

19F solid-state NMR approaches to probe antimicrobial peptide interactions with membranes in whole cells

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Kiran Kumar, Alexandre Arnold, Raphaël Gauthier, Marius Mamone, Jean-François Paquin, et al.. 19F solid-state NMR approaches to probe antimicrobial peptide interactions with membranes in whole cells. Biochimica et Biophysica Acta:Biomembranes, 2024, 1866 (3), pp.184269. 10.1016/j.bbamem.2023.184269. hal-04386127

HAL Id: hal-04386127 https://hal.science/hal-04386127

Submitted on 10 Jan2024

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1	¹⁹ F solid-state NMR approaches to probe antimicrobial peptide interactions
2	with membranes in whole cells
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	Antimicrobial peptide + erythrocyte ghosts ¹⁹ E-ssNMR of intact cells



25 Abstract

To address the global problem of bacterial antibiotic resistance, antimicrobial peptides (AMPs) 26 are considered promising therapeutic candidates due to their broad-spectrum and membrane-lytic 27 activity. As preferential interactions with bacteria are crucial, it is equally important to 28 investigate and understand their impact on eukaryotic cells. In this study, we employed ¹⁹F solid-29 state nuclear magnetic resonance (ssNMR) as a novel approach to examine the interaction of 30 31 AMPs with whole red blood cells (RBCs). We used RBC ghosts (devoid of hemoglobin) and developed a protocol to label their lipid membranes with palmitic acid (PA) monofluorinated at 32 carbon positions 4, 8, or 14 on the acyl chain, allowing us to probe different locations in model 33 and intact RBC ghost membranes. Our work revealed that changes in the ¹⁹F chemical shift 34 anisotropy, monitored through a C-F bond order parameter (S_{CF}), can provide insights into lipid 35 bilayer dynamics. This information was also obtained using magic-angle spinning (MAS) ¹⁹F 36 ssNMR spectra with and without ¹H decoupling, by studying alterations in the second spectral 37 moment (M_2) as well as the ¹⁹F isotropic chemical shift, linewidth, T_1 , and T_2 relaxation 38 times. The appearance of an additional isotropic peak with a smaller CSA, a narrower linewidth, 39 and a shorter T₁ induced by the AMP caerin 1.1, supports the presence of high-curvature regions 40 in RBCs indicative of pore formation, analogous to its antimicrobial mechanism. In summary, 41 the straightforward incorporation of monofluorinated FAs and rapid signal acquisition offer 42 promising avenues for the study of whole cells using ¹⁹F ssNMR. 43

Keywords: Host defense peptides, red blood cells, lipid bilayer, isotopic labeling, in-cell NMR,
 model membranes.

Abbreviations: AMP: antimicrobial peptide; CSA: chemical shift anisotropy; DPPC:
dipalmitoylphosphatidylcholine; DPPC-d₆₂: perdeuterated dipalmitoylphosphatidylcholine; GCMS: gas chromatography coupled to mass spectrometry; L/P: lipid-to-peptide; MAS: magicangle spinning; MFFA: monofluorinated fatty acid; MLV: multilamellar vesicle; PA: palmitic
acid; PA-F(4), (F8) or (F14): palmitic acid fluorinated at positions 4, 8 or 14; S/N: signal-tonoise; ssNMR: solid-state nuclear magnetic resonance; SSB: spinning sideband; T_m: melting
temperature.

53 1. Introduction

Antimicrobial resistance by disease-causing bacteria has become a global health threat with 54 ensuing socioeconomic issues [1,2]. In recent decades, bacterial resistance has increased at an 55 alarming rate, sustained by an extensive usage of antibiotics. There is thus an urgent need for 56 57 alternative solutions to combat bacterial resistance and in this context, antimicrobial peptides (AMPs) and their mimics stand as promising antibiotic molecules. Indeed, their action 58 mechanisms involving bacterial membrane disruption, damage to intracellular biomolecules and 59 other oxidative damages [3], are nonspecific and thus hard to evolve against. More than 24,000 60 61 AMP sequences have been identified so far, both from natural and synthetic origins [4] and only 62 a small number AMPs or their mimics have reached clinical trials [5] or are already commercialized as antimicrobial agents (e.g. bacitracin, gramicidin D, polymyxin B) [5]. In vivo 63 64 stability and low toxicity towards human cells are key factors to access the pharmaceutical 65 market.

Since many AMPs target the bacterial lipid membrane, considerable efforts have been put into 66 characterizing their interaction with cell membranes [6,7]. Solid-state NMR (ssNMR) has proved 67 to be a unique tool to study such interactions at a nanoscopic level. Different NMR-active nuclei 68 such as ¹H, ²H, ¹³C, and ³¹P are used to study the biophysical properties of lipid membranes in the 69 70 presence of AMPs [8-10]. Deuterium is mostly used to characterize lipid chain order and molecular events happening in the core of the bilayer environment, while phosphorus allows 71 72 probing the lipid headgroups or assessing the overall topology of the lipid organization [11,12]. After some initial work on whole cells in the late 70s and early 80s, the field has heavily relied 73 on model membranes until the early 2010s when our group and others reintroduced the use of 74 whole-cell ²H labelling [13-15]. Our laboratory has also developed the use of magic-angle 75 spinning (MAS) combined to ²H ssNMR to reduce the experimental time and ensure that cells 76 remain intact [16]. 77

¹⁹F is an NMR-active nucleus with interesting properties, notably high sensitivity with low background in biological systems, strong dipolar couplings and a large chemical shift range. This sparked interest in its use to study native membranes already in the late seventies and early eighties, with papers showing that the incorporation of fluorinated fatty acid (FA) probes in native membranes was feasible [17,18]. Wild-type and auxotroph *Escherichia coli* strains were successfully labelled [19,20] as well as *Acholeplasma laidlawii* membranes [17], with FAs fluorinated at

different positions, and with only weak perturbations to the membrane [17,19]. McDonough *et* 84 al. [18] also showed that monofluorinated palmitic acids (H-C-F) had a less perturbing effect 85 than difluorinated analogues (F-C-F), and also reported that a single fluorine atom in the bilayer 86 was likely to have a less perturbing effect compared to bulkier electron spin or fluorescent 87 probes. In all these studies, it was shown that an order profile along the acyl chains - similar to 88 the one obtained by ²H ssNMR - could be determined using ¹⁹F ssNMR. However, except for one 89 application from the group of Auger [21,22], the use of ¹⁹F ssNMR has not been extensively de-90 veloped to study AMP interactions from a membrane point of view. Indeed, the high sensitivity 91 and almost complete absence of this nucleus in nature has stemmed a significant body of work 92 on fluorinated peptides [23-26], but not on fluorinated membranes. 93

