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1 Reflectance spectra of mascagnite and salammoniac minerals

2 with varying viewing geometry.

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7 Highlights

- Mascagnite-(NH₄)₂SO₄ and salammoniac-NH₄Cl minerals were characterized by means of
 BRDF spectroscopy with varying viewing geometries in term of incidence (*i*) and emergence
 (*e*) angles.
- Relationship between the viewing geometry variations and the spectral feature parameters
 (band area, depth, slope) of NH₄⁺ features were observed.
- High spectral variability was observed for $i \ge 30^\circ$ and $e \ge 60^\circ$ angles.
- Results from such investigation will be useful to interpret occurrence of ammonium salts on
 planetary surfaces, especially small bodies.

16 Abstract

We analyse the quantitative effects of viewing geometry variations on the near-infrared reflectance 17 spectra of mascagnite-(NH₄)₂SO₄ and salammoniac-NH₄Cl samples. Bi-directional reflectance 18 19 spectra are collected in the 1 - 4.2 μ m range considering a set of 3 incidence (*i*) angles (*i* = 0°; 30°; 20 60°) and 9 emergence (e) angles between -70° and 70° at room temperature and computed with 21 respect to the normal direction. The NH_4^+ overtone and combinations bands located at ~ 1.09, 1.32, 1.62, 2.04, 2.2 and 3.05 µm are experimentally investigated. The bidirectional reflectance spectra 22 of these samples show significant variations with the observation geometry in terms of reflectance 23 24 values, spectral slope, and absorption bands parameters. The band positions remain essentially 25 unchanged by varying the incident and emergence angles. On the other hand, bands' area and depth 26 show the highest variability for $i \ge 30^{\circ}$ and *e* greater than $\pm 40^{\circ}$ (up to a factor 2.3 in relative mean variation). The area and depth parameters of these bands show a dual behaviour: (i) for the weak-27 28 medium spectral features below 2 μ m the area and depth decrease as the phase angle increases. 29 (ii) The strong spectral features above 2 μ m increase their values only at phase angles above 90°, 30 but also at low phase angles for high incidences, $i \ge 30^\circ$. This behaviour is linked both to the nonlinear radiative transfer in particulate media and to the way the band depth and area are defined, 31 32 relative to the local continuum. We observe important dependence (up to ~60% relative mean variation) of band depth and area on the incidence angle, up to 60°, compared to moderate variation 33 34 with emergence angles (up to ~20% relative mean variation). Furthermore, the ~ 3 μ m features 35 becomes more saturated at ± 70° emergence angles. A general trend of spectral bluing with change in observation geometry is observed. The current dataset is a contribution in the framework of 36 37 present and future space missions focused on understanding the nature and quantification of 38 ammonium-bearing minerals on icy bodies. The NH₄⁺ – bearing minerals identification could provide

information on: (i) ocean/brine compositions, (ii) possible explanations of geological phenomenaand (iii) implications for biological activity.

41 **Keywords:** mascagnite, salammoniac, reflectance spectra, icy bodies, viewing geometry.

42 1 Introduction

43 Planetary surface remote sensing by visible and near-infrared reflectance spectroscopy (0.4 -5 μ m), 44 from telescopic observations and spacecrafts, is a widely applied technique for investigating and 45 deciphering their composition. One way to resolve and identify which are the best candidates to 46 represent surface compositions of planetary bodies is to collect mineral spectra in the laboratory 47 and compare them with planetary data. In addition, by varying different physical parameters we can 48 extend the existing spectral libraries such as temperature, granulometry and observation geometry 49 on these minerals. In planetary sciences, reflectance spectroscopy in the visible and near-infrared 50 has proven its considerable power to disclose the mineralogy of Solar System bodies (Bishop et al., 51 2019; Pelkey et al., 2007; Singer et al., 1979).

52 The reflectance spectra of minerals in the near infrared and part of mid infrared region (1 -5 μ m, SWIR 1-3µm + LWIR 3-5µm) have diagnostic absorption features, due to electronic and vibration 53 54 transitions, that are essential in determining surface mineralogy and consequently to understand 55 planetary formation and evolution (Mustard and Glotch, 2019). Mineral spectral bands are closely linked to the different configurations of atoms and molecules inside the crystal structure, but they 56 57 can vary in terms of positions and shapes also as a function of different environmental conditions 58 such as temperature or pressure (e.g. (Comodi et al., 2017, 1990) which affect the structural 59 evolutions, sometimes in anomalous way (Gatta et al., 2005). Furthermore, grain size and 60 observation geometry strongly influence the spectral features (e.g. Beck et al., 2012; Comodi et al., 61 2021; De Angelis et al., 2021).

Regarding the viewing geometry, the reflectance spectra collected by spacecraft missions are 62 63 generally bidirectional, that is, the incidence of the sun light has a specific angle relative to the 64 normal direction of the surface and the reflected light has itself a proper emergence direction 65 (Pommerol and Schmitt, 2008a). Numerous bidirectional radiative transfer models have been 66 created that take into account the physical properties of the surface of planetary bodies, e.g., particle size and shape, roughness of the surface (Hapke, 2012, 1981); however, as surfaces are 67 68 complex systems and often difficult to model, these models need to be tested with experimental 69 data to confirm their reliability and suggest how they can be improved. For this reason, it is essential to collect laboratory Bidirectional Reflectance Distribution Function (BRDF) spectra of well-70 71 characterised samples. Furthermore, with the creation of hyperspectral cameras with increasingly 72 high spatial resolution, the effect of surface roughness and observation geometry can be also 73 considered.

Given the presence of surface roughness, topography and variations in morphology, the surface reflectance spectra of every Solar System bodies are affected by observation geometries, some examples include Mercury (Warell and Bergfors, 2008), Uranus' moons (Nelson et al., 1987) and asteroids (Clark et al., 2002; Magrin et al., 2012). For example on icy bodies, chaos terrain on Europa's surface, that represents resurfacing process, can be mostly identified at incidence angles > 70° (Neish et al., 2012). On Pluto's surface, the changes in viewing geometry during the New Horizons flyby generated significant spectral changes in the N₂, CO and CH₄ absorption bands (Cruikshank et al., 2015). The dwarf planet Ceres displays a negative slope in the 0.5-2.5 μ m spectral region at fresh impact craters which is related to possible changes in physical properties and composition (Stephan et al., 2017).

