

Influence of typology and management practices on water pCO 2 and atmospheric CO2 fluxes over two temperate shelf–estuary–marsh water continuums

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1 Abstract

2 Within the coastal zone, salt marshes often behave as atmospheric CO₂ sinks, allowing 3 for blue carbon (C) sequestration associated with intense autotrophic metabolism. However, C 4 dynamics over salt marshes are complex since various biogeochemical processes and fluxes 5 take place at different terrestrial - aquatic - atmospheric exchange interfaces and 6 spatiotemporal scales. This study focuses on seasonal, tidal and diurnal variations of water 7 pCO₂, estimated water-air CO₂ fluxes and controlling factors along two temperate shelf – estuary – marsh continuums. The latter include typical coastal systems (shelf, estuary and 8 9 channel) with artificial salt marshes that have contrasting management practices. Our high-10 frequency (seasonal 24-hour cycles) biogeochemical measurements at the various stations 11 highlighted a strong control of ecosystem typology on inorganic C dynamics with lower water pCO₂ values in the artificial salt marshes, due to stronger biological activity and longer water 12 13 residence times, than in the tidal estuary. In this marine-dominated estuary, water pCO₂ 14 variations (267 - 569 ppmv) were strongly controlled by tidal effects and phytoplankton activity 15 particularly in spring/summer. On the contrary, the greatest amplitudes in water pCO₂ were recorded in the artificial salt marshes (6 - 721 ppmv) due to intense macrophyte activity. 16 However, in the rewilded artificial marsh, spring/summer fast-growing macroalgae produced, 17 18 in turn, strong fall atmospheric CO₂ outgassing from degraded algae waters and thus a net annual source of CO₂ to the atmosphere (17.5 g C m⁻² yr⁻¹). Conversely, management practices 19 20 at the working artificial marsh for salt-farming activity with longer winter water residence times 21 favoured rather slow-growing macrophytes which greatly contribute to the yearly observed atmospheric CO₂ sink (-97.7 g C m⁻² yr⁻¹) and, more generally, blue C sequestration. We suggest 22 23 that artificial salt marsh management can be used to control the contribution of primary producers to marsh C budget as atmospheric CO₂ (sink and/or source). 24

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26	Key words: shelf – estuary – marsh continuums; water pCO ₂ ; air-water CO ₂ fluxes; diurnal,
27	tidal, seasonal scales; marsh management practices; macrophytes.
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44 Graphical Abstract



Water management practices and primary producer types influence the marsh CO₂ behaviour (sink/source)

51 **1. Introduction**

52 Marine coastal environments, which only account for 7% of the global ocean, perform 53 major ecological functions such as primary production, bacterial mineralization, organic matter burial and calcium carbonate deposition (Gattuso et al. 1998). Comprised of mixed continental 54 55 and oceanic waters, the coastal zone presents a wide diversity of geomorphological types and 56 ecosystems (shelves, estuaries, bays, tidal wetlands) shaping the biogeochemical cycle coupling between the land, ocean and atmosphere (Aufdenkampe et al. 2011, Bauer et al. 2013). On the 57 58 one hand, these dynamic and heterogeneous ecosystems behave as an active horizontal pipe carrying and processing large quantities of carbon (~ 0.90 ± 0.05 Pg C yr⁻¹, Cole et al. 2007) 59 60 from the drainage basin to the open ocean (Cai 2011, Najjar et al. 2018). On the other hand, the 61 coastal zone vertically exchanges large and variable quantities of carbon (C) with both the atmosphere and sediments (Cole et al. 2007, Polsenaere et al. 2012). For instance, at the global 62 scale, continental shelves behave as atmospheric CO₂ sinks and absorb 0.25 ± 0.05 Pg C yr⁻¹ 63 (Bauer et al. 2013, Dai et al. 2022) due to phytoplankton primary production (Cloern et al. 64 2014). On the contrary, CO₂ supersaturated estuarine waters emit 0.25 ± 0.05 Pg C yr⁻¹ to the 65 66 atmosphere (Bauer et al. 2013) due to a high mineralization of organic matter from the land 67 (Frankignoulle et al. 1998, Borges & Abril 2011). These atmospheric C exchanges within the 68 coastal zone are heterogeneous (Borges et al. 2005) and need to be better taken into account in 69 regional and global C budgets (Polsenaere 2011, Najjar et al. 2018). For instance, coastal 70 wetlands, including salt marshes located along inner shelf – estuary – marsh continuums, absorb 0.55 ± 0.05 Pg C yr⁻¹ from the atmosphere (Bauer et al. 2013) and may play a major role in 71 72 atmospheric CO₂ uptake and associated organic C burial on Earth (Cai 2011, Mcleod et al. 73 2011).

74 In salt marshes, inorganic C dynamics and water pCO₂ in particular are influenced by 75 several physicochemical and biological processes within and between each ecosystem 76 compartment such as tidal exchanges, calcium carbonate precipitation/dissolution, benthic-77 pelagic couplings, air-water exchanges and photosynthesis/respiration balance (Cai 2011, 78 Bauer et al. 2013, Macreadie et al. 2017). Due to high photoautotrophy rates of both aquatic 79 (micro-algae and macrophytes) and terrestrial (vascular plants) primary producers, these highly 80 productive ecosystems mostly behave as net atmospheric C sinks (Schäfer et al. 2014, Artigas 81 et al. 2015, Forbrich & Giblin 2015). For instance, net primary production (NPP) values in Spartina alterniflora salt marshes ranged from 100 to > 2500 g C m⁻² yr⁻¹ (Duarte & Cebrián 82 83 1996, Tobias & Neubauer 2019) allowing for significant CO₂ uptake (Wang & Cai 2004). A 84 refractory part of organic C produced through photosynthesis in these vegetated coastal 85 ecosystems can then be largely sequestered in sediments (Chmura et al. 2003) and stored as blue C at a rate of 218 ± 24 g C m⁻² yr⁻¹, and greatly contribute to the regional/global C cycle 86 87 in comparison with terrestrial ecosystems (Mcleod et al. 2011). Salt marshes also produce and 88 horizontally export significant quantities of C through tidal water advection (Najjar et al. 2018) 89 which could, in turn, strongly influence the C balance of the system itself as well as the estuary 90 and shelf systems (Cai 2011). The "marsh CO₂ pump" hypothesis proposes that atmospheric 91 CO₂ fixation by plants and phytoplankton in marshes and the export of part of the associated C 92 may be one of the major mechanisms making adjacent coastal waters sources of CO₂ to the 93 atmosphere (Wang & Cai 2004). To illustrate this, Wang et al. (2016) used continuous in situ 94 measurements of relevant biogeochemical parameters and water fluxes in an intertidal salt 95 marsh in northeast region of the United States (12300 km²) to estimate the total net CO₂ fixation (NPP) of 13.2 Tg C yr⁻¹. Of this total NPP value, 56% was exported to the coastal ocean through 96 97 the outwelling phenomenon (horizontal advection) with inorganic and organic C accounting for

98 39% and 17%, respectively; this means that the export of C from marsh tides could represent a major proportion (414 g C m^{-2} yr⁻¹) in the overall marsh budget (Wang et al. 2016). 99 100 Nevertheless, despite these major ecological potentials (storm protection, nursery areas, long-101 term C storage), these interface zones are the most threatened in the world by land-use changes, 102 climate changes and sea level rise as described by Gu et al. (2018) in salt marshes in China. 103 Moreover, coastal eutrophication causes the loss of salt marshes by decreasing the below-104 ground biomass of plant roots through microbial degradation thereby producing a decrease in 105 the geomorphic stability of marshes (Deegan et al. 2012). Since the 1800s, salt marshes have 106 lost about 25% of their global area with negative effects on the atmospheric CO₂ sink and the 107 associated C sequestration (Mcleod et al. 2011). Their importance as ecosystem service 108 reservoirs has made it possible to implement protection and restoration policies that contribute 109 to their better management and to the development of their ecological and economic potentials 110 (Gu et al. 2018, Adam 2019).

111 The high heterogeneity in biogeochemical processes within coastal systems at spatial and 112 temporal scales (Cai 2011, Bauer et al. 2013) requires more integrative C process and exchange 113 measurements at the various terrestrial - aquatic - atmospheric interfaces over different time 114 scales (tidal, diurnal and seasonal) to better understand the ecological functioning of these 115 ecosystems facing global changes. Some studies in coastal wetlands such as salt marshes or 116 tidal estuaries have taken water pCO₂ measurements at different temporal scales allowing the 117 study of in situ CO₂ dynamics in relation to other biotic and abiotic processes. For instance, in 118 Guanabara Bay (Brazil), water pCO₂ measured at high frequency, varying from 22 to 3715 119 ppmv, was negatively correlated with Chlorophyll a (Chl a) in surface waters, indicating that 120 phytoplankton activity has an influence on CO₂ uptake and, more broadly, on inorganic C 121 dynamics (Cotovicz Jr. et al. 2015). Moreover, Borges (2003) performed a 24-hour cycle in

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122 June 2001 in an intertidal mangrove system (Gaderu Creek, India) and showed that water pCO₂ 123 ranged from 1380 to 4770 ppmv with a strong control of these fluctuations by tides and 124 biological activity (primary production and respiration). Within a small tidal creek of the Duplin 125 River salt marsh-estuary coastal ecosystem (Georgia, USA), water pCO₂ calculated by Wang 126 et al. (2018) showed strong seasonal and tidal/diurnal variations with winter values ranging 127 from 500 to 4000 ppmv between high and low tide respectively, and summer values ranging 128 from 1600 to 12000 ppmv between high and low tide respectively. However, still too few 129 studies have taken high-frequency water pCO₂ measurements in salt marshes at the diurnal 130 scale. These temporal variations in water pCO₂ strongly affect associated air-water CO₂ fluxes 131 (FCO₂) that can, in turn, be estimated from the CO₂ gas transfer velocity, CO₂ solubility in the 132 water and air-water CO₂ gradient (Borges 2003, Crosswell et al. 2017). The atmospheric Eddy 133 Covariance (EC) technique represents an alternative way to directly measure in situ atmospheric 134 CO₂ fluxes at the ecosystem scale (Baldocchi et al. 1988, Baldocchi 2003, Schäfer et al. 2014). 135 This direct and non-intrusive micrometeorological method shows a growing interest in studying 136 the metabolism of coastal ecosystems (sink or source) under real field conditions and to 137 integrate them into regional C budgets (Polsenaere et al. 2012, Van Dam et al. 2021).

