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# Controlled Growth of 3D Topological Insulator BiSb(Te<sub>1-y</sub>Se<sub>y</sub>)<sub>3</sub> Nanocrystals by Chemical Vapor Transport

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The structural and electrical properties of thin nanocrystals of the 3D topological insulator  $BiSb(Te_{1-y}Se_y)_3$  (y = 0.0, 0.01, 0.02,....0.09) have been investigated. The nanostructures were synthesized from a bulk parent  $BiSb(Te_{1-y}Se_y)_3$  polycrystalline powder on different substrate materials using the bottom-up chemical vapor transport method without the addition of transport agents, resulting in well-faceted and thin single crystals with dimensions of about 20  $\mu$ m in length and about 20 nm in height. Thermodynamic calculations were performed to optimize the growth process. The chemical composition and morphology of the nanocrystals were analyzed by energy dispersive X-ray spectroscopy, scanning electron microscopy, and atomic force microscopy. The  $R\overline{3}m$  crystal structure of individual nanocrystals and their high crystalline quality were studied by high-resolution transmission electron microscopy. Magnetotransport measurements confirm that bulk-charge compensation could be achieved by adding a small amount of Se to the ternary compound  $BiSbTe_3$ , and the transport properties of thin flakes further reveal the enhanced carrier mobility of topological surface-state carriers.

#### 1. Introduction

Since their discovery, three-dimensional (3D) topological insulators (TIs) have attracted attention for their unique spin-momentum locked surface states, with some potential for applications in electronics, 1,2 spintronics, 3,4 and quantum computing. 5,6 Ideally, 3D TI materials are characterized by an insulating gap in the bulk but gapless surface states that are protected by time-reversal symmetry, due to a band inversion caused by strong spin-orbit interactions. 7-10 The scattering of these surface states by disorder is reduced compared to the case of massive quasi-particles in bulk conductors. 11-15 While many materials have been identified as 3D TIs, such as the prototypical Bi<sub>2</sub>Te<sub>3</sub>, Bi<sub>2</sub>Se<sub>3</sub>, and Sb<sub>2</sub>Te<sub>3</sub> compounds, all exhibit a finite bulk doping caused by vacancies and antisite defects, thus leading to a shift of the Fermi level into the bulk bands. 15-19 Moreover, the Dirac point in the surface band structure is often well below or well above the fermi level<sup>15</sup>. Combining these binary materials, which share the same crystal structure  $(R\overline{3}m)$ , to form a solid solution is a successful strategy to favor some bulk-charge compensation and ideally to have the surface-state Dirac point located within the bulk band gap, as shown for bulk crystals or ultrathin films grown by molecular-beam epitaxy<sup>20</sup>. This route has been followed by growing ternary TIs such as (Bi<sub>1-x</sub>Sb<sub>x</sub>)<sub>2</sub>Te<sub>3</sub> or Bi<sub>2</sub>(SeTe)<sub>3</sub><sup>21-</sup> <sup>24</sup> or in the more complicated quaternary compounds (Bi<sub>1-x</sub>Sb<sub>x</sub>)<sub>2</sub>Te<sub>1-</sub> <sub>y</sub>Se<sub>y</sub>)<sub>3</sub> (BSTS).<sup>25,26</sup> This remains largely unexplored for singlecrystalline nanostructures grown by chemical vapor transport. In particular, the quaternary system BSTS is an intersting candidate for studying surface-state properties because different point defects act as donors or acceptors, which can result in an efficient bulk-charge compensation through controlling the proportion of pnictogen (Bi and Sb) and chalcogen (Se and Te) atoms. For charge transport

measurements, very thin nanocrystals are required to study topological surface states close to the Dirac point. Nanostructures provide several advantages in comparison to their bulk counterparts. The different morphologies of nanostructures, such as nanoflakes, nanoribbons, and nanowires are associated with different surface transport properties, with either diffusive transport in wide quasi-ballistic transport in nanostructures or nanostructures.<sup>27-28</sup> This is important in order to produce tailored properties for specific applications, e.g. 2D thin films for spin-charge interconversion, 1D quantum wires for dissipationless interconnects or topological Majorana bound states<sup>29</sup>. The high surface to volume ratio of nanostructures, as compared to bulk materials, increases the relative contribution of surface states to charge transport. However, in nanosturctures of the binary compounds Bi<sub>2</sub>Te<sub>3</sub>, Bi<sub>2</sub>Se<sub>3</sub>, and Sb<sub>2</sub>Se<sub>3</sub>, the large carrier density induced by impurities or point defects created during the crystal growth process results in a large bulk-state conductivity, making it more challenging to explore the topological surface states. 30,31 This important issue could be overcome either by bulk-charge compensation in ternary or quaternary compounds of this family, or by considering thin nanostructures with a large bulkcarrier depletion and band bending close to interfaces<sup>32</sup>. Therefore, thin crystals of BSTS are of particular interest since their bulk conductivity can be reduced by both optimizing their composition and by reducing their thickness. For ternary (Bi<sub>1-x</sub>Sb<sub>x</sub>)<sub>2</sub>Te<sub>3</sub>, we have previously reported a reduced bulk-carrier density for x = 0.5, with a p-type residual conductivity. Since Se vacancies acts as donors, adding a small amout of Se to BiSbTe<sub>3</sub> nanocrystals could be an efficient route to larger bulk-charge compensation<sup>21</sup>.

