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# **Visible-light-mediated Divergent and Regioselective Vicinal Difunctionalization of Styrenes with Arylazo Sulfones**

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**ABSTRACT:** Activated by visible light, arylazo sulfones can serve as multifaceted reactants and employed in diazenylation, sulfonylation, and arylation reactions under (photo)catalyst free conditions. Such versatile reactivity enabled us to develop an operationally simple, regioselective and tunable difunctionalization of styrenes with arylazo sulfones to produce α-sulfonyl arylhydrazones and 1,2-alkoxyarylated products in moderate to excellent yields. Furthermore, such difunctionalized products have been exploited as key building blocks for the synthesis of various heterocycles.

Radical-mediated vicinal difunctionalizations of alkenes in which two units are simultaneously installed across a double bond in a single operation are powerful approaches to rapidly produce complex compounds with high step economies.<sup>1</sup> In this field, visible-light catalysis<sup>1-3</sup> has recently attracted much attention for developing facile intermolecular difunctionalizations of readily accessible alkenes,<sup>4</sup> and two main strategies have been successfully applied. The first one involves the addition of a radical to an alkene and a subsequent capture of the intermediate by a suitable reactant.<sup>5</sup> The second one relies on the use of bifunctional reagents, which, under photocatalytic conditions, result in the formation of two radicals that are concomitantly incorporated into the alkenes.<sup>6</sup> Although various efficient difunctionalizations have been reported based on these complementary methods, the majority needs exogenous photocatalysts which are often costly and difficult to recycle.<sup>5-6</sup> Therefore, the development of more sustainable approaches is still highly necessary.

Recently, a photocatalyst free strategy built on dyedauxiliary group7 able to mediate radical has attracted several organic chemists, notably focusing on promising photoactive arylazo sulfones.<sup>7c, 8</sup> As illustrated in Scheme 1 (eq 1), the photoinduced sequential cleavage of N-S and C-N bonds in these compounds can form three different radicals, namely diazenyl (I),<sup>9</sup> sulfonyl  $(III)$ <sup>10</sup> and aryl  $(III)$ <sup>11</sup> radicals, which can each react with partners. Produces in various visible-light-driven process, these species allowed successful alkene monofunctionalizations. For example, we developed an efficient arylation procedure of 1,1-

## **Scheme 1. Examples of Functionalizations of Alkenes with Arylazo Sulfones.**



diaryl ethylenes<sup>12a</sup> and allyl sulfones<sup>12b</sup> with arylazo sulfones yielding triaryl ethylenes and allyl arenes, respectively (eq 2). Furthermore, the diazenylation of enol silyl ether derivatives <sup>13</sup> and ketene silyl acetals (eq  $3$ )<sup>14</sup> was also achieved through direct

trapping of aryl diazenyl radicals **I** generated from arylazo sulfones. In 2020, Chawla and co-workers used arylazo sulfones as the precursors of sulfonyl radicals and reported the sulfonylation of cinnamic acids yielding (*E*)-vinyl sulfones in good yields.15 Thereafter, the same group carried out the monofunctionalization of nitrostyrenes with arylazo sulfones. The reaction proceeds through the addition of aryl or sulfonyl radicals to the double bonds followed by a denitration step, and the structure of the resulting products depends on the reaction conditions (eq 4).16 Despite these achievements, the adoption of reaction conditions that enable the use of these substrates in the 1,2-difunctionalization of alkenes has been scarcely reported. Wei and co-workers  $17$  reported the only example of 1,2 difunctionnalization, achieving an oxysulfonylation, from arylazo sulfones. In view of these premises, we envisioned that the divergent reactivities of arylazo sulfones could be applied to the vicinal difunctionalization of alkenes by carefully adjusting the reaction conditions. Herein, we report a visiblelight-mediated divergent and regioselective synthesis of  $\alpha$ sulfonyl arylhydrazones and 1,2-alkoxyarylated products from readily available styrenes (eq. 5,6).



