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## Cationic disorder: Governing the spin-insulatronic properties of nanocrystalline ZnFe<sub>2</sub>O<sub>4</sub> thin films

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#### ABSTRACT

Controlling cation order/disorder in spinel offers a highly effective means of tailoring material properties by modifying inter- and intra-sub-lattice ionic interactions. In this study, we conducted high-temperature magnetization measurements (300–1000 K) to determine average Curie temperatures ( $T_C$ ) for nanocrystalline  $ZnFe_2O_4$  thin films. Thermodynamical stability of these films under extreme conditions was investigated by assessing lattice structures, oxygen vacancies, magnetization, and electric resistivity for spin-insulatronic applications. Reversible cation inversion via heat treatments (in-situ and ex-situ) yields tunable ferrimagnetic (FiM) order in  $ZnFe_2O_4$ , with  $T_C$  ranging from 425 to 710 K. First-principles calculations highlight effective cation inversion mitigating magnetic frustration, promoting collinear FiM ordering, and elevating  $T_C$ . Oxygen vacancies further reinforce ferrimagnetic nanophase diagram elucidates dominant competing magnetic ground states (cluster spin-glassy state, FiM, and antiferromagnetic) with increasing growth temperature, fostering innovative homo-architectures from multiple  $ZnFe_2O_4$  thin films with diverse functionalities.

#### 1. Introduction

Generation and detection of spin current is crucial for spintronicbased information transfer and processing for future low-power "green" electronics [1]. Pure spin current stems from forcing electrons with different spins to flow in opposite directions, which can be achieved via spin pumping (SP) or the spin Seebeck effect (SSE) among other methods [2,3]. Magnetic insulators are frontrunners [3] in producing pure spin currents via magnonics in both SP and SSE processes due to their low eddy current losses and negligible anomalous Nernst effects. Epitaxial Ni<sub>0.65</sub>Zn<sub>0.35</sub>Al<sub>0.8</sub>Fe<sub>1.2</sub>O<sub>4</sub>, MgAl<sub>0.5</sub>Fe<sub>1.5</sub>O<sub>4</sub>, Li<sub>0.34</sub>Mn<sub>0.1</sub>Zn<sub>0.32</sub>Fe<sub>2.19</sub>O<sub>4</sub> and Li<sub>0.5</sub>Fe<sub>2.5</sub>O<sub>4</sub> spinel thin films with low damping ( $\sim$ 10<sup>-3</sup>) have been recently explored [2,4–6] as alternatives to garnet thin films. However, the former thin films require multicomponent targets, single-crystal substrates, and high growth temperatures, making them still expensive for integration with existing spintronic technology.

Affordable and low-temperature-processed Zn-ferrite (ZnFe<sub>2</sub>O<sub>4</sub>) thin films can be ideal candidates for the aforementioned applications due to the presence of only trivalent iron (Fe<sup>3+</sup>) magnetic ions, providing tunable magnetic and insulating properties with minimal microwave losses [7,8]. However, Zn<sub>x</sub>Fe<sub>3-x</sub>O<sub>4</sub> and/or ZnFeO<sub>4- $\delta$ </sub> thin films, when deposited in low oxygen pressures/vacuum [9–11], have exhibited resistivity values five orders of magnitude lower ( $\rho_{300 K} = 0.05 - 5 \times 10^{-3}$  $\Omega - \text{cm}$ ) than bulk ZnFe<sub>2</sub>O<sub>4</sub>, while also displaying high magnetizations ( $4\pi M_S = 5.6$  kG) and Curie temperatures ( $T_C \ge 700$  K). Therefore, simultaneous control of the electronic and magnetic properties of Zn-ferrite thin films continues to be a significant challenge. This

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necessitates the engineering of nanocrystalline Zn-ferrite thin films through cation inversion in an oxygen-rich environment [7,8]. Previously, we have demonstrated the ability to control a wide range of magnetic properties (superparamagnetic (SPM), ferrimagnetic (FiM), and paramagnetic (PM)) in nanocrystalline Zn-ferrite thin films depending on substrate temperature ( $T_S$ ) [12]. The FiM phase shows an initial increase in  $4\pi M_S$  values with  $T_S$ , which reach a maximum (3 kG) [7] and then decrease to PM bulk ZnFe<sub>2</sub>O<sub>4</sub> values [13], attributed to reversible cation inversion. These FiM films also exhibit a low microwave linewidth (40–200 Oe) [8], paving the way for spin-insulatronics.

Here, built on prior work on Zn-ferrite thin films [7,8,12], we performed high-temperature magnetization measurements (300–1000 K) to investigate average  $T_C$  values and gain insights into cation inversion, grain size, and spin-insulatronic properties. Density functional theory (DFT) revealed the suppression of geometrical magnetic frustration and formation of defect states near Fermi levels, leading to high-temperature FiM order and slightly reduced resistivity. A detailed magnetic nanophase diagram was constructed, unveiling diverse magnetic ground states.

