Version of Record: <https://www.sciencedirect.com/science/article/pii/S001670371300375X> Manuscript_86fa173b1a83edfdcae20bd3b672f61e

Silicon isotope variations in the inner solar system:

Implications for planetary formation, differentiation and composition

-
- 4 Thomas Zambardi a,b*, Franck Poitrasson ^a, Alexandre Corgne ^{c,d}, Merlin Méheut ^a, Ghylaine
- 5 Quitté ^e and Mahesh Anand f,g
-
- * Corresponding author:
- Tel: 00337.86.27.29.84; e-mail: zambardi@illinois.edu
-
- ^a *Géosciences Environnement Toulouse, CNRS Université de Toulouse IRD, 14 avenue*
- *Edouard Belin, 31400 Toulouse, France.*
- ^b*Department of Geology Natural History Building, University of Illinois at Urbana-*
- *Champaign, 1301 W. Green Street, 61801 Urbana, IL, USA.*
- ^c *Institut de Recherche en Astrophysique et Planétologie, CNRS Université de Toulouse, 14*
- *avenue Edouard Belin, 31400 Toulouse, France.*
- *^d Instituto de Geociencias, Universidad Austral de Chile, Casilla 567, Valdivia, Chile*
- ^e *Laboratoire de Géologie de Lyon : Terre, Planètes, Environnement, CNRS, ENS de Lyon,*
- *Université Lyon 1, 46 allée d'Italie, 69364 Lyon, France.*
- ^f *Department of Physical Sciences, The Open University, Milton Keynes, MK7 6AA, UK.*
- ^g *Department of Mineralogy, The Natural History Museum, London SW7 5BD, UK.*

Abstract

Accurate and precise Si isotope measurements were obtained using magnesium doping 24 and high-resolution plasma source mass spectrometry for samples representative of the Earth, as well as lunar samples, meteorites from Mars (SNC), eucrites, a howardite, carbonaceous 26 chondrites (CC), ordinary chondrites (OC) and enstatite chondrites (EC). Our data confirm that significant Si isotope fractionations exist among the inner solar system planetary bodies. They show that the Earth and the Moon share the same Si isotopic composition, which is heavier than all other measured bodies, in agreement with most of previous studies. At the other end of the spectrum, enstatite chondrites have the lightest Si isotope compositions. In order to precisely estimate the amount of Si that may have entered the Earth's core, we developed a refined model of Si partitioning based on continuous planetary accretion that 33 takes into account the likely variations in *T*, *P* and $fO₂$ during the Earth's accretion, as well as isotopic constraints involving metal-silicate partitioning derived from both experimental and natural sample data sets.

Assuming that the difference between the isotopic signature of the bulk silicate Earth (BSE) and chondrites solely results from Si isotope fractionation during core formation, our model implies that at least ~12wt% Si has entered the Earth's core, which is greater than most of the estimates based on physical constraints on core density or geochemical mass balance calculations.

This result leads us to propose two hypotheses to explain this apparent contradiction: 1) At least part of the Earth's building blocks had a Si isotope composition heavier than that 43 observed in chondrites (*i.e.*, δ^{30} Si > -0.39%o). 2) If on the contrary the Earth accreted only 44 from material having chondritic δ^{30} Si, then an additional process besides mantle-core differentiation is required to generate a stronger isotope fractionation and lead to the observed heavy isotope composition of the bulk silicate Earth. It may be the loss of light Si isotopes

during partial planetary vaporization in the aftermath of the Moon-forming giant impact. This process, which may have affected metallic cores, required a thorough isotopic re-equilibration between core and silicate to explain the similar heavy isotope composition of the silicate portions of the Earth and the Moon.

-
- Keywords: silicon isotopes, terrestrial planets, core formation, inner solar system

1. INTRODUCTION

The core of the Earth has been shown to have a 5-10% density deficit relative to a pure Fe-Ni alloy (Birch, 1968; Boehler, 2000; Anderson and Isaak, 2002), suggesting significant incorporation of light elements such as C, O, S and/or Si. Given the superchondritic Mg/Si ratio of the silicate Earth and assuming that terrestrial planets have a bulk composition similar to those of chondrites (*e.g.*, Jagoutz et al., 1979; Allègre et al., 1995; McDonough and Sun, 1995), Si has been considered among the most likely candidate light elements to occur in the core (*e.g.*, MacDonald and Knopoff, 1958; Allègre et al., 1995). However, the Si contribution to the light element budget of the Earth's core remains poorly constrained, reflecting the conflicting evidence from cosmochemical, experimental and theoretical studies concerning which light element(s) is/are present (*e.g.*, Poirier, 1994; Allègre et al., 1995; Sherman, 1997; Kilburn and Wood 1997; Gessmann et al., 2001; Li et al., 2001; Alfè et al., 2002; Rubie et al., 2004; Asahara et al., 2007; Badro et al., 2007; Côté et al., 2008; Corgne et al., 2009; Antonangeli et al., 2010). The determination of the Si isotope signatures (from here on 67 expressed as permil δ^{30} Si) of chondritic and Earth materials has been proposed to provide additional constraints on the amount of Si present in the Earth's core (*e.g.*, Georg et al., 2007): for a given Si isotopic fractionation between the Earth's core and the Earth's mantle, 70 the difference in δ^{30} Si between the silicate portion of the Earth and the bulk Earth is indeed proportional to the fraction of total Si going into the core. Assuming that the mean Earth signature can be estimated by measuring chondritic meteorites, the difference in isotopic composition between rock samples representative of the bulk silicate Earth and chondritic 74 materials $(\Delta^{30} Si_{BSE-Bulk Earth} = \Delta^{30} Si_{BSE-Chond.})$ can thus be used to estimate the Si content of the core.

With the exception of Chakrabarti and Jacobsen (2010a), all recent studies indicate that chondritic Si isotope compositions are clearly distinct from the Earth's mantle signature

(Georg et al., 2007; Fitoussi et al., 2009; Zambardi et al., 2009; Ziegler et al., 2010; Armytage 79 et al., 2011; Fitoussi and Bourdon, 2012). All these studies consider that the Δ^{30} SiBSE-Chond. values directly result from the Si incorporation into the Earth's core during accretion. 81 However two issues remain: (1) the measured Δ^{30} Si_{BSE-Chond} values vary from one study to another, which leads to significant variations in estimated core contents of Si and (2) the Si fractionation during terrestrial accretion must be modeled more realistically in order to better 84 understand the relationship linking Δ^{30} Si_{BSE-Chond} to the Si content of the Earth's core. To further explore the usefulness of Si isotopes in this context, and in particular to be able to compare the Si-core content inferred from isotope data with other independent constraints, it is crucial to address these two points.

In addition to potential analytical bias during isotopic measurements that may be adequately addressed through interlaboratory comparison of geostandard analyses (Reynolds et al., 2007; Zambardi and Poitrasson, 2011), part of the uncertainty in estimating the Si content of the core resides in the choice of which chondritic reference to use as a proxy for the bulk Earth. This is because different chondrites types do not necessarily share the same Si 93 isotope signature, with carbonaceous and ordinary chondrites showing δ^{30} Si values notably higher than enstatite chondrites (Georg et al., 2007; Fitoussi et al., 2009; Armytage et al., 2011; Savage and Moynier, 2013). Furthermore, the nature of the terrestrial building blocks remain hypothetical, and different Earth forming models have been proposed, requiring the use of variable chondritic sources (*e.g.*, Allègre et al., 1995, 2001; Palme and O'Neill, 2003; Trinquier et al., 2007; Carlson et al., 2007; Leya et al., 2008; Simon and DePaolo, 2010; Javoy et al., 2010; Schönbächler et al., 2010).

Last but not least, another potential source of uncertainty in estimating the Si content of the Earth's core comes from the use of single-stage core formation models (Georg et al., 2007; Shahar et al., 2009; Ziegler et al., 2010; Armytage et al., 2011) that involve fixed

103 conditions of temperature (T) , pressure (P) and oxygen fugacity (fO_2) throughout this protracted event. Such models neglect the continuous variations of these three parameters during core-mantle differentiation, while they are known to significantly affect Si metal-silicate partitioning (*e.g.*, Gessmann et al., 2001; Corgne et al., 2008). Furthermore, *T* variations are also known to affect isotope fractionation during metal-silicate melt partitioning (*e.g.*, Shahar et al., 2009, 2011; Ziegler et al., 2010). For these reasons, it is more appropriate to consider continuous core formation models (Wade and Wood, 2005; Corgne et al., 2008; Chakrabarti and Jacobsen, 2010a).

To address these issues, we have produced new Si isotope measurements for a number of terrestrial and lunar rocks, as well as Martian meteorites, eucrites, a howardite and different types of chondrites. Our analytical approach combines Mg-doping and high-resolution mass spectrometry, which improves both data accuracy and reproducibility (Zambardi and 115 Poitrasson, 2011). A Δ^{30} SiBSE-Chond. value is inferred from our data set, and incorporated into a continuous accretion elemental partitioning model (Wade and Wood, 2005; Corgne et al. 2008). Our model also considers the constraints on Si isotope metal-silicate fractionation (Ziegler et al., 2010; Shahar et al., 2009, 2011). This approach allows us to refine the estimates of Si incorporated into the Earth's core as a function of the selected chondrite type(s) that contributed matter to the growing Earth. We eventually compare the expected Si-core concentrations inferred from our model to those derived from purely physical constraints and/or geochemical mass balance calculations, and then discuss the range of possible processes and materials involved in terrestrial planet formation during the early stages of the solar system.

2. SAMPLES AND METHODS

This study focuses on the Si isotope compositions of silicate samples spanning a large

range of igneous rock compositions from the inner solar system. They include samples from the Earth (11 peridotites, 4 basalts, 1 gabbro, 1 diorite, 1 dolerite and 1 komatiite), Moon (4 basalts, and 1 norite), Mars (5 shergottites, 4 nakhlites and 1 chassignite), as well as HEDs (6 eucrites and 1 howardite), 6 ordinary chondrites, 3 carbonaceous chondrites and 2 enstatite chondrites. Each sample group consists of a variety of rocks that reflect as much as possible the geochemical diversity of their respective parent bodies. The full sample set is listed in Table 1.

For meteorites and lunar samples, the aliquots used for the analyses were taken from at least 1g of well-homogenized bulk-rock powders (or at least 0.8g for Martian meteorites) to ensure that the sample is representative of the bulk rock as far as possible even for these rare samples. For the more abundant terrestrial rock samples, powders were prepared by crushing more than 100g of starting material. Since a complete description of the analytical procedure is already given in Zambardi and Poitrasson (2011), only the main steps are summarized hereafter.