 $SO<sub>2</sub>Me$ 

$N^{\geq N}$ $\ddot{}$ $\dot{C}N$ 1a	2a	hv (Blue Light) Solvent, 18h	$N$ <sup>NHAr</sup> 3a SO <sub>2</sub> Me	$N$ <sup>MAr</sup> 4a SO <sub>2</sub> Me
entry	1a (mmol)	$2a \pmod{2}$	solvent	3a (yield $\%$ ) <sup>b</sup>
1 <sup>c</sup>	0.1	0.1	<b>DCM</b>	73
$\overline{c}$	0.2	0.2	<b>DCM</b>	90
3 <sup>d</sup>	0.2	0.2	<b>DCM</b>	90
$\overline{4}$	0.2	0.2	MeCN	90
5	0.4	0.4	<b>DCM</b>	90 <sup>e</sup>
6 <sup>f</sup>	0.4	0.2	<b>DCM</b>	96
7 <sup>g</sup>	0.4	0.4	<b>DCM</b>	

*a* Reaction conditions: **1a** and **2a** in 2.0ml solvent, blue LED ( $\lambda_{\text{max}} = 456$  nm, 40W, 100% Power) 10°C, 18 h; degassed using three freeze−evacuate−thaw cycles. *<sup>b</sup>* Yields refer to chromatographically pure product. *c*Blue LED ( $\lambda_{\text{max}}$  = 456 nm, 40W, 25% Power), argon steamdegassed. *<sup>d</sup>* Blue LED (λmax = 450 nm, 18W, 100% Power). *<sup>e</sup>* **4a** Isolated in 5% yield.  $f$  24 h.  $g$  In the dark.

We initially studied the functionalization of arylazo sulfone **1a** and 2-vinylnaphthalene (**2a**) upon light irradiation as a model reaction and the results are depicted in Table 1. In the first attempt, a product was isolated in 73% yield in degassed DCM (entry 1). The structure of product was identified by NMR studies as the  $\alpha$ -sulfonyl arylhydrazone **3a** with (*Z*)configuration. Such results encouraged us to further investigate this regioselective difunctionalization of styrenes by tuning conditions (Table 1).18,19 Increasing the scale of **1a** and **2a** led to a better yield independently of light source used (entries 2 and 3). A solvent screening pointed out that DCM and MeCN provided identical results (entry 4), and then DCM was selected for further studies. Performing the reaction at a higher concentration (0.4 mmol of **1a**), the desired product **3a** precipitated in less than 16 h and was isolated by simple filtration (entry 5). Further analysis of the filtrate revealed the formation of another product (5% yield) corresponding to the

α-sulfonyl aryl diazene **4a**. 20 We surmised that compound **4a** could be an intermediate to compound **3a**. Indeed, when the irradiation was performed for 24 h by reducing the ratio of **2a**, hydrazone **3a** was the sole product formed in 96% yield (entry 6). To explore the scope of this difunctionalization, various arylazo sulfones **1** were studied under the optimized reaction conditions (Scheme 2). We found that the efficiency of the process was sensitive to the electronic properties of the aromatic substituents in **1**. Thus, different arylazo sulfones bearing electron-withdrawing substituents (such as  $CN$ ,  $CO<sub>2</sub>Et$ ,  $NO<sub>2</sub>$ , CF3 or COCH3) in *orho-, para-* or *meta* position reacted efficiently with 4-vinylanisole **2b** to form the corresponding αsulfonyl arylhydrazones (**3b-h, 3j,k**). Disubstition was also tolerated (**3l-n**).

**Scheme 2. Scope of the Difunctionalization of Styrenes with Arylazo Sulfones.** 



Reaction conditions: Table 1, entry 6, <sup>3</sup>36 h; <sup>b</sup> Reaction carried out on 1 mmol of 1.<sup>c</sup> Natural sunlight was used as light source (77 h).

Surprisingly, in some cases, only the corresponding  $\alpha$ sulfonyl aryldiazenes **4** were isolated under the optimized conditions (e.g. **4i**), but a prolonged irradiation (36 h) resulted in the almost quantitative phototautomerization of **4** to the

desired arylhydrazones **3**. The nature of the sulfone group was found to have neglecting effect on the reaction since an alkyl or aryl substituted azosulfones were synthesized in high yields (**3o-3s**). Finally, the scope of styrenes **2** was screened, pointing out that various substituents at the ortho-, meta- and para positions were compatible to afford corresponding products **3w-3ai** in excellent yields. Nearly quantitative yield of **3x** was obtained after 77 h of exposition of the reaction mixture to natural sunlight. 2-Vinylpyridine was also transformed to the corresponding product **3aj**, albeit in low yield. The reaction was found ineffective with α-substituted styrenes, leading to a complex mixture of products under the presently optimised condition. We considered that formation of a stable benzyl radical, generated *via* a β-addition on α-methyl styrenes, could hamper the process. However, hypothesizing that such radical could be oxidized, to deliver a cation, we reasoned that it could be exploited in a coupling reaction with a nucleophilic counterpart, resulting in a new difunctionalization strategy.





