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1 **Comparison of hydration water properties of common and durum wheat**
2 **brans upon grinding with different loading modes**

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ABSTRACT

Wheat bran brings healthy properties in food products. However, its incorporation requires a first milling step during which it is subject to various loading modes which have an influence on its properties. This study investigated the influence of the loading modes (high shear or impact) generated by grinders during the milling on the hydration properties of common and durum wheat brans. An original study at molecular scale to target the distribution and intensity of hydration water bonds was carried out by gravimetric and spectroscopic methods. Results were analyzed in regards to the particle size distribution and shape factors as well as biochemical composition to highlight the process-structure-function relationships at macro and microscales. Impact grinder has a stronger effect on water vapor sorption capacity and FTIR multimer water at 3600 cm^{-1} of durum wheat bran as well as a red shift in the band at 1030 cm^{-1} related to a high decrease in residual starch crystallinity degree (about 17%). High shear grinder tends to increase the proportion of water strongly bound. Low field NMR analysis revealed differences in the high mobility water peak and a significant lower relaxation time T2 for native and ground durum wheat bran (up to 1.7 fold less).

Keywords: Mechanical loading mode, grinding, water mobility, bran structure.

42 **Abbreviations**

ATR-FTIR	Attenuated Total Reflectance-Fourier Transform Infrared spectroscopy
NMR	Nuclear magnetic resonance
BET	Brunauer-Emmett-Teller
GAB	Guggenheim, Anderson and de Boer
SEM	Scanning electron microscopy
FID	Free induced decay

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1. INTRODUCTION

58 Wheat bran is a by-product of conventional wheat milling which is currently undervalued as only
59 10% is currently used for food applications (Onipe et al., 2015). It has been demonstrated that bran
60 consumption has a positive effect on health by decreasing the risk of cardiovascular diseases (Wu
61 et al., 2015), increasing fecal bulk and reducing intestinal transit time (Stevenson et al., 2012).
62 These physiological benefits are often related to the strong water binding capacity of the bran.

63 To incorporate bran into food product, wheat bran is often reduced into the form of a fine powder
64 (Onipe et al., 2015) involving a further fine milling step. This step has to face the difficulty of
65 wheat bran being a multilayered material made-up of different tissues including the outer pericarp,
66 the inner pericarp, the hyaline band and testa, the aleurone layer and residual starchy endosperm.
67 These different tissues have distinct compositions, nutritional and functional properties and
68 contrasting mechanical properties (Ciccoritti et al., 2017). During the milling, the bran is subjected
69 to mechanical stresses generated by the grinder such as compression, impact, friction, shear and
70 extension, (Mayer-Laigle et al. 2018) and can undergo chemical or physicochemical that influence
71 its hydration properties and so the rheological properties of the dough, and the quality of the final
72 food products in which the bran is incorporated (Deroover et al. 2020). A better understanding of
73 the influence of the milling step in ground bran hydration properties could bring new insight for a
74 better processing of healthy food products. It is the aim of the present study in which the effect of
75 different mechanical loading modes (impact or shear) on the hydration water properties of common
76 and durum wheat brans during milling are compared.

77 Hydration properties are often assessed in a functional way by swelling, water retention, water
78 absorption and porosity. These methods are quick but do not allow to evaluate the binding strengths

79 between water and the bran matrix. Other methods based on thermogravimetric analyses have also
80 been reported to quantify the weakly bound water in bran (i.e in between the particles and the
81 micropores of empty pericarp cells) and the strongly bound water in bran (i.e bound in cell wall
82 nanopores and through hydrogen bonding) (Jacobs et al., 2015). In this study, we describe
83 hydration properties of ground brans by a functional approach (water vapor sorption isotherms)
84 and a molecular approach (using FTIR and low field NMR analysis respectively). This analysis is
85 carried out in close relation to mechanical loading of the grinders and the intrinsic properties
86 (morphology, microscopic structure and biochemical composition) of each type of bran.

87 **2. MATERIALS AND METHODS**

88 **2.1. Raw materials**

89 The wheat bran was produced by a French industrial miller from Fructidor variety, harvest in 2019.
90 The durum wheat bran was produced by a French industrial miller (Panzani group) from a mixture
91 of different durum wheat grain varieties (Miradou, Anvergur, Voilur, Claudio) without prior
92 dehulling.

93 **2.2. Grinding operations**

94 The two native brans were ground using two grinding technologies in two different configurations:
95 An impact mill (Hosokawa-Alpine, type 100 UPZ, Augsburg, Germany), with a selection grid of
96 500 or 300 μm operating at a speed of 18 000 rpm. Milling was carried out at ambient temperature.
97 In these configurations impact is the predominant loading mode.
98 An ultra-centrifugal mill ZM 200 (Retsch, Germany), a high speed rotor mill operated at a speed
99 of 18000 min^{-1} , with a selection grid of 500 or 250 μm . In this device shear is the predominant
100 loading mode but impact is also present to a lesser extent. Prior to the milling raw bran was frozen

101 and during the grinding a cyclone was used in order to additionally cool the sample by the air
102 stream and favor the discharged from the grinding chamber.

103 **2.3. Bran characterization**

104 2.3.1. Particle size measurement by laser diffraction

105 The particle size distribution of the ground samples was measured by laser diffraction using a
106 Mastersizer 2000 equipped with a Scirocco 2000 dry dispersion unit Scirocco 2000 (Malvern
107 Instruments, Worcestershire, UK). The particle size distribution is expressed as % of total particle
108 volume. The median particle size (D_{50}) and span ($D_{90}-D_{10}/D_{50}$) were used as the primary
109 descriptors. This measurement was performed only on ground material since the size of initial
110 brans was outside the measurement range of the device. All measurements were performed in
111 triplicate.

112

113 2.3.2. Morphological characteristics of starting material

114 Morphologi 4 (Malvern, United Kingdom) provides detailed morphological descriptions of
115 particulate samples through static image analysis. Native bran particles were either manually
116 dispersed (common wheat bran) or automatically with a sample dispersion unit (durum wheat bran)
117 on a glass plate. Results are expressed by combining data over 5 plates to have a greater
118 representativeness of the sample (with a total number of about 71 000 and 330 000 particles for
119 common and durum wheat bran, respectively). Optic with a magnification of 2.5 (recommended
120 for particle size range of 8.5 - 1300 μm) was selected. The device captures a two dimensional
121 image of the particles and circle equivalent diameter (diameter of a circle which has the same area
122 as the particle) and high sensitivity circularity (HS : ratio of the particle's projected area to the

123 particle's perimeter squared (eq. 1) were calculated from the image. HS circularity expresses the
124 surface roughness of the particle. A value of 1 corresponds to a perfectly smooth spherical particle

$$125 \quad \text{HS circularity} = 4\pi \text{Area} / \text{Perimeter}^2 \quad (\text{eq. 1})$$

126 The distributions were expressed in volume. Some samples were also analyzed using an SEM
127 (Model Phenom, Thermo Fisher Scientific, US) with an accelerating voltage (10 kV).

