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Conception and Evaluation of a Library of Cleavable Mass Tags for Digital Polymers Sequencing

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Abstract: A library of phosphoramidite monomers containing a main-chain cleavable alkoxyamine and a side-chain substituent of variable molar mass (i.e. mass tag) was prepared in this work. These monomers can be used in automated solid-phase phosphoramidite chemistry and therefore incorporated periodically as spacers inside digitally-encoded poly(phosphodiester) chains. Consequently, the formed polymers contain tagged cleavable sites that guide their fragmentation in mass spectrometry sequencing and enhance their digital readability. The spacers were all prepared via a seven steps synthetic procedure. They were afterwards tested for the synthesis and sequencing of model digital polymers. Uniform digitally-encoded polymers were obtained as major species in all cases, even though some minor defects were sometimes detected. Furthermore, the polymers were decoded in pseudo- $MS³$ conditions, thus confirming the reliability and versatility of the spacers library.

Introduction

Synthetic digital polymers are a new class of functional sequence-defined macromolecules enabling molecular information storage.^[1] They are typically synthesized by a sequence-controlled multistep process^[2] and deciphered by a

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sequencing tool.^[3] Since the first report in 2014 ,^[4] about 20 different families of synthetic digital polymers have been described.^[1c,5] The relevance of such information-containing polymers in applications such as cold data storage, anticounterfeiting technologies, cryptography, materials traceability and plastic recycling has been evidenced.^[6] Despite this recent progress, synthetic polymers are still less explored than DNA for data storage.^[7] Yet, one major advantage of synthetic digital polymers is that their storage properties can be controlled by macromolecular design.^[8] For example, the molecular structure of synthetic polymers can be tailored to facilitate sequencing by nanopore sensing,^[9] controlled chain degradation^[10] or mass spectrometry.[11]

For mass spectrometry, which is currently the most used sequencing method for synthetic polymers, it was recently reported that digital poly(phosphodiester)s with a storage capacity as high as 440 bits/chain can be decoded in a routine instrument.[12] To achieve such a record, the digital chains were carefully engineered to undergo programmed fragmentations in the mass spectrometer. As reported earlier and as schematized in Figure 1 ,^[13] the inclusion of periodically-distributed cleavable alkoxyamine sites inside a digital poly(phosphodiester) facilitates its decoding. This is due to the fact that low dissociation energy NO–C bonds can be selectively cleaved in the presence of phosphate repeat units. $[14]$ For example, when these weak links are placed between bytes, the tandem mass spectrometry (MS/ MS) fragmentation of the macromolecule leads to a library of cleaved oligomers (Figure 1a). Yet, in order to retrieve the positioning of each byte in the initial sequence, a mass tag with a specific MS signature is incorporated in all bytes with the exception of the first one.^[13] Each cleaved byte is then subjected to a further fragmentation $(MS³)$ and decoded. Ultimately, the entire sequence of the initial polymer can be reconstructed either manually or using a specific software.^[15] Main-chain NO–C bonds also influence the stability of digital polymers, which may degrade thermally. This interesting property can be exploited for the design of erasable or editable polymers.[5a]

Digital poly(phosphodiester)s are synthesized by phosphoramidite polymer chemistry (PPC), a technique that allows the use of classical phosphoramidite nucleosides^[16] but also of a wide variety of non-natural monomers.^[17] They are usually binary-encoded using two different monomers 0 and 1, leading to propyl- and 2,2-dimethylpropyl- phospho-

Figure 1. (a) General concept used in previous works for the mass spectrometry sequencing of digital poly(phosphodiester)s.[13] The polymers contain periodically-distributed cleavable spacers. Thus, in MS/MS conditions, they break into a library of predictable coded fragments. Except the first one, all fragments contain a mass tag that allows their identification. They are sequenced individually in pseudo MS³ conditions and ultimately the complete information sequence can be reconstructed, either manually or using a software. (b) Molecular structure of previously-reported phosphoramidite spacers allowing the incorporation of cleavable alkoxyamine spacers in the digital poly(phosphodiester) chains.[13,18] (c) Molecular structure of the phosphoramidite spacers prepared in the present work that contain both a mass tag and a cleavable site.

