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The astrochemical evolutionary traits of phospholipid membrane homochirality

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Abstract

 Compartmentalization is a key prerequisite for the evolution of life. Present-day phospholipid membranes exhibit a high level of complexity and species-dependent homochirality, the so- called *lipid divide*. However, it is very likely that less stable yet more dynamic systems, promoting out-of-equilibrium environments, facilitated the evolution of life at its early stages. The composition of the preceding primitive membranes and the evolutionary route towards complexity and homochirality, however, still remain unexplained. Organic-rich carbonaceous chondrites are evidence of the ample diversity of interstellar chemistry, which may have largely enriched the prebiotic milieu on the early Earth. This Review thoroughly evaluates the detections of simple amphiphiles — likely ancestors of membrane phospholipids — in extra- terrestrial samples and analogues, along with potential pathways to form primitive compartments on the primeval Earth. The chiroptical properties of chiral backbones of phospholipids provide a guide for future investigation into the origins of phospholipid membrane homochirality. We highlight a plausible common pathway towards homochirality 24 of lipids, amino acids, and sugars starting from enantioenriched monomers. Finally, given their high recalcitrance and resistance to degradation, lipids are one of the best candidate biomarkers in exobiology.

 Key words: chirality, membrane, lipids, amphiphiles, compartments, biomarkers, circularly polarized light, extant and extinct extra-terrestrial life

I. Introduction

 The origin of biological membranes is a complex and intriguing topic closely tied to the broader question of the origin of life itself. Given the wide diversity of lipids, the ease with which they self-assemble, their ability to compartmentalise along with the potential of autocatalytic activity, lipids would have been pivotal at the dawn of self-organization and 37 early metabolism, both necessary for life to arise and develop.¹ While numerous experiments have explored different challenges on how life originated on Earth, there is a lack of experimental studies addressing the dichotomy of biological membrane structure and their inherent chirality. Finding an answer to this question is crucial not only for understanding life on Earth but also for detecting life, whether extant or extinct, in space.

42 Carbonaceous chondrites, rich in chemical compounds composed of C, H, N, O, P, and/or $S₁²$ are evidence of the complexity of interstellar chemistry. This rich chemistry occurs in denser regions of the interstellar medium, so-called molecular clouds–nurseries of stars and planets– 45 where interstellar dust, water ices, and molecules are detected. $3-5$ Interstellar dust grains are primarily composed of carbonaceous and silicate material of varying size between 1 nm and $10₁$ m.⁶ Making up only about 1% of the total mass of the interstellar medium, dust grains are the cradle for the synthesis of complex organic molecules (COMs) in space that cannot be produced by gas-phase chemistry. Most complex organics, including prebiotically relevant species, are proposed to form through radical chemistry in icy mantles mixed with porous 51 grains.^{3,7} During planetary system formation, the solid cloud material containing embedded COMs becomes incorporated into various celestial bodies, including planetesimals, comets, 53 and asteroids.³ Thence, the exogenous delivery may have largely contributed to the pool of 54 organic molecules on the early Earth. However, it also implies that any monomeric building block of life is very likely to be present in extra-terrestrial bodies, which hinders the identification of biosignatures of putative life in space. Any specific biomolecule detected in space therefore cannot be considered a biosignature *per se*. A recent review by Glavin et al.⁹ distinguished the following attributes: *(1) chiral asymmetry*, *(2) light isotopic composition,* and *(3) strong structural isomer preference*, the combination of which may provide a strong hint for the presence of past extra-terrestrial life. Nevertheless, since abiogenesis was an evolutionary process rather than a one-off event, unambiguous classification of signatures between biotic or abiotic appears even more intricate.

 Fig. 1 Schematic timeline of the geological longevity of major molecular biosignatures. The timeline shows the oldest recovery ages in the geological records. Lipids and kerogen outlast DNA, amino acids' enantiomeric excess (*ee*) and proteins by more than an order of magnitude, making them the most valuable biosignatures when 68 searching for past life in space.

 Studying contemporary life forms on Earth has enabled scientists to identify four common features of life: *(1) nucleic acids*, *(2) proteins*, *(3) membranes* composed of amphiphilic molecules—lipids, and *(4) biological homochirality*. Lipids exhibit exceptional resistance to degradation under various environmental conditions, surpassing both DNA and proteins, 74 along with their monomeric building blocks-sugars and amino acids. (Fig. 1).^{10,11} This characteristic positions lipids as promising biomarkers for ancient life. However, was the molecular machinery of the most primitive organisms the same as we know it from present- day life? This is particularly debatable in the case of membrane lipids which are known to be species specific. While in general bacterial and eukaryal membranes' phospholipids are fatty acid esters linked to *sn*-glycerol-3-phosphate (G3P), archaeal phospholipids are commonly built on the backbone of the opposite handedness, namely *sn*-glycerol-1-phosphate (G1P), 81 and contain ether-linked isoprenoid derivatives (20-40 carbons), which if chiral are present 82 exclusively in the *R*-configuration (Fig. 2a).¹² Despite the high level of scientific interest, the membrane composition of the last universal common ancestor (LUCA) and the so-called *lipid- divide* remain an unresolved puzzle so far. A growing body of evidence suggests that the first 85 membranes were built of simple amphiphiles rather than complex phospholipids. $13-15$ 86 However, their composition, chirality, and the transition from the primitive membranes to 87 the ones of present-day life is underexplored. This is where the organic content analyses of 88 carbonaceous chondrites, as well as experiments simulating prebiotic species synthesis in interstellar ices and on the early Earth, become of high importance. These can help us to not only elucidate what prebiotic material with potential to form primitive membranes could have been available on the primitive Earth, but also to identify what lipid-like species of 92 abiotic origin can be expected to be found in the search for life on Mars¹⁶ or in space in general. Furthermore, studying the self-assembly and chirality interplay between primitive amphiphiles, complex phospholipids as well as amino acids and sugars, is inescapable for unravelling the origins of species-dependent homochirality of phospholipid membranes.