The sample solutions were prepared using an alkali fusion method adapted from Georg et al. (2006). Silicon purification was achieved by column chromatography using Biorad AG50W-X12 cationic resins. Typical Si recovery was between 95 and 100% after the decomposition and purification steps. Procedural blanks for Si were ~30ng, corresponding to less than 0.2% of the total amount of Si analyzed per sample and were thus negligible. All measurements were performed at the Observatoire Midi-Pyrénées (Toulouse, France) and at the Open University (Milton Keynes, United Kingdom) using a Neptune (Thermo) MC-ICP-MS in medium or high-resolution mode, with wet plasma conditions and slightly acidic HCl solutions (0.05M for both samples and standard). The instrumental mass bias drift was corrected simultaneously using the sample standard bracketing (SSB) technique and by adding magnesium (Mg) as an internal standard in both standard and samples. Measurements 153 of $25Mg/24Mg$ ratios were performed in the dynamic mode. The mass bias drift correction was 154 then applied using the Russel's exponential law (Russel et al., 1978). All the data are reported 155 in Table 1 as δ^{30} Si and δ^{29} Si (‰) relative to the NIST international standard reference 156 material SRM 8546 (NBS-28) following the relation:

$$
157 \qquad \delta^{xx} \text{Si} = \left(\frac{\left(\frac{xx}{\text{Si}} \right)^{28} \text{Si}}{\left(\frac{xx}{\text{Si}} \right)^{28} \text{Si}} \right)_{\text{NBS}-28} - 1 \right) \times 1000 \tag{1}
$$

158 where the superscript *xx* denotes the Si isotope with either mass number 30 or 29.

159 The long-term external reproducibility for δ^{30} Si and δ^{29} Si was given by the standard deviation of 54 measurements of BHVO-2 over three years. It yielded 0.07‰ and 0.04‰ 161 (2SD) for δ^{30} Si and δ^{29} Si, respectively. In the following, uncertainties of planetary and meteorite parent body means are computed as two standard error of the mean (2SE) rather than two standard deviation (2SD) since the purpose is to indicate how well we know the mean rather than to give a measure of the spread of a set of individual measurements around the mean (see *e.g.*, Miller and Miller, 1993). Noteworthy, these two standard error uncertainties represent a direct proxy of *t*-tests (see *e.g.,* Poitrasson et al., 2004, 2005). All literature data have been recomputed this way to be comparable. However, whether computed as 2SE or 2SD, it is important to keep in mind that these uncertainty estimates only include random errors but not systematic ones potentially associated to analytical or sampling biases.

170

171 **3. RESULTS**

172 All data are given in Table 1. Bulk rock δ^{30} Si values for inner solar system bodies 173 range from -0.19‰ to -0.64‰. Comparison between δ^{30} Si and δ^{29} Si in a three-isotope plot 174 shows that these Si isotope fractionations are mass dependent (Fig. 1), in agreement with 175 previous data (*e.g.*, Douthitt, 1982; Molini-Velsko, 1986; Ding 1996; Georg et al., 2007; 176 Fitoussi et al., 2009; Savage et al., 2010; Ziegler et al., 2010; Armytage et al., 2011).

177 Terrestrial mafic rocks yield δ^{30} Si ranging from -0.19‰ to -0.33‰ (Table 1; Fig 2), which 178 includes the smaller variation range displayed by lunar mafic samples $(\delta^{30}Si$ ranging from -0.24‰ to -0.29‰) and therefore suggests the two bodies share similar isotopic features. Meteoritic samples including chondrites, rocks thought to originate from Mars and 4-Vesta all 181 yield lighter δ^{30} Si values, ranging from -0.33‰ to -0.64‰ (Table 1; Fig. 2). Among 182 chondrites, enstatite chondrites clearly display the lightest isotope compositions $(\delta^{30}Si_{EC} = -$ 183 0.59‰ and -0.64‰) relative to both ordinary chondrites $\delta^{30} \text{Si}_{\text{OC}} = -0.47 \pm 0.01\%$ (2SE, 184 n=6)) themselves not statistically different from carbonaceous chondrites $(\delta^{30}S_{\text{ICC}} = -0.43 \pm \sqrt{2}$ 0.05‰ (2SE, n=3)). As the signatures of CC and OC overlap within uncertainties, their 186 averaged signature is reported in the following as $\delta^{30}Si_{O/CC} = -0.46 \pm 0.02\%$ (2SE, n=9). We note that Savage and Moynier (2013) found significant differences between EL and EH chondrites, with EH being enriched in light isotopes compared to EL. Although the present study does not reproduce such an extensive E-chondrite set, it should be noticed that our two EH-chondrites do not yield comparable negative values. In fact, the two EH we measured rather plot within the range of the EL chondrites analyzed by Savage and Moynier (2013). This issue thus requires additional measurements of E-chondrites to be solved.

A review of previously published data reveals significant differences between laboratories, even for the same extra-terrestrial samples and terrestrial geostandards (Georg et al., 2007; Fitoussi et al., 2009; Zambardi et al., 2009; Ziegler et al., 2010; Chakrabarti and Jacobsen, 2010a; Armytage et al., 2011, 2012; Zambardi and Poitrasson, 2011; Fitoussi and Bourdon 2012; Savage et al., 2012, Savage and Moynier, 2013, see Fig. 3). These variations may have multiple origins that remain complex to identify with certainty. Among them, sample heterogeneity is a plausible explanation, especially for rare samples like meteorites and lunar rocks given the limited amount of sample available for analyses. Armytage et al. (2011) previously noticed the effect of sample heterogeneity for the Martian meteorite

202 Governador Valadares, with two different preparations producing a 0.1% difference in δ^{30} Si. In the present study, sample heterogeneity was avoided as much as possible by powdering at least 0.8g of bulk SNC or 1g of bulk chondrites, achondrites and lunar samples, prior to sample preparation. Furthermore, several methodological aspects that may influence the final results are still a subject of debate in the literature (*e.g.*, Zambardi and Poitrasson, 2011). 207 These include for example the use of HCl or $HNO₃$ sample matrices (e.g. Cardinal et al., 2003; van den Boorn et al., 2009), removal of S from samples (van den Boorn et al., 2009), the use of a collision cell (Chakrabarti and Jacobsen, 2010a), operating a MC-ICP-MS under dry or wet plasma conditions and more broadly, the contrasted architectures of the plasma source mass spectrometer used. Altogether, the present data are well consistent with those of Armytage et al., (2011), especially for extra-terrestrial Si isotopic compositions (Fig. 4).

Given that the models estimating the planetary Si-core contents are highly sensitive to 214 the Δ^{30} SiBSE-Chond. value (see details in section 4.3), it is extremely important to estimate each 215 planetary δ^{30} Si average as accurately and precisely as possible. In this study, the analytical 216 protocol includes the use of Mg as an internal standard. This technique has been demonstrated to correct for isotope fractionation caused by rapid mass bias variations and matrix effects (Cardinal et al., 2003; Engström et al., 2006; Zambardi and Poitrasson, 2011). Combined with high-resolution mass spectrometry, it helps to significantly improve the reproducibility and accuracy of Si isotope measurements.

4. DISCUSSION

4.1. δ **³⁰ Si variability among planetary bodies**

224 Planetary δ^{30} Si averages calculated from our new results as well as from recent literature data are presented in Figure 4. For terrestrial samples, the effect of magmatic differentiation has been shown to increase the amount of heavy Si isotopes in the most

differentiated materials (Douthitt 1982; Ding et al., 1996; Savage et al., 2011). These previous results suggest that, assuming a homogeneous mantle source (see Savage et al., 2010), only mafic and ultramafic materials should be considered as representative of the bulk silicate parts of the solar system planetary bodies. We thus computed the isotopic composition of the bulk 231 silicate Earth (BSE) following this rule. This yields for our measurements δ^{30} Si_{BSE} = -0.27 ± 0.01‰ (2SE, n=19; see Table 1, Fig.4), in excellent agreement with values recently proposed by Savage et al. (2010, 2011), and Fitoussi and Bourdon (2012), but heavier by ~0.1‰ than that of Georg et al. (2007) and Chakrabarti and Jacobsen (2010a).

Four basalts and a norite sample allowed us to estimate the isotopic composition of the 236 Iunar mantle (BSMo). We found δ^{30} Si_{BSMo} = -0.27 \pm 0.02‰ (2SE, n=5), which is identical to 237 the Si isotope signature of the BSE (Fig. 4), and in agreement with the average computed by Georg et al. (2007), Armytage et al. (2012) and Fitoussi and Bourdon (2012). However this 239 signature is significantly heavier than the signature of a lunar breccia $(-0.45 \pm 0.05\%)$, 2SE, n=1) measured by Chakrabarti and Jacobsen (2010a).

Among the Martian meteorites studied here, the Los Angeles shergottite represents the most differentiated and felspar-rich basaltic rock (Xirouchakis et al., 2002; Papike et al., 2009). Like differentiated terrestrial samples, this more felsic sample was found to have a heavier Si isotopic signature (Table 1). Los Angeles was thus omitted from the bulk silicate Mars (BSMa) calculation with only less evolved rocks (meaning all samples but Los Angeles) 246 used to estimate the Martian mantle δ^{30} Si. We thus derive a bulk silicate Mars signature 247 (δ^{30} Si_{BSMa}) of -0.49 ± 0.03‰ (2SE, n=9), significantly lighter than both the δ^{30} Si_{BSE} and δ^{30} Si_{BSMo} (Table 1, Fig. 4). Since Nakhlites may contain up to 1wt% of secondary minerals (*e.g.*, smectite) (Bunch and Reid, 1975), their original isotopic signature could be shifted 250 toward lower δ^{30} Si values, given the generally accepted effects of low-temperature alteration on Si isotopes (*e.g.*, Ziegler et al., 2005a). However, incorporation of 1% of smectite with a

252 light δ^{30} Si signature of -2‰ (*e.g.*, Ziegler et al., 2005b) would induce a maximum shift of 253 0.02‰ on bulk Nakhlites samples, which does not lead to a significantly different δ^{30} Si_{BSMa} 254 value. Our δ^{30} Si_{BSMa} value is in excellent agreement with the -0.48‰ ± 0.06 (2SE, n=5) estimate proposed by Armytage et al. (2011), is identical within uncertainties with the result of Georg et al. (2007) and again significantly differs from the value reported by Chakrabarti and Jacobsen, 2010a).

The study of HED meteorites helps to constrain the bulk isotope composition of their parent body, most probably asteroid 4-Vesta (Consolmagno and Drake, 1977). Eucrites are thought to be fragments of the asteroid's upper crust, whereas diogenites probably originated from deeper levels. Howardites may be mixed breccias of eucrites and diogenites (*e.g.*, 262 Mittlefehldt, 1994). The lack of a resolvable difference in δ^{30} Si between the howardite and eucrites measured in the present study suggests a homogeneous Si isotope composition for the silicate portion of the HED parent body, in contrast with Li isotope compositions (Magna et 265 al., 2006). The HED signature of $-0.41 \pm 0.03\%$ (2SE, n= 7) that we obtained is in agreement with the data from Fitoussi et al. (2009), Armytage et al. (2011) and Chakrabarti and Jacobsen (2010a), but differs from data by Georg et al. (2007).