diester repeat units, respectively (Figure 1a).^[5b] In our original report,^[13] the phosphoramidite spacer \mathbf{L}_1 (Figure 1b) and commercial phosphoramidite nucleosides were respectively used to install alkoxyamine cleavable sites and mass tags in poly(phosphodiester) chains. However, the spacer **L1** leads to reactive radical fragments that form intense secondary peaks series and prevent the use of a decoding software. It was therefore recently replaced by the spacer L_2 (also known as RISC2), which leads to cleaner fragmentations and therefore allowed automated sequencing, as reported recently in this journal.^[18]

Yet, the use of nucleoside mass tags was still required. In other words, when using a simple binary alphabet based on two different comonomers,^[5b,8] 10 phosphoramidite buildingblocks are required to synthesize a byte (i.e. 1 spacer, 1 mass tag and 8 coded monomers). One important issue in the field of digital polymers is atom economy, as exemplified by the recent development of expanded coding alphabets.[5f,i,19] In this context, it would be simpler and more useful if the alkoxyamine group and the mass tag could be combined in a single molecule. In the present communication, we report the design, synthesis and use of phosphoramidite spacers **2**– **10** (Figure 1c) containing both a cleavable site and a mass tag.

involving a nucleophilic substitution on a secondary amine. It relies on tagged TEMPO intermediates that are derived in two steps from 4-oxo-TEMPO (Scheme 1). First, the intermediate 4-((3-hydroxypropyl)amino)-TEMPO **T1** was prepared. The secondary amine of **T1** was then reacted with bromo-derivatives to afford the tagged intermediates **T2**- **T10** in good yields. The mass tags (i.e. *N*-substituents) of these derivatives were carefully selected in order to fulfil the

(T4) T = $CH_2-C_6H_4-CF_3$ (T7) T = CH_2 -pyrenyl

 $(T2)T = CH_2 - C_6H_5$

(T3) T = $CH_2-C_6H_4F$

(TEMPO)-based alkoxyamine.[13] Another important design in \mathbf{L}_2 is the presence of a rigid aromatic linker that prevents back-biting radical reactions and subsequent rearrangements after MS fragmentation.[18] All these features were kept in the structures prepared in this work. Different types of chemistries can be considered for including mass tags in such spacers. Still, the chosen chemistry shall be easy and versatile to allow incorporation of a wide variety of substituents. To this end, we developed a synthetic route

Results and Discussion

The design of the cleavable mass tags was derived from $L₂$, which contains both a dimethoxytrityl (DMT)-protected OH group and a reactive phosphoramidite moiety, as requested for PPC.[17c] Furthermore, in order to induce controlled MS fragmentations, \mathbf{L}_2 also contains a tetramethylpiperidinyloxy

(T5) T = $CH_2-C_6H_4-C_6H_5$

(T6) $T = CH_2$ -naphthyl

(T8) T = $CH_2-C_6H_3-(CF_3)_2$

 $(T10)T = CH_2-C_6H_3-(tBu)_2$

 $(T9)T = CH_2-C_6H_4-tBu$

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previously-reported mass requirements for optimal MS³ sequencing.[13]

Afterwards, the synthesis of the alkoxyamine-containing phosphoramidite spacers was investigated. At first, a molecular design involving a main-chain alkyne was considered (**11**, Figure 2). This alkyne function was included because it could bring further rigidity in the final molecule. Furthermore, the synthetic route leading to **11** (Scheme S1) was tempting because it involves an easy Sonogashira coupling step. As a proof-of-feasibility, this route was only studied with intermediate **T2**. The spacer **11** was synthesized and incorporated in model poly(phosphodiester)s (data not shown) for MS sequencing. However, it was observed that the presence of an acetylene group in the cleavable alkoxyamine spacer complicated sequencing rather than simplifying it (data not shown). Therefore, a revised route

Figure 2. Molecular design originally considered for the cleavable mass tags. Molecular structures of the phosphoramidite spacers **11** and **12**.

(Scheme S1) involving the reduction of the alkyne was explored.

