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Influence of typology and management practices on water pCO₂ and atmospheric CO₂ fluxes over two temperate shelf – estuary – marsh water continuums

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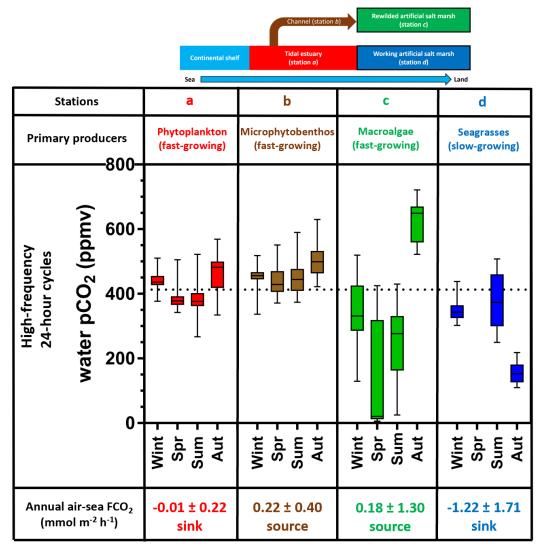
Abstract

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2 Within the coastal zone, salt marshes often behave as atmospheric CO₂ sinks, allowing 3 for blue carbon (C) sequestration associated with intense autotrophic metabolism. However, C 4 dynamics over salt marshes are complex since various biogeochemical processes and fluxes 5 take place at different terrestrial - aquatic - atmospheric exchange interfaces and 6 spatiotemporal scales. This study focuses on seasonal, tidal and diurnal variations of water 7 pCO₂, estimated water-air CO₂ fluxes and controlling factors along two temperate shelf – estuary – marsh continuums. The latter include typical coastal systems (shelf, estuary and 8 9 channel) with artificial salt marshes that have contrasting management practices. Our high-10 frequency (seasonal 24-hour cycles) biogeochemical measurements at the various stations 11 highlighted a strong control of ecosystem typology on inorganic C dynamics with lower water pCO₂ values in the artificial salt marshes, due to stronger biological activity and longer water 12 13 residence times, than in the tidal estuary. In this marine-dominated estuary, water pCO₂ 14 variations (267 - 569 ppmv) were strongly controlled by tidal effects and phytoplankton activity 15 particularly in spring/summer. On the contrary, the greatest amplitudes in water pCO₂ were recorded in the artificial salt marshes (6 - 721 ppmv) due to intense macrophyte activity. 16 However, in the rewilded artificial marsh, spring/summer fast-growing macroalgae produced, 17 18 in turn, strong fall atmospheric CO₂ outgassing from degraded algae waters and thus a net annual source of CO₂ to the atmosphere (17.5 g C m⁻² yr⁻¹). Conversely, management practices 19 20 at the working artificial marsh for salt-farming activity with longer winter water residence times 21 favoured rather slow-growing macrophytes which greatly contribute to the yearly observed atmospheric CO₂ sink (-97.7 g C m⁻² yr⁻¹) and, more generally, blue C sequestration. We suggest 22 23 that artificial salt marsh management can be used to control the contribution of primary producers to marsh C budget as atmospheric CO₂ (sink and/or source). 24

Key words: shelf – estuary – marsh continuums; water pCO ₂ ; air-water CO ₂ fluxes; diurnal
tidal, seasonal scales; marsh management practices; macrophytes.

44 Graphical Abstract



Water management practices and primary producer types influence the marsh CO₂ behaviour (sink/source)

1. Introduction

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Marine coastal environments, which only account for 7% of the global ocean, perform major ecological functions such as primary production, bacterial mineralization, organic matter burial and calcium carbonate deposition (Gattuso et al. 1998). Comprised of mixed continental and oceanic waters, the coastal zone presents a wide diversity of geomorphological types and ecosystems (shelves, estuaries, bays, tidal wetlands) shaping the biogeochemical cycle coupling between the land, ocean and atmosphere (Aufdenkampe et al. 2011, Bauer et al. 2013). On the one hand, these dynamic and heterogeneous ecosystems behave as an active horizontal pipe carrying and processing large quantities of carbon ($\sim 0.90 \pm 0.05$ Pg C yr⁻¹, Cole et al. 2007) from the drainage basin to the open ocean (Cai 2011, Najjar et al. 2018). On the other hand, the coastal zone vertically exchanges large and variable quantities of carbon (C) with both the atmosphere and sediments (Cole et al. 2007, Polsenaere et al. 2012). For instance, at the global scale, continental shelves behave as atmospheric CO₂ sinks and absorb 0.25 ± 0.05 Pg C yr⁻¹ (Bauer et al. 2013, Dai et al. 2022) due to phytoplankton primary production (Cloern et al. 2014). On the contrary, CO₂ supersaturated estuarine waters emit 0.25 ± 0.05 Pg C yr⁻¹ to the atmosphere (Bauer et al. 2013) due to a high mineralization of organic matter from the land (Frankignoulle et al. 1998, Borges & Abril 2011). These atmospheric C exchanges within the coastal zone are heterogeneous (Borges et al. 2005) and need to be better taken into account in regional and global C budgets (Polsenaere 2011, Najjar et al. 2018). For instance, coastal wetlands, including salt marshes located along inner shelf – estuary – marsh continuums, absorb 0.55 ± 0.05 Pg C yr⁻¹ from the atmosphere (Bauer et al. 2013) and may play a major role in atmospheric CO₂ uptake and associated organic C burial on Earth (Cai 2011, Mcleod et al. 2011).

In salt marshes, inorganic C dynamics and water pCO₂ in particular are influenced by several physicochemical and biological processes within and between each ecosystem compartment such as tidal exchanges, calcium carbonate precipitation/dissolution, benthicpelagic couplings, air-water exchanges and photosynthesis/respiration balance (Cai 2011, Bauer et al. 2013, Macreadie et al. 2017). Due to high photoautotrophy rates of both aquatic (micro-algae and macrophytes) and terrestrial (vascular plants) primary producers, these highly productive ecosystems mostly behave as net atmospheric C sinks (Schäfer et al. 2014, Artigas et al. 2015, Forbrich & Giblin 2015). For instance, net primary production (NPP) values in Spartina alterniflora salt marshes ranged from 100 to > 2500 g C m⁻² yr⁻¹ (Duarte & Cebrián 1996, Tobias & Neubauer 2019) allowing for significant CO₂ uptake (Wang & Cai 2004). A refractory part of organic C produced through photosynthesis in these vegetated coastal ecosystems can then be largely sequestered in sediments (Chmura et al. 2003) and stored as blue C at a rate of 218 ± 24 g C m⁻² yr⁻¹, and greatly contribute to the regional/global C cycle in comparison with terrestrial ecosystems (Mcleod et al. 2011). Salt marshes also produce and horizontally export significant quantities of C through tidal water advection (Najjar et al. 2018) which could, in turn, strongly influence the C balance of the system itself as well as the estuary and shelf systems (Cai 2011). The "marsh CO₂ pump" hypothesis proposes that atmospheric CO₂ fixation by plants and phytoplankton in marshes and the export of part of the associated C may be one of the major mechanisms making adjacent coastal waters sources of CO₂ to the atmosphere (Wang & Cai 2004). To illustrate this, Wang et al. (2016) used continuous in situ measurements of relevant biogeochemical parameters and water fluxes in an intertidal salt marsh in northeast region of the United States (12300 km²) to estimate the total net CO₂ fixation (NPP) of 13.2 Tg C yr⁻¹. Of this total NPP value, 56% was exported to the coastal ocean through the outwelling phenomenon (horizontal advection) with inorganic and organic C accounting for

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39% and 17%, respectively; this means that the export of C from marsh tides could represent a major proportion (414 g C m⁻² yr⁻¹) in the overall marsh budget (Wang et al. 2016). Nevertheless, despite these major ecological potentials (storm protection, nursery areas, long-term C storage), these interface zones are the most threatened in the world by land-use changes, climate changes and sea level rise as described by Gu et al. (2018) in salt marshes in China. Moreover, coastal eutrophication causes the loss of salt marshes by decreasing the below-ground biomass of plant roots through microbial degradation thereby producing a decrease in the geomorphic stability of marshes (Deegan et al. 2012). Since the 1800s, salt marshes have lost about 25% of their global area with negative effects on the atmospheric CO₂ sink and the associated C sequestration (Mcleod et al. 2011). Their importance as ecosystem service reservoirs has made it possible to implement protection and restoration policies that contribute to their better management and to the development of their ecological and economic potentials (Gu et al. 2018, Adam 2019).

The high heterogeneity in biogeochemical processes within coastal systems at spatial and temporal scales (Cai 2011, Bauer et al. 2013) requires more integrative C process and exchange measurements at the various terrestrial – aquatic – atmospheric interfaces over different time scales (tidal, diurnal and seasonal) to better understand the ecological functioning of these ecosystems facing global changes. Some studies in coastal wetlands such as salt marshes or tidal estuaries have taken water pCO₂ measurements at different temporal scales allowing the study of in situ CO₂ dynamics in relation to other biotic and abiotic processes. For instance, in Guanabara Bay (Brazil), water pCO₂ measured at high frequency, varying from 22 to 3715 ppmv, was negatively correlated with Chlorophyll *a* (Chl *a*) in surface waters, indicating that phytoplankton activity has an influence on CO₂ uptake and, more broadly, on inorganic C dynamics (Cotovicz Jr. et al. 2015). Moreover, Borges (2003) performed a 24-hour cycle in

June 2001 in an intertidal mangrove system (Gaderu Creek, India) and showed that water pCO₂ ranged from 1380 to 4770 ppmv with a strong control of these fluctuations by tides and biological activity (primary production and respiration). Within a small tidal creek of the Duplin River salt marsh-estuary coastal ecosystem (Georgia, USA), water pCO₂ calculated by Wang et al. (2018) showed strong seasonal and tidal/diurnal variations with winter values ranging from 500 to 4000 ppmv between high and low tide respectively, and summer values ranging from 1600 to 12000 ppmv between high and low tide respectively. However, still too few studies have taken high-frequency water pCO₂ measurements in salt marshes at the diurnal scale. These temporal variations in water pCO₂ strongly affect associated air-water CO₂ fluxes (FCO₂) that can, in turn, be estimated from the CO₂ gas transfer velocity, CO₂ solubility in the water and air-water CO₂ gradient (Borges 2003, Crosswell et al. 2017). The atmospheric Eddy Covariance (EC) technique represents an alternative way to directly measure in situ atmospheric CO₂ fluxes at the ecosystem scale (Baldocchi et al. 1988, Baldocchi 2003, Schäfer et al. 2014). This direct and non-intrusive micrometeorological method shows a growing interest in studying the metabolism of coastal ecosystems (sink or source) under real field conditions and to integrate them into regional C budgets (Polsenaere et al. 2012, Van Dam et al. 2021).

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The purpose of this study is to better understand CO₂ dynamics at different temporal scales and locations over two aquatic sea – land continuums along the Atlantic French coast on Ré Island. These continuums include typical coastal systems (shelf, estuary, marsh) such as those studied elsewhere by Cai (2011) and Bauer et al. (2013) with regards to horizontal and vertical C exchanges in the coastal ocean. Unlike tidal salt marshes, which are more generally discussed in the literature, we studied two artificial salt marshes (i.e. salt ponds; Fig. A2) in which water exchanges are controlled by dykes and locks for human uses (biodiversity protection or anthropogenic activities). Through in situ high-frequency measurements of biogeochemical

parameters in waters and estimations of atmospheric CO₂ fluxes from 2018 to 2020, we sought to (1) describe water pCO₂ and associated CO₂ exchange variations at the diurnal, tidal and seasonal scales along the studied aquatic continuums, (2) identify biophysical drivers and the potential role of station typologies and salt marsh management practices on water pCO₂ and CO₂ budgets and (3) contextualize the associated continuum metabolism among other studied systems from a C dynamic and budget point of view.

