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Fast, furious, and gassy: Etna's explosive eruption from the mantle

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- Magma in Etna's Fall Stratified eruption was extremely volatile-rich (9600 ppm CO₂)
- Magma ascended from storage at the Moho to the surface in <1-5 days
- High volatiles and fast ascent likely caused explosive intensity of this picritic magma

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30 occurred extremely fast at ~ 17.5 m/s. This eruption may provide a link between primary
31 magma composition and eruption intensity: we propose that the unusually explosive nature of
32 this picritic eruption was driven by high H₂O and CO₂ concentrations, which led to
33 continuously rapid ascent without stalling, all the way from the Moho.

34 **1. Introduction**

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35 Basaltic volcanism occurs over a vast range of scales and styles, from effusive lava flows, to
36 Strombolian, Hawaiian, and up to Plinian eruption styles (Houghton & Gonnermann, 2008).
37 Etna is a prime example of a volcano which hosts all of these eruption styles (Branca & Del
38 Carlo, 2005; Coltelli et al., 1998; Zuccarello et al., 2022). The controls on eruption intensity
39 remains poorly understood, especially for subplinian and plinian eruptions (Houghton &
40 Gonnermann, 2008; Parfitt & Wilson, 1994). Subplinian and plinian eruptions are more
41 typical for silicic magmas, where the explosive nature of the eruption is thought to be
42 associated with brittle magma fragmentation. The latter is unlikely to occur in basaltic
43 magmas because of their low viscosity (e.g. Giordano & Dingwell, 2003; Houghton &
44 Gonnermann, 2008), yet several volcanoes are known to have hosted plinian eruptions of
45 basaltic magma (Sunset craters – Allison et al., 2021; Masaya – Bamber et al., 2020; Etna –
46 Coltelli et al., 1998; Tarawera – Houghton et al., 2004; Kīlauea – McPhie et al., 1990).

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In this context, the 3930 BP subplinian Fall Stratified (FS) eruption of picritic magma
at Etna represents an extreme example of this mystery. The magma's high temperatures and
H₂O concentration, and low SiO₂ concentrations and crystallinity, all act to lower viscosity
(Coltelli et al., 2005; Kamenetsky et al., 2007), and yet this eruption was one of Etna's most
explosive. The mystery of how to explosively erupt a low viscosity picritic magma is
highlighted by their rarity; Coltelli et al. (2005) point out that the FS eruption is the only
known subplinian eruption of picritic magma.

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54 The uniformly primitive population of olivines within the FS magma (Fo 90–91)
55 requires relatively fast magma ascent from at or near the Moho to avoid reequilibration
56 toward lower forsterite compositions (e.g. Ruprecht & Plank, 2013). Furthermore, the high
57 density of picritic basalt makes it difficult to ascend through the entire span of the crust
58 (Anderson, 1995), and requires that the vapor phase remains coupled with the melt in order to
59 avoid stalling in the mid crust (Corsaro and Pompilio, 2004).

60 It has been proposed on the basis of experiments, modeling, and field observations of
61 geysers, that the plume height (and explosivity) of an eruption is positively correlated with its
62 source depth (Namiki & Manga, 2006; Reed et al., 2021). Shock tube experiments on bubble-
63 bearing viscoelastic fluids show that more violent fragmentation occurs for larger pressure
64 drops and initial bubble volume fractions (Namiki & Manga, 2006). Greater source depths
65 and higher vesicularity mean more potential energy to be converted into kinetic energy and
66 consequently faster decompression.

67 Coltelli et al. (2005) propose that the FS eruption was propelled by high volatile
68 concentrations (in particular CO₂) causing early exsolution deep in the crust and consequently
69 high rates of magma decompression. We aim to test this hypothesis by determining the
70 primary volatile concentrations and decompression rates of the magma. Ultimately, we want
71 to understand how a picritic magma is able to ascend through the entire crust without stalling,
72 mixing with other magmas, or crucially, losing its vapor, in order to erupt with the most
73 explosive intensity ever documented for a picritic magma.

74 **1.1 Background**

75 The Fall Stratified eruption of Etna (3930 BP) produced voluminous pyroclastic flow
76 deposits (0.055 km³ dense rock equivalent – Coltelli et al., 2005), which attained thicknesses
77 of 110 cm at distances of 7 km from the vent. The plume height has been estimated as 18–20
78 km from isopleth mapping corresponding to estimated mass eruption rates of $3.4 - 5.2 \times 10^7$

79 kg/s and a high explosive energy (VEI 4, subplinian – Coltelli et al., 2005). As suggested by
80 its name, the Fall Stratified eruption deposit consists of fine scale layering between finer and
81 coarser lapilli. Coltelli et al. (2005) argue that the deposit reflects “the formation of a
82 sustained eruptive column” and interpret the layering as reflecting oscillations in plume
83 height and that the lack of lithics and fine particles supports a magmatic rather than phreatic
84 origin of the eruption. The FS eruption marks an important event in Etna’s history – the
85 productivity of explosive eruptions doubled after the FS eruption (Coltelli et al., 2000) (Fig.
86 1b), suggesting that it may have fundamentally altered the magmatic plumbing system
87 beneath Etna and/or signaled a change in the mantle melting process (Fig. 1a). This is
88 supported by the FS magma’s anomalously high Rb/Nb ratios, which is typically high in
89 subducting slab fluids/melts and thus thought to signify high input from subducted material
90 (Fig. 1a).

91 Previous work has established key compositional features of the FS magma (Correale
92 et al., 2014; Corsaro & Métrich, 2016; Gennaro et al., 2019; Kamenetsky et al., 2007). The
93 magma is notable for its picritic composition – unusual for Mt. Etna and not erupted since the
94 FS eruption. Whole rocks have 12–17 wt% MgO, melt inclusions (MIs) have 9–10 wt.%
95 MgO, and most olivines are Fo 89–91. Note that Coltelli et al. (2005) find Fo as low as 83 but
96 values <88 are restricted to microlites, extreme rims, or cores in rarely occurring reverse
97 zoned crystals. Other studies (Kamenetsky et al., 2007; Gennaro et al., 2019) find exclusively
98 Fo 89–91 olivines. Most olivines are unzoned (Coltelli et al., 2005). Another unusual aspect
99 of the FS whole rocks and MIs is the high CaO/Al₂O₃ ratios (1.1–1.5; Kamenetsky et al.,
100 2007), higher than any other eruptions at Etna and any subduction-adjacent magma in the
101 world (e.g. Turner & Langmuir, 2015). The tephra contains high Mg olivines, clinopyroxenes
102 (Mg# 90–91), and Cr spinel. Inclusions of high Mg # clinopyroxenes within olivines indicate
103 early crystallization – a sign of high pressures of crystallization (Kamenetsky et al., 2007)

104 and also consistent with the high CaO/Al₂O₃. Furthermore, high volatiles concentrations (~ 4
105 wt% H₂O, 3500 ppm CO₂) have been measured in MIs, giving equilibrium pressures of ~ 500
106 MPa (Gennaro et al., 2019; Kamenetsky et al., 2007).

107 A major issue with the FS samples is that MIs contain a vapor bubble that is typically
108 coated in Ca-Mg carbonates (Kamenetsky et al., 2007). Therefore, despite the already high
109 measured CO₂ concentrations in the MIs' glass phase, true CO₂ concentrations and inferred
110 equilibrium pressures are likely to be even higher. Previous work over the past years has
111 demonstrated that the CO₂ concentrations in MI bubbles may contain > 80% of the bulk melt
112 CO₂ (e.g. Moore et al., 2015), and so it is of paramount importance to reconstitute the CO₂ in
113 MIs before being able to estimate both the depth of MI entrapment and the primary volatile
114 contents of the magma.

115 2. Sample Description

116 Tephra from the FS eruption was collected by Silvio Rotolo at the Torrente Fontanelle (see
117 IGSN:10.58052/IEACB000Y for further details), and is the same sample as studied by
118 Correale et al. (2014) and Gennaro et al. (2019). The tephra contains large (up to 5 mm) and
119 euhedral olivine and clinopyroxene crystals, along with microcrystals of Cr-spinel (Coltelli et
120 al., 2005; Gennaro et al., 2019; Kamenetsky et al., 2007). Olivine-hosted MIs show clear
121 faceting with scalloped edges (Fig.2). When viewed down the **b** axis MIs are typically oval to
122 spherical, while along **a** or **c** they show a hexagonal shape. The 'tips' of the hexagons
123 correspond to a plane (010) with a halo of micro fluid inclusions (present around all MIs,
124 Fig.2b). This texture has been described elsewhere and may indicate partial decrepitation
125 (Anderson JR, 1974; Portnyagin et al., 2005). Cr-spinels are common inclusions in olivine,
126 and occur in more than 70% of the studied MIs (Fig.2, supplementary data table).
127 Clinopyroxene can also be seen as inclusions within olivine (Fig. 2c) and were measured to
128 have high Mg # (~90) by Kamenetsky et al. (2007).

129 All MIs contain a vapor bubble accounting for 3–10 vol% of the MI volume
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2 130 (Supplementary data table). There are no anomalously large bubbles, and bubble volume
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4 131 fraction correlates with MI size, typical of shrinkage bubbles and not those that are co-
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6 132 entrapped (Steele-MacInnis et al., 2017). The bubbles are typically coated in Ca-Mg-bearing
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8 133 carbonates (Fig. 2b and Kamenetsky et al., 2007). Many olivines contain clearly decrepitated
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10 134 MIs, characterized by an abnormally large vapor bubble and patterns of fluid inclusions
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12 135 radiating from the MI along planar cracks (Fig. 2a) – these were avoided for further study.
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14 136 For modeling H⁺ and Fo profiles within the olivine, we selected inclusion-free, euhedral
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16 137 olivines. Crystal orientation was determined from morphology and confirmed for 28 crystals
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18 138 with electron backscatter diffraction.

24 139 3. **Methods**

26 140 We used a suite of analytical and experimental methods. Novel techniques are highlighted
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28 141 here, while routine methods are detailed in the supplement.

31 142 **Rehomogenization experiments**

33 143 In order to redissolve the shrinkage bubble and associated carbonate back into the melt, we
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35 144 attempted to rehomogenize the MIs using a piston-cylinder, as in Rasmussen et al. (2020).
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37 145 These attempts were unsuccessful, likely because the quench rate was not fast enough
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39 146 (supplement). We had better success using a vertically oriented internally heated pressure
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41 147 vessel (IHPV) at the Institute of Mineralogy, Leibniz University in Hannover, which has
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43 148 faster quench rates and proved an effective tool for high-pressure rehomogenization
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45 149 experiments with MIs (e.g. Mironov et al., 2015). Pt capsules were filled with MI-bearing
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47 150 olivines with and without de-ionised water and held at ~ 4 kbar and temperatures of 1250,
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49 151 1280, and 1330 °C for ten minutes (supplement).

56 152 **FTIR – glasses**

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153 H₂O and CO₂ MI concentrations (n=36) were measured on the Thermo Nicolet iN10 at
154 Lamont-Doherty Earth Observatory. We paid particular attention to the baseline under the
155 carbonate peaks (Shi et al., in revision). Full details are in the supplement.

156 **FTIR – olivine**

157 H₂O concentration profiles (n=18) were measured in doubly polished olivine wafers on the
158 Thermo Nicolet iN10 at Lamont-Doherty Earth Observatory (Fig. 3). Barth et al. (2023)
159 found good agreement between different peak-specific diffusivities so we consider bulk H₂O
160 by summing the total area under the baseline-subtracted spectra. Full details are in the
161 supplement.

162 **SIMS**

163 Rehomogenized MIs were fragile and developed cracks when intersected by polishing.
164 Therefore, FTIR was not an option for the rehomogenized MIs, which were singly polished
165 and analyzed by SIMS. Volatile abundances in rehomogenized MIs (n=26) were measured
166 using a Cameca IMS 1280 multi-collection ion microprobe at CRPG-CNRS-Nancy. The
167 calibration for C during this session was noisy and deviated significantly from typical
168 calibration curves for the same instrument (Supplementary data table). To check the
169 calibration, one rehomogenized MI and two check standard glasses were removed from the
170 indium mount, doubly polished, and measured by FTIR. These values were then used as the
171 calibration line (Supplementary data table). A second SIMS session was conducted at Caltech
172 in 2022, primarily to verify the H₂O data, and a reasonably good agreement was found
173 (Supplementary data table).

174 **EPMA**

175 Major elements for MIs and adjacent spots in their host olivines (n=49) were collected by
176 electron microprobe analysis (EMPA) at the Smithsonian Institution using a JEOL JXA-
177 8530F Hyperprobe. Major, minor and trace element profiles in olivine (n=8) were measured

178 using EPMA by a Cameca SX-5 with 5 WDS spectrometers at the American Museum of
179 Natural History (AMNH).

180 **Diffusion Modeling – H₂O**

181 In order to determine magma ascent rates, we performed diffusion modeling of H₂O
182 concentration profiles in olivines, which is sensitive to ascent rate in the depth region in
183 which H₂O is degassing from the magma (~0–15 km depth). Our methods mostly follow
184 those in Newcombe et al. (2020), except that we do not fix the final (shallowest) pressure.
185 Instead, we solve for both the best-fit decompression rate and final pressure. Full details are
186 in the supplement.