Lastly, the chondrite averages often vary among authors (Fig. 4), notably for the O/C groups. Given that the ordinary chondrites do not seem to show much isotopic differences among their types (*e.g.*, H, L or LL; Table 1; Armytage et al., 2011), we tentatively attribute these discrepancies to inter-laboratory biases.

4.2. Planetary bulk Si isotopic compositions and chemical heterogeneity of the inner solar system

As previously shown in most of the recent studies, the lunar mantle and BSE have 276 similar δ^{30} Si, both have significantly heavier isotope compositions than Mars, 4-Vesta and

chondrite parent bodies (*e.g.*, Georg et al., 2007; Armytage et al., 2011, 2012, Fitoussi and Bourdon 2012, see Fig. 4). Similarities between the Earth and the Moon were also observed for other isotope systems such as O (*e.g.*, Wiechert et al., 2003; Spicuzza et al., 2007; Hallis et al., 2010), Cr (*e.g.*, Lugmair and Shukolyukov, 1998), Ti (Zhang et al., 2012), Li (*e.g.*, Magna et al., 2006), W (*e.g.*, Touboul et al., 2007) and Sr (Moynier et al., 2010), thereby supporting a genetic link between the Moon and the Earth's mantle. Assuming that chondrites are the building blocks from which the terrestrial planets formed (*e.g.*, Jagoutz et al., 1979; Allègre et al., 1995; McDonough and Sun, 1995), the heavy Si isotope composition of the Earth's mantle relative to chondrites is therefore not representative of the bulk Earth. As previously discussed in the literature, incorporation of isotopically light Si into the Earth's core is a possible explanation for this "heavy mantle" (Georg et al., 2007; Shahar et al., 2009; Fitoussi et al., 2009; Ziegler et al., 2010; Chakrabarti and Jacobsen, 2010a; Armytage et al., 2011; Fitoussi and Bourdon, 2012).

290 The δ^{30} Si signatures of Mars and Vesta do not show such a shift with respect to chondrites. A possible explanation (further discussed in section 4.4) is that these bodies have not segregated a significant amount of Si into their core reflecting their relatively small size and different redox conditions of formation, in agreement with models of planetary differentiation (*e.g.*, Wade and Wood, 2005; Corgne et al., 2008).

To identify the process that led to the heavy Si isotope composition of the Earth's mantle, and eventually better constrain the amount of Si segregated into the Earth's core, an important objective is to obtain a precise value of the bulk-Earth signature. However current geochemical models indicate that the Earth may have accreted from C-chondrites (*e.g.*, Allègre et al., 1995, 2001; Palme and O'Neill, 2003), E-chondrites (*e.g.*, Javoy, 1995; Trinquier et al., 2007; Javoy et al., 2010), O-chondrites (*e.g.*, Carlson et al., 2007; Leya et al., 2008; Simon and de Paolo, 2010) or mixed types of chondrites (*e.g.*, Schönbächler et al.,

2010; Fitoussi and Bourdon 2012). Given that chondrites types do not all share the same Si isotope signature (Fig. 4), the bulk-Earth signature may then vary from -0.62‰ assuming a pure E-chondrites end-member, to -0.46‰ considering a pure O/C-chondrites end-member.

Given that the HED parent body and Mars did not experience significant core partitioning of Si during metal segregation (*e.g.*, Wade and Wood, 2005; Corgne et al., 2008), their mantle signature should then correspond to that of the bulk planetary body. Since the Si isotopic signature for HED samples is indistinguishable within uncertainties from that of O/C-chondrites, we infer the bulk composition of their parent body is consistent with a starting material dominated by O/C-chondrite-like materials. In contrast, our data show that the Martian isotopic composition appears slightly but significantly lighter than 4-Vesta. While this result is compatible with an O/C-chondrite-dominated Mars composition, it may betray the contribution of a significant amount of E-chondrites as well. A simple mass balance calculation can be used to give a rough estimate of the proportions of different chondrite types in the starting composition of Mars. Doing so leads to a range of 0 to 40wt% (with an average at 20wt%) E-chondrites for Mars' bulk composition. Interestingly, Sanloup et al. (1999) and Mohapatra and Murty (2003) proposed even larger contributions of E-chondrites for Mars building blocks (∼70wt% E-chondrites) based on O isotope data. Despite the uncertainties of such models, these data suggest that E-chondrites could have significantly contributed to building Mars but only played a limited role in contributing to the HED parent body. Consequently, the possible distinct nature of the building blocks of 4-Vesta and Mars would argue in favor of the existence of chemical heterogeneities in the early solar system. While still a matter of debate (*e.g.*, Lewis, 1974; Morgan and Anders, 1980; Palme 2000; Taylor and Scott, 2003; Poitrasson et al., 2005), the existence of a temperature gradient in the solar nebula may have induced a chemical gradient following early condensation, with more oxidized and volatile-rich materials (C-chondrites-like) occurring preferably farther from the Sun and more reduced materials (E-chondrites-like) occurring in the inner parts of the protoplanetary accretion disk.

4.3. Constraints on the Earth's core composition

331 Using the signature of the BSE estimated above $(\delta^{30}Si = -0.27\%)$ and assuming the Earth accreted from chondritic materials, the difference in isotopic composition between the 333 bulk silicate Earth and its chondritic building blocks can be calculated using Δ^{30} SiBSE-Chond. = δ^{30} Si_{BSE}- δ^{30} Si_{Chond}. Assuming that this difference is due to metal-silicate partitioning during 335 mantle-core differentiation, Δ^{30} Si_{BSE-Chond} can be used to constrain both the Si concentration and isotopic composition of the core. However, the conditions under which mantle-core differentiation occurred must be known. Temperature (*T*) is a critical parameter for isotope 338 fractionation and typical equilibrium fractionations scale with $1/T²$ for high temperatures (Bigeleisen and Mayer, 1947). It is therefore generally accepted that the temperature dependence of the fractionation law can be written as:

$$
341 \qquad \varepsilon_{\rm Si}(T) = \delta^{30} \mathbf{Si}_{\rm Silicate}(T) - \delta^{30} \mathbf{Si}_{\rm Metal}(T) = \frac{B \cdot 10^6}{T^2} \tag{2}
$$

342 where $\epsilon_{\rm Si}$ denotes the fractionation factor of Si given in permil and defined as $\epsilon = 1000 \times$ 343 $ln(\alpha)$.

344 The value of the constant *B* has been estimated using either natural samples $(7.64 \pm 0.47,$ 345 Ziegler et al., 2010) or from metal-silicate fractionation experiments (7.45 ± 0.41) , Shahar et al., 2011). As mantle-core differentiation occurred under high-pressure conditions, pressure (*P*) may also have influenced Si fractionation, in particular through changes in the silicate melt structure. However, theoretical calculations performed by Shahar et al. (2009) suggest that the effect of *P* is only secondary. In the following, we will therefore consider that the isotope fractionation during accretion and core formation is only temperature dependent and 351 follows Equation (2).

Most recent work (Georg et al., 2007; Fitoussi et al., 2009; Shahar et al., 2009; Armytage et al., 2011; Fitoussi and Bourdon, 2012) has modeled Si isotope fractionation during mantle-core differentiation as a steady-state process at constant *T*, which gives the following relation:

$$
356 \t\t \varepsilon_{\rm Si}(T) = \frac{\Delta^{30} \rm Si_{BSE-Chond.}}{F_{\rm Core}}
$$
\t(3)

357 where *F*_{Core} represents the Si mass fraction of the Earth present in the core. The concentration 358 of Si in the core (C_{Si}^{Core}) is then derived from:

$$
359 \t C_{\text{Si}}^{\text{Core}} = \left(\frac{m_{\text{BSE}}}{m_{\text{Core}}}\right) \cdot \left(\frac{F_{\text{Core}}}{1 - F_{\text{Core}}}\right) \cdot C_{\text{Si}}^{\text{BSE}} \tag{4}
$$

360 knowing that $m_{\text{BSE}}/m_{\text{core}}$, the BSE/core mass ratio, is 2.1 (see *e.g.*, Yoder, 1995) and that the 361 concentration of Si in BSE (C_{Si}^{BSE}) is 21wt% (see *e.g.*, McDonough and Sun, 1995; O'Neill 362 and Palme, 1998; Palme and O'Neill, 2003).

It should be noted that, in an effort to account for varying conditions of core-mantle differentiation, some authors have also considered two-stages models (Fitoussi et al. 2009), each stage corresponding to a steady-state process. The major problem with the single- or two-stage(s) approach is that geochemical and dynamical constraints indicate that core formation was a continuous process rather than a discrete one (*e.g.*, Wetherill, 1986; Canup and Asphaug, 2001; Halliday, 2008; Wood and Halliday, 2010). It is therefore more appropriate to use continuous core formation models such as the ones proposed by Wade and Wood (2005) and Corgne et al. (2008). In these models based on metal-silicate partitioning of 371 siderophile elements, *T*, *P* and oxygen fugacity (fO_2) are allowed to continuously evolve as core formation proceeds. As these parameters have a major influence on elemental Si metal-silicate partitioning (*e.g.*, Corgne et al., 2008), it is crucial to consider their effects in details to obtain the most precise estimate of the Si content in the Earth's core. We thus use an approach similar to that of Chakrabarti and Jacobsen (2010a), but in the present study all parameters (*T*, *P* and $fO₂$ are constrained by siderophile element abundances in the BSE and their metal-silicate partitioning behavior.

4.4. Model of continuous core formation

- Continuous core formation is modeled as follows:
- a. Core formation takes place during Earth's accretion, with final metal-silicate equilibration at the base of a homogeneous magma ocean.

b. Growth of the Earth is linearly modeled in 100 steps. At each step, added metal represents 32.3% of the added mass and is equilibrated with the total mass of the mantle. The metal then migrates to the core without further re-equilibration.

c. The pressure of final equilibration at the base of the magma ocean is assumed to increase proportionally to the pressure at the core-mantle boundary (CMB) of the growing Earth. The pressure at the base of the magma ocean is fixed at 40% of the pressure at the CMB for nickel (Ni) partitioning data (essentially pressure-dependent) to satisfy the well-constrained Ni content of the BSE (Corgne et al., 2008). This means that, at the end of accretion, the pressure at the CMB is ~135 GPa while the pressure at the base of the magma ocean is ~54 GPa.

d. The peridotite solidus as defined by Fiquet et al. (2010) provides the variation of *T* at the base of the magma ocean during accretion (hence during core formation (Fig. 5a)). Average *T* and *P* conditions of core formation are about 2900 K and 33 GPa, respectively.