84 Up to now, several laboratory studies have analysed in detail ammonium bearing minerals using 85 visible and near-infrared spectroscopy at various temperatures and considering different particle sizes (e.g., (Berg et al., 2016; Ciarniello et al., 2017; De Angelis et al., 2021a; Ehlmann et al., 2018; 86 87 Fastelli et al., 2020; Krohn and Altaner, 1987). In all these works, the variation in viewing geometry has not been examined for the here reported samples. Exception made for the work of Ciarniello et 88 89 al., 2017 that examined the BRDF effect of NH₄Cl sample. Thus, reflectance spectra of NH₄⁺ minerals 90 with different observation geometries seem to be lacking in the bibliographic record and a 91 systematic work, dedicated to a comparative study of different effects of emergence and incidence 92 angles on NH₄⁺ features, is here proposed.

93 Ammonium bearing minerals have been hypothesised on the surface of primordial objects, such as 94 comets (Poch et al., 2020), on evolved telluric planets such as Mars (Quinn et al., 2011), in some icy 95 bodies like Ceres and Pluto (Dalle Ore et al., 2019; King et al., 1992) and asteroids, like Ryugu 96 (Pilorget et al., 2022). The interest and identification of these minerals is linked to the possible 97 emergence of life. Nitrogen is a key element in many biological compounds and in providing 98 explanations for evolution and geological phenomena as they are potential environmental markers. 99 The unambiguous detection and quantification of ammonium bearing minerals on icy bodies and 100 small object like comets has a significant impact on the presence of either liquid salt water or 101 phenomena such as cryovolcanism (Kargel, 1992) but also in the incorporation and evolution of 102 nitrogen in the early Solar System (Füri and Marty, 2015). Salammoniac-NH₄Cl was identified on the 103 surface of Ceres in relation to the absorption feature at \sim 2.21 μ m in a bright spot area (De Sanctis et al., 2016). The genesis of this mineral is linked to the presence of either ancient brine reservoirs 104 105 or brines induced by impact heating (De Sanctis et al., 2020). In fact, the formation of salammoniac 106 is strictly related to the presence of liquid water combined with hydrothermal activity. Ammonium 107 spectral signatures, at 1.65 and 2.2 μm, were also identified on Pluto, in a region of past extensional tectonic activity called Virgil Fossae (Cruikshank et al., 2019). These bands can be attributed to the 108 109 presence of ammonium chloride salts in analogy to the one reported by Cook et al. (2018) who modelled the surface of Charon, Nyx and Hydra Pluto's satellites, observed by New Horizons, using 110 111 NH₄Cl as endmembers.

Mascagnite- $(NH_4)_2SO_4$ has been suggested to be one of the rock forming minerals of the interior of 112 Titan generated by the interactions of primordial aqueous ammonia with sulphate solutions leaking 113 114 from the rocky core (Fortes et al., 2007). The Titan's internal model, from outside to inside, consists 115 of a shell of methane clathrate, a layer of aqueous ammonium sulphates and a serpentinite core 116 overlaid by a high pressure icy mantle (Fortes et al., 2007). Furthermore, (NH₄)₂SO₄ was proposed 117 on Mars Iano Chaos area due to the presence of absorption bands at ~1.07, 1.31 and 1.57 μ m consistent with its spectrum (Sefton-Nash et al., 2012). Salammoniac and mascagnite minerals are 118 known for their low temperature phase transitions characterized by a reordering of hydrogen bonds 119 inside the crystal structure. Salammoniac undergoes to a λ-transition defined as order-disorder type 120 121 at ~242K (Levy and Peterson, 1952). After the phase transition, the mineral has a disordered CsCltype crystal structure in which the NH4⁺ ions are distributed arbitrarily between two energy-122

123 equivalent orientations (Wang and Wright, 1974). Mascagnite, on the other hand , undergoes a phase transition upon cooling at ~224K from a para-electric crystal structure with space group Pnma 124 towards a ferro-electric crystal structure with space group *Pna*2₁ (Hasebe, 1981). In the para-electric 125 phase, H-bonds are bifurcated and trifurcated while in the ferro-electric phase they turn in a classic, 126 much stronger, hydrogen bond. The presence of these phase transitions may have a strong effect 127 128 on the convective flow and consequently on the results of internal planetary models (Fortes and 129 Choukroun, 2010). The two selected ammonium bearing minerals were previously characterised and tested in Fastelli et al. (2022), where the low temperature spectral behaviour was discussed. 130 131 The two samples analysed in this work could represent the two most promising candidates as 132 ammonium-bearing minerals. Also their sensitivity to solar ray alterations (Dalle Ore et al., 2019) 133 and phase transitions should have direct consequences on the internal structure and dynamic of the bodies. If ammonium-bearing minerals are detected at the surface of icy bodies, they can play a key 134 role as freezing compounds, depending on their abundance, that reduce the solidification 135 temperature of water, allowing it to remain liquid in the conditions beneath the crust of frozen 136 137 bodies (Neveu et al., 2017). Ammonium bearing minerals mixed with liquid H₂O can also participate in potential prebiotic reactions (e.g; Burcar et al., 2016). In addition, the spectra shown and analysed 138 139 in this work will be used for future interpretation of reflectance spectra of Pluto's surface as well as 140 on other planetary bodies such as icy satellites, asteroids, Kuiper Belt Objects (KBOs) and comets.

141 In this work the BRDF spectra of salammoniac and mascagnite minerals were collected with a set of 142 three incidence angles $i = 0^{\circ}$, 30° and 60° and nine emergence angles in the $e = \pm 70^{\circ}$ range, to 143 investigate how the observation geometries influence the NH₄⁺ signatures: what is the behaviour of 144 band parameters such as their area, full width at half maximum (FWHM) and depth under varying 145 observation geometries. The effect of geometry variation on the spectral slope was also described.

146 2 Materials and Methods

147 2.1 Ammonium bearing minerals

148 The two selected mascagnite-(NH₄)₂SO₄ and salammoniac-NH₄Cl samples were checked and 149 characterized using X-ray powder diffraction (XRPD) technique at the Department of Physics and 150 Geology, University of Perugia (Italy), using a Philips PW 1830 diffractometer, with a graphite monochromator, and using the CuK α radiation (λ = 1.54184 Å). Data were taken with a step scan of 151 0.02°/step and a step time of 20 s/step. Quantitative phase analysis were performed using the 152 Rietveld method (Rietveld, 1969) through EXP-GUI GSAS software (Larson and Von Dreele, 1987; 153 Toby, 2001) following the procedure described in (Guidobaldi et al., 2018). The results show that 154 the salammoniac sample is 100 wt% pure while the mascagnite sample contains a small percentage 155 156 of mohrite mineral [(NH₄)₂Fe(SO₄)₂·6H₂O] about 6.4 wt% (Fastelli et al., 2022), see table 1 for details. Particle size, for the analysed samples, ranges between 32 – 80 µm. Samples were first finely ground 157 158 with an agate mortar and then dry sieved.