128

129 2.3.3. Content in dry matter, ash, starch and damaged starch

130 The dry matter and ash contents of the native brans and ground fractions were determined
131 according to approved AACC methods 44-19 and 08-12 respectively. Total starch content of
132 fractions was measured in duplicate using Megazyme kit (Megazyme International Ireland Ltd.,
133 Ireland) according to approved AACC method 76-13. Damaged starch was also determined on
134 initial and ground fractions with a Megazyme kit (K-SDAM starch damage assay kit, Megazyme
135 Int., Ireland) according to method AACCI N76-31.01.

136 2.3.4. Monosaccharides content

137 The polysaccharides were depolymerized before monosaccharides being quantified individually.
138 The protocol from Barteau et al. (2021) was followed using inositol as an internal standard. After
139 an acidic hydrolysis step, neutral monosaccharides were analyzed as their alditol acetate
140 derivatives by gas chromatography-flame ionization detection (GC-FID). Standards of
141 carbohydrate solutions were used for calibration. Analyses were performed in three independent
142 assays. The total monosaccharide content is the sum of each monosaccharide amount and is
143 expressed as the percentage of the dry matter mass.

144 2.3.5. Uronic acids content

145 Quantification of uronic acids was performed after wheat bran acid hydrolysis according to
146 Beaugrand et al. (2004a). Uronic acid in acid hydrolysates were then quantified using the
147 methahydroxydiphenyl colorimetric method (Blumenkr and Asboehan, 1973). All tests were done
148 in triplicate.

149

150 2.3.6. Lignin content

151 The lignin content was determined from the powdered samples by the Klason method. For this
152 gravimetric method, the lignin as the acid insoluble residue was determined using the non-
153 hydrolyzable acid residue remaining after sulfuric acid hydrolysis (Monties, 1984).

154

155 2.3.7. Water vapor adsorption isotherms

156 Bran samples were dried to constant weight over phosphorus pentoxide in a vacuum climatic
157 chamber at 40°C, then placed over saturated salt solutions (NaCl, K₂SO₄, KCl, KOH, NaBr,
158 K₂CO₃) in desiccators at constant temperature (25°C) to provide different water activities (0.08,
159 0.43, 0.57, 0.75, 0.85, and 0.97, respectively). Final water contents were determined when
160 moisture equilibrium had been obtained (after 3-6 weeks). The water content of the samples (g
161 water / 100 g dry solids) was plotted against relative humidity for the two different native brans
162 and the different ground brans. Experiments were conducted in duplicate. Guggenheim, Anderson
163 and de Boer (GAB) equation was used to model water sorption curves (eq.2):

164

$$165 \quad W = \frac{(m_0 C k a_w)}{(1 - k a_w)(1 - k a_w + C k a_w)} \quad (\text{eq.2})$$

166 Where W (g water / g dry matter) is the water content of the sample; a_w is the water activity; m₀
167 (g water / g dry matter) is the water content corresponding to the saturation of all primary

168 adsorption sites by one water molecule; C is the Guggenheim constant. k is the factor correcting
169 properties of the multilayer molecules with respect to the bulk liquid. The experimental data were
170 fit with the software Microsoft Excel 2016 by using the method of the least squares methods.
171 The solid surface area (A) of samples can be calculated from the monolayer water content (M_0)
172 according to eq. 3 (Mazza and Le Maguer, 1978):

$$173 \quad A = M_0 \left(\frac{1}{M_{H_2O}} \right) (N) (A_{H_2O}) = (3.5 \times 10^6) M_0 \quad (\text{eq.3})$$

174 Where A is the solid surface area (m^2/kg solid); M_{H_2O} is the molecular weight of water molecule
175 (18 kg/kmole); N is the Avogadro's number (6×10^{26} molecules/kmole); A_{H_2O} is the area of a water
176 molecule (10.6 \AA).

177 The quality of the fit of the GAB model was judged from the value of the root-mean-square error
178 (RMSE) calculated as follows:

$$179 \quad RMSE = \sqrt{\sum \frac{[W_i - W_{i*}]^2}{N}} \quad (\text{eq.4})$$

180 with N : the number of experimental points; W_i : the average experimental water content; W_{i*} : the
181 calculated water content.

182

183 2.3.8. ATR-FTIR spectra

184 The FTIR spectra were collected in the $400\text{-}4000 \text{ cm}^{-1}$ wavenumber range on a Bruker Vertex 70v
185 Fourier Transform spectrometer (Bruker Optik GmbH, Germany) operating with a Globar source
186 in combination with a KBr beamsplitter and a DigiTect DLaTGS detector with integrated
187 preamplifier. The optical cell was a Golden Gate diamond ATR system. The spectra were recorded
188 with a resolution of 4 cm^{-1} , automatically adding 128 repetitive scans. Scans were corrected for
189 the air contribution and were preprocessed using OPUS software (version 7.0) by performing a

190 baseline correction, a 9 points smoothing and a vector normalization for taking into account the
191 effective number of absorbers. Experiments were conducted on powders of native and ground
192 brans previously equilibrated at 40 or 58% relative humidities in specific climatic chambers (at
193 25°C). The spectra of freeze dried brans were subtracted following the direct difference method
194 (Poole and Finney, 1982) in order to analyze the band of hydration water. In order to better explore
195 the impact of grinding load mode on hydration water, we recorded FTIR spectra of initial and
196 freeze dried brans and calculated the following spectra ($A_{\text{initial bran}} - A_{\text{freeze dried bran}}$). The resulting
197 spectrum is supposed to correspond to interfacial hydration water.

198 ATR-FTIR spectra were also analyzed to determine the crystallinity of starch granules. The bands
199 in the region 1100–900 cm^{-1} have been shown to be sensitive to changes in starch structure, in
200 particular bands at 1000, 1022 and 1047 cm^{-1} have been widely studied. The band around 1022
201 cm^{-1} seems to increase in more amorphous samples, while the bands around 1000 and 1047 cm^{-1}
202 become more defined in more crystalline samples.

203

204 2.3.9 Low field NMR measurements

205 Transversal relaxation times (T_2) were obtained from time domain nuclear magnetic resonance
206 (TD-NMR) measurements performed using a Minispec mq spectrometer (minispec mq20, Bruker,
207 France) operating at a frequency of 20MHz (0.47 Tesla). A Free Induced Decay (FID) Carrer-
208 Purcell-Meiboom-Gill (CPMG) sequence with an acquisition time of 0.15 ms and 16 scans were
209 used for the FID signal, and, a recycle delay of 2s was used for the CPMG signal. Bran samples
210 were weighted and placed in climatic chambers at 25°C under controlled relative humidity of 58%
211 until mass stabilization. Samples were carefully introduced into Bruker NMR tubes and sealed to
212 prevent water evaporation. Less mobile proton populations, i.e. those having T_2 relaxation times

213 between 0.014 ms and 0.2 ms, were analyzed with the FID measurements, whereas more mobile
214 protons, i.e. those having T2 relaxation times between 1 ms and 60 ms, were analyzed using the
215 CPMG pulse sequence. The amplitude of proton populations, proportional to their relative
216 quantities is expressed in arbitrary unit and normalized per sample's weight in gram. In the results
217 section, where the proton distributions of the samples are shown, a representative continuous
218 proton distribution of the triplicate measurements was selected. Data were fitted using SigmaPlot
219 (version) and a combination of Sinus cardinal and exponential functions according to the relation:

$$220 \quad y(t) = A_i e^{(-t/T_{21})^2} \frac{\sin(B*t)}{B*t} + \sum_{i=0}^n A_i e^{-\frac{t}{T_{2i}}} \quad (\text{eq.5})$$

221 with, A the amplitude; T2, the transverse relaxation time; B, a constant; and t, the time.

