

## Ce- and Ni-codoped Double PrBaMn<sub>2</sub>O<sub>5</sub> Perovskite as a New Ceramic SOFC Anode

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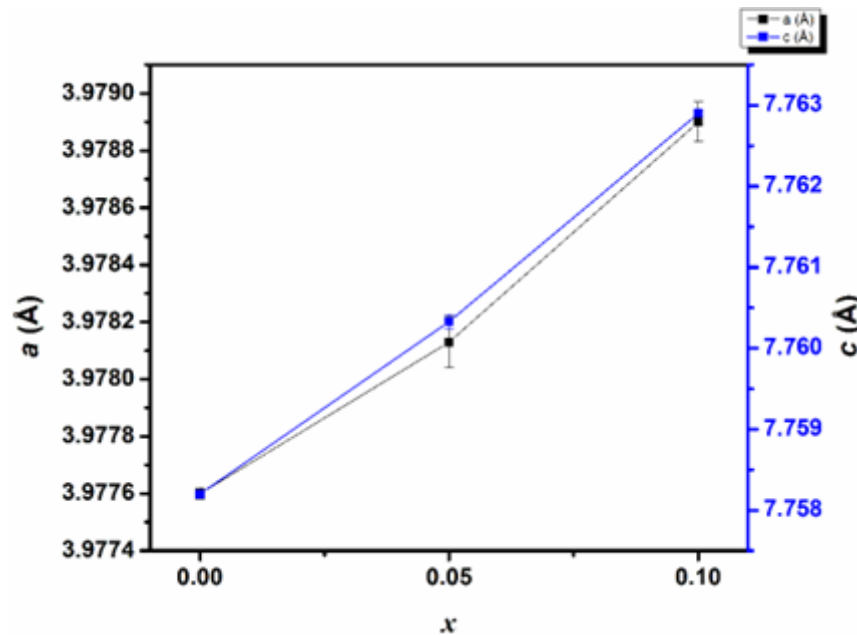


Figure S1. Variation of  $a$  and  $c$  lattice parameters for as-prepared Pr<sub>1-x</sub>Ce<sub>x</sub>BaMnO<sub>5</sub> as a function of cerium content.

Table S1. Lattice parameters of the as prepared layered perovskites (S.G.  $P4/mmm$ ).

	$a$ (Å)	$c$ (Å)	$V$ (Å <sup>3</sup> )	O1-Occupancy
<b>PBM-H</b>	3.9776 (2)	7.7582 (3)	122.74 (1)	0.1 (1)
<b>PrCe5-H</b>	3.97813 (6)	7.76033 (9)	122.811 (3)	0.03 (6)
<b>PrCe10-H</b>	3.97890 (7)	7.7629 (2)	122.899 (4)	0.04 (8)

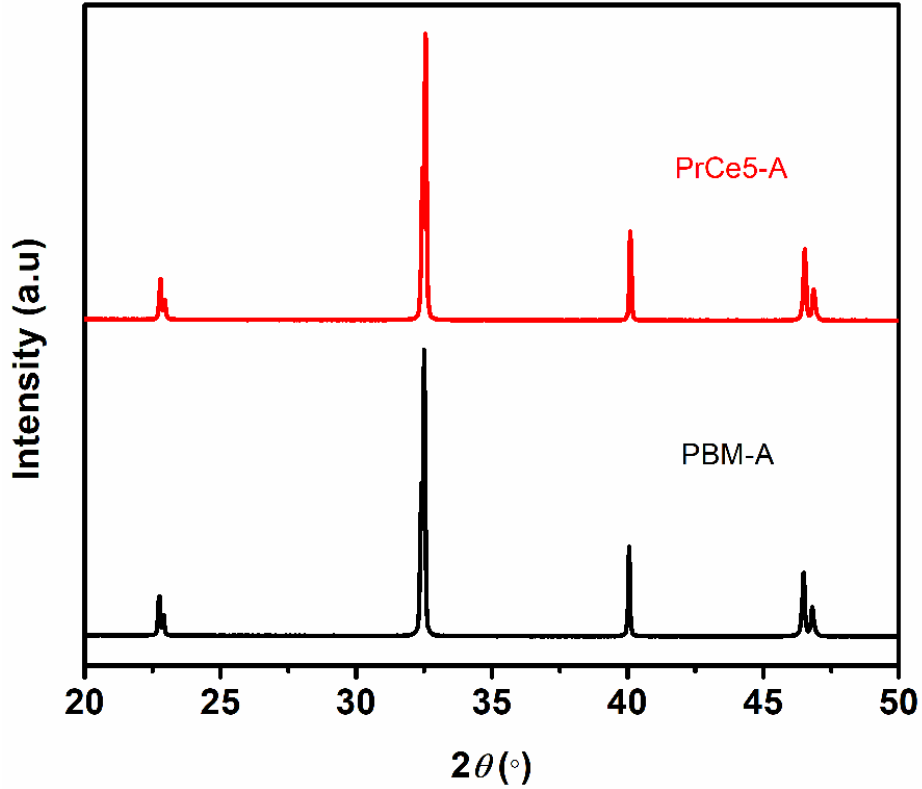


Figure S2. XRD patterns of the layered double perovskites oxidized in air at 800 °C.

Table S2. Structural parameters of the oxidized layered double perovskites (S.G.  $P4/mmm$ ).