138 The purpose of this study is to better understand CO₂ dynamics at different temporal scales 139 and locations over two aquatic sea – land continuums along the Atlantic French coast on Ré 140 Island. These continuums include typical coastal systems (shelf, estuary, marsh) such as those 141 studied elsewhere by Cai (2011) and Bauer et al. (2013) with regards to horizontal and vertical 142 C exchanges in the coastal ocean. Unlike tidal salt marshes, which are more generally discussed 143 in the literature, we studied two artificial salt marshes (i.e. salt ponds; Fig. A2) in which water 144 exchanges are controlled by dykes and locks for human uses (biodiversity protection or 145 anthropogenic activities). Through in situ high-frequency measurements of biogeochemical parameters in waters and estimations of atmospheric CO_2 fluxes from 2018 to 2020, we sought to (1) describe water p CO_2 and associated CO_2 exchange variations at the diurnal, tidal and seasonal scales along the studied aquatic continuums, (2) identify biophysical drivers and the potential role of station typologies and salt marsh management practices on water p CO_2 and CO_2 budgets and (3) contextualize the associated continuum metabolism among other studied systems from a C dynamic and budget point of view.

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153 **2. Materials and methods**

154 **2.1. Study sites**

155 **2.1.1.** Estuary (station *a*)

156 The Fier d'Ars represents a 750-ha maritime area within the French Atlantic Ocean 157 (north of Ré Island), connected to the Breton Sound continental shelf by an opening that is 700 158 m wide (Fig. 1). It corresponds to a type II temperate tidal estuary according to Dürr et al.'s 159 (2011) coastal system typology. With a maximum water height of 6.5 m, this tidal estuary 160 exchanges between 2.4 and 10.2 million m³ of coastal waters with the adjacent continental shelf 161 depending on the tidal amplitudes (Bel Hassen 2001). At low tide (LT), its subtidal zone (in 162 light blue; Fig. 1) is composed of mudflats (slikke) and tidal salt marshes (schorre) traversed 163 by numerous channels converging toward a main channel measuring 300 m wide (Fig. 1). At 164 high tide (HT), the subtidal zone is flooded by coastal waters up to the dykes (in red; Fig. 1), 165 managed to control water exchanges between the estuary and artificial salt marsh complexes. 166 These artificial salt marshes are old tidal salt marshes divided into multiple ponds mainly 167 located along European coasts for which water residence times (from a few hours to fifteen 168 days according to the management practices; Bel Hassen 2001) were originally controlled for

169 salt-farming (Tortajada et al. 2011). Nowadays, some of them are intended for other uses with 170 different water management practices based on their socio-economic activities (salt-, oyster-171 and fish-farming or natural management; Fig. A2). The studied station *a* is located within this 172 tidal estuary, inside the subtidal zone of Fier d'Ars at its entry along the main channel connected 173 to the slikke (Fig. 1).

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2.1.2. Channel (station *b*)

Station *b* is a secondary tidal channel associated to the schorre and located at the back
of the Fier d'Ars estuary just before the dyke (250 m wide at HT and 20 m wide at LT; Fig. 1).
With a maximum water height of 5.3 m at HT, it is connected to the tidal estuary station *a*(distance of 1.6 km between stations *a* and *b*) enabling the supply of coastal water to artificial
salt marshes upstream from the dyke (Fig. 1).

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2.1.3. Rewilded artificial salt marsh (station *c*)

183 For forty years, 121 ha of salt marshes within the Fier d'Ars system have been protected 184 and managed inside a National Natural Reserve (NNR) to encourage a high level of biodiversity 185 of migratory birds and vascular phanerogam and cryptogam developments (Ruppia maritima 186 and Ruppia cirrhosa in artificial salt marshes and Zostera noltei on mudflats; Champion et al. 2012). Within the terrestrial area of the NNR, artificial salt marshes communicate with Fier 187 188 d'Ars waters (station a) through lock management practices to promote biodiversity protection. To the west, station c (surface area of 40100 m², depth of 60 cm) inside the NNR is a rewilded 189 190 artificial salt marsh supplied indirectly with coastal waters from the Fier d'Ars estuary by the

191 station b channel (distance of 500 m between stations b and c; Fig. 1) through a specific lock 192 management ensured by the NNR (former salt farm that has now been rewilded; Fig. A2). From 193 November until March (winter period), the marsh lock is open only at each full moon phase 194 when tidal amplitudes are higher than 70 in order to have the best compromise between salt-195 and fresh-mixing waters (salinity around 30) to make it easier for aquatic fauna to pass from 196 the continental shelf to the marsh. On the contrary, from April to October when tidal amplitudes 197 are lower, the lock is permanently open to favour the development of Ruppia sp. seagrass beds 198 in the marsh with salinity varying between 30 and 46 (Champion et al. 2012). In past years, this 199 artificial salt marsh is also characterized by significant macroalgae development 200 (Enteromorpha sp. and Ulva sp.) at the subsurface and on sediments (Fig. A1) from early spring 201 to late summer each year thereby preventing seagrass development (Champion et al. 2012).

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2.1.4. Working artificial salt marsh (station d)

To the east of the Fier d'Ars estuary, station d (surface area of 8500 m², depth of 75 cm) 204 205 also corresponds to an artificial salt marsh upstream from the dyke even though it directly 206 communicates (no channel in between) with the coastal waters of the Fier d'Ars estuary by a 207 dyke lock (distance of 2 km between stations a and d; Fig. 1). This second artificial salt marsh 208 was chosen for its specific lock management practice depending on the salt-farming activities 209 (working marsh) contrary to station c (rewilded marsh). In spring and summer, this working 210 marsh is particularly used for the storage of salt water in the context the salt-farming activity. 211 In this way, the lock is regularly open at this time of the year allowing upstream pond succession 212 supply to produce salt via evaporation between June and September (Fig. A2). Moreover, the 213 use of this artificial marsh for salt-farming activities requires a drying up and a cleaning once a 214 year in early spring before the start of the salt production period to remove seagrass, macroalgae215 and organic matter in the marsh (Poitevin, personal communication).

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217 2.2.5. Other associated stations: continental shelf (station *Filiere W*) and tidal salt 218 marsh (station *e*)

219 The Fier d'Ars estuary communicates with the Breton Sound which corresponds to a coastal 220 maritime area on the French continental shelf between Ré Island and the continent (Fig. 1), characterized by a surface area of 425 km² and volume of 4920 million m³ (Stanisiere et al. 221 222 2006, Soletchnik et al. 2014). The Breton Sound continental shelf exchanges large quantities of 223 salt water with the Atlantic Ocean to the west at each semi-diurnal tidal cycle and it receives 224 fresh water inputs through the Sèvre and Lay rivers within the Aiguillon Bay to the east (annual mean discharges of 44.4 and 14.0 m³ s⁻¹, respectively) depending on hydrodynamic and 225 meteorological conditions (Stanisiere et al. 2006, Soletchnik et al. 2014, Polsenaere et al. 2017). 226 227 The highest and lowest river water flows were recorded in winter and summer, respectively, 228 influencing the water salinity of the Breton Sound differently. Station Filiere W in the centre of 229 the Breton Sound (Fig. 1) is located in a predominantly marine environment with a freshwater 230 contribution of approximately 30 ml/l (i.e. 3%) from the Sèvre and Lay rivers (Stanisiere et al. 231 2006, Soletchnik et al. 2014). This additional station, studied by Coignot et al. (2020) using a 232 different measurement strategy, was included in this study to contextualize the influence of the 233 Breton Sound on our studied stations. At each HT, the Breton Sound supplies our studied 234 coastal stations with various water masses based on the tidal amplitudes and seasonal periods along two aquatic sea – land continuums: (1) continental shelf (station Filiere W) – estuary 235 236 (station a) – channel (station b) – rewilded artificial salt marsh (station c) and (2) continental

shelf (station *Filiere W*) – estuary (station a) – working artificial salt marsh (station d) (Fig. 1). Conversely, at each LT, different water masses from the artificial salt marshes are exported (indirectly through the station b channel for station c or directly for station d) to the Fier d'Ars estuary and then to the Breton Sound (Fig. A2).