187 **Diffusion Modeling – Forsterite**

188 To estimate timescales of ascent from storage to eruption, we performed diffusion modeling
189 of forsterite in eight unhomogenized olivine crystals. All profiles show a decrease from a
190 central plateau towards the rim over a lengthscale of 50 – 100 μm for all analyzed elements
191 (Ni, Mn, Ca, Al, forsterite), regardless of their diffusivity (Fig. 4). Al is considered to be
192 almost immobile in olivine at magmatic temperatures over timescales of decades (Spandler &
193 O'Neill, 2010), while Ni, Mn, Ca and Fe-Mg all diffuse faster. Thus, the lack of decoupling
194 between elements indicates that the zoning we observe was dominantly generated by crystal
195 growth, and that there was insufficient time to modify these profiles by diffusion.

196 Furthermore, due to diffusive anisotropy in olivine, where Fe-Mg interdiffusion is 6 times
197 faster along the **c** than the **a** axis (Dohmen & Chakraborty, 2007), we would expect to see
198 different diffusive lengthscales along different crystallographic directions. The similar zoning
199 lengthscales of forsterite content along **a** and **c** also argues for growth-dominated zoning.

200 We used a combination of forward and inverse modeling to obtain the maximum
201 timescale in which Fe-Mg growth-zoning can be preserved for each measured profile along
202 the **c** axis (supplement). This would give the maximum residence time at magmatic

203 temperatures between crystal entrainment near the Moho and quenching upon eruption. To
204 calculate this diffusive decoupling timescales for Fe-Mg interdiffusion, we used the measured
205 growth profile as the initial condition. This assumes that the whole rim grew instantaneously,
206 where the start of the growth rim is marked by the decrease of Al from core values. We then
207 modeled Fe-Mg diffusion until the model curve can be considered sufficiently decoupled
208 from the initial growth conditions, accounting for analytical uncertainty. Modeling was
209 performed using DFENS (Mutch et al., 2019, 2021) – full details in the supplement.

210 **4. Results**

211 **4.1. IHPV rehomogenization experiments**

212 After ten minutes of heating at 1250, 1280, and 1330 °C in the IHPV, the MIs exhibit a range
213 of textures. All MIs become darker-colored after heating (Fig. 5), which may reflect the
214 presence of Fe-bearing nanolites or an increased $Fe^{3+}/\Sigma Fe$ (Lerner et al., 2021). Some MIs
215 are glassy and still contain a single vapor bubble, although comparing photos taken before
216 and after the experiment shows that the bubbles have shrunk and moved during the
217 experiment (Fig. 5a,b). Some MIs become very dark and speckled (with μm -scale bubbles or
218 crystals) throughout (Fig. 5i,j), while others are glassy inside but have speckled surfaces (Fig.
219 5c-h).

220 For further analysis, MIs which had resorbed their bubble were prioritized – no MIs at
221 1250 °C, and only two MIs at 1280 °C (O18, O129) were chosen. All available MIs from the
222 1330 °C runs were analyzed.

223 **4.2. Major and minor elements**

224 The measured chemical compositions of the MIs and their host olivines agree well with those
225 of Kamenetsky et al. (2007), Corsaro and Métrich (2016), and Gennaro et al. (2019) (Fig. 6).
226 We correct for PEC using MIMiC (Rasmussen et al., 2020) calculating the partition

227 coefficient, K_d , from Toplis (2005) fixing Fe^{3+}/Fe_T to 0.32 based on microXANES
228 measurements in Gennaro et al. (2020).

229 These calculations suggest minor PEC – up to 8%, with an average of 3%. Several
230 MIs appear to have excess olivine – a possible sign of reheating. MIs show a limited range in
231 all major and minor elements, and host olivine forsterite contents are equally restricted (90.3
232 – 91.2 %) – there is no evidence for significant fractionation (Fig. 6). Entrapment
233 temperatures calculated from MIMiC are $1196 \pm 11^\circ C$ (supplementary data table).

234 One remarkable characteristic of the Etna FS MIs is their extremely low Al_2O_3 (9.1 –
235 11. wt%) and high CaO (13.5 – 15.5 wt%) concentrations. While low Al_2O_3 can be achieved
236 by high pressures of melting, the combination of high CaO and low Al_2O_3 points to a high
237 CO_2 concentration in the source (Dasgupta et al., 2007; Lara & Dasgupta, 2022) or the
238 involvement of clinopyroxene reaction processes (Danyushevsky et al., 2004).

239 The homogenized MIs show evidence for significant melting of the olivine host, as
240 well as Fe-Mg exchange (MgO concentrations up to 18.5 wt%), which is common in these
241 types of experiments that overheat the MIs (e.g. Rasmussen et al., 2020). When they are
242 corrected for these processes, they overlap with the compositions of the unhomogenized MIs
243 (Fig. S7).

244 **4.3. Volatiles**

245 Unhomogenized MIs show a weak positive correlation between H_2O and CO_2 with ranges of
246 2.5–4.9 wt% and 1910–3190 ppm, respectively (Fig. 7c). After PEC correction, these values
247 are slightly lower due to the effect of dilution and are in the range 2.3–4.9 wt% and 1890–
248 3110 ppm. In agreement with Gennaro et al., 2019, we do not find a correlation between H_2O
249 concentrations and MI size (Fig. 6d, see supplement for details on size calculation),
250 suggesting that the MIs have not undergone significant water loss during ascent (e.g. Qin et
251 al., 1992). FTIR values for H_2O and CO_2 overlap with those in Gennaro et al. (2019),

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252 although their SIMS measurements are offset to higher concentrations for both volatiles
253 (supplement).

254 Homogenized MIs show a range of CO₂ concentrations from 4900 to 9600 ppm – a
255 clear increase from the unhomogenized MIs (Fig. 7c) that is on average two-fold. Four of the
256 measured MIs have a single remaining vapor bubble, and these MIs fall to the low end of the
257 range of CO₂ concentrations (open black circles, Fig. 7c). Nine homogenized MIs contain
258 multiple small bubbles at their edge; however, this does not appear to affect their CO₂
259 concentrations (purple vs. red circles, Fig. 7c), as found in Rasmussen et al. (2020). All of the
260 homogenized MIs have μm -scale phases (crystals and/or bubbles), primarily lining the MI
261 wall, but in some cases distributed in the interior of the MI. At present, we do not know what
262 these phases are, and it is possible that they contain some CO₂, which would make our
263 measurements lower bounds on the total amount of CO₂ in the MIs. Furthermore, the ‘haloes’
264 of secondary fluid inclusions surrounding all inclusions even before homogenization (Fig. 2)
265 may indicate partial decrepitation, in which case the amount of CO₂ in the MIs at the time of
266 entrapment may have been higher still.

267 Homogenized MIs have a narrower range and higher H₂O concentrations than
268 unheated (Fig. 7c). This is unlikely to be the result of H₂O diffusing through the olivine from
269 the experimental capsule since the experiments only last ten minutes, and we observe no
270 systematic difference between dry and wet experiments. We cannot rule out that this is an
271 analytical discrepancy – three unhomogenized and one homogenized MIs were measured by
272 both SIMS and FTIR and the methods show discrepancies up to 0.85 wt.%, with the FTIR
273 values exhibiting a wider range than the SIMS (supplement). It is possible that the FTIR
274 measurements on unheated MIs underestimate H₂O concentration due to saturation of the
275 detector. However, this would not explain the observed (albeit weak) H₂O-CO₂ correlation
276 (Fig. 4c). Furthermore, we would expect the thickest wafers to give the lowest H₂O values,

277 which is not observed (Fig. S3). Alternatively, there may be hydrous phases on the walls of
278 the MIs (although we note that none were found by Raman spectroscopy – Kamenetsky et al.,
279 2007) or H₂O in the vapor bubble prior to homogenization, which is added back into the melt
280 during homogenization (Esposito et al., 2016; Portnyagin et al., 2007).

281 The homogenization experiments do not appear to have altered the MI S or Cl
282 concentrations (Fig. 7e), suggesting that they are not stored in the vapor bubble in significant
283 quantities. In agreement with the results of Gennaro et al. (2019) and Corsaro and Métrich,
284 (2016), we find that S does not correlate with CO₂ or H₂O, although it strongly correlates
285 with Cl for both the homogenized and unhomogenized MIs (Figs. 7e,f,g). The S-Cl
286 correlation is surprising since Cl and S are not expected to degas at such high pressures (>
287 500 MPa) (e.g. Ding et al., 2023). Neither correlate with Ni or Fo of the host olivine, or any
288 indices of fractionation (e.g. K₂O). Cl and S variations may thus be indicative of primary melt
289 heterogeneity (as is present in other trace elements – Kamenetsky et al., 2007).

4.4. Diffusion modeling

4.4.1. H₂O

292 All olivines appear zoned in H₂O concentration along a with no central plateaus, indicating
293 that diffusion has reached the center of each olivine. Central H₂O concentration ranges from
294 14 to 33 ppm. Edge concentrations range from 6 to 25 ppm. An example Monte Carlo model
295 run can be seen in Fig. 8. There is a positive correlation between the diffusivity and
296 decompression rate while other parameters exhibit weak or no correlation with
297 decompression rate. Therefore, within the parameter ranges studied here, the uncertainty in
298 diffusivity accounts for the largest source of uncertainty in decompression rate. We
299 emphasize, however, that the diffusivity used was determined specifically for Etna FS
300 olivines in a companion study (Barth et al., 2023). The best-fit decompression rate for all 18
301 olivines is remarkably uniform (Fig. 8b). A histogram of all best-fit decompression rates

302 shows a lognormal distribution centered at 0.47 MPa/s with 95 % confidence bounds of 0.16
303 and 1.28 MPa/s. Assuming an average crustal density from the surface to 15 km depth of
304 2600 kg/m³ from Corsaro & Pompilio (2004), this corresponds to 17.5 m/s with 95%
305 confidence bounds of 6 and 49 m/s.

306 4.4.2. *Forsterite*

307 The median diffusive decoupling timescales determined from the Bayesian inversion (Mutch
308 et al., 2021) range from one to five days for the different olivines. Uncertainties on these
309 timescales propagated from those in temperature, fO_2 , pressure, and diffusion coefficients
310 vary for the different olivines and are -2 days and +4 days on average (full results in
311 supplementary data table). Note that we can only constrain an upper bound on timescales
312 since there are no diffusion profiles to model.

313 These results indicate that the olivines must have ascended from storage at near Moho
314 depths to the surface at rates of 0.05 – 0.35 m/s or faster. If ascent were slower, the olivines
315 would show signs of reequilibration with the surrounding olivine-free granulitic crust
316 (Corsaro & Pompilio, 2004; Tonarini et al., 1996). Recent experimental results from Shea et
317 al. (2023) show that the presence of a surrounding melt may enhance cation diffusion in
318 olivines by up to a factor of 10. While the mechanism for this effect is still not well
319 understood, it is thus possible that the true ascent velocities are even faster than the above
320 estimates.

321 5. Discussion

322 5.1 *Reconstructed CO₂ concentrations*

323 5.1.1 *Comparison of MIMiC and homogenization experiments*

324 Since the discovery of the importance of MI vapor bubbles, several methods have been
325 developed to account for the CO₂ stored within them. These techniques fall broadly into three
326 categories: (1) dissolving the vapor bubble back into the MI with rehomogenization

1 327 experiments before measurement, (2) using an equation of state and measuring or modeling
2 328 the size of the bubble, and (3) measuring the CO₂ in the vapor bubble with Raman
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4 329 spectroscopy. These methods are described and compared more fully elsewhere (Moore et al.,
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7 330 2015; Rasmussen et al., 2020; Wallace et al., 2021). We compare the results of our
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10 331 rehomogenization experiments with the modeling approach of Rasmussen et al. (2020) using
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12 332 the publically available MIMiC code. MIMiC calculates bubble volume at the closure
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14 333 temperature for CO₂ (the temperature at which CO₂ diffusion effectively ceases), using an
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17 334 equation of state approach, and taking into account deformation of the host olivine. The
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19 335 model input includes the measured chemistry and sizes of the MIs and olivine hosts and an
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22 336 estimate of cooling rate (we assume a cooling rate of 10 °C/s, as recommended for lapilli-
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24 337 sized clasts – Rasmussen et al., 2020).

26 338 There is close agreement between the CO₂ concentrations predicted by MIMiC using
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29 339 the unheated MIs and those measured by SIMS in the homogenized MIs, giving a spread of
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31 340 4400–8200 and 4900–9600 ppm, respectively (Fig. 7c). This suggests that minimal CO₂ is
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34 341 stored in the tiny phases lining the walls of some of the rehomogenized MIs (Fig. 7a,b).