	$a$ (Å)	$c$ (Å)	$V$ (Å <sup>3</sup> )
<b>PBM-A</b>	3.90151 (3)	7.75061 (6)	117.978 (3)
<b>PrCe5-A</b>	3.90201(3)	7.75075 (6)	118.010 (3)

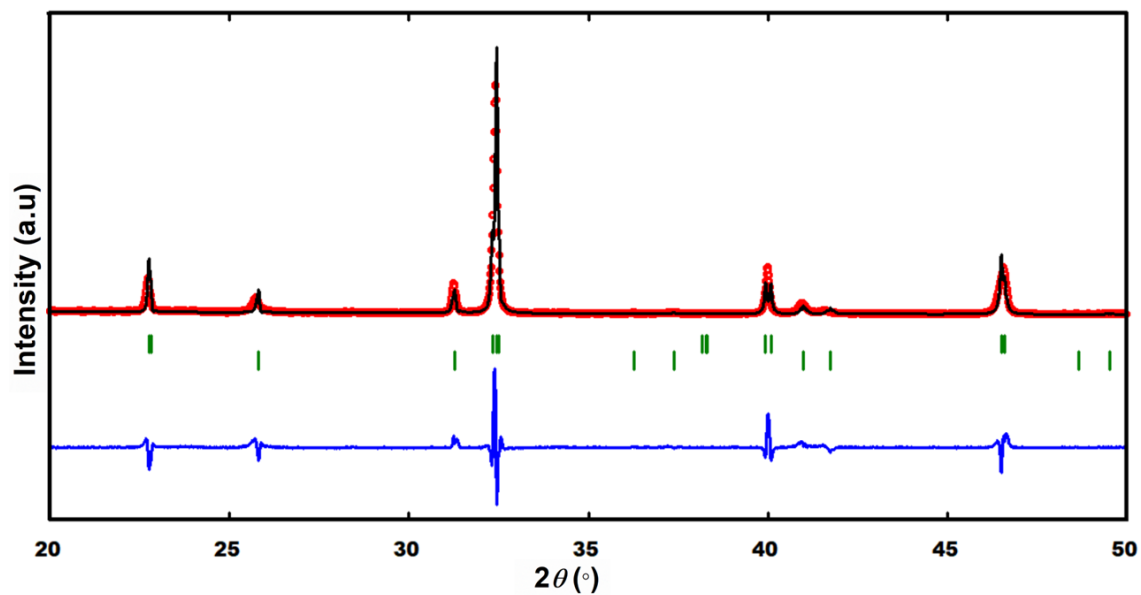


Figure S3. Rietveld refinement for as-prepared PrCe5Ni. Tick marks refer to (upper) (Pr,Ba)MnO<sub>3-δ</sub>, S.G. *Ibmm*, and (lower) 2H-BaMnO<sub>3-δ</sub>, S.G. *P6<sub>3</sub>/mmc* perovskites.

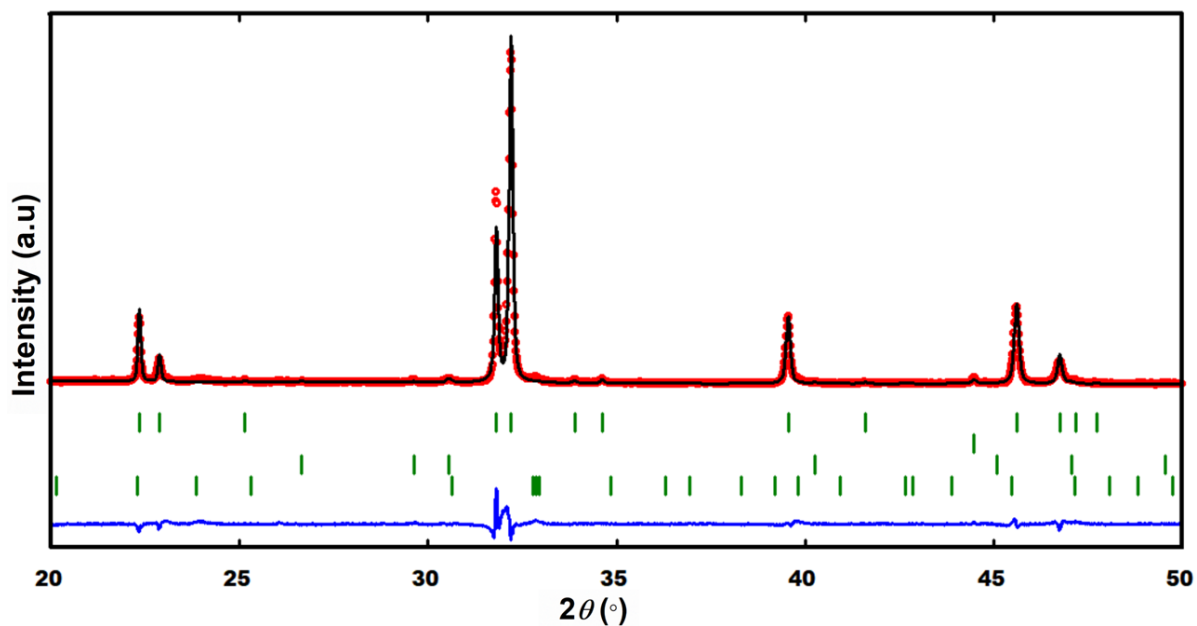


Figure S4. Rietveld refinement for PrCe5Ni after hydrogen reduction at  $T = 875$  °C. Tick marks refer to (upper row) PrCe5, (second row) Ni metal, (third row) Pr<sub>2</sub>O<sub>3</sub>, and (lower row) PrMnO<sub>3</sub>.