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Table 1: Compositional data, source and description for selected samples used in this study fromFastelli et al., 2022.

Mineral Name	Chemical Formula	Purity	Source
Mascagnite / Ammonium	(NH4)2SO4	XRPD checked: small	Pozzuoli, Italy
sulfate		amount of mohrite 6.4 wt%	
		[(NH ₄) ₂ Fe(SO ₄) ₂ ·6H ₂ O]	
Salammoniac / Ammonium	NH ₄ Cl	XRPD checked: 100 wt%	Pozzuoli, Italy
chloride			

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163 2.2 Bidirectional Reflectance Distribution Function spectra

The reflectance spectra analysed in this study were collected using the custom-made bidirectional 164 165 reflectance spectro-gonio-radiometer SHINE (SpectropHotometer with variable INcidence and Emergence) (Brissaud et al., 2004) at the Cold Surface Spectroscopy facility (CSS) of the IPAG 166 167 laboratory (https://cold-spectro.sshade.eu). This set up permits the collection of spectra in the visible (VIS) and near-infrared (NIR) ranges $(0.35 - 4.5 \,\mu\text{m})$ over a wide set of illumination $(0 - 80^\circ)$ 168 169 and emergence angles ($e = 0.80^{\circ}$). We recorded spectra over the $1 - 4.2 \,\mu m$ range with a spectral sampling of 10 nm up to 3.2 µm and 20 nm above with effective spectral resolution varying in 3 170 171 steps (3 gratings with fixed slit width) between 7.5 and 30 nm over the range. The illumination spot 172 was 7.5 mm in diameter. The illumination is monochromatic and modulated (~970 Hz) with a synchronous detection so thermal emission from the sample and surrounding (not modulated) did 173 174 not contribute to the reflected light (modulated) measurement (Bonnefoy 2001, Brissaud et al., 175 2004).

176 Measurements were taken in the principal plane considering a set of 3 incidence angles $i = 0^{\circ}$, 30° , 177 60° and 9 emergence angles $e = -70^{\circ}$, -60° , -40° , -20° , 0° , 20° , 40° , 60° , 70° at room temperature. 178 Both illumination and emergence angles are zenithal angles (measured from the normal direction 179 to the surface) and the emergence is positive in the same quadrant as illumination in the principal 180 plane (i.e, this corresponds to azimuth = 0° for positive emergence values, and 180° for negative values). The phase angle g is thus simply: g = |i - e|. Furthermore, we define (*i*=0°, *e*=20°) as our

reference configuration and compare the spectra measured at all other configurations with the spectrum measured at this configuration. Figure 1 presents a schematic view of the different geometrical parameters analysed.



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Fig. 1: Schematic representation of the different geometrical parameters analysed in this work.Pictures modified from Potin et al., 2019.

Approximately 2 g of powdered samples were put in an aluminum sample holder cup, 34 mm in diameter and 4 mm deep. The density (0.54 and 0.64 g/cm³, respectively) and porosity (~64% for both) of the salammoniac and mascagnite samples were calculated using the measured volume and mass of the samples and the bulk density of the minerals.

The surfaces were accurately prepared, as the macroscopic structure (roughness) of the surface has a significant effect on the collected spectra and particularly on the reflectance changes between different measurement geometries (Cord et al., 2003; Pommerol and Schmitt, 2008a).

195 2.3 Data reduction

The recorded spectra were firstly calibrated in reflectance factor (i.e. relative to a perfectly 196 197 Lambertian surface) by measuring the BRDF of a Spectralon[©] 99% reference target at the same set 198 of geometries as well as an Infragold[©] reference target measured under the standard calibration 199 geometry (i=0, $e=30^{\circ}$) of the instrument (spectrally equivalent to the directional-hemispheric ($i=8^{\circ}$) calibration provided by Labsphere). The reflectance is obtained by dividing the sample 200 measurement by the Spectralon[©] measurement at the same geometry and by multiplying by the 201 value of the spectral reflectance of Spectralon[©] at that geometry, from a previously calibrated 202 203 absolute BRDF evaluation of the reference target (Bonnefoy, 2001). The spectral behavior of the 204 part of the spectrum above 1.3 µm is obtained by dividing by the Infragold[©] measurement at the 205 standard calibration geometry (i=0, e=30°) of the instrument and by multiplying by its spectral directional-hemispheric reflectance. The absolute reflectance of this part of the spectrum is then 206

obtained by adjusting (with a multiplying factor) the value of the spectrum at 1.3 μm, to that of the
 spectrum absolutely calibrated with Spectralon.

209 Regarding the study of band parameters in relation to angle variations, the procedure used to homogenize the data is detailed in the following. For each examined band, the removal of the 210 continuum was taken equal and kept constant (same anchor point for the continuum) for all spectra. 211 The position of the absorption bands was then measured using a second-order polynomial fit around 212 the minimum. The depth of the band was calculated according to the following equation of: $D = (R_c$ 213 - R_b / R_c in which R_b is the reflectance at the centre of the band and R_c is the reflectance of the 214 continuum at the centre of the band (Clark and Roush, 1984). The band area was determined by 215 216 numerical integration over the band after continuum removal. The full width at half maximum 217 (FWHM) was derived by cross cutting the band removed from the continuum with a horizontal line that passes through mid-depth. The band area and depth were computed for the entire band 218 219 complex and not for all relative minima. More detailed information on the spectral parameter analysis procedure can be found in Comodi et al. (2021) and Fastelli et al. (2021). The spectra 220 221 reduction was performed for the selected absorption bands of interest for remote sensing identification of NH₄⁺- bearing minerals located at ~1.07, ~1.3, ~1.58, ~2.07, ~2.17 and ~3.03 µm 222

223 3 Results

Figures 2 and 3 report the bi-directional reflectance spectra of the two selected ammonium bearing minerals mascagnite and salammoniac, respectively, collected at different illumination-emergence geometries. Table 2 provides the absorption bands positions at reference geometry for this work $(i=0^{\circ}, e=20^{\circ})$ analysed in this work and their assignment.



