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Vitalii Shtender, Volodymyr Smetana, Jean-Claude Crivello, Anatolii Kravets, Łukasz Gondek, et al.. Intermetallics of 4:4:1 and 3:3:1 series in La–(Co,Ni)–M (M = Bi, Pb, Te, Sb, Sn and Ga, Al) systems and their properties. *Journal of Alloys and Compounds*, 2024, 982, pp.173767. 10.1016/j.jallcom.2024.173767 . hal-04563665

HAL Id: hal-04563665

<https://hal.science/hal-04563665>

Submitted on 30 Apr 2024

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Intermetallics of 4:4:1 and 3:3:1 series in La–(Co,Ni)–*M* (*M* = Bi, Pb, Te, Sb, Sn and Ga, Al) systems and their properties

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Dedicated to the memory of Prof. Vitalij K. Pecharsky

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Abstract

Two series of isostructural intermetallics have been discovered in our search for new compounds with fused honeycomb motifs, both stable at elevated temperatures (1073 K). Both crystallize with orthorhombic unit cells – La₄Co₄*M* (*M* = Sn, Sb, Te, Pb, Bi, SG *Pbam*, *a* = 8.247–8.315(2), *b* = 21.913–22.137(7), *c* = 4.750–4.664(2) Å, *V* = 850.5–869.5(4) Å³, *Z* = 4) and La₃Ni₃*M* (*M* = Al, Ga, SG *Cmcm*, *a* = 4.1790–4.2395(1), *b* = 10.4921–10.6426(6), *c* = 13.6399–13.7616(8) Å, *V* = 606.72–612.05(7), *Z* = 3). The crystal structures represent interesting variations of semiregular tilings of corrugated anionic layers and predominantly cationic zigzag motifs. The La₄Co₄*M* compounds reveal a complex type of ordering with a high degree of frustration as could be expected for the Kagomé-related lattices, while magnetic ordering in the La₃Ni₃*M* series is less evident. Electronic structure calculations have been performed for multiple compounds within both series revealing metallic character and visible local minima around the Fermi level. The bonding picture is characterized by nearly equal contributions from the anionic and the cationic components.

Keywords: *Intermetallics, crystal structure, electronic structure, magnetic properties.*

1. Introduction

Polar intermetallic compounds have been extensively investigated exhibiting broad compositional variability and enormous structural complexity [1]. While usually polar intermetallics include a wide range of compounds extending from the metal alloys to Zintl compounds [2], considerable attention has been paid to the middle range, compounds with the valence electron count of around 2 [3, 4]. This group is mostly represented by ternary compounds including active or rare-earth metals, transition metals and a *p*-block element including some metalloids [5-8]. Among them, trielides and tetrelides are the most frequent and offer a plethora of structural motifs of all dimensionalities, *i.e.* from clusters to networks and peculiar bonding features [1]. Besides compositional wealth such element combination is also interesting in terms of properties. Superconductivity has been detected for many *p* elements and their compounds, *e.g.* the majority of the La-Sn binary compounds [9-11]. On the other hand, rare earth elements, with their high number of unpaired *f*-electrons, are most important for introducing magnetism. Adding transition metals allows, aside from magnetism, to add additional features including electronic and geometric adjustments leading to interesting properties and unique property combinations. For example, though containing a magnetic element Co, LaCoSi exhibits superconductivity 4 K [12]. It is also known as a catalyst for NH₃ production [13].

Until now, none of La-(Co,Ni)-*M* (*M* = Bi, Pb, Te, Sb, Sn and Ga, Al) ternary systems have been fully explored [14]. So far, systems with Bi and Te have only one representative each: La₆Co₁₃Bi [15] and La₆CoTe₂ [16]. Systems with Pb and Sb are presented with three and four compounds, respectively: La₆Co₁₃Pb [15], La₅CoPb₃ [17], La₁₂Co₆Pb [18], and La₆Co₁₃Sb [15], LaCo_{*x*}Sb₂ [19], LaCoSb₃ [20], La_{*x*}Co₄Sb_{12-*y*} [21]. For the Sn-based system, six compounds were already reported: La₆Co₁₃Sn [15], La₄Co₂Sn₅ [22], La₁₁₇Co₆₇Sn₁₁₃ [23], LaCo_{*x*}Sn₂, La₃Co₂Sn₇ [24] and La₃Co₄Sn₁₃ [25]. In contrast, the La-Co-*M* (*M* = Bi, Pb, Te, Sb and Sn) systems, and the La-Ni-*M* (*M* = Ga and Al) systems were studied to a wider extent. For instance, the Ga-based systems presented by: La₃Ni₂Ga₂, La₃Ni₄Ga₂ [26], La₂Ni_{2.25}Ga_{0.75} [27], LaNi₄Ga [28], LaNiGa, LaNi₁₁Ga₂ [29], LaNi₃Ga₂, LaNiGa₂ [30], LaNi_{2.4}Ga_{2.6} [31], LaNi_{2-*x*}Ga_{2+*x*}, LaNi₉Ga₄ and LaNi_{5.5}Ga_{7.5} [29], LaNi_{7.25}Ga_{5.75} [29], La₂NiGa₁₂ [32], La₂NiGa₁₀ [33] and La₂Ni₁₃Ga₁₃ [29]. Finally, the system with Al so far is presented with: LaNi_{5-*x*}Al_{*x*} [34], La₂Ni_{2.25}Al_{0.75} [27], La₅Ni₂Al₃ [35], La₃Ni₄Al₂ [36], LaNiAl [37], LaNiAl₂ [38], LaNi₂Al₃ [39], LaNi₂Al₅ [40], LaNi_{1.44}Al_{5.56} [41], and La₂NiAl₇ [42].

Despite the number of discovered compounds, their physical or chemical properties have been investigated to an even lesser extent. The most interesting are ferromagnetism in La₆Co₁₃*M* [15] and La₄Co₂Sn₅ [22], low thermal conductivity in the La-filled cobalt antimonide Skutterudites La_{*x*}Co₄Sb₁₂ [43], superconductivity in La₃Co₄Sn₁₃, and largely positive and

nonsaturating magnetoresistance of $\text{La}_2\text{NiGa}_{12}$ [32]. Some other compounds, e.g. LaNi_4Ga [44] or LaNi_4Al [45] showed good hydrogen sorption efficiency. In the latter case, hydrogenation influenced the magnetic properties leading to ferromagnetism [46]. LaNi_3Ga_2 reveals spin fluctuation [47] while excellent MCE (magnetocaloric effect) performance has been observed in a series of R_2T_2X [48]. Magnetic properties have also been inspected in $\text{LaNi}_{0.75}\text{Ga}_{3.25}$ [49], LaNiAl_2 [50], $R_2\text{Ni}_2\text{Ga}$ and $R_2\text{Ni}_2\text{Al}$ [27].

In the present work, we explored two series of isotypic ternary compounds in the $\text{La}-(\text{Co},\text{Ni})-M$ systems *i.e.* La_4Co_4M and La_3Ni_3M ($M = \text{Bi}, \text{Pb}, \text{Te}, \text{Sb}, \text{Sn}$ and Ga, Al). Although isocompositional compounds are quite common in these systems, it is interesting to note that the discovery of the La_4Co_4M series brought us to the fact that such types of compounds can be obtained with ‘antagonistic pairs’ [51] (like Co-Pb or Co-Bi). Following the structural studies, magnetic behavior as well as electronic structures have been investigated for both series.

2. Experimental part and theoretical methods

The $\text{La}_4\text{Co}_4\text{Sn}$ and La_3Ni_3M ($M = \text{Ga}$ and Al) alloys (1g each) were synthesized by arc-melting of high purity La (99.9%), Ni and Co (99.99%), Sn , Ga and Al (99.999%) with further annealing at 800 °C in vacuum. The observed weight losses never exceeded 0.5 %. The La_4Co_4M ($M = \text{Bi}, \text{Pb}, \text{Te}$ and Sb) alloys were prepared from arc-melted La_4Co_4 ligature and $\text{Bi}, \text{Pb}, \text{Te}$ and Sb (all with 99.9% purity) powders to avoid evaporation losses during heating. The ligatures were crushed and mixed with the respective M -metals. Such mixtures were loaded into Ta tubes and sealed under Ar atmosphere. Ta tubes were afterward placed in stainless steel tubes to protect them from oxidation. The samples were heated up to 1073 K for 10h, kept for 10h at 1073 K and cooled to 773 K at 3 K/h. Final cooling has been performed by switching off the furnace.

