

# Thermo-Hydro-Glycol Ageing of Polyamide 6,6: Microstructure-Properties Relationships

Clément Laügt, Jean-Luc Bouvard, Gilles Robert, Noëlle Billon

### ▶ To cite this version:

Clément Laügt, Jean-Luc Bouvard, Gilles Robert, Noëlle Billon. Thermo-Hydro-Glycol Ageing of Polyamide 6,6: Microstructure-Properties Relationships. Polymers, 2022, 14 (19), pp.4097. 10.3390/polym14194097. hal-04389118

## HAL Id: hal-04389118 https://hal.science/hal-04389118

Submitted on 11 Jan2024

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Article



1

2

3

4

5

6 7

8

9

10

25 26

27

# Thermo-hydro-glycol ageing of Polyamide 6,6. Microstructureproperties relationships.

Clément Laügt 1,\*, Jean-Luc Bouvard 1, Gilles Robert<sup>2</sup> and Noëlle Billon 1

- <sup>1</sup> Ecole des Mines de Paris, PSL Reaserch University, Centre de Mise en Forme des Matériaux (CEMEF), CNRS UMR 7635, 06904, Sophia Antipolis Cedex, France ; <u>jean-luc.bouvard@minesparis.psl.eu</u> (J.L.B.) ; <u>noelle.billon@minesparis.psl.eu</u> (N.B.)
- <sup>2</sup> DOMO Chemicals, POLYTECHNYL France SAS, Avenue Albert Ramboz, 69190, Saint Fons, France ; gilles.robert@domo.org
- \* Correspondence: clement.laugt@gmail.com

Abstract: The microstructural evolutions occurring during the thermo-hydro-glycol ageing of an 11 injection molded PA66 were studied. They were correlated to the evolutions of its mechanical prop-12 erties. The aged samples were immersed in an antifreeze fluid - mainly composed of water and 13 ethylene glycol - at varying times and temperatures. The aim was to combine an as exhaustive as 14 possible microstructural investigation and a rigorous mechanical analysis. Consequently, the mi-15 crostructure of the aged and unaged PA66 was assessed through the average molar mass, the diam-16 eter of the spherulites, the lamellae thickness, the crystallites apparent size, a crystal perfection in-17 dex, and a crystallinity index. Moreover, a core-skin approach was set up. The mechanical conse-18 quences of the microstructural changes were investigated by DMA and tensile testing. The local true 19 strain fields were measured with a digital image correlation system. The temperatures and strain 20 rates of the tests were chosen by referring to the time-temperature superposition principle. It is con-21 cluded that the water and ethylene glycol intake resulted in an intense plasticization, the loss of the 22 molar mass resulted in the embrittling of the polymer, and finally, it was identified that the changes 23 of the crystalline structure have an influence on the stiffness of PA66. 24

Keywords: PA66; Microstructure; Mechanical Behavior; Thermo-hydro-glycol ageing

#### 1. Introduction

Polyamide 6,6 (PA66) is a semicrystalline polymer, widely used in the automotive 28 industry, as a matrix for reinforced thermoplastic parts. It is especially used in the cooling 29 systems of the engines. There, it faces high temperatures in the presence of the antifreeze 30 fluid, which is mainly composed of water and ethylene glycol. Over time, these conditions 31 may operate as a physical or chemical ageing. They also may impact the crystalline mi-32 crostructure, and eventually the mechanical properties. The present survey had two main 33 goals. The first, to assess the microstructure evolution of an injection molded PA66 when 34 it undergoes high temperatures in the presence of water and ethylene glycol. The second, 35 to investigate the eventual mechanical consequences of the microstructure changes. Ac-36 cording to the literature, three main phenomena were expected to occur: the diffusion of 37 water and ethylene glycol molecules into the amorphous phase, the chemical interactions 38 between these molecules and the polymer, and eventually, as suggested, some crystal 39 modifications. 40

The amide functions of polyamides favor H-bounds, especially with species like water and ethylene glycol. The absorption of water and ethylene glycol by PA66, is a welldocumented phenomenon [1–3]. The interactions between these molecules and the amide functions were confirmed with the FT-IR method [4–6]. In parallel, the kinetics of the diffusion process is often described with a Fick law [7–11]. As far as the mechanical behavior 45

Citation: Lastname, F.; Lastname, F.; Lastname, F. Title. *Polymers* **2022**, *14*, x. https://doi.org/10.3390/xxxx

Academic Editor: Firstname Lastname

Received: date Accepted: date Published: date

**Publisher's Note:** MDPI stays neutral with regard to jurisdictional claims in published maps and institutional affiliations.



**Copyright:** © 2022 by the authors. Submitted for possible open access publication under the terms and conditions of the Creative Commons Attribution (CC BY) license (https://creativecommons.org/license s/by/4.0/). is concerned, the interactions between the amide functions and the water or ethylene glycol molecules have an influence on the apparent mobility of the macromolecular chains. 47 It results in the well-known plasticizing effect, which is characterized by a decrease of the  $\alpha$  relaxation temperature  $T_{\alpha}$  [8–10]. The intensity of this decrease depends on plasticizer content [12–14]. 50

In addition to the above-mentioned physical interactions, some chemical interactions 51 between the polymer and the absorbed species can also be observed. PA66 is indeed cre-52 ated by a polycondensation/hydrolysis balanced reaction. The polycondensation involves 53 a diamine and a diacide, and it generates water as a sub-product. The presence of water 54 into the amorphous phase of the polymer may shift the equilibrium of this balanced reac-55 tion in favor of the hydrolysis reaction, resulting in the scission of the macromolecular 56 chains and in the decrease of the average molar mass [8,15–18] [15] [16]. In that field, 57 Ledieu [8] and Jacques et al. [15] suggested a kinetic model for the hydrolysis of PA66 and 58 PA11, respectively. The main impact of the molar mass decrease is the embrittlement of 59 the polymer [8,17,19] characterized by a decrease of the ultimate strain in tension. The 60 amide functions of the PA66 can also react with the ethylene glycol molecules. This reac-61 tion called glycolysis could break the macromolecular chains, creating some ester and 62 amine functions. Some authors identified the occurrence of this reaction through the evo-63 lution of the FT-IR peaks corresponding to the ester and amine functions [20–22]. Kim et 64 al. [20], and Huckowski et al. [22] also measured the decrease of the PA66 average molar 65 mass within the glycolysis time and suggested a model for this reaction. However, the 66 glycolysis reaction was not observed at temperatures below 200°C without the use of a 67 catalyst. Such temperatures were not reached in the present study. Hence, the glycolysis 68 reaction was not expected. Finally, when immersed in water with oxygen, polyamide can 69 also undergo a chemical degradation due to oxidation [23]. This reaction results in the 70 random chain scission, and in the crosslinking of the polymer. This latter phenomenon 71 can be observed through the increase of the polydispersity index [23,24]. It should be 72 noted that the present paper neither focuses on which specific chemical reaction takes 73 place nor on their kinetic, but on their microstructural consequences. 74

As far as the crystal phase is concerned, annealing processes can generally have an impact on the crystal properties of semicrystalline polymers such as crystallinity, apparent size of the crystallites, lamellae thickness and phase organization. These crystal properties may have an influence on the polymer mechanical behavior, especially on the stiffness and ductility [25–27]. 79

Finally, the injection molding process does not generate a homogeneous microstructure. Depending on its location in the mold, the material does not undergo the same thermo-mechanical history in terms of temperature, cooling rate, pressure, or shear rate. These differences in the thermo-mechanical conditions may have an influence on the microstructure and on the mechanical properties [26,28,29].