222

223 **2.4. Statistical analysis**

224 The statistical significance of results was assessed using Tukey's HSD test. Multiple comparisons
225 were performed by calculating the least significant difference by using the Statgraphics Stratus
226 software. All tests were conducted at 5% significance level.

227

228 **3. RESULTS & DISCUSSION**

229 **3.1. Particle size distribution and shape factor**

230 The native brans are both characterized by monomodal distribution of particle diameters but the
231 common wheat bran has significantly higher values of median diameter and span, than the durum
232 wheat bran (figure 1 and table 1).

233

FIGURE 1.

234 HS circularity values are higher for durum wheat bran (0.36) than for common wheat bran (0.29)
235 but are both lower than the disk reference value equal to 1. The difference of circularity, highlights

236 more irregular surface and shape for common wheat bran probably related to difference in the
237 mechanical properties of the bran.

238 **TABLE 1.**

239 Grinding operation of brans induced a significant decrease in median diameter and an increase of
240 the span. The particle size of ground durum wheat bran remains always lower than that of common
241 wheat bran, independently of grinder's loading mode. This difference of size among both brans
242 suggests a higher stiffness at ambient conditions, for durum bran in comparison to common bran
243 which leads, for the same amount of milling energy, to smaller particle size.

244 The particle size distribution of the ground bran are no longer monomodal (Figure 1) and 2
245 populations of particles may be observed, with the presence of fine particles with diameters less
246 than 100 μm (table 1). These fine particles could be attributed to starch granules as it can be seen
247 from their circular shape and size domain around 40 μm (table 1 and figure 1-A1 to B3).

248 A data analysis was performed to determine the proportion of the 2 populations in ground bran
249 particles and their characteristic criteria (Table 1) by deconvolution of bands into fine or large
250 particles populations. Significant differences in terms of proportions, median diameters and span
251 values are observed between the samples. Compared to the impact mill, the centrifugal mill
252 generates, irrespective of the wheat bran type and the mill size grid setting, a higher proportion of
253 fine particles with a small median diameter, and coarse particles with smaller median diameter.

254 Irrespective of mill grinding mode and wheat bran type, finer milling (grid of 250 or 300 μm)
255 produces a higher proportion of fine particles with a small median diameter and coarse particles
256 with a smaller median diameter.

257 The span of each population (i.e small particles or coarse particles) is not significantly impacted
258 by wheat type, mill type and settings. Cadden (1987) reported upon roller grinding of wheat bran

259 that the wheat bran cell wall is disrupted upon grinding and the matrix structure is more prone to
260 collapse affecting thus the porous structure. This effect on the matrix structure was confirmed by
261 SEM (figure 1). Ground samples appeared as small fragments of few structures that could be
262 recognized. The shearing forces in centrifugal grinder, applied tangentially to fibers, seem to limit
263 the collapse of the porous structures.

264

265 **3.2. Composition**

266 The composition of native and ground durum wheat or common wheat brans is reported in table
267 2. The two types of brans have initial water contents in the same range. They differ by their starch
268 content more than two time higher in the case of durum wheat.

269

TABLE 2.

270 Both brans also exhibit differences in terms of neutral monosaccharides as reported in table 2.
271 Durum wheat bran has a slightly higher content of neutral monosaccharides than common wheat
272 bran, due to a more important rate of glucose. This high value of glucose in durum wheat bran is
273 also related to the higher content of starch. Cell walls of bran are known to have glucose monomers
274 as a mix of two forms, β -glucan in the aleurone, and cellulose, mostly located in the inner and
275 outer pericarp (Hemery et al., 2007) in proportion of about 3 to 4 more cellulose than β -glucan.
276 The other major difference is in the xylose and arabinose content with common wheat bran being
277 richer in both constituents suggesting a higher content of arabinoxylan (Beaugrand et al., 2004b).
278 The value of the A/X ratio (arabinose/xylose) reflects the substitution degree and so the relative
279 enzymatic recalcitrance of arabinoxylan (Beaugrand et al., 2004c). The values of 0.63 and 0.55
280 obtained for durum and common wheat, respectively suggest that the arabinoxylan of durum wheat
281 is more branched. Interestingly, durum bran has also more glucuronic acid (table 1), known to
282 participate as arabinose do to the branching of the xylan backbone (Beaugrand et al., 2004c),

283 revealing definitively that durum hemicellulose is more branched. In this study, the galacturonic
284 acid (GalA) content generally associated with rhamno-galacturonan I (RGI) pectins is also higher
285 for durum wheat bran, suggesting a higher pectin level in durum wheat bran. If this amount of
286 GalA can be seen as minor (about 0.5 %) in term of the total dry mass, this is significantly enough
287 to have an impact on the functional properties significant and in this study a significantly higher
288 amount in durum wheat bran is quantified compared to common bran (about 10 time more).
289 Regarding the variation of the composition with the grinding mode, the strongest difference is
290 observed in the damaged starch content especially for durum wheat bran. A significant increase of
291 45% and 75% occurs in the case of impact and centrifugal grinders respectively for the lowest
292 grinder grid of 250/300 μm . In comparison, the increasing levels are 28% and 42% in the case of
293 common wheat bran.

294 *Discussion* - The structural composition of the durum cell walls seems more complex and further
295 suggests a reinforced protection structure of physical cell walls. Indeed, the glucuronoarabinoxylan
296 seems more branched (table 2) which can help crosslinking between polysaccharide chains.
297 Regarding pectins, we can hypothesize that RGI side chains can act as plasticizers in cell walls
298 that undergo large physical remodeling. Regarding the global composition, similar compositions
299 were reported for wheat bran with around 55% of the dietary fibre in wheat bran being
300 arabinoxylan, while the remaining was cellulose (9-12%), lignin (3-5%), fructan (3-4%) and mixed
301 linked β -glucan (2.2-2.6%) (Chalamacharla et al., 2018). The reported ash content of wheat bran
302 samples was 5.5-6.5%, in line with our values. In the pericarp-seed coat, the higher amount of
303 galactose in common wheat bran suggests the occurrence of complex heteroxylans, previously
304 identified in wheat bran (Brillouet and Joseleau, 1987). Lignin content measured by Klason's
305 method (table 2) indicates that common wheat bran displays a slightly higher content compared to

306 durum. Lignin distribution has been reported to affect the position of fracture within various cell
307 wall layers (Donaldson, 1995). Increased levels of damaged starch upon grinding is in accordance
308 with previous works as in the case of wheat flour, starch damage degree depending on raw
309 materials, particle coarseness and milling conditions (Wang et al., 2020). The impact of these 3
310 factors was observed here, as shear centrifugal grinders induced the most damaged starch, at lowest
311 grid size for durum wheat bran.

312 **3.3. Water vapor adsorption isotherms**

313 Water vapor adsorption isotherm of the two native brans (common and durum wheat) display a
314 classical sigmoidal shaped curve (fig.A.1A).

