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► To cite this version:

Olivier Berné, Marie-Aline Martin-Drumel, Ilane Schroetter, Javier Goicoechea, Ugo Jacovella, et al.. Formation of the methyl cation by photochemistry in a protoplanetary disk. *Nature*, 2023, 621 (7977), pp.56-59. 10.1038/s41586-023-06307-x . hal-04539419

HAL Id: hal-04539419

<https://hal.science/hal-04539419>

Submitted on 5 Jul 2024

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Formation of the Methyl Cation by Photochemistry in a Protoplanetary Disk

Olivier Berné¹, Marie-Aline Martin-Drumel², Ilane Schroetter¹, Javier R. Goicoechea³, Ugo Jacovella², Bérenger Gans², Emmanuel Dartois², Laurent Coudert², Edwin Bergin⁴, Felipe Alarcon⁴, Jan Cami^{5,6,7}, Evelyne Roueff⁸, John H. Black⁹, Oskar Asvany¹⁰, Emilie Habart¹¹, Els Peeters^{5,6,7}, Amelie Canin¹, Boris Trahin¹¹, Christine Joblin¹, Stephan Schlemmer¹⁰, Sven Thorwirth¹⁰, Jose Cernicharo³, Maryvonne Gerin⁸, Alexander Tielens^{12,13}, Marion Zannese¹¹, Alain Abergel¹¹, Jeronimo Bernard-Salas^{14,15}, Christiaan Boersma¹⁶, Emeric Bron⁸, Ryan Chown^{5,6}, Sara Cuadrado³, Daniel Dicken¹¹, Meriem Elyajouri¹¹, Asunción Fuente¹⁷, Karl D. Gordon¹⁸, Lina Issa¹, Olga Kannavou¹¹, Baria Khan⁵, Ozan Lacinbala²⁷, David Languignon⁸, Romane Le Gal^{19,20}, Alexandros Maragkoudakis¹⁶, Raphael Meshaka⁸, Yoko Okada¹⁰, Takashi Onaka^{21,22}, Sofia Pasquini⁵, Marc W. Pound¹³, Massimo Robberto¹⁸, Markus Röllig^{23,24}, Bethany Schefter⁵, Thiébaud Schirmer^{11,25}, Aamek Sidhu^{5,6}, Benoit Tabone¹¹, Dries Van De Putte¹⁸, Sílvia Vicente²⁶, Mark G. Wolfire¹³.

¹Institut de Recherche en Astrophysique et Planétologie, Université de Toulouse, CNRS, CNES, UPS, 9 Av. du colonel Roche, 31028 Toulouse Cedex 04, France

²Université Paris-Saclay, CNRS, Institut des Sciences Moléculaires d'Orsay, 91405 Orsay, France

³Instituto de Física Fundamental (CSIC). Calle Serrano 121-123, 28006, Madrid, Spain.

⁴Department of Astronomy, University of Michigan, 1085 South University Avenue, Ann Arbor, MI 48109, USA

⁵Department of Physics & Astronomy, The University of Western Ontario, London ON N6A 3K7, Canada

⁶Institute for Earth and Space Exploration, The University of Western Ontario, London ON N6A 3K7, Canada

⁷Carl Sagan Center, SETI Institute, 339 Bernardo Avenue, Suite 200, Mountain View, CA 94043, USA

⁸LERMA, Observatoire de Paris, PSL University, Sorbonne Université, CNRS, 5 place Janssen, 92190, Meudon Cedex, France.

⁹Department of Space, Earth, and Environment, Chalmers University of Technology, Onsala Space Observatory, 43992 Onsala, Sweden.

¹⁰I. Physikalisches Institut, Universität zu Köln, Zùlpicher Str. 77, 50937 Köln, Germany

¹¹Institut d'Astrophysique Spatiale, Université Paris-Saclay, CNRS, Bâtiment 121, 91405 Orsay Cedex, France

¹²Leiden Observatory, Leiden University, P.O. Box 9513, 2300 RA Leiden, The Netherlands

¹³Astronomy Department, University of Maryland, College Park, MD 20742, USA

¹⁴ACRI-ST, Centre d'Etudes et de Recherche de Grasse (CERGA), 10 Av. Nicolas Copernic, F-06130 Grasse, France

¹⁵INCLASS Common Laboratory., 10 Av. Nicolas Copernic, 06130 Grasse, France

¹⁶NASA Ames Research Center, MS 245-6, Moffett Field, CA 94035-1000, USA

¹⁷Observatorio Astronómico Nacional (OAN,IGN), Alfonso XII, 3, E-28014 Madrid, Spain

¹⁸Space Telescope Science Institute, 3700 San Martin Drive, Baltimore, MD 21218, USA

¹⁹Institut de Planétologie et d'Astrophysique de Grenoble (IPAG), Université Grenoble Alpes, CNRS,

F-38000 Grenoble, France

²⁰Institut de Radioastronomie Millimétrique (IRAM), 300 Rue de la Piscine, F-38406 Saint-Martin d'Hères, France

²¹Department of Physics, Faculty of Science and Engineering, Meisei University, 2-1-1 Hodokubo, Hino, Tokyo 191-8506, Japan

²²Department of Astronomy, Graduate School of Science, The University of Tokyo, 7-3-1 Bunkyo-ku, Tokyo 113-0033, Japan

²³ Physikalischer Verein - Gesellschaft für Bildung und Wissenschaft, Robert-Mayer-Str. 2, 60325 Frankfurt, Germany

²⁴ Goethe-Universität, Physikalisches Institut, Frankfurt am Main, Germany

²⁵Department of Space, Earth and Environment, Chalmers University of Technology, Onsala Space Observatory, SE-439 92 Onsala, Sweden

²⁶Instituto de Astrofísica e Ciências do Espaço, Tapada da Ajuda, Edifício Leste, 2 ° Piso, P-1349-018 Lisboa, Portugal

²⁷ KU Leuven Quantum Solid State Physics (QSP) Celestijnenlaan 200d - box 2414 3001 Leuven

Forty years ago it was proposed that gas phase organic chemistry in the interstellar medium was initiated by the methyl cation CH_3^+ (1–3), but hitherto it has not been observed outside the Solar System (4, 5). Alternative routes involving processes on grain surfaces have been invoked (6, 7). Here we report JWST observations of CH_3^+ in a protoplanetary disk in the Orion star forming region. We find that gas-phase organic chemistry is activated by UV irradiation.

As part of the PDRs4All Early Release Science program on the JWST¹ (8), we have obtained observations of the protoplanetary disk d203-506 (9). This object is situated in the Orion Bar, at about 0.25 pc from the massive, strongly UV emitting Trapezium stars which are at 414 pc from Earth (10), inside the Orion Nebula. The disk is about 100 au in radius, and has an estimated mass of $\sim 10M_{\text{Jup}}$ (Berné et al. in prep.). The central star of d203-506 has a estimated mass of $M_{\star} = 0.2 \pm 0.1M_{\odot}$ (Berné et al. in prep.), typical for stars of the Orion Nebula Cluster (11). This star is obscured by the flared disk that is seen nearly edge-on (9). Fig. 1 shows integrated intensity images of the d203-506 disk (see Methods for details on JWST data reduction). This includes the emission of vibrationally and rotationally

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excited H_2 and CH^+ , and fine-structure emission of Oxygen ([OI]) and ionized iron ([FeII]). The molecular emission arises from a hot ($T_{\text{gas}} \sim 1000$ K) and dense ($n_{\text{H}} > 10^5 \text{ cm}^{-3}$) wind that is produced by photoevaporation from the disk due to irradiation by far-UV photons (FUV; $6 < E < 13.6$ eV) from the Trapezium stars (Berné et al. in prep.). The [FeII] image shows the emission associated with a collimated jet. Some of the wind emission is co-spatial with this jet, but overall the wind is more extended and creates a “halo” around the disk.

The mid infrared spectrum of d203-506 was obtained using the MIRI-MRS spectrometer onboard JWST (see Methods for details) and is shown in Fig. 2. In the spectrum, we detect pure rotational lines of H_2 [0-0 S(1) to 0-0 S(8)] from which we derive an excitation temperature $T_{\text{ex}} = 923 \pm 48$ K (Methods). The straight line observed in the excitation diagram derived from these lines (Extended Data Fig. 3) indicates that the excitation temperature of H_2 is close to the gas kinetic temperature, and thus confirms the presence of hot molecular gas in the wind of d203-506.

In addition to identified H_2 and H I emission lines (see Methods), a strong residual emission consisting in a series of lines in the 6.5–8.0 μm range is observed (Fig. 2). This emission, as seen in the lower middle panel of Fig. 1, is spatially resolved and only present in d203-506. It is co-spatial with H_2 and CH^+ emission, with the best spatial correlation observed with the 2.12 μm line of vibrationally excited H_2 (see Fig. 1). We conclude that the observed features in Fig. 2 are an astrophysical signal associated with emission from the wind of the d203-506 protoplanetary disk. We note however, that given that the angular resolution of MIRI is at the limit to resolve the structures in d203-506, we cannot fully exclude an emission contribution from the jet.

The 7 μm band visible in Fig. 2 is composed of a succession of narrow features corresponding to ro-vibrational transitions of a molecular carrier. Such insight into the detailed structure of the band is enabled by the unprecedented high spectral resolution and high sensitivity provided by JWST in that spectral region. The presence of these resolved structures, and their spectral span, is compatible with a light molecular carrier. The wavelength coincidence between the observed emission features around

7 μm (Fig. 2) and the ν_2 (out-of-plane bending, “umbrella” motion) and ν_4 (in-plane bending) bands of CH_3^+ (12) is striking. Besides CH_3^+ , not a single match has been found for a molecule that would possess its shortest wavelength emission signature at 7 μm (see details in the Supplementary information). Recent laboratory work on the low temperature vibrational spectroscopy of CH_3^+ (13) finds the vibrational bands match the observed wavelengths. Two additional spectroscopic analyses further strengthen the CH_3^+ assignment. First, the intensity pattern of successive emission lines (Fig. 3) is characteristic of the spin statistics of a molecular carrier suggesting three equivalent non-zero-spin atoms (e.g., hydrogen atoms), as expected for CH_3^+ . Second, the observed emission spectrum can be satisfactorily simulated (see Fig. 3) using sets of spectroscopic constants taking values within the range of what is expected from available calculations (Refs. (14, 15) and this work, see Table 1) and laboratory measurements (13). A detailed description of the spectroscopic analysis procedure is given in the Methods. Quantum number assignments to the ro-vibrational transitions in this spectral range will require both additional theoretical input (notably investigating the Coriolis interaction between the two bands (14)) and laboratory measurements, in particular at even higher resolution and ideally at cold temperature to reduce the spectral density. The higher energy ν_3 band of CH_3^+ , situated near 3 μm , has been measured at high resolution in the laboratory (16). Some of the expected lines from the ν_3 band in this spectral range are tentatively detected with NIRSpec in d203-506 (see Supplementary Information and Extended Data Fig. 6). In summary, by spectroscopic standards, CH_3^+ is the best candidate to explain the 7 μm spectral band observed towards the d203-506 protoplanetary disk.

The presence of CH_3^+ in d203-506 raises the question of its origin. Carbon chemistry in these environments typically starts by the radiative association $\text{C}^+ + \text{H}_2 \rightarrow \text{CH}_2^+ + h\nu$, which is a very slow process. The alternative bimolecular reaction $\text{C}^+ + \text{H}_2(v=0) \rightarrow \text{CH}^+ + \text{H}$ [1], is endothermic by $\Delta E/k_B = 4300 \text{ K}$ (17, 18), and thus very slow in cold ($T \sim 100\text{K}$) interstellar gas where $T \ll \Delta E/k_B$. However, strong external FUV radiation fields combined with high gas densities as found in protoplanetary disks open new routes for chemistry. The irradiated gas reaches high temperatures (near 1000

Table 1: Spectroscopic parameters of CH_3^+ in the two excited states ν_2 and ν_4 from experiment and theory, and comparison with constants from our best model of the observed signatures. The ground state values ($v = 0$) are kept fixed to the experimental values determined in Ref. (16). See Table 3 for a more complete set of parameters.

Parameter	Unit	$\nu_2 = 1$		$\nu_4 = 1$	
		Prediction	Model	Prediction	Model
ν	cm^{-1}	1372–1412 ^a	1391	1373–1393 ^b	1375
	μm	7.289–7.082	7.133	7.283–7.179	7.273
B	cm^{-1}	9.06–9.49 ^c	9.37	9.48–9.52 ^c	9.50
C	cm^{-1}	4.61–4.66 ^c	4.66	4.55–4.65 ^c	4.57

^a 1σ confidence interval from the experimental values of Ref. (13) ($1402 \pm 10 \text{ cm}^{-1}$) and Ref. (12) ($1387 \pm 15 \text{ cm}^{-1}$).