36 342 ***5.1.2 Pressure of magma storage***

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39 343 One of the primary ways to estimate magma storage depth is by calculating saturation
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41 344 pressures based on the H₂O and CO₂ concentrations within MIs. Therefore, restoring the CO₂
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43 345 concentrations in the vapor bubbles has implications for the depths of magma storage. Note
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46 346 that these pressures are minima because they assume that the magma is saturated with a H₂O-
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49 347 CO₂ vapor phase. There are a range of solubility models for mixed volatiles, which have been
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51 348 calibrated over different pressure, temperature, and compositional space. MagmaSat (Ghiorso
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53 349 & Gualda, 2015) has the best coverage near the Etna FS composition and temperatures (Fig.
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56 350 S4,5). Duan (2014) also has good coverage, although is not implemented in the python
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58 351 package VESIcal (Iacovino et al., 2021). The models from Iacono-Marziano et al. (2012),
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352 Allison et al. (2019), and Allison et al. (2022) are calibrated on Etna magma but because the
353 FS magma is so unusual for Etna it lies outside their compositional range (Fig. S4,5). The
354 model of Dixon (1997) has good compositional coverage but does not extend to pressures
355 higher than 500 MPa, and Iacono-Marziano et al., 2012 suggested that the Dixon (1997)
356 calibration range was only reliable up to 100 MPa.

357 MagmaSat and Duan (2014) give the highest saturation pressures (676–819 and 645–
358 1032 MPa, respectively, corresponding to the range in CO₂ concentrations in homogenized
359 MIs without single vapor bubbles), while the solubility models in Iacono-Marziano et al.
360 (2012), Allison et al. (2019), and Allison et al. (2022) give lower pressures (405–524, 415–
361 649, and 343–547 MPa, respectively). Based on the different lithologies with depth beneath
362 Etna (Corsaro & Pompilio, 2004), we iteratively solve for the magma storage depth and
363 average density of the overburden. This gives an average overburden density of 2720 kg/m³
364 and depth of 30 km for the highest initial pressure of 819 MPa, and average density and depth
365 of 2650 kg/m³ and 24 km for the shallowest initial pressure of 676 MPa. These depths closely
366 align with entrapment depths determined from olivine-hosted CO₂-rich fluid inclusions of
367 21–24 km of samples from Aci Castello and Mt. Maletto (Kamenetsky & Clocchiatti, 1996).
368 We now examine how these depths relate to the crustal structure beneath Etna.

369 Although the regional Moho beneath eastern Sicily is on the order of 30 km (e.g.
370 Accaino et al., 2011), detailed seismic imaging offshore of Etna finds a mantle upwarp
371 (Nicolich et al., 2000) associated with a major tectonic feature, the Alfeo Fault System
372 (Dellong et al., 2018). Nicolich et al. (2000) map a thinning of the Moho from 25 to 17 km
373 immediately offshore of Etna, and suggest that the location of Etna is related to this tectonic
374 feature. We thus consider the Moho beneath Etna to be in the range of 17–25 km and that the
375 FS magma ascended from storage at or beneath the Moho. Viewed another way, our volatile
376 saturation pressures can help to constrain the Moho depth, since the high and uniform

377 forsterite content (90–91) of the FS olivines requires that the olivines must be coming from
378 storage at or below the Moho, since they cannot have equilibrated with the olivine-free mafic
379 granulites of the lower crust (Corsaro & Pompilio, 2004; Tonarini et al., 1996). As noted by
380 Corsaro & Pompilio (2004) the FS magma “represents a primitive melt in equilibrium with
381 the mantle”. Crucially, the entrapment depths of the FS MIs means that the high CO₂
382 concentrations cannot have been the result of assimilation of the shallow carbonate crust,
383 which is thought to extend to depths of ~ 10 km (Fig. 9, Corsaro and Pompilio, 2004).

384 ***5.2 Carbonate in the mantle source feeding the FS eruption***

385 Here, we address the question of what generated the unusually high CaO/Al₂O₃ ratios (1.1–
386 1.5) in Etna FS MIs and whole rocks (Fig. 10a, Kamenetsky et al., 2007; Corsaro and
387 Métrich, 2016). The two primary hypotheses are involvement of 1) clinopyroxene-bearing
388 lithologies or 2) carbonate in the magma’s source.

389 High CaO MIs have been explained by melting/dissolution of clinopyroxene-bearing
390 lithologies in general terms (Danyushevsky et al., 2004; Schiano et al., 2000) and specifically
391 for the Etna FS magma (Correale et al., 2014; Corsaro & Métrich, 2016). However, the
392 approach of Correale et al. (2014) relied on assumptions about melting and crystallization
393 based on major element modeling which was not able to fit the FS compositions (see their
394 Fig. 10). Corsaro and Métrich (2016) argue that the high Zr/Nb and low Ce/Y requires the
395 melting of a clinopyroxene-bearing lithology, however measured ratios in FS whole rocks
396 overlap with measured ratios for Hyblean peridotite xenoliths (Correale et al., 2014).

397 On the other hand, experiments have shown that addition of CO₂ to the mantle source
398 can generate high CaO/Al₂O₃ ratios in the magma (up to 2.7), both by lowering Al₂O₃ and
399 raising CaO in the melt (e.g. Dasgupta et al., 2007). In volatile-free peridotite, this level of
400 CaO-enrichment and Al₂O₃ depletion can only be generated by melting at pressures above 4–
401 5 GPa (Walter, 1998) – greater than typical arc melting pressures. Recent experiments with

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402 H₂O- and CO₂-bearing fluids (Lara & Dasgupta, 2022) suggest that the Etna FS magma
403 composition could be explained by the presence of a fluid with XCO₂ (molar
404 CO₂/(CO₂+H₂O)) of ~ 0.1. Furthermore, the olivine compositions (namely their Ni, Mn, and
405 Ca concentrations) are consistent with carbonate metasomatism of the magma's source, rather
406 than involvement of pyroxenite (Fig. 10a; Ammannati et al., 2016).

407 Finally, there is a regional signature of carbonate metasomatism in the mantle. At
408 Stromboli volcano, 115 km from Etna, isotopically heavy C ($\delta^{13}\text{C}$ -2.7‰ to -1.0‰),
409 indicative of carbonate, was measured in bulk separates of high forsterite olivines from
410 xenoliths of ultramafic cumulates thought to derive from the mantle (Gennaro et al., 2017).
411 The authors propose that this reflects contamination of the mantle by C-rich sediments from
412 the subducting slab. Magmas from the Hyblean district, ~60 km S of Etna show trace element
413 and Sr-Nd-Pb isotopic signatures consistent with carbonatite metasomatism (e.g. Trua et al.,
414 1998). C-rich fluid inclusions in wehrlitic xenoliths from Monte Vulture (~350 km N of Etna)
415 further support carbonate metasomatism of the mantle (Carnevale et al., 2022). Below, we
416 outline further support for the involvement of C-rich subducting sediments in the source of
417 the Fall Stratified eruption.

418 Etna is situated away from the main Aeolian arc, and its location may be related to a
419 tear in the subducting slab (e.g. Gvirtzman & Nur, 1999). A trace element subduction
420 signature (enrichment in large-ion lithophile (LILE), depletion in high field strength elements
421 (HFSE)) has been documented for Etna magmas since ~100 ka (Corsaro & Métrich, 2016;
422 Schiano et al., 2001; Tonarini et al., 2001), including for the FS magma (Kamenetsky et al.,
423 2007). Using the average H₂O value from our MIs (~ 4.7 wt%) and the Ce concentration in
424 the whole rock (Correale et al., 2014) and MIs (Kamenetsky et al., 2007) gives a H₂O/Ce
425 ratio of 750 and 500–1300, respectively – clearly above typical MORB values (100–300;
426 Dixon et al., 2002), suggesting that a significant portion of the magma's H₂O was derived

427 from the subducting slab. Corsaro and Métrich (2016) show that the Cl/K₂O ratios of FS MIs
428 are well above typical mantle values, suggesting that most of the Cl is derived from the slab
429 (Fig. 10). High Cl/K₂O and CaO/Al₂O₃ ratios have also been found for other historical
430 primitive eruptions at Etna: Mt. Spagnolo (15–4 ka) and Mt. Maletto (7 ka). Furthermore, Mt.
431 Spagnolo and FS show a correlation between CaO/Al₂O₃ and degree of Cl enrichment. They
432 are also the only two eruptions to show a positive correlation in S and Cl (supplement).
433 Therefore, we propose that the high CaO/Al₂O₃ ratios and enrichment in Cl and S are
434 reflecting a common process, likely input into the mantle of a fluid or melt that is rich in Cl,
435 S, H₂O and CO₂ from a subducting slab, whether it was added just prior to the FS eruption or
436 stored on long timescales in metasomatic phases in the mantle (Viccaro & Cristofolini, 2008).

437 *5.3 Magma decompression rate*

438 The combination of forsterite and H₂O diffusion modeling allows us to piece together the
439 magma ascent rate all the way from source to surface (Fig. 9). The lack of diffusive zonation
440 of forsterite within the olivine crystals is highly unusual and implies rapid ascent (> 0.1 m/s)
441 from near Moho depths (24 – 30 km) to the surface in less than 1–5 days (Fig. 4). These rates
442 are faster than those estimated for ascent from the Moho in the Borgarhraun eruption of
443 Iceland (0.02–0.1 m/s; Mutch et al., 2019), which could be tied to the lower CO₂
444 concentrations (~600 ppm) and consequently shallower vapor saturation (~3 km) of the
445 Borgarhraun magma.

446 The H₂O zonation profiles within the olivine crystals require average decompression
447 rates of 0.47 MPa/s from depths of ~15 km to the surface. This is among the highest
448 measured by diffusion chronometry for basaltic-intermediate eruptions (Barth et al., 2019;
449 Cassidy et al., 2018), although faster rates have been found in the final few kilometers of
450 ascent beneath Ambrym and Ambae/Aoba volcanoes (Moussallam et al., 2019, 2021) and for
451 recent paroxysms at Etna (Giuffrida et al., 2018; Zuccarello et al., 2022). The most explosive

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452 of these recent paroxysmal events attained plume heights of ~15 km, although it was
453 significantly lower in volume (by a factor of ~25 – Coltelli et al., 2005; Corsaro et al., 2017)
454 and was sourced from shallower depths (~1 km b.s.l.) than the FS eruption. Crucially, the
455 ascent rates that we determine exceed those which have been experimentally shown to induce
456 CO₂ supersaturation, which has been suggested as a mechanism for fragmentation of low-
457 viscosity basaltic magmas (Pichavant et al., 2013).

458 In keeping with the uniformity of olivine and MI compositions, the decompression
459 rates are also highly uniform. This supports the observation by Newcombe et al. (2020) that
460 more explosive eruptions may have less variable decompression rate than lower energy
461 eruptions, for which unsteadiness in flow rate may be more pronounced.

462 To assess whether the ascent rates that we derive are physically plausible, we apply
463 the analytical expressions for the propagation of fluid-filled fractures driven by fluid
464 buoyancy given by Davis et al., 2023. In their formulation, ascent rate is a function of fluid
465 volume, density, viscosity and the elastic parameters of the host medium. Using a Monte
466 Carlo approach, we calculate ascent rate and uncertainty by drawing parameters from normal
467 distributions (Fig. S11) with the following mean and standard deviation: volume =
468 $0.055 \pm 0.005 \text{ km}^3$ (Coltelli et al., 2005), average density contrast between host crust and
469 magma = $285 \pm 100 \text{ kg/m}^3$ (supplement, fig. S10), and viscosity = $4.6 \pm 0.5 \text{ Pas}$ (calculated
470 from the model by Giordano et al., 2008 on the average unheated MI composition). Note that
471 this viscosity model includes the effect of H₂O but not CO₂, which could make the viscosity
472 even lower (e.g. Di Genova et al., 2014). The Young's modulus of the crust is the least well-
473 constrained parameter; we calculate a depth-average from the surface to 30 km of 89 GPa
474 from Currenti et al., 2007 and use a large uncertainty of $\pm 30 \text{ GPa}$. If ascent starts at 24–30
475 km depth, the model predicts average ascent rates of 8–11 m/s (Fig S9). This suggests that

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476 our calculated ascent rates from H₂O diffusion (mean of 17.5 m/s) and our lower bound from
477 Fe-Mg interdiffusion (> 0.1 m/s) are reasonable.

478 ***5.4 Uniqueness of FS Eruption***

479 Other primitive Etna magmas (Mt. Spagnolo and Mt. Maletto) show similar
480 geochemical characteristics as the FS magma (Corsaro & Métrich, 2016), implying that these
481 magmas may not be rare at Moho depths beneath Etna but that their signature gets destroyed
482 with mixing and fractionation in the plumbing system. Unlike the FS magma, Mt. Spagnolo
483 and Mt. Maletto magmas both show a significant range in geochemical indicators of
484 fractionation (e.g. K₂O, MgO), indicating that they either mixed with more fractionated
485 magma during ascent, or crystallized extensively themselves. Either scenario implies slower
486 ascent for these two magmas compared to the FS magma, which may have allowed the
487 escape of their volatile phases and hence lowered the eruption intensity.

488 The high CO₂ concentration of the FS magma and a consequently deep exsolved
489 vapor phase may have initiated diking at Moho depths. Thus, the magma was still hot and had
490 low viscosity when it began ascending, allowing it to propagate fast (8–11 m/s, as above).
491 Either the transporting dyke fortuitously avoided shallow magma reservoirs, or else it was
492 able to pass through relatively unhindered, perhaps because of its fast speed and/or the long
493 hiatus of volcanic activity that preceded the FS eruption (Fig. 1b), which may have allowed
494 shallow magma bodies to cool and solidify. The total pressure-drop and vesicularity during
495 ascent is thought to affect the expansion velocity of the magma, and consequently the
496 eruptive intensity (e.g. Namiki and Manga, 2006). Therefore, the FS magma's unhindered
497 ascent from its deep storage pressures and its high volatile concentrations (particularly CO₂)
498 may have been responsible for its high explosive energy.

