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## Phenolic Dienes as Highly Selective Dienophiles in Asymmetric Organocatalyzed Three-Component Vinylogous Povarov Reaction

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#### ABSTRACT



Our study presents a novel enantioselective route for 1,2,3,4-tetrahydroquinolines (THQs) synthesis via a chiral phosphoric acid-catalyzed 3-component Povarov reaction, employing phenolic dienes as dienophiles. This approach produces a diverse array of 2,3,4-trisubstituted tetrahydroquinolines, each featuring a styryl group at the 4-position, in high yields with excellent regio-, diastereo- and enantioselectivities (>95:5 dr and up to >99% ee).

#### **INTRODUCTION**

1,2,3,4-tetrahydroquinolines (THQs) stand as significant heterocycles in medicinal chemistry, highlighted by their prevalence in a wide array of natural and synthetic compounds exhibiting biologically active properties (Figure 1).<sup>1,2</sup> Among the array of available synthetic methodologies the Povarov reaction, operating as an inverseelectron demand aza-Diels-Alder (IEDDA)<sup>3</sup> reaction between electron-rich dienophiles and *N*-arylimines, has emerged as a powerful synthetic method for constructing THQs with high structural diversity and excellent diastereoselectivity in the majority of instances.<sup>4</sup> It's therefore no surprise that numerous research groups directed their efforts toward developing asymmetric variants. Following seminal work by Kobayashi,<sup>5</sup> which introduced the first asymmetric version using a Yb/BINOL catalyst, alternative efficient methods based on chiral Lewis acids have been explored for producing enantioenriched THQs.<sup>6</sup> Moreover, a diverse array of enantioselective organocatalytic approaches have been investigated. Notably, chiral ureas and proline derivatives have shown high efficiency.<sup>7</sup> However, chiral phosphoric acid (CPA) catalysts, particularly those based on a BINOL or SPINOL skeleton,<sup>4a</sup> have demonstrated broader applicability in asymmetric Povarov reactions, following Akiyama's pioneering work showcasing their high efficiency in achieving highly enantioselective reactions between vinyl ethers and *N*-arylimines.<sup>8</sup> For instance, our group developed a three-component (3-C) Povarov reaction catalyzed by chiral phosphoric acid, utilizing enecarbamates or enethioureas as dienophiles to access 4-amino-THQs with excellent enantioselectivities and complete diastereoselectivity (Scheme 1a).<sup>9</sup> Subsequently, other research groups have further developed this Povarov reaction by utilizing enecarbamates to synthesize significant heterocycles derived from 4-amino-THQs.<sup>4c,10</sup> Then, our group,<sup>11</sup> as well as the Shi group,<sup>12</sup> independently demonstrated that phenolic styrenes bearing either a *para-* or *ortho*-phenol were also suitable dienophiles in the CPA-catalyzed enantioselective cycloaddition, producing 4-aryl-THQs with notable stereoselectivities (Scheme 1a). Inspired by Seidel's work,<sup>7c,13</sup> we subsequently described a CPA-catalyzed enantioselective intramolecular Povarov reaction with styrenic dienophiles,<sup>14</sup> leading to the formation of various highly enantioenriched polycycles. Additionally, many others electron rich dienophiles (including indoles, enamines) were used in CPA-catalyzed Povarov reactions.<sup>15,16</sup>





However, their asymmetric vinylogous counterparts using activated polyenes as dienophiles are rare in the literature. In 2013, Bernardi and co-workers demonstrated the ability of  $\delta$ -unsubstituted dienecarbamates to act as efficient dienophiles in such cycloaddition with high regio-, diastereo-, and enantioselectivities in presence of CPA catalysts.<sup>17,18</sup> Later on, the group of Shi showed that 2- and 3-vinylindoles were also successful substrates in CPA-catalyzed enantioselective Povarov reaction.<sup>16c,16e,17-19</sup> Despite notable achievements, research has predominantly focused on nitrogen-containing polyenes as dienophiles in vinylogous asymmetric Povarov reactions. Surprisingly, while any dienes have been widely used as dienes in asymmetric Diels-Alder reactions,<sup>20</sup> there is currently no reported example of an enantioselective Povarov reaction using aryl dienes as dienophiles. Our research seeks to address this gap by developing an enantioselective variant of the Povarov reaction employing aryl dienes as dienophiles. Drawing upon our previous work, we hypothesized that phenolic dienes could serve as suitable partners for an enantioselective Povarov reaction, leveraging chiral phosphoric acid catalysts.<sup>11</sup> Indeed, we anticipated that the simultaneous dual activation through hydrogen bonding of the bifunctional chiral phosphoric acid with both the N-arylimine and the free hydroxyl group of the phenolic diene could ensure high reactivity and stereoselectivity. Investigating such an enantioselective Povarov reaction raises the question of regioselectivity, possibly resulting in two distinct regioisomers depending on the double bond involved in the cycloaddition. While the terminal double bond is expected to be the most reactive, examples in

