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# Accurate $a b$ initio potential energy surface, rovibrational energy levels and resonance interactions of triplet ( $\widetilde{X}^{3} B_{1}$ ) methylene 

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#### Abstract

In this work, we report rovibrational energy levels for four isotopologues of methylene $\left(\mathrm{CH}_{2}, \mathrm{CHD}\right.$, $\mathrm{CD}_{2}$, and ${ }^{13} \mathrm{CH}_{2}$ ) in their ground triplet electronic state ( $\tilde{X}^{3} B_{1}$ ) from variational calculation up to $\sim 10000 \mathrm{~cm}^{-1}$ and using a new accurate $a b$ initio potential energy surface (PES). Triplet methylene exhibits a large-amplitude bending vibration and can reach a quasilinear configuration due to its low barrier ( $\sim 2000 \mathrm{~cm}^{-1}$ ). To construct the ab initio PES, the Dunning's augmented correlation-consistent core-valence orbital basis sets were employed up to the sextuple- $\zeta$ quality [aug-cc-pCVXZ, $\mathrm{X}=\mathrm{T}, \mathrm{Q}$, 5 , and 6] combined with the single- and double-excitation unrestricted coupled cluster approach with a perturbative treatment of triple excitations [RHF-UCCSD(T)]. We have shown that the accuracy of the $a b$ initio energies is further improved by including the corrections due to the scalar relativistic effects, DBOC and high-order electronic correlations. For the first time, all the available experimental rovibrational transitions were reproduced with errors less than $0.12 \mathrm{~cm}^{-1}$, without any empirical corrections. Unlike more "traditional" nonlinear triatomic molecules, we have shown that even the energies of the ground vibrational state (000) with rather small rotational quantum numbers are strongly affected by the very pronounced rovibrational resonance interactions. Accordingly, the polyad structure of the vibrational levels of $\mathrm{CH}_{2}$ and $\mathrm{CD}_{2}$ was analyzed and discussed. The comparison between the energy levels obtained from the effective Watson $A$-reduced Hamiltonian, from the generating-function approach and from a variational calculation was given.


Keywords: $\mathrm{CH}_{2}$, triplet methylene, ab initio, rovibrational levels, large-amplitude motion, centrifugal distortion, resonance interactions


Graphical Abstract
Overtones of the bending mode of triplet methylene predicted from the developed ab initio PES

## 1. INTRODUCTION

Methylene $\left(\mathrm{CH}_{2}\right)$ is a simple hydrocarbon radical with a complicated rovibrational structure in its ground triplet electronic state ( $\tilde{X}^{3} B_{1}$, hereafter denoted as $\tilde{X}$ ) due to the low barrier to linearity $\left(\sim 2000 \mathrm{~cm}^{-1}\right)$ and the anomalous centrifugal distortion. Triplet methylene $\left[\mathrm{CH}_{2}(\tilde{X})\right]$ was the subject of many works, mainly devoted to the rovibronic spectra within the ground and first excited vibrational states. Fundamental stretching bands, overtones and combination bands were not assigned from high-resolution spectra so far. This is partly explained by the presence of the low-lying excited singlet state ( $\tilde{a}^{1} A_{1}$, hereafter denoted by $\tilde{a}$ ) which is responsible for the strongest absorption above $3000 \mathrm{~cm}^{-1}$. Moreover, the extremely short lifetime of $\mathrm{CH}_{2}$ due to its high reactivity makes the spectra analysis very difficult.

Since $\mathrm{CH}_{2}$ is one of the primary products involved in the methane photolysis (see e.g. Blitz et $a l .{ }^{1}$ ), it is likely that it exists in methane-rich planetary atmospheres and in the interstellar medium. According to Gans et al., ${ }^{2}$ the quantum yields of $\mathrm{CH}_{2}$ are comparable with that of the methyl radical $\left(\mathrm{CH}_{3}\right)$ while the amount of $\mathrm{CH}_{2}(\tilde{X})$ and $\mathrm{CH}_{2}(\tilde{a})$ strongly depends on the excitation wavelength. If the wavelength differs from the Lyman- $\alpha$ line at 121.6 nm , then it can lead to the formation of both $\mathrm{CH}_{3}$ and $\mathrm{CH}_{2}(\tilde{X})$. An additional formation channel of $\mathrm{CH}_{2}(\tilde{X})$ can be due to the inelastic collisions of $\mathrm{CH}_{2}($ $\tilde{a})$ with atmospheric species. The rate coefficients measured by Douglas et al. ${ }^{3}$ demonstrated a dominance of the relaxation to the ground electronic state when the temperature is below 100 K .

At the present time, $\mathrm{CH}_{2}(\tilde{X})$ was observed in the interstellar medium near the Orion-KL and the W51 M sources by Hollis et al., ${ }^{4}$ and near molecular cloud complexes Sagittarius B2 and W49 N by Polehampton et al. ${ }^{5}$ In both cases, the unambiguous detection became possible with the help of accurate spectroscopic data for which fine and hyperfine structures were measured for several pure rotational transitions of $\mathrm{CH}_{2}(\tilde{X})$ (see e.g. Lovas et al., ${ }^{6}$ Michael et al., ${ }^{7}$ and Brünken et al. ${ }^{8,9}$ ).

This work was motivated by the following two reasons.

1. No accurate $a b$ initio potential energy surfaces (PES) of $\mathrm{CH}_{2}(\tilde{X})$ constructed from highlevel orbital basis sets and including corrections such as the diagonal Born-Oppenheimer correction (DBOC), scalar relativistic effects and high-order electronic correlations were available so far. The most recent theoretical calculations to date using the aug-cc-pCVQZ basis set were performed by Furtenbacher et al. ${ }^{10}$ Their band origin for the bending band $v_{2}$ was predicted with an error of $4 \mathrm{~cm}^{-1}$, which is generally not enough for atmospheric applications. According to the recent paper by Coudert, ${ }^{11}$ the global fit of the $\mathrm{CH}_{2}(\tilde{X})$ transitions is not unambiguous and "... a more accurate potential energy surface is needed to understand the nature of these discrepancies...".
2. No accurate variationally-computed highly-excited energy levels of $\mathrm{CH}_{2}(\tilde{X})$ were available so far, though they are required to construct comprehensive line lists. In a series of papers by Bunker
and Jensen, ${ }^{12-15}$ the perturbation-based approach ("Nonrigid bender Hamiltonian") as well as the kinetic energy operator implemented in the MORBID computer code of Jensen ${ }^{16}$ were employed, together with empirically-refined PESs. However, as it was demonstrated previously by Barclay et al., ${ }^{17}$ the truncated kinetic energy operator involved in MORBID may introduce errors of several wavenumbers during energy levels calculation of quasilinear molecules such as $\mathrm{CH}_{2}(\tilde{X})$.

Currently, accuracy of the rovibrational transitions is typically within a wavenumber using the best available high level $a b$ initio methods. ${ }^{18,19}$ One of the advantages of using ab initio PESs in conjunction with variational calculations is that all vibrational degrees of freedom are properly taken into account. In the effective approach, only a small set of interacting vibrational states is generally studied and beyond the range of observed data the extrapolation power is often limited. Consequently, performing variational calculations is more suitable for $\mathrm{CH}_{2}(\tilde{X})$ because of the lack of experimental data for high energy levels. Császár et al. ${ }^{20}$ demonstrated that the band origin of $v_{2}$ can be predicted with an error of $1.5 \mathrm{~cm}^{-1}$. In this work, we have shown that such accuracy can be improved by a factor 20. Moreover, we focused not only on the fundamental bands but also on the simultaneous calculation of all vibrational bands up to $\sim 10000 \mathrm{~cm}^{-1}$ for the four isotopologues $\mathrm{CH}_{2}, \mathrm{CHD}, \mathrm{CD}_{2}$, and ${ }^{13} \mathrm{CH}_{2}$. To this end, a new accurate ab initio PES will be presented in this paper.

The article is organized as follows. In the next Section 2 the state-of-the-art in the spectroscopy of triplet methylene will be given. Section 3 will present the $a b$ initio methods used in the present work and give the analytical form of the PES allowing a good description of both the equilibrium geometry and height of the barrier to linearity. The nuclear-motion Hamiltonian described in Ref. 21 will be also briefly discussed. In Section 4, the calculated rovibrational energy levels will be compared to both available experimental data and previous $a b$ initio results. Unusual resonances and centrifugal distortions will be discussed before concluding.

To perform $a b$ initio calculations, the MOLPRO, ${ }^{22,23}$ MRCC, ${ }^{24,25}$ and CFOUR ${ }^{26,27}$ packages were used. The nuclear motion calculations were carried out using our home-made variational computer code TENSOR ${ }^{28-30}$ applied with success in Refs. 31-43 for computing accurate rovibrational energy levels and line intensities of different polyatomic semirigid and nonrigid molecules.

## 2. SPECTROSCOPY OF $\mathrm{CH}_{2}(\widetilde{X})$ : STATE-OF-THE-ART

### 2.1. Experimental studies

First successful spectroscopic studies of $\mathrm{CH}_{2}$ were done by Herzberg et al. ${ }^{44-46}$ by the flash photolysis of diazomethane $\left(\mathrm{CH}_{2} \mathrm{~N}_{2}\right)$ in the presence of the inert gas $\mathrm{N}_{2}$. Several experiments were carried out with deuterated diazomethane. Two band systems belonging to $\mathrm{CH}_{2}$ were thus located: one in the ultraviolet region (at 141.5 nm ) and another one in the red region at $819,731.5$, and 653.1
nm . The spectral structures of the ultraviolet band suggested that $\mathrm{CH}_{2}$ had to be close to the linear configuration in its ground triplet electronic state while the observed $\Delta K$ series in the red region were typical for a bent singlet configuration. Since the intensity of the ultraviolet band increases with the concentration of the inert gas, more collisions were needed to reach the triplet state which probably corresponds to the "true" ground electronic state of $\mathrm{CH}_{2}$, although the energy difference between the two lower states was not accurately determined.

Further theoretical and experimental works allowed to determine how much $\mathrm{CH}_{2}(\tilde{X})$ deviates from linearity. According to the electron paramagnetic resonance spectra measured by Bernheim et al..$^{47,48}$ and Wasserman et al., ${ }^{49-52} \mathrm{CH}_{2}(\tilde{X})$ has a strongly bent configuration with an angle around $130^{\circ}-140^{\circ}$. This was confirmed from ab initio calculations performed by Harrison et al., , ${ }^{53,54}$ Bender et al., ${ }^{55,57}$ O'Neil et al., ${ }^{58}$ and Hay et al. ${ }^{59}$ The anomalously large values of the rotational constants were corroborated by the ultraviolet spectrum at 141.5 nm observed by Herzberg et al. ${ }^{44-46}$ The missing transitions for $K>0$ can be explained by the predissociation properties of the upper electronic state, as discussed in Ref. 60.

First observation of pure rotational transitions of $\mathrm{CH}_{2}(\tilde{X})$ was reported by Mucha et al. ${ }^{61}$ using the laser magnetic resonance (LMR) spectroscopy technique. High-resolution spectra were then systematically studied by Sears, McKellar et al. using both the LMR $^{62-65}$ and diode laser spectroscopy ${ }^{66-68}$ Fine and hyperfine structures for the pure rotational and fundamental vibrational bands were then characterized while the fit of molecular effective parameters for the (000) and (010) vibrational states allowed to obtain an accurate equilibrium geometry of $\mathrm{CH}_{2}(\tilde{X})$ as well as the band center of $v_{2}$. These values were also obtained for $\mathrm{CD}_{2}(\tilde{X})^{66,69,70}$ and ${ }^{13} \mathrm{CH}_{2}(\tilde{X}) .{ }^{71}$

In addition, McKellar et al. ${ }^{65}$ measured the rotational transitions for the ground vibrational state of $\mathrm{CH}_{2}(\tilde{a})$ as well as some allowed singlet-triplet transitions due to the non-zero values of the spin-orbital coupling (SOC) matrix elements. The energy difference (in terms of the $T_{0}$ and $T_{e}$ parameters) between the triplet and singlet states - the so called singlet-triplet splitting - turned out to be more precise than previous ab initio calculations ${ }^{54,59,72-83}$ and experimental results. ${ }^{84-93}$ Note that the interpretation of the high-resolution spectra was made possible using the ab initio fine structure parameters and SOC matrix elements calculated by Langhoff et al. ${ }^{94-96}$

The ground electronic state of methylene was also studied from photoelectron spectra of the methylene anions $\left(\mathrm{CH}_{2}{ }^{-}\right.$or $\left.\mathrm{CD}_{2}{ }^{-}\right)$by Sears and Bunker, ${ }^{97,98}$ by Leopold et al., ${ }^{99}$ and more recently by Coudert et al. ${ }^{100}$ The frequencies of the bending overtones up to $v_{2}=7$ were reported in Ref. 99 with an accuracy of $\sim 30-40 \mathrm{~cm}^{-1}$. In a series of papers by Petek et al. ${ }^{101-104}$ devoted to the study of rovibrational spectra for the singlet $\mathrm{CH}_{2}(\tilde{a})$ methylene, the manifestation of perturbation on the energy levels of both singlet and triplet methylene, caused by the singlet-triplet coupling, was analyzed.

### 2.2. Theoretical description

A new set of experimental data led to a series of works by Bunker and Jensen devoted to the construction of an accurate empirical PES for $\mathrm{CH}_{2}(\tilde{X})$. In early works, ${ }^{12}$ the parameters of the nonrigid bender Hamiltonian ${ }^{105-107}$ were fitted to experimental transitions. In this Hamiltonian, the kinetic energy part is based on the Hougen-Bunker-Johns (HBJ) formalism ${ }^{108}$ where the expansion parameters of the reciprocal $\mu_{\alpha \beta}$ tensor ( $\mu_{\alpha \beta}$ is the inverse of $I_{\alpha \beta}^{\prime}$ which is related to the inertia matrix $I_{\alpha \beta}$ in accordance with Eq. (4.8) of Ref. 109) depend explicitly on the bending angle while the effects of the centrifugal distortion and anharmonicity coming from the stretching vibrations are treated using Van Vleck perturbation theory. This model allowed to characterize observed rovibrational transitions of $\mathrm{CH}_{2}(\tilde{X}), \mathrm{CD}_{2}(\tilde{X})$, and ${ }^{13} \mathrm{CH}_{2}(\tilde{X})$ and to determine the empirical values of the equilibrium geometry as well as the height of the barrier to linearity from fitted PES parameters. In order to treat the singularity, perturbation theory was improved in Refs. 109-111 by including in the energy denominators of the perturbation series the dependence on the bending vibration quantum number $\left(v_{2}\right)$ and on the rotational quantum number $(K)$. This led to two versions of the nonrigid bender Hamiltonian (called "NRB1" and "NRB2") given in Ref. 14 with quite close results obtained by data fitting but with rather different energy levels for the stretching modes for which experimental information was missing.

A first variational calculation for $\mathrm{CH}_{2}(\tilde{X})$ was presented in Ref. 15 using the MORBID Hamiltonian developed by Jensen. ${ }^{16}$ Unlike the nonrigid bender Hamiltonian, all available experimental data were included into the fit, particularly those associated with the rotational transitions within the (100) stretching state. Almost at the same time, McLean et al. ${ }^{112}$ and Comeau et al. ${ }^{113}$ published one of the first pure ab initio PESs of $\mathrm{CH}_{2}(\tilde{X})$ that were then applied to calculate its rovibrational energy levels. According to Barclay et al., ${ }^{17}$ the error on the energy levels between MORBID and Discrete Variable Representation (DVR) calculations is of several wavenumbers. Kozin et al. ${ }^{114}$ implemented the spin-spin and spin-rotational terms in the MORBID computer code for fine structure calculations. The corresponding parameters were directly determined by data fitting.