This alternative route was also only studied with intermediate **T2**. The corresponding saturated spacer **12** was obtained. However, the overall yield was low and, on the whole, this second route did not seem so practical. Consequently, a third molecular design, which is closer to the one of \mathbf{L}_2 , was explored. Scheme 2 shows the synthetic route that was used in the present work for the synthesis of the library of tagged spacers **2**–**10**. Steps *(ii)* to *(v)* have already been reported for the synthesis of the spacer \mathbf{L}_{2} ^[18] However, instead of starting the synthesis from commercial 4-acetylphenylacetic acid **a** as done in our prior work, compound **a** was first synthesized in step *(i)* from 4-iodoacetophenone following a literature procedure.[20] Afterwards, steps *(ii)*, *(iii)*, *(iv) and (v)* were performed, as already reported, $[18]$ to afford the intermediates **b**, **c**, **d** and **e**, respectively. For the ATRA step *(vi)*, the tagged TEMPO derivatives **T2**–**T10** were reacted with **e** to afford the DMT-protected intermediates **f2**–**f10**. Ultimately, the phosphoramidite spacers **2**–**10** were obtained in step *(vii).*

The reactivity of spacers **2**–**10** was then investigated. To do so, each spacer was incorporated in the middle of a short oligo(phosphodiester) sequence and its coupling efficiency was assessed by UV titration of DMT deprotection.^[21] The oligomers were synthesized by automated PPC on a thymidine-loaded solid support, leading after cleavage to a thymidine nucleotide residue noted $T_[21]$ As described in the introduction, the polymers were all encoded with a twosymbols alphabet, in which propyl phosphate and 2,2 dimethylpropyl phosphate motifs represent 0 and 1 bits, respectively.[5b] The model oligomers all contain the sequence T1010-x-1010 (the sequence is written in the syn-

Scheme 2. Synthetic route studied herein for the synthesis of the tagged phosphoramidite spacers **2**–**10**. Experimental conditions: *(i)* PdCl2, AgOAc, NaOAc, AcOH, 130°C; (ii) NaBH₄, ethanol, RT, overnight; (iii) HCl 37%, THF, 0°C, 4 h; (iv) BH₃·SMe₂, anhydrous THF, RT, overnight; (v) anhydrous pyridine, anhydrous THF, RT, 1 h; *(vi)* CuBr, Cu(0), PMDETA, anhydrous THF, RT, overnight; *(vii)* DIPEA, anhydrous DCM, RT, 1 h.

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thesis direction), where x denotes a spacer of the TISP series **2**–**10**. The coupling efficiency was calculated by comparing the absorbance A_x of the DMT released after spacer coupling to the absorbance A_{ref} of the DMT released after coupling the previous monomer of the sequence (i.e. monomer 0 located in the fourth position from the thymidine nucleotide end-group). Table 1 shows the coupling efficiency estimated for spacers **2**–**10**. The original UV spectra are displayed in Figure S1. On the whole, the spacers exhibited markedly different reactivities, depending on the chosen mass tag. Spacers **2**, **3**, **5** and **9** led to near quantitative coupling yields in standard PPC protocols. In comparison, derivatives **6** and **8** exhibited a lower efficiency. Still, the lowest results were obtained with **4**, **7** and **10**. Different reasons may explain the observed differences in coupling efficiencies. First, the bulkiness of the mass tag seems to influence phosphoramidite activation and coupling. However, other parameters may play a role such as solubility. For instance, derivative **7** has a low solubility in acetonitrile due to classical pyrene aggregation.Thus, some drops of DCM were always added in the reaction solution to improve the solubility of **7**. Furthemore, optimized protocols can be used to improve the reactivity of bulky phosphoramidite monomers.[19b] One possibility is for example to increase the concentration of the spacer. Entries 6–8 in Table 2 compare the reactivity of **7** at a concentration of 0.1, 0.15 and 0.2 mol \cdot 1⁻¹.^[22] At the highest concentration, a near quantitative coupling efficiency was observed. Using a similar strategy, it was also possible to increase the reactivity

Table 1: Coupling efficiencies measured for spacers **2**–**10**.

Entry	Spacer ^[a]	Coupling efficiency [%]
	2	$> 99^{[a]}$
2	3	$>$ 99[a]
3	4	$7.5^{[a]}$
4	5	$> 99^{[a]}$
5	6	$89^{[a]}$
6	7	$24^{[a]}$
	7	$91^{[b]}$
8	7	$99^{[c]}$
9	8	$78^{[a]}$
10	9	$>$ 99[a]
11	10	$9.5^{[a]}$
12	10	$17^{[b]}$

Spacer concentration: [a] 0.1 mol·l⁻¹; [b] 0.15 mol·l⁻¹; [c] 0.2 mol·l⁻¹.