^b 1σ confidence interval from the experimental value of Ref. (13) ($1383 \pm 10 \text{ cm}^{-1}$)

^c from Refs. (14, 15) and this work, and scaled to the ground state parameters of Ref. (16), see Table 3

K, (19)) and a significant fraction of the H_2 molecules are radiatively pumped through fluorescence to vibrationally excited states (20), $\text{H}_2^*(v>0)$. This suprathreshold excitation overcomes the endothermicity of reaction [1], allowing H_2^* to react with C^+ , leading to the formation of abundant CH^+ (21–23). Subsequent fast and exothermic hydrogen abstraction reactions $\text{CH}^+ \xrightarrow{(2)} \text{CH}_2^+ \xrightarrow{(3)} \text{CH}_3^+$ then efficiently lead to CH_3^+ . In the Methods section, we quantitatively assess these processes using models, and show that for a wide range of acceptable parameters, CH_3^+ is formed efficiently in FUV irradiated environments. The formed CH_3^+ reacts very slowly with H_2 (through radiative association) and is mainly destroyed by dissociative recombination with electrons, leading to CH_2 , CH , and C in comparable amounts (24). CH_3^+ can also be destroyed by reactions with neutral oxygen producing HCO^+ and with neutral molecules producing molecular ions. These undergo dissociative recombination with electrons yielding complex organic molecules. Therefore, in the presence of UV radiation, gas-phase organic chemistry is initiated through CH_3^+ (2, 3, 25).

The ongoing chemistry in d203-506 described above differs greatly from what has been observed in disks that are not exposed to external UV irradiation where the freeze out of H_2O and CO_2 control the

gas composition. In such disks, high abundances of water, HCN, CH₄, C₂H₂, etc. are observed (26, 27), species which are not detected in d203-506. In the last decades, the formation of organic molecules in space has been considered to happen mostly at the surface of grains (6, 7). The detection of CH₃⁺ indicates that alternative gas-phase routes are available to activate the organic chemistry, when UV radiation is present. Far from being anecdotal, external UV irradiation is expected to occur during the early life of most protoplanetary disks (28), making UV-driven organic chemistry common for the chemical evolution of most protoplanetary disks and of the early the Solar System (29, 30). More generally, this chemistry can be active in any environment providing sufficiently high gas density and FUV irradiation ($n_{\text{H}} \gtrsim 10^5 \text{ cm}^{-3}$, $G_0 \gtrsim 10^4$). This can include, for instance, star-forming regions, the envelopes of planetary nebulae, the inner regions of disks around T-Tauri stars, and the interstellar medium of star-forming galaxies near and far. While the CH₃⁺ detection presented here is a promising achievement, there are still numerous unanswered questions surrounding the excitation, chemistry, and spectroscopic properties of this species (Methods and Supplementary Information). These topics shall be addressed thanks to interdisciplinary scientific efforts that incorporate the expertise of astronomers, physicists, and spectroscopists (both laboratory and theory) in order to fully understand the role of CH₃⁺ in organic chemistry in space.

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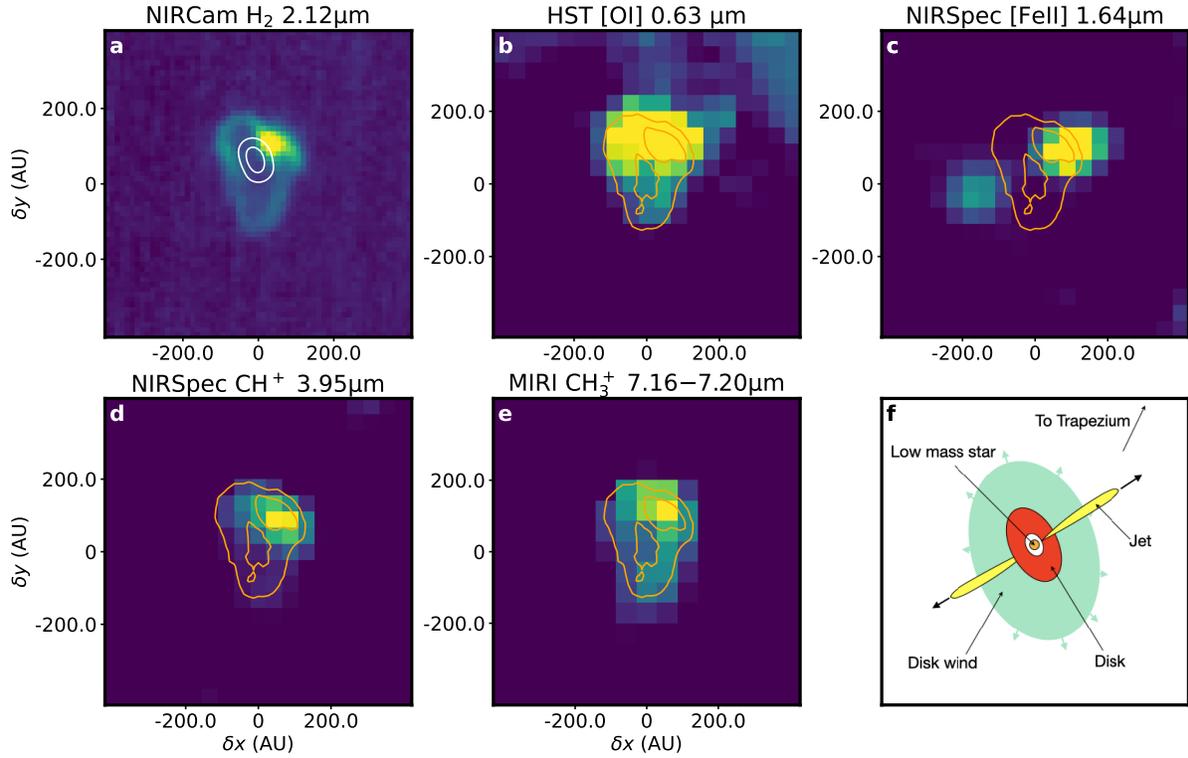


Figure 1: Overview of the d203-506 externally irradiated protoplanetary disk. Integrated intensity images from NIRCcam F212N filter (a), Hubble Space Telescope [OI] (b), NIRSpec [FeII] (c), NIRSpec CH⁺ 1-0 *P*(7) at 3.95 μm (d), and MIRI MRS integrated from 7.16 to 7.20 μm assigned to CH₃⁺ (e). Each panel is centered at $\alpha=5:35:20.318$ and $\delta=-5:25:05.662$ and are $2'' \times 2''$ wide. Contours of NIRCcam vibrationally excited H₂ (2.12 μm band) are represented in orange, and ALMA dust continuum emission at 344 GHz from the disk in white contours. (f) Sketch of the d203-506 disk, jet, and wind. We note that the low mass star is not seen in the images because of the disk flaring. All JWST images are from the PDRs4All program, while the Hubble image is from (9).

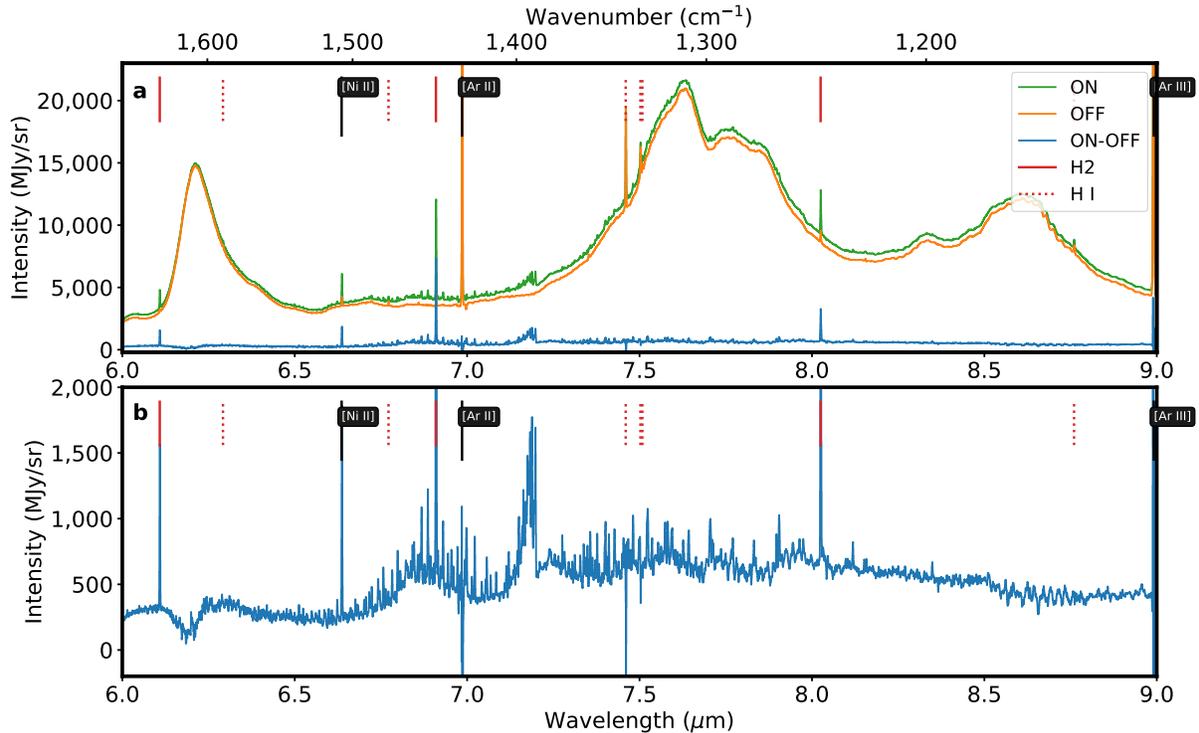


Figure 2: JWST-MRIRI spectra of d203-506. **a**, Spectrum on the position of d203-506 (ON, green) and close to 203-506 (OFF, orange) over the 6-9 μm MIRI-MRS spectral range. The OFF spectrum is dominated by emission of the Orion Nebula: The broad features at 6.2, 7.7 and 8.6 μm are due to the emission of UV excited polycyclic aromatic hydrocarbons (PAHs, see Chown et al. in prep. for a detailed discussion). **b**, Spectrum of d203-506 after subtraction of the nebular emission (ON – OFF).

Acknowledgments

OB is funded by a CNES APR program. MIRI data reduction is performed at the French MIRI centre of expertise with the support of CNES and the ANR-labcom INCLASS between IAS and the company ACRI-ST. Part of this work was supported by the Programme National “Physique et Chimie du Milieu Interstellaire” (PCMI) of CNRS/INSU with INC/INP co-funded by CEA and CNES. Quantum chemical calculations were performed using HPC resources from the “Mésocentre” computing center of CentraleSupélec and École Normale Supérieure Paris-Saclay supported by CNRS and Région Île-de-France (<http://mesocentre.centralesupelec.fr/>). JRG and SC thank the Spanish MCINN for funding support un-

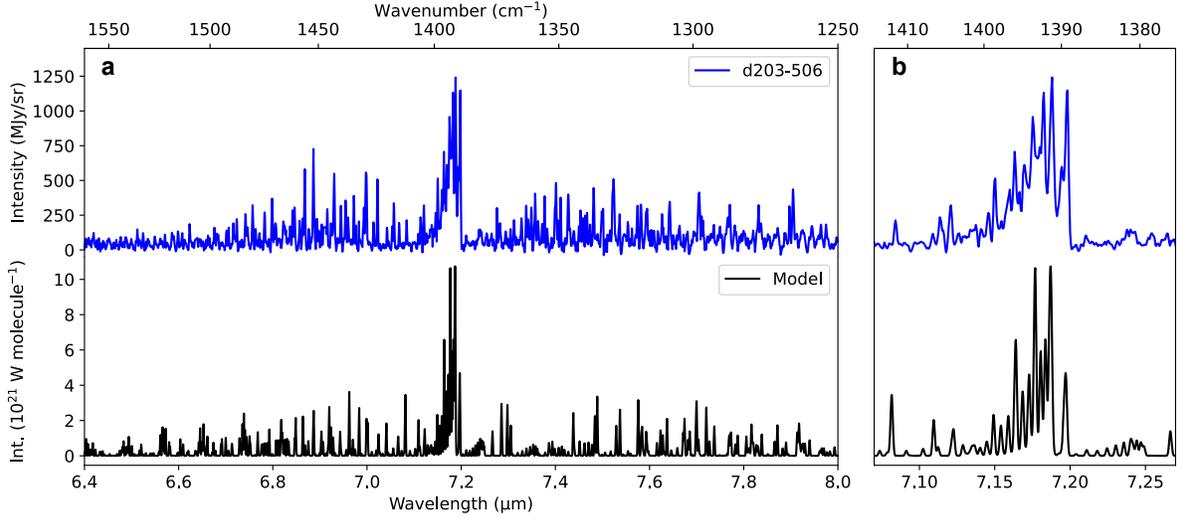


Figure 3: Comparison between the observed JWST spectrum of d203-506 and modeled CH_3^+ spectrum. **a**, Full spectrum. **b**, Zoom on the strongest lines. The model (black curve) for the ν_2 and ν_4 vibrations of CH_3^+ was obtained using the constants of Table 1 (see Methods for a more detailed description of the simulation). A Gaussian linewidth of 0.35 cm^{-1} , corresponding to the MIRI MRS resolution in this range ($\frac{\lambda}{\Delta\lambda} \sim 3800$), has been used in the simulation performed using PGOPHER (31). For clarity, the observational spectrum has been baseline corrected using a spline function and strong individual lines have been removed from the plot : [Ni II] at 6.63, H_2 at 6.92, [Ar II] at 6.99, He I at 7.47, and H_2 at 8.02 μm . The standard deviation (1σ) of the noise level is $\sim 10 \text{ MJy sr}^{-1}$ in this range. The observed lines are 10 to 100 times this noise level.

