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Diastereo- and Enantioselective (3+2) Cycloaddition of a New Palladium-Aza-Oxyallyl Intermediate

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Supporting Information Placeholder

ABSTRACT: Cycloaddition of azaoxyallyl cations or other C–(C=O)–N synthon precursors is a well-established route towards lactams and other *N*-heterocycles but, despite the wide synthetic scope of this approach, enantioselective versions remain scarce. We herein report 5-vinyloxazolidine-2,4-diones (VOxD) as a suitable precursor of chiral palladium-aza-oxyallyl intermediates. In the presence of electrophilic alkenes, (3+2) γ-lactam cycloadducts could be formed with high level of diastero- and enantioselectivity.

Nitrogen heterocycles are ubiquitous scaffolds in bioactive compounds and can display a wide range of activities.¹ Among them, lactams are a unique class of heterocycles which continue to draw the interest of medicinal chemists and pharmaceutical industries (Scheme 1a).²

A plethora of methods are available to the synthetic chemist who desires to access these heterocyclic structures and cycloadditions, allowing the formation of two σ -bonds in one step,³ naturally emerge as an advantageous strategy. Dipolar cycloadditions of synthetic equivalents of the crucial C–(C=O)–N synthon have therefore attracted considerable interest. While the reactivity of α-haloamides has been widely documented, 4 only two enantioselective dipolar cycloadditions have been reported and rely on the activation of the electrophilic cycloaddition partner by a squaramide organocatalyst.⁵ The group of Hamada exploited the dipolar properties of acrylamides to design an enantioselective $(3+2)$ cycloaddition based on a chiral iminium activation method.⁶ α-Ketoamides have also been involved as a synthetic equivalent of the same synthon and can act as a nucleophilic cycloaddition partner through the same iminium catalysis approach⁷ or as the electrophilic partner with cinnamaldehydes activated by a chiral NHC catalyst (Scheme $2b)$.⁸

It is worth pointing out that all these strategies do not rely on the catalytic activation of the C–(C=O)–N synthon's equivalent but rather of the other cycloaddition partner, hampering their generalization to the preparation of other *N*-heterocycles. To answer this shortfall, we envisioned to build on the research interest of group⁹ and develop a cycloaddition involving a π -allyl- Pd^H zwitterionic dipolar intermediate. This strategy has drawn considerable interest over the last decade¹⁰ and Pd-stabilized aza-dipoles have notably been utilized to access various enantioenriched nitrogenated heterocycles in the presence of a chiral palladium ligand.¹¹⁻¹³

Scheme 1. Asymmetric cycloadditions of the C–(C=O)–N synthon.

We identified the original vinyloxazolidine-2,4-diones **A** (VOxD, Scheme 1c) as promising new synthetic equivalents of the desired C–(C=O)–N synthon. Under catalysis with a PdL*

complex, an oxidative addition would furnish, after decarboxylation, the putative dipolar intermediate **B** bearing a nucleophilic nitrogen and an electrophilic π-allyl-Pd^{II} moiety. In this manuscript, we report the first use of a VOxD precursors to access relevant γ-lactam heterocycles with high diastero- and enantioselectivities (Scheme 1c).

After designing a straightforward multi-gram scale preparation of these original starting materials,¹⁴ the feasibility of an asymmetric palladium-catalyzed cycloaddition was explored with vinyloxazolidin-2,4-dione **1a** and 2-benzylidenemalononitrile **2a** as test substrates. During the finalization of this manuscript, the group of Guo also reported the use of VOxD for Pd-catalyzed asymmetric $(5+4)$ and $(3+2)$ cycloadditions.¹⁵

A first screening of chiral ligands was performed (THF, rt) but no reactivity was observed with common diphosphine ligands (Table 1, entries 1-3). Gratifyingly, we then identified phosphoramidites as adequate ligands for this original transformation and (*S*)-**L4** furnished **3aa** in moderate yield and good diastereoselectivity albeit with low enantioselectivity (Table 1, entry 4). The enantioselectivity could be drastically improved using sterically hindered phosphoramidite ligand (*S*)-**L5** derived from di*iso*propylamine (Table 1, entry 5). After a first screening (Table 1, entries 5-7), toluene was identified as a promising solvent and the enantioselectivity could then be enhanced using the bulkier phosphoramidite (*S*,*R*,*R*)-**L6** (Table 1, entry 9, 96:4 *er*) albeit with a sluggish reactivity (41%).