the literature have shown that 1,2-disubstituted alkenes can also be engaged as dienophiles in cycloadditions with various imines under Brønsted acid catalysis (Scheme 1b, products **B1** *vs* **B2**).<sup>11</sup> Furthermore, phenolic dienes could potentially react with *N*-aryl imines in a conventional Diels-Alder reaction, as we previously observed in a CPA-catalyzed Diels-Alder reaction between diene-carbamates and *N*-heteroaryl imines, to generate enantioenriched tetrahydropyridines (Scheme 1b, product **B3**).<sup>21</sup>







#### **RESULTS AND DISCUSSION**

In this context, our aim was to develop an enantioselective and regioselective vinylogous Povarov reaction utilizing phenolic dienes as dienophiles (Scheme 1c). We began our investigation by evaluating diene 2a, using 3.0 equivalents, with imine 1a in presence of catalytic amounts of various CPAs (Table 1). Encouragingly, in our initial trials, after 72 hours of reaction at room temperature, we observed the formation of THQ 4a as a single regioisomer and *cis*-diastereomer with 5 mol % of **3a**, exhibiting encouraging enantiocontrol (82% ee, entry 1). Moving to bulkier catalysts, the 2,4,6-<sup>*i*</sup>Pr<sub>3</sub>-C<sub>6</sub>H<sub>2</sub> substituted CPA (**3b**) led to an increased yield (54%) and enantiomeric excess (87%, entry 2). In order to enhance both enantioselectivity and yield, we further refined the condition. However, other catalysts derived from H<sub>8</sub>-BINOL and SPINOL, namely 3c and 3d, did not yield better results than 3b (entries 3 and 4). The solvent screening identified toluene as a better solvent for the reaction, resulting in an 86% yield of compound 4a with a 97% enantiomeric excess and a reduced reaction time from 72 to 48 h (entries 5 and 6). Unexpectedly, the use of a green solvent like ethyl acetate (EtOAc) resulted in comparable yields and enantiomeric excess (ee) to those obtained with toluene (entry 7). Generally, noncoordinating solvents like toluene and  $CH_2Cl_2$  are preferred in this type of reaction to prevent competitive secondary hydrogen bonding interactions between the solvent, the catalyst, and/or the substrates, which can disrupt a well-organized transition state required for high enantio-induction. Although polar solvents have been used in CPA catalysis,<sup>22</sup> this reaction represents one of the rare examples of the successful use of EtOAc in enantioselective IEDDA reactions.<sup>22g</sup> To enhance the efficiency of the process, we decided to investigate the 3-C version of the reaction. Initially, we conducted the reaction of anisidine (5a), benzaldehyde (6a) and diene 2a (3 eq.) in the presence of 5 mol% of **3b** for 48 h at room temperature in EtOAc. We were delighted to achieve **4a** with the same level of enantioselectivity as in the 2-component process, despite observing a decrease in yield (entry 8). However, by extending the reaction time to 72 h, we successfully improved the yield without compromising enantioselectivity (entry 9). In an effort to reduce reaction times, we investigated the effect of temperature. To our satisfaction, at 50°C, we isolated the product with a good yield in just 24 h while maintaining excellent enantioselectivity (entry 10). Furthermore, upon refluxing the solvent, the product was obtained with 85% yield and 97% ee within 24 hours (entry 11). Additionally, i-PrOAc was also tested as an alternative green solvent yielding the desired product 4a in similar efficiency (79% yield, 97% ee, entry 12). Finally, we observed that exposing a 1:1 Z:E mixture of diene 2a led to the formation of a single product, the Z isomer being unreactive in these conditions (entry 13).