Intensity data sets for powder X-ray diffraction (PXRD) were recorded at room temperature using a Bruker D8 X-ray diffractometer with a Lynx-eye position sensitive detector and $\text{Cu-K}\alpha$ radiation on a zero-background single crystal Si sample holder. Phase analyses using the Rietveld method [52] of the powder X-ray data were performed using Topas 6 software [53]. Rietveld refinements of selected samples are presented in Supporting Information (SI) **Figure S1**.

Single-crystal X-ray diffraction (SCXRD) data were collected at 293 K on a Bruker D8 Venture diffractometer (Bruker, USA; Photon 100 CMOS detector, $\text{I}\mu\text{S}$ microfocus source: $\text{Mo K}\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$, 2–5). Intensity data sets of reflections and scaling were integrated within the APEX3 software package by using SAINT [54]. Absorption corrections were conducted with SADABS [55] and crystal structure solutions with SHELXT [56]. For subsequent difference, Fourier analyses and least-squares refinements, SHELXL-2013 [57] was

used. The experimental details of the crystal structure determination and refinement as well as the atomic coordinates for $\text{La}_4\text{Co}_4\text{Bi}$ and $\text{La}_3\text{Ni}_3\text{Ga}$ as representative examples have been collected in **Tables 1** and **2**.

The microstructure was evaluated with a Zeiss Merlin SEM equipped with a secondary electron (SE) detector and an energy-dispersive X-ray (EDX) spectrometer. The samples for electron microscopy analysis were prepared by standard metallographic techniques through grinding with SiC paper. For the final polishing, a mixture of SiO_2 and H_2O was used. SEM micrographs are presented in **Figure S2**.

Magnetization measurements were conducted using a Physical Property Measurement System (PPMS, Quantum Design, USA). Vibrating Sample Magnetometer (VSM) options were utilized to measure zero-field cooled (ZFC) and field-cooled (FC) magnetization between 2 and 300 K in static fields (DC) up to 7 T. Isothermal magnetization was acquired in applied fields up to 7 T. Polycrystalline samples were filled into polypropylene (PP) capsules, which were mounted into a brass sample holder.

The First-Principles calculations have been led in a similar way to the previous work [58], using the VASP code [59, 60], *i.e.* the electronic properties were determined through Density Functional Theory (DFT) using a pseudopotential approach within the projector augmented-wave (PAW) method, employing collinear spin polarization. The Perdew-Burke-Ernzerhof (PBE) functional [61] in the generalized gradient approximation (GGA), was employed with a cutoff energy of 600 eV. Structural relaxations, encompassing volume, cell shape, and atomic positions, were performed while maintaining the original symmetry, with a convergence criterion set at forces less than 0.05 meV per Å. Crystal orbital Hamiltonian population (COHP) curves were extracted by the Lobster program [62]. Subsequently, phonon calculations were conducted within the harmonic approximation for $\text{La}_3\text{Ni}_3\text{Ga}$ and $\text{La}_4\text{Co}_4\text{Bi}$ using a supercell approach ($2\times 2\times 1$, $1\times 1\times 2$ respectively), as implemented with the Phonopy code [63]. Charge transfer analyses were carried out following Bader's method [64].

Table 1. Crystallographic data and experimental details of the structure determination for the La_4Co_4M and La_3Ni_3M ($M = \text{Bi, Pb, Te, Sb, Sn}$ and Ga, Al). Experiments were carried out at 293 K with Mo $K\alpha$ radiation.

Empirical formula	$\text{La}_4\text{Co}_4\text{Bi}$	$\text{La}_4\text{Co}_4\text{Pb}$	$\text{La}_4\text{Co}_4\text{Te}$	$\text{La}_4\text{Co}_4\text{Sb}$	$\text{La}_4\text{Co}_4\text{Sn}$	$\text{La}_3\text{Ni}_3\text{Al}$	$\text{La}_3\text{Ni}_3\text{Ga}$
CSD	2301799	2301803	2301805	2301804	2301802	2301800	2301801
Composition at.% from EDX	$\text{La}_{45.3(3)}\text{Co}_{43.4(3)}\text{Bi}_{11.3(1)}$	$\text{La}_{45.3(3)}\text{Co}_{42.7(6)}\text{Pb}_{12.0(6)}$	$\text{La}_{45.4(5)}\text{Co}_{43.4(3)}\text{Te}_{11.3(3)}$	$\text{La}_{45.1(2)}\text{Co}_{43.6(4)}\text{Sb}_{11.3(3)}$	$\text{La}_{45.0(2)}\text{Co}_{43.8(2)}\text{Sn}_{11.3(2)}$	$\text{La}_{44.0(4)}\text{Ni}_{41.7(6)}\text{Al}_{14.3(3)}$	$\text{La}_{42.8(5)}\text{Ni}_{42.8(4)}\text{Ga}_{14.4(5)}$
Structure type	Own	$\text{La}_4\text{Co}_4\text{Bi}$	$\text{La}_4\text{Co}_4\text{Bi}$	$\text{La}_4\text{Co}_4\text{Bi}$	$\text{La}_4\text{Co}_4\text{Bi}$	Er_3Ge_4	Er_3Ge_4
Formula weight, Mr (g/mol)	1000.34	998.55	918.96	913.11	910.05	619.84	662.58
Space group (No.)	<i>Pbam</i> (55)	<i>Pbam</i> (55)	<i>Pbam</i> (55)	<i>Pbam</i> (55)	<i>Pbam</i> (55)	<i>Cmcm</i> (63)	<i>Cmcm</i> (63)
Pearson symbol, Z	<i>oP</i> 36, 4	<i>oP</i> 36, 4	<i>oP</i> 36, 4	<i>oP</i> 36, 4	<i>oP</i> 36, 4	<i>oS</i> 28, 4	<i>oS</i> 28, 4
<i>a</i> , Å	8.315(1)	8.269(3)	8.274(3)	8.249(2)	8.247(2)	4.1790(1)	4.2395(3)
<i>b</i> , Å	22.137(4)	21.961(7)	22.0394(4)	21.930(3)	21.913(7)	10.6426(3)	10.4921(6)
<i>c</i> , Å	4.724(1)	4.749(2)	4.6638(8)	4.7060(8)	4.745(1)	13.7616(4)	13.6399(8)
<i>V</i> , Å ³	869.5(3)	862.5(5)	850.5(4)	851.3(3)	857.5(4)	612.05(3)	606.72(7)
Calculated density, ρ (g·cm ⁻³)	7.64	7.69	7.18	7.12	7.05	6.73	7.25
Absorption coefficient, μ (mm ⁻¹)	46.67	46.17	30.48	30.19	29.74	29.50	33.96
Theta range (°)	2.617 ÷ 27.494	2.632 ÷ 30.000	1.848 ÷ 32.679	2.638 ÷ 30.505	2.639 ÷ 30.664	2.960 ÷ 29.978	2.987 ÷ 29.975
<i>F</i> (000)	1676	1672	1552	1548	1544	1072	1144
Range in <i>h k l</i>	-10 ≤ <i>h</i> ≤ 8, -28 ≤ <i>k</i> ≤ 23, -4 ≤ <i>l</i> ≤ 6	-11 ≤ <i>h</i> ≤ 11, -26 ≤ <i>k</i> ≤ 29, -6 ≤ <i>l</i> ≤ 6	-12 ≤ <i>h</i> ≤ 12, -33 ≤ <i>k</i> ≤ 33, -6 ≤ <i>l</i> ≤ 7	-9 ≤ <i>h</i> ≤ 11, -24 ≤ <i>k</i> ≤ 31, -6 ≤ <i>l</i> ≤ 6	-11 ≤ <i>h</i> ≤ 10, -31 ≤ <i>k</i> ≤ 31, -6 ≤ <i>l</i> ≤ 5	-5 ≤ <i>h</i> ≤ 5, -14 ≤ <i>k</i> ≤ 14, -19 ≤ <i>l</i> ≤ 19	-5 ≤ <i>h</i> ≤ 5, -14 ≤ <i>k</i> ≤ 14, -18 ≤ <i>l</i> ≤ 19
Total No. of reflections	6252	10397	12773	9467	10437	4265	4195
No. of independent reflections	1115 ($R_{\text{eq}} = 0.0640$)	1387 ($R_{\text{eq}} = 0.0823$)	1717 ($R_{\text{eq}} = 0.0147$)	1439 ($R_{\text{eq}} = 0.0747$)	1470 ($R_{\text{eq}} = 0.0584$)	517 ($R_{\text{eq}} = 0.0276$)	508 ($R_{\text{eq}} = 0.0483$)
No. of reflections with $I > 2\sigma(I)$	918 ($R_{\sigma} = 0.0495$)	1104 ($R_{\sigma} = 0.0548$)	1684 ($R_{\sigma} = 0.0088$)	1175 ($R_{\sigma} = 0.0523$)	1224 ($R_{\sigma} = 0.0407$)	516 ($R_{\sigma} = 0.0208$)	507 ($R_{\sigma} = 0.0282$)
Data/parameters	1115/55	1387/55	1717/55	1439/55	1470/55	517/26	508/26
Goodness-of-fit on F^2	1.054	1.103	1.080	1.052	1.088	1.273	1.344
Final <i>R</i> indices [$I > 2\sigma(I)$]	$R_1 = 0.0311$; $wR_2 = 0.0375$	$R_1 = 0.0364$; $wR_2 = 0.0446$	$R_1 = 0.0110$; $wR_2 = 0.0215$	$R_1 = 0.0349$; $wR_2 = 0.0548$	$R_1 = 0.0291$; $wR_2 = 0.0368$	$R_1 = 0.0145$; $wR_2 = 0.0443$	$R_1 = 0.0206$; $wR_2 = 0.0481$
<i>R</i> indices (all data)	$R_1 = 0.0464$; $wR_2 = 0.0394$	$R_1 = 0.0597$; $wR_2 = 0.0471$	$R_1 = 0.0116$; $wR_2 = 0.0219$	$R_1 = 0.0526$; $wR_2 = 0.0586$	$R_1 = 0.0447$; $wR_2 = 0.0390$	$R_1 = 0.0145$; $wR_2 = 0.0443$	$R_1 = 0.0208$; $wR_2 = 0.0482$
Extinction coefficient	0.00134(4)	0.00059(3)	0.00032(2)	0.00171(9)	0.00086(4)	0.0012(1)	0.0010(2)

