

#### In vitro fortification of N $\epsilon$ -carboxymethyl-lysine in bovine serum albumin model systems

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#### Poster 17



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# *In vitro* fortification of Nε-carboxymethyl-lysine in bovine serum albumin model systems

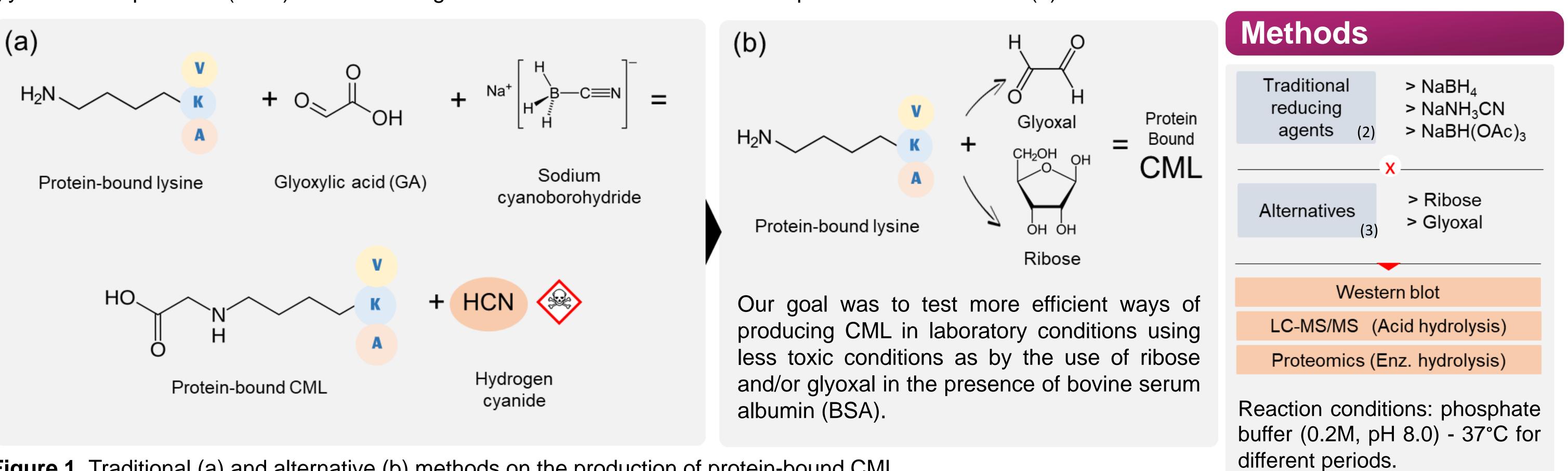
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## Introduction