In the literature, only few studies deal with the microstructure evolution of polyam-85 ide during a thermo-hydro-glycol ageing, and these focused on the average molar masses. 86 The aim of this survey was then to give a thorough assessment of the phenomena involved 87 in the thermo-hydro-glycol ageing of an injection molded PA66. The microstructure was 88 described by the average molar mass and the crystal properties. Several methods were 89 involved to evaluate the microstructure. The molar mass was measured by size-exclusion 90 chromatography and a crystal index was measured by wide angle X-ray scattering 91 (WAXS). Moreover, a multiscale approach was set up to describe the crystalline phase. At 92 the spherulite scale, the spherulite diameters were observed and measured by polarized 93 light microscopy. At a lower scale, the lamellae thickness was evaluated by differential 94 scanning calorimetry (DSC) and the crystal apparent size was measured by WAXS. Fi-95 nally, at the crystal cell scale, a crystal perfection index was estimated by WAXS. A signif-96 icant part of this survey was also to evaluate the impact of the microstructure evolution 97 on the mechanical behavior. It is a unique feature of this study, as in the literature, only 98 few studies handle at the same time a complete microstructural and a strong mechanical 99

108

109

evaluation. Here, the viscoelastic properties were assessed by dynamic mechanical anal-100 ysis (DMA) measurements. The large strain mechanical behavior was assessed through 101 tensile testing, taking advantage of time-temperature superposition principle as already 102 suggested in the literature [30–32]. The local true strain was measured with a digital image 103 correlation (DIC) system. Finally, to take into account that the injection-molding process 104 does not generate a homogeneous microstructure in the thickness of the samples, a core-105 skin approach was set up to consider the likely microstructural and mechanical gradient 106 in the thickness of the PA66 samples. 107

#### 2. Materials and Methods

#### 2.1. Materials

The material was a PA66 Technyl grade. Plaques were processed from the same batch 110 and provided by the company DOMO Chemicals. The samples were machined in the 111 thickness of PA66 injection molded 3.2 mm-thick plaques. Reference samples were dried 112 under vacuum at 80°C during 48h. The thermo-hydro-glycol ageing were performed in 113 an autoclave where samples were immerged in an antifreeze bath at different tempera-114 tures from 120°C to 140°C, during times from 48h to 1000h. The antifreeze was a 50 % - 50 115 % volume mixture of water and ethylene glycol. The glycol was supplied by the company 116 DOMO Chemicals. In the autoclave, the pressure only depended on the process tempera-117 ture: no pressure instruction was set. Core/skin measurements were performed by reduc-118 ing the thickness of the samples with a milling machine: 1 mm thick from the surface in 119 case of "skin" samples and 1 mm thick in the core for "core" samples. The sampling strat-120 egy is illustrated in Figure 1 121



Entire thickness of an injection-molded sample

Figure 1. Illustration of the skin-core sampling strategy. The core and the skin samples were tooled 123 from the whole thickness of the plates with a milling machine. 124

#### 2.2. Steric Exclusion Chromatography

The average molar masses were determined by the company DOMO Chemicals with 126 the size-exclusion chromatography (SEC) method. The samples were taken in the whole 127 thickness of the plaques, including the core and skin areas. Due to confidentiality reasons, 128 the protocol cannot be described in more detail, and the initial molar mass of the unaged 129 PA66 cannot be released. Moreover, only the standardized molar masses are displayed in this paper. They were standardized from the initial molar mass of the unaged PA66. 131

122

125

A Leica polarized light microscope fitted with a gypsum plate, was used to observe 133 the spherulites from the core to the surface of the molded samples. Thin PA66 slices were 134 cut with a glass-blade microtome. Their thickness was around 100 μm. 135

#### 2.4. Wide Angle X-Ray Scattering

(100)

(002)

15

The WAXS analysis were performed with a Philips X'Pert Pro, using CuK $\alpha$  radiation 137 with a wavelength  $\lambda$  = 1.540 Å. A goniometer with a  $\theta$ - $\theta$  set-up was used. The tests were 138 performed with a 12° - 40° angular scan and an exposure time of 12 min. After the meas-139 urements, the diffractograms were decomposed to identify the amorphous and the crystal 140 phases contributions. A flat background was fixed at 80% of the minimum intensity. A 141 typical decomposition is depicted in Figure 2. The peaks can be indexed to their corre-142 sponding plan families thanks to the works of Bunn & Garner [33], Starkweather [34,35] 143 and Haberkorn [36]. The peaks were calculated with Pearson VII symmetric functions. 144The mid-height width of the (100) peak was measured, as well as the angular gap between 145 the (100), and the (010),(110) peaks. 146

(017)

(010),(110)



25

2θ (°)

Amorphous

contribution

Figure 2. Typical decomposition of a PA66 diffractogram, with the identified plan families.

35

30

### 2.5. Differential Scanning Calorimetry

20

The thermal analyses were performed with a PerkinElmer DSC4000. Samples (ranging between 2 and 5 mg) were tooled from the surface and the core of the plates, and then placed in some aluminum pans. The samples were heated with a 10°C/min rate, from 20°C to 300°C, in order to reach their complete fusion. Calibrations were performed with indium and zinc. The typical mass of the samples was around 2 mg. 154

#### 2.6. Dynamic Mechanical Analysis

Dynamic mechanical analyses were performed with a tensile set-up on a PerkinElmer 156 DMA 8000. The  $\alpha$  relaxation temperature ( $T_{\alpha}$ ) was determined from temperature scans at 157 1 Hz with a 2°C/min temperature rate. Some master curves used for the time-temperature 158 superposition principle were built from frequency scans between 0.1 Hz and 10 Hz. To 159 build the master curves, frequency sweeps were performed every 5°C at 40 stabilized temperatures from the glassy to the rubbery plateau. For each ageing condition, the  $T_{\alpha}$  temperatures were chosen as the reference temperature for the master curves. 162

136

149

155

147

163

#### 2.7. Tensile testing

The large strain mechanical behavior was explored by performing some temperature 164 controlled tensile tests with a 3400 Serie Instron electromechanical device equipped with 165 a 30 kN load cell and an oven. The tests were performed with a  $5.0 \times 10^{-4} s^{-1}$  strain rate. 166 A digital image correlation (DIC) system was used to track the local displacement and 167 true strain fields ( $\varepsilon$ ) over time. The DIC system was composed of two cameras in stereo-168 correlation. For each of the tensile specimens, one face was speckled with black and white 169 temperature-resistant paints. The local displacement and strain fields were calculated by 170 analyzing the evolution of the random speckle patterns with the analysis software VIC-171 3D - Correlated Solutions. A tracking point was then positioned in the maximum strain 172 area. The mechanical behavior was investigated by measuring the strain at the tracking 173 point, i.e.: at the maximum of the strain field. A typical strain field, and the location of the 174 tracking point are depicted in Figure 3 (a) where  $\varepsilon_{yy}$  is the strain in the tensile direction. 175 The true stress was calculated within the frame of the transverse isotropy assumption, 176 given by equation 1. 177

$$\sigma = \frac{F(t)}{S(t)} = \frac{F(t)}{l_0 \times e_0 \times exp(2\varepsilon_{rr})} \tag{1}$$

Where  $\sigma$  is the stress, F(t) is the strength, S(t) is the section of the tensile test spec-178 imen,  $l_0$  and  $e_0$  are respectively the initial width and the initial thickness of the speci-179 men.  $\varepsilon_{xx}$  is the local true strain measured by the tracking point, in the width direction of 180 the specimen. As depicted in Figure 3 (b), for every tensile test, an elasticity modulus was 181 measured, as well as the true stress at a 0.2 strain. As the stress vs strain curves do not 182 display a real linear elastic domain, the elastic modulus was calculated by a linear regres-183 sion in a range of true strain arbitrarily set from 0 to 0.01. A partial unload was introduced 184between 0.35 and 0.65. On the stress-strain curves, the partial unload leads to the occur-185 rence of a hysteresis loop. Finally, the master curves built by DMA were used to set the 186 loading temperatures and strain rates by referring to the time-temperature superposition 187 principle [30-32]. 188



