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1 *Mineral-melt vanadium oxybarometry for primitive arc magmas:*

2 *Effect of hydrous melt composition on fO_2 estimates*

3 *Submitted for publication to Contributions to Mineralogy and Petrology*

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11
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13
14 **AUTHOR CONTRIBUTIONS**

15 All authors contributed to the study conception. Samples were provided by Michel Pichavant. Data
16 collection and analysis were performed by Saskia Erdmann. Data interpretation involved all
17 authors. The draft of the manuscript was written by Saskia Erdmann, while Michel Pichavant and
18 Fabrice Gaillard commented on all versions of the manuscript.

19
20 **STATEMENTS AND DECLARATIONS**

21 The authors have no relevant financial or non-financial interests to disclose

23 **ABSTRACT**

24 If primitive arc magmas are primarily oxidized or if they acquire their oxidizing character during
25 crustal evolution remains debated. Mineral-melt V partitioning is extremely redox sensitive and
26 has the potential to resolve this debate, but critical low-temperature, hydrous experimental
27 partitioning data are limited. We present new experimental V partitioning data for olivine-melt,
28 spinel-melt, and clinopyroxene-melt based on the ≤ 1200 °C, hydrous basalt phase equilibrium
29 experiments of Pichavant et al. (2002) and Pichavant & Macdonald (2007). Combined with
30 published experimental data, we use our olivine-melt V partitioning coefficients to show that –
31 contrary to previous conclusions – hydrous melt composition, i.e. melt H₂O concentration in
32 addition to the concentration of other melt components, affects V partitioning and thus calculated
33 fO_2 , systematically overestimating fO_2 for olivine-melt equilibrated at high melt H₂O concentration.
34 False absolute fO_2 values and false oxidation or reduction trends may be inferred. Based on these
35 findings, we have calibrated a set of new olivine-melt oxybarometers applicable to hydrous arc
36 magmas partially crystallized at ~1025-1350 °C. In a case study on a high-Mg basalt from St.
37 Vincent, Lesser Antilles, we show that olivine-melt V oxybarometry records oxidizing near-
38 liquidus conditions (~QFM+1.8) and possibly limited oxidation (~0.3 log units) during intermittent
39 crustal magma storage.

40

41

42 INTRODUCTION

43 Most arc magmas are more oxidized than mid-ocean ridge magmas, but it remains debated if their
44 more oxidized character derives from oxidized primary melts or if it results from differentiation at
45 crustal level (Canil 2002; Lee et al. 2005; Mallmann & O'Neill 2009; Kelley & Cottrell 2012;
46 Brounce et al. 2014; Laubier et al. 2014; Grocke et al. 2016; Prytulak et al. 2017; Ulmer et al. 2018;
47 Tang et al. 2018, 2019; Bucholz & Kelemen 2019; Wang et al. 2019). Several whole-rock
48 geochemical models (for V/Sc, Eu/Eu*, Zn/Fe) as well as phase equilibrium experiments have been
49 used to suggest that the majority of mantle-derived magmas initially has an oxygen fugacity (fO_2)
50 near the quartz-fayalite-magnetite (QFM) buffer, regardless of their geodynamic origin, and that
51 more oxidized conditions are acquired through fractionation, crustal assimilation, or degassing
52 (e.g., Lee et al. 2005; Mallmann & O'Neill 2009; Ulmer et al. 2018; Tang et al. 2018, 2019). Most
53 mineral-melt oxybarometry, experimental and recent whole-rock geochemical modelling studies,
54 in contrast, support the notion that primitive mantle-derived magmas have variable fO_2 , where
55 primitive arc magmas are generally more oxidized (commonly \sim QFM+0.5 to QFM+3.5) than
56 primitive mid-ocean ridge magmas (commonly \sim QFM \pm 0.5) (e.g., Carmichael 1991; Pichavant &
57 Macdonald 2007; Kelley & Cottrell 2012; Brounce et al. 2014; Gaillard et al. 2015; Bucholz &
58 Kelemen 2019). Oxidation of primitive arc magmas through fractionation, crustal assimilation, or
59 degassing is not ruled out, but invoked to be superimposed on relatively oxidized primary melts.

60 Consensus exists in that accurately constraining the fO_2 of primary magmas and their
61 possible redox evolution is critical for resolving the debate and for improving our understanding of
62 Earth's redox reservoirs through time. Vanadium partitioning between minerals and melts is known
63 to be extremely redox sensitive, particularly between \sim QFM-5 and QFM+5 (Canil 2002; Lee et al.
64 2005; Mallmann & O'Neill 2009) and thus over a large fO_2 range that covers most estimated

65 terrestrial magma redox conditions (Gaillard et al. 2015; Righter et al. 2020). A number of mineral-
66 melt V oxybarometers exists for olivine, clinopyroxene, and spinel (Canil & Fedotchouk 2001;
67 Mallmann & O'Neill 2013; Shishkina et al. 2018; Wang et al. 2019), which are increasingly
68 applied, given that V compositions can be easily and rapidly acquired by laser ablation inductively
69 coupled plasma mass spectrometry and for some minerals by electron microprobe. Existing
70 experimental V mineral-melt partitioning data are, however, biased towards high-temperature,
71 anhydrous equilibria. To date, few experimental studies – those of Laubier et al. (2014), Shishkina
72 et al. (2018), and Wang et al. (2019) – have systematically characterized V mineral-melt
73 partitioning at relatively low temperatures (≤ 1200 °C) for compositions approaching those of
74 primitive arc magmas and at anhydrous to hydrous conditions. To add to the limited database, we
75 have acquired new partitioning data for olivine-melt, clinopyroxene-melt, and spinel-melt for
76 hydrous, 1050-1200 °C basalt phase equilibrium experiments of Pichavant et al. (2002) and
77 Pichavant & Macdonald (2007). Using our and published data, we first test the recent V olivine-
78 melt, clinopyroxene-melt, and spinel-melt oxybarometers of Mallmann & O'Neill (2013),
79 Shishkina et al. (2018), and Wang et al. (2019) and then derive a new set of olivine-melt V
80 oxybarometers for arc magmas, which we apply to constrain fO_2 and its evolution for a primitive,
81 high-Mg magma from St. Vincent, Lesser Antilles.

82

83 **METHODS BACKGROUND**

84 Vanadium can occur in valence states V^0 , V^{2+} , V^{3+} , V^{4+} , or V^{5+} (Sutton et al. 2005; Wood et al.
85 2008), with V^{3+} and V^{4+} being the dominant species at terrestrial magma redox conditions (Canil
86 1997; Lee et al. 2005; Mallmann & O'Neill 2009). In olivine, clinopyroxene, and spinel, V^{3+} is
87 significantly more compatible than V^{4+} (e.g., Canil 1997; Righter et al. 2006; Mallmann & O'Neill

88 2009; Shishkina et al. 2018; Wang et al. 2019), and olivine/clinopyroxene/spinel-melt V
89 partitioning coefficients therefore strongly decrease with increasing oxidation state, while other
90 intensive parameters also have a minor, but nevertheless relevant effect on mineral-melt V
91 partitioning. Olivine-melt V partitioning coefficients are inferred to increase with decreasing
92 temperature, increasing olivine forsterite content, increasing proportions of melt network formers
93 SiO₂ and Al₂O₃ relative to network modifiers CaO, Na₂O, and K₂O, or decreasing anhydrous melt
94 NBO/T (i.e. the ratio of non-bridging oxygen to tetrahedral cations in the melt) (Mallmann &
95 O'Neill 2013; Wang et al. 2019). Pressure and melt H₂O concentration, in contrast, have been
96 inferred to have no significant effect on V olivine-melt partitioning (Canil & Fedortchouck 2001;
97 Shishkina et al. 2018; Wang et al. 2019). Vanadium partitioning coefficients for clinopyroxene-
98 melt and spinel-melt are inferred to also increase with decreasing temperature and to vary with
99 mineral and melt composition, which in turn are controlled by other variables including pressure
100 (Horn et al. 1994; Toplis & Corgne 2002; Wang et al. 2019; Holycross & Cottrell 2022).

101 The olivine-melt V oxybarometer of Mallmann & O'Neill (2013) has been calibrated for a
102 large fO_2 range (~QFM-4 to QFM+4), but at high temperature (1200-1530 °C) and anhydrous,
103 atmospheric conditions. Effects of temperature, olivine forsterite content, and melt SiO₂, Al₂O₃,
104 CaO, Na₂O, and K₂O in addition to fO_2 on V partitioning are considered in the calibration, which
105 has an estimated uncertainty of ~0.5 log units at 2 σ level. The olivine-melt V oxybarometer of
106 Wang et al. (2019) has also been calibrated for a large fO_2 range (~QFM-4 to QFM+4) and at
107 relatively high temperature (1150-1470 °C), yet for anhydrous to hydrous, low- to high-pressure
108 conditions. Effects of temperature and anhydrous NBO/T in addition to fO_2 on V partitioning are
109 considered in the calibration with an apparent uncertainty of ~0.5 log units at 2 σ level (cf. their
110 Fig. S5). The olivine-melt V oxybarometer of Shishkina et al (2018) (equation 4) has been

111 calibrated for primitive arc magmas crystallized at ≤ 1250 °C and at hydrous to anhydrous, low- to
112 intermediate-pressure conditions and for a relatively large fO_2 range (\sim QFM-2 to QFM+3). Effects
113 of intensive parameters in addition to fO_2 on olivine-melt V partitioning were not ruled out, but
114 considered insignificant for the calibration range. The uncertainty of the oxybarometer is estimated
115 to increase with decreasing V partitioning coefficient (given that V concentrations in olivine
116 decrease and analytical uncertainties thus increase), ranging between \sim 0.5 and 1.0 log units at 2σ
117 level. The clinopyroxene-melt and spinel-melt V oxybarometers of Wang et al. (2019) have been
118 calibrated for the same fO_2 range (\sim QFM-4 to QFM+4) as their olivine-melt oxybarometer, also at
119 high temperature (\sim 1150-1350 °C) and at anhydrous to hydrous, low- to high-pressure conditions.
120 Effects of temperature and tetrahedral Al (for clinopyroxene) or Cr number (for spinel) in addition
121 to fO_2 on V partitioning are considered in the calibrations, which have apparent uncertainties of
122 \sim 0.5 log units at 2σ level (cf. their Fig. S5).

123

124 **GEOLOGICAL BACKGROUND**

125 Primitive high-Mg basaltic magmas (i.e. primary or near-primary mantle-derived melts) erupt in
126 arcs, but they are not commonplace (e.g. Grove et al. 2012), with some exceptions including the
127 southern section of the Lesser Antilles arc. The Lesser Antilles arc is a mature ocean-ocean
128 subduction zones that extends for \sim 750 km, resulting from the westward subduction of the Atlantic
129 beneath the Caribbean Plate, which has been active since the early Miocene, following earlier
130 subduction systems in the eastern Caribbean (Macdonald et al. 2000; Evain et al. 2013; Allen et al.
131 2019). It comprises the islands of Saba, Statia, St. Kitts, Nevis, Montserrat, Guadeloupe, Dominica,
132 Martinique, St. Lucia, St. Vincent, the Grenadines, and Grenada, which all have active or
133 potentially active volcanic centres. A wide range of magma types have erupted, ranging from low-

134 K tholeiites over calc-alkaline basalts to alkalic basalts. The primitive, high-Mg basalts have mostly
135 erupted on St. Vincent, the Grenadines and Grenada, which are interpreted as parental magmas of
136 low-Mg, high-Al basalts and other more evolved magmas of the arc (Heath et al. 1998; Macdonald
137 et al. 2000; Pichavant & Macdonald 2007; Melekhova et al. 2015). They provide an opportunity to
138 study the fO_2 and other intensive parameters of mantle-derived magmas that have been largely
139 unaffected by fractionation or contamination at crustal level.

140 The island of St. Vincent, from which the sample of our study derives, is dominated by
141 Soufriere volcano in the northern part of the island, which is largely composed of calc-alkaline
142 basalts and basaltic andesites (Heath et al., 1998). Its early activity (~0.6 Ma-10 ka) was dominated
143 by the eruption of basaltic and basaltic andesitic lavas, the so-called Pre-Somma lavas. Later
144 activity produced the Yellow Tuff Formation (black scoria to yellow lapilli and tuff; 3600-4500
145 years bp), the Crater Lava Formation (ponded basaltic and basaltic andesitic lavas), and the
146 Pyroclastic Formation (succession of pyroclastic fall and flow deposits) with at least five major
147 historic eruptions. Sample STV301 of Heath et al. (1998), which we have used, is a high-Mg basalt
148 that derives from the Pre-Somma unit sampled at Black Point (13°15.65'N, 61°07.05'W). High-
149 Mg basalt STV301 and other high-Mg basalts from the southern part of the arc are inferred to
150 represent slightly enriched, near-primary hydrous mantle melts, which have last equilibrated with
151 peridotite at ~1130-1240°C and ~11-17 GPa (Heath et al. 1998; Pichavant et al. 2002; Bouvier et
152 al. 2008; Melekhova et al. 2015). Experimental studies and olivine-spinel oxybarometry have
153 inferred an oxidized redox state with an fO_2 of ~QFM+0.9 to ~QFM+2.6 for the primitive St.
154 Vincent magmas (Heath et al. 1998; Macdonald et al. 2000; Pichavant et al. 2002; Pichavant &
155 Macdonald 2007), which is within the range of fO_2 conditions (~QFM+0.5 to +3.5) typically
156 inferred for arc magmas using petrological methods (e.g., Carmichael 1991; Kelley & Cottrell

157 2012; Brounce et al. 2014; Gaillard et al. 2015; Bucholz & Kelemen 2019). Melt H₂O
158 concentrations of high-Mg basalts from St. Vincent, including STV301, are inferred to have been
159 ~5 wt% (Pichavant & Macdonald 2007; Bouvier et al. 2008). Melt CO₂ concentrations and thus
160 direct constraints on melt volatile saturation remain unconstrained for the Pre-Somma lavas, but
161 are high (up to >3000 ppm CO₂) for dacitic melt inclusions from basaltic andesites erupted in 1477
162 and 1902 AD at St. Vincent (Balcone-Boissard et al. 2023). Some originally volatile-undersaturated
163 high-Mg basalts may have fractionated to high-Al basalts and more evolved magma with ~7-8 wt%
164 H₂O (Pichavant & Macdonald 2007; Balcone-Boissard et al. 2023), but most St. Vincent and Lesser
165 Antilles melt inclusion H₂O and K₂O compositions show no or negative correlation (Bouvier et al.
166 2008; Balcone-Boissard et al. 2023), which indicates degassing-driven crystallization and thus
167 volatile saturation during partial crystallization at crustal level.

168

169 **SAMPLES, METHODS, NEW DATA**

170 We have investigated the natural STV301 high-Mg basalt in one thin section and selected run
171 products of phase equilibrium experiments of Pichavant et al. (2002) and Pichavant & Macdonald
172 (2007) that have used STV301 as the starting material. The experimental run products of Pichavant
173 et al. (2002) and Pichavant & Macdonald (2007) were all considered for our study, but only phases
174 from a subset of eleven charges were feasible for trace-element analysis by laser ablation
175 inductively coupled plasma mass spectrometry (LA-ICP-MS). Eight of the charges were partially
176 crystallized in internally heated pressure vessels (IHPV; experiments of Pichavant & Macdonald
177 2007), whereas three of the charges were partially crystallized in piston cylinder presses (PC;
178 experiments of Pichavant et al. 2002). The experiments were performed at conditions ranging from
179 ~QFM+0.7 to ~QFM+3.1, ~1050 to ~1200 °C, and ~400 to 1450 MPa, where melt H₂O

180 concentration varied between ~2.3 and 7.7 wt% (Table 1). Details on the control of the
181 experimental intensive parameters are provided below, in the supplementary text, and in the
182 publications of Pichavant et al. (2002) and Pichavant & Macdonald (2007).