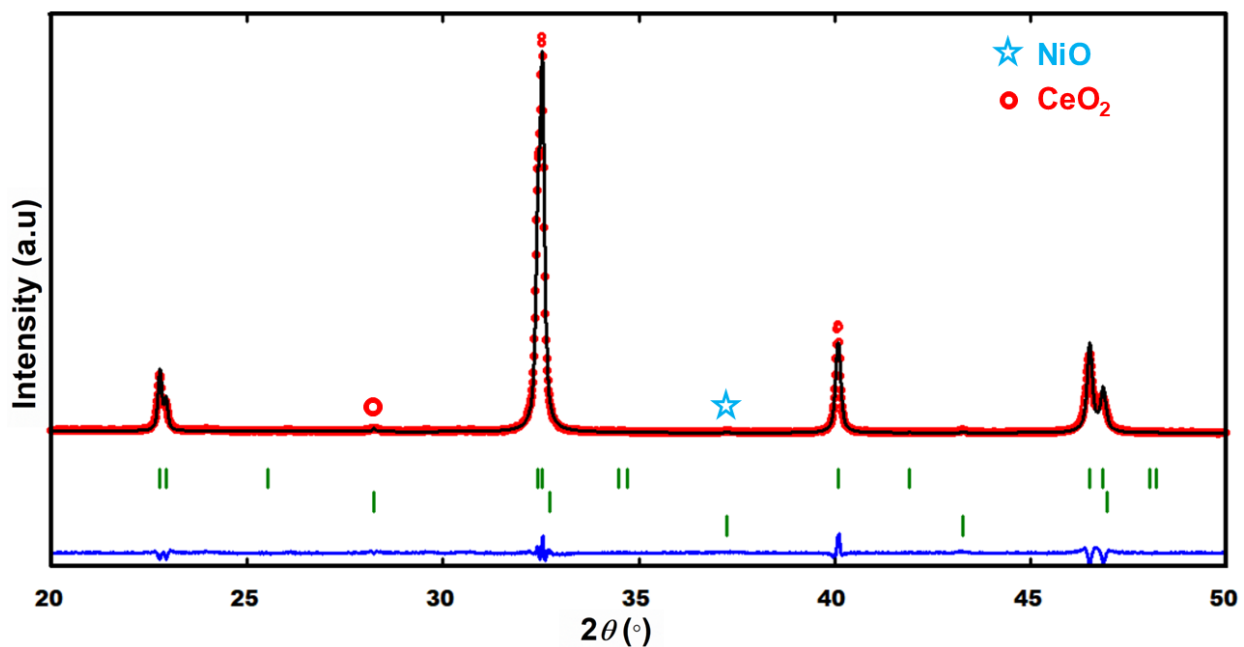


Figure S5. Rietveld refinement for the oxidized PrCe<sub>5</sub>Ni sample indicating the presence of trace amounts of (red circle) CeO<sub>2</sub> and (blue star) NiO.

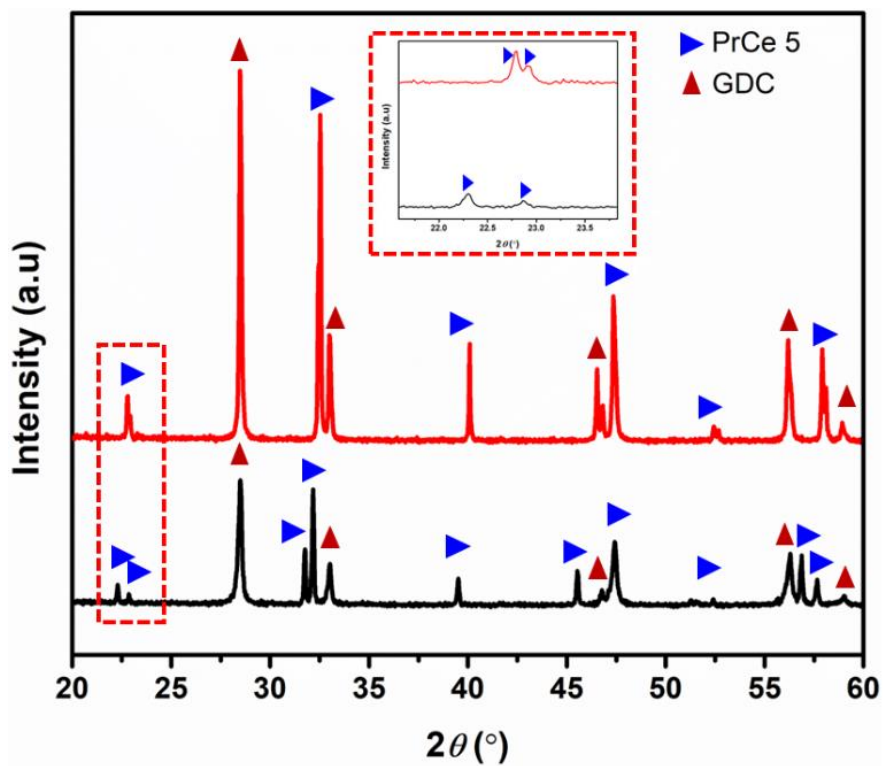


Figure S6. XRD patterns of PrCe<sub>5</sub>/GDC 50/50w/w (black) before and (red) after heating in air at 1200 °C.

