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Crystal structure of new quaternary intermetallic compounds R_2 MoSi₂C (R = Y, Gd)

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The discovery of novel quaternary intermetallic compounds R_2 MoSi₂C (R = Y, Gd) in the R-Mo-Si-C system is reported. Their crystal structure was investigated using single-crystal X-ray diffraction. It is shown that the Y₂MoSi₂C and Gd₂MoSi₂C compounds crystallize in the tetragonal space group *P4/mbm* (Pearson symbol tP12) with unit-cell parameters a = 6.9525 (18) Å, c = 4.1962 (16) Å, and a = 6.9945 (16) Å, c = 4.2020 (14) Å, respectively.

1. Introduction

Among solid materials, intermetallic compounds occupy a special place, characterized by great diversity in both their physical properties and crystal chemistry. Within the ternary intermetallics RT_2Si_2 (where R is a rare earth metal and T is a transition metal), the chromium-based compounds were found to be stable for R = Nd, Sm and Gd–Lu (Rieger & Parthé, 1969; Dommann et al., 1988; Dirken et al., 1989; Prots' et al., 1994; Morozkin, 2004) and exhibit strong antiferromagnetic behavior ($T_{\rm N} \ge 600$ K) due to a large magnetic moment on the Cr sublattice (~1.96 $\mu_{\rm B}$), whereas the lanthanide sublattice orders antiferromagnetically at very low temperatures (< 5 K) (Moze et al., 2003; Janatová et al., 2010). The quaternary RCr_2Si_2C compounds (with R = Y, La–Sm, Gd–Er) crystallize in the carbon-filled CeMg₂Si₂-type structure (space group P4/mmm) (Tang et al., 2000; Pohlkamp & Jeitschko, 2001; Janatová et al., 2008; Ma et al., 2022). These carbides exhibit ferromagnetic behavior on the lanthanide sublattice at low temperature for R = Pr, Nd, Gd–Dy while the Cr sublattice seems to be non-magnetically ordered unlike the magnetic behavior previously described for the RCr₂Si₂ compounds (Janatová et al., 2008; Klosek et al., 2008; Ma et al., 2022). The magnetism of CeCr₂Si₂C remains controversial (Tang et al., 2000; Klosek et al., 2008; Mukherjee et al., 2010; Wang et al., 2023).

In this context, the R-Mo–Si–C quaternary equilibrium diagram represents an interesting system, demonstrating intermediate valence phenomena in the CeMo₂Si₂C compound and, more recently, superconductivity in its thorium-based parent compound (Paramanik *et al.*, 2013*a*,*b*; Liu *et al.*, 2021).

In our exploratory research to investigate the possible extension of the layered RMo_2Si_2C (with R = Y, Gd) family, we have unexpectedly discovered two new compounds Y_2MoSi_2C and Gd_2MoSi_2C . In the present paper, we report on their crystal structure.

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Table 1Experimental details.

	Y ₂ MoSi ₂ C	Gd ₂ MoSi ₂ C
Crystal data		
M _r	341.91	478.44
Crystal system, space group	Tetragonal, P4/mbm (No. 127)	Tetragonal, P4/mbm (No. 127)
Phase prototype	Y_2MoSi_2C (tP12)	Y_2MoSi_2C (tP12)
Temperature (K)	293	293
<i>a</i> , <i>c</i> (Å)	6.9525 (18), 4.1962 (16)	6.9945 (16), 4.2020 (14)
$V(\text{\AA}^3)$	202.83 (13)	205.57 (12)
Ζ	2	2
$\rho (\text{g cm}^{-3})$	5.642	7.732
Radiation type	Μο Κα	Μο Κα
θ_{\max} (°)	36.08	35.88
<i>F</i> (000)	308	408
$\mu \ (\mathrm{mm}^{-1})$	31.78	35.25
Crystal size (mm)	$0.03 \times 0.02 \times 0.02$	$0.03\times0.02\times0.02$
Data collection		
Diffractometer	Bruker APEX II QUAZAR CCD	Bruker APEX II QUAZAR CCD
Absorption correction	Multi-scan	Multi-scan
T_{\min}, T_{\max}	0.7, 0.9	0.7, 0.9
No. of measured, independent and observed $[I > 2\sigma(I)]$ reflections	2125, 281, 251	1359, 287, 274
R _{int}	0.052	0.039
$(\sin \theta / \lambda)_{max} (\text{\AA}^{-1})$	0.829	0.824
Refinement		
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.026, 0.059, 1.19	0.019, 0.038, 1.13
Final <i>R</i> 1 ($I > 4\sigma$)	0.0258	0.0198
Final $wR2 \ (I > 4\sigma)$	0.059	0.0733
No. of independent reflections	281	287
No. of observed reflections, restraints, parameters	251, 0, 16	274, 0, 14
$\Delta \rho_{\rm max}, \Delta \rho_{\rm min} \ ({\rm e} \ {\rm \AA}^{-3})$	1.00, -2.23	1.24, -1.66