Siderophile element abundances in the BSE have been shown to be consistent with a continuous accretion model (Wade and Wood, 2005; Corgne et al., 2008; Wood et al., 2008). In particular, nickel and vanadium provide tight constraints on the required conditions of *T*, *P* and *f*O2 during accretion and core formation, as shown in Fig. 5b. Variations of the core/BSE Si elemental ratio during accretion can be calculated with the *T-P*-*f*O2 conditions and the parameterization of the Si metal-silicate partition coefficient (*DSi*) given by Corgne et al. 403 (2008):

404
$$
\log D_{\text{Si}} = 2.32(\pm 0.3) - \frac{21800}{T} - \frac{11(\pm 33) \cdot P}{T} - 2\log \left(\frac{x_{\text{FeO}}^{\text{Silicate}}}{x_{\text{Fe}}^{\text{ Metal}}}\right) - \log \gamma_{\text{Si}}^{\text{ Metal}} \tag{5}
$$

405 where *T* is the temperature (K), *P* is the pressure (GPa), $x_{\text{FeO}}^{\text{Silicate}}$ is the molar fraction of FeO in 406 the silicate mantle, x_{Fe}^{Meta} is the molar fraction of Fe in the metallic droplet, and $\gamma_{Si}^{\text{Meta}}$ is the activity coefficient in the metallic phase calculated from Dresler (1989). As shown in Figure 5b for a *f*O2 path satisfying siderophile element constraints (from Wood et al., 2008), the core/BSE concentration ratio of Si is poorly modeled due to large uncertainties associated with parameterizing the partitioning (Eq. 5). It is thus impossible to derive an accurate Si content of the core using this approach (Corgne et al., 2008). Values inferred this way span a huge and partly unrealistic range from 1 to 50wt%, considering 1SD uncertainty on the regression coefficients. As shown below, isotopic constraints on BSE and chondrites derived in section 3 can greatly reduce the range of estimates for the Si content of the core.

In the following, we present the procedure used to derive the isotopic signatures of core and BSE. At step *i*, the isotope signature of the added metal (AM) subsequently going to 417 the core $(\delta^{30}Si_{AM(i)})$ can be calculated knowing the BSE isotope signature and the metal-silicate fractionation law at the base of the magma ocean:

$$
419 \qquad \delta^{30} \text{Si}_{\text{AM}(i)} = \delta^{30} \text{Si}_{\text{BSE}(i)} - \varepsilon_{\text{Si}}(T_{(i)}) \tag{6}
$$

From mass-balance considerations, the following relationship can be derived between the ⁴²⁰ 421 isotope signatures at step *i* and step $i+1$:

$$
422 \t\delta^{30} \t\mathbf{Si}_{BSE(i)} \cdot m\mathbf{Si}_{BSE(i)} + \delta^{30} \t\mathbf{Si}_{\text{Chond.}} \cdot m\mathbf{Si}_{\text{Chond.}} = \delta^{30} \t\mathbf{Si}_{BSE(i+1)} \cdot m\mathbf{Si}_{BSE(i+1)} + \delta^{30} \t\mathbf{Si}_{AM(i+1)} \cdot m\mathbf{Si}_{AM(i+1)}(7)
$$

423 where $mSis_{BSE(i)}$, $mSis_{AM(i)}$ and $mSic_{hond}$ are the Si mass in BSE and added metal at step *i*, and

424 the added Si mass from chondrites at each step, respectively. Combining (6) and (7) yields:

425
$$
\delta^{30} \text{Si}_{BSE(i+1)} = \frac{\delta^{30} \text{Si}_{BSE(i)} \cdot m \text{Si}_{BSE(i)} + \varepsilon_{Si}(T_{(i+1)}) \cdot m \text{Si}_{AM(i+1)} + \delta^{30} \text{Si}_{Chond} \cdot m \text{Si}_{Chond.}}{m \text{Si}_{BSE(i+1)} + m \text{Si}_{AM(i+1)}}
$$
(8)

426 Since we are interested in isotope signatures with respect to the initial chondritic material, 427 δ^{30} Six can be replaced by Δ^{30} Six-chond. in equation (8), with Δ^{30} Six-chond. = δ^{30} Six – δ^{30} Sichond. 428 giving:

429
$$
\Delta^{30} \text{Si}_{BSE-Chond.(i+1)} = \frac{\Delta^{30} \text{Si}_{BSE-Chond.(i)} \cdot m \text{Si}_{BSE(i)} + \varepsilon_{Si}(T_{(i+1)}) \cdot m \text{Si}_{AM(i+1)}}{m \text{Si}_{BSE(i+1)} + m \text{Si}_{AM(i+1)}}
$$
(9)

430 The value of Δ^{30} SiBSE-Chond. at the end of core formation is calculated using Eq. (9) with the 431 initial conditions Δ^{30} Si_{BSE-Chond.(0)} = 0 and *m*Si_{BSE(0)} = 0 and the final condition *m*Si_{BSE(100)} 432 given by current BSE concentration models (21wt% Si, e.g., McDonough and Sun, 1995; 433 O'Neill and Palme, 1998), and using Eq. (5) to derive Si mass partitioning between the added 434 metal (mSi_{AM}) and silicate (mSi_{BSE}) . The isotopic signature of the core at step *i* is then given 435 by the equation:

436
$$
\Delta^{30} \text{Si}_{\text{Core-Chond.}(i)} = \frac{\sum_{j=1}^{i} \Delta^{30} \text{Si}_{\text{AM-Chond.}(j)} \cdot m \text{Si}_{\text{AM}(j)}}{\sum_{j=1}^{i} m \text{Si}_{\text{AM}(j)}}
$$
(10)

437 Figure 5c presents the theoretical variations of Δ^{30} SiBSE-Chond. during accretion for a typical 438 oxygen fugacity path satisfying siderophile element constraints, and with the fractionation law 439 given by Ziegler et al. (2010) (Eq. (2) with B=7.64). Given the large uncertainties on 440 elemental partitioning which are incorporated into the model (see for instance Fig. 5b), the 441 predicted difference in isotopic signature between BSE and chondrites is also highly 442 uncertain, ranging between 0.03‰ and 0.5‰ in the example considered here (Fig. 5c). To 443 circumvent this difficulty, we calculated the variation during accretion of the Δ^{30} SiBSE-444 Chond (*i*)/ $F_{\text{Core}}(i)$ ratio, where $F_{\text{Core}}(i)$ is the Si fraction of the Earth contained in the core at step *i*. 445 The following relationship links $F_{\text{Core}(i)}$ to the partition coefficient D_{Si} :

446
$$
\frac{1}{F_{\text{Core}(i)}} = \frac{1}{D_{\text{Si}(i)}} \frac{m_{\text{BSE}}}{m_{\text{Core}}} + 1
$$
 (11)

447 where $m_{\text{BSE}}/m_{\text{core}}$, the BSE/core mass ratio, equals 2.1 (e.g. Yoder, 1995). For a single step 448 steady-state process (a single-stage of core formation at a fixed *T*), Δ^{30} SiBSE-Chond./*F*Core is 449 simply equal to ε_{Si} (Eq. 2). Considering a continuous accretion scenario, the $\Delta^{30}Si_{BSE}$. 450 Chond.(*i*)/*F*Core(*i*) ratio (Fig. 6) can be seen as the integrated fractionation factor during accretion, 451 up to step *i* (Δ^{30} SiBSE-Chond.(100)/*F*Core(100) = ε si) and taking into account all relevant effects 452 (effect of *T* on isotope fractionation, *T*-*P*-*f*O2 effects on elemental Si partitioning, effect of 453 isotope dilution in the mass-increasing BSE). An important outcome of the model is that the 454 value of Δ^{30} Si_{BSE-Chond.(100)}/ $F_{\text{Core}(100)}$ (= ε_{Si}) at the end of accretion varies little (0.88 ± 0.06), 455 regardless of the *f*O2 path considered (all paths span a realistic range (see insert of Fig. 6), 456 while remaining consistent with siderophile element constraints). To evaluate the total 457 uncertainty on ε_{Si} inferred from the model, it is important to also take into account the 458 uncertainty on *T* through Earth's accretion. Given that *T* is constrained by the peridotite 459 solidus (Fiquet et al., 2010), we can evaluate the impact of the *T* uncertainty on the final ε_{Si} 460 (or Δ^{30} SiBSE-Chond.(100)^{/*F*}Core(100)</sub>) using the uncertainties from their experiments (±150K, see 461 Fiquet et al., 2010). This leads to an additional -0.053/+0.047 uncertainty (approximated to 462 \pm 0.05), resulting in a final estimate of Δ^{30} SiBSE-Chond.(100)/*F*Core(100) = 0.88 ± 0.11. In a steady 463 state model at constant *T* (Eq. (3)), this Δ^{30} Si_{BSE-Chond.(100)}/ $F_{\text{Core}(100)}$ value would correspond to 464 core formation at a fixed *T* of 2940K, close to the 3000K value proposed by Ziegler et al. 465 (2010).

466 In the approach detailed above, we have used the fractionation law of Ziegler et al. 467 (2010). More generally, it can be shown that Δ^{30} SiBSE-Chond./*F*Core final is proportional to *B* (as in 468 Equation (3)), yielding:

469
$$
\Delta^{30} \text{Sig}_{\text{BSE-Chond}} / F_{\text{Core final}} = (0.115 \pm 0.008)^* B
$$
 (12)

470 The integrated fractionation factor can then be used together with the isotope 471 signatures of BSE and chondrites to estimate F_{Core} , and hence the amount of Si present in the 472 core, using Eq. (4). Thus we derive a minimum estimated Si core content of 12_{-2}^{+4} wt% using 473 O/C-chondrites as building material for the Earth (Δ^{30} Si_{BSE-O/CC} = 0.19), and a maximal 29_{-9}^{+21} 474 wt% Si using E-chondrites (Δ^{30} Si_{BSE-EC} = 0.35). Any mix of chondrite types as starting 475 materials will thus result in a $0.19 \le \Delta^{30}$ SiBSE-Chond. ≤ 0.35 , and therefore produce an estimated 476 Si-core concentration in-between 12 and 29wt% (Fig. 7). The corresponding δ^{30} Si isotope 477 signature of the core is then approximately $-1.1 \pm 0.2\%$ according to Eq. (10).

