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High resolution spectroscopic investigation of the CH₂CHO radical in the sub-millimeter region

Rosemonde Chahbazian, Marie-Aline Martin-Drumel, and Olivier Pirali*

*Université Paris-Saclay, CNRS, Institut des Sciences Moléculaire d'Orsay, 91400 Orsay,
France*

E-mail: rosemonde.chahbazian@universite-paris-saclay.fr

Abstract

In this work, the pure rotational spectrum of the vinoxy radical (CH₂CHO) has been studied at millimeter- and sub-millimeter wavelengths (110–860 GHz). CH₂CHO was produced by H abstraction from acetaldehyde (CH₃CHO) using atomic fluorine in a double-pass absorption cell at room temperature. A Zeeman-modulation spectrometer, in which an external magnetic field generated inside the absorption cell is amplitude modulated, has been used to record the pure rotational transitions of the radical. The recorded spectra are deconvoluted of signals from closed-shell species, allowing relatively fast acquisitions over wide spectral windows. Transitions involving values of the rotational quantum numbers N'' and K_a'' up to 41 and 18, respectively, were measured and combined with all available high-resolution literature data (both pure rotation and ground state combination differences from ro-vibration) to greatly improve the modeling of the CH₂CHO spectrum. The combined experimental linelist is fit using a semi-rigid rotor Hamiltonian and the results are compared to quantum chemical calculations. This laboratory study provides the spectroscopic information needed to search for CH₂CHO in various interstellar environments, from cold (e.g., typically 10 K for dense molecular clouds) to warm (e.g., \sim 200 K for hot corinos) objects.

Introduction

Radicals are key species for chemical reactions occurring in combustions, atmospheric and astrophysical environments^{1,2}. As a primary product of the reaction between oxygen and olefins—as ethylene—of relevance for combustion processes, the vinoxy (CH_2CHO) radical has attracted much interest (see, e.g., Refs.³⁻⁵). In astrophysical environments, the oxidation reaction of C_2H_4 has been proposed to lead to the production of CH_2CHO with a 30% yield in high-temperature sources such as hot cores ($\sim 200 - 300$ K)⁶. The unambiguous interstellar detection and column density determination of CH_2CHO , whose structure is represented in Fig. 1, would validate this model. However, the radical remains elusive in the ISM, preventing a deep understanding of its formation route and reactivity in outer space.

Spectroscopic investigations on CH_2CHO were initiated by the observation of the $\tilde{\text{B}}-\tilde{\text{X}}$ band of the radical by Ramsay⁷, subsequently investigated by Inoue and Akimoto⁸. The later study was rapidly followed by infrared vibrational spectroscopy in Ar isolated matrix⁹, which included an isotopic substitution of the precursor to support the vibrational band assignments in the ground electronic state. The first rotationally-resolved spectra were obtained for the $\tilde{\text{B}}-\tilde{\text{X}}$ band by Heaven et al.¹⁰ and DiMauro et al.¹¹ using laser-induced fluorescence spectroscopy, providing insights into the rotational structure and spin splittings of the radical. Other studies focused on the $\tilde{\text{B}}-\tilde{\text{X}}$ ¹²⁻¹⁵ and $\tilde{\text{A}}-\tilde{\text{X}}$ ^{3,16-18} electronic bands. Using millimeter-wave spectroscopy, the pure rotational spectrum of CH_2CHO was first observed by Endo et al.⁴. At these wavelengths, the hyperfine structure was resolved for the two protons (but not for the third proton, noted H_3 hereafter, of the $-\text{CHO}$ moiety, see Fig. 1). Because only *a*-type transitions were observed in that work, the *K* dependant centrifugal distortion constant Δ_K could not be determined. In what remains today the only high-resolution vibrational investigation of the radical, Utkin et al.¹⁹ recorded the rotationally-resolved spectrum of the ν_3 band around 2800 cm^{-1} . From the assigned *b*-type transitions, the authors were able to manually adjust (but not fit) the value of the Δ_K constant in the ground state. Hansen et al.²⁰ pursued the investigation of the rotational spectrum of CH_2CHO in the microwave domain but their assignments are in disagreement with those subsequently reported by

Endo and Nakajima²¹. In the latter study, four *b*-type rotational lines were also measured which led to a significant refinement of the *A* rotational constant while keeping the Δ_K fixed to the value determined by Utkin et al.¹⁹.

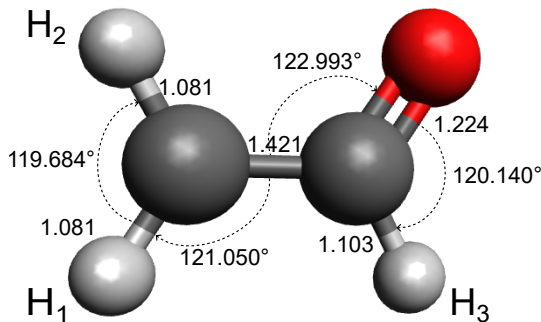


Figure 1: **Representation of the structure of CH_2CHO .** The numbering of the three protons— H_1 , H_2 and H_3 —is indicated. The values of the distances (in Å) and angles (in degrees) are those derived from our calculations.