315 The GAB model fits well the experimental data ($R^2 > 0.995$) for all samples. The values obtained
316 from the fit are reported in table 3. The water vapor adsorption ability of common wheat bran
317 water is more pronounced than the lower-sized durum wheat bran and differences are more
318 pronounced in the intermediate “multilayer region” as shown by the higher values of solid surface
319 area, A, for common wheat bran (table 3).

320 **TABLE 3.**

321 Solid surface area refers to the total surface area available for hydrophilic binding. It differs from
322 “theoretical surface area” by taking material porosity as well as particle size reduction into account.
323 The calculated values of the GAB model for the monolayer content M_0 for bran were reported to
324 be around 6.6-6.8 g/100 g dm (Li et al., 2021). In our case, these monolayer values were lower for
325 native durum wheat bran indicating lower water adsorption capacities under low relative humidity
326 conditions. The chemical composition is known to influence its water vapor adsorption capacity
327 (Li et al., 2021) and the water holding capacity was positively correlated with the amount of
328 insoluble noncellulosic polysaccharides, and negatively with cellulose and lignin (Dural and

329 Hines, 1993). The transposition to our results must be done with care as the water holding capacity
330 is based on the addition of water in liquid form while water vapor sorption isotherms are based on
331 equilibration in a controlled water vapor atmosphere. With water vapor adsorption isotherms, the
332 direct interactions of wheat bran with the water molecules can be estimated, while with liquid
333 addition only the ability to trap large amounts of water inside macromolecular complexes is taken
334 into account. Our results suggest that the ratio of cellulose-lignin content to hemicellulose content
335 is negatively correlated with its water sorption capacity. The comparative analysis of lignin
336 revealed a comparable amount in common and durum wheat brans. In addition, the higher amount
337 of hemicellulose and pectin in common wheat bran and on another hand the higher amount of
338 glucose-polymer for durum wheat bran are experimental proofs that confirm the coherence of a
339 higher water sorption capacity for common wheat bran.

340 Grinding tends to decrease the M_0 values as well as the surface area of common wheat brans
341 powders. It tended to increase the value of the k constant in the GAB equation for common wheat
342 bran (Table 3). Similar trend was reported for ultrafine grinding of wheat bran (Li et al., 2021).
343 This was not that obvious in the case of ground durum wheat bran. The centrifugal grinder with a
344 predominant mechanical shear stress maintains the same initial differences observed between the
345 two types of bran. The impact grinder with the grid size of 500 μm has the most effect upon
346 grinding of durum wheat bran reducing the water vapor adsorption differences (fig.A.1B) with
347 solid surface area, M_0 , c and k . For all other milling conditions, these values remain in similar
348 range for both types of brans.

349 Since milling may cause the damage of hydrogen bonding (Wang et al., 2020), the exposition of
350 hydroxyls can be modified upon grinding and may explain the difference observed during impact
351 milling of durum wheat bran. This is the case for impact milling and durum wheat bran in our case.

352 Discussion - Durum wheat bran appears to be more prone to variations in its sorption behavior as
353 function of mechanical loading mode (shearing – centrifugal mill or impact- impact mill) than
354 common wheat bran. As discussed previously, these variations are more pronounced with impact
355 grinders, with an increase of sorption aptitude independently of the grid size highlighted by a
356 higher variation of M_0 and A_0 values for durum wheat bran in comparison with common wheat
357 bran (table 3). Dural and Hines (1993) indicated that the differences in particle size between the
358 ground sample (related to the specific surface area) do not allow to explain the variations of the
359 sorption behavior. It can be noted in the SEM images (figure 1) that the structure of durum wheat
360 bran seems more porous than that of durum bran. We also pointed out earlier that the bran structure
361 collapsed more with impact milling than with shear milling. This modification of durum wheat
362 bran structure during impact milling may explain the difference observed in water vapor adsorption
363 behavior.

364

365 **3.4. ATR-FTIR measurements**

366 ATR-FTIR experiments were conducted on native and ground wheat brans in order to estimate the
367 impact of grinding loading mode on hydration water structuring. In particular, we attached to
368 differentiate the structure and dynamics of bulk water from that of hydration water (Bellissent-
369 Funel, 2001), which may result in a three-dimensional transient network in which bonds are
370 constantly breaking and reforming. This approach is not much reported in literature for granular
371 media such as wheat bran.

372 We focus on the O-H stretching mode $\nu(\text{OH})$ ($3700\text{-}3000\text{ cm}^{-1}$) that is especially sensitive to the
373 surrounding hydrogen bond environment and may provide insights on the structure of hydration
374 water (Grossutti et Dutcher, 2016). There exists a variety of water molecules that are differently

375 H bonded to each other, with H-bond coordination numbers ranging from 0 to 4. The low-
376 wavenumber oscillators (centered around 3310 cm^{-1}), called hereafter "network water" (NW),
377 correspond to strongly H-bonded water molecules having a coordination number close to four. The
378 high-wavenumber component (around 3600 cm^{-1}), called "multimer water" (MW), is assigned to
379 water molecules with a small number of H-bonds to other water molecules (giving dimers or
380 trimers). In between lies the component centered at 3450 cm^{-1} corresponding to "intermediate
381 water" molecules (labelled IW) (Enev et al., 2019).

382 Figures 2A-B show that the two types of brans differ in the $3700\text{-}3100\text{ cm}^{-1}$ region with native
383 common wheat bran having higher absorbance intensity for the 3200 cm^{-1} band, corresponding to
384 strongly bonded water molecules.

385 **FIGURE 2.**

386
387 Figures 2A-B show that multimer water at 3600 cm^{-1} in the case of durum wheat (fig 2A) is more
388 disturbed by the grinding loading mode than in the case of common wheat (fig 2B). The absorbance
389 of this band decreases mostly with the particle size (grinding grid of $300\text{ }\mu\text{m}$) for the centrifugal
390 shear grinder and we observed a clear redistribution of the 3 water sub-populations: ordered at
391 3200 cm^{-1} , intermediate at 3400 cm^{-1} and disturbed multimer water at 3600 cm^{-1} . In addition, there
392 is a 17 cm^{-1} shift for the multimer band after all grinding steps implicating a change in interaction
393 energy of hydration water upon grinding. This significant decrease in band's wavenumber is
394 indicative of an increase in the number and strength of hydrogen bonds upon durum wheat bran
395 grinding (Enev et al., 2019). It was reported in literature that this high frequency band tends to be
396 displaced and less broad in the case of pore size reduction (Le Caër et al., 2011). This is noted in
397 figure 2A for the smallest particles (grid sizes of 250 and $300\text{ }\mu\text{m}$). The gain in the intermediate

398 water zone (at 3400 cm^{-1}) occurs therefore to the detriment of network water (NW) molecules (at
399 3200 cm^{-1}) for durum wheat bran upon grinding.