One method to obtain thin nanocrystals is mechanical exfoliation. <sup>33-</sup> Despite simple and easy fabrication, exfoliation methods have

some limitations in terms of limited scalability, reproducibility, and low yield. In addition, exfoliation is a poorly-controlled process because the in-plane crystalline orientation is usually unknown and irregularly shaped crystals with rough edges are obtained. Highquality ultra-thin films can be prepared by molecular beam epitaxy (MBE),20,38-40 but MBE requires a high-priced apparatus, and the growth mechanism results in percolation of nanoislands with boundaries, so that the charge mobility is even lower than for thin flakes exfoliated from bulk crystals, with little influence of the substrates. In contrast to the previously mentioned methods, chemical vapor deposition (CVD) is a scalable convenient technique.  $^{\rm 41\text{-}46}$  For example, CVD allows direct growth on the substrate without a catalyst and produces good quality single-crystal nanostructures with well-defined morphology. However this method requires carrier gases, suitable available precursors, and careful control of crystal growth. Another method based on vapor deposition is chemical vapor transport (CVT). 47-48 CVT offers several advantages over other methods used to prepare TI nanostructures. The CVT setup is generally less complex than a CVD setup, as no carrier gas is required. Currently, the preparation of TI nanocrystals of the binary compounds (Bi<sub>2</sub>Se<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub>) and ternary compounds (Bi<sub>1-x</sub>Sb<sub>x</sub>)<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>(SeTe)<sub>3</sub> have been extensively studied using various preparation methods including the CVT technique. <sup>21,49,50</sup> In this work, we present a first systematic study of the CVT growth process of thin  $BiSb(Te_{1-y}Se_y)_3$  (y = 0.0, 0.01 ,0.02...,0.09) nanocrystals obtained from bulk  $BiSb(Te_{1-y}Se_y)_3$  polycrystal powder with a detailed discussion of the effect of the growth conditions on morphology, size, thickness and magnetotransport properties of the obtained nanocrystals. Thermodynamic calculations are performed to facilitate a systematic planning of the experimental procedure for BSTS crystal growth. This work provides the structural details of the synthesized bulk starting materials, and the grown nanocrystals. In addition to AFM, Raman spectra and magnetotransport data are discussed.

# 2. Experimental part

## 2.1 Thermodynamic Simulations

For successful rational synthesis plan, a modified version of the program package TRAGMIN 5.1<sup>51</sup> was used for thermodynamic simulation. Thermodynamic data (enthalpy, entropy, and heat capacity) of all known relevant gaseous and condensed species involved in the system were taken from the Fact PS database  $^{52}$ . A list of the incorporated species can be found in the Supplementary Information (SI. 1 and SI. 2) . Based on this, we simulated the partial pressures of all gaseous species over temperature range of 400 °C to 800 °C. In each simulation we took into consideration the presence of 5  $\times$  10<sup>-9</sup> mol  $\,$  H<sub>2</sub>O, and 1× 10<sup>-9</sup> argon, using 9.4 mL ampoule volume, thus simulating close to the actual performed experiments.

#### 2.2 BSTS Bulk Starting Material Preparation

All preparation process was performed in a glovebox (M. Braun). BSTB bulk materials have been synthesized by mixing stoichiometric amounts of Bi<sub>2</sub>Te<sub>3</sub>, Bi<sub>2</sub>Se<sub>3</sub>, and Sb<sub>2</sub>Te<sub>3</sub> (> 99.9%,

Alfa Aesar) to achieve desired compositions BiSb( $Te_{1-y}Se_y$ )<sub>3</sub> ( $y=0.0,\ 0.01,\ 0.02,\ ....,0.09$ ), The completely grounded starting materials were filled in evacuated one-chamber ampoule (11 cm long, 1 cm inner diameter, 7.8 ml volume), and sealed at a pressure of  $\sim 3\times 10^{-3}$  mbar. The ampoule was heated to 500 °C (15 K/min) in a standard muffle furnace and held at that temperature for 72 h, naturally cooled down to room temperature, and opened inside the glovebox.

#### 2.3 Substrates Preparation

All substrates were purchased from Crystal GmbH with one-side polished. Four types of 5 × 20 mm² substrates were used: Si (111), Si/SiO $_2$  (100), BaF $_2$  (111), and A-plane sapphire (11 $\overline{2}$ 0). All substrates were ultrasonically cleaned in acetone and isopropyl alcohol for 10 min each to remove the photoresist layer, then rinsed in deionized water, and dried under nitrogen flow.

#### 2.4 BSTS Nanocrystals Growth

CVT experiments were performed as follows: around 1 mg of the bulk starting material was placed at the source side of twochamber ampoule (12 cm long, 1 cm inner diameter, 0.5 mm narrow diameter, ca. 9.4 ml volume), while positioning the substrate at the sink side correspondingly. The ampoule was sealed at pressure of ca. 3×10<sup>-3</sup> mbar, then heated in a horizontal tube furnace to specific temperature gradients performed for each individual experiment. To provide stable conditions during crystal growth, a precise control of various parameters is needed. CVT experiments were performed using temperature gradients of  $\Delta T$  = 80 ( $T_1$  = 500 °C,  $T_2$  = 420 °C),  $\Delta T$  = 120 ( $T_1$  = 500 °C,  $T_2$  = 330 °C), and  $\Delta T$  = 170 ( $T_1$  = 500 °C,  $T_2$  = 380 °C). Also, experiments were carried out by varying the source and substrate temperatures keeping  $\Delta T$  = 170 ( $T_1$  = 630 °C,  $T_2$  = 460 °C), ( $T_1$  = 560 °C,  $T_2$  = 390 °C), and ( $T_1$  = 500 °C,  $T_2$  = 330 °C) with a duration time between t = 4 h and 16 h for all the mentioned conditions. Several experiments were performed using different ampule length (8 cm, 12 cm, 16 cm) during the optimization process. An inverse temperature gradient was applied ( $T_1 = 370 \,^{\circ}\text{C}$ ,  $T_2 = 550 \,^{\circ}\text{C}$ ) before starting the actual CVT crystal growth to remove any material that adhered to the ampoule wall and the substrate during preparation and sealing process, returning it to the source side.