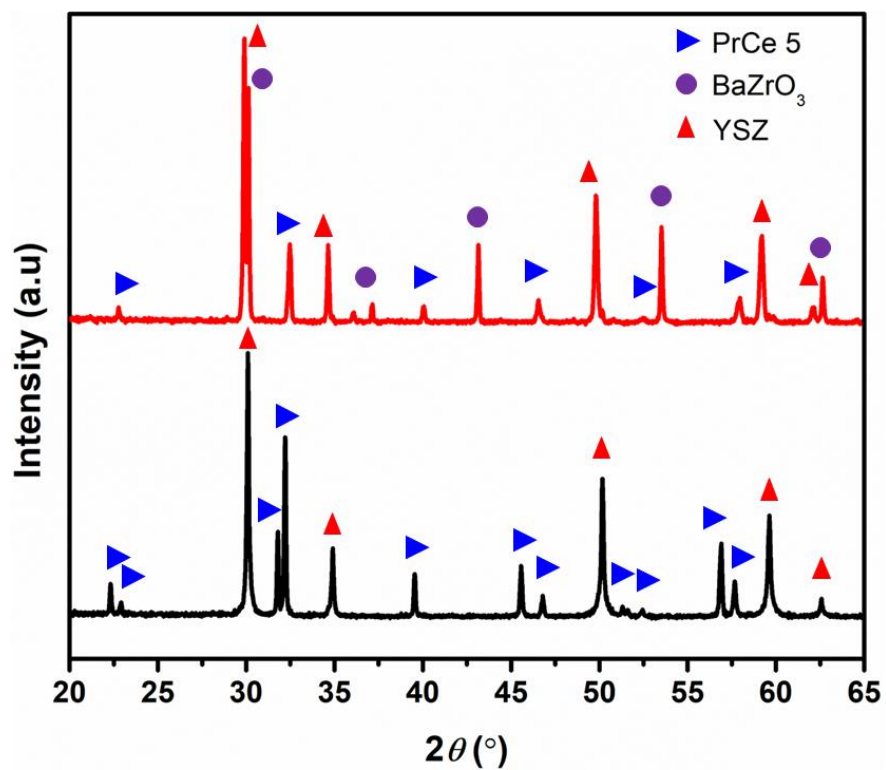


Figure S7. XRD patterns of reactivity tests for PrCe5/YSZ 50/50w/w (black) before and (red) after heating in air at 1200 °C for 3 h.

Table S3. Structural Parameters for PrCe5 determined by Rietveld analysis of NPD at 20 °C in S.G.  $P4/mmm$ ,  $a = 3.9653$  (1) Å,  $c = 7.7473$  (3) Å.

<i>atom</i>	<i>site</i>	<i>x</i>	<i>y</i>	<i>z</i>	$B_{iso}$ (Å <sup>2</sup> )	<i>occ.</i>
Pr/Ce	<i>1a</i>	0	0	0	1.0(2)	0.97/0.03
Ba	<i>1b</i>	0	0	0.5	2.2(2)	1
Mn	<i>2h</i>	0.5	0.5	0.2418(7)	0.9(1)	1
O1	<i>1c</i>	0.5	0.5	0	1.4(2) *	0.024(1)
O2	<i>1d</i>	0.5	0.5	0.5	1.4(2) *	1
O3	<i>4i</i>	0.5	0	0.1961(3)	1.1(0)	0.99(2)

\* Constrained to be the same.

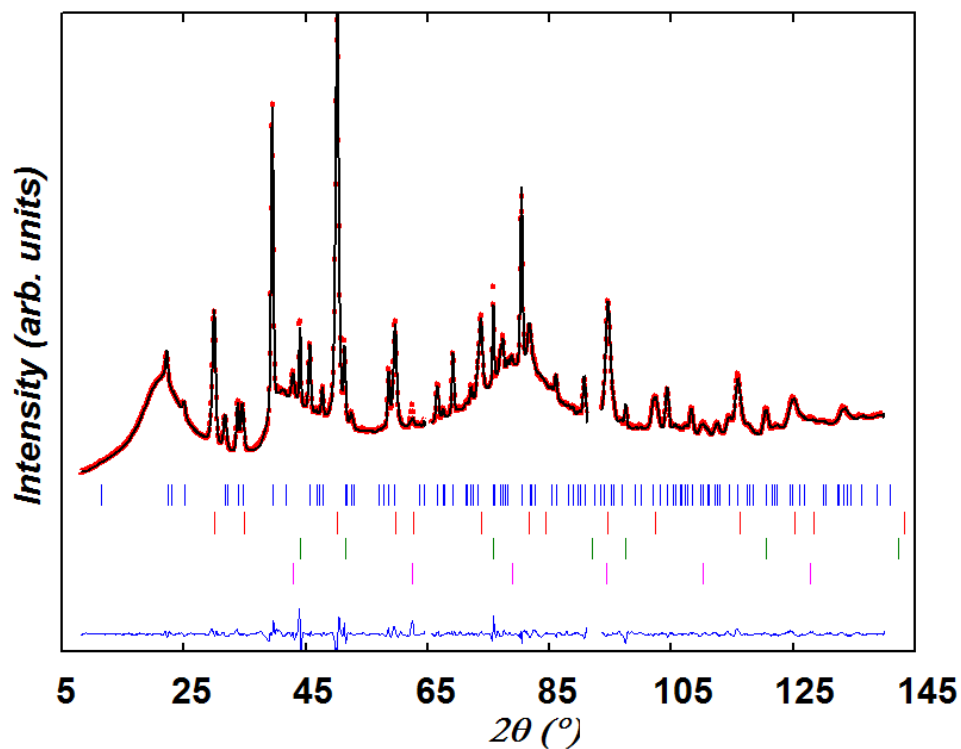


Figure S8. Neutron Powder Diffraction (NPD) pattern of PrCe<sub>5/8</sub>YSZ composite collected at 20 °C, using a wavelength of 1.54 Å. Tick marks indicate phases from PrCe<sub>5</sub> (first row) and YSZ (second row). The presence of the Chromel-Alumel thermocouple (S.G. *Fm-3m*, third row) and the vanadium heating element (S.G. *Im-3m*, fourth row) is also noted.