2. Materials and methods

2.1. Study sites

2.1.1. Estuary (station *a*)

The Fier d'Ars represents a 750-ha maritime area within the French Atlantic Ocean (north of Ré Island), connected to the Breton Sound continental shelf by an opening that is 700 m wide (Fig. 1). It corresponds to a type II temperate tidal estuary according to Dürr et al.'s (2011) coastal system typology. With a maximum water height of 6.5 m, this tidal estuary exchanges between 2.4 and 10.2 million m³ of coastal waters with the adjacent continental shelf depending on the tidal amplitudes (Bel Hassen 2001). At low tide (LT), its subtidal zone (in light blue; Fig. 1) is composed of mudflats (slikke) and tidal salt marshes (schorre) traversed by numerous channels converging toward a main channel measuring 300 m wide (Fig. 1). At high tide (HT), the subtidal zone is flooded by coastal waters up to the dykes (in red; Fig. 1), managed to control water exchanges between the estuary and artificial salt marsh complexes. These artificial salt marshes are old tidal salt marshes divided into multiple ponds mainly located along European coasts for which water residence times (from a few hours to fifteen days according to the management practices; Bel Hassen 2001) were originally controlled for

salt-farming (Tortajada et al. 2011). Nowadays, some of them are intended for other uses with different water management practices based on their socio-economic activities (salt-, oyster- and fish-farming or natural management; Fig. A2). The studied station *a* is located within this tidal estuary, inside the subtidal zone of Fier d'Ars at its entry along the main channel connected to the slikke (Fig. 1).

2.1.2. Channel (station *b*)

Station b is a secondary tidal channel associated to the schorre and located at the back of the Fier d'Ars estuary just before the dyke (250 m wide at HT and 20 m wide at LT; Fig. 1). With a maximum water height of 5.3 m at HT, it is connected to the tidal estuary station a (distance of 1.6 km between stations a and b) enabling the supply of coastal water to artificial salt marshes upstream from the dyke (Fig. 1).

2.1.3. Rewilded artificial salt marsh (station *c*)

For forty years, 121 ha of salt marshes within the Fier d'Ars system have been protected and managed inside a National Natural Reserve (NNR) to encourage a high level of biodiversity of migratory birds and vascular phanerogam and cryptogam developments ($Ruppia\ maritima$ and $Ruppia\ cirrhosa$ in artificial salt marshes and $Zostera\ noltei$ on mudflats; Champion et al. 2012). Within the terrestrial area of the NNR, artificial salt marshes communicate with Fier d'Ars waters (station a) through lock management practices to promote biodiversity protection. To the west, station c (surface area of 40100 m², depth of 60 cm) inside the NNR is a rewilded artificial salt marsh supplied indirectly with coastal waters from the Fier d'Ars estuary by the

station *b* channel (distance of 500 m between stations *b* and *c*; Fig. 1) through a specific lock management ensured by the NNR (former salt farm that has now been rewilded; Fig. A2). From November until March (winter period), the marsh lock is open only at each full moon phase when tidal amplitudes are higher than 70 in order to have the best compromise between saltand fresh-mixing waters (salinity around 30) to make it easier for aquatic fauna to pass from the continental shelf to the marsh. On the contrary, from April to October when tidal amplitudes are lower, the lock is permanently open to favour the development of *Ruppia sp.* seagrass beds in the marsh with salinity varying between 30 and 46 (Champion et al. 2012). In past years, this artificial salt marsh is also characterized by significant macroalgae development (*Enteromorpha sp.* and *Ulva sp.*) at the subsurface and on sediments (Fig. A1) from early spring to late summer each year thereby preventing seagrass development (Champion et al. 2012).

2.1.4. Working artificial salt marsh (station *d*)

To the east of the Fier d'Ars estuary, station d (surface area of 8500 m², depth of 75 cm) also corresponds to an artificial salt marsh upstream from the dyke even though it directly communicates (no channel in between) with the coastal waters of the Fier d'Ars estuary by a dyke lock (distance of 2 km between stations a and d; Fig. 1). This second artificial salt marsh was chosen for its specific lock management practice depending on the salt-farming activities (working marsh) contrary to station c (rewilded marsh). In spring and summer, this working marsh is particularly used for the storage of salt water in the context the salt-farming activity. In this way, the lock is regularly open at this time of the year allowing upstream pond succession supply to produce salt via evaporation between June and September (Fig. A2). Moreover, the use of this artificial marsh for salt-farming activities requires a drying up and a cleaning once a

year in early spring before the start of the salt production period to remove seagrass, macroalgae and organic matter in the marsh (Poitevin, personal communication).

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2.2.5. Other associated stations: continental shelf (station $Filiere\ W$) and tidal salt marsh (station e)

The Fier d'Ars estuary communicates with the Breton Sound which corresponds to a coastal maritime area on the French continental shelf between Ré Island and the continent (Fig. 1), characterized by a surface area of 425 km² and volume of 4920 million m³ (Stanisiere et al. 2006, Soletchnik et al. 2014). The Breton Sound continental shelf exchanges large quantities of salt water with the Atlantic Ocean to the west at each semi-diurnal tidal cycle and it receives fresh water inputs through the Sèvre and Lay rivers within the Aiguillon Bay to the east (annual mean discharges of 44.4 and 14.0 m³ s⁻¹, respectively) depending on hydrodynamic and meteorological conditions (Stanisiere et al. 2006, Soletchnik et al. 2014, Polsenaere et al. 2017). The highest and lowest river water flows were recorded in winter and summer, respectively, influencing the water salinity of the Breton Sound differently. Station Filiere W in the centre of the Breton Sound (Fig. 1) is located in a predominantly marine environment with a freshwater contribution of approximately 30 ml/l (i.e. 3%) from the Sèvre and Lay rivers (Stanisiere et al. 2006, Soletchnik et al. 2014). This additional station, studied by Coignot et al. (2020) using a different measurement strategy, was included in this study to contextualize the influence of the Breton Sound on our studied stations. At each HT, the Breton Sound supplies our studied coastal stations with various water masses based on the tidal amplitudes and seasonal periods along two aquatic sea – land continuums: (1) continental shelf (station Filiere W) – estuary (station a) – channel (station b) – rewilded artificial salt marsh (station c) and (2) continental

shelf (station $Filiere\ W$) – estuary (station a) – working artificial salt marsh (station d) (Fig. 1). Conversely, at each LT, different water masses from the artificial salt marshes are exported (indirectly through the station b channel for station c or directly for station d) to the Fier d'Ars estuary and then to the Breton Sound (Fig. A2).

Since June 2019, an atmospheric Eddy Covariance (EC) station (Campbell Scientific) has been deployed at station e (Fig. 1) to continuously measure in situ CO₂ fluxes at both the marsh-atmosphere and water-atmosphere interface at the ecosystem scale (Mayen et al. in prep). This tidal salt marsh station is located within the maritime NNR area with typical marsh vegetation (*Halimione sp.*, *Suaeda sp.* and *Spartina sp.*) emerged for 70% of time, during low tides and neap tides. In this study, only meteorological parameters (air temperature, rain, wind speed) corresponding to data from our four station measurement cycles were used from the EC station.

2.2. Measurement strategy and probes

 30/01/2018 13:55; 15/02/2018 14:03; 03/03/2018 13:40), four samplings in spring (17/04/2018 14:25; 26/04/2018 12:20; 29/05/2018 13:40; 14/06/2018 14:00); four samplings in summer (28/06/2018 13:05; 11/07/2018 12:45; 09/08/2018 10:45; 11/09/2018 08:50) and five samplings in autumn (26/09/2018 13:45; 09/10/2018 13:25; 24/10/2018 12:50; 08/11/2018 08:15; 11/12/2018 09:50). At stations b and c in summer 2019, the water height, temperature and salinity were simultaneously measured every 10 min. by two STPS sensors (NKE Instrumentation) (unpublished results).

An autonomous pCO₂ underwater sensor (C-SenseTM pCO₂ sensor, PME/Turner Designs), an EXO₂ multiparameter probe (YSI) and a C3-submersible fluorimeter (Turner Designs) were deployed to measure water pCO₂, physicochemical parameters and fluorescence, respectively. Chlorophyll *a* concentrations were derived from the fluorescence data measured by the C3-fluorimeter every 10 min. as a proxy to study the dynamics of phytoplankton biomass over time (Aminot & Kérouel, 2004). This sensor was used only for summer 2019 and winter 2020. The measurement range of the C-Sense probe is 0-2000 ppmv with an absolute accuracy of 60 ppmv (3% of the full scale; Turner Designs). The EXO₂ probe was used to measure the pH (NBS scale), temperature (in °C), salinity, dissolved oxygen concentration (DO in μmol l⁻¹), oxygen saturation percentage (O₂-sat. in %) and turbidity (in NTU). The maximum permissible errors of the different EXO₂ sensors are 0.2 and 0.5 unit for pH and salinity respectively, 0.25 °C for temperature, 5 NTU for turbidity and 25 μmol l⁻¹ for O₂ (YSI). Between each station, the EXO₂ pH sensor was calibrated with three buffer solutions (pH 3, 7 and 10). It was not possible to measure pH values at stations *a* and *b* in 2018 (Table 2).

Water pCO₂ values measured by the C-Sense probe are influenced by the total dissolved gas pressure (TDGP) which corresponds to the total pressure exhibited by all gases within the

water column. When this pressure greatly exceeded the pressure at which the C-Sense was calibrated, the output needed to be corrected. Then, a pCO₂ correction was applied taking both TDGP and atmospheric pressure during sensor calibration (1009 hPa) and the measured pCO₂ by the C-Sense probe (gross values) into account, as per equation (pCO_{2meas}. \times 1009) / TDGP (Turner Designs). Over the fifteen 24-hour cycles performed at the seasonal and spatial scales, the corrected pCO₂ with TDGP were $2.6 \pm 0.9\%$ lower than the raw pCO₂ values.

2.3. Temperature and non-temperature effects on pCO₂ variations

To distinguish between the temperature and non-temperature effects on in situ pCO₂ variations at the seasonal and diurnal scales, TpCO₂ (pCO₂ variations related to temperature physical effects, in ppmv) and NpCO₂ (pCO₂ variations related to non-temperature effects, in ppmv) were calculated respectively, following (Eq. 1) and (Eq. 2) from Takahashi et al. (2002):

$$TpCO2 = pCO2mean \times exp[0.0423 \times (Tobs - Tmean)]$$
 (1)

where T_{obs} and pCO_{2obs} are the temperature and pCO_2 values measured by the probes at each time step (1 min.), respectively. T_{mean} and pCO_{2mean} are the temperature and pCO_2 averaged either at the seasonal (annual mean) or diurnal scale (means per cycle). Whereas $TpCO_2$ is only associated with the physical pump, $NpCO_2$ is associated with biological processes, tidal advection and benthos-pelagos couplings that may be important in shallow coastal systems (Cotovicz Jr. et al. 2015, Polsenaere et al. 2022).