Since June 2019, an atmospheric Eddy Covariance (EC) station (Campbell Scientific) has been deployed at station e (Fig. 1) to continuously measure in situ CO₂ fluxes at both the marshatmosphere and water-atmosphere interface at the ecosystem scale (Mayen et al. in prep). This tidal salt marsh station is located within the maritime NNR area with typical marsh vegetation (*Halimione sp., Suaeda sp.* and *Spartina sp.*) emerged for 70% of time, during low tides and neap tides. In this study, only meteorological parameters (air temperature, rain, wind speed) corresponding to data from our four station measurement cycles were used from the EC station.

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249 **2.2. Measurement strategy and probes**

250 In the subsurface water (~ 30 cm depth), partial pressures of CO₂ (pCO₂) and 251 biogeochemical parameters (temperature, salinity, turbidity, dissolved oxygen and pH) were 252 autonomously measured each minute with in situ probes deployed during fifteen 24-hour cycles 253 at stations a, b, c and d during each season to record relevant temporal (diurnal, tidal and 254 seasonal) and spatial (continuums) variations (Table A1). It should be noted that these seasonal 255 measurement cycles were performed at one-day intervals (i) between stations a and b in 2018 256 and (ii) between stations c and d in 2019/2020 (Table A1). In situ measurements could not be 257 taken at station d in spring 2020 due to the Covid pandemic. Over the Breton Sound continental 258 shelf at station Filiere W, the same biogeochemical measurements were taken biweekly by 259 Coignot et al. (2020) over the year 2018: four samplings in winter (17/01/2018 14:15; 260 30/01/2018 13:55; 15/02/2018 14:03; 03/03/2018 13:40), four samplings in spring (17/04/2018 261 14:25; 26/04/2018 12:20; 29/05/2018 13:40; 14/06/2018 14:00); four samplings in summer 262 (28/06/2018 13:05; 11/07/2018 12:45; 09/08/2018 10:45; 11/09/2018 08:50) and five samplings 263 in autumn (26/09/2018 13:45; 09/10/2018 13:25; 24/10/2018 12:50; 08/11/2018 08:15; 264 11/12/2018 09:50). At stations *b* and *c* in summer 2019, the water height, temperature and 265 salinity were simultaneously measured every 10 min. by two STPS sensors (NKE 266 Instrumentation) (unpublished results).

An autonomous pCO₂ underwater sensor (C-SenseTM pCO₂ sensor, PME/Turner Designs), 267 268 an EXO2 multiparameter probe (YSI) and a C3-submersible fluorimeter (Turner Designs) were 269 deployed to measure water pCO₂, physicochemical parameters and fluorescence, respectively. 270 Chlorophyll a concentrations were derived from the fluorescence data measured by the C3-271 fluorimeter every 10 min. as a proxy to study the dynamics of phytoplankton biomass over time 272 (Aminot & Kérouel, 2004). This sensor was used only for summer 2019 and winter 2020. The 273 measurement range of the C-Sense probe is 0-2000 ppmv with an absolute accuracy of 60 ppmv 274 (3% of the full scale; Turner Designs). The EXO2 probe was used to measure the pH (NBS scale), temperature (in °C), salinity, dissolved oxygen concentration (DO in µmol l⁻¹), oxygen 275 276 saturation percentage (O₂-sat. in %) and turbidity (in NTU). The maximum permissible errors 277 of the different EXO2 sensors are 0.2 and 0.5 unit for pH and salinity respectively, 0.25 °C for temperature, 5 NTU for turbidity and 25 μ mol l⁻¹ for O₂ (YSI). Between each station, the EXO2 278 279 pH sensor was calibrated with three buffer solutions (pH 3, 7 and 10). It was not possible to 280 measure pH values at stations a and b in 2018 (Table 2).

Water pCO₂ values measured by the C-Sense probe are influenced by the total dissolved gas pressure (TDGP) which corresponds to the total pressure exhibited by all gases within the water column. When this pressure greatly exceeded the pressure at which the C-Sense was calibrated, the output needed to be corrected. Then, a pCO₂ correction was applied taking both TDGP and atmospheric pressure during sensor calibration (1009 hPa) and the measured pCO₂ by the C-Sense probe (gross values) into account, as per equation (pCO_{2meas}. \times 1009) / TDGP (Turner Designs). Over the fifteen 24-hour cycles performed at the seasonal and spatial scales, the corrected pCO₂ with TDGP were 2.6 ± 0.9% lower than the raw pCO₂ values.

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290 **2.3. Temperature and non-temperature effects on pCO₂ variations**

To distinguish between the temperature and non-temperature effects on in situ pCO₂ variations at the seasonal and diurnal scales, $TpCO_2$ (pCO₂ variations related to temperature physical effects, in ppmv) and NpCO₂ (pCO₂ variations related to non-temperature effects, in ppmv) were calculated respectively, following (Eq. 1) and (Eq. 2) from Takahashi et al. (2002):

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$$TpCO_2 = pCO_{2mean} \times exp[0.0423 \times (T_{obs} - T_{mean})]$$
(1)

where T_{obs} and pCO_{2obs} are the temperature and pCO₂ values measured by the probes at each time step (1 min.), respectively. T_{mean} and pCO_{2mean} are the temperature and pCO₂ averaged either at the seasonal (annual mean) or diurnal scale (means per cycle). Whereas TpCO₂ is only associated with the physical pump, NpCO₂ is associated with biological processes, tidal advection and benthos-pelagos couplings that may be important in shallow coastal systems (Cotovicz Jr. et al. 2015, Polsenaere et al. 2022).

303 **2.4. Calculations of air-water CO₂ fluxes**

For all our measurement periods (Table A1), the gas transfer velocity (k_{600}) and hourly CO₂ fluxes (FCO₂) at the air-water interface were estimated following, for instance, Ribas-Ribas et al. (2011) and Polsenaere et al. (2022) in coastal environments. At stations *a* and *b*, only airwater FCO₂ during HT (four hours around each HT) were calculated, whereas at stations *c* and *d*, all hourly FCO₂ were calculated using the following formula (Eq. 3):

$$309 \quad FCO_2 = \alpha \times k \times \Delta pCO_2 \tag{3}$$

where FCO₂ is the estimated CO₂ fluxes at the air-water interface (mmol m⁻² h⁻¹), α is the CO₂ 310 solubility coefficient in saltwater (mol kg⁻¹ atm⁻¹), k is the gas transfer velocity of CO₂ (in cm 311 312 h^{-1}) and ΔpCO_2 is the gradient between mean water and air pCO₂. Water pCO₂ (ppmv) were 313 measured by the in situ C-Sense probe and atmospheric CO₂ concentrations (ppm) were 314 measured by the EC station (station e; Fig. 1) during summer 2019 and winter 2020. For all 315 other periods, atmospheric CO₂ data were obtained from the National Oceanic and Atmospheric 316 Administration (NOAA) Mauna Observatory at the Loa in Hawaii 317 (https://gml.noaa.gov/ccgg/trends/). The CO2 solubility coefficient (a) depends on water 318 temperature and salinity and was calculated according to Weiss (1974). The gas transfer 319 velocity (k) also significantly controls air-water FCO₂ since it directly takes turbulence 320 processes at the air-water exchange interface into account (Polsenaere et al. 2013). In this study, 321 k (or k_{660}) was calculated according to both Raymond & Cole (2001) (RC01; Eq. 4) and 322 Wanninkhof (1992) (W92; Eq. 5) corresponding to closed environments and more open coastal 323 environments, respectively. These two parametrization methods for the k exchange coefficient 324 were applied to the four stations in order to compare the results.

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326 For closed freshwater environments (Raymond & Cole 2001),

327
$$k_{600} = 1.91 \times \exp[0.35 \times U10]$$

328 For more open coastal environments (Wanninkhof 1992),

$$329 k_{600} = 0.31 \times (U_{10})^2 (5)$$

The gas transfer coefficients normalized to a Schmidt number of 600 (k_{600}) obtained with the two parametrization were then converted to the gas transfer velocity of CO₂ at the in situ temperature and salinity (k_{660}) according to Jähne et al. (1987) as per the equation (6):

333
$$k_{660} = k_{600} / (660/\text{Sc})^{-0.5}$$
 (6)

334 where k_{660} is the gas transfer velocity of CO₂ at the in situ temperature and salinity (cm h⁻¹) according to the parametrizations of RC01 or W92, U_{10} is the wind speed normalized to 10 m 335 336 (m s⁻¹) using the relationship of Amorocho & DeVries (1980) and Sc is the Schmidt number which describes both the water viscosity and the molecular diffusion of the subsurface layer 337 338 (Bade 2009). In summer 2019 and winter 2020, wind speed data were measured by the EC 339 station at a height of 3.15 m (station e; Fig. 1); for all other periods, wind data were obtained 340 from the "Infoclimat" station on Ré Island (Fig. 1) measured at a height of 10 m (6.20, 4.85, 341 4.30 and 8.40 km from stations a, b, c and d respectively).

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343 **2.5.** In situ Chl *a* concentrations and fluorimeter data calibration

In situ Chl *a* concentrations were measured from samples collected at stations *c* and *d* following Aminot & Kérouel (2004). Water samples (50 ml) were filtrated (47 mm Whatman GF/F filter) and Chl *a* filters were kept at -18° C up to analysis. In the laboratory, in the dark, pigments were extracted in a 90% acetone solvent at the same time the filter was crushed with a glass rod. After one night of stirring at 4 °C to continue the extraction and a 10-min centrifugation at 3000 rpm, the supernatant absorbance was measured by spectrophotometry at 665 nm to obtain the Chl *a* level (its absorbance peak is at 665 nm). Chl *a* were measured from eight water samples except at station *c* in spring. Chl *a* could not be measured at stations *a* and *b* in 2018.