der grant PID2019-106110GB-I00. JC and EP acknowledge support from the University of Western Ontario, the Institute for Earth and Space Exploration, the Canadian Space Agency, and the Natural Sciences and Engineering Research Council of Canada. The Cologne spectroscopy group acknowledges funding by the Deutsche Forschungsgemeinschaft DFG (CRC956, sub-project B2, ID 184018867) and the ERC AdG Missions (ID: 101020583). Work by YO and MR is carried out within the Collaborative Research Centre 956, sub-project C1, funded by the DFG – project ID 184018867. CB is grateful for an appointment at NASA Ames Research Center through the San José State University Research Foundation (80NSSC22M0107). TO acknowledges support from JSPS Bilateral Program, Grant Number 120219939.

Author contributions

O.B. found the signal in the data and led the analysis of the data and write-up of the article. M.A.M.D., I.S., U.J., B.G., E.D., L.C., E.B., F.A., J.C., E.R., J.B., O.A., C.J., S.S., S.T., J.C., M.G., A.T., T.O., M.Z. conducted the spectroscopic analysis and participated in the write-up. M.A.M.D. created figures 3, S4, S5. I.S. created figures 1, 2, S1, S2, S3. I.S. and O.B. created figure S6. J.G. performed the chemical models and figures S7, S8 and participated in the write-up. O.B., E.H., E.P. led the JWST observing program. I.S., J.G., E.D., E.B., F.A., J.C., A.C., B.T., C.J., A.T., M.Z., A.A., J.B.S., C.B., E.B., R.C., S.C., D.D., M.E., A.F., K.D., L.I., O.K., B.K., O.L., D.L., R.L.G., A.M., R.M., Y.O., T.O., S.P., M.P., M.R., M.R., B.S., T.S., A.S., B.T., D.V.P., S.V. and M.W. contributed to the observing program with JWST. I.S., A.C., R.C., A.S., B.T., F.A., D.V.P. reduced the data. E.D., M.A.M.D., L.C., J.G. and O.B. conducted the column density analysis. J.H.B. wrote the section on the excitation of CH_3^+ . M.W. and J.H.B. corrected the English throughout the manuscript. All authors contributed to the discussions and provided feedback on the manuscript.

Methods

Observations and data reduction

The JWST/MIRI Medium Resolution Spectroscopy (MRS) Integral Field Unit (IFU) data were obtained on 2023 January 31 as part of the JWST ERS 1288 program (PI: O. Berné, E. Habart, E. Peeters, (32)) referred to as “PDRs4All”. All four channels and the three sub-channels were used, covering a wavelength range of 4.9–28 μm at a spectral resolution of 4000–1500 (33). The observations are centered on $\text{RA}=05^{\text{h}}35^{\text{m}}20^{\text{s}}.4749$ $\text{DEC}= -05^{\circ}25', 10''.45$ and span a mosaic of 9 pointings. The overall science exposure time is 14086.11s for the whole mosaic. We used the FASTR1 readout mode with 4-point dithering. We reduced the data using the JWST Data Reduction pipeline version 1.9.5. The stage 2 residual fringe correction was applied in addition to the standard fringe correction step. A master background

subtraction was applied in stage 3 of the reduction. At the end of the data reduction, we obtained four MIRI datacubes, one for each channel, each channel containing its three corresponding sub-channels (short, medium and long). The details of the data reduction for MIRI-IFU is part of a dedicated paper by the PDRs4All team (Chown et al. in preparation).

In this paper we also use one narrow band image of the NIRSpec observations of the same object as well as a NIRCам filter. Specifically, we use the NIRCам F212N filter image and the NIRSpec spectral cube corresponding to the F290LP filter which spans from ~ 2.9 to $5 \mu\text{m}$. Observations were obtained with JWST-NIRSpec (JWST-NIRCам) on 2022 September 10 and reduced using the JWST pipeline version 1.9.4 (1.7.1) with Calibration Reference Data System (CRDS) context file `iwst_1014.pmap` (`iwst_0969.pmap`). For the NIRCам observations, no OFF emission was subtracted. For the NIRSpec observations, a dedicated OFF observation was subtracted. The details of the data reduction for these two instruments is discussed in a dedicated paper by the PDRs4All team on NIRCам (Habart et al. in prep) and NIRSpec (Peeters et al. in prep). The details on the observing strategy can be found in (34).

Data analysis

In order to work with complete spectra spanning all the MIRI MRS wavelength range of $4.9\text{--}28 \mu\text{m}$, we first stitch the spectra of the four channels to remove jumps between spectral orders. For each spectrum, we arbitrarily choose the one from Channel 2 long as a reference. Shorter and longer wavelengths (Channels 1, 3, and 4) are thus scaled accordingly. We create (with numpy) an array of wavelengths spanning the full wavelength range, i.e. between 4.900 and $27.901 \mu\text{m}$, with $50\,000$ points and a constant step. Each channel spectrum is then interpolated onto this grid and jumps are removed by scaling each spectral order based on the average flux in the overlap region. The integrated intensity is conserved in this procedure, and uncertainties propagated.

The two extracted spectra shown in blue and orange in Fig. 2 are from the “ON” and “OFF”

position, respectively. The ON (OFF) spectrum were extracted from an ellipse (circular aperture) centered on the position $\alpha = 5:35:20.357$, $\delta = -5:25:05.81$ ($\alpha = 5:35:20.370$, $\delta = -5:25:04.97$), with dimension $l = 0.52''$, $h = 0.38''$ (of radius $r = 0.365''$) and a position angle PA= $+33$ (0) degrees (trigonometric) with respect to North. In order to have the emission of d203-506 we choose to subtract the nebula emission by evaluating ON–OFF. The ON and OFF spectra over the full MIRI-MRS range are shown in Extended Data Fig. 1, and the full subtracted spectrum is shown in Extended Data Fig. 2. In Extended Data Fig. 2, some lines are negative due to the over-subtraction of ionized emission lines which dominate in the nebula but are absent in d203-506. Some PAH bands are seen in negative in the ON–OFF spectrum; this is due to intrinsic variation of the PAH bands due to e.g. size or ionization, and this cannot be interpreted as PAH absorption.

Line identification

Using the ON-OFF spectrum described previously and the line list provided by the PDRs4All ERS team (8), we identified the strong emission lines present in the data. The main emission lines are from H I and H₂ and are listed in Table 1 and 2, respectively. The H₂ lines intensities presented in the latter table are measured using a Gaussian fitting to the observed lines, and the wavelength of the H₂ transition from (35). The approach to fit the observed lines is presented in (36). Nebular emission lines from atomic ions are also identified and are shown in Figures 2 and 2 with black vertical lines with their name in the attached box. In addition, a number of OH lines are also identified between 9 and 11 μm and are shown as green vertical lines on the same Figures. For this wavelength range we used OH wavelengths from (37) and (38). The study of OH emission in d203-506 will be the subject of a forthcoming paper (Zannese et al. in prep.).

From the H₂ lines listed in Table 2 we derive an excitation diagram, using the H₂ Toolbox (39) developed as part of the PDRs4All project science enabling products (see <https://pdrs4all.org/seps/> and). This is a tool for fitting temperature, column density, and ortho-to-para ratio in

H₂ excitation diagrams. A one or two temperature model is assumed, and the fit finds the excitation temperatures and column densities, and optionally ortho-to-para ratio. The source code is available at <https://dustem.astro.umd.edu/tools.html>. (?)he result of this analysis is shown in Extended Data Fig. 3. The derived excitation temperature is $T_{\text{ex}} = 923 \pm 48.2K$.

Other candidate molecules

There are no unassigned series of lines observed in the 5.2–6.2 μm range, corresponding to C=O or C \equiv N vibrations (Fig. 2) thus excluding most small species containing these chemical functions as carrier of the observed 7 μm features. At longer wavelengths (8–17 μm , see Extended Data Fig. 2), the spectrum is devoid of strong unassigned emission lines. Many hydrocarbon molecules, radicals, and ions (e.g., CH₂, CH₂⁺, ...) possess low frequency modes and would thus emit in that range. Instead, the lowest vibrational modes of CH₃⁺ lie at 7 μm . We thoroughly inspected the literature data on other hydrocarbons and known interstellar species (both neutral and charged) for possible matches and used local thermodynamic equilibrium (LTE) models to predict the emission from molecules in this spectral range. The tested molecules include—not exhaustively—H₂O (and isotopologues), H₂O⁺, NH₄⁺, C₂H₂, CH₃, HCN, SO₂, all hydrocarbons present in the HITRAN database (40), NH₃, CH₃OH, and CH₃CN.

Spectroscopy of the Methyl cation

CH₃⁺ is a planar molecule belonging to the D_{3h} group of symmetry. It possesses four fundamental modes of vibration following an irreducible representation $\Gamma = 1A'_1 + 1A''_2 + 2E'$ (the two E' modes are doubly degenerate). It is an oblate symmetric-top molecule for which rotational energy levels of non-degenerate vibrational states are described with two quantum numbers, J , the total-angular-momentum quantum number excluding nuclear spin, and K , the projection of \vec{N} (the total angular momentum excluding nuclear and electron spins) along the principal axis of symmetry; an additional l quantum number accounts for Coriolis-coupling in degenerate vibrational states ($l = \pm 1$ in $v_4 = 1$). The energy levels in a given vibronic state can be calculated using the energy of the vibronic state, two rotational constants ($A = B$

and C), centrifugal distortion parameters ($D_J, D_{JK}, D_{K\dots}$), and, for states of E' symmetry, additional Coriolis-coupling and l -doubling parameters (ζ, η_J, q). The molecule possesses three equivalent hydrogen atoms (fermions, $I_{\text{H}} = 1/2$) resulting in spin-statistical weights of (0, 0, 4, 4, 2, 2) for the levels of the states of symmetries ($A'_1, A''_1, A'_2, A''_2, E', E''$) (41). In the following, we used the PGOPHER software (31) to simulate the rovibrational spectrum of CH_3^+ .

Since no experimental information is available on the $v_2 = 1$ (A''_2 symmetry) and $v_4 = 1$ (E' symmetry) states of CH_3^+ , except the band positions (12, 13), we rely on quantum chemical calculations to estimate the rotational constants in these excited states. We carried out geometry optimization and anharmonic frequency analysis at the $\omega\text{B97X-D/cc-pVQZ}$ level of theory (42–44) using the Gaussian 16 suite of electronic structure programs (45). The main results from these calculations are reported in Table 3. The table also contains the calculated constants reported in the literature (14, 15). Besides the rotational constants, the quantum chemical calculations also give us insights about the transition moments of the ν_2 (out-of-plane bending, “umbrella” motion) and ν_4 (in-plane bending) bands. Our calculations predict transition moments of 0.084 D and 0.064 D for ν_2 and ν_4 , respectively. Such low values have to be taken with caution as experimental values can differ significantly, by 0.1 D or more. Nevertheless, these values are consistent with the small values obtained by CCSD(T)/cc-pwCVTZ calculations performed in course of the work presented in Ref. (13) (0.06 D and 0.07 D, respectively), as well as those reported in Ref. (46) (0.049 D and 0.111 D) and Ref. (47) (0.10 D and 0.16 D).