2.4. Calculations of air-water CO₂ fluxes

For all our measurement periods (Table A1), the gas transfer velocity (k_{600}) and hourly CO₂ fluxes (FCO₂) at the air-water interface were estimated following, for instance, Ribas-Ribas et al. (2011) and Polsenaere et al. (2022) in coastal environments. At stations a and b, only air-water FCO₂ during HT (four hours around each HT) were calculated, whereas at stations c and d, all hourly FCO₂ were calculated using the following formula (Eq. 3):

$$309 \quad FCO_2 = \alpha \times k \times \Delta pCO_2 \tag{3}$$

where FCO₂ is the estimated CO₂ fluxes at the air-water interface (mmol m⁻² h⁻¹), α is the CO₂ solubility coefficient in saltwater (mol kg $^{-1}$ atm $^{-1}$), k is the gas transfer velocity of CO₂ (in cm h⁻¹) and ΔpCO₂ is the gradient between mean water and air pCO₂. Water pCO₂ (ppmv) were measured by the in situ C-Sense probe and atmospheric CO₂ concentrations (ppm) were measured by the EC station (station e; Fig. 1) during summer 2019 and winter 2020. For all other periods, atmospheric CO₂ data were obtained from the National Oceanic and Atmospheric Administration (NOAA) Mauna Observatory at the Loa in Hawaii (https://gml.noaa.gov/ccgg/trends/). The CO₂ solubility coefficient (α) depends on water temperature and salinity and was calculated according to Weiss (1974). The gas transfer velocity (k) also significantly controls air-water FCO₂ since it directly takes turbulence processes at the air-water exchange interface into account (Polsenaere et al. 2013). In this study, k (or k_{660}) was calculated according to both Raymond & Cole (2001) (RC01; Eq. 4) and Wanninkhof (1992) (W92; Eq. 5) corresponding to closed environments and more open coastal environments, respectively. These two parametrization methods for the k exchange coefficient were applied to the four stations in order to compare the results.

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For closed freshwater environments (Raymond & Cole 2001),

$$327 k_{600} = 1.91 \times \exp[0.35 \times \text{U}10] (4)$$

328 For more open coastal environments (Wanninkhof 1992),

$$329 k_{600} = 0.31 \times (U_{10})^2 (5)$$

The gas transfer coefficients normalized to a Schmidt number of 600 (k_{600}) obtained with the two parametrization were then converted to the gas transfer velocity of CO₂ at the in situ temperature and salinity (k_{660}) according to Jähne et al. (1987) as per the equation (6):

$$k_{660} = k_{600} / (660/\text{Sc})^{-0.5}$$
 (6)

where k_{660} is the gas transfer velocity of CO₂ at the in situ temperature and salinity (cm h⁻¹) according to the parametrizations of RC01 or W92, U₁₀ is the wind speed normalized to 10 m (m s⁻¹) using the relationship of Amorocho & DeVries (1980) and Sc is the Schmidt number which describes both the water viscosity and the molecular diffusion of the subsurface layer (Bade 2009). In summer 2019 and winter 2020, wind speed data were measured by the EC station at a height of 3.15 m (station e; Fig. 1); for all other periods, wind data were obtained from the "Infoclimat" station on Ré Island (Fig. 1) measured at a height of 10 m (6.20, 4.85, 4.30 and 8.40 km from stations a, b, c and d respectively).

2.5. In situ Chl a concentrations and fluorimeter data calibration

In situ Chl a concentrations were measured from samples collected at stations c and d following Aminot & Kérouel (2004). Water samples (50 ml) were filtrated (47 mm Whatman GF/F filter) and Chl a filters were kept at -18°C up to analysis. In the laboratory, in the dark, pigments were extracted in a 90% acetone solvent at the same time the filter was crushed with a glass rod. After one night of stirring at 4 °C to continue the extraction and a 10-min

centrifugation at 3000 rpm, the supernatant absorbance was measured by spectrophotometry at 665 nm to obtain the Chl a level (its absorbance peak is at 665 nm). Chl a were measured from eight water samples except at station c in spring. Chl a could not be measured at stations a and b in 2018.

For the fluorimeter data, the calibration procedure was applied to derive Chl a concentrations from our 10 min-water fluorescence measurements. Chl a concentrations were calculated through the significant (p < 0.05) linear regressions obtained for each deployment (summer 2019 and winter 2020) between C3-submersible fluorimeter data and the in situ Chl a concentration analysis from sub-surface waters (eight samplings on average per each 24 hourcycle) sampled simultaneously right at the fluorescence probe.

2.6. Statistical tools and analysis

For all measured variables, the data did not respect a normal distribution (Shapiro-Wilk, p < 0.05). Shapiro-Wilk tests and non-parametric comparison tests such as the Mann-Whitney and Kruskal-Wallis tests were carried out with 0.05 level of significance. A Dunn test was used to perform a post-hoc multiple comparison of the Kruskal-Wallis test to detect significant differences among groups. The R-studio software was used to perform the principal component analysis ("FactoMineR" package; Lê et al. 2008) and the Spearman correlation matrices ("corrplot" package; Wei and Simko 2017). A PCA was carried out to study the contribution of the variables to the data at the spatial and seasonal scales. It is based on the mean values of the physicochemical and biogeochemical parameters of water (temperature, salinity, turbidity, dissolved oxygen, oxygen saturation and water pCO₂) for each 24-hour cycle at stations *a*, *b*, *c*

and d. Temporal graphs, linear regressions, boxplots and barplot were performed with the GraphPad Prism 7 software.

Total alkalinity (TA) were estimated at the diurnal scale from the measured salinity, temperature, pH and water pCO₂ using the carbonic acid constant from Mehrbach et al. (1973) as modified by Dickson and Millero (1987), the K_{HSO4} constant from Dickson (1990) and the borate acidity constant from Lee et al. (2010).The CO₂ system calculation programme (version 2.1.) was used to perform these calculations (Lewis and Wallace, 1998).

3. Results

3.1. Biogeochemical overview of the shelf – estuary – marsh water continuums

Over our measurement periods, meteorological conditions for the years 2018 and 2019 were similar with regard to air temperatures. The thermal conditions followed a classical seasonal trend with the highest and lowest temperatures measured in summer (July 2018 and 2019) and in winter (March 2018 and February 2020), respectively (Table 1). However, the months of July 2018, July 2019 and February 2020 were warmer (\pm 2.1, \pm 2.4 and \pm 3.2 °C, respectively) than the reference period (1981-2010). Annual cumulative precipitations were lower in 2018 than in 2019 (786 and 827 mm, respectively); March 2018 and October 2019 were the rainiest months (\pm 82 and \pm 41%, respectively, compared with the 1981-2010 period; Table 1). Salinity values measured by the French Phytoplankton Monitoring Network REPHY at station *Filiere W* as the water source flowing into the two aquatic continuums did not vary significantly between the years 2018, 2019, 2020 and the 2000-2017 period (Kruskal-Wallis test, p = 0.77; Fig. A3).

At the station *Filiere W* continental shelf, the highest and lowest Chl a concentrations were recorded in winter $(2.0 \pm 0.0 \,\mu g \, l^{-1})$ and in summer $(3.0 \pm 1.2 \,\mu g \, l^{-1})$, respectively (Coignot et al. 2020). In the artifical salt marshes, in summer 2019, autumn 2019 and winter 2020 (Table A1), the Chl a concentration varied significantly at the spatial scale (Mann-Whitney tests, p < 0.05). For example, at stations c and d, in situ Chl a averaged 5.4 ± 1.6 and $1.9 \pm 0.3 \,\mu g \, l^{-1}$, respectively in summer 2019, and 1.3 ± 0.3 and $3.4 \pm 0.4 \,\mu g \, l^{-1}$, respectively in winter 2020. At station c during spring 2019, an important macroalgae development (*Enteromorpha sp.* and $Ulva \, sp.$) was observed in the subsurface waters and on sediments in the marsh up to autumn (Fig. A1). On the contrary, at station d in 2019 and 2020, no macroalgae development occurred, allowing for the growth of $Ruppia \, sp.$ seagrasses in the marsh (Fig. A1).

At station *Filiere W* in 2018, over the biweekly measurement periods, water temperature values varied from 7.5 (winter) to 21.7 °C (summer) whereas at station a over our seasonal 24-hour cycles (high-frequency measurements), values varied from 9.1 (winter) to 26.9 °C (summer). Along the studied aquatic continuums, the water temperature varied significantly between stations a and b (Mann-Whitney test, p < 0.05) and between stations a and c (p < 0.05) but no significant variation was recorded between stations b and c (p = 0.23) (Table 2). Salinity ranged from 28.9 (winter) to 35.4 (autumn) at station *Filiere W*, whereas values varied from 31.4 (winter) to 35.7 (autumn) at station a, from 27.5 (winter) to 36.9 (autumn) at station a, from 27.0 (winter) to 42.6 (summer) at station a and a0 (artificial salt marshes) than at stations a1 and a2 (tidal estuary and channel). In average over the year 2018, the tidal estuary (station a3 was slightly oversaturated in oxygen compared to the atmosphere (106 a2 14%) with O2-sat. values ranging between 70% (LT during dawn) and 147% (receving tide the day) during the summer cycle (Fig. 3). Along the continuum, the channel (station a3 was close to the

saturaturion value with the atmosphere over the same measurement periods in 2018 with a lower maximum value (120%) than at station a measured in summer (Fig. 4). Larger amplitudes of oxygen saturation percentages were recorded in the artificial salt marshes with O₂-sat. values ranging from 36 to 176% at station c (over the summer cycle; Fig. 5) and from 49 to 150% at station d (over the summer cycle; Fig. 6). The annual percentage of pCO₂ undersaturation compared to the atmosphere were 47.8%, 15.9%, 64.9% and 85.9% at stations a, b, c and d respectively, with a strong annual CO₂ oversaturation at station b (Table 2). Moreover, along aquatic continuums, the greatest amplitude in water pCO₂ was recorded at station c with values varying between 6 ppmv in spring 2019 (macroalgae bloom and undersaturated in CO₂) and 721 ppmv in autumn 2019 (macroalgae degradation and oversaturated in CO₂).

3.2. Spatial and seasonal dynamics of studied continuum waters

Significant spatial variations in water pCO₂ and associated biogeochemical parameters (salinity, temperature, turbidity and DO) were recorded along shelf – estuary – marsh water continuums (Kruskall-Wallis tests, p < 0.05; Table 2). The PCA reveals that stations are distinguished vertically according to pCO₂, turbidity and oxygen saturation (pCO₂ and O₂-sat. negatively correlated) within PC2 explaining 35.5% of the total variance (Fig. 2). We can cleary observe that the water pCO₂ values measured biweekly at station *Filiere W* were higher than the station *a* values measured over our 24-hour cycles (Fig. 2). On average over the year, the station *b* channel had the highest pCO₂ values (462 \pm 51 ppmv) compared to the three other stations studied (418 \pm 57, 335 \pm 214 and 293 \pm 113 ppmv, respectively at stations *a*, *c* and *d*; Fig. 2). Station *b* was also characterized by the highest water turbidity with values ranging from 1.6 to 59.7 NTU (median of 8.2 NTU) whereas at station *a*, the values only varied from 0.1 to

32.6 NTU (median of 1.9 NTU). The PCA also shows seasonal data are distinguished horizontally according to salinity and temperature within PC1 explaining 42.3% of the total variance (Fig. 2). Generally, for each station, the highest and lowest temperature and salinity values were measured in summer and winter, respectively (Fig. 2) except at station a where the lowest salinity values were recorded in spring (Table 2). Within each station, significant differences in the temperature and salinity values were recorded at the seasonal scale (Kruskall-Wallis tests, p < 0.0001).