Early work on ¹⁹F-labelled membranes either by a fluorinated fatty acid or phospholipid fo-94 95 cused on static (no magic-angle spinning, MAS) samples and without ¹H decoupling [17-20,27]. The order profile in the acyl chains was, however, shown to be accessible and described by order 96 97 parameters obtained through spectral fitting and normalizing by a rigid linewidth. More recently, ¹⁹F-labelled phospholipids synthesis was achieved and these lipids were incorporated into model 98 membranes [22,28-31]. ¹⁹F positions spanning the whole acyl chain length and showed by ¹⁹F ss-99 NMR that the ¹⁹F chemical shift anisotropy (CSA) decreased monotonically from the headgroup 100 101 to the end of the acyl chain, thus informing on motion throughout the membrane. They also reported the strong ¹⁹F isotropic chemical shift variation along the acyl chain. Protons were decou-102 103 pled, but MAS was not used in this study.

¹⁹F NMR thus appears as an excellent candidate to investigate membrane-AMP interactions in 104 intact cells. In this work, we reintroduce the use of monofluorinated FAs (MFFAs) to label intact 105 red blood cell (RBC) membranes and establish a ¹⁹F ssNMR approach to study AMP-membrane 106 107 interactions. More specifically, we incorporated into erythrocyte ghosts MFFAs labelled at positions 4, 8 and 14 on the acyl chains (Fig. 1A). Ghosts are erythrocytes from which 108 hemoglobin has been removed, thus reducing potential interference of the paramagnetic heme 109 group with NMR signals. After an initial assessment on a model membrane, we show how both 110 in-cell static and MAS ¹⁹F ssNMR can be exploited to study the pore formation mechanism of 111 112 AMPs using caerin 1.1 - a natural AMPs isolated from the skin secretions of Australian tree frogs as a model. 113

114 **2. Materials and methods**

115 **2.1. Materials**

Caerin 1.1 was purchased from GenScript Corporation (Piscataway, NJ, USA) with >98 % 116 purity. Protonated and deuterated dipalmitoylphosphatidylcholine (DPPC and DPPC-d₆₂) were 117 obtained from Avanti Polar Lipids (Alabaster, AL, USA). Ethylenediaminetetraacetic free acid 118 (EDTA) was bought from Fisher Scientific (Fair Lawn, NJ, USA), while unlabelled palmitic 119 120 acid, deuterium-depleted water, Triton X-100, fatty acid methyl ester mix C4-C24 (FAME mix), [16-[(7-nitro-2-1,3-benzoxadiazol-4-yl)amino] palmitic acid (NBD-PA), 1,1'-dilinoleyl-3,3,3',3'-121 tetramethylindocarbocyanine 4-chlorobenzenesulfonate (Fast DiI) and all other solvents and 122 123 chemicals were purchased from Sigma Aldrich (Oakville, ON, Canada). Fresh horse RBCs 124 packed 100% were purchased from Cedarlane Laboratories (Burlington, ON, Canada). Deionized 18.2 MΩ.cm Milli-Q water was used in all experiments (Millipore-Sigma, Oakville, 125 126 ON, Canada). Fluorine-labelled fatty acids were synthesized following a protocol detailed in the 127 SI section.

128 2.2. Multilamellar vesicles (MLVs) preparation

MLVs were prepared using dry film method as described by Warschawski *et al.* [32]. The lipid mixture (including ¹⁹F-labelled PAs) was dissolved in 1:2 methanol/CHCl₃ solution and dried under nitrogen stream. Remaining traces of organic solvent in the lipid film were removed by high vacuum for at least 2 h. The lipid film was then hydrated with a physiologically-relevant solution of 150 mM NaCl (pH 7.0) prepared with ²H-depleted water. The lipid dispersion was vortexed and freeze-thawed 3 to 5 times (10 min at -20 °C, followed by 10 min above 40-55 °C) and transferred directly into a 4-mm rotor.

136 **2.3. Preparation of ¹⁹F-labelled erythrocyte ghosts**

Erythrocyte ghosts were prepared as described by Kumar *et al.* [9]. Briefly, 3-4 mL of concentrated horse RBCs were suspended in a 40 mL round bottom centrifugation tube with 25 mL isotonic HEPES buffer (20 mM HEPES, 150 mM NaCl, pH 7.4), and centrifuged at 500 g for 5 min at 4 °C. Then three supplementary washes were done with the same buffer until the supernatant became clear. After the final wash, the pellet was resuspended in 20 mL hypotonic HEPES buffer (20 mM HEPES, pH 7.4) and centrifuged at 25,000 g for 40 min at 4 °C (rotor JA-20, Beckman). The supernatant was then removed, and the pellet transferred into new 144 centrifuge tubes, leaving behind the "red button" that contains proteases. Additional washes (3 to145 4) with the same buffer were carried out to obtain a hemoglobin-free white ghosts pellet.

A mixed micelles solution of 0.5 mM Triton X-100 / 0.25 mM of ¹⁹F-labelled PA in isotonic 146 buffer was prepared in a sealed glass vial with three freeze (-20 °C)/thaw (95 °C)/vortex shaking 147 cycles. The white ghost pellet was then resuspended in 320 mL isotonic buffer to which 80 mL 148 of the mixed micelles solution were transferred, and incubated for 15 min at 37°C. The mixture 149 of ghosts and micellized ¹⁹F-labelled PA was then centrifuged at 25,000 g for 20 min at 4 °C in 150 151 different 20 mL aliquots. The excess detergent was washed away twice by centrifugation in isotonic buffer at 25,000 g for 20 min at 4 °C. The ghost pellets were pooled together in a 1.5 mL 152 Eppendorf tube and centrifuged at 20,000 g for 20 min at 4 °C. The pellet was then washed with 153 an isotonic buffer and centrifuged at 100,000 g and 4 °C for 20 min. This concentrated pellet was 154 155 collected and stored at 4°C prior to the experiments and used within 3-4 days.