Table 2. Atomic coordinates and equivalent isotropic displacement parameters for the $\text{La}_4\text{Co}_4\text{Bi}$ and $\text{La}_3\text{Ni}_3\text{Ga}$ compounds as representative models for the corresponding isostructural series.

Atom	Site	x	y	z	$U_{\text{eq}} (\text{\AA}^2)$
$\text{La}_4\text{Co}_4\text{Bi}$					
La1	4h	0.27768(8)	0.15043(3)	$\frac{1}{2}$	0.0101(2)
La2	4h	0.40848(8)	0.42431(3)	$\frac{1}{2}$	0.0082(2)
La3	4g	0.06761(8)	0.42052(3)	0	0.0106(2)
La4	4g	0.44744(8)	0.27701(3)	0	0.0079(2)
Co1	8i	0.25659(13)	0.02919(5)	0.2492(2)	0.0086(2)
Co2	4g	0.1037(2)	0.10918(7)	0	0.0129(4)
Co3	2d	0	$\frac{1}{2}$	$\frac{1}{2}$	0.0089(5)
Co4	2a	0	0	0	0.0099(5)
Bi	4h	0.16016(5)	0.30108(2)	$\frac{1}{2}$	0.0080(1)
$\text{La}_3\text{Ni}_3\text{Ga}$					
La1	8f	0	0.28250(3)	0.60309(2)	0.01215(6)
La2	4c	0	0.00101(5)	$\frac{1}{4}$	0.01310(7)
Ni1	8f	0	0.43151(8)	0.08559(6)	0.01262(9)
Ni2	4c	0	0.30527(9)	$\frac{1}{4}$	0.0140(1)
Ga	4a	0	0	0	0.0098(1)

3. Results and discussion

3.1 Crystal structures

As to the extent of our knowledge, the intriguing hexagonal honeycomb-like structure motifs extending along the 6(3)-fold axis were first observed in $\text{La}_{15}\text{Ni}_{13}\text{Bi}_5$ [58], we extended our explorations onto chemically closely related systems. Although identical motifs could not be observed in any of them, two isostructural series were identified, *i.e.* La_4Co_4M ($M = \text{Bi}, \text{Pb}, \text{Te}, \text{Sb}$ and Sn) and La_3Ni_3M ($M = \text{Ga}$ and Al). Both of them crystallize with orthorhombic unit cells though different space groups, $Pbam$ and $Cmcm$, respectively (**Table 1**). While La_3Ni_3M are ordered ternary representatives of the Er_3Ge_4 type [65], La_4Co_4M are the first representatives of their own type of structure.

The crystal structure of $\text{La}_4\text{Co}_4\text{Bi}$ (as a representative of the La_4Co_4M series) is best described in terms of relatively separated Co and Bi layers well in line with the antagonistic nature [51] of this pair stacking along the b axis (**Figure 1**). Bi atoms form zigzag chains along the a axis being surrounded by La equatorially tricapped prisms. Such a zigzag sequence of the prisms also leaves tetrahedral voids appearing like a distorted $(3.6)^2$ Kagomé-like ribbon. It is worth noting that such a p -element-centered trigonal prism is a common structural element in an extended family of intermetallics, though they usually condense forming larger triangular conglomerates with an edge size of up to six units [66, 67]. It is interesting that unit cell of La_4Co_4M contracts from Sn to Sb and Te, however, the unit cell shape change is strongly

anisotropic where the a parameter is the least affected, the b expands and the c contracts (**Table 1**).

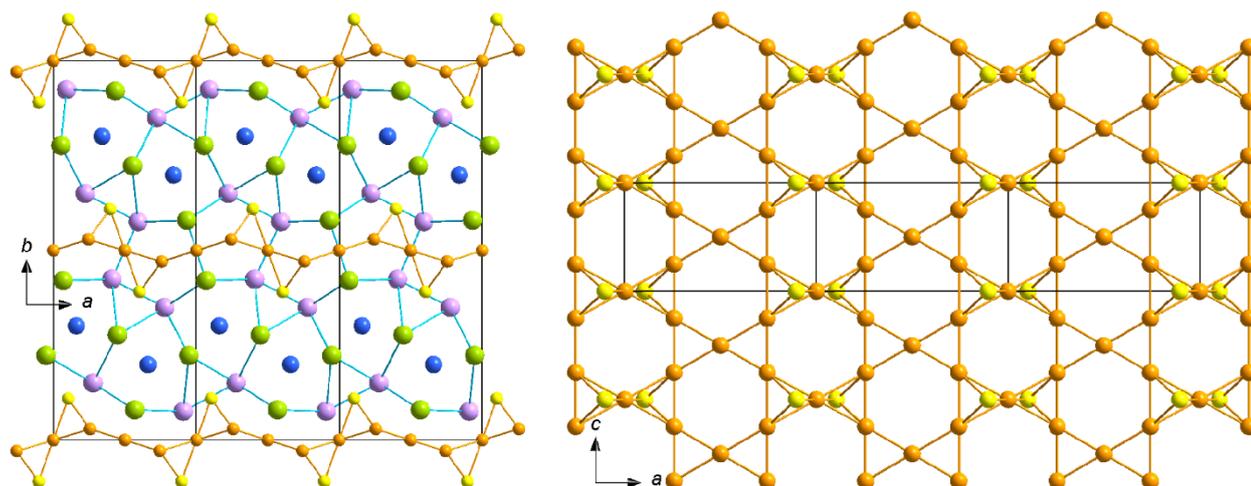


Figure 1. Projection of the crystal structure of $\text{La}_4\text{Co}_4\text{Bi}$ on the ab plane (left) and Co Kagomé lattice (right). La atoms are green and lavender, Co – orange and yellow, and Bi – blue.

The cobalt layer represents a slightly corrugated augmented Kagomé lattice (**Figure 1 (right)**) separating the triangular La/Bi motifs. Due to additional capping Co atoms (yellow), Co hexagons and triangles show a little different degree of distortion with Co-Co distances 2.3544–2.4326(8) Å and 2.3544–2.5211(8) Å, respectively. It is worth noting that none of these distances exceed the sum of two covalent radii of Co (1.27 Å) [68]. The Co-Co-Co angles within the hexagons do not deviate significantly from the ideal 120° angle – 119.3–121.7(1) and 117.8–124.3(1)°, however, the angles between the neighboring hexagons along the a axis are 143.4(1)°, quite noticeably deviating from planarity.