As a model molecule for resonant structures, the vinyloxy radical has been the subject of discussions in the literature. Indeed, it possesses two isoelectronic structures: the formyl methyl structure (involving a C=O bond, $\bullet\text{CH}_2\text{CH}=\text{O}$) and the ethenyloxy—or vinyloxy—structure (involving a C=C bond, $\text{CH}_2=\text{CHO}\bullet$). These two forms are indistinguishable because of electronic resonance CH_2CHO (see, e.g., Ref.⁸), but many literature studies have aimed at determining the dominant nature. In the ground electronic state, it is now well-established that the formyl methyl resonance form is dominant (although the common name of the radical remains vinyloxy): it was predicted by calculations (see, e.g., Refs.^{22,23}) and confirmed by experiments by Jacox⁹ using isotopic substitutions and by Endo et al.⁴ using the hyperfine coupling constants of the protons. The structural parameters of the skeleton were first determined by DiMauro et al.¹¹ using electronic spectroscopy and then by Endo and Hirota²⁴ using isotopic substitution millimeter-wave spectroscopy.

In the literature, high-resolution pure rotational data are limited to relatively low values of the rotational quantum numbers ($N_{\text{max}} = 7$ and $K_{a\text{max}} = 5$ in the work from Endo et al.²⁵ between 81 and 176 GHz). The observed transitions allow the search for the yet-to-be-detected vinyloxy radical in cold sources of the interstellar medium using highly sensitive millimeter-wave radiotelescopes

(such as NOEMA²⁶ and ALMA²⁷). However, high-resolution measurements covering a broader spectral range and involving higher values of N and K_a are mandatory to search for this intermediate species in warmer environments, such as hot cores, knowing that at higher temperatures the most intense lines are around 176 GHz. In the present work, we recorded the high-resolution spectrum of CH₂CHO in a spectral range spanning from 140 to 850 GHz. Most of the observed transitions are *a*-type transitions involving values of N'' and K_a'' up to 41 and 18, respectively, while *b*-type transitions involve values of N'' and K_a'' up to 22 and 3, respectively. These new observations allow a significant refinement of the model describing the rotational structure of the lowest vibronic state of the radical and enable interstellar searches in the sub-millimeter domain.

The paper is built as follows: the performances of our experimental set-up, which allows for efficiently broad spectral recordings of radicals lines, is described in the following section; the results of the spectral analysis as well as the comparison of our model with the literature are given in the second section; astronomical implications are discussed in the last section.

Laboratory methods

Quantum chemical calculations

Geometry optimization and anharmonic frequency analysis have been carried out at the ω B97X-D/cc-pVQZ level of theory^{28–30} using the Gaussian 16 suite of electronic structure programs³¹. The resulting equilibrium structural parameters of CH₂CHO are reported in Fig. 1.

Synthesis of CH₂CHO

Inoue and Akimoto⁸ and Endo et al.²⁵ have shown that H-abstraction from acetaldehyde (CH₃CHO) by F atoms (themselves synthesized by a microwave discharge in CF₄) leads to a relatively abundant production of CH₂CHO radicals. These authors observed that this synthesis technique yields a radical concentration five times larger than that obtained by the reaction of ethylene (C₂H₄) with oxygen atoms. We applied a similar technique in our work, replacing CF₄

with a mixture of F_2 diluted in He (5 % dilution) similar to what we previously used to study the CH_2OH ^{32,33} and CH_2CN ³⁴ radicals in the sub-millimeter region. The technique has proven to be very efficient in selectively producing the targeted paramagnetic species. The details of the reaction cell design are given in Ref.³⁴. Shortly, this cell is a two-meter-long Pyrex tube with its internal surface covered by a fluorinated wax and equipped with three F injection inlets. The optical configuration allows for a double pass of the radiation in the cell, although in this study the spectra in the 140–220 GHz range were recorded in a single pass configuration (higher frequency transitions were acquired in a double pass configuration). The signal of CH_2CHO was optimized on the transition centered at 153785.5 MHz, originally measured by Endo et al.²⁵. The optimum signal was obtained with about 40 μ bar of CH_3CHO and 50 μ bar of the F_2/He mixture, measured in a continuous gas flow maintained by a roots blower.

Zeeman Modulated (Sub-)millimeter Spectrometer

In the present work, the transitions of CH_2CHO were recorded using a (sub-)millimeter wave spectrometer previously described in Chitarra et al.³⁴. It consists of a frequency multiplication chain (from Virginia Diode Inc.) fed by the RF frequency of a synthesizer (Rhode & Schwarz) for which a 10 MHz reference signal is provided by a Rubidium clock (Stanford Research). We used Schottky diode detectors (VDI) to measure the transitions in the 140–220 GHz and 220–330 GHz spectral ranges and a mechanically-cooled Indium antimonide (InSb) bolometer (QMC) between 330 GHz and 850 GHz. The main difference between the present study and our previous works^{32–34} is that instead of a source frequency modulation (FM) or a double modulation (FM together with Zeeman modulation³⁵), a single Zeeman modulation (ZM) scheme was used here. This was achieved using an alternating magnetic field generated by an 18 A alternating current (AC) circulating in a coil surrounding the cell (about 400 spirals/m, with a total resistance of about 5.5 Ω). The sinusoidal signal was generated by a waveform generator operating at 16.9 kHz followed by an audio amplifier providing the required current to obtain a 14.5 Gauss magnetic field. As the entire sinusoidal waveform feeds the coil (no rectification nor half-rectification of the AC signal), the

rotational levels of open-shell species are split by the Zeeman effect twice per modulation period. The detection is performed using a phase-sensitive lock-in amplifier for which the reference signal is the magnetic field modulation frequency (demodulation at the first harmonic). The frequency steps ranged from 100 to 500 kHz according to the Doppler line broadening and the time constant was adjusted to values ranging from 20 ms to 2 s depending on the line intensity. The line frequency accuracy varies from 50 to 200 kHz in these conditions.