High-level ab initio calculations of enthalpy of the $\mathrm{CH}_{2}(\tilde{X})$ formation were carried out by Császár et al. ${ }^{20}$ and Tajti et al. ${ }^{115}$ The corrections due to the high-order electronic correlations, DBOC and relativistic effects were included. As already mentioned above, the excited energy levels of $\mathrm{CH}_{2}(\tilde{X})$ were not studied in Ref. 20. Soon after, Furtenbacher et al. ${ }^{10}$ performed DVR calculations for $\mathrm{CH}_{2}(\tilde{X})$ with a deviation of $4 \mathrm{~cm}^{-1}$ for $v_{2}$ while accuracy of their rovibrational transitions varied from 0.2 to $4.5 \mathrm{~cm}^{-1}$. Again, only the fundamental bands were considered in Ref. 10. At the same time, a global ab initio PES was constructed by Medvedev et al. ${ }^{116}$ in order to describe the chemical reaction
between the $\mathrm{CH}_{2}(\tilde{X})$ and $\mathrm{CH}_{3}\left(\tilde{X}^{2} A_{2}^{\prime \prime}\right)$ radicals. The difference between the calculated vibrational levels in the $\mathrm{H}+\mathrm{CH}_{2}(\tilde{X})$ dissociation channel using this global PES and those predicted from the empirically refined PES by Jensen et al. ${ }^{15}$ was of a few tenths of wavenumbers.

### 2.3. Fine and hyperfine structures

Last but not least, an important aspect of the $\mathrm{CH}_{2}(\tilde{X})$ spectroscopy concerns its hyperfine structure. Since the two hydrogen atoms of $\mathrm{CH}_{2}(\tilde{X})$ have nonzero nuclear spin, $I\left(H_{1}\right)=I\left(H_{2}\right)=1 / 2$, para $\left(I_{\text {total }}=0\right)$ and ortho $\left(I_{\text {total }}=1\right)$ configurations are possible. In the case of ortho $-\mathrm{CH}_{2}(\tilde{X})$, each fine energy level further splits into three hyperfine sublevels. Lovas et al. ${ }^{6}$ resolved the hyperfine splitting in the pure rotational transitions without using an external magnetic field (zero-field splitting). The information on the hyperfine splitting made it possible the unambiguous identification of methylene in the interstellar medium near the Orion-KL and the W51 M sources by Hollis et al. ${ }^{4}$ New measurements with resolved fine and hyperfine structures by Ozeki et al., ${ }^{117,118}$ Michael et al., ${ }^{7}$ and Brünken et al. ${ }^{8,9}$ extended the spectral region to lower frequencies, which are more suitable for cold observations of dense interstellar clouds. Polehampton et al. ${ }^{5}$ used these new data to detect orthoand para- $\mathrm{CH}_{2}(\tilde{X})$ in the emission spectra toward the molecular cloud complexes Sagittarius B2 and W49 N. For the less studied deuterated isotopologue $\operatorname{CHD}\left(\tilde{X}^{3} A^{\prime \prime}\right)$, several pure rotational transitions were measured by Nolte et al. ${ }^{119}$ using the LMR spectroscopy and by Ozeki et al. ${ }^{120}$ in field-free conditions.

### 3.4. Anomalous centrifugal distortion effects

The experimental transitions of $\mathrm{CH}_{2}(\tilde{X})$ were generally fitted by using Watson $A$-reduced Hamiltonians and including spin-spin, spin-rotational, and electron-spin-nuclear-spin coupling terms. However, the centrifugal distortion terms are anomalously large, even for transitions with small values of the rotational quantum numbers. To improve the convergence of the Hamiltonian, the Euler expansion of the Hamiltonian was applied by Brünken et al., ${ }^{9}$ both for the rotational and spinrotational parts. Recently, Coudert ${ }^{11}$ performed a global fit of all transitions involved in the ground and first excited vibrational states. To this end, a modified version of the "Bending-Rotation" approach was used, assuming the dependence of the fine coupling terms on the large-amplitude bending coordinate (for further details, see Ref. 11 and references therein).

As it will be also shown, the ground vibrational state $(000)$ of $\mathrm{CH}_{2}(\tilde{X})$ is no longer isolated for $K_{a} \geq 5$ values. In that case, fitting $\mathrm{CH}_{2}(\tilde{X})$ transitions using standard effective Hamiltonians is a tricky task because of strong rovibrational resonance interactions. Performing a variational calculation is thus fully justified in the present work.

## 3. METHOD

## 3.1. $A b$ initio calculations

Since the electronic structure of $\mathrm{CH}_{2}(\tilde{X})$ is dominated by one electronic configuration, the single reference approach can be used to compute accurately the potential. For this reason, the spinunrestricted coupled cluster (CC) method [RHF-UCCSD(T)] implemented in MOLPRO by Knowles et al. ${ }^{121,122}$ was employed. The validity of the CC method was confirmed by the T1 and D1 diagnostic numbers extracted from the output files of MOLPRO. These numbers depend on the amplitudes of the excitations and are used to estimate the multi-reference character of the problem. For all the nuclear configurations of our grid, the maximum values of T1 and D1 were about two times less than their critical values of 0.02 and 0.05 , respectively, established by Lee et al. ${ }^{123}$ and Janssen et al. ${ }^{124}$

The ground electronic state of methylene has different multiplicity and symmetry (under the $C_{\mathrm{s}}$ point group) compared to the first excited singlet state. This allows to explicitly distinguish these two electronic states during ab initio calculations. In this work, we do not consider the singlet-triplet transitions caused by the nonzero SOC matrix elements and do not focus on fine and hyperfine structures of the rovibrational levels of $\mathrm{CH}_{2}(\tilde{X})$. Indeed, the spin-spin and spin-rotational parameters allowing an accurate description of the fine structure can be easily obtained by direct data fitting (see e.g. Ref. 114). Instead, the present study aims at computing an accurate PES for $\mathrm{CH}_{2}(\tilde{X})$ and performing nuclear-motion calculations.


FIGURE 1 Plot of the 1D cuts for the three lowest electronic states of $\mathrm{CH}_{2}$ calculated at the fixed bond length of $r_{1}\left(\mathrm{C}-\mathrm{H}_{1}\right)=r_{2}\left(\mathrm{C}-\mathrm{H}_{2}\right)=1.076$ Ang. The blue curve corresponds to the biggest SOC matrix element between the ground triplet and first excited singlet states

Contrary to the energy levels of the ground electronic triplet state, the excited levels of the singlet methylene $\mathrm{CH}_{2}(\tilde{a})$ are strongly perturbed by a Renner-Teller effect, caused by the degeneracy of $\tilde{a}$ and of the second excited singlet state $\left(\tilde{b}^{1} B_{1}\right.$, hereafter $\left.\tilde{b}\right)$ at the linear configuration. Since the $\tilde{b}$ state has a rather shallow minimum of potential (less than $1000 \mathrm{~cm}^{-1}$ ) with a quasilinear equilibrium bond angle of $\approx 143^{\circ}$ (versus $102^{\circ}$ for the $\tilde{a}$ state), the strong manifestation of the Renner-Teller effect occurs only for levels with $v_{2} \geq 4$ for $\mathrm{CH}_{2}(\tilde{a})$. For further details on the Renner-Teller effect, the reader can refer to the corresponding theoretical and experimental works by Duxbury et al., ${ }^{125}$ Alijah et al., ${ }^{126}$ Green et al., ${ }^{127}$ Xie et al., ${ }^{128}$ Hartland et al., ${ }^{129}$ and Gu et al. ${ }^{130}$

In order to show the structure of the three lowest electronic states of $\mathrm{CH}_{2}$, the 1 D cuts calculated at the icMRCI(Q)/aug-cc-pCVQZ level of the theory using the state-average $\operatorname{CASSCF}(8,5)$ are displayed in Figure 1. The blue curve in Figure 1 describes the angular dependence of the largest SOC matrix element responsible for the singlet-triplet transitions. As it can be seen, the maximum value of SOC is not big $\left(\approx 10 \mathrm{~cm}^{-1}\right)$, compared to the potential energies, and equals zero at the linear configuration. Accordingly, the effect of SOC is mainly determined by the diagonal elements of the spin-orbit matrix. For example, at $r_{1}\left(\mathrm{C}-\mathrm{H}_{1}\right)=r_{2}\left(\mathrm{C}-\mathrm{H}_{2}\right)=1.076$ Ang. the potential curves for the $\tilde{X}$ and $\tilde{a}$ states cross at the bond angle of $103-104^{\circ}$. The direct diagonalization of the spin-orbit matrix in MOLPRO led to the following shifts in the potential energies of the $\tilde{X}$ state: 1.16 and $-2.5 \mathrm{~cm}^{-1}$ at $103^{\circ}$ and $104^{\circ}$, respectively. Out of this region, the absolute shift was of 0.05 $\mathrm{cm}^{-1}$ in average, with a shift of $0.02 \mathrm{~cm}^{-1}$ at the bottom of the $\tilde{X}$ curve. Note that the description of the singlet-triplet transitions would require the evaluation of the spin-orbit matrix which is beyond the scope of the present work.

To analyze the convergence of the potential energies using the RHF-UCCSD(T) method, the Dunning's augmented correlation-consistent core-valence orbital basis sets ${ }^{131}$ were applied up to the sextuple- $\zeta$ quality [aug-cc-pCVXZ, hereafter denoted as ACVXZ , where $\mathrm{X}=\mathrm{T}, \mathrm{Q}, 5$, and 6] and all electrons were correlated (valence-valence, core-valence, and core-core contributions were included). Although the ab initio energies could be further extrapolated to the Complete Basis Set (CBS) limit using different extrapolation formula (see e.g. Ref. 132), our results prove that the ACV6Z basis set is enough to get a Root Mean Square (RMS) deviation better than $0.1 \mathrm{~cm}^{-1}$ for the band origins and better than $0.12 \mathrm{~cm}^{-1}$ for the rovibrational transitions.

In addition, we have tested the simplified explicitly correlated methods [RHF-UCCSD(T)F12x $\{\mathrm{x}=\mathrm{a}, \mathrm{b}\}]$ of Knizia et al. ${ }^{133}$ available in MOLPRO. The explicitly correlated methods can approach the CBS limit with less computational effort and is thus very relevant for polyatomic molecules with many electrons where the application of the "standard" CC method is difficult. Here, the benchmark calculation for the F12a,b methods was carried out using the correlation-consistent core-valence cc-pCVQZ-F12 basis set (hereafter, CVQZ-F12). For the specially optimized
correlation-consistent F12 basis sets, the appropriate defaults options are automatically chosen by MOLPRO (see Section 22.6 of the manual book ${ }^{23}$ ). Particularly, the suitable density fitting (DF) and resolution of the identity (RI) basis sets were automatically chosen as VQZ-F12/JKFIT and CVQZF12/OPTRI, respectively. The default Slater exponents were used for valence-valence, core-valence, and core-core pairs $\left(G E M \_B E T A=[1,1,1]\right)$. The contribution of core orbitals to the singles energy was not included $($ CORE_SINGLES $=0)$ whereas the associated complementary auxiliary basis set (CABS) singles correction to the reference energy was taken into account (CABS_SINGLES=1).

We have shown that accuracy of the potential energies could be significantly improved by including corrections due to (i) scalar relativistic effects, (ii) DBOC as well as (iii) contributions to the correlation energy from highly-excited Slater determinants.
(i) The relativistic correction was obtained by calculating the RHF-UCCSD(T)/ACVQZ-DK energies from the Douglas-Kroll-Hess (DKH) scalar relativistic Hamiltonian expanded up to the fourth order. Higher-order transformations provide contributions that are less than $10^{-6} \mathrm{~cm}^{-1}$. The energy difference, namely [RHF-UCCSD(T)+DKH/ACVQZ-DK] - [RHF$\operatorname{UCCSD}(\mathrm{T}) / \mathrm{ACVQZ}]$, was then computed and added to the PESs.
(ii) The DBOC was calculated at the UHF-CCSD/cc-pVTZ level of the theory using the CFOUR computer code. It turns out that the CCSD approach is enough for most molecules, including $\mathrm{CH}_{2}$, as it was already demonstrated by Gauss et al. ${ }^{134}$ By default, CFOUR operates with nuclear masses for the main isotopologue. According to our estimations, if the nuclear masses of $\mathrm{CD}_{2}$ are applied instead of those of ${ }^{12} \mathrm{CH}_{2}$ then the difference between the observed and calculated band center for $\nu_{2}$ changes from $0.07 \mathrm{~cm}^{-}$ ${ }^{1}$ to $-0.09 \mathrm{~cm}^{-1}$, which remains rather small in absolute value. It is also likely that the isotopic dependent DBOC allows a better agreement with observation for higher energies but it was not possible to check due to the absence of experimental data. Finally, we decided to construct one PES for all the isotopologues by including the DBOC contribution coming from ${ }^{12} \mathrm{CH}_{2}$.
(iii) One of the advantages of using the single reference CC method is that the accuracy of the correlation energy can be substantially improved by including contributions from highorder excitations. Since the number of the excited Slater determinants grows rapidly as the excitation order increases, the calculations become more and more time-demanding. However, in order to make calculations tractable, the high-order CC methods can be combined with less costly basis sets because they converge more rapidly with respect to the basis set size (see e.g. Ref. 135). For example, the total number of determinants for the biggest ACV6Z basis set is of 3084335 when the CC method restricted to single and double excitations [CCSD] is used. A similar number (2955831) is obtained after
including the contributions from the triple excitations [CCSDT], combined with the ccpVQZ basis set. For the quadruple CC calculations [CCSDTQ], this number becomes two times bigger (6046929), even when the cc-pVTZ basis set is considered. Finally, the following basis sets were used in the MRCC code when making scaling from $\operatorname{CCSD}(\mathrm{T})$ to $\operatorname{CCSDTQP:\quad [\operatorname {CSDT}-\operatorname {CSD}(T)]/cc-pVQZ,\quad [CCSDT(Q)-CCSDT]/cc-pVQZ,~}$ [CCSDTQ-CCSDT(Q)]/cc-pVTZ, and [CCSDTQP-CCSDTQ]/cc-pVDZ.
Finally, all the ab initio calculations presented here have been performed on the whole grid of the nuclear configurations (see next subsection) with a global energy threshold of $10^{-10}$ Hartree. The computed ab initio energies can be found in Supplementary materials.