Reaction conditions: **1** (0.60 mmol) and **2** (1.20 mmol) and ROH (2.40 mmol) in 3 mL of argon saturated MeCN at rt for 20 h, blue LED ( $\lambda_{\text{max}}$ = 450 nm, 18W, 100% Power) irradiation. Isolated yield**. <sup>a</sup>** Starting from 1.2 mmol of **1a** in 6.0 mL of MeCN**.**

With this in mind, we focused on  $\alpha$ -methylstyrene 2*j* as the model substrate. During the screening process, we found that, in MeOH, a different outcome led to 1,2-methoxyarylated product **5a**21 as the single regioisomer in 32% yield (See SI). After surveying the reaction conditions by varying the solvent, and stoichiometry, we found that the reaction performed in MeCN in the presence of 4 eq of MeOH afforded the product **5a** in 63% yield (Scheme 3, see SI for details). With the optimized conditions in our hands, we subsequently studied the scope of the three-component alkene difunctionalization (Scheme 3). An array of arylazo sulfones efficiently reacted with α-methyl styrenes and smoothly delivered the corresponding methoxyarylated products **5b-5i** in moderate to good yields. Both electron-donating groups (such as OMe, Me) and electron-withdrawing groups (such as CN, F, Br, NO2) on the arylazo sulfones were compatible. The reaction of α-methyl styrenes bearing electron-rich as well as electron-poor substituents in *ortho*, *meta*, and *para* positions of the aryl ring proceeded well to afford the corresponding products **5g-5m** in good yields. α-Ethyl styrene reacted to yield **5n**, albeit with a s lower yield. Notably, the three-component reaction was successfully applied to 1,1-diarylsubstituted ethylene leading to the desired methoxyarylated product **5o**, instead of the triarylethylene generated in dry acetonitrile.<sup>12a</sup> Various alcohols such as EtOH, *n*BuOH, and *i*PrOH as well as water participated smoothly in the reaction to afford the products (**5p-5s)** in moderate to good yields. In addition, trifluoroethanol decently performed as a nucleophile and allowed to easily introduce the highly electronegative and lipophilic trifluoroethoxy group in product **5t**. The three component difunctionalization can be performed with deuterated methanol and propargyl alcohol (**5u** and **5v**). Gratefully, the chiral (*R*)-1,2-*O*isopropylideneglycerol can also be used, providing the desired product **5w** as a mixture of diastereomers (1:1 dr).

Control experiments were conducted to obtain mechanistic insights (see SI, section 1.4 for details), including on–off experiments (see Figures S9 and S16) and irradiation in the presence of the radical scavenger  $(2,2,6,6-$ Tetramethylpiperidin-1-yl)oxyl (TEMPO), where the efficiency of both reactions was inhibited and where LC-MS analyses revealed the presence of a TEMPO adduct resulting from the trapping of an aryl radical. On the basis of the obtained results, and on available literature,<sup>7c,8,12-16</sup> we suggested two competing radical pathways (Scheme 4), that are mainly controlled by the nature of the styrene employed.

**Scheme 4. Plausible Reaction Mechanism** 



First, irradiation with visible light ( $\lambda = 456$  nm) promotes the arylazo sulfones into  $\frac{1}{n}\pi^*$ -state 1<sup>\*</sup> (step a) which undergoes a homolytic cleavage of the S–N bond to generate the diazenyl radical **I** and sulfonyl radical **II** (step b). In absence of alcohol and in fully deoxygenated conditions (pathway A), **II** is in turn trapped by non-α-substituted styrene **2** to form the benzylic radical species **IV** (step c) which thus adds itself onto the N=N bond of arylazo sulfone **1** and release a new sulfonyl radical **II** (by following a chain propagation pathway) furnishing αsulfonyl aryldiazene **4** (step d)**,** that in turn undergoes (photo)tautomerization to form  $3$  (step e).<sup>22</sup> On the other hand in (pathway B), in presence of alcohol the diazenyl radical **I**  rapidly leads to aryl radical **III** which selectively reacts with αsubstituted styrene **2** (step f).23 Then, the tertiary radical **V** could be oxidized into carbocation **VI** by sulfonyl radical **II** (step g) or by residual traces of  $O_2$  (step g'). Final trapping of the

cationic intermediate by nucleophilic alcohol (step h), affords the corresponding alkyloxyarylated compound **5**.