The crystal structure of $\text{La}_3\text{Ni}_3\text{Ga}$ exhibits analogous puckered but polyanionic layers separated by the layers of edge-sharing square pyramids. The latter extend along the a axis but alternate their orientation along the c (**Figure 2**). More interesting are the polyanionic motifs exhibiting a mix of a planar arrangement of $\text{Y}_2\text{Co}_2\text{Ga}$ [69] and zigzag corrugation of $\text{La}_4\text{Co}_4\text{Bi}$. Although the Er_3Ge_4 structure type is not rare with the smaller rare earth metals, *e.g.* Y [70], our compound is just the second representative of this type with La following $\text{La}_3\text{Ni}_3\text{Zn}$ [71]. The tiling of the anionic layers is now 3^28^2 ; however, the 8-membered corrugated Ni_6Ga_2 octagon is in fact a truncated version of two fused hexagons (Figure S3) with the mechanism observed in $\text{La}_{15}\text{Ni}_{13}\text{Bi}_5$ [58]. The truncated vertices are analogously compensated with the La atoms. All angles within the octagons show a slightly more significant deviation from 120° but only due to the presence of the second type atoms in the network (112.8–123.5(1)°) where the angles between the hexagons are even more regular – 118.5(1)°. It is though interesting that Ni-Ni distances are relatively long (>2.6 Å) exceeding the sum of the covalent radii (1.26 Å) indicating significant influence of the heteroatomic anionic component.

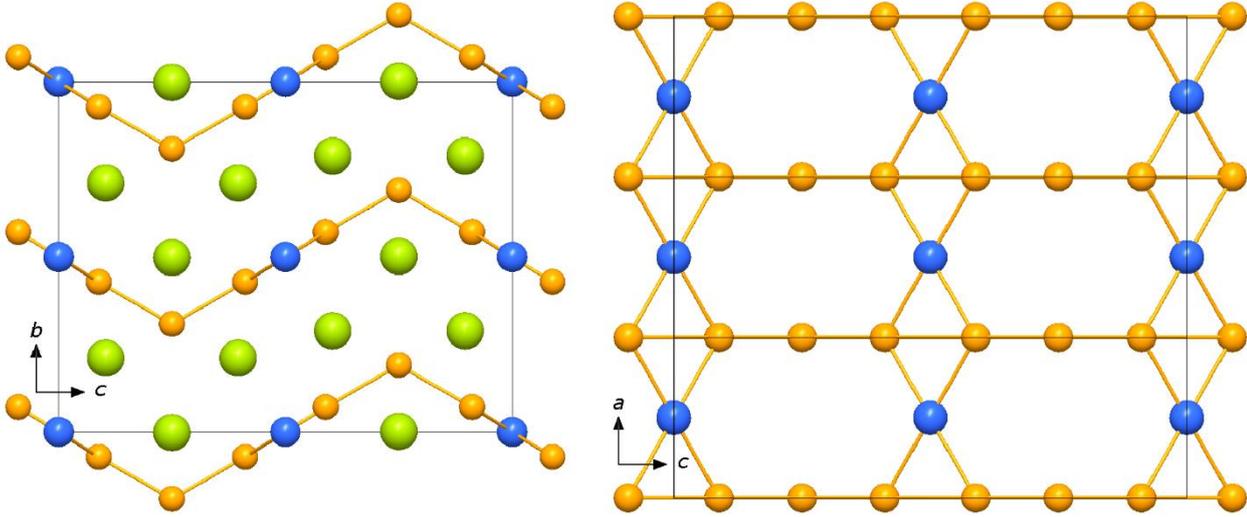


Figure 2. Projection of the crystal structure of La₃Ni₃Ga on the *bc* plane (left) and Co/Ga 3²8² lattice (right). La atoms are green and lavender, Ni – orange and Ga – blue.

Kagomé lattice compounds, particularly but not limited to intermetallics ones, attracted a lot of attention in recent years due to the fascinating physics they can reveal [72-77]. Although the La₄Co₄Bi type is new, an identical corrugated kagomé lattice has been observed in the binary La₂Co₃ [78]. From the common features, we must outline that the homoatomic Co network corrugation in both structures is interconnected with the cationic zigzag motifs. La₂Co₃ exhibits similar hexagonal motifs though in the form of homogeneous slightly distorted honeycombs along the *a* axis. Contrary, the lattice in Y₂Co₂Ga [69] is planar but composed of two elements and shows a bit different tiling scheme – 3²6² (**Figure 3**). Apparently, such a tiling scheme is more resistant and can accommodate corrugated cationic motifs keeping the network planarity. More complex augmented or even sandwiched kagomé lattices have been observed in the series of intermetallics EuT₅In (*T* = Cu, Ni) [79, 80] also revealing complex magnetic structures.

Related corrugated tilings with truncated hexagonal fragments may extend to 3²10² or even 3²12² (**Figure 3**) as observed in the crystal structures of Y₄Co₄Ga and Y₅Co₅Ga or their derivatives, respectively [81, 82]. Their corrugations involve multiple truncated hexagons including those with two missing vertices in Y₅Co₅Ga with even more differentiated shapes. Each missing vertex is again compensated by a rare-earth atom from the interlayer space. In this series, the planar tiling in Y₂Co₂Ga shows the highest degree of distortion and extremely short Co-Co distances suggesting possible underoccupation of some positions. Following the general tendency, all Co-Co-Co angles in the corrugated layers show a minimal degree of distortion not exceeding 3° from the ideal hexagonal angle. Contrary, the heteroatomic component always leads to a more significant distortion and some stretching of the ribbon ($\angle_{\text{Co-Ga-Co}} = 112.5\text{--}115.5^\circ$).

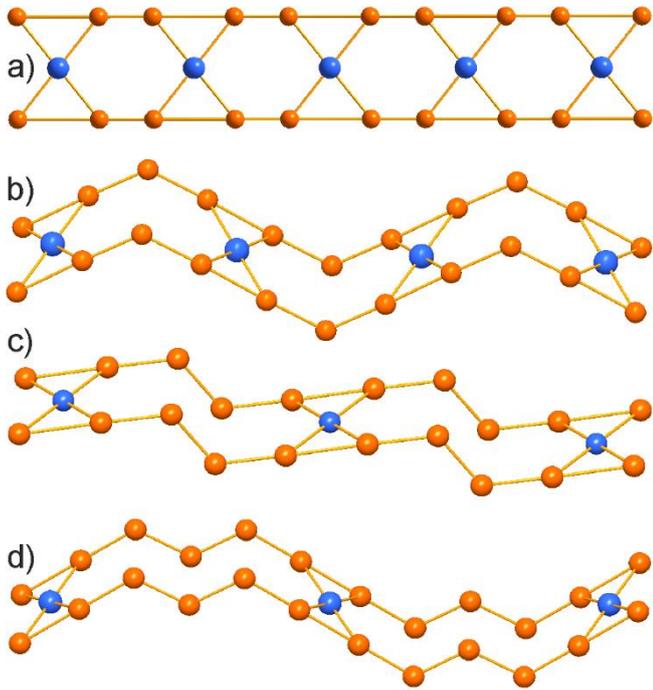


Figure 3. 3^26^2 , 3^28^2 , 3^210^2 and 3^212^2 type ribbons stacking into layers in the crystal structures of $\text{Y}_2\text{Co}_2\text{Ga}$ (a), $\text{La}_3\text{Ni}_3\text{Ga}/\text{Y}_3\text{Co}_3\text{Ga}$ (b), $\text{Y}_4\text{Co}_4\text{Ga}$ (c) and $\text{Y}_5\text{Co}_5\text{Ga}$ (d), respectively. Transition metals are colored orange and Ga – blue.

