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Transformation products generated by oxidation of carbamazepine during wastewater treatment processes - Review and Modelling -



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6th IWA International Conference on eco-Technologies for Wastewater Treatment

Introduction – Context

Carbamazepine (CBZ) is an anticonvulsant drug, used for the treatment of epilepsy, bipolar disorder and trigeminal neuralgia.

- Global consumption > 1000 t/year ^[1].
- Poorly removed (< 25%) by conventional secondary processes (500-1000 ng.L⁻¹ in WWTP effluents ^[2])

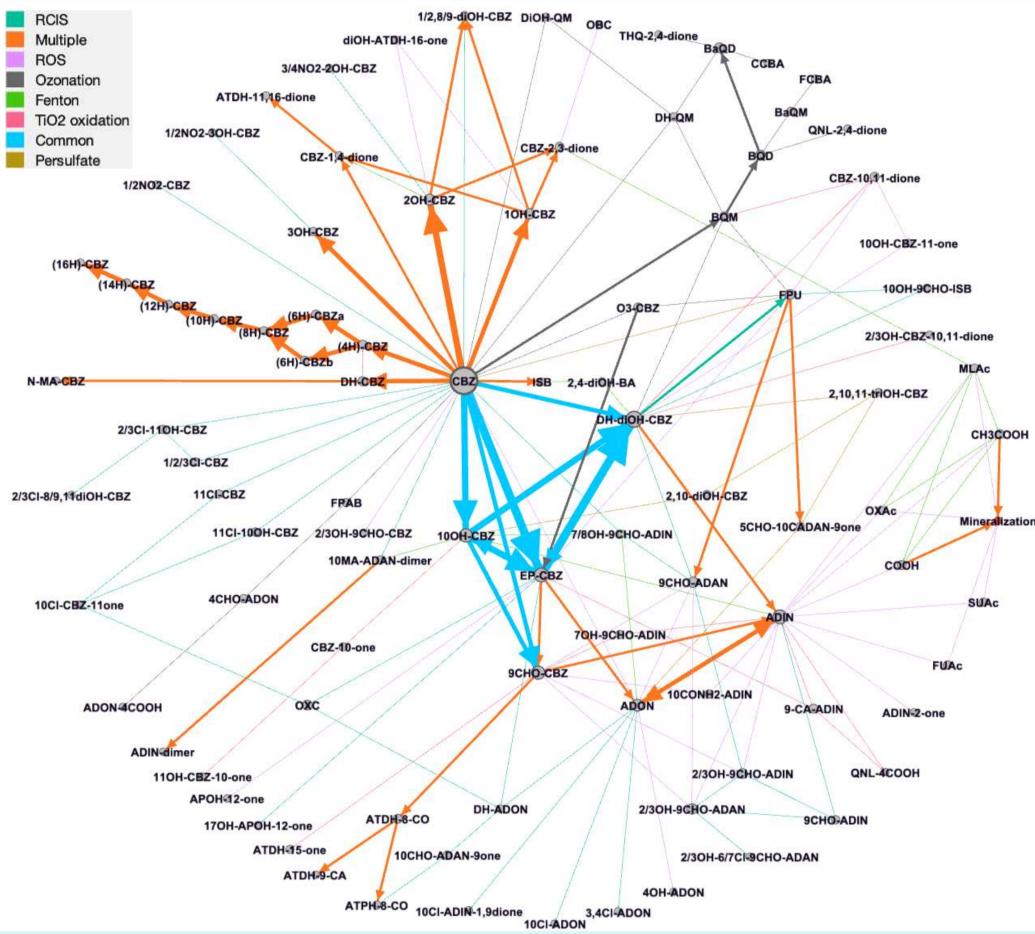
CBZ can be eliminated by polishing treatment processes, such as ozonation or <u>advanced oxidation processes</u> (AOP), known to be responsible for transformation products (TPs) generation ^[3, 4].

- Various TPs (structures, physico-chemicals characteristics, concentrations).
- Many unknown about the fate and behavior of such diversity of TPs

<u>computer-based modelling</u> as a promising tool to predict the fate of MPs and their TPs ^[5, 6].

Aim: improve the knowledge on the nature, fate and dynamics of CBZ and its TPs.