Traditionally when investigating radicals in the (sub-)millimeter range, acquisitions are mostly made line-by-line. Here, the ZM detection scheme allows probing wide spectral windows (typically 1–10 GHz) relatively fast, with good SNR and almost no interference from precursor lines. For example, a spectral survey of the 335 GHz to 385 GHz range, showing the achievable broad spectral coverage, is given on the upper panel of Fig. 2. The 50 GHz were recorded with 30 ms per point and 200 kHz frequency steps resulting in a relatively fast acquisition (of about 10 hours for the full range plotted in the upper panel in Fig. 2). Additionally, specific portions of the spectrum, where transitions of CH₂CHO were expected, were recorded with deeper integration conditions. For example, the portion of the spectrum displayed in the lower panel of Fig. 2 was recorded with an acquisition time of 300 ms/point and 100 kHz frequency steps (150 minutes acquisition time). Hundreds of lines of the CH₂CHO radical are detected with a signal-to-noise ratio (SNR) greater than 10 for the most intense lines. This survey also identified numerous lines of HCO radical with good SNR. These lines are shown on the lower panel of Fig. 2, where we compare the experimental spectrum with the predictions of HCO (using the catalog from the JPL database³⁶) and our prediction of CH₂CHO lines at 300 K.

Results and discussion

Spectroscopic Description

The vinoxy radical is a near-prolate asymmetric top molecule¹⁴ with a ²A'' electronic ground state²². The two permanent dipole moments lie along the *a*-axis and *b*-axis of inertia and take

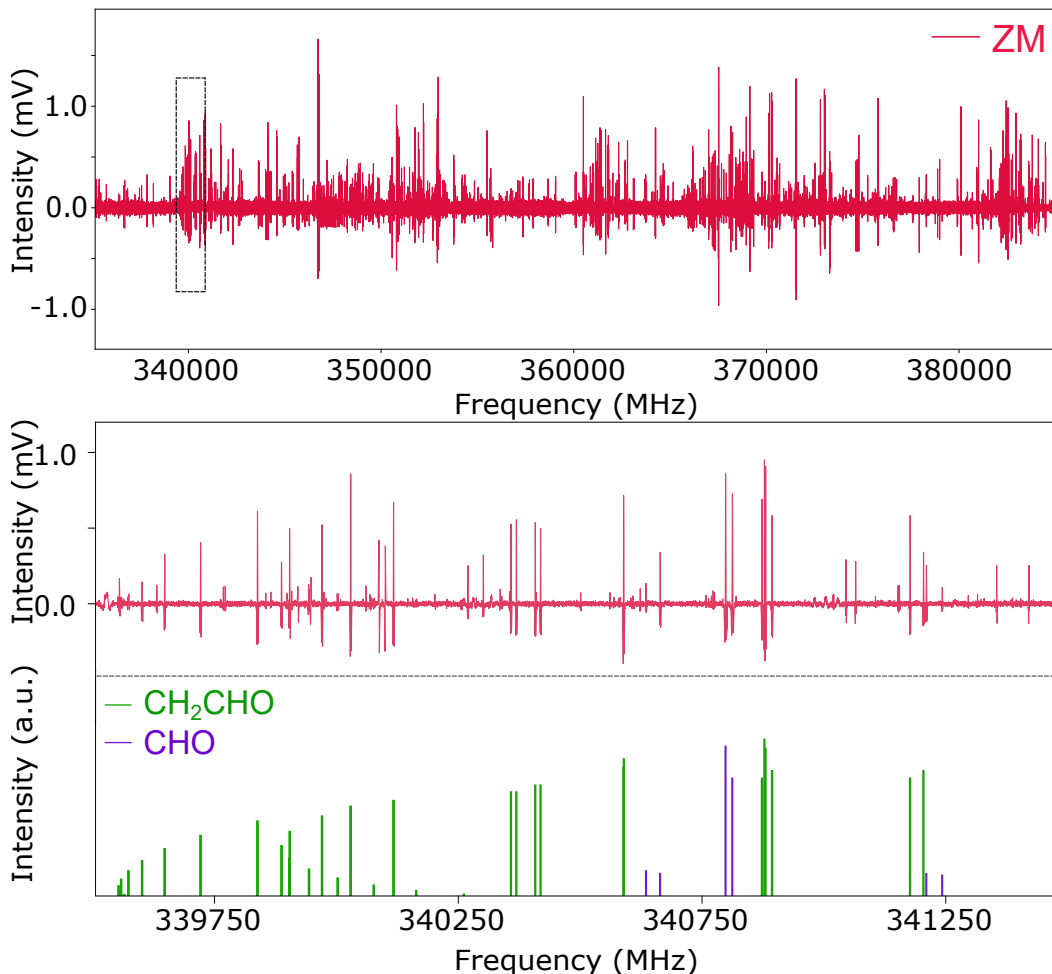


Figure 2: **Portion of the experimental spectrum recorded with ZM.** The top trace shows the survey from 335 to 385 GHz, where numerous lines are detected. Thanks to the use of ZM, no saturation of the precursor lines is observed. Only lines from radical species appear in the spectrum. The lower trace shows the region of the $N'' = 16$ a -type cluster of transitions of the CH_2CHO radical (in green) simulated with the final set of spectroscopic parameters obtained in this work (see later in text). Strong lines of the CHO radical are shown in purple (frequencies extracted from the JPL database³⁶). Residual lines may belong to vibrational satellites of these two radicals or to the CH_3CO radical also synthesized in this experiment (analysis in progress).