Figure 3. General molecular structure of the polymers synthesized in this work. The Greek letters α and ω indicate the end-groups and chaindirectionality. Synthesis starts from α whereas decoding starts from ω.

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of **10**, however with moderate success. Overall, spacers **2**–**3** and **5**–**9** seem appropriate for the preparation of digital polymers, whereas derivatives **4** and **10** are less suitable.

The spacers were then tested for the synthesis and MS decoding of digital poly(phosphodiester)s. Figure 3 shows the general molecular structure of the synthesized polymers and Table 2 lists the different samples **P1**–**P6** that have been studied herein. All polymers were prepared by automated PPC.[21] As described in the previous section, the polymers were prepared using comonomers θ and 1 .^[5b] Two different model sequences were investigated in this work, a triblock $(0)_8 - b - (1)_8 - b - (0)_8$ and an ASCII-encoded digital sequence 01010000–01001101–01000011. Following our previously-established convention, $[13]$ the reading direction of the polymers was set opposite to the synthesis direction. After each sub-sequence of 8 coded monomers, a cleavable spacer was incorporated in the chain. The size of blocks was selected to be long enough to contain one byte (i.e. 8 bits) but short enough to prevent extensive charging which complicates recovery of block sequence in pseudo- $MS³$.^[13] Thus, each byte is labelled with a tag with the exception of the last synthesized one (i.e. first byte for sequencing). The combination and sequence of the tags were selected according to pre-established rules for pseudo-MS³ sequencing.^[13] Another important feature in these polymers is the chain-end of the first synthesized byte (i.e. last byte for sequencing). In previous works,[12–13,19b,21] a thymidine-loaded solid support was used. After cleavage, it leads to poly(phosphodiester)s with a thymidine chain-end (see previous section). Although the initial reason for this choice was UV detection in $HPLC₁^[21]$ the thymidine nucleoside was afterwards exploited as a byte tag for pseudo MS³ sequencing.^[13] However, since nucleoside tags are no longer used in this work, the presence of a terminal thymidine is not mandatory. Therefore, a universal solid support was selected herein. We chose to work with commercial UnySupport, which is a methylated version of the classical UnyLinker.[23] Upon cleavage, it leads to a terminal dephosphorylation.^[23–24] Consequently, the cleaved poly(phosphodiester)s do not contain a residual end-group inherited from the solid support, as shown in Figure 3.

All the formed polymers were characterized by negative mode ESI-MS (Figures 5 and S2–S6). Some issues have arisen during the synthesis of the polymers. The first obstacle was due to the utilization of the commercial Unysupport. The standard cleavage procedure with this support includes a 2 h heating treatment at 55 °C. However, when applying this procedure, an abundant species with an

Figure 4. Different types of α-chain-ends observed for the formed polymers after cleavage from commercial UnySupport. Left: targeted hydroxyl chain-ends. Right: chains containing a residual group due to incomplete dephosphorylation.

[a] Sequences are written in the reading direction (i.e. from ω to α, which is opposite to the synthesis direction). See Figure 3 for the meaning of the numbers as well as the Greek letters $α$ and $ω$. [b] Measured as [Pi-6H]⁶⁻ at monoisotopic peak.

Figure 5. Mass spectrometry analysis of polymer **P3**. (a) Molecular structure of **P3**. (b) Negative mode ESI-MS spectrum of **P3**. (c) MS/MS spectrum (11 eV laboratory frame) of P3 obtained by collision-induced dissociation of the [P3-6H]⁶⁻ precursor ion.

increment of mass of $+293$ Da as compared to the targeted polymers was often detected in mass spectrometry.This is probably due to an incomplete dephosphorylation and corresponds to polymers in which the leaving group of the support was still attached to the main-chain, as proposed in Figure 4. Therefore, the cleavage procedure was modified. A more efficient dephosphorylation could be achieved using slightly more elevated temperatures (60 or 70 $^{\circ}$ C), as exemplified in Figure S2a for **P1** and Figure S4a for **P4**. However, these conditions are not optimal for the mainchain alkoxyamines of the polymers and may result in partial strand degradation. In this context, the best compromise was to achieve cleavage overnight at 50 °C (Figure S3a for **P2**).

