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# Synthesis of Heteroarylated Pyridines via a Double C–H Bond Functionalization using Palladium-catalyzed 1,4-Migration Combined with Direct Arylation

Mohamed Niyaz Vellala Syed Ali,<sup>[a]</sup> Linhao Liu,<sup>[a]</sup> Monika Ravi,<sup>[a]</sup> and Henri Doucet<sup>\*[a]</sup>

The C–H bond functionalization of pyridyl C3-position of 4arylpyridines using Pd 1,4-migration has been investigated. Under suitable reaction conditions, oxidative addition of 4-(2bromoaryl)pyridines to palladium followed by a Pd 1,4-migration activates these pyridyl C3-positions. Next, the Pd-catalyzed coupling *via* C–H bond activation of the heteroarenes affords the C3-heteroarylated 4-arylpyridines. The new C–C bond created in this coupling reaction results from the functionaliza-

#### Introduction

Pyridine derivatives are ubiquitous in pharmaceutical compounds and organic materials.<sup>[1]</sup> Consequently, the discovery of new effective routes to these compounds, enabling the functionalization of specific positions, is a very important aspect of current research involving pyridine derivatives.<sup>[2]</sup> Metalcatalyzed C–H bond activation/functionalization of (hetero)arenes currently represents one of the simplest method for the preparation of bi(hetero)aryls.<sup>[3–8]</sup> However, to our knowledge, the synthesis of 4-aryl-3-heteroarylpyridines *via* activation of the C–H bond at the C3-position of pyridines has not been reported to date.

We recently reported that during the Pd-catalyzed direct heteroarylation of 2-(2-bromoaryl)thiophenes, partial Pd 1,4-migration<sup>[9,10]</sup> occurred at the thienyl  $\beta$ -positions,<sup>[11]</sup> leading to a mixture of 1,2-diheteroarylated benzenes and also to  $\beta$ -heteroarylated 2-arylthiophene derivatives such as 2'-aryl-2,3'-bithiophenes. This Pd 1,4-migration methodology is an appealing method for functionalizing C–H bonds in proximity to 2-bromoaryl units. We have therefore tried to extend this methodology to (2-haloaryl)pyridines. To date, little is known about Pd-catalyzed 1,4-migration of these substrates. Gallagher et al. reported that the Heck reaction of 4-(2-bromoaryl)pyridines with an acrylate gave a mixture of the classical Heck type product and also the Pd 1,4-migration

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tion of two C–H bonds. For this coupling reaction, a bromosubstituent on the aryl unit of 4-arylpyridines was employed as a traceless directing group. This heteroarylation method tolerates a range of substituents on the benzene ring as well as several heteroarenes. Moreover, from 3-arylpyridines, C2- or C4functionalization of the pyridine ring also proved possible. In addition, this procedure uses an air-stable catalyst and an inexpensive base.

vinylation product in a ratio of 2:1 to 3:1 (Scheme 1, a).<sup>[12]</sup> Activation of the C–H bond in the C3 position of the pyridyl ring of 4-arylpyridines using a phosphine or phenol as directing groups has also recently been described (Scheme 1, b and c).<sup>[13]</sup> However, in general the current preparation of 4-aryl-3-heteroarylpyridine derivatives still requires a multi-step syn-



e) This work: C-H bond activation using bromide as a traceless directing group



Scheme 1. Pd-catalyzed functionalization of the C3-position of the pyridyl unit of 4-arylpyridines.



thesis. In many cases, this involves the preparation of 4-halo-3heteroarylpyridines, which is rather difficult, as the last step in their synthesis generally consists of Suzuki coupling between bromopyridines and heteroaryl boronic acids (Scheme 1, d).<sup>[14]</sup> Consequently, the outcome of Pd-catalyzed direct arylation reaction of 4-(2-bromoaryl)pyridines needed to be explored. Our aim was to find conditions favoring Pd 1,4-migration associated with direct arylation for access to 4-aryl-3-heteroarylpyridines via a double C-H bond functionalization. Here, we report i) the conditions for Pd-catalyzed regioselective direct heteroarylations of the pyridyl unit of 4-(2-bromoaryl)pyridines using bromide substituent as a traceless directing group,<sup>[15]</sup> ii) and on the scope the Pd 1,4-migration associated with direct arylation for the formation of 4-aryl-3-heteroarylpyridines via a double C-H bond functionalization, iii) on the reactivity of 3-(2bromoaryl)pyridines (Scheme 1, e).