<sup>*a*</sup> General condition: imine **1a** (0.10 mmol), diene **2a** (0.30 mmol), and **3** (0.005 mmol), in solvent (1.0 mL) at rt under Ar atmosphere for 72 h. <sup>*b*</sup> Yields refer to chromatographically pure cis-isomer. <sup>*c*</sup> dr was determined to be higher than 95:5 by <sup>1</sup>H NMR analysis of the crude reaction mixture. The *cis*-diastereoselectivity of **4a** was assumed by analogy of **4e** determined by X-ray analysis (cf Supporting information). <sup>*d*</sup> Enantiomeric excess was determined by HPLC analysis on chiral stationary phase. <sup>*e*</sup> Catalyst derived from (S)-BINOL. <sup>*f*</sup> Catalyst derived from (S)-BINOL. <sup>*f*</sup> Reaction conducted for 48 h. <sup>*i*</sup> Reaction performed through a 3-C condition: anisidine (**5a**, 0.10 mmol), benzaldehyde (**6a**, 0.10 mmol), diene **2a** (0.30 mmol), and **3b** (0.005 mmol), in solvent (1.0 mL) at rt under Ar atmosphere. <sup>*j*</sup> Reaction conducted for 72 h. <sup>*k*</sup> Reaction at 50°C. <sup>*m*</sup> Reaction at 77°C. <sup>*n*</sup> Reaction at 90 °C. <sup>*o*</sup> A E:Z mixture (1:1) of diene **2a** was used (0.30 mmol of E isomer).

With the optimized reaction conditions in hand, we next investigated the scope of this 3-C reaction. Various ortho-, meta- and para- substituted benzaldehydes were found to be suitable for this reaction, yielding a wide range of 2-aryl-THQ derivatives with excellent diastereo- and enantioselectivities (4b-4j and 4m, Scheme 2). Notably, 2-azadienes 1 derived from benzaldehydes carrying electron-withdrawing group (EWG) substituents such as halogens, NO<sub>2</sub>, and CN exhibited heightened reactivity compared to their counterparts derived from electron-rich aldehydes, attributed to the enhanced electrophilicity of the imine. The latter necessitated prolonged reaction times of 48 hours, resulting in diminished yields observed in selected examples (4f and 4j). The 3-C-Povarov reaction efficiently produced compound 4k from heteroaromatic aldehyde, achieving high yields and enantiomeric excess (4k, Scheme 2). We were also delighted to observe the formation of aliphatic derivative 4l in 68% yield and 98% ee demonstrating that contrary to unsubstituted aliphatic aldehydes that did not prove to be suitable substrates, hindered aliphatic aldehyde were fully compatible with the transformation. Next, the reaction scope of THQs was investigated relative to the aniline partner (5). The electronic nature of substituents at the *para*-position on the anilines minimally affected the reaction outcome (4n-4o and 4q-4r) except for strong electron-withdrawing groups such as NO<sub>2</sub>, which led to a lower yield of **4p** while the enantioselectivity remained consistently high. Then, the 3C-protocol was tested with ortho-substituted aniline derivative, giving the THQ 4s in moderate yield and nearly racemic ee. This observation suggests that the presence of the ortho substituent could disrupt or even prevent hydrogen bonding between the catalyst and the imines due to steric clash. Consequently, the bifunctional activation of proposed intermediate I-TS in Scheme 4 no longer occurs, and only the dienophilic phenol could interact with the chiral catalyst, resulting in a complete loss of enantioselectivity during the attack of the dienophile on the diene, as well as a decrease in reactivity. Interestingly, the metasubstituted anilines exclusively yielded one regioisomer out of the two possible options (compound 4t-4v, scheme 2). However, while the 5-amino-1,3-benzodioxole produced the less congested isomer 4t with high enantioselectivity, m-chloroaniline as well as 2-naphtylamine formed the most congested isomer 4u and 4v with a lower enantiomeric excess of 70%. These results suggest that at electronic properties not only dictate regioselectivity but also impact the transition state stability in the Friedel-Crafts reaction. Specifically, electronneutral or electron-poor substituents like chlorine or aryl groups seem to favor ortho substitution, aligning with their observed propensity to form the less hindered product. In contrast, the presence of oxygen at position 1 in 5amino-1,3-benzodioxole might hypothetically enhance para substitution due to specific electronic effects, thereby contributing to the formation of the less hindered product. Additionally, it appears that there is a delicate balance between electronic and steric effects to govern enantioselectivity. It is hypothesized that unfavorable interactions may occur between the catalyst with azadiene and dienophile disrupting the transition state and thereby reducing selectivity.