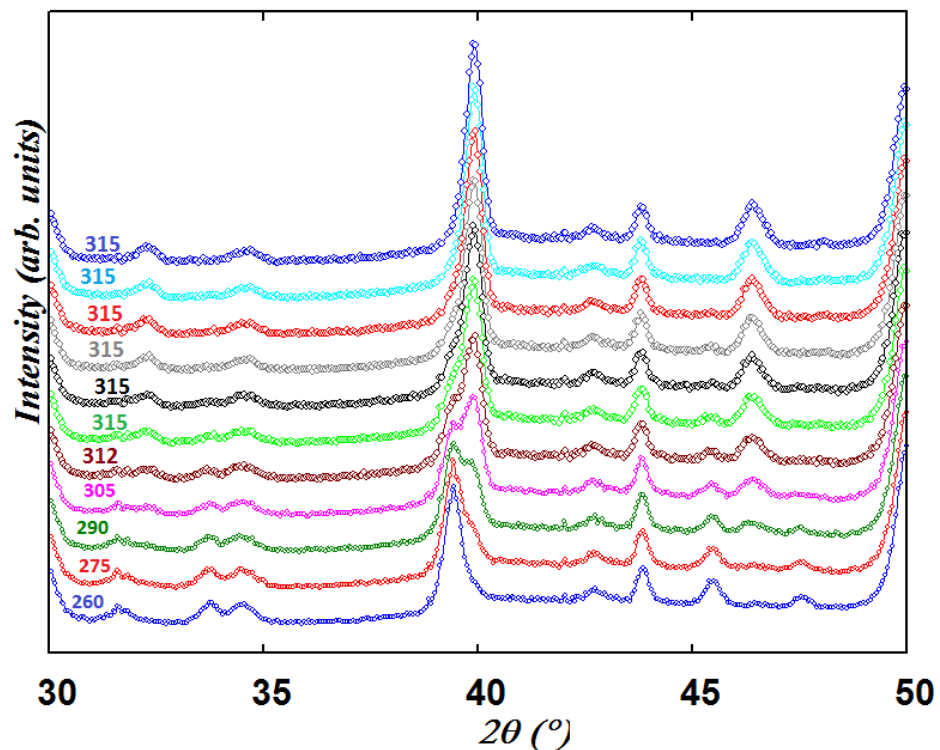


Figure S9. Neutron Powder Diffraction (NPD patterns of PrCe5/8YSZ composite collected in air within the oxidation temperature range using a wavelength of 1.54 Å. Temperature increases upwards.

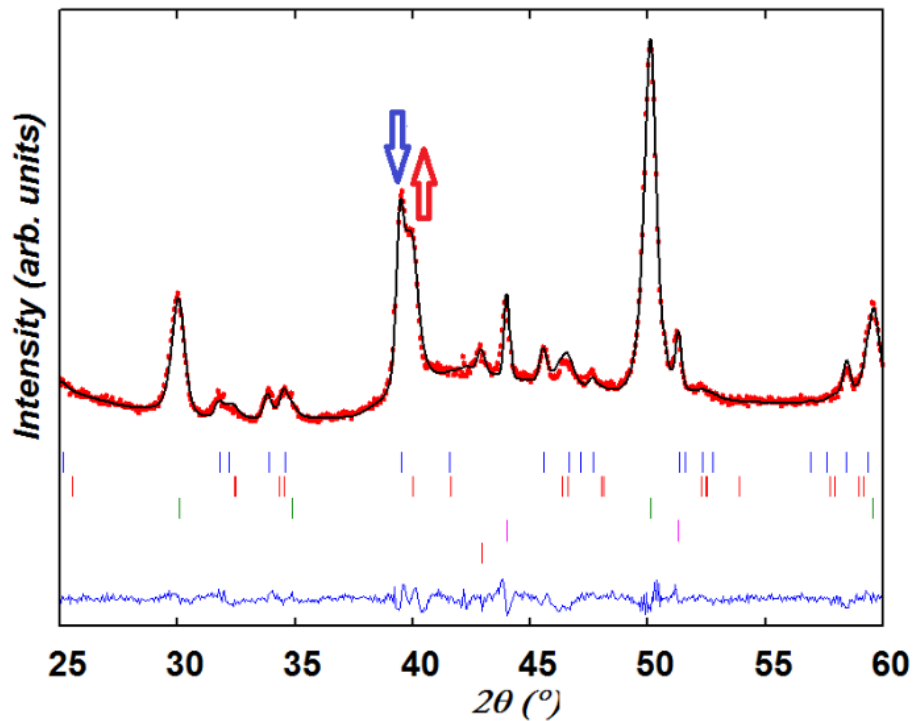


Figure S10. Rietveld analysis of the neutron data ( $\lambda = 1.54 \text{ \AA}$ ) collected at 290 °C upon heating in air the progression of reduced (blue, first row) and oxidized (red, second row) PrCe5 phases. Contributions from the YSZ electrolyte (3<sup>rd</sup> row), the thermocouple (4<sup>th</sup> row) and vanadium heating element (5<sup>th</sup> row) are shown.