3.2 Magnetic properties

Magnetic properties of the investigated La_4Co_4M with $M = \text{Sb}, \text{Pb}$ and Bi are presented in **Figure 4**. For low measuring fields the properties of $\text{La}_4\text{Co}_4\text{Sb}$ were biased by small impurity of $\text{La}_2\text{Co}_{1.7}$ [83] (**Table S1**). Therefore, on the **Figure 4a** susceptibility measured at 5 T is presented. It shows a distinct anomaly at 170 K suggesting occurrence of antiferromagnetic ordering below that temperature. At lowest temperatures the susceptibility shows a small increase, which presumably is connected to some reorientation of magnetic moments. Inverse susceptibility follows the Curie-Weiss (CW) law down to 220 K, with the paramagnetic Curie temperature (θ_{CW}) as low as $-79.8(1)$ K, the effective magnetic moment was found to be $7.0(0)$ μ_{B} per f.u. Assuming that only Co atoms are magnetic, it means that mean value per Co atoms is about 1.75 μ_{B} , which is close to typical values for low-spin Co^{2+} configuration. Isothermal magnetization measured at 2 K shows soft magnetic contribution overlaid on the antiferromagnetic (AF)-like, non-saturating behavior. The soft magnetic properties are likely related to aforementioned rise of magnetic susceptibility at low temperatures. In general, soft magnetic properties are not very surprising as strong AF interaction is usually restricted to Kagomé lattices, whereas inter-planar coupling can be way weaker, yet even ferromagnetic.

To some extent, the situation evidenced for $\text{La}_4\text{Co}_4\text{Pb}$ is similar (**Figure 4b**). The departure of ZFC and FC (field cooled) curves observed at 170 K is likely connected to the occurrence of a minority ferromagnetic phase $\text{La}_6\text{Co}_{13}\text{Pb}$ [15], while the real ordering temperature seems to be

30 K. Hysteresis loop collected at 2K shows an interesting wasp-shape (inset to the **Figure 4b**), which occurs for materials with coexistence of components with contrasting coercivities. In our case the exchange bias was not observed, hence interaction with the impurity phase could be excluded. For this reason, this could be attributed to the distribution of magnetic domain sizes, which is likely for heavily frustrated systems.

In contrast to the previous cases, the CW behavior of $\text{La}_4\text{Co}_4\text{Bi}$ (**Figure 4c**) could not be observed due to the antiferromagnetic impurity phase La_2Co_3 ($T_N = 315$ K [83]). However, at 50 K irreversibility in FC and ZFC curves arises. As apparent the magnetic susceptibility at small fields as well as magnetization collected at 2 K in the magnetic field of 7 T are very small. The isothermal magnetization shows a small ferromagnetic-like contribution overlaid onto an antiferromagnetic signal. Owing to the saturation of the small hysteresis loop of $0.015 \mu_B$ only, one may expect that this suggests non-magnetic (van Vleck paramagnetism) or spin glass properties of the $\text{La}_4\text{Co}_4\text{Bi}$.

Magnetic properties of compounds with triangular arrangements of magnetic atoms (hexagonal, pyrochlore or Kagomé lattices) are usually complex as magnetic frustration may favor second or further neighbors to play a crucial role in the RKKY exchange. This is the case for 4:4:1 family, which exhibits decorated Kagomé motifs built from Co atoms stacked within the ac plane (see **Figure 1**). The decoration by out-of-plane Co atoms may introduce additional Dzyaloshinskii-Moriya interaction, which favors complex non-collinear, modulated magnetic structures. Usually, cobalt compounds are ferromagnetic, as long as Co is the only magnetic atom in the lattice. Some exceptions are known mainly in the binary $R\text{-Co}$ ($R =$ rare earth) system [83, 84]. In light of the above, we suggest that the compounds do not exhibit magnetic ordering down to the lowest temperatures studied due to the complexity of the magnetic interactions.

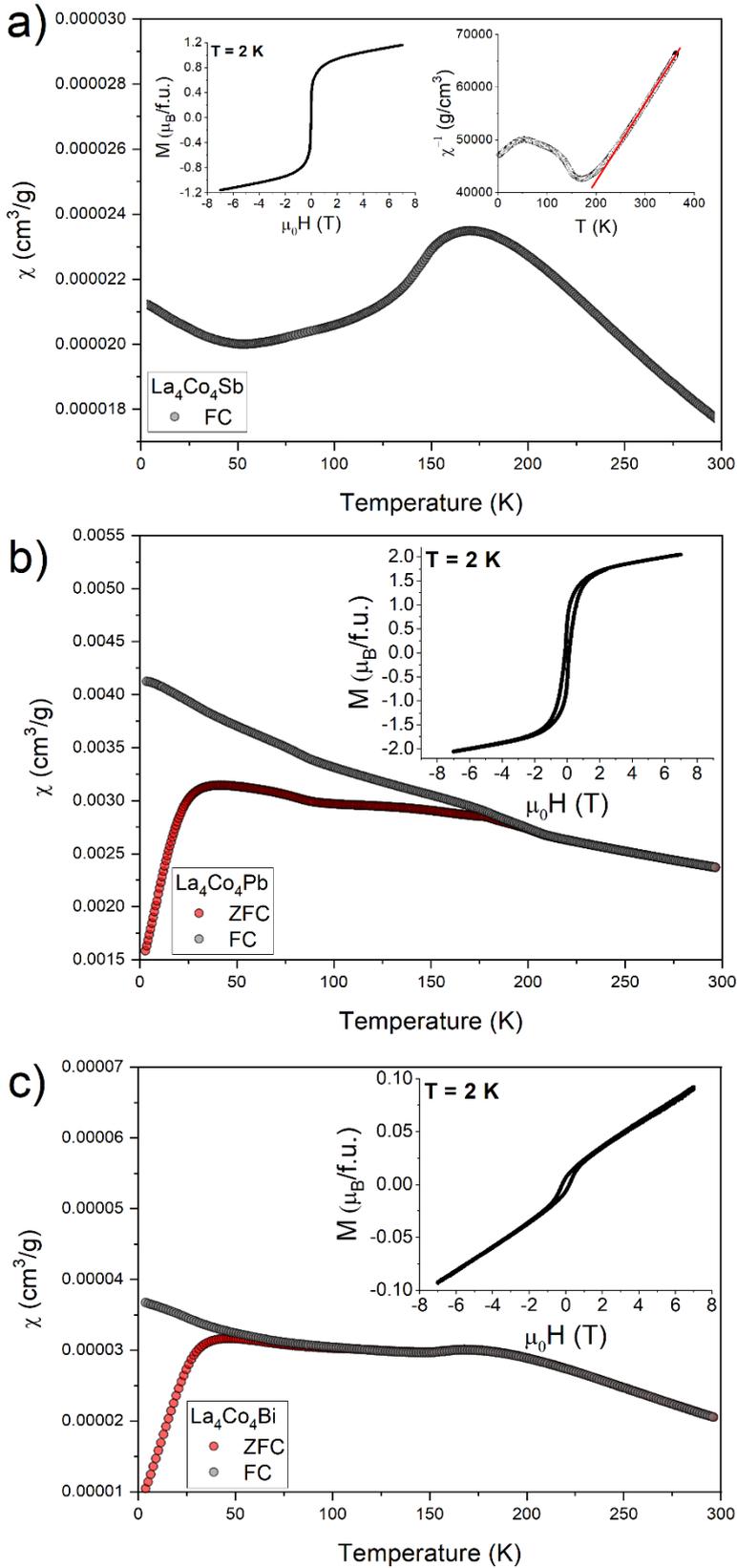


Figure 4. Magnetic properties of La_4Co_4M with $M=\text{Sb}$ (a), Pb (b), and Bi (c). Field-cooled (FC) and Zero field-cooled (ZFC) curves were measured in the field of 0.01 T, except for $\text{La}_4\text{Co}_4\text{Sb}$, where 5T magnetic field was used. Isothermal magnetization loops collected at 2 K are presented in the insets. For the $\text{La}_4\text{Co}_4\text{Sb}$ compound inverse susceptibility was plotted in the inset to part (a), along with the Curie-Weiss fit (red line).

For the $\text{La}_3\text{Ni}_3\text{Al}$ compound, the magnetic susceptibility shows no Curie-Weiss behavior, as well as the modified one, in the investigated temperature range (**Figure 5a**). The susceptibility shows a broad maximum at about 37 and 22 K for ZFC and FC curves, respectively. Additionally, small departure between the FC and ZFC curves can be noticed below roughly 250 K. However, the collected magnetization curves (inset to **Figure 5a**) show an extremely small signal. Measurements performed at higher magnetic fields of 5T revealed that the sample shows temperature independent behavior above 25 K. The discrepancy between low and high field data is a clear sign that the above-mentioned anomalous behavior originates from spurious phases. This is supported by isothermal magnetization curves, which shows small ferromagnetic signal with saturation of $0.005 \mu_B$ only. It can be presumably associated with traces of pure nickel, hence, ferromagnetic precipitations. Taking above into account the sample can be classified as the van Vleck paramagnet.