478

479 **4.5 Application of the model to literature data**

480 Applying our model to previously published data sets from various laboratories 481 (Georg et al., 2007; Fitoussi et al., 2009; Chakrabarti and Jacobsen 2010a; Armytage et al., 482 2011, 2012; Fitoussi and Bourdon 2012, Savage and Moynier 2013) yields estimated Si-core 483 concentrations varying between about 1 and 13wt% using O/C-chondrites, and between 484 11wt% and 35wt% using E-chondrites (see Table 2 for details). Since the model is very 485 sensitive to the Δ^{30} Si_{BSE-Chond.} value, it is extremely important to find the most accurate 486 estimates of both δ^{30} Si_{BSE} and δ^{30} Si_{Chond}. In order to improve the statistical robustness of our 487 model, below we computed a Δ^{30} Si_{BSE-Chond.} that incorporates the previously published data. 488 Given the significant offsets in both the Georg et al. (2007) and Chakrabarti and Jacobsen 489 (2010a) results from all other recently published data including ours, we omit the former from 490 the calculation. Thus using data from Fitoussi et al., (2009), Savage et al., (2010), Armytage 491 et al., (2011), Fitoussi et Bourdon (2012), Savage and Moynier (2013) and this study, we 492 estimate an average δ^{30} Si_{BSE} = -0.28 ± 0.01‰, (2SE, n=62), δ^{30} Si_O/cc. = -0.46 ± 0.01‰, 493 (2SE, n=37), and δ^{30} Si_{EC.} = -0.64 ± 0.03‰, (2SE, n=26). This result in a Δ^{30} Si_{BSE-O/CC} = 0.18, 494 which leads to minimum estimate of 12_{-3}^{+3} wt% Si expected in the core, and a Δ^{30} SiBSE-EC =

495 0.36, leading to a maximum estimate of 31^{+19}_{-10} wt% Si expected in the core (see Table 2, Fig. 7). It is worth noting that, although absolute values are apparently shifted, data from Georg et 497 al. (2007) lead to similar results (Δ^{30} Si_{BSE-O/CC} = 0.20 and Si-core = 13wt%, and/or Δ^{30} Si_{BSE-EC} 498 = 0.31 and Si-core = $24wt\%$, see Table 2) on average.

4.6. Did the Earth accrete from chondrites only?

Assuming that mantle-core differentiation is the sole process responsible for Si isotope fractionation during Earth's formation, our model suggests that the minimum expected Si core concentration is ∼12wt%, using most of the available isotopic data (Table 2, Fig. 7). Such a value exceeds most of the estimates based on the expected density of Fe-Si alloys at core conditions (*e.g.*, Sherman, 1997; Alfè et al., 2002; Dobson et al., 2003; Badro et al. 2007; Brosh et al., 2009; Antonangeli et al., 2010; Ricolleau et al., 2011; Bouhifd et al., 2012) or geochemical mass balance calculations (Allègre et al., 1995; 2001; Javoy et al., 2010). The latter usually limit the Si core content at between 1 and 7wt% (in most cases between 2 and 5wt%), with the exception of a few models (Ricolleau et al., 2011) that predict higher Si core 510 contents, up to $11wt\%$.

Consequently, if the actual Si concentration of the Earth's core does not exceed 5 to 7wt%, and if mantle-core differentiation is effectively the sole process responsible for the heavy isotopic signature of the BSE, then our model does not argue in favor of an Earth purely made of chondrites, but rather implies that the Earth has at least partly accreted from non-chondritic materials. Such a scenario is worth being considered further since it has already been suggested by several other studies on the basis of, for example, non-chondritic Nd/¹⁴⁴Nd ratio (Caro et al., 2008, 2010). Hence, assuming the Earth has formed from non-chondritic materials, these latter should have had a heavier Si isotope composition than that of 519 chondrites (*i.e.*, δ^{30} Si > -0.39‰).

4.7. Silicon isotope fractionation by vaporization during a giant interplanetary impact?

Preferential loss of Si to outer space during the Earth's accretion may provide an alternative explanation for both the superchondritic terrestrial Mg/Si ratio and the Si isotope fractionation observed between the BSE and chondrites. A study by Knight et al. (2009) shows that Si vaporization from calcium-aluminum-rich inclusions (CAIs) in meteorites results in very strong Si isotope fractionation. These authors found a constant fractionation 527 factor α^{29} _{Si} = 0.98985 ± 0.00044 for the 1600-1900^oC temperature range, with the gas phase being enriched in light isotopes. Considering a ~2000K Si degassing source (*e.g.*, Poitrasson et al., 2004; Pahlevan and Stevenson, 2007), data from Knight et al. (2009) suggest that even small amounts of vaporized Si can significantly fractionate Si isotopes. Following a simple Rayleigh distillation model, we can estimate that loss of between 0.9 and 1.6wt% of the bulk Earth's Si loss through volatilization can explain the isotope fractionation we measured between the BSE and chondrites, without the need of Si isotope fractionation during mantle-core differentiation (Fig. 8A). If one now considers a chondritic Earth with 7wt% Si core 535 content (which can be translated into a 0.12‰ decrease of the starting ∆³⁰Si_{BSE-Chond} *via* mantle-core differentiation; see Fig. 7), a loss of 0.4 to 1.1wt% bulk Earth's Si is sufficient to reconcile our partitioning model with isotopic observations (Fig. 8B). Interestingly, this result is almost identical to the Fe loss required through partial vaporization (between 0.5 and 1wt%) to explain the observed Fe isotope differences between the Earth-Moon couple and the other solar system planetary bodies (Poitrasson et al., 2004).

The most widely accepted model for lunar formation involves a violent impact between the proto-Earth and a Mars-sized body (*e.g.*, Canup and Asphaug 2001; Poitrasson et al., 2004; Touboul et al., 2007). In this scenario, the ejected materials should have suffered significant vaporization during and after the impact, accompanied by isotope fractionation.

Elliptical trajectories of the ejecta (Cameron, 2000; Canup and Asphaug, 2001) would have facilitated the loss of light isotopes escaping from the Earth-Moon system (Poitrasson et al., 2004). This event would have been followed by a thorough homogenization process between the material that subsequently re-accreted to form the Earth and the Moon, yielding undistinguishable Si isotope compositions for the two bodies (Nimmo and Agnor, 2006; Pahlevan and Stevenson, 2007, 2011; Touboul et al., 2007). Noteworthy, the gas pressure during CAI formation was most likely different – and lower – than in the aftermath of the impact, so that the fractionation factor provided by the CAI data is possibly too high for the conditions of Earth accretion (*e.g.*, Young et al., 2002). Unfortunately, the "true" Si isotope fractionation factor during large impacts is unknown, hence this model should be considered as an end-member scenario.

While Fe and Si show significant mass-dependent isotopic differences between the Earth–Moon system, Mars and 4-Vesta (*e.g.*, Poitrasson et al., 2004; Georg et al., 2007), this is not the case for elements such as O, K and Mg (*e.g.*, Humayun and Clayton, 1995; Wiechert et al., 2001; Bourdon et al., 2010; Teng et al., 2010). The lack of observable mass-dependent interplanetary isotope fractionation may result from the large fractionations observed within planets for O, or because of analytical issues for K (see Poitrasson et al., 2004). A possible explanation for the difference between Mg on one hand and Fe and Si on the other hand could be the fact that Si and Fe occur in significant quantities in the Earth's core (*e.g.*, Shahar et al., 2009, 2011; Corgne et al., 2009; Ricolleau et al., 2011). One could thus hypothesize that the elemental losses and isotopic fractionation in Si and Fe preferentially result from the partial vaporization of metallic core components in the aftermath of the giant impact (see Poitrasson et al., 2004). Such a scenario suggests that the ejected Si-rich core particles have lost light Si isotopes to space before re-accretion. These core particles should then have acquired a heavier isotopic signature and subsequently re-equilibrated with

mantle components within the turbulent accretion disk of the new Earth-Moon system (*e.g.*, Pahlevan and Stevenson, 2007). This process may therefore have ultimately led to an isotopically heavier mantle signature. This hypothesis is in agreement with numerical simulations of interplanetary impacts that show significant ejection of metallic core particles (*e.g.*, Cameron, 2000). Additionally, experiments have demonstrated that the vaporization flux of elements is much more effective if they are initially present in metal rather than in oxide form (Wang et al., 1994).

The feasibility of this process can be evaluated by mass balance calculations. Assuming the Earth's core contains 7wt% Si prior to the impact, and considering the need to vaporize 1.6% (worst possible case, *i.e.,* using only E-chondrites, see Fig. 8) of the bulk Earth's Si preferentially from the core, this means that ∼10% of the Si present in the core has to be lost in the aftermath of the impact, which will eventually drive the Si-core concentration to 6.2wt% after re-accretion. This concentration remains very close to the initially assumed 7wt% Si-core content. If correct, this evaluation thus suggests that only marginal amounts of Si are lost from the core after the impact, given the current uncertainties on the Si-core content. This result also suggests that the minimum amount of bulk core that needs to be processed in such a case is 10%. Although no upper limit can be given, this still implies that a relatively limited core proportion may have been processed through this mechanism. In summary, the combination of a volatilization-driven isotopic fractionation process and mantle-core differentiation can produce the observed heavier Si isotope signature of the silicate Earth and Moon relative to chondrites.

5. CONCLUSIONS

Significant differences in Si isotope composition were found among planetary bodies of the inner solar system, in agreement with most previous studies. The bulk silicate Earth and the bulk silicate Moon have the same Si isotope composition, both are isotopically heavier than O/C-chondrites, E-chondrites, achondrites from 4-Vesta and Martian meteorites. This argues in favor of a common origin for the Earth and the Moon.

Coupling Si isotope constraints with a realistic continuous accretion core formation model allows us to estimate the amount of Si incorporated into the Earth's core. Assuming that mantle-core differentiation is the sole process that fractionates Si isotopes, our model suggests that the Earth's core has incorporated at least ∼12wt% Si. While these results are consistent with Si being a major light element in the Earth's core, this refined model requires higher Si core contents than permitted by geophysical and experimental constraints.

We propose two explanations to reconcile the geochemical, dynamical and geophysical data:

1) The Earth has – at least partly – accreted from non-chondritic materials, the latter 607 having on average heavier Si isotopic signature than chondrites (*i.e.*, δ^{30} Si > -0.39‰). Such 608 materials would be required to reduce the isotopic difference Δ^{30} SiBSE-Bulk Earth to 0.12‰ or less, hence limiting the Si core concentration to a reasonable value (*i.e.,* < 7wt%).

2) If instead the Earth accreted only from chondritic materials and the Si core content is <12wt% as dictated by geophysical constraints, then mantle-core differentiation cannot be the sole mechanism from which planetary Si isotopes variations originate. An additional process is then required, particularly if the Earth accreted significant amount of E-chondrite material during its formation. We propose that volatilization-driven isotopic fractionation in the aftermath of the Moon's forming giant impact could be this process. This mechanism could have preferentially affected the metallic core, which explains the lack of isotopic variations observed for lithophile elements, such as Mg, among planetary bodies. This scenario requires a final isotopic re-equilibration on a planetary scale, leading to the heavier isotopic compositions of the lunar and terrestrial mantles.