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Along the continuums, the PCA recorded contrasted seasonal variations of water pCO₂, particularly between the two artifical salt marshes (Fig. 2). At the station Filiere W continental shelf in 2018, no significant differences in water pCO₂ were recorded at the seasonal scale (Kruskall-Wallis test, p = 0.13), although the highest and lowest seasonal means were recorded in winter and spring, respectively (Table 2). At the station a estuary in 2018, water pCO₂ showed the same seasonal pattern decreasing from winter (441 \pm 21 ppmv) to spring-summer $(390 \pm 40 \text{ and } 385 \pm 60 \text{ ppmv}, \text{ respectively})$ before increasing in autumn $(460 \pm 58 \text{ ppmv})$, whereas the station b channel showed lower seasonal variations over the same measurement periods (Table 2 and Fig. 2). In contrast to the tidal stations (Filiere, a and b), artificial salt marshes (c and d) showed larger seasonal pCO₂ variations (Fig. 2). Station c waters were undersaturated in CO_2 with respect to the atmosphere both in spring 2019 (135 \pm 165 ppmv), summer 2019 (242 \pm 116 ppmv) and winter 2020 (343 \pm 87 ppmv) but oversaturated in autumn 2019 (622 \pm 57 ppmv). At the same time and on average, station d was undersaturated in CO₂ in summer, autumn and winter with the largest water CO2 undersaturation recorded in autumn $(155 \pm 30 \text{ ppmv})$ in contrast to station c. At each studied station (a, b, c and d), water pCO₂ significantly differed at the seasonal scale (Kruskall-Wallis tests, p < 0.001), except for station a between spring and summer (Dunn's post-test, p > 0.999).

At stations *a* and *b* in 2018, the same seasonal NpCO₂ variations were observed, decreasing from winter (595 and 624 ppmv, respectively) to summer (296 and 347 ppmv, respectively) and then increasing towards autumn (420 and 439 ppmv, respectively). At station *c*, the seasonal mean NpCO₂ value increased sharply from summer 2019 (193 ppmv) to autumn 2019 (630 ppmv) and then decreased towards winter 2020 (441 ppmv), contrarily to station *d* where, values decreased from summer (286 ppmv) to autumn 2019 (160 ppmv) before increasing towards winter 2020 (453 ppmv) to reach a similar seasonal NpCO₂ value as station *c*. Regarding temperature effects on water pCO₂, the lowest and highest seasonal TpCO₂ values were measured in winter and summer, respectively, with seasonal TpCO₂ values followed systematically seasonal water temperature variations. At stations *a* and *b*, weak seasonal pCO₂ variations were recorded with the TpCO₂ effects offset by the NpCO₂ effects, particularly in winter and summer. Conversely, in artificial salt marshes, the water NpCO₂ and pCO₂ values followed the same seasonal patterns in spring, summer and autumn at station *c* and only in autumn at station *d* (Fig. A4).

3.3. Biogeochemical variations at diurnal and tidal scales

Over all 24-hour measurement cycles, the measured parameters showed strong variations at the diurnal/tidal scales (Figs. 3-6). At station a in winter, only small tidal salinity variations were measured (31.9 - 32.9; Fig. 3), whereas at station b over the same measurement period, the largest tidal salinity variations occurred with values varying from 27.5 at LT to 32.5 at HT (Fig. 4). On the contrary, at stations a and b during summer and autumn, lower salinity gradients were measured and salinity values were higher at LT than at HT (Figs. 3 and 4). In the two artificial salt marshes, the highest salinity gradients were recorded in summer with

values decreasing through coastal water inflows from 42.6 to 33.5 at station c (Fig. 5) and from 38.1 to 33.8 at station d (Fig. 6). During the other seasons at the diurnal/tidal scales, salinity values varied more slightly with, for instance, a difference in salinity units of 2.7, 0.7 and 1.1 at station c in winter, spring and autumn, respectively (Fig. 5). It should be noted that, at station c, coastal water inflows from station d led to an increase in salinity only in winter (Fig. 5); in autumn, a rainfall event occurred (5 mm) between 14:00 and 15:00 leading to a decrease in salinity of 1 unit, an increase in turbidity from 2.7 to 7.3 NTU (peak at 15.7 NTU) but without any water pCO₂ variation (Fig. 5). At station d, turbidity and salinity did not vary both in autumn and winter cycles (Fig. 6).

The largest diurnal/tidal variations in the water pCO₂ and DO concentrations occurred during summer with pCO₂ gradients of 255, 216, 405 and 258 ppmv at stations a, b, c and d respectively, and DO gradients of 153.2, 152.2, 262.8 and 205.6 µmol Γ^1 at stations a, b, c and d, respectively (Figs. 3-6). At stations a and b, the low tide periods during the day (LT/D) occurring at dawn showed higher water pCO₂ and lower O₂-sat. values than the low tide periods during the night (LT/N) occurring at dusk, particularly in summer (Figs. 3 and 4). In general, our diurnal cycles showed a decrease in pCO₂ that was negatively correlated to an increase in DO during the daytime (except at station c in spring; Fig. 5) and an opposite pattern during the night-time (except at station c in summer; Fig. 5). For instance, at station c in winter during the marsh confinement, water pCO₂ decreased from 519 to 129 ppmv during the day (from 09:00 to 17:00) and increased from 212 to 442 ppmv during the night (from 20:00 to 05:00), while DO increased from 273.4 to 350.0 µmol Γ^1 and decreased from 286.5 to 257.5 µmol Γ^1 , respectively (Fig. 5). It should be noted that, at station c in summer, the lowest DO concentration and O₂-sat. percentage were reached during the night (76.9 µmol Γ^1 and 35.6%, respectively; Fig. 5). At station d, the same diurnal water pCO₂ and DO patterns were also

observed (Fig. 6). However, these measured diurnal pCO₂ and DO variations were significantly disrupted once coastal water horizontal advection and artificial salt marsh management practices occurred (Figs. 3-6).

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Strong tidal variations in water pCO₂ were generally recorded during all seasons except at station d both in autumn and winter (Fig. 6). At stations a and b, incoming tides during the day produced rapid decreases in water pCO₂ from an oversaturation to a slight undersaturation of subsurface waters compared to the atmosphere, particularly in spring (-121 and -167 ppmv, respectively) and summer (-139 and -115 ppmv, respectively; Figs. 3 and 4). Only at station a, receding tides during the day generated an additional decrease in pCO₂ to reach the lowest values measured within each season (270 and 335 ppmv in summer and autumn, respectively; Fig. 3). Then, at station a in winter and spring, incoming tides during the night produced slight increases in water p CO_2 whereas in summer and autumn at the same station and at station b over the four seasons, incoming tides during the night-time produced strong increases in pCO₂ leading to oversaturation periods (Figs. 3 and 4). Through simultaneous salinity measurements at stations b and c and a cross-correlation function analysis, we estimated the horizontal advection of water masses from station b to station c during a spring tide period at 90 minutes (periods from 29/06 to 08/07/2019). At station c during the night, higher pCO₂ were measured at HT than at LT (i) in spring (363 \pm 85 and 16 \pm 5 ppmv at HT/N and LT/N, respectively), (ii) in summer (295 \pm 46 and 258 \pm 37 ppmv at HT/N and LT/N, respectively) and (iii) in winter $(431 \pm 6 \text{ and } 323 \pm 53 \text{ ppmv at HT/N} \text{ and LT/N}, \text{ respectively; Fig. 7})$. The same tidal pCO₂ pattern was also recorded at this station in summer 2019 during the day (323 \pm 88 and 197 \pm 141 ppmv at HT/D and LT/D, respectively; Fig. 7). In spring, the rewilded artificial marsh (station c) recorded very low water pCO₂ values both the day and the night during the marsh confinement but coastal water inflows from the station b channel instantly produced a large and rapid increase in water pCO₂ (+390 ppmv) mostly within a two-hour period (Fig. 5).

For all 24-hour cycles, there were strong positive correlations between pCO₂ and NpCO₂ showing that non-temperature effects have a strong control over water pCO₂ at the diurnal/tidal scale along the two studied continuums (Figs. 3-6). Similarly, water pCO₂ values were strongly negatively correlated with oxygen saturation, especially in autumn with diurnal correlations between pCO₂ and O₂-sat. ranging from -0.67 (winter) to -0.97 (autumn) at station a, from -0.63 (summer) to -0.87 (autumn) at station b, from -0.54 (winter) to -0.86 (autumn) at station c and from -0.59 (winter) to -0.84 (summer) at station d (Figs. 3-6). At the artificial salt marsh stations, correlations between pCO₂ and in situ Chl a were -0.62 and -0.47, respectively in winter and summer at station d (Fig. 6) and -0.41 in winter at station c (Fig. 5). At station c in summer, correlations between pCO₂ vs. DO and pCO₂ vs. Chl a were not significant but the pCO₂ vs. salinity coefficient was -0.95 (Fig. 5). At station b, significant linear regressions were calculated between pCO₂ and salinity, i.e. negative in spring (R² = 0.49) and positive in summer (R² = 0.53; Fig. 7). Lastly, at station c, a negative pCO₂ vs. salinity regression was also calculated both in spring and summer (R² = 0.83 and R² = 0.87, respectively) with a positive water pCO₂ vs. water height correlation (R² = 0.72) observed at station c in summer (Fig. 7).

3.4. Air-water CO₂ flux variations and associated salt marsh metabolism

Mean air-water CO₂ fluxes (FCO₂) according to the W92 parametrization were estimated to be -0.01 \pm 0.22, 0.22 \pm 0.40, 0.18 \pm 1.37 and -1.22 \pm 1.71 mmol m⁻² h⁻¹ at stations a (sink), b (source), c (source) and d (sink), respectively, whereas the downstream point of each studied continuum (station *Filiere W*) behaved as a net CO₂ source (0.30 \pm 1.04 mmol m⁻² h⁻¹).

Large seasonal variations were observed at the four studied stations (Fig. 8). On average, station a showed positive FCO₂ values in both winter and autumn (slight CO₂ source) but negative means in spring and summer (slight CO₂ sink; Table 3 and Fig. 8). At the station b channel, positive FCO₂ values were estimated, with maximun and minium FCO₂ mean values occurring in winter $(0.52 \pm 0.65 \text{ mmol m}^{-2} \text{ h}^{-1})$ and summer $(0.06 \pm 0.08 \text{ mmol m}^{-2} \text{ h}^{-1})$, respectively (Fig. 8). Station c behaved as a CO₂ sink in spring, summer and winter whereas it emitted large quantities of CO₂ to the atmosphere in autumn $(2.03 \pm 1.17 \text{ mmol m}^{-2} \text{ h}^{-1}; \text{ Fig. 8})$. FCO₂ values at station c varied between -3.00 and 0.03 mmol m⁻² h⁻¹ in spring and between 0.61 and 4.61 mmol m⁻² h⁻¹ in autumn. Lastly, station d behaved as a CO₂ sink in summer, autumn and winter with the largest atmospheric CO₂ uptake in autumn (-3.43 \pm 1.09 mmol m⁻² h⁻¹) where fluxes varied between -6.03 and -1.79 mmol m⁻² h⁻¹ (Fig. 8). At the global scale (stations a, b, c and d), the Fier d'Ars system behaved as a CO₂ sink over the year (-0.20 \pm 1.32 mmol m⁻² h⁻¹) with (i) the lowest FCO₂ values recorded during the growing season in spring and (ii) higher FCO₂ values during the night than during the day.