### 3.2. Description of the PES

For energy calculations, the following three internal coordinates can be used in MOLPRO for $\mathrm{CH}_{2}(\tilde{X})$ : two bond lengths, $r_{1}\left(\mathrm{C}-\mathrm{H}_{1}\right)$ and $r_{2}\left(\mathrm{C}-\mathrm{H}_{2}\right)$, and one bond angle, $\alpha\left(\mathrm{H}_{1}-\mathrm{C}-\mathrm{H}_{2}\right)$. However, the grid of points in this coordinate system was generated using symmetry coordinates transforming as the $A_{1}$ and $B_{2}$ irreducible representations of the $C_{2 v}$ point group, allowing smoother variations for the three internal coordinates. The symmetry coordinates are given by

$$
\left\{\begin{array}{l}
\mathrm{s}_{1}^{A_{1}}=\left(\Delta r_{1}+\Delta r_{2}\right) / \sqrt{2}  \tag{1}\\
\mathrm{~s}_{2}^{A_{1}}=\Delta \alpha \\
\mathrm{s}_{3}^{B_{2}}=\left(\Delta r_{1}-\Delta r_{2}\right) / \sqrt{2}
\end{array}\right.
$$

where " $\Delta$ " means the deviation from the equilibrium values $r_{\mathrm{e}}$ and $\alpha_{\mathrm{e}}$, respectively.
Our grid of nuclear configurations consists of 1150 points with an energy range going from 0 to $15000 \mathrm{~cm}^{-1}$. A special attention was made for the evaluation of the barrier to linearity: the bond angle was gradually scanned from $76^{\circ}$ to $180^{\circ}$ by step of $2^{\circ}$ around its equilibrium value of $\approx 134^{\circ}$ and for each value of $\alpha, s_{1}^{A_{1}}$ and $s_{3}^{B_{2}}$ varied simultaneously. In particular, our grid consists of 84 different values of $r_{1}$ and $r_{2}$ at $\alpha=180^{\circ}$, which is enough to determine accurately the equilibrium bond length at the linear configuration (hereafter, $r_{\text {linear }}$ ) as well as the barrier to linearity (hereafter, $E_{\text {barrier }}$ ).

Following our recent studies, ${ }^{41,42}$ the PES of $\mathrm{CH}_{2}(\tilde{X})$ was expressed as a Taylor power series expansion in terms of the symmetry coordinates

$$
\begin{equation*}
V\left(S_{1}^{A_{1}}, S_{2}^{A_{1}}, S_{3}^{B_{2}}\right)=\sum_{i, j, k} C_{i j k} \cdot\left[S_{1}^{A_{1}}\right]^{i} \cdot\left[S_{2}^{A_{1}}\right]^{j} \cdot\left[S_{3}^{B_{2}}\right]^{2 k} \tag{2}
\end{equation*}
$$

where $C_{i j k}$ are expansion coefficient to be determined from the fit of the $a b$ initio potential energies. The even powers for $S_{3}^{B_{2}}$ ensure that the potential transforms as the totally symmetric irreducible representation $A_{1}$. Following the MORBID approach, ${ }^{16}$ Morse-cosine coordinates were preferred instead of (1) to define the symmetry coordinates in Eq. (2),
where $y_{1}, y_{2}$, and $a$ are asymptotic functions depending on the internal coordinates as follows:

$$
\left\{\begin{array}{l}
y_{1}=1-\exp \left[-b \cdot\left(r_{1}-r_{e}\right)\right]  \tag{4}\\
y_{2}=1-\exp \left[-b \cdot\left(r_{2}-r_{e}\right)\right] \\
a=\cos (\alpha)-\cos \left(\alpha_{e}\right)
\end{array}\right.
$$

where $b=1.5 \AA^{-1}$. It is also worth mentioning that the supplement angle $\bar{\rho}=180^{\circ}-\alpha$ is introduced in the HBJ formalism to describe the potential energy function, while an "effective" coordinate $\rho$ is introduced in the kinetic energy operator. It was shown that $\bar{\rho}$ and $\rho$ can be related a through coordinate transformation. ${ }^{109,136}$

TABLE 1 Impact of the basis set size on the structural parameters of the $\mathrm{CH}_{2}(\tilde{X})$ PES at the RHF-UCCSD(T) level of the theory

| Basis set | $r_{\mathrm{e}}$ (Ang.) | $\alpha_{\mathrm{e}}$ (Deg.) | $r_{\text {linear }}($ Ang.) | $E_{\text {barrier }}\left(\mathrm{cm}^{-1}\right)$ | $E_{0}($ Hartree $)$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| ACVTZ | 1.077818 | 133.74601 | 1.067323 | 1981.511 | -39.1301794830 |
| ACVQZ | 1.075978 | 133.84392 | 1.065630 | 1947.313 | -39.1423138808 |
| ACV5Z | 1.075417 | 133.86762 | 1.065073 | 1937.292 | -39.1457118077 |
| ACV6Z | 1.075300 | 133.88001 | 1.064952 | 1933.144 | -39.1469010781 |
| CVQZ-F12b | 1.075216 | 133.87635 | 1.064844 | 1935.930 | -39.1471016435 |
| CVQZ-F12a | 1.075190 | 133.92376 | 1.064870 | 1919.419 | -39.1487912991 |

## Notes:

All these values were determined from a fit of the ab initio PES.
$E_{0}$ is the total energy (Hartree-Fock + correlation) taken at the bottom of the PES, in Hartree.

TABLE 2 Contributions of the corrections to the equilibrium geometry and to the height of the barrier to linearity of $\mathrm{CH}_{2}(\tilde{X})$

| Ab initio level | $r_{\mathrm{e}}$ (Ang.) | $\alpha_{\mathrm{e}}($ (Deg. $)$ | $r_{\text {linear }}($ Ang.) | $E_{\text {barrier }}\left(\mathrm{cm}^{-1}\right)$ |
| :---: | :---: | :---: | :---: | :---: |
| RHF-UCCSD(T)/ACV6Z | 1.075300 | 133.88001 | 1.064952 | 1933.144 |
| +Relativistic/ACVQZ-DK | 1.075187 | 133.88167 | 1.064805 | $1934.300(+1.156)$ |
| +DBOC/VTZ | 1.075332 | 133.90224 | 1.064933 | $1933.431(-0.869)$ |
| +CCSDT/VQZ | 1.075582 | 133.91237 | 1.065180 | $1928.319(-5.112)$ |
| +CCSDT(Q)/VQZ | 1.075649 | 133.92160 | 1.065262 | $1924.969(-3.350)$ |
| +CCSDTQ/VTZ | 1.075656 | 133.92230 | 1.065270 | $1924.690(-0.279)$ |
| +CCSDTQP/VDZ | 1.075660 | 133.92256 | 1.065274 | $1924.592(-0.098)$ |

## Notes:

The $a b$ initio value of the barrier lies between the two last Bunker and Jensen's results: $1931 \pm 30 \mathrm{~cm}^{-1}$ and $1916 \mathrm{~cm}^{-1}$ taken from Refs. 14 and 15.
The final values of the equilibrium coordinates are consistent with those determined from the empirically refined PESs, see Ref. 14: $r_{\mathrm{e}}=1.0766 \pm 0.0014$ Ang., $\alpha_{\mathrm{e}}=134.037^{\circ} \pm 0.045^{\circ}$, and $r_{\text {linear }}=1.065$ Ang.

The $a b$ initio energies of our grid were fitted with a RMS deviation of $0.226 \mathrm{~cm}^{-1}$ using 66 expansion coefficients in Eq. (2). To this end, potential terms up to the $8^{\text {th }}$ order were considered.

The values of the internal coordinates for each stationary point (global minimum and saddle point at the linear configuration) were found numerically through the first derivatives. As it can be seen from Tables $\mathbf{1}$ and 2, and according to the height of the barrier to linearity, the shape of the PES is very sensitive to both the basis set size and correction beyond the $\operatorname{RHF}-\operatorname{UCCSD}(\mathrm{T})$ level of the theory. Such a dependence was already noticed during calculation of the height of the inversion barrier in $\mathrm{NH}_{3}$ (see, Ref. 41 for further details).

Increasing the basis set size from ACVTZ to ACV6Z allows to gradually converge the correlation energy from the RHF-UCCSD(T) method. This becomes quite clear after a brief inspection of the height of the barrier: in the ACV6Z PES, the height is of $1933 \mathrm{~cm}^{-1}$, which is about $4 \mathrm{~cm}^{-1}$ lower than that in the ACV5Z PES, while it is lowered by $10 \mathrm{~cm}^{-1}$ from ACVQZ to ACV5Z and by $34 \mathrm{~cm}^{-1}$ from ACVTZ to ACVQZ.

Concerning the explicitly correlated F12b method, the values of the structural parameters of the PES fall between those obtained from ACV5Z and ACV6Z. Hence, we can assume a similar behavior for the band origins. The F12a method, on the contrary, provides overestimated values of the correlation energy (see column " $E_{0}$ " in Table 1). However, such a behavior is relevant for this method when it is combined with the CVQZ-F12 basis set (see e.g. the recent $\mathrm{CH}_{3}$ work ${ }^{42}$ ). Finally, the F12a,b methods are relatively fast and quite consistent to check the convergence with respect to the size of the basis set. Indeed, computing one point requires about one hundred times less computational time than RHF-UCCSD(T)/ACV6Z.

Among all the corrections we proposed, the largest contributions in Table 2 are sorted in this order: high-order electronic correlations, the relativistic corrections and DBOC. Finally, the height of the barrier was shifted by $-8.552 \mathrm{~cm}^{-1}$ after adding simultaneously all these corrections. For the first time, a good agreement was reached between the ab initio height of the barrier $\left(1924.6 \mathrm{~cm}^{-1}\right)$ and the empirical estimation ( $1931 \pm 30 \mathrm{~cm}^{-1}$ ) obtained almost 40 years ago. ${ }^{14}$

### 3.3. Variational calculations

The nuclear motion calculations presented in this work were performed using our homemade variational computer code TENSOR based on the construction of irreducible tensor operators for the rotational, small amplitude and large amplitude vibrational parts. Following our previous works on nonrigid molecules, ${ }^{41,42}$ a hybrid nonrigid Hamiltonian model based on the HBJ formalism was employed. This model is described in detail in Ref. 21. The kinetic energy and potential parts were both Taylor expanded in terms of two normal mode coordinates describing the "small" vibrations (associated here with $v_{1}$ and $v_{3}$ ) for each point of a numerical grid for the nonrigid coordinate $\rho$ describing the large amplitude motion (associated here with $v_{2}$ ). The procedure ${ }^{136}$ linking the effective coordinate $\rho$ involved in the kinetic energy operator to the geometrically defined "real" coordinate
$\bar{\rho}$ involved in the PES was implemented in TENSOR. In order to approximately include the nonadiabatic effects, the atomic masses were employed in the kinetic energy operator (see the work by Kutzelnigg ${ }^{137}$ for more details).

In the variational calculation, 1729 and 1547 basis functions of symmetry $A_{1}$ and $B_{2}$, respectively, was needed to achieve a convergence better than $10^{-5} \mathrm{~cm}^{-1}$ up to $10000 \mathrm{~cm}^{-1}$. In order to check consistency of our calculation, the DVR3D computer code by Tennyson et al. ${ }^{138}$ was also used. The difference between the energy levels obtained from the two computer codes is around $10^{-5} \mathrm{~cm}^{-1}$ up to $10000 \mathrm{~cm}^{-1}$ (see Ref. 21), but contrary to DVR3D, we are able to provide quite easily the vibrational and rotational quantum numbers.

## 4. RESULTS

### 4.1. Vibrational levels

Before comparing our results with experimental data, let us first analyze how our predicted band origins depend on the size of the orbital basis set. According to Table 3, the differences between the ACV5Z and ACV6Z PESs for the three fundamental bands of $\mathrm{CH}_{2}(\tilde{X})$, namely $v_{2}\left(A_{1}\right), v_{1}\left(A_{1}\right)$, and $v_{3}\left(B_{2}\right)$, are of $-0.632 \mathrm{~cm}^{-1},+0.452 \mathrm{~cm}^{-1}$, and $+0.734 \mathrm{~cm}^{-1}$, respectively. We can conclude that the error for the three fundamental band centers is better than $1 \mathrm{~cm}^{-1}$ when using the ACV6Z PES. The reader can also see the correspondence between the height of the barrier (see Tables 1 and $\mathbf{2}$ ) and the value of the band origin for $v_{2}\left(A_{1}\right)$ : this latter decreases as the height of the barrier decreases. As expected, the band origin of $v_{2}\left(A_{1}\right)$ obtained from the F12b PES lie between those provided by ACV5Z and ACV6Z calculations while it is underestimated when using the F12a PES.

| TABLE 3 Impact of the orbital basis set size on the band origins of $\mathrm{CH}_{2}(\tilde{X})$ |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Band | ACVTZ | ACVQZ | ACV5Z | ACV6Z | F12b | F12a |  |
| $v_{2}\left(A_{1}\right)$ | 968.577 | 965.081 | 963.790 | 963.158 | 963.616 | 960.791 |  |
| $2 v_{2}\left(A_{1}\right)$ | 1837.448 | 1831.871 | 1830.016 | 1829.128 | 1830.125 | 1825.599 |  |
| $3 v_{2}\left(A_{1}\right)$ | 2820.716 | 2820.048 | 2819.718 | 2819.336 | 2820.392 | 2817.168 |  |
| $v_{1}\left(A_{1}\right)$ | 3021.146 | 3028.522 | 3031.185 | 3031.637 | 3032.565 | 3032.647 |  |
| $v_{3}\left(B_{2}\right)$ | 3237.492 | 3245.849 | 3249.144 | 3249.878 | 3250.932 | 3251.319 |  |
| $v_{1}+v_{2}\left(A_{1}\right)$ | 3969.381 | 3973.766 | 3975.172 | 3975.161 | 3976.461 | 3974.155 |  |
| $4 v_{2}\left(A_{1}\right)^{*}$ | 4017.326 | 4021.664 | 4022.821 | 4022.792 | 4024.017 | 4021.555 |  |
| $v_{2}+v_{3}\left(B_{2}\right)$ | 4219.593 | 4225.209 | 4227.437 | 4227.644 | 4229.166 | 4226.911 |  |
| $v_{1}+2 v_{2}\left(A_{1}\right)$ | 4860.846 | 4862.977 | 4863.695 | 4863.294 | 4865.334 | 4860.714 |  |
| $2 v_{2}+v_{3}\left(B_{2}\right)$ | 5089.476 | 5092.337 | 5093.787 | 5093.663 | 5095.830 | 5091.347 |  |

Notes:
All ab initio PESs included the corrections beyond the RHF-UCCSD(T) level of the theory that were demonstrated in Table 2.
*for this band the contributions of the (040) and (110) basis wavefunctions were of $30 \%$ and $34 \%$, respectively.

The F12a,b methods yield the results near the basis set limit accuracy already with the CVQZF12 basis set. Concerning the stretching $v_{1}\left(A_{1}\right)$ and $v_{3}\left(B_{2}\right)$ bands, the difference between ACV6Z and

F12a,b is below $1.5 \mathrm{~cm}^{-1}$, which is rather good for such methods requiring low computational effort. Note that these methods involve approximations and do not yield the exact CCSD-F12 energies. Some improvement of the results can be obtained by scaling the triples energy contribution (see Section 22.10 in Ref. 23), which was not considered in this work where the default options were applied.

According to Table 4, the variational calculation using the ACV6Z PES allowed to predict accurately the band origin for the $v_{2}$ band of the three triplet isotopologues $\mathrm{CH}_{2}, \mathrm{CD}_{2}$, and ${ }^{13} \mathrm{CH}_{2}$ with the following deviations (in $\mathrm{cm}^{-1}$ ) to observation: $-0.0585,0.0738$, and -0.0636 , respectively. Our results are thus 20 times more accurate than best available ones by Császár et al. ${ }^{20}$ Finally, among all the PESs we have tested in this work, the ACV6Z PES gives the most accurate results (see Table 3). Concerning the asymmetric CHD molecule, for which no observed data are available, the accuracy of our predictions is presumably the same as for the other species.

