^* = 0.9$ to 1.0 (average = 0.96 ± 0.04). $\text{Y}/\text{Ho} = 32$ to
357 34 (average = 33.5 ± 0.6) (Fig. 2) and absence of correlation with Zr ($R^2 = 0.085$); Th
358 ($R^2 = 0.16$); Ni ($R^2 = 0.096$); Pb ($R^2 = 0.39$).

359 5.1.1.2 Cap Limestone Tangará Quarry

360 The cap limestone shows higher Sr concentrations between 549 to 1115 ppm
361 (average = 634 ± 196 ; $n = 10$). The cap limestone is also associated with an elevated

362 presence of detrital components compared to the cap dolostone, such as Al = 17.15 to
363 45.42 ppm (average = 34.3 ± 9.3), Zr = 18 to 48 ppm (average = 28 ± 10), and Th = 1.14
364 to 2.68 ppm (average = 1.86 ± 0.48) (Fig. 2). Cap limestone has (REE + Y)_{SN} patterns
365 (Fig. 4 D) characterized by a higher concentration of Σ REE + Y, ranging from 34.9 to 60.4
366 ppm (average = 47.2 ± 8.1 ; n = 10) with a depleted LREE pattern (average (Pr/Yb)_{SN} =
367 0.87 ± 0.07) (Fig. 6A, C, and E), and show the following attributes: 1) negative La
368 anomalies, La/La* = 0.8 to 0.9 (average = 0.79 ± 0.06); 2) weakly negative Ce anomalies
369 Ce/Ce* = 0.7 to 0.8, (average = 0.76 ± 0.02); 3) Eu anomalies are minor to absent,
370 Eu/Eu* = 0.9 to 1.0, (average = 1.04 ± 0.05). The Y/Ho ratio exhibits a super-chondritic
371 value when contrasted with the cap dolostone between 33 to 38 (average = 35 ± 1.6) (Fig.
372 7) and occurs with an absence of Zr ($R^2 = 0.069$); Th ($R^2 = 0.35$); Pb ($R^2 = 0.10$), and Ni
373 ($R^2 = 0.13$).

374 **5.1.2. Terconi Quarry**

375 *5.1.2.1 Cap dolostone from Terconi Quarry*

376

377 5.1.2.1.1 Microbialites Terconi

378 The microbialites (stratiform doloboundstones with gypsum cement and domic
379 doloboundstones with tubestone) of the cap dolostone in the Terconi quarry section are
380 similar to those of the Calcário Tangará Section, as follows: 1) Mg/Ca ratio ranging from
381 0.4 to 0.6, (average = 0.6 ± 0.05 ; n = 18); 2) low Sr concentrations of between 40 to 60
382 ppm (average = $49 \text{ ppm} \pm 6$; n = 17); 3) Zr = 1.5 to 12.6 ppm (average = 3.5 ± 3.0); 4) Th
383 = 0.03 to 0.75 (average = 0.19 ± 0.17); and 5) Al = 1.69 to 42.30 ppm (average = $7.9 \pm$
384 9.7) (Fig. 3). The microbialites at the basal section occurs with low Σ REE + Y
385 concentrations, ranging from 1.4 to 6.6 ppm (average = $5.80 \text{ ppm} \pm 3.4$; n = 18) and have

386 flat (REE + Y)_{SN} patterns (Fig. 5A) with a depleted LREE pattern (average (Pr/Yb)_{SN} =
387 0.82 ± 0.15 ; n=17) (Fig. 6B, D, and F), and show the following attributes: 1) weakly
388 negative Ce anomalies $Ce/Ce^* = 0.7$ to 1.1 (average = 0.93 ± 0.10), 2) Eu anomalies are
389 small to absent $Eu/Eu^* = 0.9$ to 1.2 (average = 1.08 ± 0.04), and 3) weakly negative La
390 anomalies, $La/La^* = 0.7$ to 1.0 , (average = 0.94 ± 0.09). The Y/Ho ratio exhibits a super-
391 chondritic value when contrasted with the medium values of the microbialites from
392 Tangará da Serra, ratios between 32 to 49 ppm (average = 35.2 ± 0.09) (Fig. 3). These
393 ratios showed an absence of correlation with (R² = 0.085); Th (R² = 0.16); Ni (R² = 0.096),
394 and Pb (R² = 0.39).

395 5.1.2.1.2 Wave-dominated dolostone facies Terconi

396

397 The wave-dominated dolostone facies (e.g., peloidal dolopacktone with
398 megaripples beds) has Mg/Ca = 0.5 to 0.6 (average = 0.6 ± 0.01 n= 8) and low Sr
399 concentrations similar at the base of the section, between 63 and 87 ppm (average = $59 \pm$
400 8.4 ppm). In these facies, there is an increase in detrital components compared to the basal
401 microbialites, such as Al = 10.59 to 83.65 ppm (average = 36.6 ± 24.1); Zr = 3.14 to 27.13
402 ppm (average = 12.6 ± 9.25), and Th = 0.24 to 1.54 ppm (average = 0.92 ± 0.55) (Fig. 3).

403 These facies occur with higher $\Sigma REE + Y$ concentrations than those of
404 microbialites, ranging from 8.32 to 42.26 ppm (average = $15.6 \text{ ppm} \pm 12.5$; n = 8) and
405 have flat (REE + Y)_{SN} patterns (Fig. 5B) with a depleted LREE pattern (average (Pr/Yb)
406 $_{SN} = 0.9 \pm 0.1$; n = 8) (Fig. 6B, D, and F), and show the following attributes: 1) weakly
407 negative Ce anomalies, $Ce/Ce^* = 0.8$ to 0.9 , (average = 0.87 ± 0.02), 2) Eu anomalies are
408 minor to absent, $Eu/Eu^* = 0.9$ to 1.1 (average = 0.95 ± 0.06), and 3) weakly negative La
409 anomalies, $La/La^* = 0.7$ to 1.1 , (average = 0.96 ± 0.05) and slightly super-chondritic

410 Y/Ho ratios between 32 to 34 (average = 34.2 ± 1.6) (Fig. 8B) and showed an absence of
411 a correlation between Zr ($R^2 = 0.14$), Th ($R^2 = 0.03$), Ni ($R^2 = 0.0002$) and Pb ($R^2 = 0.074$).

412 5.1.2.2 Cap Limestone Terconi Quarry

413

414 The cap limestone shows higher Sr concentrations between 549 to 1115 ppm
415 (average = 427 ± 347 ppm; $n = 12$). The cap limestone is also associated with an elevated
416 presence of detrital components compared to the cap dolostone, such as Al = 10.59 to
417 83.65 ppm (average = 71.5 ± 39 ppm), Zr = 63.13 to 194.71 ppm (average = 35.14 ± 17.4
418 ppm) and Th = 1.5 to 3.8 ppm (average = 1.8 ± 0.7 ppm) (Fig. 3). Cap limestone has (REE
419 + Y)_{SN} patterns (Fig. 5C) characterized by a significantly higher concentration of Σ REE
420 +Y, ranging from 36,4 to 97,6 ppm (average = 46.3 ± 17 ppm, $n = 12$) with a depleted
421 LREE pattern (average (Pr/Yb)_{SN} = 0.9 ± 0.05 , $n = 12$) (Fig. 6B, D, and F), and show the
422 following attributes; 1) weakly negative Ce anomalies, Ce/Ce* = 0.9 (average = $0.9 \pm$
423 0.01); 2) Eu anomalies are small to absent, Eu/Eu* = 0.9 to 1.0 (average = 0.964 ± 0.03),
424 and 3) weakly negative La anomalies La/La* = 0.8 to 1.1 (average = 0.86 ± 0.07). Y/Ho
425 ratios range from 31 to 37 (average = 33 ± 2) and show a low correlation between Zr (R^2
426 = 0.29); Th $R^2 = 0.47$, Ni ($R^2 = 0.34$) and Pb ($R^2 = 8.10^{-6}$).

427 5.2 Carbon and oxygen isotope stratigraphy in the Tangará Quarry cap dolostone

428

429 The $\delta^{13}\text{C}$ and $\delta^{18}\text{O}$ results of the Calcário Terconi Quarry section are presented in
430 Fig. 8 and detailed in Table 2. The isotopic composition is relatively homogeneous. In
431 the dolostone basal (stratiform doloboundstones with gypsum pseudomorphs), $\delta^{13}\text{C}$
432 values from -4.1 to - 8.2 (average = $-4.8\% \pm 1.1$ $n = 33$) and $\delta^{18}\text{O}$ values from -4.1‰ to -
433 7.3‰ (average = $-5.1\% \pm 0.7$ $n = 33$) occur, followed by $\delta^{13}\text{C}$ values from -4.8‰ to

434 -5.7‰ (average = $-5.0\text{‰} \pm 0.3$ n = 10) and $\delta^{18}\text{O}$ values from -3.7‰ to -4.9‰ (average
435 = $-4.7\text{‰} \pm 0.4$ n = 10) in the domoic dolostones with tubestone associated. Our data
436 complete the isotopic curve of the dolostone succession (Fig. 7) made by Sansjofre et al.
437 2011 and Nogueira et al. 2019 show an absence of covariation between $\delta^{13}\text{C}$ and $\delta^{18}\text{O}$
438 ($R^2=6.10^{-5}$) isotope compositions. The isotopic trends are similar to those obtained for
439 Puga cap carbonates from the southern Amazon craton (Terconi, Fig. 7B) and Calcário
440 Tangará Quarry (Fig. 7A) by Nogueira et al. 2003;2019; Font et al. 2006;2010; Sansjofre
441 et al. 2011, which resemble similar trends for other Ediacaran carbonate platforms.

442 **6. DISCUSSION**

443 6.1 Assessment of detrital input to Puga cap carbonate deposits

444 Previous microfacies studies have suggested that the Puga cap carbonate sequence
445 within the southern Amazon Craton occurs with low detritic content in the matrix,
446 especially the cap dolostone (Nogueira et al. 2003; 2019; Font et al. 2004; Sansjofre et al.
447 2014; Soares et al. 2020; Santos et al. 2021). To assess the impact of detrital contributions
448 on cap dolostone geochemistry (REE+Y anomalies), we evaluated the data using two
449 dissolution techniques: total bulk rock dissolution and an acetic acid leach. The cap
450 dolostone exhibited high purity in both sections, Tangará Quarry (Fig. 2) and Terconi
451 Quarry (Fig. 3); the results are presented below. Our data suggest a clear covariance
452 between the increase in insoluble particles (Zr, Al, Th abundances) vs. $\sum\text{REE+Y}$. Zr
453 concentrations are an excellent marker of siliciclastic contamination, as they are abundant
454 in different detrital materials, and contamination will drive carbonate [Zr] far above
455 typical values of ≈ 4 ppm (Frimmel, 2009). Similarly, [Th] > 1 ppm may indicate
456 significant contributions from insoluble residues and thus are considered to be another

457 marker of detrital contamination (Webb and Kamber, 2000; Zhao and Zheng, 2014,
458 2017).

459 In this study, binary diagrams were generated to determine the correlation between
460 different element ratios and element concentrations to assess the degree of diagenetic
461 alteration and contamination by siliciclastic materials on $\Sigma\text{REE}+\text{Y}$ distribution patterns
462 (Kamber and Webb, 2001; Nothdurft et al. 2004; Bolhar and Van Kranendonk, 2007;
463 Allwood et al. 2010; Wang et al. 2014; Zhao and Zheng, 2014, and Uhlein et al. 2021).
464 Despite the apparent influence on $\Sigma\text{REE}+\text{Y}$, there is a lack of anti-correlation, for
465 example, between Y/Ho ratios, which would be a predicted behavior in carbonates
466 significantly contaminated by terrigenous particles. The increase in the insoluble content
467 of the cap dolostone is observed over the transition from microbialite facies to wave-
468 dominated dolostone (Figs. 2 and 3). Conservative mixing lines between end members
469 are based on Y/Ho vs. (Zr, Th, Al, and $\Sigma\text{REE} + \text{Y}$) (Fig. 8) data using calcite cement from
470 Nothdurft et al. (2004) and PAAS from Taylor and McLennan (1985). The calcite cement
471 by Nothdurft et al. 2004 followed the criterion of purity and not its early diagenetic
472 history.