4. Discussion and conclusions

4.1. Water pCO₂ dynamics according to the typologies of the shelf – estuary – marsh

continuums

Along the studied continuum stations from the coastal ocean to artificial salt marshes, significant water pCO₂ variations were recorded at the spatial scale (Fig. 9). A strong influence of ecosystem typology (continental shelf, estuary, marsh) on inorganic C dynamics was observed, as described more generally by Bauer et al. (2013) for the coastal ocean. In 2018 and 2019, the continental shelf (station *Filiere W*), influenced by the Aiguillon Bay and its

associated watershed, was characterized by salinity ranges below those from the Atlantic Ocean (35.6; Vandermeirsch 2012). This was confirmed by the French Phytoplankton Monitoring Network REPHY for the 2000-2017 period with values between 26.2 and 35.8 (Fig. A3; Belin et al. 2021). At station Filiere W, the lowest salinity values were measured in winter when terrestrial river inputs from the Aiguillon Bay watershed were the highest leading to strong nutrient inputs into coastal waters during this time period (Belin et al. 2021). However, in 2018, the non-significant relationship found between salinity and pCO₂ (p = 0.88) measured biweekly at station Filiere W showed a rather weak influence of terrestrial inputs on pCO₂ dynamics. At this shelf station, phytoplankton blooms generally occur in spring and summer (Coignot et al. 2020), inducing an undersaturation of water CO₂, followed by pCO₂ decreases observed at station a along the continuum. Station a, found at the entry of the Fier d'Ars estuary, is influenced by buffered marine waters from the shelf with the percentage of CO₂ oversaturation over a 24-hour cycle varying from 96% (winter) to 18% (spring/summer). This decrease in water pCO₂ could be attributed to phytoplankton development in the estuarine waters during this period. The fact that the seasonal NpCO₂ (non-temperature effects on pCO₂) varied from heterotrophy in winter to autotrophy in summer confirmed the phytoplankton activity on the pCO₂ dynamics. Another study carried out close to station a measured Chl a concentrations between 0.2 (winter) and 3.5 µg l⁻¹ (spring/summer) and a net Chl a export suggestive of a net primary production within this tidal estuary (Bel Hassen 2001). Moreover, microphytobenthos (MBP) may also contribute to water CO_2 undersaturation at station a and to the overall water column Chl a concentration through tidal resuspensions as shown by Savelli et al. (2019) in a nearby intertidal zone. Due to the small insulary catchement area (1200 ha) consisting only of salt marshes (no terrestrial water input), the CO₂ dynamics in the Fier d'Ars estuary (station a) is different from other estuaries worldwide (Borges & Abril 2011). Low water pCO₂ values

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(270 - 569 ppmv) with no significant relationship between pCO₂ and salinity were found, like in the present study and elsewhere on the Atlantic coast of the United States (Jiang et al. 2008), thereby enabling potential CO₂ sink behaviours (Maher & Eyre 2012). Similarly, the marine-dominated estuary of Sapelo Sound (USA) was also characterized by lower water pCO₂ values than river-dominated estuaries (Borges & Abril 2011). However, the water pCO₂ values measured at this estuary (390 - 2400 ppmv) were higher than those measured at station *a* in our study due to high bacterial mineralization rates of organic carbon (OC) produced by *Spartina* in nearby tidal salt marshes (Jiang et al. 2008). On the contrary, river-dominated estuaries characterized by strong freshwater inputs generally show much higher pCO₂ values. The heterotrophic status due to the microbial degradation of OC from rivers produces large CO₂ degassing into the atmosphere (Borges & Abril 2011, Bauer et al. 2013, Najjar et al. 2018) as observed nearby in the Aiguillon Bay (Coignot et al. 2020), the Gironde estuary (Frankignoulle et al. 1998) and the Loire estuary (Abril et al. 2009).

Overall, the studied channel (station b) between the tidal estuary and artificial salt marshes (Fig. 9) showed longer periods of CO₂ oversaturation over the year 2018 (90, 73 and 100% in winter, spring/summer and autumn, respectively) with significantly higher water pCO₂ values than those at the station a estuary. At station b, strong hydrodynamic forcings during incoming and receding tides produced more turbid waters due to organic matter (OM) resuspension from channel muds (Fig. 4). Then, it probably limited the activity of primary producers (phytoplankton, algae, plants) and, on the contrary, favoured heterotrophic processes (Polsenaere et al. 2022). In the Fier d'Ars waters close to station b, Tortajada (2011) measured POC/Chl a > 200 on average over the year and POC/Chl a > 600 in autumn. This may confirm a microbial loop-type trophic network in channel waters (Tortajada 2011) and a potential influence on the C dynamics leading to OM mineralization processes from MPB and, in turn,

water CO_2 oversaturation periods measured at station b. However, channel waters showed lower pCO₂ values compared with other coastal channel systems probably due to very low terrestrial water inputs upstream/downstream over the Fier d'Ars estuary. In the Arcachon lagoon, Polsenaere et al. (2022) recorded at a tidal channel similar to stations a and b according to typology, long periods of CO_2 oversaturation with seasonal means ranging from 461 ± 14 in July 2008 to 530 ± 39 ppmv in September 2009. Another study showed that the Sancti Petri Channel waters and the adjacent salt marsh system linking the Atlantic Ocean to the Cadiz Bay were mainly CO_2 oversaturated similarly to our station b, with an annual p CO_2 mean of 564 ± 134 ppmv (281 - 862 ppmv), due to OM diagenetic processes in mudflats that could constitute a DIC source in the water column and thus increased water p CO_2 (Burgos et al. 2018). Within the Duplin River salt marsh-estuary coastal system, higher summer p CO_2 values and DIC concentrations were recorded at low tide in channel waters (12000 ppmv and 4300 μ mol 1⁻¹, respectively) than at high tide in marsh waters (1600 ppmv and 2200 μ mol 1⁻¹, respectively; Wang et al. 2018).

Contrary to tidal stations (a and b), significant and more pronounced periods of CO_2 undersaturation were recorded in artificial salt marsh waters (76 and 87% at stations c and d, respectively), once again indicating that marsh typology has a strong influence on CO_2 dynamics. In shallow coastal wetlands such as artificial salt marshes, lower hydrodynamic conditions and longer water residence times promote the development of primary producers such as macrophytes and phytoplankton and as a result, biological CO_2 uptake (Wang & Cai 2004, Bauer et al. 2013, Tobias & Neubauer 2019). The artificial salt marsh waters studied here showed lower pCO_2 values and longer CO_2 undersaturation periods, particularly in spring and summer due to strong macrophyte activity (i.e. macroalgae at station c and seagrasses at station d) compared with other marshes and wetland types (Cotovicz Jr. et al. 2015, Wang et al. 2018,

Berg et al. 2019). By comparison, Ternon et al. (2018) measured higher water pCO₂ in nearby fresh- and brackish-water artificial marshes on the French Atlantic coast in summer 2018 with values varying between 227 and 1925 ppmv and between 349 and 2000 ppmv, respectively. Unlike seagrass beds (Mcleod et al. 2011), fast-growing macroalgae developing in coastal wetlands such as at station c have a limited capacity to store C over the long-term. However, other studies have shown their potential contribution to coastal blue C by (i) storing large OM quantities in their living biomass through their high primary production (Raven 2018) and (ii) transferring it to adjacent systems through tides and storage in coastal sediments (Duarte & Cebrián 1996, Hill et al. 2015, Krause-Jensen & Duarte 2016).

4.2. Marsh management practices module temporal carbon dynamics

Management practices at the studied artificial salt marshes (stations c and d) correspond to specific water lock management approaches linked to economic activities. They can strongly modulate coastal water fluxes from the Fier d'Ars estuary (station a) and thereby influence salinity and marsh pCO₂ dynamics. At station c in 2019, the specific management practice undertaken by the NNR (see M&M section), along with higher air temperatures and lower precipitations in summer, produced favourable conditions for macroalgae development from early spring to late summer favoured by low marsh hydrodynamics and large temperature fluctuations (Newton & Thornber 2013). To explain the observed macroalgae bloom, we also assumed there were excess nutrient inputs, as described for other coastal ecosystems (Teichberg et al. 2010). Nearby marsh aquafarming activities occurring upstream from the Ré Island watershed that can communicate through the station b channel to station c may result in high nutrient intake and could explain the macroalgae development (Paticat 2007, Tortajada 2011).

At station b in September 2018, DIN > 55 μ mol l⁻¹ and DIP > 5 μ mol l⁻¹ were measured at ebbing tide (unpublished results). Moreover, shelf waters influenced by terrestrial inputs (see the above paragraphs in the Discussion) could also lead to high nutrient inputs at marsh station c through incoming tides. At station Filiere W, NO₃- values ranged between 29 and 107 µmol 1⁻¹ in winter 2019 and between 0.9 and 17.4 μmol 1⁻¹ in spring 2019 (Belin et al. 2021). Consequently, at station c from spring, macroalgae communities probably prevented the development of phytoplankton and seagrasses by marsh nutrient depletion and by light limitation in the water column (Gouazé 2019). However, these fast-growing macrophytes induce intense CO₂ uptake as observed in spring and summer 2019 (Fig. 5). During this period, strong NpCO₂ effects confirmed the major influence of macroalgae biological autotrophy on inorganic C dynamics (Fig. A4); large CO₂ undersaturation periods over tidal/diurnal cycles were maintained although modulated by occasional coastal water inflows under weak tidal amplitudes from station b into the marsh (Fig. 5). This result is confirmed by higher salinity values at station c than at station b measured in spring and summer during our sampling periods (Table 2). Other strong CO_2 undersaturation periods were recorded at station c in summer 2019 with a water pCO₂ mean of 297 \pm 150 ppmv over a 96-hour cycle in early August (2 < pCO₂ < 723 ppmy; unpublished results). Therefore, these marsh waters were characterized by very low pCO₂ values rarely observed over other similar wetland typologies (Borges 2003, Wang et al. 2018, Burgos et al. 2018, Berg et al. 2019). On the contrary, in autumn 2019, macroalgae degradation probably by microbial mineralization processes produced the highest pCO₂ values and largest water oversaturation periods recorded in the marsh waters. This result can be explained by high NH₄⁺ concentrations recorded in November 2019 (62 µmol l⁻¹; unpublished results) and the lowest oxygen saturation values (Table 2).