156 **2.4 Sample preparation for confocal microscopy**

A total of 1 % (v/v) of ¹⁹F-labelled ghosts were labelled with the fluorophore Fast-DiI 157 (1/1000) using a 1 h incubation at 37 °C, followed by pelleting at 16,000 g for 15 min, or with 158 NBD-PA directly incorporated into ghosts along with ¹⁹F-labelled PA as described above in 159 section (2.3). The pellet was resuspended in 500 µL of HEPES buffer then transferred to Sarstedt 160 8-well microscopy slides (300 µL) and left to stand for 30 min to allow ghosts immobilization. 161 Samples were incubated for 1 h at 37 °C after addition of an appropriate amount of caerin 1.1 162 into 1 % (v/v) ghosts, and single frame per second images were generated using a Nikon 163 confocal microscope with a 60x oil-immersion lens. Images were processed using ImageJ 164 software. 165

166 2.5. Fatty acid profile and labelling efficiency

FA analyses were carried out using gas chromatography coupled to mass spectrometry (GCMS) as described by Laydevant et al. [33] following lipid extraction and transesterification. Briefly, lipids were extracted (triplicates) using the Folch protocol[14], then transesterified using 2 mL of H_2SO_4 (2% in methanol) and 0.8 mL of toluene for 10 min at 100 °C. A polar HP-5MS column (30 m length 250 µm diameter 0.25 µm film thickness) was used with an injection volume of 1 µL, and the oven temperature was programmed to heat at 140 °C for 5 min followed by a 4 °C /min ramp up to 300 °C, prior to electron ionization and detection with a Agilent
quadrupole MS. Data acquisition and processing were done with the Chemstation software.

175 2.6. Solid-state NMR and spectral analysis

All ssNMR spectra were recorded using a Bruker Avance III-HD wide-bore 400 MHz 176 spectrometer (Milton, ON, Canada) equipped with a double tuned 4mm-HFX probe and a ¹H-¹⁹F 177 filter. Static ¹⁹F and ³¹P ssNMR spectra were obtained using a phase-cycled Hahn echo pulse 178 sequence, with an inter pulse delay of 35 µs and high-power (50 kHz) ¹H decoupling during 179 acquisition. The 90° pulse length was 4 µs for ¹⁹F ssNMR and 3 µs for ³¹P ssNMR. Data were 180 collected using 2048 points for ¹⁹F ssNMR and 1024 points for ³¹P ssNMR. The recycle delay 181 was 2 s for ¹⁹F ssNMR and 3 s for ³¹P ssNMR. A total of 12 k scans per ¹⁹F spectra for ghosts and 182 183 1 k scans for MLVs were collected, amounting to 7 hours of acquisition for ghosts and 35 mins for MLVs. And a total of 14 k scans per ³¹P spectra for ghosts and 1 k scans for MLVs were 184 collected, amounting to 12 hours of acquisition for ghosts and 50 mins for MLVs. The ³¹P 185 chemical shifts were referenced to the phosphoric acid (H₃PO₄) signal at 0 ppm, while the ¹⁹F 186 187 chemical shifts were referenced to the trifluoroacetic acid (TFA) signal at -76.5 ppm.

²H and ¹⁹F ssNMR experiments were also carried out at 10 kHz MAS frequency. A phase-188 cycled Hahn echo sequence was used for both MAS ²H ssNMR experiments with MLVs samples 189 and MAS ¹⁹F ssNMR analysis of ghosts. ²H ssNMR spectra were recorded with 100 k data 190 191 points, a 4 µs 90° pulse length, rotor-synchronized interpulse delay of 96 µs and recycle time of 500 ms. A total of 1024 scans per spectra were collected, amounting to 8 min acquisition time. 192 ¹⁹F ssMAS spectra were collected with 3 k data points, a 4 µs 90° pulse length, rotor-193 synchronized interpulse delay of 94 µs and recycle time of 2 s. In model membrane sample a 194 total of 512 scans per spectra were collected, amounting to 18 minutes of acquisition time with a 195 recycle time of 2 s and in ghost sample a total of 6k scans per spectra were collected, amounting 196 to 3.5 hours of acquisition time with a recycle time of 2 s. 197

³¹P and ¹⁹F CSA values were determined by line fitting using the Bruker Topspin 4.0.6 software with Sola (Solid Lineshape Analysis) program and values from minimum two replicates are reported. Since the CSA is proportional to the order parameter, a bond order parameter S_{CF} was calculated using equation (1)[19]. The ¹⁹F CSA measured on pure and dry fluorinated PAs at -30 °C was taken as the rigid CSA (\approx 18 ppm):

203
$$S_{CF} = \frac{Observed^{19} F CSA}{Rigid^{19} F CSA}$$
(Equation 1)

²H spectral moment analysis was performed using MestRenova software V6.0 (Mestrelab
 Research, Santiago de Compostela, Spain). Second spectral moments, M₂, were calculated using
 equation (2) [34,35] and values from minimum two replicates are reported.

207
$$M_{2} = \omega_{r}^{2} \frac{\sum_{N=0}^{\infty} N^{2} A_{N}}{\sum_{N=0}^{\infty} A_{N}} = \frac{9 \Pi^{2} \square_{Q}^{2}}{20} \langle S_{CD}^{2} \rangle \quad \text{(Equation 2)}$$

where ω_r is the angular spinning frequency, N is the side band number, and A_N is the area of each sideband obtained by spectral integration, S^2_{CD} is the mean square order parameter, and Ψ_Q is the static quadrupolar coupling constant equal to 168 kHz for a C-D bond in acyl chains. The M₂ value provides a quantitative description of the membrane lipid ordering and is particularly sensitive to the gel-to-fluid phase transition. In the case of ¹⁹F ssNMR, M₂ can also be determined using the first part of equation 2.

214 **3. Results**

215 **3.1. Fluorinated FAs as reporters of membrane structure and order**

216 FAs are molecular probes that can be readily incorporated into biological membranes to study their biophysical properties in native cellular conditions [9]. The incorporation of exogenous FAs 217 into living cells is often easier than other labelling strategies exploiting the cell's biochemistry 218 219 [36,37]. Moreover, free FAs are natural constituents of cell membranes where they are involved in various cellular processes and play a role in regulating cellular functions and membrane 220 221 fluidity [38]. However, incorporating high concentrations of FAs can change the bilayer phase behavior - an effect that has been addressed in previous works [39-42]. For example, the gel 222 (L_{β}) -to-fluid (L_{α}) phase transition temperature (T_m) of a lipid membrane is increased by the 223 presence of FAs, and a coexistence of gel and fluid phases can occur at certain phospholipid/FA 224 225 ratios [41]. Nevertheless, FAs remain accurate reporters of variations in the lipid order in a membrane system [41]. Therefore, before incorporating MFFAs in whole cells' membranes, we 226 227 characterized their effect on the membrane fluidity and as a function of the fluorine atom position on the acyl chain. 228

We used model dipalmitoylphosphatidylcholine (DPPC) membranes to assess the effect of palmitic acid (PA) fluorinated at positions 4, 8 or 14, referred to as PA-F(4), PA-F(8) and PA-F(14) (**Fig. 1A**). The addition of perdeuterated DPPC (DPPC-d₆₂) allowed monitoring the membrane order profile by measuring the variation of the second spectral moment (M_2) calculated from ²H ssNMR spectra as a function of temperature, as described elsewhere [16,43]. Examples of such spectra, can be found in **Fig. S1**.



