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Chemical Disassembly and Recycling of Electronic Devices Manufactured with Biosourced Polymers

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38th International Conference of the Polymer Processing Society







PhD work at the INSA Lyon labs



3D Plastronics lab (AMPERE), Villeurbanne Plastics processing site (IMP), Bellignat Polymers characterization (IMP), Villeurbanne



Polymer materials processing, characterization and recycling







Antennas development and characterization



3D Plastronics



Funding from AURA Region





- 1. Context and objectives of the BIOANTENNA project
- 2. Polymer materials and IME technology
- 3. Scientific challenges
- 4. Experimental results
 - IME manufacturing
 - Polymer recycling

5. Conclusion







- IoT (Internet of Things) revolution towards direct interaction between objects thanks to an internet network
 - ➤ 30 billions IoT worldwide in 2020¹
- Why plastronic IoT antennas for energy storage and harvesting ?



<u>2019, France</u>: 1,5 billion batteries \approx **32 500 tons of toxic wastes**

Less than **50% collected 35% recycled 30% thrown** in domestic wastes → increasing environmental pollution



[1] Agence de la transition écologique (ADEME), *Piles et Accumulateurs – Rapport annuel 2020– Données 2019*, accessed 13/05/2021

[2] Maurizio CAPRA et al., *Edge Computing: A Survey On the Hardware Requirements in the Internet of Things World*, Department of Electrical, Electronics and Telecommunication Engineering, Politecnico di Torino, 10129 Torino TO, Italy, accessed 13/05/2021









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Polymer materials



- Two thermoplastic polymers:
 - Poly(Lactic Acid) (PLA), sustainable alternative for the environment
 - **PolyCarbonate (PC)**, reference material in IME
- PLA already used for:
 - Extrusion
 - 3D printing
 - Injection molding
- \rightarrow Interesting choice for IME



- Main objective: determine the compatibility of PLA at each step of the IME process
- <u>Secondary objective</u>: determine the recycling capability of plastronic devices made of PLA

Pictures sources accessed 28/03/2022: <u>https://www.3dnatives.com/asa-impression-3d-26082020/</u> https://french.alibaba.com/product-detail/high-quality-stretch-PLA-wrapping-film









- Used to give a 3D shape to the device which enables better integration of the desired electronic functions to the object
- Main steps of the process: screen printing, thermoforming and injection molding



- Process in late-stage development with amorphous PC
- Not studied with semi-crystalline PLA

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MANUFACTURING

- Compatibility of inks with PLA films (adhesion, low Tg of PLA \rightarrow During curing after inking?)
- Adhesion "polymer/ink+electronic components" during thermoforming?

Main scientific challenges

- Adhesion "polymer/polymer" and "circuit/polymer" during injection molding?
- Thermoformability of PLA with screen printed inks and glued components (cracks, delamination of inks on PLA)?
- Injection molding: thermal shrinkage inducing warping, early deterioration, ink delamination or component shearing?
- DISASSEMBLY and RECYCLING
 - Which method ? Mechanical, chemical, both?
 - What kind of value-added products do we want?
 - Solvents, reactants?

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[3] N. J. Teh et al., "Statistical optimisation of thermoplastic injection moulding process for the encapsulation of electronic subassembly," J. Electron. Manuf., Sep. 2000



Example of warping

after injection molding³











The IME process steps

Experimental work for manufacturing



38th International Conference of PPS, 22-26.05.2023

Before

 $1 \, \text{mm}$

 $1 \, \text{mm}$

10

 $1 \, \text{mm}$

Adhesion of screen-printed inks on PLA

- Conductive ink (silver particles)
- Drying on PC1: 120°C 20 min
- Drying on PLA1: 55°C 2h30
- Standardized cross-cut test ISO 2409
 - 6-blade cutter

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- Adhesive tape
- Very good adhesion
 - Category 0 in classification
 - Edges of the cuts smooth
 - No square of the lattice detached







Thermoforming of PLA



11



Thermoforming simulation with 375 µm thick PC film (T-SIM software)

Effect of geometrical parameters on film stretching

- Reduction in thickness to 174 μm
- Experimental:
 - PLA film well thermoformed: no cracks or holes
 - Ink slightly cracked in areas of greater deformations
 (→ smaller draft angle)



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- Cylinder-shaped
 - Fillet radius: 0,5 mm or 1 mm
 - Draft angle: 10° or 30°
- 15 mm-high double cone-shaped mold