TABLE 4 Fundamental band origins of four isotopologues ("Iso") of triplet methylene computed from the different ab initio PESs

| Iso | Band | I | II | III | IV | V | This work | Exp. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $v_{1}\left(A_{1}\right)$ | 2985.0 | 3013.0 | 3035.6 | 3067.5 | 3036.0 | 3031.637 | - |
|  | $v_{2}\left(A_{1}\right)$ | 967.0 | 969.0 | 964.6 | 942.8 | 967.0 | 963.158 | $963.0995(2)^{a}$ |
|  | $v_{3}\left(B_{2}\right)$ | 3205.0 | 3235.0 | 3248.9 | 3200.2 | 3252.0 | 3249.878 | - |
| CHD | $\mathrm{v}_{1}\left(A^{\prime}\right)$ | - | 3134.0 | - | - | - | 3146.318 | - |
|  | $v_{2}\left(A^{\prime}\right)$ | - | 876.0 | - | - | - | 867.282 | - |
|  | $v_{3}\left(A^{\prime}\right)$ | - | 2297.0 | - | - | - | 2308.324 | - |
| $\mathrm{CD}_{2}$ | $v_{1}\left(A_{1}\right)$ | 2157.0 | 2177.0 | - | - | - | 2190.588 | - |
|  | $v_{2}\left(A_{1}\right)$ | 754.0 | 755.0 | - | - | - | 752.301 | $752.3748(16)^{b}$ |
|  | $v_{3}\left(B_{2}\right)$ | 2428.0 | 2449.0 | - | - | - | 2460.944 | - |
| ${ }^{13} \mathrm{CH}_{2}$ | $v_{1}\left(A_{1}\right)$ | - | $v_{2}\left(A_{1}\right)$ | - | 9010.0 | - | - | - |

Notes:
I - McLean et al.; ${ }^{112} \mathrm{II}$ - Comeau et al.; ${ }^{; 13} \mathrm{III}$ - Csázár et al.; ${ }^{20} \mathrm{IV}$ - Medvedev et al.; ${ }^{116} \mathrm{~V}$ - Furtenbacher et al. ${ }^{10}$ Sources of experimental data: $a$ - Marshall et al.; ${ }^{67} b$ and $c$-McKellar et al. ${ }^{66,71}$
This work result is based on the RHF-UCCSD(T)/ACV6Z PES

The low accuracy of the previous works could be explained by as follows.
First, from the quality of the ab initio calculations during the construction of the PES. In the previous works by McLean et al. and Comeau et al. (columns I and II in Table 4), the $\operatorname{SOCI}(\mathrm{Q})$ and $\operatorname{MRCI}(\mathrm{Q})$ approaches were combined with relatively small orbital basis sets comparable to the current VTZ, VQZ or V5Z basis sets with no both augmenting polarization and core correlation functions. Accordingly, their values for the two stretching modes were underestimated because the correlation energy was poorly described. Similarly, their value for the bending mode was overestimated because of a negative contribution of the correlation energy to the height of the barrier (see Tables $\mathbf{1}$ and $\mathbf{2}$ ).

Second, from the completeness of the PES. Although the ab initio calculations by Császár et al. ${ }^{20}$ and Furtenbacher et al. ${ }^{10}$ (see columns III and V in Table 4) were done using the restricted open-
shell CC method $[\mathrm{R}(\mathrm{O}) \operatorname{CCSD}(\mathrm{T})]$ and the ACVQZ basis set, the anharmonic contribution was included only from the quartic force field representation of the PES. Obviously, this is not enough for quasilinear nonrigid molecules like $\mathrm{CH}_{2}(\tilde{X})$. Let us note that performing calculations at the $\operatorname{CCSD}(\mathrm{T}) / \mathrm{ACVQZ}$ level of the theory remains a good choice for many molecules when corrections are not considered. In that case, the accuracy is between 5 and $10 \mathrm{~cm}^{-1}$ for the fundamental bands. However, to get an agreement within $1 \mathrm{~cm}^{-1}$ it is necessary to converge calculations with respect to the basis set and to include corrections.

The calculations of Medvedev et al. ${ }^{116}$ (column IV in Table 4) stand apart because they were carried out for $\mathrm{CH}_{2}(\tilde{X})$ in the dissociation channel of $\mathrm{CH}_{3}: \mathrm{CH}_{2}(\tilde{X})+\mathrm{H}$. The band origin obtained from icMRCI $(\mathrm{Q}) / \mathrm{AVTZ}$ was predicted at $942 \mathrm{~cm}^{-1}$ for the bending mode and is in contradiction with the height of the barrier to linearity of $2039 \mathrm{~cm}^{-1}$ given in Table 3 of Ref. 116. One explanation could be the quality of the fit.

Another explanation concerning the difference between the results of Császár et al. ${ }^{20}$ and Furtenbacher et al. ${ }^{10}$ could be the use of different nuclear-motion computer codes because their $a b$ initio calculations were made using the same level of the theory. In the first paper in 2003, the socalled nonrigid-rotation large-amplitude-internal-motion Hamiltonian (NRLH) of Szalay et al. ${ }^{139-141}$ was used, while the DVR3D code was used in the second paper in 2006. A different construction of the kinetic energy operator (NRLH versus DVR3D) may thus deviate the energy levels by several wavenumbers. In that case, it is difficult to estimate the actual accuracy of the ab initio PES.

TABLE 5 Vibrational energy levels (in $\mathrm{cm}^{-1}$ ) for the four isotopologues of triplet methylene computed variationally from the $a b$ initio RHF-UCCSD(T)/ACV6Z PES

| $\mathrm{CH}_{2}$ | CHD | $\mathrm{CD}_{2}$ | ${ }^{13} \mathrm{CH}_{2}$ |
| :---: | :---: | :---: | :---: |
| 963.158 (0 1 0) | 867.282 (0 1 0) | 752.301 (0 1 0) | 959.231 (0 1 0) |
| 1829.128 (020) | 1628.341 (0 2 0) | 1414.854 (020) | 1820.333 (020) |
| 2819.336 (030) | 2308.324 (0 0 1) | 2048.251 (0 30 ) | 2803.044 (0 30 ) |
| 3031.637 (100) | 2441.980 (0 30 ) | 2190.588 (100) | 3029.183 (100) |
| 3249.878 (001) | 3146.318 (100) | 2460.944 (0 0 1) | 3234.765 (0 0 1) |
| 3975.161 (110) | 3191.922 (0 1 1) | 2792.553 (0 4 0) | 3958.688 (0 40 ) |
| 4022.792 (0 40 )* | 3416.499 (0 4 0) | 2945.893 (110) | 4008.766 (1 1 0) |
| 4227.644 (0 1 1) | 3941.422 (0 21 ) | 3221.260 (0 1 1) | 4208.197 (0 1 1) |
| 4863.294 (120) | 4040.141 (110) | 3598.240 (120) | 4852.394 (120) |
| 5093.663 (0 2 1) | 4505.821 (0 50) | 3663.395 (050) | 5069.745 (0 21 ) |
| 5306.542 (0 50) | 4557.371 (0 0 2) | 3898.977 (0 21 1) | 5274.580 (0 50) |
| 5838.471 (130) | 4741.115 (0 3 1) | 4238.582 (130) | 5819.444 (130) |
| 5997.505 (200) | 4817.642 (120) | 4349.813 (200) | $5992.152(200)$ |
| 6039.756 (0 3 1) | 5414.847 (0 1 2) | 4522.088 (0 3 1) | 6009.535 (0 3 1) |
| 6168.038 (101) | 5472.738 (101) | 4588.939 (060) | 6150.681 (101) |
| 6434.046 (002) | 5577.665 (130) | 4593.235 (101) | 6405.276 (002) |
| 6696.045 (0 6 0) | 5644.320 (060) | 4879.700 (0 0 2) | 6657.723 (060) |
| 6950.998 (2 1 0) | 5771.169 (060) | 4980.427 (140) | 6937.435 (2 1 0) |
| 7033.175 (140) | 6188.779 (200) | 5109.304 (2 1 0) | 7008.601 (140) |
| 7132.469 (1 1 1) | 6204.277 (0 2 2) | 5233.579 (0 4 1) | 7106.279 (1 1 1) |


| 7189.571 (0 4 1) | 6346.426 (1 1 1) | 5356.258 (111) | 7156.830 (0 4 1) |
| :---: | :---: | :---: | :---: |
| $7423.094(012)$ | 6520.229 (140) | 5575.236 (070) | 7389.680 (0 1 2) |
| 7832.176 (2 20 ) | 6743.853 (0 0 3) | 5645.782 (0 1 2) | 7818.846 (2 20 ) |
| 8014.679 (12 1) | 6752.570 (0 5 1) | 5758.838 (2 20 ) | 7989.050 (12 1) |
| 8126.487 (0 70 ) | 6966.698 (070) | 5862.870 (150) | 8085.000 (070) |
| 8295.211 (0 2 2) | 7005.925 (0 3 2) | 6018.614 (12 1) | 8257.369 (0 2 2) |
| 8351.228 (150) | 7065.676 (2 1 0) | 6079.688 (0 5 1) | 8309.765 (150) |
| 8439.110 (051) | 7143.316 (12 1) | 6337.194 (0 2 2) | 8394.664 (05 1) |
| 8790.667 (2 30 ) | 7578.653 (150) | 6397.123 (2 3 0) | 8769.080 (2 30 ) |
| 8893.598 (300) | 7613.463 (013) | 6477.820 (300) | 8883.925 (300) |
| 8943.319 (1 3 1) | 7704.967 (102) | 6596.252 (0 80 ) | 8911.894 (13 1) |
| 9013.339 (301) | 7819.808 (2 20 ) | 6649.657 (1 3 1) | 8992.172 (301) |
| 9201.718 (0 3 2) | 7858.110 (061) | 6693.657 (201) | 9159.372 (0 3 2) |
| 9261.157 (102) | 7911.864 (13 1) | 6803.520 (160) | 9230.898 (102) |
| 9549.291 (160) | 8035.611 (0 4 1) | 6946.683 (102) | 9506.370 (160) |
| 9549.423 (003) | 8282.274 (0 8 0) | 6969.295 (0 3 2) | 9507.941 (0 0 3) |
| 9784.662 (06 1) | 8397.529 (02 2) | 6974.499 (06 1) | 9734.861 (06 1) |
| 9803.446 (0 80 ) | 8468.812 (201) | 7135.630 (2 40 ) | 9756.717 (0 80 ) |
| 9860.452 (4 1 0) | 8594.750 (1 1 2) | 7243.191 (3 10) | 9842.370 (410) |
| 9979.943 (3 1 1) | 8610.132 (2 30 ) | 7256.595 (003) | 9954.239 (3 1 1) |
| 9991.433 (140) | 8700.453 (160) | 7360.137 (14 1) | 9956.232 (140) |
| Notes: <br> Vibrational assignment is given in brackets for each isotopologue as $\left(v_{1} v_{2} v_{3}\right)$. The total lists of the theoretical rovibrational energy levels can be found in Supplementary Materials. <br> ZPE (in $\mathrm{cm}^{-1}$ ): $3731.613\left(\mathrm{CH}_{2}\right), 3250.525(\mathrm{CHD}), 2758.755\left(\mathrm{CD}_{2}\right)$, and $3720.242\left({ }^{13} \mathrm{CH}_{2}\right)$. <br> *for this band the contributions of the $(040)$ and $(110)$ basis wavefunctions were of $30 \%$ and $34 \%$, respectively. |  |  |  |
|  |  |  |  |

Finally, our full-dimensional ab initio RHF-UCCSD(T)/ACV6Z PES was used to predict the band origins for the four isotopologues of triplet methylene below $10000 \mathrm{~cm}^{-1}$ (see Table 5). As expected, the vibrational levels are quite close for $\mathrm{CH}_{2}$ and ${ }^{13} \mathrm{CH}_{2}$. It is worth mentioning that the gap between the overtones of the bending mode gradually increases. This is quite unusual for "standard" triatomic molecules where the barrier to linearity is generally one order of the magnitude larger than the frequency of the bending mode. For example, for $\mathrm{H}_{2} \mathrm{O}$ the gap between the (010) and (020) levels is of $390 \mathrm{~cm}^{-1}$ and is bigger than the gap between (070) and (080) because of the anharmonicity. In $\mathrm{CH}_{2}$, on the contrary, the gap between (070) and (080) is about two times bigger than that between the first two vibrational levels.

### 4.2. The polyad structure and rovibrational resonances

For most of the polyatomic molecules, the energy levels are organized as small groups of strong interacting vibrational levels, called polyads. In that case, the idea of introducing so-called polyad effective Hamiltonians for the analysis of absorption spectra is well established. The effective approach still plays a key role in the assignment of high-resolution spectra of many molecules (see e.g. the HITRAN ${ }^{142}$ and GEISA ${ }^{143}$ databases). Recently, a novel methodology for obtaining the parameters of both the effective Hamiltonians and dipole moment operators has been proposed in Ref. 144 for semirigid molecules, but not yet for nonrigid molecules.


FIGURE 2 (Panels $a$ and $b$ ) Polyad structure for the vibrational levels $(N=0)$ of $\mathrm{CH}_{2}(\tilde{X})$ and $\mathrm{CD}_{2}(\tilde{X})$ assuming the polyad number, $P=3 v_{1}+v_{2}+3 v_{3}$. (Panels $c$ and $d$ ) Rovibrational energies as a function of $N$.
The curly brackets are used to show the energy differences between the levels with $K_{a}=0$ and $N=K_{a}$ at assuming the polyad number, $P=3 v_{1}+v_{2}+3 v_{3}$. (Panels $c$ and $d$ ) Rovibrational energies as a function of $N$.
The curly brackets are used to show the energy differences between the levels with $K_{a}=0$ and $N=K_{a}$ at $N=1$. We can see the various overlapping between the energy levels of the (000) and (010) states after $N=5$

Each polyad formed by groups of nearly degenerate vibrational states is characterized by the same number $P$ defined by multiple ratios between the normal mode frequencies. According to Table 5, a quite clear polyad scheme occurs for $\mathrm{CH}_{2}: v_{1} \approx 3 v_{2} \approx v_{3}$. As a result, the vibrational levels inside a given polyad can be labeled by the additional polyad number, $P=3 v_{1}+v_{2}+3 v_{3}$. Such a polyad structure is quite unusual for semirigid nonlinear triatomic molecules like $\mathrm{H}_{2} \mathrm{O}, \mathrm{H}_{2} \mathrm{~S}, \mathrm{SO}_{2}, \mathrm{NO}_{2}$, etc. for which we have $P=2 v_{1}+v_{2}+2 v_{3}$.

As displayed in Figures $\mathbf{2 a}$ and $\mathbf{2 b}$, the first three polyads of $\mathrm{CH}_{2}(\tilde{X})$ and $\mathrm{CD}_{2}(\tilde{X})$ consist of single vibrational states and the next three polyads consist of three vibrational states (also called triads). The seventh polyad ( $P=6$ ) corresponds to an hexad (six vibrational states) for both
isotopologues, while the polyad structure is becoming unclear above for $\mathrm{CH}_{2}(\tilde{X})$, because of various overlapping. For $\mathrm{CD}_{2}(\tilde{X})$ the inter-polyad couplings occur from the ninth polyad $(P=8)$.

Thus, it is possible to define quite easily a polyad scheme for $\mathrm{CH}_{2}(\tilde{X})$ up to $P=5$ and build the corresponding effective vibrational Hamiltonian. However, due to the large values of the rotational constants, the difficulty grows very rapidly as $N$ increases. The rotational constant $A$ also increases with the vibrational excitation of the bending mode. Its value is proportional to the energy difference between the levels with $K_{a}=0$ and $K_{a}=1$ at $N=1$. Concerning the heaviest molecule, $\mathrm{CD}_{2}(\tilde{X})$, the value of the rotational constant $A$ is about two times less than that of $\mathrm{CH}_{2}(\tilde{X})$, which is confirmed by the observations (see e.g. Tables II and III in Refs. 66 and 67).

