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Benjamin Le Delliou, Olivier Vitrac, Sandra Domenek. Nucleation of biodegradable poly(hydroxybutyrateco-hydroxyvalerate) and poly(butylene succinate-co-adipate) blends. JCAT 53 - 53ième Journée de Calorimétrie et Analyse Thermique, May 2023, PALAISEAU, France. https://jcat53.sciencesconf.org/477018. hal-04375416

HAL Id: hal-04375416 https://hal.science/hal-04375416

Submitted on 5 Jan 2024

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Nucleation of biodegradable poly(hydroxybutyrateco-hydroxyvalerate) and poly(butylene succinate-co-adipate) blends

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Introduction

Objective: Nucleation of PHBV in biobased and biodegradable polymer blends for the packaging of frozen food



Polyhydroxyalkanoates (PHAs) are thermoplastic polyesters produced as carbon and energy storage compounds by microorganisms under nutritional stress. Short-chain-length PHAs, mainly represented by poly(3-hydroxybutyrate) P(3HB) and the copolymer poly(3hydroxybutyrate-co-3-hydroxyvalerate)P(3HB-co-3HV) are already commercially available. PHBV can be produced using carbon sources from waste or by-products of agriculture and food industries. This would further promote circular economy because local food waste could be used to make food containers. PHBV is a stiff, brittle, highly crystalline polymer with low thermal stability and a narrow processing window. One strategy that can be adopted to overcome these drawbacks is based on physical melt-blending using compatibilizers. We developed PHBV/PBSA blends compatibilized with dicumyl peroxide (DCP). PHBV forms large spherulites during annealing treatment, which are not beneficial for its mechanical properties. Therefore, nucleating agents, such as boron nitride, BN, are commonly used. They increase the nucleation density and help the yield more and smaller spherulites. Here the nucleation of PHBV in compatibilized blends with PBSA with BN and its consequences were investigated.

Blend processing

Development of the compatibilized blend using an internal mixer

Addition of BN as nucleating agent



Results

Thermal stability of PHBV/PBSA+ 0.1 phr DCP with BN



Analysis of non-isothermal crystallization kinetics







Morphology of PHBV/PBSA/DCP blends

PHBV/PBSA 70/30

PHBV/PBSA 70/30 + 0.1 phr DCP





DCP decreased size of PBSA nodules in PHBV.

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Morphology of fracture surfaces after tensile testing



PHBV/PBSA 70/30 + 0.1 phr DCP + 0.5 phr BN

- Fracture surfaces show stretched fibers. • DCP increased surface adhesion.
- Two distinct glass transition temperatures: **PBSA and PHBV**
- T_g almost unchanged whatever blend ratio
- -> PHBV and PBSA are immiscible.
- At the tested concentrations, BN had no supplementary nucleating effect on the PHBV phase.

Cooling rate (°C/min)

Mechanical properties of the films in service situation





The use of BN did not change the mechanical properties.

Conclusions

This study investigated the *in-situ* compatibilization and nucleation of PHBV/PBSA blends at a weight ratio of 70:30 using the free radical initiator dicumyl peroxide (DCP) and the nucleant BN. Batch mixing was used for the fabrication of different formulations and test samples were produced by thermo-compression. PHBV and PBSA are immiscible. The addition of DCP induced reaction between PHBV and PBSA leading to the formation of a cross-linked PHBV-g-PBSA structure acting as compatibilisant. The size of the PBSA nodules in the compatibilized blends decreased. The use of DCP had no impact on the thermal properties of the polymers. Nucleation by BN did not affect the half-time of crystallization. The results of tensile testing showed that the elongation at break slightly was increased with addition of DCP. The observation of the fracture surface showed some stretched fibers. However, no evolution of the elastic modulus or the ultimate stress was found. The use of BN in the tested concentration was disadvantageous for the mechanical properties. In conclusion, the compatibilizing of PHBV/PBSA blends with DCP was promising, but no advantage of supplementary nucleation for the properties of the blends was observed.



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