Scheme 2. Substrate scope of the enantioselective 3-CR synthesis of THQs.<sup>a-d</sup>



**4r**: X = Me, 89%, 95% ee

<sup>*a*</sup> General conditions: aniline **5** (0.10 mmol), aldehyde **6** (0.10 mmol), diene **2a** (0.30 mmol), and (*R*)-**3b** (0.005 mmol), in ethyl acetate (1.0 mL) at reflux under Ar atmosphere for 24 h. <sup>*b*</sup> Yields refer to chromatographically pure cis-isomer. <sup>*c*</sup> dr was determined to be higher than 95:5 by 1H NMR analysis of the crude reaction mixture. The *cis*-diastereoselectivity was assumed by analogy of **4e** determined by X-ray analysis (cf Supporting information). Enantiomeric excesses were determined by HPLC analysis on chiral stationary phase.<sup>*d*</sup> Performed on 1 mmol scale. <sup>*e*</sup> Reaction conducted with (*S*)-**3b** instead of (*R*)-**3b**.<sup>*f*</sup> Reaction conducted for 48 h. <sup>*g*</sup> 0.15 mmol of **2a** were used.

The scope of the enantioselective vinylogous Povarov reaction was expanded to include dienes with diverse substitution patterns on the aryl ring and alkene substituents. The phenolic dienes were employed exclusively in their (E) isomer form, following an E/Z isomerization performed in refluxing toluene in the final step of their synthesis. The di-*ortho*-methoxy phenolic diene **2b** led to cycloadduct **4w** with a 60% yield and an excellent enantiomeric excess of 99%. When less electron-rich phenolic dienes **2c** and **2d** were employed, the resulting THQs **4x** and **4y** were obtained with moderate yields and enantiomeric excess. To our satisfaction, the terminally substituted diene **2e**, used as a mixture of E and Z isomers (4:1), successfully underwent the reaction with

benzaldehyde and anisidine to yield THQ 4z with three contiguous stereocenters as a single *trans-trans* diastereomer<sup>23</sup> maintaining a high enantiomeric excess, albeit with a reduced yield compared to the unsubstituted diene 2a. Unfortunately, substituting the terminal methyl group of 2e with an isobutyl or phenyl group did not yield the desired cycloadduct product; only starting materials were recovered, underscoring the reaction's sensitivity to both steric and electronic properties of dienophiles. Efforts to synthesize 1,2,3,4-tetrahydroquinolines featuring a quaternary center at position 4 through catalytic enantioselective Povarov reactions have encountered limitations in previous studies.<sup>6c,12</sup> However, our investigation produced promising results with the internally substituted diene 2f, which displayed reactivity as a dienophile in the asymmetric 3-C Povarov reaction under established conditions. Interestingly, when reacted with various aromatic aldehydes, the target THQs 4aa, 4ab and 4ac were obtained with moderate to high yields and high enantioselectivities (96 to >99% ee).



Scheme 3. Phenolic diene scope for enantioselective 3-CR Povarov reaction.<sup>a-c</sup>

<sup>*a*</sup> General condition: aniline **5** (0.10 mmol), aldehyde **6** (0.10 mmol), diene **2** (0.30 mmol), and (*R*)-**3a** (0.005 mmol), in ethyl acetate (1.0 mL) at reflux under Ar atmosphere for 24 h. <sup>*b*</sup> Yields refer to chromatographically pure cis-isomer. <sup>*c*</sup> dr was determined to be higher than 95:5 by 1H NMR analysis of the crude reaction mixture. The *cis*-diastereoselectivity of **4a** was assumed by analogy of **4e** determined by X-ray analysis (cf Supporting information). Enantiomeric excesses were determined by HPLC analysis on chiral stationary phase vs racemic. <sup>*d*</sup> Reaction conducted with (*S*)-**3a** instead of (*R*)-**3a**.