#### 2.5 Characterization

Powder X-ray diffraction (pXRD). The prepared starting materials were analysed by pXRD using "STOE STADI-P" diffractometer (Stoe, Germany) in transmission mode with a Cu-K $\alpha$ 1 radiation source ( $\lambda$  = 1.5418 Å) and a Mythen 1 K detector. A small amount of silicon powder (99.5%, Alfa Aesar) was added to the samples as an internal standard.

Digital microscopy: A "VHX-7000" (Keyence Deutschland GmbH, Germany) with high-resolution lenses, a 4K CMOS camera and powerful lighting, was used for taking high resolution images for the grown crystals.

Scanning electron microscopy (SEM): A "FEI Nova NanoSEM 200" with 10-15 acceleration voltage, coupled with Energy dispersive X-ray spectroscopy with a QUANTA 200/400 (AMATEX) was used to investigate the morphology and the composition of the grown crystals. The software "EDXGenesis" version 6.32 was applied for quantification analysis. At least five measurements of different areas or crystals were averaged to calculate the compositions of each sample or substrate.

Atomic force microscopy (AFM): The measurements were performed using an Icon Dimension (Bruker) in tapping mode (Si single crystal cantilever). Data analysis was done by "Nasoscope Analysis" version 1.8.

Transmission electron microscopy (TEM): High resolution transmission electron microscopy (HR-TEM) analysis as well selected area electron diffraction (SAED) measurements were conducted at "FEI Titan 80–300", fitted with a Cs corrector using an electron acceleration voltage of 300 keV. For sample preparation, substrates with grown BSTS crystals were ultrasonicated in (0.5-1~mL) ethanol for 5 min. A few drops of the resulting dispersion were poured on a lacey-carbon copper grid, allowed for the successful transfer of several crystals onto the grid. The software package "Single-Crystal Version 3.1.5" was utilized for indexing the SAED pattern with the corresponding structure model.

Micro-Raman spectra were recorded in back-scattering configuration using "T64000 spectrometer" (Horiba Jobin Yvon, USA), laser excitation at 660 nm with Torus Laser (Laser Quantum) and "Symphony II 1024x256 Cryogenic Back-Illuminated Deep-Depletion CCD Detector" (Horiba Jobin Yvon). The measurements were performed at room temperature. For non-polarized measurements, the laser polarization was scrambled by a depolarizer.

Magneto-transport measurements were performed on individual nanostructures connected with Ti/Au contacts patterned in a Hall-bar geometry (e-beam lithography and lift-off processes), using low-frequency lock-in amplifiers and an 18 T superconducting magnet. The samples were mounted onto the cold finger of a <sup>3</sup>He Oxford Heliox refrigerator, with a 250 mK base temperature.

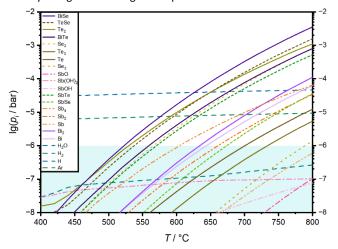
#### 3. Results and discussion

#### 3.1 Thermodynamic Calculations

To understand the gas phase composition and the vapor-solid growth mechanism, thermodynamic simulations have been performed. Hence, successful experiments for the CVT BSTS synthesis can be carried out in sealed silica ampoules without transport agent. The conditions for BSTS nanocrystals growth were derived from the theoretical calculations, the gas phase composition was conducted for BiSb $(Te_{1-y}Se_y)_3$  in a temperature range between 400 °C and 800 °C in 20 K steps (Fig. 1). The starting components for the simulation were selected for a solid solution with x = 0.5, y = 1.5. The standard thermodynamic data of BSTS has been calculated based on the values of the binaries and assuming an ideal mixing

behavior. The thermodynamic data used are listed in the supplementary information. Calculations that involved hydroxides or oxides formed from residual moisture, such as the gaseous molecules Sb (OH)2 and SbO, suggest that the presence of moisture has negligible effects on the vapor transport process. In order for vapor species to contribute in a relevant manner to the diffusionbased transport processes of the CVT, a partial pressure of ca. 10-<sup>6</sup> bar is required. Based on the data presented in Fig. 1, it can be observed that the gas species BiSe, TeSe, Te2, Se2 and SbTe exhibit partial pressures that fulfill these criteria at the lowest possible temperatures for CVT in the range of ca. 550 °C to 630 °C. As a result, BiSe, TeSe, Te<sub>2</sub>, Se<sub>2</sub> and SbTe are the predominant driving agents for the vapor transport, and thus the BSTS crystal growth. BiSe and SbTe BiSe(g) are the major volatile species for the transport of Bi and Sb, respectively, during the CVT. The partial pressure of BiSe is higher than SbTe, which can lead to an enrichment of Bismuth in the mixed crystals during the CVT. However, this behavior was not observable in the performed experiments, and the cation ratio Bi/Sb remains close to one in the nanostructures grown. Vapor species BiSe, TeSe, and Se<sub>2</sub> with their high pressures are the main vapor species responsible for the transport of Se, potentially resulting in enrichment of selenium in the crystal composition, as found for some growth conditions. According to the calculations, a simplified example of a suggested main transport reaction occurring during the CVT process for BSTS can be described as the congruent decomposition-sublimation:

BiSb( $Te_{0.5}Se_{0.5}$ )<sub>3 (s)</sub>  $\leftrightarrows$  BiSe  $_{(g)} + \%$  TeSe  $_{(g)} + \text{SbTe }_{(g)}$ . However, in reality and especially for BSTS with compositions that deviate from this stoichiometry, the occurs transport reactions are more complex and involve also the other vapor species with high vapor pressures like  $Se_2$  or  $Te_2$ , whereas also the individual equilibrium partial pressures are depending on the initial composition of the BSTS. The thermodynamic simulations provide insights into the appropriate parameters for a rational and successful synthesis of BSTS nanocrystals growth during CVT experiments.



**Fig. 1** Simulated temperature dependent equilibrium partial pressures above the solid BSTS with equimolar cation and anion ratio. The blue area highlights partial pressures that are too low for a significant contribution to vapor transport.

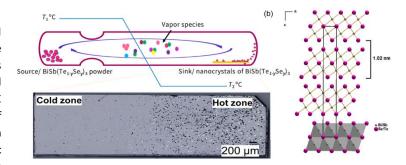
#### 3.2 BSTS Bulk Materials Investigation

To investigate the elemental composition of the prepared starting materials, SEM and EDX was used. (SI. 3) in the Supporting Information. shows the result of the EDX analysis performed on bulk BiSb(Te<sub>0.98</sub>Se<sub>0.02</sub>). The error bars presented are within the accepted values of the uncertainty of the EDX measurements. All measured samples confirm the existence of the elements Bi, Sb, Te, and Se in the samples, with an amplitude ratio corresponding to the stoichiometric compounds  $BiSb(Te_{1-y}Se_y)_3$  (y =0, 0.01, 0.02, ....0.09). These starting materials serve as source for our CVT experiments to grow BSTS nanocrystals. XRD pattern were measured at room temperature for the BiSb( $Te_{1-y}Se_y$ )<sub>3</sub> (y = 0, 0.01, 0.02, ......0.09) bulk starting materials in their powder form, as shown in the Supporting Information (SI. 4). The observed Bragg reflexes correspond to BiSbTe( $_{1-y}$ Se $_y$ )<sub>3</sub> in a trigonal structure with R3m space group. The absence of any additional reflections confirms the successful phase-pure synthesis for all prepared compositions. A shift of Bragg peaks toward higher angles, possibly expected for a solid solution, with an increasing content of selenium was not clearly observed. This is due to the small differences of selenium amount ( $\Delta y = 0.01$ ) between the all prepared compositions.

#### 3.3 BSTS Nanocrystals Investigation

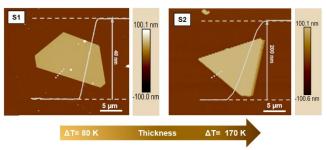
A scheme of a general CVT process of BSTS is illustrated in Fig. 2(a). The growth happens in a sealed quartz ampule placed an oven with defined temperature gradient. It proceeds as a sublimation of the source at temperature  $T_1$ , followed by vapor transport to a lower temperature region and the condensation on a substrate at temperature  $T_2$ . For a long-enough substrate, the local temperature gradient during the growth result in two zones with different densities of nanostructures (hot and cold zones), see Fig. 2(c). In general, the CVT-grown Sb( $Te_{1-y}Se_y$ )<sub>3</sub> nanocrystals were found predominantly in the form of agglomerated crystalline material at the hot side of the substrate and as individual and spatially separated nanocrystals at the cold side of the substrate.

Importantly, it is possible to alter the morphology and the thickness of the grown nanocrystals by varying the growth conditions (table SI.5 in the Supporting Information): (i) the temperature gradient, (ii) the source and substrate temperatures, (iii) the source-substrate distance, (iv) the growth duration, and (v) the substrate material. We obtained various morphologies of nanostructures deposited on the substrate such as nanoflakes (triangles, hexagons and rhombohedrons), nanoribbons, and nanowires, with nanometres heights from 20 to 40 nm and lateral sizes ranging from  $^{\sim}$  0.5 to 20  $\mu m$ , when particular parameters are changed while keeping the other parameters constant.



**Fig. 2** Overview of CVT growth of BiSb( $Te_{1-y}Se_y$ )<sub>3</sub> nanocrystals. (a) Schematics of the CVT setup and the synthesis process. The starting material and the substrate are place inside two chambers of a silica ampoule. The deposition was assisted by temperature gradient (see the text for details). (b) Layered crystal structure of BiSb( $Te_{1-y}Se_y$ )<sub>3.</sub> (c) The substrate is divided into two temperature zones (cold, and hot).