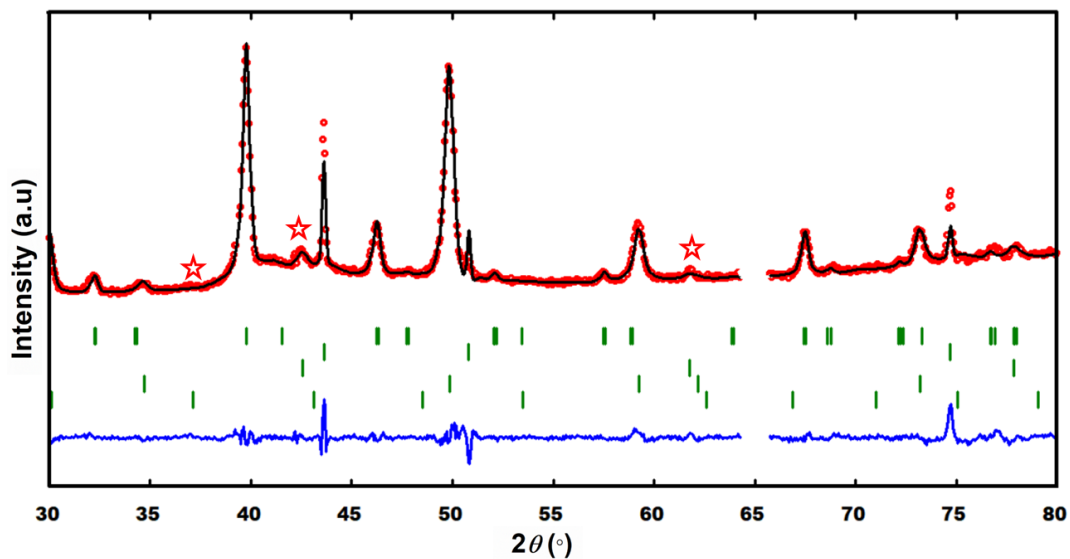


Figure S11. NPD patterns for PrCe<sub>5</sub>/8YSZ at  $T = 950\text{ }^{\circ}\text{C}$  highlighting main peaks from BaZrO<sub>3</sub> (4<sup>th</sup> row). The tick marks for PrCe<sub>5</sub> (1<sup>st</sup> row), 8YSZ (2<sup>nd</sup> row), and the thermocouple (3<sup>rd</sup> row) are delineated.

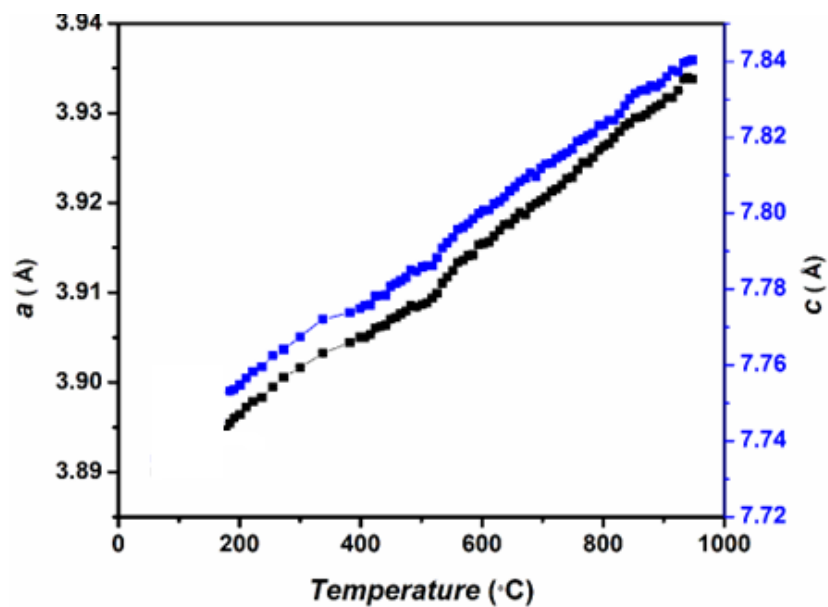


Figure S12. Lattice parameters  $a$  (in black) and  $c$  (in blue) for PrCe<sub>5</sub>, as determined from Neutron Powder Diffraction measurements during cooling in air.



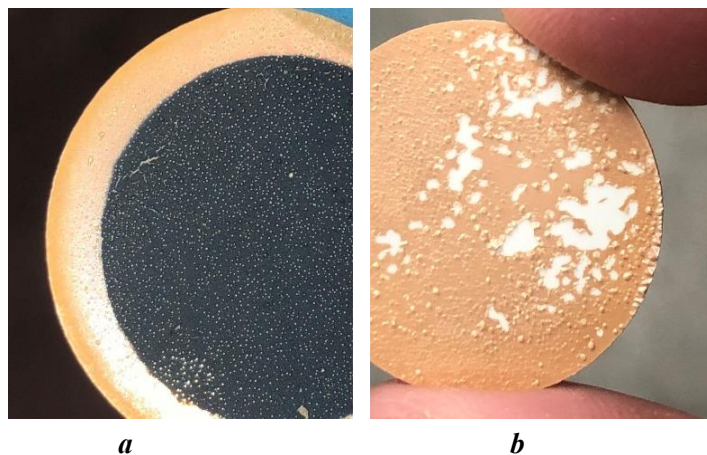


Figure S13. Comparison of 8  $\mu\text{m}$ -GDC buffer layer on YSZ substrate: (a) intact and uniform layer pre-testing, (b) delamination and peeling post-testing due to thermal expansion mismatch.