473 When assessing the impact of detrital materials on total bulk rock dissolution
474 analyses, we find that the cap dolostone is contaminated by 1%–5% detrital particles (Fig.
475 8A and 8B) (Fig. 8E and 8F). However, Y/Ho vs. Al suggests up to <1% contamination
476 (Fig. 8C and 8D), while Y/Ho vs. Th indicates up to 10% contamination (Fig. 8G and
477 8H). Our data display a minimal contribution of insoluble particles in the cap dolostone,
478 especially within the microbialites intervals, and the limestone succession of the cap
479 records the maximum detrital contribution. When analyzing the covariance of detrital
480 components between Y/Ho and Zr and Th, for example, low correlation is demonstrated

481 throughout the succession in total bulk rock dissolution. Another critical factor is that the
482 anomalies remain unaltered, as Y/Ho, Ce/Ce, and Eu/Eu show similarities throughout the
483 entire succession; we further discuss these observations in detail below. However, when
484 applying conservative mixing lines between end members based on Y/Ho vs. Zr, Th, Al,
485 and $\Sigma\text{REE} + \text{Y}$ (Fig. 8), the cap limestone displays up to 30% contamination (Fig. 8A and
486 8B) (Fig. 8C, E and F), while Y/Ho vs. Al (Fig. 8 B, C and D) is up to 1% detrital
487 contribution, and Y/Ho vs. Th indicates up to 30% contamination (Fig. 8G and 8H).

488 6.2 Diagenetic influences on the Puga cap carbonate

489

490 Petrographic observations reveal that the matrix of the cap dolostone is
491 predominantly composed of dolomicrite. It consists of replacive micritic pervasive
492 dolomite with irregular intercrystalline boundaries (non-planar) and exhibits bright red to
493 orange luminescence in cathodoluminescence. Texturally, a predominance of the
494 grumose pattern is observed, manifested in diffuse peloids clustered in no individualized
495 micro aggregates or isolated grumous. In rare instances, these peloids partly resemble
496 shrubs, like those described by Frasier and Corsetti (2003). Additionally, replacive
497 dolomite is present in all facies, forming a fine to medium planar-e to planar-s crystal
498 mosaic with sizes ranging from 0.1 to 0.2 mm. The gypsum pseudomorph filling the
499 fenestrae is well-developed. It has sizes ranging from 0.5 to 0.8 mm, occupying the
500 majority of fenestral porosities within the rock, with its occurrence limited to the base of
501 the section. The cement exhibits a fibrous habit, characterized by euhedral crystals that
502 display two distinct growth forms: 1) fascicular and 2) radial/radiaxial, typical of
503 swallowtail twinning. Subordinately, Fe-rich calcites occur mainly in fenestral,
504 interpeloidal, and vug porosity. Furthermore, post-depositional alteration of the cap
505 dolostone primarily occurred through processes such as dolomitization of gypsum,

506 neomorphism of dolomicrite, development of vugs and intercrystalline porosity, euhedral
507 dolomite precipitation, syndimentary fracturing, calcite and quartz precipitation,
508 chemical compaction, Fe-Mn-oxide substitution, and precipitation (for more details, see
509 Soares et al. 2020, and Santos et al. 2021; Santos et al. 2023). The diagenetic processes
510 that the cap limestone was subjected to are suggested to have occurred in the following
511 sequence: pyrite precipitation, calcite cementation, neomorphism of micrite and crystal
512 fan, partial crystal fan dissolution, dolomitization, chemical compaction, fracturing,
513 ferrous dolomite and calcite precipitation, pyrite (pseudomorph) and micrite replacement
514 with Fe-oxide and replacement of rhombohedral dolomite by Mn-oxides (Soares et
515 al.2020; Santos et al. 2023).

516 The paragenetic relationships of the Puga cap carbonate demonstrate consistent
517 evidence of early diagenetic processes, with lesser influence from later mesodiagenetic
518 events (Soares et al., 2020; Santos et al., 2023). The cap dolostone shows a more
519 significant impact of syndepositional cementation, which may have enhanced the
520 preservation of the cap dolostone (Santos et al. 2021., 2023). This exceptional
521 preservation is observed in the well-preserved primary texture and what is interpreted to
522 be primary $d^{13}C$ and $d^{18}O$ signals from the new and previously published data (Nogueira
523 et al. 2003, 2019; Font et al. 2006,2010; Sansjofre et al. 2011; Soares et al.2020, Santos
524 et al. 2021,2023). We discuss possible influences of late fluid-flow interaction in the Puga
525 cap carbonate using mobile trace elements, $\delta^{13}C$, and $\delta^{18}O$ isotopic compositions to
526 evaluate *syn*- or post-diagenetic fluid-rock interactions (Derry, 2010; Brand and Veizer,
527 1981; Derry et al. 1989; Banner and Hanson, 1990; Jacobsen and Kaufman, 1999;
528 McArthur, 1994; Hohl et al. 2015). The overprint of these fluids can mask the pristine
529 geochemical compositions and thus must be thoroughly monitored (Hohl et al.
530 2015;2022). All samples analyzed present negative $\delta^{18}O$ values without systematic

531 correlations with $\delta^{13}\text{C}$ values, which is suggestive of the preservation of primary signals
532 (Fig. 9). The low correlation ($R^2 = 6 \times 10^{-5}$) between $\delta^{13}\text{C}$ and $\delta^{18}\text{O}$ suggests that the
533 carbon isotope system was not significantly disturbed (Folling and Frimmel, 2002).
534 Postdepositional fluids can modify $\delta^{13}\text{C}$ and $\delta^{18}\text{O}$ values (Jacobsen and Kaufman, 1999;
535 Higgins et al., 2018; Crockford et al., 2021). Many mechanisms have been proposed to
536 explain this close covariance in altered $\delta^{13}\text{C}$ and $\delta^{18}\text{O}$, including early authigenesis in
537 shallow marine sediments (Cui et al. 2017) and post-depositional processes such as
538 interaction with meteoric water and diagenetic fluids during burial diagenesis (Taylor
539 1977; Banner and Hanson 1999; Knauth et al. 2017; Kennedy, 2009; Zao et al. 2020;
540 Derry, 2010). Therefore, the $\delta^{13}\text{C}$ and $\delta^{18}\text{O}$ signals of the occurrences of the southern
541 region of the Amazon Craton are interpreted as pristine carbonate, although we
542 acknowledge that the application of additional diagenetic tracers such as $\delta^{44}\text{Ca}$, $\delta^{26}\text{Mg}$,
543 or $\delta^7\text{Li}$ would potentially be insightful (Higgins et al., 2018; Ahm et al., 2018; 2019;
544 Crockford et al., 2021). In most samples analyzed, Mg/Ca ratios averaged 0.8 ± 0.18 ,
545 suggesting that dolomitization partially influenced the succession (Derry et al. 1994).
546 However, the cross plot of $\delta^{18}\text{O}$ vs. Mg/Ca shows a low correlation $R^2 = 0.0182$ (Fig. 9),
547 suggesting that late dolomitization processes did not change the values of $\delta^{18}\text{O}$ (Lu and
548 Meyers, 1998). This low correlation between Mg/Ca vs. $\delta^{18}\text{O}$ and $\delta^{13}\text{C}$ vs. $\delta^{18}\text{O}$ suggests
549 that isotopes may have maintained pristine signals. Likewise, the low correlation of the
550 Mg/Ca ratio with the Mn/Sr and Fe ratios ($R^2 = 0.015$) indicated that the trace elements
551 did not change during dolomitization (Chen et al. 2022).

552 Mn/Sr and Sr/Ca variations can have two causes: postdepositional alteration or
553 paleoenvironmental changes. However, our data demonstrated an anti-correlation. The
554 Mn/Sr vs. $\delta^{18}\text{O}$ of the cap dolostone showed an absence correlation ($R^2 = 0.017$), and the
555 Sr/Ca vs. $\delta^{18}\text{O}$ occurred with $R^2 = 0.43$, while Sr/Ca vs. Mg/Ca occurred with $R^2 = 0.40$.

556 Despite the low correlation, our data have a more significant relationship with
557 syndepositional dolomitization (Sr/Ca and Mg/Ca) than post-stabilization changes in
558 dolomite (Late diagenesis). Therefore, Mn/Sr and Sr/Ca ratios may result from specific
559 paleoceanographic conditions or the development of alteration processes by diagenetic
560 fluids (Asmerom et al., 1991; Derry et al., 1992; Kaufman et al., 1993; Kaufman and
561 Knoll, 1995; Melezhik et al. 2001; Folling and Frimmel, 2002; Halverson et al. 2007).
562 Thus, the high Mn/Sr ratios obtained in many cap dolostones may have been caused by
563 increased Mn concentrations in bottom waters or anoxic porewaters. (Hohl et al. 2017;
564 Chen et al. 2022). Another possible explanation for the high Mn concentration in cap
565 carbonates is that Fe-rich hydrothermal fluids in the coeval anoxic or suboxic deep ocean
566 (Yoshioka et al. 2003; Nédélec et al. 2007; Shen et al. 2005; Font et al. 2006; Hurtgen et
567 al. 2006).

568 The relationship between our petrographic and isotopic data suggests that
569 diagenetic processes may have occurred due to fluid (possibly seawater) rock interactions
570 during syndepositional diagenesis or shallow burial (Nogueira et al. 2003; Font et al.
571 2004; Gilleaudeau et al. 2018; Soares et al. 2020; Santos et al. 2021, 2023). The evidence
572 of early diagenesis demonstrates an early stabilization of the Puga cap carbonate. This
573 evidence is confirmed in the dating of aragonite crystals from the cap limestone,
574 which indicates a U-Pb age of 623 million years (Carvalho et al. 2023), suggesting that
575 post-depositional alteration occurred within the first ≈ 10 million years. In contrast,
576 besides the majority of cap carbonate having the same textural and compositional
577 framework, the diagenetic history seems to be regional and completely dependent on the
578 tectonic, thermal, and burial history of the depositional basins where they were deposited
579 (Soares et al. 2020, Ahm et al., 2019, Santos et al., 2023). However, we suggested that the
580 cap carbonate succession preserves its syndepositional signature. We show that most

581 post-depositional modifications occurred early, and there is no evidence of effective
582 overprints in the succession. This exceptional preservation qualifies the Puga cap
583 carbonate succession as an excellent candidate for tracing the paleoceanographic
584 dynamics after the Marinoan glaciation.

585 **6.2.1 Diagenetic overprint interference in REE+Y distributions**

586 Previous diagenetic evaluations of the Puga cap carbonate (Soares et al. 2020;
587 Santos et al. 2021) demonstrate a predominance of early diagenesis process, which may
588 have increased the potential preservation of cap dolostone. Diagenetic studies of REE+Y
589 distributions in ancient carbonates suggest that $\sum\text{REE+Y}$ patterns may be stable even
590 during relatively intense diagenesis (Webb and Kamber, 2000; Liu et al. 2019) because
591 $\sum\text{REE+Y}$ concentrations are low in diagenetic fluids (10^{-6} to 10^{-4} ppm) (Sholkovitz et
592 al. 1989; Banner and Hanson, 1990; McLennan et al. 2003), and the high permeability
593 and porosity of sedimentary carbonates are strongly reduced by calcite cement formation
594 during early diagenesis (Tanaka and Kawabe, 2006).

595 We analyzed Ce and Eu variations within binary diagrams. Ce and Eu were chosen
596 for investigation due to their high sensitivity to fluid mobilization processes. Ce^{3+} has the
597 potential to oxidize to Ce^{4+} , while concurrently, Eu^{3+} can undergo reduction to Eu^{2+}
598 (Brookins, 1989). Diagenetic overprinting of carbonate deposits will lead to REE
599 enrichment and a decrease in (Dy/Sm). They would show strong covariance between Ce
600 anomalies and $(\text{Dy/Sm})_{\text{SN}}$ (German and Elderfield (1990), Shields and Stille, 2001).
601 They could show a positive correlation between Eu/Eu^* anomalies and $(\text{Pr/Sm})_{\text{SN}}$
602 (Bolhar and Van Kranendonk, 2007. However, Ce/Ce^* anomalies would not covary with
603 $(\text{Dy/Sm})_{\text{SN}}$ (Fig. 11), and anticorrelation between Eu/Eu^* anomalies and $(\text{Pr/Sm})_{\text{SN}}$
604 suggesting low interference of postdepositional diagenetic fluids in the $\sum\text{REE+Y}$ of the

605 carbonates of the Puga cap carbonate. The Puga cap carbonate sequence shows strong
606 Eu/Eu* anomalies ranging up to 3.2 (average = 1.56 ± 0.31). Eu anomalies could be an
607 analytical artifact due to interference by high Ba concentrations during ICP–MS
608 measurements (Jiang et al. 2007). Our data show a low correlation with Ba ($R^2 = 0.35$).
609 Also, the Eu/Eu* anomalies strongly correlate with Gd and Sm ($R^2 = 0.9$), which has
610 nothing to do with isobaric interference with Ba. These positive Eu anomalies can be
611 explained by plagioclase dissolution. Deng et al. (2017) and Langarica et al. (2020)
612 discussed the influences of these variables on Eu anomalies and the flattening of the
613 REE+Y patterns. However, this is not the case in the present study since we demonstrated
614 in section 5.1 the purity of cap dolostone. Our data show low covariance between Eu/Eu
615 anomalies and Zr, Al, and Th. Another possible explanation for the anomalies could be
616 syn- and postdepositional hydrothermal processes because the hydrothermal fluids are a
617 potentially significant secondary flow of REE+Y to the ocean, which carry flat or heavy
618 rare earth (HREE)-depleted REE+Y signal and excessive concentrations of europium
619 (Bau and Dulski, 1996; Douville et al. 1999; Bau e Dulski, 1999; Douville et al. 1999;
620 Barrat et al. 2000; Cui et al. 2019). Furthermore, the previous discussion noted the low
621 covariance of $\delta^{13}\text{C}$ vs. $\delta^{18}\text{O}$ and the absence of a correlation of Mg/Ca vs. $\delta^{18}\text{O}$ (Fig. 9),
622 which suggests minor modification after dolomitization, which agrees with the diagenetic
623 studies by Soares et al. (2020) and Santos et al. (2023). These observations suggest that
624 REE+Y patterns are most consistent with syn or penecontemporaneous processes before
625 cap dolostone stabilization.