As shown in Figures $\mathbf{2 c}$ and $\mathbf{2 d}$, a clear polyad structure is rapidly lost from $N>4$, making problematic the construction of effective Hamiltonians. It is worth stressing the fact that such an overlapping between rovibrational states exists for most of the molecules, but occurs generally for higher excited polyads.


FIGURE 3 Plot of the ratio of the first $\left(C_{1}\right)$ and second $\left(C_{2}\right)$ biggest expansion coefficients of the rovibrational wavefunction of the (000) (a) and (010) (b) states of triplet $\mathrm{CH}_{2}$ and $\mathrm{CD}_{2}$ in comparison with $\mathrm{H}_{2} \mathrm{O}$

In order to show the strong coupling between rotation-bending states, we have plotted in Figure 3 the ratio of the two largest mixing coefficients in the eigenfunction decomposition obtained from the variational calculation as the function of $K_{a}$. When the ratio is close to 1 , the coupling between the rotation-bending states is strong and makes assignments ambiguous. As we can see, the mixing grows rapidly as a function of $K_{a}$ for the (000) state of $\mathrm{CH}_{2}(\tilde{X})$. For example, at $N=K_{a}=5$ the magnitudes of the two largest coefficients are of -0.792 and -0.509 while at $N=K_{a}=7$ they are much closer, -0.661 and -0.590 , respectively. This behavior for the ( 010 ) state is more pronounced. As an illustration, we have also plotted in Figure 3 these ratios for the $\mathrm{H}_{2} \mathrm{O}$ molecule. Up to $K_{a}=7$, needless
to say that $\mathrm{H}_{2} \mathrm{O}$ exhibits a smooth behavior compared to $\mathrm{CH}_{2}$ and $\mathrm{CD}_{2}$ and could be considered as a "normal" semirigid molecule. Note that some energy levels with $N=K_{a}>4$ are missing in Figure $\mathbf{3 b}$ because of an unclear assignment in our variational calculation (see Supplementary materials for more details about the assignments).

### 4.3. Rovibrational transitions

In Table 6, we compare the calculated rovibrational transitions with available empirical data. Our ab initio PES allows to reach a RMS deviation of $0.112 \mathrm{~cm}^{-1}$, which is almost twenty times better than the previous $a b$ initio results by Furtenbacher et al. ${ }^{10}$ Our RMS error is comparable with the RMS deviation of $0.11 \mathrm{~cm}^{-1}$ obtained by Jensen et al. ${ }^{15}$ from an empirically refined PES. Moreover, our calculations cover a wider range of vibrational and rotational quantum numbers and include the main isotopologues.

TABLE 6 A $b$ initio rovibrational transitions of triplet methylene calculated in this work and by Furtenbacher et al. ${ }^{10}$ in comparison with the available empirical data (Emp.)
$\mathrm{CH}_{2}:(000) \leftarrow(\mathbf{0 0 0}) \quad \mathrm{CH}_{2}:(010) \leftarrow(000)$

| $N_{K_{a}^{*} K_{c}^{*}}^{\prime \prime} \leftarrow N_{K_{a}^{\prime} K_{c}^{\prime}}^{\prime}$ | Emp. | Emp.-Calc. |  | $N_{K_{a}^{d} K_{c}^{*}}^{\prime \prime} \leftarrow N_{K_{d}^{\prime} K_{c}^{\prime}}^{\prime}$ | Emp. | Emp.-Calc. |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Ref. 10 | This work |  |  | Ref. 10 | This work |
| $4_{04} \leftarrow 3_{13}$ | 2.3104 | -0.566 | -0.085 | $808 \leftarrow 7_{17}$ | 1032.4728 | - | -0.131 |
| $2_{12} \leftarrow 3_{03}$ | 14.7873 | 0.558 | 0.084 | $4_{14} \leftarrow 5_{05}$ | 1032.9928 | -1.735 | 0.223 |
| $5_{05} \leftarrow 4_{14}$ | 19.7808 | -0.574 | -0.085 | $9_{09} \leftarrow 818$ | 1050.1838 | - | -0.129 |
| $1_{11} \leftarrow 202$ | 31.4467 | 0.553 | 0.083 | $10_{010} \leftarrow 9_{19}$ | 1068.0581 | - | -0.122 |
| $1_{10} \leftarrow 1_{01}$ | 63.8826 | 0.549 | 0.083 | $2_{12} \leftarrow 3_{03}$ | 1068.2735 | -1.733 | 0.218 |
| $2_{11} \leftarrow 2_{02}$ | 65.1649 | 0.545 | 0.082 | $1_{11} \leftarrow 202$ | 1085.1470 | -1.731 | 0.218 |
| $3_{12} \leftarrow 3_{03}$ | 67.1149 | 0.538 | 0.080 | $2_{11} \leftarrow 202$ | 1118.8696 | -1.750 | 0.213 |
| $4_{13} \leftarrow 4_{04}$ | 69.7632 | 0.529 | 0.079 | $3_{12} \leftarrow 3_{03}$ | 1120.7315 | -1.766 | 0.215 |
| $6_{15} \leftarrow 66_{06}$ | 77.3110 | 0.501 | 0.073 | $5_{14} \leftarrow 5_{05}$ | 1126.5534 | -1.824 | 0.215 |
| $1_{11} \leftarrow 0_{00}$ | 78.3211 | 0.553 | 0.083 | $6_{15} \leftarrow 6_{06}$ | 1130.6264 | - | 0.217 |
| $2_{12} \leftarrow 1_{01}$ | 92.8254 | 0.558 | 0.083 | $1_{11} \leftarrow 0_{00}$ | 1132.0200 | -1.855 | 0.216 |
| $4_{22} \leftarrow 5_{15}$ | 97.5430 | 0.975 | 0.163 | $7_{16} \leftarrow 7_{07}$ | 1135.5658 | - | 0.215 |
| $3_{22 \leftarrow 4}{ }_{13}$ | 97.7342 | 1.037 | 0.174 | $8_{17} \leftarrow 808$ | 1141.4324 | - | 0.213 |
| $5_{23} \leftarrow 5_{14}$ | 158.5792 | 1.045 | 0.173 | $2_{12} \leftarrow 1_{01}$ | 1146.3117 | -1.730 | 0.218 |
| $5_{42} \leftarrow 4_{31}$ | 382.4364 | 1.094 | 0.236 | $3_{13} \leftarrow 202$ | 1159.9406 | -1.732 | 0.220 |
| $6_{43} \leftarrow 5_{32}$ | 398.5730 | 1.101 | 0.241 | $4_{14} \leftarrow 3_{03}$ | 1172.9343 | -1.731 | 0.222 |
| $6_{42} \leftarrow 5_{33}$ | 398.5674 | 1.089 | 0.229 | $\mathrm{CD}_{2}:(000) \leftarrow(\mathbf{0 0 0})$ |  |  |  |
| $744 \leftarrow 633$ | 414.7015 | 1.098 | 0.236 | $1_{11} \leftarrow 202$ | 17.1920 | 0.230 | 0.014 |
| $7_{43} \leftarrow 6{ }_{34}$ | 414.7175 | 1.096 | 0.235 | $1_{10} \leftarrow 1_{01}$ | 33.5554 | 0.238 | 0.017 |
| $\mathbf{C H}_{2}:(100) \leftarrow(100)$ |  |  |  | $2_{11} \leftarrow 202$ | 34.1038 | 0.238 | 0.018 |
| $1_{10 \leftarrow} \leftarrow 1_{01}$ | 59.9423 | - | 0.076 | $3_{12} \leftarrow 3_{03}$ | 34.9365 | 0.236 | 0.018 |
| $211 \leftarrow 202$ | 61.2095 | - | 0.075 | $1_{11} \leftarrow 0_{00}$ | 40.9556 | 0.246 | 0.019 |
| $3_{12} \leftarrow 3_{03}$ | 63.1389 | - | 0.073 | $2_{12} \leftarrow 1_{01}$ | 48.3728 | 0.252 | 0.021 |
| $2_{12} \leftarrow 1_{01}$ | 88.4156 | - | 0.076 | $3_{13 \leftarrow 202}$ | 55.5373 | 0.260 | 0.023 |
| $\mathbf{C H}_{2}$ : (010) $\leftarrow(000)$ |  |  |  | $4_{22} \leftarrow 5_{15}$ | 58.9443 | 0.527 | 0.036 |
| $4_{14} \leftarrow 5_{23}$ | 801.2638 | -3.301 | -0.029 | $3_{22} \leftarrow 4_{13}$ | 60.1329 | 0.553 | 0.037 |
| $606 \leftarrow 7_{17}$ | 802.5901 | - | -0.143 | $22_{21} \leftarrow 3_{12}$ | 69.0165 | 0.554 | 0.039 |
| $4_{13}$ ¢ $5_{24}$ | 814.4573 | -3.366 | -0.032 | $5_{15} \leftarrow 4_{04}$ | 69.1425 | 0.278 | 0.027 |
| $5_{05}^{6} \leftarrow 6_{16}$ | 815.3156 | - | -0.145 | $5_{23} \leftarrow 5_{14}$ | 90.9683 | 0.575 | 0.044 |


| $5_{24} \leftarrow 6_{33}$ | 815.4803 | -2.174 | 0.023 | $4_{22} \leftarrow 4_{13}$ | 92.0210 | 0.572 | 0.045 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $3_{13} \leftarrow 4_{22}$ | 820.2705 | -3.303 | -0.032 | $3_{21} \leftarrow 3_{12}$ | 92.9054 | 0.569 | 0.045 |
| $3_{12} \leftarrow 4_{23}$ | 828.1105 | -3.337 | -0.033 | $2_{20} \leftarrow 2_{11}$ | 93.5938 | 0.567 | 0.045 |
| $4_{04 \leftarrow 5} 5_{15}$ | 828.4198 | -4.410 | -0.146 | $3_{22} \leftarrow 3_{13}$ | 96.0123 | 0.560 | 0.046 |
| $4_{22} \leftarrow 5_{33}$ | 831.7019 | -2.167 | 0.017 | $4_{23} \leftarrow 4_{14}$ | 97.1510 | 0.558 | 0.046 |
| $4_{23} \leftarrow 5_{32}$ | 831.6805 | -2.162 | 0.017 | $3_{22} \leftarrow 2_{11}$ | 117.4458 | 0.583 | 0.051 |
| $3_{03} \leftarrow 4_{14}$ | 841.9163 | - | -0.143 | $\mathrm{CD}_{2}$ : (010) $\leftarrow(010)$ |  |  |  |
| $2_{11} \leftarrow 3_{22}$ | 842.2978 | -3.318 | -0.037 | $1_{11} \leftarrow 2_{02}$ | 45.8637 | 0.347 | 0.041 |
| $3_{22} \leftarrow 44_{31}$ | 847.8123 | -2.150 | 0.016 | $2_{11} \leftarrow 2_{02}$ | 62.7231 | 0.887 | 0.046 |
| $3_{21} \leftarrow 4_{32}$ | 847.8195 | -2.152 | 0.016 | $3_{12} \leftarrow 3_{03}$ | 63.6418 | 0.898 | 0.046 |
| $10_{010} \leftarrow 10_{19}$ | 855.5925 | - | -0.092 | $1_{11} \leftarrow 0_{00}$ | 69.4368 | 0.903 | 0.048 |
| $1_{11} \leftarrow 2_{20}$ | 855.7320 | -3.298 | -0.035 | $2_{12} \leftarrow 1_{01}$ | 76.8260 | 0.901 | 0.049 |
| $202 \leftarrow 3_{13}$ | 855.8054 | -4.365 | -0.143 | $3_{13} \leftarrow 2_{02}$ | 83.9800 | 0.907 | 0.051 |
| $1_{10} \leftarrow 2_{21}$ | 857.0167 | -3.304 | -0.036 | $4_{14} \leftarrow 3_{03}$ | 90.9033 | 0.929 | 0.053 |
| $9_{09} \leftarrow 9_{18}$ | 863.5033 | - | -0.104 | $2_{21} \leftarrow 3_{12}$ | 117.0708 | 1.372 | 0.058 |
| $22_{20} \leftarrow 3_{31}$ | 863.8719 | -2.140 | 0.016 | $3_{21} \leftarrow 3_{12}$ | 140.8969 | 1.385 | 0.065 |
| $7_{17} \leftarrow 7_{26}$ | 866.8280 | - | -0.016 | $2_{20} \leftarrow 2_{11}$ | 141.5486 | 1.483 | 0.064 |
| $1_{01} \leftarrow 2_{12}$ | 870.0952 | -4.348 | -0.141 | $2_{21} \leftarrow 2_{12}$ | 143.1560 | 1.371 | 0.065 |
| $8_{08} \leftarrow 8_{17}$ | 870.6924 | - | -0.111 | $2_{21} \leftarrow 1_{10}$ | 157.8676 | 1.383 | 0.068 |
| $6_{16} \leftarrow 6_{25}$ | 872.0183 | - | -0.021 | $22_{20} \leftarrow 1_{11}$ | 158.4080 | 1.384 | 0.069 |
| $5_{15} \leftarrow 5_{24}$ | 876.5200 | - | -0.026 | $\mathrm{CD}_{2}$ : (100) $\leftarrow(100)$ |  |  |  |
| $77_{07} \leftarrow 7_{16}$ | 877.1265 | - | -0.116 | $3_{22} \leftarrow 2_{11}$ | 117.1872 | - | 0.048 |
| $4_{14} \leftarrow 4_{23}$ | 880.3110 | -3.303 | -0.028 | $\mathrm{CD}_{2}$ : (010) $¢(000)$ |  |  |  |
| $6_{06} \leftarrow 6_{15}$ | 882.7805 | - | -0.124 | $3_{03} \leftarrow 3_{12}$ | 717.0699 | - | 0.062 |
| $3_{13} \leftarrow 3_{22}$ | 883.3664 | -3.303 | -0.032 | $4_{04 \leftarrow} \leftarrow 4_{13}$ | 715.7102 | - | 0.064 |
| $0_{00} \leftarrow 1_{11}$ | 884.7780 | - | -0.143 | $5_{05} \leftarrow 5_{14}$ | 713.9986 | - | 0.064 |
| $2{ }_{12} \leftarrow 2_{21}$ | 885.6730 | -3.299 | -0.035 | $6_{06}^{5} 6_{15}$ | 711.9297 | - | 0.063 |
| $5_{05} \leftarrow 5_{14}$ | 887.6467 | - | -0.128 | $7_{07} \leftarrow 7_{16}$ | 709.4992 | - | 0.063 |
| $2_{11} \leftarrow 2_{20}$ | 889.4580 | -3.314 | -0.036 | ${ }^{13} \mathrm{CH}_{2}: \mathbf{( 0 1 0 ) \leftarrow ( 0 0 0 )}$ |  |  |  |
| $3_{12} \leftarrow 3_{21}$ | 890.8735 | -3.333 | -0.036 | $3_{13} \leftarrow 3_{22}$ | 879.8830 | - | -0.050 |
| $4_{04} \leftarrow 4_{13}$ | 891.7090 | -4.344 | -0.133 | $0_{00} \leftarrow 1_{11}$ | 881.7530 | - | -0.166 |
| $4_{13 \leftarrow 4} 4_{22}$ | 892.6835 | - | -0.035 | $2{ }_{11} \leftarrow 2_{20}$ | 886.0130 | - | -0.053 |
| $5_{14} \leftarrow 5_{23}$ | 894.8265 | -3.388 | -0.034 | $3_{12} \leftarrow 3_{21}$ | 887.4390 | - | -0.054 |
| $202 \leftarrow 2111$ | 897.4080 | -4.340 | -0.139 | $4_{04} \leftarrow 4_{13}$ | 888.6370 | - | -0.156 |
| $1_{01} \leftarrow 1_{10}$ | 899.0370 | - | -0.142 | $3_{03} \leftarrow 3_{12}$ | 891.9050 | - | -0.160 |
| $5_{24} \leftarrow 5_{33}$ | 910.1390 | -2.168 | 0.021 | $202 \leftarrow 2_{11}$ | 894.3580 | - | -0.163 |
| $5_{23} \leftarrow 5_{32}$ | 910.1660 | -2.178 | 0.020 | $1_{01} \leftarrow 1_{10}$ | 895.9940 | - | -0.165 |
| $4_{23} \leftarrow 4_{32}$ | 910.6279 | -2.153 | 0.020 | $2{ }_{02} \leftarrow 1_{11}$ | 928.0780 | - | -0.165 |
| $4_{22} \leftarrow 4_{31}$ | 910.6406 | -2.159 | 0.018 | $3_{03 \leftarrow 2}{ }_{12}$ | 944.2500 | - | -0.164 |
| $3_{22} \leftarrow 3_{31}$ | 911.0095 | -2.144 | 0.016 | CHD: (000) $\leftarrow(000)$ |  |  |  |
| $3_{13} \leftarrow 2_{20}$ | 930.5250 | -3.300 | -0.033 | $3_{13} \leftarrow 4_{04}$ | $4.4189^{a}$ | - | 0.030 |
| $2_{02} \leftarrow 1_{11}$ | 931.1260 | -4.348 | -0.141 | $5_{05} \leftarrow 4_{14}$ | $7.3681^{a}$ | - | -0.041 |
| $3_{03} \leftarrow 2_{12}$ | 947.2910 | -4.376 | -0.141 | $2_{12} \leftarrow 3_{03}$ | $15.9644^{a}$ | - | 0.028 |
| $8_{17} \leftarrow 7_{26}$ | 1030.7219 | - | -0.030 | $6_{06 \leftarrow 515}$ | $19.3912^{b}$ | - | -0.031 |
|  |  |  |  | $707 \leftarrow 6_{16}$ | $31.5825^{\text {b }}$ | - | -0.030 |
| RMS |  |  |  |  |  | 2.091 | 0.112 |