499 **6. Conclusions**

1 500 The subplinian FS eruption of Etna was remarkable in terms of its primitive composition and
2 501 high volatile concentrations. Previous studies underestimated the magma's CO₂ concentration
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4 502 by neglecting the MI vapor bubbles. In this study, we rehomogenized olivine-hosted MIs to
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6 503 resorb their vapor bubbles and carbonate back into the MIs. SIMS and FTIR measurements
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8 504 show that these rehomogenized MIs contain between 4900 and 9600 ppm CO₂,
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10 505 corresponding to magma storage pressures of at least 676–819 MPa (24–30 km). These
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12 506 depths are comparable to the seismologically estimated Moho (17–25 km) suggesting that
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14 507 MIs were entrapped in the mantle, and cannot have gained their CO₂ contents by dissolving
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16 508 the carbonate crust, which only extends to ~ 10 km depth. Instead, we argue that the MIs'
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18 509 high CO₂ concentration is derived from carbonate in the mantle source of Etna on the basis of
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20 510 MI CaO/Al₂O₃ ratios, olivine Ni contents, and isotopic measurements of C in high forsterite
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22 511 olivines from the nearby Stromboli volcano.

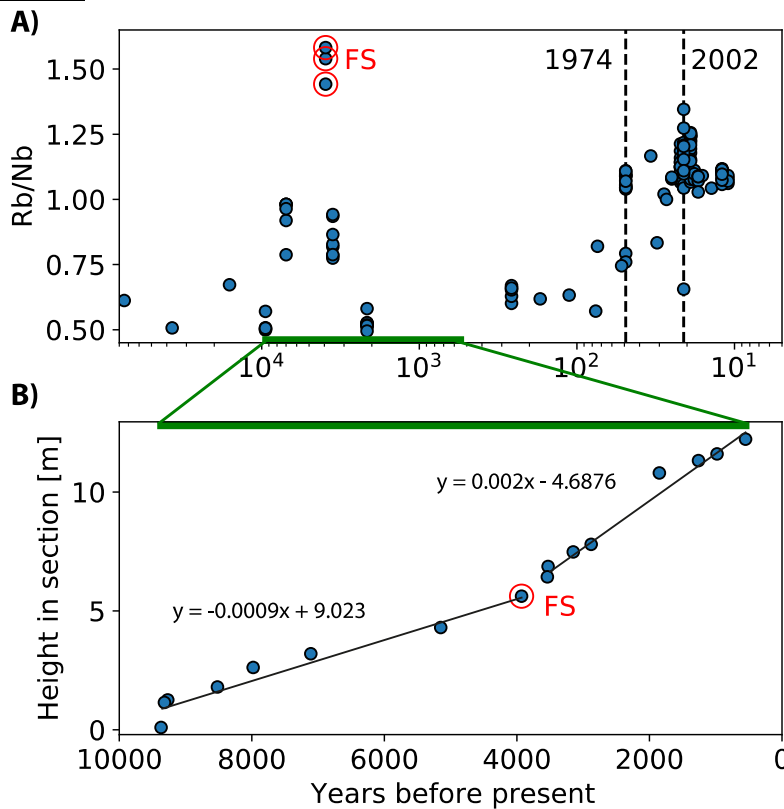
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29 512 Diffusion modeling of H₂O and forsterite profiles in olivine crystals demonstrates the
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31 513 remarkably fast ascent of the FS magma. The forsterite profiles require that the magma
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33 514 ascended from near Moho depths to the surface in less than 1–5 days. Modeling H₂O profiles,
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35 515 which are sensitive to the region in which H₂O is degassing from the magma (<15 km depth),
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37 516 gives average decompression rates of 0.47 MPa/s or 17.5 m/s (assuming a lithostatic pressure
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39 517 gradient).

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43 518 We propose that this eruption may record a link between the primary magma
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45 519 composition and eruption style; the unusually explosive nature of this picritic eruption was
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47 520 driven by the high CO₂ concentrations, imparted to the magma by subducted carbonate in the
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49 521 mantle source. This CO₂ caused deep exsolution of a vapor phase which may have initiated
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51 522 fast diking all the way from the Moho to the surface, relatively uninterrupted by fractionation
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53 523 and mixing in the crust. The magma's deep origin, high volatile contents, and consequently
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55 524 rapid ascent, all acted to increase the explosive intensity of this eruption.
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525 **Acknowledgements**

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532 by NSF grant number EAR-1524542 to TP.

533 **Figures**



534
535 *Figure 1a) Rb/Nb ratios over time highlighting uniqueness of FS eruption. Note the log scale of time, in order to show data*
536 *from recent eruptions clearly. b) Height and age of Etna eruption deposits in stratigraphic section showing doubling of bulk*
537 *sedimentation rate after the Fall Stratified Eruption. Also note hiatus of large eruptions for the ~1000 years prior to FS. Data*
538 *and references listed in supplementary data table.*

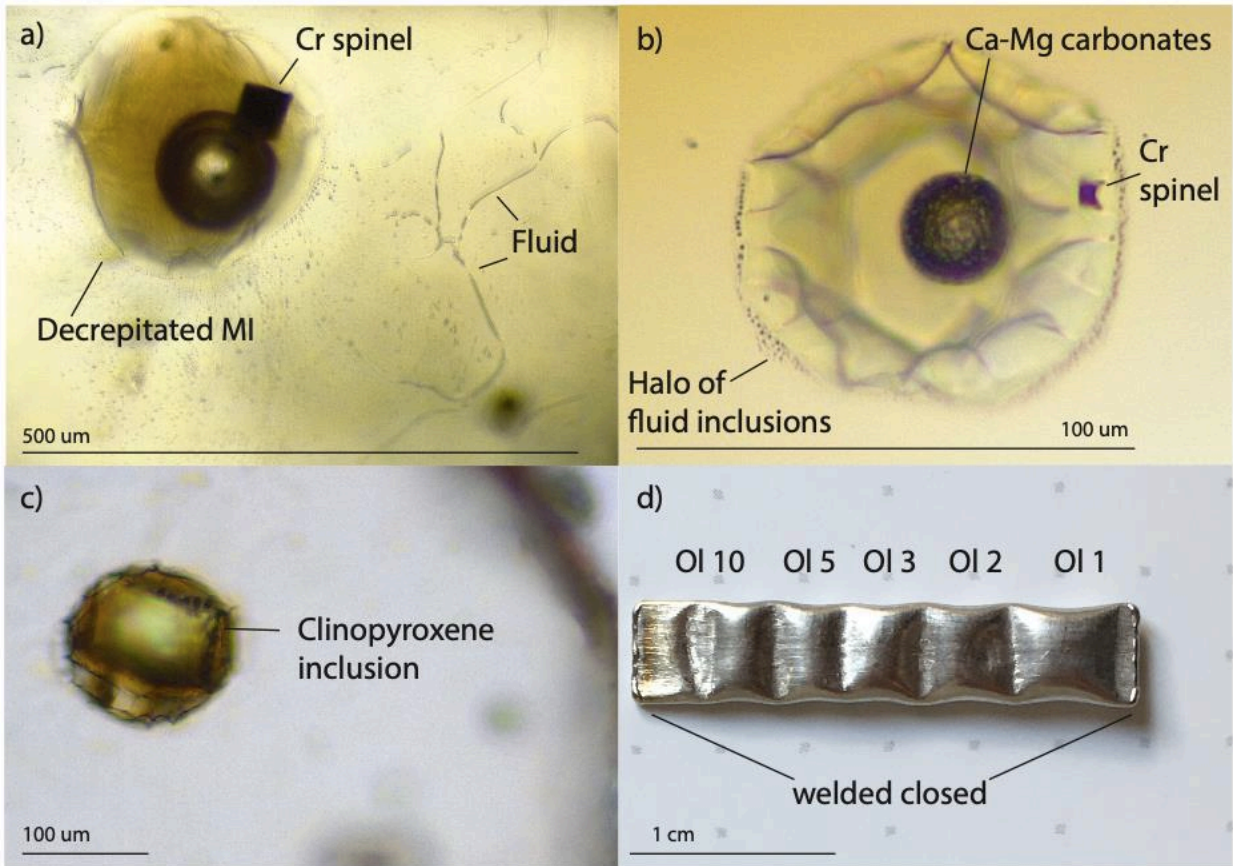
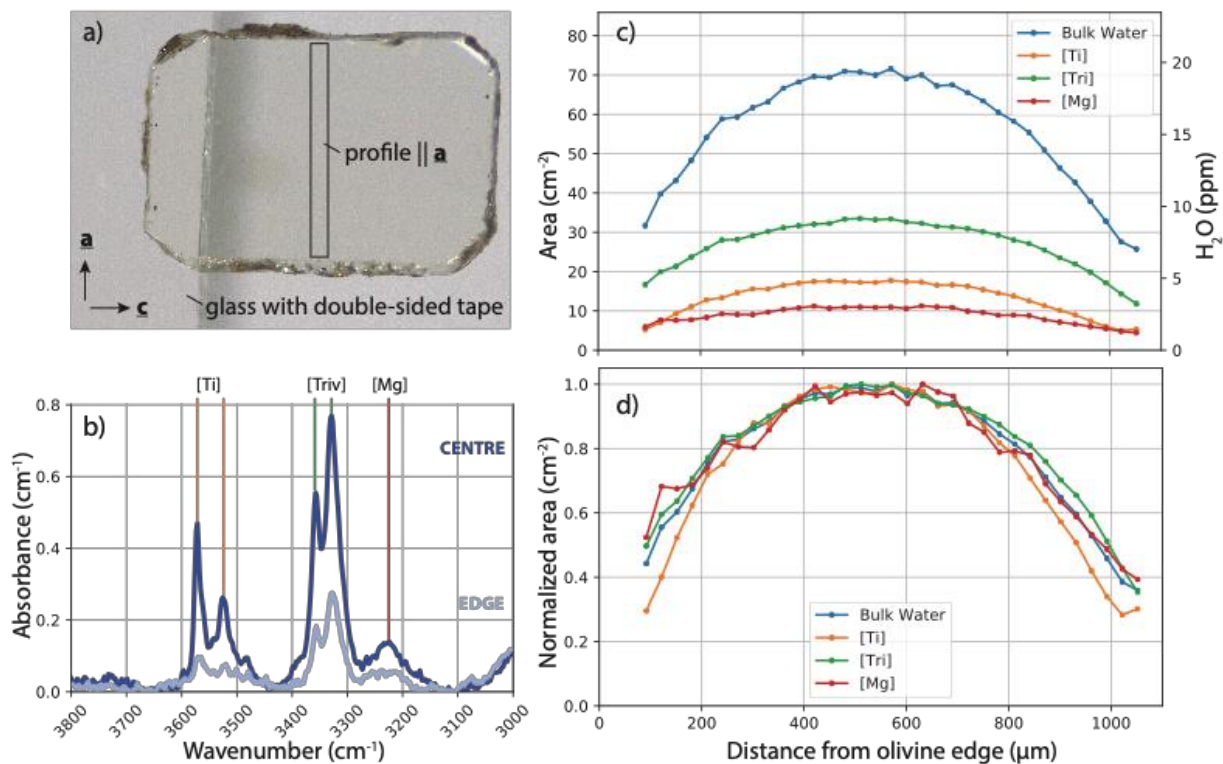


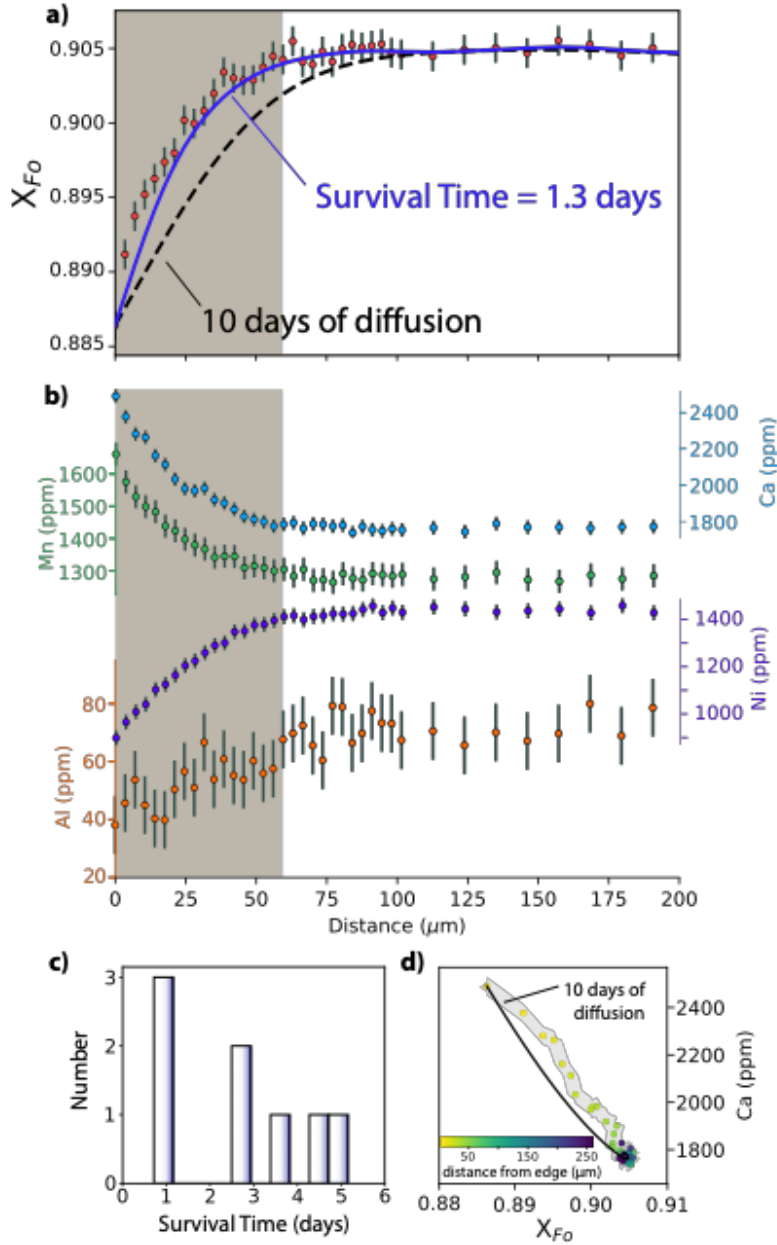
Figure 2. (a) Decrepitated MI containing a Cr-spinel with fluid inclusions radiating away from the MI along a planar crack. Note the large bubble size, which is a feature of these decrepitated MIs. (b) Typical Etina FS MI containing tiny Cr-spinel and vapor bubble lined with Ca-Mg carbonates. Photo is looking down b. The halo of tiny fluid inclusions around the MI is ubiquitous in this sample. (c) Inclusion of melt and clinopyroxene within olivine. (d) capsule for IHPV experiments. Individual olivines are separated by gently crimping around them in order to compare before and after photos.