#### 2. Methods

Nanocrystalline Zn-ferrite thin films with a thickness of 200 nm were laser ablated in a pure oxygen atmosphere ( $P(O_2) = 0.16$  mbar) on amorphous fused quartz substrates [7]. These thin films were grown at different  $T_S$ , ranging from room temperature (RT) to 850 °C (1123 K). The crystalline structure and chemical composition were analyzed using X-ray based techniques, including X-ray diffraction (XRD) performed on a Rigaku SmartLab X-ray diffractometer with a Cu K<sub>a</sub> X-ray source, and X-ray photoelectron spectroscopy (XPS) conducted on a Kratos AXIS Ultra DLD Photoelectron spectrometer with an Al  $K_{\alpha}$  anode. The phase purity of each thin film was determined by micro Raman spectra in the range of  $100 - 1000 \text{ cm}^{-1}$ . Grain size distribution was determined by analyzing bright-field micrographs obtained from transmission electron microscopy (TEM) using a Philips CM 200 microscope and analyzed using ImageJ 1.46 analysis software. Magnetic properties were measured using a vibrating sample magnetometer (VSM) in an in-plane layer configuration, connected to a physical property measurement system (PPMS), where the sample was kept under a vacuum of  $10^{-5}$ mbar within the temperature range of up to 1000 K. The high-temperature magnetization measurement (HTMM) process is equivalent to vacuum annealing and includes in-situ measurements. Two measurement cycles were applied to the samples. Cycle-I measured temperatures from 300 to 800 K and then returned to 300 K, with a ramp rate of 12 K/min and a fixed field of 5 kOe. Cycle-II covered the temperature range from 300 K to 1000 K without removing the sample of Cycle-I from the chamber, using the same ramp rate and applied field. Afterward, the sample was cooled to 5 K, and a low-temperature magnetization measurement (LTMM) was conducted while maintaining a constant flow of helium gas between 5 and 300 K, along with the same ramp rate and applied field. The diamagnetic contribution of the quartz substrate and sample holder was subtracted from the magnetization data.

DFT calculations were performed using the Vienna *Ab initio* Simulation Package (VASP) [14,15] and the projector augmented-wave (PAW) method [16], with a cut-off energy of 600 eV. We used the exchange-correlation energy functional parametrized by Perdew, Burke, and Ernzerhof (PBE) [17] with a "+*U*"-dependent correction [18] added to the *3d* bands of Fe atoms and a  $U_{eff}$  parameter of 4.5 eV. The first Brillouin zone associated to the conventional spinel cubic cell was sampled with a Monkhorst-Pack grid [19] of  $6 \times 6 \times 6$  *k*-vectors. The spinel structure is formed by a distorted face-centered lattice of oxygen atoms. When the structure is stoichiometric and adopts a normal cation distribution, the divalent  $Zn^{2+}$  cations are located in tetrahedral atomic sites (A), while all trivalent Fe<sup>3+</sup> cations occupy octahedral sites (B). The presence of cation disorder is mostly attributed to  $Zn^{2+}/Fe^{3+}$  cation

exchanges between A and B sites, according to the chemical formula  $[Zn_{1-x}^{2+}, Fe_x^{3+}]_A [Zn_x^{2+}, Fe_{2-x}^{3+}]_B O_4^{2-}$ , where *x* is called the inversion degree. When the spinel structure is perfectly inverse, the most stable  $Zn^{2+}/Fe^{3+}$  cation ordering in B sites corresponds to the space group  $P4_122$  (No. 91). Structures with an intermediate inversion degree were generated by keeping this cation distribution in B sites fixed [20]. For every structure, different magnetic configurations (FM, AFM, or FiM) were tested. The Heisenberg exchange integrals  $J_{ij}$  were calculated using the Liechtenstein-Katsnelson-Antropov-Gubanov formula [21] implemented in the QuantumATK code [22,23]. With this method, all the exchange integrals between a specific atom and the neighboring atoms included in a range of 3 conventional cells (i.e., with a maximum nearest-neighbor distance of  $\approx 20$  Å) were calculated. These integrals were then used to obtain the Curie Temperature  $T_C^{MFA}$  within the mean-field approximation (MFA) [24].

#### 3. Results and discussion

In the present study, three sets of nanocrystalline Zn-ferrite thin films, namely *ZFRT* ( $T_S$  of RT), *ZF350* ( $T_S$  of 350 °C), and *ZF850* ( $T_S$  of 850 °C), were selected for detailed magnetization studies. These films exhibit single-phase ZnFe<sub>2</sub>O<sub>4</sub> (Supplementary Material Fig. S1) with three types of grain size distributions (Supplementary Material Fig. S2). The average grain sizes (estimated from TEM micrograph and Scherrer formula), lattice constants (estimated from Rietveld analysis), and magnetic transition temperatures [12] are presented in Table 1. The *ZFRT* layer predominantly contained small-sized grains (10–20 nm) and exhibited a dominant SPM behavior at 300 K, with a blocking temperature ( $T_B$ ) of 100 K. In contrast, the *ZF350* layer, composed mainly of intermediate-sized grains (25–35 nm), showed a dominant FiM order. Finally, the *ZF850* layer, with larger grain sizes (50–70 nm), displayed a dominant bulk ZnFe<sub>2</sub>O<sub>4</sub> PM behavior with N é el temperature  $T_N$  of 15 K.

The M-T curves for the *ZFRT*, *ZF350*, and *ZF850* thin films under HTMM (305 K  $\leq T \leq 1000$  K) at a fixed field of 5 kOe are shown in Fig. 1(a-d) for both heating and cooling cycles. No impurity phases (Supplementary Material Fig. S1) were detected after HTMM; only grain growth and slight increment in lattice parameters (Table 1). Distinct M-T behaviors can be observed in all three thin films. Susceptibility ( $\chi$ ) vs. *T* curves of *ZF850* thin films follow Curie-Weiss law, *C/T-θ* of PM behavior, during reversible heating–cooling cycles (Fig. 1(a)) with slight presence of ferromagnetic interactions below Weiss temperature  $\theta$ =226 K. This feature is expected because the maximum temperature of HTMM (800 K) is not higher than  $T_{S}$ , resulting in no discernible change in the crystalline structure in Cycle-I (Supplementary Material Fig. S1),

Table 1

Spontaneous magnetization ( $4\pi M_S$ ), magnetic ordering temperatures (average Curie temperature  $\overline{T_{c}}$ , Néel temperature  $T_N$ , and blocking temperature  $T_B$ ), lattice constants, average grain size [7] and resistivity ( $\rho_{300 \text{ k}}$ ) measured at 300 K for Zn-ferrite thin films at different  $T_S$  before and after HTMM.