values of $\mu_a = 2.76$ Debye and $\mu_b = 0.92$ Debye²¹. CH₂CHO has three nonequivalent protons and one unpaired valence electron. The Hamiltonian employed to model the energy levels of the vibronic ground state is built as follows (1):

$$\hat{H} = \hat{H}_{\text{rot}} + \hat{H}_{\text{sr}} + \hat{H}_{\text{hfs},1} + \hat{H}_{\text{hfs},2} + \hat{H}_{\text{hfs},3} \quad (1)$$

where \hat{H}_{rot} refers to the rotational Hamiltonian, \hat{H}_{sr} refers to the spin-rotation Hamiltonian involving the unpaired electron, and $\hat{H}_{\text{hfs},1}$, $\hat{H}_{\text{hfs},2}$ and $\hat{H}_{\text{hfs},3}$ are the hyperfine structure Hamiltonian accounting for the non-zero nuclear spin of the three protons H₁, H₂ and H₃, with $I_{\text{H}_1} = I_{\text{H}_2} = I_{\text{H}_3} = 1/2$ (see Fig. 1). The standard N , K_a , and K_c quantum numbers label the rotational levels. Due to the electron spin-rotation coupling, each level is split into two J sub-levels, with $\mathbf{J} = \mathbf{N} + \mathbf{S}$ ($S = 1/2$). Then, each sub-level is further split into hyperfine sub-levels because of the spin-spin coupling of the three hydrogens. This leads to sixteen hyperfine sub-levels as the three protons are non-equivalent. Each sub-level is labeled using the F_1 , F_2 and F quantum numbers, defined as : $\mathbf{F}_1 = \mathbf{J} + \mathbf{I}_{\text{H}_1}$, $\mathbf{F}_2 = \mathbf{F}_1 + \mathbf{I}_{\text{H}_2}$ and $\mathbf{F} = \mathbf{F}_2 + \mathbf{I}_{\text{H}_3}$. In the present work, we partially resolved the hyperfine structure involving H₁ and H₂ because Doppler broadening limits the resolution. All lines were assigned using the PGOPHER program³⁷ and the fit was performed using the Pickett program³⁸.

Performances of the Zeeman modulation scheme

The ZM method is well known to be efficient to study paramagnetic species, as described by Guelachvili³⁹. If we compare a double modulation scheme (using FM followed by ZM, as implemented in Chitarra et al.³⁴) with pure ZM at kHz modulation frequencies, ZM is advantageous because it allows filtering of the precursor lines (as well as of the lines of all the other closed-shell molecules produced in the experiment) which often saturates the lock-in amplifier and obliterates wide parts of the spectrum, preventing the observation of numerous paramagnetic lines. This is illustrated in Fig. 3 where three lines of CH₃CHO are observed in FM and saturate the lock-in (black trace, upper panel). In such conditions, in the regions where the FM signal is saturated, a

FM + ZM detection scheme would result in the absence of any magnetic field-dependant signal at the second demodulation stage, preventing the observation of the underlying CH_2CHO transitions. In contrast, a relatively fast single ZM detection scheme allows to easily retrieve the signal from the radical species (red trace, bottom panel). The signal-to-noise ratio (SNR) is the same for both methods at low frequencies. However, at high frequencies (~ 600 GHz), the amplitude of the magnetic field generated by our set-up is not sufficient to split the rotational transitions out of the Doppler line width by the Zeeman effect. This translates to a drastic loss of the spectrometer sensibility.