626 6.3 Constraints on the addition of meltwater in shallow platforms during the aftermath 627 of the Marinoan glaciation 628

629 The REE + Y patterns of the Puga cap carbonate have distinct shale-normalized
630 patterns compared to modern marine environments, which have uniform LREE depletion

631 and HREE enrichment, positive La anomalies, negative Ce anomalies, and high Y/Ho
632 ratios (Bau and Dulski, 1996; Webb and Kamber, 2000; Kamber and Webb, 2001; Van
633 Kranendonk et al. 2003; Bolhar et al. 2004; Shields and Webb, 2004; Zhang et al. 2016;
634 Tepe e Bau, 2016). The Y/Ho ratio can differentiate between successions influenced by
635 freshwater vs. seawater (Bau et al., 1996; Nozaki et al., 1997; Wang et al., 2014;
636 Hodgskiss et al., 2021). As a result, marine carbonates often exhibit elevated values, i.e.,
637 they are considered superchondritic with Y/Ho ratios > 36 (Bau et al., 1996; Zhang and
638 Nazaki, 1996; Nozaki et al., 1997). Conversely, REE + Y ratios remain relatively constant
639 in freshwater environments, typically hovering around $Y/Ho \approx 26$ (Lawrence et al., 2006;
640 Bolhar and Van Kranendonk, 2007). Intermediate ratios are commonly associated with
641 inland waters or mixed systems (Webb and Kamber, 2000; Van Kranendonk et al., 2003;
642 Bolhar et al., 2004; Shields and Webb, 2004; Lawrence and Kamber, 2006). However, it
643 has been frequently reported that marine carbonates may exhibit distinct REE + Y
644 patterns compared to seawater, primarily due to the input of freshwater into shallow
645 seawater (Frimmel, 2009; Zhao et al., 2009; Bolhar et al., 2015). Except for Ce, the typical
646 marine rare earth element features can thus be attributed to estuarine or mixing water
647 processes (Lawrence and Kamber, 2006).

648 The cap dolostone occurs with Y/Ho ratios ranging from 24.50 to 76.67. Our Y/Ho
649 values in the Tangará and Terconi sections are consistent with a mixture of waters leading
650 to an intermediate salinity, for example, in the basal microbialite (stratiform
651 doloboundstones with gypsum pseudomorphs) with an average value of 32 ± 10 ; $n = 32$.
652 However, intermittent elevated ratios (up to 70) of typical seawater are observed,
653 implying sporadic episodes of salinity increase (Lawrence and Kamber, 2006). The domic
654 doloboundstone stromatolites with tubestone occur with values between 26 and 33
655 (average = 29 ± 1), suggesting the mixture of waters (Bolhar and Van Kranendonk, 2007).
656 The wave-dominated dolostone facies (e.g., Peloid dolopacktone with quasi-planar
657 lamination) occur with Y/Ho = 32 to 34 (average = 33.5 ± 0.6) close to marine values
658 coastal. These ratios are common in glacial meltwater and seawater, characterized by
659 generally low Y/Ho values relative to terrigenous REE + Y patterns (Frimmel, 2009; Zhao
660 and Zheng, 2010). These patterns are widely recorded in cap dolostone deposits (REFS).
661 They are commonly interpreted as reflecting the dilution of the seawater signal in shallow
662 marine environments by the influx of meltwater derived from continents in the aftermath
663 of the Marinoan glaciation (~635 Ma). The plume world hypothesis proposed that during
664 the massive Marinoan deglaciation, a thick layer of floating freshwater plume was
665 produced from continental ice melting, creating physical and chemical stratification of
666 the oceans (Shields, 2005; Yang et al., 2017). Upwelling events (e.g., oceanic circulation)
667 disrupt sea stratification (Liu et al., 2014; Rame and Marotzke, 2022). Our mineralogical
668 and geochemical data support this interpretation. During these events, hypersaline deep
669 water reaches shallow water, causing increased salinity, high values of Y/Ho (up to 71),
670 and gypsum precipitation in the fenestral porosities of basal cap dolostone.

671

672 **6.3.1 Upwelling of hydrothermal and hypersaline fluids in the shallow**
673 **platform**

674 The Puga cap carbonate exhibits mostly weakly negative Eu anomalies throughout
675 both sequences; however, elevated values are documented at the base of the cap dolostone
676 (microbialites level). High Eu anomalies are absent from the general characteristics of
677 marine carbonates (Lawrence et al. 2006) and are generally regarded as strong indicators
678 of high temperature-reducing fluids (Bau, 1991; Klinkhammer et al. 1994; Barrat et al.
679 2000; Alexander et al., 2008; Parsapoor et al. 2009). REE-depleted patterns with positive
680 Eu anomalies indicate the contribution of hydrothermal fluids (Cui et al. 2019). Our data
681 are consistent with positive Eu/Eu values, which are generally considered a sign of
682 hydrothermal components in previous studies of post-Marinoan cap dolostones (Font et
683 al. 2006; Nédélec et al. 2007; Huang et al. 2011; Meyer et al. 2013; Wang et al. 2014;
684 Rodler et al. 2016; Hohl et al. 2017; Xiao et al. 2020).

685 We tested the hydrothermal influence on cap carbonates using Fe/Ti vs. Al/ (Al +
686 Fe + Mn) (Bostrom, 1973; Barrett, 1981). Deposits with a strong hydrothermal influence
687 don't typically occur with high concentrations of Al, resulting in high Al/Ti ratios
688 (Marchig et al. 1982). The contamination of these deposits by pelagic and terrigenous
689 seabed sediments enriches them in components such as Ti and Al, resulting in a drastic
690 reduction of Fe/Ti ratios and an increase in the Al/ (Al + Fe + Mn) ratio. High FeT/Al
691 ratios but deficient concentrations of extractable pyrites suggest ferruginous rather than
692 euxinic conditions (Huang et al. 2011). In addition, REE absorption by Fe oxides is
693 typically associated with submarine hydrothermal plume activity (German et al. 1991;
694 Ling et al. 2013).

695 Our data suggest a robust hydrothermal influence with respect to siliciclastic
696 contamination (Fig. 10A). The Eu/Eu anomalies are strongly correlated with Fe/Al ratios
697 (Fig. 10B). This relation of the increase in Fe accompanying the strong Eu anomalies
698 reinforces the interpretation of a hydrothermal contribution for the Puga cap carbonate,
699 similar to what was described by Jing Huang in 2011, which suggests a hydrothermal
700 origin of elevated iron and manganese for the Doushantou cap carbonate from China.
701 Furthermore, we assessed the hydrothermal impact by applying conservative mixing lines
702 (see Fig. 10 D, E, and F). This approach aligns with the model proposed by Alexander et
703 al. (2008), which considers the interaction of high T-hydrothermal fluids with seawater
704 during sediment deposition using the MREE/HREE ratios. To achieve this, we integrated
705 data from high T-hydrothermal sources (>350 °C, as per Bau and Dulski, 1999) and
706 shallow seawater components (as per Alibo and Nozaki, 1999). Conservative mixing
707 calculations between end members based on Sm/Yb vs. Eu/Sm (Fig. 10 D) indicate that
708 a high-T hydrothermal fluid contribution of less than 1% could have been responsible for
709 the anomalies recorded in the cap dolostone (Alexander et al. 2009). The Eu anomalies
710 and the strong Sm/Yb and La/Yb ratios suggest interaction with hydrothermal
711 syndepositional fluids (Fig. 10). The studied samples were plotted distant from the mixing
712 line (between 0.1 and 1% hydrothermal fluids). This scenario suggests no influence of
713 high-temperature fluids and that the depositional environment of carbonate rocks was
714 distant from fumaroles. However, positive Eu anomalies obtained from Archean and
715 Paleoproterozoic chemical sediments are interpreted as mixtures of seawater and
716 hydrothermal fluids (Kamber and Webb, 2001; Frei et al., 2008; Hodgskiss et al., 2021).
717 For the Ediacaran, positive Eu anomalies are typical for the deep anoxic ocean (Kamber
718 and Webb, 2011; Bolhar et al., 2005; Allwood et al., 2010; Rodler et al., 2016). In
719 addition, our interpretations are similar to recent interpretations for the origin of the

720 Doushantou cap carbonates, which are mainly based on $\delta^{13}\text{C}$, $\delta^{34}\text{S}_{\text{CAS}}$, $\delta^{17}\text{O}_{\text{CAS}}$, $\delta^{44}\text{Ca}$,
721 $^{87}\text{Sr}/^{86}\text{Sr}$, rare earth, elements, and metal trace elements. These data suggest that they may
722 have been formed under the influence of deglacial meltwater, hydrothermal fluids, and
723 upwelling (Huang et al. 2009; Huang et al. 2011; Huang et al. 2013; Wang et al. 2014;
724 Hohl et al. 2017; Zhao et al. 2018; Wei et al. 2019a; Peng et al., 2022).

725 Carbonates associated with deep ocean brine are characterized by Fe and trace
726 metal enrichments with positive Eu anomalies (Meyer et al. 2012). Liu et al. (2013)
727 extensively discussed the potential origin of radiogenic $^{87}\text{Sr}/^{86}\text{Sr}$ in Marinoan cap
728 carbonates of the Nuccaleena Formation, South Australia. They found that the interaction
729 of the carbonates with brines or hydrothermal fluids could better explain these values, or
730 they may reflect values from glacial meltwater mixing with seawater. This evidence
731 strengthens previous interpretations that the Marinoan cap carbonate precipitated in a
732 mixture of coastal meltwater with saline Mg-rich deeper waters through the upwelling
733 process (Grotzinger and Knoll, 1995; Ridgwell et al., 2003; Shields, 2006; Yang et al.,
734 2017; Liu et al. 2014;2018; Hoffman et al. 2017; Santos et al. 2021; Rame and Marotzke,
735 2022). Through modeling, a glacial hydrothermal flow during the Marinoan snowball
736 glaciation was demonstrated using a binary mixture of Ge/Si and Eu/Eu* values (Cui et
737 al., 2016). Marinoan deposits were accumulated during the drifting stage of post-rift
738 subsidence (Hoffman et al. 2017), and this hydrothermal flow during Snowball Earth
739 events could potentially enrich the oceans in certain elements in the aftermath of the
740 Marinoan glaciation.

741 The syndepositional hydrothermal event was unlikely, and these observations
742 align with our interpretations of rapid precipitation of the Marinoan cap carbonate
743 (Hoffman et al. 1998; Nogueira et al. 2003; 2019; Cui et al.2016 Zhou et al. 2019; Santos
744 et al. 2021). Our data suggests some samples with typical seawater values (Y/Ho up to

745 70) others displaying strong indications of hydrothermal fluid interaction (Eu/Eu* up to
746 3.2). Nevertheless, most of our samples exhibit Y/Ho ratios typical of mixwater,
747 approximately ~ 33 (a combination of glacial meltwater and seawater). This finding aligns
748 consistently with interpretations of upwelling events during the precipitation of the cap
749 dolostone. These deep Cryogenian waters would have interacted with high-temperature
750 hydrothermal fluids responsible for the elemental enrichments (e.g., Eu, Fe, Mn). At this
751 time, deep waters would have interacted with high-temperature hydrothermal fluids
752 responsible for the elemental enrichments (e.g., Eu, Fe, Mn). This evidence is consistent
753 with the diachronous precipitation model for the Marinoan cap dolostone from the melt-
754 water layer during syn-deglacial transgression (Hoffman et al. 2017).