Figure 3. Typical analysis of a tensile test. PA66 taken at the skin of the plaque, aged during 168h 190 at 120°C; (a) Strain field measured by DIC, and location of the tracking point; (b) Measurement of 191 the elasticity modulus and the stress for a 0.2 strain. The strain was measured on the tracking 192 point. 193

189

#### 3. Results and discussion

#### 3.1. Microstructural evolution

#### 3.1.1. Average molar mass

As mentioned before, during the ageing process, the water and ethylene glycol mol-199 ecules spread into the amorphous phase of PA66 until saturation. In the range of consid-200 ered temperatures, the hydrolysis reaction occurred (but not glycolysis), resulting in the 201 chain scission of the polymer. Its effects were measured by the decrease of the average 202 molar mass with the ageing time (Figure 4). In addition to the hydrolysis reaction, the 203 chain scission can also be a consequence of the oxidation reaction. As already mentioned, 204 the identification of the chemical reactions and the study of their kinetic was not the ob-205 jective of this paper. The standardized average molar masses measured by the company 206 DOMO Chemicals are depicted in Figure 4. Whatever the chemical reaction (hydrolysis 207 and/or oxidation), the chain scission is faster as the ageing temperature increases. These 208 results are consistent with the literature [8,15]. At the maximum, the molar mass de-209 creased by 65 %. This decrease was reached after a 336h long ageing at 140°C. For all the 210 ageing conditions, the polydispersity index remained close to 2. It reveals a homogeneous 211 random chain scission phenomenon and the absence of a crosslinking effect due to oxida-212 tion. 213



Figure 4. Reduction of the average molar masses due to the hydrolysis reaction during the ageing processes. 215

All the results displayed in Figure 4 arise from samples taken in the whole thickness 217 of the plaques, containing core and skin areas. However, after a 336h ageing at 120°C, 218 specific measurements were also performed in the core and skin areas separately. The 219 resulting standardized molar masses are 0.660 at the core and 0.657 at the skin. The gap 220 between these two measurements is under 0.5% indicating that after ageing, the molar 221 mass is homogeneous in the thickness of the plaques. Although oxidation is shown to be 222 a diffusion limited reaction [37], the present result means that for 1 mm thick areas (core 223 and skin areas), the possible diffusion limited phenomena are neglectable if the whole 224 bulk of the samples are considered. 225

#### 3.1.2. Microscopy observations

Microscopy was used to assess the microstructure at the crystallites scale. A typical 227 polarized light microscopy picture is shown on Figure 5. The spherulites can be easily 228 identified by their Maltese cross pattern. The picture brings out a microstructure gradient: 229 spherulite globally get smaller from the core to the surface. 230

225 226

214

197



Figure 5. Polarized light microscopy observation of the spherulites for a 140°C 168h aged PA66. The232picture on the left depicts half of the thickness of the plaque.233

To estimate sizes of spherulites, around 15 spherulites were manually selected near 234 the core and 15 near the surface. Their area was calculated with ImageJ software, and then 235 their diameters for the same area were deduced. This was reproduced for each condition 236 (the initial one and the aged ones). The diameter of the core and the skin spherulites are 237 reported in Figure 6. The differences observed between aged samples must not be inter-238 preted as resulting from ageing. They may arise from variations in the measurements. 239 Since these variations are quite small, it ensures that the selected spherulites were repre-240 sentative of the whole spherulite population. Hence, an average diameter over 165 test 241 points was estimated at the core and at the skin of the plaques (15 measurements per age-242 ing conditions). The average diameter is around 40  $\mu$ m at the core, and around 20  $\mu$ m at 243 the skin. This gradient clearly arises from the injection molding process. Indeed, the dif-244ferences in the shear rate and cooling rate between the core and the surface may have an 245 influence on the nucleation and spherulites growth phenomena [38,39]. 246



Figure 6. Spherulites diameters measured from microscopy observations.

#### 3.1.3. DSC results

Typical PA66 fusion peaks are displayed in Figure 7. The melting temperature (tem-250 perature at the maximum of the peak) ranges between 261°C and 263.5°C. Near 240°C, a 251 small exotherm can be observed but its origin is not well determined. It is sometimes at-252 tributed to a crystalline reorganization from unstable crystals to more stable ones [25,40]. 253 This exotherm progressively disappears when the ageing temperature increases. It could 254 be interpreted as a reorganization of the crystalline phase occurring during ageing. In that 255 case higher temperatures should favor a greater reorganization. Still in Figure 7, a shoul-256 der can be observed on the peaks around 255°C. This well-known asymmetry is some-257 times attributed to the fusion of smaller and less perfect crystalline entities, or to the trans-258 formation of a mesophase to a crystalline phase [41-43]. Nevertheless, whatever the un-259 derlying phenomenon, the evolution of this shoulder after ageing, highlights that a mi-260 crostructural reorganization occurred during the ageing process. More specifically, it 261 seems that the shoulder is less pronounced when increasing the ageing temperature. It 262 means that the microstructural reorganization is related to the ageing temperature. 263

249

247



Figure 7. Typical fusion peaks of unaged and 168h aged PA66.

In a more quantitative standpoint, the microstructure was assessed at the lamellae 266 scale by considering the lamellae thickness. It was calculated from the DSC thermograms 267 with the Gibbs-Thomson equation (equation 2). 268

$$T_f = T_f^0 \left( 1 - \frac{2\sigma_e}{\Delta H_f^0 \cdot \rho_c \cdot L_c} \right)$$
(2)

Where  $\sigma_e$  is the basal surface energy per unit area,  $\rho_c$  is the density of the crystal, 269  $\Delta H_f^0$  is the melting enthalpy of a perfect crystal and  $T_f^0$  is the melting temperature of an 270 infinite crystal. For calculation, the following values were used:  $\sigma_e = 0.0296 J. m^{-2}$ ,  $T_f^0 = 271$  $270 \ ^\circ C$ ,  $\rho_c = 1240 \ kg. m^{-3}$  and  $\Delta H_f^0 = 196 J. g^{-1}$  [26]. 272

The evolution of the lamellae thickness is gathered in Figure 8. If one focuses on the 273 average values, the lamellae are initially slightly thicker at the core than at the skin, but 274 the difference is contained in the error bars. It means that the lamellae thickness is not 275 concerned by the microstructural gradient generated by the injection molding process. 276 Although the initial thickness is similar in the core and skin areas, Figure 8 display an 277 increase of the thickness at the core and a decrease at the skin. These results point out that 278 the thermo-hydro-glycol ageing can generate opposite phenomena depending on where 279 the material is located in the thickness of the sample. It was not shown yet in the literature. 280 The basic phenomena underlying the evolution of the thickness are difficult to identify, 281 nevertheless, the pre-melting or the refolding of the fold surfaces could be mentioned as 282 well as the melting of the thinner unperfect lamellae [44,45]. 283



Figure 8. Influence of the ageing conditions on the lamellae thickness at the core and the surface of the PA66 plates.

#### 3.1.4. WAXS results

Typical diffractograms of unaged and 168h aged PA66 are gathered in Figure 9. Only 287 a restricted angular window is displayed so that the attention is focused on the two main peaks of the diffractograms. For the unaged polymer, the first main peaks near 20° is higher and wider at the skin than at the core. It may reveal a smaller crystallites apparent size. In the core and skin areas, the first main peak gets sharper after ageing. Its angular position is also shifted to smaller angles whereas the angular position of the second main peak (near 23°) is shifted to higher angles. These observations are the consequences of 293 microstructural changes occurring during ageing, that will be discussed later. 294



Figure 9. Typical diffractograms of unaged and aged PA66 during 168h.