Finally, it is clear from our modeling that the mantle-core differentiation process alone cannot explain the observed terrestrial planet Si isotope compositions. Our current level of knowledge however, does not allow yet favoring one of the two hypotheses (*i.e.,* non-chondritic Earth and/or impact generated Si vaporization) proposed in this study.

Acknowledgements

This study would not have been possible without the allocation by the NASA's CAPTEM committee of lunar rocks samples to FP. The Natural History Museum of London and Meteorite Working Group is gratefully acknowledged for the provision of Martian meteorites to MA, as well as the Museum National d'Histoire Naturelle de Paris for providing eucrites samples to GQ. We also thank the Museum d'Histoire Naturelle de Toulouse for providing aliquots of Orgueil and St Sauveur meteorites and the Field Museum of Chicago for a piece of meteorite Abee. Michel Gregoire is thanked for providing part of the terrestrial peridotite samples. Carole Boucayrand, Manuel Henry, Sébastien Gardoll and Jonathan Prunier are gratefully acknowledged for assistance during clean-lab work, as well as Jérôme Chmeleff, Rémi Freydier and Manuela Fehr for analytical assistance on Neptune spectrometers. Craig Lundstrom is warmly thanked for fruitful discussions as well as for English language corrections. MA thanks RCUK for a research fellowship during which this collaborative project was carried out. This work has been funded by a grant from the Programme National de Planétologie (PNP) of CNRS-INSU to FP. Anonymous reviewers are thanked for in depth reviews that helped improving both the clarity and the quality of this manuscript. Associate Editor Mark Rehkämper is also acknowledged for his work on this manuscript and helpful comments.

References

- Alfè D., Gillan M.J. and Price G.D. (2002) Composition and temperature of the Earth's core constrained by combining ab initio calculations and seismic data. *Earth Planet. Sci. Lett.* **195**, 91-98.
- Allègre C.J., Manhès G. and Lewin E. (2001) Chemical composition of the Earth and the volatility control on planetary genetics. *Earth Planet. Sci. Lett.* **185**, 49-69.
- Allègre C.J., Poirier J.P., Humler E. and Hofmann A.W. (1995) The chemical composition of the Earth. *Earth Planet. Sci. Lett.* **134**, 515-526.
- Anderson O.L. and Isaak D.G. (2002) Another look at the core density deficit of Earth's outer core. *Phys. Earth Planet. Inter.* **131**, 19-27.
- Antonangeli D., Siebert J., Badro J., Farber D.L., Fiquet G., Morard G. and Ryerson F.J.
- (2010). Composition of the Earth's inner core from high-pressure sound velocity measurements in Fe-Ni-Si alloys. *Earth Planet. Sci. Lett.* **295**, 292-296.
- Armytage R.M.G., Georg R.B., Savage P.S., Williams H.M. and Halliday A.M. (2011)
- Silicon isotopes in meteorites and planetary core formation. *Geochim. Cosmochim. Acta* **75**, 3662-3676.
- Armytage R.M.G., Georg R.B., Williams H.M. and Halliday A.M. (2012). Silicon isotopes in lunar rocks : implications for the Moon's formation and the early history of the Earth. *Geochim. Cosmochim. Acta* **77**, 504-514.
- Asahara Y., Frost D.J. and Rubie D.C. (2007) Partitioning of FeO between magnesiowüstite
- and liquid iron at high pressures and temperatures: implications for the composition of the Earth's outer core. *Earth Planet. Sci . Lett.* **257**, 435-449.
- Badro J., Fiquet G., Guyot F., Gregoryanz E., Occelli F., Antonangeli D. and d'Astuto M. (2007) Effect of light elements on the sound velocities in solid iron: Implications for the composition of Earth's core. *Earth Planet. Sci. Lett.* **254**, 233-238.