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Contrarily to station c, the working artificial marsh (station d) is managed for salt production in the upstream ponds along the continuum (Fig. A2) and is directly connected to the Fier d'Ars estuary (station a) with no channel connecting the two (Fig. 9). Salt production requires a subtle lock hydraulic management of the salt marsh depending on the frequency of the coastal water supplies that are mainly controlled by the salt manufacturer and meteorological conditions (rainfall, sunshine and wind) for evaporation (Paticat 2007). Therefore, contrary to station c, coastal water inflows to station d were generally performed sparingly with relatively small daily volumes to limit these water mixing effects (i.e. rapid accumulation of large water volumes through rainfall events or spring tides that stop the increase in temperature and salinity of the water already present in the marsh; Paticat 2007). At station d in summer 2019, the measured water pCO₂ values were significantly higher than those at the same period at station c but were very similar to those from station a in summer 2018 (Table 2). This result could be attributed to low activity of the primary producers during this period dedicated to salt production and confirmed by the significant thermal effects on water pCO₂ observed at this station unlike at station c (Fig. A4). A higher frequency of repetitive and direct water inflows through a specific lock management approach from station a to station d (similar mean summer salinity values at stations a and d; Table 2) could also explain these higher pCO₂ values. In autumn 2019 and winter 2020, lower hydrodynamic conditions due to lock closure (the salt farming activity was at a standstill; see M&M section) may have led to lower water turbidity values, lower nutrient inputs from shelf waters and then growth of Ruppia sp. seagrasses and phytoplankton in the marsh instead of macroalgae, thereby producing very low pCO₂ values.

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Anthropogenic management practices in artificial salt marshes can therefore strongly influence the contribution and turnover of macrophytes and, consequently, the marsh CO_2 behaviour (sink/source). Due to eutrophication in the rewilded marsh (station c), development

of the macroalgae communities favoured an atmospheric CO₂ sink during the growing season but an overall annual net atmospheric CO₂ source through their degradation. Their presence prevents phytoplankton blooms and the establishment of seagrass beds as well as economic activities requiring specific management practices. Our work suggests a confinement of artificial marshes in winter and a drying up like at station *d* to avoid the nutrient inputs and the macroalgae development favouring rather slow-growing macrophytes such as seagrasses which could ultimately contribute to blue C sequestration as more generally described by Mcleod et al. (2011). Similarly, other studies have suggested that the coastal ecosystem management by reducing anthropogenic nutrients could favour blue C ecosystems such as seagrasses, salt marshes and C sequestration (Macreadie et al. 2017, Palacios et al. 2021). Few studies on the links existing between the functioning and C biogeochemical processes of artificial salt marshes have been carried out as in our study. For instance, in Mediterranean poly-euhaline lagoon waters, Le Fur et al. (2018) confirmed that eutrophication levels can strongly favour perennial seagrass species and, to the contrary, fast-growing macroalgae in oligotrophic and eutrophic waters, respectively.

4.3. Influence of biological activity on diurnal water pCO₂ dynamics

At our studied continuum stations (*a*, *b*, *c* and *d*), negative correlations between water pCO₂ and DO, associated with large oxygen saturation ranges and strong NpCO₂ effects on the measured pCO₂ values, were calculated at the diurnal scale, particularly at the artificial marsh stations (Figs. 5 and 6). It partly demonstrates the strong biological influence of both autotrophic (macrophytes and phytoplankton) and heterotrophic (bacteria) organisms on the diurnal dynamics of inorganic C. Dai et al. (2009) confirmed that C biogeochemical processes

in coastal environments such as the Fier d'Ars system are generally controlled by nontemperature effects (biological and tidal effects) compared to more open systems like the Atlantic Ocean. Several studies have shown a major biological control on diurnal pCO₂ variations in coastal systems such as the temperate Bay of Brest (France; Bozec et al. 2011), the Arcachon tidal flat (France; Polsenaere et al. 2022), a shallow subtropical estuary in Tampa Bay (USA; Yates et al. 2007) and the tropical coastal embayment at Guanabara Bay (Brazil; Cotovicz Jr. et al. 2015). At the Fier d'Ars estuary (station a) in spring and summer 2018, we calculated non-significant correlations between the measured pCO₂ and estimated TA values and between the measured pH and estimated TA values. We also calculated significant positive correlations between the measured DO and pH values ($R^2 = 0.75$ and $R^2 = 0.50$, respectively; p < 0.0001; regressions not shown) and between the measured DO and pCO₂ values ($R^2 = 0.86$ and $R^2 = 0.91$, respectively; p < 0.0001; regressions not shown). These results confirmed the significant control of the photosynthesis vs. respiration balance and the weaker influence of the carbonate system on diurnal pCO2 variations as shown by Yates et al. (2007) in Tampa Bay. At station c, in spring and summer 2019 during the day, macroalgae CO₂ uptake induced large CO₂ undersaturation periods associated with high oxygen saturations (> 170%) in the water column; during the night, macroalgae respiration consumed large oxygen quantities and therefore very low oxygen saturations were observed (< 60%; Fig. 5). In winter 2020, during the marsh confinement, the highest diurnal variations in water pCO₂ were recorded with a pCO₂ decrease of 386 ppmv during the day and an increase of 119 ppmv during the night; again, this may confirm significant photosynthesis and respiration processes in these shallow marsh waters (Fig. 5). By comparison, in a Zostera marina meadow (South Bay, USA) in spring and summer 2015, Berg et al. (2019) measured similar diurnal fluctuations of water pCO₂ that were directly controlled by seagrass metabolism with diurnal ranges of 528 and 603 ppmv in late April and

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late June, respectively. At station *d*, during periods when the lock was closed, the temperature and non-temperature effects on water pCO₂ were 9 and 115 ppmv, respectively in autumn 2019 and 34 and 134 ppmv, respectively in winter 2020 showing also an influence of biological processes on the diurnal pCO₂ dynamics (Fig. 6). In tidal estuaries and salt marshes, diurnal variations in water pCO₂ due to biological influence must therefore be measured and better accounted for CO₂ budget studies to avoid overestimating their CO₂ sink potential through daytime pCO₂ measurements only (Cotovicz Jr. et al. 2015).

4.4. Biogeochemical station interconnections through tidal advection

Along the studied aquatic continuums, horizontal advection can significantly control water pCO₂ variations at the tidal scale. As seen during the present study, flooding of the *Filiere W* shelf waters induced a salinity increase and a decrease in winter and from spring to autumn, respectively, at stations *a*, *b*, *c* and *d* along with significant water pCO₂ variations. For instance, at tidal stations (*a* and *b*) in spring and summer during the day, flooding of the shelf waters created a significant decrease in water pCO₂ since the advected waters were CO₂ undersaturated compared with estuarine waters (Figs. 3 and 4). This observation was confirmed by significant negative correlations calculated between pCO₂ and water heights during this period, highlighting the strong control of tidal rhythm on the coastal waters pCO₂. Along the French Atlantic coast, Polsenaere et al. (2022) also showed a strong control of the tidal rhythm in the Arcachon lagoon with higher pCO₂ values measured at LT than at HT during each season in 2008 and 2009. Similarly, in the Sancti Petri Channel and its adjacent salt marshes, Burgos et al. (2018) recorded the highest and lowest pCO₂ values (1059 and 754 ppmv) at LT and HT, respectively. Even stronger tidal influences on in situ water pCO₂ (from 1380 to 4770 ppmv

between HT and LT) were observed by Borges (2003) in summer 2001 in a mangrove system (Gaderu Creek). Interestingly, at station a in summer 2018, a significant water pCO₂ decrease from 370 ppmv (15:00) to 267 ppmv (19:30) associated with an oxygen saturation increase from 120 to 140% (characteristic of macrophyte biological activity) was observed in estuary waters during a receding tide, certainly due to CO₂ undersaturated upstream waters from the productive artificial salt marsh waters (Fig. 3). In spring and summer, marsh waters at station c were largely CO₂ undersaturated due to the high primary production of macroalgae and in particular, the long water residence times; however, more CO₂-enriched water inflows from stations *Filiere W*, a and b instantly produced significant water pCO₂ increases at this marsh station (Fig. 5). Therefore, in each season (except at station d in autumn and winter), variations in marsh water pCO₂ at the tidal scale were dependent on the biogeochemical state of the advected waters downstream from the shelf, estuary and channel with respect to CO₂ saturation.

4.5. Metabolism assessment of the Fier d'Ars continuums

Over the year 2018, the Fier d'Ars estuary (station *a*) behaved on average as a yearly CO₂ sink close to the atmospheric equilibrium, although a significant sink was measured in the spring (Table 3) due to potential phytoplankton blooms occurring in coastal waters (see previous sections in the Discussion). Conversely, over the same meteorological periods, the channel (station *b*) was a net annual source from its turbid waters to the atmosphere due to several oversaturation periods, particularly in winter, characterized by the highest wind speed and CO₂ exchange coefficient values (Table 3 and Fig. 8). Over the subtidal Bay of Brest, Bozec et al. (2011) estimated slightly higher air-water FCO₂ values ranging from -0.38 to 0.01 mmol m⁻² h⁻¹ during spring/summer and from 0.04 to 0.91 mmol m⁻² h⁻¹ during autumn/winter (Table

3). In the Arcachon tidal lagoon (similar to our stations a and b) using the same 24-hour cycle approach, an annual air-water FCO₂ was estimated to be 0.27 ± 0.22 mmol m⁻² h⁻¹, the highest and the lowest CO₂ degassing values recorded in September and January 2009, respectively (Polsenaere et al. 2022). In the present study, rewilded artificial salt marsh waters (station c) behaved as a yearly source of atmospheric CO2 when macroalgae degradation and mineralization processes produced strong CO₂ effluxes to the atmosphere, as in autumn 2019 (Fig. 8; see previous sections in the Discussion). On the contrary, the working artificial salt marsh (station d) behaved as the largest yearly CO₂ sink particularly favoured by inexistent tidal variations (lock closure) in the absence of salt-farming activities (from fall to winter; Fig. 8). Within the Duplin River salt marsh-estuary coastal ecosystem, both channel and marsh waters degassed CO₂ to the atmosphere and, unlike our studied stations, the highest and lowest sources were recorded in summer (5.50 and 3.90 mmol m⁻² h⁻¹ from channel and marsh waters, respectively) and in winter (0.70 and 0.60 mmol m⁻² h⁻¹ from channel and marsh waters, respectively), respectively (Wang et al. 2018). Overall, the whole Duplin system emits more CO₂ into atmosphere than the Fier d'Ars system, probably due to its more intense estuarine heterotrophic metabolism.

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In autumn, the lack of significant variations in wind speeds between stations a and b in 2018 and between stations c and d in 2019, whereas atmospheric CO_2 exchanges significantly changed, highlighting the predominance of air-water CO_2 gradients in the control of CO_2 flux directions either as a sink or a source (Table 3). However, at the seasonal scale, the turbulence processes measured at the air-water interface also played an important role in flux variability and magnitude; for instance, at station a between spring and summer and at station b between winter and summer, wind speed variability produced significant FCO_2 variations although no significant air-water CO_2 gradients were measured (Table 3). Atmospheric exchanges in

marshes are therefore dependent on the CO₂ saturation state of the water column (air-water gradient) considering that the wind only acts as a driver of the flux (Polsenaere et al. 2022). Moreover, the (site specific) methodological calculations and associated differences chosen for the exchange coefficient parameterizations (higher fluxes with RC01 than with W92 methods; Table 3) may produce even more contrasts in the estimated air-water FCO₂ values (Cotovicz Jr. et al. 2015, Polsenaere et al. 2022).