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As-grown thin films	4πM <sub>S</sub> (kG)	Lattice constant (Å)	Transition temperature (K)	Grain sizes (nm)	Resistivity <sup>ρ</sup> 300 κ (Ω-cm)			
ZFRT ZF350	0.4 3	- 8.416	$\frac{T_B = 100}{T_C} = \sim 589$	14 31	$>10^4$ $2.5 imes10^2$			
ZF850	0.2	8.446	$T_N = 15$	68	$> 10^{4}$			
Bulk	0.02	8.44	$T_N = 15$	-	$2  imes 10^2$			
After HTMM								
ZFRT (Cvcle-I)	2.1	8.433	$\overline{T_C} = \sim 710$	46	$1.3\times10^{1}$			
ZF350 (Cycle-I)	3.3	8.418	$\overline{T_C} = \sim 557$	46	$3.1\times10^{1}$			
ZF350 (Cycle- II)	1.3	8.421	$\overline{T_C} = \sim 425$	56	2.1			



**Fig. 1.**  $\chi$  vs.T curves of ZF850 layer (a), M–T curves of ZF350 layer (b-d), and ZFRT (c) during HTMM heating and cooling cycle measured at 5 kOe. In Cycle-I, the ZF350 layer was heated up to 800 K, and in Cycle-II it was heated up to 1000 K. The inset (b) shows the M–T curve at 5 kOe in LTMM (5 – 300 K) (green open circle) and the fitted curve (red solid line) using Bloch T<sup>3/2</sup> law. The inset (c) shows a larger view of the heating cycle M–T curve of ZFRT. The T<sub>C</sub> probability distribution functions were obtained from fitted M–T curves in equation-1(e).

so we will not discuss it further.

A notable disparity in the M-T behavior can be observed between the heating and cooling stages of the ZFRT thin films (Fig. 1(c)). During the heating process up to a maximum temperature of ~800 K (>  $T_s$ ), the thin films initially exhibit PM behavior with a decrease in magnetization. However, at around 700 K, a peak develops (inset of Fig. 1(c)). In the cooling stage, the magnetization increases with decreasing temperature, until reaching 305 K. This behavior suggests the crystallization of amorphous/disordered ZnFe2O4 phases during the heating stage Cycle-I (Supplementary Material Fig. S1), leading to FiM behavior. Interestingly, during the heating of ZF350 thin films in Cycle-I (Fig. 1(b)), a gradual decrease in magnetization is observed over a broad temperature range, and magnetization vanishes at around 650 K, indicating progressive grain growth during the heating stage (Table 1). While in the cooling stage, the M-T curve lies above that of the heating stage, with slight hysteresis, suggesting that heating up to 750 K ( $> T_S$ ) causes grain growth. This conclusion is further supported by the extrapolated  $T_C$ values of 550 K obtained from Bloch  $T^{3/2}$  law fitting of the lowtemperature M-T curve (inset of Fig. 1(b)), assuming a fixed composition and grain sizes. This FiM behavior can be attributed to a cation inversion [25], with the displacement of some of  $Fe^{3+}$  ions from the octahedral sites to the tetrahedral sites. Notably, the magnetization curve of ZF350 thin films (Cycle-II) (Fig. 1(d)) during the cooling stage is lower than those of the heating stage, with magnetization vanishing around 500 K. This indicates that further heating up to 1000 K ( $> T_S$ ) leads to larger grain sizes approaching the bulk's paramagnetic structure (Table 1). Thus, high-temperature treatments, no matter whether during high-temperature growth or during HTMM, lead to grain growth and distributions of grain sizes. Moreover, the HTMM process was conducted in vacuum, so the presence of oxygen vacancies and the transformation

of a few Fe<sup>3+</sup> (5  $\mu_B$ ) ions into Fe<sup>2+</sup> (4  $\mu_B$ ) ions at this stage cannot be ruled out, which may lead to lower electric resistivity values (Table 1) compared with the bulk ZnFe<sub>2</sub>O<sub>4</sub>. The average magnetization in a heterogeneous FiM granular system can be expressed by assuming a Gaussian distribution function for the sample-specific  $T_C$  values based on the Central Limit Theorem [26,27] as:

$$< M(T) > = \int_{0}^{\infty} M_0 (\frac{T_c - T}{T_c})^{\beta} \quad \theta(T_c - T) \frac{1}{\sqrt{2\pi\Delta T_c}} e^{\frac{-(T_c - T_c)^2}{2\Delta T_c^2}} dT_c$$
 (1)

Here,  $\overline{T_C}$  represents the average Curie temperature and  $\Delta T_C$  is the width of the Curie temperature distribution.  $M_0$  is the saturation magnetization,  $\beta$  is the critical exponent (1.2 –1.5), and  $\theta$  is the Heaviside function, ensuring that M(T) equals zero for  $T > T_C$ . The experimental M-T curves were fitted near  $T_C$  using the least-squares method (Supplementary Material Fig. S3). The  $\overline{T_C}$  values of our nanocrystalline Zn-ferrite thin films are given in Table 1. The as-grown ZF350 layer shows  $\overline{T_c} = \sim 589$  K, which is reduced after HTMM to 557 K and 425 K in Cycle-I and Cycle-II, respectively. The drift of  $\overline{T_C}$  values of the ZF350 thin films toward lower temperatures is more clearly demonstrated in their probability distribution functions (PDFs), as shown in Fig. 1(e), after Cycle-I and II. The higher  $\Delta T_C$  in Cycle-II (95 K) compared to Cycle-I (45 K) can be attributed to the larger differences in nanograin size distribution within the thin films. After HTMM, ZFRT thin films transform into FiM with an increased  $\overline{T_C} = \sim 710$  K, along with a distribution width  $\Delta T_C = 150$  K. These thin films show large  $\overline{T_C}$  values with  $4\pi M_S$ values (2.1 kG). In Cycle-I of ZF350 thin films,  $\overline{T_c}$  is lower (~557 K) but show higher  $4\pi M_S$  values (3.3 kG), indicating different inter- and intrasublattice magnetic couplings.