### **Results and Discussion**

First, we prepared a set of 3- or 4-(2-bromoaryl)pyridines *via* Suzuki coupling of 3- and 4-pyridineboronic acids with 1-bromo-2-iodobenzenes (Scheme 2).<sup>[13]</sup> The desired arylated pyridines 1a-1p were obtained in good yields in all cases. 5-(2-Bromophenyl)pyrimidine 1q was prepared from pyrimidin-5-ylboronic acid and 1-bromo-2-iodobenzene by the same method.

Next, the selectivity of the Pd-catalyzed reaction of 4-(2bromophenyl)pyridine **1a** with 2-isopropyl-4-methylthiazole was investigated (Table 1). A high selectivity of 87% for the desired Pd-catalyzed 1,4-migration product **2a** was obtained using 2 mol% Pd(OAc)<sub>2</sub> catalyst combined with KOAc base in DMA at 150 °C; while the product **2b** resulting from the heteroarylation of the aryl unit was obtained with a selectivity of 13%. (Table 1, entry 1). However, with this phosphine-free





Table 1. Influence of the reaction conditions on the Pd-catalyzed coupling of 4-(2-bromophenyl)pyridine 1 a with 2-isopropyl-4-methylthiazole. <sup>[a]</sup>						
H + H + S + H + S + I = 150 °C, 16 h						
Entry	Catalyst	Solvent	Base	Conv. (%)	Ratio 2a:2b	Yield in <b>2 a</b> (%)
1	Pd(OAc) <sub>2</sub>	DMA	KOAc	94	87:13	60
2	PdCI(C <sub>3</sub> H₅)(dppb)	DMA	KOAc	100	87:13	61
3	PdCI(C₃H₅)(dppb)	DMA	Cs <sub>2</sub> CO <sub>3</sub>	45	-	-
4	PdCI(C <sub>3</sub> H₅)(dppb)	DMA	K <sub>2</sub> CO <sub>3</sub>	15	-	-
5	PdCI(C <sub>3</sub> H₅)(dppb)	DMA	NaOAc	100	88:12	58
6	PdCI(C₃H₅)(dppb)	DMA	CsOAc	56	78:22	-
7	PdCI(C <sub>3</sub> H <sub>5</sub> )(dppb)	DMA	KOPiv	100	89:11	63
8	PdCI(C <sub>3</sub> H <sub>5</sub> )(dppb)	DMF	KOAc	100	89:11	60
9	PdCI(C <sub>3</sub> H <sub>5</sub> )(dppb)	NMP	KOAc	100	85:15	57
10	PdCI(C <sub>3</sub> H <sub>5</sub> )(dppb)	xylene	KOAc	0	-	-
11	PdCI(C <sub>3</sub> H <sub>5</sub> )(dppb)	xylene	CsOAc	0	-	-
12	PdCI(C <sub>3</sub> H <sub>5</sub> )(dppb)	xylene	KOPiv	0	-	-
13	PdCI(C <sub>3</sub> H <sub>5</sub> )(dppb)	DEC	KOAc	0	-	-
14	PdCI(C <sub>3</sub> H <sub>5</sub> )(dppb)	TBAA <sup>[b]</sup>	KOAc	100	88:12	61
15	PdCl(C₃H₅)(dppb)	DMA	КОАс	33	91:9	_[c]

[a] [Pd] (0.02 equiv.), 4-(2-bromophenyl)pyridine **1a** (1 equiv.), 2-isopropyl-4-methylthiazole (2 equiv.), base (2 equiv.), 150 °C, 16 h, **2a:2b** ratios determined by <sup>1</sup>H NMR and GC/MS analysis of the crude mixtures, isolated yields. [b] TBAA: tetrabutylammonium acetate (2 mL). [c] 130 °C.