183 All samples were imaged in backscatter-electron mode using a ZEISS Merlin Compact
184 SEM at the Institut des Sciences de la Terre d'Orléans (ISTO-CNRS-Université d'Orléans-
185 BRGM), France. Phase proportions of the natural sample were determined by point-counting (this
186 study), while phase proportions of the experimental run products have been previously determined
187 by mass balance (Pichavant et al. 2002 and Pichavant & Macdonald 2007). For the experimental
188 samples, we have used the major- and minor-element phase compositions reported by Pichavant et
189 al. (2002) and Pichavant & Macdonald (2007). For the natural sample, we have determined new
190 major- and minor-element compositions for olivine, spinel, clinopyroxene, and matrix using a
191 Cameca SX five electron microprobe in wavelength dispersive mode (details are provided in the
192 supplementary text). The redox-sensitive V concentrations of olivine, spinel, clinopyroxene, and
193 glass/matrix were determined for the natural and the experimental samples using LA-ICP-MS.
194 Temperature-sensitive concentrations of Sc, Y, and Ni were also determined for all phases by LA-
195 ICP-MS. For the experimental samples, the data showed zoning in Sc for some olivine (equilibrated
196 at low-temperature) and contamination problems for Ni and Y for other charges (equilibrated in
197 PCs). Given these problems, we report the experimental Sc, Y, and Ni data only in the
198 supplementary text, figures, and Table S1. For the natural samples, the Sc, Y, and Ni concentrations
199 are reported in the main text, which we use to interpret and to quantify crystallization conditions.
200 The four elements of interest (V, Sc, Y, Ni) as well as Si and Al as internal standards were measured
201 as masses ^{27}Al , ^{29}Si , ^{45}Sc , ^{51}V , ^{60}Ni , and ^{89}Y using a RESolution SE ArF excimer laser ablation
202 system coupled to an Agilent 8900 QQQ mass spectrometer. Instrument parameters were set to

203 common conditions, and calibration and acquisition followed established procedures, which are
204 outlined in the supplementary text and in Table S1. Phase compositions were determined by point
205 analyses, while zoning was evaluated by LA-ICP-MS compositional mapping. Raw LA-ICPMS
206 data were reduced using Glitter (point analyses; version 4.4.4; van Achterbergh et al. 2000) and
207 Iolite (compositional maps; version 4.7.1; Paton et al. 2011; Petrus et al. 2017). Accuracy and
208 precision determined for V, Sc, Y, and Ni concentrations in BCR-2G, BHVO-2G, and BIR-1G
209 reference materials were mostly better than 15 and 5%, respectively (Table S2).

210

211 **OXYBAROMETER TEST AND CALIBRATION DATA**

212 Our aim was to test how the recent olivine-melt V oxybarometers of Mallmann & O'Neill (2013),
213 Shishkina et al. (2018) (equation 4), and Wang et al. (2019) as well as the clinopyroxene-melt and
214 spinel-melt V oxybarometers of Wang et al. (2019) perform for primitive, hydrous arc magmas and
215 to improve on them if possible. We have therefore focused on mineral-melt V partitioning
216 coefficients determined for experiments performed at hydrous conditions and low temperatures
217 with starting material compositions approaching those of primitive arc magmas, but we have also
218 integrated relevant anhydrous V partitioning data (Fig. 1). In addition to our new data presented
219 below, we have considered the experimental studies of Laubier et al. (2014), Shishkina et al. (2018),
220 and Wang et al. (2019), for which a close approach to equilibrium was inferred in the original
221 publications (to which the reader is referred for details). However, to consider only the highest
222 quality data, we have selected a subset of the published data (1) from experiments with estimated
223 melt H₂O concentrations of <10 wt% (which can be quenched to a homogeneous glass); and (2)
224 from experiments with $\leq \pm 20\%$ Fe loss/gain (which can be assumed to have approached chemical
225 and fO_2 equilibrium), which excluded data from 17 experiments of Wang et al. (2019).

226 The hydrous partitioning coefficients of Shishkina et al. (2018), Wang et al. (2019), and our
227 new data were supplemented by anhydrous, low-temperature V partitioning data from Laubier et
228 al. (2014) for the same fO_2 range and similar starting material compositions. This selection yielded
229 an anhydrous to hydrous partitioning dataset equilibrated at an fO_2 of $\sim\Delta QFM-2.0$ to $\sim\Delta QFM+4.0$,
230 at melt H_2O concentrations of 0-9.5 wt%, and at temperatures of 1025-1350 °C but mostly 1025-
231 1200 °C (Fig. 1) with 54 partitioning coefficients for olivine-melt (Table S4), 13 partitioning
232 coefficients for spinel-melt (Table S5) and 11 partitioning coefficients for clinopyroxene-melt
233 (Table S6). Vanadium partitioning data determined at higher temperatures or for compositions that
234 strongly deviate from the composition of primitive arc magmas (e.g., those of Canil &
235 Fedortchouck 2001; Toplis & Corgne 2002; Papike et al. 2013; Mallmann & O'Neill 2009, 2013;
236 Holycross & Cottrell 2022) were not used. The fO_2 conditions of the considered studies were
237 variably expressed in log units or relative to various QFM or NNO buffer calibrations, which we
238 have recalculated relative to the QFM buffer of O'Neill (1987) (Tables S4-S6).

239 Our focus on hydrous partitioning data for arc-type magma compositions is different from
240 the approach taken by Shishkina et al. (2018) and Wang et al. (2019) for their V oxybarometers for
241 arc magmas, who have integrated limited hydrous olivine-melt V partitioning data ($n=12$ or $n=26$)
242 with anhydrous partitioning data ($n=47$ or $n=169$) from experiments extending over a larger
243 temperature range and also a larger compositional range than the hydrous equivalents (Fig. 1). We
244 posit that focusing our dataset on hydrous to anhydrous V partitioning data for compositions,
245 temperatures ($\sim 1025-1350$ °C), and fO_2 conditions ($\sim QFM-2$ to $QFM+4$) relevant to partial
246 crystallization of most primitive arc magmas (Carmichael 1991; Grove et al. 2012; Kelley &
247 Cottrell 2012; Brounce et al. 2014; Gaillard et al. 2015; Bucholz & Kelemen 2019) permits us to

248 best assess relevant effects of their various intensive parameters on V partitioning and thus on fO_2
249 estimates.

250

251 **EXPERIMENTAL fO_2 AND MELT H_2O CONSTRAINTS**

252 For atmospheric pressure experiments in gas mixing furnaces (i.e. for the considered experiments
253 of Laubier et al. 2014), control of intensive parameters is relatively straightforward with melt H_2O
254 concentrations of <0.1 wt% and an accuracy of better than ± 0.1 log units for fO_2 . For high-pressure
255 experiments, fO_2 and melt H_2O concentration are more difficult to maintain and to determine, given
256 that the systems are partly open. Oxygen fugacity may be imposed through pressurizing gas
257 mixtures (commonly Ar- H_2), experimental assemblies, double-capsule buffer techniques, and fluid
258 concentrations (H_2O - CO_2) in individual charges. The fO_2 imposed by gas mixtures may be directly
259 measured by Shaw membrane (experiments of Shishkina et al. (2018)), but is more commonly
260 determined using solid buffer capsules or redox-sensitive run products (experiments of Pichavant
261 et al. (2002), Pichavant & Macdonald (2007), and Wang et al. (2019)). For experiments in internally
262 heated pressure vessels (experiments of Pichavant & Macdonald (2007) and Shishkina et al. (2018))
263 and piston cylinder presses (experiments of Pichavant et al. (2002) and Wang et al. (2019)),
264 experimental fO_2 is typically constrained with accuracies of better than $\sim \pm 0.3$ to 0.5 and $\sim \pm 0.5$ to
265 1.0 log units, respectively. Melt H_2O concentrations can be constrained by various techniques, but
266 only indirectly, i.e. for the quenched glasses. Water concentrations measured by secondary ion
267 mass spectrometry (some data from Pichavant et al. (2002) and Pichavant & Macdonald (2007)) or
268 Fourier-transform infrared spectroscopy (some data of Shishkina et al. (2019)) are in most cases
269 accurate to within ± 0.5 wt%, whereas melt H_2O concentrations inferred on the basis of saturation
270 models are typically within ± 0.5 -1.0 wt% (some data of Shishkina et al. (2019)). Melt H_2O

271 concentrations calculated by difference from electron microprobe totals (some data from Pichavant
272 et al. (2002) and Pichavant & Macdonald (2007) and all data from Wang et al. (2019)) have
273 accuracies of ideally $\sim\pm 1.0$ wt% (Devine et al. 1995) but may overestimate concentrations by up
274 to $\sim+2$ wt% (Hughes et al. 2019) or more (e.g. Holycross & Cottrell (2022), especially for high
275 melt H₂O concentrations. These uncertainties are indicated and considered below; further details
276 are provided in the published experimental studies of Pichavant et al. (2002), Pichavant &
277 Macdonald (2007), Laubier et al. (2014), Shishkina et al. (2018), and Wang et al. (2019).

278

279 **STUDY OUTLINE**

280 We first present a textural-compositional characterization of olivine, spinel, clinopyroxene and
281 glass of the STV301 phase equilibrium experiments of Pichavant et al. (2002) and Pichavant &
282 Macdonald (2007). Together with the selected literature data (Tables S3 to S5), we then use our
283 compositions to evaluate (1) the olivine-melt V oxybarometers of Mallmann & O'Neill (2013),
284 Shishkina et al. (2018) (equation 4), and Wang et al. (2019); and (2) the spinel-melt and
285 clinopyroxene-melt V oxybarometers of Wang et al. (2019). Following the evaluation, we
286 formulate a set of new olivine-melt V oxybarometers for hydrous arc magmas partially crystallized
287 at ~ 1050 - 1350 °C, which we then apply in a petrographic case study on the natural STV301 basalt
288 from St. Vincent, in combination with all tested oxybarometers and the Sc/Y and Ni thermometers
289 of Mallmann & O'Neill (2013) and Pu et al. (2017). A challenge for applying the mineral-melt V
290 oxybarometers and thermometers to the natural STV301 sample was that a record of melt
291 compositions in the form of glassy melt inclusions or glassy matrix is lacking. Possible crystallized
292 melt inclusions (i.e. inclusions in 2D but not necessarily in 3D) large enough for LA-ICP-MS
293 analysis are also rare, which precluded meaningful characterization, and we have therefore (1)

294 analyzed the crystallized matrix/groundmass of the natural sample and (2) estimated melt
295 compositions as outlined later in the text.

296

297 **RESULTS AND DISCUSSION**

298 **Experimental constraints**

299 *Phase relations and compositions*

300 The STV301 experimental run products – which have been previously characterized by Pichavant
301 et al. (2002) and Pichavant & Macdonald (2007) in regards to their assemblage, textures, and major-
302 and minor-element phase compositions – comprise olivine-spinel-glass, olivine-spinel-
303 clinopyroxene-glass, olivine-spinel-clinopyroxene-plagioclase-glass, or orthopyroxene-
304 clinopyroxene-spinel-glass and quench crystals (Table 1; Fig. S1). Olivine, spinel, clinopyroxene,
305 and glass were successfully analyzed by LA-ICP-MS in 9, 3, 2, and 12 charges, respectively (Table
306 1; Tables S4-S6). Olivine (~2-17 wt%) forms subhedral to euhedral crystals, which are up to ~100
307 μm but typically ~10-20 μm in size. Olivine crystals of the ≥ 1180 °C experiments are unzoned and
308 largely inclusion-free (Fig. S2a-c). Olivine formed at ≤ 1090 °C, however, becomes partly zoned
309 and has locally abundant small (<1-2 μm) spinel, glass, and/or fluid inclusions especially in crystal
310 cores (Fig. S2d-i). Unzoned inclusion-free olivine cores and unzoned inclusion-free near-rim zones
311 have low V concentrations (~1.7-12.7 ppm), which decrease with increasing $f\text{O}_2$ (Table 1). The
312 inclusion-rich cores, in contrast, yield relatively V-rich compositions, representing mixtures of
313 olivine \pm spinel \pm glass inclusion analyses, which we have removed from the dataset. The variation
314 of V in olivine of individual charges is relatively low (mostly \sim <15% 2RSDs; Table 1), indicating
315 that inclusions were successfully eliminated from our data set.

316 Spinel (~1-2 wt%) in the experimental charges forms euhedral to subhedral crystals and
317 inclusions in olivine, which are commonly ≤ 10 μm in size (Fig. S2a). In all experimental run
318 products, some spinel crystals show core-rim zoning from BSE-bright cores to BSE-dark rims,
319 where zoning is sharp in the low-temperature, low- H_2O experiments and gradational in the high-
320 temperature, high- H_2O experiments. Cores of the zoned crystals were not analyzed, as they are
321 inherited from the starting material (cf. Pichavant et al. (2002) and Pichavant & Macdonald (2007);
322 Fig. S1a). Cores of the unzoned spinel crystals and near-rim zones of the zoned spinel crystals have
323 high V concentrations (~320-1080 ppm) which strongly decrease with increasing experimental $f\text{O}_2$
324 with limited variation in spinel of individual charges ($\leq 25\%$ 2RSDs) (Table 1). The experimental
325 clinopyroxene crystals (~6-25 wt%) are subhedral-euhedral, typically clustered in the experimental
326 charges, and commonly $< 5-10$ μm large (Fig. S2g-i). They show no obvious core-rim zoning in
327 BSE images, but they were too small to assess trace-element zoning by LA-ICP-MS mapping. The
328 crystals have intermediate V (~130-450 ppm), showing limited variation in crystals within
329 individual charges ($\leq 15\%$ 2RSDs), but only few crystals were large enough for analysis (Table 1).

330 Glasses (~51-96 wt%) of the experimental run products are largely homogeneous in BSE
331 imaging mode, but commonly comprise quench crystals, especially those of the H_2O -rich
332 experimental charges (Fig. S2a,d,g). Glasses around quench crystals (Fig. S2a-f) and pyroxene
333 (Fig. S1b,c,d,f,j, S2g) show $< 5-10$ μm wide, BSE-dark boundary layers, which were too narrow
334 for quantitative analysis (Fig. S2d-i). Glasses away from boundary layers have high V (~250-370
335 ppm), where concentrations primarily vary with phase assemblage and proportions and $f\text{O}_2$ (Table
336 1). They are relatively homogeneous in V ($< 20\%$ and mostly $< 3\%$ 2RSDs), where pairs of the
337 homogeneous glass and inclusion-free, near-rim mineral compositions yield V partitioning
338 coefficients for olivine-glass (~0.006-0.041), spinel-glass (~1.1-3.3), and clinopyroxene-glass

339 (~0.3-1.8) that compare to partition coefficients previously determined in other studies, and
340 particularly to those of other low-temperature experiments (Fig. 2a-c; Table 1).

341

342 *Approach to equilibrium for mineral-melt V partitioning*

343 Pichavant et al. (2002) and Pichavant & Macdonald (2007) have inferred a close approach to total
344 equilibrium for most of their STV301 experimental assemblages using established tests (i.e.,
345 experimental duration, textural evidence, homogeneous major- and minor-element phase
346 compositions in individual charges as well as systematic compositional variation between charges,
347 mineral-melt major-element partitioning coefficients in agreement with literature values, and low
348 residuals in mass balance calculations), but highlighted partial disequilibrium for spinel and
349 clinopyroxene. For our study, it is most important to identify partial equilibrium, i.e. equilibrium
350 between the V compositions of selected mineral-glass compositional pairs.

351 (1) Homogenous textures and the lack of zoning in minerals and glasses of most charges is
352 consistent with an approach towards equilibrium. Glass boundary layers around
353 clinopyroxene (<5-10 μm wide), however, indicate partial disequilibrium. Clinopyroxene
354 appears unzoned in BSE images, but crystals were too small to investigate trace-element
355 zoning. The presence of core-rim zoning of spinel in all experiments and core-rim zoning
356 of olivine in ≤ 1155 °C experiments also records partial disequilibrium. The cores of the
357 zoned spinel crystals are inherited from the starting material (Fig. S1a), which have failed
358 to dissolve completely during preparation of the starting glasses and during the experiments,
359 as reported by Pichavant et al. (2002) and Pichavant & Macdonald (2007). The cores of the
360 zoned olivine crystals (Fig. S2g-i), in contrast, have formed during the experiments, which
361 are interpreted to reflect fast crystallization relative to diffusion and incomplete subsequent

362 equilibration. The presence of compositionally distinct cores of spinel and olivine had a
363 limited effect on phase compositions (e.g. some elements were preferentially retained in the
364 crystal cores), but it had no effect on our calculated partition coefficients and calculated
365 intensive parameters, as long as the unzoned near-rim mineral and main melt compositions
366 were in equilibrium.