Room temperature Raman spectra were obtained to investigate cation disorder in Zn-ferrite thin films before and after HTMM, as shown in Fig. 2(a-b). The bulk ZnFe<sub>2</sub>O<sub>4</sub> Raman data are also included in Fig. 2



**Fig. 2.** Raman spectra of ZFRT, ZF350 and ZF850 thin films along with deconvoluted Raman data using Lorentzian peak functions before (a) and after HTMM (b).

(a) for comparative study. We observed fewer Raman modes  $(A_{1g}, T_{2g}^2)$ and  $T_{2g}^3$ ) than the theoretically predicted five modes ( $A_{1g}, E_g$  and three  $T_{2g}$ ), but no extra peaks from impurity phases were detected [28]. In spinel ferrites, the  $A_{1,q}$  modes above 600 cm<sup>-1</sup> correspond to the motion of oxygen atoms in tetrahedral AO<sub>4</sub> sites, while the low-frequency modes are associated with vibrations within the octahedral BO<sub>6</sub> sites. The Raman spectra were fitted with five distinct peaks using Lorentzian functions. The peak positions of the  $A_{1g}$  and  $T_{2g}^2$  modes in the as-grown Zn-ferrite thin films (Fig. 2(a)) exhibited a significant red shift, indicating a shift towards smaller wavenumbers compared to the bulk data. In ZFRT and ZF350 thin films, the peak at 352 cm<sup>-1</sup> ( $T_{2g}^2$ ) broadens, and the peak at 447 cm<sup>-1</sup> ( $T_{2g}^3$ ) shows a broad shoulder at higher wavenumbers. In ZF850 thin films, however, these two peaks closely resemble bulk spectra. This observation suggests that smaller grain sizes cause cation disorder [28]. The broad peak at 650 cm<sup>-1</sup> ( $A_{1,q}$ ) observed in Zn-ferrite thin films can be separated into two Lorentzian peaks, ranging between around  $610 - 650 \text{ cm}^{-1}$  and  $685 - 698 \text{ cm}^{-1}$ . The former Raman modes can be attributed to oxygen breathing vibrations against  $Zn^{2+}$ , while the latter modes are associated with oxygen vibrations against  $Fe^{3+}$ . This suggests the coexistence of  $ZnO_4$  and  $FeO_4$ groups. If the vibrational frequencies of the two modes were very close, only a single broad peak would be observed. Otherwise, vibrations related to different cations at the same site may exhibit two distinct first-order Raman modes. In ZFRT and ZF350 thin films, the increased intensity ratio of the FeO<sub>4</sub> vibration compared with the ZnO<sub>4</sub> vibration suggests the incorporation of more Fe ions into the tetrahedral A sites. This rearrangement of cations results in the formation of two magnetic sublattices, which contribute to the enhanced magnetization compared to bulk ZnFe<sub>2</sub>O<sub>4</sub>. Similarly, the  $A_{1g}$  peak (650 cm<sup>-1</sup>) in ZF850 thin films is located at a similar position as in bulk ZnFe<sub>2</sub>O<sub>4</sub>, with a shoulder around 610 cm<sup>-1</sup>. Comparing the Raman spectra measured after HTMM, we observe further broadening of the  $T_{2g}^2$  and  $T_{2g}^3$  peaks, as depicted in Fig. 2(b). In ZFRT and ZF350 thin films heated at 750 - 800 K (Cycle-I), the  $T_{2g}^2$  peaks exhibit a noticeable red-shift, and the intensity of the shoulder of the  $T_{2g}^3$  peak at higher wavenumbers increases, directly resulting from the replacement of Fe atoms by heavier Zn atoms at the octahedral BO<sub>6</sub> sites. The intensity ratio of the FeO<sub>4</sub> vibration to the ZnO4 vibration, associated with the A1 g peaks, changes from 1:1 to 3:1 in ZFRT and ZF350 thin films, indicating an increase in cation inversion. However, this ratio is reduced in ZF350 thin films upon further heating at 1000 K (Cycle-II), as the cation distribution starts to revert back to the normal bulk structures, which is supported by XRD and magnetic studies. The ratio between the two  $T_{2g}^2$  and  $A_{1g}$  Raman scattering peaks appears to depend on the degree of inversion; the former signal is dominant in normal structure, while the latter is dominant in inverted structure.

We observed that cation disorder can lead to significant changes in symmetrical vibrations. Furthermore, heating under vacuum can introduce structural defects, such as oxygen vacancies, which can alter the local electronic properties and influence the electron-phonon interaction in Zn-ferrite thin films. Allen's formula [29] was utilized to estimate the strength of electron-phonon coupling ( $\lambda_i$ ) associated with the decay of phonons into electron-hole pair excitations and the resulting additional broadening. This estimation was based on the observed line shape parameters, including the peak position ( $\omega$ ) and the full width at half maximum ( $\Gamma_i$ ). The relationship is given by:

$$\lambda_i = \frac{\Gamma_i}{\omega^2} \left[ \frac{g_i}{2\pi N(E_F)} \right] \tag{3}$$

where  $g_i$  represents the degeneracy of the  $i^{\text{th}}$  mode and  $N(E_F)$  is the density of states at the Fermi level. Though the precise values of  $\lambda_i$  cannot be estimated, we have presented the relative variations in  $\lambda_i$  values for the high-intensity  $A_{Ig}$  and  $T_{2g}^2$  modes in Table 2. We also include the data for bulk ZnFe<sub>2</sub>O<sub>4</sub>, using a theoretical  $N(E_F)$  (3 states eV<sup>-1</sup> f.u.<sup>-1</sup>) at 300 K. The enhanced  $\lambda_i$  values observed in the as-grown

#### Table 2

Line shape parameters; peak position ( $\omega$ ) and full width at half maximum ( $\Gamma_i$ ) and electron-phonon coupling strength ( $\lambda_i$ ) for A<sub>1 g</sub> and T<sup>2</sup><sub>2 g</sub> Raman modes for Zn-ferrite thin films before and after HTMM.