## Notes:

All empirical data except for CHD were taken from Table I of Jensen et al. ${ }^{15}$ These rovibrational transitions were obtained by removing the fine and hyperfine structures (see e.g. Refs. 14 and 15 for further details).
The empirical rovibrational transitions for CHD were obtained in this work using the empirical parameters from the following papers: $a$ - Ref. 119 and $b$-Ref. 120 (see details in the text).
The total lists of the theoretical rovibrational energy levels are given in Supplementary Materials.

Note that the differences between the "observed" and calculated transitions in Table 6 are not regular. Concerning the heaviest $\mathrm{CD}_{2}(\tilde{X})$ molecule, the agreement is undoubtedly better. For example, deviations of 0.174 and $0.037 \mathrm{~cm}^{-1}$ may occur for the same pure rotational $3_{22} \leftarrow 4_{13}$ transition of $\mathrm{CH}_{2}(\tilde{X})$ and $\mathrm{CD}_{2}(\tilde{X})$, respectively, and cannot be explained by some resonances. According to Figure 3, the mixing coefficients are close for both isotopologues. Even for the (010)-(000) transitions of $\mathrm{CH}_{2}(\tilde{X})$, the deviations are far to be regular, going from -0.016 to $-0.141 \mathrm{~cm}^{-1}$ for $7_{17} \leftarrow 7_{26}$ and $1_{01} \leftarrow 2_{12}$, respectively. These two transitions were measured by Marshall \& McKellar ${ }^{67}$ and we can note from their Table I that the splitting due to the fine structure was several times bigger for $1_{01} \leftarrow 2_{12}$ than for $7_{17} \leftarrow 7_{26}$. For example, the separation between the $F_{2}(S=0)$ and $F_{3}(S=-1)$ components were of -0.1054 and $+0.5021 \mathrm{~cm}^{-1}$ for $7_{17}^{\leftarrow} 7_{26}$ and $1_{01} \leftarrow 2_{1}$, while the $F_{1}(S=+1)$ component was shifted from $F_{2}$ by -0.0919 and $+0.1233 \mathrm{~cm}^{-1}$, respectively.

Consequently, the irregularities between the "observed" and calculated transitions can be partly explained by the difference in the magnitude of the electron spin splitting, which was removed in the empirical rovibrational transitions presented in Table 6. Indeed, the empirical transitions were obtained by setting to zero the spin-spin and spin-rotation parameters in the effective Hamiltonian. This corroborates the fact that the spin-rotation parameters ( $\varepsilon_{a a}, \varepsilon_{b b}$, and $\varepsilon_{c c}$ ) of $\mathrm{CD}_{2}(\tilde{X})$ are about two times smaller than those of $\mathrm{CH}_{2}(\tilde{X})$ (see e.g. Table 2 in Ref. 114).

For this work, we have also calculated the empirical rovibrational transitions for the $\operatorname{CHD}(\tilde{X})$ isotopologue. To this end, the SPCAT program of Pickett ${ }^{145}$ was employed and the empirical parameters of the effective Hamiltonian determined by Nolte et al. ${ }^{119}$ and Ozeki et al. ${ }^{120}$ were used. The fine and hyperfine contributions were not included here. The obtained empirical line positions of the five rovibrational transitions are in very good agreement with our calculations (see the end of

## Table 6).

Before making predictions for $\operatorname{CHD}(\tilde{X})$, we first made trial tests using the empirical parameters of $\mathrm{CH}_{2}(\tilde{X})$ given in Table III of Ref. 63 in order to obtain the same results as those presented in the column "Observed" of Table IV of Ref. 63. Moreover, this corroborated the fact that the spin-spin and spin-rotation parameters were not considered to build the "Emp." column of Table 6.

### 4.4. Comparison with the generating-function approach

Although the effective Watson Hamiltonian was used to fit the observed line positions of triplet methylene (see Section 2), the anomalously large centrifugal distortion of $\mathrm{CH}_{2}$ makes the predictions rapidly unreliable. Indeed, the conventional power series expansion of the Hamiltonian rapidly fails. For that reason, Bunker, Jensen et al. preferred using the variational approach to compute rovibrational levels.

7 illustrative example, we have compared the ground vibrational state energy levels of $\mathrm{CH}_{2}(\tilde{X})$ obtained
Alternative methods were proposed to improve the convergence of the effective Hamiltonian, based for example on the Padé and Borel approximants (see e.g. Burenin et al., ${ }^{146}$ Polyansky et al.,, ${ }^{147}$ and Belov et al. ${ }^{148}$ ). In the SPFIT and SPCAT programs of Pickett, the Euler series was implemented for this purpose (more details are given in Ref. 149). Another approach based on the use of generating functions was proposed by Tyuterev and Starikov. ${ }^{150,151}$ In Ref. 152, the generating functions combined with the Borel-type summation were used to predict $\mathrm{CH}_{2}(\tilde{X})$ energy levels. As an from variational calculations (This work), from the generating function approach and from the 9 Watson Hamiltonian (see Table 7).

TABLE 7 Rotational energy levels of the ground vibrational state of $\mathrm{CH}_{2}(\tilde{X})$ calculated in this work and compared to those predicted by Starikov and Mikhailenko ${ }^{152}$ using different effective approaches

| $N_{K_{a} K_{c}}$ | This work |  | This work - Ref. 152 |  |
| :---: | :---: | :---: | :---: | :---: |
|  | Energy | $C_{2} / C_{1}$ | "standard" Watson | generating function |
| 101 | 15.63 | 0.157 | 0.00 | 0.00 |
| $1{ }_{11}$ | 78.24 | 0.180 | -0.10 | -0.10 |
| $1{ }_{10}$ | 79.43 | 0.180 | 1.09 | -0.09 |
| 202 | 46.87 | 0.158 | 0.01 | -0.01 |
| 212 | 108.38 | 0.180 | -0.08 | -0.10 |
| 211 | 111.96 | 0.178 | -0.06 | -0.07 |
| 221 | 276.02 | 0.276 | -0.11 | -0.26 |
| 220 | 276.04 | 0.276 | -0.11 | -0.25 |
| 303 | 93.67 | 0.159 | 0.03 | -0.01 |
| 313 | 153.56 | 0.179 | -0.04 | -0.10 |
| 312 | 160.71 | 0.176 | -0.02 | -0.06 |
| 322 | 323.20 | 0.273 | -0.10 | -0.23 |
| 321 | 323.28 | 0.274 | -0.11 | -0.24 |
| 331 | 566.64 | 0.391 | 18.96 | 1.40 |
| 330 | 566.64 | 0.391 | 18.95 | 1.40 |
| 404 | 155.95 | 0.161 | 0.05 | -0.01 |
| 414 | 213.75 | 0.178 | 0.01 | -0.09 |
| 413 | 225.64 | 0.173 | 0.01 | -0.05 |
| 423 | 386.05 | 0.272 | -0.09 | -0.19 |
| 422 | 386.29 | 0.272 | -0.12 | -0.23 |
| 432 | 629.83 | 0.390 | 18.87 | 1.45 |
| 431 | 629.84 | 0.390 | 18.88 | 1.46 |
| 441 | 932.73 | 0.516 | 18.67 | 1.13 |
| 440 | 932.73 | 0.516 | 18.67 | 1.13 |
| 505 | 233.61 | 0.164 | 0.06 | -0.01 |
| 515 | 288.91 | 0.176 | 0.08 | -0.07 |
| 514 | 306.69 | 0.168 | 0.04 | -0.06 |
| $5{ }_{24}$ | 464.52 | 0.269 | -0.09 | -0.16 |
| 523 | 465.09 | 0.269 | -0.16 | -0.23 |
| 533 | 708.77 | 0.388 | 18.75 | 1.51 |
| 532 | 708.78 | 0.388 | 18.75 | 1.51 |
| $5_{42}$ | 1012.04 | 0.514 | 18.63 | 1.21 |


| $5_{41}$ | 1012.04 | 0.514 | 18.63 | 1.21 |
| :---: | :---: | :---: | :---: | :---: |
| $5_{51}$ | 1361.62 | 0.643 | -239.08 | -5.61 |
| $5_{50}$ | 1361.62 | 0.643 | -239.08 | -5.61 |
| $6_{06}$ | 326.54 | 0.167 | 0.09 | 0.00 |
| $6_{16}$ | 379.00 | 0.174 | 0.20 | -0.01 |
| $6_{15}$ | 403.78 | 0.172 | 0.04 | -0.10 |
| $6_{25}$ | 558.57 | 0.266 | -0.09 | -0.11 |
| $6_{24}$ | 559.70 | 0.267 | -0.24 | -0.25 |
| $6_{34}$ | 803.42 | 0.384 | 18.56 | 1.56 |
| $6_{33}$ | 803.43 | 0.384 | 18.55 | 1.55 |
| $6_{43}$ | 1107.11 | 0.511 | 18.52 | 1.28 |
| $6_{42}$ | 1107.11 | 0.511 | 18.52 | 1.28 |
| $6_{52}$ | 1457.14 | 0.641 | -238.47 | -5.54 |
| $6_{51}$ | 1457.14 | 0.640 | -238.47 | -5.54 |
| $6_{61}$ | 1844.06 | 0.770 | -1390.88 | -22.83 |
| $6_{60}$ | 1844.06 | 0.770 | -1390.88 | -22.83 |

## Notes:

Here "standard" Watson means the effective $A$-reduced Watson Hamiltonian. For further details concerning the generating function approach see Ref. 152.
To demonstrate the mixing between the basis states we give the $C_{2} / C_{1}$ ratio, where $C_{1}$ and $C_{2}$ are the first and second biggest expansion coefficients of the rovibrational wavefunction of the ground vibrational state.

As expected, the "standard" Watson Hamiltonian dramatically failed to predict the pure rotational energy levels. Even at $N=K_{a}=3$, the disagreement is of $-19 \mathrm{~cm}^{-1}$ while at $N=K_{a}=6$ it is of $-1391 \mathrm{~cm}^{-1}$. The generating function approach gives more consistent results, with a smoother behavior, though the disagreements with our calculations remains quite large: $-23 \mathrm{~cm}^{-1}$ at $N=K_{a}=6$. We did not compare our results to those obtained using the Euler series approach of Pickett. However, according to Figure 3 of Ref. 9, we can conclude that the Euler series approach does not allow to drastically improve the quality of the predictions.

In Table 7 we can see that the mixing between eigenfunctions (estimated as $C_{2} / C_{1}$ ) strongly depends on $K_{a}$. The ratio is quite similar for different states with the same value of $K_{a}$. Similar results were obtained using the effective approaches with for example a deviation of $-5.54 \mathrm{~cm}^{-1}$ for the state $6_{51}$ and a deviation of $-5.61 \mathrm{~cm}^{-1}$ for the state $5_{51}$. As expected, there is clearly a correlation between the magnitude of the deviation and that of the mixing.

## 5. CONCLUSIONS

In this work, an accurate $a b$ initio PES for the quasilinear triplet ( $\tilde{X}^{3} B_{1}$ ) methylene was built using the RHF-UCCSD(T)/aug-cc-pCV6Z method and including the corrections due to the scalar relativistic effects, DBOC and high-order electronic correlations. For the first time, the ab initio height of the barrier to linearity ( $1924.6 \mathrm{~cm}^{-1}$ ) agrees well with that obtained from the empirical PES published almost 40 years ago $\left(1931 \pm 30 \mathrm{~cm}^{-1}\right)$. All the available experimental band origins for the three isotopologues $\mathrm{CH}_{2}, \mathrm{CD}_{2}$, and ${ }^{13} \mathrm{CH}_{2}$ were reproduced with an accuracy below $0.1 \mathrm{~cm}^{-1}$ while
the RMS deviation between calculated and empirical transitions, including those for CHD, was of $0.112 \mathrm{~cm}^{-1}$.

Moreover, our variational calculations for rovibrational energy levels up to $\sim 10000 \mathrm{~cm}^{-1}$ allowed to analyze the overall polyad structure, and the polyad number $P=3 v_{1}+v_{2}+3 v_{3}$ was proposed. However, the rovibrational resonance interactions between the polyads grow very rapidly as $K_{a}$ increases because of the extremely large rotational and centrifugal distortion constants of triplet methylene. We have noted a strong mixing from $K_{a}=5$, even for the ground vibrational state (000) making unusable traditional Watson-based effective Hamiltonians. We can thus conclude that the variational approach is probably better suited than the effective approach to compute the $\mathrm{CH}_{2}\left(\tilde{X}^{3} B_{1}\right)$ spectra.

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## SUPPLEMENTARY MATERIALS

We provide the complete list of the rovibrational energy levels of the four isotopologues of triplet methylene calculated variationally up to $\sim 10000 \mathrm{~cm}^{-1}$. The $\mathrm{C}++$ and Fortran codes for computing potential energies from the final $a b$ initio PES. The $a b$ initio points of the potential energies for 1150 nuclear configurations calculated in this work.