PdCI(C<sub>3</sub>H<sub>5</sub>)(dppb) 2 mol% KOPiv (2 equiv.), DMA 150 °C, 16 h 2 equiv C Ratio 4a:4b 89:11 Ratio 3a:3b 92:8 Ratio 5a:5b 87:13 3a 72% 4a 78%. 64% 5a 54% EtC Ratio 6a:6b 75:25 Ratio 7a:7b 79:21 Ratio 8a:8b 81:19 6a 60% 7a 64% 8a 62% nBu NC Ratio 9a:9b 100:0 Ratio 10a:10b 100:0 Ratio 11a:11b 95:5 Ratio 12a:12b 100:0 10a 62% **9a** 68% 11a 66% 12a 91% Me Ratio 15a:15b 77:23 Ratio 13a:13b 100:0 Ratio 14a:14b 100:0 Ratio 16a:16b 100:0 **13a** 78% 14a 76% 15a 50% 16a 48%

Ratio 17a:17b 80:20 Ratio 18a:87b 100:0 Ratio 19a:19b 100:0 17a 62% 18a 80% \*: Pd(OAc)<sub>2</sub> 2 mol% as catalyst Scheme 3. C3-heteroarylations of the pyridyl ring of 4-(2bromophenyl)pyridine 1 a via Pd 1,4-migration combined with direct

arvlation.

arylpyridines 1c and 1e substituted with 4-fluoro and 4trifluoromethyl, were also well tolerated, with the selective formation of Pd 1,4-migration products 22 a and 24 a in similar yields. Conversely, a 4-trifluoromethoxy substituent (1f) gave a mixture of products 25a and 25b in 80:20 ratio, and an ester substituent (1g) gave no coupling products 26a or 26b but decomposition of 4-arylpyridine 1 g was observed. The low selectivity for product 25 a is possibly due to the coordination of the OCF<sub>3</sub> group to palladium. Methyl or fluoro substituents at meta-position of the 4-arylpyridine (1h and 1i) gave products 27 a and 28 a with complete selectivity and yield of 71% and 74%, respectively. An ortho-fluoro substituent was also well tolerated and furnished product 29a in similar yield. The selectivity of the reaction with a 4-(2-bromophenyl)pyridine bearing a 2-fluoro substituent on the pyridine ring 1k was also

catalyst, conversion of 1 a was not complete, therefore we used  $PdCl(C_3H_5)(dppb)^{[16]}$  catalyst which gave a similar 2a:2b selectivity of 87:13 and a 2a yield of 61% with full conversion of 1a (Table 1, entry 2). The use of carbonate bases Cs<sub>2</sub>CO<sub>3</sub> and K<sub>2</sub>CO<sub>3</sub> proved ineffective, and the formation of several unidentified products was observed (Table 1, entries 3 and 4). In contrast, NaOAc and CsOAc led to good selectivities in 2a and KOPiv gave 2a in 89% selectivity and 63% yield (Table 1, entries 5-7). We also determined the influence of solvent, but DMF and NMP led to slightly lower yields in 2a; while xylene and DEC were ineffective (Table 1, entries 8–13).<sup>[17]</sup> Using tetrabutylammonium acetate (TBAA) as solvent, 2a was obtained with a good selectivity of 88% and a yield of 61% (Table 1, entry 14).<sup>[18]</sup> This solvent could therefore replace DMA, which is undesirable from the point of view of sustainable chemistry. A lower reaction temperature of 130°C instead of 150°C provided 2a:2b with similar selectivity (91:9) but low yield due to partial conversion of **1a** (Table 1, entry 15).