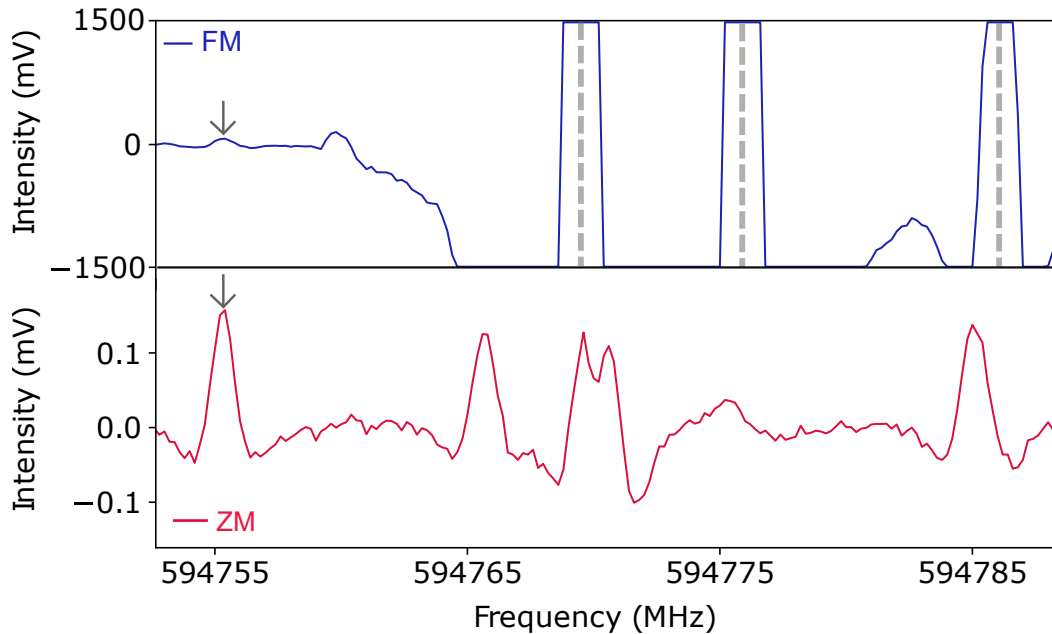


Figure 3: **Comparison between frequency modulation (FM, top trace) and Zeeman modulation (ZM, bottom trace).** The FM spectrum shows three saturated lines for a lock-in sensitivity of 0.5 V, assigned to the CH_3COH precursor. In a standard FM+ZM detection method, no information can be extracted around these three saturated lines and only one transition of CH_2CHO is visible in the region plotted in the Fig. (the one marked with an arrow). The three precursor lines are well filtered in a pure ZM technique, revealing weaker paramagnetic lines assigned to CH_2CHO . Here, the acquisition time was 10 times longer for the ZM than for FM measurement because the Zeeman effect is less efficient in this range (see text).

Spectroscopic analysis

Using the experimental set-up described above, we recorded large spectral surveys in the 110–860 GHz spectral range, avoiding spectral contamination of close-shell species as well as saturation issues of the lock-in amplifier. These portions of spectra allow us to perform a thorough characterization of the (sub-)millimeter wave spectra of the CH_2CHO and CH_3CO radicals, the latter being under analysis. We measured almost 3000 transitions of CH_2CHO , involving relatively high values of the rotational quantum numbers N'' and K_a'' up to 41 and 18 respectively.

We initiated this study by refitting the literature data using a Watson A-reduced Hamiltonian with the SPFIT program³⁸. This data set contains all pure rotational transitions of Endo et al.²⁵ and Endo and Nakajima²¹ with frequency errors ranging from 40 kHz to 400 kHz and 8 kHz to 80 kHz, respectively, based on the weights of the individual lines reported by the authors. In the set of parameters used to adjust the data, Δ_K was fixed to the value determined by IR spectroscopy¹⁹. All lines are reproduced within 5σ of their experimental accuracy, with a weighted standard deviation of 1.064 and an RMS of 59 kHz (see Table 1). However at 3σ rejection factor, the fit does not reproduce all lines of the literature, and 4 lines, at frequencies 95853.419 MHz, 81383.025 MHz, 42368.966 MHz, and 42357.564 MHz, are rejected (weighted standard deviation of 0.98 and RMS of 56.5 kHz). All parameters are in agreement with Endo and Nakajima²¹, except for $T_{ab}(\text{H}_2)$ which is larger (see Table 1), probably due to the difference between the two models. Concerning $T_{ab}(\text{H}_3)$, when adjusted it takes the value of 1.51(72) MHz while not influencing the overall fit quality; we thus decided to remove it from the parameters at this stage.

We assigned 2960 rotational transitions (1013 different frequencies), mostly a -type R-branch transitions. The newly measured a -type lines involve values of N'' and K_a'' up to 41 and 18, respectively (against 7 and 5, respectively, in the previous study). The frequency uncertainties have been chosen based on the line width and the signal-to-noise ratio of each transition, ranging from 50 to 200 kHz. Up to about 500 GHz, many lines show partly resolved hyperfine structure (that results from the H_1 and H_2 atoms). Considering the large number of transitions with no hyperfine structure resolved, we introduced a 'dummy' vibrational state, without any hyperfine

splitting, in the model; this state is described with rotational, centrifugal distortion, and spin-rotation constants fixed to the values in the main hyperfine-split state. To reproduce all these transitions up to 850 GHz, we added only 10 parameters both for the rotation (all sextics diagonal terms with Φ_K , ϕ_K and ϕ_{NK} fixed to the calculated values) and for the spin-rotation (Δ_N^s , Δ_{KN}^s and δ_N^s), which is a reasonable number of parameters comparing to the extension in N'' and K_a'' quantum numbers. Nevertheless, as the number of attributed b -type is low, we decided to include ground state combination difference (GSCD) from Utkin et al.¹⁹ in the fit to improve the determination of A and Δ_K . We calculated 541 GSCD and their weights were attributed according to Utkin et al.¹⁹ at 0.004 cm^{-1} for transitions measured in the jet and at 0.006 cm^{-1} for transitions measured at room temperature. This gives a fit with Δ_K taking the value of $1316.3(19) \text{ kHz}$, very close to the value estimated by Utkin et al.¹⁹. At that point, the fit reproduced all the rotational transitions and the GSCD at their experimental accuracy with a weighted standard deviation of 1.09 and an RMS of 98 kHz for the pure rotational data (with a factor rejection of 5σ) and 0.0035 cm^{-1} for the GSCD.

