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Modelling the fate of the transformation products generated by the degradation of carbamazepine in various water treatment processes

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Carbamazepine and water treatment

Carbamazepine (CBZ) = anticonvulsant drug (treatment of epilepsy, bipolar disorder and trigeminal neuralgia)

- Worldwide consumption > 1000 t/year
- Poor removal (< 25%) by conventional secondary processes
- 500-1000 ng.L⁻¹ in WWTP effluents

AOP: radicals attack on CBZ structure

Generation of transformation products (TPs)

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Various TPs (structures, physico-chemicals

Many **unknown** about the fate and

behavior of such diversity of TPs

characteristics, concentrations)



Partial or complete elimination, cyclisation of NH₂ with C6, methylation

Trognon, Albasi and Choubert, Sci. Tot. Env. (2024)

Computer-based modelling

<u>Computer-based modelling</u> as a promising tool to predict the fate of MPs and their TPs

Control MP release



Risk evaluation (environmental impact assessment)

Processes optimization (operational conditions)

Aim: improve the knowledge on the nature, fate and dynamics of CBZ and its TPs

- 1) Identification of CBZ transformation pathways in various oxidation processes and
- 2) Modelling of CBZ transformation products to improve the knowledge of their dynamics

1) Trognon et al, Sci. Tot. Env., 2024

2) Trognon et al, in prep.

Scientific approach: development of individual models



Scientific approach: development of individual models

Source of data	Process	[CBZ] ₀ (µM)	[oxidant]	Time	Data for the TPs	Reaction scheme	
Jelic et al., 2013	UV/TiO ₂	42.3	100 mg.L ⁻¹	120 min	Normalized peak area (A _{TPs} /A _{CBZ})	CBZ \rightarrow CBZ-10,11-dione CBZ \rightarrow 2OH-CBZ 3OH-CBZ	
Suara et al., 2022	UV/CI	42.3	10 mg.L ⁻¹	10 min	[TPs]	CBZ 10OH-CBZ 2OH-CBZ	
Pan et al., 2017	UV/CI	2.1	5 mg.L ⁻¹	15 min	[TPs]	$CBZ \longrightarrow EP-CBZ \xrightarrow{\bullet} trans DH-diOH-CBZ \xrightarrow{\bullet} ADIN$	
Lu & Hu, 2019	UV/H ₂ O ₂	169	3 g.L⁻¹	30 min	Normalized [TPs]/[CBZ] ₀	CBZ ← EP-CBZ → ADIN 9CHO-CBZ → 2/3OH-9CHO-ADIN	
	 3 different processes Diversity of operational conditions				Challenge :	experimental [TPs]	
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Scientific approach: development of individual models



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Results: modelling of CBZ degradation



- good adequation between predicted and experimental data ($R^2 > 86\%$)
- k_{CBZ} resulting from model and k_{CBZ} originally calculated in datasets are comparable
- Pseudo-first order equation appropriate for ³/₄ datasets

Results: modelling of TPs formation/degradation

Reference	TPs	R ² _{TPs} (%)	k _{TPs} (min⁻¹)	k _{stoi}
	DH-diOH-CBZ	94.2	0.134	0.54
Jolia et al. 2012	2OH-CBZ	99.9	0.071	0.24
Jenc et al., 2013	3OH-CBZ	90.3	0.053	0.29
	CBZ-10.11-dione	86.0	0.025	0.12
Sucre et al. 2022	100H-CBZ	77.5	0.003	0.009
Suara et al., 2022	2OH-CBZ	94.3	0.001	0.013
	EP-CBZ	98.8	0.161	0.24
	trans-diOH-CBZ	63.1	0.439	1.00
Pan et al., 2017	cis-diOH-CBZ	83.9	1.230	1.00
	ADIN	17.3	1.110	0.19
	EP-CBZ	18.5	1.001	0.12
	9CHO-CBZ	42.8	2.060	0.05
Lu & Hu, 2019	DH-diOH-CBZ	96.1	0.302	1.00
	2/30H-9CHO-ADIN	85.5	0.152	0.96
	ADIN	22.9	2.312	0.32

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- 11/15 TPs predicted and experimental data fit-well (R² > 63%)
- Quality of the fit decreases for secondary and final TPs (reaction scheme) or lowest concentrations
- Predicted concentrations consistent with the reaction pathway



Results: mass balance calculations



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Scientific approach: development of a global model



Results: modelling of CBZ and TPs

Global model based on an overall reaction scheme – CBZ and TPs results

Mean nRMSE (%) of each TP, calculated using data from all four datasets. Primary, intermediate and final TPs are represented in blue, green and pink, respectively.



Results: modelling of CBZ and TPs

Global model based on an overall reaction scheme – sensitivity analysis

<u>Global model</u> = 12 molecules, 13 differential equations, 23 parameters (kstoi, k)

Which TPs ? Which parameters ?

Help future calibration



Observing the changes in the CBZ and primary TPs prediction upon variations of their parameters

For each parameter studied, 5 different values were used: 0.1, 0.3, 0.5, 0.7 and 0.9 for k_{stoj} , 10⁻⁴, 10⁻³, 10⁻², 10⁻¹ and 10⁰ for *k*

Results: modelling of CBZ and TPs

Global model based on an overall reaction scheme – sensitivity analysis

		Global Model Quality	R ² TPs		
kstoi	EP-CBZ	+++	++		
	2OH-CBZ	+++	++		
	3OH-CBZ	++	+		
	10OH-CBZ	+++	++		Strong influence
	9CHO-CBZ	+++	+	TTT	
	DH-diOH-CBZ	++	+	++	Moderated influence
k (min-1)	CBZ	+	+	+	Low influence
	EP-CBZ	++	++		No influence
	2OH-CBZ	-	++	-	
	3OH-CBZ	-	+		
	10OH-CBZ	-	+		
	9CHO-CBZ	++	+		
	DH-diOH-CBZ	-	++		

- k_{stoi} have more influence on the model quality than k
- 4 Primary TPs appear as important for the overall quality of the model (EP-CBZ, 9CHO-CBZ, 2OH-CBZ and 10OH-CBZ)

Conclusions & Outlooks

- Development of a model resolving ordinary differential equations (ODEs) based on formation/degradation kinetic equations and mass balance approach
- Overall reaction scheme
- Promising results with good correlation between the model predictions and the experimental data for both CBZ and TPs and for both the individual and the global model
- **4 primary TPs** are instrumental for the quality of the predictions: EP-CBZ, 9CHO-CBZ, 2OH-CBZ and 10OH-CBZ. Future model calibrations should focus on the **stoichiometric k**_{stoi} **parameters**, rather than the kinetic constants k.

Outlooks:

Model validation Investigation of the UTP fraction Add more parameters in the model such as macroscopic properties of the TPs (hydrophobicity, ...)

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Thank you for your attention !