367 (2) Unzoned olivine and unzoned spinel core and rim compositions and glass compositions of
368 individual charges show low V variation (<20% and mostly <3% 2RSDs) and calculated V
369 partition coefficients that are within the range of published values (Fig. 2a,b; Table 1),
370 which decrease with increasing fO_2 as expected (e.g., compared to data of Laubier et al.
371 2014; Shishkina et al. 2018; Wang et al. 2019), suggesting that they reflect equilibrium
372 partitioning. The calculated clinopyroxene-melt V partition coefficients also broadly
373 compare to published low-temperature data (Fig. 2c), falling between the partitioning
374 coefficients of Laubier et al. (2014) and Wang et al. (2019) and those of Toplis & Corgne
375 (2002) and Holycross & Cottrell 2022, and they decrease with increasing fO_2 as expected.
376 The presence of melt (now glass) boundary layers around clinopyroxene crystals
377 nevertheless indicates partial disequilibrium, but V compositions may not have been
378 significantly affected with partitioning coefficients close to unity (~0.3 and ~1.5) and thus
379 limited if any V enrichment or depletion in the melt boundary layers. Further below, we
380 therefore continue to assess if our clinopyroxene-melt V partitioning coefficients
381 approached equilibrium, evaluating them against inferred equilibrium partitioning data of
382 Laubier et al. (2014) and Wang et al. (2019).

383

384

385 *Vanadium oxybarometry tests*

386 The focus of our study was on olivine-melt V oxybarometers, but we have also considered spinel-
387 melt and clinopyroxene-melt V oxybarometers, which may extend the range of applications. We
388 have tested overall performance of the calibrations and evaluated bias for calculated compared to
389 experimental fO_2 relative to experimental temperature, pressure, as well as mineral and melt
390 composition. Melt composition was considered through melt H₂O concentration and NBO/T
391 composition (i.e. the ratio of non-bridging oxygen to tetrahedral cations). Melt NBO/T is a first-
392 order approximation of melt polymerization, which was calculated as $NBO/T = (2X_O - 4X_T) / X_T$ (cf.
393 Mysen et al. 1982), where X are the atomic proportions of oxygen (O) and tetrahedrally coordinated
394 cations (T=Si⁴⁺, Ti⁴⁺, and P⁵⁺, plus Al³⁺ and Fe³⁺ charge balanced by excess alkalis, with Fe³⁺
395 calculated using the equation of Kress & Carmichael (1991)). In addition to anhydrous NBO/T
396 (commonly used, calculated from anhydrous glass compositions), we have calculated hydrous
397 NBO/T (calculated from hydrous glass compositions considering molecular H₂O concentrations
398 and formation of T-OH bonds, e.g. following Di Muro et al. 2006). Accounting for analytical
399 uncertainties in determining melt composition and especially melt H₂O concentration, we assign
400 uncertainties of 0.1 and 0.2, respectively, to the calculated anhydrous and hydrous NBO/T values.
401 We highlight that the structure of multi-component hydrous melts cannot yet be fully quantified
402 (e.g., Behrens 2020), and that the hydrous NBO/T values may not give a true image of melt
403 structure, but that they permit quantitative comparison of variably hydrous melts, accounting for
404 the fact that hydrous melt component are known to significantly affect melt polymerization.

405 To test the olivine-melt oxybarometers of Mallmann & O'Neill (2013), Shishkina et al.
406 (2018) (equation 4), and Wang et al. (2019) for primitive arc magmas, we have used the selected
407 hydrous to anhydrous experimental V partitioning data equilibrated at conditions (~ΔQFM-2.0 to

408 $\Delta\text{QFM}+4.0$, ~ 1025 to 1350 °C, <10 wt% melt H_2O , 0 to 2000 MPa) that match those typical for
409 partial crystallization within the crust and upper mantle (Table S4). The starting materials of the
410 considered experimental studies were natural high-Mg basalts (our data and the data of Shishkina
411 et al. 2018), natural high-Mg basaltic andesite and mid-ocean ridge basalt (Laubier et al. 2014),
412 and synthetic high-Mg basalt mixed with peridotite (Wang et al. 2019). None of the test data were
413 part of the Mallmann & O'Neill (2013) calibration. However, the data of Laubier et al. (2014) and
414 Shishkina et al. (2018) were part of the Shishkina et al. (2018) calibration, and the data of Laubier
415 et al. (2014) and Wang et al. (2019) were part of the Wang et al. (2019) calibrations. Using some
416 of the oxybarometer calibration data enabled us to test the oxybarometers for the full range of
417 hydrous to anhydrous V partitioning data relevant for arc magmas, permitting direct comparison of
418 the different calibrations. The Shishkina et al. (2018) V oxybarometer is the simplest of the
419 evaluated calibrations in that it considers only the effect of $f\text{O}_2$ on olivine-melt partitioning, thus
420 showing most directly the possible effect of other intensive parameters on V partitioning and
421 oxybarometric estimates.

422 For the test data, the three olivine-melt oxybarometers calculate $f\text{O}_2$ values that are close to
423 the experimental $f\text{O}_2$ values considering uncertainties (Fig. 3a-c). The calculated and experimental
424 $f\text{O}_2$ values are (1) mostly within ± 0.8 and 0.7 log units (2SDs) of each other for the Wang et al.
425 (2019) and Mallmann & O'Neill (2013) calibrations (Fig. 3a,b); and (2) mostly within ± 0.4 log
426 units (2 SDs) of each other for the Shishkina et al. (2018) calibration, which was specifically
427 derived for hydrous, low-temperature equilibration (Fig. 3c). However, calculated compared to
428 experimental $f\text{O}_2$ values (expressed as ΔQFM (Calc-Exp) show systematic bias as a function of
429 various intensive parameters for all three calibrations (Figs 3d-l). Most of the data show concurring
430 trends, but the data of Wang et al. (2019) deviate from the correlation trends in some instances,

431 which we describe below. Test data outside the temperature calibration ranges (i.e. data equilibrated
432 at <1200 °C for MO2013, data equilibrated at <1150 °C for W2019, and data equilibrated at >1250
433 °C for S2018) or the two high-temperature (1350 °C) data, in contrast, do not systematically deviate
434 from correlation trends (Fig. 3). Using the Mallmann & O'Neill (2013) calibration, ΔQFM (Calc-
435 Exp) shows negative correlation with temperature (Fig. 3d; $R^2=0.65$) and no correlation with
436 pressure (Fig. S3a, $R^2=0.06$). ΔQFM (Calc-Exp) also shows no correlation with olivine forsterite
437 content (Fig. S3b), with anhydrous or hydrous NBO/T (Fig. 3g,j), or melt H₂O concentration (Fig.
438 S3c) for the entire test data. However, significant positive correlation is observed with hydrous
439 NBO/T (Fig. 3j) or melt H₂O concentration (Fig. S3c) without the Wang et al. (2019) data, which
440 show a subparallel, but offset trend. Using the Wang et al. (2019) calibration, ΔQFM (Calc-Exp)
441 shows negative correlation with temperature (Fig. 3e; $R^2=0.35$) and no correlation with pressure
442 (Fig. S3d; $R^2=0.10$). ΔQFM (Calc-Exp) and olivine forsterite content (Fig. S3e) or anhydrous
443 NBO/T (Fig. 3h; $R^2=0.08$) are uncorrelated, whereas ΔQFM (Calc-Exp) and hydrous NBO/T or
444 melt H₂O concentration are positively correlated (Fig. 3k and S3f). As for the Mallmann & O'Neill
445 (2013) calibration, the data of Wang et al. (2019) deviate from the hydrous NBO/T and melt H₂O
446 concentration correlation trends, but less significantly. Using the Shishkina et al. (2018) calibration,
447 ΔQFM (Calc-Exp) shows no correlation with temperature (Fig. 3f; $R^2=0.03$), but a weak positive
448 correlation with pressure (Fig. S3g; $R^2=0.47$). ΔQFM (Calc-Exp) and olivine forsterite content
449 (Fig. S3h) or anhydrous NBO/T (Fig. 3i) are uncorrelated, whereas ΔQFM (Calc-Exp) and hydrous
450 NBO/T (Fig. 3i; $R^2=0.58$) or melt H₂O concentration (Fig. S3i; $R^2=0.49$) are positively correlated.
451 As for the other two calibrations, the Wang et al. (2019) data slightly deviate from the hydrous
452 NBO/T and melt H₂O concentration correlation trends (Fig. 3l). That the Wang et al. (2019) data
453 tend to fall off the NBO/T and H₂O concentration correlation trends likely reflects (1) that their

454 melt H₂O concentrations are slightly overestimated (causing the limited offset visible in Fig. 3l),
455 and (2) variable performance of the three calibrations for their relatively Na- and K-rich melt
456 compositions (e.g. Figs 3j and k versus l).

457 The bias in ΔQFM (Calc-Exp) as a function of the various tested intensive parameters
458 (typically ≤ 0.5 -1.0 log units) may appear minor, but is important as it is on the order of the
459 characteristic difference in $f\text{O}_2$ inferred for most arc ($\sim\text{QFM}+0.5$ to $+3.5$) compared to mid-ocean-
460 ridge magmas ($\sim\text{QFM}\pm 0.5$). The bias may moreover result in misinterpretation of the redox
461 evolution of individual magma systems, such that false oxidation or reduction trends are calculated.
462 Some of the observed correlations between $f\text{O}_2$ bias and various intensive parameters may be
463 spurious (e.g. weak correlation with pressure may reflect co-correlation with hydrous NBO/T).
464 However, temperature, pressure, and anhydrous as well as hydrous melt compositions are all known
465 to affect mineral-melt element partitioning to some degree (Blundy & Wood 2003), and most of
466 the parameters – i.e., temperature, olivine composition, and anhydrous NBO/T or anhydrous SiO₂-
467 Al₂O₃-Na₂O-K₂O melt compositions – have been inferred to have a significant effect on olivine-
468 melt V partitioning. An effect of melt H₂O concentration or hydrous NBO/T on olivine-melt V
469 partitioning and thus calculated $f\text{O}_2$ values (Fig. 3j-l), which appears to be relevant for all three
470 considered calibrations, in contrast, was ruled out by Shishkina et al. (2018) and Wang et al. (2019).

471 Like Shishkina et al. (2018) and Wang et al. (2019), we see only a weak correlation between
472 ΔQFM (Calc-Exp) and melt H₂O concentration or hydrous NBO/T if we consider small datasets
473 (e.g. only the data of Shishkina et al. (2018) or our new data). However, if we consider ΔQFM
474 (Calc-Exp) and hydrous NBO/T or melt H₂O content for the to date largest possible high-quality
475 dataset, as we have done above, or if we consider subsets of our data – i.e. compositions from
476 experiments that were performed in the same IHPV run (at the same temperature and different melt

477 H₂O concentrations; e.g., our experiments 3-1 and 3-2, 4-1 and 4-2 etc.) – then we can see that
478 Δ QFM (Calc-Exp) generally increases with increasing melt H₂O concentration and hydrous NBO/T
479 (Fig. 3j-l, S3c,f,i; Table S4) Our conclusion is new for olivine-melt V partitioning, but in agreement
480 with other studies who have experimentally shown that trace-element partitioning into silicate
481 minerals is lower in equilibrium with hydrous than in equilibrium with anhydrous melt, which has
482 been linked to the lower degree of polymerization of the hydrous relative to the anhydrous melts
483 (Green et al. 2000; Gaetani et al. 2003; McDade et al. 2003).

484 Spinel-melt and clinopyroxene-melt V oxybarometers for arc magmas have recently been
485 calibrated by Wang et al. (2019), which may provide an independent test relative to olivine
486 oxybarometers and enlarge the application range, but all calibration data are from experiments
487 equilibrated at >1150 °C, and they should thus be used with caution for arc magmas, as most
488 clinopyroxene will have crystallized and most spinel will have equilibrated at lower temperature
489 (e.g., Heath et al. 1998; Pichavant & Macdonald 2007). As for olivine above, we have used the
490 selected hydrous to anhydrous test data, but our new data for clinopyroxene (n=2) and spinel (n=3)
491 are too limited to conclusively assess oxybarometer accuracy or bias related to various intensive
492 parameters. Our clinopyroxene-melt compositions and the selected subset of data from Wang et al.
493 (2019) and data from Laubier et al. (2014), calculate redox state to mostly within $\sim\pm 1.0$ log units
494 of the experimentally constrained conditions (Fig. S4a-c; Table S6). The estimates are thus
495 moderately accurate for the test dataset, but there moreover appears to be bias, overestimating fO_2
496 for Al-rich clinopyroxene and underestimating fO_2 for Al-poor clinopyroxene compositions (Fig.
497 S4b), indicating that the oxybarometer calibration overestimates the effect of clinopyroxene Al
498 concentration, as was previously suspected by Holycross & Cottrell (2022). Our clinopyroxene-
499 melt compositions are interpreted to have been in partial disequilibrium, but their V concentrations

500 seem to have approached equilibrium as much as the other test data, showing similar ΔQFM (Calc-
501 Exp). Our spinel-melt compositions and the selected subset of data from Wang et al. (2019)
502 calculate redox state to mostly within $\sim\pm 0.8$ log units of the experimentally constrained conditions
503 (Fig. S4d-f; Table S5). The estimates are thus reasonably accurate for the test dataset, but there also
504 appears to be bias, overestimating $f\text{O}_2$ for spinel crystallized at oxidizing conditions and
505 underestimating $f\text{O}_2$ for spinel crystallized at reducing conditions, likely relating to spinel
506 compositional variation in addition to Cr# (Fig. S4e; Table S4). For the small test datasets, which
507 are dominated by data employed in the calibrations, the ΔQFM (Calc-Exp) values show no clear
508 correlation with anhydrous or hydrous NBO/T (Fig. S4c,f; Tables S5 and S6). However, for
509 clinopyroxene, a subset of the data show positive correlation between ΔQFM (Calc-Exp) and
510 hydrous NBO/T, which we would suspect from our findings for olivine and from previous studies
511 (Green et al. 2000; Gaetani et al. 2003; McDade et al. 2003).

512

513 *New olivine V oxybarometer calibrations*

514 Following, we now calibrate a new set of olivine-melt V oxybarometers with the aim to eliminate
515 the ΔQFM (Calc-Exp) bias of the evaluated olivine-melt V oxybarometers, using the same dataset
516 as for our tests above (Table S3). Step-wise linear regression for all intensive parameters considered
517 (i.e., temperature, pressure, olivine forsterite content, and anhydrous as well as hydrous NBO/T,
518 and melt H_2O concentration) confirmed a significant effect of hydrous NBO/T on V partitioning
519 and thus on ΔQFM (Calc-Exp). Effects from intensive parameters other than $f\text{O}_2$ and hydrous
520 NBO/T are not ruled out, but they are insignificant (with regression p-values > 0.05) for the

521 considered calibration range, including temperature, or directly correlated with hydrous NBO/T
522 (e.g. melt H₂O concentration).

523 In a first step, we use simple linear regression to define an oxybarometer as a function of olivine-
524 melt V partitioning (Table 2):

$$525 fO_2 (\Delta QFM) = -3.2848 - 2.9893 \times \text{Log}(D_V) \quad (R^2=0.94, 2SD=0.4; \text{Eqn 1}).$$

526 In a second step, we use multiple linear regression to yield a more complex oxybarometer as a
527 function of olivine-melt V partitioning and hydrous melt NBO/T (Table 2):

528

$$529 fO_2 (\Delta QFM) = -2.8438 - 3.1960 \times \text{Log}(D_V) - 0.8781 \times \text{h-NBO/T} \quad (R^2=0.97, 2SD=0.3; \text{Eqn 2}),$$

530 where $fO_2 (\Delta QFM)$ is the oxygen fugacity relative to the quartz-fayalite-magnetite buffer of O'Neill
531 (1987) in log units, D_V is the olivine-melt V partitioning coefficient and h-NBO/T is the hydrous
532 NBO/T value. The equation 1 oxybarometer shows a relatively high degree of correlation ($R^2=0.94$)
533 and relatively low uncertainty ($2SD=0.4$ log units), but ΔQFM (Calc-Exp) bias as a function of
534 hydrous NBO/T (Table S4). The oxybarometer closely compares to the oxybarometer of Shishkina
535 et al. (2018), which can be used for estimating fO_2 in cases where melt composition including melt
536 H₂O concentration is weakly constrained. More importantly, we suggest that the equation 1
537 oxybarometer is used for estimating Fe₂O₃/FeO concentrations needed to calculate melt NBO/T
538 values (instead of assuming Fe₂O₃/FeO as is commonly done). The equation 2 calibration removes
539 the ΔQFM (Calc-Exp) bias relative to hydrous NBO/T, but also relative to all other considered
540 intensive parameters (e.g., removing spurious correlation for pressure or individual melt
541 components), showing matching correlation trends for all individual test datasets. The equation

542 shows a high degree of correlation ($R^2=0.97$) and relatively low uncertainty ($2SD=0.3$ log units)
543 (Fig. 4). Errors on estimates for unknowns can be calculated as:

544 σ (Eqn 1) = $(0.0279 + 0.0116 \times (\text{Log}(D_V))^2 + 8.9359 \times (\sigma(\text{Log}(D_V)))^2)^{0.5}$ and

545 σ (Eqn 2) = $(0.0159 + 0.0060 \times (\text{Log}(D_V))^2 + 10.2144 \times (\sigma(\text{Log}(D_V)))^2 + 0.0127 \times (\text{h-NBO}/T)^2 +$
546 $0.7711 \times \sigma(\text{h-NBO}/T)^2)^{0.5}$, accounting for regression errors as provided in Table 2 and requiring
547 users to estimate σ for their $\text{Log}(D_V)$ and hydrous melt NBO/T input values.