With this new set of spectroscopic parameters, we were able to search for b -type transitions in our surveys. We measured 59 b -type rotational transitions (32 different frequencies), between 210 GHz and 520 GHz, including 38 Q-branch transitions and 21 R-branch transitions. We did not measure b -type transitions at higher frequencies because above 500 GHz we performed mainly line-by-line acquisitions targeting the a -type transitions. The hyperfine structure of all b -type transitions was unresolved, except for four partially resolved transitions. Fig. 4 shows an example of such a partially resolved b -type transition. The splittings of the energy levels are due to the spin-spin coupling of H_1 , H_2 , and H_3 with the spin of the unpaired electron. Only the hyperfine structure due to H_1 and H_2 is resolved here.

Our final set of parameters shows better determination for all high-order terms of the rotational (quartics, sextics constants) and spin-rotation parameters (Δ_K^s , Δ_N^s , Δ_{KN}^s and δ_N^s), in comparison with the literature values. By adding the b -type transitions to the fit, Δ_K takes the value $1313.10(13) \text{ kHz}$, and is 10 times better determined than previously. The values of the weighted

standard deviation and the RMS are respectively 0.81 and 80 kHz for Endo et al.²⁵, 1.63 and 14.02 kHz for Endo and Nakajima²¹ and 1.07 and 102.8 kHz for the lines measured in this study. We note that the final experimental rotational, centrifugal distortion, and hyperfine parameters are in excellent agreement with the calculated ones. Larger discrepancies are seen with the spin-rotation parameters, although the fitted values remain in excellent agreement with the values derived by Endo and Nakajima²¹.

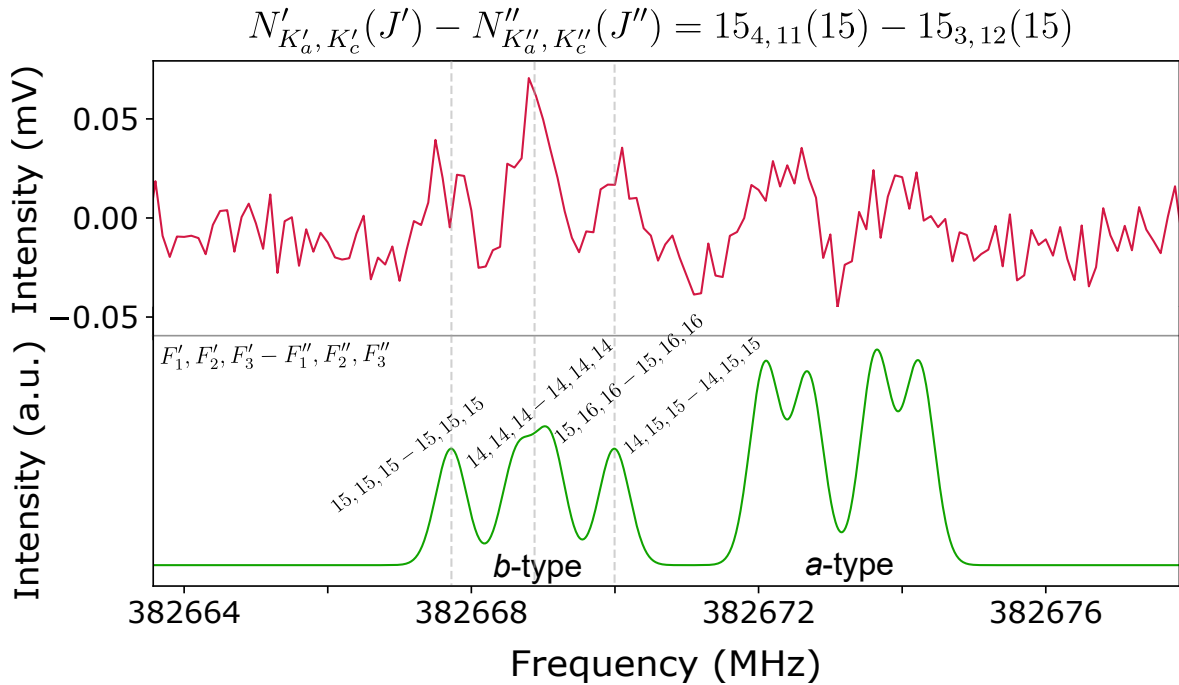


Figure 4: **Partially resolved hyperfine structure of a *b*-type transition of CH₂CHO.** The upper panel contains the experimental trace, and the lower panel is the simulated spectrum using the final set of spectroscopic parameters. Here the hyperfine structure is partially resolved, showing splittings due to the spin-spin interaction of H₁ and H₂ with the unpaired electron.

Astronomical implications

The new set of spectroscopic parameters allows to accurately predict the rotational spectrum of CH₂CHO from centimeter wavelengths up to 850 GHz. Such broad spectral coverage is crucial for light species as CH₂CHO whose rotational spectrum extends to high frequencies, even under low-temperature conditions. For example, at temperatures relevant for hot corinos (e.g., ~ 225 K⁶), the spectrum of CH₂CHO peaks around 550 GHz (see Fig. 5, panel **a**). In this region, transitions