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By scaling-up and considering stations a and b together along the continuum (estuary and channel), an annual source of atmospheric CO₂ of 7.3 g C m⁻² yr⁻¹ was calculated in 2018. The rewilded artificial salt marsh (station c) emitted 17.5 g C m⁻² yr⁻¹ (702.0 kg C yr⁻¹) to the atmosphere, when the working artificial marsh (station d) absorbed 97.7 g C m⁻² yr⁻¹ (828.6 kg C yr⁻¹) from the atmosphere. A larger scale study of C along three shelf – estuary – tidal wetland continuums on the Atlantic coast of the United States also showed strong spatial variations in atmospheric CO₂ exchanges with CO₂ uptake for wetland and shelf waters of 5.3 ± 1.5 and 4.0 \pm 0.7 Tg C yr⁻¹, respectively and a CO₂ source from estuarine waters of 4.2 \pm 1.7 Tg C yr⁻¹ (Najjar et al. 2018). During our study, contrasting stations along the continuums were sampled via seasonal 24-hour cycles to estimate the air-water CO₂ exchanges. However, longer seasonal measurement periods up to several days would be more representative of the strong temporal variability in k₆₆₀, water pCO₂ and other biogeochemical parameters. At the Bossys perdus salt marsh (station e; Fig. 1), another flux methodology using the atmospheric EC technique was deployed to continuously measure year-round in situ CO₂ fluxes at the ecosystem scale under real field conditions. Over the year 2020, this preserved salty meadow fixed 483 g C m⁻² yr⁻¹ from the atmosphere, which is much higher than the studied artificial salt marshes indicating a potential stronger atmospheric CO₂ sink in tidal salt marshes (Mayen et al. in prep.). However, it is also important to study the whole marsh metabolism taking terrestrial and aquatic compartments into account and distinguishing their respective contributions to atmospheric fluxes and the regional C budgets of the associated marshes. To do this, we measured the atmospheric CO₂ exchanges at the ecosystem scale at both the water-air and soil-air interfaces, biogeochemical parameters of the channel column water, metabolic balance of planktonic communities and horizontal C export during seasonal 24-hour cycles to be more integrative from a C budget point of view over the salt marshes (Mayen et al. in prep.).

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380	Abril G, Commarieu M-V, Solionemo A, Breier P, Guerin F (2009) Turbidity limits gas
887	exchange in a large macrotidal estuary. Estuarine, Coastal and Shelf Science 83:342-
888	348.
889	Adam P (2019) Salt Marsh Restoration. In: Coastal Wetlands. Elsevier, p 817–861
890	Aminot A, Kerouel R (2004). Hydrologie des écosystèmes marins. Paramètres et analyses. Ed.
891	Ifremer
892	Amorocho J, DeVries JJ (1980) A new evaluation of the wind stress coefficient over water
893	surfaces. J Geophys Res 85:433.
894	Artigas F, Shin JY, Hobble C, Marti-Donati A, Schäfer KVR, Pechmann I (2015) Long term
395	carbon storage potential and CO ₂ sink strength of a restored salt marsh in New Jersey.
396	Agricultural and Forest Meteorology 200:313–321.
397	Aufdenkampe AK, Mayorga E, Raymond PA, Melack JM, Doney SC, Alin SR, Aalto RE, Yoo
898	K (2011) Riverine coupling of biogeochemical cycles between land, oceans, and
399	atmosphere. Frontiers in Ecology and the Environment 9:53-60.
900	Bade DL (2009) Gas Exchange at the Air–Water Interface. In: <i>Encyclopedia of Inland Waters</i> .
901	Elsevier, p 70–78
902	Baldocchi DD (2003) Assessing the eddy covariance technique for evaluating carbon dioxide
903	exchange rates of ecosystems: past, present and future: CARBON BALANCE and
904	EDDY COVARIANCE. Global Change Biology 9:479–492.
905	Baldocchi DD, Hincks BB, Meyers TP (1988) Measuring Biosphere-Atmosphere Exchanges
906	of Biologically Related Gases with Micrometeorological Methods. Ecology 69:1331-
907	1340.

- 908 Bauer JE, Cai W-J, Raymond PA, Bianchi TS, Hopkinson CS, Regnier PAG (2013) The 909 changing carbon cycle of the coastal ocean. Nature 504:61–70.
- 910 Bel Hassen M (2001) Spatial and Temporal Variability in Nutrients and Suspended Material
- Processing in the Fier d'Ars Bay (France). Estuarine, Coastal and Shelf Science 52:457–
- 912 469.
- 913 Belin C, Soudant D, Amzil Z (2021) Three decades of data on phytoplankton and phycotoxins
- on the French coast: Lessons from REPHY and REPHYTOX. Harmful Algae
- 915 102:101733.
- 916 Berg P, Delgard ML, Polsenaere P, McGlathery KJ, Doney SC, Berger AC (2019) Dynamics
- of benthic metabolism, O₂, and pCO₂ in a temperate seagrass meadow. Limnol
- 918 Oceanogr 64:2586–2604.
- 919 Borges AV (2003) Atmospheric CO₂ flux from mangrove surrounding waters. Geophys Res
- 920 Lett 30:1558.
- 921 Borges AV, Abril G (2011) Carbon Dioxide and Methane Dynamics in Estuaries. In: *Treatise*
- 922 on Estuarine and Coastal Science. Elsevier, p 119–161
- 923 Borges AV, Delille B, Frankignoulle M (2005) Budgeting sinks and sources of CO₂ in the
- ocastal ocean: Diversity of ecosystems counts: COASTAL CO₂ SINKS AND
- 925 SOURCES. Geophys Res Lett 32.
- 926 Bozec Y, Merlivat L, Baudoux A-C, Beaumont L, Blain S, Bucciarelli E, Danguy T,
- Grossteffan E, Guillot A, Guillou J, Répécaud M, Tréguer P (2011) Diurnal to inter-
- annual dynamics of pCO₂ recorded by a CARIOCA sensor in a temperate coastal
- 929 ecosystem (2003–2009). Marine Chemistry 126:13–26.

930	Burgos M, Orlega 1, Forja J (2018) Carbon Dioxide and Methane Dynamics in Three Coastai
931	Systems of Cadiz Bay (SW Spain). Estuaries and Coasts 41:1069–1088.
932	Cai W-J (2011) Estuarine and Coastal Ocean Carbon Paradox: CO ₂ Sinks or Sites of Terrestrial
933	Carbon Incineration? Annu Rev Mar Sci 3:123–145.
934	Champion E, Gernigon J, Lemesle J-C, Terrisse J, Maisonhaute S (2012) 3ème Plan de gestion
935	2013-2017 de la réserve naturelle nationale de Lilleau des Niges.
936	Chmura GL, Anisfeld SC, Cahoon DR, Lynch JC (2003) Global carbon sequestration in tidal,
937	saline wetland soils. Global Biogeochem Cycles 17(4), 1111,
938	doi:10.1029/2002GB001917, 2003.
939	Cloern JE, Foster SQ, Kleckner AE (2014) Phytoplankton primary production in the world's
940	estuarine-coastal ecosystems. Biogeosciences 11:2477–2501.
941	Coignot E, Polsenaere P, Soletchnik P, Le Moine O, Souchu P, Joyeux E, Le Roy Y, Guéret J-
942	P, Froud L, Gallais R, Chourré E, Chaigneau L (2020) Variabilité spatio-temporelle des
943	nutriments et du carbone et flux associés le long d'un continuum terrestre-aquatique
944	tempéré (Marais poitevin – Baie de l'Aiguillon – Pertuis Breton). Ifremer.
945	Cole JJ, Prairie YT, Caraco NF, McDowell WH, Tranvik LJ, Striegl RG, Duarte CM,
946	Kortelainen P, Downing JA, Middelburg JJ, Melack J (2007) Plumbing the Global
947	Carbon Cycle: Integrating Inland Waters into the Terrestrial Carbon Budget.
948	Ecosystems 10:172–185.
949	Cotovicz Jr. LC, Knoppers BA, Brandini N, Costa Santos SJ, Abril G (2015) A strong CO ₂ sink
950	enhanced by eutrophication in a tropical coastal embayment (Guanabara Bay, Rio de
951	Janeiro, Brazil). Biogeosciences 12:6125–6146.

952	Crosswell JR,	, Anderson IC,	Stanhope JW	, Dam BV,	, Brush MJ,	, Ensign S,	Piehler MF,	McKee
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- B, Bost M, Paerl HW (2017) Carbon budget of a shallow, lagoonal estuary:
- Transformations and source-sink dynamics along the river-estuary-ocean continuum.
- Limnology and Oceanography 62:S29–S45.
- Dai M, Lu Z, Zhai W, Chen B, Cao Z, Zhou K, Cai W-J, Chenc C-TA (2009) Diurnal variations
- of surface seawater pCO₂ in contrasting coastal environments. Limnol Oceanogr
- 958 54:735–745.
- 959 Dai M, Su J, Zhao Y, Hofmann EE, Cao Z, Cai W-J, Gan J, Lacroix F, Laruelle GG, Meng F,
- 960 Müller JD, Regnier PAG, Wang G, Wang Z (2022) Carbon Fluxes in the Coastal Ocean:
- 961 Synthesis, Boundary Processes, and Future Trends. Annu Rev Earth Planet Sci 50:593–
- 962 626.
- Deegan LA, Johnson DS, Warren RS, Peterson BJ, Fleeger JW, Fagherazzi S, Wollheim WM
- 964 (2012) Coastal eutrophication as a driver of salt marsh loss. Nature 490:388–392.
- 965 Dickson AG, Millero FJ (1987) A comparison of the equilibrium constants for the dissociation
- 966 of carbonic acid in seawater media. *Deep-Sea Research* 34: 1733–1743.
- Dickson AG (1990) Standard potential of the reaction: AgCl(s) + 1/2H2(g) = Ag(s) + HCl(aq),
- and the standard acidity constant of the ion HSO₄⁻ in synthetic sea water from 273.15
- to 318.15 K. Journal of Chemical Thermodynamics 22: 113–127.
- 970 Duarte CM, Cebrián J (1996) The fate of marine autotrophic production. Limnol Oceanogr
- 971 41:1758–1766.
- 972 Dürr HH, Laruelle GG, van Kempen CM, Slomp CP, Meybeck M, Middelkoop H (2011)
- Worldwide Typology of Nearshore Coastal Systems: Defining the Estuarine Filter of
- 974 River Inputs to the Oceans. Estuaries and Coasts 34:441–458.

975	Forbrich I, Giblin AE (2015) Marsh-atmosphere CO ₂ exchange in a New England salt marsh. J
976	Geophys Res Biogeosci 120:1825–1838.
977	Frankignoulle M, Abril G, Borges A, Bourge I, Canon C, Delille B, Libert E, Théate J-M (1998)
978	Carbon Dioxide Emission from European Estuaries. Science, New Series 282:434–436.
979	Gattuso J-P, Frankignoulle M, Wollast R (1998) CARBON AND CARBONATE
980	METABOLISM IN COASTAL AQUATIC ECOSYSTEMS. Annu Rev Ecol Syst
981	29:405–434.
982	Gouazé M (2019) Étude du couple benthos-pélagos dans les marais littoraux et rétro-littoraux
983	de Charente Maritime. La Rochelle Université, La Rochelle.
984	Gu J, Luo M, Zhang X, Christakos G, Agusti S, Duarte CM, Wu J (2018) Losses of salt marsh
985	in China: Trends, threats and management. Estuarine, Coastal and Shelf Science
986	214:98–109.
987	Hill R, Bellgrove A, Macreadie PI, Petrou K, Beardall J, Steven A, Ralph PJ (2015) Can
988	macroalgae contribute to blue carbon? An Australian perspective: Can macroalgae
989	contribute to blue carbon? Limnol Oceanogr 60:1689–1706.
990	Jähne B, Münnich KO, Bösinger R, Dutzi A, Huber W, Libner P (1987) On the parameters
991	influencing air-water gas exchange. J Geophys Res 92:1937.
992	Jiang L-Q, Cai W-J, Wang Y (2008) A comparative study of carbon dioxide degassing in river-
993	and marine-dominated estuaries. Limnol Oceanogr 53:2603–2615.
994	Krause-Jensen D, Duarte CM (2016) Substantial role of macroalgae in marine carbon
995	sequestration. Nature Geosci 9:737–742.