2. Experimental

Synthesis in a high-frequency induction furnace was undertaken to obtain the quaternary polycrystalline compounds RMo_2Si_2C (with R = Y, Gd). High-purity starting elements ($\geq 99.9\%$) were used. Stoichiometric amounts of elements, for a total sample mass of about 5 g, were melted under a highly purified argon gas atmosphere. Pieces of Y, Gd and Mo were used while Si and C powders were introduced as pellets. Under the synthesis conditions used, the YMo_2Si_2C and GdMo_2Si_2C compounds could not be obtained.

However, the presence of single crystals was detected when observing the as-cast ingots, and then they were subsequently extracted. Very small single crystals were selected under a binocular microscope, glued to MiTeGen MicroLoops and then mounted on a four-circle diffractometer. Data acquisition was performed at 300 K using a Bruker APEX-II Quazar CCD diffractometer equipped with a CCD detector using Mo K α radiation ($\lambda = 0.71073$ Å). Cell refinement and data reduction were carried out using the *APEX2* (Bruker, 2006) software. The structure solution was obtained using direct methods and refined using full-matrix least-squares techniques with *Sir97* (Altomare *et al*, 1999) and *SHELXL2018/3-97* (Sheldrick, 2015) software, respectively, implemented by employing the *WinGX* suite (Farrugia, 2012). Semi-empirical absorption correction using *SADABS* (Bruker, 2004) was applied to the data. The illustrations were made using *VESTA* (Momma & Izumi, 2011).

3. Crystal structure determination and discussion

The crystal structure was determined in the tetragonal lattice system in centrosymmetric space group P4/mbm (No. 127). The refinement parameters and main crystallographic data are



(a) 3D-representation and (b) projection along the b axis of the crystal structure of R_2 MoSi₂C compounds.

Table 2

Atomic coordinates, site occupancy (m_j) and atomic displacement parameters of Y_2MoSi_2C at room temperature.

Site	Atom	x/a	y/b	z/c	mj	U_{11}	U ₃₃	U_{12}	$U_{ m eq}$
4h	Y	0.17653 (5)	$x + \frac{1}{2}$	$\frac{1}{2}$	1.0	0.00328 (19)	0.006(1)	-0.00053(16)	0.00417 (15)
2a	Мо	0	0 2	Õ	1.0	0.0021 (2)	0.0117 (3)	-	0.0053 (2)
4g	Si	0.62546 (16)	$x + \frac{1}{2}$	0	1.0	0.0046 (5)	0.0059(7)	-0.0006(5)	0.0050 (3)
2b	С	0	0 2	$\frac{1}{2}$	1.0	0.008 (2)	0.012 (4)	_	0.009 (2)

 $\bar{T}_{i} = \exp\left[-2\pi^{2} \left(U_{11}h^{2}a^{*2} + U_{22}k^{2}b^{*2} + U_{33}h^{2}c^{*2} + 2U_{12}hka^{*}b^{*} + 2U_{13}hla^{*}c^{*} + 2U_{23}klb^{*}c^{*}\right)\right]; U_{22} = 2U_{12}; U_{13} = U_{23} = 0.$

Table 3

Atomic coordinates, site occupancy (m_i) and atomic displacement parameters of Gd₂MoSi₂C at room temperature.