Table 1: Molecular constants for the vinoxy radical

	units	Calculated ^a	Ref ²¹	Refit Lit.	Exp.
A	/MHz	66856.48	66677.85679(159)	66677.8578(17)	66677.8581(13)
B	/MHz	11405.79	11447.0460(55)	11447.0474(69)	11447.04684(41)
C	/MHz	9743.53	9758.9065(53)	9758.9050(68)	9758.90748(36)
Δ_N	/kHz	9.3427847	9.6468(22)	9.642(28)	9.67571(23)
Δ_{NK}	/kHz	-80.607172	-83.045(137)	-83.01(17)	-83.1828(23)
Δ_K	/kHz	1288.6753	[1307.]	[1307.]	1313.10(13)
δ_N	/kHz	2.0122412	2.1215(101)	2.169(14)	2.09528(31)
δ_K	/kHz	37.634069	38.4(26)	38.3(33)	40.669(15)
Φ_N	/Hz	0.0167965			0.017303(99)
Φ_{NK}	/Hz	-0.00289155			-0.0144(11)
Φ_{KN}	/Hz	-7.383695425			-7.6141(83)
Φ_K	/Hz	77.45606136			[77.456061]
ϕ_N	/Hz	0.007414007			0.00812(14)
ϕ_{NK}	/Hz	0.084506752			[0.084506752]
ϕ_K	/Hz	13.43581186			[13.435812]
ϵ_{aa}	/MHz	-969.18773	-899.0720(63)	-899.0677(62)	-899.0777(51)
ϵ_{bb}	/MHz	-75.467329	-66.2053(44)	-66.1975(47)	-66.1995(41)
ϵ_{cc}	/MHz	3.1831907	-0.6128(33)	-0.6150(35)	-0.6250(32)
$(\epsilon_{ab} + \epsilon_{ba})/2^b$	/MHz	-180.28534	84.0021(183)	83.716(35)	83.671(13)
Δ_K^s	/MHz		0.0385(32)	0.0384(26)	0.05216(35)
Δ_N^s	/MHz				0.0002448(37)
Δ_{KN}^s	/MHz				0.00375(14)
δ_N^s	/kHz				0.1519(43)
$a_F(\text{H}_1)$	/MHz	-53.04061	-55.7658(48)	-55.7587(51)	-55.7517(47)
$T_{aa}(\text{H}_1)$	/MHz	13.9884	11.9275(116)	11.917(12)	11.895(11)
$T_{bb}(\text{H}_1)$	/MHz	-13.7458	-13.6902(118)	-13.700(12)	-13.717(12)
$T_{cc}(\text{H}_1)^c$	/MHz	-0.2426		1.783	1.822
$T_{ab}(\text{H}_1)^b$	/MHz	29.917866	-26.97(25)	-26.50(30)	-26.94(24)
$a_F(\text{H}_2)$	/MHz	-51.892733	-54.1511(46)	-54.1548(49)	-54.1543(46)
$T_{aa}(\text{H}_2)$	/MHz	-29.8938	-28.1502(78)	-28.1394(84)	-28.1364(79)
$T_{bb}(\text{H}_2)$	/MHz	31.6087	28.3807(95)	28.369(10)	28.3687(97)
$T_{cc}(\text{H}_2)^c$	/MHz	-1.7149		-0.2296	-0.2323
$T_{ab}(\text{H}_2)^b$	/MHz	-5.5437919	4.91(25)	8.30(67)	8.36(38)
$a_F(\text{H}_3)$	/MHz	1.3706762	-0.5579(32)	-0.5563(34)	-0.5559(33)
$T_{aa}(\text{H}_3)$	/MHz	3.6502	2.2312(78)	2.2115(82)	2.2031(78)
$T_{bb}(\text{H}_3)$	/MHz	1.7105	2.7792(94)	2.7989(99)	2.8134(95)
$T_{cc}(\text{H}_3)^c$	/MHz	-5.3607		-5.0104	-5.0165
$T_{ab}(\text{H}_3)^b$	/MHz	-4.9996091	5.30(25)		[-4.9996091]
$\#/n^d$			585/567	585/567	4142/1774
$N''_{\max}, K''_{a\max}$			7, 5	7, 5	41, 18
RMS ^e	/MHz			0.059	0.096
RMS ^f	/cm ⁻¹				0.00323
σ^g				1.064	1.012

^a The calculated rotational constants have been Bayesian corrected by a 0.9866 factor (see Lee and McCarthy ⁴⁰)

^b The absolute sign cannot be retrieved in the fit (see Endo and Nakajima ²¹)

^c Calculated from $T_{aa} + T_{bb} + T_{cc} = 0$

^d Number of lines ($\#$) and number of lines with different frequencies (n), unitless

^e Root mean square of the pure rotation data, in MHz

^f Root mean square of the GSCD data ¹⁹, in cm⁻¹

^g Weighted standard deviation, unitless

involve relatively large values of the N'' and K_a'' quantum numbers for which the predicting power of the model is often poor when the experimental dataset is limited (in terms of spectral coverage or probed quantum numbers). This deficiency in extrapolating accurate measurements to higher frequency is shown in the panels (b) and (c) of Fig. 5. Near 550 GHz, transitions involving $N'' = 24$ and $K_a'' = 4$ are relatively well predicted using the refit model which contains the literature dataset^{21,25}, with a shift between our measurements and the refit of the literature of the order of ~ 1 MHz (see Fig. 5 (b)). However, for higher values of N'' (Fig. 5 (c)), the shift becomes much larger (~ 36 MHz), avoiding any possible interstellar detection of the radical at these frequencies.