## REFERENCES

[1] M. A. Blitz, P. W. Seakins, Chem. Soc. Rev. 2012, 41, 6318. https://doi.org/10.1039/C2CS35204D
[2] B. Gans, S. Boyé-Péronne, M. Broquier, M. Delsaut, S. Douin, C. E. Fellows, P. Halvick, J.-C. Loison, R. R. Lucchese, D. Gauyacq, Phys. Chem. Chem. Phys. 2011, 13, 8140. https://doi.org/10.1039/C0CP02627A
[3] K. M. Douglas, M. A. Blitz, W. Feng, D. E. Heard, J. M.C. Plane, H. Rashid, P. W. Seakins, Icarus 2019, 321, 752. https://doi.org/10.1016/j.icarus.2018.12.027
[4] J. M. Hollis, P. R. Jewell, F. J Lovas, The Astrophysical Journal 1995, 438, 259. https://doi.org/10.1086/175070
[5] E. T. Polehampton, K. M. Menten, S. Brünken, G. Winnewisser, J.-P. Baluteau, A\&A 2005, 431, 203. https://doi.org/10.1051/0004-6361:20041598
[6] F. J. Lovas, R. D. Suenram, K. M. Evenson, Astrophys. J. Lett. 1983, 267, L131. https://doi.org/10.1086/184017
[7] E. A. Michael, F. Lewen, G. Winnewisser, H. Ozeki, H. Habara, E. Herbst, The Astrophysical Journal 2003, 596, 1356. https://doi.org/10.1086/378090
[8] S. Brünken, E. A. Michael, F. Lewen, Th. Giesen, H. Ozeki, G. Winnewisser, P. Jensen, E. Herbst, Can. J. Chem. 2004, 82, 676. https://doi.org/10.1139/v04-034
[9] S. Brünken, H. S. P. Müller, F. Lewen, T. F. Giesen, J. Chem. Phys. 2005, 123, 164315. https://doi.org/10.1063/1.2074467
[10] T. Furtenbacher, G. Czakó, B. T. Sutcliffe, A. G. Császár, V. Szalay, J. Mol. Structure. 2006, 780781, 283. https://doi.org/10.1016/j.molstruc.2005.06.052
[11] L. H. Coudert, J. Chem. Phys. 2020, 153, 144115 https://doi.org/10.1063/5.0026162
[12] P. Jensen, P. R. Bunker, A. R. Hoy, J. Chem. Phys. 1982, 77, 5370 https://doi.org/10.1063/1.443785
[13] P. R. Bunker, P. Jensen, J. Chem. Phys. 79, 1224 (1983) http://dx.doi.org/10.1063/1.445927
[14] P. R. Bunker, P. Jensen, W. P. Kraemer, R. Beardsworth, J. Chem. Phys. 85, 3724 (1986) https://doi.org/10.1063/1.450944
[15] P. Jensen, P. R. Bunker, J. Chem. Phys. 89, 1327 (1988) https://doi.org/10.1063/1.455184
[16] P. Jensen, J. Mol. Spectrosc. 128, 478 (1988) https://doi.org/10.1016/0022-2852(88)90164-6
[17] V. J. Barclay, I. P. Hamilton, P. Jensen, J. Chem. Phys. 99, 9709 (1993) https://doi.org/10.1063/1.465453
[18] Recent advances in coupled-cluster methods, edited by R. J. Bartlett, World Scientific Publishing Company 1997, 340. https://doi.org/10.1142/3455
[19] Recent Progress in Coupled Cluster Methods: Theory and Applications in Challenges and Advances in Computational Chemistry and Physics, edited by P. Cársky, J. Paldus, J. Pittner, Springer 2010, 657. https://doi.org/10.1007/978-90-481-2885-3
[20] A. G. Császár, M. L. Leininger, V. Szalay, J. Chem. Phys. 2003, 118, 10631. https://doi.org/10.1063/1.1573180
[21] M. Rey, D. Viglaska, O. Egorov, A. V. Nikitin, J. Chem. Phys. 2023 (under revision).
[22] H.-J. Werner, P. J. Knowles, F. R. Manby, J. A. Black, K. Doll, A. Heßelmann, D. Kats, A. Köhn, T. Korona, D. A. Kreplin, Q. Ma, T. F. Miller, A. Mitrushchenkov, K. A. Peterson, I. Polyak, G. Rauhut, M. Sibaev, J. Chem. Phys. 2020, 152, 144107. https://doi.org/10.1063/5.0005081
[23] H.-J. Werner, P. J. Knowles, G. Knizia, F. R. Manby, M. Schütz et al. MOLPRO, version 2019.2, a package of ab initio programs, https://www.molpro.net (accessed April 6, 2023).
[24] M. Kállay, P. R. Nagy, D. Mester, Z. Rolik, G. Samu, J. Csontos, J. Csóka, P. B. Szabó, L. GyeviNagy, B. Hégely, I. Ladjánszki, L. Szegedy, B. Ladóczki, K. Petrov, M. Farkas, P. D. Mezei, A. Ganyecz, J. Chem. Phys. 2020, 152, 074107. https://doi.org/10.1063/1.5142048
[25] M. Kállay, P. R. Nagy, D. Mester, Z. Rolik, G. Samu, J. Csontos, J. Csóka, P. B. Szabó, L. GyeviNagy, B. Hégely, I. Ladjánszki, L. Szegedy, B. Ladóczki, K. Petrov, M. Farkas, P. D. Mezei, Á. Ganyecz, MRCC, a quantum chemical program suite, https://www.mrcc.hu (accessed April 6, 2023).
[26] D. A. Matthews, L. Cheng, M. E. Harding, F. Lipparini, S. Stopkowicz, T. -C. Jagau, P. G. Szalay, J. Gauss, J. F. Stanton, J. Chem. Phys. 2020, 152, 214108. https://doi.org/10.1063/5.0004837
[27] J. F. Stanton, J. Gauss, L. Cheng, M. E. Harding, D. A. Matthews, P. G. Szalay, A. A. Auer, R. J. Bartlett, U. Benedikt, C. Berger et al. CFOUR, a quantum chemical program package, http://www.cfour.de (accessed April 6, 2023).
[28] M. Rey, A. V. Nikitin, V1. G. Tyuterev, Mol. Phys. 2010, 108, 2121 https://doi.org/10.1080/00268976.2010.506892
[29] M. Rey, A. V. Nikitin, Vl. G. Tyuterev, J. Chem. Phys. 2012, 136, 244106. https://doi.org/10.1063/1.4730030
[30] M. Rey, J. Chem. Phys. 2019, 151, 024101. https://doi.org/10.1063/1.5109482
[31] T. Delahaye, A. Nikitin, M. Rey, P. G. Szalay, Vl. G. Tyuterev, J. Chem. Phys. 2014, 141, 104301. https://doi.org/10.1063/1.4894419
[32] T. Delahaye, A. Nikitin, M. Rey, P. G. Szalay, V1. G. Tyuterev, Chem. Phys. Lett. 2015, 639, 275. https://doi.org/10.1016/j.cplett.2015.09.042
[33] M. Rey, A. Nikitin, A. Campargue, S. Kassi, D. Mondelain, V. Tyuterev, Phys. Chem. Chem. Phys. 2016, 18, 176. https://doi.org/10.1039/C5CP05265C
[34] M. Rey, A. V. Nikitin, Y. L. Babikov, V. G. Tyuterev, J. Mol. Spectrosc. 2016, 327, 138. https://doi.org/10.1016/j.jms.2016.04.006
[35] M. Rey, A.V. Nikitin, V.G. Tyuterev, The Astrophysical Journal 2017, 847, 105. https://doi.org/10.3847/1538-4357/aa8909
[36] M. Rey, I. S. Chizhmakova, A. V. Nikitin, V. G. Tyuterev, Phys. Chem. Chem. Phys. 2018, 20, 21008. https://doi.org/10.1039/C8CP03252A
[37] D. Viglaska, M. Rey, A.V. Nikitin, V. G. Tyuterev, J. Chem. Phys. 2018, 149, 174305. https://doi.org/10.1063/1.5045525
[38] O. Egorov, A. Nikitin, M. Rey, A. Rodina, S. Tashkun, V. Tyuterev, J. Quant. Spectrosc. Radiat. Transfer 2019, 239, 106668. https://doi.org/10.1016/j.jqsrt.2019.106668
[39] D. Viglaska-Aflalo, M. Rey, A. Nikitin, T. Delahay, Phys. Chem. Chem. Phys. 2020, 22, 3204. https://doi.org/10.1039/C9CP06383H
[40] M. Rey, I.S. Chizhmakova, A.V. Nikitin, V.G. Tyuterev, Phys. Chem. Chem. Phys. 2021, 23, 12115. https://doi.org/10.1039/D0CP05727D
[41] O. Egorov, M. Rey, A. Nikitin, V. Dominika, J. Phys. Chem. A 2021, 125, 10568. https://doi.org/10.1021/acs.jpca.1c08717
[42] O. Egorov, M. Rey, A.V. Nikitin, D. Viglaska, J. Phys. Chem. A 2022, 126, 6429. https://doi.org/10.1021/acs.jpca.2c04822
[43] O. Egorov, M. Rey, R. V. Kochanov, A. V. Nikitin, V. Tyuterev, Chem. Phys. Lett. 2023, 811, 140216. https://doi.org/10.1016/j.cplett.2022.140216
[44] G. Herzberg, J. Shoosmith, Nature 1959, 183, 1801. https://doi.org/10.1038/1831801a0
[45] G. Herzberg, Proc. R. Soc. London A 1961, 262, 291. https://doi.org/10.1098/rspa.1961.0120
[46] G. Herzberg, J. W. C. Johns, Proc. R. Soc. London A 1966, 295, 107. https://doi.org/10.1098/rspa.1966.0229
[47] R. A. Bernheim, H. W. Bernard, P. S. Wang, L. S. Wood, P. S. Skell, J. Chem. Phys. 1970, 53, 1280. https://doi.org/10.1063/1.1674129
[48] R. A. Bernheim, H. W. Bernard, P. S. Wang, L. S. Wood, P. S. Skell, J. Chem. Phys. 1971, 54, 3223. https://doi.org/10.1063/1.1675313
[49] E. Wasserman, W.A. Yager, V.J. Kuck, Chem. Phys. Lett. 1970, 7, 409. https://doi.org/10.1016/0009-2614(70)80320-7
[50] E. Wasserman, V. J. Kuck, R. S. Hutton, W. A. Yager, J. Am. Chem. Soc. 1970, 92, 7491. https://doi.org/10.1021/ja00728a054
[51] E. Wasserman, V. J. Kuck, R. S. Hutton, E. D. Anderson, W. A. Yager, J. Chem. Phys. 1971, 54, 4120. https://doi.org/10.1063/1.1675475
[52] E. Wasserman, R. S. Hutton, V. J. Kuck, W. A. Yager, J. Chem. Phys. 1971, 55, 2593. https://doi.org/10.1063/1.1676452
[53] J. F. Harrison, L. C. Allen, J. Am. Chem. Soc. 1969, 91, 807. https://doi.org/10.1021/ja01032a0004
[54] J. F. Harrison, Acc. Chem. Res. 7, 378 (1974) https://doi.org/10.1021/ar50083a003
[55] C. F. Bender, H. F. Schaefer III, J. Am. Chem. Soc. 1970, 92, 4984. https://doi.org/10.1021/ja00719a039
[56] C. F. Bender, H. F. Schaefer III, D. R. Franceschetti, L. C. Allen, J. Am. Chem. Soc. 1972, 94, 6888. https://doi.org/10.1021/ja00775a004
[57] D. R. McLaughlin, C. F. Bender, H. F. Schaefer III, Theoret. Chim. Acta 1972, 25, 352. https://doi.org/10.1007/BF00526567
[58] S. V. O'Neil, H. F. Schaefer III, C. F. Bender, J. Chem. Phys. 1971, 55, 162. https://doi.org/10.1063/1.1675503
[59] P. J. Hay, W. J. Hunt, W. A. Goddard III, Chem. Phys. Lett. 1972, 13, 30. https://doi.org/10.1016/0009-2614(72)80035-6
[60] G. Herzberg, J. W. C. Johns, J. Chem. Phys. 1971, 54, 2276. https://doi.org/10.1063/1.1675164
[61] J. A. Mucha, K. M. Evenson, D. A. Jennings, G. B. Ellison, C. J. Howard, Chem. Phys. Lett. 1979, 66, 244. https://doi.org/10.1016/0009-2614(79)85007-1
[62] T. J. Sears, P. R. Bunker, A. R. W. McKellar, J. Chem. Phys. 1981, 75, 4731. https://doi.org/10.1063/1.442592
[63] T. J. Sears, P. R. Bunker, A. R. W. McKellar, J. Chem. Phys. 1982, 77, 5363. https://doi.org/10.1063/1.443784
[64] T. J. Sears, P. R. Bunker, A. R. W. McKellar, K. M. Evenson, D. A. Jennings, J. M. Brown, J. Chem. Phys. 1982, 77, 5348. https://doi.org/10.1063/1.443783
[65] A. R. W. McKellar, P. R. Bunker, T. J. Sears, K. M. Evenson, R. J. Saykally, S. R. Langhoff, J. Chem. Phys. 1983, 79, 5251. https://doi.org/10.1063/1.445713
[66] A. R. W. McKellar, C. Yamada, E. Hirota, J. Chem. Phys. 1983, 79, 1220. https://doi.org/10.1063/1.445926
[67] M. D. Marshall, A. R. W. McKellar, J. Chem. Phys. 1986, 85, 3716. https://doi.org/10.1063/1.450943
[68] T. J. Sears, J. Chem. Phys. 1986, 85, 3711. https://doi.org/10.1063/1.450942
[69] P. R. Bunker, T. J. Sears, A. R. W. McKellar, K. M. Evenson, F. J. Lovas, J. Chem. Phys. 1983, 79, 1211. http://dx.doi.org/10.1063/1.445925
[70] K. M. Evenson, T. J. Sears, A. R. W. McKellar, J. Opt. Soc. America B 1984, 1, 15. https://doi.org/10.1364/JOSAB.1.000015
[71] A. R. W. McKellar, T. J. Sears, Can. J. Phys. 1983, 61, 480. https://doi.org/10.1139/p83-060
[72] V.Staemmler, Theoret. Chim. Acta 1973, 31, 49. https://doi.org/10.1007/BF00527438
[73] V.Staemmler, Theoret. Chim. Acta 1974, 35, 309. https://doi.org/10.1007/BF00548481
[74] A.H. Pakiari, N.C. Handy, Theoret. Chim. Acta 1975, 40, 17. https://doi.org/10.1007/BF00547909
[75] M. J. S. Dewar, R. C. Haddon, W.-K. Li, W. Thiel, P. K. Weiner, J. Am. Chem. Soc. 1975, 97, 4540. https://doi.org/10.1021/ja00849a015
[76] J. H. Meadows, H. F. Schaefer III, J. Am. Chem. Soc. 1976, 98, 4383. https://doi.org/10.1021/ja00431a006
[77] L. B. Harding, W. A. Goddard III, J. Chem. Phys. 1977, 67, 1777. https://doi.org/10.1063/1.435043
[78] L. B. Harding, W. A. Goddard III, Chem. Phys. Lett. 1978, 55, 217. https://doi.org/10.1016/0009-2614(78)87005-5