 At the emergence or evolution of life, the stereospecificity of amino acid–phospholipid interactions might have guided a pre-selection of life's machineries. Stereosensitivity is a hallmark of ligand–receptor recognition in many biological processes, due to the specific and complementary bonds between the ligand and the receptor binding site. Consequently, enantiospecific interactions are broadly exploited in miscellaneous pharmaceutical concepts. Thorough understanding of interactions of drugs with chiral biological molecules is needed to not only efficiently direct the drug delivery, but also to avoid unwanted side-effects arising 103 from non-target stereospecific interactions.¹⁷ Nanoencapsulation of drugs into vesicles and micelles represents an efficient means for protecting the active compounds throughout the 105 transit, channelling the drug release and decreasing potential side toxicity.¹⁸ The most common building blocks of drug delivery vectors are phospholipids. Analogously to biological membranes, phospholipids' structure critically affects the biophysical properties of liposomes and hence their functionality. The type of headgroups, carbon chain length, branching, symmetry, degree of saturation as well as intra- and intermolecular interactions affect the membrane thickness, fluidity, phase transition temperature, and drug release rate. Generally, the shorter hydrocarbon chain, the higher degree of unsaturation or branching are likely to 112 result in reduced drug retention times due to looser membrane packing.^{19,20} Archaeal phospoholipid based liposomes–archaeosomes–have been increasingly investigated for potential application in pharmaceutics thanks to their robustness against extreme 115 temperatures, pH, oxidative conditions, pressure, etc. $21,22$ Probing enantioselectivity and other properties of chiral amphiphilic membranes such as permeability, stability, etc. has, 117 therefore, important implications not only for elucidating the origin of life, but also in e.g. designing new drug delivery systems, biosensing, as well as for testing drug effects or

119 antibiotic resistance and selectivity.

 Fig. 2 The membrane lipid divide. a Characteristic phospholipid composition of archaea vs. bacteria and eukarya. **b** Scheme of the phylogenetic tree of three clades of life. The chirality of the phospholipid backbones, the 124 ester/ether linkages, and side chains are, however, not strictly delineated within the three domains. Th ester/ether linkages, and side chains are, however, not strictly delineated within the three domains. The *R*- isoprenoids, found ubiquitously in all three domains, are synthesized through distinct enzymatic pathways: the 126 mevalonate pathway (MVA) and the 2-C-methyl-D-erythritol 4-phosphate/1-deoxy-D-xylulose 5-phosphate
127 (MEP-DOXP) pathway. c The most common branched hydrocarbons. Farnesane (C15) is an ancient analogue of (MEP-DOXP) pathway. **c** The most common branched hydrocarbons. Farnesane (C15) is an ancient analogue of farnesol, esterifying alcohol in bacterial chlorophylls. Norpristane (C18), pristane (C19), and phytane (C20) were 129 previously detected in the Murchison meteorite.²⁴ Most chlorophylls contain phytol (C20) side chains (*R*- configurations), and are considered the main source of pristane (*meso*-configuration after separation) and phytane on Earth. Another source of pristane (*meso*-configuration after separation) are tocopherols possessing 132 phytyl tails (*R*-configurations).²⁵

II. Species-specific traits of membrane phospholipids

Terrestrial life is based on carbon chemistry operating in aqueous media. While liquid water

- appears indispensable for driving prebiotic reactions, the emergence of life was very likely
- conditioned by compartmentalization. Membranes not only create protective boundaries to

 the outer environment sequestering critical cellular biopolymers, they also secure energy gradients, communication, controlled transport of molecules across the membrane, and are 140 involved in metabolism. Such rich functionality stems from their complex organization, which is a consequence of their protracted evolution. A clear trait of the evolution is a distinct composition of archaeal, versus bacterial and eukaryal membrane phospholipids encoded in their biosynthetic machineries. In general, glycerol-1-phosphate dehydrogenase (G1PDH) catalyses the formation of G1P from dihydroxy acetone phosphate (DHAP) in archaea, while G3PDH and glycerol kinase (GK) catalyse the formation of G3P from DHAP and glycerol, 146 respectively, in bacteria and eukarya.²⁷ This distinction is, however, not mutually exclusive. 147 Fatty acid phospholipid synthesis was identified in certain types of archaea, $28,29$ while some 148 bacteria were found to produce ether-linked lipids^{30,31} and even possess genes for synthetic 149 pathways of archaeal ether-bound isoprenoid lipids.³² Given these interlinks between the two branches of lipid membranes, it was hypothesised that membranes were first of mixed chirality and it was the intrinsic instability of heterochiral membranes which triggered the *lipid divide*.^{33,34} However, it was proven later on that heterochiral membranes can be even 153 more robust than homochiral ones.^{35,36}

