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# L-PBF processing and characterization of a $\text{Ti}_{35}\text{Nb}_{30}\text{Zr}_{29}\text{Mo}_3\text{Ta}_3$ multiprincipal element alloy for medical implants

Farzin Arjmand,<sup>1\*</sup> Adrien Mourgout,<sup>2</sup> Alima Chali,<sup>1</sup> Madjid Djemai,<sup>3</sup> Credson-Yannick Languet,<sup>4</sup> Olivier Monasson,<sup>5,6</sup> Elisa Peroni,<sup>5,6</sup> Michel Boissiere,<sup>4</sup> Souad Ammar,<sup>1</sup> Guy Dirras<sup>2</sup>

<sup>1</sup> Université Paris Cité, CNRS UR 7086, ITODYS, 75205 Paris, France

<sup>2</sup> Université Sorbonne Paris Nord, CNRS UPR 3407, LSPM, 93430 Villetaneuse, France

<sup>3</sup> Z3DLAB sas, 95270 Chaumontel, France

<sup>4</sup> CY Cergy Paris Université, ERRMECe, 95000 Cergy Pontoise, France

<sup>5</sup> CY Cergy Paris Université, CNRS, BioCIS, 95000 Cergy Pontoise, France

<sup>6</sup> Université Paris-Saclay, CNRS, BioCIS, 92290 Châtenay-Malabry, France

\*Corresponding author E-mail address: farzin.arjmand@gmail.com

## ABSTRACT

This letter reports on the processing, mechanical properties, electrochemical performance and toxicity behavior of a multiprincipal element  $\text{Ti}_{35}\text{Nb}_{30}\text{Zr}_{29}\text{Mo}_3\text{Ta}_3$  alloy. The recovered centimeter-sized cubes or parallelepipeds were polished before analysis. It was found that the homogeneity and yield stress of the alloys get increased when higher volumetric energy densities (VEDs) are applied during processing. The synthesized alloys also exhibited a Young modulus closer to those of bones than those usually reported on the conventional orthopedic metals (60-80 GPa for this alloy against 110 and 4-30 GPa for the Ti-6Al-4V and bone, respectively). The electrochemical corrosion assays performed at 25°C and 37°C, in sodium chloride (NaCl, 0.1 M), deoxygenated high glucose Dulbecco's Modified Eagle Medium (DMEM) and deoxygenated DMEM implemented with Fetal Bovine Serum (FBS) showed corrosion potentials ( $E_{\text{corr}}$ ) ranged in the order  $\text{NaCl}_{(25^\circ\text{C})} \approx \text{DMEM} + \text{PBS}_{(37^\circ\text{C})} > \text{DMEM}_{(37^\circ\text{C})}$ , and impedances much larger in the DMEM and DMEM + PBS media than in NaCl. Finally, pre-osteoblasts cells cultured in the medium conditioned by the alloy did not evidence cytotoxicity effect, because no effects on the cell proliferation and morphology were observed. These data suggest the biocompatibility of this alloy, indicating no acute toxicity, which is a prerequisite for the integration of an alloy as a bioimplant.

Faced with the aging population in our modern societies and the increased risk of bone diseases and accidents in these populations, the development of long-lasting implant materials is a significant

35 public health issue. It is, nowadays, essential to adapt advanced structural materials to extend the net  
36 duration of the revision surgery beyond the current limit of ten years.<sup>1</sup> According to the recent United  
37 Nations reports, the over 60s population of the world is growing, and the proportion of the old persons  
38 is expected to increase by 45% by 2050, with a life expectancy that could gain another five additional  
39 years in the developed countries,<sup>2</sup> making this period of ten years before the surgical revision  
40 extremely short. The problem of the longer-lasting implants also concerns people with acute and  
41 chronic bone diseases while maintaining an acceptable perspective and quality of life.

42 Multiprincipal element alloys (MPEAs) are a new class of materials that offer a range of unique  
43 properties, such as high strength and corrosion resistance.<sup>3,4</sup> These properties make them ideal  
44 candidates for designing medical devices. Moreover, by adjusting the composition of the MPEAs, it is  
45 possible to create materials that are optimized for specific applications. For instance, their Young's  
46 modulus can be reduced to values close to those of bones (20 to 30 GPa), without losing their yield  
47 strength, thereby reducing the probability of the occurrence of osteoporosis near the implant.  
48 Consequently, the MPEAs are increasingly being used in the medical industry for various  
49 applications.<sup>5</sup>

50 In parallel, the emergence of the additive manufacturing (AM) has revolutionized the production of  
51 complex and intricate structures, previously impossible to be manufactured<sup>6</sup>; opening thus real  
52 opportunities for medical implant processing, meaning that AM allows the production of medical  
53 implants tailored to the patient's needs, matching correctly the targeted function.

54 The technology behind the AM is based on the principles of the layer-by-layer deposition of materials.  
55 This means that the implant is built up from a series of thin layers, and each layer is deposited on top  
56 of the previous one. In this way, the AM eliminates the need for the expensive and time-consuming  
57 tooling. It also enables the production of porous structures, which can serve as channels for drug  
58 storage and then controlled delivery,<sup>7</sup> as it allows surface texturing, which can be useful for tissue  
59 integration and rejection risk reduction.<sup>8</sup>

60 Altogether, the AM designed MPEAs-based biomedical implants have the potential to revolutionize  
61 healthcare.<sup>9</sup> Combining these two innovative concepts, the implants composition and morphology

62 would be easily tailored to meet patients' anatomy, age, and gender, laying the ground for personalized  
63 treatment.