Figure 1: Effect of fluorination on DPPC-d₆₂/PA model membranes. **(A)** PAs were fluorinated at different positions along the acyl chain. Second spectral moment (M₂) as a function of temperature. M₂ values were obtained from the ²H ssNMR MAS spectra (Fig. S1) of DPPC/DPPC-d₆₂/PA at a molar ratio of (1:1:1) with fluorinated and non-fluorinated PA analogues, **(B)** with PA-F(4), **(C)** PA-F(8), **(D)** PA-F(14) and **(E)** PA. M₂ reports on acyl chain order and values are indicated with standard deviations.

The DPPC/PA mixture results in a complex phase diagram in which, at intermediate temperatures, PA-rich gel and DPPC-rich fluid phases coexist [42]. As shown in **Fig. 1B to 1E**, the order profile as well as T_m of DPPC varies very little when comparing fluorinated and nonfluorinated samples. The average T_m determined by ²H ssNMR is centered at \approx 48-53 °C. Our results are consistent with previous studies of the phase behavior of PA-containing DPPC bilayers [39,42]. In their investigation of the pseudo-binary phase diagram of DPPC/PA mixtures, Inoue *et al.* [42] reported a $T_m \sim 53-54$ °C at 2:1 DPPC/PA molar ratio. The melting of the acyl chain can also be monitored using the ¹H ssNMR intensity of the main-chain CH₂ peak, as well as other lipid resonances (**Fig. S2**). In this case, T_m is centered at ~53-54 °C (**Fig. S3**) - a difference expected between protonated and deuterated phospholipid analogues [44]. Overall, our results show that the insertion of one fluorine atom in the bilayer (positions 4, 8 and 14 on PA acyl chain), has a very weak perturbing effect.

The pioneering work on ¹⁹F-labelled biological membranes from the late 70's and early 80's 254 was technically limited to non-spinning samples and ¹⁹F detection without ¹H decoupling. The 255 single recent work on membranes labelled with monofluorinated phospholipids carried out by 256 Gagnon et al. [22] showed that, with ¹H decoupling, the ¹⁹F lineshape could be used to assess the 257 local order at a given acyl chain position. They also reported a strong ¹⁹F isotropic chemical shift 258 difference along the phospholipid acyl chains - a property that could be exploited to determine 259 260 the acyl chain order in multiply labelled FAs or lipids using MAS. In their study, similarities are observable between ³¹P and ¹⁹F lineshapes, indicating that ¹⁹F could be exploited in a similar 261 fashion to ³¹P to study the formation of fast tumbling structures that can be induced by 262 263 membrane-active AMPs [45]. This is notably the case of AMPs that act through a carpet 264 mechanism [9,45].

We verified whether ¹⁹F ssNMR spectra of model membranes labelled with MFFAs could 265 reflect changes in membrane structure and acyl-chain local order. Fig. 2 compares the static ³¹P 266 and ¹⁹F ssNMR spectra of DPPC membranes incorporating PA-F(4), PA-F(8) or PA-F(14). ³¹P 267 spectra indicate changes in the phospholipids headgroup dynamics while ¹⁹F spectra probe the 268 hydrophobic region of the bilayer. The ³¹P spectra lineshapes in **Fig. 2** are characteristic of lipids 269 270 in a lamellar phase with axial symmetry, and clearly reveal a gel (L_{β})-to-fluid (L_{α}) phase 271 transition. For such systems, this transition can be quantified by measuring the chemical shift anisotropy (CSA), which value is related to the spectrum width and decreases with increasing 272 molecular dynamics. As seen in **Fig. 3A**, characteristic gel phases were detected up to 30 °C with 273 274 a ³¹P CSA value of 36 ppm that gradually dropped to 25 ppm at 45 °C, as the first components of 275 the bilayer reach their phase transition temperature. In comparison, we determined CSA of 36-40 ppm in pure DPPC/DPPC-d₆₂ membranes (**Fig. S4**) at a gel phase (up to 25 °C), which gradually 276

dropped to 29 ppm at the fluid phase (42 °C). The addition of protonated PA shifted T_m to ~45-50 °C, as expected from the DPPC-PA phase diagram [42].

¹⁹F ssNMR lineshapes of the MFFAs show strong similarities with ³¹P ssNMR spectra, in 279 particular in the gel phase with powder patterns characteristic of multilamellar vesicles (Fig. 2). 280 As shown in Fig. 3A, the change in CSA values through the gel-to-fluid phase transition is 281 analogous to the one observed with ³¹P ssNMR. The gel phase is maintained up to 25 °C with ¹⁹F 282 CSA values of 14-16.5 ppm, and a noticeable larger CSA of 16.5 ppm when the ¹⁹F-label is on 283 the 8th carbon position, indicating a higher degree of order at the center of the lipid monolayer. 284 Although ¹⁹F CSA values are smaller than those of ³¹P, they are very sensitive to changes in 285 membrane fluidity. Indeed, when transitioning from the gel to the fluid phase, ¹⁹F CSA values 286 are reduced by 87% on average. Above 25 °C up to about 42 °C, a coexistence of gel and fluid 287 phases is observed in the ¹⁹F spectra (Fig. 2 and 3A-C), consistent with the DPPC-PA phase 288 diagram [42]. In the fluid phase, the ¹⁹F CSA values range from ~1 ppm (PA-F(14)) to 4 ppm 289 (PA-F(4)), depending on the fluorine atom position. This result is in agreement with the 290 conformation and order profile along the acyl chains, with position 4 close to the headgroup 291 undergoing less *trans-gauche* isomerisations than the 14th position, closer to the terminal methyl 292 293 group.



Figure 2: Static ³¹P and ¹⁹F ssNMR spectra of DPPC model membranes incorporating monofluorinated PAs, at a DPPC/DPPC-d₆₂/PA molar ratio of 1:1:1. (A) with PA-F(4), (B) PA-F(8) and (C) PA-F(14). Experiments were performed at different temperatures with ¹H decoupling. Fitted spectra are shown as dashed lines. Average CSA values are indicated, with standard deviation.