 Interestingly, the branched hydrocarbon chains with the *R*-configuration on their stereocentres are present in all three clades of life (**Fig. 2b,c**), and are found in chlorophylls and tocopherols of cyanobacteria, algae, and higher plants, as well as in steroids, carotenoids, 157 ubiquinones etc., $25,37,38$ which points towards their common origins. The isoprenoid chains are in present life synthesised from isopentenyl pyrophosphate (IPP) and the isomeric form dimethylallyl pyrophosphate (DMAPP). Two distinct enzymatic pathways lead to these precursors, notably the mevalonic (MVA) pathway named after its intermediate *R*- mevalonate and the 2-*C*-methyl-D-erythritol 4-phosphate/1-deoxy-D-xylulose-5-phosphate (MEP-DOXP) pathway starting from D-glycerladehyde-3-phosphate and pyruvate. While the MVA channel is fully responsible for the synthesis of IPP and DMAPP in archaea and non- photosynthetic eukarya, the MEP-DOXP is present in bacteria, algae, and plants (**Fig. 2b**). Interestingly, the three final steps of the MVA synthetic route in archaea and eukarya differ as the three corresponding enzymes in archaea do not share a common sequence homology 167 with those of eukarya³⁸. It is, therefore, likely that the distinct enzymatic pathways for the synthesis of branched hydrocarbons always leading to the *R*-configurations are a result of speciation, while the roots of ubiquitous homochiral isoprenoid chains were already anchored abiotically or in an alternative enzymatic route of a common ancestor.

171 It is widely accepted that archaea and bacteria evolved from the last universal common 172 ancestor. Despite extensive research into the genetic traits and membrane compositions of archaea and bacteria, a clear-cut theory about the composition of LUCA's membrane is still missing. It also remains open whether LUCA's membrane phospholipid synthesis was enzymatic or not, the elucidation of which could help to resolve the question about the origin 176 of the chiral lipid divide.¹⁷ Several hypotheses and reviews have been centred around 177 discussing these research questions and the reader is referred to references 12,17,27,34,38,39 .

III. Amphiphilic compartments as primitive membranes

 Given the distinct membrane phospholipid suits, archaea and bacteria employ different 181 mechanisms to maintain cell integrity upon extreme conditions.³⁸ Such robust membrane morphology is undoubtedly a result of long evolution and is optimal for extant life. However, it is very likely that more dynamic systems facilitating growth, division, recombination, and transfer of molecules aided the emergence and/or evolution of life at the early stages by creating out-of-equilibrium environments. Amphiphilic molecules (**Fig. 3**) were found to spontaneously form vesicles in aqueous environments. This is driven by hydrophilic interactions between polar head groups, which also form hydrogen bonds with water molecules, and van der Waals forces between hydrophobic hydrocarbon tails, which are repelled by water molecules. The formation and stability of vesicles is affected by the type of 190 amphiphiles, their concentration, temperature, pH, and salinity of the solution.⁴⁰⁻⁴⁴ As opposed to double chain phospholipid vesicles, single-chain fatty acid vesicles exhibit enhanced dynamics, mainly thanks to their lower hydrophobicity. This, however, comes at 193 the expense of lower stability especially at low concentrations.⁴⁵ The addition of small amounts of alcohol or glycerol monoesters to fatty acid vesicles was found to enhance their stability, manifested by their self-assembly at lower concentrations, and better resistance to 196 basic pH^{42,46} and elevated temperatures.^{44,46} Prebiotic nucleobases⁴⁷ and amino acids⁴⁸ were 197 shown to bind to and stabilise decanoic acid vesicles in the presence of 10 mM Mg^{2+} salt. Isoprenoids were also found to improve the stability of fatty acid vesicles in the presence of salts, however, about 30-times higher concentrations are necessary to allow membrane 200 formation.⁴⁹ The level of binding and the effect of amino acids or dipeptides on the growth, size, leakiness, and susceptibility to flocculation of prebiotic vesicles further depends on chirality, structure/sequence, or length, thence creating selective environments for Darwinian evolution at the cellular level to start.

204 An alternative scenario for the emergence of the first proto-cellular systems is based on liquid-liquid phase separation of oppositely charged polyelectrolytes in aqueous solution, so- called coacervation. Depending on their charge with respect to the charge of these membrane-less structures, phospholipid liposomes were found to coat coacervates (same 208 charge) or form fibre networks within them (opposite charge).^{50,51} However, given phospholipids' relative structural complexity, the transition from membrane-less coacervates would have been more likely driven by a self-assembly of fatty-acid membranes around them 211 as a potential intermediate step towards present-day phospholipid bilayers.^{52,53}