For the fluorimeter data, the calibration procedure was applied to derive Chl *a* concentrations from our 10 min-water fluorescence measurements. Chl *a* concentrations were calculated through the significant (p < 0.05) linear regressions obtained for each deployment (summer 2019 and winter 2020) between C3-submersible fluorimeter data and the in situ Chl *a* concentration analysis from sub-surface waters (eight samplings on average per each 24 hourcycle) sampled simultaneously right at the fluorescence probe.

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2.6. Statistical tools and analysis

361 For all measured variables, the data did not respect a normal distribution (Shapiro-Wilk, p 362 < 0.05). Shapiro-Wilk tests and non-parametric comparison tests such as the Mann-Whitney 363 and Kruskal-Wallis tests were carried out with 0.05 level of significance. A Dunn test was used 364 to perform a post-hoc multiple comparison of the Kruskal-Wallis test to detect significant 365 differences among groups. The R-studio software was used to perform the principal component 366 analysis ("FactoMineR" package; Lê et al. 2008) and the Spearman correlation matrices 367 ("corrplot" package; Wei and Simko 2017). A PCA was carried out to study the contribution of 368 the variables to the data at the spatial and seasonal scales. It is based on the mean values of the 369 physicochemical and biogeochemical parameters of water (temperature, salinity, turbidity, 370 dissolved oxygen, oxygen saturation and water pCO_2) for each 24-hour cycle at stations a, b, c

and *d*. Temporal graphs, linear regressions, boxplots and barplot were performed with theGraphPad Prism 7 software.

Total alkalinity (TA) were estimated at the diurnal scale from the measured salinity, temperature, pH and water pCO₂ using the carbonic acid constant from Mehrbach et al. (1973) as modified by Dickson and Millero (1987), the K_{HSO4} constant from Dickson (1990) and the borate acidity constant from Lee et al. (2010). The CO₂ system calculation programme (version 2.1.) was used to perform these calculations (Lewis and Wallace, 1998).

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379 3. Results
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380 3.1. Biogeochemical overview of the shelf – estuary – marsh water continuums

381 Over our measurement periods, meteorological conditions for the years 2018 and 2019 were 382 similar with regard to air temperatures. The thermal conditions followed a classical seasonal 383 trend with the highest and lowest temperatures measured in summer (July 2018 and 2019) and 384 in winter (March 2018 and February 2020), respectively (Table 1). However, the months of 385 July 2018, July 2019 and February 2020 were warmer (+2.1, +2.4 and +3.2 °C, respectively) 386 than the reference period (1981-2010). Annual cumulative precipitations were lower in 2018 387 than in 2019 (786 and 827 mm, respectively); March 2018 and October 2019 were the rainiest 388 months (+82 and +41%, respectively, compared with the 1981-2010 period; Table 1). Salinity 389 values measured by the French Phytoplankton Monitoring Network REPHY at station Filiere 390 W as the water source flowing into the two aquatic continuums did not vary significantly 391 between the years 2018, 2019, 2020 and the 2000-2017 period (Kruskal-Wallis test, p = 0.77; 392 Fig. A3).

393 At the station Filiere W continental shelf, the highest and lowest Chl a concentrations were recorded in winter $(2.0 \pm 0.0 \text{ µg l}^{-1})$ and in summer $(3.0 \pm 1.2 \text{ µg l}^{-1})$, respectively (Coignot et 394 395 al. 2020). In the artifical salt marshes, in summer 2019, autumn 2019 and winter 2020 (Table 396 A1), the Chl *a* concentration varied significantly at the spatial scale (Mann-Whitney tests, $p < d^{2}$ 0.05). For example, at stations c and d, in situ Chl a averaged 5.4 \pm 1.6 and 1.9 \pm 0.3 μ g l⁻¹, 397 respectively in summer 2019, and 1.3 ± 0.3 and $3.4 \pm 0.4 \ \mu g \ l^{-1}$, respectively in winter 2020. At 398 399 station c during spring 2019, an important macroalgae development (Enteromorpha sp. and 400 Ulva sp.) was observed in the subsurface waters and on sediments in the marsh up to autumn 401 (Fig. A1). On the contrary, at station d in 2019 and 2020, no macroalgae development occurred, 402 allowing for the growth of Ruppia sp. seagrasses in the marsh (Fig. A1).

403 At station *Filiere W* in 2018, over the biweekly measurement periods, water temperature 404 values varied from 7.5 (winter) to 21.7 °C (summer) whereas at station a over our seasonal 24-405 hour cycles (high-frequency measurements), values varied from 9.1 (winter) to 26.9 °C 406 (summer). Along the studied aquatic continuums, the water temperature varied significantly 407 between stations a and b (Mann-Whitney test, p < 0.05) and between stations a and c (p < 0.05) 408 but no significant variation was recorded between stations b and c (p = 0.23) (Table 2). Salinity 409 ranged from 28.9 (winter) to 35.4 (autumn) at station Filiere W, whereas values varied from 410 31.4 (winter) to 35.7 (autumn) at station a, from 27.5 (winter) to 36.9 (autumn) at station b, 411 from 27.0 (winter) to 42.6 (summer) at station c and from 21.3 (winter) to 38.4 (autumn) at 412 station d with higher salinity gradients at stations c and d (artificial salt marshes) than at stations 413 a and b (tidal estuary and channel). In average over the year 2018, the tidal estuary (station a) 414 was slightly oversaturated in oxygen compared to the atmosphere ($106 \pm 14\%$) with O₂-sat. 415 values ranging between 70% (LT during dawn) and 147% (receving tide the day) during the 416 summer cycle (Fig. 3). Along the continuum, the channel (station b) was close to the 417 saturaturion value with the atmosphere over the same measurement periods in 2018 with a lower 418 maximum value (120%) than at station a measured in summer (Fig. 4). Larger amplitudes of 419 oxygen saturation percentages were recorded in the artificial salt marshes with O₂-sat. values 420 ranging from 36 to 176% at station c (over the summer cycle; Fig. 5) and from 49 to 150% at 421 station d (over the summer cycle; Fig. 6). The annual percentage of pCO₂ undersaturation 422 compared to the atmosphere were 47.8%, 15.9%, 64.9% and 85.9% at stations a, b, c and d423 respectively, with a strong annual CO_2 oversaturation at station b (Table 2). Moreover, along 424 aquatic continuums, the greatest amplitude in water pCO_2 was recorded at station c with values 425 varying between 6 ppmv in spring 2019 (macroalgae bloom and undersaturated in CO₂) and 426 721 ppmv in autumn 2019 (macroalgae degradation and oversaturated in CO₂).

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428 **3.2. Spatial and seasonal dynamics of studied continuum waters**

429 Significant spatial variations in water pCO₂ and associated biogeochemical parameters 430 (salinity, temperature, turbidity and DO) were recorded along shelf – estuary – marsh water 431 continuums (Kruskall-Wallis tests, p < 0.05; Table 2). The PCA reveals that stations are distinguished vertically according to pCO₂, turbidity and oxygen saturation (pCO₂ and O₂-sat. 432 433 negatively correlated) within PC2 explaining 35.5% of the total variance (Fig. 2). We can cleary 434 observe that the water pCO₂ values measured biweekly at station *Filiere W* were higher than 435 the station a values measured over our 24-hour cycles (Fig. 2). On average over the year, the 436 station b channel had the highest pCO₂ values (462 ± 51 ppmv) compared to the three other 437 stations studied (418 \pm 57, 335 \pm 214 and 293 \pm 113 ppmv, respectively at stations a, c and d; 438 Fig. 2). Station b was also characterized by the highest water turbidity with values ranging from 439 1.6 to 59.7 NTU (median of 8.2 NTU) whereas at station a, the values only varied from 0.1 to

440 32.6 NTU (median of 1.9 NTU). The PCA also shows seasonal data are distinguished 441 horizontally according to salinity and temperature within PC1 explaining 42.3% of the total 442 variance (Fig. 2). Generally, for each station, the highest and lowest temperature and salinity 443 values were measured in summer and winter, respectively (Fig. 2) except at station *a* where the 444 lowest salinity values were recorded in spring (Table 2). Within each station, significant 445 differences in the temperature and salinity values were recorded at the seasonal scale (Kruskall-446 Wallis tests, p < 0.0001).

