



## Kinetic limitations of a graphite electrode

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# Kinetic limitations of a graphite electrode

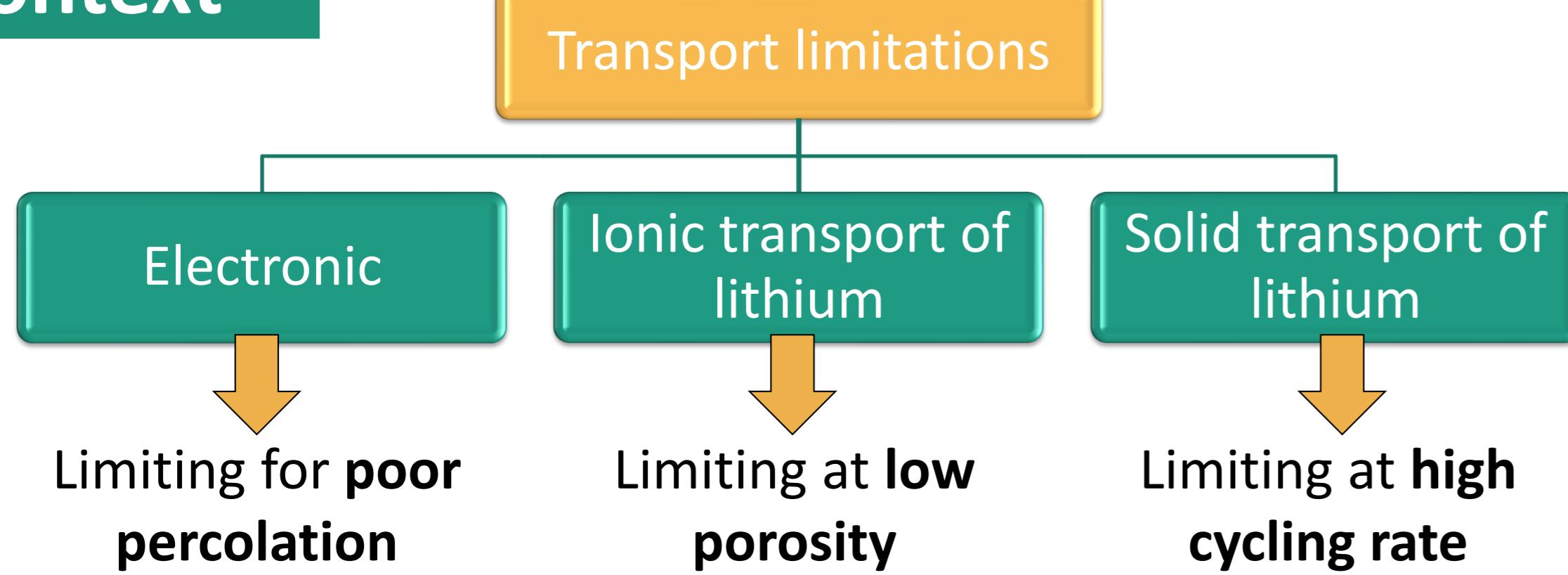