64 In this study, a series of the MPEA with the nominal composition of  $\text{Ti}_{39}\text{Nb}_{28}\text{Zr}_{28}\text{Mo}_{2.5}\text{Ta}_{2.5}$  (at. %) were elaborated by the laser powder bed fusion (L-PBF), using the selected laser melting (SLM) technology (see Table 1S for the actual achieved composition). By adjusting the fabrication parameters, especially the volumetric energy density (VED) during the SLM processing, the Young's modulus of the alloy was significantly reduced, without losing its yield strength, reducing thereby the probability of osteoporosis occurrence near the implant.

70 Through the processing optimization, several types of microstructural defects were identified (experimental details in the supplementary data). These were mainly unmelted powders, cracks, and pits.

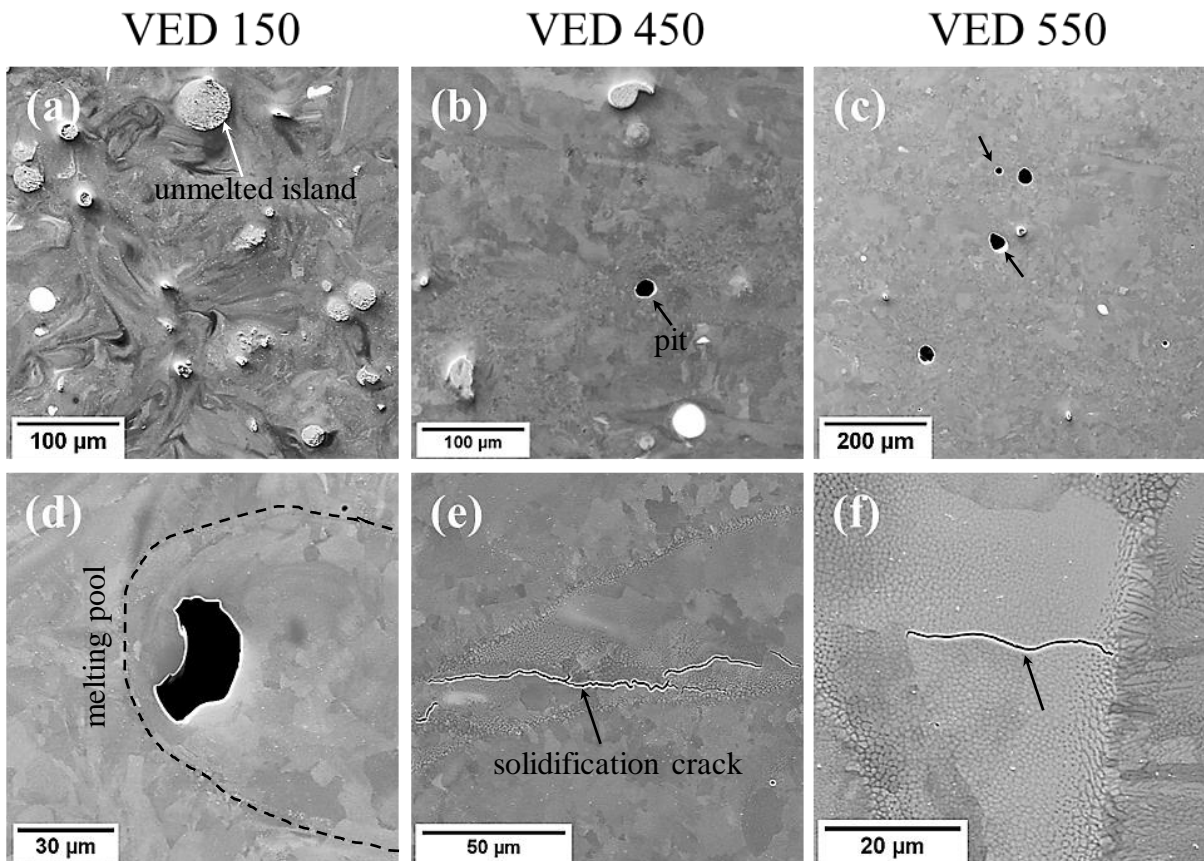
73 Melting elements like Ta, Mo and Nb with high melting points (3020, 2623 and 2477°C, respectively) is obviously arduous. The low VED resulted in the accumulation of unmelted Nb/Mo/Ta-rich islands. Scanning electron microscopy (SEM) images clearly evidence these entities by their after polishing relief and light grey contrast (as well as the Ta/Ta-rich islands by their bright light contrast), as shown in Figs. 1(a) and 1S.

78 Increasing the VED induced the Nb/Mo/Ta-rich islands solubilization, improving the homogeneity of the final alloy (Fig. 1(b, c)). Nevertheless, the solubilization of Nb/Mo/Ta into the matrix triggers the vaporization of Ti, since the melting point of Ta is close to the boiling point of Ti (3287°C). As a consequence, typical processing defects, such as keyhole porosities form. The higher the VED, the higher the keyhole porosities fraction was in the structure of the alloy. Their appearance frequency, means the coalescence of the cavities while porosities facilitate the degradation of the mechanical properties of the alloy.<sup>12</sup> Contrariwise, cracks cause less dramatic effects on the mechanical properties of the material. Cracking occurs either as a result of the residual stress relaxations (induced by the high cooling rate of the SLM) or the complex solidification mechanisms (involving the solidification range and liquid viscosity) (Fig. 1(e, f)). The second case usually forms short/thin intergranular cracks, primarily along the high-angle grain boundaries (HAGBs).