To assess the reliability of the spectroscopic constants derived from the different quantum chemical calculations (both from this work and the literature), the calculated values in $v = 0$ and $v_3 = 1$ were compared to the experimental values of the ν_3 band. Transitions within the ν_3 band (asymmetric stretching) observed by Crofton et al. (16) were fitted in the present work using PGOPHER (so as to use the same model for ν_3 and ν_2/ν_4) leading to the spectroscopic parameters reported in Table 3. These derived parameters are in excellent agreement with those reported in Ref. (16). They also are in very good agreement with the parameters obtained by the anharmonic calculations from this study. These

results were used to scale the calculated constants in $v_2 = 1$ and $v_4 = 1$ according to the formula $B_{\text{scaled } v_i} = B_{\text{calc } v_i} \times B_{\text{exp } v_3=1} / B_{\text{calc } v_3=1}$ (and similarly for C). For the rotational constants calculated in Ref. (14), since no calculated values in $v_3 = 1$ are reported, the scaling was made using values in $v = 0$. The scaled values appear in blue in Table 3; they serve as a range of confidence for the spectral simulations.

Despite our best efforts, no definite spectroscopic analysis of the bands observed by JWST was achieved, i.e., we could not assign with confidence quantum numbers to the observed transitions. Instead, we performed several simulations by varying the rotational constants of $v_2 = 1$ and $v_4 = 1$ until qualitative agreement with the observational data was achieved. Such qualitative agreement was assessed using the following criteria (by decreasing order of importance): i) Q-branch line position and intensity; ii) P- and R-branches spread and spectral line density; iii) line positions and intensities in the P- and R-branches. In all cases, the Coriolis interaction constants were kept fixed to the calculated values of those in $v_3 = 1$, and so were the transition moments of the ν_2 and ν_4 bands. This led us to produce four models, I to IV (see Table 3 and Figures 4-5; model III being the model presented in the main article), that reasonably reproduce the astronomical data. It is worth noting, however, that the spectral density is greater on the spectrum of d203-506 than in our simulations, which could either reflect the presence of another species, or some discrepancies in the rotational constants used in models I to IV. Regarding the temperature, for all models, a rotational temperature of 400 K seems to adequately simulate the astronomical features (i.e., the spread of the P-, Q- and R-branches). Higher temperatures lead to P- and R-branches that spread further than what is observed in the astronomical spectrum. While the simulations were performed under the assumption of thermal equilibrium, astronomical excitation conditions may differ significantly (see detailed discussion in the following section). This is particularly relevant to the relative intensities of the ν_2 and ν_4 bands, since the $v_2 = 1$ and $v_4 = 1$ vibrational states may be populated differently in d203-506.

We have also investigated the presence of lines emanating from the ν_3 band of CH_3^+ near $3 \mu\text{m}$ in

the NIRSpec data of d203-506 presented in Berné et al.(in prep.). The ν_3 band has been observed at high resolution in the laboratory by Refs. (16, 48). Using the constants derived from these observations, we computed a synthetic spectrum of the ν_3 band at 400 K and compare it to the NIRSpec background subtracted spectrum (using the same apertures as for MIRI, see Methods) in Extended Data Fig. 6. The predicted lines are found to coincide with lines in the observations.

In summary, high-resolution laboratory infrared spectra of CH_3^+ in the 7 μm region are needed to identify individual transitions in the astronomical data. First laboratory measurements at low temperatures will initiate this process, hopefully deciphering the strong Coriolis coupling between the degenerate ν_2 and ν_4 vibrations of CH_3^+ . In order to support a quantitative analysis of the astronomical spectra, laboratory works also conducted at higher temperatures are required.

Chemistry of the Methyl cation

We modeled the chemistry in the strongly FUV-irradiated photoevaporative wind and upper disk layers of d203-506 using the Meudon Photodissociation Region (PDR hereafter) code (49). The code solves the FUV radiative transfer in a medium of gas and dust (50), as well as the steady-state heating, cooling, chemistry, and $\text{H}_2(v, J)$ level populations as a function of depth into the neutral disk (in mag of visual extinction A_V). Berné et al.(in prep.) previously used this code to reproduce the H_2 line intensities detected by JWST/NIRSpec and obtained a radiation field intensity $G_0 \simeq 4 \times 10^4$ and a gas density $n_{\text{H}} = n(\text{H}) + 2n(\text{H}_2) \simeq 3.5 \times 10^5 \text{ cm}^{-3}$ as the best-fit. We first adopt the same physical parameters and use an extinction law suited to Orion molecular cloud (51), and consistent with dust grains ($R_V = 5.62$ and $N_{\text{H}}/E(B - V) = 1.05 \times 10^{22} \text{ cm}^{-2}$) bigger than in standard diffuse interstellar clouds ($R_V = 3.1$). This choice leads to a FUV dust extinction cross-section, $\sigma_{1000\text{\AA}}(\text{FUV}) = 1.1 \times 10^{-21} \text{ cm}^{-2}/\text{H}$, which is in the upper range of the cross-sections estimated by Berné et al. (in prep.) for this object, and a factor of about two smaller than standard ISM grains. In addition, we run models adopting "bigger grains" (by a factor of about four), leading to $\sigma_{1000\text{\AA}}(\text{FUV}) = 7 \times 10^{-22} \text{ cm}^{-2}/\text{H}$. This smaller FUV cross-section is

still compatible with the wind models presented by Berné et al. (in prep.) and with the kind of dust grains expected in the upper layers of protoplanetary disks (52).

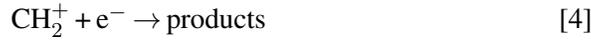
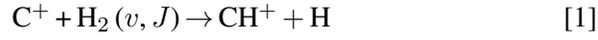
Extended Data Fig. 7 shows the predicted density and temperature structure of the wind and upper disk layers (upper panels) as well as the C^+ , CH^+ , CH_2^+ , CH_3^+ , CO , and HCO^+ abundance profiles (lower panels). Figure 7a refers to models using “Orion grains” and $n_H = 3.5 \times 10^5 \text{ cm}^{-3}$ (Berné et al. in prep.). The other plots refer to models adopting “bigger grains” and densities of $3.5 \times 10^5 \text{ cm}^{-3}$ (Extended Data Fig. 7b), $3.5 \times 10^6 \text{ cm}^{-3}$ (Extended Data Fig. 7c), and 10^7 cm^{-3} (Extended Data Fig. 7d). The latter ones are more representative of the outer layers of a disk (53).

In all these models, reaction $C^+ + H_2(v, J) \rightarrow CH^+ + H$ [1] drives the formation of CH^+ as soon as the H_2 abundance rises. In these conditions, high temperatures and presence of FUV-pumped vibrationally excited H_2^* , reaction [1] is much faster than the slow radiative association reactions $C^+ + H_2 \rightarrow CH_2^+ + \text{photon}$ and $C^+ + H \rightarrow CH^+ + \text{photon}$. These radiative associations produce small amounts of CH_2^+ and CH^+ in cold gas ($T < 100 \text{ K}$). Our models include an $H_2(v, J)$ state-dependent treatment of reaction [1] (18), appropriate to the nonthermal populations of H_2^* in FUV-irradiated environments. In particular, the CH^+ formation rate is computed by summing over all formation rates for each specific state of H_2 . Once CH^+ is formed, fast and exoergic hydrogen abstraction reactions $CH^+ \xrightarrow{H_2} CH_2^+ \xrightarrow{H_2} CH_3^+$ lead to CH_3^+ . The *efficiency* of this chemical pipe to CH_3^+ depends on the abundance of H atoms in the gas (because they readily react with CH^+ ; (54)) and that of electrons (because they destroy CH_2^+ and CH_3^+). Reactions with atomic hydrogen dominate CH^+ destruction when the molecular gas fraction, defined as $f_{H_2} = 2n(H_2)/n_H$, is $\lesssim 0.5$ ($f_{H_2} = 1$ when all hydrogen is in molecular form). All models in Extended Data Fig. 7 predict that the CH_3^+ abundance peaks close to the H/ H_2 transition, at $A_V \simeq 1 \text{ mag}$, where $T \simeq 1000\text{-}800 \text{ K}$. Extended Data Fig. 8 summarizes the dominant chemical reactions at the CH_3^+ abundance peak. We note that the CH^+ and CH_3^+ abundance profiles roughly follow the density profile of vibrationally excited H_2^* (dotted black curve in the upper panel of Extended Data Fig. 7). Hence, irrespective of the exact gas density value, the infrared H_2 lines detected with JWST in d203-506 probe

strongly FUV-irradiated hot gas, where CH^+ and CH_3^+ efficiently form.

Our models predict $N(\text{CH}_3^+)/N(\text{CH}^+)$ column density ratios of $\simeq 1\text{--}15$ (increasing as n_{H} increases). We note that the bulk column density of these species stem from FUV-illuminated gas at $A_V < 3$ mag. Deeper inside, their abundances drop by orders of magnitude. Hence, both CH^+ and CH_3^+ are chemical tracers of the most irradiated wind and upper disk layers.

The local $\text{CH}^+/\text{CH}_3^+$ abundance ratio can be analytically estimated from the following network of gas-phase chemical reactions:



We note that adopting the photodissociation rate of (55), CH_3^+ photodissociation is expected to be much slower than dissociative recombination with electrons even in strong UV fields. Therefore, in steady-state one obtains:

$$\frac{x(\text{CH}_3^+)}{x(\text{CH}^+)} = \frac{k_{2b} f_{\text{H}_2}^2}{2k_5 x_e (f_{\text{H}_2} + 2x_e k_4/k_3)} \simeq \frac{k_{2b}}{2k_5 x_e} f_{\text{H}_2} \quad (1)$$

where $x_e = n(\text{e}^-)/n_{\text{H}}$ is the electron abundance. In the last step we assumed that CH_2^+ destruction by reactions with H_2 are much faster than dissociative recombinations with electrons at the CH_3^+ abundance peak (as confirmed by the model). In our model we used the following reaction rate coefficients: $k_{2b} = 1.2 \times 10^{-9} \text{ cm}^3 \text{ s}^{-1}$ (56), $k_3 = 1.6 \times 10^{-9} \text{ cm}^3 \text{ s}^{-1}$ (57), $k_4 = 6.40 \times 10^{-7} (T/300)^{-0.60} \text{ cm}^3 \text{ s}^{-1}$ (58), and $k_5 = 6.97 \times 10^{-7} (T/300)^{-0.61} \text{ cm}^3 \text{ s}^{-1}$ (24). Using representative values for the wind and upper disk layers of d203-506; $T = 900 \text{ K}$ and $x_e \simeq x(\text{C}^+) \simeq 1.4 \times 10^{-4}$, one obtains $x(\text{CH}_3^+)/x(\text{CH}^+) \simeq 12 f_{\text{H}_2}$ from Eq. (1). This analytical abundance ratio agrees with the detailed predictions of our photochemical

models. That is, the above chemical reactions dominate the formation of CH_3^+ in FUV-irradiated gas. In particular, models predict that the CH_3^+ abundance peaks at gas molecular fractions of $f_{\text{H}_2} = 0.3\text{--}0.5$ (pink dotted curves in Figs. 7).

Reaction [5] produces CH_2 , CH , and C in similar amounts (24). These are key reactive intermediate species that trigger the chemistry of carbon species (25). In addition, reaction $\text{O} + \text{CH}_3^+ \rightarrow \text{HCO}^+ + \text{H}_2$ (59) is a dominant source of HCO^+ , and thus of CO , in these irradiated hot gas layers (Extended Data Fig. 7 also shows the predicted HCO^+ abundance profile in the photoevaporative wind and upper disk layers). The morphology of the observed HCO^+ $J=4\text{--}3$ line emission (first detected by (19) and then mapped with ALMA at high angular resolution by (Berné et al. in prep.)) resembles that of vibrationally excited H_2 and CH_3^+ observed with JWST (Berné et al. in prep.). It will be difficult to explain the presence of extended HCO^+ emission in these strongly irradiated gas layers without the FUV-driven chemistry described here and tested by the presence of CH_3^+ . We note that this hot HCO^+ linked to the extended H_2^* emission is different from the HCO^+ present in lower and denser layers of protoplanetary disks and formed by standard ion-molecule chemistry (53).

Specific 2D models, better adapted to the geometry of the upper disk layers and FUV-irradiated wind, will be needed to fully understand the density structure and abundance distribution of the observed molecular emission with JWST.

Data Availability

The JWST data presented in this paper is publicly available through the MAST online archive (<http://mast.stsci.edu>) using the PID 1288. The MIRI spectra presented in Fig. 2 and Extended Data Figs. 1,2 are available in ASCII format at <https://doi.org/10.5281/zenodo.7989669>. The p-Gopher files to create the model spectra of CH_3^+ are available via : <https://doi.org/10.5281/zenodo.7993330>.