For the $\text{La}_3\text{Ni}_3\text{Ga}$ alloy, the magnetic susceptibility shows temperature-independent behavior between 200–380 K with hysteresis loops typical of weak paramagnet with extremely small values of magnetic moment (**Figure 5b**). The low-temperature part of the susceptibility shows a small maximum at about 51 K, which is presumably related to the traces of the La_2Ni_7 impurity phase, which was reported to order antiferromagnetically at that temperature [85]. The measurements performed at higher magnetic fields did not reveal substantial changes, apart from flattening the curve above 100 K. Additionally, the inverse susceptibility does not follow the Curie-Weiss behavior, which suggests that the sample is van Vleck paramagnet, as well. Below 25 K the magnetic susceptibility shows some rise, which may originate from some traces of pure Ni precipitations at the grains boundaries, as the isothermal magnetization shows some small ferromagnetic component with the saturation moment less than $0.0007 \mu_B$. For both compounds, the lack of the long-range magnetic ordering along with Curie-Weiss behavior is in agreement with electronic structure calculations discussed below.

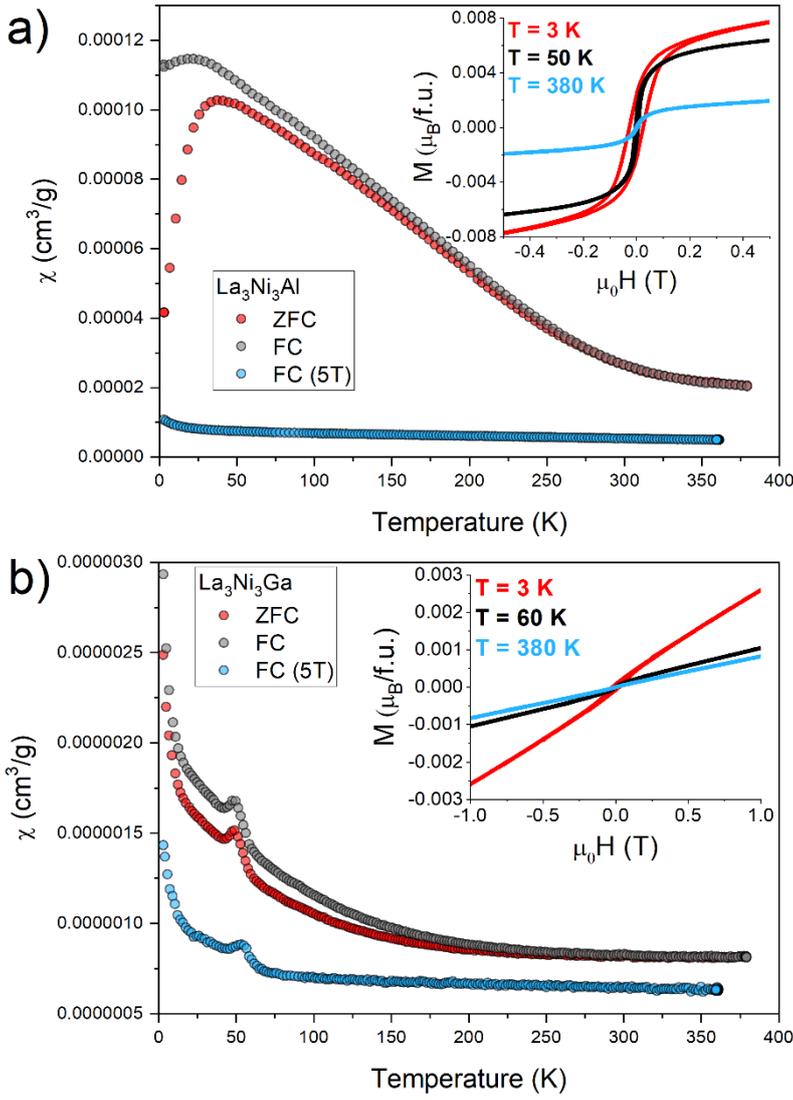


Figure 5. Magnetic properties of $\text{La}_3\text{Ni}_3\text{Al}$ (a) and $\text{La}_3\text{Ni}_3\text{Ga}$ (b). Zero field cooled (ZFC) curves were measured in a field of 0.01 T and 5 T. Inserts show isothermal magnetization curves measured at several temperatures.

3.3 Electronic structures

To gain a better understanding of the composition-structure-property relationships in the investigated compounds, band structure calculations have been performed. The initial sets of structural parameters for the selected La_4Co_4M ($M = \text{Bi}, \text{Sb}$ and Pb) and La_3Ni_3M ($M = \text{Ga}$ and Al) have been relaxed and the optimized data have been provided in **Table 3**. The optimized average volume is comparable to the experimental ones, with a small underestimation (usually not exceeding 1%), though not expected when using the PBE exchange and correlation functional.

Table 3. Relaxed cell parameters after DFT Calculation, Total magnetic moment, Heat of formation and average Bader charge by atom.

Compound	SG	Cell Parameters			$V/\text{\AA}^3$	Magnetic Moment, μ_B/at	$\Delta_f H$, kJ/mol	Bader charges		
		$a/\text{\AA}$	$b/\text{\AA}$	$c/\text{\AA}$				La	Co/Ni	M
$\text{La}_4\text{Co}_4\text{Bi}$	<i>Pbam</i>	8.240	21.600	4.800	854.32	0.17	-32.69	1.065	-0.686	-1.518
$\text{La}_4\text{Co}_4\text{Sb}$	<i>Pbam</i>	8.312	21.838	4.646	843.33	0.20	-41.57	1.098	-0.692	-1.622
$\text{La}_4\text{Co}_4\text{Pb}$	<i>Pbam</i>	8.354	21.924	4.683	857.71	0.24	-24.40	1.079	-0.678	-1.605
$\text{La}_3\text{Ni}_3\text{Ga}$	<i>Cmcm</i>	4.177	10.533	13.699	602.71	0.00	-42.16	1.043	-0.886	-0.469
$\text{La}_3\text{Ni}_3\text{Al}$	<i>Cmcm</i>	4.150	10.642	13.756	607.52	0.00	-39.94	1.026	-1.185	0.478

Using spin relaxation, only the Co-based compounds converged to a distinguishable ferromagnetic ordering with a weak magnetic moment of about $0.2 \mu_B/\text{at}$, mainly held by the Co atoms in $8i$ and $2d$ positions ($0.9 \mu_B/\text{at}$), with a small ferrimagnetic contribution from La. Several magnetic orders, such as antiferromagnetic, have been tested on the Ni-based compounds, all falling to a Pauli paramagnetic, within the pseudo-potential approximation. All these spin polarization results show a good agreement with the experimental measurement, where the Ni-based compounds hardly reveal any magnetic ordering. For the Co-based compound, as in the measurement, the calculation shows that the Co atoms hold the major magnetic contribution. The total moment is slightly different, which can be explained by the pseudopotential approximation used.

The electronic DOS is shown in **Figure 6**. For all compounds, a small structure around 6 to 8 eV below the Fermi level corresponds to the hybridized sp bands of the M element, while the valence structure is mainly composed of the $3d$ elements. The filling of these dense bands could be clarified with the Bader analysis, and the corresponding electronic charge transfer is given in **Table 3**. While La allows to give about 1 electron, the $3d$ element fills its bands by grabbing 0.7 electrons for Co and 0.9–1.2 for Ni. Since Ni reveals practically filled $3d$ bands in both spin directions, it serves as a good explanation for the absence of magnetism for both Ni-based compounds. It can be noted that the $s-p$ elements such as Sb, Bi and Pb capture about 1.5 electrons thanks to their highest electronegativity, whereas Al, the worst electronegative, even gives electrons to the system, especially to fill the Ni- $3d$ bands as explained earlier.