To demonstrate the synthetic potentials of our protocols, the 1,2-sulfonyl-arylhydrazination and 1,2-alkyloxy arylation of styrenes were scaled up (1 mmol). The results are comparable (99% for **3n** and 63% for **5c**, Scheme 2 and 3), if not better (65% and 99% for **3w** and **3x**, respectively, Scheme 2). Thereafter, a series of transformations was carried out on these difunctionalized products (Scheme 5). For instance, the reaction of **3x** with the Vilsmeier–Haack reagent readily afforded the valuable pyrazole 7 in decent yield.<sup>24a</sup> In addition, the arylhydrazone was also converted to benzopyrazole **8** by PIFA-mediated oxidative cyclization of **3x** in HFIP.24b Finally, the Cu-mediated click reaction of **5v** with benzyl azide **9** gave the corresponding triazole **10** in quantitative yield.25

### **Scheme 5. Synthetic Transformations of 3 and 5.**



In conclusion, we have developed two sets of conditions for the visible-light mediated difunctionalization of styrenes with readily available arylazo sulfones to afford either α-sulfonyl arylhydrazones or 1,2-alkoxyarylated products, depending on the styrenes used. This protocol offers novel syntheses of α-sulfonyl arylhydrazones or 1,2-alkoxyarylated products which are versatile building blocks for the synthesis of various heterocycles.

# **ASSOCIATED CONTENT**

#### **Data Availability Statements**

The data underlying this study are available in the published article and its Supporting Information

### **Supporting Information**

The Supporting Information is available free of charge on the ACS Publications website. Experimental procedures and characterization data. (PDF)

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#### **Notes**

The authors declare no competing financial interest.

#### **Author Contributions**

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## **REFERENCES**

(1) (a) Engl, S.; Reiser, O. Copper-photocatalyzed ATRA reactions: concepts, applications, and opportunities. *Chem. Soc. Rev.* **2022**, *51*, 5287. (b) Wang, Y.; Bao, Y.; Tang, M.; Ye, Z.; Yuan, Z.; Zhu*,* G. Recent advances in difunctionalization of alkenes using pyridinium salts as radical precursors. *Chem. Commun*. **2022**, *58*, 3847. (c) Coppola, G. A.; Pillitteri, S.; Van der Eycken, E. V.; You, S.-L.; Sharma, U. K. Multicomponent reactions and photo/electrochemistry join forces: atom economy meets energy efficiency. *Chem. Soc. Rev*. **2022**, *51*, 2313. (d) Zhao, B.; Yin, X.; Li, H.; Cheng, K.; Wan, J.-P. Transition metal-free, photocatalytic arylation and dioxygenation for vicinal diketone synthesis using alkynes and arene diazonium salts. *Org. Chem. Front*. **2023**, *10,*1942.

(2) (a) Kärkäs, M. D.; Matsuura, B. S.; Stephenson, C. R. J. Enchained by visible light–mediated photoredox catalysis. *Science* **2015**, *349*, 1285. (b) Skubi, K. L.; Blum, T. R.; Yoon, T. P. Dual Catalysis Strategies in Photochemical Synthesis. *Chem. Rev.* **2016**, *116*, 10035. (c) Matsui, J. K.; Lang, S. B.; Heitz, D. R.; Molander, G. A. Photoredox-Mediated Routes to Radicals: The Value of Catalytic Radical Generation in Synthetic Methods Development. *ACS Catal.* **2017**, *7,* 2563. (d) Crisenza, G. E. M.; Melchiorre, P. Chemistry glows green with photoredox catalysis. *Nat. Commun.* **2020**, *11*, 803.

(3) (a) Strieth-Kalthoff, F.; James, M, J.; Teders, M.; Pitzer, L.; Glorius, F. Energy transfer catalysis mediated by visible light: principles, applications, directions. *Chem. Soc. Rev*. **2018**, *47*, 7190. (b) Zhou, Q.- Q.; Zou, Y.-Q.; Lu, L.-Q.; Xiao, W.-J. Visible-Light-Induced Organic Photochemical Reactions through Energy-Transfer Pathways. *Angew. Chem. Int. Ed.* **2019**, *58*, 1586.