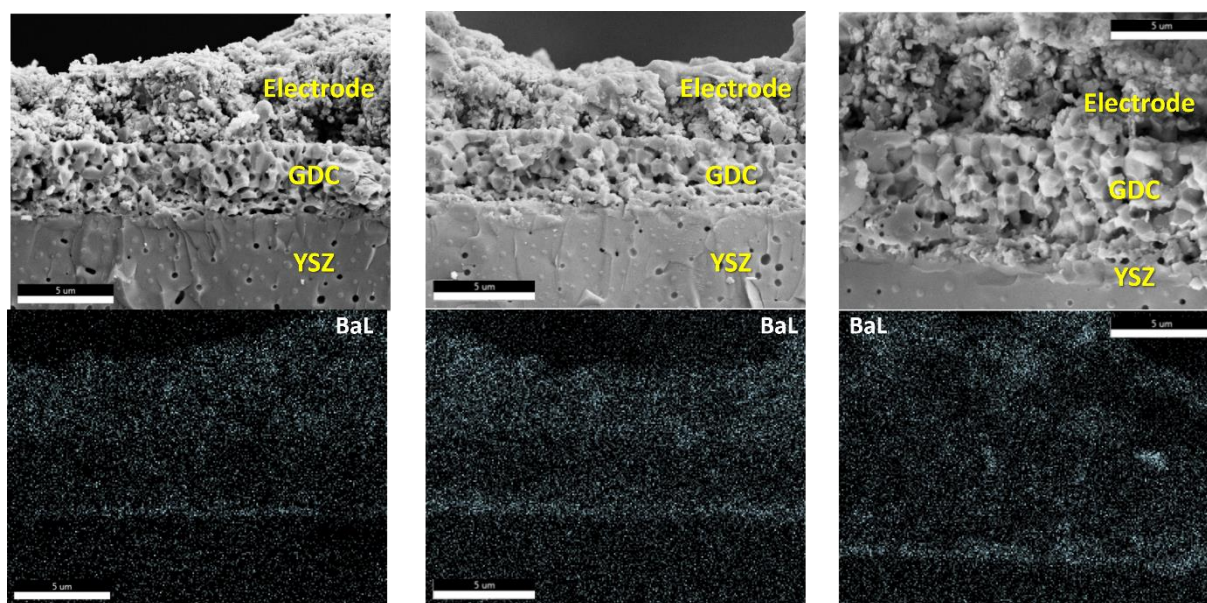
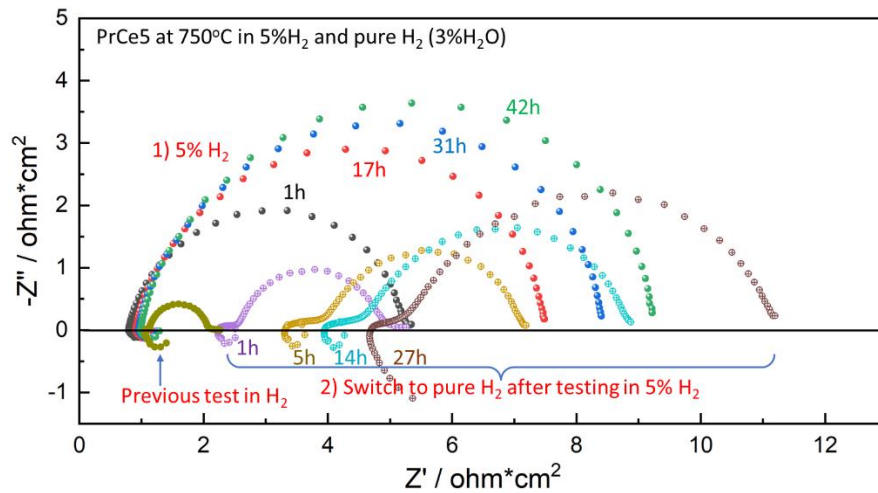
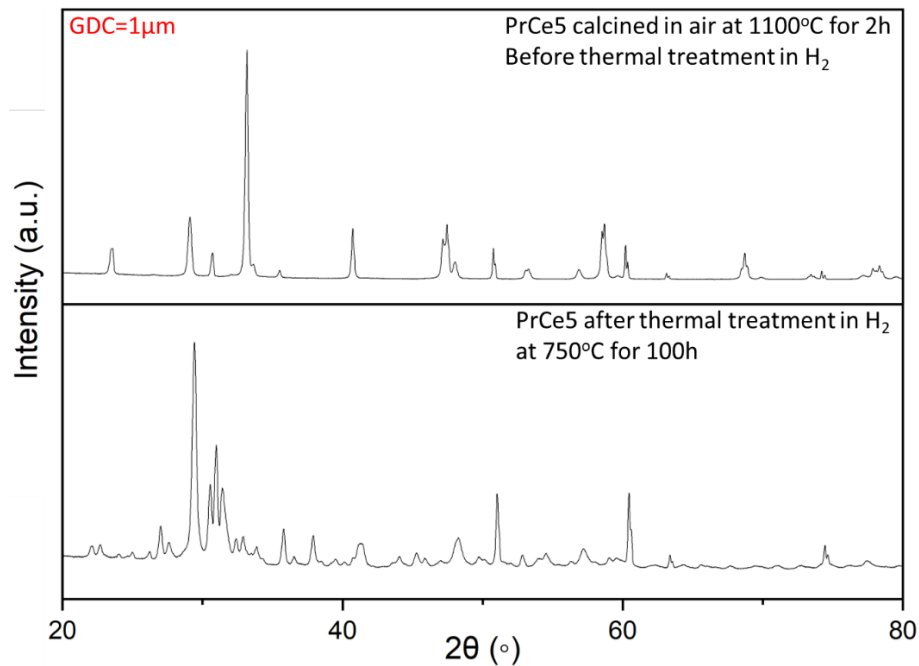


Figure S14. PBM electrode cross-section SEM images on 3.5-GDC coupled with EDS barium mapping taken from various baseline locations after 200-h hydrogen testing. The extent of Ba migration across the GDC buffer layer underscores potential impact on the electrode's stability and performance.



**Figure S15.** Nyquist plot for PrCe5 electrode screen-printed on a 1  $\mu\text{m}$ -GDC suggest that switching from pure  $\text{H}_2$  to 5%  $\text{H}_2$  at 750  $^\circ\text{C}$  results in increased polarization resistance and irreversible degradation, as evidenced by persistent ASR elevation and spectral shifts.