Positioning ¹⁹F atoms at different places along the FA acyl chain enables assessing the local order of the bilayer's hydrophobic region, from the interface to the core [19,22,27]. Providing that FAs undergo fast rotational motion along their long axis, this order can be quantified by introducing an order parameter, which describes the dynamic reorientation of a bond with respect to the main axis. This parameter is obtained by normalizing a given NMR-measured quantity in a mobile segment by its value in the static case. In the case of ²H ssNMR for example, this is the quadrupolar coupling for a static C-D bond [43].



Figure 3: Temperature dependence of the static ³¹P (dotted lines) and ¹⁹F (solid and dashed lines) CSA values for DPPC membranes with monofluorinated PAs at a molar ratio of 1:1, calculated from Fig. 2. (**A**) PA-F(4), (**B**) PA-F(8) and (**C**) PA-F(14) with filled symbols corresponding to ³¹P ssNMR and empty ones to ¹⁹F ssNMR. Note the presence of two components detected by ¹⁹F ssNMR in the coexistence region. (**D**) ²H (S_{CD}) and ¹⁹F (S_{CF}) order parameter profile of the acyl chain in ¹⁹F-labelled DPPC membranes as a function of the fluorine atom position, at 45 and 65 °C, with standard deviation.

In early ¹⁹F ssNMR studies, spectra were recorded without ¹H decoupling; the CSA and 315 dipolar couplings interactions could therefore not be isolated, and were estimated by computer-316 assisted line fitting [19,27]. When employing high-power ¹H decoupling, it is reasonable to 317 assume that only the CSA persists, simplifying the determination of the order parameter [46]. 318 Here, we introduced a C-F bond order parameter, S_{CF}, defined as the ratio between the 319 experimentally-measured CSA and its value for an immobile C-F bond (Equation 1). By 320 definition a rigid molecule with an all-trans conformation has an order parameter of $S_{CF} = 1$ 321 while this value is 0 in the case of fast isotropic motions. The ¹⁹F CSA tensor is not aligned along 322 the C-F bond, and not necessarily axially symmetric, but in the fluid phase it becomes pseudo-323 axially symmetric with respect to the FA's long axis, as reflected in the spectra. Fig. 3D shows 324 the S_{CF} bond order parameters for PA-F(4), -F(8) and -F(14) in model membranes. As a means of 325 326 comparison, the S_{CD} order parameter of DPPC-d₆₂ measured in the same sample is also presented (Fig. S5). The similarity of S_{CF} and S_{CD} values suggests that the relative orientation of the ¹⁹F 327 328 CSA tensor with respect to the molecular frame is factored out when the dry FA at -30°C is used 329 as the reference spectrum. The evolution of the two order parameters adequately reflect dynamic 330 changes with carbon position and temperature, confirming that the S_{CF} of MFFAs can be efficiently employed to measure the order profile along the acyl chains. 331

332 In the case of intact cells, measuring the acyl chain order at each carbon position is usually impossible. When perdeuterated lipids are used, the distribution of quadrupolar couplings, which 333 334 reflects the overall dynamics in the membrane, can be assessed by measuring the second spectral moment, M₂, of the ²H ssNMR spectra [43]. As we showed in a previous work, M₂ can also be 335 determined under MAS conditions, which enhances the signal-to-noise (S/N) ratio and decreases 336 the acquisition time [16]. Fig. 4 shows ¹⁹F ssNMR spectra obtained in the gel and fluid states 337 338 under 10 kHz MAS. The resulting increase in S/N ratio with respect to the static spectra is *ca*. \approx 339 5-6-fold, corresponding to a reduction in acquisition time by a factor 25-36 for the same S/N ratio. This substantial gain in sensitivity enables recording spectra without ¹H decoupling (Fig. 340 S6) - an important alternative considering that not all laboratories are equipped with probes 341 capable of detecting ¹⁹F while decoupling ¹H. Additionally, it allows working with significantly 342 lower proportions of FAs in the membrane, as demonstrated in Fig. S7. At an MAS frequency of 343 10 kHz, the ¹⁹F CSA (ranging between 0.3 kHz in the fluid phase and 6.5 kHz in the gel phase) 344 should be entirely averaged out. Note that the CSA varies linearly with the static magnetic field 345

strength, necessitating faster spin rates at higher fields. While spectra are dominated by an 346 isotropic peak (δ_{iso}), 1-2 spinning sidebands (SSBs) are observed even in the ¹H-decoupled 347 spectra albeit with lower intensity as compared to non-decoupled spectra. In principle, ¹H-¹⁹F 348 dipolar couplings should be eliminated by the strong ¹H decoupling, and the intermolecular ¹⁹F-349 ¹⁹F dipole-dipole interactions significantly reduced by MAS. However, it is possible that residual 350 dipolar couplings (¹H-¹⁹F and/or ¹⁹F-¹⁹F) remain, or that internal acyl chain motions result in 351 incomplete MAS averaging. In both cases, a change in membrane fluidity would lead to a 352 corresponding alteration in the SSBs' intensities - as is indeed observed when temperature is 353 varied. This feature can be exploited to assess the overall membrane dynamics, as will be shown 354 below. It would be interesting to record the spectra at different magnetic fields in order to 355 determine the relative contribution of field-dependent interactions, such as CSA, to the spinning 356 357 sideband intensities.

The isotropic chemical shift (Table S1) is sensitive to the ¹⁹F position on the FA chain, and 358 359 also to the membrane lipid phase for positions 8 and 14 (Fig. 4 and Fig. S6). Indeed, two peaks are observed in the gel-fluid coexistence region (between 25 °C and 52 °C) separated by up to 360 361 0.8 ppm for these positions. The isotropic resonances are separated by up to 1.1 ppm between the 362 gel (5 °C) and fluid (60 °C) phases (Fig. 4A). These peaks are easier to distinguish in the ¹H-363 decoupled spectra where they are almost baseline-resolved. As mentioned, the presence of SSBs allows measuring the ¹⁹F spectral moment value (M₂), which reports on the membrane fluidity. 364 365 Fig. 4B shows the evolution of M₂ as a function of temperature for both ¹H-coupled and decoupled spectra. Spinning sidebands were more intense when no decoupling was applied and 366 the M₂ values varied between 38 10⁶ s⁻² in the gel phase down to 2-3 10⁶ s⁻² at 60 °C (fluid 367 phase) for ¹⁹F at positions 4 and 8. When the 14th carbon was ¹⁹F-labelled (closer to the terminal 368 methyl group) the M₂ value was smaller in the gel phase (25-32 10^6 s⁻²) as compared to PA-F(4) 369 and PA-(F8), but their M₂ values in the fluid phase were all very similar (2-3 10⁶ s⁻²). When ¹H 370 decoupling was applied, the SSBs' intensity was significantly reduced, as previously mentioned. 371 The M_2 values dropped to 6.8 and 11.7 10^6 s⁻² in the gel phase for positions 8 and 14, 372 respectively. In the fluid phase, the SSBs almost disappeared and the M₂ values were further 373 reduced to 1 10^6 s⁻². Overall, our results show that M₂ values vary sufficiently between the gel 374 and fluid phases for both ¹H-coupled and decoupled spectra to be used to assess membrane 375 fluidity and detect phase transition temperatures. When feasible, we recommend recording the 376



377 spectra with and without ¹H decoupling. In favorable cases, ¹H decoupling might allow 378 determining the gel phase proportion by a simple integration of the isotropic lines since the ¹⁹F 379 isotropic chemical shift varies with membrane melting. On the other hand, the second spectral 380 moment variation is larger on spectra without ¹H decoupling and membrane fluidity can thus be 381 assessed from the M₂ with a higher precision.