From the diffractograms of the aged and unaged PA66, the microstructure was again 296 assessed at the lamellae scale bay calculating the crystallites apparent size (CAS). It is cal-297 culated from mid-height width of the diffraction peaks by means of Scherrer equation 298 (equation 3), assuming that,  $\theta$  is the angular position of a crystal peak and  $\Delta 2\theta$  is its 299

286

284

285

mid-height width (Figure 2). It corresponds to the dimension of the diffracting crystal entities in the direction normal to the diffracting plans associated to  $\theta$ . 300

$$CAS = \frac{0.9 \lambda}{\Delta 2\theta \cos \theta}$$
(3)

The (100) plans and their corresponding peak were chosen to calculate the CAS. 302 Hence the CAS was measured in the H bounding direction, i.e.: perpendicular to the chain 303 axis. Figure 10 points out that the diffracting elements are initially and after ageing, 304 around 30 Å bigger at the core than at the skin. As for the diameter of the spherulites, 305 these results indicate a microstructural gradient in the thickness of the plaques caused by 306 the injection molding process. After ageing, an increase of the CAS can be noticed both at 307 the core and at the skin. It testifies of a reduction of the crystal defects. The resulting la-308 mellae are then composed of bigger diffracting entities. Hence at the lamellae scale, be-309 sides modifying the lamellae thickness, the thermo-hydro-glycol ageing also modified the 310 structure of the crystal. More precisely, it resulted in a "perfectioning" of the crystal. The 311 shape of the curves (Figure 10) suggests that the perfectioning reaches an equilibrium over 312 time, and especially after 168h for the skin. In this specific area the value of the equilibrium 313 is besides related to the ageing temperature. 314



**Figure 10.** Influence of the ageing conditions on the CAS at the core and at the surface of the PA66 315 plaques. 316

At the crystal cell scale, the microstructure was assessed by measuring a perfection 317 index, referring to the compacity of the crystal. To define it, let us notice that the PA66 318 crystalline entities are mainly composed of a triclinic  $\alpha$  phase [33] which can be refined 319 into a  $\alpha_1$  phase and a  $\alpha_2$  phase [34,35]. The density of the  $\alpha_1$  phase is higher. In other 320 words, the  $\alpha_1$  phase is more perfect than the  $\alpha_2$  phase. Because of their different com-321 pacity, the diffracting peaks created by these two phases do not appear at the same angu-322 lar position. Hence Haberkorn et al. [36] identified that the angular gap between the (100) 323 and the (010),(001) peaks (Figure 2) is proportionally related to the amount of  $\alpha_1$  phase 324 against  $\alpha_2$  phase. The increase of the angular gap between the (100) and the (010),(001) 325 peaks highlights an increase of the proportion of the  $\alpha_1$  phase, that is to say, an increase 326 of the packing perfection in crystalline phase. This angular gap can hence be considered 327 as a crystal perfection index. The measurements brought out that at the dry unaged state, 328 the crystal entities are more perfect at the core than at the skin (Figure 11). Once again, it 329 is a consequence of the injection molding process. After ageing, the perfection index in-330 creased both at the core and at the skin. The increase is related to the ageing temperature, 331 and it seems to reach an equilibrium after a 168h ageing. Hence, the results still conclude 332 that the thermo-hydro-glycol ageing can modify the structure of the crystal of PA66, and 333 more precisely increase its perfection. 334



**Figure 11.** Influence of the ageing conditions on the crystal perfection index: the angular gap between the (100) peak and the (010),(001) peak of WAXS diffractograms. 336

The WAXS method was also used to assess the crystallinity of the aged and unaged 337 PA66. A crystallinity index ( $\chi_c$ ) was defined as the ratio of the area of the crystalline peaks 338 ( $A_c$ ) over the total area of the diffractograms, including the crystalline and amorphous 339 contributions ( $A_a + A_c$ ) (equation 4). 340

$$\chi_c = \frac{A_c}{A_a + A_c} \tag{4}$$



**Figure 12.** Influence of the ageing conditions on the crystallinity index at the core and the skin of the PA66 plaques. 343

All the measurements of the crystallinity index are gathered in Figure 12. Before age-344 ing, the crystallinity index is very close at the core and at the skin. They both lie around 345 0.4. Initially, the crystallinity index is a little bit lower at the core, but the difference with 346 the skin is lower than the error bars. After ageing, the results bring out some different 347 crystal growth processes occurring at the core and at the skin (Figure 12). At the core, the 348 evolution of the crystallinity index is consistent with the evolution of the lamellae thick-349 ness (Figure 8): the growth of the lamellae in their thickness direction results in a higher 350 crystallinity index. At the skin, the results are more difficult to construe. They could be 351 interpreted as the dissolution of the external part of the lamellae into the amorphous 352 phase. If these parts have a too high density of defects, one could assume that they are not 353 detected with the WAXS method. According to this assumption, at the skin, the dissolu-354 tion of the external part of the lamellae may results in the decrease of the lamellae thick-355 ness without changing the crystallinity index. Nevertheless, despite these different 356

363

tendencies, the values gathered in Figure 12 above all point out that the crystallinity index
does not change a lot during the ageing. Indeed, the average values only change of few
percent, and these changes are not significant when compared with the error bars. Hence,
as far as the crystalline phase is concerned, the thermo-hydro-glycol ageing does not genaction ageing in the amount of crystal, but in the very structure of the crystalline
action ageing
actio

3.1.5. Microstructural summary

Finally, microstructural changes caused by a thermo-hydro-glycol ageing were as-364 sessed. A wide range of microstructural parameters was considered, which is a novelty 365 concerning thermo-hydro-glycol ageing. The ageing results in the decrease of the average 366 molar mass due to the hydrolysis and/or oxidation reaction. Concerning the crystalline 367 phase, a core-skin gradient was observed in the PA66 plaques. It is a consequence of the 368 injection molding process. After ageing only few changes of the crystallinity index were 369 observed, but the structure of the crystal changed significantly. More precisely significant 370 evolutions of the lamellae thickness were observed, as well as a "perfectioning" of the 371 crystal entities. The main microstructural observations are summarized in Table 1. 372

**Table 1.** Summary of the microstructural changes occurring in an injection molded PA66 during a373thermo-hydro-glycol ageing.374

| Measurement scale          | Main results   |
|----------------------------|--|
|                            | Reduction of the average molar mass due to the hy-     |
| Molar mass                 | drolysis reaction                                      |
|                            | No core/skin gradient.                                 |
| Spherulites diameter       | Core/skin gradient.                                    |
| Crystallites apparent size | Increase of the crystallites apparent size at the core |
|                            | and at the skin.                                       |
|                            | Core/skin gradient.                                    |
| Lamellae thickness         | Thickening of the lamellae at the core of the          |
|                            | plaques.   |
|                            | Reduction of the thickness at the skin.                |
| Crystal perfection index   | Increase of the perfection at the core and at the      |
|                            | skin.  |
|                            | Core/skin gradient before ageing.                      |
| Crystallinity index        | Few changes of the crystallinity index.                |
|                            | Small increase at the core of the plaques.             |

#### 3.2. Mechanical consequences

#### 3.2.1. Plasticization

The intake of small molecules such as water and ethylene glycol, is well known for 377 resulting in a plasticizing effect. It is characterized by the decrease of the alpha relaxation 378 temperature  $T_{\alpha}$ . Moreover, some models were developed, to foresee the evolution of  $T_{\alpha}$ 379 in a polymer-plasticizer or a polymer-polymer blend [13,14,46-48]. Hence after the ageing 380 processes, a decrease of  $T_{\alpha}$  was measured by DMA temperature scans.  $T_{\alpha}$  was identified 381 as the temperature of the damping factor peak. This phenomenon is illustrated in Figure 382 13 a, where after a specific ageing condition, the damping factor peak is shifted to a lower 383 temperature. All the measured  $T_{\alpha}$  are reported in Table 2. Hence, after ageing  $T_{\alpha}$  de-384 creases from around 70°C for a dry unaged PA66, to -20°C for the aged ones. The exact 385 amount of water and ethylene glycol in the aged samples was not calculated. Karls-Fisher 386 analysis could be set up to do so. Nevertheless, as all the aged PA66 got approximately a 387