**Figure S16.** XRD before and after treatment in  $\text{H}_2$  for PrCe5 calcined on 1  $\mu\text{m}$ -GDC at 1100  $^\circ\text{C}$ . The degradation suggests the migration of the  $\text{Ba}^{2+}$  ions to the YSZ electrolyte facilitated by the slim GDC buffer layer.

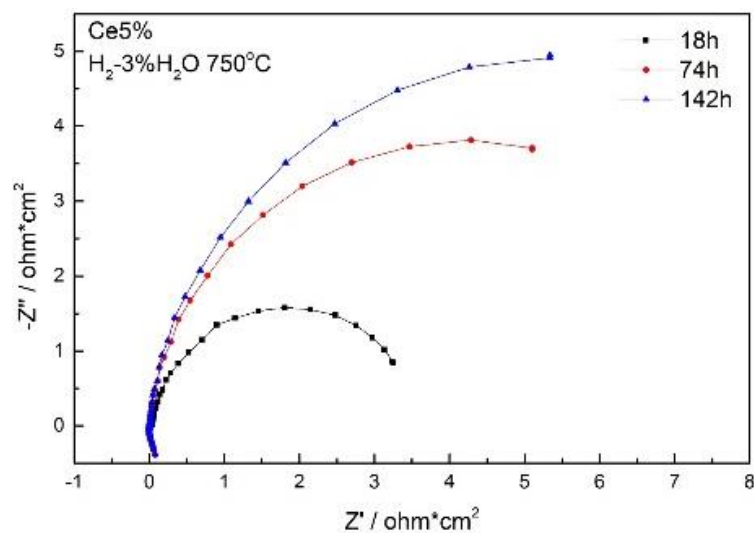


Figure S17. Time evolution of the electrochemical performance of PrCe5/8  $\mu\text{m}$ -GDC. The semicircular shape of the plots indicates a single relaxation process or a single time constant but the shift to the right as the exposure time increases suggests an increase in the material's resistance.

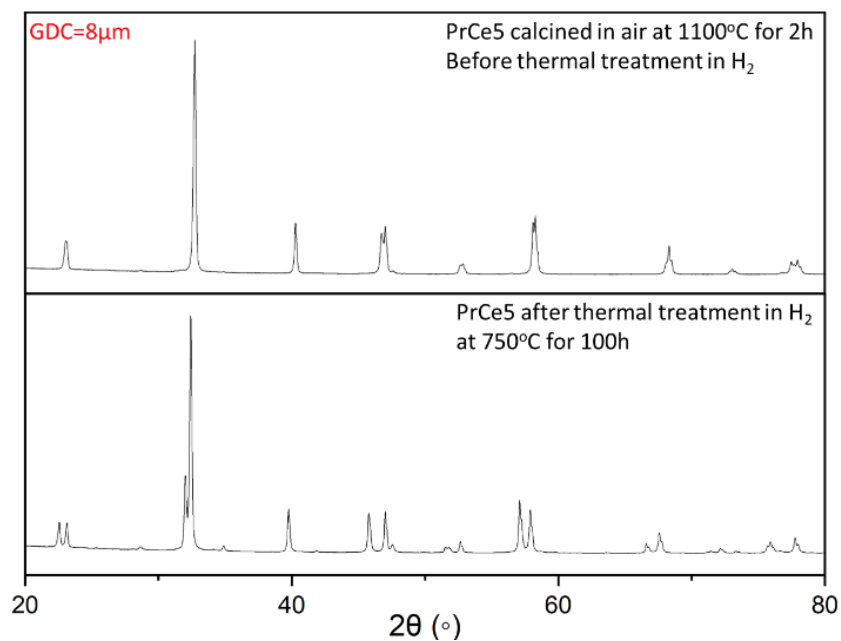
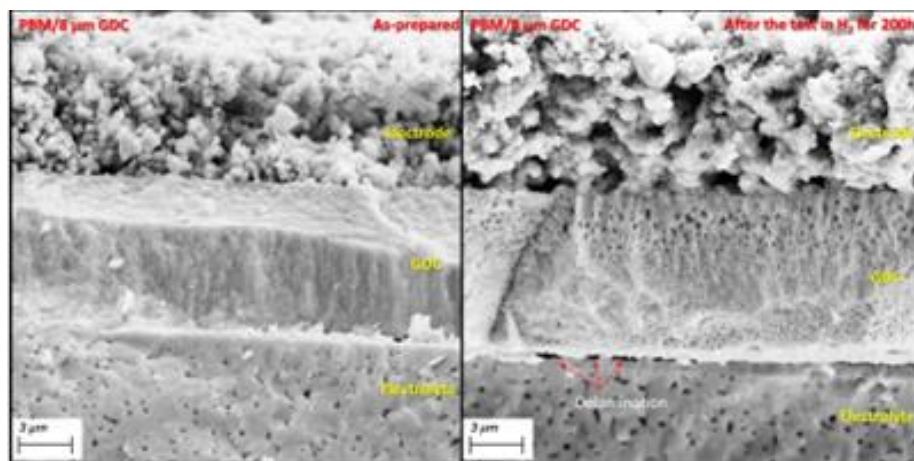


Figure S18. XRD patterns for PrCe5 electrode sintered on 8  $\mu\text{m}$ -GDC before and after treatment in H<sub>2</sub>. The comparable array of peaks suggests that the crystalline structure is retained and absence of new phases.



*a*

*b*

Figure S19. Cross-section SEM for PBM sintered on 8 μm-GDC before and after 200h H<sub>2</sub> testing.