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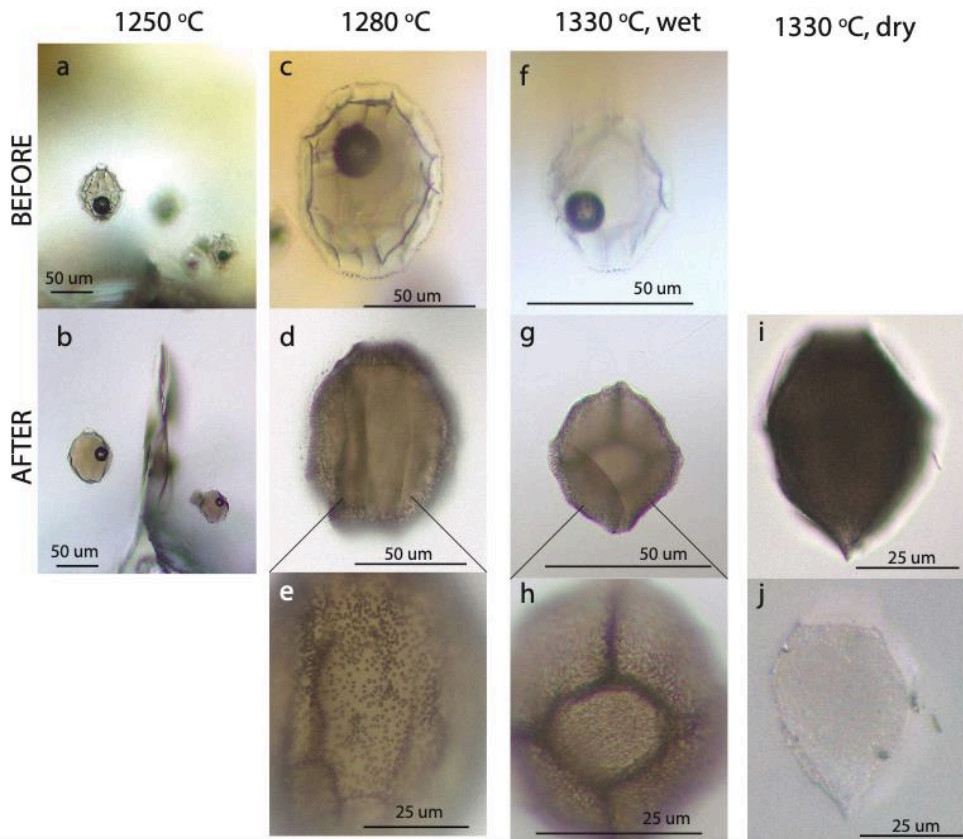


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 Figure 3. H₂O concentration profile in olivine along the crystallographic **a** direction measured with FTIR. (a) Photograph of olivine wafer taken down **b** [010]. (b) FTIR spectra with baseline subtracted taken in the centre (dark blue) and edge (light blue) of the olivine. Absorption peaks represent OH⁻ bonds associated with different point defects. Bulk H⁺ concentration estimated by summing area under spectra from 3600 – 3100 cm⁻¹. (c) Defect-specific profiles along crystallographic **a** direction. Area is converted to H₂O concentration using the Withers et al. (2012) calibration and multiplying by a correction factor to account for the light which would be absorbed in the direction of the raypath. (d) Normalised defect-specific profiles.

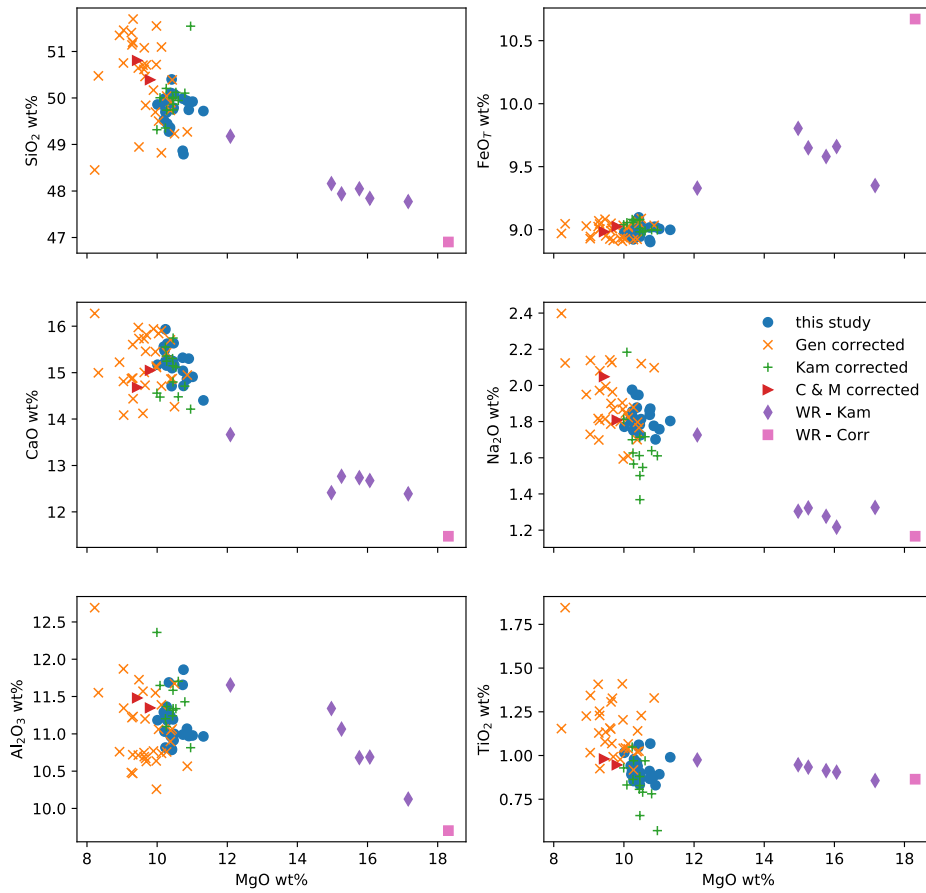
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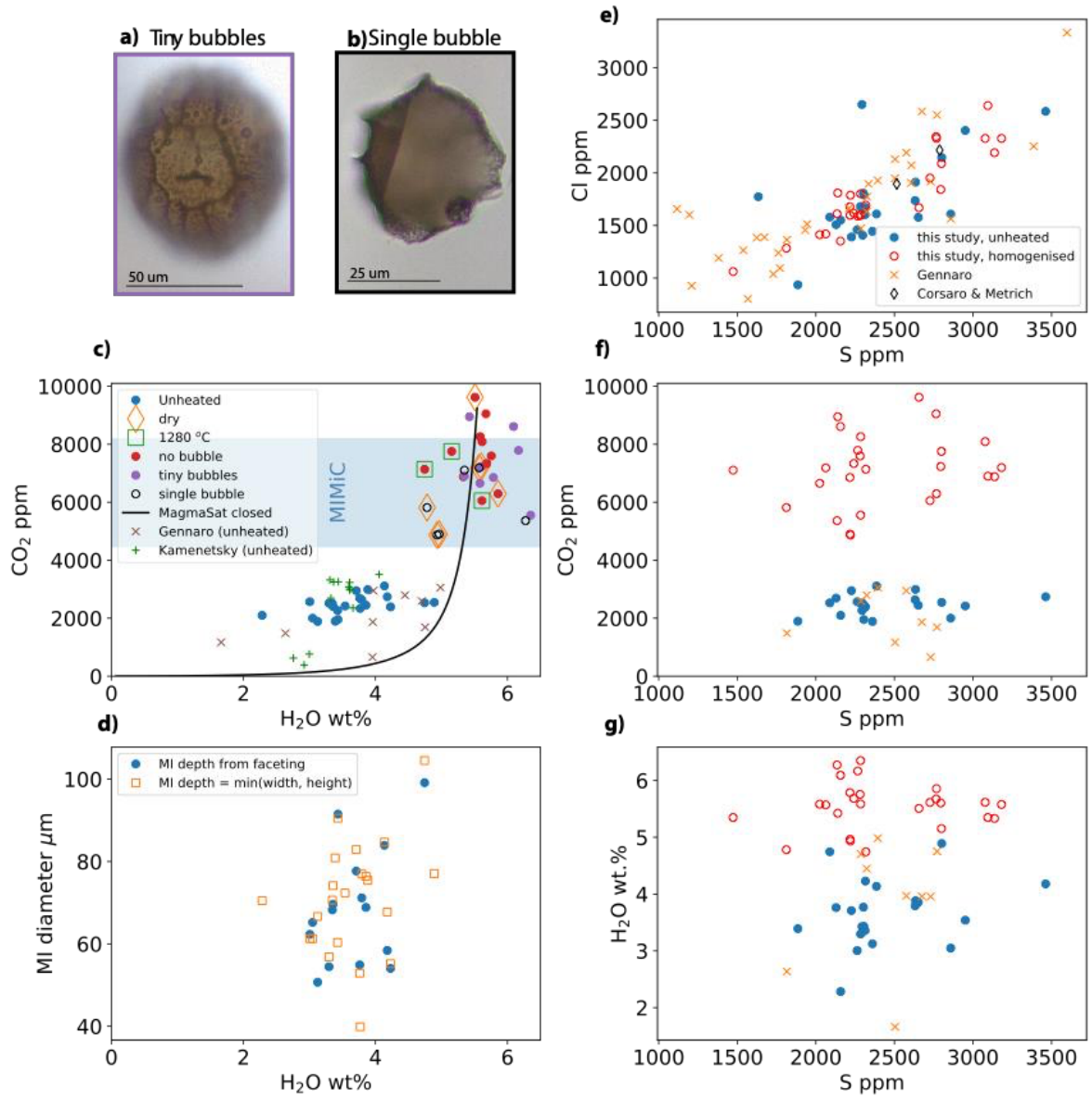
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 Figure 4. a, b) Major and trace element diffusion profiles in one olivine (O152) along [001], the fast direction for Fe-Mg diffusion. Note the similar lengthscales of zonation profiles for all elements (shaded region), regardless of their diffusivity. c) histogram of survival times based on forsterite zonation profiles (see text for description of calculation). d) Initial condition for Fo diffusion modeling calculated from linear relationship between Fo and Ca. Black reference curve in a) and c) shows effect of 10 days of diffusion.