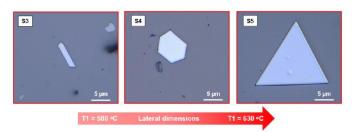
(i) Nanocrystals with different thicknesses can be obtained by changing the temperature gradient (or the growth duration). Using a lower temperature gradient of  $\Delta T = 80 \text{ K}$  leads to the formation of thinner crystals with height ranging from 20 to 40 nm (Fig. 3. sample S1), since the low temperature gradient limits the mass transport of the starting material to the deposition zone. Thus, the availability of materials at the substrate surface becomes limited, causing a slow growth rate as well as the formation of thinner crystals. Moreover, a lower temperature gradient can provide better nucleation control, with the reduction of material species flow rate. Nucleation sites may be more preciously determined, potentially resulting in flatter and thinner crystals growth. A high temperature gradient ( $\Delta T$  = 170 K) can result in faster mass transport of the material species to the sink side and a higher density of nucleation sites, resulting in faster growth rates and the growth of thick crystals with visible facets (~ 200 nm), see (Fig. 3, sample S2).



**Fig. 3** AFM images of BiSb( $Te_{1-y}Se_y$ )<sub>3</sub> flakes show thickness -dependent on the temperature gradient. S1, and S2 were grown at ( $T_1 = 500 \, ^{\circ}\text{C}$ ,  $T_2 = 420 \, ^{\circ}\text{C}$ ,  $\Delta T = 80 \, \text{K}$ ) and ( $T_1 = 500 \, ^{\circ}\text{C}$ ,  $T_2 = 330 \, ^{\circ}\text{C}$ .

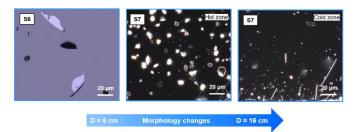
(ii) Based on thermodynamic simulations, three different temperature ranges were used to investigate the effect of the source and sink temperatures on the crystal growth. These are range R1 ( $T_1$  = 630 °C,  $T_2$  = 460 °C), range R2 ( $T_1$  = 560 °C,  $T_2$  = 390 °C), and range R3 ( $T_1$  = 500 °C,  $T_2$  = 330 °C), keeping the temperature gradient constant ( $\Delta T$  = 170 K) in all cases. At

higher temperature range R1 (Fig.4, sample S5), larger regularly-shaped crystals with a larger lateral size of  $^{\sim}$  20  $\mu m$ , as compared to the size of the crystals grown at lower temperatures, (R2 and R3) (Fig. 4, samples S4 and S3,) with a length of  $^{\sim}$  10, 5  $\mu m$ , respectively, are obtained. Some thin crystals were observed in all samples with a height of  $^{\sim}$  20 nm. However, many thick agglomerated crystals extending up to a few micrometres in height were found at the hot side of the substrate. The higher temperature range promotes the formation of substantial quantity of viable nucleus and facilitated higher mass flow rates, thus resulting in a higher amount of material available for CVT growth. Consequently, the nucleus formed can be covered by sequent particle depositions, eventually causing the nanocrystals to aggregate.



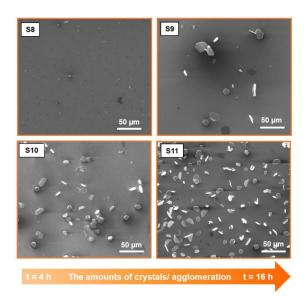
**Fig. 4** Optical-microscope images of BiSb( $Te_{1-y}Se_y$ )<sub>3</sub> flakes show size - dependent on the source temperature. S3, S4, and S5 were grown at  $T_1 = 500$  °C,  $T_2 = 330$  °C,  $T_1 = 560$  °C,  $T_2 = 390$  °C and  $T_1 = 630$  °C,  $T_2 = 460$  °C, respectively.

(iii) Changing the distance between the source and substrate can have a significant impact on the morphology of the resulting nanostructures. When the substrate is placed as close as 8 cm to the source (Fig. 5, sample S6,), the obtained crystals are irregularly shaped, sized and distributed over the substrate. Due to the rapid and non-equilibrium nature of the crystal growth process, the short distance allows the starting material species reach the substrate quickly. As a result, the system may not have the required time for the crystal growth leading to irregularities in crystal structure and morphology. Increasing of the source-substrate distance to 16 cm results in a decrease of the nuclei number in the cold zone (Fig. 5, sample S7,) as well as the change in crystal morphology. On the hot side of the substrate randomly distributed nanoflakes were grown, while nanowires and nanoribbons were found on the cold zone edges of the substrate. The formation of nuclei can be influenced by material availability with limited amount material, nuclei that initiate nanowire growth maybe favoured due to their smaller size compared to those required for nanoflakes formation.



**Fig. 5** Optical-microscope images of BiSb( $Te_{1-y}Se_y$ )<sub>3</sub> flakes show a morphology - dependent on the ampoule length . The distance between the source and the sink is 8 cm for S6 and 16 cm for S7.

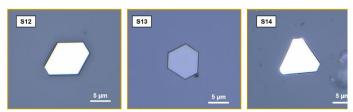
(iv) Long duration growth results in an increase of the size in all directions (X, Y, Z) and the number of deposited crystals. At short reaction time (4 h), there is no significant crystal growth on the substrate (Fig. 6, sample S8,). Only very few crystals with predominant nanowire morphology can be observed. A long growth duration (36 h) (Fig. 6, S 11), many crystals are grown on the substrate. A longer growth time can promote the continuous expansion of more nuclei into different growth directions, rather than being limited to one specific direction. It is important to note that the influence of reaction time on crystal growth varies by changing the temperature. For example, a larger crystal size can be obtained during a short reaction time (4 h) when a high source temperature is applied ( $T_1 = 630$  °C) (Fig. 4, sample S5).