Code Availability

The JWST pipeline used to produce the final data products presented in this article is available at <https://github.com/spacetelescope/jwst>. The MEUDON PDR code is publicly available at https://ism.obspm.fr/pdr_download.html

Author information

Supplementary Information is available for this paper.

Correspondence and requests for materials should be addressed to Olivier Berné.

The authors declare no financial or non-financial competing interest.

Supplementary information

Supplementary methods

ν_3 emission of CH_3^+

CH_3^+ has a band (ν_3) near $3\mu\text{m}$. This spectral range was observed as part of the PDRs4All ERS program with the NIRSpec instrument. These data are subject of other papers in preparation, however we present here an analysis of these data for d203-506 in the context of the CH_3^+ detection. The NIRSpec spectrum in the spectral range of the ν_3 band is shown in Fig. 6. The shaded region represents the $\pm 3\sigma$ error interval of the data. This error interval has been computed as the sum of the errors provided by the JWST pipeline and an empirical error. The empirical error is three times the standard deviation of noise in the 3.102 to 3.126 spectral range of NIRSpec, where no lines are present. In the same figure we show a best fit model of the OH lines which are present in this spectral range. This is an LTE model at a temperature of 800K. A detailed model of the OH emission will be presented in a forthcoming paper (Zannese et al. in prep). The objective here is not to obtain the best possible model, but rather to assess the contribution of OH in this spectral range. The H_2 lines present in the spectral range have been fitted individually with Gaussians using the wavelengths provided by (35). The width is set by the instrumental resolution. The CH_3^+ model used here is Model III., at a temperature of 400K. All the models have been convolved to the spectral resolution of NIRSpec which is 2100 in this range, and re-sampled on the NIRSpec wavelength grid. The total model is the sum of the OH, H_2 and CH_3^+ models. It can be seen in Fig. 6 that there is a good agreement between the total model and the NIRSpec data. We note that the intensity of the detectable CH_3^+ signal is comparable with the 3σ noise of NIRSpec. Several strong lines of CH_3^+ , in particular at 3.157 or 3.223 μm do appear clearly in the data. However, at this stage, the noise level in the data is too high to provide a definitive identification of specific lines of CH_3^+ with NIRSpec. We are currently working on obtaining a higher SNR spectrum by improving the data reduction, and by performing additional observations.

Excitation of CH_3^+ infrared bands

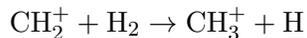
Hereafter we estimate the rates of the competing processes that might influence the excitation and line emission of CH_3^+ .

COLLISIONAL EXCITATION. A vibrationally excited state of the molecule can be excited by inelastic collisions with H or H_2 with an upward rate coefficient

$$q_{\ell u} = q_0 \exp(-hc\bar{\nu}/kT_k) g_u/g_\ell \text{ cm}^3 \text{ s}^{-1}$$

where T_k is the kinetic temperature of the gas, q_0 is the corresponding collisional quenching rate of the band $u \rightarrow \ell$, with degeneracies g_u and g_ℓ of the upper and lower states, respectively. The wavenumber of the transition is $\bar{\nu}$. In the disk wind, the H_2 pure rotational lines, which have a critical density lower than that of the gas density, are thermalized, hence $T_{\text{rot}}(\text{H}_2) = T_k \approx 900 \text{ K}$. The quenching rate coefficient might be as large as $10^{-11} \text{ cm}^3 \text{ s}^{-1}$; therefore, at the gas density of the reference PDR model shown in Fig. 7, $n_{\text{H}} = 3.5 \times 10^5 \text{ cm}^{-3}$, the collisional excitation rates of the CH_3^+ ν_4 and ν_3 fundamental bands would be at most $7 \times 10^{-7} \text{ s}^{-1}$ and $5 \times 10^{-8} \text{ s}^{-1}$ per ion, respectively. For a higher density of $n_{\text{H}} = 10^7 \text{ cm}^{-3}$, these rates would be 2×10^{-5} and $1 \times 10^{-6} \text{ s}^{-1}$ per ion, respectively. The spontaneous transition probability of the ν_4 fundamental band is calculated to be $A = 12 \text{ s}^{-1}$. This means that the vibrationally excited state of CH_3^+ cannot be thermally populated at the T_k : the collisional rates are many orders of magnitude too small, and thus $T_{\text{ex}} < T_k$.

FORMATION-PUMPING. As shown in previous sections, CH_3^+ is thought to form mainly through a sequence of exothermic ion-neutral reactions. The change of enthalpy in the reaction



is 65.6 kJ/mol , which corresponds to 5481 cm^{-1} . Energetically, this is sufficient to leave the product ion in the $\nu_4 = 1$ (1403 cm^{-1}) or $\nu_3 = 1$ (3108 cm^{-1}) vibrationally excited states; however, this is likely to happen in only a small fraction f_u of the reactions. In steady state, the total formation rate $F(\text{CH}_3^+)$ in

$\text{cm}^{-3} \text{ s}^{-1}$ is balanced by a total destruction rate $n(\text{CH}_3^+)D$. The principal destruction processes in the PDR include dissociative recombination with electrons, slow radiative association reactions with H_2 , reaction with atomic O to form HCO^+ , and photodissociation. Photodissociation of CH_3^+ was investigated through *ab initio* computations (55), but there has been no recent treatment of the process. As a result, photodissociation of CH_3^+ is absent from the Leiden database of photodissociation and photoionization of astrophysically relevant molecules. (55) identified the electronic states of CH_3^+ that would participate in its photodissociation and found only one $^1\text{E}''$ state with an energy below 13.6 eV with the right symmetry. They estimated that the oscillator strength for transitions to this state must be small, $f \sim 10^{-3}$, implying that the photodissociation rate in the background Galactic radiation field must be of the order of 10^{-11} s^{-1} . The reference PDR model in Fig. 7 has $G_0 = 4 \times 10^4$, so that that the corresponding photodissociation rate is $\sim 4 \times 10^{-7} \text{ s}^{-1}$. With reference to the zone of the PDR model where CH_3^+ is most abundant, $n_{\text{H}} = 3.5 \times 10^5 \text{ cm}^{-3}$, $T_{\text{k}} = 900 \text{ K}$, and the electron fraction $x_e \approx 10^{-4}$. Adopting the rate coefficients of (24) for dissociative recombination, we estimate that the destruction rate of CH_3^+ by electrons is $1.3 \times 10^{-5} \text{ s}^{-1}$, which completely dominates over reactions with H_2 and O. For higher densities - as discussed in the previous section, this destruction process will be even more efficient. For a given density, CH_3^+ is likely to be destroyed at least 100 times faster than it can be vibrationally excited by collisions at 900 K. This means that the excited vibrational states cannot be populated in equilibrium. It also implies that the excitation rate by the formation process might be of the order of $\sim 10^{-5} f_u \text{ s}^{-1}$ per ion. Thus if the ion-neutral source reaction yields $\sim 1\%$ of the product CH_3^+ ions in $v_4 = 1$, the formation process could be at least as important as collisional excitation.

RADIATIVE EXCITATION. The rate of radiative pumping by absorption of continuous radiation is given by

$$\frac{A_{ul}g_u/g_l}{\exp(hc\bar{\nu}/kT_b) - 1} \text{ s}^{-1}$$

where A_{ul} is the spontaneous transition probability and T_b is the brightness temperature of radiation at

frequency $\bar{\nu}$ defined by the Planck function

$$B_\nu(T_b) = \frac{2hc\bar{\nu}^3}{\exp(hc\bar{\nu}/kT_b) - 1} .$$

As shown in Fig 2, the surface brightness of the continuum in the ON-OFF difference spectrum is 500 MJy sr⁻¹ ($= 5 \times 10^{-18}$ W m⁻² Hz⁻¹ sr⁻¹) near the wavelength of the ν_4 fundamental band. This corresponds to $T_b = 105$ K. In the calculation of Ref. (47), this band has a transition dipole moment of $\mu = 0.165$ Debye and $g_u/g_\ell = 2$, so that the corresponding transition probability is $A = 12$ s⁻¹. If the continuum source is roughly co-extensive with the CH₃⁺-emitting region, then the molecules "feel" the same surface brightness that we see. Thus the radiative pumping rate in the ν_4 band itself is of the order of 5×10^{-8} s⁻¹ per ion. The total continuum brightness in the PDR spectrum (Fig. 2), is 10 times larger than that of the ON-OFF spectrum, which suggests that the molecules might be exposed to an even larger radiative excitation rate. The surface brightness in the NIRSpc spectrum of d203 – 506 is approximately 450 MJy sr⁻¹ at 3150 cm⁻¹ in the vicinity of the ν_3 band, corresponding to $T_b = 208$ K at this frequency. For a calculated transition probability $A = 309$ s⁻¹ (dipole transition moment 0.181 Debye from (47)), this implies a radiative excitation rate of 1.1×10^{-7} s⁻¹ per ion. Both estimates of radiative excitation omit the diluted starlight from the stars that energize the PDR.

COMPARISON. The rates of collisional excitation, hot-molecule formation, and infrared-radiative excitation of the observed CH₃⁺ bands can be comparable to order-of-magnitude in the case of a moderate density ($n_H = 3.5 \times 10^5$ cm⁻³). For higher densities ($n_H = 1 \times 10^7$ cm⁻³) collisions are likely to dominate the excitation. This example illustrates the possible non-LTE excitation mechanisms of the molecular vibrational bands that JWST may detect in related interstellar environments.

Column density estimates

Because the excitation processes that lead to the observed emission of CH₃⁺ remain to be determined (see previous section), it is difficult at this stage to provide an accurate determination of the total column density of the species. We however provide an order of magnitude estimate hereafter, and also point to

the main limiting factors in this derivation to guide future studies.

The integration over frequency of the emission band in the JWST observations in the 1200–1600 cm^{-1} range, after local baseline subtraction (spline fit on points outside the strong bands), is of $J_{\text{tot}} \sim 8 \times 10^{-6} \text{ W/m}^2/\text{sr}$. To get an estimate of the column density of CH_3^+ , we operate under the assumption that the ν_2 and ν_4 bands of this ion are the major contributors to the observed features (once the strong isolated atomic and H_2 lines are removed). Using the Einstein coefficients reported in Ref. (24) of $A(\nu_2) = 2.65 \text{ s}^{-1}$ and $A(\nu_4) = 4.43 \text{ s}^{-1}$, the column density in these upper emitting states can be derived by:

$$N_{\nu_2, \nu_4} \approx J_{\text{tot}} \frac{4\pi}{h\nu \sum_{\nu_2, \nu_4} A}$$

i.e., $N_{\nu_2, \nu_4} \sim 4.7 \times 10^{10} \text{ cm}^{-2}$. From PGOPHER models using dipole moments values 0.084 and 0.064 D from our best model (III.) for the vibrational modes ν_2 and ν_4 , respectively, we derive a total emission in the 7 μm band of $1.13 \times 10^{-19} \text{ W/CH}_3^+$. Thus, under such assumptions a column density of $N_{\nu_2, \nu_4} \sim 7.8 \times 10^{10} \text{ cm}^{-2}$ is estimated. These numbers must be taken with caution however, and are provided to the reader as an order of magnitude on the column densities for this species in this protoplanetary disk, as the line strength are not determined experimentally yet, resulting in possible change by large factors in the derivation. In addition, as the excitation is unlikely to be at LTE (see previous section), the determination of $N_{\text{tot}}(\text{CH}_3^+)$, the total column density of CH_3^+ , cannot be safely determined by using simple Boltzmann factors to relate ν_2 and ν_4 populations to the ground state one. Within LTE approximation, this factor is given by the ratio of states degeneracies multiplied by $\exp(-h(\nu_i \rightarrow 0)/kT)$. For the considered transitions, and excitation temperatures in the 800 – 300 K range, it implies total column densities a factor of ten to thousand times the above quoted N_{ν_2, ν_4} , respectively. At 400 K at LTE, $N_{\text{tot}}(\text{CH}_3^+) \sim 4 \times 10^{12} \text{ cm}^{-2}$. A model including lines spectroscopic assignment and collisional rates, when they will be determined, is to be developed in the future to better constrain the CH_3^+ excitation (see section above) and the total column densities, under the observed astronomical conditions. Finally, an additional uncertainty

in this derivation stems from the beam dilution, as the emission from the CH_3^+ emitting region is likely not spatially resolved with MIRI. Hence – put aside the spectroscopic uncertainties describe hereabove, the determined column density value should be considered as a lower limit. Overall, this discussion further demonstrates the absolute need for collaborative actions coordinating astrophysicists observations and spectroscopists dedicated laboratory astrophysics experiments to interpret such sets of data.