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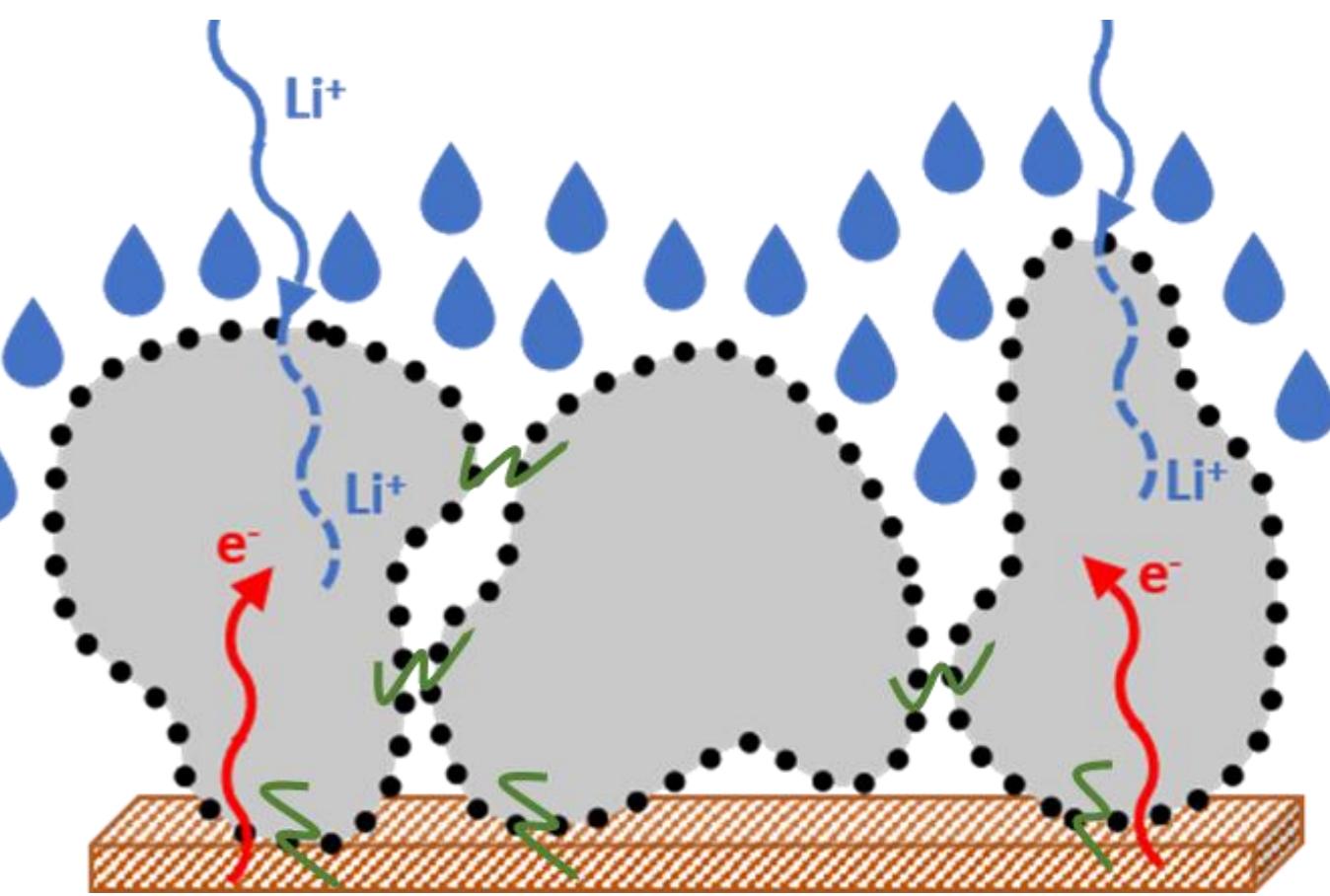
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## Context