Table 1. Optimization of reaction conditions.^a

^aReaction conditions: **1a** (0.075 mmol), **2a** (0.050 mmol), Pd₂dba₃ (5 mol\%) , ligand LX (x mol%) in solvent $(0.5 \text{ mL}, 0.1 \text{M})$ b Isolated yield. "Determined by ¹H NMR analysis of the crude product. ^dDetermined by chiral HPLC analysis.

It is worth noting that the diastereomer (*S*,*S*,*S*)-**L7** only led to the recovery of starting materials (Table 1, entry 10). Finally, a second screening of solvents furnished **3aa** with better enantioselectivity when the reaction was performed in *o*-xylene (Table 1, entry 11) and a better reactivity was restored by increasing slightly the amount of ligand while working at 70 °C (Table 1, entries 12 and 13).

Having these optimized conditions in hand, we first explored the reactivity of VOxD bearing different substituents on the nitrogen atom $(R¹$ group) as the putative transformation mechanism highlights the critical influence of its electronic properties (Scheme 2). Indeed, an electron withdrawing group will result in a more stabilized zwitterionic intermediate **B**, favoring the decarboxylation step but also diminishing its nucleophilicity. On the other hand, a more nucleophilic intermediate could be formed by installing a more electron-donating $R¹$ group but its formation would require a more energy demanding decarboxylation process.

When compared to our initial optimization study using $1a(R¹)$ $=$ Ph), VOxD **1b** (R ¹ = Bn) and **1c** (R ¹ = PMB) possess more electron-donating R^1 groups but the corresponding (3+2) cycloadducts **3ba** and **3ca**, respectively, were generated with excellent enantioselectivities under the same reaction conditions. However, no reactivity was observed with VOxD 1d $(R¹ =$ C_3H_5) and **1e** (R^1 = Me) and increasing the temperature to 90 °C and 110 °C, respectively, was necessary to obtain γ-lactams **3da** (45%) and **3ea** (59%) while retaining a similar stereoselectivity. VOxD **1f** bearing a more electron-donating cyclohexyl group was not found to be a suitable substrate as the decarboxylation step might be hampered even at higher temperatures (*o*-xylene, 130 °C). On the other hand, the decarboxylation step would be favored with VOxD **1g** substituted by an electron-withdrawing tosyl group but the nucleophilicity of the resulting zwitterionic intermediate might be insufficient for the cycloaddition to proceed as cycloadduct **3ga** was not observed.

Vinyloxazolidine-2,4-diones **1h-1k** derived from various anilines (R^1 = Ar) were applicable and similar yields and stereoselectivity were observed. A steric hindrance at this position resulted in a lower yield for the preparation of γ-lactam **3ha** (2-O-MeC6H4) while maintaining high level of enantioselectivity (98:2 *er*). To further explore the tolerance of this methodology, different aryl groups were installed at the allylic position. Steric hindrance slightly decreased the diastereoselectivity as γ-lactam **3la** (2-Me) was obtained with a 80:20 diastereoisomeric ratio but we were pleased to observe the enantioselective formation of cycloadducts **3ma** (4-OMe), **3na** (4-Cl) and **3oa** (4-CF3) bearing electron-withdrawing or donating groups at the *para* position (Scheme 2).

We then began to explore the scope of the electrophilic cycloaddition partner, starting with a variety of benzylidenemalononitrile derivatives. Electron-rich phenyl groups were well tolerated and **3ab**, **3ac** and **3ad** bearing a methoxy group at the *ortho*, *meta* and *para* position, respectively, could be readily obtained from the corresponding activated alkenes with very good enantioselectivities.The presence of electron-withdrawing groups on the aryl group resulted in a slight loss of diastereoselectivity, but excellent enantioselectivities (98:2 *er*) were obtained for the cycloaddition of VOxD **1a** with alkenes **2e** (3- CNC_6H_4) and **2f** (4- $CF_3C_6H_4$).