Fig. 2: BRDF spectra of the Mascagnite- $(NH_4)_2SO_4$ sample. Each of the three plots presents spectra measured at zenithal incidence angle 0° (a), 30° (b) and 60° (c) and different emergence angles in the ± 70° range. Spectra with negative emergence "e" angle (i.e. azimuth = 180°) are reported with dashed lines retaining the same colour for same absolute value of the positive emergence (i.e. azimuth = 0°). BRDF data at 1.06, 1.28, 1.57, 2.14 and 3.01 µm wavelengths are analysed (marked

234 by vertical dash lines).



Fig. 3: BRDF spectra of the Salammoniac-NH₄Cl sample. Each of the three plots presents spectra measured at zenithal incidence angle 0° (a), 30° (b) and 60° (c) and different emergence angles in the \pm 70° range. Spectra with negative emergence "e" angle (i.e. azimuth = 180°) are reported with dashed lines retaining the same colour for same absolute value of the positive emergence (i.e. azimuth = 0°). BRDF data at 1.09, 1.33, 1.62, 2.06, 2.2 and 3.05 µm wavelengths are analysed (marked by vertical dash lines).

Table 2: Positions and vibration mode assignments of the reflectance spectra collected with reference geometry (i=0°; e=20°) for the selected bands analysed in this work. Intensity labels are: w = weak, m-s = moderately-strong, m=medium and s=strong. Band assignments were made following (Berg et al., 2016; Fastelli et al., 2022; Krohn and Altaner, 1987).

Mineral	Band position (µm)	Vibrational modes	Intensity
Mascagnite	1.06	NH4 ⁺ 3v3	W
(NH4)2SO4	1.28	$NH_4^+ 2v_3 + v_4$	W
	1.57	NH4 ⁺ 2v3	m-s
	2.18	$NH_4^+ V_3 + V_4$	S
	3.01	$NH_4^+ v_3$	S
Salammoniac	1.09	NH4 ⁺ 3v3	W
NH₄Cl	1.32	$NH_4^+ 2v_3 + v_4$	W
	1.62	NH4 ⁺ 2v3	m-s
	2.06	$NH_4^+ V_2 + V_3$	S
	2.2	$NH_4^+ v_1 + v_4$ and/or $NH_4^+ v_3 + v_4$	5
	3.05	$NH_4^+ v_3$	S

246

247 The absorption bands identified in the spectra of both samples in the 1-2.5 µm range are mainly due to overtones and combinations of the fundamental internal vibrations of the NH_4^+ group. These 248 features are located at ~1.07, ~1.3, ~1.58, ~2.07, ~2.17 µm. On the right side of the spectra, between 249 250 2.8 and 3.8 µm, we notice several saturated and overlapping NH4⁺ bands except for the salammoniac 251 sample which shows two clear features at 3.05 and 3.25 µm. Due to the high hygroscopicity of these 252 minerals, although they were kept inside a desiccator before measurement, some spectral features of adsorbed H₂O are present at ~1.47, ~1.94, ~2.62, ~2.83 and ~3.0 μ m that, in some cases, overlap 253 254 with the NH₄⁺ bands. In addition, the water molecules absorb in wide wavelength ranges around 255 each of the central wavelengths listed above. These positions refer to spectra collected with the reference configuration geometry (*i*=0°, *e*=20°). 256

Ammonium sulphates NH₄⁺ absorption features are located near 1 μ m, at ~1.56 μ m, and ~2.2 μ m. Over 3 μ m, bands are broad and saturated at reference configuration geometry. Ammonium chloride (NH₄Cl) presents clear absorption features in the 1.0 – 2.12 μ m range with distinct absorption feature at 1.09, 1.33, 1.38 1.62 and 1.67 μ m and three bands in the 2- 2.4 μ m range attributed to NH₄⁺ groups.

The selected absorption bands can be ascribed to the following detailed vibrational modes of NH₄⁺ group: ~ 1.09 µm (NH₄⁺ 3v₃); ~1.3 µm (NH₄⁺ 2v₃ + v₄); ~ 1.6 µm (NH₄⁺ 2v₃); ~2.02 µm (NH₄⁺ v₂ + v₃); ~2.2 µm (NH₄⁺ v₃ + v₄); ~ 3 µm (NH₄⁺ v₃) (Berg et al., 2016; Fastelli et al., 2021; Krohn and Altaner, 1987).

The effect of increase in reflectivity towards high incidence and emergence angles due to 'forward scattering' is evident from the plots shown in Figures 2 and 3 for both samples. In fact, the highest reflectance values are found at the highest incidence and negative emergence angles (high phase angles). A second clear effect is the reduction of the negative slope with respect to wavelength in the continuum of salammoniac and mascagnite samples at high emergence and incidence angles. Spectra acquired at an incident angle of 0° are characterised by a lower variability of this parameter relative to emergence angles. On the contrary, spectra collected at an incident angle of 60° show significant variations depending on the angle of emergence, in particular at both low (g \leq 20°, i.e., e = 40 - 70°) and high phase angles (g > 100°, i.e., e < -40°).

275 Another interesting behaviour is shown by the NH_4^+ absorption feature in mascagnite sample 276 located at ~3.0 µm. In particular, as the incident and emergence angles increase, these features 277 become less saturated and better defined (Figure 2). In order to clarify the behaviour of the selected 278 bands, for both samples, figures 4 and 5 show the variation in reflectance factor in relation to phase 279 angles (g).



280

Figure 4: Reflectance factor as a function of the phase angle g for mascagnite-(NH₄)₂SO₄ sample.

282 Colored symbols refer to each incidence angle: blue squares ($i = 0^{\circ}$), red circles ($i = 30^{\circ}$) and green 283 triangles ($i = 60^{\circ}$). Panels a, b, c, d and e correspond to 1.06, 1.28, 1.58, 2.18 and 3.01 μ m

284 wavelengths, respectively. Note the different scale of the y axis.



Figure 5: Reflectance factor as a function of the phase angle g for salammoniac-NH₄Cl sample. Colored symbols refer to each incidence angle: blue squares ($i = 0^{\circ}$), red circles ($i = 30^{\circ}$) and green triangles ($i = 60^{\circ}$). Panels a, b, c, d, e and f correspond to 1.09, 1.32, 1.62, 2.06, 2.2 and 3.05 µm wavelengths, respectively. Note the different scale of the y axis.