As-grown thin films	$T_{2g}^2$ mode			A <sub>1 g</sub> mode			
	$\Gamma$ (cm <sup>-1</sup> )	$\omega$ (cm <sup>-1</sup> )	λ	$\Gamma$ (cm <sup>-1</sup> )	$\omega$ (cm <sup>-1</sup> )	λ	
ZFRT	83.1	347.7	0.88	96.7	658	0.09	
ZF350	121.8	329.9	1.38	126.3	656	0.14	
ZF850	70.5	347.3	0.74	80.9	647	0.06	
Bulk	-	-	0.13	-	-	0.05	
After HTMM							
ZFRT (Cycle-I)	99.6	336.9	1.07	169.5	669.5	0.16	
ZF350 (Cycle-I)	134.01	335.9	1.58	180.4	664	0.17	
ZF350 (Cycle-II)	94.1	326.9	1.13	108.7	654.5	0.11	

ZFRT ( $A_{1g}$ -0.09,  $T_{2g}^2$ -0.88) and ZF350 ( $A_{1g}$ -0.14,  $T_{2g}^2$ -1.38) thin films compared to bulk ZnFe<sub>2</sub>O<sub>4</sub> ( $A_{1g}$ -0.05,  $T_{2g}^2$ -0.13) and ZF850 layer  $(A_{1g}-0.06, T_{2g}^2-0.74)$  suggest an increase in spin disorder resulting from the redistribution of cations between the interstitial A and B sites and the formation of an inverse spinel structure [30]. The spin disorder in the ZFRT ( $A_{1g}$ -0.16,  $T_{2g}^2$ -1.07) and ZF350 ( $A_{1g}$ -0.17,  $T_{2g}^2$ -1.58) thin films increases after Cycle-I of HTMM. The changes in the  $T_{2g}^2$  mode result from the symmetric and asymmetric bending of Fe-O bonds at the octahedral B sites, which affect both the electronic and structural properties [30,31]. Consequently, the highest  $\lambda_i$  values can be attributed to the electronic disorder arising from defects/oxygen vacancies and the partial transformation of Fe<sup>3+</sup> ions into Fe<sup>2+</sup> at the B sites. However, if this had been the sole cause, the  $\lambda_i$  values would have continued to increase after Cycle-II of HTMM. The observed trend (A1g-0.11,  $T_{2g}^2$  (-1.13) suggests that cation disorder plays a dominant role over the electronic disorder and that it is reduced after Cycle-II, thus approaching the cation distribution of the normal spinel structure.

XPS analysis further confirms the cation inversion and the presence of only Zn<sup>2+</sup> and Fe<sup>3+</sup> ions in as-grown Zn-ferrite thin films (Supplementary Material Fig. S4). The influence of HTMM in Cycle-I and II on the surface chemical composition and ionic state of ZFRT and ZF350 thin films are investigated in Fig. 3(a-b). The Fe 2p XPS spectra exhibit two peaks, Fe  $2p_{3/2}$  (~711.4 eV) and Fe  $2p_{1/2}$  (~725 eV), of  $\rm ZnFe_2O_4$ without any iron oxides (FeO,  $\alpha/\gamma$  – Fe<sub>2</sub>O<sub>3</sub> and Fe<sub>3</sub>O<sub>4</sub>) impurity phases [32-34]. The satellite peak at binding energies (BE) = 718-719 eV indicates the dominance of  $Fe^{3+}$  ions [32–34].  $Fe^{2+}$  ions, which have a satellite peak at 714–715 eV and an Fe  $2p_{3/2}$  peak around 709.5 eV, are absent [35]. The asymmetric Fe  $2p_{3/2}$  peaks suggest that the inversion parameters are greater than zero. Specifically, the Fe<sup>3+</sup> ions occupying tetrahedral sites exhibit higher BE for their 2p electrons compared to Fe<sup>3+</sup> ions occupying octahedral sites. This is because tetrahedral coordination involves fewer O<sup>2-</sup> anions, resulting in lower electron density within their coordination sphere [36]. The higher BE peaks observed in the as-grown ZFRT thin films may indicate incomplete ferrite formation [34], which is supported by the XRD results. However, after HTMM Cycle-I, the width of the Fe  $2p_{3/2}$  peak narrowed down, although it remained broader than that of bulk ZnFe<sub>2</sub>O<sub>4</sub>. Conversely, the Fe 2p peaks of ZF350 layer after HTMM Cycle-II approached the BE values of the bulk, and their intensity was significantly low.

In the insets of Fig. 3(a-b), we also plot the O-1 *s* edge spectra of both thin films before and after HTMM. These asymmetric curves of O-1 *s* can be divided into two peaks around 529.5 eV and 531.8 eV, indicating different oxygen environments [37–40]. The peak, centered around 529.5 eV, corresponds to lattice oxygen ions ( $O_L$ ), while the higher binding energy peak at 531.8 eV is attributed to oxygen vacancies ( $O_V$ ). After Cycle-I and Cycle-II, the LTMM *M*-*T* curves (5- 300 K) of the Zn-ferrite thin films plotted in Fig. 3(c-d) do not exhibit a Verwey transition, which is typically associated with the presence of both Fe<sup>3+</sup> and Fe<sup>2+</sup>ions at octahedral sites of Fe<sub>3</sub>O<sub>4</sub>, confirming the absence of the



Fig. 3. XPS spectra of  $Fe_{2p}$  before and after HTMM for ZFRT(a) and ZF350 (b) thin films. Insets show their  $O_{1\,s}$  spectra. Corresponding LTMM, M-T curves (5 – 300 K) for ZFRT (c) and ZF350 thin films(d) at 5 kOe before and after HTMM.