 Wet-dry cycling processes representative of night-day cycles on the early Earth have manifested unprecedented efficiency in the emergence of membrane-forming proto-214 amphiphilic systems, notably polyesters synthesised from hydroxycarboxylic acids,⁵⁴ N-acyl 215 amino acids synthesised from amino acids and phospholipids⁵⁵ or amino acids and 216 monoglycerides,⁵⁶ and amino acid-fatty alcohol esters synthesised from homonymous 217 components.¹⁴ Importantly, *N*-acyl amino acids can also serve as a substrate for peptide 218 formation similarly to depsipeptides comprised of amino acids and hydroxycarboxylic acids⁵⁷ 219 or thiodepsipeptides formed from amino acids and mercaptoacids.⁵⁸ Therefore, such one-pot reactions from a common set of prebiotically relevant reactants would have spontaneously and simultaneously driven the emergence of two key pre-requisites of life, proto-peptides 222 and primitive membranes. The robustness of proto-amphiphilic vesicles towards extreme pH or the presence of metal ions can be enhanced in the presence of oleic acid or 224 monoglyceride.⁵⁶ Such a scenario of mixed amphiphilic membranes appears prebiotically 225 more plausible considering the diversity of organics delivered by meteorite impacts^{2,59,60} or 226 possibly synthesised on the early Earth. $61-64$

 Fig. 3 Schematic structures of simple amphiphiles. a Monocarboxylic acids, **b** amino acids, **c** hydroxycarboxylic acids, **d** alcohols, **e** amines, **f** phosphates, and **g** sulphates.

232 Lipids comprise more than 60% of the soluble organic matter in meteorites,⁶⁵ likely enhancing 233 the importance of their contribution to the pool of organics synthesised on the primitive 234 Earth. Microscopic studies by Dworkin et al.⁶⁶ manifested the potential of forming vesicles from self-assembling amphiphilic molecules synthesised upon astrophysical conditions in UV 236 irradiated interstellar/pre-cometary ice analogues composed of H_2O , CH_3OH , NH_3 , and CO, as well as from those found in the extracts of Murchison. Unfortunately, insufficient amount of material prevented compositional analysis in the study. Nevertheless, numerous amphiphiles have already been identified, both in carbonaceous chondrites and simulated interstellar/pre- cometary ices. **Supplementary Tables 1–7** list the detections of aliphatic monocarboxylic acids, hydroxycarboxylic acids, alcohols, amines, and alkyl phosphorus- and sulphur-bearing molecules. An exhaustive review of amino acid detections has been recently reported by 243 Glavin et al. While a rich variety of straight-chain and branched monocarboxylic acids, ranging from C1 to C20 units, has been identified, only short chain alcohols (**Supplementary Table 3**) have been reported so far. The presence and formation pathways of their long-chain counterparts remain to be elucidated. However, the synthesis of long-chain aliphatic hydrocarbons, extending up to C27, under astrophysical conditions has already been manifested by numerous detections in carbonaceous chondrites (**Supplementary Table 7**). 249 Among prebiotically relevant molecules, amino acids have been extensively studied, followed by hydroxycarboxylic acids (**Supplementary Table 2**). In contrast, the identification of amines in simulated interstellar ices (**Supplementary Table 4**), as well as alkyl phosphorus (**Supplementary Table 5**) and sulphur-bearing (**Supplementary Table 6**) compounds, both in extra-terrestrial samples and ice analogues, lags behind.

254 **IV. Emergence of phospholipid membrane homochirality**

255 Homochirality is often considered a pre-requisite for life based on its critical role for molecular 256 recognition and replication.⁶⁷ Importantly, the presence of enantiomerically enriched species⁹ 257 in carbonaceous chondrites points to the astrophysical roots of symmetry breaking. So far, 258 the *astrophysical circularly polarized light (CPL) scenario* provides the most conclusive proof 259 for explaining the origin of chiral bias in space. 68 Despite significant scientific interest, the

- 260 species-dependent homochirality of phospholipid membranes has remained unexplained.
- 261

262 **Box 1**

263 Astrophysical circularly polarized light scenario

264 Cosmic rays and stellar radiation liven up cold dark interstellar regions transforming simple 265 | molecules such as water, methanol, ammonia, methane, carbon monoxide or dioxide, frozen 266 onto interstellar dust particles into complex organic molecules. Throughout their lifetime 267 these molecules can undergo phase transitions as indicated in the figure, while concurrently 268 | being subject to further chemical transformations triggered by radiation.^{3,6,69}

269 Circularly polarised light (CPL) in the interstellar medium is a likely trigger of symmetry breaking of chiral organics in space.¹²⁷ Organics synthesised **Stereospecific interactions of** ultraviolet CPL with chiral amino acids or their precursors, known to generate a subtle chiral bias, are broadly discussed to be linked to the L-enantiomeric excesses of amino acids detected in multiple carbonaceous chondrites. 9 It is plausible that the Solar System remnants containing CPL enantioenriched chiral molecules seeded the early 285 Earth during the late heavy

 286 bombardment period, with a systematic chiral imbalance that would have shaped the 287 evolution of life's homochirality.^{60,70} The single chirality of amino acids in proteins (L-) and 288 sugars in nucleic acids (D -) is the basis for the build-up of functional biopolymers necessary 289 \mid for life.