89 The VED affects also the grain morphology, size and size distribution. The inverse pole figure (IPF)  
90 electron backscatter diffraction (EBSD) images recorded on the specimens, elaborated at the VED =  
91 150, 450, and 550 J.mm<sup>-3</sup>, clearly evidence it (Fig. 2(a-c)). These images show heterogeneous grain  
92 shapes. The degree of heterogeneity decreases with the applied VED. For the VED = 550 J.mm<sup>-3</sup>, a  
93 relatively bimodal character was observed, while for the VED = 450 J.mm<sup>-3</sup>, a slight <100> texture  
94 was noted.

95 On a finer scale, the grains appeared as made up of dendritic cells whose sizes increased with  
96 increasing the VED, from 429 up to 516, and 740 nm, for the VED = 150, 450 and 550 J.mm<sup>-3</sup>,  
97 respectively (Fig. 2(d-f)). The statistical details of the microstructure of the alloys can be seen in Fig.  
98 2(g, h). These observations are in agreement with the existing literature data,<sup>10,11</sup> where it has been  
99 reported that the walls between the cells are made of accumulated geometrically necessary  
100 dislocations, inducing the disorientations in the order of 2 degrees. Furthermore, the dendritic cell  
101 walls were rich in Zr, while Mo was accumulated mostly inside the cells (Fig. 2S).

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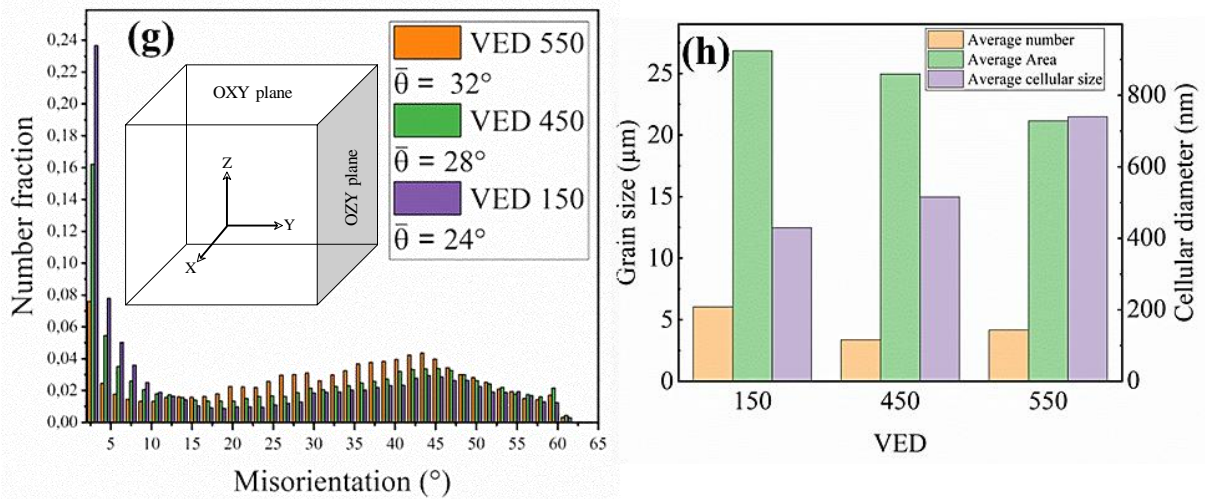
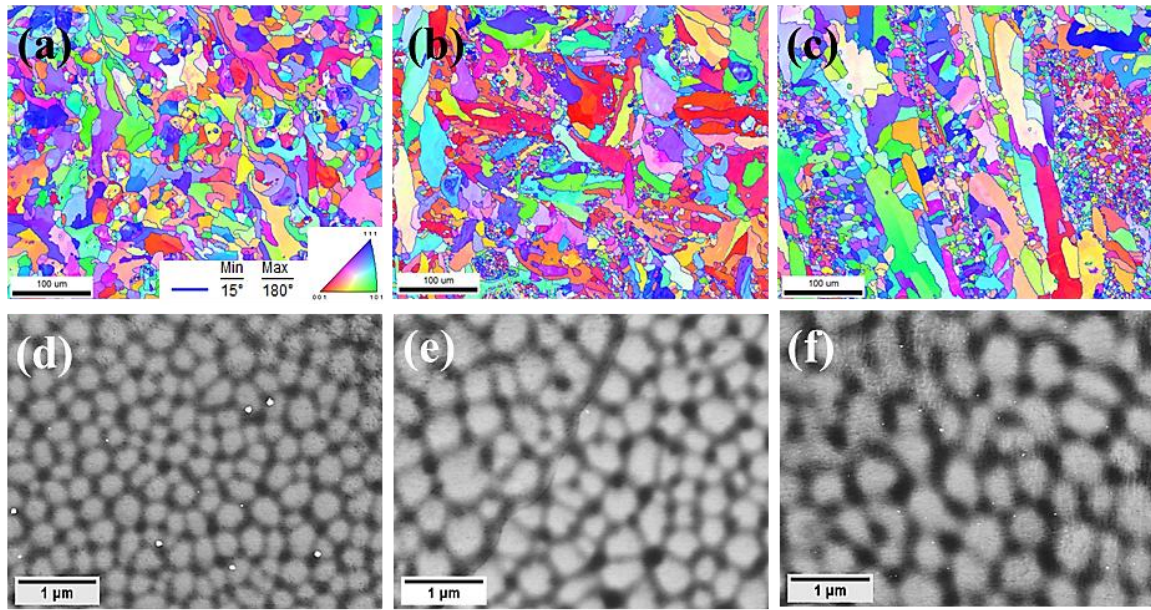
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104 **FIG. 1.** Nature of the processing defects observed as a function of the applied VED. (a, d)  $150 \text{ J.mm}^{-3}$ ,  
105 (b, e)  $450 \text{ J.mm}^{-3}$  and (c, f)  $550 \text{ J.mm}^{-3}$ .  
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108 The number of the HAGBs increased with increasing the VED. The average degrees of  
109 misorientations were 24, 28, and  $32^\circ$  for the VED = 150, 450, and  $550 \text{ J.mm}^{-3}$ , respectively (Fig. 2(g)).  
110 The high fraction of low angle grain boundary (LAGB) misorientation at low VED is in line with the  
111 cells size given in Fig. 2(h). Indeed, smaller cell sizes (longer cell walls) increased the degree of the  
112 low angle misorientation. The increase in the HAGBs' percentage coincided with the solidification  
113 cracking probably due to the increase in the disorientation between the contiguous grains, in  
114 agreement with the literature data.<sup>12</sup>