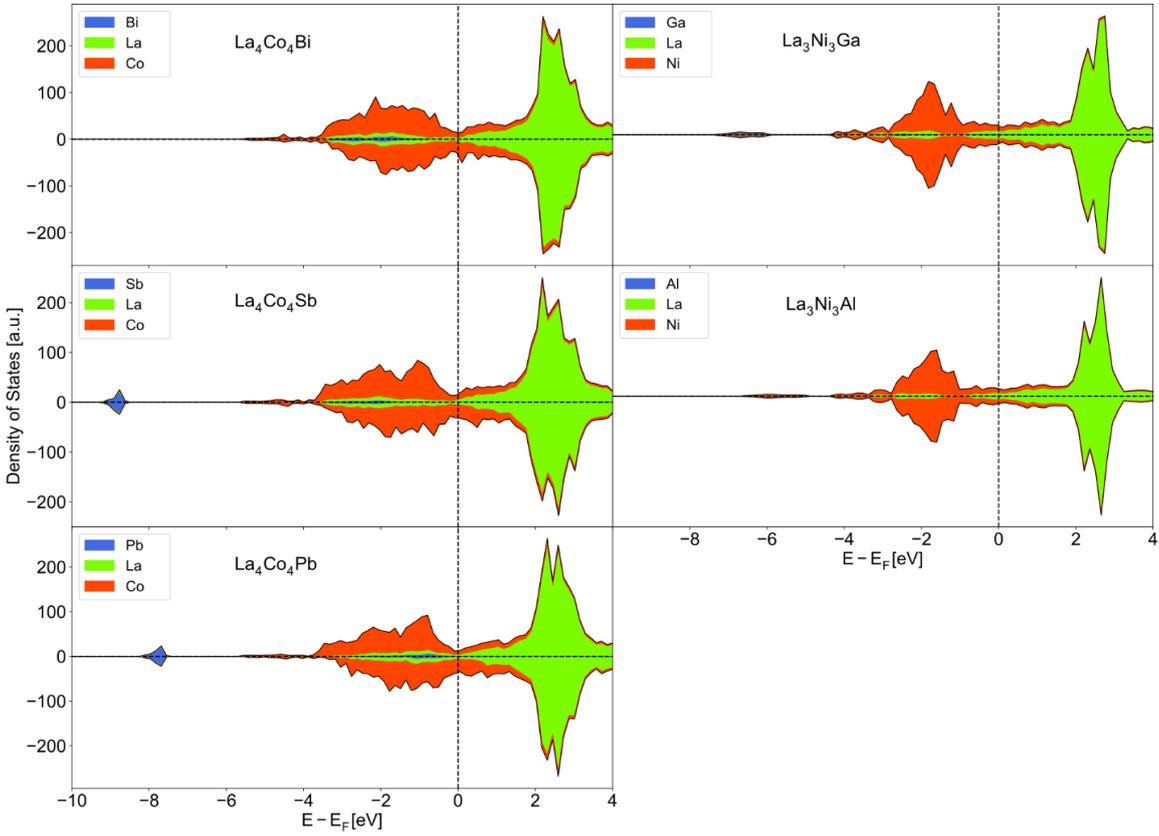


Figure 6. Electronic density of states with partial contribution of each atom for both spin directions of La_4Co_4M ($M=\text{Bi}, \text{Sb}, \text{Pb}$) and La_3Ni_3M ($M=\text{Ga}, \text{Al}$) compounds.

The heat of formation, $\Delta_f H$, for all compounds has been calculated and shows a value around -40 kJ/mol, as usual for the intermetallics of La with transition elements [86]. The mechanical stability has been studied in the harmonic approximation. As an example, the phonon dispersion curves of $\text{La}_4\text{Co}_4\text{Bi}$ and $\text{La}_3\text{Ni}_3\text{Ga}$ present no imaginary frequencies (**Figure 7**). The M element has acoustic branches in common with Co/Ni, while the heavier atom, La, vibrates at lower frequencies. The $\text{La}_4\text{Co}_4\text{Bi}$ presents a vibrational contribution center of the Bi, as the s - p element, located at lower frequencies compared with the Ga, due to the respective weight.

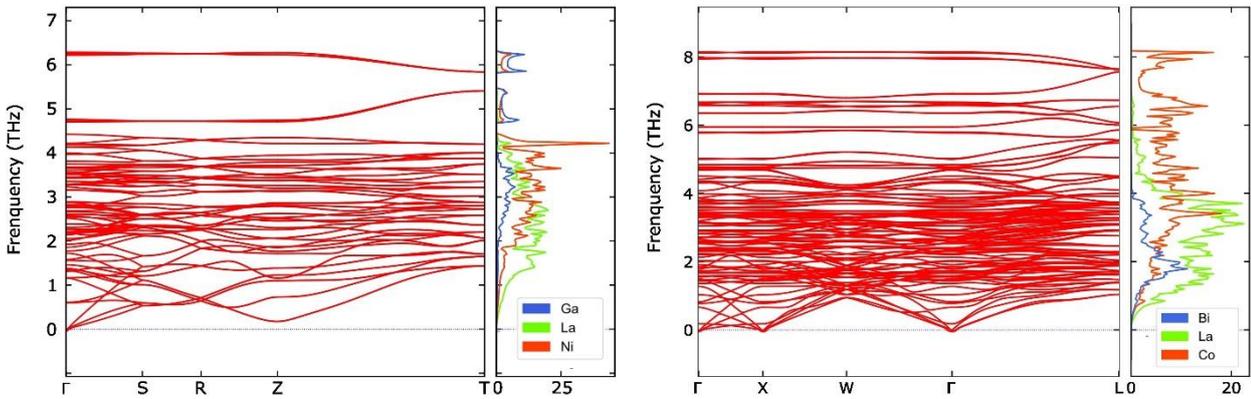


Figure 7. Phonon dispersion curves and associated DOS of $\text{La}_3\text{Ni}_3\text{Ga}$ (left) and $\text{La}_4\text{Co}_4\text{Bi}$ (right).

3.4 Chemical Bonding

Bonding analysis by the COHP method for $\text{La}_3\text{Ni}_3\text{Ga}$ and $\text{La}_4\text{Co}_4\text{Sn}$ as representative examples of the corresponding series (**Figures 8, 9** and S4-6, **Table 4**) revealed plenty of common features, although compounds exhibit different composition-related structural and bonding motifs. The principal difference comes from the participation of the second anionic component. While in the crystal structure of 3:3:1 the p element is involved in the formation of the kagomé-like layers, in 4:4:1 it stays separately leading to the absence of heteroanionic contacts.

Ni-Ni contacts in the octagons are somewhat shorter compared to those in the Ni_2Ga triangles and correspondingly show a higher bond population. The latter are naturally affected by strong heteroatomic Ni-Ga interactions. Weak antibonding contributions for the Ni-Ni pairs, especially longer ones, could be observed at 0.1-2 eV below E_F due to possible repulsion between the filled d states [87-89]. With a little exception for those longer Ni-Ni contacts, all interactions are significantly bonding at the Fermi level. It is also worth noting La atoms participate actively in the covalent bonding interactions providing almost 60% of the total bonding population and even La-La interactions are not negligible (**Table 4**).

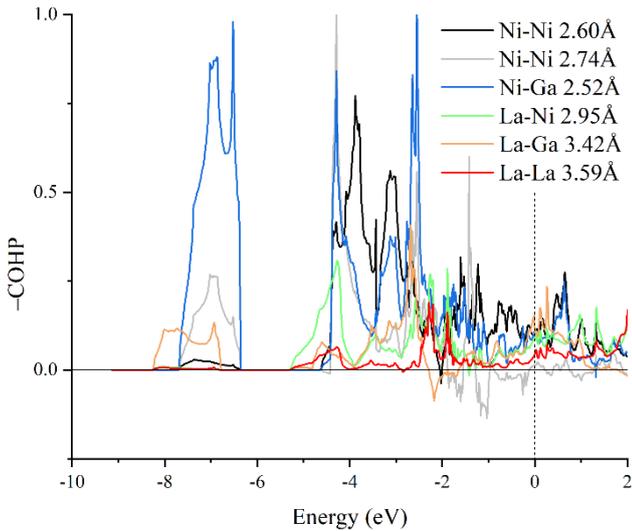


Figure 8. $-\text{COHP}$ curves of the six types of bonds in the crystal structure of $\text{La}_3\text{Ni}_3\text{Ga}$. Only selected La-Ni, La-Ga and La-La bonds have been plotted for clarity purposes. The Fermi level is indicated by a dotted line.