(4) (a) Koike; T.; Akita, M. Fine Design of Photoredox Systems for Catalytic Fluoromethylation of Carbon–Carbon Multiple Bonds. *Acc. Chem. Res.* **2016**, *49*, 1937. (b) Courant, T.; Masson, G. Recent Progress in Visible-Light Photoredox-Catalyzed Intermolecular 1,2-Difunctionalization of Double Bonds via an ATRA-Type Mechanism. *J. Org. Chem.* **2016**, *81*, 6945. (c) Coppola, G. A.; Pillitteri, S.; Van der Eycken, E. V.; You, S.-L.; Sharma, U. K. Multicomponent reactions and photo/electrochemistry join forces: atom economy meets energy efficiency. *Chem. Soc. Rev*. **2022**, *51*, 2313.

(5) (a) Garbarino, S.; Ravelli, D.; Protti, S.; Basso, A. Photoinduced Multicomponent Reactions. *Angew. Chem. Int. Ed.* **2016**, *55*, 15476. (b) Pitzer, L.; Schwarz, J. L.; Glorius, F. Reductive radical-polar crossover: traditional electrophiles in modern radical reactions. *Chem. Sci.* **2019**,10, 8285. (c) Jiang, H.; Studer, A. Intermolecular radical carboamination of alkenes. *Chem. Soc. Rev*. **2020**, *49*, 1790. Examples from our lab: (d) Jarrige, L.; Carboni, A; Dagousset, G.; Levitre, G.; Magnier, E.; Masson, G. Photoredox-Catalyzed Three-Component Tandem Process: An Assembly of Complex Trifluoromethylated Phthalans and Isoindolines. *Org. Lett.* **2016**, *18*, 2906. (e) Levitre, G.; Dagousset, G.; Anselmi, E.; Tuccio, B.; Magnier, E.; Masson, G. Four-Component Photoredox-Mediated Azidoalkoxy-trifluoromethylation of Alkenes. *Org. Lett.* **2019**, *21*, 6005.

(6) (a) Wang, H.; Bellotti, P.; Zhang, X.; Paulisch, T. O.; Glorius, F. A base-controlled switch of SO<sub>2</sub> reincorporation in photocatalyzed radical difunctionalization of alkenes. *Chem* **2021**, *7*, 3412. (b) Huang, H.- M.; Bellotti, P.; Ma, J.; Dalton, T.; Glorius, F. Bifunctional reagents in organic synthesis. *Nat. Rev. Chem.* **2021**, *5*, 301. Recent selected examples: (c) Tan, G.; Das, K.; Keum, H.; Bellotti, P.; Daniliuc, C.; Glorius, F. Photochemical single-step synthesis of β-amino acid derivatives from alkenes and (hetero)arenes. *Nat. Chem.* **2022**, *14*, 1174. (d) Zheng, Y.; Liao, Z.; Xie, Z.; Chen, H.; Chen, K.; Xiang, H.; Yang, H. Photochemical Alkene Trifluoromethylimination Enabled by Trifluoromethylsulfonylamide as a Bifunctional Reagent. *Org. Lett.* **2023**, *25*, 2129.

(7) (a) Saraiva, M. F.; Couri, M. R. C.; Le Hyaric, M.; de Almeida, M. V. The Barton ester free-radical reaction: a brief review of applications. *Tetrahedron* **2009**, *65*, 3563. (b) Goti, G. B.; Bieszczad, B.; Vega-Peñaloza, A.; Melchiorre, P. Stereocontrolled Synthesis of 1,4- Dicarbonyl Compounds by Photochemical Organocatalytic Acyl Radical Addition to Enals. *Angew. Chem., Int. Ed.* **2019**, *58*, 1213. (c) Qiu, D.; Lian, C.; Mao, J.; Fagnoni, M.; Protti, S. Dyedauxiliary Groups, an Emerging Approach in Organic Chemistry. The Case of Arylazo Sulfones. *J. Org. Chem.* **2020**, *85*, 12813.