400 In the case of common wheat bran the major difference upon grinding is the reduction of the low
401 wavenumber component's intensity at 3200 cm^{-1} , indicating an increase in the number of network
402 water molecules. The spectral parameter R_{multimer} ($R_{\text{multimer}} = A_{3600}/A_{3200}$) was calculated for each
403 spectrum according to Grossutti and Dutcher (2016). This parameter presented in figure 2C-D
404 provides information about the relative population of water molecules in a disrupted hydrogen
405 bond network to those in a well-ordered hydrogen bond network. The spectra were recorded for
406 bran samples equilibrated at 40 or 58% of relative humidity, as to be in the water vapor adsorption
407 isotherms zone where more differences among brans were revealed. R_{multimer} varies more steeply
408 among the 2 types of wheat bran at 40% relative humidity (figures 2C-D), indicating the significant
409 impact of early stages of water sorption. Farshchi et al. (2019) demonstrated for spray dried
410 detergent powders that a relative humidity increase induces a higher occupation of polar available
411 sites by water molecules which in turn increases the order of the system and mechanisms of water
412 sorption. At 58% relative humidity, we can assume the formation of subsequent layers of
413 adsorbate. At low relative humidity, the mechanism of surface adsorption related to the powder
414 micro-structure may be predominant. However, at medium and high relative humidity values, bulk
415 water sorption mainly predominates.

416 Discussion - The R_{multimer} is more elevated for durum wheat bran at a larger particle size ($500\text{ }\mu\text{m}$
417 for both impact mill and centrifugal mill). This is mostly related to the initial differences in raw
418 material properties rather than size reduction or loading mode. Differences are more marked for
419 the centrifugal shear miller as previously observed on the adsorption isotherms. The difference in
420 the evolution of the R_{multimer} between the 2 types of bran suggests potential changes in

421 physicochemical and hydration properties upon milling. The functional groups and/or their
422 accessibility can be altered by the grinding loading mode impacting the network water component
423 situated at low wavenumbers. Regarding damaged starch, durum wheat bran had the highest values
424 initially and after all grinding processing (impact and centrifugal shear grinders at both grid sizes)
425 (table 2). Durum wheat bran ground with UPZ grinders expressed the greater reduction of the
426 1030:997 cm^{-1} ratio band (a reduction of about 17% at both grid sizes *vs* 1.5% for common wheat
427 bran) and an additional red shift of 8-12 cm^{-1} in the band at 1030 cm^{-1} upon grinding (*vs* no shift
428 for common wheat bran). This is in line with water vapor adsorption isotherms trends for durum
429 wheat bran. Roa et al. (2014) also demonstrated that ball-milling treatment of amaranth grain
430 significantly decreased the intensity ratios of the band 1039 and 1014 cm^{-1} corresponding to the
431 crystalline/amorphous part of starch structure.

432 The centrifugal shear miller increases the FTIR signal associated with the relative content of the
433 water strongly bound in the hydration layer of bran: less water molecules are accessible to network
434 water-water interactions. This effect seems more pronounced for the durum wheat bran with the
435 centrifugal miller and a grid of 500 μm . Rupturing of the cell walls in bran through the actions of
436 impact forces and friction (close to surface shear) upon milling might result in cleavage of covalent
437 bonds (Jacobs et al., 2015). The same forces can be held responsible for the increase in damaged
438 starch with this centrifugal shear grinder (table 2).

439

440 **3.5. Low field NMR measurements**

441 Different states of water binding (little, medium, or very mobile water) can be distinguished using
442 low field NMR methods. The use of Carr–Purcell–Meiboom–Gill (CPMG) sequences allows the
443 determination of the transverse relaxation time (T_2) that describes the interactions between water

444 and components, their intensity and the amount of water involved. Systems with low proton
445 mobility have short T2 of the order of a few μ s. Systems with high proton mobility have shorter
446 interaction times to exchange energy, and therefore longer T2 of the order of one second.

447 NMR results (figure 3) show two FID populations around 0.014 and 0.15 ms (peaks A and B) and
448 three CPMG populations at 1, 4-6 ms and 30-60 ms (peaks C, D, and E). Based on the work on
449 wheat bran of Hemdane et al. (2017), population A could be attributed to protons of densely packed
450 arabinoxylan and probably also to those of crystalline cellulose. Populations B and C are less
451 straight-forward to define, but could comprise most likely CH-protons of arabinoxylan, given their
452 comparable distribution with arabinoxylan and their low mobility. Population D can be assigned
453 to OH protons of water interacting with bran-related biopolymers such as arabinoxylan and
454 cellulose as well as protons of intragranular water. Population E representing the most mobile
455 proton population observed in this study can only be assigned to water held through improper
456 stacking, and not by intragranular water.

457 The major differences between the two brans ground in the different milling conditions are in the
458 high mobility water (observed in peak E). Relaxation time T2 of peak E (figure 3) is significantly
459 lower for the initial and ground durum wheat bran (all processing conditions), the lowest value
460 being obtained with the centrifugal shear grinder with a 250 μ m grid.

461 **FIGURE 3.**

462 In addition, T2 relaxation times are always lower for native and ground durum wheat bran for the
463 high mobility water peak (peak E). Less mobile water is thus present for durum wheat bran. This
464 is in line with higher $R_{multimer}$ as seen in ATR-FTIR measurements, indicative of more interactions
465 between bran matrix and water.

466

467 *Discussion* - Similar results of differences in high mobility water were observed by Hemdane et
468 al. (2017) on coarse and ground wheat brans. They suggest less interactions between water
469 molecules and the bran's backbone related to the mechanical loading mode constraints which
470 create disturbances during the grinding. These observations are in accordance with those from the
471 water vapor adsorption isotherms of durum wheat bran for values in the region 0.4-0.8
472 corresponding to the intermediate and most mobile water. For the comparison of bran between
473 them, the main differences are evidenced at the lowest particle sizes of 250 μm and 300 μm in
474 centrifugal grinders and impact grinders respectively.

475 The mechanisms responsible of the interaction between water and bran components are not yet
476 fully elucidated. In wheat bran, Jacobs et al. (2015) suggested that strongly bound water may be
477 related to capillary action (nano-pores) and strong hydrogen bonds, whereas weakly bound water
478 mainly arises from the filling of micropores and stacking phenomena.

479

480 **4. CONCLUSION**

481 The present study investigates the effect of grinding loading modes on intrinsic hydration
482 properties as function of type of bran (common vs durum wheat bran) with the aim to develop a
483 better mechanistic understanding of their origin in ground wheat brans. The results from water
484 vapor adsorption isotherms, FTIR and NMR experiments suggested that grinding is not limited to
485 size reduction and also affect the physical structure of the bran particles and generated
486 modifications of bran structure that can be evidenced by the strength of the bran-water interactions.
487 These effects are depending on the type of bran, and on the type of main loading mode generated
488 by the milling device and their mechanical work intensities (related to the size grid). Native durum
489 wheat bran has lower water vapor adsorption capacities than common wheat bran due to

490 differences in chemical composition and particle characteristics. The water vapor adsorption
491 capacities of durum wheat bran are more affected by the milling process.

492 During the milling process of common wheat bran, shear with a middle intensity mechanical work
493 (size grid of 500 μm) tends to increase the proportion of strongly bound water, probably related to
494 the opening of cell wall. By contrast impact milling tends to reduce the water vapor adsorption
495 capacity by collapsing the porous structure of the wheat bran particles. For durum wheat bran, the
496 effect for shear milling with high mechanical work (size grid of 300 μm) and for impact milling
497 are in between these two behaviors as they probably result of a competition between the opening
498 of new cell wall during the grinding and the collapsing of the structure durum wheat structure.
499 Complementary studies are needed to clarify the relation between the effect of the grinding mode
500 on the dissociation of the tissues and on the accessibility of the constituents and their impact on
501 the hydration properties. This will bring new elements to optimize the milling processability of
502 wheat bran to target specific hydration properties to produced cereals products enriched in bran
503 and converges towards a better valorization of common and durum wheat bran with fractions
504 enriched in compounds of interest.