 $\bar{T}_{i} = \exp\left[-2\pi^{2}\left(U_{11}h^{2}a^{*2} + U_{22}k^{2}b^{*2} + U_{33}l^{2}c^{*2} + 2U_{12}hka^{*}b^{*} + 2U_{13}hla^{*}c^{*} + 2U_{23}klb^{*}c^{*}\right)\right]; \quad U_{22} = 2U_{12}; \\ U_{13} = U_{23} = 0.$

Site	Atom	x/a	y/b	z/c	m _j	U_{11}	U_{33}	U_{12}	$U_{ m eq}$
4h	Gd	0.17668 (3)	$x + \frac{1}{2}$	$\frac{1}{2}$	1.0	0.00396 (11)	0.00376 (13)	-0.00033(9)	0.00390 (9)
2a	Мо	0	0 2	Õ	1.0	0.0043 (2)	0.0055 (3)	-	0.00472 (14)
4g	Si	0.62525 (21)	$x + \frac{1}{2}$	0	1.0	0.0050 (4)	0.0048 (7)	-0.0010(6)	0.0049 (3)
$2\tilde{b}$	С	0	0	$\frac{1}{2}$	1.0	0.018 (19)	0.005 (3)	-	0.0028 (13)

summarized in Table 1. The fractional atomic coordinates, site occupancy and displacement parameters are given in Tables 2 and 3. A perspective representation of the unit cell is depicted in Fig. 1(a) and a projection along the b axis is shown in Fig. 1(b). In this structure, all atoms occupy single crystallographic sites (Tables 2 and 3). The R atoms reside on the 4h site with the point symmetry *m*.2*m*, generated by $(x, x + \frac{1}{2}, \frac{1}{2})$. The Mo and C atoms occupy the 2a (0,0,0) and 2b (0,0, $\frac{1}{2}$) Wyckoff positions, respectively, with an axial symmetry 4/m. The silicon atoms are located on the 4g crystal position with the point symmetry *m.2m*, generated by $(x, x + \frac{1}{2}, 0)$. The tetragonal unit cell consists of two formula units, and altogether it contains 12 atoms. This structure can be described as a C-filled variant of the U₃Si₂ type (Zachariasen, 1948), more precisely it belongs to C-filled Mo₂FeB₂-type (Lukachuk & Pöttgen, 2003) with Y and Gd on the U 4h site, molybdenum on the 2a one and silicon on the silicon 4g site.

This tetragonal structure is characterized by the stacking of alternating R_2C and $MoSi_2$ square planes along the [001] direction.

Interatomic distances between atoms considered as nearest neighbors are gathered in Tables 4 and 5 for Y_2MoSi_2C and Gd_2MoSi_2C , respectively. They have been calculated from single-crystal refinement of the two new quaternary carbides. Their relative evolutions have been evaluated from the relationship:

$$\Delta_{ij} = \left(100 \left[d_{ij} - (r_i + r_j) \right] \right) / (r_i + r_j)$$
(1)

where d_{ij} is the interatomic distance between atoms *i* and *j*, and r_i and r_j , the metallic radius of the corresponding atoms (Teatum *et al.*, 1972; Venturini, 1989).

The results of these calculations show, for both compounds, short Mo–C distances ($\Delta_{Mo-C} = -9.4\%$ in YMo₂Si₂C and -9.3% in Gd₂MoSi₂C). Short bonds between rare earth and silicon atoms ($\Delta_{Y-Si} = -7.1\%$ and $\Delta_{Gd-Si} = -7.6\%$) and rare earth and carbon atoms ($\Delta_{Y-C} = -4.3\%$ and $\Delta_{Gd-C} = -5.5\%$) are also present.