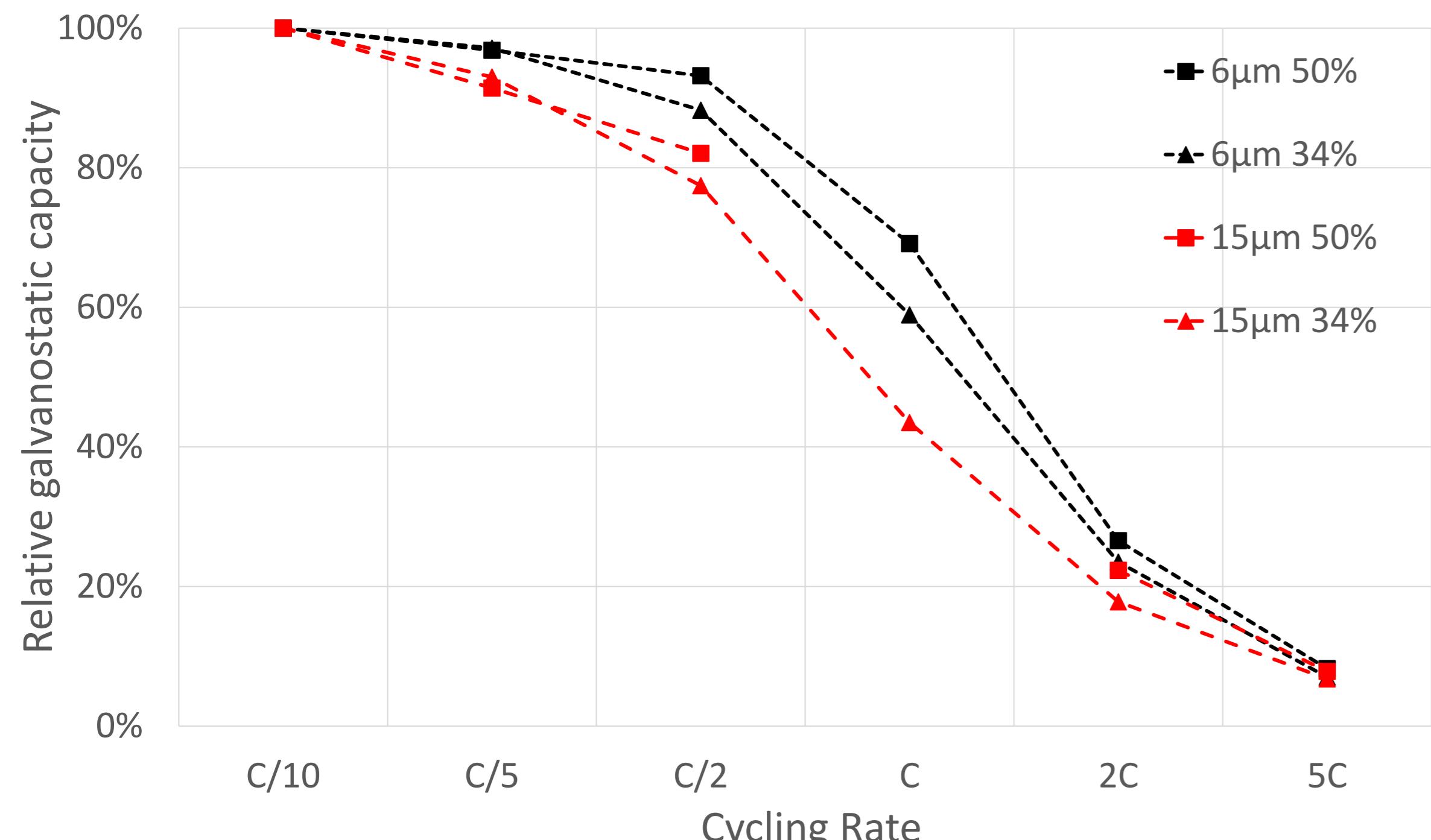


Every system has its limitations ... **kinetic limitations** are the one limiting Li-ion batteries, especially the graphite electrode !

Which transport processes are limiting the system ?