447 Along the continuums, the PCA recorded contrasted seasonal variations of water pCO₂, 448 particularly between the two artifical salt marshes (Fig. 2). At the station Filiere W continental 449 shelf in 2018, no significant differences in water pCO₂ were recorded at the seasonal scale 450 (Kruskall-Wallis test, p = 0.13), although the highest and lowest seasonal means were recorded 451 in winter and spring, respectively (Table 2). At the station a estuary in 2018, water pCO₂ 452 showed the same seasonal pattern decreasing from winter (441 ± 21 ppmv) to spring-summer 453 $(390 \pm 40 \text{ and } 385 \pm 60 \text{ ppmv}, \text{ respectively})$ before increasing in autumn (460 ± 58 ppmv), 454 whereas the station b channel showed lower seasonal variations over the same measurement 455 periods (Table 2 and Fig. 2). In contrast to the tidal stations (*Filiere*, a and b), artificial salt 456 marshes (c and d) showed larger seasonal pCO_2 variations (Fig. 2). Station c waters were 457 undersaturated in CO₂ with respect to the atmosphere both in spring 2019 (135 \pm 165 ppmv), 458 summer 2019 (242 \pm 116 ppmv) and winter 2020 (343 \pm 87 ppmv) but oversaturated in autumn 459 2019 (622 \pm 57 ppmv). At the same time and on average, station d was undersaturated in CO₂ 460 in summer, autumn and winter with the largest water CO₂ undersaturation recorded in autumn 461 $(155 \pm 30 \text{ ppmv})$ in contrast to station c. At each studied station (a, b, c and d), water pCO₂ 462 significantly differed at the seasonal scale (Kruskall-Wallis tests, p < 0.001), except for station 463 *a* between spring and summer (Dunn's post-test, p > 0.999).

464 At stations a and b in 2018, the same seasonal NpCO₂ variations were observed, decreasing 465 from winter (595 and 624 ppmv, respectively) to summer (296 and 347 ppmv, respectively) and then increasing towards autumn (420 and 439 ppmv, respectively). At station c, the seasonal 466 467 mean NpCO₂ value increased sharply from summer 2019 (193 ppmv) to autumn 2019 (630 468 ppmv) and then decreased towards winter 2020 (441 ppmv), contrarily to station d where, 469 values decreased from summer (286 ppmv) to autumn 2019 (160 ppmv) before increasing 470 towards winter 2020 (453 ppmv) to reach a similar seasonal NpCO₂ value as station c. 471 Regarding temperature effects on water pCO_2 , the lowest and highest seasonal TpCO₂ values 472 were measured in winter and summer, respectively, with seasonal TpCO₂ values followed 473 systematically seasonal water temperature variations. At stations a and b, weak seasonal pCO_2 474 variations were recorded with the TpCO₂ effects offset by the NpCO₂ effects, particularly in 475 winter and summer. Conversely, in artificial salt marshes, the water NpCO₂ and pCO₂ values 476 followed the same seasonal patterns in spring, summer and autumn at station c and only in 477 autumn at station d (Fig. A4).

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3.3. Biogeochemical variations at diurnal and tidal scales

Over all 24-hour measurement cycles, the measured parameters showed strong variations at the diurnal/tidal scales (Figs. 3-6). At station *a* in winter, only small tidal salinity variations were measured (31.9 - 32.9; Fig. 3), whereas at station *b* over the same measurement period, the largest tidal salinity variations occurred with values varying from 27.5 at LT to 32.5 at HT (Fig. 4). On the contrary, at stations *a* and *b* during summer and autumn, lower salinity gradients were measured and salinity values were higher at LT than at HT (Figs. 3 and 4). In the two artificial salt marshes, the highest salinity gradients were recorded in summer with 487 values decreasing through coastal water inflows from 42.6 to 33.5 at station c (Fig. 5) and from 488 38.1 to 33.8 at station d (Fig. 6). During the other seasons at the diurnal/tidal scales, salinity 489 values varied more slightly with, for instance, a difference in salinity units of 2.7, 0.7 and 1.1 490 at station c in winter, spring and autumn, respectively (Fig. 5). It should be noted that, at station 491 c, coastal water inflows from station b led to an increase in salinity only in winter (Fig. 5); in 492 autumn, a rainfall event occurred (5 mm) between 14:00 and 15:00 leading to a decrease in 493 salinity of 1 unit, an increase in turbidity from 2.7 to 7.3 NTU (peak at 15.7 NTU) but without 494 any water pCO₂ variation (Fig. 5). At station d, turbidity and salinity did not vary both in autumn 495 and winter cycles (Fig. 6).

496 The largest diurnal/tidal variations in the water pCO₂ and DO concentrations occurred 497 during summer with pCO₂ gradients of 255, 216, 405 and 258 ppmv at stations a, b, c and d respectively, and DO gradients of 153.2, 152.2, 262.8 and 205.6 μ mol l⁻¹ at stations a, b, c and 498 499 d, respectively (Figs. 3-6). At stations a and b, the low tide periods during the day (LT/D) 500 occurring at dawn showed higher water pCO₂ and lower O₂-sat. values than the low tide periods 501 during the night (LT/N) occurring at dusk, particularly in summer (Figs. 3 and 4). In general, 502 our diurnal cycles showed a decrease in pCO₂ that was negatively correlated to an increase in 503 DO during the daytime (except at station c in spring; Fig. 5) and an opposite pattern during the 504 night-time (except at station c in summer; Fig. 5). For instance, at station c in winter during the 505 marsh confinement, water pCO₂ decreased from 519 to 129 ppmv during the day (from 09:00 506 to 17:00) and increased from 212 to 442 ppmv during the night (from 20:00 to 05:00), while DO increased from 273.4 to 350.0 µmol 1⁻¹ and decreased from 286.5 to 257.5 µmol 1⁻¹, 507 508 respectively (Fig. 5). It should be noted that, at station c in summer, the lowest DO concentration and O₂-sat. percentage were reached during the night (76.9 μ mol l⁻¹ and 35.6%, 509 510 respectively; Fig. 5). At station d, the same diurnal water pCO₂ and DO patterns were also

511 observed (Fig. 6). However, these measured diurnal pCO₂ and DO variations were significantly 512 disrupted once coastal water horizontal advection and artificial salt marsh management 513 practices occurred (Figs. 3-6).

514 Strong tidal variations in water pCO₂ were generally recorded during all seasons except 515 at station d both in autumn and winter (Fig. 6). At stations a and b, incoming tides during the 516 day produced rapid decreases in water pCO₂ from an oversaturation to a slight undersaturation 517 of subsurface waters compared to the atmosphere, particularly in spring (-121 and -167 ppmv, 518 respectively) and summer (-139 and -115 ppmv, respectively; Figs. 3 and 4). Only at station a, 519 receding tides during the day generated an additional decrease in pCO₂ to reach the lowest 520 values measured within each season (270 and 335 ppmv in summer and autumn, respectively; 521 Fig. 3). Then, at station *a* in winter and spring, incoming tides during the night produced slight 522 increases in water pCO_2 whereas in summer and autumn at the same station and at station b 523 over the four seasons, incoming tides during the night-time produced strong increases in pCO_2 524 leading to oversaturation periods (Figs. 3 and 4). Through simultaneous salinity measurements 525 at stations b and c and a cross-correlation function analysis, we estimated the horizontal 526 advection of water masses from station b to station c during a spring tide period at 90 minutes 527 (periods from 29/06 to 08/07/2019). At station c during the night, higher pCO₂ were measured 528 at HT than at LT (i) in spring $(363 \pm 85 \text{ and } 16 \pm 5 \text{ ppmv} \text{ at HT/N} \text{ and LT/N}, \text{ respectively}), (ii)$ 529 in summer (295 \pm 46 and 258 \pm 37 ppmv at HT/N and LT/N, respectively) and (iii) in winter 530 $(431 \pm 6 \text{ and } 323 \pm 53 \text{ ppmv at HT/N and LT/N, respectively; Fig. 7)}$. The same tidal pCO₂ 531 pattern was also recorded at this station in summer 2019 during the day (323 \pm 88 and 197 \pm 532 141 ppmv at HT/D and LT/D, respectively; Fig. 7). In spring, the rewilded artificial marsh 533 (station c) recorded very low water pCO_2 values both the day and the night during the marsh 534 confinement but coastal water inflows from the station *b* channel instantly produced a large and 535 rapid increase in water pCO_2 (+390 ppmv) mostly within a two-hour period (Fig. 5).

536 For all 24-hour cycles, there were strong positive correlations between pCO₂ and NpCO₂ 537 showing that non-temperature effects have a strong control over water pCO₂ at the diurnal/tidal 538 scale along the two studied continuums (Figs. 3-6). Similarly, water pCO₂ values were strongly 539 negatively correlated with oxygen saturation, especially in autumn with diurnal correlations 540 between pCO₂ and O₂-sat. ranging from -0.67 (winter) to -0.97 (autumn) at station a, from -541 0.63 (summer) to -0.87 (autumn) at station b, from -0.54 (winter) to -0.86 (autumn) at station c542 and from -0.59 (winter) to -0.84 (summer) at station d (Figs. 3-6). At the artificial salt marsh 543 stations, correlations between pCO_2 and in situ Chl *a* were -0.62 and -0.47, respectively in 544 winter and summer at station d (Fig. 6) and -0.41 in winter at station c (Fig. 5). At station c in 545 summer, correlations between pCO_2 vs. DO and pCO_2 vs. Chl a were not significant but the 546 pCO₂ vs. salinity coefficient was -0.95 (Fig. 5). At station b, significant linear regressions were 547 calculated between pCO₂ and salinity, i.e. negative in spring ($R^2 = 0.49$) and positive in summer $(R^2 = 0.53; Fig. 7)$. Lastly, at station c, a negative pCO₂ vs. salinity regression was also 548 calculated both in spring and summer ($R^2 = 0.83$ and $R^2 = 0.87$, respectively) with a positive 549 water pCO₂ vs. water height correlation ($R^2 = 0.72$) observed at station c in summer (Fig. 7). 550

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552 **3.4.** Air-water CO₂ flux variations and associated salt marsh metabolism