(8) (a) Crespi, S.; Protti, S.; Fagnoni, M. Wavelength Selective Generation of Aryl Radicals and Aryl Cations for Metal-Free Photoarylations. *J. Org. Chem.* **2016**, *81*, 9612. (b) Protti, S.; Ravelli, D.; Fagnoni, M. Wavelength dependence and wavelength selectivity in photochemical reactions. *Photochem. Photobiol. Sci.* **2019**, *18*, 2094.

(9) (a) Porter, N. A.; Green, J. G.  $^{15}N$  study of the phenyldiazenyl radical: A search for phenyl migration. *Tetrahedron Lett.* **1975**, *16*, 2667. (b) Porter, N. A.; Dubay, G. R. Diazenyl radicals: a nitrogen-15 CIDNP and radical trapping study of unsymmetric diazenes. *J. Am. Chem. Soc.* **1978**, *100*, 920.

(10) Reisenauer, H. P.; Schreiner, P. R.; Romanski, J.; Mloston, G. Gas-Phase Generation and Matrix Isolation of the MethylsulfonylRadical CH3SO2 • from Allylmethylsulfone, *J. Phys. Chem. A,* **2015**, *119*, 2211.

(11) (a) Hari, D. P.; König, B. The Photocatalyzed Meerwein Arylation: Classic Reaction of Aryl Diazonium Salts in a New Light. *Angew. Chem. Int. Ed.* **2013**, *52*, 4734. (b) Mo, F.; Qiu, D.; Zhang, L.; Wang, J. Recent Development of Aryl Diazonium Chemistry for the Derivatization of Aromatic Compounds. *Chem. Rev.* **2021**, *121*, 5741.

(12) (a) Onuigbo, L.; Raviola, C.; Di Fonzo, A.; Protti, S.; Fagnoni, M. Sunlight-Driven Synthesis of Triarylethylenes (TAEs) via Metal-Free Mizoroki–Heck-Type Coupling. *Eur. J. Org. Chem.* **2018**, 5297. (b) Dossena, A.; Sampaolesi, S.; Palmieri, A.; Protti, S.; M. Fagnoni Visible Light Promoted Metal- and Photocatalyst-Free Synthesis of Allylarenes. *J. Org. Chem.* **2017**, *82*, 10687.

(13) Abdulla, H. O.; Scaringi, S.; Amin, A. A.; Mella, M.; Protti, S.; Fagnoni, M. Aryldiazenyl Radicals from Arylazo Sulfones: Visible Light-Driven Diazenylation of Enol Silyl Ethers. *Adv. Synth. Catal*. **2019**, *362*, 2150.

(14) Di Terlizzi, L.; Cola, I.; Raviola, C.; Fagnoni, M.; Protti, S. Dyedauxiliary Group Strategy for the α-Functionalization of Ketones and Esters. *ACS Org. Inorg. Au* **2021**, *1*, 68.

(15) Chawla, R.; Jaiswal, S.; Dutta, P. K.; Yadav, L. D. S. Photocatalyst-free visible light driven synthesis of (*E*)-vinyl sulfones from cinnamic acids and arylazo sulfones. *Tetrahedron Lett***. 2020**, *61*, 151898.

(16) Chawla, R.; Jaiswal, S; Dutta, P. K.; Yadav, L. D. S. A photocatalyst-free visible-light-mediated solvent-switchable route to stilbenes/vinyl sulfones from β-nitrostyrenes and arylazo sulfones. *Org. Biomol. Chem*. **2021**, *19*, 6487.

(17) (a) Liu, Q.; Liu, F.; Yue, H.; Zhao, X.; Xiaohui; L.; Li, J.; Wei, W. Photocatalyst-Free Visible Light-Induced Synthesis of β-Oxo Sulfones via Oxysulfonylation of Alkenes with Arylazo Sulfones and Dioxygen in Air. *Adv. Synth. Catal*. **2019**, *361*, 5277. (b) Lv, Y.; Liu, Q.; Liu, F.; Yue, H.; Li, J.-S.; Wei, W. Visible-light-promoted aerobic oxidative synthesis of β-ketosulfones under photocatalyst-free conditions. *Tetrahedron Lett*. **2020**, *61*,151335.