Extended Data

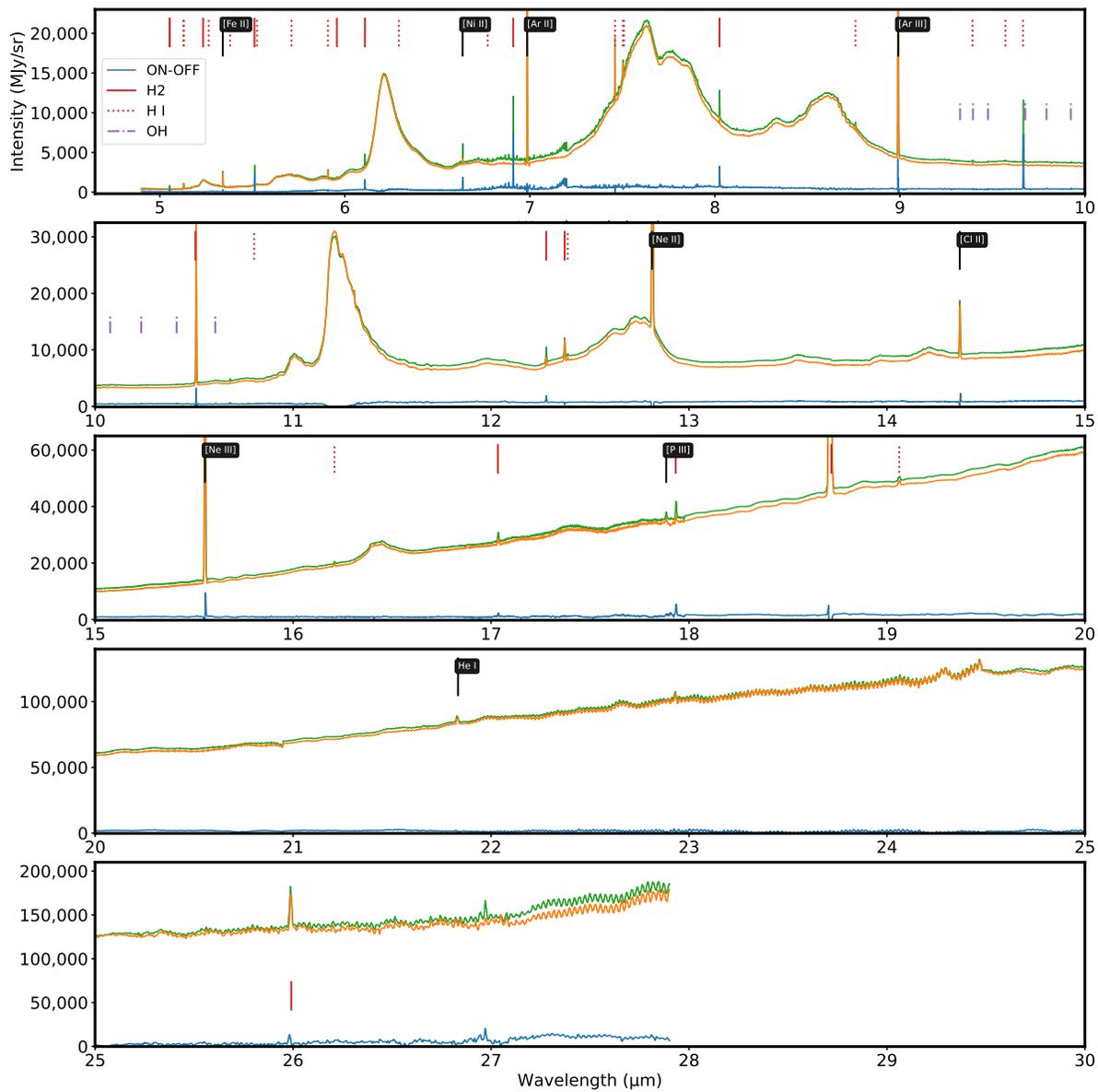


Figure 1: On and off spectra of d203-506 over the full MIRI-MRS spectral range. The On-Off spectrum is also shown. Main atomic and H₂ lines are indicated.

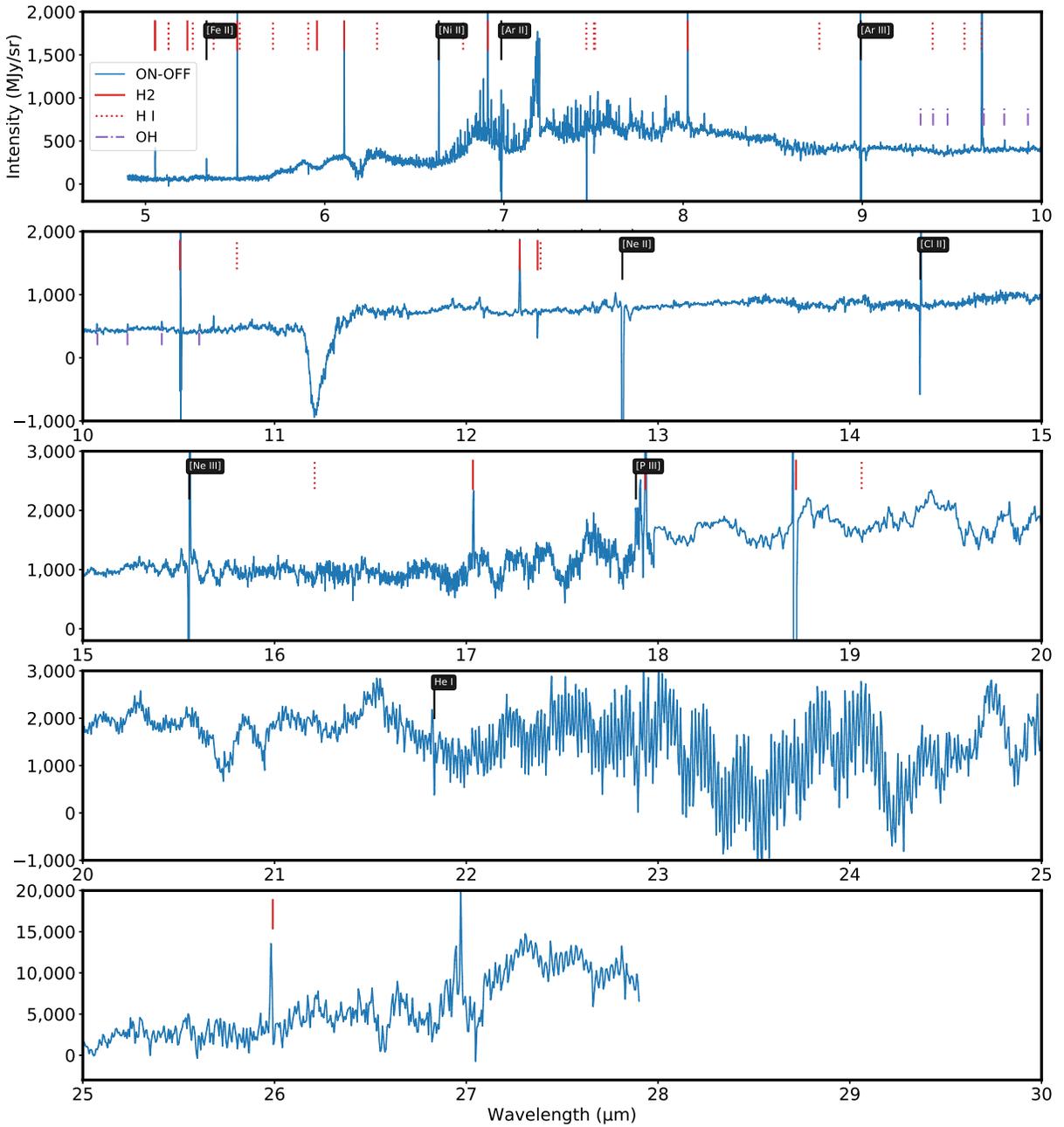


Figure 2: On-Off spectrum of d203-506 over the full MIRI-MRS spectral range. Main atomic and H₂ lines are indicated.

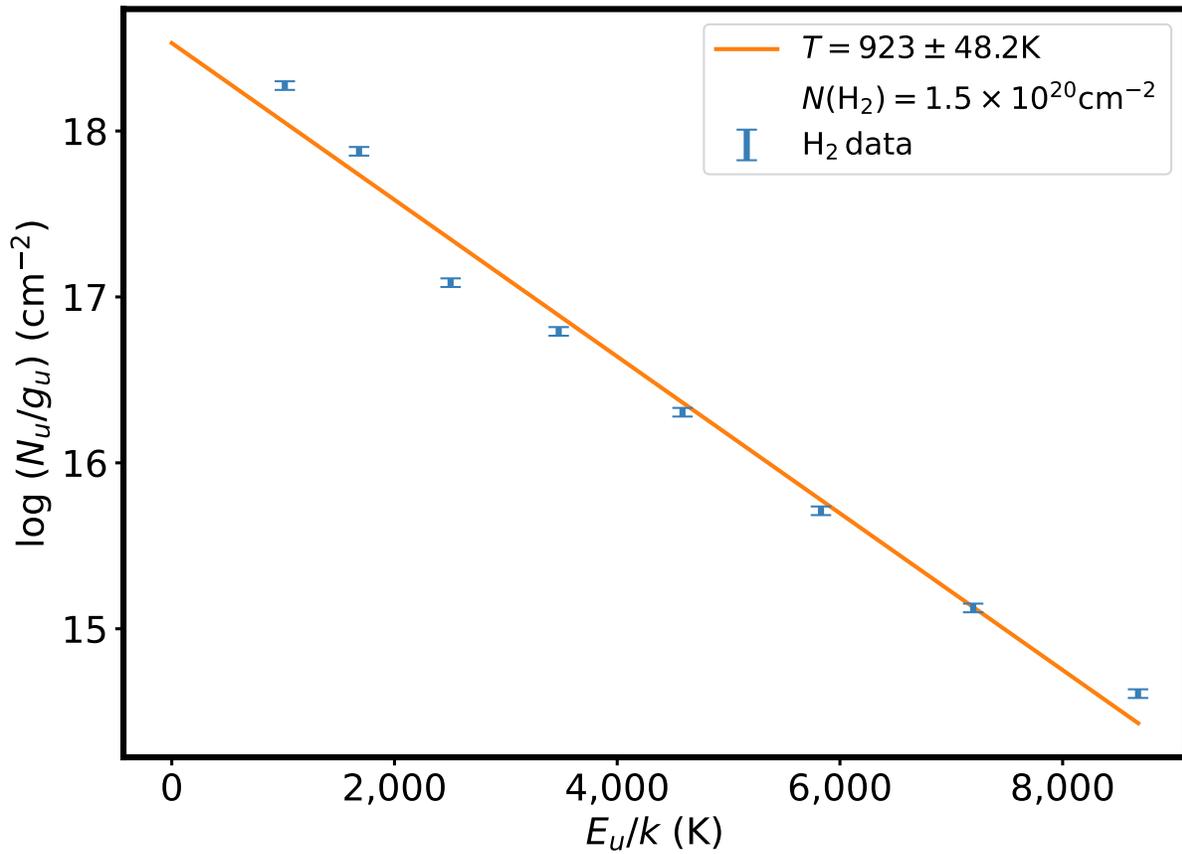


Figure 3: H_2 excitation diagram derived from the line intensities in Table 2 using the H_2 toolbox (39) developed within the PDRs4All team as one of their Science Enabling Products. Error bars result from the propagation of the absolute calibration error of MIRI, which we take from (60).

Table 1: H I detected emission lines

λ (μm)	E_{up}	A	Transition
(1)	(2)	(3)	(4)
5.128662	156225.1	$3.6881 \cdot 10^4$	6–10
5.263685	157316.1	$1.3121 \cdot 10^3$	7–18
5.379776	157257.1	$1.7812 \cdot 10^3$	7–17
5.525190	157186.7	$2.4709 \cdot 10^3$	7–16
5.711464	157101.8	$3.51558 \cdot 10^3$	7–15
5.908220	155854.9	$7.0652 \cdot 10^4$	6–9
6.291918	156869.4	$7.8457 \cdot 10^3$	7–13
6.771993	156707.3	$1.2503 \cdot 10^4$	7–12
7.45984	153419.7	$1.0254 \cdot 10^6$	5–6
7.502502	155337.5	$1.5609 \cdot 10^5$	6–8
7.508107	156498.9	$2.1174 \cdot 10^4$	7–11
8.760068	156225.1	$3.9049 \cdot 10^4$	7–10
9.392013	156869.4	$7.8037 \cdot 10^3$	8–13
10.803593	157186.7	$2.2679 \cdot 10^3$	9–16
12.387158	156498.9	$2.3007 \cdot 10^4$	8–11
16.20909	156225.1	$4.6762 \cdot 10^4$	8–10
19.06192	155337.5	$2.272 \cdot 10^5$	7–8

(1) Emission line wavelength (μm); (2) upper level energy (K); (3) Einstein A coefficient; (s^{-1})
(4) Transition label.