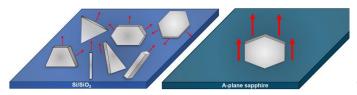
**Fig. 6** SEM images of BiSb( $Te_{1-y}Se_y$ )<sub>3</sub> flakes show agglomeration-dependent on the duration growth . S8, S9, S10, and S11 were grown for 4 h, 8 h, 12 h, and 16 h respectively.

(v) Besides Si/SiO<sub>2</sub>, other substrates, such as A-plane sapphire  $(11\overline{2}0)$ , Si, and BaF<sub>2</sub> were also used to grow BSTS crystals. (Fig. 7, S12, S13, S14). The ability to grow on various amorphous or crystalline substrates allow for exploration of the interaction between the BSTS-substrate and associated properties for diverse applications. Nanocrystals with different morphologies

and orientations can be obtained. For example, nanocrystals on A-plane sapphire substrate preferably grow perpendicular to the substrate surface, while nanocrystals on  $Si/SiO_2$  did not have a preferential growth and were deposited randomly. In general, the growth direction of nanocrystals is mainly determined based on crystallographic orientation of the substrate. However, other factors such as varying surface energy of the substrate material, and growth conditions may influence the growth direction. The different growth processes of  $BiSb(Te_{1-y}Se_y)_3$  nanocrystals deposited on  $Si/SiO_2$ , and sapphire are schematically depicted in Fig. 8. See more figures in the Supporting Information.



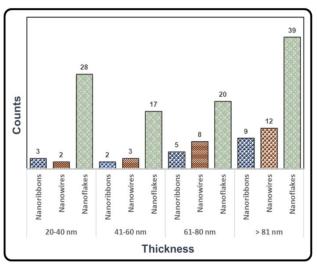
**Fig. 7** Optical-microscope images of BiSb( $Te_{1-y}Se_y$ )<sub>3</sub> flakes grown on different substrates, S12, S13, and S14 were deposited on  $Al_2O_3$ , Si, and  $BaF_2$  respectively.



**Fig. 8** A growth process sketch of BiSb( $Te_{1-y}Se_y$ )<sub>3</sub> nanocrystals grown in different directions Si/SiO<sub>2</sub>, and A-plane sapphire substrates.

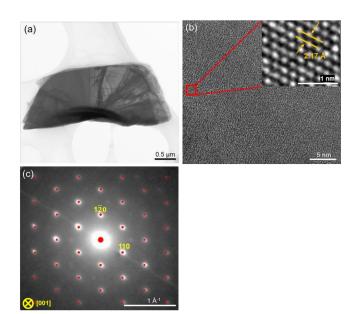
SEM-EDX measurements indicate signals of bismuth, antimony, tellurium, and selenium. The expected elemental ratio could be confirmed. One example of this is shown in the Supplementary Information (SI. 10) for BiSb(Te<sub>0.99</sub>Se<sub>0.01</sub>)<sub>3</sub>. Due to overlapping of the relevant Sb and Te peaks in the EDX, using the Bi content is the most reliable method of quantifying the results. An external standard calibration series was made to increase the precision of the SEM-EDX quantification results for grown crystals.

Thin nanostructures with thickness between 20 and 40 nm found in all compositions, ( $y = 0, 0.01, 0.02, \ldots, 0.09$ ), crystals as thin as h = 20 nm, see SI. 11 in the supporting Information, were majority present in samples prepared at  $T_1 = 500$  °C,  $T_2 = 330$  °C,  $\Delta T = 80$  K. To quantify the reproducibility of optimal growth conditions, we determined thickness of BiSb ( $Te_{1-y}Se_y$ )<sub>3</sub> crystals from multiple samples (Fig. 9). Nanoflakes are the most common morphology with heights mostly between 20 and 40 nm or higher than 80 nm. The thickness of nanowires and nanoribbons are found to be mostly between 60 and 80 nm or higher than 80 nm.



**Fig. 9** Thicknesses of BiSb(Te<sub>1-y</sub>Se<sub>y</sub>)<sub>3</sub> nanostructures deposited by CVT. Several samples were considered for these statistics, samples prepared at  $T_1$  = 500 °C,  $T_2$  = 420 °C,  $\Delta T$  = 80 K, t = 4 h.

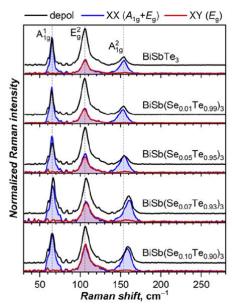
The local crystal structure of individual BSTS nanocrystals grown by CVT was confirmed by transmission electron microscopy. In particular, high-resolution transmission electron microscopy (HR-TEM) and selected-area electron diffraction (SAED) were performed. The results of BiSb(Te<sub>0.99</sub>Se<sub>0.01</sub>)<sub>3</sub> nanocrystals are shown in Fig. 10(a-c). Fig. 10(a) shows a bright-field TEM image revealing the morphology of high crystal quality with partially faceted edges. The different grey levels of certain regions indicate different layers thicknesses, whereas the dark lines can be attributed to bending contours, which might be caused by the transfer process of the crystals from the substrate to the TEM grid. The HRTEM image in Fig. 10(b) depicts a highly periodic arrangement of atomic columns in BiSb (Te<sub>0.99</sub>Se<sub>0.01</sub>)<sub>3</sub> with a determined lattice spacing of ~ 2.17 Å for the {110} lattice plane fringes, which is slightly larger than the expected value from XRD for BiSbTe<sub>3</sub> (2.165 Å). This slight difference may be attributed to the smaller ionic size of Se compared to Te. The good agreement between the experimental and the simulated SAED patterns a long zone-axis orientation [001] for  $BiSb(Te_{0.99}Se_{0.01})_3$  nanocrystals ( Fig. 10(c) ) confirms the high quality of the crystals. Consequently, HR-TEM and SAED investigations verify the hexagonal symmetry of the synthesized BiSb(Te<sub>0.99</sub>Se<sub>0.01</sub>)<sub>3</sub> that can be assigned to space group  $R\overline{3}m$ .