Then, the scope of the coupling reaction of 4-(2bromophenyl)pyridine 1a with a set of heteroarenes was studied (Scheme 3). From 2-isopropylthiazole and 1a, we obtained an a:b ratio of 92:8 with the formation of the Pd 1,4migration product 3a in 72% yield. In the presence of 2methylthiophene and 2-chlorothiophene, very high a isomers selectivities were also observed, and products 4a and 5a were isolated in 78% and 54% yield, respectively. No cleavage of the C-Cl bond of the thienyl ring was observed, allowing further transformations. Slightly lower a isomers selectivities were achieved using 2-acetylthiophene, ethyl thiophene-2-carboxylate and cyclopropyl-2-thienylketone affording products 6a-8a in yields of 60-64%. In contrast, in the presence of 2-methyl-2-(thiophen-2-yl)-1,3-dioxolane and 2-(thiophen-2-yl)acetonitrile, only the 9a and 10a isomers were detected. 3-Methylbenzothiophene gave 11a with 95% selectivity and 66% yield. Furan derivatives were also suitable substrates for this reaction. 2-Butylfuran and 1-furan-2-ylbutan-1-one afforded products 12a and 13a with 100% selectivity and in good yields. The natural product menthofuran was also used successfully, affording product 14a with complete selectivity and 76% yield. Conversely, quite sluggish reactions were observed with a 2acetylpyrrole and 1,2-dimethylimidazole, with the formation of 15 a and 16 a in only 50% and 48% yield, respectively due to the formation of some unidentified side-products. Imidazo[1,2*a*]pyrazine, imidazo[1,2-c]pyrimidine and imidazo[1,2b]pyridazine afforded products 17a-19a in 62-80% yields.

observed We had previously that with 2-(2bromoaryl)thiophenes, Pd 1,4-migration was favored by the presence of electron-donating substituents on their aryl moiety, whereas electron-withdrawing substituents led to mixtures of products due to partial Pd 1,4-migration.[11] Therefore, the selectivity of the reaction using a set of 4-(2bromophenyl)pyridine bearing a para-substituent on the aryl unit was investigated (Scheme 4). As expected, the presence of a methyl para-substituent (1 b) using 2-methylthiophene and 2butylfuran as reaction partners had practically no influence, affording the products 20a and 21a with 100% selectivity and 78% and 76% yield, respectively. The electron-deficient 419a 76%



Scheme 4. C3-heteroarylations of the pyridyl ring of 4-(2-bromophenyl)pyridine derivatives 1 b-1 k via Pd 1,4-migration combined with direct arylation.

studied. The heteroarylation was regioselective in favor of the C5-position of the pyridyl ring to give **30 a**.

To determine the influence of the position of the nitrogen atom on the pyridine ring on the reactivity and selectivity of the Pd 1,4-migration/direct arylation reaction, the result of the reaction with 3-(2-bromophenyl)pyridine **1j** was examined (Scheme 5). Using 2-methylthiophene as the reaction partner, no formation of coupling products **31a**-**c** was detected by GC/ MS analysis with  $Pd(OAc)_2$  (2 mol%) or  $PdCl(C_3H_5)(dppb)$ 



Scheme 5. Reactivity of 3-(2-bromophenyl)pyridine 11 with 2-methylthiophene. (5 mol%) catalysts, and 3-(2-bromophenyl)pyridine 11 was recovered unreacted. A similar result was observed with 2-butylfuran as the heteroaryl source. This lack of reactivity could be explained by the formation of a stable Pd complex by reaction with 3-(2-bromophenyl)pyridine 11 or 31a, as 2-(thiophen-2-yl)pyridines can give stable palladacycles by C–H activation.<sup>[19]</sup> With 5-(2-bromophenyl)pyrimidine 1q in the presence of 2-methylthiophene, an identical result was observed. The starting material 1q was recovered and no coupling product was detected, which is consistent with the formation of a stable palladacycle.