_		
1 a	h	A
10		~

4

Interatomic distances and their relative evolutions in Y2MoSi2C.

		d _{inter} (Å)	Δ (%)
Y (4h)	4 Y	3.616 (2)	+2.6
	1 Y	3.463 (1)	-1.7
	4 Mo	3.3115 (7)	+4.7
	2 Si	2.8622 (13)	-7.1
	4 Si	2.9894 (10)	-3.0
	2 C	2.5620 (7)	-4.3
Mo (2 <i>a</i>)	6 Y	3.3115 (7)	+4.7
	4 Si	2.7462 (10)	+1.0
	2 C	2.0981 (8)	-9.4
Si (4g)	4 Y	2.9894 (10)	-3.0
	2 Y	2.8622 (13)	-7.1
	2 Mo	2.7462 (10)	+1.0
	1 Si	2.4672 (33)	-6.5
C (2b)	4 Y	2.5620 (7)	-4.3
~ /	2 Mo	2.0981 (8)	-9.4

Table 5

Interatomic distances and their relative evolutions in Gd₂MoSi₂C.

		d _{inter} (Å)	Δ (%)
Gd (4h)	4 Gd	3.645 (2)	+1.8
· /	1 Gd	3.495 (1)	-2.4
	4 Mo	3.3250 (7)	+4.2
	2 Si	2.8748 (14)	-7.6
	4 Si	2.9994 (10)	-3.6
	2 C	2.5771 (6)	-5.5
Mo (2 <i>a</i>)	6 Gd	3.3250 (7)	+4.2
	4 Si	2.7650 (10)	+1.7
	2 C	2.1010(7)	-9.3
Si (4g)	4 Gd	2.9994 (10)	-3.6
	2 Gd	2.8748 (14)	-7.6
	2 Mo	2.7650 (10)	+1.7
	1 Si	2.472 (4)	-6.3
C (2b)	4 Gd	2.5771 (6)	-5.5
	2 Mo	2.1010 (7)	-9.3

The coordination polyhedra of each atom were established based on the shortest interatomic distances, and their relative evolutions were determined from single-crystal refinements.



Figure 2 Coordination polyhedra in the R_2 MoSi₂C compounds.

Two types of polyhedra describe the structure: tricapped deformed trigonal prisms and deformed octahedra (Fig. 2). The R atom is located at the center of a deformed trigonal prism of Si atoms, tricapped by two C atoms and one R atom. The deformed trigonal prism around Si is formed by six R atoms and is tricapped by two Mo atoms and one Si atom. The Mo atom is localized at the center of a deformed octahedron,



Figure 3 View of the chain of $MoSi_4C_2$ octahedra along the *c* axis.

with a distorted silicon square plane and two C atoms at the axial positions. Finally, the C atoms are positioned at the center of irregular octahedra, composed of R atoms in square planes and two Mo atoms at the top.

This new crystal structure can be described as chains of $MoSi_4C_2$ octahedra along the *c* axis. These chains of octahedra form channels in which rare earth atoms are found (Fig. 3). The environment of molybdenum atoms is the same as that in the quaternary carbides RMo_2Si_2C structure (Dashjav *et al.*, 2006). This type of environment is also characteristic for chromium both in the RCr_2Si_2C and RCr_3Si_2C compounds (Klosek *et al.*, 2008; Lemoine *et al.*, 2013). All these results show that the stability of the novel structure is mainly based on the existence of strong bonds between Mo and C atoms like in the RMo_2Si_2C compounds without forgetting the R-Si and R-C bonds.

4. Conclusion

Novel intermetallic compounds, Y_2MoSi_2C and Gd_2MoSi_2C , have been discovered during a survey of the *R*-Mo-Si-C phase diagram. Single-crystal X-ray diffraction analyses revealed the crystal structure parameters of a tetragonal cell with space group *P4/mbm*.

Further studies are planned in order to examine both the magnetic behavior of these new quaternary carbides and the existence of this new intermetallic phase with the other rare earth elements.

Conflict of interest

The authors declare no competing interests.

Data availability

The authors confirm that the data supporting the findings of this study are available within the article and its supplementary materials.

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