Building on previous research and precedent mechanistic studies,<sup>4c,9b,11,12</sup> we propose a plausible stepwise process involving a Mannich reaction followed by Friedel-Crafts alkylation (Scheme 4). To elucidate the observed enantioselectivity, several control experiments were conducted. Under standard conditions, the reaction involving dienes containing *meta-* and *ortho-*phenolic groups **2g** and **2h** resulted in the formation of THQs **4ad** and **4ae** with low yields and enantiomeric excess values of 29% and 68%, respectively (**4ad** was obtained in 55% yield along with an inseparable impurity). Additionally, the di-methoxy diene **2i** produced the corresponding cycloadduct **4af** with high enantioselectivity (96% ee) albeit with a low yield.

Based on these experimental findings, we proposed that activation of the imine occurs through interaction with the chiral phosphoric acid (*R*)-**3b**. Subsequently, hydrogen bonding takes place between the phenol and the Lewis basic phosphoryl oxygen. Following this and according to the Simón and Goodman model,<sup>24,25</sup> a pseudo-intramolecular Si-face attack of the dienophile **2** onto the imine *via* transition state **I-TS** is expected to form (S)-amine **II**. Finally, an intramolecular aza-Friedel-Crafts reaction *via* the less sterically congested conformer **II** would result in the observed *trans*, *trans* trisubstituted tetrahydroquinoline **4z** with a defined absolute configuration, as depicted in Scheme 3. It is difficult to rely on a bifunctional activation mode when considering the use of di-methoxy diene **2i** which leads to compound **4af** with high enantioselectivity (96% ee). In this case, a simple mono activation of imine (possibly an ion pair if full protonation of the imine is taking place) allows for facial discrimination when reacting with the rich dimethoxy diene **2i** (**I**<sup>1</sup>). Presence of two hydroxy unit (OH and OMe or 2 OMe) on the arene, seems crucial for achieving reactivity. We believe that, when possible, establishment of a hydrogen bond between Lewis basic phosphoryl oxygen and the phenol group is the rule thereby increasing reactivity and enhancing yield. However, position of the hydroxy unit is of critical importance for high enantioselectivity, where contrary to *meta-* and *ortho-* position (phenolic groups **2g** and **2h**) only the *para* position allows for a geometrically well-defined transition state and high induction.



Scheme 4. Experimental Control<sup>a</sup>, Potential Reaction Mechanisms, and Activation Modes

To demonstrate the synthetic applicability of this transformation and to further modify the enantioenriched THQ products, we conducted several post-transformations (Scheme 5). The hydrogenation of THQ **4a** using a flow hydrogenation apparatus (H-Cube) proceeded smoothly, affording **7** in 97% yield in high purity without purification and with no erosion of enantioselectivity. Chemoselective modification of the phenol residue could also be demonstrated. The phenol group was protected as a triflate group, followed by a Pd-mediated reduction using formic acid, resulting in compound **9** in 78% yield over 2 steps without compromising enantioselectivity. Additionally, the Suzuki coupling of THQ **8** with phenylboronic acid afforded compound **10** in 55% yield with no erosion of enantioselectivity.



*a.* H<sub>2</sub>, Pd/C, H-cube, Full H<sub>2</sub> mode, 1 mL/min, 20 °C, MeOH; *b.* Tf<sub>2</sub>NPh, DMAP (cat.), Et<sub>3</sub>N, DCM, rt, 1h; *c.* Pd(dppf)Cl<sub>2</sub>.DCM (10 mol %), PhB(OH)<sub>2</sub>, Cs<sub>2</sub>CO<sub>3</sub>, H<sub>2</sub>O, toluene, THF, 65 °C, 90 °C, 5h ; *d.* Pd(OAc)<sub>2</sub> (10 mol %), dppf (10 mol %), Et<sub>3</sub>N, HCOOH, THF, reflux, 2h.