Since Co-Sn interactions are not present in the crystal structure of $\text{La}_4\text{Co}_4\text{Sn}$, all interanionic interactions are represented by Co-Co bonds, although their total contribution is comparable to those of Ni-Ni and Ni-Ga together in $\text{La}_3\text{Ni}_3\text{Ga}$. Most of the interactions are non-bonding at the Fermi level. Some specific interactions reveal antibonding character particularly the Co-Co pairs within Co_4 tetrahedra, particularly those involving Co atoms capping triangles of the kagomé tiling. Interestingly, another Co-Co bond in the same tetrahedron outside the kagomé layer is the longest, the least populated and optimized at 0.5 eV below E_F , while the remaining Co-Co bonds

of the hexagons are always bonding. Similarly to $\text{La}_3\text{Ni}_3\text{Ga}$, La-Co and La-Sn contributions are significant (almost 50%) and significantly bonding below E_F , though their proportion is more steady (**Table 4**) due to the isolation of the Sn positions. La-La contributions in both structures are comparable as well.

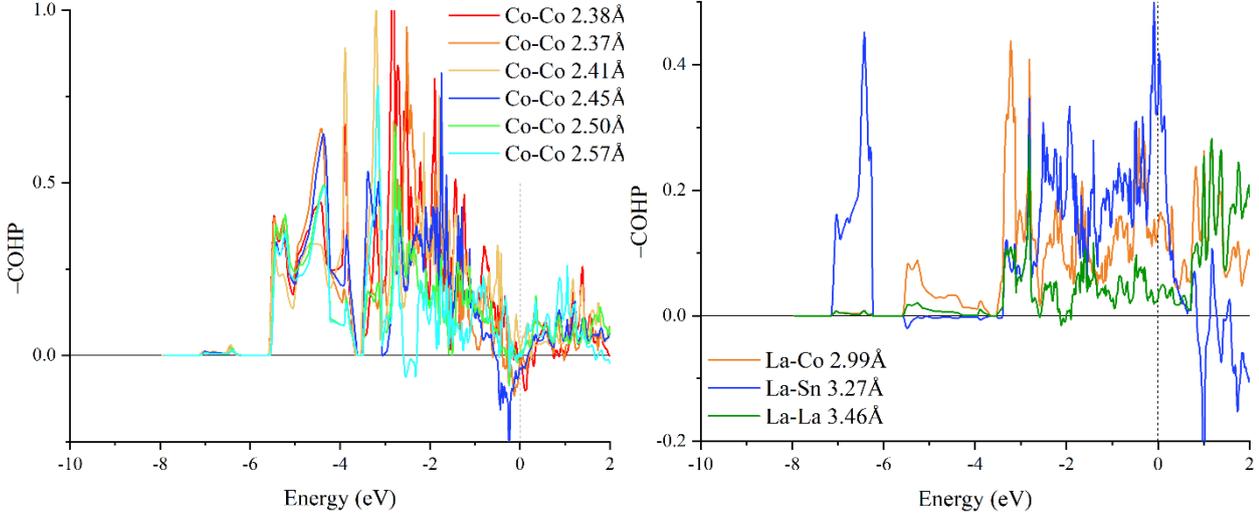


Figure 9. $-\text{COHP}$ curves of (left) all Co-Co bonds and (right) three representative La-Co, La-Sn and La-La bonds in the crystal structure of $\text{La}_4\text{Co}_4\text{Sn}$. The Fermi levels are indicated by dotted lines.

Table 4. Bond length ranges, average $-\text{ICOHP}$ values and total contributions to bonding interactions in $\text{La}_3\text{Ni}_3\text{Ga}$ and $\text{La}_4\text{Co}_4\text{Sn}$.

Bond type	Length (\AA)	$-\text{ICOHP}$	No./cell	$-\text{ICOHP}$	Contribution
$\text{La}_3\text{Ni}_3\text{Ga}$					
Ga-Ni	2.52	1.65	16	26.4	26.0
Ni-Ni	2.60	1.10	12	13.2	13.0
	2.74	0.69	4	2.8	2.7
La-Ni	2.95	0.41	8	3.3	3.2
	3.01–3.06	0.48	40	19.2	18.9
	3.10–3.17	0.45	32	14.3	14.1
	3.19	0.56	4	2.2	2.2
La-Ga	3.28	0.45	12	5.4	5.3
	3.41–3.42	0.37	24	8.8	8.7
La-La	3.59	0.14	16	2.2	2.2
	3.70	0.20	16	3.2	3.1
	4.00	0.14	4	0.6	0.6
$\text{La}_4\text{Co}_4\text{Sn}$					
Co-Co	2.37–2.41	1.44	16	23.0	20.4
	2.45	1.29	12	15.4	13.6
	2.50–2.57	0.99	12	11.9	10.5
La-Co	2.90	0.76	12	9.1	8.1
	2.93–3.05	0.43	44	18.9	16.7
	3.29	0.28	8	2.2	2.0
La-Sn	3.27	0.79	4	3.2	2.8
	3.36–3.42	0.71	24	17.0	15.1
	3.53	0.62	8	5.0	4.4
La-La	3.46	0.18	4	0.7	0.6

3.70–3.78	0.14	20	2.8	2.5
3.80–4.02	0.16	24	3.8	3.4

4. Conclusions

Two isostructural series, La_4Co_4M ($M = \text{Sn, Sb, Te, Pb, Bi}$) and La_3Ni_3M ($M = \text{Al, Ga}$) have been synthesized. Both series are characterized by interesting variations of the semiregular tilings of the corrugated anionic layers combined with predominantly cationic zigzag motifs. The layers in La_4Co_4M are of augmented Kagomé type with a minimal degree of corrugation, while those in La_3Ni_3M have a more complex tiling (3^28^2) and a more significant degree of corrugation. In some sense larger (8^2) fragments exist of two truncated hexagons and the corrugation is required to keep the optimized Co-Co-Co bonding angles ($\sim 120^\circ$) within the polyanionic layer. The discovery of the La_4Co_4M series brought us to the fact that such types of compounds can be obtained with ‘antagonistic pairs’ *i.e.* Co-Pb or Co-Bi. As for the La_3Ni_3M compounds, it is only the second observation of such a structure containing La and the first one within $R\text{-Ni-M}$ ($R = \text{rare earth element, } M = \text{Ga and Al}$) systems.

Magnetic characterization has been performed for both series. For La_4Co_4M compounds, we were unable to prove long-range magnetic ordering due to strong magnetism from impurity phases. Only for the $\text{La}_4\text{Co}_4\text{Sb}$, complex antiferromagnetic ordering below 170 K was suggested. However, it seems that the other compounds show rather weak, frustrated magnetism, if any. No magnetic ordering in the La_3Ni_3M series was evidenced. The magnetic susceptibility was found to be temperature independent with no Curie-Weiss behavior, which is likely a sign of van Vleck paramagnetism. From the electronic structure study, the partial charge transfer from La to the other elements explains the filling of the $3d\text{-Ni}$ bands and the corresponding absence of magnetism for Ni-based compounds.

The electronic densities of states of La_4Co_4M show extensive overlaps, large mostly Co- d states dominating at 0–4 eV below the Fermi level and local minima at E_F . Predominantly $M\text{-}p$ states could be observed at -8 eV or below. As for La_3Ni_3M large Ni- d states are less broad dominating at 1.5–2.5 eV, while the $M\text{-}p$ states are located at 6–7 eV below E_F . The bonding picture in both series is qualitatively and quantitatively similar showing enhanced contribution from the La pairs. Even though heteroanionic pairs could not be observed in La_4Co_4M total contribution from both anionic pairs in La_3Ni_3M is equal to the Co-Co contribution. Similarly, contributions from the La-La pairs are not negligible in both compounds giving $\sim 6\%$ in total.

Acknowledgments

We acknowledge Myfab Uppsala for providing facilities. Myfab is funded by the Swedish Research Council as a national research infrastructure. VSh and MS acknowledges financial support from the Foundation for Strategic Research (SSF), grant #EM16-0039, as well as from

ÅForsk, grant # 21-196, is gratefully acknowledged. DFT calculations were performed using HPC resources from GENCI-CINES (Grant A0060906175). AK acknowledges support from the Swedish Strategic Research Council (SSF UKR22-0050). AVM acknowledges DFF (Danmarks Frie Forskningsfind) for support.

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