## In-lab insight of kinetic limitations



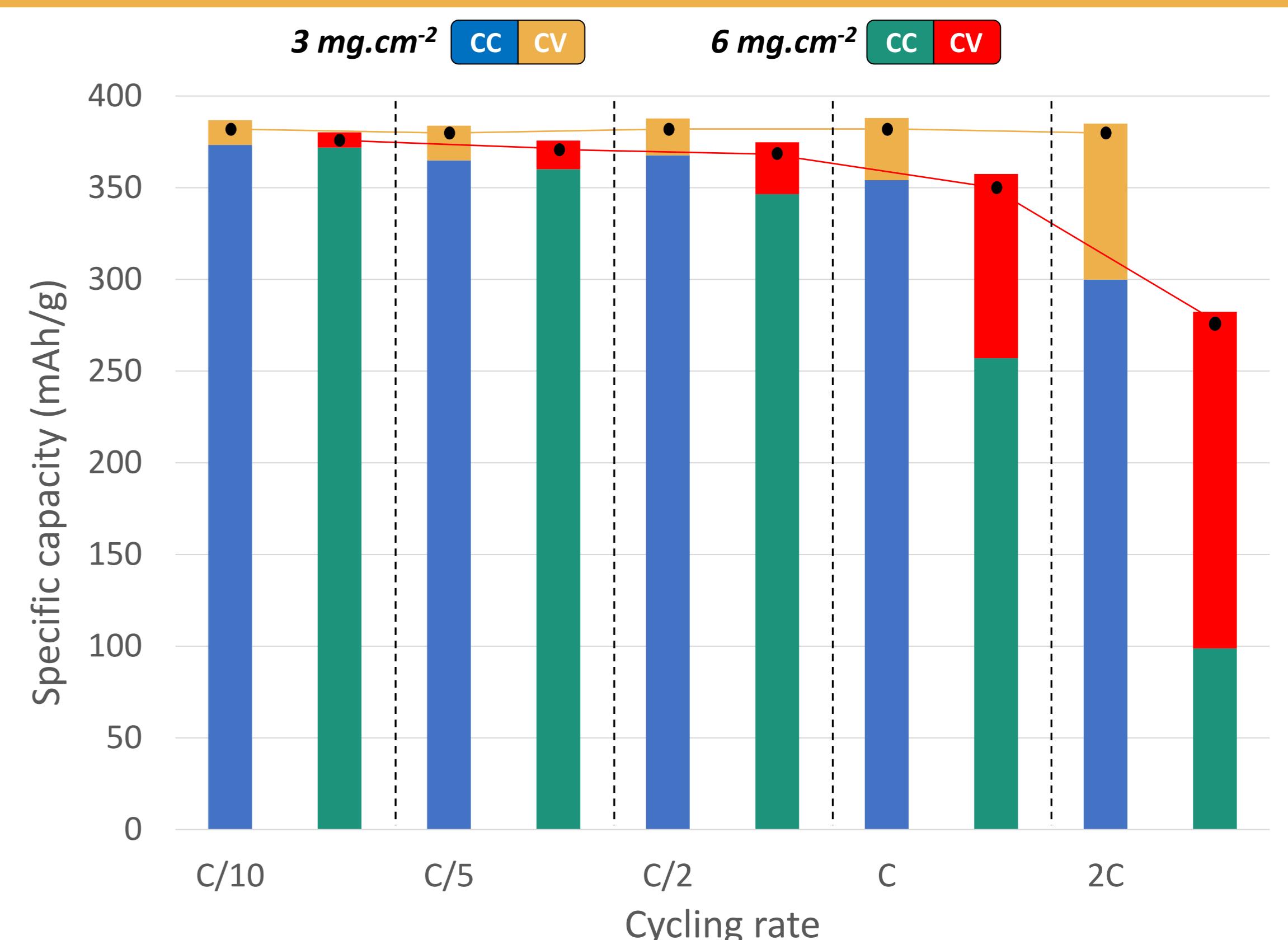
❖ Decreasing the porosity complexifies **electronic** and **ionic transport**

### CC – CV cycling

- 1<sup>st</sup>: **galvanostatic step (CC)**  
→ 1,2V – 10 mV vs  $\text{Li}^+/\text{Li}$   
→ sensitivity to **kinetic limitations**
- 2<sup>nd</sup>: **potentiostatic step (CV)**  
→ 30 min at 10 mV vs  $\text{Li}^+/\text{Li}$   
→ **thermodynamic conditions**

Ionic transport      Electronic transport

Which one is limiting ?