To confirm the poisoning of the Pd catalyst by 3-(2bromophenyl)pyridine 11, we performed a reaction using an equimolar mixture of 3- and 4-(2-bromophenyl)pyridines 1a and 11 with 2-methylthiophene as the heteroaryl coupling partner (Scheme 6). No coupling product formation was detected and arylpyridines 1a and 11 were recovered.

The influence of substituents on the adjacent positions of the nitrogen atom of 3-(2-bromophenyl)pyridines was then examined, as their electronic or blocking properties may result in a different reactivity to 3-(2-bromophenyl)pyridine **11** (Scheme 7). 5-(2-Bromophenyl)-2-methoxypyridine **1m** gave an inseparable mixture of products in very low yields, and a large amount of **1m** was recovered. Conversely, from 5-(2-bromo-



Scheme 6. Reactivity of an equimolar mixture of 3- and 4-(2bromophenyl)pyridines 1a and 1l with 2-methylthiophene.



Scheme 7. Reactivity of substituted 3-(2-bromophenyl)pyridines 1 m-1 p.

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phenyl)-2-fluoropyridine 1n, we were pleased to observe that regioselective arylation at the C6 position on the pyridine ring was observed to give product 33a in 41% yield. Both 3-(2bromophenyl)-2-methoxypyridine 1o and 3-(2-bromophenyl)-2fluoropyridine 1p also reacted nicely to give C4-heteroarylated products 34a and 35a in good yields. With these two substrates, the presence of a substituent at the C2-position on the pyridine ring may prevent catalyst poisoning by the formation of palladacycles with 2-(thiophen-2-yl)pyridine derivatives.

The selectivity and yield of the reaction of **1a** with (5methylthiophen-2-yl)boronic acid instead of 2-methylthiophene using the Pd(OAc)<sub>2</sub>/dppf catalyst and KOPiv base were also determined (Scheme 8). Selective formation of the product **4a** from 1,4-migration reaction with Suzuki coupling<sup>[20]</sup> was also observed, albeit in very low yield (< 10%). Consequently, the use of the Pd 1,4-migration reaction in combination with direct arylation seems more attractive than Pd 1,4-migration reaction in combination with Suzuki type coupling for the preparation of such compounds.

A possible mechanism for access to 3-heteroarylated pyridines **a** is described in Scheme 9. The first step is the oxidative addition of aryl bromide to palladium to form intermediate **A**. Next, ligand exchange with KOPiv and Pd-1,4-migration provides intermediate **B**. After a concerted metallation deprotonation  $(CMD)^{[21]}$  of the 5-membered ring heteroarene coupling partner, intermediate **C** is produced. Finally, reductive elimination affords 4-aryl-3-heteroarylpyridines **a** with catalyst regeneration.



Scheme 8. Selectivity of the Suzuki coupling with 4-(2-bromophenyl)pyridine 1 a and a thiophen-2-ylboronic acid.



Scheme 9. Proposed catalytic cycle for the access to a products.

# Conclusions

In summary, bromo-substituent as a traceless directing group was successfully employed to access heteroarylated pyridine derivatives. From both 3- and 4-(2-bromophenyl)pyridines, Pd 1,4-migration combined with direct arylation enabled the introduction of heteroarenes at positions 2, 3 or 4 on the pyridine rings. The new C–C bond created during this coupling reaction, results from the functionalization of two C-H bonds. The reaction proceeds nicely with a wide variety of heteroarenes and tolerates several substituents on the phenyl and pyridine rings of (2-bromophenyl)pyridines. In addition, this procedure employs an air stable palladium source combined with an inexpensive base and readily available 3- or 4pyridineboronic acids and dihalobenzenes. Consequently, this methodology constitutes a powerful tool for the heteroarylation of pyridines and could certainly be applied to introduce other functional groups on C-H bonds of pyridines.