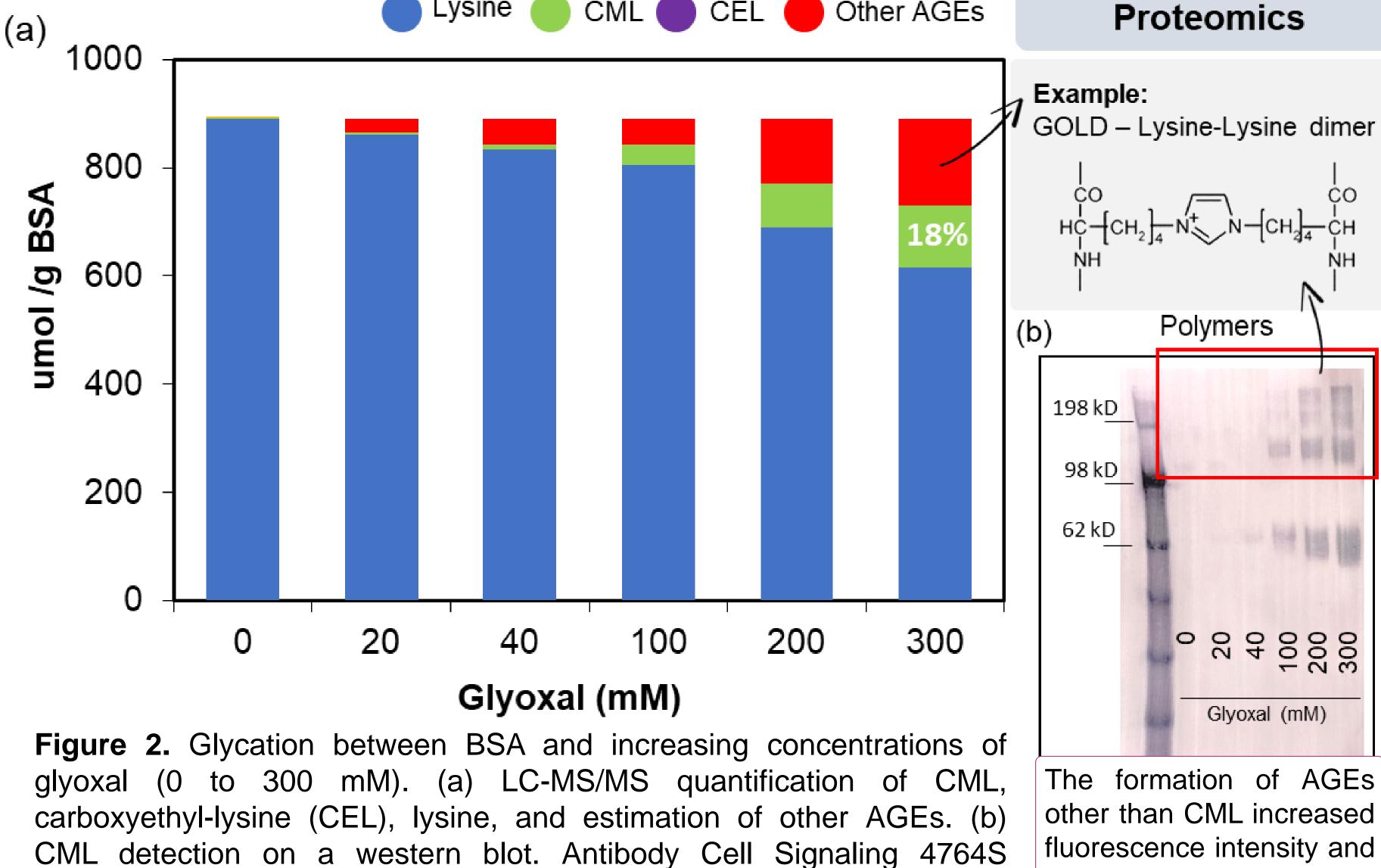
The heat treatment of foods results in the production of a heterogeneous group of glycated molecules related to the flavor and aroma of foods, but associated with potential negative health effects. Carboxymethyl-lysine (CML), result of the glycation of lysine residues, is a major dietary advanced glycation end-products (AGE) for the investigation of the effects of AGE consumption on human health (1).