#### CONCLUSION

In summary, we have developed an enantioselective 3-C Povarov reaction employing phenolic dienes as dienophiles. This innovative method enables the synthesis of 1,2,3,4-tetrahydroquinolines containing a styryl group at the 4-position with excellent levels of enantioselectivity, regioselectivity, and diastereoselectivity. We propose a plausible mechanism, supported by experimental evidence, to account for the stereochemical outcomes. Furthermore, the pivotal role of the *para*-positioned free phenol group combined with a *meta*-positioned methoxy group on the phenolic dienes in facilitating this transformation was demonstrated.

#### **EXPERIMENTAL SECTION**

**Materials and General methods.** All reactions requiring anhydrous conditions and inert atmosphere were carried out under argon atmosphere in dried glassware. EtOAc was distilled over CaCl<sub>2</sub>. Immersion coolers IMC-40 and TC100E fitted with a temperature probe and a flexible cooling tube were used to perform reactions at low temperature. The heat source employed for the synthesis of catalyst and substrates was an oil bath. Solvents were distilled by standard methods using the appropriate drying agent and stored over molecular sieves under argon.

All other reagents were obtained from commercial suppliers and directly used unless otherwise noted. Flash column chromatography was carried out using 40-63 µm particle sized silica gel with air pressure. Analytical thin layer chromatography (TLC) plates (silica gel 60 F254) were visualized either with a UV lamp (254 nm), or by submersion in potassium permanganate, ninhydrine or iodine. Melting points were recorded using a melting point apparatus (Büchi B-540) and are uncorrected. Specific rotations for chiral compounds were recorded on a Perkin Elmer 141 using sodium D ray (589 nm) at room temperature in a 700-µL cell with a path length of 1 dm in CH<sub>2</sub>Cl<sub>2</sub>. Infrared spectra were recorded on a Perkin Elmer Spectrum BX FT-IR spectrometer and the characteristic IR absorption frequencies are reported in cm<sup>-1</sup>. Proton NMR (<sup>1</sup>H) spectra were recorded on a Bruker Avance 500 MHz or 300 MHz and proton-decoupled carbon 13 NMR (<sup>13</sup>C{<sup>1</sup>H}) spectra were recorded at 75 MHz. NMR experiments were carried out in deuterochloroform (CDCl<sub>3</sub>) or in trideuteroacetonitrile (CD<sub>3</sub>CN). Proton-decoupled fluorine <sup>19</sup>F NMR (<sup>19</sup>F{<sup>1</sup>H}) spectra were recorded at 282 MHz. Chemical shifts ( $\delta$ ) are reported in parts per million (ppm) relative to residual solvent as an internal reference (<sup>1</sup>H: 7.26, <sup>13</sup>C: 77.16 ppm for CHCl<sub>3</sub>/CDCl<sub>3</sub> and <sup>1</sup>H: 1.94, <sup>13</sup>C: 1.32 ppm for CHD<sub>2</sub>CN/CD<sub>3</sub>CN). The following abbreviations are used for the multiplicities: s: singlet; d: doublet; t: triplet; q: quadruplet; quint: quintuplet; m: multiplet or overlap of nonequivalent resonances; br s: broad singlet; app: apparent; rot: rotamer. Coupling constants (J) are reported in Hertz (Hz). Structural assignments were made with additional information from gCOSY, gHSQC, and gHMBC experiments. High resolution mass spectra were determined on an AEI MS-9 using electrospray ionization (ESI) and a time-of-flight (TOF) analyzer. Enantiomeric excesses were determined by high performance liquid chromatography (HPLC) equipped with diode array UV detectors, using Chiralpak® AD-H, OD-H, AS-H, OJ-H, IA and IB columns.

