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▶ To cite this version:

Eduard Emil Iojoiu, Vincent Lauga, Johnny Abboud, Guillaume Legros, Alexis Matynia, et al.. Biofuel impact on Diesel engine after-treatment: deactivation mechanisms and soot reactivity. ETH conference "Combustion Generated Nanoparticles", Jun 2017, Zurich (CH), Switzerland. hal-04595928

HAL Id: hal-04595928 https://hal.science/hal-04595928v1

Submitted on 3 Jun 2024

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Biofuel impact on Diesel engine after-treatment: deactivation mechanisms and soot reactivity

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BACKGROUND: The new emission standards for diesel engines empower the need of complex and high efficient after-treatment systems, the durability being a crucial aspect. When biofuel is used, the after-treatment catalytic system is exposed to large amounts of poisons, the particles composition being as well impacted. The comprehension of the involved deactivation mechanisms as well as soot reactivity is a complex and multidisciplinary challenge.

METHODOLOGY: One focus was the study of the deactivation of the DOC and SCR catalysts through poisoning. Limited information is available about the physics and chemistry of the particles formed when biodiesel is used. The second focus was therefore devoted to the impact on soot reactivity in mechanistic and kinetic terms using model and real soot.



HOW SOOT LOADING & REGENERATION ARE IMPACTED BY USING BIOFUEL ?

Reactivity – Durability/ Characterization – Mechanisms – Ageing – Regeneration

The **DOC was exposed to Na and P impurities**, which are specific of the use of biofuel, being studied their impact on the catalytic performance.

BET specific surface area decreased and the total pore volume increased after addition of Na, due to the second hydrothermal treatment. In presence of P, partial blocking of the smallest mesopores decreasing pore volume was detected. TEM images showed a homogenous distribution of Na on the alumina bulk, which could modify the Al_2O_3 acid sites. On the other hand, bigger Pd-Pt particles were formed and a layer of phosphorus coated on the alumina surface was observed in the case of P-poisoned catalyst.

Catalytic results have shown that Na have a negative impact on CO oxidation, whereas an improvement can be observed in presence of P. In addition, P poisoned catalyst enhance C_3H_6 conversion, while Na impurities have the contrary effect. All elements have shown a negative effect on NO oxidation.





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The real soot samples were collected from filters operated on a **medium-duty truck in real driving conditions** or from engine bench using standard or 100% biofuel as well as **doped biofuel**. The use of biodiesel significantly reduces the soot formation. There is no significant impact of biodiesel on the soot specific surface, the higher value being obtained for an accelerated soot loading. Experiments performed at laboratory scale showed no relationship between volatile fraction and real soot reactivity under passive regeneration conditions. Adding alkali metals to the real Diesel soot enhances soot oxidative reactivity in the whole temperature range (200 – 600 °C), regardless the cycle of production applied. Those species act as catalyst for soot oxidation process. The kinetic of soot oxidation in presence of water in the feed gas, at low temperature (≤ 400 °C), is significantly increased in the presence of phosphorus.



DOC

Diesel

Oxidation

DPF

Diesel

Particulate

Filter

Model soot were collected on a glass microfiber filter in the post flame region of an atmospheric axisymmetric co-flow diffusion flame burner. Particle size distribution and oxidative reactivity of model soot from the burner are in the same range as real soot derived from Diesel engine.

	B7-Customer Cycle	1,84	3,91	0,05	0,1
	B100-Customer Cycle	7,78	5,56	0,05	0,1
	B100-High Loading	1,53	0,75	0,19	0,03
	B100(K+P)-High Loading	4,62	1,00	0,89	1,17





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The impact of Na and P on the SCR activity of Cu/FER catalyst was studied depending on the mineral loading (until $\approx 2_{wt}\%$) and the impregnation solvent (H_2O or ethanol). Acidity (measured by NH₃ adsorption) was poisoned after Na addition which directly affected the NOx conversion at low temperature (250°C). Na impregnation in water led to a stronger catalyst deactivation than in ethanol, because water favors the migration of the Cu exchanged species, leading to the formation of CuO extra framework species. It appears that the deNOx efficiency at high temperature (500°C) is clearly related to the amount of active exchanged copper. **Cu-P interactions were evidenced** after phosphorus addition, leading to a decrease in redox behaviors (NO oxidation and NH₃ oxidation) and consequently in the SCR activity (especially at low temperature and in Standard SCR condition). Again, deactivations were observed when the wet lower impregnations were performed in ethanol.



Catalytic Reduction

SCR

Selective

CONCLUSIONS

CONCLUSIONS		Global kinetic model -		NO & hydrocarbon
Impact of biofuel poisoning elements to DOC & SCR performance has been identified, Na having the bighest deactivation notential: SCR deactivation mechanism has been proposed	Collected data by academic partners combined with results obtained on engine bench and vehicles will be used to build	biofuel impact		oxidation
Comparison between real and model soot as well as the impact of Na, K and P on soot	kinetic models that will be integrated to empower an ageing	Chemical composition of	AGEING PREDICTION	NOx
oxidation have been successfully studied.	predictive model taking into account the use of biodiesel;	biofuel (Na, K, P, etc)		N ₂ O
Biofuel use leads to more poisoning of the oxidation catalysts, especially due to Na, with a direct impact on DOC performance in terms of NO formation which might indirectly affect	Adapted/innovative systems with improved poisoning resistance	Biofuel consumption		Soot oxidation
soot oxidation rate and NO_x reduction.				Asir ueposit

PERSPECTIVES

Financial support: French National Agency for Research ANR (Appibio Project, Ref. ANR-14-CE22-0003)