553 Mean air-water CO₂ fluxes (FCO₂) according to the W92 parametrization were 554 estimated to be -0.01 ± 0.22 , 0.22 ± 0.40 , 0.18 ± 1.37 and -1.22 ± 1.71 mmol m⁻² h⁻¹ at stations 555 *a* (sink), *b* (source), *c* (source) and *d* (sink), respectively, whereas the downstream point of each 556 studied continuum (station *Filiere W*) behaved as a net CO₂ source (0.30 ± 1.04 mmol m⁻² h⁻¹). 557 Large seasonal variations were observed at the four studied stations (Fig. 8). On average, station 558 a showed positive FCO_2 values in both winter and autumn (slight CO_2 source) but negative 559 means in spring and summer (slight CO_2 sink; Table 3 and Fig. 8). At the station b channel, 560 positive FCO₂ values were estimated, with maximun and minium FCO₂ mean values occurring in winter $(0.52 \pm 0.65 \text{ mmol m}^{-2} \text{ h}^{-1})$ and summer $(0.06 \pm 0.08 \text{ mmol m}^{-2} \text{ h}^{-1})$, respectively (Fig. 561 562 8). Station c behaved as a CO_2 sink in spring, summer and winter whereas it emitted large quantities of CO₂ to the atmosphere in autumn (2.03 \pm 1.17 mmol m⁻² h⁻¹; Fig. 8). FCO₂ values 563 at station c varied between -3.00 and 0.03 mmol $m^{-2} h^{-1}$ in spring and between 0.61 and 4.61 564 mmol m⁻² h⁻¹ in autumn. Lastly, station *d* behaved as a CO₂ sink in summer, autumn and winter 565 with the largest atmospheric CO₂ uptake in autumn (-3.43 \pm 1.09 mmol m⁻² h⁻¹) where fluxes 566 varied between -6.03 and -1.79 mmol m⁻² h⁻¹ (Fig. 8). At the global scale (stations a, b, c and 567 *d*), the Fier d'Ars system behaved as a CO₂ sink over the year (-0.20 \pm 1.32 mmol m⁻² h⁻¹) with 568 569 (i) the lowest FCO₂ values recorded during the growing season in spring and (ii) higher FCO₂ 570 values during the night than during the day.

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572 **4. Discussion and conclusions**

4.1. Water pCO₂ dynamics according to the typologies of the shelf – estuary – marsh continuums

Along the studied continuum stations from the coastal ocean to artificial salt marshes, significant water pCO_2 variations were recorded at the spatial scale (Fig. 9). A strong influence of ecosystem typology (continental shelf, estuary, marsh) on inorganic C dynamics was observed, as described more generally by Bauer et al. (2013) for the coastal ocean. In 2018 and 2019, the continental shelf (station *Filiere W*), influenced by the Aiguillon Bay and its 580 associated watershed, was characterized by salinity ranges below those from the Atlantic Ocean 581 (35.6; Vandermeirsch 2012). This was confirmed by the French Phytoplankton Monitoring 582 Network REPHY for the 2000-2017 period with values between 26.2 and 35.8 (Fig. A3; Belin 583 et al. 2021). At station Filiere W, the lowest salinity values were measured in winter when 584 terrestrial river inputs from the Aiguillon Bay watershed were the highest leading to strong 585 nutrient inputs into coastal waters during this time period (Belin et al. 2021). However, in 2018, 586 the non-significant relationship found between salinity and pCO_2 (p = 0.88) measured biweekly 587 at station Filiere W showed a rather weak influence of terrestrial inputs on pCO₂ dynamics. At 588 this shelf station, phytoplankton blooms generally occur in spring and summer (Coignot et al. 589 2020), inducing an undersaturation of water CO₂, followed by pCO₂ decreases observed at 590 station a along the continuum. Station a, found at the entry of the Fier d'Ars estuary, is 591 influenced by buffered marine waters from the shelf with the percentage of CO₂ oversaturation 592 over a 24-hour cycle varying from 96% (winter) to 18% (spring/summer). This decrease in 593 water pCO₂ could be attributed to phytoplankton development in the estuarine waters during 594 this period. The fact that the seasonal NpCO₂ (non-temperature effects on pCO₂) varied from 595 heterotrophy in winter to autotrophy in summer confirmed the phytoplankton activity on the 596 pCO₂ dynamics. Another study carried out close to station a measured Chl a concentrations between 0.2 (winter) and 3.5 μ g l⁻¹ (spring/summer) and a net Chl *a* export suggestive of a net 597 598 primary production within this tidal estuary (Bel Hassen 2001). Moreover, microphytobenthos 599 (MBP) may also contribute to water CO_2 undersaturation at station *a* and to the overall water 600 column Chl a concentration through tidal resuspensions as shown by Savelli et al. (2019) in a 601 nearby intertidal zone. Due to the small insulary catchement area (1200 ha) consisting only of 602 salt marshes (no terrestrial water input), the CO_2 dynamics in the Fier d'Ars estuary (station *a*) 603 is different from other estuaries worldwide (Borges & Abril 2011). Low water pCO₂ values

(270 - 569 ppmv) with no significant relationship between pCO₂ and salinity were found, like 604 605 in the present study and elsewhere on the Atlantic coast of the United States (Jiang et al. 2008), 606 thereby enabling potential CO₂ sink behaviours (Maher & Eyre 2012). Similarly, the marine-607 dominated estuary of Sapelo Sound (USA) was also characterized by lower water pCO₂ values than river-dominated estuaries (Borges & Abril 2011). However, the water pCO₂ values 608 609 measured at this estuary (390 - 2400 ppmv) were higher than those measured at station a in our 610 study due to high bacterial mineralization rates of organic carbon (OC) produced by Spartina 611 in nearby tidal salt marshes (Jiang et al. 2008). On the contrary, river-dominated estuaries 612 characterized by strong freshwater inputs generally show much higher pCO₂ values. The 613 heterotrophic status due to the microbial degradation of OC from rivers produces large CO₂ 614 degassing into the atmosphere (Borges & Abril 2011, Bauer et al. 2013, Najjar et al. 2018) as 615 observed nearby in the Aiguillon Bay (Coignot et al. 2020), the Gironde estuary (Frankignoulle 616 et al. 1998) and the Loire estuary (Abril et al. 2009).

617 Overall, the studied channel (station *b*) between the tidal estuary and artificial salt marshes 618 (Fig. 9) showed longer periods of CO₂ oversaturation over the year 2018 (90, 73 and 100% in 619 winter, spring/summer and autumn, respectively) with significantly higher water pCO₂ values 620 than those at the station a estuary. At station b, strong hydrodynamic forcings during incoming 621 and receding tides produced more turbid waters due to organic matter (OM) resuspension from 622 channel muds (Fig. 4). Then, it probably limited the activity of primary producers 623 (phytoplankton, algae, plants) and, on the contrary, favoured heterotrophic processes 624 (Polsenaere et al. 2022). In the Fier d'Ars waters close to station b, Tortajada (2011) measured 625 POC/Chl a > 200 on average over the year and POC/Chl a > 600 in autumn. This may confirm 626 a microbial loop-type trophic network in channel waters (Tortajada 2011) and a potential 627 influence on the C dynamics leading to OM mineralization processes from MPB and, in turn,

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water CO₂ oversaturation periods measured at station b. However, channel waters showed 628 lower pCO₂ values compared with other coastal channel systems probably due to very low 629 630 terrestrial water inputs upstream/downstream over the Fier d'Ars estuary. In the Arcachon 631 lagoon, Polsenaere et al. (2022) recorded at a tidal channel similar to stations a and b according 632 to typology, long periods of CO₂ oversaturation with seasonal means ranging from 461 ± 14 in 633 July 2008 to 530 \pm 39 ppmv in September 2009. Another study showed that the Sancti Petri 634 Channel waters and the adjacent salt marsh system linking the Atlantic Ocean to the Cadiz Bay 635 were mainly CO₂ oversaturated similarly to our station b, with an annual pCO₂ mean of 564 \pm 636 134 ppmv (281 - 862 ppmv), due to OM diagenetic processes in mudflats that could constitute 637 a DIC source in the water column and thus increased water pCO₂ (Burgos et al. 2018). Within 638 the Duplin River salt marsh-estuary coastal system, higher summer pCO₂ values and DIC 639 concentrations were recorded at low tide in channel waters (12000 ppmv and 4300 μ mol 1⁻¹, 640 respectively) than at high tide in marsh waters (1600 ppmv and 2200 µmol l⁻¹, respectively; 641 Wang et al. 2018).

642 Contrary to tidal stations (a and b), significant and more pronounced periods of CO_2 643 undersaturation were recorded in artificial salt marsh waters (76 and 87% at stations c and d, 644 respectively), once again indicating that marsh typology has a strong influence on CO₂ 645 dynamics. In shallow coastal wetlands such as artificial salt marshes, lower hydrodynamic 646 conditions and longer water residence times promote the development of primary producers 647 such as macrophytes and phytoplankton and as a result, biological CO₂ uptake (Wang & Cai 648 2004, Bauer et al. 2013, Tobias & Neubauer 2019). The artificial salt marsh waters studied here 649 showed lower pCO₂ values and longer CO₂ undersaturation periods, particularly in spring and 650 summer due to strong macrophyte activity (i.e. macroalgae at station c and seagrasses at station 651 d) compared with other marshes and wetland types (Cotovicz Jr. et al. 2015, Wang et al. 2018,

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Berg et al. 2019). By comparison, Ternon et al. (2018) measured higher water pCO_2 in nearby 652 653 fresh- and brackish-water artificial marshes on the French Atlantic coast in summer 2018 with 654 values varying between 227 and 1925 ppmv and between 349 and 2000 ppmv, respectively. 655 Unlike seagrass beds (Mcleod et al. 2011), fast-growing macroalgae developing in coastal 656 wetlands such as at station c have a limited capacity to store C over the long-term. However, 657 other studies have shown their potential contribution to coastal blue C by (i) storing large OM 658 quantities in their living biomass through their high primary production (Raven 2018) and (ii) 659 transferring it to adjacent systems through tides and storage in coastal sediments (Duarte & 660 Cebrián 1996, Hill et al. 2015, Krause-Jensen & Duarte 2016).