### **Experimental section**

Procedure for the preparation of arylated arylpyridines **2a–35a**: As a typical experiment, the reaction of the (2-bromoaryl)pyridine derivative **1a–1p** (1 mmol), heteroarene (2 mmol) and KOPiv (0.280 g, 2 mmol) at 150 °C during 16 h in DMA (4 mL) in the presence of PdCl(C<sub>3</sub>H<sub>3</sub>)(dppb) (12.2 mg, 0.02 mmol) under argon affords the coupling product after evaporation of the solvent and purification on silica gel using a CombiFlash NextGen 300 with Buchi FlashPure cartridges containing 40 µm irregular silica. The **a**:**b** ratios were determined by <sup>1</sup>H NMR and GC/MS analysis of the crude mixtures.

## **Supporting Information**

The authors have cited additional references within the Supporting Information.  $^{\scriptscriptstyle [22,23]}$ 

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## **Conflict of Interests**

The authors declare no conflict of interest.

# Data Availability Statement

The data that support the findings of this study are available in the supplementary material of this article.

**Keywords:** C–H bond activation · Heteroarylation Homogeneous catalysis · Migration · Palladium · Pyridines 4, 3. Downloaded from https://chemistry-europe.onlinelibrary.wiley.com/doi/10.1002/cetc.202301226 by Universite De Rennes, Wiley Online Library on [10/10/2024]. See the Terms and Conditions (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable Creative Commons License