996	Le Fur I, De Wit R, Plus M, Oheix J, Simier M, Ouisse V (2018) Submerged benthic
997	macrophytes in Mediterranean lagoons: distribution patterns in relation to water
998	chemistry and depth. Hydrobiologia 808:175-200.
999	Lê S, Josse J, Husson F (2008) FactoMineR: An R Package for Multivariate Analysis. J Stat
1000	Soft 25.
1001	Lee K, Kim T-W, Byrne RH, Millero FJ, Feely RA, Liu Y-M (2010) The universal ratio of
1002	boron to chlorinity for the North Pacific and North Atlantic oceans. Geochimica et
1003	Cosmochimica Acta 74:1801–1811.
1004	Lewis E, Wallace D (1998) Program developed for CO ₂ system calculations. <i>Carbon dioxide</i>
1005	information analysis center. Oak Ridge National Laboratory
1006	Macreadie PI, Nielsen DA, Kelleway JJ, Atwood TB, Seymour JR, Petrou K, Connolly RM,
1007	Thomson AC, Trevathan-Tackett SM, Ralph PJ (2017) Can we manage coastal
1008	ecosystems to sequester more blue carbon? 8.
1009	Maher DT and Eyre BD (2012) Carbon budgets for three autotrophic Australian estuaries:
1010	Implications for global estimates of the coastal air-water CO ₂ flux, Global Biogeochem.
1011	Cycles, 26, GB1032, doi:10.1029/2011GB004075
1012	Mcleod E, Chmura GL, Bouillon S, Salm R, Björk M, Duarte CM, Lovelock CE, Schlesinger
1013	WH, Silliman BR (2011) A blueprint for blue carbon: toward an improved
1014	understanding of the role of vegetated coastal habitats in sequestering CO ₂ . Frontiers in
1015	Ecology and the Environment 9:552–560.
1016	Mehrbach C, Culberson CH, Hawley JE, Pytkowicz RM (1973) Measurement of the Apparent
1017	Dissociation Constants of Carbonic Acid in Seawater at Atmospheric Pressure.
1018	Limnology and Oceanography 18:897–907.

1019	Najjar RG, Herrmann M, Alexander R, Boyer EW, Burdige DJ, Butman D, Cai W -J., Canuel
1020	EA, Chen RF, Friedrichs MAM, Feagin RA, Griffith PC, Hinson AL, Holmquist JR,
1021	Hu X, Kemp WM, Kroeger KD, Mannino A, McCallister SL, McGillis WR, Mulholland
1022	MR, Pilskaln CH, Salisbury J, Signorini SR, St-Laurent P, Tian H, Tzortziou M, Vlahos
1023	P, Wang ZA, Zimmerman RC (2018) Carbon Budget of Tidal Wetlands, Estuaries, and
1024	Shelf Waters of Eastern North America. Global Biogeochem Cycles 32:389–416.
1025	Newton C, Thornber C (2013) Ecological Impacts of Macroalgal Blooms on Salt Marsh
1026	Communities. Estuaries and Coasts 36:365–376.
1027	Palacios MM, Trevathan-Tackett SM, Malerba ME, Macreadie PI (2021) Effects of a nutrient
1028	enrichment pulse on blue carbon ecosystems. Marine Pollution Bulletin 165:112024.
1029	Paticat F (2007) Flux et usages de l'eaude mer dans les marais salés endigués Charentais: Cas
1030	du marais salé endigué de l'île de Ré. Thèse, Nantes
1031	Polsenaere P (2011) Echange de CO ₂ atmospherique dans la lagune d'Arcachon et relations
1032	avec le métabolisme intertidal. Thèse, Bordeaux
1033	Polsenaere P, Lamaud E, Lafon V, Bonnefond J-M, Bretel P, Delille B, Deborde J, Loustau D,
1034	Abril G (2012) Spatial and temporal CO ₂ exchanges measured by Eddy Covariance over
1035	a temperate intertidal flat and their relationships to net ecosystem production.
1036	Biogeosciences 9:249–268.
1037	Polsenaere P, Deborde J, Detandt G, Vidal LO, Pérez MAP, Marieu V, Abril G (2013) Thermal
1038	enhancement of gas transfer velocity of CO2 in an Amazon floodplain lake revealed by
1039	eddy covariance measurements: GAS TRANSFER VELOCITY IN AN AMAZON
1040	LAKE. Geophys Res Lett 40:1734–1740.

1041	Poisenaere P, Soleichnik P, Le Moine O, Gonin F, Robert S, Pepin J-F, Stamsiere J-Y, Dumas
1042	F, Béchemin C, Goulletquer P (2017) Potential environmental drivers of a regional blue
1043	mussel mass mortality event (winter of 2014, Breton Sound, France). Journal of Sea
1044	Research 123:39–50.
1045	Polsenaere P, Delille B, Poirier D, Charbonnier C, Deborde J, Mouret A, Abril G (2022)
1046	Seasonal, Diurnal, and Tidal Variations of Dissolved Inorganic Carbon and pCO2 in
1047	Surface Waters of a Temperate Coastal Lagoon (Arcachon, SW France). Estuaries and
1048	Coasts.
1049	Raven J (2018) Blue carbon: past, present and future, with emphasis on macroalgae. Biol Lett
1050	14:20180336.
1051	Raymond PA, Cole JJ (2001) Gas Exchange in Rivers and Estuaries: Choosing a Gas Transfer
1052	Velocity. Estuaries 24:312.
1053	REPHY - French Observation and Monitoring program for Phytoplankton and Hydrology in
1054	coastal waters (2021). REPHY dataset - French Observation and Monitoring program
1055	for Phytoplankton and Hydrology in coastal waters. Metropolitan data. SEANOE.
1056	https://doi.org/10.17882/47248
1057	Ribas-Ribas M, Gómez-Parra A, Forja JM (2011) Air–sea CO ₂ fluxes in the north-eastern shelf
1058	of the Gulf of Cádiz (southwest Iberian Peninsula). Marine Chemistry 123:56-66.
1059	Savelli R, Bertin X, Orvain F, Gernez P, Dale A, Coulombier T, Pineau P, Lachaussée N,
1060	Polsenaere P, Dupuy C, Le Fouest V (2019) Impact of Chronic and Massive
1061	Resuspension Mechanisms on the Microphytobenthos Dynamics in a Temperate
1062	Intertidal Mudflat. J Geophys Res Biogeosci 124:3752–3777.

1063	Schäfer KVR, Tripathee R, Artigas F, Morin TH, Bohrer G (2014) Carbon dioxide fluxes of an
1064	urban tidal marsh in the Hudson-Raritan estuary: Carbon dioxide fluxes of an wetland.
1065	J Geophys Res Biogeosci 119:2065–2081.
1066	Soletchnik P, Polsenaere P, Le Moine O, Guesdon S, Bechemin C (2014) Interactions between
1067	river freshwater inputs and the shellfish farming (oysters and mussels) in the Pertuis
1068	Charentais (France).
1069	Stanisiere J-Y, Dumas F, Plus M, Maurer D, Robert S (2006) Hydrodynmaic characterization
1070	of a semi-enclosed coastal system: Marennes Oleron (France) basin.
1071	Takahashi T, Sutherland SC, Sweeney C, Poisson A, Metzl N, Tilbrook B, Bates N,
1072	Wanninkhof R, Feely RA, Sabine C, Olafsson J, Nojiri Y (2002) Global sea-air CO ₂
1073	flux based on climatological surface ocean pCO2, and seasonal biological and
1074	temperature effects. Deep Sea Research Part II: Topical Studies in Oceanography
1075	49:1601–1622.
1076	Teichberg M, Fox SE, Olsen YS, Valiela I, Martinetto P, Iribarne O, Muto EY, Petti MAV,
1077	Corbisier TN, Soto-Jiménez M, Páez-Osuna F, Castro P, Freitas H, Zitelli A,
1078	Cardinaletti M, Tagliapietra D (2010) Eutrophication and macroalgal blooms in
1079	temperate and tropical coastal waters: nutrient enrichment experiments with Ulva spp.
1080	Global Change Biology 16:2624–2637.
1081	Ternon Q, Polsenaere P, Le Fouest V, Favier J-B, Philippine O, Chabirand J-M, Grizon J,
1082	Dupuy C (2018) Étude des pressions partielles et flux de CO ₂ au sein de la Communauté
1083	d'Agglomération de La Rochelle. 32.

1084	Tobias C, Neubauer SC (2019) Sait Marsh Biogeochemistry—An Overview. in: Coastai
1085	Wetlands. Elsevier, p 539–596
1086	Tortajada S (2011) De l'étude du fonctionnement des réseaux trophiques planctoniques des
1087	marais de Charente Maritime vers la recherche d'indicateurs. Thèse, Université La
1088	Rochelle, La Rochelle
1089	Tortajada S, David V, Brahmia A, Dupuy C, Laniesse T, Parinet B, Pouget F, Rousseau F,
1090	Simon-Bouhet B, Robin F-X (2011) Variability of fresh- and salt-water marshes
1091	characteristics on the west coast of France: A spatio-temporal assessment. Water
1092	Research 45:4152–4168.
1093	Van Dam B, Polsenaere P, Barreras-Apodaca A, Lopes C, Sanchez-Mejia Z, Tokoro T, Kuwae
1094	T, Loza LG, Rutgersson A, Fourqurean J, Thomas H (2021) Global Trends in Air-Water
1095	CO ₂ Exchange Over Seagrass Meadows Revealed by Atmospheric Eddy Covariance.
1096	Global Biogeochem Cycles 35.
1097	Vandermeirsch F (2012) ÉTAT PHYSIQUE ET CHIMIQUE Caractéristiques physiques.
1098	Wang SR, Di Iorio D, Cai W, Hopkinson CS (2018) Inorganic carbon and oxygen dynamics in
1099	a marsh-dominated estuary. Limnol Oceanogr 63:47–71.
1100	Wang ZA, Cai W-J (2004) Carbon dioxide degassing and inorganic carbon export from a
1101	marsh-dominated estuary (the Duplin River): A marsh CO2 pump. Limnol Oceanogr
1102	49:341–354.
1103	Wang ZA, Kroeger KD, Ganju NK, Gonneea ME, Chu SN (2016) Intertidal salt marshes as an
1104	important source of inorganic carbon to the coastal ocean. Limnol Oceanogr 61:1916-
1105	1931.

1106	Wanninkhof R (1992) Relationship between wind speed and gas exchange over the ocean. J
1107	Geophys Res 97:7373.
1108	Wei, T., & Simko, V. (2017). R package "corrplot": Visualization of a Correlation Matrix.
1109	https://github.com/taiyun/corrplot
1110	Weiss RF (1974) Carbon dioxide in water and seawater: the solubility of a non-ideal gas.
1111	Marine Chemistry 2:203–215.
1112	Yates KK, Dufore C, Smiley N, Jackson C, Halley RB (2007) Diurnal variation of oxygen and
1113	carbonate system parameters in Tampa Bay and Florida Bay. Marine Chemistry
1114	104:110–124.
1115	
1116	