- Bigeleisen J. and Mayer M.G. (1947) Calculation of equilibrium constants for isotopic exchange reactions. *J. Chem. Phys.* **15**, 261-267.
- Birch F. (1968) Density and composition of mantle and core. *J. Geophys. Res.* **69**, 4377-4388.
- Boehler R. (2000) High-pressure experiments and the phase diagram of lower mantle and core materials. *Rev. Geophys.* **38**, 221-245.
- Bourdon B., Tipper E.T., Fitoussi C. and Stracke A., (2010) Chondritic Mg isotope composition of the Earth. *Geochim. Cosmochim. Acta* **74**, 5069-5083.
- Brosh E., Makov G. and Shneck R.Z. (2009) Thermodynamic analysis of high-pressure phase
- equilibria in Fe–Si alloys, implications for the inner-core. *Phys. Earth Planet. Inter.* **172**, 289-298.
- Bunch T.E. and Reid A.M. (1975) The Nakhlites Part 1. Petrography and mineral chemistry. *Meteoritics* **10**, 303-315.
- Cameron A.G.W. (1997) The origin of the Moon and the single impact hypothesis. *Icarus* **126**, 126-137.
- Cameron, A.G.W., 2000. Higher-resolution simulations of the giant impact. In: Canup, R.M., Righter, K. (Eds.), Origin of the Earth and Moon. The University of Arizona Press, Tucson, pp. 133-144.
- Canup R.M. and Asphaug E. (2001) Origin of the Moon in a giant impact near the end of the Earth's formation. *Nature* **412**, 708-712.
- Cardinal D., Alleman L.Y., de Jong J., Ziegler K. and Andre L. (2003) Isotopic composition
- of silicon measured by multicollector plasma source mass spectrometry in dry plasma mode. *J. Anal. Atom. Spectro.* **18**, 213–218.
- Carlson R.W., Boyet M. and Horan M. (2007) Chondrite barium, neodymium, and samarium isotopic heterogeneity and early earth differentiation. *Science* **316**, 1175-1178.
- Chakrabarti R. and Jacobsen S.B. (2010a). Silicon isotopes in the inner Solar System:
- implications for core formation, solar nebular processes and partial melting. *Geochim. Cosmochim. Acta* **74**, 6921-6933.
- Chakrabarti R. and Jacobsen S.B. (2010b) The isotopic composition of magnesium in the inner Solar System. *Earth. Planet. Sci. Lett.* **293**, 349-358.
- Consolmagno, G.J., Drake, M.J., 1977. Composition and evolution of the eucrite parent body:
- Evidence from rare earth elements. Geochimica et Cosmochimica Acta, 41: 1271-1282.
- Corgne A., Keshav S., Wood B.J., McDonough W.F. and Fei Y. (2008) Metal–silicate partitioning and constraints on core composition and oxygen fugacity during Earth accretion. *Geochim. Cosmochim. Acta* **72**, 574-589.
- Corgne A., Siebert J. and Badro J. (2009) Oxygen as a light element: a solution to single-stage core formation. *Earth Planet. Sci. Lett.* **288**, 108-114.
- Côté A.S., Vocadlo L. and Brodholt J.P. (2008) The effect of silicon impurities on the phase diagram of iron and possible implications for the Earth's core structure. *J. Phys. Chem. Solids* **69**, 2177-2181.
- Ding T., Jiang S., Wan D., Li Y., Li J., Song H., Liu Z. and Yao X. (1996) Silicon isotope geochemistry. *Geological Publishing House*, Beijing, 125 p.
- Dobson D.P., Crichton W.A., Bouvier P., Vocadlo L. and Wood I.G. (2003) The equation of state of CsCl-structure FeSi to 40 GPa: implications for silicon in the Earth's core. *Geophys. Res. Lett.* **30**, 1014.
- Douthitt C.B. (1982) The geochemistry of the stable isotopes of silicon. *Geochim. Cosmochim. Acta* **46**, 1449-1458.
- Dresler W. (1989) Activities of silicon and carbon in the Fe-C-Si liquid system. Ironmaking *Conf. Proc.* **48**, 83-88.
- Engström E., Rodushkin I., Baxter D.C. and Öhlander B. (2006) Chromatographic purification for the determination of dissolved silicon isotopic compositions in natural
- waters by high resolution multicollector inductively coupled plasma mass spectrometry. *Anal. Chem.* **78**, 250-257.
- Fiquet G., Auzende A.L., Siebert J., Corgne A., Bureau H., Ozawa H. and Garbarino G. (2010) Melting of peridotite to 140 GPa. *Science* **329**, 1516-1518.
- Fitoussi C., Bourdon B., Kleine T., Oberli F. and Reynolds B.C. (2009) Si isotope systematics
- of meteorites and terrestrial peridotites: implications for Mg/Si fractionation in the solar nebula and for Si in the Earth's core. *Earth Planet. Sci. Lett.* **287**, 77-81.
- Fitoussi C., Bourdon B. (2012) Silicon isotope evidence against an enstatites chondrite Earth. *Sciencexpress*, 1 March 2012, 1-4.
- Georg R.B., Halliday A.N., Schauble E.A. and Reynolds B.C. (2007) Silicon in the Earth's 730 core. Nature 447, 1102-1106.
- Georg R.B., Reynolds B.C., Frank M. and Halliday A.N. (2006). New sample preparation techniques for the precise determination of the Si isotope composition of natural samples using MC-ICP-MS. *Chem. Geol.* **235**, 95-104.
- Gessmann C.K., Wood B.J., Rubie D.C. and Kilburn M.R. (2001) Solubility of silicon in liquid metal at high pressure: implications for the composition of the Earth's core. *Earth Planet. Sci. Lett.* **184**, 367-376.
- Hallis L.J., Anand M., Greenwood R.C., Miller M.F., Franchi I.A. and Russell S.S. (2010) The oxygen isotope composition, petrology and geochemistry of mare basalts : Evidence for large-scale compositional variation in the lunar mantle. *Geochim. Cosmochim. Acta* **74**, 6885-6899.
- Halliday A.N. (2008) A young Moon-forming giant impact at 70–110 million years accompanied by late-stage mixing, core formation and degassing of the Earth. *Phil.Trans. R. Soc. A.* **366**, 4163-4181.
- Jagoutz E., Palme H., Baddenhausen K., Blum M., Cendales G., Dreibus B., Spettel V.,
- Lorenz V. and Wanke H. (1979). The abundances of major, minor and trace elements in the Earth's mantle as derived from primitive ultramafic nodules. *Proc. Lunar Planet. Sci. Conf.* **10**, 2031-2050.
- Javoy M. (1995) The integral enstatite chondrite model of the Earth. *Geophys. Res. Lett.* **22**, 2219-2222.
- Javoy M., Kaminski E., Guyot F., Andrault D., Sanloup C., Moreira M., Labrosse S., Jambon A., Agrinier P., Davaille A. and Jaupart C. (2010) The chemical composition of the Earth: Enstatite chondrite models. *Earth Planet. Sci. Lett.* **293**, 259-268.
- Kilburn M.R., Wood B.J. (1997) Metal-silicate partitioning and the incompatibility of S and Si during core formation. *Earth Planet. Sci. Lett.* **152**, 139-148.
- Knight K.B., Kita N.T., Mendybaev R.A., Richter F.M., Davis A.M. and Valley J.W. (2009)
- Silicon isotopic fractionation of CAI-like vacuum evaporation residues. *Geochim. Cosmochim. Acta* **73**, 6390-6401.
- Lewis J.S. (1974) The temperature gradient in the solar nebula. *Science* **186**, 440-443.
- Leya I., Schönbächler M., Wiechert U., Krähenbühl U. and Halliday A.N. (2008) Titanium isotopes and the radial heterogeneity of the solar system. *Earth Planet. Sci. Lett.* **266**, 233-244.
- Li J., Fei Y., Mao H.K., Hirose K., Shieh S.R. (2001) Sulfur in the Earth's inner core. *Earth Planet. Sci. Lett.* **193**, 509-514.
- Liu Y., Spicuzza M.J., Craddock P.R., Day J.M.D., Valley J.W., Dauphas N. and Taylor L.A.
- (2010) Oxygen and iron isotope constraints on near-surface fractionation effects and the composition of lunar mare basalt source regions. *Geochim. Cosmochim. Acta* **74**, 6249- 6262.
- 768 Lugmair G.W. and Shukolyukov A. (1998) Early solar system timescales according to Mn-⁵³ Cr systematics. *Geochim. Cosmochim. Acta* **62**, 2863-2886.
- MacDonald G.J.F. and Knopoff L. (1958) On the chemical composition of the outer core. *Geophys. J. Astron. Soc.* **1**, 284-297.
- Magna T., Wiechert U., Halliday A. N. (2006) New constraints on the lithium isotope compositions of the Moon and terrestrial planets. *Earth. Planet. Sci. Lett.* **243**, 336-353.
- McDonough W.F., Sun S.-S. (1995) The composition of the Earth. *Chem. Geol.* **120**, 223- 253.
- Miller J.C., Miller, J.N., (1993) Statistics for analytical chemistry. Ellis Horwood, New York, 233 pp.
- Mittlefehldt D.W. (1994) The genesis of diogenite and HED parent body petrogenesis. *Geochim. Cosmochim. Acta* **58**, 1537-1552.
- Molini-Velsko C., Mayeda T.K., Clayton R.N., (1986) Isotopic composition of silicon in meteorites. *Geochim. Cosmochim. Acta* **50**, 2719-2726.
- Mohapatra R.K. and Murty S.V.S. (2003) Precursors of Mars: constraints from nitrogen and oxygen isotopic compositions of martian meteorites. *Meteor. Planet. Sci.* **38**, 225-241.
- Morgan J.W. and Anders E. (1980) Chemical composition of Earth, Venus, and Mercury. *Proc. Natl. Acad. Sci.* **77**, 6973-6977.
- Moynier F., Agranier A., Hezel D.C. and Bouvier A. (2010) Sr stable isotope composition of Earth, the Moon, Mars, Vesta and meteorites. *Earth Planet. Sci. Let*. **300**, 359-366.
- Nimmo F. and Agnor C.B. (2006) Isotopic outcome of N-body accretion simulations: constraints on equilibration processes during large impacts from Hf/W observations.
- *Earth Planet. Sci. Lett.* **243**, 26-43.
- O'Neill H.S.C. and Palme H. (1998) Composition of the silicate Earth: implications for accretion and core formation. In: Jackson, I. (Ed.), The Earth's mantle: composition,
- structure, and evolution. Cambridge University Press, Cambridge, pp. 3-126.
- Pahlevan K. and Stevenson D.J. (2007) Equilibration in the aftermath of the lunar-forming
- giant impact. *Earth Planet. Sci. Lett.* **262**, 438-449.
- Pahlevan K., Stevenson D.J., Eiler J.M. (2011) Chemical fractionation in the silicate vapor atmosphere of the Earth. *Earth Planet. Sci. Lett.* **301**, 433-443.
- Palme H. (2000) Are there chemical gradients in the inner solar system? *Space Sci. Rev.* **92**, 237-262.
- 800 Palme H. and O'Neill H.S.C (2003) Cosmochemical estimates of mantle composition. In:
- Carlson R.W. (Ed.), Treatise on Geochemistry, The mantle and core, vol. 2, Elsevier Press, New York, pp. 1-38.
- Papike J.J., Karner J.M., Shearer C.K. and Burger P.V. (2009) Silicate mineralogy of martian meteorites. *Geochim. Cosmochim. Acta* **73**, 7443-7485.
- Poirier J.-P. (1994) Light elements in the Earth's outer core: a critical review. *Phys. Earth Planet. Inter.* **85**, 319-337.
- Poitrasson, F., 2009. Probes of the ancient and the inaccessible. Science, 323: 882-883.
- Poitrasson F., Halliday A.N., Lee D.C., Levasseur S. and Teutsch N. (2004) Iron isotope differences between Earth, Moon, Mars and Vesta as possible records of contrasted accretion mechanisms. *Earth Planet. Sci. Lett.* **223**, 253-266.
- Poitrasson F., Levasseur S. and Teutsch N. (2005) Significance of iron isotope mineral fractionation in pallasites and iron meteorites for the core–mantle differentiation of terrestrial planets. *Earth Planet. Sci. Lett.* **234**, 151-164.
- Reynolds B.C., Aggarwal J., André L., Baxter D., Beucher C., Brzezinski M.A., Engström E.,
- Georg B., Land M., Leng M.J., Opfergelt S., Rodushkin I., Sloane H.J., Van den Boorn
- S.H.J.M., Vroon P.Z. and Cardinal D. (2007). An inter-laboratory comparison of Si
- isotope reference materials. *J. Anal. Atom. Spectrom.* **22**, 561-568.
- Ricolleau A., Fei Y., Corgne A., Siebert J., Badro J. (2011) Oxygen and silicon contents of
- Earth's core from high pressure metal–silicate partitioning experiments. *Earth Planet. Sci. Lett.* **310**, 409-421.
- Rubie D.C., Gessmann C.K. and Frost D.J. (2004) Partitioning of oxygen during core formation on the Earth and Mars. *Nature* **429**, 58-61.
- Russell W.A., Papanastassiou D.A. and Tombrello T.A. (1978) Ca isotope fractionation on the Earth and other solar system materials. *Geochim. Cosmochim. Acta* **42**, 1075-1090.
- Sanloup C., Jambon A. and Gillet P. (1999) A simple chondritic model of Mars. *Phys. Earth Planet. Inter.* **112**, 43-54.
- Savage P.S., Georg R.B., Armytage R.M.G, Williams H.M. and Halliday A.N. (2010) Silicon isotope homogeneity in the mantle. *Earth Planet. Sci. Lett.* **295**, 139-146.
- Savage P.S., Georg R.B., Williams H.M., Burton K.W. and Halliday A.N. (2011) Silicon isotope fractionation during magmatic differenciation. *Geochim. Cosmochim. Acta* **75**, 6124-6139.
- Savage P.S., Moynier F. (2013) Silicon isotopic variation in enstatites meteorites: clues to their origin. *Earth Planet. Sci. Lett.*, *accepted manuscript*.
- Schauble E.A. (2004) Applying stable isotope fractionation theory to new systems. *Review Mineral. Geochem.* **55**, 65-111.
- Shahar A., Ziegler K., Young E.D., Ricolleau A., Schauble E.A. and Fei Y. (2009) Experimentally determined Si isotope fractionation between silicate and Fe metal and implications for Earth's core formation. *Earth Planet. Sci. Lett.* **288**, 228-234.
- Shahar A., Hillgren V.J., Young E.D., Fei Y., Macris C.A. and Deng L. (2011) High temperature Si isotope fractionation between iron metal and silicate. *Geochim. Cosmochim. Acta* **75**, 7688-7697.
- Sherman D.M. (1997) The composition of the Earth's core: constraints on S and Si vs.
- temperature. *Earth Planet. Sci. Lett.* **153**, 149-155.
- Schönbächler M., Carlson R.W., Horan M.F., Mock T.D. and Hauri E.H. (2010) Heterogeneous accretion and the moderatly volatile element budget of Earth. *Science* **328**, 884-887.
- Simon J.I. and DePaolo D.J. (2010) Stable calcium isotopic composition of meteorites and rocky planets. *Earth Planet. Sci. Lett.* **289**, 457–466.
- Spicuzza M.J., Day J.M.D., Taylor L.A. and Valley J.W. (2007) Oxygen isotope constraints on the origin and differentiation of the Moon. *Earth Planet. Sci. Lett.* **253**, 254-265.
- Taylor G.J. and Scott E.R. (2003) Mercury. In: Davis, A.M. (Ed.), Treatise on Geochemistry, Meteorites, Comets, and Planets, vol. 1., Elsevier Press, New York, pp. 477-485.
- Teng F.Z., Li W.Y., Ke S., Marty B., Dauphas N., Huang S.C., Wu F.Y., Pourmand, A.
- (2010) Magnesium isotopic composition of the Earth and chondrites. *Geochim. Cosmochim. Acta* **74**, 4150-4166.
- Touboul M., Kleine T., Bourdon B., Palme H. and Wieler R. (2007) Late formation and prolonged differentiation of the Moon inferred from W isotopes in lunar metals. *Nature* **450**, 1206-1209.
- 859 Trinquier A., Birck J.-L. and Allègre C.J. (2007) Widespread Cr heterogeneity in the inner solar system. *Astrophys. J.* **655**, 1179-1185.
- Van den Boorn S.H.J.M., Vroon P.Z., Van Bergen M.J. (2009) Sulfur-induced offset in MC-ICP-MS silicon-isotopes measurements. *J. Anal. Atom. Spect.* **24**, 1111-1114.
- Wade J. and Wood B.J. (2005) Core formation and the oxidation state of the Earth. *Earth Planet. Sci. Lett.* **236**, 78-95.
- Wang J., Davis A.M., Clayton R.N., Mayeda T.K. (1994) Kinetic isotopic fractionation during the evaporation of the iron oxide from the liquid state. *Lunar Planet. Sci.* **24**, 1479-1480.