290 The most likely sources of CPL in space, detected several times in the infrared range in 291 different star-forming regions, $71,72$ are reflection nebulae associated with high-mass young 292 $\frac{1}{2}$ stellar objects.⁷³ Modelling of dichroic extinction or scattering of light on magnetically aligned 293 | non-spherical dust particles has proven the potential to generate circular polarization.^{74,75} It 294 is assumed that our Sun was formed alongside a heavier star which might have acted as a 295 Source of CPL for asymmetric photochemistry during the early stages of protoplanetary disk 296 \parallel evolution. The peripheral regions of the protoplanetary disk would have received the highest 297 flux of external UV CPL, while being well shielded from the unpolarized UV radiation of the \parallel young Sun. Mixing and diffusion of the dust within the disk would have secured exposure of \parallel a greater amount of material to chiral photons eliciting more asymmetric interactions, while shielding the already CPL-enantioenriched species from the destruction due to UV overexposure.⁷³ Our recent study on propylene oxide, the first chiral molecule detected in space, has indicated that our presolar nebula would have likely been irradiated by left-handed 303 | CPL.⁷⁶

 The potential of UV CPL in generating chiral enrichment to amino acids has been experimentally confirmed in two types of simulation experiments. The analysis of room-306 temperature residues of interstellar ice analogues, prepared from a mixture of H₂O, CH₃OH and NH₃ cooled down to 80 K and irradiated by monochromatic UV circularly polarized light, demonstrated the formation of a number of chiral amino acids with enantiomeric 309 enrichments of up to 2.5%.^{73,77} Irradiation of racemic amino acids by monochromatic CPL in 310 \parallel aqueous solution,^{78,79} approximating water-rich solvation shell in interstellar ices, as well as 311 in solid phase thin films, $80-82$ mimicking molecules condensed on bare interstellar grains, 312 confirmed the asymmetric photolysis reaching up to about 5% enantioenrichment of the 313 Fremaining material. Further investigations of CPL interactions with other chiral organics and 314 deeper elucidation of the circular polarisation of stellar radiation will hone our understanding of the conditions of the astrophysical circularly polarized light scenario and its implications for the origin and evolution of biological homochirality.

 Exogeneous delivery would very likely have been an important source of organo-phosphorous compounds, given constrained bioavailable phosphorus in terrestrial water and the thermodynamically unfavourable nature of phosphorylation reactions on biomolecules in an 321 agueous environment in the absence of enzymes. In support of this are several detections 322 of phosphorus-bearing molecules in space, notably phosphine (PH₃), $84-86$ PO, and PN $87,88$ as well as organophosphorus compounds in carbonaceous chondrites and simulated astrophysical ices (**Supplementary Table 5**). Finally, the abiotic synthesis of glycerol phosphate has been confirmed in electron irradiated simulated interstellar ices composed of 326 PH₃, CH₃OH, CO₂, and H₂O.⁸⁹ Given the relative complexity of present membrane phospholipids, it is unlikely that these would have been synthesised in extra-terrestrial environments. Instead, chiral backbones and shorter chain phospholipid derivatives could have been delivered, preceding their function in the early stages of life evolution. Several potential paths for the prebiotic phospholipid synthesis on the early Earth have been unveiled to date, encompassing the synthesis of (lyso)phosphatidic acids and naturally occurring 2- amino alcohol-bearing head groups (choline, ethanolamine, serine, threonine, etc.), which are crucial intermediates in the build-up of complex phospholipids. In particular, it has recently been discovered that stepwise *N*-methylation of ethanolamine leads to the 335 formation of choline.⁹⁰ Ethanolamine has been found toward the molecular cloud G+0.693-336 0.027 in the Sagittarius B2 complex in the Galactic Center⁹¹ as well as in the Almahata Sitta 337 meteorite⁹² and in a UV irradiated interstellar ice analogue.⁹³ Furthermore, several detections of serine and threonine in carbonaceous chondrites and simulated interstellar ices are 339 reviewed in reference ⁹⁴.

 To ultimately commence an experimental search of the potential astrophysical origins of homochirality of membrane phospholipids, we recorded electronic circular dichroism (ECD) and anisotropy spectra of the phospholipid backbones *sn*-glycerol-1/3-phosphate 343 (G1P/G3P).⁹⁵ While ECD provides information about the wavelength-dependent differential absorption of UV circularly polarized light by enantiomers of chiral molecules (Δ*ε* = *ε*^L - *ε*R; *ε*^L 345 and ε_R are extinction coefficients of left- and right-CPL, respectively), the fractional difference 346 in the absorption given by the anisotropy factor g ($g = \Delta \varepsilon / \varepsilon$, where $\varepsilon = (\varepsilon_L + \varepsilon_R)/2$ is the extinction coefficient) reflects on the enantiomeric excess (*ee*) inducible by preferential UV-348 photon induced photolysis of the more absorbing enantiomer. Consequently, the enantiomeric excess (*ee*) inducible by asymmetric photodecomposition by monochromatic UV CPL can be approximated as

$$
|%ee| \ge \left(1 - (1 - \xi)^{\frac{|g|}{2}}\right) \times 100\% \tag{1}
$$

352 where ξ is the extent of reaction and the sign of *ee* follows the sign of $g^{.96}$ Based on the 353 anisotropy spectra of G1P/G3P reported in Bocková et al.⁹⁵, the asymmetric photolysis of G1P/G3P by monochromatic circularly polarized light at wavelengths corresponding to the anisotropy spectra extrema would introduce a chiral bias of less than 0.5%, even at photolysis rates as high as 99.99%. This suggests rather low efficiency of chirality transfer from chiral photons to phospholipid backbones, especially when compared to the experimentally induced optical purities in amino acids via UV CPL. In an astrophysical scenario, stellar circularly polarized light covers a broad range of wavelengths and is only partially polarized, thus averaging out to even smaller net chiral photon induced *ee*-s. Thence, it is rather improbable that the homochirality of phospholipid backbones would stem from the direct asymmetric interactions of G1P/G3P with circularly polarized light. It is more plausible that alternative processes have been involved in the development of lipid homochirality.