382

Figure 4: (A) MAS (10 kHz) ¹⁹F ssNMR spectra of DPPC/PA-F(8) model membranes, with change in isotropic chemical shift ($\Delta \delta_{iso}$) values as function of temperature. **(B)** Second spectral

moment with standard deviation, calculated from the ¹⁹F ssNMR MAS spectra (Fig. S6) of DPPC/DPPC-d₆₂ with fluorinated PAs analogues in a 1:1:1 molar ratio. Experiments were carried out at different temperatures with (black, dashed line) and without (grey, solid line) ¹H decoupling and symbols are (\bigcirc): PA-F(4), (\triangle): PA-F(8) and (\bigtriangledown): PA-F(14).

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390 Since MAS considerably enhances the signal intensity, we further evaluated the membrane dynamics in a site-resolved fashion by measuring the spin-lattice (T_1) and spin-spin relaxation 391 (T_2) times at each position by ¹⁹F ssNMR (**Table S1**). Both T_1 and T_2 values increased as the ¹⁹F 392 atom position moved from the headgroup region to the center of the bilayer. The same trend was 393 observed with increasing temperature (data not shown), confirming that T₁ and T₂ values indeed 394 report changes in dynamics. The concomitant increase in T₁ with dynamics indicate that the 395 correlation times of the motions contributing to longitudinal relaxation are in the nanosecond 396 timescale, i.e. faster than the inverse of the Larmor frequency ($\tau_{\rm C} < 1/\omega_0 \approx 3$ ns). Motions with 397 this type of correlation times are most likely trans-gauche isomerisations. If such motions are 398 399 affected by the interaction with AMPs, a corresponding change in T₁ should thus be observed.

400 **3.2. Monofluorinated fatty acids to study whole cell membranes**

Once the ¹⁹F ssNMR methods were established in model membranes, we assessed their 401 402 application to study membranes in intact cells. Early work on bacterial membranes established the possibility of labelling intact bacteria with fluorinated FAs [19,27]. Here, we focused on 403 erythrocyte ghosts, which we had successfully labelled with deuterated PA (PA- d_{31}) in a previous 404 study [9]. Using a similar protocol, we achieved the incorporation of MFFAs into ghosts with 405 labelling levels up to 30%, as determined by GC-MS (Fig. S8). Moreover, fluorescence 406 microscopy images of the ¹⁹F-labelled ghosts obtained with the lipophilic tracer FAST-DiI and 407 the NBD-PA fluorophore, which is structurally similar to PA, show that they retain their original 408 RBC morphology (Fig. 5A). 409

Fig. 5B shows static ¹⁹F ssNMR spectra of RBC ghosts incorporating MFPAs at 20 °C, which are all characteristic of lipids in a lamellar phase. ³¹P ssNMR spectra are also shown for comparison. Note that all the different phospholipid headgroups in ghosts, which contain ca. 43% PC, 23% phosphatidylethanolamine, 6% phosphatidylserine and 22% sphingomyelin, contribute to the broad ³¹P NMR spectra [9]. As was the case in model membranes, ¹⁹F spectra are characteristic of the acyl chain order at each carbon position. Indeed, the ¹⁹F CSA value at the 416 4^{th} position close to PA's headgroup is 7.7 ppm, which is smaller than at the 8^{th} position (10.1 417 ppm). The small CSA value of 4.5 ppm at position 14 reflects the high mobility in the middle of 418 the bilayer. As established in section 3.1, we used the CSA values to determine the order 419 parameter as a function of the fluorine atom position, and found S_{CF} values of 0.43, 0.56 and 0.27 420 for positions 4, 8 and 14, respectively. the higher order determined for PA-F(8) is in line with the 421 characteristics expected in a membrane with a high cholesterol content, where the central portion 422 of the acyl chains is known to be more rigid compared to the chain's beginning or end [47].

¹⁹F ssNMR MAS spectra (**Fig. 5B**) show that the trend of the isotropic linewidth as a function 423 of the fluorine atom position is similar to that of the CSA values. The more rigid 8th position has 424 the broadest full width at half maximum (FWHM=180 Hz), followed by the 4th position (158 Hz) 425 then the more mobile 14th position (103 Hz). We further explored the membrane dynamics by 426 monitoring the ¹⁹F relaxation times of the MFFAs' isotropic resonance in ghosts (which we will 427 refer to as P₁ peak). As reported in Table 1, T₁ values vary with the ¹⁹F atom position, similarly to 428 model DPPC membranes. Values of 328, 414 and 607 ms were respectively determined for 429 positions 4, 8 and 14, i.e., shorter than in DPPC bilayers. This suggests that fluorinated PA 430 431 probes experience an increase in motions on the nanosecond timescale when incorporated in the ghost membranes, revealing the higher fluidity of ghost membranes as compared to those formed 432 by DPPC. T₂ values in ghosts, which range from 1.5 ms for F(4) to 2.4 ms for F(14), do not differ 433 434 significantly from those determined in model membranes, implying that millisecond timescale motions are similar in these two types of bilayers. The T_2 value for PA-F(8) is the lowest (1.3 435 ms), consistent with the lowest FWHM value. 436

The isotropic peaks observed on ¹⁹F ssNMR MAS spectra are sufficiently well separated to be resolved in a mixture of the three FAs, or if a triply labelled FA was prepared. The acyl chain order could then be studied in a single experiment while benefiting from the high sensitivity offered by MAS, providing that a triple labelling would not affect the membrane assembly. Altogether, our results show that both the membrane structure and dynamics are measurable by labelling whole cells such as RBC ghosts with MFPAs. Their physicochemical properties can thus be investigated, enabling the study of their interaction with membrane-active molecules.