1- TPs generation from CBZ oxidation



Reaction pathways of TPs generated from CBZ oxidation (Drawn with Gephi software 0.9.5, review of n= 34 papers, between 2000 and 2022)

> 100 TPs identified:

4 TPs common to all processes

Key intermediary TPs emerge (ADIN, ADON)

Various, complex reaction pathways:

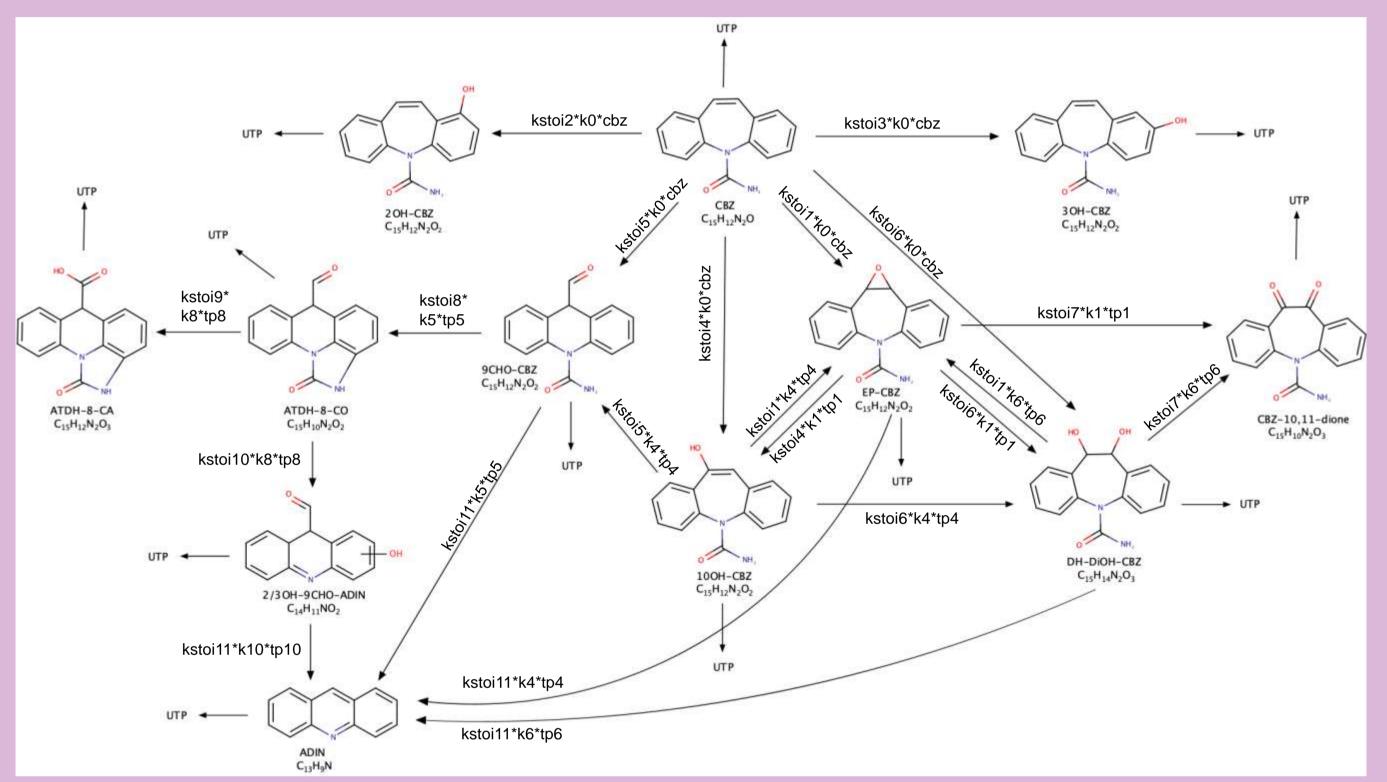
TPs generated from different processes and pathways

Inverse reactions

Diversity of structures and characteristics (hydrophobicity, permeability, ionization state)

<u>3- Unique degradation scheme</u>

One reaction pathway by compilating the 5 datasets New formation and degradation kinetic equations Unique model applied for all datasets