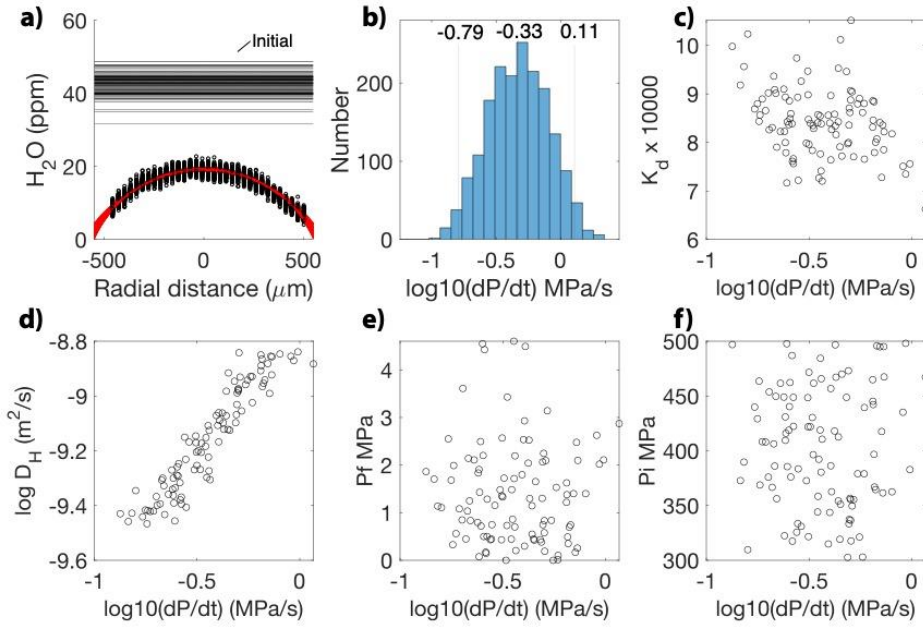
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 Figure 5. Before and after photographs of MIs at different experimental temperatures and conditions. There was no 'before' photo for the MI heated in dry capsule at 1330 °C. (a) and (b) show photos of MIs before and after heating at 1250 °C under wet conditions. Note the MI color becomes darker, the faceting less distinct, the carbonate lining the bubble dissolves, and the bubble shrinks and moves. (c) and (d) show photos for MI before and after heating at 1280 °C under wet conditions. Bubble is fully dissolved and MI interior is glassy. (e) Photograph focused on bottom surface of MI and zoomed in to show tiny bubbles and/or crystals lining MI wall. (f-h) are same as (c-e) but for 1330 °C. (i) and (j) show transmitted and reflected photographs of MI heated at 1330 °C under dry conditions. MI becomes dark and speckles can be seen in reflected light throughout interior of MI.



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 570 *Figure 6. PEC-corrected major element composition of unheated MIs from this study (blue circles). Data from literature for*
 571 *Etna FS shown for comparison (Gennaro et al. (2019), orange crosses; Kamenetsky et al. (2007), green plus; Corsaro and*
 572 *Métrich (2016), red triangle). Whole rock (WR) from Kamenetsky et al., 2007 (purple diamond) and Correale et al., 2014*
 573 *(red square) and shows variable degrees of crystal accumulation. All data is normalized to 100% volatile-free. Note that*
 574 *data for other MI studies was corrected for PEC using MIMiC with the same parameters as the data in this study (see*
 575 *supplementary data table).*

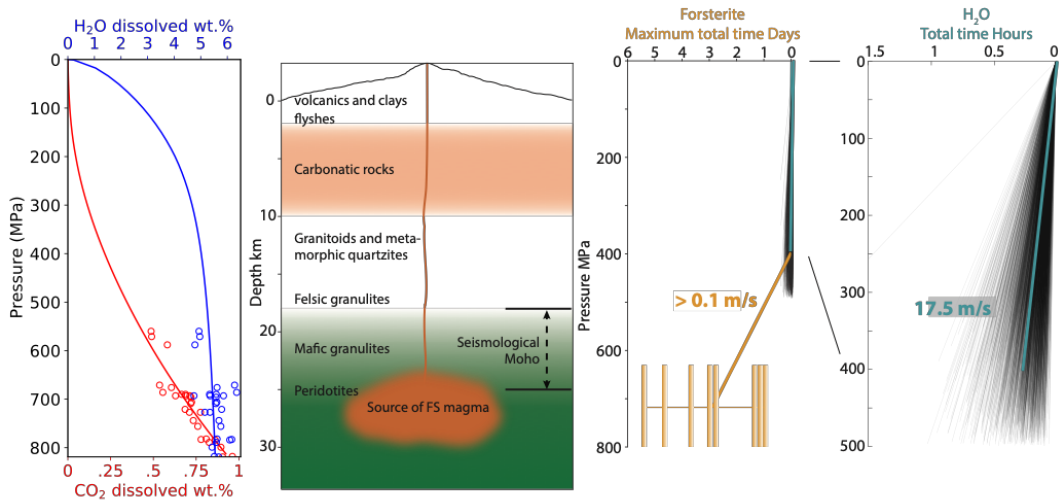


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 577 *Figure 7. Volatiles in homogenized and unheated MIs. (a, b) Examples of MIs with a single vapor bubble and multiple tiny*
 578 *bubbles. (c) H₂O versus CO₂ for homogenized and unheated MIs. Dry experiments shown with orange diamonds.*
 579 *Experiments at 1280 °C shown with green squares (all others are at 1330 °C). MIs with a single vapor bubble (open circles)*
 580 *plot to lower CO₂ concentrations. However, MIs with multiple tiny bubbles (purple circles) do not have less CO₂ than those*
 581 *without (red circles). Note that unheated MIs show scattered positive trend. CO₂ predicted by MIMiC modeling of unheated*
 582 *MIs (shaded blue) is similar to measurements of homogenized MIs. MagmaSat closed system degassing trend used in*
 583 *diffusion modeling of H⁺ in olivine profiles is overlain. (d) H₂O versus MI diameter measured from photographs in two ways*
 584 *for unheated MIs. The first method is detailed in the supplement and uses MI faceting to predict MI depth from*
 585 *photographs taken down the **b** axis (blue circles). This method is only possible for MIs which were photographed down the **b***
 586 *axis. The second method assumes that MI depth is equal to the shortest MI dimension measured in the photograph (orange*
 587 *squares). (e) Cl versus S measured by EPMA. Note the strong, approximately linear, correlation between S and Cl in all three*
 588 *studies (Gennaro et al., 2019; Corsaro and Metrich, 2016; this study), and for both unheated and homogenized MIs. (f) CO₂*
 589 *versus S, symbols same as in (e). CO₂ measured by FTIR for unheated MIs and SIMS for homogenized MIs. (g) H₂O versus S,*
 590 *symbols same as in (e). Note that for data from Gennaro et al., 2019, only H₂O and CO₂ determined by FTIR are shown.*



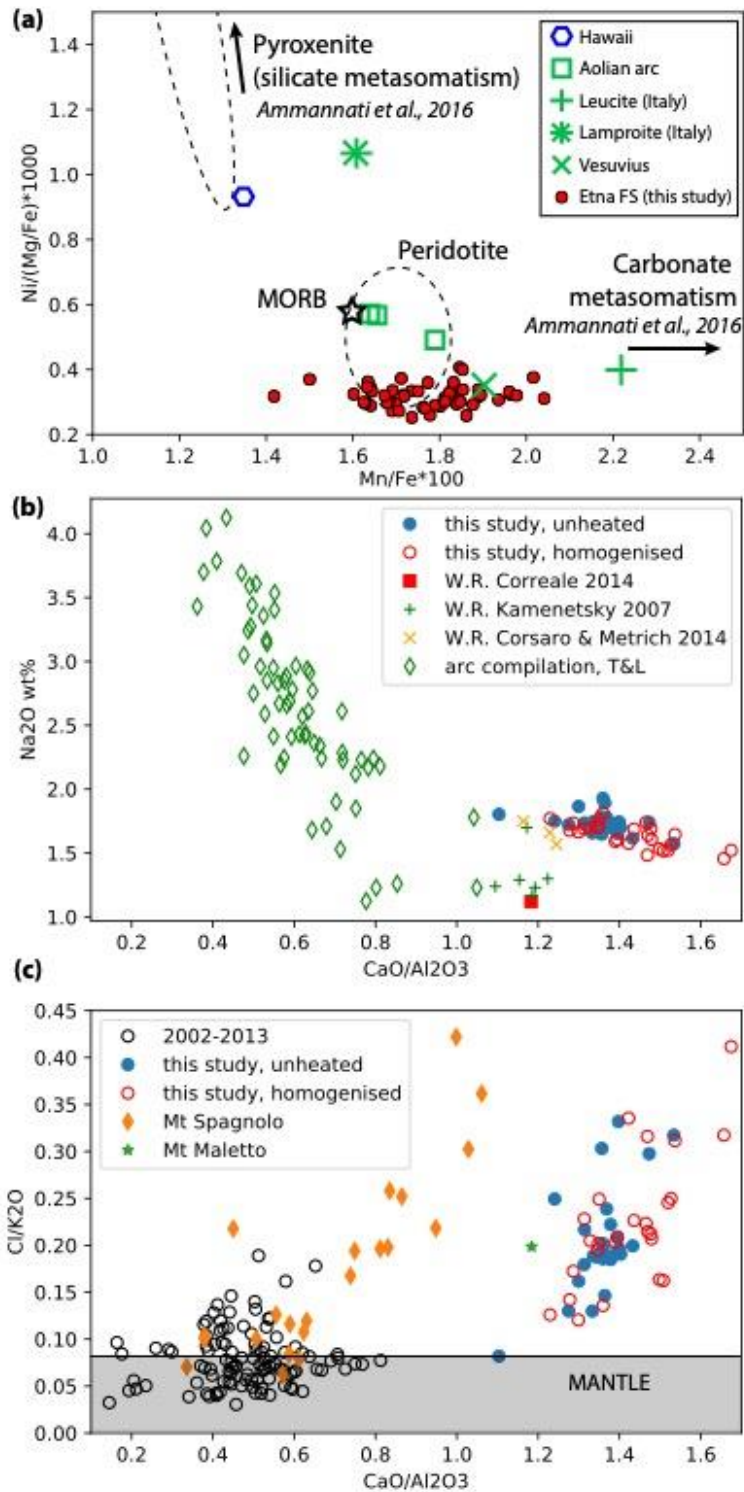
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Figure 8. Example Monte Carlo run to constrain decompression rate and final pressure by modeling 1D H^+ diffusion along crystallographic a direction. (a) FTIR profile with noise added in black circles, model fits (red lines). Initial H_2O concentration chosen from normal distribution with mean of 42 ppm and standard deviation of 3 ppm (see text for details). (b) Histogram of best-fit decompression rates for all Monte Carlo runs for different olivines. Vertical lines show mean and 95 % confidence intervals (c) Partition coefficient, K_d , versus decompression rate. K_d is calculated as the ratio of initial H_2O concentration in olivine (shown in (a)) relative to magma, which is set by MagmaSat solubility model and initial pressure, P_i . (d) Diffusivity versus decompression rate shows a clear positive correlation – most of the uncertainty in decompression rate comes from uncertainty in diffusivity. Upper bound of diffusivity is set by fastest dehydrating rate measured (Ferriss et al., 2018), Lower bound is set by initial dehydration rate measured in Barth et al. (2023). Final, P_f , (e) and initial, P_i , (f) pressure versus decompression rate show no relationship. Range of initial pressure chosen as the region in which H_2O begins significant degassing



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Figure 9. a) Degassing path modeled with MagmaSat for FS magma with MI saturation pressures determined with MagmaSat (Ghiorso & Gualda, 2015) overlain in blue (H_2O) and red (CO_2) circles. Crustal section beneath Etna with lithologies from Corsaro and Pompilio 2004. Likely range of Moho depth from seismological observations shown by dashed arrow (see text for references). c) Maximum ascent times from storage to surface calculated from forsterite zonation modeling shown as vertical yellow bars. Vertical span of yellow bars represents uncertainty in storage pressure from range of MI volatile contents. Average ascent rate is 0.1 m/s and is shown by bold yellow line. Note that each bar represents a maximum, not best-fit, ascent time for one olivine. Ascent rate from H_2O Monte Carlo diffusion modeling shown by black lines, with mean ascent rate overlain in blue d) Zoom-in of H_2O diffusion model. Each black line represents results from a single Monte Carlo run. Average ascent rate is 17.5 m/s and is shown by bold blue line.