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511

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638 **Table 1.** Proportion of fine particles and parameters (median value and span) of distribution
 639 curves of particle size for the different milled brans from common or durum wheat. *Circle
 640 equivalent diameter indicated as D50 was acquired by granulomorphological analysis for native
 641 brans. Cumulated values of 5 plates are represented.
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	Proportion of fine particles (%)	Finer particles		Larger particles	
		D ₅₀ (µm)	Span	D ₅₀ (µm)	Span
<i>Common wheat</i>					
<i>Native bran</i> (*)	--	--	--	3225 (±271)	0.75 (±0.01)
Impact mill - 300 µm	16.0 (±1.0) ^d	34.4 (± 0.4) ^c	2.01 (±0.13) ^{bc}	340 (±11) ^c	1.50 (±0.07) ^{ab}
Impact mill - 500 µm	8.0 (±0.3) ^a	36.0 (±0.4) ^d	2.20 (±0.13) ^c	525 (±6) ^h	1.74 (±0.03) ^c
Centrifugal mill - 250 µm	25.9 (±0.3) ^g	28.2 (±0.2) ^b	1.97 (±0.02) ^{ab}	202 (±1) ^b	1.57 (±0.01) ^b
Centrifugal mill - 500 µm	14.0 (±0.7) ^c	42.9 (±1.8) ^f	1.89 (±0.06) ^a	364 (±13) ^f	1.54 (± 0.07) ^{ab}
<i>Durum wheat</i>					
<i>Native bran</i> (*)	--	--	--	1125 (±72)	1.36 (±0.02)
Impact mill - 300 µm	21.9 (±0.4) ^f	36.0 (±0.6) ^d	2.17 (± 0.03) ^{dc}	293 (±5) ^c	1.41 (±0.04) ^a
Impact mill - 500 µm	10.3 (±1.1) ^b	36.4 (± 0.5) ^d	2.11 (±0.03) ^{cde}	433 (±13) ^g	1.64 (±0.03) ^{bc}
Centrifugal mill - 250 µm	29.0 (±1.8) ^h	24.8 (±0.4) ^a	1.92 (± 0.03) ^{ab}	179 (± 9) ^a	1.58 (±0.27) ^b
Centrifugal mill - 500 µm	20.1 (±0.4) ^e	37.6 (±0.8) ^e	2.08 (±0.04) ^{cd}	315 (±4) ^d	1.52 (±0.02) ^{ab}

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Means in the same column with different letters are significantly different (P<0.05) by Tukey's HSD test.

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Table 2. Composition of common and durum wheat grains and their brans.

	Water (g/100g dm)	Ash (g/100g dm)	Total starch (g/100g dm)	Damaged starch (g/100g dm)
<i>Common wheat</i>				
Native bran	12.8 ± 0.2	6.24 ± 0.04	8.0 ± 0.4	0.7 ± 0.02
Impact mill - 300 µm	10.1 ± 0.01	6.72 ± 0.02	8.5 ± 0.6	0.9 ± 0.02
Impact mill - 500 µm	10.8 ± 0.1	6.75 ± 0.02	7.1 ± 0.1	0.7 ± 0.01
Centrifugal mill - 250 µm	8.2 ± 0.03	7.07 ± 0.01	8.3 ± 0.05	1.0 ± 0.01
Centrifugal mill - 500 µm	10.2 ± 0.01	7.03 ± 0.03	8.9 ± 0.5	0.9 ± 0.01
<i>Durum wheat</i>				
Native bran	11.5 ± 0.2	5.05 ± 0.04	18.6 ± 0.3	2.0 ± 0.1
Impact mill - 300 µm	10.9 ± 0.03	5.04 ± 0.03	19.4 ± 0.2	2.9 ± 0.1
Impact mill - 500 µm	11.6 ± 0.1	5.04 ± 0.01	21.6 ± 1.0	2.2 ± 0.02

Centrifugal mill - 250 μm	8.7 ± 0.1	4.82 ± 0.07	18.7 ± 0.3	3.5 ± 0.01
Centrifugal mill - 500 μm	10.9 ± 0.1	5.09 ± 0.02	21.6 ± 0.5	2.8 ± 0.1
	Native Common wheat bran		Native Durum wheat bran	
Total monosaccharide (g/100g dm)	51.7 ± 0.41		54.8 ± 1.86	
GalA (g/100g dm)	0.04 ± 0.03		0.54 ± 0.14	
GlcA (g/100g dm)	0.84 ± 0.01		0.76 ± 0.05	
Man (g/100g dm)	0.68 ± 0.04		0.77 ± 0.03	
Xyl (g/100g dm)	16.4 ± 0.11		12.9 ± 0.25	
Glc (g/100g dm)	24.24 ± 0.70		31.47 ± 1.60	
Gal (g/100g dm)	1.10 ± 0.06		1.26 ± 0.08	
Rha (g/100g dm)	0.19 ± 0.05		0.19 ± 0.02	
Ara (g/100g dm)	9.09 ± 0.25		8.22 ± 0.08	
Lignin (g/100g dm)	5.97 ± 0.01		5.28 ± 4.10^{-4}	

647

648 **Table 3.** GAB equation parameters and root-mean-square-error (RMSE) for the common and
649 durum wheat brans treated with several grinding processes.
650

	M_0	c	k	$A(m^2/g)$	RMSE	R^2
Common wheat						
Native bran	7.69	15.19	0.66	269.18	0.12	0.999
Impact mill - 300 μm	7.53	15.37	0.68	263.55	0.11	0.999
Impact mill - 500 μm	7.46	15.32	0.68	261.09	0.10	0.998
Centrifugal mill - 250 μm	7.30	15.53	0.68	255.58	0.13	0.999
Centrifugal mill - 500 μm	7.67	15.41	0.66	268.38	0.10	0.999
Durum wheat						
Native bran	6.97	17.82	0.69	244.08	0.13	0.998
Impact mill - 300 μm	6.88	17.67	0.70	240.66	0.16	0.998
Impact mill - 500 μm	7.52	15.31	0.66	263.25	0.13	0.996
Centrifugal mill - 250 μm	6.54	16.20	0.72	228.75	0.13	0.996
Centrifugal mill - 500 μm	7.12	16.22	0.69	249.23	0.57	0.998

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FIGURE CAPTIONS

Figure 1. Particle size distributions for the different native and milled brans from common or durum wheat. SEM observations of native common (A) or durum (B) wheat brans, in native form (A1-B1), ground with impact miller - 300 μm (A2 - B2) or ground with centrifugal grinder - 250 μm (A3-B3).

Figure 2. ATR-FTIR spectra of interface hydration water at different grinding processes of durum (A) and common wheat bran (B) equilibrated at a relative humidity of 58% obtained after subtraction of freeze dried spectra. R_{multimer} calculated for durum wheat and common wheat brans at different grinding processes at a relative humidity of 58% (C) and 40% (D).