755 This upwelling interaction may have also lowered the Y/Ho values of the
756 dolostone samples in the basal succession, as hydrothermal fluids tend to possess low
757 Y/Ho ratios. (Allwood et al. 2010). In addition, as discussed in section 5.1, we
758 demonstrated the absence of interference of siliciclastic components at lower Y/Ho
759 values. The base of the cap dolostone shows a fluctuation in the Y/Ho values and high
760 values of Eu anomalies, suggesting a strong contribution from hydrothermal sources of
761 REE+Y to the cap dolostone of the southeast margin of the Amazon Craton.

762 6.4 Paleoredox indications

763 Ce/Ce* anomalies in the Puga cap carbonate indicate oxygen deficiency
764 throughout the section (Fig. 11), but these anomalies are weaker in the microbialite
765 interval. In modern seawater, REE+Y (Rare Earth Elements + Yttrium) can record the
766 evolution of seawater redox, and the negative Ce anomaly marks oxidized conditions
767 (Byrne and Sholkovitz, 1996; Webb and Kamber, 2000; Ling et al., 2013), while the
768 absence of anomalies is indicative of anoxic conditions (Lawrence et al., 2006).
769 Furthermore, soluble Ce³⁺ is easily oxidized to Ce⁴⁺, which is relatively insoluble in oxic

770 seawater (McLennan, 1989; Alibo and Nozaki, 1999; Pattan et al., 2005; Frimmel, 2009).
771 Therefore, the Ce anomaly in marine sediments distinguishes oxic from anoxic
772 depositional environments throughout the sedimentary record (Bau et al., 1996; de Baar
773 et al., 2018).

774 However, the scenario presented thus far demonstrates the influence of water
775 mixing, particularly a strong influence of meltwater. The Y/Ho ratio exhibited similar
776 behavior to estuarine dynamics (freshwater and saltwater). Except for Ce, typical marine
777 rare earth element features can be attributed to estuarine or mixing water processes since
778 Ce anomalies only form in salinities greater than 2‰ (the ocean average is 3.5‰)
779 (Lawrence and Kamber, 2006). Therefore, only samples with a Y/Ho ratio ≥ 36 will be
780 considered for interpretation. Three samples fall into the oxygenated category, but most
781 of these eligible samples exhibit values suggesting dysoxic conditions. Our results have
782 been extensively tested for post-depositional changes; at least, the samples with high
783 Y/Ho ratios may reflect the redox conditions of our post-Marinoan scenario.

784 The interpretation of a dysoxic to anoxic water environment in the Puga cap
785 carbonate is consistent with other post-Marinoan deposits (Huang et al., 2009; Huang et
786 al., 2011; Wang et al., 2014; Lang et al., 2016; Rodler et al., 2016; Hohl et al., 2017;
787 Caxito et al., 2018; Wu et al., 2019; Hu et al., 2023; Ren and Li, 2023). In this sense, it
788 could reflect the biochemical conditions of this anomalous environment, as Ce anomalies
789 are known to be affected by environmental parameters such as pH, alkalinity, and
790 microbial activity (De Baar et al., 1985a; Alemão and Elderfield, 1990; Byrne and Kim,
791 1993; Pourret et al., 2008; Zhao et al., 2022). The Ce/Ce* values of the studied samples
792 (0.70–1.05, mostly below 0.95) could indicate a small amount of oxygen deficiency in
793 the water column (Fig. 11). Dysoxic waters would have likely provided even more
794 stressful conditions for microbial life after the Marinoan glaciation. Generally, oxidizing

795 conditions prevailed in the atmosphere and shallow seawater during the post-Marinoan
796 phase (Fike et al., 2004; Sahoo et al., 2012; Sansjofre et al., 2014; Hohl et al., 2015; 2017;
797 Wy et al., 20xx; Ader et al., 2014). Sansjofre et al. (2014), based on trace metals,
798 suggested that a water column must have been essentially oxic during the cap dolostone
799 precipitation, with anoxia only sporadically reaching the sediment–water column
800 interface in the deepest parts of the platform on the Amazon craton. The interpretation of
801 these authors differs in part from our findings, which indicate predominantly dysoxic
802 water. However, other proxies must be applied to understand microbial communities' role
803 and evaluate the reservoir effect due to water mixing.

804 6.5 CAP CARBONATE PRECIPITATION MODEL AND GLOBAL CORRELATION

805 The cap carbonate immediately precipitated over the Puga diamictite in a coastal-
806 marine environment influenced by microbial mat proliferation during the transition from
807 icehouse to greenhouse conditions (Nogueira et al. 2003; 2019; 2019; Font et al.
808 2006, 2010; Nédélec et al. 2007; Sansjofre et al. 2014; Santos et al. 2021, 2023). The
809 deformed dolomite-diamictite contact indicates rapid accumulation of dolomicrite
810 deposited over a still plastic diamicton linked to a regional scale event compatible with
811 the glacial isostatic adjustments on the coastal zone, forming a substrate with irregular
812 morphology (Nogueira et al. 2003; 2019). Rapid precipitation ($\approx 10^3 - 10^4$ years) for cap
813 dolostones is consistent with our sedimentological evidence and other models, such as
814 enhanced continental weathering alkalinity addition, gas hydrate destabilization, and
815 persistent freshwater surface layer (Shields, 2005; Hoffman et al., 2017; Crockford et al.
816 2021). Complete glacier retreat and ocean warming were succeeded by isostatic
817 adjustments, characterized by an uplift of continental areas and eustatic subsidence on the
818 coastal zone, concomitant with relative post-glacial sea-level rise (Creveling and
819 Mitrovica, 2014) and during the advance of the syn-deglacial transgression, meltwater

820 production and surface warming created stable density stratification (Liu et al. 2014;2018;
821 Yang et al. 2017; Hoffman et al. 2017). The diamicton substrate was colonized by
822 microbial communities in stratified and shallow water, and the high organic productivity
823 due to the maximum nutrient input from mixed water culminated in the giant biostromas
824 built on the coast (Santos et al. 2021). The microbial communities that colonized the
825 diamicton recorded not only the influences of the chemistry of the environment but also
826 the effects of the hydrodynamics of the environment. According to Yang et al. (2017),
827 water stagnation and high density led to the development of lamination (Santos et al.
828 2021).

829 The negative $\delta^{13}\text{C}$ isotope signatures can be considered a consequence of
830 microbial activity, producing organically derivatived ^{13}C -depleted bicarbonate (Nedeléc
831 et al. 2007). These interpretations are consistent with our $\delta^{13}\text{C}$ data ranging from -8.2%
832 to -4.1% (average = -4.9%); as discussed earlier, our tests suggested values that express
833 Early Ediacaran seawater chemistry, as well as previous data for Puga cap carbonate
834 (Nogueira et al.2003;2019; Fonte et al. 2006; Sansjofre et al. 2011) similar to Marinoan
835 cap dolostone successions worldwide (Nedeléc et al 2007; Hoffman et al., 2021; Peng et
836 al., 2022). Thus, the negative carbon isotope values may reflect the high productivity after
837 the Marinoan glaciation in a dysoxic water column, with intense microbial activity on the
838 paleocoast due to rising temperatures and significant inputs of nutrients provided by
839 reactivation of ocean biogeochemical cycling and an increased hydrological cycle
840 (Kunzmann 2012; Sansjofre et al. 2014; He et al. 2021). These conditions favored the
841 proliferation and preservation of a giant biostroma laterally continuous for kilometers in
842 the paleocoast (Nogueira et al. 2003;2022; Font et al. 2006; Romero et al. 2020; Santos
843 et al.2021,2023).

844 During deglaciation, continental ice sheets trapped detrital materials (Hoffman et
845 al. 2017; Yu et al. 2020). Estimating how long the sediments were trapped in the ice sheets
846 in the distal zone before reaching the coastal area is impossible. However, rapid
847 precipitation for cap dolostones is consistent with our sedimentological evidence. Our
848 data show intense participation with microbial activity (giant stromatolites), an abrupt
849 increase in alkalinity (cap dolostone-cap limestone transitions), plume world (Mixwater
850 and gypsum cement), and siliciclastic starvation (pure carbonates). Rapid precipitation is
851 consistent with other models, such as continental weathering alkalinity, gas hydrate
852 destabilization, and plume world, as well as sedimentary features and geochemical
853 models ($\sim 10^3 - 10^4$ years) (Shields, 2005; Crockford et al., 2016; Hoffman et al., 2017;
854 Yang et al., 2017; Crockford et al. 2021). However, a single model does not explain the
855 precipitation of the Puga cap carbonate despite the similarity between different sequences
856 of this age. A reason for this heterogeneity is that there is an endemic distinct preservation
857 between locations, primarily related to the paleogeographical position that dictates the
858 coastal/marine dynamics.

859 We have previously demonstrated the purity of cap dolostone, between 1 and 5%
860 contamination, based on Y/Ho vs. Zr and Al (see item 5.1). These conditions of high
861 density, up to 3 times higher than current rates, according to the model by Yang et al.
862 2017, may have controlled the sediment inflow to the shallow carbonate platform,
863 indicated by the low content of detrital components in the cap dolostone (figs). The
864 density between meltwater and seawater or even sediment starvation can explain the low
865 content of detrital constituents. The starvation of sediments indicates precipitation of the
866 cap dolostone over $\sim 10^5 - 10^6$ years (Spence et al. 2016; Wallace et al. 2019). A larger
867 landward shoreline migration in the siliciclastic starvation model in cap deposition is not
868 directly associated with final Snowball Earth conditions (Nordsvan et al. 2019). During

869 deglaciation, detrital materials were trapped in continental ice sheets (Hoffman et al.
870 2017; Yu et al. 2020). In modern shelves, the area starved of sediment is associated with
871 detritic trapping that can occur at the terrestrial limit with the saline wedge, producing
872 maximum estuarine turbidity (Swift, 1968; Kidwell, 1991; Geyer et al. 2004; McHugh et
873 al., 2010). The generation of hyperpycnal flows to transport fine sediments distributed in
874 surficial plumes that only occur in very shallow waters can extend for thousands of
875 kilometers (Geyer et al. 2004; Dowdeswell et al. 2015. The model by Yang et al. (2017)
876 postulates a salinity three times higher than the current level. In this sense, the detrital
877 content in meltwaters from the Marinoan deglaciation could be trapped during the linkage
878 of freshwater and hypersaline seawater, forming extensive plumes with thousands of
879 kilometers.

880 However, our geochemical data does not agree with a higher salinity during fast
881 cap dolostone precipitation in mixwaters as previously demonstrated (Y/Ho ratios of the
882 mixwater patterns). Another possible explanation for siliciclastic starvation could have
883 arisen due to the restricted conditions with persistently shallow waters in the coastal zone.
884 This depositional scenario was strongly influenced by glacial isostatic adjustment (GIA)
885 or rebound but affected only the initial meters of cap dolostone (e.g., Nogueira et al.
886 2003). This process led to continuous uplift in conjunction with a rapid post-glacial sea-
887 level rise, maintaining a shallow shelf isolated from direct oceanic processes. Barrier-
888 inland facies were not observed in the studied succession, indicating that irregular
889 diamicton morphologies may have created protected areas against waves, promoting
890 stratiform microbial proliferation and inducing dolomite precipitation. This scenario
891 resulted in a consistently shallow water column, following a significant landward
892 shoreline migration depositional pattern with a high potential for preserving the same

893 signature, devoid of detritic components, found for hundreds of kilometers along the
894 southern margin of the Amazon Craton.

895 The conditions outlined above likely lasted for approximately ~ 8 ka (Liu et al.
896 2014) or would have occurred on a scale of 10^4 and 10^5 years following a unidirectional
897 mixing model associated with thermal expansion (Yang et al. 2017). Another hypothesis
898 suggests only a few thousand years for the mixing waters based on a three-dimensional
899 model considering oceanic circulation as a crucial factor for the end of stratification by
900 meltwater (Rame and Marotzke, 2022). Although the glacial-isostatic adjustment (GIA)
901 could be the driving force influencing the mixing of the waters and paleoceanography,
902 the time spent developing this event depends on the ice cap thickness and melting
903 velocity. The lack of terrigenous components in cap dolostone may be evidence of rapid
904 precipitation (keep-up) before the main detritic flow on the coast of the Amazon craton
905 or lasted at least as long as the stratified oceans.

906 6.6 Implications for the Carbonatic Factory Change

907

908 Water turbidity increases partially inhibit carbonate precipitation, causing marl
909 deposition and suppressing microbial life, a prelude to marine stratification declines and
910 thermal expansion (Romero et al. 2020; Santos et al. 2021; Nogueira et al. 2022). The
911 mixing waters and massive detritic flow generated a geochemical change, causing CaCO_3
912 over-saturation, which contributed to the precipitation of poloidal limestones with calcite
913 crystal fans (pseudomorphs after aragonite) (Higgins and Schag, 2003; Nogueira et al.
914 2003, 2007, 2019; Hoffman et al. 2017,2021; Soares et al. 2020; Santos et al. 2021,2023).
915 The cap limestone presents an abrupt increase in the Sr concentration that accompanies a
916 rise in Zr, Al, and Th in covariance with REE+Y), indicating a direct relationship between

917 the siliciclastic input and the increases in alkalinity and, as a consequence, a change in
918 the dolomitic-aragonitic features.