365
366

 Fig. 4 Chiroptical properties of *sn***-glycerol-1-phosphate lithium salt (blue) in aqueous solution reveal the small enantioselective sensitivity of the chiral backbone of glycerophospholipids to circularly polarized light as opposed to D-glyceraldehyde-3-phoshate (purple). a** Electronic circular dichroism, and **b** anisotropy spectra (thick lines) and the *ee* inducible by asymmetric photolysis by monochromatic circularly polarized light at the 370 extent of reaction ζ =0.9999 (thin lines).⁹⁵

 Interestingly, replacing the terminal carbon with a hydroxy group of the G1P moiety by an aldehyde group was found to significantly affect the chiroptical properties (**Fig. 4**), enhancing the magnitude of the asymmetry factor up to ~0.03. This can lead to *ee*-s of up to ~14% at the 375 asymmetric photolysis rate of 99.99%.⁹⁵ Thus, the delivery of enantioenriched glyceraldehyde-phosphate to the early Earth might have shaped the evolution of homochiral phospholipids via synthetic pathways that have yet to be explored. The structural analogues of glyceraldehyde-phosphate, notably glycerol phosphate and glyceraldehyde, were 379 previously detected in simulated interstellar ices^{89,97,98} and carbonaceous chondrites⁹⁹, suggesting that astrochemical synthesis and extra-terrestrial supply of glyceraldehyde- phosphate is highly plausible. So far reported prebiotically relevant synthetic routes leading 382 to phospholipids were recently reviewed by Fiore et al.¹⁰⁰ and Aleksandrova et al.⁹⁰

 An alternative route by which phospholipids could have acquired their homochirality is 384 through interactions with simpler amphiphiles.¹⁵ Although enantioselective adsorption of 385 amino acids on homochiral liposomes¹⁰¹ and their oligomerization¹⁰² have been observed previously, a major shortcoming of the scenario in which the chiral information is transferred from phospholipids to amino acids is how the phospholipids' homochirality would have arisen in the first place. Nevertheless, chirality-specific interactions between amino acids and phospholipids have been clearly evidenced, indicating the potential of the assembly of phospholipids around polypeptide chains. It is, however, very likely that the chirality transfer would have been initiated already before the synthesis of homochiral peptides.

 As discussed in Section III, wet-dry cycling appears to be an efficient means for the synthesis of vesicle forming proto-amphiphilic systems that can simultaneously serve as substrates for primitive peptides. Moreover, polyester microdroplets comprised of hydroxycarboxylic acids 395 were found to scaffold self-assembly of phospholipid amphiphiles around them.⁵⁴ Our recent findings showed that UV CPL of the same helicity and photon energy can simultaneously induce L-enantiomeric excesses in amino acids and hydroxycarboxylic acids in the conditions approximating water-rich interstellar ices' environment as a cradle of prebiotic organic 399 species.¹⁰³ In addition to numerous detections of L-enriched amino acids in carbonaceous 400 chondrites, L-excesses of much less studied hydroxycarboxylic acids were for lactic acid also 401 detected in several meteoritic samples.¹⁰⁴ Taken together, a potential route towards the emergence of protocells might have started from CPL L-enriched amino and hydroxycarboxylic 403 acids brought to the early Earth via meteorites, micrometeorites, or interplanetary dust 404 particles.^{60,70} A small chiral excess generated by CPL could have been potentially amplified 405 during parent bodies alteration processes.¹⁰⁵

 In wet-dry cycles on the primeval Earth, L-enriched depsipeptides would spontaneously 407 emerge from L-enriched amino and hydroxycarboxylic acids (Fig. 5a).^{57,106,107} A combination of ester bond-formation and ester–amide bond exchange during the dry/hot phase and dispersion of reactants and ester bond hydrolysis during the wet/cool phase would have led 410 to the enrichment of depsipeptides with amino acids.⁵⁷ A preference for homochiral chain 411 growth as opposed to racemic was demonstrated by Blocher et al.¹⁰⁸ by studying polycondensation from a pool of racemic tryptophan *N*-carboxyanhydrides. In experiments

by Blair and Bonner,¹⁰⁹ polymerization of a scalemic mixture of leucine *N*-carboxyanhydrides yielded an oligomer with a higher enantiomeric excess of the dominant enantiomer than in 415 the original scalemic mixture. The oligomer was then partially hydrolysed, leaving even more enantiomerically enriched unhydrolyzed polyleucine residues. Moreover, hydrolysis of 417 scalemic leucine oligomers proceeded more readily than for the homochiral chains. Despite 418 the specific conditions in the above-mentioned experiments lacking prebiotic relevance, they provide a precedent that polymerization and hydrolysis are capable of chiral amplification. A repetitive sequence of polymerizations and hydrolyses in environmentally driven wet-dry cycles, both critically involved in the formation of proto-peptides, would have been, therefore, likely favouring the L-enrichments of built-up oligomers.