375

90°C decrease of  $T_{\alpha}$ , it suggests that they all contain the same amount of water and the same amount of ethylene glycol. 389

**Table 2.**  $T_{\alpha}$  for the dry unaged and the aged PA66.

| Ageing condition | $T_{\alpha}$ in the core | $T_{\alpha}$ in the skin |
|------------------|--------------------------|--------------------------|
| Dry unaged       | 70°C                     | 70°C                     |
| 120°C 168h       | -19°C                    | -18°C                    |
| 120°C 336h       | -19°C                    | -19°C                    |
| 120°C 504h       | -20°C                    | -19°C                    |
| 120°C 1000h      | -20°C                    | -17°C                    |
| 130°C 168h       | -19°C                    | -20°C                    |
| 130°C 336h       | -21°C                    | -20°C                    |
| 130°C 504h       |                          | -19°C                    |
| 135°C 168h       | -22°C                    | -20°C                    |
| 135°C 336h       | -19°C                    | -20°C                    |
| 140°C 168h       | -20°C                    | -18°C                    |

In addition to the plasticization effect, the glassy plateau of the aged PA66 was shifted 391 to lower temperatures and much higher modulus (Figure 13 b). This phenomenon could 392 be interpreted as an antiplasticization effect [9,49]. This effect was so strong that the glassy 393 plateau was not observed for the aged PA66 in this study, even at -150°C. No lower temperatures were considered, as -150°C is already close to the minimum of the operating 395 range of the machine. 396





The two storage modulus curves displayed in Figure 13 b point out that if an aged 399 and a dry unaged PA66 are loaded at the same temperature and the same strain rate, they 400 may not be loaded in the same state. For example, at 20°C and 1 Hz the dry unaged PA66 401 is in a glassy state, whereas the aged one is in a rubbery-like state (see the vertical dashed 402 line). In order to load all the polymers in the same state, an additional drying of the aged 403 samples could have been considered. However, this method has two major drawbacks: 404

- The additional drying could again modify significatively the microstructure of the aged samples;
- It would avoid from eventual mechanical consequences due to the presence of water 407 and ethylene glycol other than the plasticization. 408

Hence, in order to load all the PA66 in the same state, without resorting to additional 409 drying, an approach referring to the time-temperature superposition principle was set up. 410 For every aged PA66 as for the dry unaged one, a master curve was built at a reference 411

temperature, for the core material and the skin materials. The master curves were built by 412 performing frequency scans at several fixed temperatures. For every temperature, a hori-413 zontal shift factor was defined to fit with the curve obtained at the reference temperature 414(Figure 14). 415



Figure 14. Illustration of some frequency scans used for building a master curve. Example of a dry 417 unaged PA66. For a better readability, only few temperatures are displayed. 418

For all the considered PA66, the reference temperature was defined as  $T_{\alpha}$ : 70°C for 419 the dry unaged PA66, and -20°C for the aged ones. Using the master curves, some loading 420 conditions were identified in order to load both the aged and dry unaged PA66 in their 421 rubbery-like state [30-32]. 422

Hence, the dry unaged PA66 were loaded 80°C and  $5.10^{-4}s^{-1}$ , and the aged PA66 423 were loaded at  $-5^{\circ}$ C and  $5 \cdot 10^{-4} s^{-1}$ . Depending on the master curves, the equivalent 424 strain rate at  $T_{ref} = T_{\alpha}$  lied between  $10^{-6}s^{-1}$  and  $10^{-5}s^{-1}$ . Figure 15 a depicts the mas-425 ter curves of a dry unaged PA66 and a PA66 aged during 168h at 120°C. They were both 426 taken at the core of the plaques. The range of the targeted equivalent strain rate is indi-427 cated on the master curves by the two vertical dotted lines. This range only represents a 428 restricted portion of the rubbery-like plateau, meaning that the loading conditions are 429 very similar. 430

The corresponding strain-stress curves are depicted in Figure 15 b. They were ob-431 tained with a dry unaged PA66 and a 120°C/168h aged PA66 taken at the core, loaded 432 with the above-mentioned conditions. As the resulting mechanical behaviors are very 433 similar, the following statements can be confirmed:

- Referring to the time-temperature superposition principle is a relevant approach to 435 foresee the mechanical behavior of PA66 and to set the loading conditions; 436
- Thanks to this approach all the PA66 considered in this survey were loaded in a rub-437 bery like state without resorting to an additional drying of the aged specimens. 438

416



Figure 15. Superposition of two typical master curves. The dry unaged PA66 and the aged one are 439 both loaded on their rubbery plateau. The corresponding strain - stress curves are plotted on the 440 right. 441

3.2.2. Impact of the molar mass reduction



Figure 16. Strain-stress curves of PA66 aged during 168h, and a dry unaged PA66. Specimens taken at the core of the plaques. 445

The large strain mechanical behavior of the aged and unaged PA66 was assessed with 446 two or three tensile tests for each condition. Figure 16 depicts the stress-strain curves of 447 PA66 aged during 168h and a dry unaged PA66, taken at the core of the plaques. As men-448tioned above, all the PA66 display almost the same mechanical behavior, corresponding 449 to the rubbery-like plateau. For the aged samples, a breaking occurred before the final 450 unload. A strain at break was then measured, and the results showed that the embrittling 451 of the polymer was related to the reduction of the molar mass. Figure 17 gathers the strain 452 at break measured on every aged specimen. They are plotted as a function of the corre-453 sponding standardized molar mass. It clearly brings out that the true strain at break is 454 driven by the average molar mass. More precisely, the molar mass reduction occurring 455 during the ageing process results in the embrittling of the material, both at the core and at 456

442

443

the surface. Moreover, this relationship appears to be linear, and almost the same slope is457obtained at the core and at the skin. It means that compared to the average molar mass,458the crystalline differences between the core and the skin have a neglectable influence on459the ductility of the polymer.460



Figure 17. Evolution of the true strain at break with the decreasing molar mass. Measurements and461tendency: (a) at the core; (b) at the skin.462

These results are consistent with those from the literature [8,19,50], although most of463the studies dealing with the thermo-hydro-glycol ageing of PA66 consider engineering464strain and not the local true strain.465

3.2.3. Impact of the crystal on the stiffness

Finally, the impact of the crystal properties on the large strain mechanical behavior 467 was also investigated. For every ageing or dry unaged condition, it was observed that the 468 core material was always stiffer than the skin material. This observation is illustrated in 469 Figure 18. In this figure, two stress-strain curves are displayed for every ageing condition: 470



one for the core material and one for the skin material. It can be observed that globally, 471 the core material is stiffer than the skin material. 472

Figure 18. Illustration of the gradient of stiffness between the core ang the skin material. Two curves474are displayed for every ageing condition: one for the core and one for the skin.475

At the beginning of the tensile tests, the gradient of stiffness can be observed by comparing the elastic modulus. They are all reported in Table 3. Except for the PA66 aged during 336h at 130°C and 135°C, the elastic modulus is always higher at the core than at the skin. However, for the above-mentioned condition, the difference between the core and the skin is lower than some tens of MPa i.e.: almost the accuracy of the measurement. 480

**Table 3.** Elastic modulus of the aged PA66 end the dry unaged one. Measurements on specimens481taken at the core and at the skin of the plaques.482

| Ageing condition | Core     | Skin    |
|------------------|----------|---------|
| Dry unaged       | 920 MPa  | 800 MPa |
| 120°C 168h       | 840 MPa  | 740 MPa |
| 120°C 336h       | 910 MPa  | 870 MPa |
| 120°C 504h       | 940 MPa  | 910 MPa |
| 120°C 1000h      | 1030 MPa | 950 MPa |
| 130°C 168h       | 880 MPa  | 840 MPa |
| 130°C 336h       | 820 MPa  | 880 MPa |
| 130°C 504h       | 890 MPa  | 840 MPa |
| 135°C 168h       | 890 MPa  | 880 MPa |
| 135°C 336h       | 840 MPa  | 870 MPa |
| 140°C 168h       | 850 MPa  | 780 MPa |