[79] C. W. Bauschlicher Jr., I. Shavitt, J. Am. Chem. Soc. 1978, 100, 739. https://doi.org/10.1021/ja00471a012
[80] S.-K. Shih, S. D. Peyerimhoff, R. J. Buenker, M. Perić, Chem. Phys. Lett. 1978, 55, 206. https://doi.org/10.1016/0009-2614(78)87003-1
[81] P. Saxe, H. F. Schaefer III, N. C. Handy, J. Phys. Chem. 1981, 85, 745. https://doi.org/10.1021/j150607a001
[82] P. R. Taylor, J. Chem. Phys. 1981, 74, 1256. https://doi.org/10.1063/1.441186
[83] H.-J. Werner, E.-A. Reinsch, J. Chem. Phys. 1982, 76, 3144 https://doi.org/10.1063/1.443357
[84] H. M. Frey, G. J. Kennedy, J. Chem. Soc., Chem. Commun. 1975, 233. https://doi.org/10.1039/C39750000233
[85] F. Lahmani, J. Phys. Chem. 1976, 80, 2623. https://doi.org/10.1021/j100565a001
[86] P. F. Zittel, G. B. Ellison, S. V ONeil, E. Herbst, W. C. Lineberger, W. P. Reinhardt, J. Am. Chem. Soc. 1976, 98, 3731. https://doi.org/10.1021/ja00428a070
[87] J.W. Simons, R. Curry, Chem. Phys. Lett. 1976, 38, 171. https://doi.org/10.1016/0009-2614(76)80283근
[88] W. L. Hase, P. M. Kelley, J. Chem. Phys. 1977, 66, 5093 https://doi.org/10.1063/1.433765
[89] H. M. Frey, G. J. Kennedy, J. Chem. Soc. Faraday Trans. 1 1977, 73, 164. https://doi.org/10.1039/F19777300164
[90] R. K. Lengel, R. N. Zare, J. Am. Chem. Soc. 1978, 100, 7495. https://doi.org/10.1021/ja00492a010
[91] D. Feldmann, K. Meier, H. Zacharias, K. H. Welge, Chem. Phys. Lett. 1978, 59, 171. https://doi.org/10.1016/0009-2614(78)85641-3
[92] P. C. Engelking, R. R. Corderman, J. J. Wendoloski, G. B. Ellison, S. V. ONeil, W. C. Lineberger, J. Chem. Phys. 1981, 74, 5460. https://doi.org/10.1063/1.440951
[93] C. C. Hayden, D. M. Neumark, K. Shobatake, R. K. Sparks, Y. T. Lee, J. Chem. Phys. 1982, 76, 3607. https://doi.org/10.1063/1.443397
[94] S. R. Langhoff, E. R. Davidson, Int. J. Quantum Chem. 1973, 7, 999. https://doi.org/10.1002/qua. 560070515
[95] S. R. Langhoff, J. Chem. Phys. 1974, 61, 3881. https://doi.org/10.1063/1.1681679
[96] S. R. Langhoff, C. W. Kern, Applications of Electronic Structure Theory. Modern Theoretical Chemistry edited by H.F. Schaefer III 1977, 4, 381. https://doi.org/10.1007/978-1-4684-8541-7_10
[97] T. J. Sears, P. R. Bunker, J. Chem. Phys. 1983, 79, 5265. https://doi.org/10.1063/1.445714
[98] P. R. Bunker, T. J. Sears, J. Chem. Phys. 1985, 83, 4866. https://doi.org/10.1063/1.449747
[99] D. G. Leopold, K. K. Murray, A. E. S. Miller, W. C. Lineberger, J. Chem. Phys. 1985, 83, 4849. https://doi.org/10.1063/1.449746
[100] L. H. Coudert, B. Gans, F. Holzmeier, J.-C. Loison, G. A. Garcia, C. Alcaraz, A. Lopes, A. Röder, J. Chem. Phys. 2018, 149, 224304. https://doi.org/10.1063/1.5062834
[101] H. Petek, D. J. Nesbitt, P. R. Ogilby, C. B. Moore, J. Phys. Chem. 1983, 87, 5367. https://doi.org/10.1021/j150644a012
[102] H. Petek, D. J. Nesbitt, D. C. Darwin, C. B. Moore, J. Chem. Phys. 1987, 86, 1172. https://doi.org/10.1063/1.452263
[103] H. Petek, D. J. Nesbitt, C. B. Moore, F. W. Birss, D. A. Ramsay, J. Chem. Phys. 1987, 86, 1189. http://dx.doi.org/10.1063/1.452264
[104] H. Petek, D. J. Nesbitt, D. C. Darwin, P. R. Ogilby, C. B. Moore, D. A. Ramsay, J. Chem. Phys. 1989, 91, 6566. https://doi.org/10.1063/1.457375
[105] A. R. Hoy, P. R. Bunker, J. Mol. Spectrosc. 1974, 52, 439. https://doi.org/10.1016/0022-2852(74)90191-X
[106] P. R. Bunker, B. M. Landsberg, J. Mol. Spectrosc. 1977, 67, 374. https://doi.org/10.1016/0022-2852(77)90048-0
[107] A. R. Hoy, P. R. Bunker, J. Mol. Spectrosc. 1979, 74, 1. https://doi.org/10.1016/0022-2852(79)90019$\underline{5}$
[108] J. T. Hougen, P. R. Bunker, J. W. C. Johns, J. Mol. Spectrosc. 1970, 34, 136. https://doi.org/10.1016/0022-2852(70)90080-9
[109] P. Jensen, Comp. Phys. Rep. 1983, 1, 1. https://doi.org/10.1016/0167-7977(83)90003-5
[110] P. Jensen, P. R. Bunker, J. Mol. Spectrosc. 1983, 99, 348. https://doi.org/10.1016/0022-2852(83)90319-3
[111] P. Jensen, P. R. Bunker, J. Mol. Spectrosc. 1986, 118, 18. https://doi.org/10.1016/0022-2852(86)90220-1
[112] A. D. McLean, P. R. Bunker, R. M. Escribano, P. Jensen, J. Chem. Phys. 1987, 87, 2166. http://dx.doi.org/10.1063/1.453141
[113] D. C. Comeau, I. Shavitt, P. Jensen, P. R. Bunker, J. Chem. Phys. 1989, 90, 6491. https://doi.org/10.1063/1.456315
[114] I. N. Kozin, P. Jensen, J. Mol. Spectrosc. 1997, 183, 398. https://doi.org/10.1006/jmsp.1997.7288
[115] A. Tajti, P. G. Szalay, A. G. Császár, M. Kállay, J. Gauss, E. F. Valeev, B. A. Flowers, J. Vázquez, J. F. Stanton, J. Chem. Phys. 2004, 121, 11599. http://dx.doi.org/10.1063/1.1811608
[116] D. M. Medvedev, L. B. Harding, S. K. Gray, Mol. Phys. 2006, 104, 73. http://dx.doi.org/10.1080/00268970500238663
[117] H. Ozeki, S. Saito, The Astrophysical Journal 1995, 451, L97. https://doi.org/10.1086/309696
[118] H. Ozeki, S. Saito, J. Chem. Phys. 1996, 104, 2167. https://doi.org/10.1063/1.470972
[119] J. Nolte, F. Temps, H. Gg. Wagner, M. Wolf, T. J. Sears, J. Chem. Phys. 1994, 100, 8706. http://dx.doi.org/10.1063/1.466726
[120] H. Ozeki, S. Bailleux, G. Wlodarczak, A\&A 2011, 527, A64. https://doi.org/10.1051/00046361/201016043
[121] P. J. Knowles, C. Hampel, H.-J. Werner, J. Chem. Phys. 1993, 99, 5219. http://dx.doi.org/10.1063/1.465990
[122] P. J. Knowles, C. Hampel, H.-J. Werner, J. Chem. Phys. 2000, 112, 3106. http://dx.doi.org/10.1063/1.480886
[123] T. J. Lee, P. R. Taylor, Int. J. Quant. Chem. 1989, 36, 199. https://doi.org/10.1002/qua. 560360824
[124] C. L. Janssen, I. M. B. Nielsen, Chem. Phys. Lett. 1998, 290, 423. https://doi.org/10.1016/S0009-2614(98)00504-1
[125] G. Duxbury, Ch. Jungen, Mol. Phys. 1988, 63, 981. https://doi.org/10.1080/00268978800100721
[126] A. Alijah, G. Duxbury, Mol. Phys. 1990, 70, 605. https://doi.org/10.1080/00268979000102621
[127] W. H. Green, N. C. Handy, P. J. Knowles, S. Carter, J. Chem. Phys. 1991, 94, 118. https://doi.org/10.1063/1.460385
[128] W. Xie, C. Harkin, H. -L. Dai, J. Chem. Phys. 1990, 93, 4615. http://dx.doi.org/10.1063/1.458701
[129] G. V. Hartland, D. Qin, H.-L. Dai, J. Chem. Phys. 1995, 102, 6641. https://doi.org/10.1063/1.469136
[130] J. -P. Gu, G. Hirsch, R. J. Buenker, M. Brumm, G. Osmann, P. R. Bunker, P. Jensen, J. Mol. Spectrosc.
2000, 517-518, 247. https://doi.org/10.1016/S0022-2860(99)00256-2
[131] T. H. Dunning, J. Chem. Phys. 1989, 90, 1007. https://doi.org/10.1063/1.456153
[132] K. A. Peterson, D. E. Woon, T. H. Dunning, J. Chem. Phys. 1994, 100, 7410. https://doi.org/10.1063/1.466884
[133] G. Knizia, T. B. Adler, H.-J. Werner, J. Chem. Phys. 2009, 130, 054104. https://doi.org/10.1063/1.3054300
[134] J. Gauss, A. Tajti, M. Kállay, J. F. Stanton, P. G. Szalay, J. Chem. Phys. 2006, 125, 144111. https://doi.org/10.1063/1.2356465
[135] A.D. Boese, M. Oren, O. Atasoylu, M.L. Jan Martin, M. Kállay, J. Gauss, J. Chem. Phys. 2004, 120, 4129. https://doi.org/10.1063/1.1638736
[136] D. Viglaska, M. Rey, A. V. Nikitin, V. G. Tyuterev, J. Chem. Phys. 2020, 153, 084102. https://doi.org/10.1063/5.0016365
[137] W. Kutzelnigg, Mol. Phys. 2007, 105, 2627. http://dx.doi.org/10.1080/00268970701604671
[138] J. Tennyson, M. A. Kostin, P. Barletta, G. J. Harris, O. L. Polyansky, J. Ramanlal, N. F. Zobov, Comp. Phys. Comm. 2004, 163, 85. https://doi.org/10.1016/j.cpc.2003.10.003
[139] V. Szalay, J. Mol. Spectrosc. 1988, 128, 24. https://doi.org/10.1016/0022-2852(88)90204-4
[140] V. Szalay, J. Chem. Phys. 1990, 92, 3633. https://doi.org/10.1063/1.457819
[141] G. Tarczay, A. G. Császár, W. Klopper, V. Szalay, W. D. Allen, H. F. Schaefer III, J. Chem. Phys. 1999, 110, 11971. http://dx.doi.org/10.1063/1.479135
[142] I. E. Gordon, L. S. Rothman, R. J. Hargreaves, R. Hashemi, E. V. Karlovets, F. M. Skinner, E. K. Conway, C. Hill, R. V. Kochanov, Y. Tan, P. Wcisłof, A. A. Finenko, K. Nelson, P. F. Bernath, M. Birk, V. Boudon, A. Campargue, K. V. Chance, A. Coustenis, B. J. Drouinm, J.-M. Flaud, R. R. Gamache, J. T. Hodges, D. Jacquemart, E. J. Mlawer, A. V. Nikitin, V. I. Perevalov, M. Rotger, J. Tennyson, G. C. Toon, H. Tran, V. G. Tyuterev, E. M. Adkins, A. Baker, A. Barbe, E. Canè, A. G. Császár, O. Egorov, A. J. Fleisher, H. Fleurbaey, A. Foltynowicz, T. Furtenbacher, J. J. Harrison, J.-M. Hartmann, V.-M. Horneman, X. Huang, T. Karman, J. Karns, S. Kassi, I. Kleiner, V. Kofman, F. Kwabia-Tchana, T. J. Lee, D. A. Long, A. A. Lukashevskaya, O. M. Lyulin, V. Yu. Makhnev, W. Matt, S. T. Massie, M. Melosso, S. N. Mikhailenko, D. Mondelain, H. S. P. Müller, O. V. Naumenko, A. Perrin, O.L. Polyansky, E. Raddaoui, P. L. Raston, Z. D. Reed, M. Rey, C. Richard, R. Tóbiás, I. Sadiek, D. W. Schwenke, E. Starikova, K. Sung, F. Tamassia, S. A. Tashkun, J. Vander Auwera, I. A. Vasilenko, A. A. Vigasin, G. L. Villanueva, B. Vispoel, G. Wagner, A. Yachmenev, and S. N. Yurchenko, J. Quant. Spectrosc. Radiat. Transfer 2022, 277, 107949. https://doi.org/10.1016/j.jqsrt.2021.107949
[143] T. Delahaye, R. Armante, A. Chédin, L. Crépeau, C. Crevoisier, V. Douet, N. Jacquinet-Husson, A. Perrin, N. A. Scott, A. Barbe, V. Boudon, A. Campargue, L. H. Coudert, V. Ebert, J.-M. Flaud, R. R. Gamache, D. Jacquemart, A. Jolly, F. Kwabia Tchana, A. Kyuberis, G. Li, O. M. Lyulin, L. Manceron, S. Mikhailenko, N. Moazzen-Ahmadi, H. S. P. Müller, O. V. Naumenko, A. Nikitin, V. I. Perevalov, C. Richard, E. Starikova, S. A. Tashkun, Vl. G. Tyuterev, J. Vander Auwera, B. Vispoel, A. Yachmenev, and S. Yurchenko, J. Mol. Spectrosc. 2021, 380, 111510. https://doi.org/10.1016/j.jms.2021.111510
[144] M. Rey, J. Chem. Phys. 2022, 156, 224103. https://doi.org/10.1063/5.0089097
[145] H. M. Pickett, J. Mol. Spectrosc. 1991, 148, 371. https://doi.org/10.1016/0022-2852(91)90393-O
[146] A. V. Burenin, T. M. Fevral'skikh, E. N. Karyakin, O. L. Polyansky, S. M. Shapin, J. Mol. Spectrosc.
1983, 100, 182. https://doi.org/10.1016/0022-2852(83)90035-8
[147] O. L. Polyansky, J. Mol. Spectrosc. 1985, 112, 79. https://doi.org/10.1016/0022-2852(85)90193-6
[148] S. P. Belov, I. N. Kozin, O. L. Polyansky, M. Yu. Tret'yakov, N. F. Zobov, J. Mol. Spectrosc. 1987, 126, 113. https://doi.org/10.1016/0022-2852(87)90081-6
[149] H. M. Pickett, J. C. Pearson, C. E. Miller, J. Mol. Spectrosc. 2005, 233, 174. https://doi.org/10.1016/j.jms.2005.06.013
[150] Vl. G. Tyuterev, J. Mol. Spectrosc. 1992, 151, 97. https://doi.org/10.1016/0022-2852(92)90009-D
[151] V. I. Starikov, S. A. Tashkun, Vl. G. Tyuterev, J. Mol. Spectrosc. 1992, 151, 130. https://doi.org/10.1016/0022-2852(92)90010-L
[152] V. I. Starikov, S. N. Mikhailenko, J. Phys. B: At. Mol. Opt. Phys. 2000, 33, 2141. https://doi.org/0.1088/0953-4075/33/11/314