 Hydroxycarboxylic acid rich depsipeptides would have very likely self-assembled into vesicles like polyesters⁵⁴ and facilitated the adsorption of first phospholipids (**Fig. 5a**), either primitive ones or those synthesised on the early Earth from both terrestrial and extra-terrestrial substrates. Various enantiospecific interaction mechanisms between enantiomers of chiral molecules that direct a plethora of biorecognition processes have been characterised to 428 date,¹¹⁰⁻¹¹² including a recent finding of a spin-dependent exchange interaction due to the 429 chiral-induced spin selectivity (CISS) effect,¹¹³ which results in a stronger binding force between same enantiomers than between optical antipodes. Furthermore, numerous reports of adsorption enantioselectivities of chiral minerals and clays, organic crystals, and metal 432 surfaces can be found in the literature.¹¹⁴ Consequently, it is likely that L-enriched depsipeptides would have transferred their chiral preference to phospholipids. Such enantioenriched phospholipid membranes could have further aided the enantioselective 435 adsorption¹⁰¹ and oligomerization¹⁰² of amino acids leading to even more L-enriched proto- peptides (**Fig. 5b**). In addition, phospholipid bilayers were found to be enantioselectively 437 permeable for amino acids and dipeptides¹¹⁵ which would have further enhanced the chiral excess of the encapsulated material (**Fig. 5b**). Since it is the headgroup chirality which governs 439 such permeation, even primitive phospholipids with shorter carbon chains could have served as such chiral filters. Racemization of the non-encapsulated free chiral moieties, possibly in combination with self-disproportionation mechanisms such as phase transitions, would have 442 ensured continuous re-supply of the L-enantiomers.¹¹⁶ The build-up of enriched oligomers would have then served as a trigger for template-directed stereospecific adsorption and oligomerization, hence continuously increasing the length and the chiral excess of newly built oligomers towards homochirality.

 Fig. 5 Schematic illustrating a potential pathway from enantio-enriched amphiphiles towards the build-up of homochiral phospholipid membranes and proto-peptides. a 1 – pre-concentration of initially CPL L-enriched 449 amphiphiles (e.g. hydroxycarboxylic/amino acids) on the primitive Earth, 2 – polymerization of the amphiphiles
450 upon wet-dry cycling, 3 – self-assembly of polymers into vesicles, 4 – scalemic amphiphilic vesicles s 450 upon wet-dry cycling, – self-assembly of polymers into vesicles, 4 – scalemic amphiphilic vesicles scaffolding
451 the enantioselective self-assembly of phospholipids. **b** An L-enriched scalemic phospholipid ves the enantioselective self-assembly of phospholipids. **b** An ^L-enriched scalemic phospholipid vesicle encapsulating an L-enriched amphiphilic vesicle. 1 – Enantioselective adsorption and 2 – enantioselective permeation of amino acids.

 Clearly, this is a simplified illustrative example of how CPL L-enriched amino and hydroxycarboxylic acids, and a racemic pool of phospholipids could have led tohomochiral peptides and phospholipid membranes. However, the prebiotic pool would have contained many more species, including sugars, incorporation of which would have been necessary for 458 the emergence of primitive genetic information carriers and metabolism. The experiments by 459 Sacerdote and Szostak¹¹⁷ exhibited varying degrees of lipid bilayer permeability to sugars and sugar alcohols. While the membrane permeability showed no chiral preference to the enantiomers of xylose, ribose permeated more readily than other aldopentoses or hexoses. A potential chirality transfer pathway from amino acids to sugars was reported by Breslow 463 and Cheng.¹¹⁸ In addition to hydroxycarboxylic acids and amino acids, other amphiphilic molecules (Section III) would very likely have been involved too, though specific mechanisms and conditions are yet to be elucidated. Still, the hitherto identified chirality interplay between lipids, amino acids, and sugars suggests that homochirality of all the three branches of key building blocks could have evolved prebiotically and simultaneously.

V. Lipids as biomarkers in the quest for extra-terrestrial life

 Life displays discernible differences in chemical composition, stereochemical preferences, and a confined chemical space when compared to non-living organic matter. This provides a 472 foundation for identifying potential biomarkers of extant or extinct life in space.^{9,119–121} On 473 Earth, even-carbon numbered fatty acids $(C_{14}-C_{20})$, odd carbon-numbered hydrocarbons, and specific acyclic isoprenoids such as C15, C20, and C25 are characteristic of biologically derived lipids. Unlike nucleic acids and proteins, lipids consist of only two basic building blocks without

476 a hydrolysable linkage, making them resistant to depolymerization.^{119,120} The presence of

 repeating constitutional sub-units, along with features like homochirality, ubiquity and geo-478 stability, particularly hydrocarbons,²⁵ positions these molecules as excellent biomarkers for

potential extra-terrestrial life.