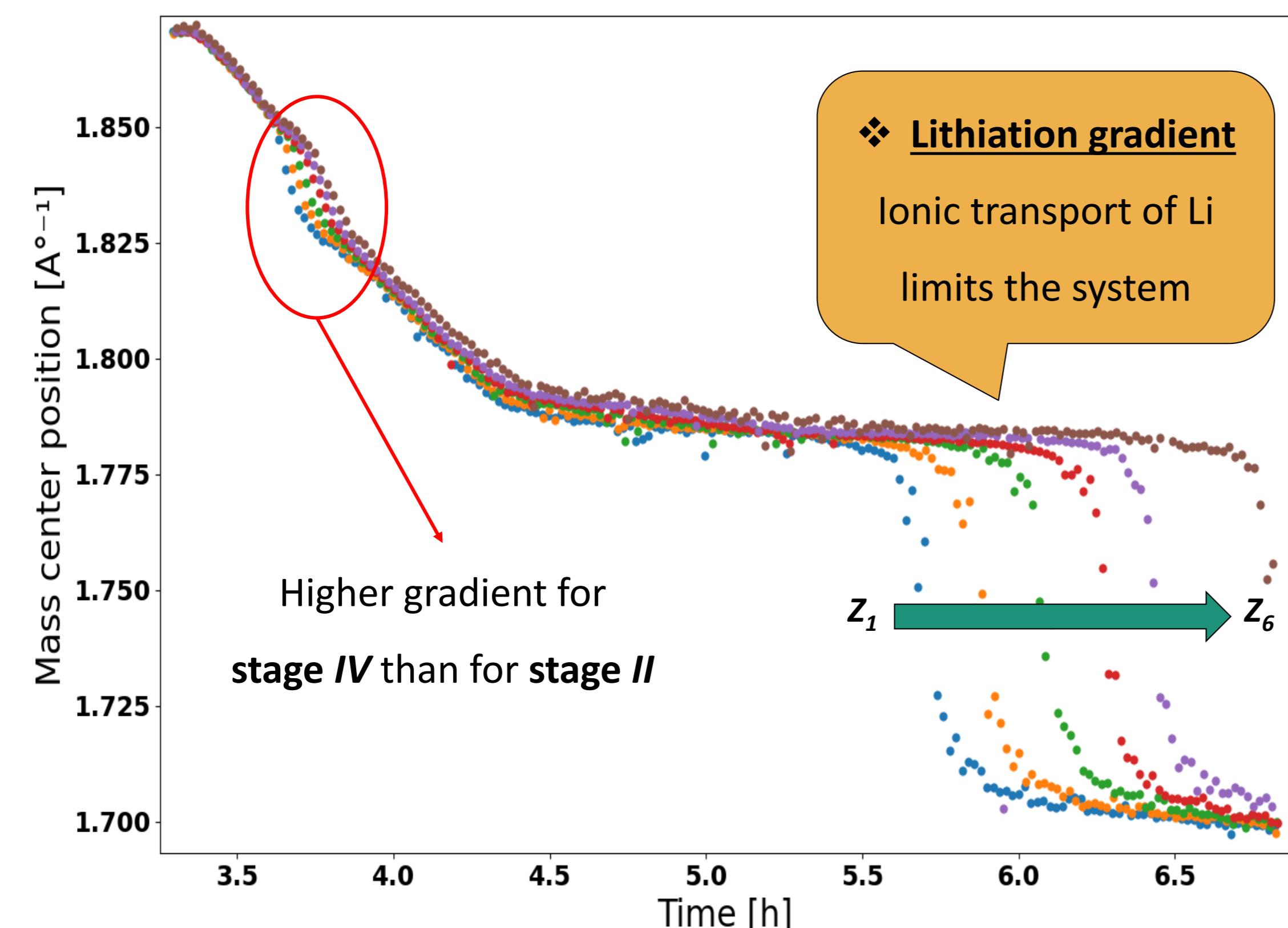
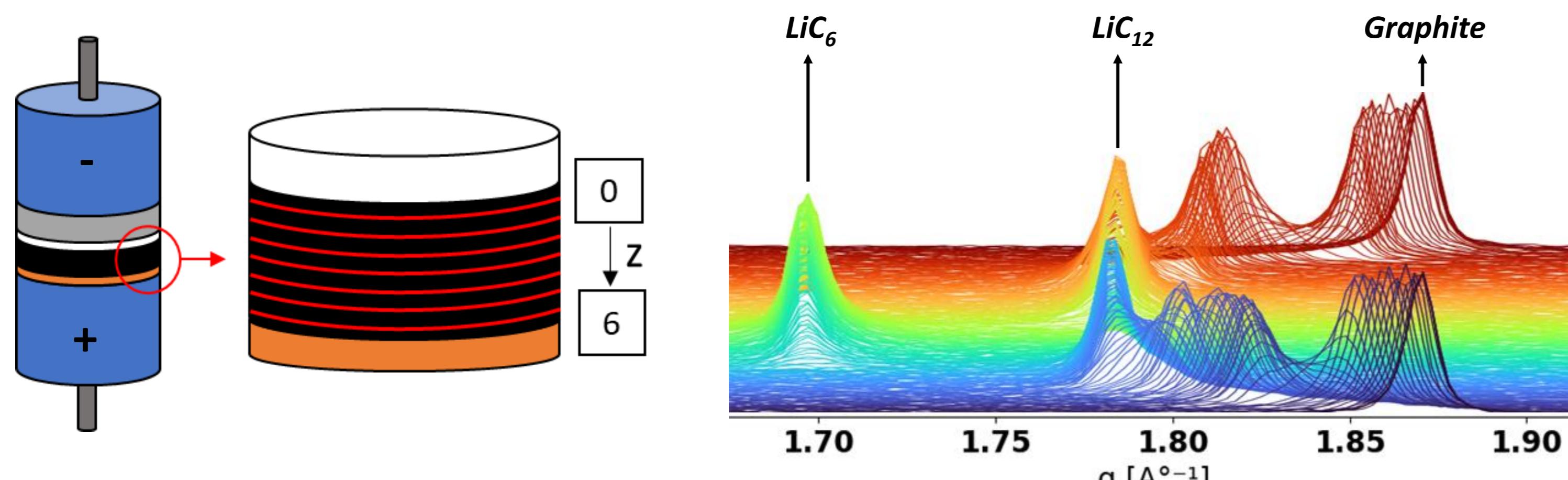