548 We stress (1) that the hydrous calibration dataset on which our olivine-melt oxybarometers
549 are built remains limited, which should be supplemented by more high-quality experimental data
550 in the future, especially for temperatures >1200 °C and hydrous NBO/T >1.2 ; and (2) that the
551 oxybarometers should only be employed to phase compositions and intensive system parameters
552 that closely match those of the calibration range, as provided in Table S3. We further suggest that
553 our and other available mineral-melt V oxybarometers are best applied in concert and that the
554 results are then interpreted in detail, as we do below in our case study of the natural St. Vincent
555 high-Mg basalts.

556

557 **The St. Vincent case study**

558 Primitive high-Mg basalts erupted at Saint Vincent are primary and near-primary mantle-derived
559 melts (Heath et al. 1998; Macdonald et al. 2000; Pichavant & Macdonald 2007; Melekhova et al.
560 2015). High-Mg basalt sample STV301 comprises olivine (~16 vol%), spinel (~2 vol%), and
561 clinopyroxene (~9 vol%) phenocrysts with rare inherited xenocrystic or antecrystic cores,
562 embedded in a crystallized, plagioclase-, clinopyroxene, olivine, and spinel-bearing matrix (Fig.
563 S5). No glassy melt inclusions or interstitial glass are present, and potential crystallized melt

564 inclusions are rare (Fig. S5). Small potential crystallized melt inclusions (Fig. S5b) may be true
565 inclusions, whereas larger inclusions may be apparent, which were connected to
566 matrix/groundmass in 3D (Fig. S5a,d,e). The sample was previously petrologically studied by
567 Heath et al. (1998), Pichavant et al. (2002), and Pichavant & Macdonald (2007), who presented
568 major- and minor-element, but no trace-element compositional data.

569

570 *Phenocryst textures and compositions*

571 Olivine phenocrysts of high-Mg basalt STV301 are subhedral to euhedral and up to ~2 mm in size.
572 They have locally spinel and potential crystallized melt inclusions and they rarely contain inherited
573 cores (~1 vol%), which are subround and devoid of inclusions (Fig. S5a,b, S6a-f). The olivine
574 phenocrysts show core-rim zoning with weakly zoned core plateau zones and significantly zoned
575 rims, or more complex zoning where inherited cores are present. The phenocryst cores have high
576 Fo (~84-90), low to intermediate CaO (~0.14-0.23 wt%) and MnO (~0.10-0.32 wt%), low V (~3.8-
577 6.3 ppm), intermediate to high Sc (~5.7-8.4 ppm), low Y (≤ 0.12 ppm), and intermediate to high Ni
578 (~1350-2600 ppm) (Fig. 5; Table S7). Their compositions match those of the experimental crystals,
579 except that the experimental crystals have overall higher CaO concentrations. The phenocryst rims
580 have lower Fo (~70-77) and Ni (~680-1060 ppm), yet higher MnO (~0.41-0.52 wt%), CaO (~0.26-
581 0.35 wt%), V (~8.0-16.1 ppm) and Y (≥ 0.15 ppm), and overall higher Sc (~7.1-8.6 ppm) than the
582 phenocryst cores and the experimental crystals (Fig. 5; Table S7). The inherited olivine cores have
583 compositions that are distinct from those of the phenocrysts and the experimental crystals with
584 moderately high Fo (~80-86), low CaO (≤ 0.07 wt%), low to intermediate MnO (0.08-0.32), and
585 low V (~2.1 ppm), Sc (~2.8), Y (~0.06 ppm), and Ni (~1000 ppm) (Fig. 5; Table S7).

586 Spinel in sample STV301 forms euhedral to skeletal phenocrysts up to ~200 μm in size and
587 small (<10 μm), or subhedral to euhedral inclusions in olivine and clinopyroxene that commonly
588 occur together with crystallized melt (Fig. S5b,d). The spinel phenocrysts and spinel inclusions
589 show gradational core-rim zoning in BSE imaging mode. The spinel inclusions have high to
590 intermediate Mg# (~45-58%), low to intermediate $\text{Fe}^{3+}/\text{R}^{3+}$ (~7-13%) and low to high $\text{Fe}^{3+}/\text{Fe}(\text{tot})$
591 (~20-36%) (Fig. 6a,b; Table S8). High-Mg# inclusions that were large enough for LA-ICP-MS
592 analysis have low V (~880 ppm), Sc (~5.7 ppm), and Y (≤ 0.04 ppm), and intermediate to high Ni
593 (~570-1150 ppm) (Fig. 6c-f; Table S8). The cores of spinel phenocrysts have overall lower Mg#
594 (~28-44%) and lower $\text{Fe}^{3+}/\text{R}^{3+}$ and $\text{Fe}^{3+}/\text{Fe}(\text{tot})$ than the spinel inclusions at equivalent Mg# (Fig.
595 6a,b; Table S8). Compared to the analyzed spinel inclusions, the phenocryst cores have equivalent
596 to higher V (~870-1520 ppm), Sc (~4.7-11.6 ppm), and Y (~0.03-0.09 ppm), yet lower Ni (~390-
597 590 ppm) (Fig. 6c-f; Table S8). Both high-Mg# spinel inclusions and phenocryst cores have
598 compositions that match those of the low-Mg# experimental crystals. Spinel phenocryst rims have
599 low Mg# (~21-34%) and high $\text{Fe}^{3+}/\text{Fe}(\text{tot})$ (~42-49%), $\text{Fe}^{3+}/\text{R}^{3+}$ (~26-49%), V (>2000 ppm) and
600 Sc (~12 ppm) relative to the phenocryst cores with concentrations that significantly also differ from
601 the experimental spinel compositions (Fig. 6; Table S8).

602 Clinopyroxene phenocrysts are subhedral to euhedral and typically <500 μm in maximum
603 dimension (Fig. S5c,d). Like olivine, the clinopyroxene phenocrysts locally contain inherited cores
604 (~2 vol%), which are anhedral, comprising coarsely crystallized melt inclusions (Fig. S5d). The
605 phenocrysts show sharp sector zoning with superimposed oscillatory zoning or more complex
606 zoning where inherited cores are present (Fig. S6j-e). The sector-zoned phenocrysts have
607 intermediate to high Mg# (~80-86%), low to high TiO_2 and Al_2O_3 (~0.5-1.4 and ~2.9-7.4 wt%) and
608 high CaO (~21.8-23.6 wt%) (Fig. 7a-c; Table S9). They also have low to high V (~210-400 ppm),

609 high Ni (~110-130 ppm) and Sc (~200-250 ppm), and low Y (~9-14 ppm) (Fig. 7d-f; Table S9).
610 Sector zoning is most prominent in Al₂O₃ and V (Fig. 7b, d-f). The phenocrysts have notably higher
611 Mg#, CaO and Sc, comparable V, Y, and Ni and lower Al₂O₃ than the experimental crystals,
612 showing no compositional match. The inherited clinopyroxene cores show complex patchy zoning
613 (Fig. S6j-l) with low Mg# (~73-76%), high TiO₂ (~0.9-1.3 wt%), low Al₂O₃ (~2.9-3.6 wt%) and
614 CaO (~20.2-21.1 wt%), and intermediate V (~360 ppm) concentrations (Fig. 7; Table S9). Their
615 Ni (~30-90 ppm) and Sc (~140-200 ppm) concentrations are low to intermediate, whereas their Y
616 concentration (~16-30 ppm) is intermediate to high. Their compositions are different from the
617 compositions of both the phenocrysts and the experimental crystals.

618

619 *Melt compositions and mineral-melt partitioning coefficients*

620 High-Mg basalt STV301 comprises no glass inclusions or glassy matrix, and only a limited number
621 of large inclusions that may represent crystallized melt (Fig. S5), which could have been connected
622 to the matrix/groundmass in three dimensions, and we have therefore estimated melt compositions
623 (Fig. 8; Table 3). We highlight that it is common for primitive arc magmas such as STV301 that
624 melt inclusions are absent from phenocrysts or that the melt inclusions are in disequilibrium with
625 their phenocryst hosts (e.g., Bouvier et al. 2008, 2022). In our case study, we show that intensive
626 parameters using mineral-melt compositions can nevertheless be estimated for these rocks, if
627 equilibrium melt compositions are rigorously constrained. Following, we estimate melt
628 compositions for early-stage magma storage (estimates E1 to E3), where primitive phenocryst cores
629 formed (olivine+spinel), and late-stage magma storage (estimates E4 to E6), where evolved
630 phenocryst cores formed (olivine+spinel+clinopyroxene). Following Pichavant & Macdonald
631 (2007), we infer early- to late-stage magma storage at depth at ~400 MPa, at volatile saturated

632 conditions, and in equilibrium with a hydrous melt with ~5 wt% H₂O. Strongly zoned phenocryst
633 rims and matrix crystals – which have equivalent size and compositions (e.g. <Fo80 olivine) – are
634 inferred to have crystallized or re-equilibrated during magma ascent to the surface and subsequent
635 to eruption, for which equilibrium melt compositions could not be estimated.

636 During early-stage storage crystallization, olivine, spinel and melt ±fluid were the stable
637 phases of the STV301 magma (Pichavant and Macdonald 2007). Following Heath et al. (1998) and
638 Pichavant & Macdonald (2007), we suggest that the most primitive olivine cores (~Fo89, ~4.0-5.6
639 ppm V, ~5.7-7.4 ppm Sc, ~0.06-0.08 ppm Y, and ~1780-2590 ppm Ni; Fig. 5; Table S7)
640 crystallized at this stage. Spinel inclusions and phenocrysts were present, but have at least partly
641 equilibrated at lower temperature (in Mg-Fe but possibly also in V), with calculated olivine-spinel
642 equilibration temperatures of ≤1130 °C (cf. Heath et al. 1998). Following common practice (e.g.,
643 Putirka 2008), we estimate that the early-stage storage melt composition has approached bulk-rock
644 composition (E1), but that it was slightly more evolved owing to minor crystallization of olivine
645 and spinel and the presence of inherited crystal cores in our sample STV301, in agreement with
646 Pichavant & Macdonald (2007), who experimentally showed that the primary STV301 melt was
647 slightly less primitive (lower in MgO) than the STV301 whole rock. Our best estimate for early-
648 stage storage melt composition are the experimental STV301 glass composition 7-2 in equilibrium
649 with ~3 vol% olivine and ~1 wt% spinel (estimate E2) and a melt composition calculated
650 considering crystallization of olivine and spinel and the presence of inherited crystal cores
651 (combined ~6 wt% olivine, ~1 wt% spinel, and ~2 wt% clinopyroxene) (estimate E3) (Tables 3
652 and S10). The estimated E2 and E3 melt compositions with ~284-291 ppm V, ~42-44 ppm Sc,
653 ~14.1-14.8 ppm Y, and ~180-230 ppm Ni at ~9.8-10.2 wt% MgO (Table 3) closely match, except
654 for their Ni concentrations (Fig. 8). Their compositions fall onto primitive Saint Vincent whole-

655 rock compositional trends (Fig. 9; Table 3) and they moreover match the MgO concentrations of
656 primitive olivine-hosted melt inclusions of other high-Mg basalts from Saint Vincent (reported by
657 Bouvier et al. 2008, 2010), but they have overall lower V and Y contents. The difference in
658 estimated STV301 melt and measured melt inclusion V and Y compositions likely reflects variable
659 magma/melt compositions and fO_2 for the studied STV301 magma from the Pre-somma unit and
660 for younger high-Mg magmas (likely from the Yellow Tuff formation) sampled by Bouvier et al.
661 (2008, 2010). Alternatively, the difference in V and Y compositions could record formation of V-
662 and Y-rich melt boundary layers around olivine that were enclosed as melt inclusions, in line with
663 olivine-melt inclusion isotopic disequilibrium detected for a large proportion of the inclusions
664 (Bouvier et al. 2022). In combination, our best estimates for the early-stage storage melt
665 composition (estimates E2 and E3) and the primitive olivine phenocryst core compositions (Table
666 3) calculate Fe(tot)-Mg partitioning coefficients of ~ 0.23 to 0.24 , which are in the range of
667 equilibrium values for moderately oxidizing crystallization conditions (cf. Blundy et al. 2020), thus
668 suggesting that the estimated melt compositions E2 and E3 and primitive olivine phenocryst cores
669 indeed represent or approach equilibrium pairs. The olivine and melt (E2 and E3) compositions
670 further calculate a limited range of V partition coefficients of ~ 0.014 - 0.020 and Sc/Y and Ni
671 partition coefficients of ~ 24 - 46 and ~ 7.8 - 15 (Table 3), which translate into a small variation in
672 calculated fO_2 (≤ 0.15 log units difference) and a limited to moderate variation in calculated
673 temperatures (< 30 °C for Sc/Y versus < 75 °C for Ni) (Table 4).

674 During late-stage storage conditions (prior to magma ascent towards eruption), olivine,
675 clinopyroxene, spinel, and melt \pm fluid were the stable phases of the STV301 magma (Pichavant
676 and Macdonald 2007). Following again the interpretation of Heath et al. (1998) and Pichavant &
677 Macdonald (2007), we suggest that the evolved olivine core ($\sim Fo_{84-86}$, ~ 4.2 - 5.1 ppm V, ~ 5.8 - 8.0

678 ppm Sc, ~0.04-0.10 ppm Y, and ~1350-2000 ppm Ni), Mg-rich spinel phenocryst core (Mg#~40-
679 50%, ~870-980 ppm V), and clinopyroxene phenocryst core (Mg#>80%, ~210-400 ppm V, ~200-
680 250 ppm Sc) compositions crystallized or equilibrated at this stage (Tables S5-S7). Following
681 common practice (e.g. Putirka 2008), we have attempted to calculate the late-stage storage melt
682 composition, subtracting phenocryst and inherited core compositions from the STV301 bulk rock
683 composition (estimate E4) (Tables 3 and S11). The estimated E4 melt composition falls along
684 primitive Saint Vincent whole-rock compositional trends (Fig. 8), but is too evolved to have been
685 in equilibrium with ~Fo84-86 olivine, calculating an Fe(tot)-Mg partitioning coefficient of ~0.16
686 (Table 3), reflecting the difficulty to accurately constrain the phenocryst and inherited crystal core
687 cargos and their average compositions. Moderately evolved experimental STV301 glass
688 compositions (averages of experimental glasses 3-1 and 3-2) and the matrix/groundmass
689 composition of the natural STV301 sample (estimates E5 and E6) are taken as alternative
690 constraints (Table 3). The average experimental glass composition was taken to account for partial
691 crystallization (~7.5 wt% olivine, ~0.5 wt% spinel, and ~7.0 wt% clinopyroxene). The composition
692 of the matrix/groundmass was analyzed with a broad, 100 μm beam away from phenocrysts,
693 averaging a total of 20 analyses. Compared to our estimates of the early-stage storage melt, the E5
694 and E6 melt estimates for the late-stage storage melt remain uncertain given the difficulty to
695 precisely estimate the proportion of phenocryst cores that have been present at this stage (especially
696 for complexly zoned clinopyroxene, which may have crystallized during late-stage storage or
697 magma ascent). However, both estimated E5 and E6 compositions fall along Saint Vincent whole-
698 rock compositional trends (Fig. 8) and they moreover match the MgO and characteristic Y
699 concentrations of evolved olivine-hosted melt inclusions of other high-Mg basalts from Saint
700 Vincent (Fig. 8). The calculated Fe(tot)-Mg partitioning coefficients for olivine and estimated melt
701 compositions E5 and E6 are ~0.23-0.25 (Table 3), which are in the range of equilibrium values for

702 moderately oxidizing crystallization conditions (cf. Blundy et al. 2020), moreover suggesting that
703 the E5 and E6 melt and evolved olivine core compositions indeed represent or approach equilibrium
704 pairs (Table 3). The compositions further calculate olivine-melt V, Sc/Y, and Ni partition
705 coefficients of ~ 0.013 - 0.018 , ~ 31 - 59 and ~ 15 - 26 , and spinel-melt V partitioning coefficients of
706 ~ 2.8 - 3.4 (Table 3). The clinopyroxene phenocrysts show complex sector and oscillatory zoning,
707 and they were therefore not used for estimating equilibrium partitioning coefficients or
708 crystallization conditions below. As for the early-stage storage conditions, our best estimates for
709 the late-stage storage melt compositions (E5 and E6) translate into a small variation in calculated
710 fO_2 (≤ 0.15 log units difference) and a limited to moderate variation in calculated temperatures (< 70
711 $^{\circ}C$ for Sc/Y versus < 30 $^{\circ}C$ for Ni) (Table 4).