(18) (a) Wang, L.; Cheng, P.; Wang, X. H.; Wang, W.; Zeng, J. G.; Liang, Y.; Reiser, O. Visible-light promoted sulfonamidation of enol acetates to α-amino ketones based on redox-neutral photocatalysis. *Org. Chem. Front*. **2019**, *6*, 3771. (b) Ni, B.; Zhang, B.; Han, J.; Peng, B.; Shan, Y.; Niu, T. Heterogeneous Carbon Nitrides Photocatalysis Multicomponent Hydrosulfonylation of Alkynes To Access β-Keto Sulfones with the Insertion of Sulfur Dioxide in Aerobic Aqueous Medium. *Org. Lett.* **2020**, 22, 670. (c) Jia, X.; Luo, L.; Huang, C.; Zhang, X.; Lian, Z. Iron-Catalyzed Sulfonylthiocyanation of α,β-Unsaturated Amides/Esters via the Insertion of Sulfur Dioxide. *Org. Lett*. **2022**, *24,* 7560.

(19) (a) Atlan, L.; Kaim, V. L. E.; Supiot, C. The use of hydrazones for efficient mannich type coupling with aldehydes and secondary amines. *Chem. Commun*. **2000,** 1385. (b) Liu, C.; Lv, J.; Luo, S.; Cheng, J.-P. Sc(OTf)<sub>3</sub>-Catalyzed Transfer Diazenylation of 1,3-Dicarbonyls with Triazenes via N–N Bond Cleavage. *Org. Lett.* **2014**, *16*, 5458.

(20) (a) Léonard, E.; Mangin, F.; Villette, C.; Billamboz, M.; Len, C. Azobenzenes and catalysis. *Catal. Sci. Technol.* **2016**, *6*, 379. Synthesis of azo compounds via a difunctionalization of alkenes: (b) Shen, J.; Xu, J.; Zhu, Q.; Zhang, P. Hypervalent iodine(III)-promoted rapid cascade reaction for the synthesis of unsymmetric azo compounds. *Org. Biomol. Chem*. **2021**, *19*, 3119. (c) Shen, J.; Xu, J.; He, L.; Ouyang, Y.; Huang, L.; Li, W.; Zhu, Q.; Zhang, P. Photoinduced Rapid Multicomponent Cascade Reaction of Aryldiazonium Salts with Unactivated Alkenes and TMSN3. *Org. Lett.* **2021**, *23*, 1204.

(21) (a) Hartmann, M.; Li, Y.; Studer, A. Transition-Metal-Free Oxyarylation of Alkenes with Aryl Diazonium Salts and TEMPONa. *J. Am. Chem. Soc.* **2012**, *134*, 16516. (b) Fumagalli, G.; Boyd, S.; Greaney, M. F. Oxyarylation and Aminoarylation of Styrenes Using Photoredox Catalysis. *Org. Lett.* **2013**, *15*, 4398. (c) Govindarajan; R.; Ahmed, J.; Swain, A. K.; Mandal S. K. Transition-Metal-Free Catalytic Carboalkoxylation of Styrenes at Room Temperature. *J. Org. Chem.* **2019**, *84*, 13490.

(22) (a) Simeth, N. A.; Crespi, S.; Fagnoni, M.; König, B. Tuning the Thermal Isomerization of Phenylazoindole Photoswitches from Days to Nanoseconds. *J. Am. Chem. Soc.* **2018**, *140*, 2940. (b) Bandara, H. M. D.; Burdette, S.C. Photoisomerization in different classes of azobenzene, *Chem. Soc. Rev*. **2012**, *41*, 1809-1825. (c) Coyle, J. D. The Photochemistry of Olefinic Compounds, *Chem. Soc. Rev*. **1976**, 329- 353.

(23) Orlandi, M.; Escudero-Casao, M.; Licini, G. Nucleophilicity Prediction via Multivariate Linear Regression Analysis. *J. Org. Chem*. **2021**, *86,* 3555.

(24) (a) Padmavathi, V.; Mahesh, K.; Thriveni, P.; Reddy, T. V. R. Phenacylsulfonylacetic acid methyl ester‐synthon for 1,2,3‐selena/thiadiazoles and 2H‐diazaphospholes. *J. Heterocycl. Chem*. **2007**, *44*, 1165. (b) Ivonin, S. P.; Kurpil', B. B.; Grygorenko, O. O.; Volochnyuk, D. M. Reaction of hydrazones derived from active methylene compounds with Vilsmeier–Haack reagent. *Monatsh. Chem.* **2014**, *145*, 2011.

(25) Devaraj, N. K.; Finn, M. G. Introduction: Click Chemistry. *Chem. Rev.* **2021**, 121, 6697