Figure 3. FID and Carr-Purcell-Meiboom-Gill (CPMG) proton distributions of common and durum wheat brans at different grinding conditions. Free induction decay proton distributions of wheat brans with relaxation time T2 and amplitude.

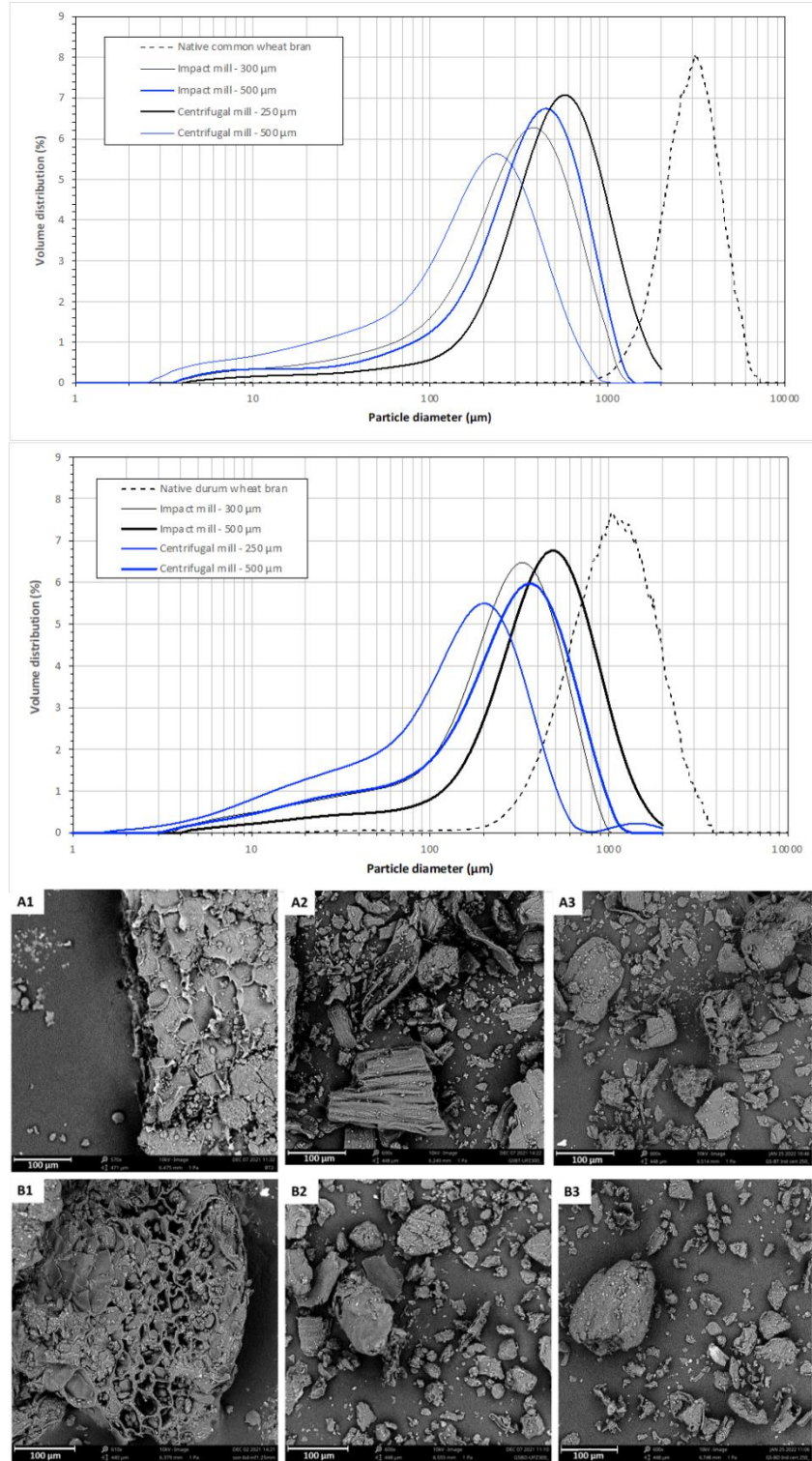
Figure A.1. Water vapor adsorption isotherms of native brans from durum wheat and common wheat (A) and of ground brans after UPZ grinder and 500 μm grid (B). Circle dots are experimental values and line is calculated curves with the GAB model. Mean values of duplicate measurements are represented.

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FIGURES

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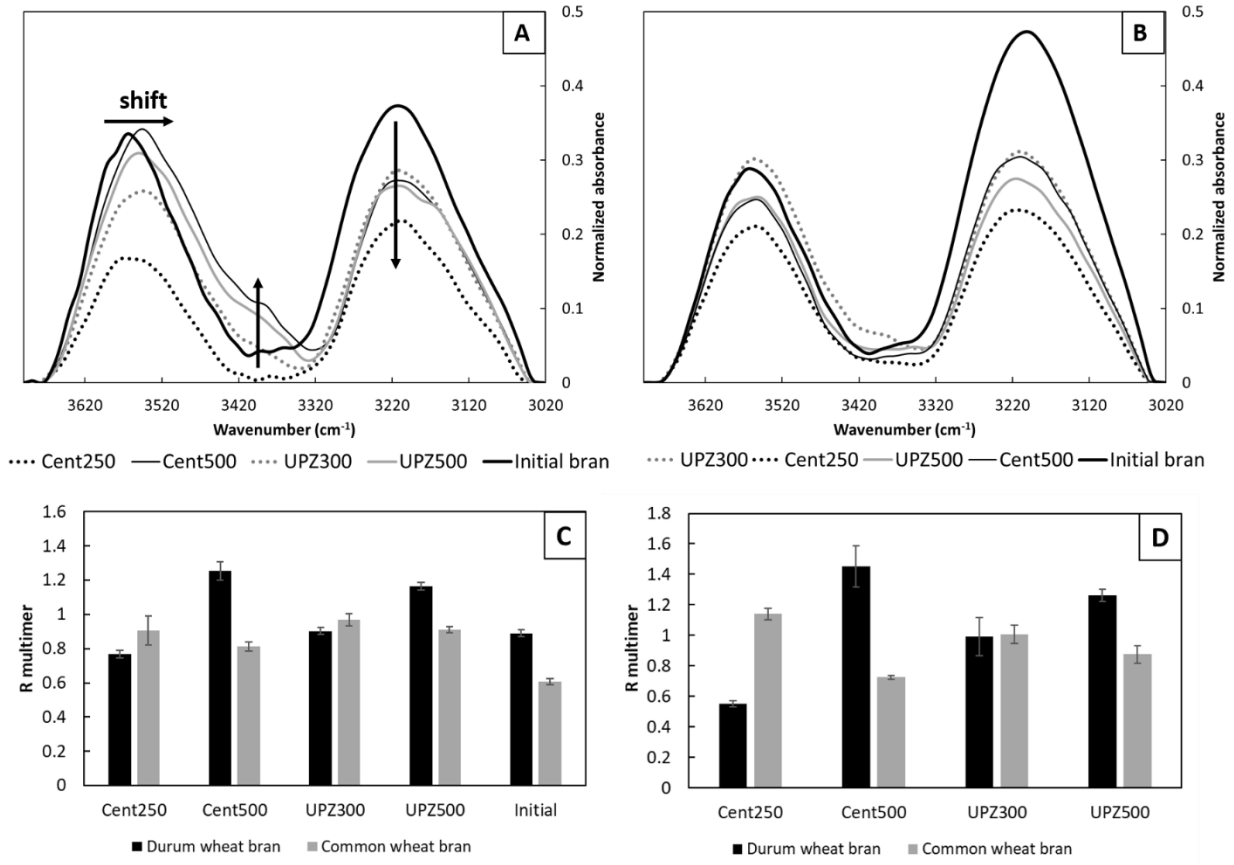
Figure 1.