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4.2. Marsh management practices module temporal carbon dynamics

663 Management practices at the studied artificial salt marshes (stations c and d) correspond to 664 specific water lock management approaches linked to economic activities. They can strongly 665 modulate coastal water fluxes from the Fier d'Ars estuary (station a) and thereby influence 666 salinity and marsh pCO₂ dynamics. At station c in 2019, the specific management practice undertaken by the NNR (see M&M section), along with higher air temperatures and lower 667 668 precipitations in summer, produced favourable conditions for macroalgae development from 669 early spring to late summer favoured by low marsh hydrodynamics and large temperature 670 fluctuations (Newton & Thornber 2013). To explain the observed macroalgae bloom, we also 671 assumed there were excess nutrient inputs, as described for other coastal ecosystems (Teichberg 672 et al. 2010). Nearby marsh aquafarming activities occurring upstream from the Ré Island 673 watershed that can communicate through the station b channel to station c may result in high 674 nutrient intake and could explain the macroalgae development (Paticat 2007, Tortajada 2011).

At station b in September 2018, DIN > 55 μ mol l⁻¹ and DIP > 5 μ mol l⁻¹ were measured at 675 676 ebbing tide (unpublished results). Moreover, shelf waters influenced by terrestrial inputs (see 677 the above paragraphs in the Discussion) could also lead to high nutrient inputs at marsh station 678 c through incoming tides. At station *Filiere W*, NO_3^- values ranged between 29 and 107 µmol l^{-1} in winter 2019 and between 0.9 and 17.4 µmol l^{-1} in spring 2019 (Belin et al. 2021). 679 680 Consequently, at station c from spring, macroalgae communities probably prevented the 681 development of phytoplankton and seagrasses by marsh nutrient depletion and by light 682 limitation in the water column (Gouazé 2019). However, these fast-growing macrophytes 683 induce intense CO₂ uptake as observed in spring and summer 2019 (Fig. 5). During this period, 684 strong NpCO₂ effects confirmed the major influence of macroalgae biological autotrophy on 685 inorganic C dynamics (Fig. A4); large CO₂ undersaturation periods over tidal/diurnal cycles 686 were maintained although modulated by occasional coastal water inflows under weak tidal 687 amplitudes from station b into the marsh (Fig. 5). This result is confirmed by higher salinity 688 values at station c than at station b measured in spring and summer during our sampling periods 689 (Table 2). Other strong CO_2 undersaturation periods were recorded at station c in summer 2019 690 with a water pCO₂ mean of 297 \pm 150 ppmv over a 96-hour cycle in early August (2 < pCO₂ < 691 723 ppmv; unpublished results). Therefore, these marsh waters were characterized by very low 692 pCO₂ values rarely observed over other similar wetland typologies (Borges 2003, Wang et al. 693 2018, Burgos et al. 2018, Berg et al. 2019). On the contrary, in autumn 2019, macroalgae 694 degradation probably by microbial mineralization processes produced the highest pCO_2 values 695 and largest water oversaturation periods recorded in the marsh waters. This result can be explained by high NH_4^+ concentrations recorded in November 2019 (62 µmol l⁻¹; unpublished 696 697 results) and the lowest oxygen saturation values (Table 2).

698 Contrarily to station c, the working artificial marsh (station d) is managed for salt production 699 in the upstream ponds along the continuum (Fig. A2) and is directly connected to the Fier d'Ars 700 estuary (station a) with no channel connecting the two (Fig. 9). Salt production requires a subtle 701 lock hydraulic management of the salt marsh depending on the frequency of the coastal water 702 supplies that are mainly controlled by the salt manufacturer and meteorological conditions 703 (rainfall, sunshine and wind) for evaporation (Paticat 2007). Therefore, contrary to station c, 704 coastal water inflows to station d were generally performed sparingly with relatively small daily 705 volumes to limit these water mixing effects (i.e. rapid accumulation of large water volumes 706 through rainfall events or spring tides that stop the increase in temperature and salinity of the 707 water already present in the marsh; Paticat 2007). At station d in summer 2019, the measured 708 water pCO_2 values were significantly higher than those at the same period at station c but were 709 very similar to those from station a in summer 2018 (Table 2). This result could be attributed 710 to low activity of the primary producers during this period dedicated to salt production and 711 confirmed by the significant thermal effects on water pCO₂ observed at this station unlike at 712 station c (Fig. A4). A higher frequency of repetitive and direct water inflows through a specific 713 lock management approach from station a to station d (similar mean summer salinity values at 714 stations a and d; Table 2) could also explain these higher pCO_2 values. In autumn 2019 and 715 winter 2020, lower hydrodynamic conditions due to lock closure (the salt farming activity was 716 at a standstill; see M&M section) may have led to lower water turbidity values, lower nutrient 717 inputs from shelf waters and then growth of Ruppia sp. seagrasses and phytoplankton in the 718 marsh instead of macroalgae, thereby producing very low pCO₂ values.

Anthropogenic management practices in artificial salt marshes can therefore strongly influence the contribution and turnover of macrophytes and, consequently, the marsh CO_2 behaviour (sink/source). Due to eutrophication in the rewilded marsh (station *c*), development 722 of the macroalgae communities favoured an atmospheric CO₂ sink during the growing season 723 but an overall annual net atmospheric CO₂ source through their degradation. Their presence 724 prevents phytoplankton blooms and the establishment of seagrass beds as well as economic 725 activities requiring specific management practices. Our work suggests a confinement of 726 artificial marshes in winter and a drying up like at station d to avoid the nutrient inputs and the 727 macroalgae development favouring rather slow-growing macrophytes such as seagrasses which 728 could ultimately contribute to blue C sequestration as more generally described by Mcleod et 729 al. (2011). Similarly, other studies have suggested that the coastal ecosystem management by 730 reducing anthropogenic nutrients could favour blue C ecosystems such as seagrasses, salt 731 marshes and C sequestration (Macreadie et al. 2017, Palacios et al. 2021). Few studies on the 732 links existing between the functioning and C biogeochemical processes of artificial salt marshes 733 have been carried out as in our study. For instance, in Mediterranean poly-euhaline lagoon 734 waters, Le Fur et al. (2018) confirmed that eutrophication levels can strongly favour perennial 735 seagrass species and, to the contrary, fast-growing macroalgae in oligotrophic and eutrophic 736 waters, respectively.

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4.3. Influence of biological activity on diurnal water pCO₂ dynamics

At our studied continuum stations (*a*, *b*, *c* and *d*), negative correlations between water pCO₂ and DO, associated with large oxygen saturation ranges and strong NpCO₂ effects on the measured pCO₂ values, were calculated at the diurnal scale, particularly at the artificial marsh stations (Figs. 5 and 6). It partly demonstrates the strong biological influence of both autotrophic (macrophytes and phytoplankton) and heterotrophic (bacteria) organisms on the diurnal dynamics of inorganic C. Dai et al. (2009) confirmed that C biogeochemical processes 745 in coastal environments such as the Fier d'Ars system are generally controlled by non-746 temperature effects (biological and tidal effects) compared to more open systems like the 747 Atlantic Ocean. Several studies have shown a major biological control on diurnal pCO₂ 748 variations in coastal systems such as the temperate Bay of Brest (France; Bozec et al. 2011), 749 the Arcachon tidal flat (France; Polsenaere et al. 2022), a shallow subtropical estuary in Tampa 750 Bay (USA; Yates et al. 2007) and the tropical coastal embayment at Guanabara Bay (Brazil; 751 Cotovicz Jr. et al. 2015). At the Fier d'Ars estuary (station *a*) in spring and summer 2018, we 752 calculated non-significant correlations between the measured pCO_2 and estimated TA values 753 and between the measured pH and estimated TA values. We also calculated significant positive correlations between the measured DO and pH values ($R^2 = 0.75$ and $R^2 = 0.50$, respectively; p 754 < 0.0001; regressions not shown) and between the measured DO and pCO₂ values (R² = 0.86 755 756 and $R^2 = 0.91$, respectively; p < 0.0001; regressions not shown). These results confirmed the 757 significant control of the photosynthesis vs. respiration balance and the weaker influence of the carbonate system on diurnal pCO₂ variations as shown by Yates et al. (2007) in Tampa Bay. At 758 759 station c, in spring and summer 2019 during the day, macroalgae CO₂ uptake induced large CO₂ 760 undersaturation periods associated with high oxygen saturations (> 170%) in the water column; 761 during the night, macroalgae respiration consumed large oxygen quantities and therefore very 762 low oxygen saturations were observed (< 60%; Fig. 5). In winter 2020, during the marsh 763 confinement, the highest diurnal variations in water pCO_2 were recorded with a pCO_2 decrease 764 of 386 ppmv during the day and an increase of 119 ppmv during the night; again, this may 765 confirm significant photosynthesis and respiration processes in these shallow marsh waters 766 (Fig. 5). By comparison, in a Zostera marina meadow (South Bay, USA) in spring and summer 767 2015, Berg et al. (2019) measured similar diurnal fluctuations of water pCO₂ that were directly 768 controlled by seagrass metabolism with diurnal ranges of 528 and 603 ppmv in late April and

late June, respectively. At station *d*, during periods when the lock was closed, the temperature and non-temperature effects on water pCO₂ were 9 and 115 ppmv, respectively in autumn 2019 and 34 and 134 ppmv, respectively in winter 2020 showing also an influence of biological processes on the diurnal pCO₂ dynamics (Fig. 6). In tidal estuaries and salt marshes, diurnal variations in water pCO₂ due to biological influence must therefore be measured and better accounted for CO₂ budget studies to avoid overestimating their CO₂ sink potential through daytime pCO₂ measurements only (Cotovicz Jr. et al. 2015).