❖ Increasing the loading impacts **electronic** and **ionic transport**

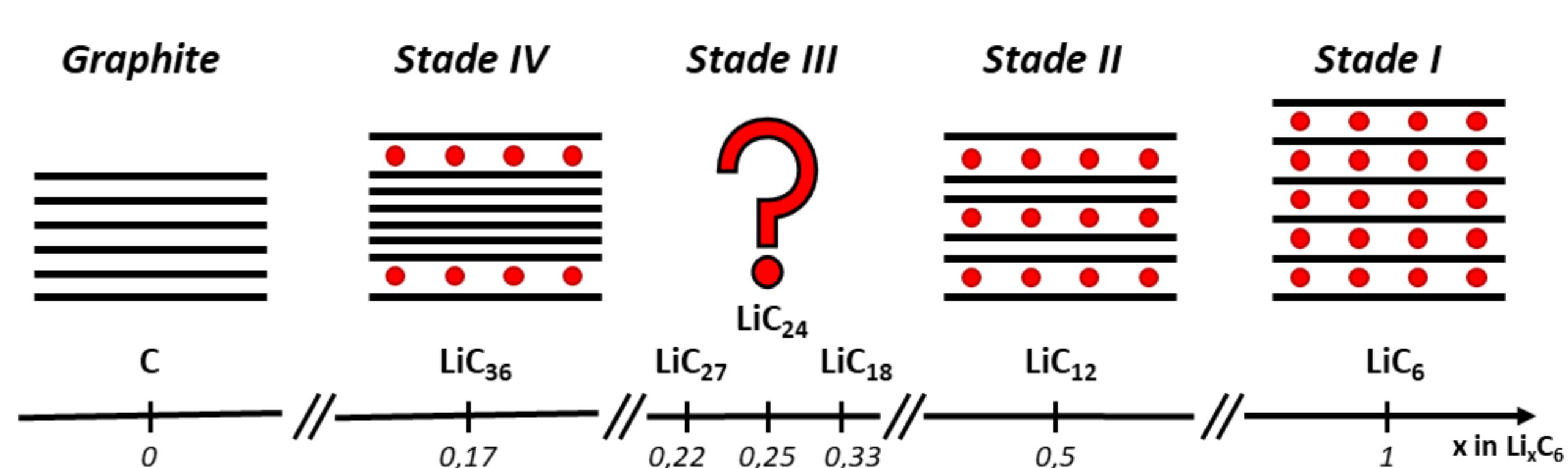
## Synchrotron insight of kinetic limitations

### ID31 – HIGH ENERGY BEAMLINE FOR BURIED INTERFACE STRUCTURE

- Operando measurement** with 220 000 diffractograms collected
- Combination of electrochemical cycling and **WAXS analysis**
- Electrode of **6 μm** graphite particles at **50%** of porosity and **6 mg.cm<sup>-2</sup>**