483

473

Moreover, this gradient of stiffness comes along with a gradient of microstructure. 484 As the crystallinity index is almost the same at the core and at the skin (Figure 12), the 485 gradient of stiffness may be caused by the very structure of the crystal. As displayed in 486 Figure 6 and Figure 19, the dimensions of the crystalline entities are always bigger at the 487 core than at the skin, especially the spherulites diameter, the lamellae thickness, and the 488 crystallites apparent size. Hence, it can be stated core/skin gradient of microstructure 489



generated by the thermo-hydro-glycol ageing results in a gradient of stiffness with a core 490 material stiffer than the skin material. 491

Figure 19. Comparison of the dimension of the crystalline entities at the core and at the surface:492Lamellae thickness and crystallites apparent size.493

The influence of the lamellae thickness was then more precisely investigated, and not 494 limited to a comparison between the core and the skin. For every ageing condition, the 495 elastic modulus, and the measured stress at 0.2 were plotted as a function of the corre-496 sponding lamellae thickness (Figure 20). It appears that an increase of the lamellae thick-497 ness results in an increase of the stiffness. A linear tendency can be identified, both for the 498 elastic modulus and the stress at 0.2. The relationship between the lamellae thickness and 499 the stiffness could be explained by referring to the constraint level of the interlamellar 500 amorphous phase. One could assume that as the lamellae are thicker, the amorphous 501 phase is more constrained, leading to a stiffer mechanical behavior 502



Figure 20. Influence of the lamellae thickness on the PA66 stiffness: Elastic modulus and true stress 503 at a 0.2 strain. 504

#### 3.2.4. Mechanical summary

The mechanical behavior of the aged and unaged PA66 was assessed by tensile testing. It seems to depend above all on the state of the polymer, which depends itself on the 507 gap with  $T_{\alpha}$ . To avoid the plasticizing effect caused by the water and ethylene glycol intake, an approach referring to the time-temperature superposition principle was set-up. All the PA66 were then the loaded in a rubbery like state. For a same state it was pointed out that the brittleness of the polymer is primarily driven by its average molar mass. Still 511 for a same state, changes in the crystalline structure have a non-negligible influence of 512 stiffness of the polymer, and especially the lamellae thickness. This influence is the reason 513 behind the observation of a core/skin gradient of stiffness. The main microstructure-prop-514 erties relationships are summarized in Table 4. 515

Table 4. Microstructure-properties relationships observed in an aged injection molded PA66.

| Microstructural observations                            | Mechanical consequences   |  |
|---|---|--|
| Water and ethylene glycol intake                        | Plasticization of PA66: decrease of $T_{\alpha}$                          |  |
| Decrease of the average molar mass                      | Embrittlement of the polymer  |  |
| Core-skin gradient: bigger crystal entities at the core | Core stiffer than the skin  |  |
| Evolution of the lamellae thickness                     | Evolution of the stiffness: thicker lamellae result in a stiffer polymer. |  |

#### 4. Conclusions

A wide range of experimental methods was used to assess the evolution of the mi-519 crostructure and the mechanical behavior of an injection molded PA66 over a thermo-520 hydro-glycol ageing. The objective was to give a complete view of the microstructural 521 evolutions occurring during the ageing, and to analyze their mechanical consequences. 522 The microstructure-properties relationships were hence investigated. To take into account 523 that the injection molding process does not generate a homogeneous microstructure, a 524 core/skin approach was set up. 525

Concerning the microstructure, a decrease of the average molar mass was measured. 526 It is related to the occurrence of the hydrolysis and/or oxidation reaction. Moreover, the 527 measures of the polydispersity index assert the random chain scission of the polymer. The 528 objective of this paper was not to determine the exact nature or kinetic of the chemical 529 reaction, but rather to focus on its microstructural consequences. 530

505 506

508 509 510

516

20 of 23

518

In addition to the molar mass measurements, the evolution of the crystalline phase 531 was investigated. The amount of crystal was discussed, as well as the structure of the 532 crystal itself, which is a novelty concerning the thermo-hydro-glycol ageing. To do so the 533 structure was assessed from the spherulite scale to the crystal cell scale. Before and after 534 ageing, a microstructural gradient was observed in the thickness of the specimens as a 535 consequence of the injection molding process. The thermo-hydro-glycol ageing caused 536 very few changes of the crystallinity index whereas the structure of the crystal changed 537 significatively. At the core the lamellae thickness increased whereas it decreased at the 538 skin. It means that in addition to the core/skin gradient, opposite phenomena can also 539 occur in the thickness of the samples during the thermo-hydro-glycol ageing, and not only 540 at the extreme surface as it is for example in the case of diffusion limited reactions [37]. 541 Finally, both at the core and at the skin, the thermo-hydro-glycol ageing resulted in the 542 "perfectioning" of the crystal attested by an increase of the crystallite apparent size and 543 the crystal perfection index. The increase of these two parameters is construed by respec-544 tively by a decrease of the density of defects in the crystal, and an increase of its compacity. 545 Even though it was not the objective of this paper, it can be said, that it was difficult to 546 identify the specific role of water, ethylene glycol and the heat treatment in the changes 547 of the crystalline phase. To do so, specific ageing in only water, ethylene glycol or in a 548 controlled atmosphere could be considered. 549

The mechanical behavior of PA66 was investigated by tensile testing. A first consequence of the ageing was the plasticization of the polymer caused by the water and ethylene glycol intake. Around a 90°C decrease of  $T_{\alpha}$  was measured for all the aged PA66. To load plasticized polymers in the same state than that of the dry unaged ones, the loading conditions were chosen by referring to the time-temperature superposition principle. Solution and unaged PA66 were then loaded in a rubbery like state. This method was found relevant to do so without resorting to additional drying. Solutions were chosen by referring to additional drying.

The results showed that the mechanical behavior of the aged and unaged PA66 is 557 above all related to the state of the polymer. Indeed, for a same rubbery like state, all the 558 tested polymers had a similar mechanical behavior. Still for a same state, the ductility of 559 PA66 is driven by its average molar mass, hence its reduction due to hydrolysis/oxidation 560 led to the embrittlement of PA66 Finally, the changes in the structure of the crystal occur-561 ring during the ageing was found to have impact on the stiffness of the polymer, especially 562 for a same rubbery like state, the stiffness is related to the lamellae thickness. This influ-563 ence of the crystalline structure is the cause of a gradient of stiffness observed in the thick-564 ness of the samples. 565

Author Contributions: Conceptualization, C.L., J.L.B, G.R. and N.B.; methodology, C.L., J.L.B, G.R.567and N.B.; validation, C.L.; formal analysis, C.L.; investigation, C.L.; resources, G.R.; writing—origi-568nal draft preparation, C.L.; writing—review and editing, J.L.B, G.R. and N.B.; visualization, C.L.;569supervision, J.L.B, G.R. and N.B.; project administration, N.B.; funding acquisition, N.B.570

**Funding:** This research was funded by the co-operative FUI project "THERMOFIP" piloted by DOMO CHEMICALS and was co-funded by the ANRT (French National Association for Research and Technology) 573

Acknowledgments: The authors gratefully acknowledge DOMO Chemicals for supporting this574work. The authors would also like to thank the ANRT (French National Association for Research575and Technology) for its support via the FUI ThermoFip.576

Conflicts of Interest: The authors declare no conflict of interest.