290 3.1 BRDF spectra and geometry

The measured reflectance factors at different incident zenithal angles are shown in polar plots in 291 Figures 6 and 7 for mascagnite and salammoniac, respectively, where the BRDF appears as a half of 292 a circle (with value of 1 for a perfectly Lambertian surface) and gives a simple overview of the 293 different viewing geometry effect. BRDF is plotted in polar coordinates at 1.06, 1.28, 1.57, 2.18 and 294 3.01 μ m for mascagnite (Figure 6) and at 1.09, 1.32, 1.60, 2.06, 2.2 and 3.05 μ m for salammoniac 295 (Figure 7). These wavelengths correspond to the minima positions of NH₄⁺ absorption bands. The 296 297 reported BRDF polar plots show a backscattering effect around the incidence quadrant (small phase angles) and forward scattering at high phase angles. The increase of reflectance values as phase 298 299 angle decreases is due to the shadows hiding opposition effect (SHOE), i.e., at angles close to 0° the effect of shadows between particles is almost negligible and the reflectance in the direction of 300 incidence increases (Hapke, 1986). The samples analysed were manually powdered and sieved, 301

302 reducing the distribution of grain sizes. Thanks to the accurate sample preparation the roughness is at the level of a few grain sizes. However, the SHOE effect is affected by the intrinsic roughness of 303 powder that creates shadowing of one particle over another (Pommerol and Schmitt, 2008a; Potin 304 et al., 2019). Mascagnite and salammoniac present relatively similar behaviours of the BRDF and 305 show an increase in reflectivity towards high phase angles (for $i \ge 30^{\circ}$ and $e \ge 60^{\circ}$) especially for 306 307 absorption bands over 2 µm. In fact, smaller particles compared to those with larger dimensions, 308 producing a larger number of reflections at their interface, have more pronounced forward scattering than larger particles, except when very small particles form fluffy aggregates, which is not 309 310 the case of our samples which are slightly packed to get a flat surface. Due to these phenomena, the two samples studied show a non-Lambertian behaviour. 311

- 312 The reflectance values variations for the analysed bands are reported in table 3 and 4 for mascagnite and salammoniac, respectively. For the mascagnite sample, the weak bands located at 1.06 and 1.28 313 μ m have slightly decreased reflectance values, for *i* = 0° and 30° and large emergence angles, *e* \geq 314 $\pm 40^{\circ}$, from 0.87 (*i* = 0°; *e* = 20°) to 0.75 (*i* = 0°; *e* = -70°) (Figures 6 a-b); The moderately strong spectral 315 feature at 1.58 µm, shows almost no change in reflectance with emergence angle for nadir 316 illumination (i = 0°), but at i = 30° the reflectance starts to increase at high phase angle (> 70°, i.e., e 317 < -40°, Figure 6 c). This trend is growing with the strong band at 2.18 µm and the very strong one at 318 319 3.01 μ m which show an increase in the reflectance values for increasing phase angles at both i = 0° 320 and 30°, a behaviour opposite to that of the 1.06 and 1.28 µm features. Furthermore, at high incidence angle (i = 60°) all bands display a clear reflectance increase at both low (g \leq 20°, i.e., e = 321 40 - 70°) and high phase angles (g > 100°, i.e., e < -40°). 322
- At low and high phase angles the relative magnitude of the increase in reflectance (relative to values at medium phase angle) rapidly grows with the absolute band strength. For the strongest band around 3 μ m we can observe the highest increase of the reflectance values, by a factor more than 3 at low phase angle, and by more than an order of magnitude at high phase angles. On the other hand the weakest 1.06 and 1.28 μ m features display only moderate increase in band center reflectance values, by factors of less than 1.3.
- The BRDF spectra of salammoniac sample exhibit similar behaviour to mascagnite. The spectral features at 1.09, 1.58 and 1.62 μ m are those characterized by less variability, in terms of reflectance reduction, with the phase angle variation (Figures 7 a-b-c). These spectral features show the maximum reflectance value for *i* = 60° and *e* = 70°. The bands located at 2.06, 2.2 and 3.05 μ m have increased reflectance values starting from *i* = 30° and *e* ≥ 40° (Figures 7 d-e-f). Maximum reflectance
- values are obtained for observation geometry configurations corresponding to $i = 30^{\circ}$ and $e = 70^{\circ}$.

Table 3: Reflectance behaviour of the analysed bands as a function of incidence (*i*) and emergence

336 (e) angles for the mascagnite sample. The arrows indicate the increase (\uparrow) or decrease (\downarrow)

variations with respect to the $e = 0^{\circ}$ angle. The symbol (-) indicates no variations.

Bands position	i = 0°								
(μm)	-70°	-60°	-40°	-20°	0°	20°	40°	60°	70°
1.06	\downarrow	\downarrow	\downarrow	-		-	\downarrow	\downarrow	\downarrow
1.28	\downarrow	\downarrow	\downarrow	-		-	\downarrow	\downarrow	\downarrow
1.57	\uparrow	\uparrow	\downarrow	-		-	\downarrow	\downarrow	\downarrow
2.18	\uparrow	\uparrow	\uparrow	-		-	\uparrow	\uparrow	\uparrow
3.01	\uparrow	\uparrow	\downarrow	-		-	\downarrow	\uparrow	\uparrow
	i = 30°								
1.06	\downarrow	\downarrow	\uparrow	\uparrow		\uparrow	\uparrow	\downarrow	\downarrow
1.28	\downarrow	\downarrow	\uparrow	-		\uparrow	\uparrow	\downarrow	\downarrow
1.57	\uparrow	\uparrow	\uparrow	-		\uparrow	\uparrow	\downarrow	\downarrow
2.18	\uparrow	\uparrow	\uparrow	-		\uparrow	\uparrow	\uparrow	\uparrow
3.01	\uparrow	\uparrow	\uparrow	\uparrow		\uparrow	\uparrow	-	\uparrow
	i = 60°								
1.06	\uparrow	\uparrow	\uparrow	\checkmark		\downarrow	\uparrow		\uparrow
1.28	\uparrow	\uparrow	\uparrow	\downarrow		\downarrow	\uparrow		\uparrow
1.57	\uparrow	\uparrow	\uparrow	\uparrow		\downarrow	\uparrow		\uparrow
2.18	\uparrow	\uparrow	\uparrow	\uparrow		\uparrow	\uparrow		\uparrow
3.01	\uparrow	\uparrow	\downarrow	\uparrow		\downarrow	\uparrow		\uparrow

338

Table 4: Reflectance behaviour of the analysed bands as a function of incidence (i) and emergence (e) angles for the salammoniac sample. The arrows indicate the increase (\uparrow) or decrease (\downarrow)

variations with respect to the $e = 0^{\circ}$ angle. The symbol (-) indicates no variations.