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614 *Figure 10. Evidence for carbonate in FS magma source (a,b) and involvement of subducting slab(c). (a) Olivine compositions*
615 *from FS have low Ni concentrations, prohibiting a large role for pyroxenite in its source. Compositions for other Italian*
616 *magmas (green) from Zamboni et al., 2017. Also shown are the compositions for peridotite and pyroxenite, as well as the*
617 *expected trend for silicate and carbonate mantle metasomatism from Ammannati et al., 2016. (b) Whole rock and melt*
618 *inclusions (this study) of the FS magma is outside of the global arc array (from samples with $Mg\# > 60$, Turner and*
619 *Langmuir, 2015) at low Na_2O and high CaO/Al_2O_3 . (c) Comparison of Cl/K_2O for modern eruptions and three primitive*
620 *eruptions at Etna (Mt. Spagnolo – Gennaro et al., 2019, Mt Maletto – Schiano et al., 2001, and Fall Stratified – this study).*
621 *Modern eruptions have only moderately higher Cl/K_2O ratios compared with MORB, and individual eruptions do not show a*
622 *correlation between CaO/Al_2O_3 versus Cl/K_2O . In contrast, all three primitive eruptions have high Cl/K_2O and CaO/Al_2O_3 , and*
623 *Mt. Spagnolo and FS show positive correlation between these ratios.*

624 **References**

- 1 625
2 626 Accaino, F., Catalano, R., Di Marzo, L., Giustiniani, M., Tinivella, U., Nicolich, R., Sulli, A.,
3 627 Valenti, V., & Manetti, P. (2011). A crustal seismic profile across Sicily.
4 628 *Tectonophysics*, 508(1–4), 52–61.
5 629 Allison, C. M., Roggensack, K., & Clarke, A. B. (2019). H₂O–CO₂ solubility in alkali-rich
6 630 mafic magmas: new experiments at mid-crustal pressures. *Contributions to Mineralogy
7 631 and Petrology*, 174(7), 1–24.
8 632 Allison, C. M., Roggensack, K., & Clarke, A. B. (2021). Highly explosive basaltic eruptions
9 633 driven by CO₂ exsolution. *Nature Communications*, 12(1), 1–10.
10 634 <https://doi.org/10.1038/s41467-020-20354-2>
11 635 Allison, C. M., Roggensack, K., & Clarke, A. B. (2022). MafiCH: a general model for H₂O–
12 636 CO₂ solubility in mafic magmas. *Contributions to Mineralogy and Petrology*, 177(3),
13 637 1–22. <https://doi.org/10.1007/s00410-022-01903-y>
14 638 Ammannati, E., Jacob, D. E., Avanzinelli, R., Foley, S. F., & Conticelli, S. (2016). Low Ni
15 639 olivine in silica-undersaturated ultrapotassic igneous rocks as evidence for carbonate
16 640 metasomatism in the mantle. *Earth and Planetary Science Letters*, 444, 64–74.
17 641 <https://doi.org/10.1016/j.epsl.2016.03.039>
18 642 Anderson, A. T. (1995). CO₂ and the eruptibility of picrite and komatiite. *Lithos*, 34(1–3),
19 643 19–25.
20 644 Anderson JR, A. T. (1974). Evidence for a picritic, volatile-rich magma beneath Mt. Shasta,
21 645 California. *Journal of Petrology*, 15(2), 243–267.
22 646 Bamber, E. C., Arzilli, F., Polacci, M., Hartley, M. E., Fellowes, J., Di Genova, D.,
23 647 Chavarría, D., Saballos, J. A., & Burton, M. R. (2020). Pre-and syn-eruptive conditions
24 648 of a basaltic Plinian eruption at Masaya Volcano, Nicaragua: the Masaya Triple Layer
25 649 (2.1 ka). *Journal of Volcanology and Geothermal Research*, 392, 106761.
26 650 Barth, A., Newcombe, M., Plank, T., Gonnermann, H., Hajimirza, S., Soto, G. J., Saballos,
27 651 A., & Hauri, E. (2019). Magma decompression rate correlates with explosivity at
28 652 basaltic volcanoes—Constraints from water diffusion in olivine. *Journal of Volcanology
29 653 and Geothermal Research*, 387, 106664.
30 654 Barth, A., Plank, T., & Towbin, H. (2023). Rates of dehydration in hydrous, high-Fo,
31 655 magmatic olivines. *Geochimica et Cosmochimica Acta*, 342, 62–73.
32 656 Branca, S., & Del Carlo, P. (2005). Types of eruptions of Etna volcano AD 1670–2003:
33 657 implications for short-term eruptive behaviour. *Bulletin of Volcanology*, 67(8), 732–742.
34 658 Carnevale, G., Caracausi, A., Rotolo, S. G., Paternoster, M., & Zanon, V. (2022). New
35 659 Inferences on Magma Dynamics in Melilitite-Carbonatite Volcanoes: The Case Study of
36 660 Mt. Vulture (Southern Italy). *Geophysical Research Letters*, 49(21).
37 661 <https://doi.org/10.1029/2022GL099075>
38 662 Cassidy, M., Manga, M., Cashman, K., & Bachmann, O. (2018). Controls on explosive-
39 663 effusive volcanic eruption styles. *Nature Communications*, 9(1).
40 664 <https://doi.org/10.1038/s41467-018-05293-3>
41 665 Coltelli, M., Del Carlo, P., Pompilio, M., & Vezzoli, L. (2005). Explosive eruption of a
42 666 picrite: The 3930 BP subplinian eruption of Etna volcano (Italy). *Geophysical Research
43 667 Letters*, 32(23), 1–4. <https://doi.org/10.1029/2005GL024271>
44 668 Coltelli, M., Del Carlo, P., & Vezzoli, L. (1998). Discovery of a Plinian basaltic eruption of
45 669 Roman age at Etna volcano, Italy. *Geology*, 26(12), 1095–1098.
46 670 [https://doi.org/10.1130/0091-7613\(1998\)026<1095:DOAPBE>2.3.CO;2](https://doi.org/10.1130/0091-7613(1998)026<1095:DOAPBE>2.3.CO;2)
47 671 Coltelli, M., Del Carlo, P., & Vezzoli, L. (2000). Stratigraphic constraints for explosive
48 672 activity in the past 100 ka at Etna Volcano, Italy. *International Journal of Earth
49 673 Sciences*, 89(3), 665–677. <https://doi.org/10.1007/s005310000117>
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

- 674 Correale, A., Paonita, A., Martelli, M., Rizzo, A., Rotolo, S. G., Anna, R., & Di, V. (2014).
1 675 Lithos A two-component mantle source feeding Mt. Etna magmatism: Insights from
2 676 the geochemistry of primitive magmas. *LITHOS*, 184–187, 243–258.
3 677 <https://doi.org/10.1016/j.lithos.2013.10.038>
- 4 678 Corsaro, R. A., Andronico, D., Behncke, B., Branca, S., Caltabiano, T., Ciancitto, F.,
5 679 Cristaldi, A., De Beni, E., La Spina, A., Lodato, L., Miraglia, L., Neri, M., Salerno, G.,
6 680 Scollo, S., & Spata, G. (2017). Monitoring the December 2015 summit eruptions of Mt.
7 681 Etna (Italy): Implications on eruptive dynamics. *Journal of Volcanology and*
8 682 *Geothermal Research*, 341(December 2015), 53–69.
9 683 <https://doi.org/10.1016/j.jvolgeores.2017.04.018>
- 10 684 Corsaro, R. A., & Métrich, N. (2016). Chemical heterogeneity of Mt. Etna magmas in the last
11 685 15 ka. Inferences on their mantle sources. *Lithos*, 252–253, 123–134.
12 686 <https://doi.org/10.1016/j.lithos.2016.02.006>
- 13 687 Corsaro, R. A., & Pompilio, M. (2004). Buoyancy-controlled eruption of magmas at Mt Etna.
14 688 *Terra Nova*, 16(1), 16–22.
- 15 689 Currenti, G., Del Negro, C., & Ganci, G. (2007). Modelling of ground deformation and
16 690 gravity fields using finite element method: An application to Etna volcano. *Geophysical*
17 691 *Journal International*, 169(2), 775–786. [https://doi.org/10.1111/j.1365-](https://doi.org/10.1111/j.1365-246X.2007.03380.x)
18 692 [246X.2007.03380.x](https://doi.org/10.1111/j.1365-246X.2007.03380.x)
- 19 693 Danyushevsky, L. V., Leslie, R. A. J., Crawford, A. J., & Durance, P. (2004). Melt inclusions
20 694 in primitive olivine phenocrysts: the role of localized reaction processes in the origin of
21 695 anomalous compositions. *Journal of Petrology*, 45(12), 2531–2553.
- 22 696 Dasgupta, R., Hirschmann, M. M., & Smith, N. D. (2007). Partial melting experiments of
23 697 peridotite+ CO₂ at 3 GPa and genesis of alkalic ocean island basalts. *Journal of*
24 698 *Petrology*, 48(11), 2093–2124.
- 25 699 Davis, T., Rivalta, E., Smittarello, D., & Katz, R. F. (2023). Ascent rates of 3-D fractures
30 700 driven by a finite batch of buoyant fluid. *Journal of Fluid Mechanics*, 954, 1–23.
31 701 <https://doi.org/10.1017/jfm.2022.986>
- 32 702 Dellong, D., Klingelhoefer, F., Kopp, H., Graindorge, D., Margheriti, L., Moretti, M.,
33 703 Murphy, S., & Gutscher, M. A. (2018). Crustal Structure of the Ionian Basin and Eastern
34 704 Sicily Margin: Results From a Wide-Angle Seismic Survey. *Journal of Geophysical*
35 705 *Research: Solid Earth*, 123(3), 2090–2114. <https://doi.org/10.1002/2017JB015312>
- 36 706 Di Genova, D., Romano, C., Alletti, M., Misiti, V., & Scarlato, P. (2014). The effect of CO₂
37 707 and H₂O on Etna and Fondo Riccio (Phlegrean Fields) liquid viscosity, glass transition
38 708 temperature and heat capacity. *Chemical Geology*, 377, 72–86.
- 39 709 Ding, S., Plank, T., Wallace, P. J., & Rasmussen, D. J. (2023). Sulfur_X: A model of sulfur
40 710 degassing during magma ascent. *Geochemistry, Geophysics, Geosystems*, 24(4),
41 711 e2022GC010552.
- 42 712 Dixon, J. E. (1997). Degassing of alkalic basalts. *American Mineralogist*, 82(3–4), 368–378.
- 43 713 Dixon, J. E., Leist, L., Langmuir, C., & Schilling, J.-G. (2002). Recycled dehydrated
44 714 lithosphere observed in plume-influenced mid-ocean-ridge basalt. *Nature*, 420(6914),
45 715 385–389. <https://doi.org/10.1038/nature01215>
- 46 716 Dohmen, R., & Chakraborty, S. (2007). Fe-Mg diffusion in olivine II: Point defect chemistry,
47 717 change of diffusion mechanisms and a model for calculation of diffusion coefficients in
48 718 natural olivine. *Physics and Chemistry of Minerals*, 34(6), 409–430.
49 719 <https://doi.org/10.1007/s00269-007-0158-6>
- 50 720 Duan, X. (2014). A general model for predicting the solubility behavior of H₂O–CO₂ fluids
51 721 in silicate melts over a wide range of pressure, temperature and compositions.
52 722 *Geochimica et Cosmochimica Acta*, 125, 582–609.
- 53 723 Esposito, R., Lamadrid, H. M., Redi, D., Steele-MacInnis, M., Bodnar, R. J., Manning, C. E.,
54
55
56
57
58
59
60
61
62
63
64
65

- 724 De Vivo, B., Cannatelli, C., & Lima, A. (2016). Detection of liquid H₂O in vapor
725 bubbles in reheated melt inclusions: Implications for magmatic fluid composition and
726 volatile budgets of magmas? *American Mineralogist*, *101*(7), 1691–1695.
- 727 Gennaro, E., Grassa, F., Martelli, M., Renzulli, A., & Rizzo, A. L. (2017). Carbon isotope
728 composition of CO₂-rich inclusions in cumulate-forming mantle minerals from
729 Stromboli volcano (Italy). *Journal of Volcanology and Geothermal Research*, *346*, 95–
730 103. <https://doi.org/https://doi.org/10.1016/j.jvolgeores.2017.04.001>
- 731 Gennaro, E., Iacono-Marziano, G., Paonita, A., Rotolo, S. G., Martel, C., Rizzo, A. L.,
732 Pichavant, M., & Liotta, M. (2019). Melt inclusions track melt evolution and degassing
733 of Etnean magmas in the last 15 ka. *Lithos*, *324–325*, 716–732.
734 <https://doi.org/10.1016/j.lithos.2018.11.023>
- 735 Gennaro, E., Paonita, A., Iacono-Marziano, G., Moussallam, Y., Pichavant, M., Peters, N., &
736 Martel, C. (2020). Sulphur behaviour and redox conditions in etnean magmas during
737 magma differentiation and degassing. *Journal of Petrology*.
738 <https://doi.org/10.1093/petrology/egaa095>
- 739 Ghiorso, M. S., & Gualda, G. A. R. (2015). An H₂O–CO₂ mixed fluid saturation model
740 compatible with rhyolite-MELTS. *Contributions to Mineralogy and Petrology*, *169*(6),
741 1–30.
- 742 Giordano, D., & Dingwell, D. (2003). Viscosity of hydrous Etna basalt: implications for
743 Plinian-style basaltic eruptions. *Bulletin of Volcanology*, *65*(1), 8–14.
744 <https://doi.org/10.1007/s00445-002-0233-2>
- 745 Giordano, D., Russell, J. K., & Dingwell, D. B. (2008). Viscosity of magmatic liquids: A
746 model. *Earth and Planetary Science Letters*, *271*(1–4), 123–134.
747 <https://doi.org/10.1016/j.epsl.2008.03.038>
- 748 Giuffrida, M., Viccaro, M., & Ottolini, L. (2018). Ultrafast syn-eruptive degassing and ascent
749 trigger high-energy basic eruptions. *Scientific Reports*, *8*(1), 1–7.
750 <https://doi.org/10.1038/s41598-017-18580-8>
- 751 Gvirtzman, Z., & Nur, A. (1999). The formation of Mount Etna as the consequence of slab
752 rollback. *Nature*, *401*(6755), 782–785. <https://doi.org/10.1038/44555>
- 753 Houghton, B. F., & Gonnermann, H. M. (2008). Basaltic explosive volcanism: Constraints
754 from deposits and models. *Chemie Der Erde - Geochemistry*, *68*(2), 117–140.
755 <https://doi.org/http://dx.doi.org/10.1016/j.chemer.2008.04.002>
- 756 Houghton, B. F., Wilson, C. J. N., Del Carlo, P., Coltelli, M., Sable, J. E., & Carey, R.
757 (2004). The influence of conduit processes on changes in style of basaltic Plinian
758 eruptions: Tarawera 1886 and Etna 122 BC. *Journal of Volcanology and Geothermal*
759 *Research*, *137*(1-3 SPEC. ISS.), 1–14. <https://doi.org/10.1016/j.jvolgeores.2004.05.009>
- 760 Iacono-Marziano, G., Morizet, Y., Le Trong, E., & Gaillard, F. (2012). New experimental
761 data and semi-empirical parameterization of H₂O–CO₂ solubility in mafic melts.
762 *Geochimica et Cosmochimica Acta*, *97*, 1–23.
- 763 Iacovino, K., Matthews, S., Wieser, P. E., Moore, G. M., & Bégué, F. (2021). VESICAL Part I:
764 An open-source thermodynamic model engine for mixed volatile (H₂O–CO₂) solubility
765 in silicate melts. *Earth and Space Science*, *8*(11), e2020EA001584.
- 766 Kamenetsky, V., & Clocchiatti, R. (1996). Primitive magmatism of Mt. Etna: Insights from
767 mineralogy and melt inclusions. *Earth and Planetary Science Letters*, *142*(3–4), 553–
768 572. [https://doi.org/10.1016/0012-821x\(96\)00115-x](https://doi.org/10.1016/0012-821x(96)00115-x)
- 769 Kamenetsky, V., Pompilio, M., Métrich, N., Sobolev, A. V., Kuzmin, D. V., & Thomas, R.
770 (2007). Arrival of extremely volatile-rich high-Mg magmas changes explosivity of
771 Mount Etna. *Geology*, *35*(3), 255–258. <https://doi.org/10.1130/G23163A.1>
- 772 Lara, M., & Dasgupta, R. (2022). Carbon recycling efficiency in subduction zones
773 constrained by the effects of H₂O–CO₂ fluids on partial melt compositions in the mantle