577 578

#### References

- Puffr, R.; Šebenda, J. On the Structure and Properties of Polyamides. XXVII. The Mechanism of Water Sorption in Polyamides. 580 J. Polym. Sci. Part C Polym. Symp. 1967, 16, 79–93, doi:10.1002/polc.5070160109. 581
- Murthy, N.S.; Akkapeddi, M.K.; Orts, W.J. Analysis of Lamellar Structure in Semicrystalline Polymers by Studying the Absorption of Water and Ethylene Glycol in Nylons Using Small-Angle Neutron Scattering. *Macromolecules* 1998, 31, 142–152, doi:10.1021/ma9707603.
- 3. Broudin, M.; Le Gac, P.Y.; Le Saux, V.; Champy, C.; Robert, G.; Charrier, P.; Marco, Y. Water Diffusivity in PA66: Experimental Characterization and Modeling Based on Free Volume Theory. *Eur. Polym. J.* **2015**, *67*, 326–334, doi:10.1016/j.eur-polymj.2015.04.015.
- 4. Iwamoto, R.; Murase, H. Infrared Spectroscopic Study of the Interactions of Nylon-6 with Water. J. Polym. Sci. Part B Polym. Phys. 2003, 41, 1722–1729, doi:10.1002/polb.10526.
- 5. Lim, L.-T.; Britt, I.J.; Tung, M.A. Sorption and Transport of Water Vapor in Nylon 6,6 Film. *J. Appl. Polym. Sci.* **1999**, *71*, 197–206, doi:10.1002/(SICI)1097-4628(19990110)71:2<197::AID-APP2>3.0.CO;2-J.
- Hong, J.H.; Dhevi, D.M.; Lee, J.S.; Kim, K.J. Origin of Deterioration in Mechanical Properties of Glass Fiber Reinforced Nylon 6,6 Composites by Aqueous Ethylene Glycol Solution. *Polym. Compos.* 2007, 28, 778–784, doi:10.1002/pc.20351.
- Thomason, J.L.; Porteus, G. Swelling of Glass-Fiber Reinforced Polyamide 66 during Conditioning in Water, Ethylene Glycol, and Antifreeze Mixture. *Polym. Compos.* 2011, 32, 639–647, doi:10.1002/pc.21054.
- 8. Ledieu, B. Aging in Water/Glycol Medium of Polyamide 66 Reinforced in Short Glass Fibers for Water End Tank of Radiator Application. Theses, Arts et Métiers ParisTech, 2010.
- 9. Thomason, J.L.; Porteus, G. An Investigation of Glass–Fiber Reinforced Polyamide 66 during Conditioning in Various Automotive Fluids. *Polym. Compos.* **2011**, *32*, 1369–1379, doi:10.1002/pc.21160.
- Thomason, J.L.; Ali, J.Z.; Anderson, J. Effects of Hydrolysis Ageing on the Performance and Dimensional Stability of Glass-Fiber Reinforced Polyamide 66. In Proceedings of the 68th Annual Technical Conference and Exhibition, ANTEC 2010; May 16 2010;
   pp. 265–271.
- Thomason, J.L.; Ali, J.Z. The Dimensional Stability of Glass–Fibre Reinforced Polyamide 66 during Hydrolysis Conditioning.
   *Compos. Part Appl. Sci. Manuf.* 2009, 40, 625–634, doi:10.1016/j.compositesa.2009.02.017.
- 12. Smith, L.S.A.; Schmitz, V. The Effect of Water on the Glass Transition Temperature of Poly(Methyl Methacrylate). *Polymer* **1988**, 605 29, 1871–1878, doi:10.1016/0032-3861(88)90405-3. 606
- Kelley, F.N.; Bueche, F. Viscosity and Glass Temperature Relations for Polymer-Diluent Systems. J. Polym. Sci. 1961, 50, 549– 556, doi:10.1002/pol.1961.1205015421.
- 14. Fabre, V.; Quandalle, G.; Billon, N.; Cantournet, S. Time-Temperature-Water Content Equivalence on Dynamic Mechanical Response of Polyamide 6,6. *Polymer* **2018**, *137*, 22–29, doi:10.1016/j.polymer.2017.10.067.
- Jacques, B.; Werth, M.; Merdas, I.; Thominette, F.; Verdu, J. Hydrolytic Ageing of Polyamide 11. 1. Hydrolysis Kinetics in Water.
   *Polymer* 2002, 43, 6439–6447, doi:10.1016/S0032-3861(02)00583-9.
- Lee, J.-Y.; Kim, K.-J. MEG Effects on Hydrolysis of Polyamide 66/Glass Fiber Composites and Mechanical Property Changes. 613 Molecules 2019, 24, 755, doi:10.3390/molecules24040755. 614
- Deshoulles, Q.; Le Gall, M.; Dreanno, C.; Arhant, M.; Stoclet, G.; Priour, D.; Le Gac, P.Y. Origin of Embrittlement in Polyamide 6 Induced by Chemical Degradations: Mechanisms and Governing Factors. *Polym. Degrad. Stab.* 2021, 191, 109657, doi:10.1016/j.polymdegradstab.2021.109657.
- Deshoulles, Q.; Le Gall, M.; Dreanno, C.; Arhant, M.; Priour, D.; Le Gac, P.-Y. Modelling Pure Polyamide 6 Hydrolysis: Influence 618 of Water Content in the Amorphous Phase. *Polym. Degrad. Stab.* 2021, *183*, 109435, doi:10.1016/j.polymdegradstab.2020.109435.
- 19. Fayolle, B.; Colin, X.; Audouin, L.; Verdu, J. Mechanism of Degradation Induced Embrittlement in Polyethylene. *Polym. Degrad. Stab.* **2007**, *92*, 231–238, doi:10.1016/j.polymdegradstab.2006.11.012.
- 20. Kim, K.J.; Dhevi, D.M.; Lee, J.S.; Cho, Y.D.; Choe, E.K. Mechanism of Glycolysis of Nylon 6,6 and Its Model Compound by Ethylene Glycol. *Polym. Degrad. Stab.* **2006**, *91*, 1545–1555, doi:10.1016/j.polymdegradstab.2005.09.019.
- 21. Datta, J.; Błażek, K.; Włoch, M.; Bukowski, R. A New Approach to Chemical Recycling of Polyamide 6.6 and Synthesis of Polyurethanes with Recovered Intermediates. *J. Polym. Environ.* **2018**, *26*, 4415–4429, doi:10.1007/s10924-018-1314-4.
- 22. Huczkowski, P.; Kapko, J.; Olesiak, R. Degradation of Nylon-6 in Ethylene Glycol. *Polymer* **1978**, *19*, 77–80, doi:10.1016/0032-3861(78)90177-5.
- 23. Deshoulles, Q.; Le Gall, M.; Dreanno, C.; Arhant, M.; Priour, D.; Le Gac, P.Y. Chemical Coupling between Oxidation and Hydrolysis in Polyamide 6 - A Key Aspect in the Understanding of Microplastic Formation. *Polym. Degrad. Stab.* **2022**, *197*, 109851, doi:10.1016/j.polymdegradstab.2022.109851.
- Okamba-Diogo, O.; Richaud, E.; Verdu, J.; Fernagut, F.; Guilment, J.; Fayolle, B. Molecular and Macromolecular Structure 631 Changes in Polyamide 11 during Thermal Oxidation. *Polym. Degrad. Stab.* 2014, 108, 123–132, doi:10.1016/j.polymdegrad-632 stab.2014.05.028.
- Bureau, M.N.; Denault, J.; Cole, K.C.; Enright, G.D. The Role of Crystallinity and Reinforcement in the Mechanical Behavior of Polyamide-6/Clay Nanocomposites. *Polym. Eng. Sci.* 2002, 42, 1897–1906, doi:10.1002/pen.11082.
- Billon, N.; Giraudeau, J.; Bouvard, J.L.; Robert, G. Mechanical Behavior Microstructure Relationships in Injection-Molded Polyamide 66. *Polymers* 2018, 10, 1047, doi:10.3390/polym10101047.