919 Our geochemical and microfacies data support an abrupt transition in the carbonate
920 factory, as enhanced continental weathering could have lowered seawater Mg/Ca ratios
921 with substantial input of Ca²⁺ to platform environments (Hoffman et al.1998, xx),
922 triggering CaCO₃ oversaturation (Nogueira et al. 2019). Marine chemistry changes are
923 commonly associated with secular modifications such as the Mg/Ca ratio, pH, redox state
924 of the ocean, climate (icehouse-greenhouse conditions), pCO₂, and the extent of
925 continental weathering (Hardie, 1996, 2003; Hood and Wallace, 2018; Ries, 2010;
926 Sandberg, 1983; Stanley and Hardie, 1998; Wood et al., 2017; Wey et al., 2022). The cap
927 dolostone-cap limestone transition saw an increase in the concentration of Sr in
928 covariance with Zr and Th (Fix and Fix). The cap dolostone in this study displayed a low
929 Sr content, ranging from 30 to 70 ppm, in contrast to the cap limestones with Sr 300-1400
930 ppm. The interpretation of primary dolomite precipitation by Nogueira et al.2003; 2019,
931 Font et al. 2006; Romero et al. 2020; Soares et al. 2020, and Santos et al. 2021, 2023 is
932 consistent with our data because ancient dolomite commonly has significantly lower Sr
933 concentrations (20–70 ppm) (Baker and Burns, 1985; Vahrenkamp and Swart, 1990).
934 This abrupt transition in Sr concentrations follows a decrease in the Mg/Ca ratio (0.9 to
935 0.0) (see Fig. x). According to Ahm et al. (2019), dolomitization related to an aragonitic
936 primary fabric is consistent with fluid flow and seawater upwelling during global
937 deglaciation. Our data agree with an upwelling event and a syn-deglacial dolomitization
938 event, as previously discussed:

- 939 1. Mixing of meltwater and seawater (Y/Ho = 27 to 71)
- 940 2. Shallow seawater interacting with deep water (high Eu/Eu anomalies, suggesting
941 the upwelling of syndepositional T-high hydrothermal fluids)

942 3. Low correlation between Mg/Ca and $\delta^{13}\text{C}$ and $\delta^{18}\text{O}$, suggesting that
943 dolomitization occurred at least early diagenesis.

944 4. The excellent preservation of gypsum pseudomorphs indicates early
945 dolomitization (see Santos et al. 2021, 2023).

946 Despite our evidence agreeing with upwelling events, our sedimentological data
947 disagree with an abrupt increase in water depth as a controller for the last dolomitization
948 event. Although limestone beds are associated with a moderately deep platform, aragonite
949 crystal fans are truncated, indicating wave action in shallow waters (Soares et al. 2020),
950 and these precipitated on the seafloor were possibly dispersed as micrite by oscillation
951 (Viera et al. 2015). The hydrodynamic conditions were similar during upper cap
952 dolostone and cap limestone precipitation in a shallow platform influenced by wave
953 actions. Despite this evidence, the product of the interaction with the mats still needs to
954 be clarified if we initially have a precipitated proto-dolomite according to the organogenic
955 model (Vasconcelos et al. 1994, Dupraz 2009; Qiu et al. 2017; Chang et al. 20) or if it was
956 initially aragonite as suggested by Ahm et al. 2019. The concentration of Mg remained
957 constant during the restriction of paleoenvironmental conditions on dolomitic platforms.
958 The rapid sea-level rise facilitated more excellent dispersion of Mg ions, leading to the
959 definitive interruption of dolomicrite precipitation, abruptly replaced by CaCO_3
960 supersaturated seas with massive aragonite precipitation influenced by a long-lasting
961 transgression, still occurring during the Lower Ediacaran.

962 **7. CONCLUSION**

963

964 The REE+Y patterns and $\delta^{13}\text{C}$ values observed in the Puga cap carbonate provide
965 valuable insights into post-snowball Earth paleoceanographic conditions. The low Y/Ho
966 ratios in the cap dolostones suggest dilution of seawater composition due to meltwater

967 influx from continents (Y/Ho ~29). Conversely, superchondritic Y/Ho ratios up to 71 in
968 the basal cap dolostones suggest upwelling of hypersaline seawater in coastal areas. In
969 addition, the flattened REE+Y pattern with a positive Eu anomaly (up to 3.2) may suggest
970 meltwater with influences from a region with volcanic rocks (e.g., Amazon Craton).
971 REE+Y mass fractions in such settings may not accurately reflect global ocean water
972 composition; instead, they primarily fractionate in response to local factors related to
973 alkalinity and freshwater mixing. These findings imply that ocean destratification is
974 associated with ocean circulation and glacial isostatic adjustment, with reduced
975 siliciclastic inflow supporting a siliciclastic starvation model or rapid precipitation. The
976 recurrence of shallow-water conditions, attributed to early evidence of relative sealevel
977 decline, is linked to local ice sheet recession (ice gravity). This led to the continuous uplift
978 of the coastal zone, the creation of an isolated shelf protected by ocean dynamics, and
979 deformation in diamictite, resulting in irregular substrate relief morphologies. This phase
980 witnessed the flourishing of extensive microbial communities and the rapid recovery of
981 primary productivity, which induced dolomite precipitation in the coastal setting. Mixing
982 water masses and massive detrital input, as indicated by the increase of insoluble elements
983 (e.g., REE, Zr, and Th), generated a geochemical change caused by CaCO₃-
984 supersaturated seas influenced by significant landward shoreline migration during the
985 longterm transgression. These observations comprehensively explain the complex
986 interplay between post-glaciation and paleoceanographic dynamics during the
987 Cryogenian-Ediacaran boundary.

988

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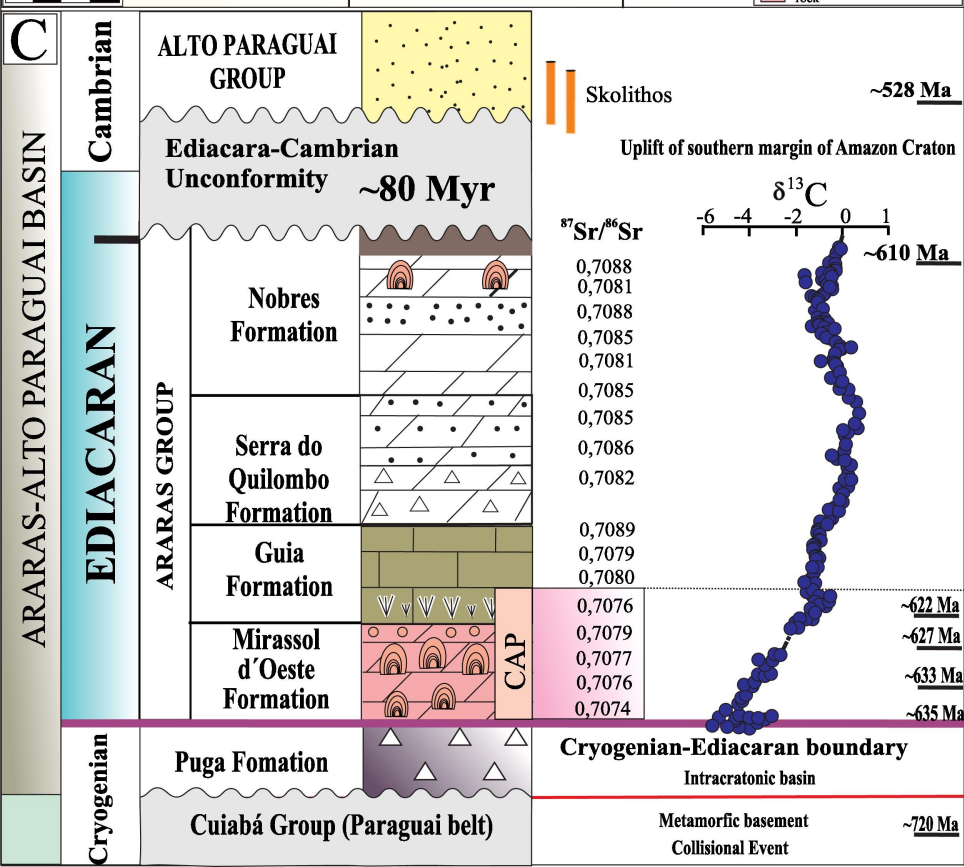
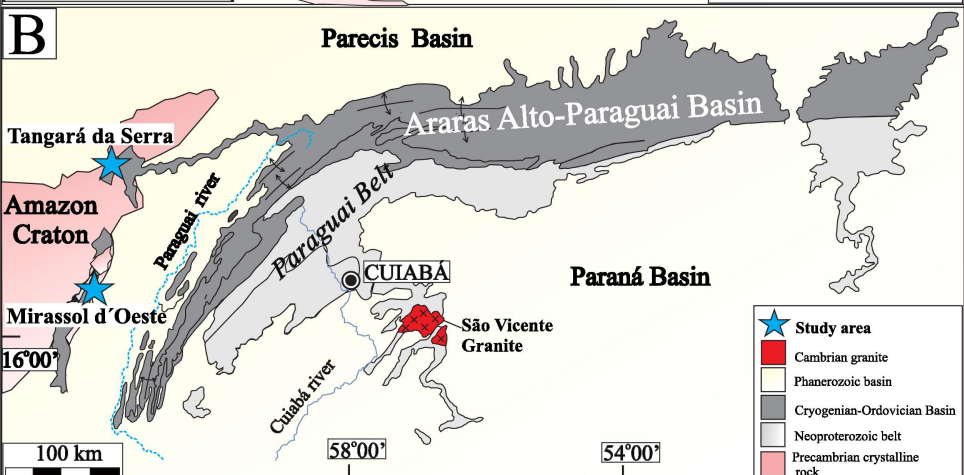
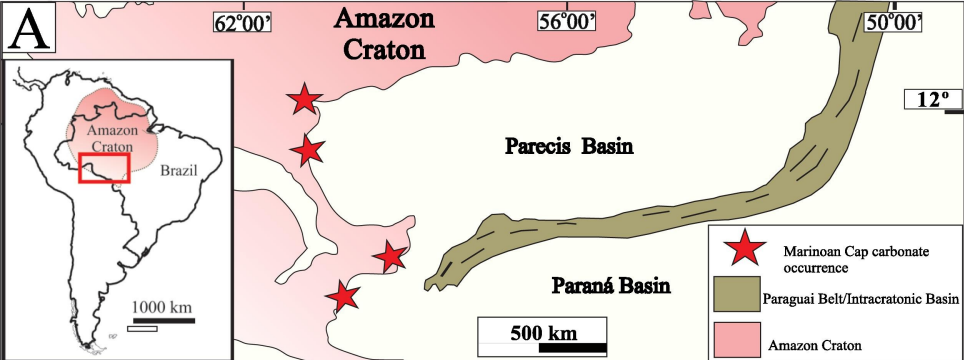
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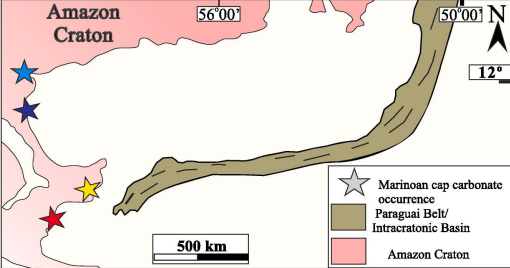
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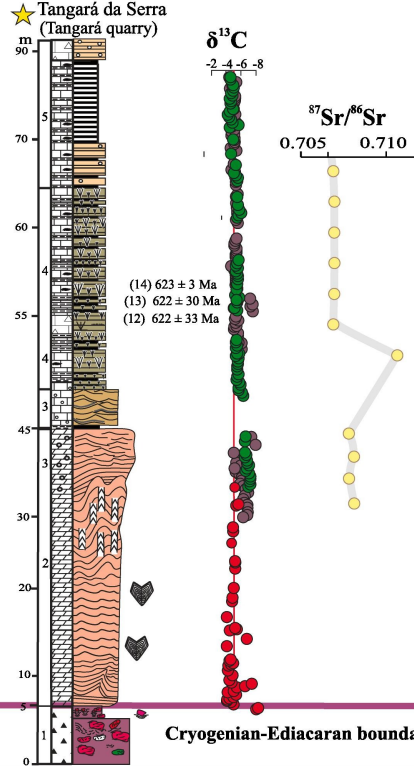
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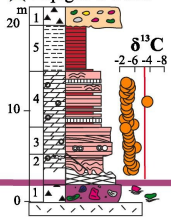
Southern Amazon Craton