712

713 *Temperature and fO_2 of the primitive St. Vincent magma during early-stage storage*

714 Average temperatures calculated for early-stage storage olivine and inferred equilibrium-melt (E2
715 and E3) compositions are $\sim 1210 \pm 30$ $^{\circ}C$ using the Sc/Y olivine-melt thermometer of Mallmann &
716 O'Neill (2013) and $\sim 1240 \pm 60$ $^{\circ}C$ using the Ni olivine-melt thermometer of Pu et al. (2017) (Table
717 4). The calculated Ni temperature appears high, but the calculated Sc/Y temperature closely
718 approaches the experimentally constrained near-liquidus temperature of ~ 1200 $^{\circ}C$ of Pichavant &
719 Macdonald (2007), which in combination suggest that the three tested olivine-melt V
720 oxybarometers and our new calibration are all applicable to the inferred mineral-melt equilibrium
721 pairs. To estimate the early-stage storage fO_2 , we then use the minimum and maximum V
722 concentrations of Fo-rich olivine, the various estimated melt compositions, and a temperature
723 estimate of ~ 1200 $^{\circ}C$ (Tables 3 and 4). We assume a melt H_2O concentration of ~ 5 wt% as
724 experimentally constrained by Pichavant & Macdonald (2007) and as has been detected in primitive

725 olivine-hosted melt inclusions from other St. Vincent high-Mg basalts (Bouvier et al. 2008), which
726 is consistent with the observation that olivine phenocryst core CaO compositions match or most
727 closely approach those crystallized in H₂O-rich experiments ($\sim \geq 5$ wt% H₂O) (Fig. 5a) (cf.
728 Gavrilenko et al. 2016). For these parameters, the various olivine-melt oxybarometers calculate fO_2
729 values of \sim QFM+1.4 to QFM+2.4 and on average \sim QFM+1.8 to QFM+2.1 \pm 0.5-0.9 log units (Fig.
730 9a; Table 4). The range of estimated melt compositions (E2 and E3) has a limited effect on the
731 inferred fO_2 ($\Delta \leq 0.15$ log units) and Sc/Y temperature ($\Delta \sim 30$ °C) (Table 4), as melt V, Sc, and Y
732 contents show limited variation after crystallization of minor olivine and traces of spinel and can
733 therefore be fairly well constrained (Fig. 8). The inferred Ni temperatures, in contrast, are more
734 variable ($\Delta < 65$ °C), reflecting the difficulty to tightly constrain olivine-melt Ni equilibrium pairs.
735 The range of detected trace-element concentrations in Fo-rich olivine ($\Delta V \sim 1.6$ ppm, $\Delta Sc \sim 1.7$
736 ppm, $\Delta Y \sim 0.02$ ppm; Table S7) are significantly larger than the analytical errors (≤ 0.24 ppm V,
737 ≤ 0.37 ppm Sc, ≤ 0.006 ppm; Table S7), thus likely recording true variation in crystallization
738 conditions, which could reflect fO_2 variation by up to 0.3 log units, temperature variation by up to
739 60 °C, or variation in other intensive parameters between magma parcels and their phenocryst
740 cargo.

741 The reason that all applied V oxybarometers calculate closely concurring fO_2 values is
742 consistent with equilibration at relatively high temperature (~ 1200 °C), but we posit that all
743 previous calibrations and our equation 1 oxybarometer nevertheless tend to systematically
744 overestimate fO_2 for the hydrous magma T(cf. Fig. 3). The estimated average fO_2 values of
745 \sim QFM+1.8 to QFM+2.1 \pm 0.5-0.9 and our best estimate of \sim QFM+1.8 \pm 0.5 (calculated using our
746 equation 2 calibration) are overall lower than but within the uncertainty of the previous fO_2 estimate
747 of \sim QFM+2.1 to +2.6 of Pichavant & Macdonald (2007) (Fig. 9a). We suggest that our new fO_2

748 estimate is more accurate for early-stage crystallization conditions than the previous estimate,
749 which was based on the composition of small, high-Mg#, high $\text{Fe}^{3+}/\text{Fe}(\text{tot})$ spinel inclusions, which
750 have partially equilibrated in Mg-Fe with their olivine hosts, consistent with olivine-spinel
751 equilibration temperatures of ≤ 1130 °C (Heath et al. 1998). Our redox estimate of $\sim\text{QFM}+1.8\pm 0.5$
752 for the early-storage, near-liquidus, mantle-derived STV301 melt is within the range of values
753 characteristic for primitive Lesser Antilles magmas, which vary between $\sim\text{QFM}+0.5$ to $\text{QFM}+4.3$
754 and are mostly $\sim\text{QFM}+1.0$ to $\text{QFM}+3.0$ (Heath et al. 1998; Pichavant & Macdonald 2007; Stamper
755 et al. 2014). Our estimate is also within the range of values considered characteristic for arc settings
756 constrained using various mineral-melt oxybarometers, phase-equilibrium experiments and recent
757 whole-rock geochemical modelling ($\sim\text{QFM}+0.5$ to $\text{QFM}+3.5$, e.g., Gaillard et al. 2015; Bucholz &
758 Kelemen 2019). The inferred early-stage storage, near-liquidus $f\text{O}_2$ is, however, significantly
759 higher than $f\text{O}_2$ values ($\sim\text{QFM}$) inferred for primitive mantle-derived arc magmas using the bulk-
760 rock V/Sc or Ga/Sc models of Lee et al. (2005) or Mallmann & O'Neill (2009).

761

762 *Temperature and $f\text{O}_2$ of the primitive St. Vincent magma during late-stage storage*

763 Average temperatures calculated for the late-stage storage olivine-melt (E4 and E5) compositional
764 pairs are $\sim 1080\pm 30$ °C using the Sc/Y olivine-melt thermometer of Mallmann & O'Neill (2013)
765 and $\sim 1130\pm 60$ °C using the Ni olivine-melt thermometer of Pu et al. (2017) (Table 4). The
766 calculated Sc/Y temperature agrees with the experimentally constrained late-stage storage
767 temperatures of ~ 1050 - 1100 °C of Pichavant & Macdonald (2007), whereas the Ni thermometer
768 appears to overestimate temperature as for the early-stage storage conditions. To estimate the late-
769 stage storage $f\text{O}_2$, we therefore take a temperature of ~ 1075 °C, which suggests that the mineral-
770 melt pairs equilibrated outside the calibration range of the Mallmann & O'Neill (2013) and Wang

771 et al. (2019) V oxybarometers. To estimate the late-stage storage fO_2 , we nevertheless employ all
772 tested and our new olivine-melt V oxybarometers and the spinel-melt V oxybarometer of Wang et
773 al. (2019) to assess and discuss the different results. We use the minimum and maximum V
774 concentrations of the evolved olivine phenocryst core compositions ($\sim Fo_{84-86}$), Mg-rich spinel
775 phenocryst core compositions ($\sim Mg\#40-60$) and estimated (E4 and E5) melt compositions (Table
776 3). As for the early-stage storage conditions, we assume a melt H_2O concentration of ~ 5 wt% (cf.
777 Pichavant & Macdonald 2007; Bouvier et al. 2008), given that melt inclusions from St. Vincent
778 and Lesser Antilles magmas in general record volatile (H_2O-CO_2) saturation at ≥ 5 wt% (cf.
779 Balcone-Boissard et al. (2023)) and given that the STV301 olivine/melt Ca contents show no
780 significant increase with decreasing olivine Fo content (Fig. 5a), which would be expected in the
781 case of significant variation of melt H_2O concentration (cf. Gavrilenko et al. 2016). For these
782 parameters, the various olivine-melt and the spinel-melt oxybarometers calculate fO_2 values of
783 $\sim QFM+1.9$ to $QFM+3.1$ and on average $\sim QFM+2.1$ to $QFM+2.9 \pm 0.5-0.9$ log units, which are
784 much more variable (differing by ~ 0.8 log units) than the fO_2 values calculated for the early-stage
785 storage condition (differing by ~ 0.3 log units) (Fig. 9a versus 9b; Table 4). Our equation 1 and 2
786 olivine-melt V oxybarometers calculate the lowest fO_2 values of $\sim QFM+2.1 \pm 0.6$ on average,
787 whereas the Wang et al. (2019) spinel-melt and the Mallmann & O'Neill (2013) and Wang et al.
788 (2019) olivine-melt oxybarometers calculate higher fO_2 values of $\sim QFM+2.6 \pm 0.5$, $\sim QFM+2.7 \pm 0.5$
789 and $\sim QFM+2.9 \pm 0.5$ on average (Fig. 9b, Table 4). The range of estimated late-stage storage melt
790 (E4 and E5) compositions has a slightly higher, but still limited effect on the calculated fO_2 ($\Delta < 0.2$
791 log units) compared to the early-stage storage melt. The reason for this is that the crystallization of
792 several weight percent of clinopyroxene in addition to olivine significantly fractionates melt V and
793 that equilibrium mineral-melt compositions become more difficult to constrain. The range of trace-

794 element concentrations in Mg#-rich spinel and in Fo-rich olivine phenocryst cores ($\Delta V \sim 100$ ppm
795 for spinel; $\Delta V \sim 0.9$ ppm, $\Delta Sc \sim 1.7$ ppm, $\Delta Y \sim 0.06$ ppm for olivine; Table S4) are significantly
796 larger than the analytical errors (≤ 30 ppm V for spinel; ≤ 0.18 ppm V, ≤ 0.29 ppm Sc, and ≤ 0.01
797 ppm Y for olivine) and for olivine on the same order as for the primitive phenocryst cores,
798 suggesting that intensive parameters varied to the same degree (e.g. fO_2 by ~ 0.3 log units or
799 temperature by ~ 60 °C), which could indicate that smaller magma parcels and their phenocryst
800 cargo were blended just prior to eruption.

801 We take an average fO_2 of $\sim QFM+2.1 \pm 0.6$, calculated using our olivine-melt V
802 oxybarometers, as the best estimate for the late-stage storage fO_2 (Fig. 9b; Table 4). That our and
803 the Shishkina et al. (2018) oxybarometers calculate closely comparable fO_2 for low-temperature
804 equilibration ($\sim QFM+2.1$ and $\sim QFM+2.2$) and relatively less oxidized fO_2 values than the
805 Mallmann & O'Neill (2013) and Wang et al. (2019) calibrations was expected (cf. Fig. 3), which
806 highlights that it is critically important to have tight temperature constraints for the crystallization
807 of the redox-sensitive assemblage, and that the Mallmann & O'Neill (2013) and Wang et al. (2019)
808 oxybarometers should not be applied outside their calibration range at below ~ 1200 - 1150 °C. That
809 our oxybarometers and the Shishkina (2018) (equation 4) calibration calculate closely matching
810 values reflects the relatively polymerized character of the estimated equilibration melt (h-NBO of
811 ~ 0.84 - 0.97 ; Table 3) with intermediate melt H_2O concentration (~ 5 wt%) yet otherwise evolved
812 composition (cf. Figs 3l, S4f). Our best estimate of $\sim QFM+2.1 \pm 0.6$ falls between the previous
813 estimates of Pichavant & Macdonald (2007) ($\sim QFM+2.1$ to $QFM+2.6$ based on their phase
814 equilibrium experiments and spinel compositions) and Heath et al. (1998) ($\sim QFM+1.5$ to $QFM+1.8$
815 using the Ballhaus et al. (1991) olivine-spinel oxybarometer) (Fig. 10b), which is slightly higher
816 than our fO_2 estimate for early-stage storage ($\sim QFM+1.8$) by ~ 0.3 log units, which may indicate a

817 limited fO_2 increase during intermittent crustal storage. If the results from the Mallmann & O'Neill
818 (2013) or Wang et al. (2019) oxybarometers would have been considered, then more significant
819 oxidation by ~ 0.8 log units may have been erroneously inferred. Results from our equation 1 or the
820 Shishkina et al. (2018) oxybarometers, which do not account for melt compositional variation other
821 than in V, in contrast would have suggested no essential oxidation (i.e. by ≤ 0.1 log units).

822

823 **Limited oxidation during intermittent crustal storage of the STV301 magma**

824 Our best fO_2 estimates for early-stage to late-stage storage suggest limited oxidation by ~ 0.3 log
825 units during intermittent crustal magma storage, but such limited redox variation is within the
826 uncertainty of our method (± 0.3 log units) and smaller than the estimated individual uncertainties
827 (± 0.5 - 0.6 log units) (Table 4). Considering the extremes of individual uncertainties, V
828 oxybarometry alone cannot rule out limited reduction (by ~ 0.8 log units, i.e. from QFM+1.8+0.5
829 to QFM+2.1-0.6) or more significant oxidation (by ~ 1.4 log units, i.e. from QFM+1.8-0.5 to
830 QFM+2.1+0.6) for the STV301 magma during intermittent crustal storage. However, limited
831 evidence for contamination (i.e. presence of minor amounts of inherited crystal cores) and lack of
832 evidence for degassing during magma storage in combination with crystallization of moderate
833 amounts of olivine (~ 7.5 - 10.5 wt%), clinopyroxene (~ 6.0 - 8.0 wt%), and spinel (~ 0.5 - 2.5 wt%), are
834 in agreement with limited oxidation. Redox variation of primitive magmas largely owing to
835 fractional crystallization of moderate amounts of olivine, clinopyroxene, and spinel (< 25 wt%) is
836 typically inferred to be ≤ 0.5 log units as estimated for natural samples (Cottrell & Kelley 2011;
837 Kelley & Cottrell 2012), but may be higher, e.g. ~ 0.8 log units as shown in fractionation
838 experiments (Ulmer et al. 2018). For single samples like the studied STV301 basalt, such limited
839 redox variation cannot be reliably resolved by V olivine-melt or other mineral-melt oxybarometry.

840 However, V mineral-melt oxybarometry for a large number of samples from a single magma
841 system, ideally with well-constrained melt compositions and large degrees of fractionation, have
842 the capacity to establish if primitive arc magmas typically experience oxidation by ~ 0.5 , ~ 1.0 , or
843 more log units. To track autooxidation for large degrees of fractional crystallization ($\gg 25$ wt%),
844 which has been suggested to be as high as ≥ 4 log units based on fractionation experiments (Ulmer
845 et al. 2018), V oxybarometers will have to be rigorously calibrated for lower-temperature equilibria
846 for clinopyroxene-melt, spinel-melt, and/or amphibole-melt.

847

848 CONCLUSIONS

849 Olivine-melt and spinel-melt V oxybarometry have the potential to further improve our
850 understanding of the redox state and redox evolution of Earth's primitive magmas. For robust
851 constraints and to resolve limited fO_2 variation, it is however essential that the effects of
852 temperature and anhydrous to hydrous melt polymerization on fO_2 estimates are considered. If they
853 are neglected, olivine-melt and spinel-melt compositions may variably over- or under-estimated
854 fO_2 by commonly up to ± 1 log unit. This has important implications for constraining fO_2 for
855 magmas generated at different geodynamic settings and for constraining redox evolution of
856 magmas during partial crystallization. For primitive, high-Mg basaltic STV301 magma erupted at
857 St. Vincent, Lesser Antilles, new olivine-melt V oxybarometry indicates oxidizing near-liquidus
858 conditions ($\sim QFM+1.8 \pm 0.5$), consistent with the notion that arc magmas are primarily oxidized.
859 During intermittent crustal storage, limited oxidation (by ~ 0.3 log units) appears to have occurred,
860 in agreement with redox variation largely controlled by < 25 wt% fractionation of olivine-
861 clinopyroxene-spinel, but such limited oxidation cannot be reliably resolved by V or other mineral-
862 melt oxybarometry for individual samples (with uncertainties of $\geq \pm 0.3$ - 0.9 log units), requiring

863 characterization of a large number of samples to establish representative averages or other
864 characterization methods.