Fe<sup>+2</sup> ions in our Zn-ferrite thin films. In the literature, Zn-ferrite thin films grown with very low oxygen content [10] resulted in the formation of mixed Fe<sup>2+</sup>/Fe<sup>3+</sup> valence phases [Zn<sub>x</sub><sup>2+</sup>Fe<sup>3+</sup><sub>1×x</sub>]<sub>A</sub>[Fe<sup>3+</sup><sub>1+x</sub>Fe<sup>2+</sup><sub>1×x</sub>]<sub>B</sub>O<sub>4</sub>, or even of binary oxides (Fe-O and Zn-O) during annealing at high temperature, which seriously jeopardizes the stability of the Zn-ferrite thin films [41,42]. Thus, the present results are significant because our Zn-ferrite thin films only lose their unique magnetic and electric properties at elevated temperatures beyond the requirements of SP and SSE applications ( $\geq$  800 K). As confirmed by our DFT calculations, discussed in the later section, defect gap states can be induced by the presence of oxygen vacancies, which may affect aforesaid properties.

The cation distribution and degree of inversion *x* are influenced by the growth and HTMM conditions, which may modify the content of oxygen vacancies  $\delta$  [42,43]. To understand how these two parameters, affect the electronic and magnetic properties, we performed DFT simulations. Table 3 summarizes the variations of lattice parameters, spin-resolved band gaps, and magnetic couplings as a function of the inversion degree *x*.

First, let us consider the case of the stoichiometric spinel ZnFe<sub>2</sub>O<sub>4</sub>

 $(\delta = 0)$  with a normal cation distribution (x = 0.0), where all Fe<sup>3+</sup> ions occupy the octahedral B sites. As Zn<sup>2+</sup> cations are in a 3d<sup>10</sup> electronic configuration, they display no magnetic moment, and the magnetic interactions thus remain confined within the octahedra ( $J_{AA} = J_{AB} = 0$ ) through their connection via edge-sharing  $O_{2p}\xspace$  orbitals. This leads to weak AFM superexchange couplings ( $J_{BB} < 0$ ) with nearly-90° o Fe<sub>B</sub><sup>3+</sup> –  $O - Fe_B^{3+}$  bridges, as schematically shown in Fig. 4(a). We calculated that the  $J_{\rm BB}$  constants display low values, with a similar order of magnitude varying between -0.60 and -1.49 k<sub>B</sub>K, for interactions between the 1<sup>st</sup>- to 4<sup>th</sup>-neighbor atoms (see Supplementary Material Table S1, which also provides a comparison with the J values calculated by Quintero et al. [44]), before falling to nearly 0 k<sub>B</sub>K. All these interactions having the same sign means that the system undergoes magnetic frustrations and that we can, in consequence, expect that the magnetic ground state of ZnFe<sub>2</sub>O<sub>4</sub> could correspond to a non-collinear magnetic ordering. Using these values, we calculated an order temperature  $T_{\rm N} = 14.8$  K, which matches our experimental measurements. By mapping several collinear orderings, we found that the more stable magnetic configuration corresponds to the antiferromagnetic ordering  $(M_{\rm S} = 0 \,\mu_{\rm B})$  that lowers the symmetries of the system to that of the space group P4122 and in which the antiferromagnetic coupling is set between pairs of ferromagnetically-coupled first-neighbor Fe atoms; this result agrees with the calculations reported by Quintero et al. [45].

In the event of cation inversion ( $x \ddagger 0$ ), even at a very low inversion degree such as x = 0.125, there is a significant alteration in magnetic ordering due to the presence of Fe<sup>3+</sup> and Zn<sup>2+</sup> cations in the A and B sites, respectively. As x varies from 0.125 to 1.0, the number of Fe<sup>3+</sup><sub>A</sub>  $- 0 - Fe^{3+}_{B}$  couplings increase, with a bond angle of nearly 125° and extremely-negative  $J_{AB}$  values. This confirms the expected remarkably strong AFM couplings between A-B sites [46]; approximately 20 times higher than the initial  $J_{BB}$  calculated for x = 0. This also results in a change of the  $J_{BB}$  parameters to positive values, signifying ferromagnetic couplings within the B sublattice, as schematically shown in Fig. 4(b).



**Fig. 4.** (a) Schematic of spin structure of the pure  $ZnFe_2O_4$ , where the magnetic  $Fe^{3+}$  cations inhabit the center of the octahedron, Oh, and couple antiferromagnetically ( $J_{BB} < 0$ ). (b) Schematic spin structure of  $(Zn_{1-x}Fe_x)(Zn_xFe_{2-x})O_4$ , for a degree of inversion. In this structure, the  $Fe^{3+}$  cations can be located in the tetrahedron sites, Td, that strongly couple antiferromagnetically with the  $Fe^{3+}$  ions at Oh sites ( $J_{AB} < 0$ ), thus forcing the  $Fe^{3+}$  ions in Oh sites to be coupled ferromagnetically ( $J_{BB} > 0$ ).