Puga cap carbonate

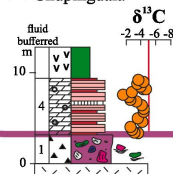


- FACIES ASSOCIATIONS**
- 1- Glacial marine
 - 2- Shallow platform with microbial activity
 - 3- Wave- and storm-dominated shallow platform
 - 4- CaCO₃-Over-saturated moderately deep platform
 - 5- Anoxic deep platform

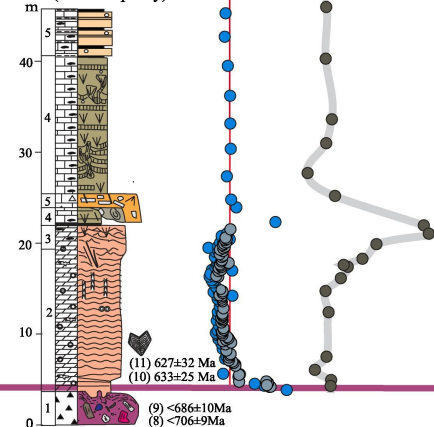
Espigão d'Oeste



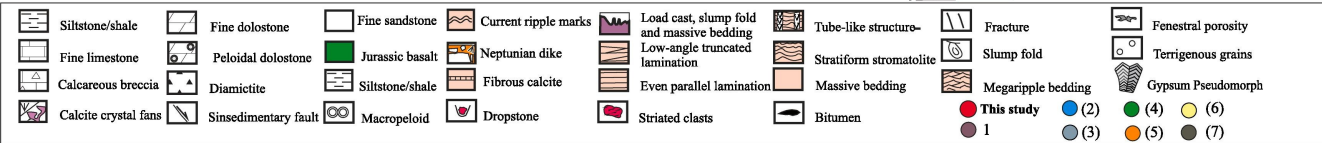
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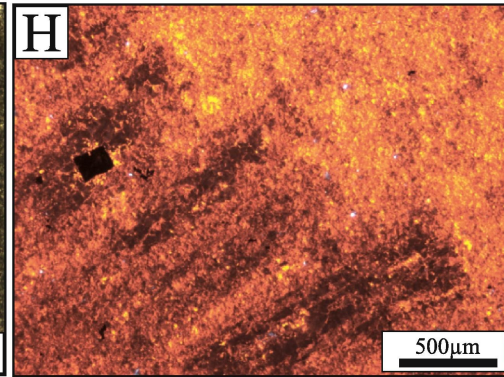
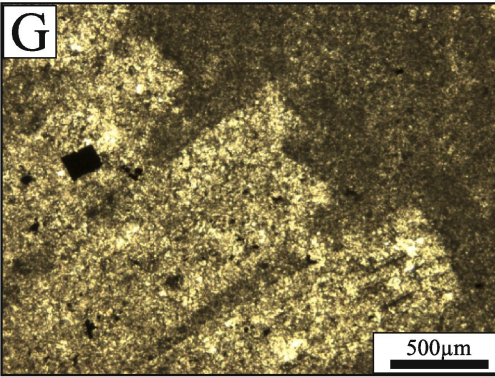
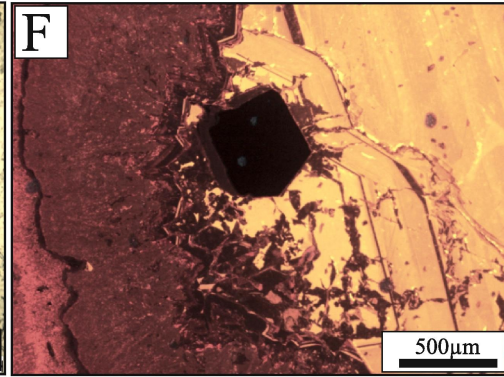
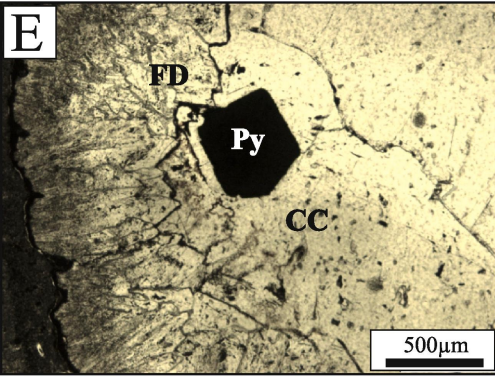
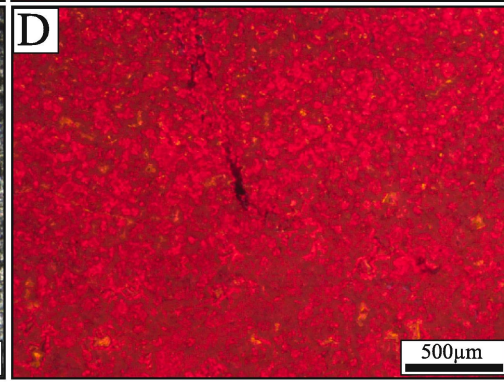
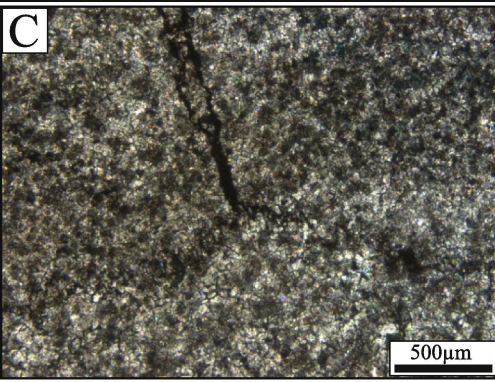
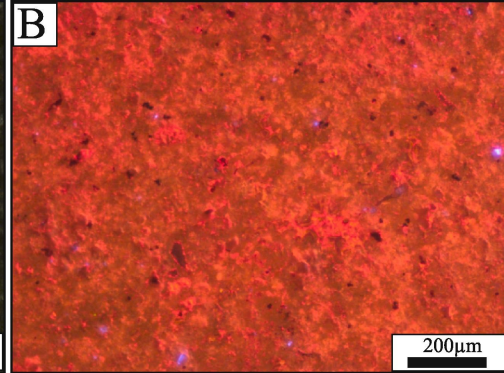
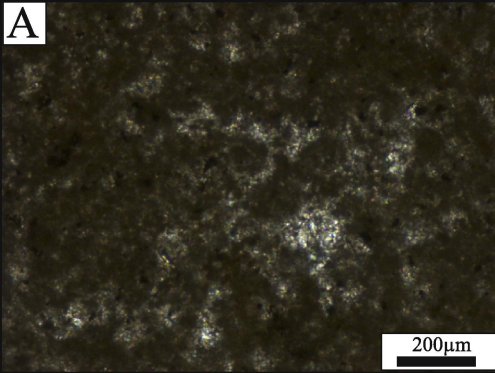


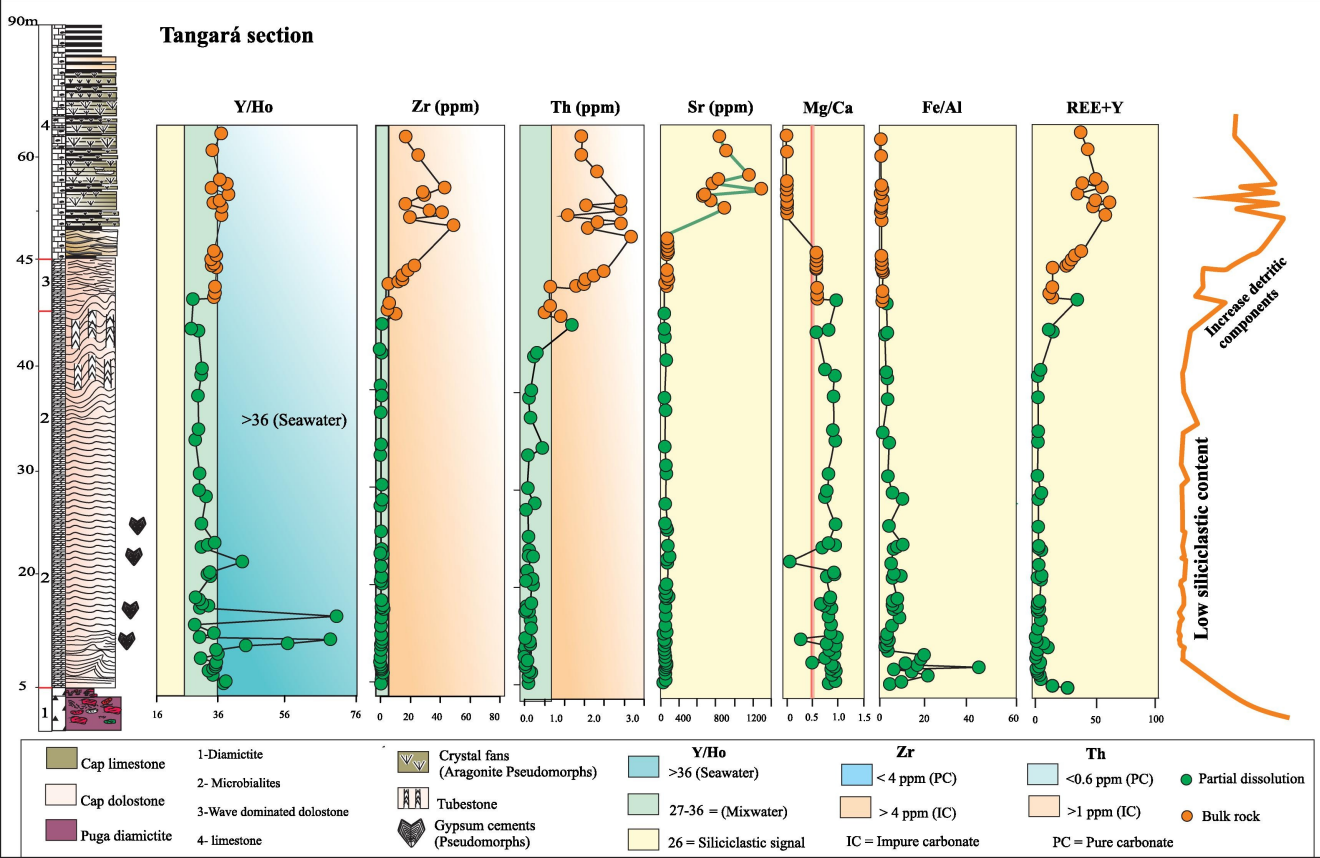
Mirassol d'Oeste (Terconi quarry)