115 Apart from the evolution of the size of the dendritic cells which increased with the VED (Fig. 2(d-f)),  
116 the average grain size of the alloys (measured as area fraction) decreased as the VED increased. The  
117 latter is attributed to the bimodality of the alloy structure (Fig. 2(a-c)).

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119

120 **FIG. 2.** Microstructure quantification details. (a-c) IP EBSD images superimposed to the grain  
 121 boundaries map of the VED = 150, 450 and 550 J.mm<sup>-3</sup> specimens, respectively, (d-f) SEM images of  
 122 the cellular substructure of each specimen and (g, h) distribution of the misorientation across the grain  
 123 boundaries, frequency of grain, and cellular size as a function of the applied VED (Z axis in Fig (g)  
 124 indicates the laser beam direction).  
 125

126 The results of the compression and tension tests can be compared in Fig. 3(a, b). See also experimental  
 127 details in the supplementary data. Fig. 3(a) depicts that all the as-processed alloys possess a high  
 128 ability to get plastically deformed. Indeed, all the specimens exhibited nominal strains well above  
 129 70%. The yield stresses were found to be in the range from 900 to 1350 MPa, increasing with the VED  
 130 increase. As expected, the above-mentioned processing defects did not compromise the use of these  
 131 alloys in the loading conditions. Contrariwise, the tensile tests showed a high sensitivity to the  
 132 microstructure. The higher the VED, the higher were the yield stress and yield strength values. For

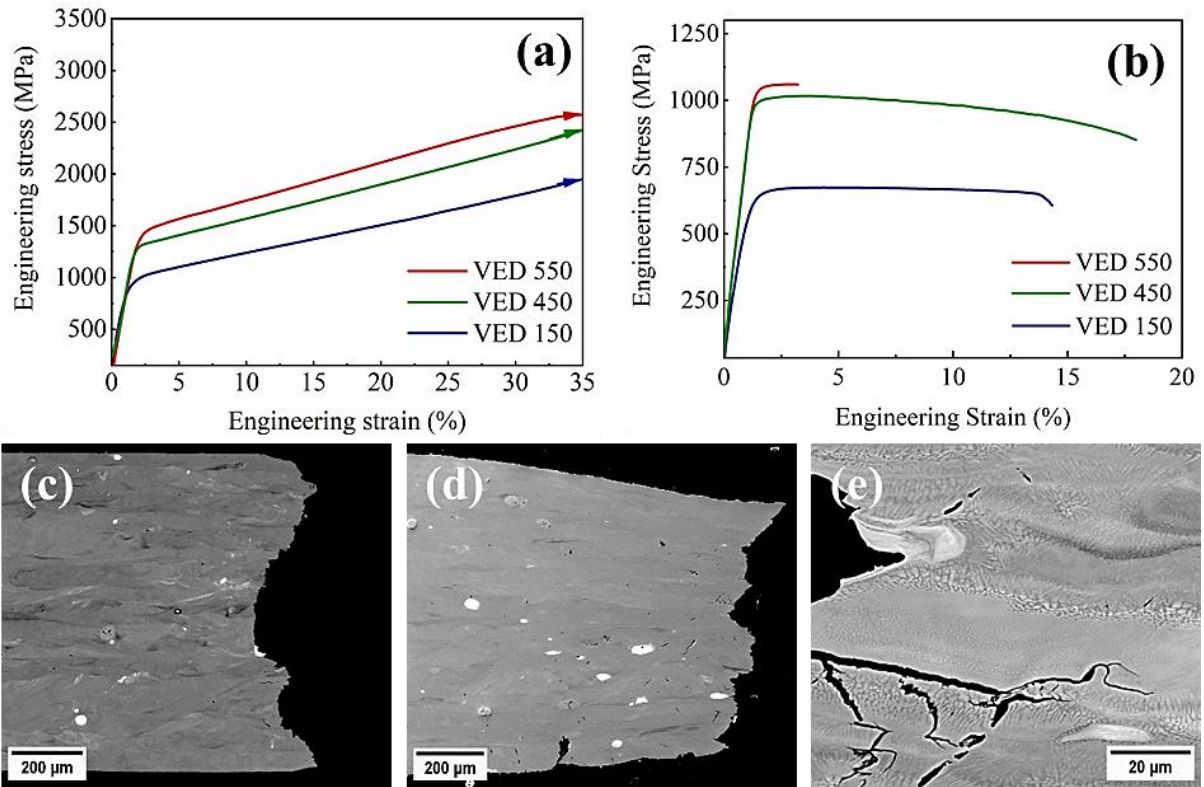
133 instance, the  $VED = 550 \text{ J.mm}^{-3}$  led to a relatively poor ductility. The related specimen is in the regime  
134 with the highest keyhole porosities frequency, along with the solidification cracking. The coalescence  
135 of these cavities during the straining and bridging of the microcracks into a three-dimensional  
136 network, contributed to the observation of the lack of ductility. Fig. 3(c, d) illustrates the necking  
137 regions of the 550 and 450  $\text{J.mm}^{-3}$  specimens, respectively. The elongation of the Ta-rich entities into  
138 a lenticular shape in the latter is evident; since opposite to the 550  $\text{J.mm}^{-3}$  specimen, here they  
139 contributed to the overall ductility of the material. Images of the generated microcracks in the 450  
140  $\text{J.mm}^{-3}$  specimen are also given in Fig. 3(d, e).