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- [1] S. De, S. K. Askol Kumar, S. Kumar Shah, S. Kazi, N. Sarkar, S. Banerjee, S. Dey, ' RSC Adv. 2022, 12, 15385–15406.
- [2] Recent Developments in the Synthesis and Applications of Pyridines, Ed. P. Singh, Elsevier Science, **2022**.
- [3] For reviews on metal-catalyzed C–H bond functionalization: a) L. Ackermann, Chem. Rev. 2011, 111, 1315–1345; b) P. B. Arockiam, C. Bruneau, P. H. Dixneuf, Chem. Rev. 2012, 112, 5879–5918; c) L. Theveau, C. Schneider, C. Fruit, C. Hoarau, ChemCatChem 2016, 8, 3183–3194; d) S. Ruiz, P. Villuendas, E. P. Urriolabeitia, Tetrahedron Lett. 2016, 57, 3413–3432; e) S. Agasti, A. Dey, D. Maiti, Chem. Commun. 2017, 53, 6544–6556; f) T. Gensch, M. J. James, T. Dalton, F. Glorius, Angew. Chem. Int. Ed. 2018, 57, 2296–2306; g) J. Kalepu, P. Gandeepan, L. Ackermann, L. T. Pilarski, Chem. Sci. 2018, 9, 4203–4216; h) K. Hirano, M. Miura, Chem. Sci. 2018, 92–32; i) A. M. Prendergast, G. P. McGlacken, Eur. J. Org. Chem. 2018, 2018, 6068–6082; j) S. Mao, H. Li, X. Shi, J.-F. Soulé, H. Doucet, ChemCatChem 2019, 11, 269–286; k) S. Rej, Y. Ano, N. Chatani, Chem. Rev. 2020, 120, 1788–1887; I) A. Saha, M. Shankar, S. Sau, A. K. Sahoo, Chem. Commun. 2022, 58, 4561–4587.
- [4] a) Y. Akita, A. Inoue, K. Yamamoto, A. Ohta, T. Kurihara, M. Shimizu, *Heterocycles* **1985**, *23*, 2327–2333; b) A. Ohta, Y. Akita, T. Ohkuwa, M. Chiba, R. Fukunaga, A. Miyafuji, T. Nakata, N. Tani, Y. Aoyagi, *Hetero-cycles* **1990**, *31*, 1951–1958.
- [5] a) R. Rossi, F. Bellina, M. Lessi, C. Manzini, Adv. Synth. Catal. 2014, 356, 17–117; b) H.-Y. Huang, A. Benzai, X. Shi, H. Doucet, Chem. Rec. 2021, 21, 343–356.
- [6] For selected examples of metal-catalyzed C–H functionalizations of pyridines: a) H.-Q. Do, R. M. K. Khan, O. Daugulis, J. Am. Chem. Soc. 2008, 130, 15185–15192; b) I. B. Seiple, S. Su, R. A. Rodriguez, R. Gianatassio, Y. Fujiwara, A. L. Sobel, P. S. Baran, J. Am. Chem. Soc. 2010, 132, 13194–13196; c) M. Wasa, B. T. Worrell, J. Q. Yu, Angew. Chem. Int. Ed. 2010, 49, 1275–1277; d) Y. Fujiwara, J. A. Dixon, F. O'Hara, E. D. Funder, D. D. Dixon, R. A. Rodriguez, R. D. Baxter, B. Herle, N. Sach, M. R. Collins, Y. Ishihara, P. S. Baran, Nature 2012, 492, 95–100; e) F. Dai, Q. Gui, J. Liu, Z. Yang, X. Chen, R. Guo, Z. Tan, Chem. Commun. 2013, 49, 4634–4636; f) G. L. Gao, W. Xia, P. Jain, J. Q. Yu, Org. Lett. 2016, 18, 744–747.
- [7] For an example of Pd-catalyzed C–H C2-arylation of a 3-arylpyridine: Y. Zeng, C. Zhang, C. Yin, M. Sun, H. Fu, X. Zheng, M. Yuan, R. Li, H. Chen, Org. Lett. 2017, 19, 1970–1973.
- For selected examples of direct arylations of 5-membered ring [8] heteroarene: Thiophenes: a) K. Masui, A. Mori, K. Okano, K. Takamura, M. Kinoshita, T. Ikeda, Org. Lett. 2004, 6, 2011-2014; b) E. David, S. Pellet-Rostaing, M. Lemaire, Tetrahedron 2007, 63, 8999-9006; c) J. Roger, F. Požgan, H. Doucet, Green Chem. 2009, 11, 425-432; d) B. Liégault, I. Petrov, S. I. Gorlesky, K. Fagnou, J. Org. Chem. 2010, 75, 1047-1060; Furans: e) M. Parisien, D. Valette, K. Fagnou, J. Org. Chem. 2005, 70, 7578-7584; f) B. Liégaut, D. Lapointe, L. Caron, A. Vlassova, K. Fagnou, J. Org. Chem. 2009, 74, 1826-1834; g) J. J. Dong, J. Roger, F. Požgan, H. Doucet, Green Chem. 2009, 11, 1832-1846; Pyrroles or indoles: h) F. Bellina, S. Cauteruccio, R. Rossi, Eur. J. Org. Chem. 2006, 2006, 1379-1382; Thiazoles: i) A. Mori, A. Sekiguchi, K. Masui, T. Shimada, M. Horie, K. Osakada, M. Kawamoto, T. Ikeda, J. Am. Chem. Soc. 2003, 125, 1700-1701; j) A. Yokooji, T. Okazawa, T. Satoh, M. Miura, M. Nomura, Tetrahedron 2003, 59, 5685-5689; k) J. J. Dong, J. Roger, C. Verrier, T. Martin, R. Le Goff, C. Hoarau, H. Doucet, Green Chem. 2010, 12, 2053-2063.