Bands position (μm)	i = 0°								
	-70°	-60°	-40°	-20°	0°	20°	40°	60°	70°
1.09	\downarrow	\downarrow	\downarrow	-		-	\downarrow	\downarrow	\downarrow
1.32	\downarrow	\downarrow	-	-		-	\downarrow	\downarrow	\downarrow
1.62	\uparrow	\uparrow	\downarrow	-		-	\downarrow	\uparrow	-
2.06	\uparrow	\uparrow	\downarrow	-		-	\downarrow	\uparrow	\uparrow
2.2	\uparrow	\uparrow	\uparrow	-		-	\uparrow	\uparrow	\uparrow
3.05	\uparrow	\uparrow	\uparrow	-		-	\uparrow	\uparrow	\uparrow

	i = 30°								
1.09	\downarrow	\downarrow	\downarrow	-		\uparrow	\uparrow	-	\downarrow
1.32	\downarrow	\downarrow	-	-		\uparrow	\uparrow	\downarrow	\downarrow
1.62	\uparrow	\uparrow	-	\downarrow		\uparrow	\uparrow	\uparrow	\uparrow
2.06	\uparrow	\uparrow	\uparrow	-		\uparrow	\uparrow	\uparrow	\uparrow
2.2	\uparrow	\uparrow	\uparrow	-		\uparrow	\uparrow	\downarrow	-
3.05	\uparrow	\uparrow	\uparrow	\uparrow		\uparrow	\uparrow	\uparrow	\uparrow
				i =	60°				
1.09	\downarrow	\uparrow	\uparrow	\uparrow		\uparrow	\uparrow		\uparrow
1.32	\downarrow	\uparrow	\uparrow	\uparrow		\uparrow	\uparrow		\uparrow
1.62	\uparrow	\uparrow	\uparrow	\uparrow		\uparrow	\uparrow		\uparrow
2.06	\uparrow	\uparrow	\uparrow	\uparrow		\uparrow	\uparrow		\uparrow
2.2	\uparrow	\uparrow	\uparrow	\uparrow		\uparrow	\uparrow		\uparrow
3.05	\uparrow	\uparrow	\uparrow	\uparrow		\uparrow	\uparrow		\uparrow



Fig.6. Polar plots in the principal plane of bidirectional reflectance distribution function for mascagnite- $(NH_4)_2SO_4$ sample. Colored symbols and lines refer to each incidence angle: blue squares ($i = 0^\circ$), red circles ($i = 30^\circ$) and green triangles ($i = 60^\circ$). Positive emergence angles correspond to azimuth = 0° (back-scattering quadrant) and negative values to azimuth = 180° (forward-scattering quadrant). Angles are given in degrees. Note the different scales of the axis. Panels a, b, c, d and e correspond to 1.06, 1.28, 1.58, 2.18 and 3.01 µm wavelengths, respectively.



Fig. 7: Polar plots in the principal plane of bidirectional reflectance distribution function for mascagnite- $(NH_4)_2SO_4$ sample. Colored symbols and lines refer to each incidence angle: blue squares (*i* = 0°), red circles (*i* = 30°) and green triangles (*i* = 60°). Positive emergence angles correspond to azimuth = 0° (back-scattering quadrant) and negative values to azimuth = 180° (forward-scattering quadrant). Angles are given in degrees. Note the different scales of the axis. Panels a, b, c, d, e and f correspond to 1.09, 1.32, 1.62, 2.06, 2.2 and 3.05 µm wavelengths, respectively.

357 3.2 Effect of observation geometry on bands parameters

350

For the two analysed samples, the bands depth, area and FWHM were analysed following Comodi et al. (2021) as reported in the methods section (section 2). The bands area, depth and FWHM of mascagnite sample are reported in Figures 8, 9, and 10, respectively, whereas the bands area, depth and FWHM of salammoniac sample are given in Figures 11, 12 and 13, respectively. The absorption features detected for the mascagnite and salammoniac samples are consistent with previouspublished spectra (e.g., Berg et al., 2016).

364 The collected data show that the band depth and area are strongly influenced by the viewing angle (Figures 8, 9 and 11, 12). On the contrary, the position of the bands remains unchanged. Similar 365 works (e.g.; Beck et al., 2012; Pommerol and Schmitt, 2008b; Potin et al., 2019) report that variation 366 in the observation geometry corresponds to a decrease in band depth at higher phase angles due 367 to the increase in photons scattered out of the sample before penetrating a few grains deep below 368 the surface. Our measurements show a decrease in depth values with increasing incidence and 369 emergence angles as reported in literature for $e \ge 40^\circ$. However, an opposite trend can be observed 370 371 for the absorption bands located over 2 µm. For both mascagnite (Figures 8 and 9) and salammoniac 372 samples (Figures 11 and 12), for $i = 60^{\circ}$ and (i) and (e) angles greater than 40°, the band depth and consequently the area tend to increase their values. This is particularly marked on the $\sim 2.2 (v_3 + v_4)$ 373 374 and ~3.0 (v_3) μ m features in both minerals.

Regarding the 3 μ m band, it exhibits a broad and shallow shape for variable emergence angles collected with *i*=0°. This spectral feature reveals a fine structure with increasing incidence and emergence angles. For incidence angle equal to 60° and emergence angles \geq 60°, the 3 μ m feature is well defined and clearly detectable (Figures 6e and 7f).

Finally, the FWHM parameter shows no clear trend with the change in observation geometry
(Figures 10 and 13). The absence of a possible trend for this band parameter was also found when
varying other physical parameters such as temperature and particle size (e.g.; De Angelis et al., 2021,
2019). However, our data for the 2.18 µm bands of mascagnite sample (Fig. 7d) and the 1.62 µm
bands of salammoniac (Fig. 10c) display clear trends with changing viewing geometry.