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4.4. Biogeochemical station interconnections through tidal advection

778 Along the studied aquatic continuums, horizontal advection can significantly control water 779 pCO₂ variations at the tidal scale. As seen during the present study, flooding of the Filiere W 780 shelf waters induced a salinity increase and a decrease in winter and from spring to autumn, 781 respectively, at stations a, b, c and d along with significant water pCO₂ variations. For instance, 782 at tidal stations (a and b) in spring and summer during the day, flooding of the shelf waters 783 created a significant decrease in water pCO₂ since the advected waters were CO₂ undersaturated 784 compared with estuarine waters (Figs. 3 and 4). This observation was confirmed by significant 785 negative correlations calculated between pCO₂ and water heights during this period, 786 highlighting the strong control of tidal rhythm on the coastal waters pCO₂. Along the French Atlantic coast, Polsenaere et al. (2022) also showed a strong control of the tidal rhythm in the 787 788 Arcachon lagoon with higher pCO₂ values measured at LT than at HT during each season in 789 2008 and 2009. Similarly, in the Sancti Petri Channel and its adjacent salt marshes, Burgos et 790 al. (2018) recorded the highest and lowest pCO_2 values (1059 and 754 ppmv) at LT and HT, 791 respectively. Even stronger tidal influences on in situ water pCO₂ (from 1380 to 4770 ppmv

792 between HT and LT) were observed by Borges (2003) in summer 2001 in a mangrove system 793 (Gaderu Creek). Interestingly, at station *a* in summer 2018, a significant water pCO₂ decrease 794 from 370 ppmv (15:00) to 267 ppmv (19:30) associated with an oxygen saturation increase 795 from 120 to 140% (characteristic of macrophyte biological activity) was observed in estuary 796 waters during a receding tide, certainly due to CO₂ undersaturated upstream waters from the 797 productive artificial salt marsh waters (Fig. 3). In spring and summer, marsh waters at station c 798 were largely CO₂ undersaturated due to the high primary production of macroalgae and in 799 particular, the long water residence times; however, more CO₂-enriched water inflows from 800 stations *Filiere W*, a and b instantly produced significant water pCO₂ increases at this marsh 801 station (Fig. 5). Therefore, in each season (except at station d in autumn and winter), variations 802 in marsh water pCO₂ at the tidal scale were dependent on the biogeochemical state of the 803 advected waters downstream from the shelf, estuary and channel with respect to CO₂ saturation.

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4.5. Metabolism assessment of the Fier d'Ars continuums

806 Over the year 2018, the Fier d'Ars estuary (station a) behaved on average as a yearly CO₂ 807 sink close to the atmospheric equilibrium, although a significant sink was measured in the 808 spring (Table 3) due to potential phytoplankton blooms occurring in coastal waters (see 809 previous sections in the Discussion). Conversely, over the same meteorological periods, the 810 channel (station b) was a net annual source from its turbid waters to the atmosphere due to 811 several oversaturation periods, particularly in winter, characterized by the highest wind speed 812 and CO₂ exchange coefficient values (Table 3 and Fig. 8). Over the subtidal Bay of Brest, Bozec 813 et al. (2011) estimated slightly higher air-water FCO₂ values ranging from -0.38 to 0.01 mmol m⁻² h⁻¹ during spring/summer and from 0.04 to 0.91 mmol m⁻² h⁻¹ during autumn/winter (Table 814

815 3). In the Arcachon tidal lagoon (similar to our stations a and b) using the same 24-hour cycle approach, an annual air-water FCO₂ was estimated to be 0.27 ± 0.22 mmol m⁻² h⁻¹, the highest 816 817 and the lowest CO₂ degassing values recorded in September and January 2009, respectively 818 (Polsenaere et al. 2022). In the present study, rewilded artificial salt marsh waters (station c) 819 behaved as a yearly source of atmospheric CO₂ when macroalgae degradation and 820 mineralization processes produced strong CO₂ effluxes to the atmosphere, as in autumn 2019 821 (Fig. 8; see previous sections in the Discussion). On the contrary, the working artificial salt 822 marsh (station d) behaved as the largest yearly CO₂ sink particularly favoured by inexistent 823 tidal variations (lock closure) in the absence of salt-farming activities (from fall to winter; Fig. 824 8). Within the Duplin River salt marsh-estuary coastal ecosystem, both channel and marsh 825 waters degassed CO₂ to the atmosphere and, unlike our studied stations, the highest and lowest sources were recorded in summer (5.50 and 3.90 mmol m⁻² h⁻¹ from channel and marsh waters, 826 respectively) and in winter (0.70 and 0.60 mmol m⁻² h⁻¹ from channel and marsh waters, 827 respectively), respectively (Wang et al. 2018). Overall, the whole Duplin system emits more 828 829 CO₂ into atmosphere than the Fier d'Ars system, probably due to its more intense estuarine 830 heterotrophic metabolism.

831 In autumn, the lack of significant variations in wind speeds between stations a and b in 2018 832 and between stations c and d in 2019, whereas atmospheric CO₂ exchanges significantly 833 changed, highlighting the predominance of air-water CO₂ gradients in the control of CO₂ flux 834 directions either as a sink or a source (Table 3). However, at the seasonal scale, the turbulence 835 processes measured at the air-water interface also played an important role in flux variability 836 and magnitude; for instance, at station *a* between spring and summer and at station *b* between 837 winter and summer, wind speed variability produced significant FCO₂ variations although no 838 significant air-water CO₂ gradients were measured (Table 3). Atmospheric exchanges in marshes are therefore dependent on the CO₂ saturation state of the water column (air-water
gradient) considering that the wind only acts as a driver of the flux (Polsenaere et al. 2022).
Moreover, the (site specific) methodological calculations and associated differences chosen for
the exchange coefficient parameterizations (higher fluxes with RC01 than with W92 methods;
Table 3) may produce even more contrasts in the estimated air-water FCO₂ values (Cotovicz
Jr. et al. 2015, Polsenaere et al. 2022).

845 By scaling-up and considering stations a and b together along the continuum (estuary and channel), an annual source of atmospheric CO₂ of 7.3 g C m⁻² yr⁻¹ was calculated in 2018. The 846 rewilded artificial salt marsh (station c) emitted 17.5 g C m⁻² yr⁻¹ (702.0 kg C yr⁻¹) to the 847 atmosphere, when the working artificial marsh (station d) absorbed 97.7 g C m⁻² yr⁻¹ (828.6 kg 848 $C yr^{-1}$) from the atmosphere. A larger scale study of C along three shelf – estuary – tidal wetland 849 850 continuums on the Atlantic coast of the United States also showed strong spatial variations in 851 atmospheric CO₂ exchanges with CO₂ uptake for wetland and shelf waters of 5.3 ± 1.5 and 4.0 \pm 0.7 Tg C yr⁻¹, respectively and a CO₂ source from estuarine waters of 4.2 \pm 1.7 Tg C yr⁻¹ 852 853 (Najjar et al. 2018). During our study, contrasting stations along the continuums were sampled 854 via seasonal 24-hour cycles to estimate the air-water CO₂ exchanges. However, longer seasonal 855 measurement periods up to several days would be more representative of the strong temporal 856 variability in k_{660} , water pCO₂ and other biogeochemical parameters. At the Bossys perdus salt 857 marsh (station e; Fig. 1), another flux methodology using the atmospheric EC technique was 858 deployed to continuously measure year-round in situ CO₂ fluxes at the ecosystem scale under real field conditions. Over the year 2020, this preserved salty meadow fixed 483 g C m⁻² yr⁻¹ 859 860 from the atmosphere, which is much higher than the studied artificial salt marshes indicating a 861 potential stronger atmospheric CO₂ sink in tidal salt marshes (Mayen et al. in prep.). However, it is also important to study the whole marsh metabolism taking terrestrial and aquatic 862

compartments into account and distinguishing their respective contributions to atmospheric fluxes and the regional C budgets of the associated marshes. To do this, we measured the atmospheric CO_2 exchanges at the ecosystem scale at both the water-air and soil-air interfaces, biogeochemical parameters of the channel column water, metabolic balance of planktonic communities and horizontal C export during seasonal 24-hour cycles to be more integrative from a C budget point of view over the salt marshes (Mayen et al. in prep.).

869

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