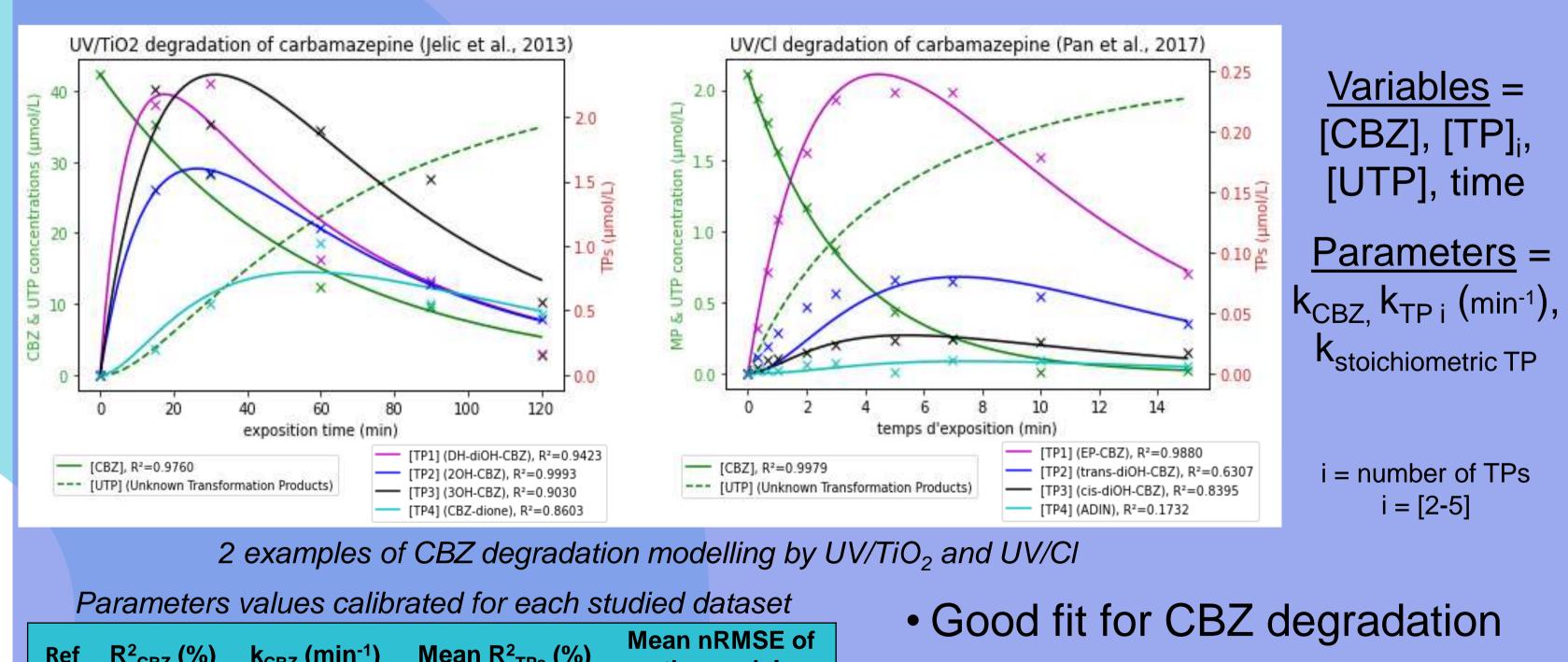
<u>2-Numerical model calibration</u>

Python program => Resolution of ordinary differential equations (ODE) based on formation and degradation kinetic equations:

For CBZ \rightarrow TP : d(TP, t) = $k_{\text{stoi TP}} * k_{\text{CBZ}} * [\text{CBZ}] (TP \text{ formation}) - k_{\text{TP}} * [\text{TP}] (TP \text{ degradation})$