Fig. 8: Band area of the 1.06 (a), 1.28 (b), 1.58 (c), 2.18 (d) and 3.01 (e) μm absorption features of
mascagnite sample for different incident and emergence angles. Note the different scales of the y
axis.



390

Fig. 9: Band depth of the 1.06 (a), 1.28 (b), 1.58 (c), 2.18 (d) and 3.01 (e) μm absorption features of
mascagnite sample for different incident and emergence angles. Note the different scales of the y
axis.



Fig.10: Band FWHM of the 1.06 (a), 1.28 (b), 1.58 (c), 2.18 (d) and 3.01 (e) μm absorption features
of mascagnite sample for different incident and emergence angles. Note the different scales of the
y axis.



398

Fig.11: Band area of the 1.09 (a), 1.32 (b), 1.62 (c), 2.06 (d), 2.2 (e) and 3.05 (f) μm absorption
features of salammoniac sample for different incident and emergence angles. Note the different
scales of the y axis.



402

Fig. 12: Band depth of the 1.09 (a), 1.32 (b), 1.62 (c), 2.06 (d), 2.2 (e) and 3.05 (f) μm absorption
features of salammoniac sample for different incident and emergence angles geometry. Note the
different scale of the y axis.



407 Fig.13: Band FWHM of the 1.09 (a), 1.32 (b), 1.62 (c), 2.06 (d), 2.2 (e) and 3.05 (f) μ m absorption

408 features of salammoniac sample for different incident and emergence angles. Note the different

409 scales of the y axis.

410 3.3 Spectral slope vs geometry

For the quantitative determination of the slope parameter, the methodology reported in Fornasier 411 412 et al. (2015) was followed. The spectral slope can be computed as the ratio of two reflectance values given at two different wavelengths (Schröder et al., 2014). For each collected spectrum, three slopes 413 414 were determined, taking into account the reflectance values located at 1.15 - 1.4 μ m, 1.4 – 1.76 μ m and 1.76 – 2.46 µm; these three regions were chosen as no absorption feature occurs within them. 415 The value of the normalized spectral slope is calculated as the ratio between the reflectance values 416 at 1.4 μ m (1) and at 1.15 μ m (2) here reported as an example. The same operation was made for 417 the other two slopes: 418

419 *norm.slope* (%
$$\mu$$
m⁻¹) = 2 * $\frac{(R1-R2)}{[(R1+R2)*(\lambda 1-\lambda 2)]}$

420 where *R* and λ indicate the values of reflectance and wavelength (μ m), respectively.

421 The variation of the spectral slope and the emergence angles is reported in Figure 14 for both 422 salammoniac and mascagnite. The reflectance spectra of salammoniac (Figure 14a) have a similar 423 negative slope (blueing) to those of mascagnite (Figure 14b). At low emergence angle, mascagnite spectra have slope value of -0.53 %μm⁻¹, whereas salammoniac displays slope value of -0.77 %μm⁻ 424 ¹. At incidence (*i* = 0) and emerges ($e = \pm 70^{\circ}$) angles, the slope of mascagnite sample reaches the 425 426 mean value of -0.30 %µm⁻¹ compared to -0.45 %µm⁻¹ for the salammoniac spectra. Generally, the 427 blueing effect is linear with increasing angles of incidence and emergence. The spectral slope is 428 strongly affected by the surface particle density and texture (Binzel et al., 2015) as well as by the 429 different grain size distribution and chemical composition (De Angelis et al., 2014; Pisello et al., 430 2022a, 2022b). Regarding the variation of the slope with different grain sizes, we expect a change from positive (red) to negative (blue) values as the grain size increases. 431





Fig.14: Effect of the geometry variation on the spectral slope of the reflectance spectra of mascagnite
(left column a-c-e) and salammoniac samples (right column b-d-f).

435 4. Discussions

436 4.1 Quantification of viewing geometry variations

437 To numerically quantify the effect of the emergence angle variations on the measured reflectance

438 spectra, we calculated the difference between reflectance with $e = 20^{\circ}$ with respect to the other

439 values of *e*. For the selected bands analysed in the previous sections, the absolute value of $(R_e - R_e)$

- 440 $R_{e=20}$) was divided by $R_{e=20}$ to express the ratio of deviation with respect to the reference geometry
- observation ($i=0^{\circ}$ and $e=20^{\circ}$). Each deviation value for mascagnite e salammoniac samples is plotted

in Figures 15 and 16, respectively, for $i=0^\circ$; 30°; 60°.



Mascagnite

Fig. 15: Histogram reports the deviation ratio relative to the spectrum collected with reference geometry (i=0° and e=20°), for incidence angles equal to 0° (a); 30° (b) and 60° (c) and emergence angles in the \pm 70° range for the mascagnite sample. Note the different scales.



Fig.16: *Histogram reports the deviation ratio from relative to the spectrum collected with* reference geometry (i=0° and e=20°), for incidence angles equal to 0° (a); 30° (b) and 60° (c) and emergence

450 angles in the \pm 70° range for the salammoniac sample. Note the different scales.

In general, the relative change in reflectance values is already evident for small emergence angle variations ($e = \pm 20^\circ$). For $e \ge 40^\circ$, the spectral variation becomes significant and much more appreciable as shown in Figures 6 and 7. At geometric configurations with $i = 60^\circ$ and $e \ge 40^\circ$ we can notice that the reflectance spectra of the inclined geometries tend to diverge from the one measured in nadir geometry.

456 Mascagnite sample displays a moderate monotonic increase of reflectance values, when varying the 457 emergence angles for the NH₄⁺ bands located at 1.06 and 1.28 μ m with the highest values for *e* = -458 70°. The bands at 2.18 and 3.01 μ m display an abrupt change when *e* \ge 60°. These behaviours are 459 the same for different incidence angles but much more marked for configurations with *i*= 60°. In 460 fact, the 1.06 and 1.28 μ m features with *i*=60° more drastically change their reflectance values 451 (Figure 15). Salammoniac bands at 1.09 and 1.32 μ m increase reflectance values reaching the 452 highest ones for *e* \ge 60° and *i* = 60°, whereas for *i* = 30° these two features do not show severe 463 changes (Figure 16). Regarding the bands above 2 μ m, they are the most sensitive to variations in 464 viewing geometries with a strong increase for $e \ge 40^\circ$, especially also for $i = 60^\circ$ (Figure 16).