Conclusion and prospects

We have extended the investigation of the pure rotational spectrum of the CH_2CHO radical, from 140 to 850 GHz. CH_2CHO was produced from acetaldehyde, by hydrogen abstraction using fluorine atoms. We used a Zeeman-modulated spectrometer to record the data, allowing us to efficiently filter lines of the precursor that saturates the lock-in amplifier, obliterating a part of the spectrum, in conventional source-frequency modulated spectroscopy. This modulation method makes scanning broad spectral ranges to study radicals possible, instead of scanning line by line. This method is promising to study new radicals never observed previously. The revised vinoxy data set provides a more accurate set of molecular constants that includes transitions at higher frequencies, where N'' and K_a'' are up to 41 and 18 respectively. It improves the quartic centrifugal distortion by about 2 orders of magnitude, and some sextics are determined for the first time. This study may have implications for future detection of CH_2CHO in hot environments of the interstellar medium and up to the THz range.

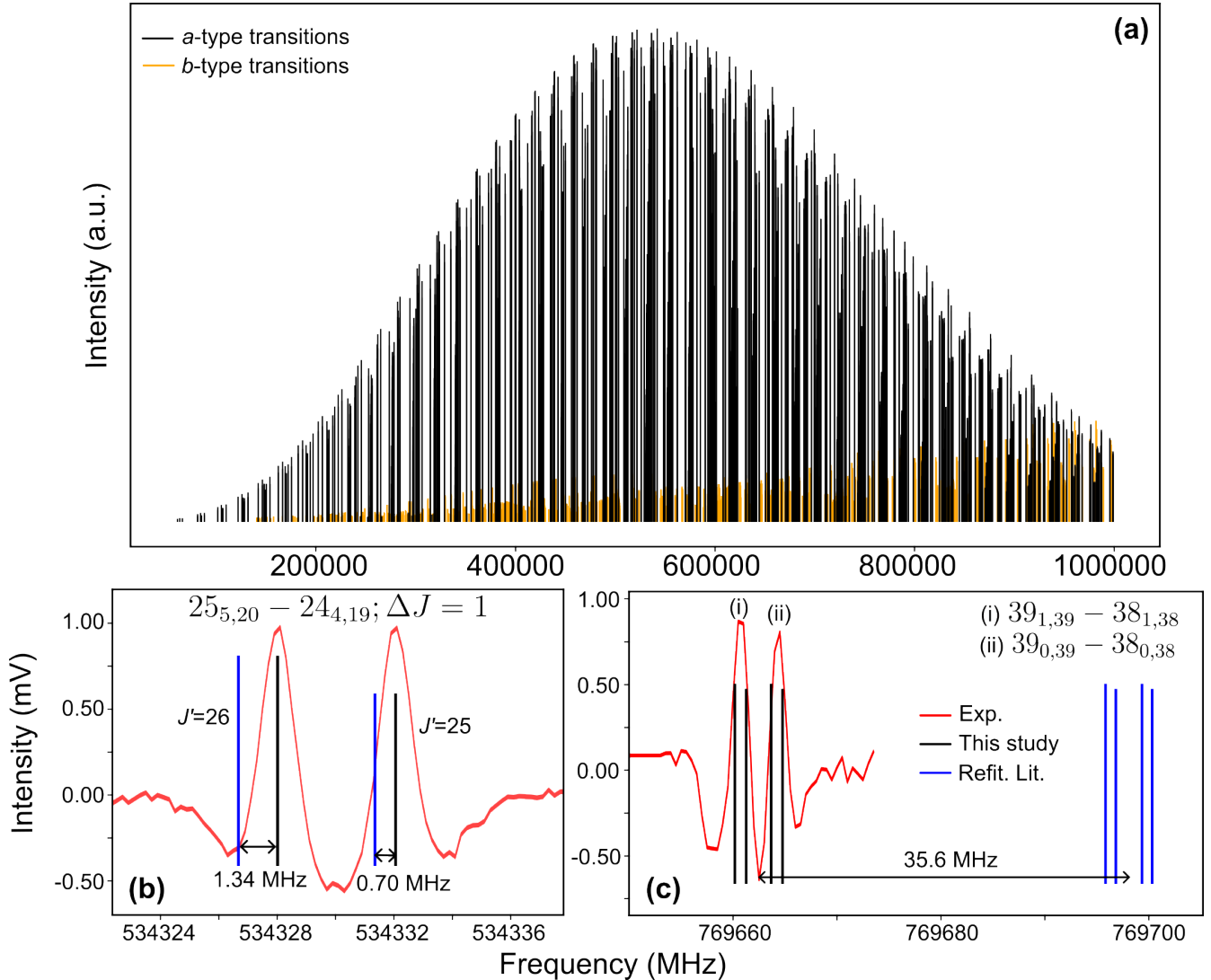


Figure 5: **Pure rotation spectrum of CH₂CHO up to 1 THz at 225 K and comparison of the experimental measurements with models.** (a) Simulation of the spectrum of the vinyoxy radical using the spectroscopic constants derived in this study at $T = 225$ K. a -type transitions are plotted in black and b -type transitions in orange. (b) Zoom around 534 GHz showing two transitions measured experimentally (in red) compared with the simulation using the new set of parameters (in black) and using the refit of the literature data^{21,25} (in blue). (c) Zoom around 769 GHz showing two rotational transitions with no fine structure (the spin-rotation is not resolved). N_{K_a, K_c} and J quantum numbers are used to label the transitions.

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Supporting Information Available

The list of experimental frequencies and all files relevant to the fit and spectral prediction (linelist, parameters, partition function) are available as ASCII files.

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