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Figure 2.



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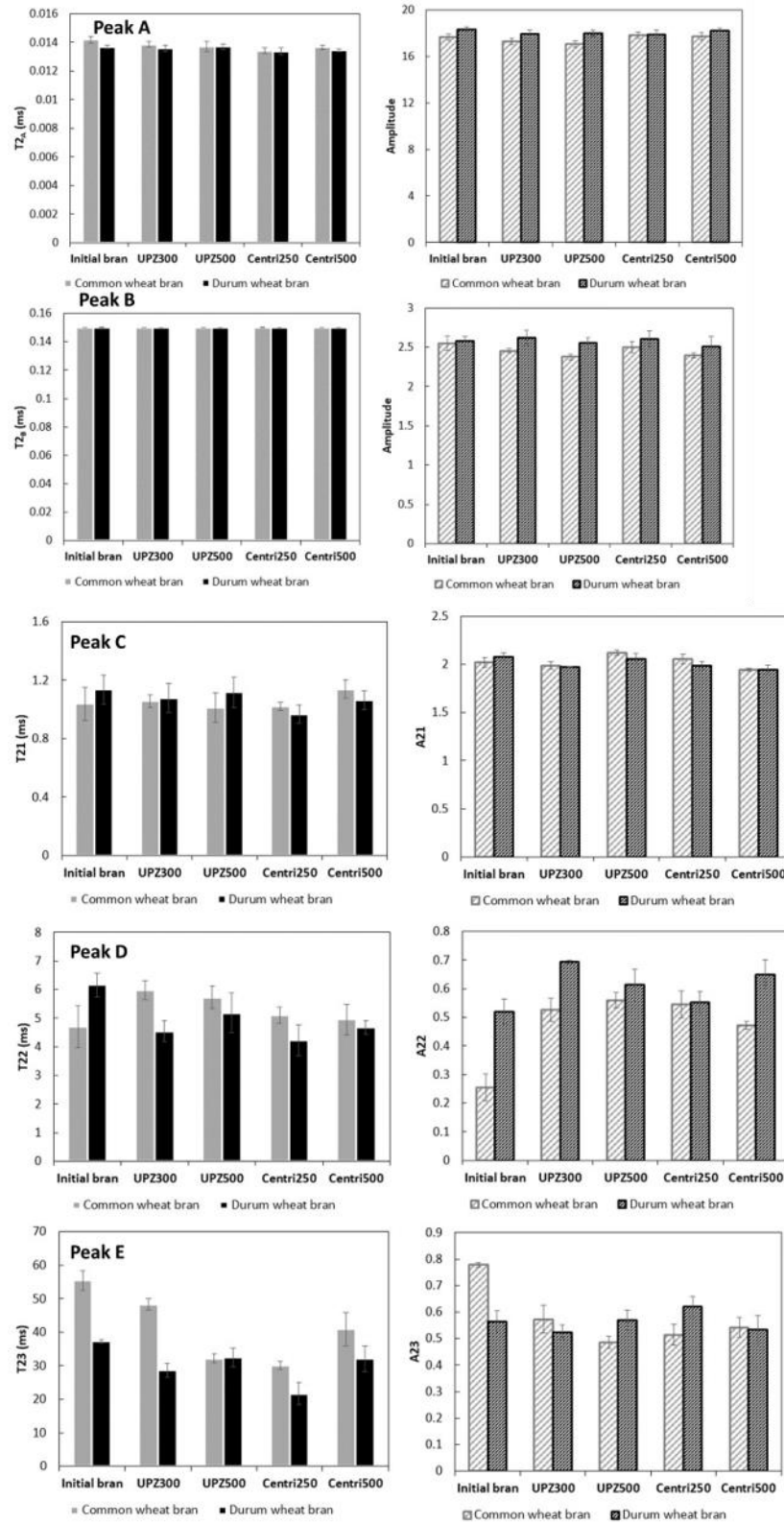
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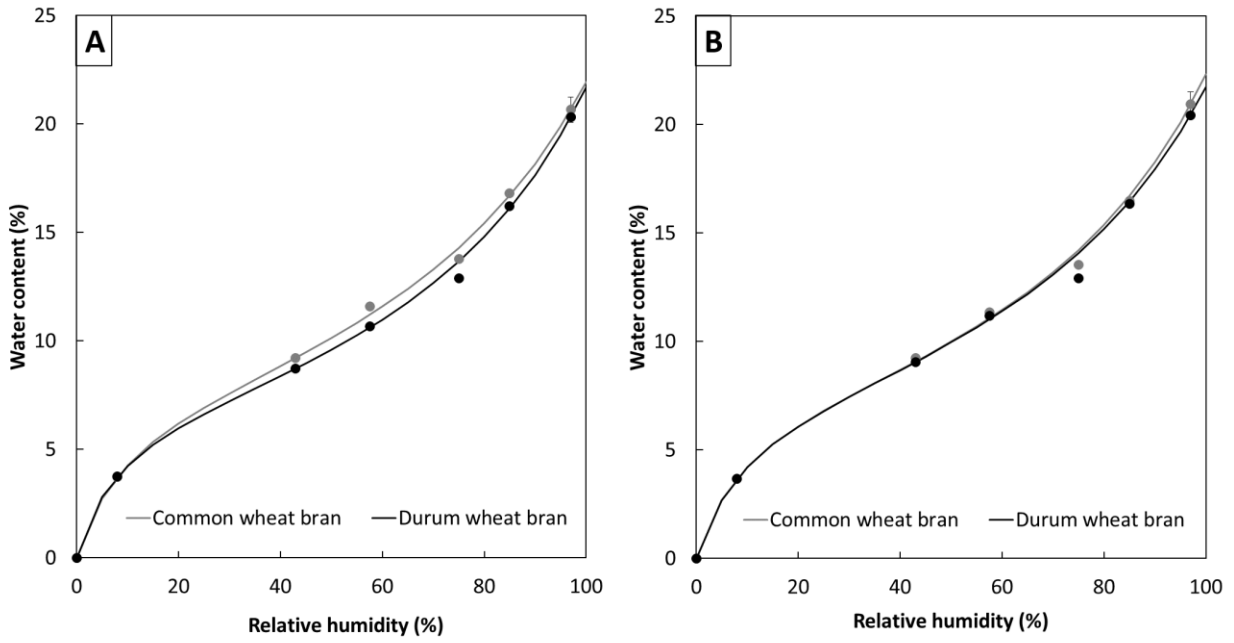
Figure 3.



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Figure A.1.

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Conflict of interest

The authors declare no conflict of interest