592

593

596

597

598

599

609

610

615

616

617

620

621

622

623

624

625

626

627

628

629

630

579

582

583

- Yuan, M.; Turng, L.-S. Microstructure and Mechanical Properties of Microcellular Injection Molded Polyamide-6 Nanocomposites. *Polymer* 2005, 46, 7273–7292, doi:10.1016/j.polymer.2005.06.054.
   638
- Freire, L.; Combeaud, C.; Monge, G.; Billon, N.; Haudin, J.-M. Transcrystallinity versus Spherulitic Crystallization in Polyamide 640
   66: An Experimental Study. *Polym. Cryst.* 2019, *2*, e10028, doi:https://doi.org/10.1002/pcr2.10028.
- Russell, D.P.; Beaumont, P.W.R. Structure and Properties of Injection-Moulded Nylon-6. J. Mater. Sci. 1980, 15, 197–207, 642 doi:10.1007/BF00552445.
- Maurel-Pantel, A.; Baquet, E.; Bikard, J.; Bouvard, J.L.; Billon, N. A Thermo-Mechanical Large Deformation Constitutive Model 644 for Polymers Based on Material Network Description: Application to a Semi-Crystalline Polyamide 66. Int. J. Plast. 2015, 67, 645 102–126, doi:10.1016/j.ijplas.2014.10.004.
- Federico, C.E.; Bouvard, J.L.; Combeaud, C.; Billon, N. Large Strain/Time Dependent Mechanical Behaviour of PMMAs of Different Chain Architectures. Application of Time-Temperature Superposition Principle. *Polymer* 2018, 139, 177–187, 648 doi:10.1016/j.polymer.2018.02.021.
- Forestier, E.; Combeaud, C.; Guigo, N.; Monge, G.; Haudin, J.-M.; Sbirrazzuoli, N.; Billon, N. Strain-Induced Crystallization of Poly(Ethylene 2,5-Furandicarboxylate). Mechanical and Crystallographic Analysis. *Polymer* 2020, 187, 122126, doi:10.1016/j.polymer.2019.122126.
- 33. Bunn Charles William; Garner E. V.; Bragg William Lawrence The Crystal Structures of Two Polyamides ('Nylons'). *Proc. R. Soc. Lond. Ser. Math. Phys. Sci.* **1947**, *189*, 39–68, doi:10.1098/rspa.1947.0028.
- 34. Starkweather, H.W.; Jones, G.A. Crystalline Transitions in Powders of Nylon 66 Crystallized from Solution. *J. Polym. Sci. Polym. Phys. Ed.* **1981**, *19*, 467–477, doi:https://doi.org/10.1002/pol.1981.180190307.
- Starkweather, H.W.; Zoller, P.; Jones, G.A. The Heat of Fusion of 66 Nylon. J. Polym. Sci. Polym. Phys. Ed. 1984, 22, 1615–1621, doi:https://doi.org/10.1002/pol.1984.180220905.
- 36. Haberkorn, H.; Illers, K.H.; Simak, P. Calorimetric, X-Ray and Infra-Red Investigations on Poly(Hexamethylene Adipamide). *Polym. Bull.* **1979**, *1*, 485–490, doi:10.1007/BF00255713.
- Pliquet, M.; Rapeaux, M.; Delange, F.; Bourgogne, D.; Gardette, J.L.; Therias, S.; Bussiere, P.O. Multiscale Analysis of Thermal Degradation of Polyamide 6,6 - Influence of Temperature on Oxygen Diffusion-Limited Oxidation Profiles. *Polym. Degrad. Stab.* 2021, 192, 109695, doi:10.1016/j.polymdegradstab.2021.109695.
- Khoury, F. The Formation of Negatively Birefringent Spherulites in Polyhexamethylene Adipamide (Nylon 66). J. Polym. Sci. 1958, 33, 389–403, doi:10.1002/pol.1958.1203312637.
- 39. Haas, T.W.; Maxwell, B. Effects of Shear Stress on the Crystallization of Linear Polyethylene and Polybutene-1. *Polym. Eng. Sci.* **1969**, *9*, 225–241, doi:10.1002/pen.760090402.
- 40. Penel-Pierron, L.; Depecker, C.; Séguéla, R.; Lefebvre, J.-M. Structural and Mechanical Behavior of Nylon 6 Films Part I. Identification and Stability of the Crystalline Phases. *J. Polym. Sci. Part B Polym. Phys.* **2001**, *39*, 484–495, doi:10.1002/1099-0488(20010301)39:5<484::AID-POLB1022>3.0.CO;2-R.
- 41. Furushima, Y.; Nakada, M.; Ishikiriyama, K.; Toda, A.; Androsch, R.; Zhuravlev, E.; Schick, C. Two Crystal Populations with Different Melting/Reorganization Kinetics of Isothermally Crystallized Polyamide 6. *J. Polym. Sci. Part B Polym. Phys.* **2016**, *54*, 2126–2138, doi:10.1002/polb.24123.
- 42. Xie, S.; Zhang, S.; Liu, H.; Chen, G.; Feng, M.; Qin, H.; Wang, F.; Yang, M. Effects of Processing History and Annealing on Polymorphic Structure of Nylon-6/Montmorillonite Nanocomposites. *Polymer* **2005**, *46*, 5417–5427, doi:10.1016/j.polymer.2005.03.094.
- 43. Quintanilla, L.; Pastor, J.M. Structural Analysis of Polyamide-6,6 Reinforced with Glass Fibre by the Use of Fourier Transform Infra-Red Spectroscopy with Photoacoustic Detection and Differential Scanning Calorimetry. *Polymer* **1994**, *35*, 5241–5246, doi:10.1016/0032-3861(94)90475-8.
- 44. Fischer, E.W. Effect of annealing and temperature on the morphological structure of polymers. *Pure Appl. Chem.* **1972**, *31*, 113–132, doi:10.1351/pac197231010113.
- Yeh, G.S.Y.; Hosemann, R.; Loboda-Čačković, J.; Čačković, H. Annealing Effects of Polymers and Their Underlying Molecular Mechanisms. *Polymer* 1976, 17, 309–318, doi:10.1016/0032-3861(76)90187-7.
- Gordon, M.; Taylor, J.S. Ideal Copolymers and the Second-Order Transitions of Synthetic Rubbers. i. Non-Crystalline Copolymers. J. Appl. Chem. 1952, 2, 493–500, doi:10.1002/jctb.5010020901.
   684
- 47. Lu, X.; Weiss, R.A. Relationship between the Glass Transition Temperature and the Interaction Parameter of Miscible Binary Polymer Blends. *Macromolecules* **1992**, *25*, 3242–3246, doi:10.1021/ma00038a033.
- 48. Brinke, G.; Karasz, F.E.; Ellis, T.S. Depression of Glass Transition Temperatures of Polymer Networks by Diluents. *Macromolecules* **1983**, *16*, 244–249, doi:10.1021/ma00236a017.
- Serpe, G.; Chaupart, N. Relaxation–Structure Relationship in Bulk and Plasticized Polyamide 11. J. Polym. Sci. Part B Polym. 690 Phys. 1996, 34, 2351–2365, doi:10.1002/(SICI)1099-0488(199610)34:14<2351::AID-POLB5>3.0.CO;2-Z. 691
- Reano, A.F.; Guinault, A.; Richaud, E.; Fayolle, B. Polyethylene Loss of Ductility during Oxidation: Effect of Initial Molar Mass Distribution. *Polym. Degrad. Stab.* 2018, 149, 78–84, doi:10.1016/j.polymdegradstab.2018.01.021.

653

654

655

656

659

660

661

662

663

664

665

666

667

668

669

670

671

672

673

674

675

676

677

678

679

680

681

686