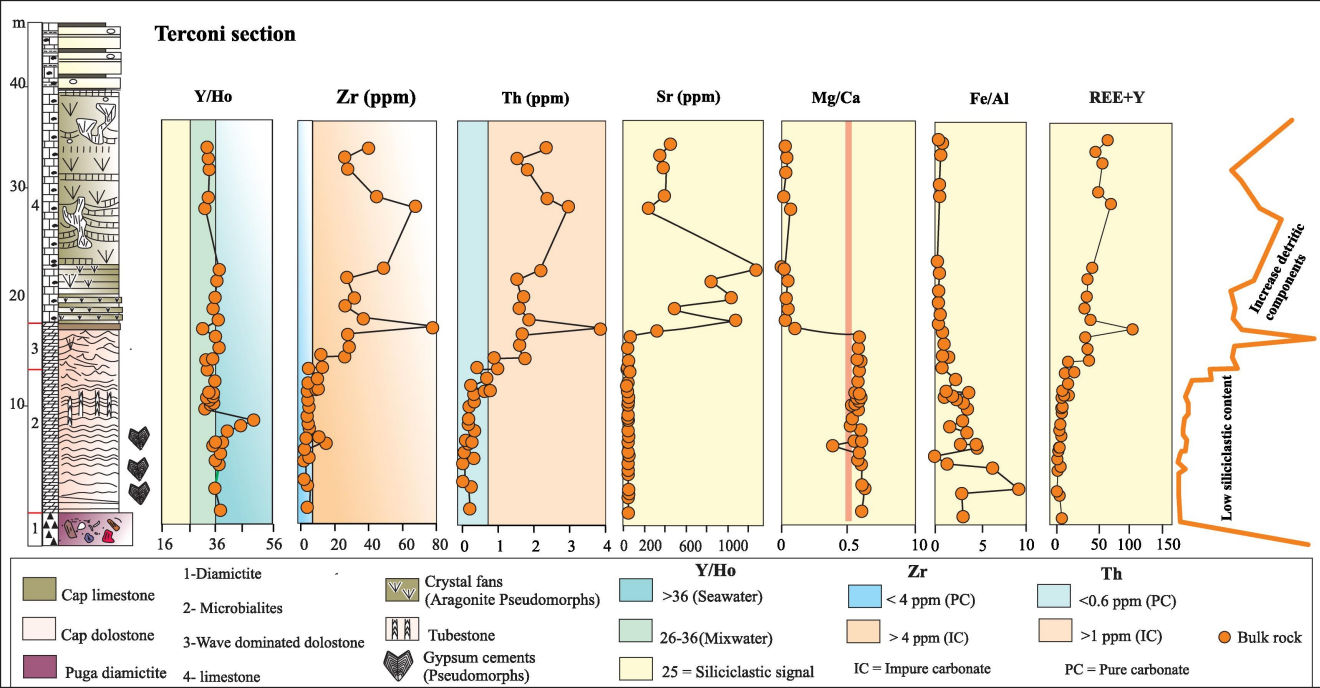


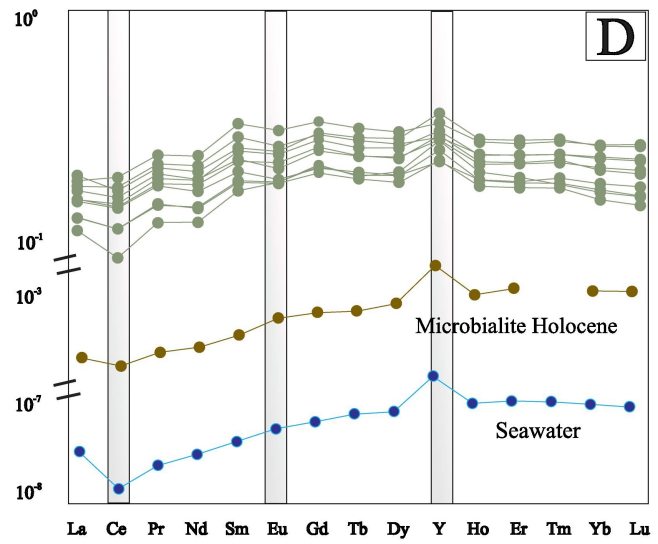
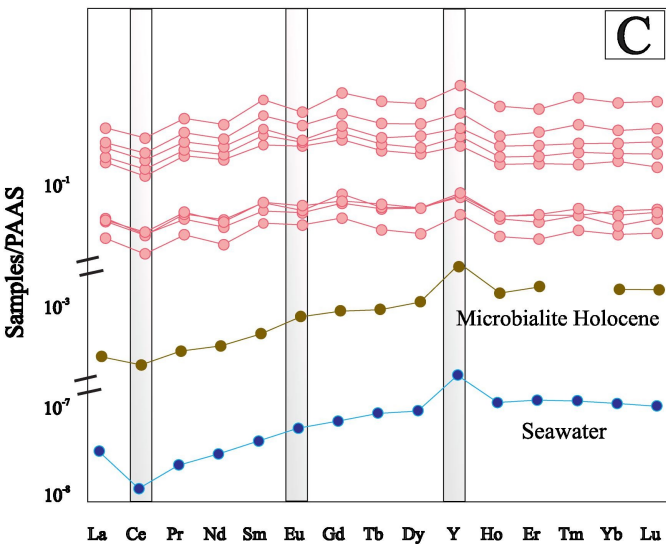
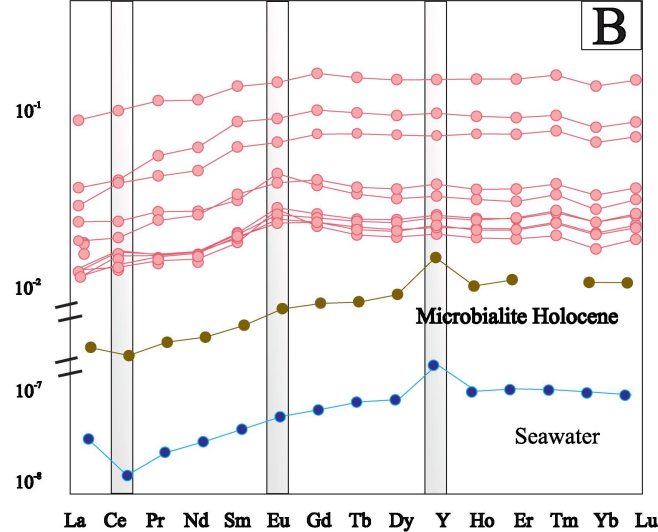
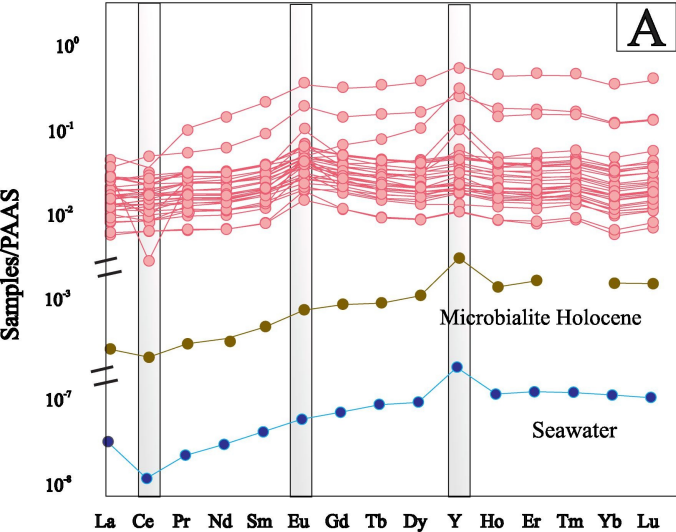
Cryogenian-Ediacaran boundary

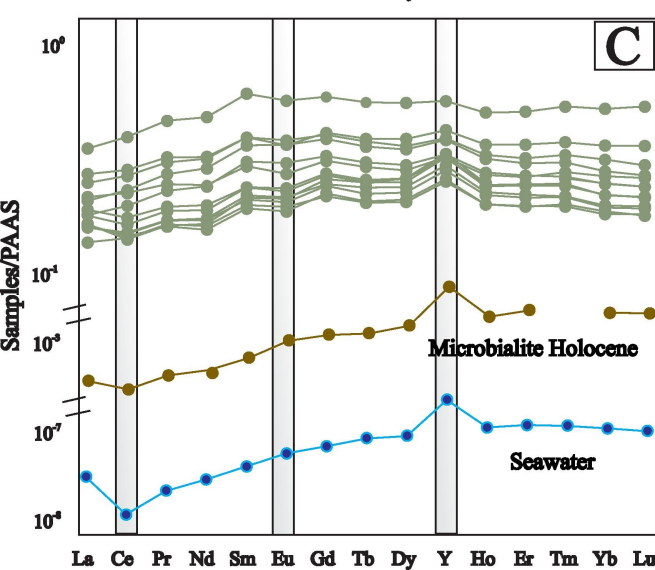
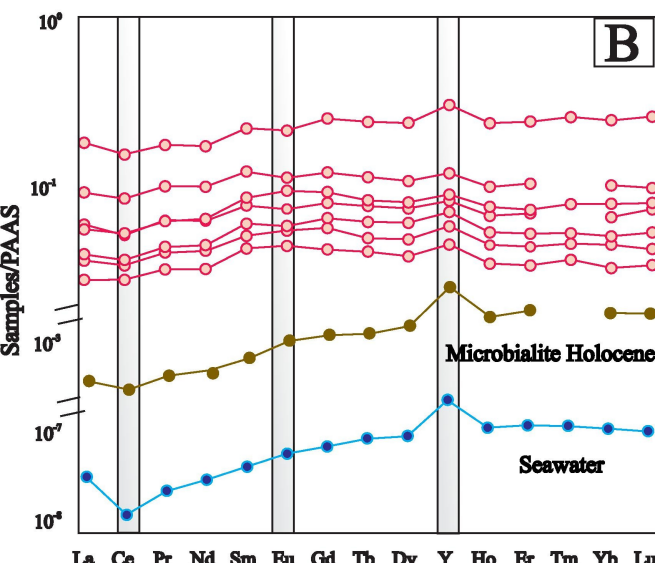
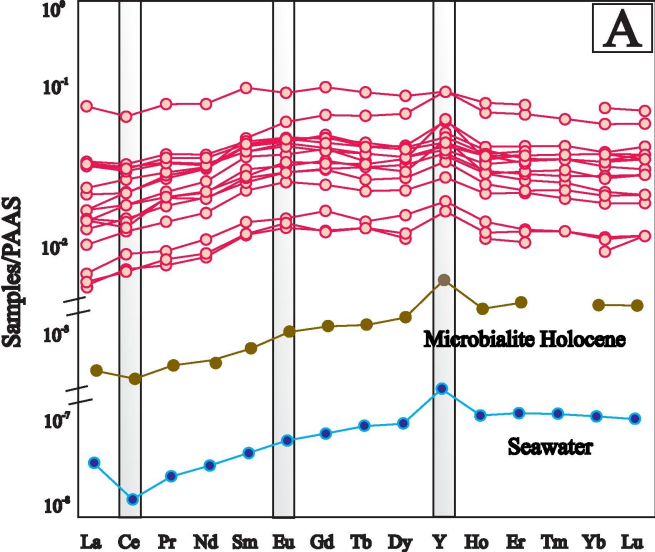


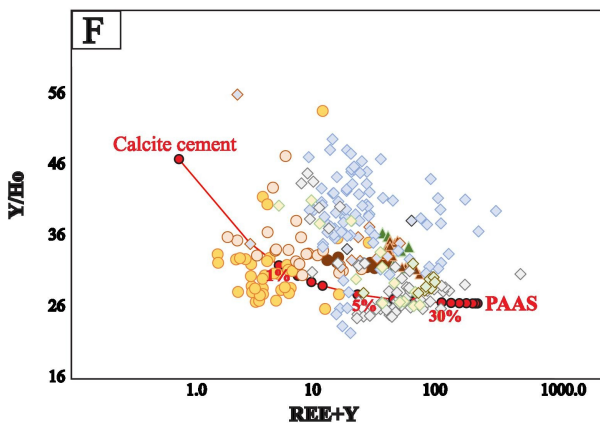
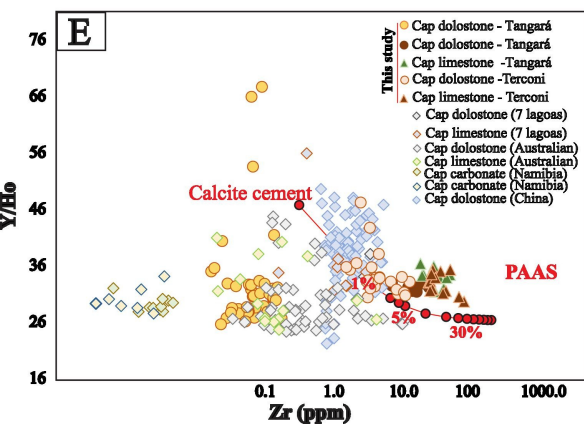
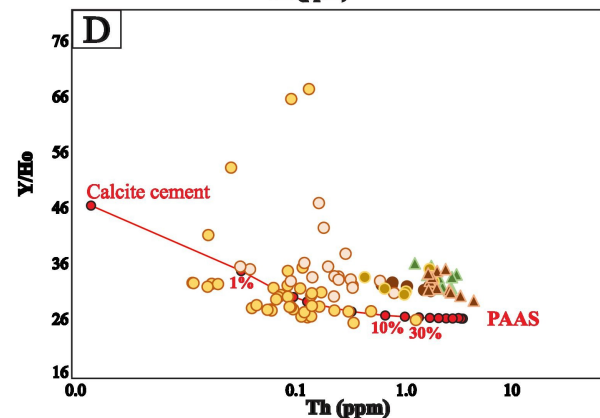
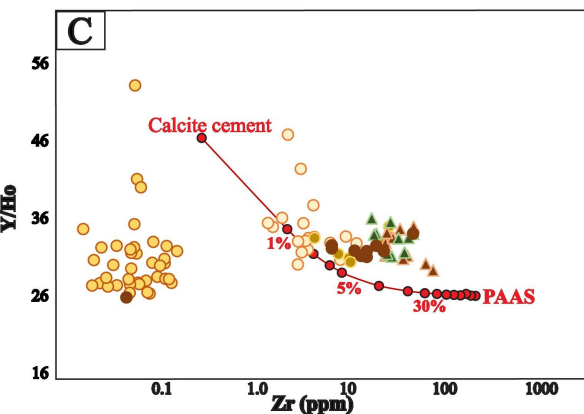
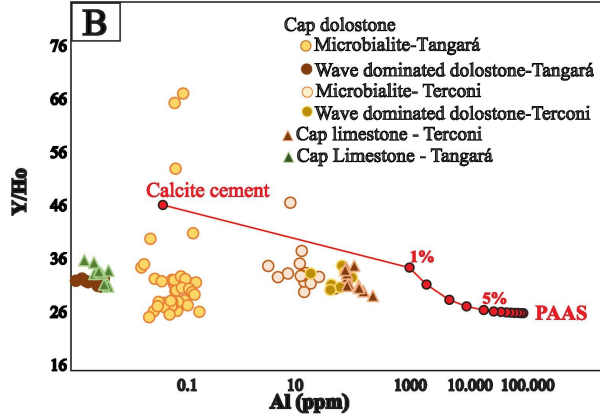
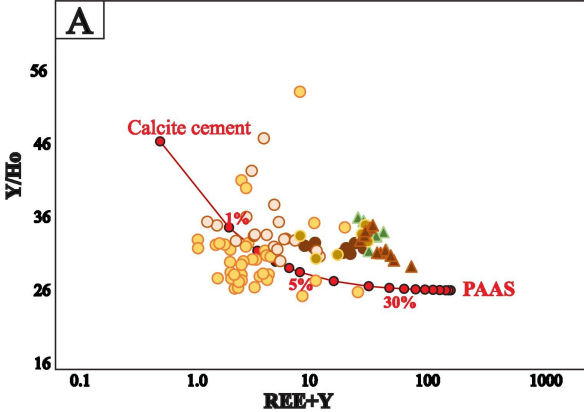