Figure 5: (A) Confocal fluorescence microscopy images of erythrocyte ghosts labelled with monofluorinated PAs. Secondary fluorophore labelling was achieved using lipophilic tracer FAST DiI (red) after fluorination, NBD-PA (green) labelling was achieved along with the fluorination protocol. Scale bars are 5 µm and Z projection image of labelled ghosts with an average aspect ratio is indicated **(B)** Static ³¹P (top), static ¹⁹F (middle) and ¹⁹F MAS (10 kHz) ssNMR spectra of erythrocyte ghosts labelled with monofluorinated PAs, recorded at 293 K with ¹H decoupling. CSA and FWHM values are indicated, with standard deviation. Note the significant improvement in S/N ratio between ³¹P and ¹⁹F NMR.

455 **3.3. Interaction of caerin 1.1 with ¹⁹F-labelled erythrocyte ghosts**

We verified the applicability of our ¹⁹F ssNMR approach to study the interaction mechanism 456 of an AMP with intact cells. To do so, we used caerin 1.1 - a 25 amino-acid cationic peptide 457 with the following primary sequence: GLLSVLGSVAKHVLPHVVPVIAEHL-NH₂. This AMP 458 was shown to create pores in RBC ghosts labelled with deuterated PA (PA- d_{31}) by ³¹P and ²H 459 ssNMR in our previous work [9]. Fig. 6A-C present the static ¹⁹F ssNMR spectra of the ghosts 460 labelled at positions 4, 8 and 14, for different lipid-to-peptide (L/P) molar ratios. The ³¹P ssNMR 461 spectra are available in the SI (Fig. S9) for comparison. The ¹⁹F spectra show the presence of a 462 central resonance at approximately 180 ppm with a linewidth ranging from 300 to 400 Hz in the 463 464 presence of caerin 1.1. Rather than resulting from small fast tumbling objects, the breadth of this 465 line could for example indicate the presence of high-curvature regions in the ghost membranes 466 (see below). ¹⁹F CSA values decrease as a function of peptide concentration for ghosts containing PA-F(8) and PA-F(14), indicating an increased mobility at these positions. An opposite trend is 467 468 observed for ghosts labelled with PA-F(4), suggesting a proximity of the AMP to the headgroup region of the membrane with local motion hindrance. 469

The membrane perturbing effect revealed by ¹⁹F static NMR (and ³¹P) correlates well with 470 caerin-induced RBC leakage assays [9]. Indeed, the appearance of the isotropic peak in the 471 472 ghosts and the RBC leakage occur at roughly the same L/P ratios. These observations are in good agreement with our previous work [9] as well as fluorescence microscopy images in Fig. S10, 473 which suggest the formation of pores in the ¹⁹F-labelled ghosts membrane. It should be noted that 474 the L/P ratio refers to the total number of molecules present in the whole sample; the local ratio 475 476 at the actual membrane is most certainly smaller due to AMP partitioning between the membrane and buffer. 477

Fig. 6D-F show the ¹⁹F MAS ssNMR spectra of fluorinated ghosts with different 478 concentrations of caerin 1.1. As was the case in model membranes, the anisotropic interactions 479 are conveniently averaged by MAS even at a moderate frequency of 10 kHz. Without peptide, 480 481 the area of each broad spectrum is concentrated into a single narrow line (noted P_1) with an 482 associated increase in the S/N ratio. Interestingly, when caerin is added, an additional resonance (P₂) emerges and its intensity increases with peptide concentration. This peak is observed with all 483 MFFA analogues. Both P₁ and P₂ resonances are best resolved when the ¹⁹F atom is located at 484 positions 8 and 14. The appearance of two isotropic peaks upon caerin addition suggests that the 485

486 MFFAs are in two different environments, which we quantified by integrating P_1 and P_2 lines. As 487 shown in **Fig. 7 (Table S2)**, the P_2 contribution to the MAS ¹⁹F ssNMR spectra is in excellent 488 agreement with the isotropic contribution to the static ¹⁹F and ³¹P ssNMR spectra.





Figure 6: Static **(A-C)** and MAS **(D-F)** ¹⁹F ssNMR spectra of erythrocyte ghosts labelled with PA-F(4) **(A and D)**, PA-F(8) **(B and E)** and PA-F(14) **(C and F)**, exposed to different concentrations of caerin 1.1. Spectra were recorded at 293 K with ¹H decoupling. CSA and FWHM values are indicated, with standard deviation. Note the changes of linewidth with fluorine position.



Figure 7: Isotropic contribution in the static ³¹P (black) and ¹⁹F (grey) ssNMR spectra compared
 to the P₂ contribution (white) in the MAS (10 kHz) ¹⁹F ssNMR spectra as a function of caerin 1.1
 concentration in erythrocyte ghosts containing monofluorinated PAs.

500 We then explored dynamics changes in the membrane in the presence of AMPs by measuring T_1 and T_2 relaxation times under MAS. At all fluorine atom positions, the T_1 relaxation time of 501 502 the P₂ resonance is consistently shorter than that of P₁ by approximately 100 ms, suggesting that 503 the presence of interacting caerin increases the motions on the nanosecond timescale that contribute to T_1 relaxation. On the other hand, the T_2 values are similar for P_1 and P_2 . When 504 comparing the T_1 and T_2 values of the ghosts/AMP systems with those of MFFAs incorporated in 505 fast reorienting isotropic dodecylphosphocholine (DPC) micelles, Table 1 shows that P₂'s T₁ 506 507 values are slightly lower, and T₂ values are an order of magnitude smaller in the AMP-perturbed ghosts. 508

509 The presence of an isotropic peak in the static spectra, a small change in T_1 and strong reduction in T₂ have been shown, in the case of ³¹P ssNMR, to be characteristic of high-curvature 510 regions such as cubic or hexagonal phases [48,49]. In full analogy, we thus assign the isotropic 511 512 peak in our static 19 F spectra and P₂ peak in the MAS spectra to high-curvature regions induced by the AMP. Our NMR and microscopy results rule out a carpet mechanism, which leads to the 513 formation of micelles. A contrario, the presence of high curvature regions could be due to the 514 formation of toroidal pores for example, a result consistent with our leakage assay and 515 fluorescence microscopy results [9]. Control experiments (not shown) rule out the possibility of 516 hexagonal phase formation. 517

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- 519

Table 1: Average ¹⁹F chemical shifts and ¹⁹F T₁ values in different systems incorporating
 monofluorinated PA probes, with standard deviation. Values calculated from MAS ¹⁹F ssNMR
 spectra recorded at 293 K with ¹H decoupling.