141 Here the increase in the yield strength by increasing the VED can be explained by a Hall-Petch-like  
142 effect combined with a solid solution strengthening. The Hall-Petch-like effect comes from the  
143 reduction of the grain size while the origin of the solid solution strengthening comes from the  
144 solubilization of Nb/Mo/Ta in the matrix.

145 Table 1 summarizes the major compression and tensile loading parameters of the alloys. Here the  
146 Young's modulus  $E$  values (60 to 80 GPa) are much lower than those of the materials that are  
147 currently used for the orthopedic applications (CP-Ti: 105, Ti-6Al-4V: 115 and Co-Cr: 240 GPa). The  
148 low  $E$  values are expected to eliminate the stress shielding effect of the material.<sup>13,14</sup> Another  
149 biomechanical indicator for the metallic implant materials, is the elastic strain deformation  $\varepsilon$ .<sup>15</sup> It was  
150 computed, for this study, from the tensile test data as  $\varepsilon = \sigma_{YS}/E$ . The obtained values were found to be  
151 varying in the range from 0.87 to 1.22%, larger than those of the CP-Ti and Ti6Al4V (0.47 and 0.78%,  
152 respectively), but closer to that of the as-cast TiZrNbTa (1.26%).<sup>16</sup>

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**FIG. 3.** (a) Typical stress-strain compression test plots of the synthesized alloys along the Z axis/OXY plane (building direction), (b) tensile test plots along the OYZ plane, (c, d) failure profiles of the tensile tested 550 and 450 J.mm<sup>-3</sup> specimens, respectively and (e) crack propagation along the HAGBs in the tensile tested 450 J.mm<sup>-3</sup> specimen.

160 **Table 1.** Some mechanical parameters of the 150, 450 and 550 J.mm<sup>-3</sup> specimens.

specimen	Young modulus [GPa]	ultimate tensile stress [MPa]	yield strength [MPa]	strain to failure [%]	compression yield strength (MPa)
VED = 150 J.mm <sup>-3</sup>	60	674	527	13.38	940
VED = 450 J.mm <sup>-3</sup>	78	1016	950	17.99	1215
VED = 550 J.mm <sup>-3</sup>	80	1061	964	3.23	1335

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Altogether, these results suggest that the 150 and 550 J.mm<sup>-3</sup> specimens are the best according to the  $E$  and  $\epsilon$  criteria, respectively; for the desired application, while the 450 J.mm<sup>-3</sup> one offering finally the best compromise. In the same time, the 550 J.mm<sup>-3</sup> specimen exhibited the best chemical homogeneity and then very probably a better corrosion performance. The persistence of a larger number of the Nb/Mo/Ta-rich islands in the 450 J.mm<sup>-3</sup> specimen compared to the 550 J.mm<sup>-3</sup> one, might favor pitting corrosion, as a consequence of the (electro)chemical reactivity differences between these islands and the matrix. Also differences between the biological behaviors might be expected between

169 the islands and matrix, when contacted to bone cells. Consequently, to pursue our investigations, a  
170 focus was made on the  $550 \text{ J.mm}^{-3}$  specimen.

171 The electrochemical measurements were thus performed on both  $150$  and  $550 \text{ J.mm}^{-3}$  specimens to  
172 investigate their corrosion performance. The specimens were exposed to a  $0.1 \text{ M NaCl}$  ( $25^\circ\text{C}$ ) solution  
173 and to a  $37^\circ\text{C}$  deoxygenated Dulbecco's Modified Eagle mammal cell culture medium (DMEM) or  
174 DMEM supplemented with  $20\%$  of fetal bovine serum (FBS) (experimental details in the  
175 supplementary data). After  $\sim 24 \text{ h}$  of exposure, their potentiodynamic polarization curves and  
176 impedance plots were recorded (Figs. 4(a, b) and 3S(a-d)). Each open circuit potential (OCP) value,  
177 given in the inset of the figures is the average potential of at least three electrodes, before being  
178 polarized. The corrosion potential ( $E_{\text{corr}}$ ), as well as the corrosion current density ( $I_{\text{corr}}$ ) values for each  
179 coupon in each electrolyte are given in these figures. Along with the potentiodynamic polarization  
180 curves, some selected SEM images of the pits nucleated during the polarization of the coupons in the  
181 DMEM and NaCl electrolytes can be seen in Figs. 4(a) and 3S(b). (more images can be found in Fig.  
182 4S).