**Fig. 10** TEM and SAED images of BiSb( $Te_{0.99}Se_{0.01}$ )<sub>3</sub>. (a) Bright-field TEM image of a representative nanoflake on lacey carbon. (b) High-resolution TEM image showing the atomic arrangement of the crystal along the [001] zone axis in which the top right image (b) represents the zoom-in of the area marked by a red square (c) SAED pattern of the nanoflake (a) oriented along [001] zone axis (white) overlapped with a corresponding simulated pattern (red).

Raman spectra of BiSb( $Te_{1-y}Se_y$ )<sub>3</sub> (y = 0, 0.01, 0.05, 0.07, 0.10) nanocrystals were studied to understand the effect of stoichiometric ratio on Raman-active modes and local symmetry. According to XRD results, all these compounds are layered crystal assigned to the space group  $R\overline{3}m$ , the same as in parent compounds. The irreducible representation of the Raman active mode is described as  $\Gamma \text{vib} = 2A_{1g} + 2E_g$ . The lowest frequency Eg1 mode is usually very weak and can be rarely observed in experimental spectra<sup>53</sup>, while three other modes are found at 63 (A1g1), 103 (Eg2), 134 (A1g2) cm-1 in Bi2Te3 and at 69, 112, 165  $cm^{-1}\,in\,Sb_2Te_3.^{54}In\,solid\,solutions$  with stochastic distribution of Bi/Sb and Te/Se atoms, the formal symmetry should be lower and can potentially lead to a splitting of the peaks. However, experimental spectra of all studied BiSb(Te<sub>1</sub>-<sub>y</sub>Se<sub>y</sub>)<sub>3</sub> compositions still show three Raman peaks, although somewhat broadened (Fig. 11). Note that earlier Raman studies of three-component solutions also showed three peaks for most of the compositions.55-57 Polarized measurements were then performed to evaluate if  $E_g$  and  $A_{1g}$  modes partially mix in BiSb $(Te_{1-y}Se_y)_3$ . A1g modes should be active only in parallel (XX) configuration, while Eg modes are active in both parallel and cross-polarized (XY) geometries. The measurements revealed that the  $A_{1g}^{1}$  mode strictly follows the expected polarization pattern, but the  $A_{1g}^{2}$  mode shows residual intensity in the XY polarization geometry. Besides,  $E_g{}^2$  and  $A_{1g}{}^2$  peaks develop an asymmetry with tails towards each other. Thus, we conclude

that effective  $R\overline{3}m$  symmetry is still operative for vibrational modes, but signatures of the mode mixing are also evident. Peak positions of  $A_{1g}^1$ ,  $E_g^2$ , and  $A_{1g}^2$  modes in BiSbTe $_3$  at 65, 106, and 154 cm $^{-1}$  are intermediate between those in Bi $_2$ Te $_3$  and Sb $_2$ Te $_3$ . Substitution of Te by Se should induce a positive frequency shift, because Se atoms are lighter, smaller in size and stronger in electronegativity compared to Te, which leads to shortening of the chemical bonds between Se and Bi. In the experimental spectra, the shifts are hardly detectable up to the composition of y = 0.05, but become more apparent for a higher content of Se (Fig. 11). Upon increasing y to 0.10, the most pronounced shift of  $\sim$  5 cm $^{-1}$  is observed for the  $A_{1g}^2$  mode,  $E_g^2$  frequency increases by 1-2 cm $^{-1}$ , while  $A_{1g}^1$  remains



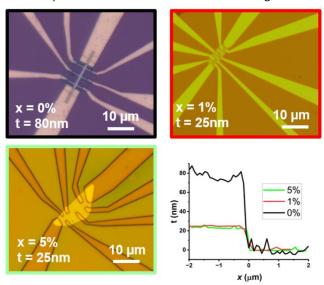
unchanged.

**Fig. 11** Raman spectra of different compositions of BiSb( $Te_{1-y}Se_y$ )<sub>3</sub> nanocrystals measured at the excitation wavelength of 660 nm. For each composition, depolarized spectrum (black) is compared to the spectra measured in parallel (XX, blue) and cross-polarized (XY, red) geometries.