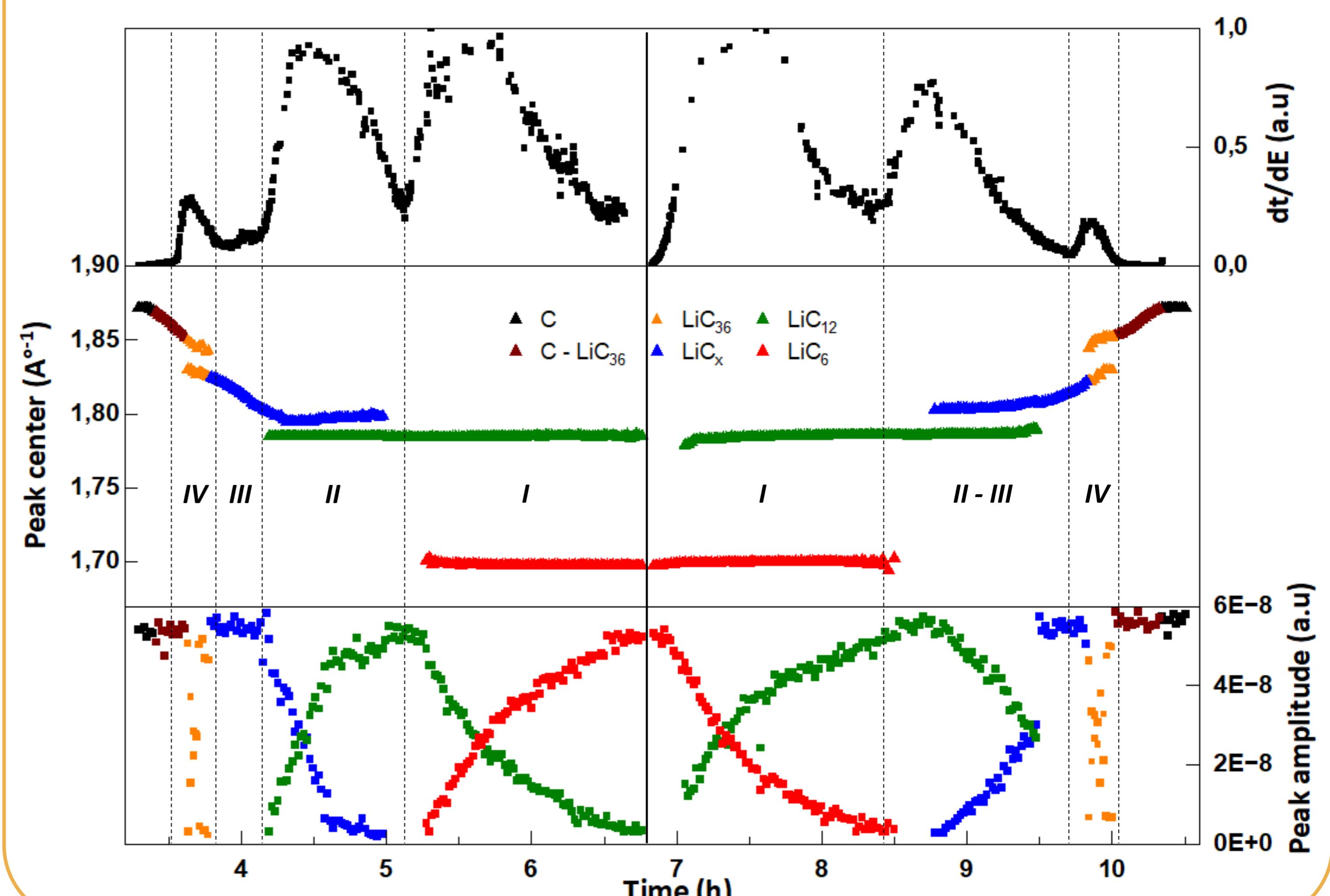
## Additional characterization



### Peak characterization by fitting

- Peak center
- Amplitude
- FWHM

❖ Biphasic behaviour for stages IV, II and I  
❖ Liquid like behaviour for stage III



## Overview

- Ionic limitation of the electrode at a porosity of 50%
- What about the limitation for **higher porosity** ?

- ❖ GITT experiment to determine **Li coefficient diffusion** for each stage
- ❖ What about the **rate capability** ?