865

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877

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- 1042

1043 **FIGURE CAPTIONS**

1044 **Figure 1:** Conditions of experiments for which V partitioning was quantified for olivine-melt,
1045 spinel-melt, or clinopyroxene-melt in this and in previous studies. Dataset used in this study
1046 (summarized in Tables S4 to S6) and datasets used by Shishkina et al. (2018) (S2018) and Wang
1047 et al. (2019) (W2019) for their V oxybarometers for arc magmas for comparison.

1048 **Figure 2:** Mineral-melt partition coefficients determined for the STV301 experimental charges
1049 (Table 1) compared to published experimental data. Redox-sensitive V partition coefficients for (a)
1050 olivine-melt, (b) spinel-melt, and (c) clinopyroxene-melt. Ol=olivine, Cpx=clinopyroxene,
1051 Spl=spinel, L=melt. Literature data are from Canil (1997) (C1997), Toplis & Corgne (2002)
1052 (TC2002), Richter et al. (2006) (R2006), Karner et al. (2008) (K2008), Mallmann & O'Neill (2009,
1053 2013) (MO2009, MO2013), Davis et al. (2013) (D2013), Papike et al. (2013) (P2013), Laubier et
1054 al. (2014) (L2014), Shishkina et al. (2018) (S2018), Wang et al. (2019) (W2019), and Holycross &
1055 Cottrell (2022) (HC2022). Errors for our partition coefficients are indicated as 2 RSDs, which are
1056 commonly smaller than the symbol size. Estimated errors on fO_2 and temperature are ± 0.5 and ± 1.0
1057 log units and ± 5 and ± 25 °C for our internally heated pressure vessel and piston cylinder
1058 experiments, respectively.

1059 **Figure 3:** Calculated ΔQFM versus experimentally-constrained ΔQFM (first row, a-c),
1060 experimental temperature (T) versus ΔQFM (Calc-Exp) (second row, d-f), calculated anhydrous
1061 NBO/T versus ΔQFM (Calc-Exp) (third row, g-i), and calculated hydrous NBO/T versus (ΔQFM
1062 (Calc-Exp) (fourth row, j-l) for the three tested oxybarometers. Calculated data shown were
1063 determined using the oxybarometers of Mallmann & O'Neill (2013) (first column (a,d,g,j), Eqn
1064 M2013), Wang et al. (2019) (second column (b,e,h,k), Eqn W2019), and Shishkina et al. (2018)
1065 (third column (c,f,i,l), Eqn S2018). Test data are from Laubier et al. (2014) (n=22), Shishkina et

1066 al. (2018) (n=12), Wang et al. (2019) (n=11), and our study (n=9) Indicated uncertainties are
1067 estimated experimental and calibration uncertainties. All input and calculated data are provided in
1068 Table S4.

1069 **Figure 4:** Calculated versus experimental fO_2 for our equation 2 olivine-melt V oxybarometer. (a)
1070 Calculated fO_2 values show limited deviation from experimental values. (b-f) Calculated compared
1071 to experimental fO_2 (ΔQFM (Calc-Exp)) values show no systematic bias in relation to
1072 crystallization temperature, pressure, olivine forsterite (Fo) content, or anhydrous or hydrous melt
1073 NBO/T (R^2 is always <0.05). The two V partitioning data from Wang et al. (2019) for olivine-melt
1074 equilibrated at 1350 °C do not significantly deviate from all other (≤ 1200 °C) data. Temperature
1075 thus does not appear to significantly affect V partitioning for the investigated temperature range
1076 between ~1025 and 1350 °C, but a limited effect (as predicted by Mallmann & O'Neill (2013) or
1077 Wang et al. (2019)) needs to be further investigated. Indicated uncertainties are – as for Fig. 3 –
1078 estimated experimental and calibration uncertainties. All data are reported in Table S4.

1079 **Figure 5:** Olivine compositions, where forsterite (Fo) content was calculated as
1080 $Fo = Mg / (Mg + Fe) * 100$. Olivine phenocryst cores have compositions that largely match the
1081 compositions of the experimental STV301 olivine, whereas inherited cores and phenocryst rim
1082 compositions differ. The STV301 olivine compositions are from this study (round symbols) and
1083 the study of Heath et al. (1998) (square symbols). Our STV301 data and their uncertainties are
1084 reported in Tables 1 and S7. Shown for comparison are olivine compositions of low-Mg basalts
1085 from St. Vincent (LMB) (Heath et al. 1998), Grenada basaltic to andesitic lavas (Stamper et al.
1086 2014), St. Vincent and Grenada cumulates (Tollan et al. 2012; Stamper et al. 2014), and peridotite
1087 xenoliths including primary and secondary olivine (Vanucci et al. 2007).

1088 **Figure 6:** Spinel compositions, where spinel Mg# was calculated as $Mg/(Mg+Fe^{2+}) \times 100$, Fe^{3+} was
1089 calculated through charge balance, and R^{3+} was calculated as the sum of Al, Cr, and Fe^{3+} . The Mg#-
1090 rich spinel phenocryst cores and inclusions are close in composition to the experimental STV301
1091 spinel, but the inclusions have overall higher Mg#, $Fe^{3+}/Fe(tot)$, and Fe^{3+}/R^{3+} . The STV301 spinel
1092 compositions are from this study (round symbols) and the study of Heath et al. (1998) (square
1093 symbols). Our STV301 data and their uncertainties are reported in Tables 1 and S8. Shown for
1094 comparison are spinel compositions of low-Mg basalts from St. Vincent (LMB) (Heath et al. 1998),
1095 Grenada basaltic to andesitic lavas (Stamper et al. 2014), St. Vincent and Grenada cumulates
1096 (Tollan et al. 2012; Stamper et al. 2014), and peridotite xenoliths (Vanucci et al. 2007).

1097 **Figure 7:** Clinopyroxene compositions, where Mg# was calculated as $Mg/(Mg+Fe) \times 100$. Sector-
1098 zoned clinopyroxene phenocrysts and inherited cores differ from each other in composition and
1099 they also differ from the composition of the experimental STV301 clinopyroxene. The STV301
1100 clinopyroxene compositions are from this study (round symbols) and the study of Heath et al.
1101 (1998) (square symbols). Our STV301 data and their uncertainties are reported in Tables 1 and S9.
1102 Shown for comparison are clinopyroxene compositions of low-Mg basalts from St. Vincent (LMB)
1103 (Heath et al. 1998), Grenada basaltic to andesitic lavas (Stamper et al. 2014), and St. Vincent and
1104 Grenada cumulates (Tollan et al. 2012; Stamper et al. 2014) and peridotite xenoliths including
1105 primary and secondary clinopyroxene (Vanucci et al. 2007).

1106 **Figure 8:** Estimated early-stage storage (E1 to E3) and late-stage storage (E4 to E6) melt
1107 compositions for the STV301 magma (see Table 3 for full compositions and uncertainties). Our
1108 best estimates are compositions E2 and E3 for early-stage storage and E5 and E6 for late-stage
1109 storage conditions. Shown for comparison are the compositions of olivine-hosted melt inclusion
1110 compositions for other high-Mg basalts from St. Vincent (Bouvier et al. 2010, possibly from a

1111 sample of the Yellow Tuff formation), and St. Vincent whole-rock compositions (extracted from
1112 the GEOROC database). (b and e) are close-ups of (a and d) as indicated.

1113 **Figure 9:** Calculated average fO_2 values for the St. Vincent STV301 magma at early-stage and
1114 late-stage storage conditions for the different melt proxies and in comparison to previous estimates
1115 (see Table 4 for all calculated values). (a) All calculated fO_2 values for early-stage, near-liquidus
1116 condition are relatively close to each other, varying between \sim QFM+1.8 and \sim QFM+2.1 with our
1117 best estimate at \sim QFM+1.8. (b) Calculated fO_2 values for late-stage storage condition vary over a
1118 large range, between \sim QFM+2.1 and \sim QFM+2.9. Indicated uncertainties (\pm 0.5 log units in (a) and
1119 \pm 0.6 log units in (b)) are for our best estimates based on equation 2. All other uncertainties (\pm 0.5-
1120 0.9) are given in Table 4. Previous estimates are from Heath et al. (1998) (H1998) and Pichavant
1121 & Macdonald (2007) (PM2007), which are based on olivine \pm spinel composition. Calibrations of
1122 Mallmann & O'Neill (2013) (MO2013), Shishkina et al. (2018) (S2018), Wang et al. (2019)
1123 (W2019), and our new calibrations (Eqn 1 and 2).

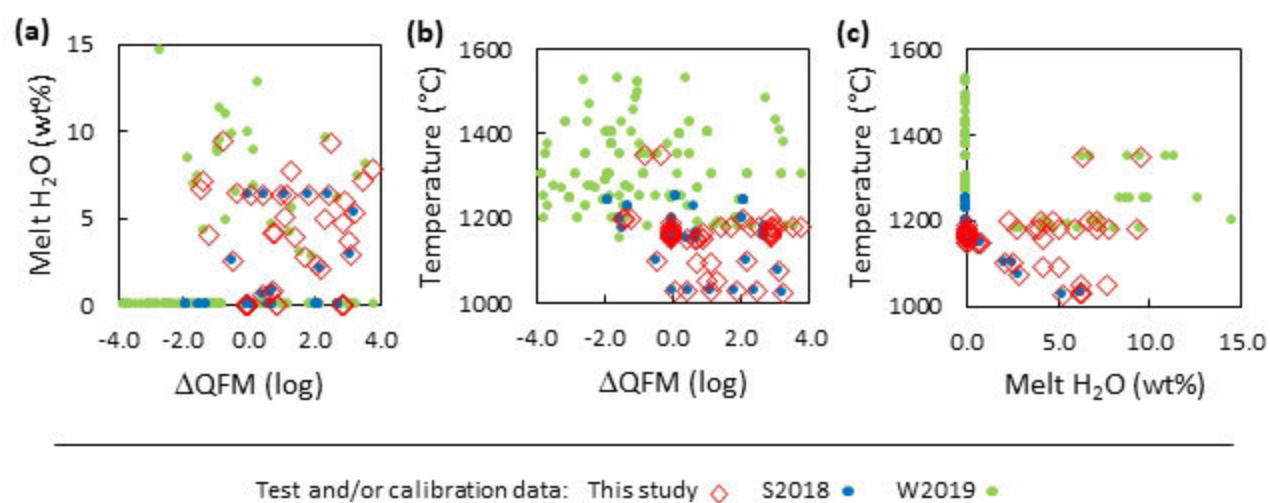


Figure 1 – Erdmann et al.

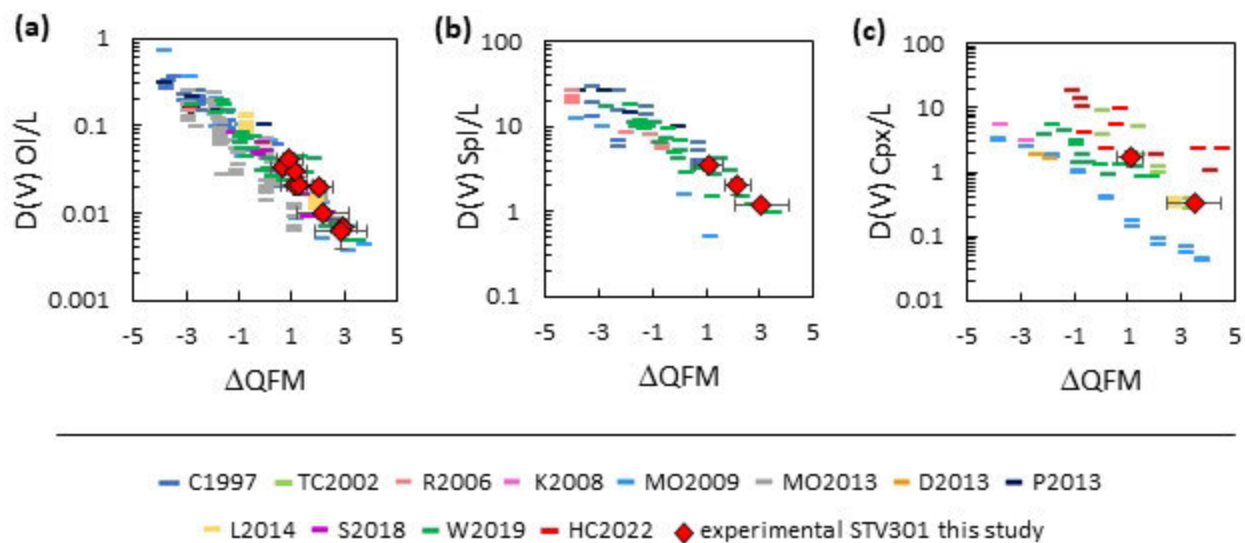
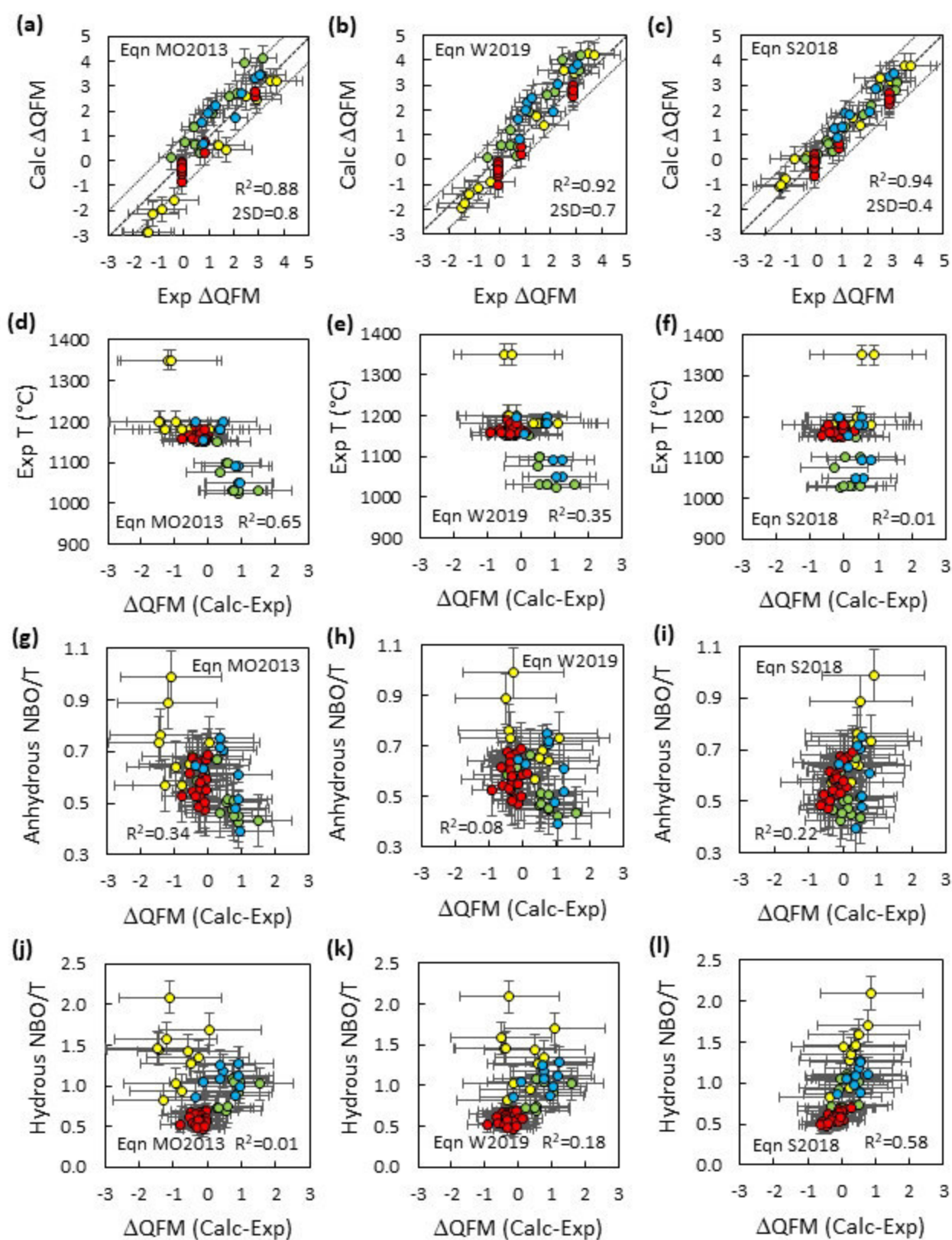
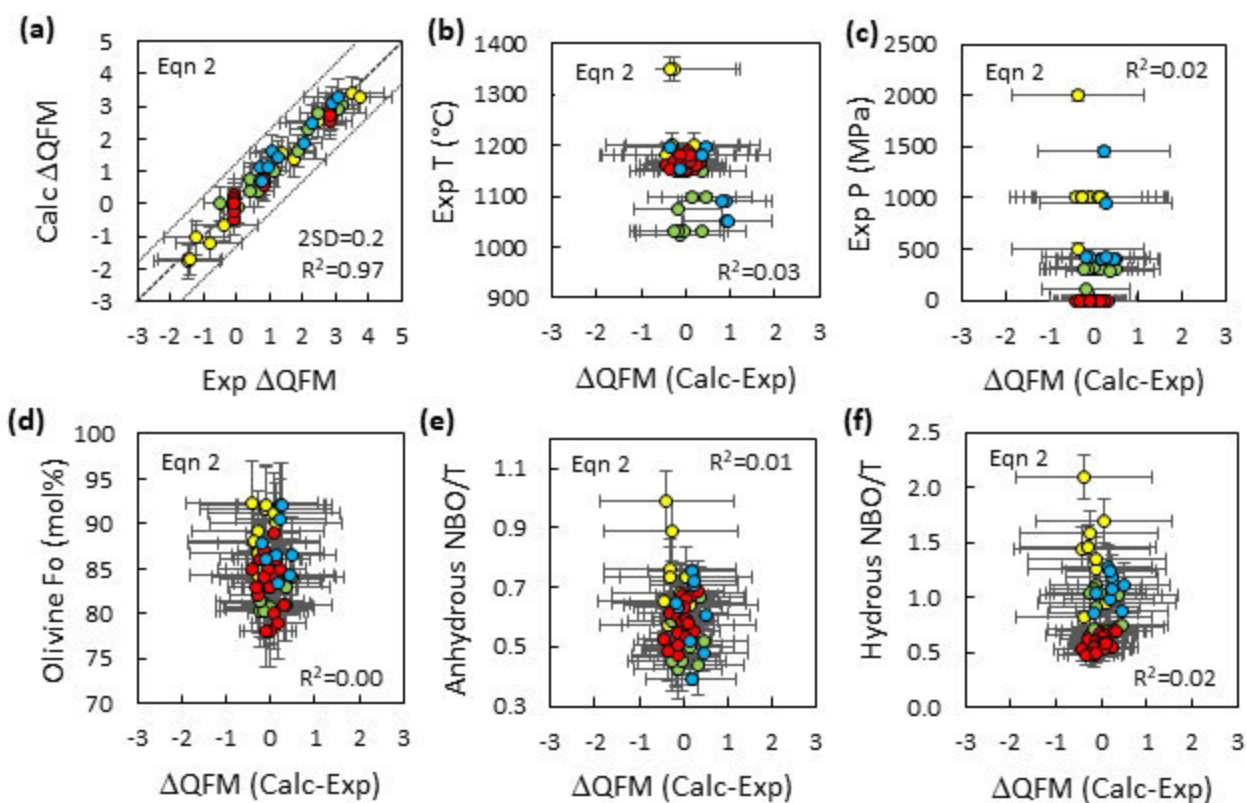