#### Table 3

Calculated variations of pseudo-cubic lattice parameter a, band gap at the Fermi level,  $E_g$ , and total spin magnetic moment,  $M_S$ , as a function of the inversion degree, x. The more stable magnetic ordering calculated with the QuantumATK code is also given, with the corresponding exchange integrals,  $J_{AB}$  and  $J_{BB}$ , between first-neighbor Fe atoms and the resulting Curie temperature,  $T_c^{MFA}$ , calculated within the MFA.

x	a (Å)	E <sub>g</sub> (eV)		Mag. config.	$M_{\rm s}$ ( <sub>µB</sub> /f.u.)	$J_{AB}/k_b$ (K)	$J_{BB}/k_b$ (K)	$T_C^{MFA}$ (K)
		Maj. spin	Min. spin					
0.0	8.518	1.88	1.88	$\left[\uparrow_{8/8},\downarrow_{8/8}\right]_{\rm B}$	0.0	-	-0.60	14.8
0.125	8.518	1.78	1.74	$\left(\downarrow_{1/8}\right)_{\text{A}}\left[\uparrow_{15/8}\right]_{\text{B}}$	8.75	-26.31	0.73	273.7
0.25	8.515	2.03	1.55	$\left(\downarrow_{2/8}\right)_{\rm A}\left[\uparrow_{14/8}\right]_{\rm B}$	7.5	-24.27	8.57	351.3
0.5	8.507	1.88	1.55	$\left(\downarrow_{4/8}\right)_{\rm A}\left[\uparrow_{12/8}\right]_{\rm B}$	5.0	-20.56	9.72	506.7
0.75	8.496	1.96	1.63	$\left(\downarrow_{6/8}\right)_{A}\left[\uparrow_{10/8}\right]_{B}$	2.5	-22.67	8.73	530.5
1.0	8.487	1.98	2.17	$\left(\downarrow_{8/8}\right)_{\rm A}\left[\uparrow_{8/8}\right]_{\rm B}$	0.0	-21.29	8.38	558.2

The ferromagnetic coupling in B-B chains is believed to be due to the dominant AFM coupling within A-B sites [41,47], which overrides the inherent AFM coupling along B-B chains. This transformation leads to the establishment of FiM ordering, accompanied by an increase in both the net  $M_S$  and  $T_C$ . The calculated maximum of the net magnetization, 8.75  $\mu_B$ , is obtained for the lowest value of x = 0.125. It is then followed by a gradual decrease of  $M_S$  when x increases, because of the higher number of Fe<sup>3+</sup> cations in A site, which have spin magnetic moments antiparallel to that of Fe<sup>3+</sup><sub>B</sub>;  $M_S$  becomes zero at x = 1.0, when there is exactly the same amount of Te<sup>3+</sup> cations in A and B sites. In addition to the decrease of the net magnetization when x > 0 increases, we calculated an enhancement of  $T_C$  because of the higher number of Fe<sup>3+</sup><sub>A</sub> -O -Fe<sup>3+</sup><sub>B</sub> bridges, which carry the dominant AFM interaction. This explains why our as-grown *ZF350* and annealed *ZFRT* and *ZF350* (Cycle-I) thin films exhibit large *Ms*.

Furthermore, with an increase in the degree of inversion, the lattice constants are found to be smaller. This phenomenon cannot be explained solely by the simple inversion effect; oxygen vacancies have to be considered. In the absence of oxygen vacancies,  $ZnFe_2O_4$  exhibits an insulating behavior due to its relatively large bandgap, ranging from 1.55 to 1.98 eV (depending on *x*), as shown in Table 3 and represented in the densities of states (DOS) of Figs. 5(a) and 5(b).

We propose now to investigate the effect of a neutral oxygen vacancy (V<sub>O</sub>) on the electronic and magnetic properties. To do so, we performed additional ab initio calculations considering an oxygen-deficient ZnFe<sub>2</sub>O<sub>4</sub> crystal ( $\delta = 0.125$ ) with a partially-inverse spinel structure (x = 0.5): [Zn<sub>0.5</sub>,Fe<sub>0.5</sub>][Zn<sub>0.5</sub>,Fe<sub>1.5</sub>]O<sub>3.875</sub>. Depending on the position of the Vo, two scenarios are possible: the V<sub>O</sub> can be located near a Zn<sub>A</sub> or near a Fe<sub>A</sub> atom in a tetrahedral site. At high temperatures, where the distribution of V<sub>O</sub> is expected to be random, the occurrence of each scenario will mainly depend on the number of Fe atoms in A sites, represented by the inversion degree *x*. In the first case, when V<sub>O</sub> is near a Zn<sub>A</sub> atom, the two electrons will be transferred solely to octahedra and

distributed among Fe<sub>B</sub> atoms, first-neighbors of the vacancy. The corresponding DOS is shown in Fig. 5(c), where two gap states are created near the Fermi level, only in the minority-spin channel, resulting in a decrease of the total spin magnetic moment by  $2 \mu_B$  per V<sub>O</sub>. On the other hand, in the second case, when Vo is near a FeA atom, one electron will reduce the oxidation state of the first-neighbor FeA cation, while the second electron will be distributed onto the nearest Fe<sub>B</sub> ions [20,48] [Fig. 5(d)], leading to the formation of one gap state in each spin channel and leaving the total spin magnetic moment unchanged, if compared to the stoichiometric compound (where  $ZnFe_2O_4$  with x = 0.5is predicted to be FiM). In either scenario (Figs. 5(c) and 5(d)), new defect states give rise to a smaller bandgap, which may explain the lowering of the resistivities by one to two orders of magnitude measured after HTMM in the ZFRT and ZF350 (Cycle-I and Cycle-II) thin films (Table 1). In their numerical study, Quintero et al., moreover predicted that reduced ZnFe<sub>2</sub>O<sub>4</sub> oxides would display higher inversion degree and that the presence of oxygen vacancies strengthens the ferrimagnetic ordering [45].