Table 2: Pure rotational H₂ detected emission lines in MIRI MRS wavelength range

λ (μm)	T_u	ν	A	Transition	Intensity ($\times 10^{-4}$)
(1)	(2)	(3)	(4)	(5)	(6)
5.0531	8677.1	1978.977	$3.236 \cdot 10^{-7}$	0-0 S(8)	0.864
5.5111	7196.7	1814.492	$2.001 \cdot 10^{-7}$	0-0 S(7)	4.367
6.1085	5829.8	1637.046	$1.142 \cdot 10^{-7}$	0-0 S(6)	2.582
6.9095	4586.1	1447.280	$5.879 \cdot 10^{-8}$	0-0 S(5)	12.217
8.0250	3474.5	1246.099	$2.643 \cdot 10^{-8}$	0-0 S(4)	4.193
9.6649	2503.7	1034.670	$9.836 \cdot 10^{-9}$	0-0 S(3)	6.468
12.278	1681.6	814.424	$2.755 \cdot 10^{-9}$	0-0 S(2)	2.408
17.034	1015.1	587.032	$4.761 \cdot 10^{-10}$	0-0 S(1)	1.742

(1) Emission line wavelength (μm), from (35); (2) Upper level energy (K); (3) Transition energy (cm^{-1}); (4) Einstein A coefficient (s^{-1}); (5) transition label; (6) $\text{erg s}^{-1} \text{cm}^{-2} \text{sr}^{-1}$

Table 3: Spectroscopic parameters of CH₃⁺.

	Experimental	Calculated ^{*,a}			Models [*]			
		This work ^b	Ref. (14)	Ref. (15)	I	II	III	IV
<i>v</i> = 0								
<i>B</i>	9.36214(28)	9.32	9.415	9.18	9.36214	9.36214	9.36214	9.36214
<i>C</i>	4.589949(35)	4.59	4.715	4.59	4.589949	4.589949	4.589949	4.589949
<i>D_J</i>	0.0007380(36)	0.00071	0.000719		0.0007380	0.0007380	0.0007380	0.0007380
<i>D_{JK}</i>	-0.0013144(79)	-0.00124	-0.001239		-0.0013144	-0.0013144	-0.0013144	-0.0013144
<i>D_K</i>	0.0004552(51)	0.00057	0.000568		0.0004552	0.0004552	0.0004552	0.0004552
<i>v</i> ₃ = 1								
<i>ν</i>	3108.3556(18)	2948						
<i>B</i>	9.27239(25)	9.21		9.00				
<i>C</i>	4.550184(29)	4.46		4.50				
<i>D_J</i>	0.0007029(30)							
<i>D_{JK}</i>	-0.0012814(71)							
<i>D_K</i>	0.0004547(47)							
<i>ζ</i>	0.110551(38)	0.115						
<i>η_J</i>	-0.0006660(80)							
<i>q</i> ⁺	0.00971(17)							
<i>v</i> ₂ = 1								
<i>ν</i>	1372 – 1412	1412	1391, 1433	1383, 1418	1392.80	1389.01	1391.22	1388.71
<i>B</i>		9.21 [9.27]	9.112 [9.06]	9.21 [9.49]	9.2270	9.3766	9.3721	9.3647
<i>C</i>		4.53 [4.61]	4.758 [4.63]	4.61 [4.66]	4.6392	4.6542	4.6560	4.6651
<i>D_J</i>			0.000715		0.002201	0.000798	.000703	.000703
<i>D_{JK}</i>			-0.001212		-0.005267	-0.00131	-.00118	-.00113
<i>D_K</i>			0.000547		0.002995	0.000488	.000455	.000455
<i>v</i> ₄ = 1								
<i>ν</i>	1373 – 1393	1331	1399	1385, 1429	1374.56	1374.46	1374.54	1396.35
<i>B</i>		9.44 [9.50]	9.574 [9.52]	9.20 [9.48]	9.50	9.50	9.5000	9.5027
<i>C</i>		4.46 [4.55]	4.743 [4.62]	4.60 [4.65]	4.5776	4.5802	4.5714	4.7534
<i>D_J</i>			0.000719		0.000747	0.000938	.000703	.000703
<i>D_{JK}</i>			-0.001240		-0.00116	-0.00146	-.00128	-.00082
<i>D_K</i>			0.000569		0.000326	0.00032	.000455	.000455
<i>ζ</i>		0.115	0.1136		0.11	0.11	0.11	0.11
<i>η_J</i>					-0.00063	-0.00063	-0.00063	-0.00063
<i>q</i> ⁺					0.0095	0.0095	0.0095	0.0095
<i><v</i> ₂ = 1 <i>v</i> ₄ = 1 > ^c								
<i>ζ</i> ₂₄		-0.66			-0.66	-0.66	-0.66	-0.66
<i><v</i> _{<i>i</i>} = 1 <i>d</i> _{<i>i</i>} <i>v</i> = 0 > ^d								
<i>d</i> ₂			0.084		0.084	0.084	0.084	0.084
<i>d</i> ₃			0.102					
<i>d</i> ₄			0.064		0.064	0.064	0.064	0.064

* Rotational, centrifugal distortion, and Coriolis constants in the ground state, *v*₃ = 1, *v*₂ = 1, and *v*₄ = 1 (in cm⁻¹), and transition moments of the *v*₂ and *v*₄ bands. Comparison of quantum calculated values, and those used to best model the spectrum of the d203-506 source around 1400 cm⁻¹. ^a Values in brackets (in blue) are scaled according to *v*₃ = 1 results when available, to *v* = 0 otherwise

^b ωB97X-D/cc-pVQZ, this work

^c Coriolis interaction parameter between *v*₂ = 1 and *v*₄ = 1, unitless

^d Transition dipole moment of the *ν*_{*i*} fundamental bands, in Debye

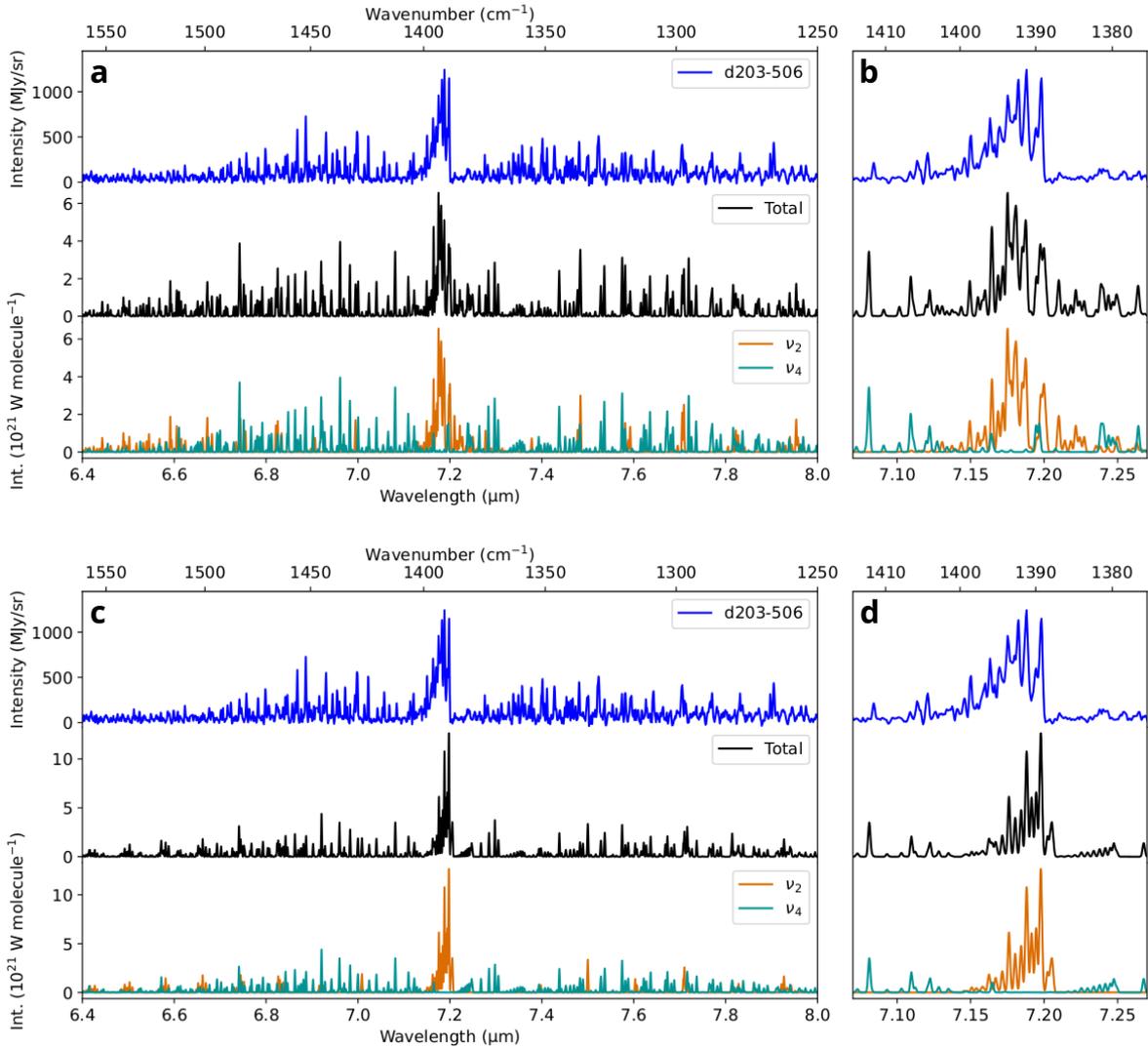


Figure 4: **a**, Model I, with zoom on the strongest lines (**b**). **c**, Model II, with zoom on strongest the lines (**d**). For these models, the excitation temperature is $T = 400$ K, and we use a Gaussian profile of 0.35 cm^{-1} full-width-at-half-maximum.

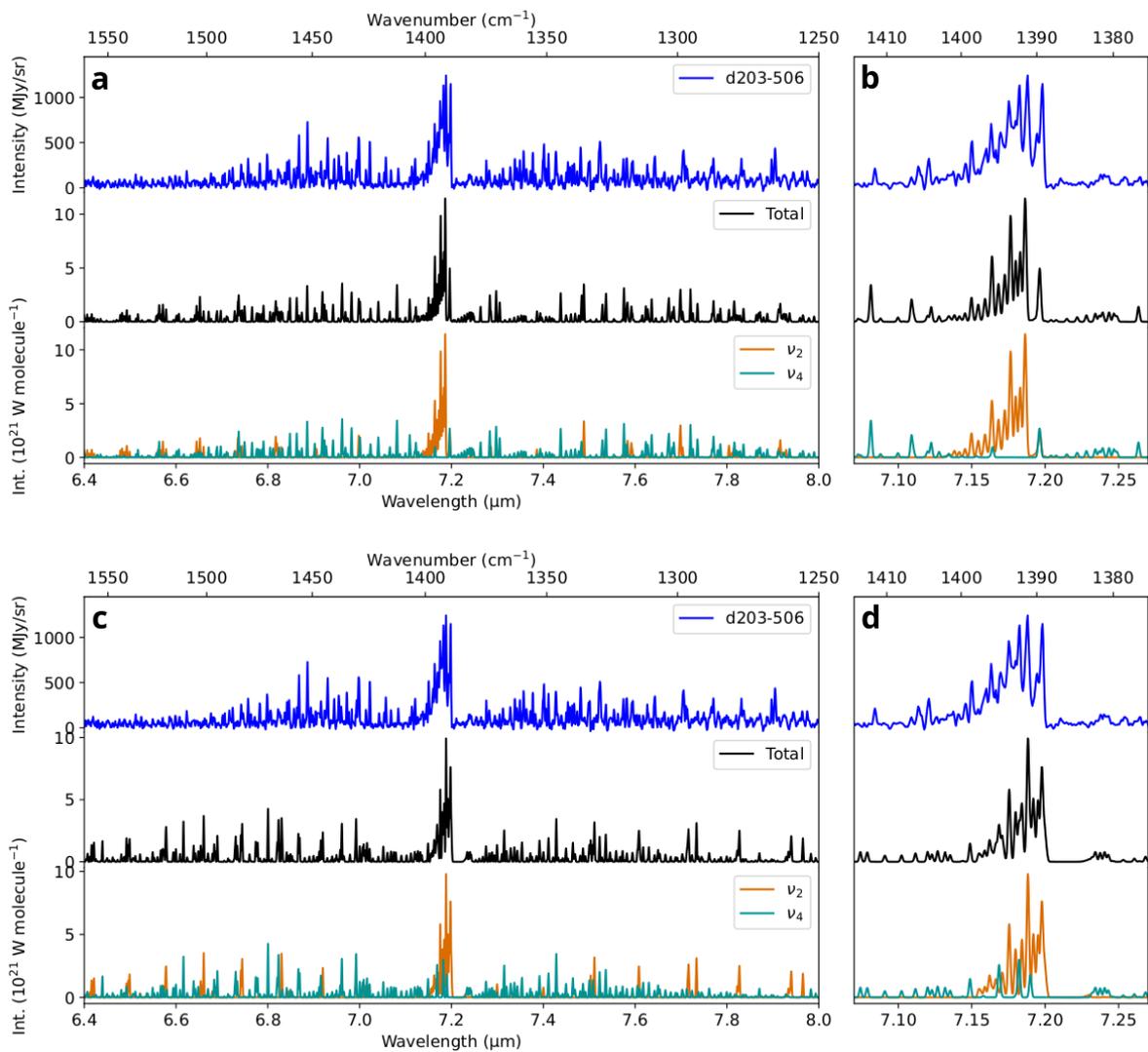


Figure 5: Model III. (a), and Model IV. (b) $T = 400$ K, Gaussian profile (0.35 cm^{-1} full-width-at-half-maximum)