Figure 2 – Erdmann et al.



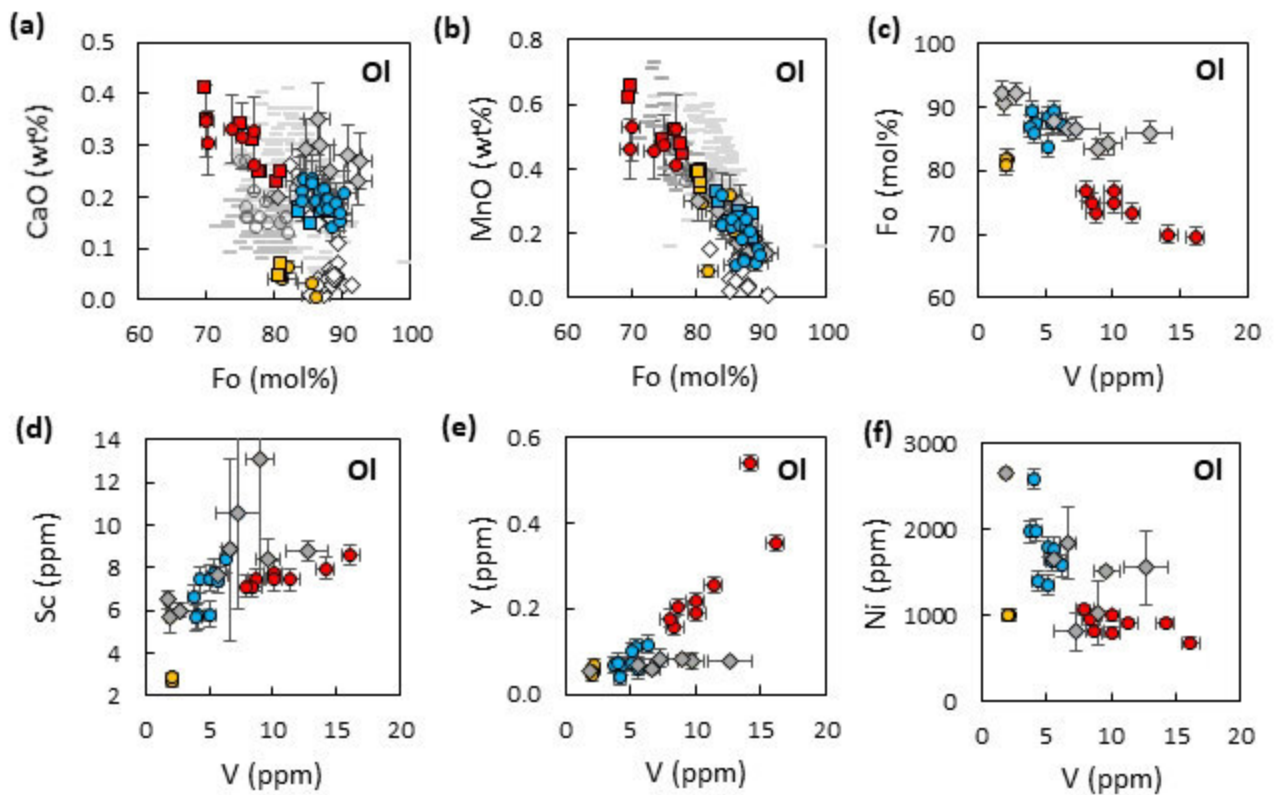
Ol data: STV301 ● L2014 ● S2018 ● W2019 ●

Figure 3 – Erdmann et al.



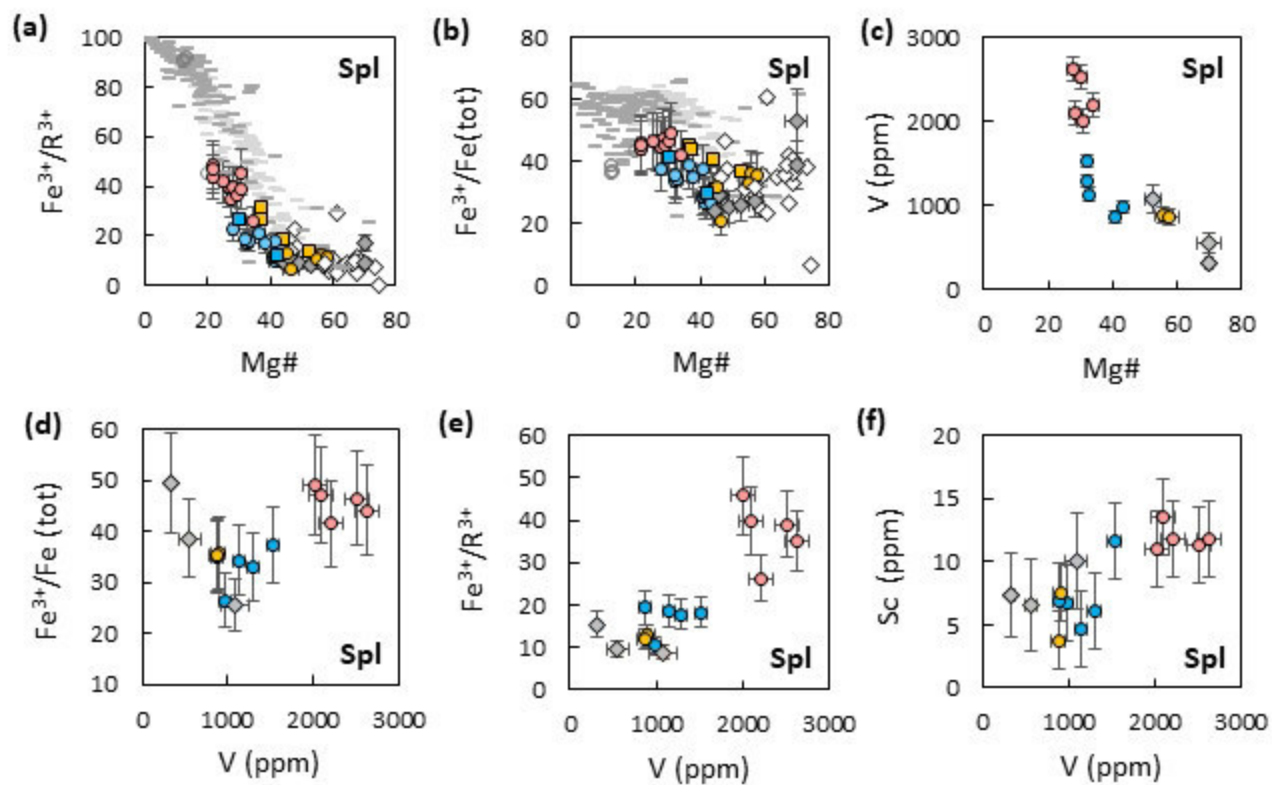
Data: STV301 ● L2014 ● S2018 ● W2019 ●

Figure 4 – Erdmann et al.



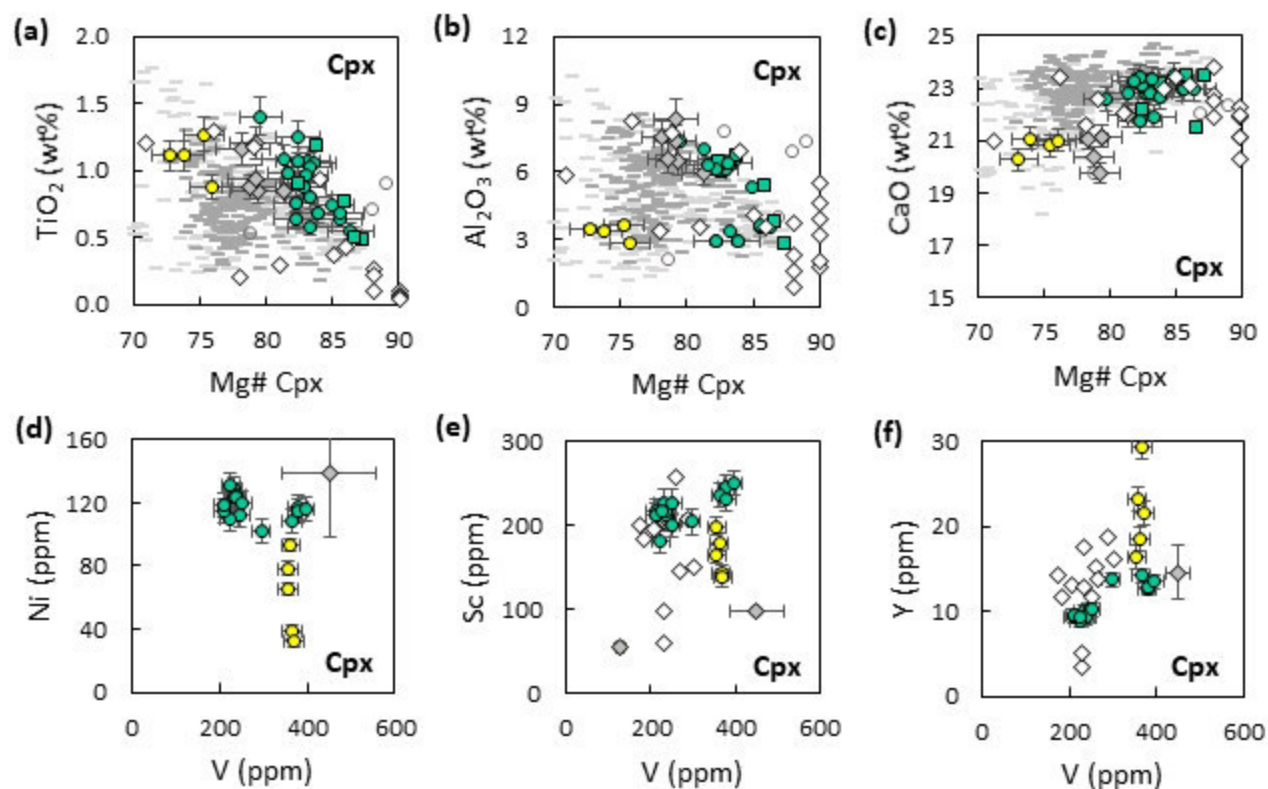
STV301: ● Phenocryst cores ● Phenocryst rims ● Inherited cores ◆ Experimental
 ○ LMB — Grenada lavas — St. Vincent-Grenada cumulates ◇ Peridotite XLTHs

Figure 5 – Erdmann et al.



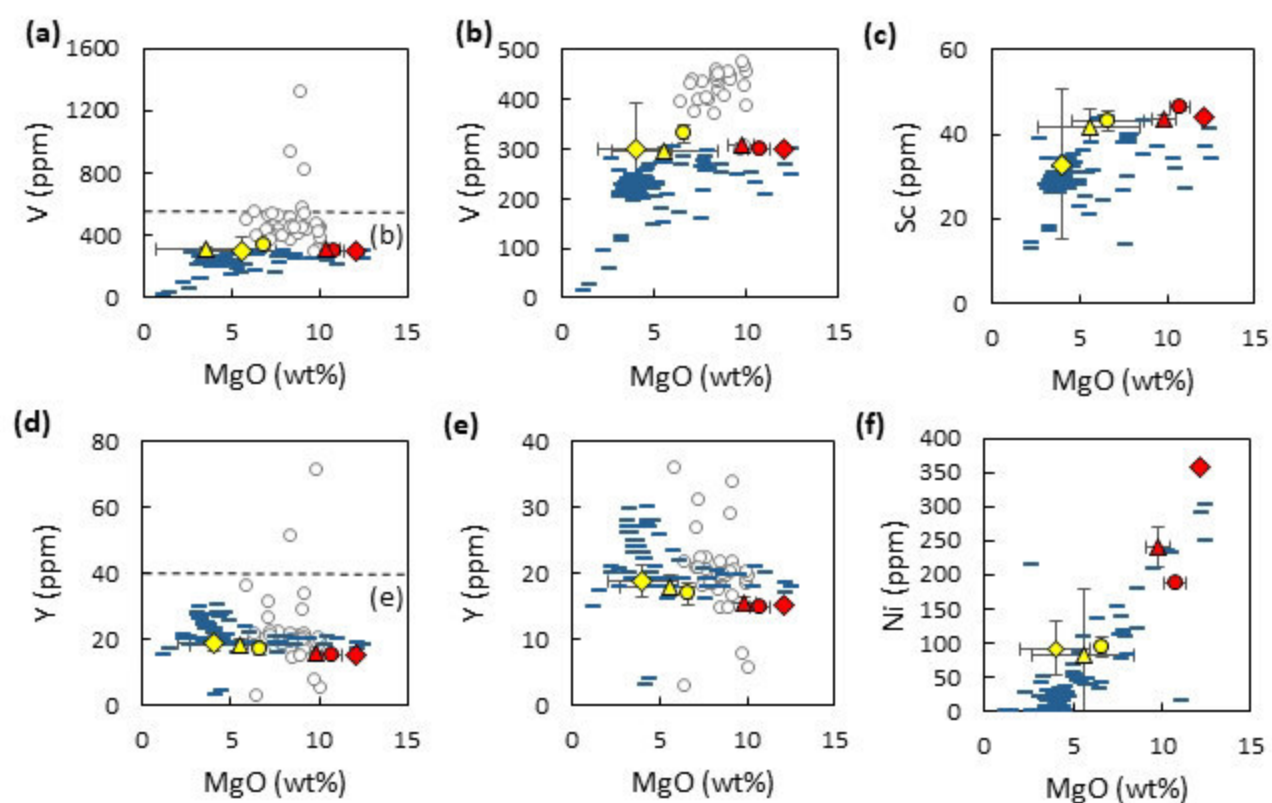
STV301: ● Inclusions in Ol ● Phenocryst cores ● Phenocryst rims ◆ Experimental
 ○ LMB — Grenada lavas — St. Vincent-Grenada cumulates ◆ Peridotite XLTHs

Figure 6 – Erdmann et al.