- 774 wedge. *Earth and Planetary Science Letters*, 588.
1 775 <https://doi.org/10.1016/j.epsl.2022.117578>
- 2 776 Lerner, A. H., Muth, M. J., Wallace, P. J., Lanzirotti, A., Newville, M., Gaetani, G. A.,
3 777 Chowdhury, P., & Dasgupta, R. (2021). Improving the reliability of Fe- and S-XANES
4 778 measurements in silicate glasses: Correcting beam damage and identifying Fe-oxide
5 779 nanolites in hydrous and anhydrous melt inclusions. *Chemical Geology*, 586, 120610.
6 780 <https://doi.org/https://doi.org/10.1016/j.chemgeo.2021.120610>
- 7 781 McPhie, J., Walker, G. P. L., & Christiansen, R. L. (1990). Phreatomagmatic and phreatic fall
8 782 and surge deposits from explosions at Kilauea volcano, Hawaii, 1790 AD: Keanakakoi
9 783 Ash Member. *Bulletin of Volcanology*, 52, 334–354.
- 10 784 Mironov, N., Portnyagin, M., Botcharnikov, R., Gurenko, A., Hoernle, K., & Holtz, F.
11 785 (2015). Quantification of the CO₂ budget and H₂O-CO₂ systematics in subduction-zone
12 786 magmas through the experimental hydration of melt inclusions in olivine at high H₂O
13 787 pressure. *Earth and Planetary Science Letters*, 425, 1–11.
14 788 <https://doi.org/10.1016/j.epsl.2015.05.043>
- 15 789 Moore, L. R., Gazel, E., Tuohy, R., Lloyd, A. S., Esposito, R., Steele-MacInnis, M., Hauri, E.
16 790 H., Wallace, P. J., Plank, T., & Bodnar, R. J. (2015). Bubbles matter: An assessment of
17 791 the contribution of vapor bubbles to melt inclusion volatile budgets. *American*
18 792 *Mineralogist*, 100(4), 806–823. <https://doi.org/10.2138/am-2015-5036>
- 19 793 Moussallam, Y., Médard, E., Georgeais, G., Rose-Koga, E. F., Koga, K. T., Pelletier, B.,
20 794 Bani, P., Shreve, T. L., Grandin, R., Boichu, M., Tari, D., & Peters, N. (2021). How to
21 795 turn off a lava lake? A petrological investigation of the 2018 intra-caldera and
22 796 submarine eruptions of Ambrym volcano. *Bulletin of Volcanology*, 83(5), 36.
23 797 <https://doi.org/10.1007/s00445-021-01455-2>
- 24 798 Moussallam, Y., Rose-Koga, E. F., Koga, K. T., Médard, E., Bani, P., Devidal, J.-L., & Tari,
25 799 D. (2019). Fast ascent rate during the 2017–2018 Plinian eruption of Ambae (Aoba)
26 800 volcano: a petrological investigation. *Contributions to Mineralogy and Petrology*,
27 801 174(11), 90. <https://doi.org/10.1007/s00410-019-1625-z>
- 28 802 Mutch, E. J. F., Maclennan, J., Shorttle, O., Edmonds, M., & Rudge, J. F. (2019). Rapid
29 803 transcrustal magma movement under Iceland. *Nature Geoscience*, 12(7), 569–574.
- 30 804 Mutch, E. J. F., Maclennan, J., Shorttle, O., Rudge, J. F., & Neave, D. A. (2021). DFENS:
31 805 Diffusion chronometry using finite elements and nested sampling. *Geochemistry,*
32 806 *Geophysics, Geosystems*, 22(4), e2020GC009303.
- 33 807 Namiki, A., & Manga, M. (2006). Influence of decompression rate on the expansion velocity
34 808 and expansion style of bubbly fluids. *Journal of Geophysical Research: Solid Earth*,
35 809 111(11), 1–17. <https://doi.org/10.1029/2005JB004132>
- 36 810 Newcombe, M. E., Plank, T., Barth, A., Asimow, P., & Hauri, E. (2020). Water-in-olivine
37 811 magma ascent chronometry: Every crystal is a clock. *Journal of Volcanology and*
38 812 *Geothermal Research*, 106872.
- 39 813 Nicolich, R., Laigle, M., Hirn, A., Cernobori, L., & Gallart, J. (2000). Crustal structure of the
40 814 ionian margin of Sicily: Etna volcano in the frame of regional evolution.
41 815 *Tectonophysics*, 329(1–4), 121–139. [https://doi.org/10.1016/S0040-1951\(00\)00192-X](https://doi.org/10.1016/S0040-1951(00)00192-X)
- 42 816 Parfitt, E. A., & Wilson, L. (1994). Modelling the transition between Hawaiiin-style lava
43 817 fountaining and Strombolian explosive volcanic activity. *Abstracts of the 25th Lunar*
44 818 *and Planetary Science Conference*, 1 995, 1049–1050.
45 819 <http://adsabs.harvard.edu/full/1994LPI....25.1049P>
- 46 820 Pichavant, M., Di Carlo, I., Rotolo, S. G., Scaillet, B., Burgisser, A., Le Gall, N., & Martel,
47 821 C. (2013). Generation of CO₂-rich melts during basalt magma ascent and degassing.
48 822 *Contributions to Mineralogy and Petrology*, 166(2), 545–561.
49 823 <https://doi.org/10.1007/s00410-013-0890-5>
- 50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

- 824 Portnyagin, M., Hoernle, K., Plechov, P., Mironov, N., & Khubunaya, S. (2007). Constraints
1 825 on mantle melting and composition and nature of slab components in volcanic arcs from
2 826 volatiles (H₂O, S, Cl, F) and trace elements in melt inclusions from the Kamchatka Arc.
3 827 *Earth and Planetary Science Letters*, 255(1–2), 53–69.
- 4 828 Portnyagin, M. V., Mironov, N. L., Matveev, S. V., & Plechov, P. Y. (2005). Petrology of
5 829 avachites, high-magnesian basalts of Avachinsky Volcano, Kamchatka: II. Melt
6 830 inclusions in olivine. *PETROLOGY C/C OF PETROLOGIJA*, 13(4), 322.
- 7 831 Qin, Z., Lu, F., & Anderson, A. T. (1992). Diffusive reequilibration of melt and fluid
8 832 inclusions. *American Mineralogist*, 77(5–6), 565–576.
- 9 833 Rasmussen, D. J., Plank, T. A., Wallace, P. J., Newcombe, M. E., & Lowenstern, J. B.
10 834 (2020). Vapor-bubble growth in olivine-hosted melt inclusions. *American Mineralogist*,
11 835 105(12), 1898–1919.
- 12 836 Reed, M. H., Munoz-Saez, C., Hajimirza, S., Wu, S.-M., Barth, A., Girona, T., Rasht-
13 837 Behesht, M., White, E. B., Karplus, M. S., & Hurwitz, S. (2021). The 2018 reawakening
14 838 and eruption dynamics of Steamboat Geyser, the world’s tallest active geyser.
15 839 *Proceedings of the National Academy of Sciences*, 118(2).
- 16 840 Ruprecht, P., & Plank, T. (2013). Feeding andesitic eruptions with a high-speed connection
17 841 from the mantle. *Nature*, 500(7460), 68–72.
- 18 842 Schiano, P., Clocchiatti, R., Ottolini, L., & Busà, T. (2001). Transition of Mount Etna lavas
19 843 from a mantle-plume to an island-arc magmatic source [6]. *Nature*, 412(6850), 900–904.
20 844 <https://doi.org/10.1038/35091056>
- 21 845 Schiano, P., Eiler, J. M., Hutcheon, I. D., & Stolper, E. M. (2000). Primitive CaO-rich, silica-
22 846 undersaturated melts in island arcs: Evidence for the involvement of clinopyroxene-rich
23 847 lithologies in the petrogenesis of arc magmas. *Geochemistry, Geophysics, Geosystems*,
24 848 1(5). <https://doi.org/10.1029/1999gc000032>
- 25 849 Shea, T., Ruth, D., Jollands, M., Ohtaki, K., Ishii, H., & Bradley, J. (2023). The presence of
26 850 silicate melt may enhance rates of cation diffusion in olivine. *Earth and Planetary
27 851 Science Letters*, 621, 118370. <https://doi.org/10.1016/j.epsl.2023.118370>
- 28 852 Spandler, C., & O’Neill, H. S. C. (2010). Diffusion and partition coefficients of minor and
29 853 trace elements in San Carlos olivine at 1,300 C with some geochemical implications.
30 854 *Contributions to Mineralogy and Petrology*, 159, 791–818.
- 31 855 Steele-MacInnis, M., Esposito, R., Moore, L. R., & Hartley, M. E. (2017). Heterogeneously
32 856 entrapped, vapor-rich melt inclusions record pre-eruptive magmatic volatile contents.
33 857 *Contributions to Mineralogy and Petrology*, 172(4), 1–13.
34 858 <https://doi.org/10.1007/s00410-017-1343-3>
- 35 859 Tonarini, S., Armienti, P., D’Orazio, M., & Innocenti, F. (2001). Subduction-like fluids in the
36 860 genesis of Mt. Etna magmas: evidence from boron isotopes and fluid mobile elements.
37 861 *Earth and Planetary Science Letters*, 192(4), 471–483.
- 38 862 Tonarini, S., D’Orazio, M., Armienti, P., Innocenti, F., & Scribano, V. (1996). Geochemical
39 863 features of eastern Sicily lithosphere as probed by Hyblean xenoliths and lavas.
40 864 *European Journal of Mineralogy*, 8(5), 1153–1173.
- 41 865 Toplis, M. J. (2005). The thermodynamics of iron and magnesium partitioning between
42 866 olivine and liquid: Criteria for assessing and predicting equilibrium in natural and
43 867 experimental systems. *Contributions to Mineralogy and Petrology*, 149(1), 22–39.
44 868 <https://doi.org/10.1007/s00410-004-0629-4>
- 45 869 Trua, T., Esperança, S., & Mazzuoli, R. (1998). The evolution of the lithospheric mantle
46 870 along the N. African Plate: Geochemical and isotopic evidence from the tholeiitic and
47 871 alkaline volcanic rocks of the Hyblean plateau, Italy. *Contributions to Mineralogy and
48 872 Petrology*, 131(4), 307–322. <https://doi.org/10.1007/s004100050395>
- 49 873 Turner, S. J., & Langmuir, C. H. (2015). The global chemical systematics of arc front
50
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52
53
54
55
56
57
58
59
60
61
62
63
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65

874 stratovolcanoes: Evaluating the role of crustal processes. *Earth and Planetary Science*
1 875 *Letters*, 422, 182–193.
2 876 Viccaro, M., & Cristofolini, R. (2008). Nature of mantle heterogeneity and its role in the
3 877 short-term geochemical and volcanological evolution of Mt. Etna (Italy). *Lithos*, 105(3–
4 878 4), 272–288. <https://doi.org/10.1016/j.lithos.2008.05.001>
5 879 Wallace, P. J., Plank, T., Bodnar, R. J., Gaetani, G. A., & Shea, T. (2021). Olivine-Hosted
6 880 Melt Inclusions: A Microscopic Perspective on a Complex Magmatic World. *Annual*
7 881 *Review of Earth and Planetary Sciences*, 49.
8 882 Walter, M. J. (1998). Melting of garnet peridotite and the origin of komatiite and depleted
9 883 lithosphere. *Journal of Petrology*, 39(1), 29–60.
10 884 Zuccarello, F., Schiavi, F., & Viccaro, M. (2022). The eruption run-up at Mt. Etna volcano:
11 885 Constraining magma decompression rates and their relationships with the final eruptive
12 886 energy. *Earth and Planetary Science Letters*, 597, 117821.
13 887 <https://doi.org/10.1016/j.epsl.2022.117821>
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