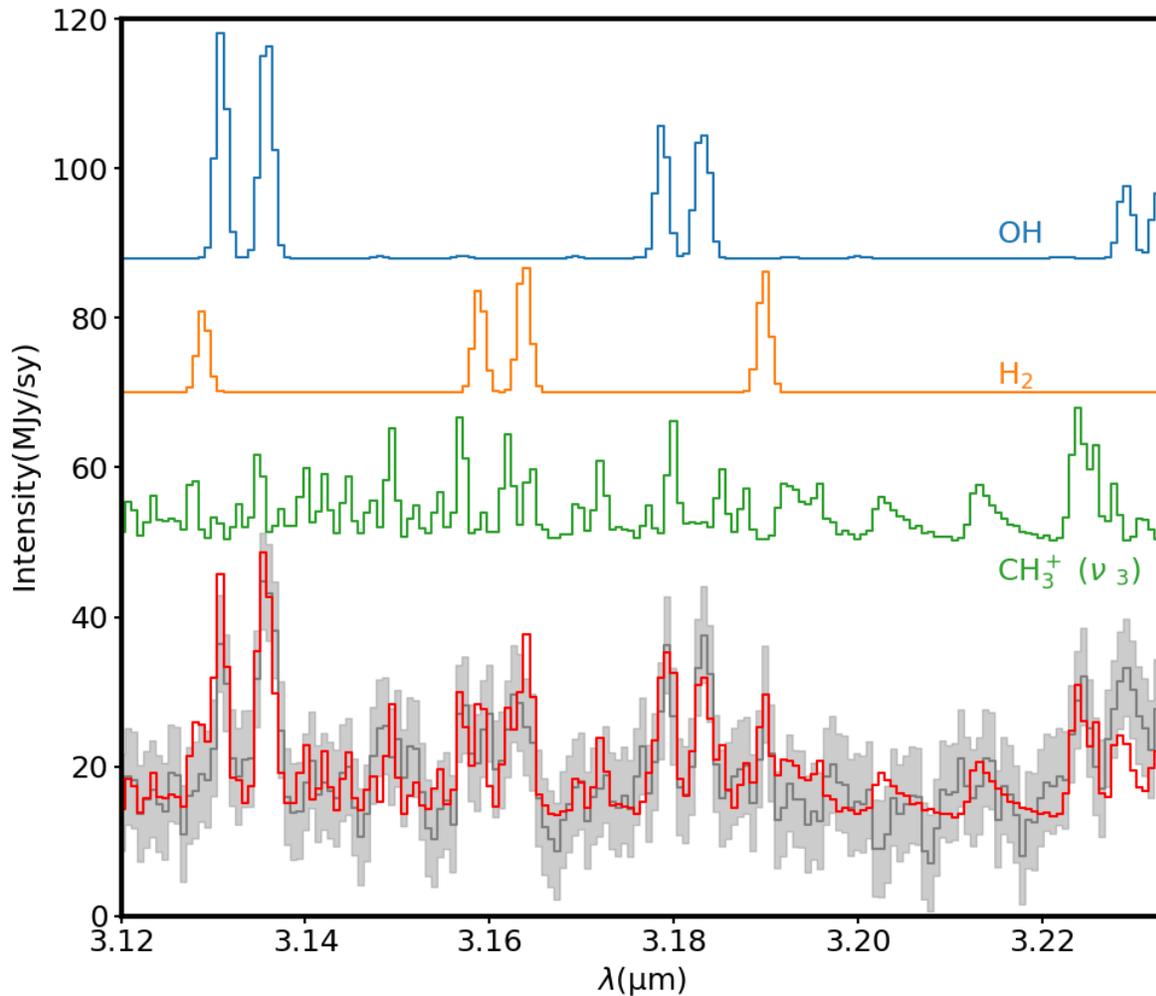


Figure 6: NIRSpec spectrum of d203-506. The spectrum is shown in gray, the shaded regions is the ± 3 sigma error interval of the data. This includes the error provided by the JWST pipeline, and error ν_3 band of CH₃⁺ in the NIRSpec spectrum of d203-506. Model of the OH emission (blue), H₂ emission (orange), CH₃⁺ emission (green), and sum of these three (red). Beyond 3.22 μm , emission due to the wings of the Aromatic Infrared Band at 3.3 μm is seen, affecting the baseline of the NIRSpec spectrum. The OH spectrum is computed with an LTE model at a temperature of 800K. A detailed model of the OH emission will be presented in a forthcoming paper (Zannese et al. in prep). The H₂ lines are fitted individually. The CH₃⁺ model used here is Model III., at a temperature of 400K.

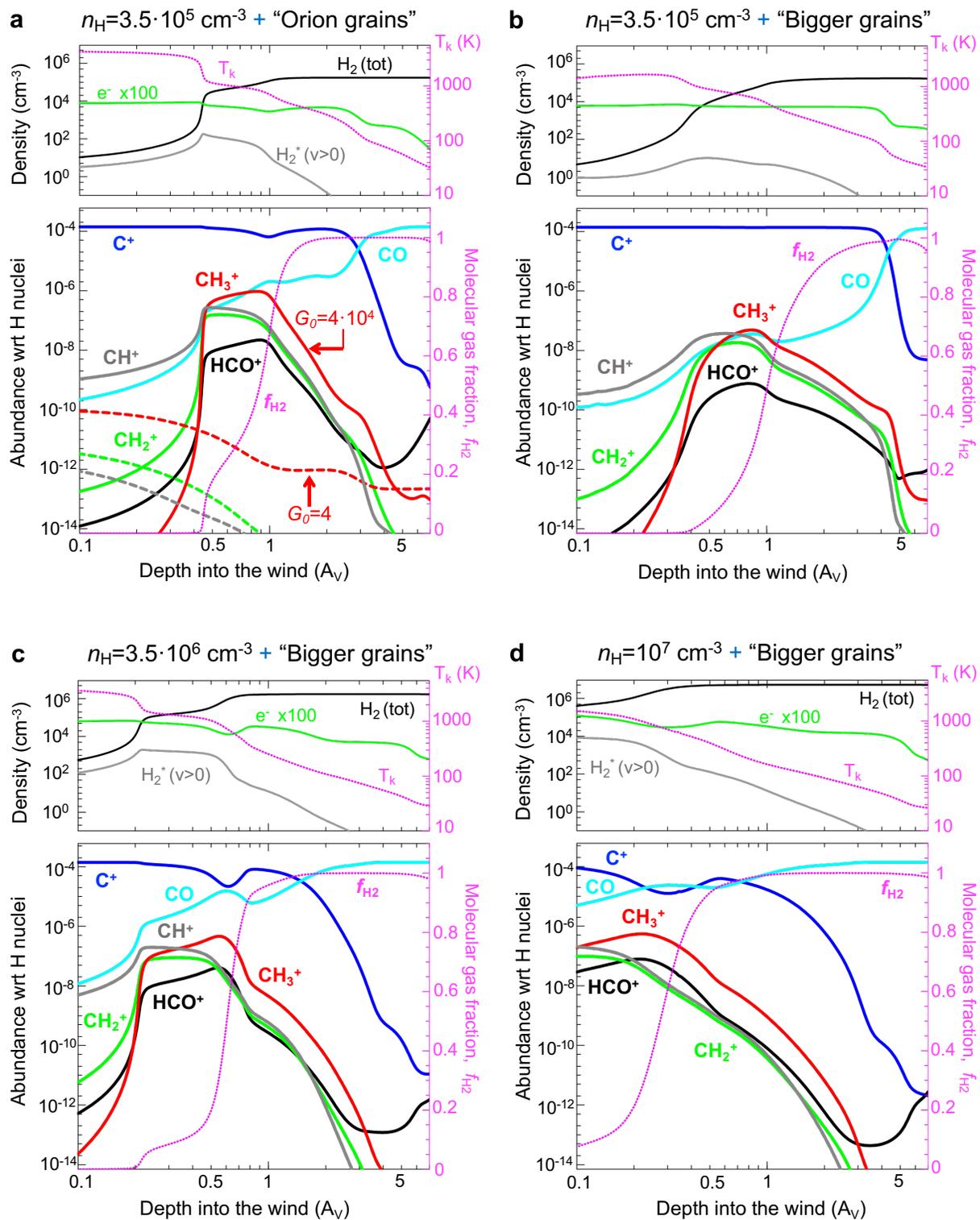


Figure 7: Photochemical model results for d203-506 adopting $G_0 = 4 \times 10^4$ and different gas densities (n_{H}) and dust grain properties. *Upper panels:* Density and gas temperature structure as a function of visual extinction (A_V) from the wind surface. The gray curve shows the density of vibrationally excited $\text{H}_2^+(v > 0)$. *Lower panels:* Abundance profiles with respect to H nuclei. The pink dotted curves show the molecular fraction f_{H_2} profile. Dashed curves in model a) refer to a model with the same gas density but G_0 lower by a factor 10^4 .

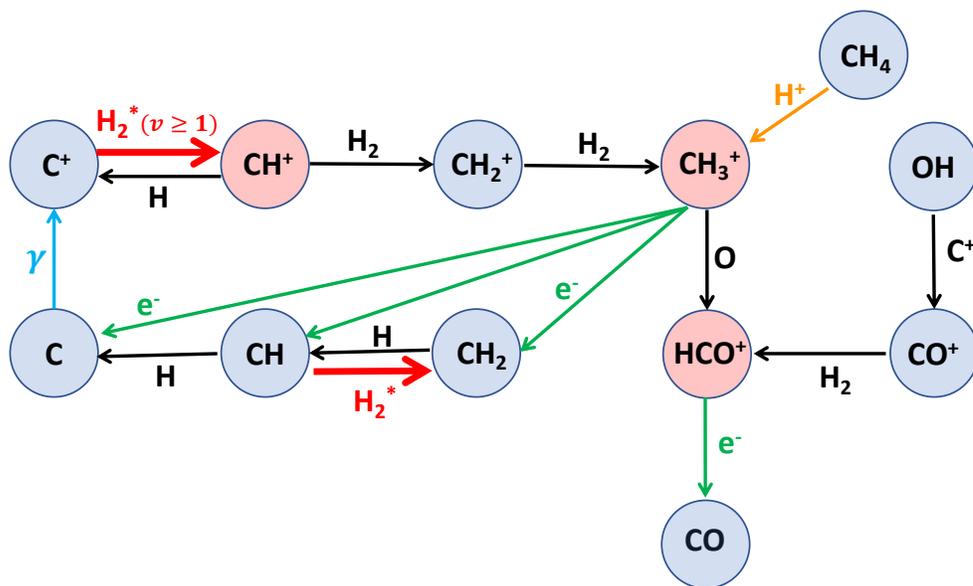


Figure 8: Dominant CH_3^+ formation and destruction reactions at the CH_3^+ abundance peak predicted by the photochemical model shown in Fig. 7. This reaction network also leads to abundant HCO^+ in FUV-irradiated gas layers where $x(\text{C}^+) > x(\text{CO})$. Red arrows show endoergic reactions when H_2 is in the ground-vibrational state $v=0$. These reactions become fast only in disk layers where the gas temperature is high (several hundred K) and/or significant vibrationally excited $\text{H}_2^*(v \geq 0)$ exists. The formation of CH_3^+ from methane will only be relevant if very high CH_4 and H^+ abundances coexist in the gas.