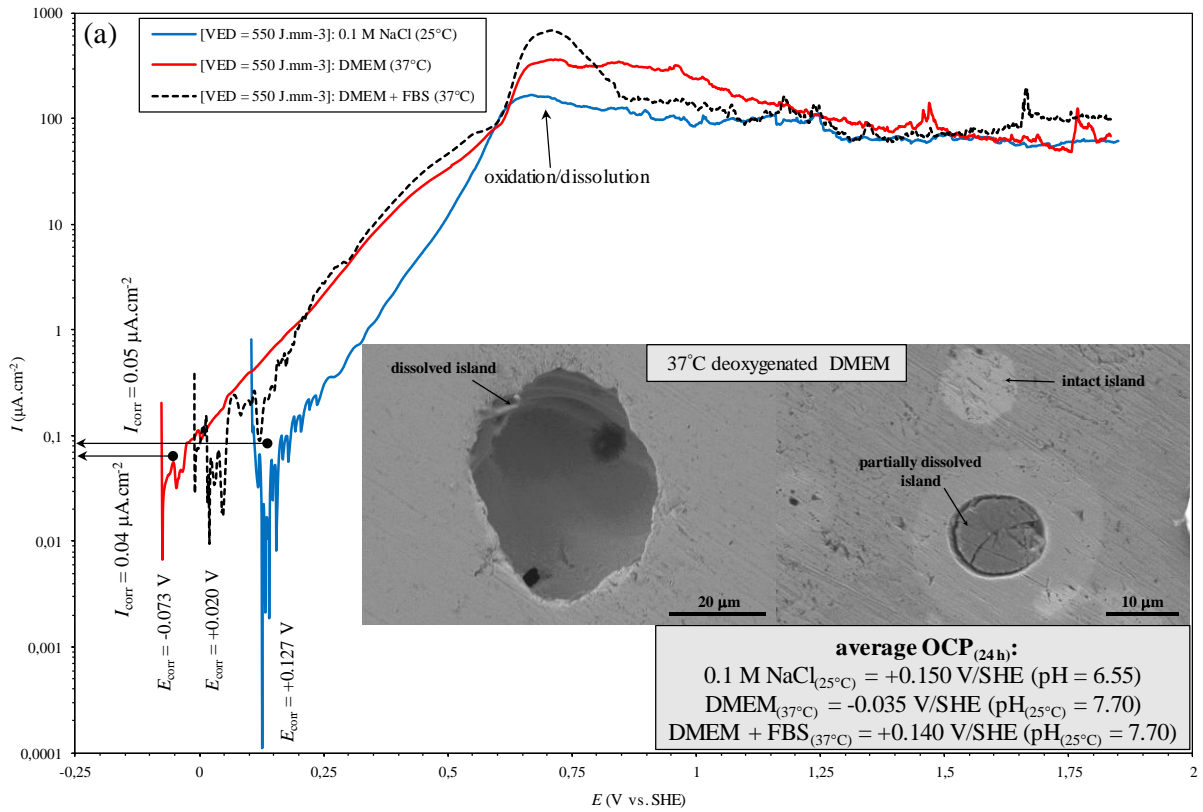
183 Interestingly, the anodic branches of all the curves (Figs. 4(a) and 3S(b)) exhibited the generation of a  
184 large oxidation current, especially in  $\sim +0.7 \text{ V/SHE}$ . This indicates that by polarizing the coupons  
185 towards more positive potentials, a considerable degree of dissolution occurs on their surfaces. The  
186 OCP of the  $550 \text{ J.mm}^{-3}$  alloy in the NaCl ( $\text{pH}_{(25^\circ\text{C})} = 6.55$ ) and DMEM + FBS ( $\text{pH}_{(25^\circ)} = 7.70$ ) solutions  
187 is the most positive ( $+0.150 \text{ V/SHE}$ ). This potential in the DMEM ( $\text{pH}_{(25^\circ\text{C})} = 7.70$ ) solution changes to  
188  $-0.035 \text{ V/SHE}$ . The curves in Fig. 4(a) also depict that although the  $I_{\text{corr}}$  recorded in the NaCl solution  
189 is slightly larger than those of the DMEM and DMEM + FBS solutions, the alloy generated a smaller  
190 oxidation current in this solution, compared to the two others.

191 The recorded Nyquist plots of the  $550 \text{ J.mm}^{-3}$  coupons, show that the resistivity of the alloy in the  
192 three solutions is in the order DMEM + FBS  $>$  DMEM  $>$  NaCl. This can be seen in the polarization  
193 resistance and imaginary impedance values in Fig. 4(b). The plot recorded in the NaCl solution has an  
194 incomplete/deformed semi-circle which can be fitted with an electrical circuit model based on a  
195 constant phase element (CPE), parallel to a charge transfer resistance ( $R_{\text{ct}}$ ) element; both in series with  
196 a solution resistivity ( $R_s$ ) element. In DMEM, the recorded plot consists of only a deformed semi-