#### General procedure for the enantioselective construction of 4-substituted tetrahydroquinolines:

A vial equipped with a Teflon stir bar was flame-dried under vacuum and backfilled with argon three times. Then, aniline (1.0 eq, 0.1 mmol), aldehyde (1.0 eq, 0.1 mmol) and (R)- or (S)-TRIP (0.05 eq, 0.005 mmol, 3.7 mg) were added to the vessel and dissolved in EtOAc (1 mL). The resulting solution was stirred at rt for 1h, followed by the addition of phenolic diene **2** (1.0-3.0 eq, 0.1-0.3 mmol). The reaction was stirred at reflux in an oil bath for 24h-72h depending on the substrates. When the reaction was deemed complete by TLC analysis, the mixture was concentrated under reduced pressure. The crude was then purified by flash column chromatography on silica gel using PET/EtOAc (8:2) as the eluent to yield the desired tetrahydroquinoline.

The racemic versions were synthesized using the same procedure, on a 0.05 mmol scale regarding the aldehyde, in 500  $\mu$ L of EtOAc, using *rac*-BINOL CPA as the catalyst (5 mol%).

**2-methoxy-4-((E)-2-((2S,4R)-6-methoxy-2-phenyl-1,2,3,4-tetrahydroquinolin-4-yl)vinyl)phenol** (4a): Synthesized according to general procedure using *p*-anisidine (1.0 eq, 0.1 mmol, 12 mg), benzaldehyde (1.0 eq, 0.1 mmol, 10 µL), diene **2a** (3.0 eq, 0.3 mmol, 53 mg) and *(R)*-TRIP. After 24h of reaction, the crude was purified to yield product **4a** as brown powder (33 mg, >95:5 *dr*, 96% *ee*) in 85% yield.  $[a]_{D}^{22} = -52$  (c = 0.1, CHCl<sub>3</sub>). **Rf:** 0.28 (PET/EA 8:2, UV & vanillin). **Mp:** 125-127 °C. **IR** (neat) *v* (cm<sup>-1</sup>): 3472, 3365, 2943, 2830, 1733, 1601, 1509, 1463, 1449, 1429, 1350, 1258, 1211, 1152, 1119, 1034. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.47 – 7.28 (m, 5H), 6.93 – 6.84 (m, 3H), 6.78 (dd, *J* = 2.8, 1.1 Hz, 1H), 6.68 (ddd, *J* = 8.7, 2.9, 0.8 Hz, 1H), 6.55 (d, *J* = 4.9 Hz, 1H), 6.51 (d, *J* = 12.0 Hz, 1H), 6.00 (dd, *J* = 15.6, 9.1 Hz, 1H), 4.47 (dd, *J* = 11.2, 2.5 Hz, 1H), 3.89 (s, 3H), 3.88 – 3.80 (m, 1H), 3.72 (s, 3H), 2.19 (ddd, *J* = 13.0, 5.8, 2.5 Hz, 1H), 2.01 (dt, *J* = 13.1, 11.5 Hz, 1H). <sup>13</sup>C{<sup>1</sup>H} NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  152.3, 146.8, 145.4, 144.3, 139.2, 131.6, 130.9, 130.1, 128.8, 127.8, 126.8, 124.8, 120.2, 115.6, 115.1, 114.5, 113.5, 108.1, 56.9, 56.1, 56.0, 42.6, 39.8. HRMS (ESI/Q-TOF) m/z: [M+H]<sup>+</sup> Calc. for C<sub>25</sub>H<sub>26</sub>NO<sub>3</sub>: 388.1892, Found: 388.1892; **HPLC** Daicel Chiralpak IA, Hept/EtOH 90:10, flow rate = 1 mL/min, 254 nm): minor isomer t<sub>R</sub> = 33.9 min, major isomer: t<sub>R</sub> = 42.1 min.

## ASSOCIATED CONTENT

#### Data Availability Statement

The data underlying this study are available in the published article and its Supporting Information.

### Accession Code

CCDC 2345645 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via <u>www.ccdc.cam.ac.uk/data\_request/cif</u>, or by emailing <u>data\_request@ccdc.cam.ac.uk</u>, or by contacting The Cambridge Crystallographic Data Centre, 12 UnionRoad, Cambridge CB21EZ, UK; fax: +44 1223336033

### NOTE

The authors declare no competing financial interest.

## **AUTHOR CONTRIBUTIONS**

All authors have given approval to the final version of the manuscript.

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