 Given the similarities in the evolutionary processes of Earth and Mars, the quest for potential 481 Martian life has been a subject of 14 space missions since NASA's Viking mission in 1976.^{11,59} Instruments such as the SAM (Sample Analysis at Mars) on the Curiosity rover, which landed in the Gale Crater in 2012, the SHERLOC (Scanning Habitable Environments with Raman & Luminescence for Organics & Chemicals) on the Perseverance rover, active in the Jezero Crater since 2021, and the MOMA (Mars Organic Molecule Analyser) on the Rosalind Franklin 486 rover, scheduled to be launched in 2028, hold the potential to detect lipids.^{11,25} In the challenging environmental conditions that Mars has endured over billions of years, the detection of any complex organic species in surface or shallow subsurface samples is likely to 489 have been thwarted.¹¹ However, the drilling capabilities of the Rosalind Franklin rover, reaching up to about 2 meters in depth, present the highest potential yet for directly 491 uncovering molecular biosignatures on Mars.¹²⁰ Adding to its uniqueness, the MOMA instrument holds enantio-resolution capabilities through its gas-chromatograph (GC), equipped with a chiral capillary column (CP Chirasil-Dex CB, Agilent). Notably, a Chirasil-Dex column has demonstrated the ability to resolve enantiomers of the branched chiral hydrocarbon 4-methyloctane in a conventional laboratory gas-chromatography mass 496 spectrometry instrument.²⁵ Further testing on the ground-based duplicate (Testbed Model) 497 is necessary to confirm the enantio-resolving capability of the MOMA Flight Model system. It 498 should be noted that lipids serve as biomarkers not only in Mars exploration, but also beyond. The LDI-MS (Laser Desorption Ionization Mass Spectrometry) instrument of the proposed Enceladus Orbilander mission allows for simultaneous detection of lipids, amino acids and 501 polycyclic aromatic hydrocarbons.¹²² Additionally, the SUDA (Surface Dust Analyzer) instrument of the Europa Clipper mission planned for launch in October 2024, based on laboratory analogue experiments, can identify bacterial lipids, DNA, metabolic 504 intermediates, as well as fatty acids, amino acids and peptides while distinguishing 505 between biotic and abiotic origins. $124,125$

 The *in-situ* detection of lipids, as well as any organics in general, faces limitations imposed by the challenges of space instruments in reliably identifying such targets in the harsh conditions 508 beyond Earth.¹¹ To gain further insights into the potential existence of past and/or present extraterrestrial life, sophisticated state-of-the-art laboratory instruments will play a crucial role in analysing samples returned from Mars (Mars Sample-Return, MSR) and its moon Phobos (Martian Moons eXploration, MMX). In contrast, the recently completed successful asteroid sample return missions, such as Hayabusa2 (from the asteroid Ryugu) and OSIRIS- REx (from the asteroid Bennu), have provided the most intact extraterrestrial pieces of the Solar System ever to reach Earth. These missions are invaluable for establishing a baseline for the complexity and diversity of lipids that can be produced abiotically in extraterrestrial environments. At the same time, they will offer the most credible information regarding the

- organics that could have contributed to the prebiotic milieu and hence enlighten our research
- into the origin of life on the early Earth.
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 Fig. 6 Schematics illustrating two conceptual space missions that could enable the identification of potential extraterrestrial lipids. a This illustration shows a concept of the Mars Sample Return Campaign for multiple robots that would team up to ferry to Earth samples of rocks and soil being collected from the Martian surface by NASA's Mars Perseverance rover. The Perseverance rover would deliver its samples directly to the Sample Retrieval Lander. The lander would also carry and deliver to Mars two NASA helicopters to serve as an alternate way to deliver a set of samples stashed on the surface by Perseverance. **Credits:** NASA/JPL-Caltech. **b** Encealdus Orbilander: An artistic representation of the spacecraft designed for orbital exploration around Saturn's moon Enceladus, poised to unveil the mysteries of its icy surface and subsurface environments. The ideal arrival times aim to strike a balance between Jupiter flyby opportunities in the late 2030s and optimal solar illumination at the high southern latitudes of Enceladus, where plume material is abundant. As a result, the mission concept is set to commence in 2030. **Credits:** NASA/Johns Hopkins APL

VI. Conclusion

 While studying contemporary cells has unprecedented importance in identifying common features of all life forms in our biosphere, the complexity of present-day life, which is a result of several billions of years of evolution, can well mask its most primitive origins. This is undoubtedly the case of phospholipid membranes, which are clearly distinguished in bacteria and eukarya as opposed to archaea. While there has been ample discussion about how and when the *lipid divide* occurred, resolving these questions seeks targeted experiments.

 The extra-terrestrial supply of phosphorous-bearing organics likely significantly contributed to the source of bioavailable phosphorous on Earth. Still, the complexity of membrane phospholipids and low efficiency of chirality transfer from chiral photons to chiral backbones of glycerophospholipids suggests that an extra-terrestrial delivery of enantioenriched phospholipids is unlikely to have occurred. In the present review, we have highlighted a potential precursor of phospholipid backbones, glyceraldehyde-phosphate, which could, after efficient enantioenrichment by stellar CPL, trigger the synthesis of enantioenriched phospholipids on the early Earth. The experiments centred at identifying prebiotically plausible synthesis pathways are the subject of future studies. In addition, we discussed an alternative path towards homochiral membranes starting from enantioenriched amphiphiles. Numerous amphiphiles are expected to have been present on the early Earth, the assembly of which upon wet-dry cycling might have shaped not only the evolution of membranes but also primitive peptides. Here, laboratory astrochemistry experiments aimed at enantioselective photosynthesis of amphiphilic species and their analysis in extra-terrestrial samples, both return and meteorites, will be central. Enantioselective interactions between simple chiral amphiphiles, amino acids, sugars, and phospholipids on the primeval Earth are likely to have simultaneously led towards homochiral protocells. Specific processes and driving forces are to be identified and will be of great importance, also for strengthening our search for traces of past and/or present primitive life in space.

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