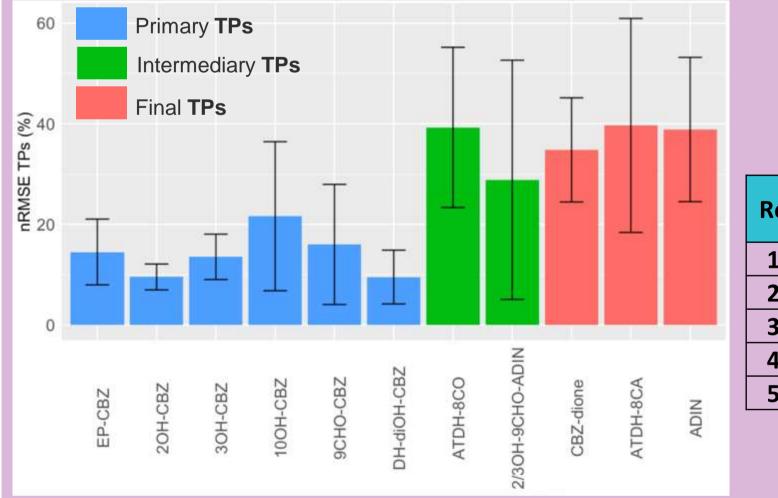
5 datasets: TiO₂ photocatalysis (n = 2), UV/CI (n = 2), UV/H₂O₂ (n = 1)

(1) Franz et al., Catalyst, 2020; 2) Jelic et al., Journal of Hazardous Materials, 2013; 3) Lu & Hu, Sciences of the Total Environment, 2019;
 4) Pan et al., Water Research, 2017; 5) Suara et al., Environmental Science and Pollution Research, 2022)



Unique reaction scheme. Obtained by compilation of the 5 studied datasets and knowledge from literature (Drawn with MarvinSketch [v22.13], UTP = unknown transformation product)

Modelling the 5 experiments with the unique pathway



Main results obtained for the compilated model, for the 5 studied datasets

Ref	nRMSE _{CBZ} (%)	k _{CBZ} (min ⁻¹)	Mean nRMSE of the model (%)
1)	9,8	5.156 x 10 ⁻²	20,3
2)	5,6	1.723 x 10 ⁻²	24,9
3)	3,1	1.876 x 10 ⁻¹	35,9
4)	1,7	3.036 x 10 ⁻¹	12,2
5)	15,6	1.730 x 10 ⁻¹	31,8

mean nRMSE (%) of each TPs, calculated from all

- nRMSE_{primary TPs} < 22% & 30 % < nRMSE_{other TPs} < 40%
- *The further the molecule is from CBZ in the reaction pathway, the harder it is to predict.*
- Mean nRMSE of the model $\leq 35\% \rightarrow$ acceptable fit

Sensitivity analysis on k_{stoi} and k (min⁻¹) parameters:

Model more sensitive towards k_{stoi} (rather than k (min⁻¹))

REI	К ⁻ СВZ (/0)	CBZ (IIIII)	TPs (70)	the model
1)	93,0	5.155 x 10 ⁻²	81 +/- 9% (n=2)	13,97%
2)	97,6	1.723 x 10 ⁻²	93 +/- 6% (n=4)	7,76%
3)	99,3	1.876 x 10 ⁻¹	53 +/- 36% (n=5)	19,08%
4)	99,8	3.027 x 10 ⁻¹	66+/- 35% (n=4)	14,17%
5)	86,6	1.730 x 10 ⁻¹	86 +/- 12% (n=2)	14,12%
	1) 2) 3) 4)	1) 93,0 2) 97,6 3) 99,3 4) 99,8	1) 93,0 5.155 x 10 ⁻² 2) 97,6 1.723 x 10 ⁻² 3) 99,3 1.876 x 10 ⁻¹ 4) 99,8 3.027 x 10 ⁻¹	1) 93,0 5.155 x 10 ⁻² 81 +/- 9% (n=2) 2) 97,6 1.723 x 10 ⁻² 93 +/- 6% (n=4) 3) 99,3 1.876 x 10 ⁻¹ 53 +/- 36% (n=5) 4) 99,8 3.027 x 10 ⁻¹ 66+/- 35% (n=4)

Resulting k_{CBZ} ≈ k literature

- Overall, good quality of TPs
 modeling
- Not sensitive to 3OH-CBZ and DiOH-CBZ parameters variations

CONCLUSION

- Modeling TPs fate through a unique degradation scheme was successful
- Good CBZ degradation modeling and prediction of TPs formation and degradation
- Sensitivity allows to choose the most interesting data to focus on and help for future model calibration
- Perspective: model validation by experimental data on CBZ degradation & TPs formation

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The lack of TPs quantification remains an important challenge