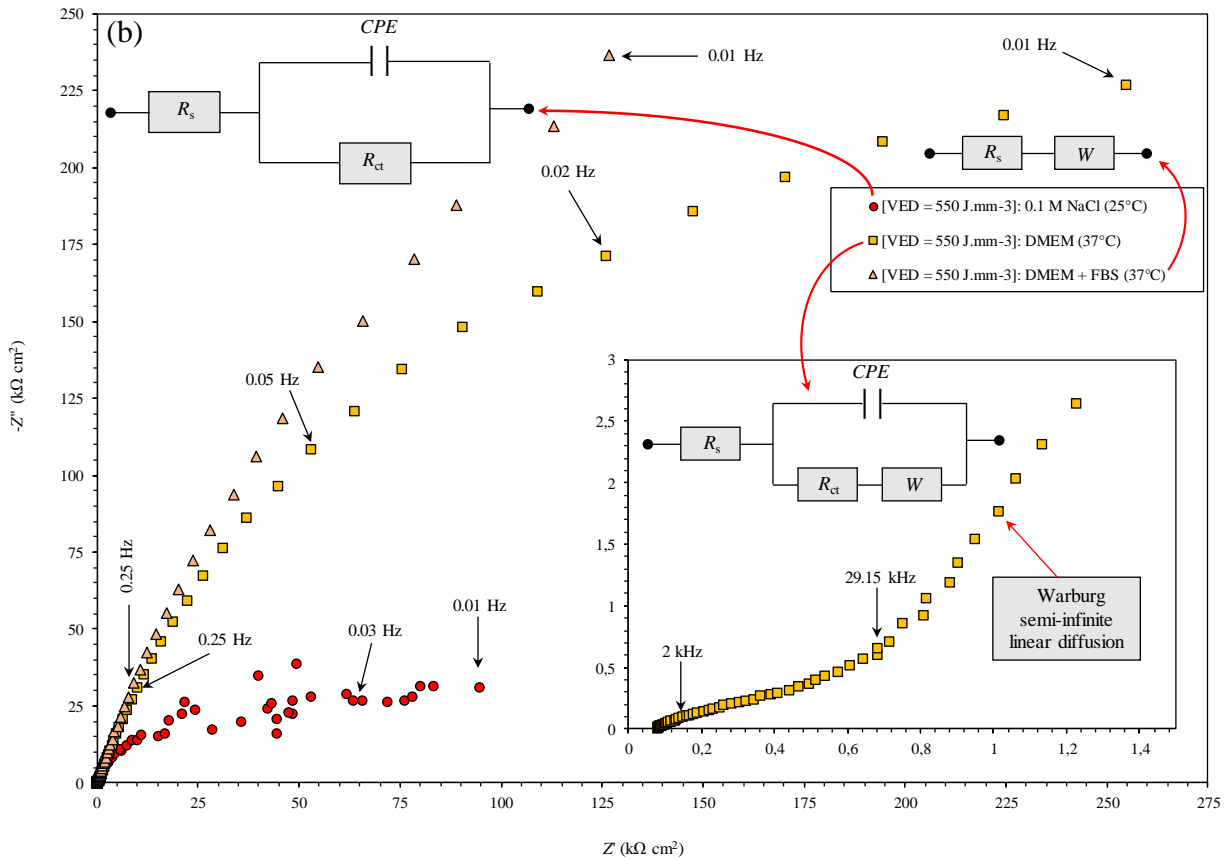
197 infinite Warburg linear diffusion impedance, that is represented with a Warburg element ( $W$ ) in series  
198 with a  $R_s$  (Fig. 4(b)). Here it should be noted that the slope of a classic Warburg plot is  $45^\circ$  and plots  
199 with degrees  $> 45^\circ$  indicate the existence of a greater mass transport limitation. In DMEM + FBS, the  
200 recorded plot has an incomplete/deformed semi-circle in the high frequency region, followed with a  
201 linear diffusion Warburg impedance, which can be modelled with an electrical circuit based on a  $W$   
202 element in series with a  $R_{ct}$  and parallel to a  $CPE$ .

203 Comparing the Bode total impedance ( $|Z|$ ) versus frequency plots of the two alloy coupons in the  
204 three electrolytes (Fig. 3S(a, d)) allows evidencing that the  $|Z|$  of the  $550 \text{ J.mm}^{-3}$  coupon at the  
205 minimum applied frequency is the maximum and minimum for the DMEM and NaCl solutions,  
206 respectively. While, in the case of the  $150 \text{ J.mm}^{-3}$  coupon the largest and smallest minimum frequency  
207  $|Z|$  belongs to the plots recorded in the DMEM and DMEM + FBS.

208 To resume the overall electrochemical results, it is clear that the  $550 \text{ J.mm}^{-3}$  alloy, has a better  
209 resistivity to corrosion compared to the  $150 \text{ J.mm}^{-3}$  alloy, especially in the DMEM and DMEM + FBS  
210 media, at least for the 24 h of exposure, which is useful for the desired application.



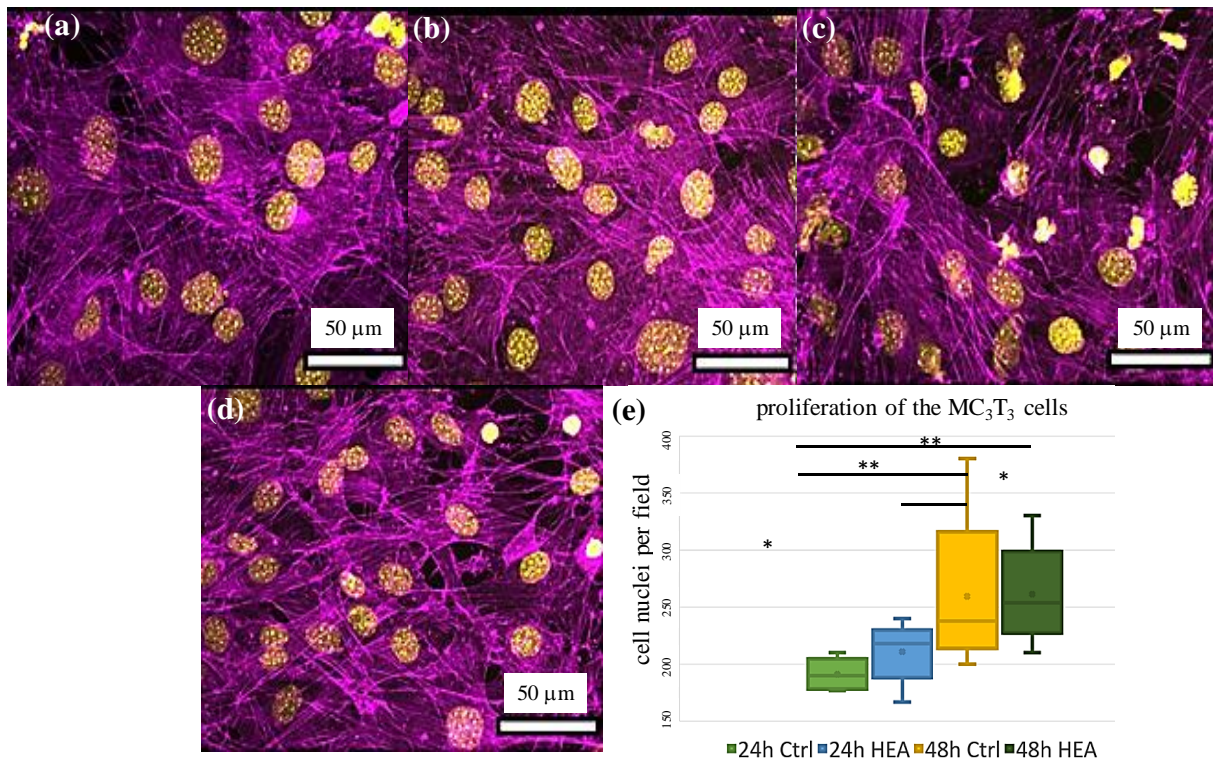
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213 **FIG. 4.** (a) Potentiodynamic polarization curves and (b) Nyquist plots of the 550 J.mm<sup>-3</sup> alloy (in the  
214 OXY plane), recorded after 24 h exposure to a 0.1 M NaCl (25°C), a deoxygenated DMEM (37°C) and