In all cases, the targeted polymers appear as dominant species, even though some defects may be detected. As written above, defects due to incomplete desphophorylation were significantly reduced in these samples. Thus, they were either not detected or appeared as minimal species. Still, some truncated sequences due do to missing (macro)monomers can be seen in some spectra. As expected, sequences containing spacers **2** and **5**–**9** could be easily achieved and detected. Despite its low solubility in acetonitrile, **7** could also be incorporated efficiently in polymers **P3** and **P6**. Furthermore, all polymers could be sequenced. First, the MS/MS analysis of samples **P1**–**P6** resulted in the controlled fragmentation of the backbone, thus indicating that the expected alkoxyamine cleavage occurs with all studied spacers. The MS/MS collision conditions were adapted in order to release all the bytes including the inner byte. As a result, MS/MS spectra highlighting a clear byte library were obtained in all cases (Figures 5 and S2–S6). For example, Figure 5c shows the MS/MS spectrum obtained for polymer **P3.** Although the same information can be obtained from any $[P3-zH]^{z-}$ ions (with $z=4-9$) observed in MS (Figure 5b), selecting the charge state of the dissociating precursor as a multiple of the number of bytes allows a single charge state for each fragment, hence highly contributing to the readability of MS/MS data. Importantly, the chosen combinations of mass tags permitted to identify unequivocally the cleaved bytes in all samples. These results indicate that the synthetic mass tags develop herein proceed as efficiently as previously-used nucleotide tags. Consequently, complete sequencing could be obtained through the $MS³$ analysis of each fragmented byte (Figure S7).

Conclusion

In summary, eleven different phosphoramidite monomers containing a cleavable alkoxyamine and a mass tag were synthesized in this work. At first, a synthetic route involving an alkyne connection was explored but was not retained

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because of synthesis and sequencing issues. An optimized route was found and allowed successful preparation of nine different tagged monomers. They were then tested individually for the stepwise phosphoramidite synthesis of oligo(phosphodiester)s and their reactivity was assessed by UV monitoring. It was observed that their coupling yields depend on the chemical nature of the mass tags. For instance, monomers containing benzyl, 4-fluoro benzyl, 4 *tert*-butylbenzyl and biphenyl mass tags exhibited near quantitative reactivities in standard PPC conditions. However, other monomers exhibited lower or significantly lower coupling yields. This could be improved by increasing their concentration in PPC. Still, alternative molecular design may also be considered in the future. For bulky substituents, for instance, the use of long linkers between the mass tag and the phosphoramidite group may improve coupling yields. Nevertheless, the best monomers were tested for the synthesis of digital polymers. Poly(phosphodiester)s with different monomer sequences were prepared and sequenced in pseudo- $MS³$ conditions. It was found that the main chain alkoxyamine of the spacers do enable controlled fragmentations and the mass tags allow decoding. Furthermore, spacers library allows some flexibility in the choice of the tags for PPC. In terms of atom economy, the cleavable mass tags reported herein correspond to synthons with a molar mass ranging from 532 (benzyl tag) to 668 Da (3,5 bis(trifluoromethyl)benzyl tag). In our previous strategy employing both a spacer and a nucleotide mass tag, the \mathbf{L}_2 synthon^[18] has a molar mass of 383 Da and the tags have a molar mass ranging from 289 (deoxycytidine monophosphate tag) to 415.9 Da (iodo-deoxyuridine monophosphate tag), $^{[13]}$ thus a global contribution ranging from 672 to 798.9 Da. This means that for each cleavable site, the present system leads to a mass reduction ranging from 4 to 266.9 Da. In addition, the use of only one phosphoramidite monomer in PPC instead of two significantly improves atom economy. Yet, the present concept is of course not restricted to the mass tags studied herein and a broader variety of spacers can be envisioned in the future. Overall, this work considerably simplifies the design of digital polyphosphodiesters and opens up new opportunities for macromolecular data storage.

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Conflict of Interest

The authors declare no conflict of interest.

Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

Keywords: Data Storage **·** Mass Spectrometry **·** Sequence-Controlled Polymers **·** Sequencing **·** Solid-Phase Chemistry

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