Overmolding by injection PLA2 on PLA1 G2ELab

Central

injection

 \otimes

point



- Injection mold: 3-mm-thick plate, central injection point
- Polymer melt temperature: 195°C
- PLA1 screen printed films overmolded with PLA2 ۲



No wash-out of conductive ink, even close to the injection point

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- Injection mold at room temperature
- Screw speed: 60 mm/s



ME603 with electronic components

No wash-out, SMDs perfectly in place in plate mold area

Proof of concept

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Electronic circuit on PLA1, conductive ink dried at 55°C – LED on



After screen printing + SMDs pasting



After thermoforming





Electronic circuit well encapsulated





Disassembly and recycling

By chemical means



State of the art on PLA recycling



- Independent works of interest published:
- P. Coszach, J-C Bogaert, J. Willocq (2010, patent in French) → Chemical recycling of PLA by hydrolysis

Four steps :

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- 1. Grinding of PLA wastes
- 2. Dissolution of PLA in solvent
- 3. Chemical recycling in mild conditions to avoid PLA degradation
- 4. Purification of lactic acid
 - Leibfarth et al. (2012), Transforming Polylactide into value-added materials → study of 14 nucleophiles for the depolymerization of PLA by transesterification
 - Majgaonkar et al. (2021) \rightarrow Chemical recycling of PLA by alcoholysis

PLA + ethanol
$$\xrightarrow{\text{Acetone; TBD}}_{50^{\circ}\text{C; 1 bar}}$$
 Ethyl lactate + ethanol
(excess)

• Yield in ethyl lactate between 60% and 80% in 150 min of reaction



Ethyl lactate + water $\stackrel{\leftarrow}{\rightarrow}$ *lactic acid + ethanol*



PLA + water $\xrightarrow{\text{Ethyl lactate}}$ Lactic acid + water + ethyl lactate 130°C; 2.4 bars



> <u>Alcoholysis</u> : ethyl lactate as main product

> <u>Hydrolysis</u> : lactic acid as main product

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Alcoholysis



- Reaction at 60°C for 120 min ; magnetic stirring
- Main difference between the 2 reactions = nucleophile
 - Comparison ethanol 96° and absolute ethanol
- Effect on the production of ethyl lactate and lactic acid after 120 min







- In literature: for better hydrolysis reaction, use temperature above 100°C
- Test at 60°C for identical reaction as alcoholysis
 - Nucleophile = deionized water
- Effect on the production of lactic acid (and oligomers) after 240 min



Reaction occurs but very slow Lactic acid yield $\sim 4\%$

- Temperature not high enough
 Necessity to get closer to 100°C
- Kinetic study depending on temperature



Published article

Conclusion

- C. Goument *et al.* "In-Mold Electronics on Poly(Lactic Acid): towards a more sustainable mass production of plastronic devices". *Int J Adv Manuf Technol* (2023). <u>https://doi.org/10.1007/s00170-023-10878-4</u>
 - For the manufacturing steps of IME devices in PLA
- Recycling of Poly(Lactic Acid) from electronic devices
 - Alcoholysis
 - Optimized for PLA with mild conditions and non-toxic solvents
 - Hydrolysis
 - Study still ongoing with promising results
- Perspective
 - Life Cycle Assessment (LCA) to determine the most interesting method from an environmental point of view







Thank you for your attention!

Do you have questions?

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Appendix 1: PLA extrusion



- Laboratory-scale extrusion of PLA films
 - SCAMEX single-screw extruder
- Profile temperature 160-190-200-210°C from zone 1 to cast film die
- Screw speed 56 rpm
- Chill roll temperature 55°C
- Obtained films about 300 μm thick



Extrusion of PLA films at IMP Bellignat





Appendix 2: Study of polymer properties



,			Film		Injection	
	Analyses	Propriété	PLA1	PC1	PLA2	PC2
Thermal –	DSC	Tg (°C)	55	150	59	144
		Tm (°C)	151	/	171	/
	TGA	Td (°C)	310	455	310	455
		Residue (%)	1	25	1	24
Mecanical –	Tensile tests	Young Modulus (MPa)	2971	2132	/	/
		Tensile strength at break (N/mm ²)	59	65	/	/
		Elongation at break (%)	3,9	128,7	/	/
	Charpy	Impact resistance (kJ/m ²)	/	/	1,7 (notched)	8,9 (notched)
Surface –	Wettability	Roughness Sa (µm)	0,37	0,19	/	/
		Surface energy (mN/m)	37,8	45,8	45,7	46,8
		Dispersive (mN/m) Polar (mN/m)	32,4 5,4	44,4 1,5	39,9 5,7	43,6 3,2





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