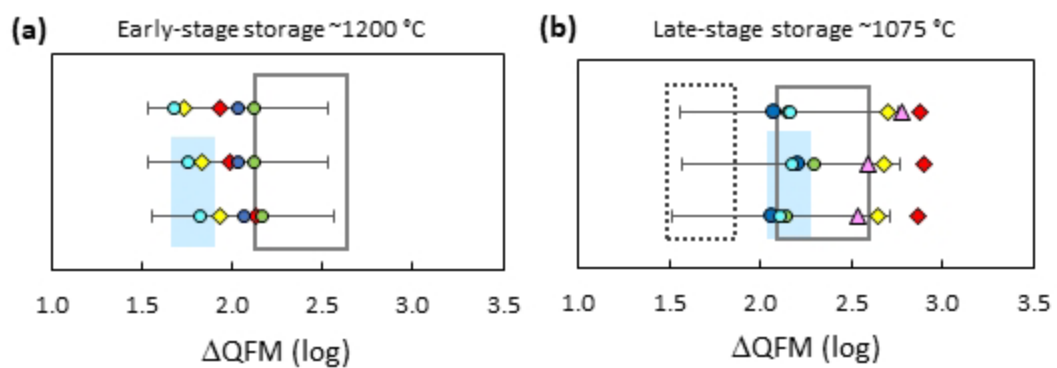
STV301: ● ■ Sector-zoned phenocrysts ● Inherited cores ◆ Experimental
 ○ LMB — Grenada lavas — St. Vincent-Grenada cumulates ◆ Grenada lavas

Figure 7 – Erdmann et al.



STV301 early-storage melt: E1 ◆ E2 ● E3 ▲ late-stage storage melt: E4 ▲ E5 ● E6 ◆
 St. Vincent whole-rock compositions ■ Olivine-hosted melt inclusions (Bouvier et al. 2010; Yellow Tuff?) ○

Figure 8 – Erdmann et al.



Calibration: ◆ MO2013 ● S2018 ◆ ▲ W2019 (Ol/Spl) ● Eqn 1 ● Eqn2
○ Our best estimates □ H1998 □ PM2007

Figure 9 – Erdmann et al.

Table 1: Experimental details, V compositions, and partitioning coefficients of minerals and glasses

Charge	Ref. Exp.	Temp. Press.	Δ QFM (log)	Melt H ₂ O (wt%)	Assemblage	Phase (n)	V ppm(2RSD%)
4-1	PM2007 IHPV	1050 °C 400 MPa	1.3	7.7	Gl-Ol-Spl-Cpx-Q	Gl(10) Ol(5) D-Ol/L	353(2.8) 7.31(23.6) 0.021(25.0)
4-2	PM2007 IHPV	1050 °C 400 MPa	1.0	6.3	Gl-Ol-Spl-Cpx-Q	Gl(10) Ol(5) D-Ol/L	297(8.0) 9.0(12.6) 0.03(12.8)
4-3	PM2007 IHPV	1050 °C 400 MPa	1.1	5.9	Gl-Ol-Spl-Cpx-Pl-Q	Gl(3) Cpx(3) D-Cpx/L	254(18.8) 451(13.8) 1.8(23.4)
3-1	PM2007 IHPV	1092 °C 402 MPa	1.1	5.1	Gl-Ol-Spl-Q	Gl(10) Ol(3) Spl(3) D-Ol/L D-Spl/L	331(0.6) 6.7(8.8) 1078(15.6) 0.02(8.8) 3.3(15.6)
3-2	PM2007 IHPV	1092 °C 402 MPa	0.7	4.2	Gl-Ol-Spl-Cpx-Q	Gl(10) Ol(3) D-Ol/L	298(11.8) 9.6(10.0) 0.03(15.4)
6-1	PM2007 IHPV	1155 °C 415 MPa	0.8	4.2	Gl-Ol-Spl-Q	Gl(8) Ol(3) D-Ol/L	310(0.6) 12.70(13.2) 0.041(13.2)
7-2	PM2007 IHPV	1198 °C 420 MPa	2.9	4.7	Gl-Ol-Spl-Q	Gl(12) Ol(5) D-Ol/L	284.4(1.4) 1.86(13.8) 0.007(13.9)
7-3	PM2007 IHPV	1198 °C 420 MPa	2.1	2.3	Gl-Ol-Spl-Q	Gl(9) Ol(8) Sp(4) D-Ol/L D-Spl/L	295.4(0.6) 5.64(16.0) 553(22.0) 0.019(16.0) 1.9(22.0)
PCW11	P2002 PC	1160 °C 1500 MPa	3.5	5.8	Gl-Cpx-Opx-Spl-Q	Gl(8) Cpx(3) D-Cpx/L	369(1.2) 128(6.0) 0.3(6.2)
PCW12	P2002 PC	1180 °C 1450 MPa	2.3	3.7	Gl-Ol-Spl-Q	Gl(9) Ol(3) Sp(3) D-Ol/L D-Spl/L	286(0.6) 2.74(39.0) 323(6.0) 0.010(39.0) 1.1(6.0)
PCW13	P2002 PC	1180 °C 950 MPa	3.1	5.0	Gl-Ol-Spl-Q	Gl(7) Ol(7) D-Ol/L	288(1.8) 1.74(13.4) 0.006(13.6)

Gl=glass, Ol=olivine, Spl=spinel, Cpx=clinopyroxene, Opx=orthopyroxene, Pl=plagioclase, Q=quench crystals, L=liquid, D-Ol/L=olivine-liquid partition coefficient; n=number of averaged analyses. Ref.=reference; Exp.=experiment; Temp.=temperature; Press.=pressure; P2002=Pichavant et al. 2002; PM2007=Pichavant & Macdonald 2007; IHPV=internally heated pressure vessel experiment; PC=piston cylinder experiment; uncertainties on compositions are given as two relative standard deviations (2RSD%). The reported fO_2 values are expressed relative to the QFM buffer defined by O'Neill (1987) in log bar units.

Table 2: Regression parameters for equation 1 and 2 olivine-melt V oxybarometers

Equation 1	Coeff.	Error	Probability		
Constant	-3.2848	0.1670	3.4955E-25	R ²	0.94
Log(D _v)	-2.9893	0.1075	4.4346E-32	Observations	52
				Regressions	1
Equation 2	Coeff	Error	Probability		
Constant	-2.8438	0.1261	1.5999E-27	R ²	0.97
Log(D _v)	-3.1960	0.0772	8.8955E-40	Observations	52
h-NBO/T	-0.8781	0.1127	3.9404E-10	Regressions	2

D_v is the V partitioning coefficient for olivine-melt; h-NBO/T is the hydrous NBO/T, calculated as outlined in the text. Coeff.=Coefficients of the regression. Uncertainties can be calculated as

Table 3: Estimated range of possible melt compositions and calculated mineral-melt partitioning coefficients for the STV301 magma during intermittent crustal storage

Estimate	Early-stage storage ^a			Late-stage storage ^b		
	E1 Bulk rock	E2* Exp. glass	E3* Calc. melt	E4 Calc. melt	E5* Exp. glass	E6* Matrix
SiO ₂	45.0(0.8)	45.9(0.4)	45.8(0.9)	48.0(1.7)	47.0(0.5)	45.5(1.4)
TiO ₂	1.1(0.6)	1.0(0.1)	1.1(0.6)	1.3(0.2)	1.2(0.1)	0.7(0.3)
Al ₂ O ₃	14.7(0.4)	15.0(0.4)	15.7(0.6)	18.8(1.8)	17.4(0.2)	20.6(4.3)
Fe ₂ O ₃ ^d	3.0(0.2)	3.1(0.2)	2.9(0.3)	2.7(0.5)	2.9(0.1)	2.7(1.8)
FeO ^d	6.3(0.4)	6.4(0.3)	6.0(0.5)	4.9(0.9)	5.6(0.2)	5.0(3.2)
MnO	0.2(0.1)	0.1(0.1)	0.2(0.1)	0.2(0.1)	0.2(0.1)	0.1(0.1)
MgO	11.5(0.2)	10.2(0.6)	9.3(0.7)	3.8(2.9)	6.4(0.2)	5.3(2.0)
CaO	10.5(0.2)	11.0(0.4)	11.1(0.3)	11.7(0.6)	11.3(0.2)	11.7(1.8)
Na ₂ O	2.2(0.1)	2.2(0.2)	2.3(0.1)	3.0(0.4)	2.5(0.1)	2.8(0.5)
K ₂ O	0.5(0.1)	0.5(0.1)	0.5(0.1)	0.7(0.2)	0.6(0.1)	0.6(0.1)
H ₂ O ^c	5.0(1.0)	5.0(1.0)	5.0(1.0)	5.0(1.0)	5.0(1.0)	5.0(1.0)
V	284(3)	284(4)	291(5)	285(7)	315(19)	283(92)
Sc	42.0(0.9)	44.0(1.0)	41.5(1.0)	31.1(4.1)	40.9(2.5)	39.8(17.8)
Y	14.4(0.3)	14.1(0.4)	14.8(0.7)	17.8(1.7)	16.0(1.7)	17.1(2.5)
Ni	340(5.0)	177(2.3)	228(29)	87(99)	89(14)	77(39)
NBO/T	0.81	0.74	0.70	0.37	0.50	0.40
h-NBO/T	1.32	1.24	1.20	0.82	0.97	0.86
K _D Ol/L(FeT-Mg)	0.27	0.24	0.23	0.16	0.25	0.23
K _D Ol/L(V)	0.014-0.020	0.014-0.020	0.014-0.019	0.015-0.018	0.013-0.016	0.015-0.018
K _D Ol/L(Sc/Y)	26-44	24-41	27-46	45-79	31-54	34-59
K _D Ol/L(Ni)	5.2-7.6	10-15	7.8-11	16-23	15-22	18-26
K _D Spl/L(V)	-	-	-	2.9-3.3	2.8-3.1	3.1-3.4

Concentrations normalized to 100 wt% for major and minor oxide components including H₂O and in ppm for trace elements V, Sc, Y, and Ni with estimated uncertainties (2std) in parentheses. a=Estimated early-stage storage melt compositions (E1 to E3) and calculated partitioning coefficients for primitive olivine phenocrysts cores (~Fo89, ~4.0-5.6 ppm V, ~5.7-7.4 ppm Sc, ~0.06-0.08 ppm Y, ~1780-2590 ppm Ni). b=Estimated late-stage storage melt compositions (E4 to E6) and calculated partitioning coefficients for evolved olivine phenocryst cores (~Fo84-86, ~4.2-5.1 ppm V, ~5.8-8.0 ppm Sc, ~0.04-0.10 ppm Y, ~1350-2000 ppm N) and spinel cores (>Mg#40, ~870-970 ppm V). c=estimated following Pichavant & Macdonald (2007). d=estimated using the formulation of Kress & Carmichael (1991) with fO_2 constrained using our equation 1 oxybarometer. (E1) Bulk-rock sample, which was analyzed as a glass fused from STV301 rock powder, which was the experimental starting material for the STV301 experiments. Oxide concentrations are from PM2007. (E2) Experimental glass composition 7-2. Oxide concentrations are from PM2007. (E3) Bulk-rock composition minus xenocryst and primitive phenocryst core compositions. (E4) Bulk-rock composition minus xenocryst and primitive to evolved phenocryst core compositions. (E5) Average of glass compositions 3-1 and 3-2. Oxide concentrations are from PM2007. (E6) Coarsely crystallized matrix composition. Indicated uncertainties for E1, E2, E4, and E5 are analytical uncertainties; indicated uncertainties for E3 and E6 are analytical uncertainties for bulk-rock composition plus and an estimated arbitrary uncertainty of $\pm 20\%$ on mineral proportions used for calculating residual melt composition. PM2007=Pichavant & Macdonald (2007). * Our best estimates. Additional details on the calculated melt compositions are provided in Tables S10 and S11.

Table 4: Calculated temperature (T) in °C and oxygen fugacity (fO₂) relative to ΔQFM in log units for the STV301 magma

Calibration	Early-stage storage						Late-stage storage						
	E1	E2*	E3*	E4	E5*	E6*	E1	E2*	E3*	E4	E5*	E6*	
T_OI/L_Sc/Y_MO2013	Av	1207	1220	1194	1059	1044	1112	1209±30	1059	1044	1112	**Av±σ(2sd)	1078±30
	Max	1262	1276	1248	1104	1090	1161		1104	1090	1161		
	Min	1151	1163	1139	1013	998	1063		1013	998	1063		
T_OI/L_Ni_P2017	Av	1365	1207	1277	1163	1144	1117	1242±60	1163	1144	1117		1130±60
	Max	1418	1250	1325	1206	1186	1157		1206	1186	1157		
	Min	1312	1164	1229	1120	1102	1076		1120	1102	1076		
fO ₂ _OI/L_V_MO2013	Av	1.73	1.83	1.93	2.70	2.68	2.64	1.9±0.5	2.70	2.68	2.64		2.7±0.5
	Max	2.02	2.12	2.19	2.88	2.86	2.81		2.88	2.86	2.81		
	Min	1.43	1.53	1.66	2.52	2.50	2.46		2.52	2.50	2.46		
fO ₂ _OI/L_V_W2019	Av	1.93	1.99	2.13	2.87	2.90	2.86	2.1±0.5	2.87	2.90	2.86		2.9±0.5
	Max	2.21	2.27	2.41	3.03	3.06	3.02		3.03	3.06	3.02		
	Min	1.64	1.70	1.85	2.71	2.74	2.69		2.71	2.74	2.69		
fO ₂ _OI/L_V_S2018	Av	2.12	2.12	2.16	2.15	2.29	2.14	2.1±0.9	2.15	2.29	2.14		2.2±0.9
	Max	2.34	2.34	2.38	2.28	2.42	2.27		2.28	2.42	2.27		
	Min	1.89	1.89	1.93	2.02	2.16	2.01		2.02	2.16	2.01		
fO ₂ _OI/L_V_Eqn1	Av	2.03	2.03	2.06	2.07	2.20	2.06	2.1±0.5	2.07	2.20	2.06		2.1±0.6
	Max	2.25	2.25	2.28	2.19	2.32	2.18		2.19	2.32	2.18		
	Min	1.81	1.81	1.84	1.94	2.07	1.93		1.94	2.07	1.93		
fO ₂ _OI/L_V_Eqn2	Av	1.68	1.75	1.82	2.16	2.17	2.11	1.8±0.5	2.16	2.17	2.11		2.1±0.6
	Max	1.91	1.98	2.05	2.29	2.30	2.24		2.29	2.30	2.24		
	Min	1.45	1.52	1.59	2.02	2.03	1.98		2.02	2.03	1.98		
fO ₂ _Spl/L_V_W2019	Av	-	-	-	2.78	2.59	2.53	-	2.78	2.59	2.53		2.6±0.5
	Max	-	-	-	3.01	2.82	2.76		3.01	2.82	2.76		
	Min	-	-	-	2.54	2.35	2.29		2.54	2.35	2.29		

Temperatures were calculated using the Sc/Y thermometer of Mallmann & O'Neill (MO2013) and the Ni thermometer of Pu et al. (2017) (P2107). The fO₂ values were calculated using the oxybarometers of Mallmann & O'Neill (2013) (MO2013), Wang et al. (2019) (W2019), Shishkina et al. (2018) (S2018), and our new equation 1 and equation 2 oxybarometers. * Our best estimates; ** average for the best estimates E2 and E3 or E5 and E6, respectively. Reported uncertainties (σ at 2SD level) are inferred standard errors for M2013, W2019, and P2017, and calculated for S2018 and our calibrations. For Eqn 1 and 2, we have assumed errors of 0.001 (E1 to E3) and 0.002 (E4 to E5) on K₀OI/L(V) and errors of 0.1 (E1 to E3) and 0.2 (E4 to E5) on estimated hydrous melt NBO/T, following from the range of compositions presented in Table 3. L=liquid, Ol=olivine, Spl=spinel.