Scheme 2. Scope of vinyloxazolidine-2,4-diones.^a

^aReaction conditions: **1a-1o** (0.30 mmol), **2a** (0.20 mmol), Pd₂dba₃ (5 mol%), **L6** (24 mol%), *o*-xylene (2 mL, 0.1M), 70 °C. ^bReaction run at 90 °C. ^cReaction run at 110 °C. ^dReaction run at 130 °C.

The cycloaddition of sterically demanding substrate **2g** was also carried out without loss of stereoselectivity (99:1 *er*) and γlactam **3ag** (93%), bearing a 1-naphthyl group, could be easily prepared. This methodology further accommodates the presence of a 2-furyl and a ferrocenyl group as the two corresponding cycloadducts **3ah** (81%, 97:3 *er*) and **3ai** (64%, 98:2 *er*) were isolated with excellent enantioselectivities. Finally, (3+2) cycloadduct **3aj** (93%), bearing a *para*-bromo-phenyl moiety, was prepared (97:3 *er*) and its absolute configuration was determined by single-crystal X-ray analysis.¹⁵ Throughout our study, we also examined the reactivity of cinnamylidene and alkylidene malononitriles **2k** and **2l**, respectively. In both cases, we were able to prepare the corresponding racemic cycloadducts, but a complex mixture was unfortunately obtained under the optimized conditions with chiral phosphoramidite ligands **L6** or **L4** (Scheme 3).

To continue our investigation for other suitable electrophiles, we involved benzylidene **4a** derived from indane-1,3-dione and benzaldehyde. Under the same optimized conditions (**L6**, *o*-xylene, 70 °C), the expected spiro-γ-lactam **7aa** (65%, 91:9 *dr*) was isolated with high enantioselectivity (98:2 *er*) and a variety of other benzylidenes were successfully engaged under these conditions. Electron-rich and electron-poor substituents were well-tolerated at the *para* position and spirocycloadducts **7ab** $(4\text{-}OMe)$ and **7ac** $(4\text{-}CF₃)$ were generated with excellent enantioselective ratios.

Scheme 3. Cycloadditions of benzylidenemalononitriles. a

^aReaction conditions: **1a** (0.30 mmol), **2b**-**2l** (0.20 mmol), Pd2dba³ (5 mol%), **L6** (24 mol%), *o*-xylene (2 mL, 0.1M), 70 °C. ^bPPh³ was used as the ligand.

This strategy also allowed the preparation of sterically hindered lactam **7ad** bearing a 1-naphthyl group. Halogen atoms are susceptible to tempering the course of the transformation as a competitive oxidative addition of the palladium(0) catalyst can occur but benzylidenes **4e** (4-Cl), **4f** (3-Br) and **4g** (4-Br) were converted in moderate to good yields to the corresponding (3+2) γ-lactams cycloadducts. The reactivity of less activated alkenes 5 (EWG = COPh) and 6 (EWG = CO₂Me) was also examined but these substrates have proven to be unreactive under the optimized conditions (Scheme 4).

Scheme 4. Scope of other C=C acceptors.^a

^aReaction conditions: **1a** (0.30 mmol), **4**-**6** (0.20 mmol), Pd2dba³ (5 mol%), **L6** (24 mol%), *o*-xylene (2 mL, 0.1M), 70 °C.

Finally, we demonstrated the utility of our method by performing further synthetic transformations. After its preparation on a mmol scale, the deprotection of γ-lactam **3ca**, bearing a *para*methoxybenzyl group (CAN, $MeCN/H₂O$) afforded the free amide **10** (90%) with an unaltered enantiomeric ratio of 97:3. Under usual conditions $(Ac₂O, 4-DMAP)$, this latter can subsequently lead to the enantioenriched acetylated product **11** which could not be directly attained with our strategy. Finally, the ozonolysis of γ-lactam **3aa** afforded efficiently aldehyde **12** without deterioration of the enantiomeric ratio. (Scheme 5).

Scheme 5. Further synthetic transformations.

In summary, we have developed a straightforward synthesis of original vinyloxazolidine-2,4-diones (VOxD) which have been engaged in enantioselective $(3+2)$ cycloadditions towards γ -lactams under palladium catalysis. This study establishes VOxD as versatile synthetic equivalents of the ubiquitous $C-(C=O)$ –N synthon in asymmetric cycloaddition reactions and further studies are ongoing in our laboratory to better understand and exploit their reactivity with other cycloaddition partners.

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