215 a deoxygenated DMEM + FBS (80:20 (v/v)) (37°C) media. Inset: SEM images of the pits, nucleated  
216 during the polarization of the alloy in DMEM.  
217  
218 Finally, the toxicity of the 550 J.mm<sup>-3</sup> alloy was evaluated by studying the cell proliferation and  
219 morphology in the alloy conditioned and normal media. The conditioned media were prepared by  
220 immersing a freshly prepared 550 J.mm<sup>-3</sup> coupon for 24 h in a 10 mL DMEM-F<sub>12</sub> medium.  
221 Cells were thus stained (their nucleus and actin cytoskeleton in yellow and purple, respectively) (the  
222 experimental details in the supplementary data) and optical images were collected and compared (Fig.  
223 5). The images in Fig. 5 revealed no morphological differences between the cells cultured in the  
224 control medium compared to those cultured in the medium conditioned by the alloy. This was  
225 consistent with the study of the cell proliferation over 48 h. The number of the viable cells is nearly  
226 identical between the cell populations, regardless of whether they were exposed to the conditioned  
227 medium or not.  
228 These observations are a strong indication of no acute toxicity of the alloy, suggesting its  
229 biocompatibility.  
230 Since it is commonly accepted that the cell proliferation performance gets enhanced by modification  
231 of the surface morphology,<sup>17,18</sup> the effect of the surface microstructure, in term of roughness, will be  
232 further investigated. Application of the bioactive peptide coatings is also currently under investigation  
233 to improve the cell viability, adhesion and proliferation.<sup>19</sup> Up to now, only nude coupons were tested.  
234 Here despite the lack of the surface optimization, the biological preliminary results supported the  
235 choice of the elaborated alloy as an implant material.  
236



237

238 **FIG. 5.** Optical microscopic images of the MC<sub>3</sub>T<sub>3</sub> A A pre-osteoblasts cells cultured in the  
 239 nonconditioned ((a) 24 h, (b) 48 h) and conditioned media ((c) 24 h, (d) 48 h) and (e) the proliferation  
 240 graphic (the line within the box represents the median). ANOVA test results: n = 3, \* p < 0.05, \*\* p <  
 241 0.01.

242

243 In summary, the L-PBF processed Ti<sub>35</sub>Nb<sub>30</sub>Zr<sub>29</sub>Mo<sub>3</sub>Ta<sub>3</sub> alloys were investigated, optimizing the  
 244 elaboration parameters, mainly VED, to tentatively reach a valuable microstructure, mechanical  
 245 properties, electrochemical corrosion parameters and a biological compatibility toward mammal  
 246 osteoblast cell proliferation, for a successful integration as bone bioimplants.

247 Increasing the VED induced the Nb/Mo/Ta solubilization, improving the chemical homogeneity of the  
 248 processed alloys. Besides, the alloys had a high ability to get plastically deformed. They also exhibited  
 249 nominal strains of >> 70%, Young modulus values much lower than those of the conventional implant  
 250 alloys, elastic strains close to those of bones and yield stresses from 900 to 1350 MPa, that increased  
 251 with increasing the VED, proving the capability of the alloy to be used as a bioimplant material.

252 Focusing on the specimens processed at high VED, their OCPs in the NaCl and DMEM + FBS media  
 253 were found to be more positive, compared to DMEM, while the impedance parameters of the alloy  
 254 were much more desirable in contact with the DMEM and DMEM + FBS. Finally, the same

255 specimen, did not induce acute toxicity and did not inhibit cell proliferation, proving its  
256 biocompatibility, supporting their valuability as an implant material.

257

## 258 **Supplementary Material**

259 The supplementary data includes the detailed information about the elaboration process,  
260 microstructure/mechanical characterization, electrochemical corrosion performance and cytotoxicity  
261 behavior of the synthesized alloys.

262

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267

## 268 **Author Contributions**

269 Equal: conceptualization, Equal: data curation, formal analysis, writing original draft and project  
270 administration.

271

## 272 **Data Availability**

273 The data that support the findings of this study are available within the letter and its supplementary  
274 material.

275

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