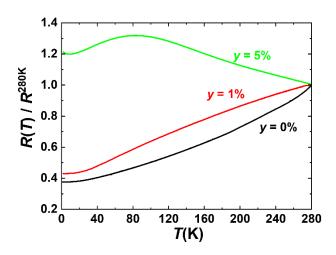
### 3.4 Magneto-Transport Measurements

Magneto-transport measurements were performed on individual nanostructures connected with Ti/Au contacts patterned in a Hall-bar geometry, down to about 2 K and applying a magnetic field up to 18 T, perpendicular to the sample plane. The transport properties of 25 nm-thin nanostructures were investigated in detail for two different compositions (y = 0.01 and 0.05) and compared to previous measurements made on a ternary BiSbTe<sub>3</sub> nanostructure.<sup>21</sup> Here we focus on three devices (Fig. 12), each one showing the typical behaviour observed for a given composition. First, the temperature dependence of the longitudinal resistance shows either a metal-like behaviour or a crossover from semiconducting to metal-like properties, depending on the relative contributions of the bulk and surface states. As shown

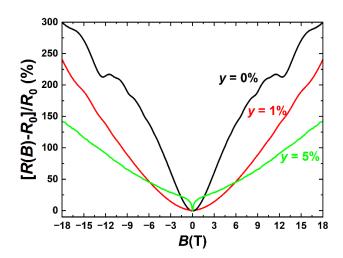
in Fig.13, the resistance ratio R(T)/R(280K), independent of geometric factors reveals the increased mobility in nanostructures with a low selenium content, giving a rather large residual resistance ratio R(280K)/R(2K). Given that the bulk density is reduced for these compounds, the metallic behaviour is dominated by surface-state transport, as confirmed by the large magneto-resistance measured at 2 K (Fig. 14) and the onset of Shubnikov-deHaas oscillations above 5 T. both revealing the enhanced mobility of charge carriers. On the contrary, nanostructures with a higher 5% selenium content have a larger bulk contribution to the total conductance, with a non-degenerate semiconductor behaviour, and a reduced average mobility. The reduction of the bulk-carrier density for BSTS nanostructures with a low Se content is directly evidenced from Hall measurements (Fig. 15), revealing the transition from p-type (for y = 0) to n-type transport (for y = 0.01), as well as a non-linear behaviour typical for multi-band transport and some plateaus in the transverse resistance due to the formation of Landau level in samples with enhanced mobilities. Small variations were found between different flakes with the same nominal composition (grown on the same substrate), as expected from their different thicknesses and partial bulkcarrier depletion at interfaces due to band bending.31



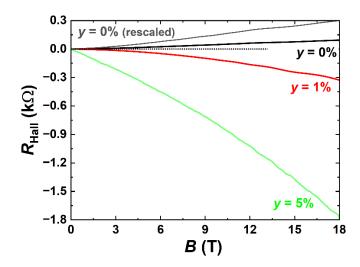
**Fig. 12** Optical-microscope images of the three electrically connected thin  $BiSb(Te_{1-y}Se_y)_3$  flakes and their respective AFM thickness profiles. Coloured frames of the images are corresponding to the respective plot colours.



**Fig. 13** Temperature dependence of the resistance, normalized to the value measured at 280 K, for three different compositions (y = 0%, y = 1%, y = 5%).



**Fig. 14** Longitudinal magneto-resistance, measured at 2 K, for three different compositions (y = 0%, y = 1%, y = 5%), showing the enhanced mobility for a small Se content (onset of Shubnikov-deHaas oscillations around 5 T).



**Fig. 15** Transverse magneto-resistance, measured at 2 K, for three different compositions (y = 0%, y = 1%, y = 5%), showing the evolution from p-type to n-type. charge transport. For comparison, the data obtained for y = 0% (80 nm) are also shown rescaled to an equivalent thickness of 25 nm.

#### 4. Conclusions

We successfully synthesized thin nanocrystals of BiSb(Te<sub>1-y</sub>Se<sub>y</sub>)<sub>3</sub> (y = 0.0,0.01,0.02...,0.09) using a transport agent-free CVT method and taking thermodynamic calculations into consideration. The structural and chemical compositions of the bulk starting materials were studied by powder X-ray diffraction and EDX, confirming a phase pure synthesis for all prepared compositions. The careful control of the synthesis conditions (source and substrate temperatures, temperature gradient, source-substrate distance, growth duration, and substrate material) allowed us to determine the optimal conditions for growing high-quality thin nanostructures. These have a smooth surface with well-faceted edges and metallic reflection, different morphologies (nanowires, nanoribbons, and planner triangular or hexagonal flakes), a thickness of 20-40 nm with lateral size range from  ${\sim}0.5\,\mu m$  to  ${\sim}20\,\mu m.$ Nanocrystals deposited on Si/SiO<sub>2</sub>, A- plane sapphire, Si, and BaF<sub>2</sub> substrates demonstrated different growth direction.

HRTEM /SAED and EDX data confirmed the high quality of the grown nanocrystals and the  $R\overline{3}m$  trigonal crystal lattice structure as well as the chemical compositions of BiSb(Te<sub>1-y</sub>Se<sub>y</sub>)<sub>3</sub> with ( y = 0, 0.01,0.02, ......,0.09). A systematic Raman spectroscopy study performed on different compositions of BiSb(Te<sub>1-y</sub>Se<sub>y</sub>)<sub>3</sub>, indicated that the vibrational modes are shifted to higher frequency for an increased in corporation of Se towards the substituting Te sites. Magneto-transport measurements revealed a large bulk-charge compensation in these quaternary-compound nanostructures, with dominant topological surface-state transport in thin nanoflakes, associated to a higher mobility of charge carriers.

# **Author Contributions**

N.A. and S.H. conceived and designed the project. N.A. conducted the synthesis of materials and analysis, optical, SEM, EDX, AFM and pXRD characterizations. S.F. completed the thermodynamic modelling. D.W. performed the HR-TEM measurements. A.P. and S.S. executed the RAMAN measurements. T.C, R.G and J.D. carried out magneto-transport measurements and analysis. The manuscript was written by N.A, based on contribution from all authors. S.H., B.B. and M.M. supervised the work.

#### Conflicts of interest

There are no conflicts to declare.

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