- Weyer S., Anbar A.D., Brey G.P., Munker C., Mezger K. and Woodland A.B. (2005) Iron isotope fractionation during planetary differentiation. *Earth. Planet. Sci Lett.* **240**, 251- 264.
- Wetherill G.W. (1986) Accumulation of the terrestrial planets and implications concerning lunar origin. In "Origin of the Moon", Lunar and Planetary Institute, Houston, pp. 519- 550.
- Wiechert U., Halliday A.N., Lee D.C., Snyder G.A., Taylor L.A. and Rumble D. (2003) Oxygen isotopes and the Moon forming giant impact. *Science* **294**, 345-348.
- Wood B.J. and Halliday A.N. (2010) The lead isotopic age of the Earth can be explained by core formation alone. *Nature* **465**, 767-770.
- Wood B.J., Wade J. and Kilburn M.R. (2008) Core formation and the oxidation state of the
- Earth: Additional constraints from Nb, V and Cr partitioning. *Geochim. Cosmochim. Acta* **72**, 1415-1426.
- Xirouchakis D., Draper D., Schwandt C.S., Lanzirotti A. (2002) Crystallization conditions of Los Angeles, a basaltic Martian meteorite. *Geochim. Cosmochim. Acta* **66**, 1867-1880.
- Yoder C.F. (1995) Astrometric and geodetic properties of the Earth and the solar system. In:
- Ahrens, T.J. (Ed.), Global Earth Physics: A Handbook of Physical Constants. American Geophysical Union, Washington D.C., pp. 1-31.
- Young E.D., Galy A., Nagahara H. (2002) Kinetic and equilibrium mass-dependant isotope fractionation laws in nature and their geochemical and cosmochemical significance. Geochim. Cosmochim. Acta 66, 1095-1104.
- Zambardi T., Poitrasson F., Quitté G., Anand M. (2009) Silicon isotope variations in the Earth and meteorites. *Geochim. Cosmochim. Acta* **73**, A1497.
- Zambardi T. and Poitrasson F. (2011) Precise determination of silicon isotopes in silicate rock
- reference materials by MC-ICP-MS. *Geostand. Geoanal. Res.* **34**, 89-99.
- Zhang J., Dauphas N., Davis A., Leya I., Fedkin A. (2012) The proto-Earth as a significant source of lunar material. *Nature Geoscience* **5**, 251-255.
- Ziegler K., Chadwick O.A., Brzezinski M.A. and Kelly E.F. (2005a) Natural variations of
- δ ³⁰ Si ratios during progressive basalt weathering, Hawaiian Islands. *Geochim. Cosmochim. Acta* **69**, 4597-4610.
- 898 Ziegler K., White A.F. and Brzezinski M.A. (2005b) δ^{30} Si systematics in a granitic saprolite, Puerto Rico. *Geol. Soc. Amer.* **33**, 817-820.
- Ziegler K., Young E.D., Schauble E.A. and Wasson J.T. (2010) Metal–silicate silicon isotope
- fractionation in enstatite meteorites and constraints on Earth's core formation. *Earth*
- *Planet. Sci. Lett.* **295**, 487-496.

Figure Captions:

905 Fig. 1: Three-isotope plot showing δ^{29} Si vs. δ^{30} Si for individual analyses of all the samples from this study. The dashed line represents the theoretical mass dependent fractionation slope 907 defined as: δ^{29} Si = δ^{30} Si × 0.5092. The long-term reproducibility (2SD) is shown in the bottom left quadrant of the figure.

910 Fig. 2: δ^{30} Si values relative to NBS-28 for terrestrial and extraterrestrial samples. 911 Uncertainties for individual samples are given as 2SE, computed as $2SD/\sqrt{(n-1)} \times t$, where *t* is the Student's correcting factor (Platzner et al., 1997). BSMa stands for bulk silicate Mars, and BSMo means bulk silicate Moon. All data are from Table 1.

915 Fig. 3: Comparison of published δ^{30} Si data relative to NBS-28 for samples from planetary 916 bodies in the inner solar system. Uncertainties are given as 2SE, computed as $2SD/\sqrt{(n-1)} \times t$, where *t* is the Student's correcting factor (Platzner et al., 1997). AGV-2 data is from Zambardi and Poitrasson, (2011). C.C. = carbonaceous chondrites, E.C. = enstatites 919 chondrites, R.M. = reference materials.

921 Fig. 4: Comparison of δ^{30} Si averages relative to NBS-28 for planetary bodies in the inner 922 solar system from this and previous work. All uncertainties are given as \pm 2SE.

Fig. 5: Model of continuous core formation during Earth's accretion (Wade and Wood, 2005; Corgne et al., 2008; Wood et al., 2008). (a) Variations of *T* of last equilibration (at the base of the magma ocean) and FeO content of the BSE during accretion. The *T* profile is defined by the peridotite solidus (Fiquet et al., 2010). FeO variations correspond to the three-step profile proposed by Wood et al. (2008). Such temperatures and FeO variations lead to modeled depletions of moderately siderophile elements in the BSE consistent with observations (as 930 shown for Ni and V in (b)). (b) Modeled core/BSE variations during accretion for Ni, V and Si using P-T-FeO conditions defined in (a) and parameterizations for metal-silicate partitioning from Corgne et al. (2008) for Ni and Si and Wood et al. (2008) for V. Note the 933 large range of final Si core/BSE ratios (yellow path). (c) Expected variations of Δ^{30} SiBSE-Chond. showing an important level of uncertainty at the end of accretion, due to large uncertainties in elemental metal-silicate partitioning. Note that the prediction is independent from the type of chondrite chosen as reference.

937

938 Fig. 6: Variations of the Δ^{30} Si_{BSE–Chond}/ F_{Core} ratio during continuous core formation for 939 several FeO_{BSE} paths (insert, same abscissa) satisfying constraints from moderately 940 siderophile elements. The blue dashed curve is the three-step model proposed by Wood et al. 941 (2008) and considered in Fig. 5. Taking into account the uncertainty on elemental metal-942 silicate Si partitioning, Δ^{30} Si_{BSE–Chond.}/*F*_{Core} lies along a restricted range at the end of accretion 943 (0.88 ± 0.11) , regardless of the chosen FeO path.

944

945 Fig. 7: Predicted Si core (wt.%) concentrations as a function of measured Δ^{30} SiBSE-Chond. The 946 model (thick curve) is computed using Δ^{30} Si_{BSE-Chond}./*F*_{Core} = 0.88 (see text and Fig. 6). Thin 947 curves show the \pm 0.11 variability of Δ^{30} Si_{BSE-Chond}./*F*_{Core}, which includes the uncertainties on 948 both fO_2 and T variations during the Earth accretion. Experimental constraints (Shahar et al., 949 2009) and metal-silicate Si isotope partitioning in natural samples (Ziegler et al., 2010) were 950 also taken into account. Any mix of chondrite (whatever their type) shall lead to a Δ^{30} Si_{BSE}. 951 chond. that falls on the red section of the curves. The model indicates that the initial Δ^{30} SiBSE-952 Chond. should have been 0.12 if one assumes a maximum possible Si core concentration of 953 7wt% after mantle-core differentiation, which is below the minimum value given by O/C-954 chondrites (0.18). Hence, mantle-core differentiation alone cannot explain the observed 955 different Si isotope composition between chondrites and the bulk silicate Earth if the accreted 956 materials had the composition of chondrites.

957

958 Fig. 8: A) Rayleigh models predicting the variations of δ^{30} Si_{Chond} during a vaporization 959 process that may have occurred along with giant interplanetary impact. The equation is 960 defined as δ^{30} Si_{Solid residue} = δ^{30} Si_{Chond}. $\times (\alpha^{30}$ _{Si}-1) \times ln (*f*), where α is the fractionation factor 961 of Si and *f* is the fraction of Si remaining in the residual solid. The computation involves a Si 962 isotope fractionation factor during vaporization α^{29} si = 0.98985 ± 0.00044 (Knight et al., 963 2009), which has been converted into α^{30} _{Si} = 0.97893 ± 0.00088, assuming the fractionation 964 process is mass dependent. Starting δ^{30} Si_{Chond.} values were calculated from sample 965 measurements and considered as δ^{30} Si_{Bulk Earth}. δ^{30} Si_{Chond.} values were calculated using E-966 chondrites (blue), or O/C-chondrites (red). Solid lines represent the corresponding Rayleigh 967 models. The paths delimited by dashed lines takes into considerations both the uncertainties 968 on δ^{30} Si_{Chond} averages and α^{30} _{Si}. The intersection between the model and the black solid line 969 $(\delta^{30}$ Si_{BSE} value) gives the fraction Si remaining in the Earth residue at the end of the

- 970 vaporization process (*i.e.*, when Δ^{30} Si_{BSE-Chond.} = 0). These 2 models do not consider Si isotope 971 fractionation from mantle-core differentiation, and therefore are considered as an extreme 972 case. B) The same Rayleigh models assuming that the mantle-core differentiation fractionated 973 Si isotopes prior to vaporization, and the Earth's core incorporated 7wt% Si (thus leading to a 974 0.12‰ decrease of the Δ^{30} Si_{BSE-Chond.} value (Fig. 7) translated on the figure into a 0.12‰ shift 975 on the starting materials). According to these latter models, between 0.4 and 1.1wt% on 976 average of Si loss by vaporization is enough to reconcile a chondritic Earth with isotopic 977 observations while keeping a reasonable amount of Si in the core.
- 978

979 **Table Captions:**

980 Table 1: Silicon concentration and isotope data for the Earth, Moon and meteorite samples.

981 Uncertainties are given as \pm 2SE, computed as 2SD/ $\sqrt{(n-1)} \times t$, where *t* is the Student's correcting factor (Platzner et al., 1997) and where n is the number of replicate analyses of a sample. With the exception of Enstatite chondrites, uncertainties on planetary average values are also given as 2SE, computed using the same relation, where n means the number of 985 individual samples (see text for justification). **Given the E-chondrite dataset only gathers 2 samples, a propagated uncertainty calculation was adopted for these data instead of the 2SE 987 calculation. °Data from Zambardi and Poitrasson (2011). *Sample not included in the bulk-silicate-body average calculation. For averages, N means the number of samples.

989

990 Table 2: Si-core concentrations inferred from our model on all available Δ^{30} SiBSE-Chond. data 991 from recent literature and this study. All Δ^{30} Si_{BSE-Chond} uncertainties were propagated from the 992 2SE of both δ^{30} Si_{BSE} and δ^{30} Si_{Chond}. All extreme Si-core concentrations include uncertainties 993 from both the model and the Δ^{30} Si_{BSE-Chond} estimates. *Data computed using the δ^{30} Si_{BSE (Mean)} 994 from Savage et al. (2010), see Table 2 in Armytage et al. (2011). **Data computed using an 995 averaged E-chondrites (EH and EL) δ^{30} Si_{EC} = -0.67 \pm 0.04 (2SE, n=13) from Savage and 996 Moynier (2013). ***Data computed from the average of all available analysis of terrestrial 997 and chondrites samples in recent literature, with the exception of Georg et al. (2007) and 998 Chakrabarti and Jacobsen (2010a) data. The δ^{30} Si_{BSE} also includes all data from Savage et al. 999 (2010).

Table 2