- [9] For reviews on Pd 1,4-migration in catalytic organometallic reactions: a) S. Ma, Z. Gu, Angew. Chem. Int. Ed. 2005, 44, 7512–7517; b) Y. Yasui, J. Syn. Org. Chem. Jpn. 2008, 66, 251–252; c) X. Dong, H. Wang, H. Liu, F. Wang, Org. Chem. Front. 2020, 7, 3530–3556.
- [10] For selected examples of Pd-catalyzed reactions involving a Pd-migration: a) R. C. Larock, Y. D. Lu, A. C. Bain, C. E. Russell, J. Org. Chem. 1991, 56, 4589–4590; b) Q. Huang, A. Fazio, G. Dai, M. A. Campo, R. C. Larock, J. Am. Chem. Soc. 2004, 126, 7460–7461; c) J. Zhao, D. Yue, M. A. Campo, R. C. Larock, J. Am. Chem. Soc. 2007, 129, 5288–5295; d) M. A. Campo, H. Zhang, T. Yao, A. Ibdah, R. D. McCulla, Q. Huang, J. Zhao, W. S. Jenks, R. C. Larock, J. Am. Chem. Soc. 2007, 129, 6298–6307.
- [11] a) L. Liu, H. Doucet, Adv. Synth. Catal. 2022, 364, 2783–2795; b) L. Liu, M. Cordier, T. Roisnel, H. Doucet, Org. Chem. Front. 2023, 10, 1441–1455.
- [12] G. Karig, M.-T. Moon, N. Thasana, T. Gallagher, *Org. Lett.* **2002**, *4*, 3115–3118.
- [13] a) J.-W. Li, L.-N. Wang, M. Li, P.-T. Tang, N.-J. Zhang, T. Li, X.-P. Luo, M. Kurmoo, Y.-J. Liu, M.-H. Zeng, *Org. Lett.* **2020**, *22*, 1331–1335; b) K. Kinoshita, M. Yamaguchi, H. Sasou, H. Konishi, K. Manabe, *Chem. Pharm. Bull.* **2023**, *71*, 175–182.
- [14] a) G. Karig, J. A. Spencer, T. Gallagher, *Org. Lett.* 2001, *3*, 835–838; b) E. Karadeniz, M. Zora, N. Z. Kilicaslan, *Tetrahedron* 2015, *71*, 8943–8952; c) J. Fan, W. Zhang, W. Gao, T. Wang, W.-L. Duan, Y. Liang, Z. Zhang, *Org. Lett.* 2019, *21*, 9183–9187.
- [15] For a review on arene C–H functionalization using directing groups: F. Zhang, D. R. Spring, *Chem. Soc. Rev.* 2014, 43, 6906–6919.
- [16] T. Cantat, E. Génin, C. Giroud, G. Meyer, A. Jutand, J. Organomet. Chem. 2003, 687, 365–376.
- [17] For the formation of Pd colloids at high temperature in polar solvents such as DMA, DMF or NMP: M. T. Reetz, J. G. de Vries, *Chem. Commun.* 2004, 1559–1563.
- [18] M. Power, K. Mackey, M. E. Light, D. J. Jones, G. P. McGlacken, *Eur. J. Org. Chem.* 2023, 2023, e202300790.
- [19] S. K. Callear, J. Spencer, H. Patel, J. J. Deadman, M. B. Hursthouse, J. Chem. Crystallogr. 2011, 41, 523–527.
- [20] N. Miyaura, A. Suzuki, Chem. Rev. 1995, 95, 2457-2483.
- [21] For concerted metalation deprotonation mechanism: a) D. L. Davies,
  S. M. A. Donald, S. A. Macgregor, J. Am. Chem. Soc. 2005, 127, 13754–13755; b) M. Lafrance, K. Fagnou, J. Am. Chem. Soc. 2006, 128, 16496–16497; c) S. I. Gorelsky, D. Lapointe, K. Fagnou, J. Am. Chem. Soc. 2008, 130, 10848–10849; d) S. I. Gorelsky, Coord. Chem. Rev. 2013, 257, 153–164; e) D. L. Davies, S. A. Macgregor, C. L. McMullin, Chem. Rev. 2017, 117, 8649–8709; f) R. A. Alharis, C. L. McMullin, D. L. Davies, K. Singh,
  S. A. Macgregor, J. Am. Chem. Soc. 2019, 141, 8896–8906.
- [22] C.-G. Dong, T.-P. Liu, Q.-S. Hu, Synlett 2009, 1081–1086.
- [23] J. Ruch, A. Aubin, G. Erbland, A. Fortunato, J.-P. Goddard, Chem. Commun. 2016, 52, 2326–2329.

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