19 F	DPC Micelles			DPPC/DPPC-d ₆₂ /PA			Ghosts* (P ₁)			Ghosts + Caerin 1.1 (P ₂)		
analogue	δ_{iso}	T_1	T_2	δ_{iso}	T_1	T_2	δ_{iso}	T_1 (ms)	T_2	δ_{iso}	T_1	T_2
	(ppm)	(ms)	(ms)	(ppm)	(ms)	(ms)	(ppm)		(ms)	(ppm)	(ms)	(ms)
	-180.4	382	23.8	-180.9	611	1.7	-180.1	328	1.5	-179.7	226	1.1
PA-F(4)		(53)	(14.6)		(47)	(0.2)		(27)	(0.3)		(24)	(0.1)
DA E(9)	-178.2	487	18.4	-179.7	760	2.2	-179.2	414	1.3	-178.6	310	0.9
PA-F(0)		(18)	(8.1)		(163)	(0.1)		(4.2)	(0.4)		(77)	(0.1)
$\mathbf{D}\mathbf{A} = \mathbf{E}(1\mathbf{A})$	-179.9	731	24.9	-181.3	883	2.2	-180.3	607	2.4	-179.9	472	1.8
PA-F(14)		(10)	(5.8)		(54)	(0.1)		(19.1)	(0.1)		(69)	(0.2)

523 P_1 has the same isotropic chemical shift and T_1 values, without and with caerin 1.1.

524 4. Discussion

525 The objective of this study was to leverage the labelling of whole cells with MFFAs to investigate specific interactions using ¹⁹F ssNMR. Fluorine offers several key advantages, 526 including high sensitivity, which reduces acquisition time (a critical factor for whole-cell and in 527 528 vivo studies), low occurrence in biological samples, which minimizes background signals, and a broad chemical shift range, allowing for sensitivity to even minor structural changes. Fluorine is 529 an isostere of hydrogen that alters the hydrogen bonding properties and dipole moment of the 530 molecule in which it is incorporated [50,51]. As a consequence, fluorination can change the 531 polarity and hydrophobicity of a molecule to an extent that will depend on the location and level 532 of fluorination [50]. Here, we showed, as have others in similar systems, that the incorporation of 533 534 MFFAs has only minor effects on the membrane structure [19,22,27], and that it can report changes in the bilayer's dynamics that are of great relevance to the study of membrane-protein 535 536 interactions [19,27,33,52]

Using model DPPC membranes incorporating PA fluorinated at three different positions along the acyl chains, we established that changes in the ¹⁹F CSA, isotropic chemical shift and linewidth, as well as T_1 and T_2 relaxation times, can inform on changes in the lipid bilayer dynamics. To further characterize these changes, we introduced the order parameter S_{CF} as well as M_2 measurements, similar to the widely used ²H ssNMR study of perdeuterated lipids. We then showed that ¹⁹F ssNMR is a valuable tool to describe the membrane state in whole erythrocyte ghosts and to monitor AMP-membrane interactions. The results obtained here using our ¹⁹F ssNMR methodology are consistent with previous work reporting a pore formation mechanism for caerin, thus validating our approach.

546 ¹⁹F static spectra enabled the distinction between lamellar and non-lamellar phases with a higher sensitivity than ³¹P ssNMR. Preliminary data also show that hexagonal phases can be 547 identified by ¹⁹F ssNMR (data not shown). In addition, ¹⁹F ssNMR with appropriate labelling 548 could replace ³¹P ssNMR in membranes deprived of phospholipids, such as plant membranes for 549 example. The spectra can be interpreted in terms of an order parameter reflecting the acyl chain 550 order in a similar way to static ²H ssNMR. The use of MAS, combined to the large chemical shift 551 range of ¹⁹F NMR enables distinguishing lamellar and high-curvature regions in a spectrum 552 where anisotropic interactions are fully averaged and all the intensity is concentrated into sharp 553 peaks. This greatly reduces the experimental time and also opens the possibility of measuring 554 555 relaxation times, allowing to probe dynamical properties of the molecules.

In summary, by adequately selecting the ¹⁹F position on the acyl chain, ¹⁹F ssNMR can 556 provide the same information as ²H and ³¹P ssNMR in lipid systems. Typically, knowledge of 557 acyl chain order necessitates the measurement of anisotropic interactions such as CSA, 558 559 quadrupolar couplings, or dipolar couplings [32,35]. While the simplest approach may involve studying static samples, it is conceivable to reintroduce these anisotropic interactions, thereby 560 561 harnessing the advantages offered by both MAS and static spectra [32]. In this regard, the broader range of ¹⁹F chemical shifts, compared to ³¹P and even more so to ²H, becomes 562 563 advantageous. Additionally, information obtained from anisotropic interactions can be readily complemented by relaxation time measurements, as demonstrated in this study. 564

565 **5. Conclusions**

In this work, we showed how ¹⁹F ssNMR is a useful tool to understand the interaction of 566 AMPs with model membranes and, more importantly, whole cells. The incorporation of 567 568 fluorinated FA probes placed at different depths in the membrane enables mapping changes in dynamics at various locations in the membrane hydrophobic core, thus complementing other 569 570 biophysical methods such as ³¹P ssNMR to describe membrane-peptide interactions. Considering its location in the middle of the acyl chain, PA-F(8) would be the preferred MFFA if a single 571 position needs to be chosen. ¹⁹F ssNMR experiments can be carried out with static samples or 572 under MAS, with and without ¹H decoupling, each of these approaches having its advantages. 573

The results presented in this work should guide the choice of a ¹⁹F ssNMR experiment to investigate lipid membranes in both model and cell systems, whether it concerns membrane structure, dynamics, or a peptide mode of action. We exemplified our methodology with the study of the cationic AMP caerin 1.1 as it interacts with erythrocyte membranes. The labelling strategy presented in this work should be widely applicable to other cells such as bacteria.

579 6. Acknowledgements

This research was funded by the Natural Sciences and Engineering Research Council (NSERC) of Canada (grant RGPIN-2018–06200 to I.M.) and the Centre National de la Recherche Scientifique (UMR 7203 to D.E.W.). K.K. would like to thank the Quebec Network for Research on Protein Function, Engineering, and Applications (PROTEO) - strategic cluster of the Fonds de recherche du Québec – Nature et technologies (FRQNT) - for the award of a scholarship. The authors thank Dr. Mathew Sebastiao (Université du Québec à Montréal) for technical assistance and Pierre Audet (Université Laval) for the loan of the HFX probe.

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