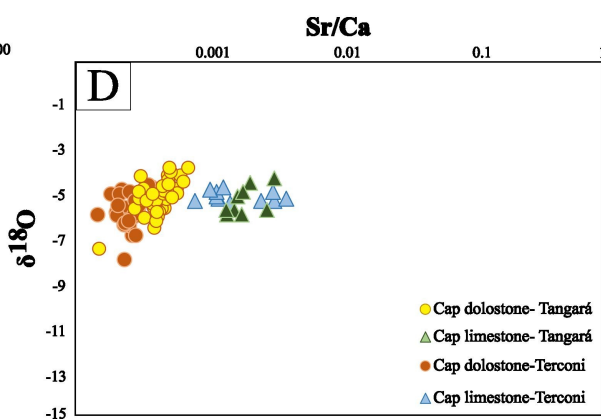
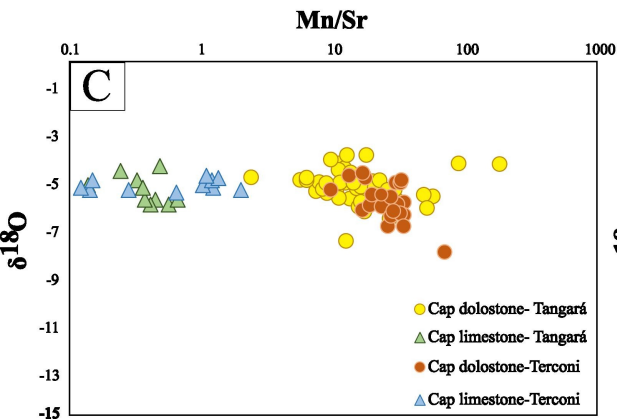
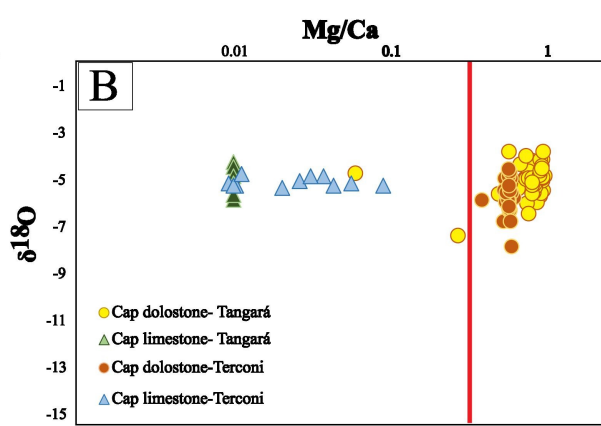
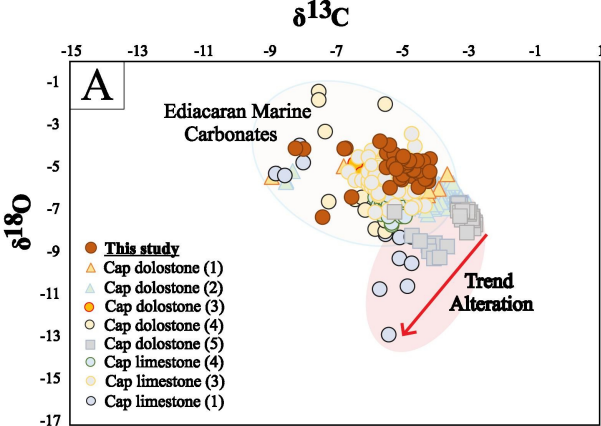


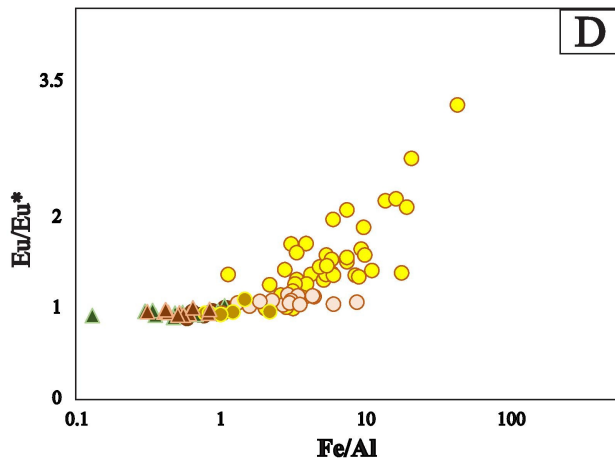
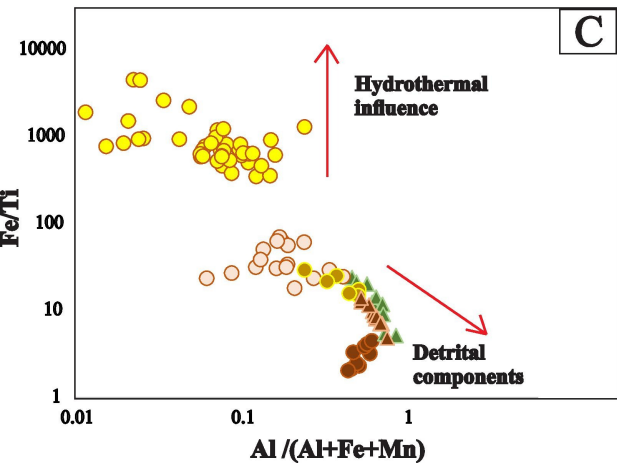
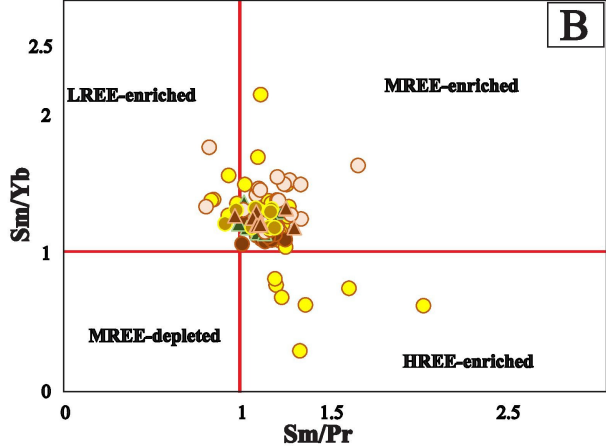
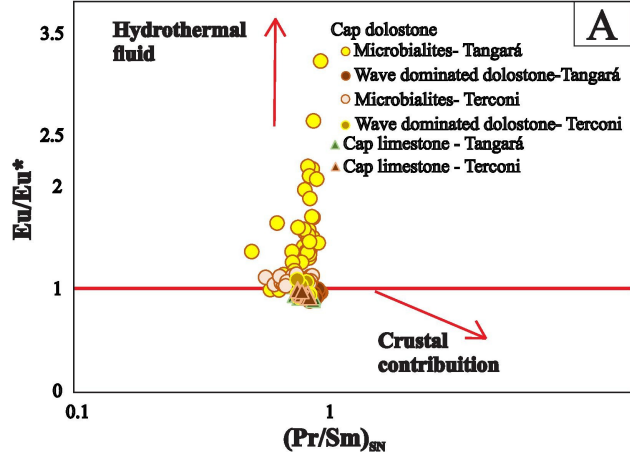






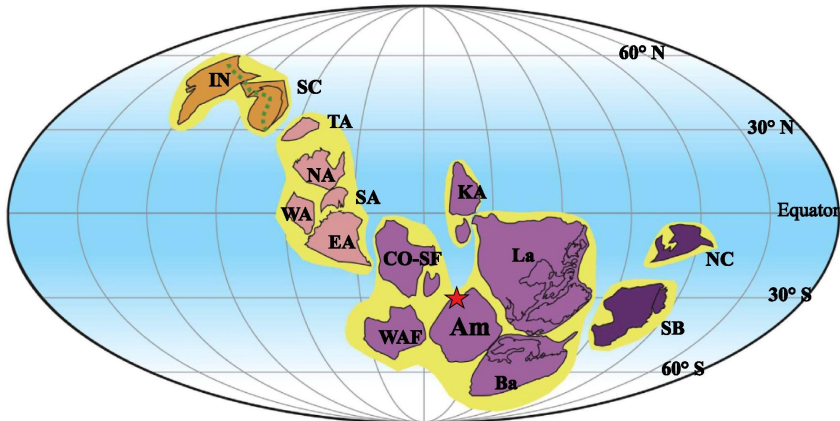




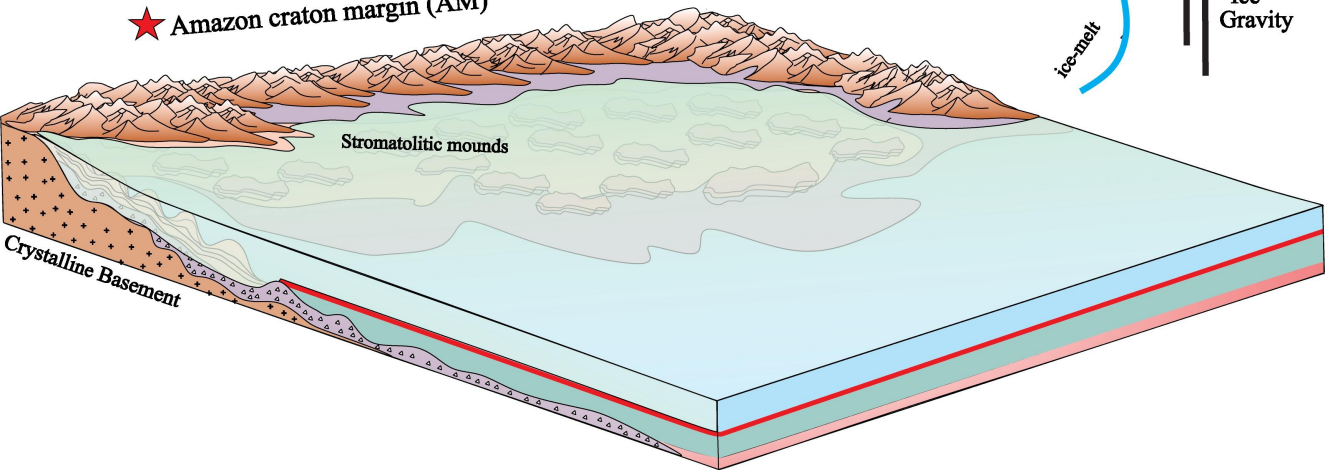


~ 635 Ma

Paleogeographic Reconstruction



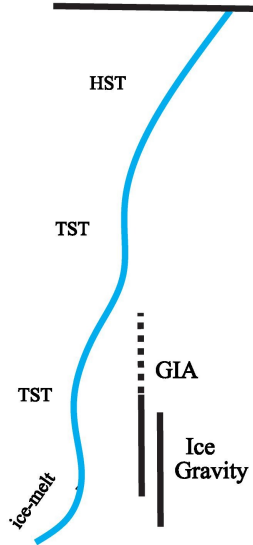
★ Amazon craton margin (AM)



★ Amazon Margin (AM)

Sea level changes

Shallow Deep



Amazon craton margin