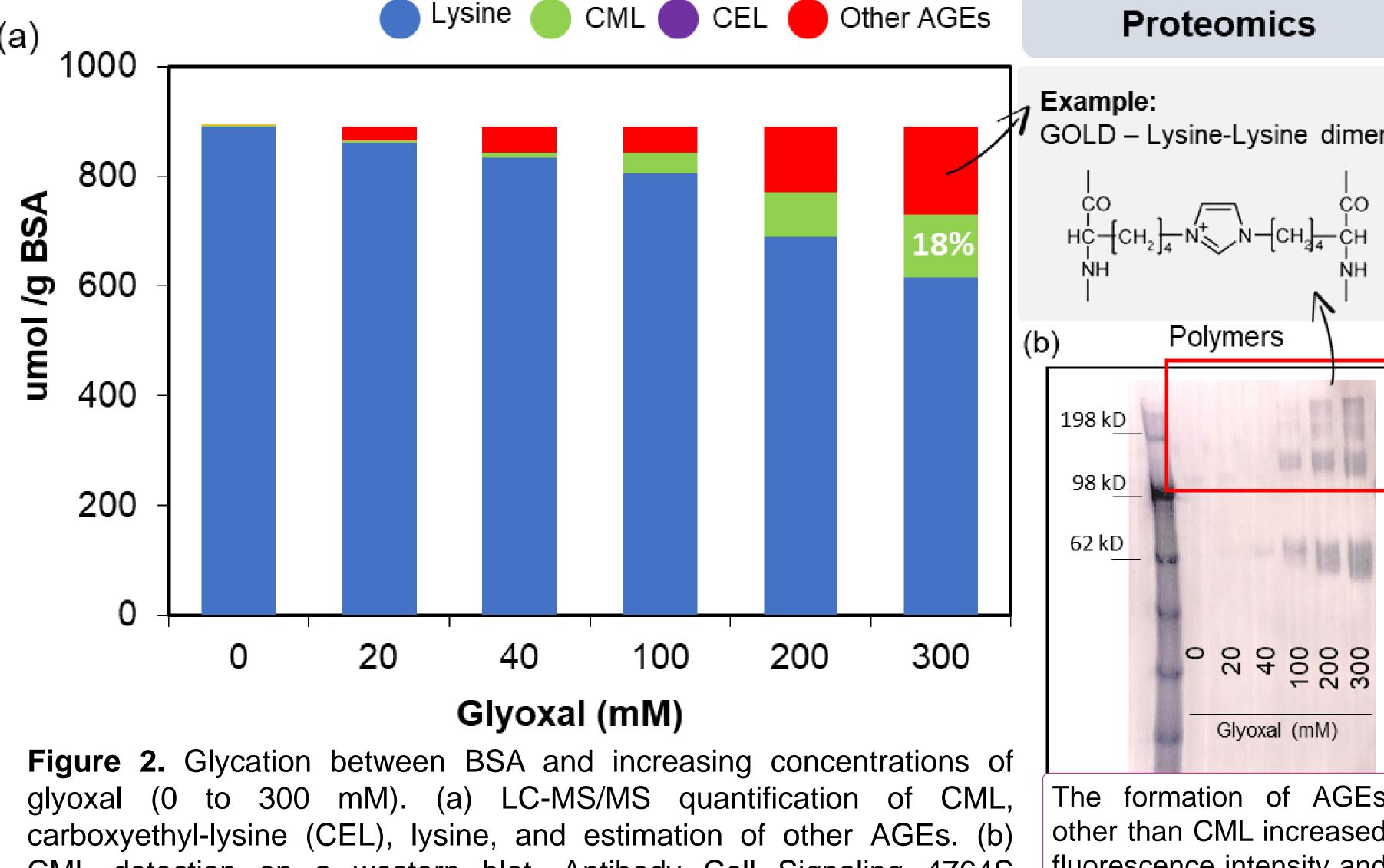


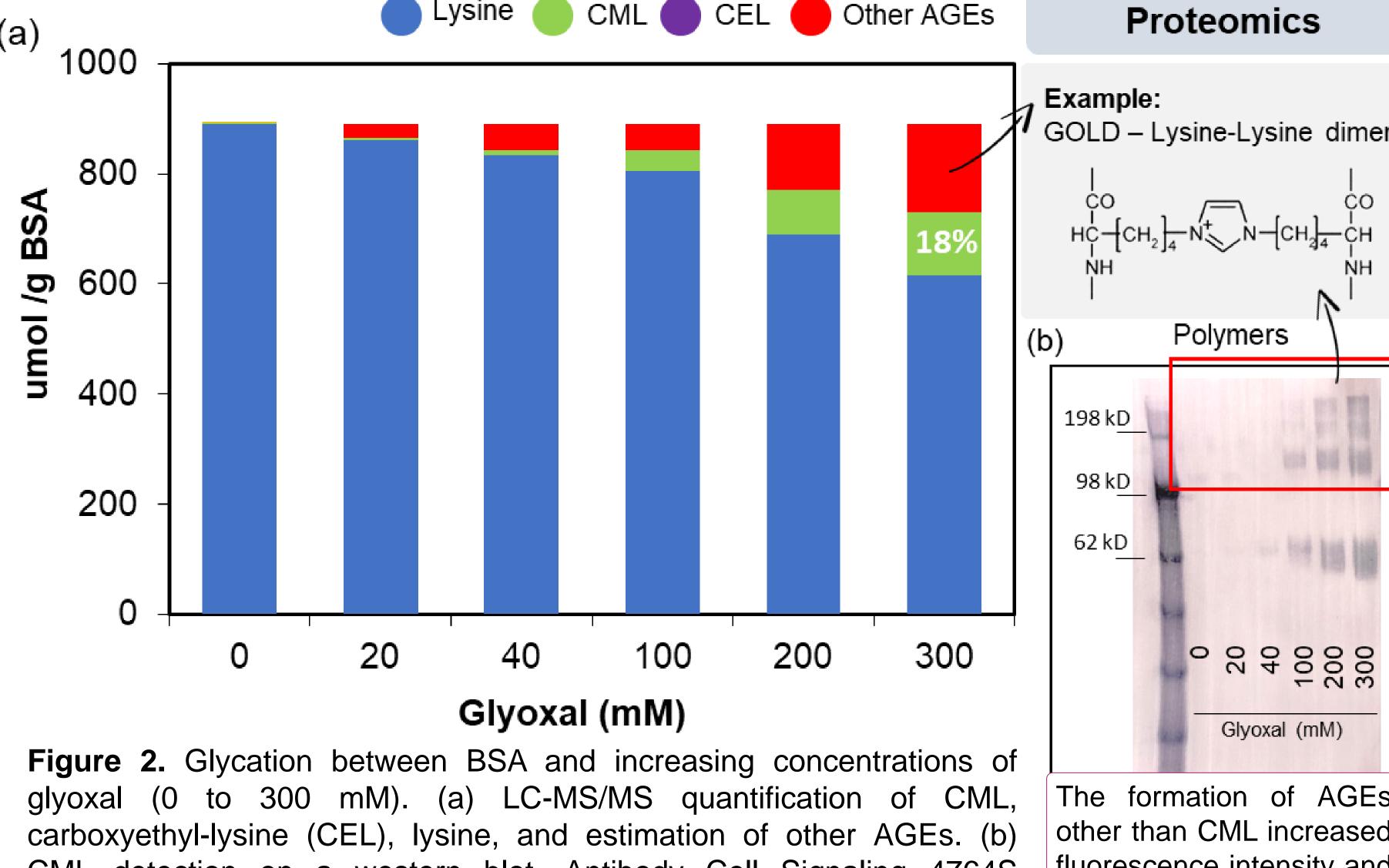
**Figure 1.** Traditional (a) and alternative (b) methods on the production of protein-bound CML.

#### **Results**

Degree of lysine modification into CML(%) in BSA using traditional methods:







Reagent	CML formation (% of lysine)
NaNH <sub>3</sub> CN	37.0
NaBH <sub>4</sub>	3.5
NaBH(OAc)	1.6

Ribose with or without glyoxal induced a loss of 35 to 57% of lysine, but low CML formation (1-2%).

The CML fortification of BSA with increasing concentrations of glyoxal alone formed CML in proportion to the glyoxal concentration: 0.5, 1.0, 4.5, 9.8 and 18% of lysine with 20, 40, 100, 200 and 300 mM glyoxal, respectively (Fig. 2a).

Further characterization:

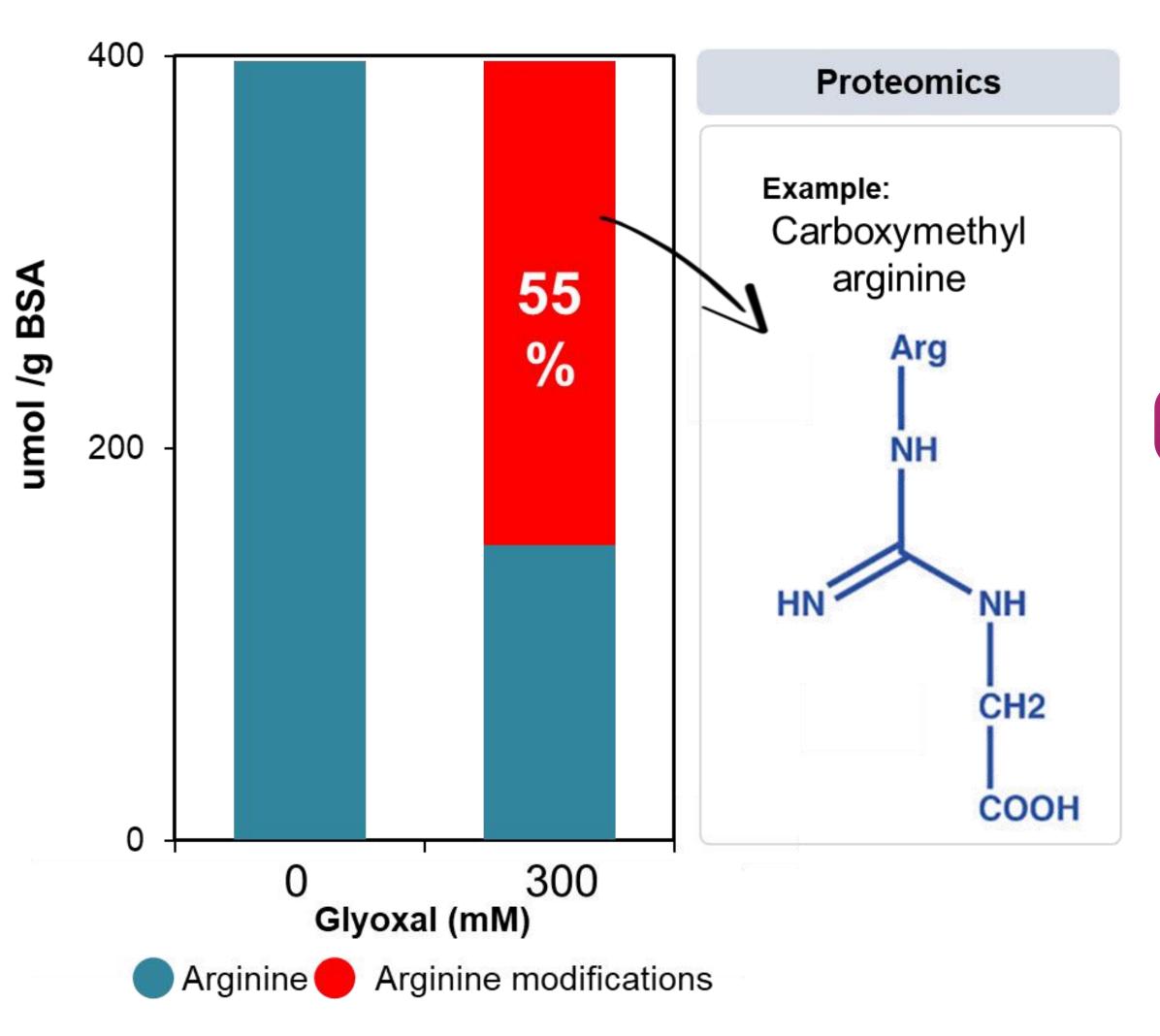


Figure 3. Glycation between BSA and glyoxal. Assessment of arginine modifications by LC-MS/MS.

Proteomic analysis identified arginine modifications as carboxymethyl-arginine in this preparation (Fig. 3).

## Conclusion

(1/5000).

#### **Alternative:**

BSA + 300mM glyoxal alone is a suitable alternative to GA with NaBH<sub>3</sub>CN Pros: > No HCN production

> Satisfactory lysine modification into CML (18%) compared to other methods Perspectives:

> Finding a model protein selectively fortified only with CML remains to be discovered; > Characterize the crosslinks between proteins.

## References

(1) Nguyen et al. (2014). *F Reviews Inte,* 30, 36. (2) Andersen et al. (2012). Synlett, 4, 531. (3) Glomb et al. (1995). *J Bio Chem*, 270, 10017.

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the formation of BSA

polymers.