Based on experimental low- and high-temperature magnetic studies, a magnetic nanophase diagram was constructed for nanocrystalline Znferrite thin films with varying  $T_{\rm S}$  at fixed magnetic field of 0.5 kOe, which indicates different dominant magnetic behaviors, as shown in Fig. 6. Of course, these correspond to prevailing magnetic phases only, and do not imply full homogeneity in the samples. By increasing  $T_S$ , the Zn-ferrite thin films undergo various magnetic states, transitioning from cluster-glass (CG) to FiM and eventually reaching a PM bulk state. Consequently, different magnetic ordering temperatures ( $T_C$ ,  $T_N$ , and  $T_G$ ) can be obtained in Zn-ferrite thin films due to the distribution of nano-grains produced at different  $T_S$ . The growth of stoichiometric Znferrite thin films through physical vapor deposition at low  $T_S$ (<500 °C) often leads to the formation of disordered ZnFe<sub>2</sub>O<sub>4</sub>. This disorder, characterized by the inversion parameter x, arises from the quenching of randomly distributed  $Zn^{2+}$  and  $Fe^{3+}$  vapor phases. It results in a strong  $J_{AB}(Fe^{3+}-Fe^{3+})$  AFM interaction. However, at low



**Fig. 5.** Densities of states (DOS) calculated for bulk  $ZnFe_2O_4$  with (a) a normal spinel structure (x = 0) and an AFM ordering, and (b,c,d) a partially inverse spinel structure (x = 0.5) and a FiM ordering. In (c) and (d), an oxygen vacancy has been introduced by removing 1/32 oxygen atoms. In (c), the tetrahedral site, first neighbor of the oxygen vacancy, is occupied by a Zn atom, while in (d) it is filled by a Fe<sub>A</sub> atom. Atom contributions to the DOS are represented in colors; light-green DOS highlight the presence of defect states. Positive and negative DOS are for majority and minority spin projections, respectively. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)



Fig. 6. Magnetic phase diagram describing various dominant magnetic phase transitions with changing  $T_{\rm S}$  from RT to 850 °C at fixed magnetic field of 0.5 kOe.

temperatures these thin films exhibit CG behavior, as confirmed by the de Almeida-Thouless relationship between  $T_G$  and the applied field H [49]. In the case of low  $T_S = RT$ , the  $T_G$  value reached approximately 100 K. The origin of the dominant high- $T_G$  CG behavior could be attributed to spin frustration in smaller grain-sized thin films, resulting from the competition between AFM interactions (Fe<sub>A</sub>-Fe<sub>B</sub> and Fe<sub>B</sub>-Fe<sub>B</sub>) within the grains, as well as the competition between AFM and FiM grains. Additionally, the presence of random oxygen deficiency may affect the overall  $J_{AB}$  and  $J_{BB}$  interactions. However, in higher  $T_S$  cases  $(500 - 850 \degree C)$ , the cation inversion begins to revert back to its original bulk  $[Zn_A^{2+}Fe_{2B}^{3+}O_4]$  structure, wherein the weaker  $J_{BB}$  AFM interaction dominates below  $T_{\rm N} = 15$  K. The  $T_S$  window (100 - 450 °C) exhibits interesting dominant FiM properties because of strong JAB coupling, with higher  $T_C$  values (350 – 650 K). Thus, the constructed magnetic phase diagram may hold significant implications for integrating various Zn-ferrite thin films of different properties into lattice compatible homo-spintronic architectures such as magnetic tunnel junctions including spin-filtering barriers, exchange bias (FM/AFM) structures, opto-magnetic switches and next-generation devices based on the use of pure spin currents. However, operating temperatures of these devices should not be higher than the transition temperatures  $T_C$ ,  $T_N$ , and  $T_G$  of the Zn-ferrite thin films, which, as demonstrated in this study, can be engineered by an appropriate choice of growth conditions.

#### 4. Conclusions

The current study presents the dependence of the structural, electronic, and magnetic properties of  $ZnFe_2O_4$  nanocrystalline films on their deposition temperature and subsequent thermal treatment, using a combination of experimental and theoretical methods. The most interesting findings include the spontaneous magnetization in most samples, with tunable average Curie temperatures ranging from 425 to 710 K (especially after annealing up to 1000 K), in contrast to the bulk antiferromagnetic order. X-ray photoemission spectroscopy and low-temperature magnetization measurements reveal no evidence of Fe<sup>2+</sup> but indicate spinel inversion, where Fe<sup>3+</sup> occupy the tetrahedral sites. Raman spectra analysis further supports the nature of cationic disorder through the weighting of vibration modes of tetrahedra FeO<sub>4</sub> over octahedra FeO<sub>6</sub>. DFT calculations suggest that structural inversion and oxygen vacancies can give rise to ferrimagnetism and slightly reduced resistivity values by forming Fe-3d gap states near the Fermi level. A

proposed magnetic nanophase diagram elucidates the dominant competing magnetic ground states (cluster spin-glassy state, ferrimagnetic, and antiferromagnetic) with increasing growth temperature. This exploration sets the stage for the development of innovative homo-architectures utilizing multiple  $ZnFe_2O_4$  thin films with diverse spin-insulator functionalities.

#### Author agreement statement

We the undersigned declare that this manuscript is original, has not been published before and is not currently being considered for publication elsewhere. We confirm that the manuscript has been read and approved by all named authors and that there are no other persons who satisfied the criteria for authorship but are not listed. We further confirm that the order of authors listed in the manuscript has been approved by all of us. We understand that the Corresponding Author is the sole contact for the Editorial process. He/she is responsible for communicating with the other authors about progress, submissions of revisions and final approval of proofs.

#### CRediT authorship contribution statement

Bohra Murtaza: Writing – original draft, Methodology, Investigation, Formal analysis, Data curation, Conceptualization. Arras Rémi: Investigation, Formal analysis. Singh Vidyadhar: Methodology. Singh Nitesh: Investigation. Annadi Anil: Writing – review & editing. Toulkeridou Evropi: Methodology, Formal analysis. Grammatikopoulos Panagiotis: Writing – review & editing. Chou Hsiung: Writing – original draft, Formal analysis.

#### **Declaration of Competing Interest**

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests: We do not have any conflict of interest. If there are other authors, they declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

#### Data availability

The data that support the findings of this study are available from the corresponding author upon reasonable request.

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#### Appendix A. Supporting information

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.mtcomm.2024.108333.

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