465 The histograms in Figures 15 and 16 also show which geometric configurations and bands have the greatest variability. It is evident that the bands located at 1.58 and 1.62 μ m, attributed to 2v₃ NH₄⁺ 466 467 vibrational modes, are the less sensitive to changes in geometry. Furthermore, when increasing the 468 incidence angle, the deviation drastically increases (Figure 15 and 16 c). The change in observation geometry affects all the selected features, in particular the bands located at 2.18 ($v_3 + v_4$) and 3.01 469 $(v_3) \mu m$ for mascagnite and 2.06 $(v_2 + v_3)$, 2.2 $(v_3 + v_4)$ and 3.05 $(v_3) \mu m$ for salammoniac. Finally, high 470 incidence and negative emergence angles (forward scattering quadrant) have the highest 471 472 reflectance deviation compared to positive equivalents (backward scattering quadrant) due to the 473 shallow penetration depth of the light in the forward grazing geometries. The highest deviation is confirmed for incidence angles \geq 30° and emergence angles \geq 60°. 474

The results obtained in this work show that observation geometry can have a strong impact on band
parameters, such as depth and area, in both ammonium-bearing minerals, in terms of shape of NH₄⁺
absoption features. Incidence and emergence angles changes affect absorption bands and spectral
slope. The different geometry configurations variously affect the continuum and absorption bands,

479 especially for high incidence and emergence angles.

480 Although in literature (Beck et al., 2012, 2011) the variation in observation geometry is followed by 481 a decrease in depth at larger phase angles, we notice a different trend related to the variations in 482 both emergence and incidence angles. Considering angles of incidence close to nadir (0° and 30°), a twofold trend can be observed as the angle of emergence changes. An initial increase in area and 483 484 depth values for small emergence angles up to about 20° and then its gradual decreases, especially for bands located above 2 μ m. On the other hand, for *i* = 60°, we notice that for bands located at 485 486 ~1.05, ~1.3, ~1.56 µm, depth and area generally decrease while the features at ~2.02, ~2.2 and ~3.0 487 μm have an opposite trend. This difference in behaviour between weak-medium bands and strong-488 to-nearly saturated bands is linked to the fact that at high incidence and phase angles the light 489 penetrates to a shallower depth than for near normal incidence and consequently escapes with less 490 absorption (smaller path through the medium). For the weak-medium bands it will thus increase 491 their reflectance at band centre but to a smaller extent the continuum wavelengths around the band which are already weakly absorbing and have high reflectance. This effect reduces the depth and 492 493 area of these bands with increasing phase angle. For strong-to-nearly saturated bands the reverse 494 occurs. The continuum wavelengths around the band, which have medium reflectance levels, will 495 significantly decrease their absorption and increase their reflectance, while the almost saturated 496 band centres see only a moderate increase of their reflectance. Consequently, the band contrast 497 ('relative' band depth and area) strongly increases when approaching grazing geometries. This 498 interpretation is thus linked both to the non-linear radiative transfer in particulate media and to the 499 way the individual band depth and area are defined, i.e. relative to their nearby reflectance maxima, called 'continuum', which have the tendency to have lower reflectance for strong bands than for 500 weaker ones. 501

502 Finally, to summarise the behaviour of the band area and depth parameters, we calculated the 503 relative percentage change from the reference observation configuration $i=0^{\circ}$ and $e=20^{\circ}$. In order 504 to provide a semi-quantitative relative percentage variation, we used the following normalisation:

- 505 $[(\Delta_{e} \Delta_{emin})/(\Delta_{emax} \Delta_{emin})]*100$
- 506 where,
- 507 $\Delta_e = X_0 X_e$, meaning X₀ the value of band parameters, either area or depth, at reference 508 observation configuration (*i*=0° and *e*=20°) and X_e the value of band parameters at each 509 emergence angle in the ±70° range.
- 510 Δ_{emax} , meaning the maximum values of Δ_e in the studied dataset.
- 511 Δ_{emin} , meaning the minimum values of Δ_e in the studied dataset.

512 This calculation was made for each absorption band analysed in this work. The results obtained show that the relative percentage change in area and depth, considering the angle of incidence i=0, 513 514 varies from ~10% (e=- 20°) to ~37% (e= ±70°). Variations are significant at angles of incidence equal 515 to 60° and emergence $\pm 70^{\circ}$ with an average variation of ~70%. Both mascagnite and salammoniac samples display single values of band area and depth percentage change of ~90% for i=0° and e=-516 517 70°. The main difference between the two samples, in terms of relative percentage change, was found for the *i*=0° configuration. In the mascagnite samples the percent variation is greater (from 518 519 ~20% at e = -20 to ~55% at $e = 70^{\circ}$) than salammoniac (from ~5% at e = -20 to ~20% at $e = 0^{\circ}$).

520 Our results indicate that variations of incidence and emergence angles generate changes in band 521 parameters and reflectance values with a band strength change non directly related to the 522 abundance of the presumed mineralogical phase. This supports the relevance of measuring the 523 BRDF of minerals and BRDF surfaces over wide wavelength ranges, both in the laboratory and for 524 the surfaces of Solar System objects.

525 4.2 Implications for icy bodies

526 The surface absorption feature of Ceres at 2.21 µm, in the remotly collected reflectance spectra, 527 has been attributed to the presence of NH₄⁺, in good agreement with absorption characteristics of the NH₄Cl mineral (De Sanctis et al., 2016). As reported in the present work, the 2.21 µm spectral 528 529 feature has a great variability in terms of depth and area under different viewing configurations, thus implying that the band parameters variations in the Ceres's spectra may be due to different 530 observation geometries during the mapping of the surface by the Dawn mission (Russell and 531 Raymond, 2012) and not only in relation to the abundance of the NH₄Cl mineral. In analogy to the 532 533 previous spectral features, the band at \sim 3.05 µm for the salammoniac sample is also relevant, 534 relatively to its variations, for the detection of NH4⁺. In fact, the shape of the 3.05 µm band varies 535 under different viewing configurations, becoming more defined at high incidence angles compared 536 to other studies on ammonium minerals that describe the 3.05 µm feature as a broad water-537 saturated band (e.g. Ehlmann et al., 2018; Ferrari et al., 2019).

Ammonium sulphate exhibits absorption characteristics located in the $3.0 - 3.3 \mu m$ spectral range, 538 consistent with the spectra of comet 67P as mapped by the Visible and InfraRed Thermal Imaging 539 540 Spectrometer, Mapping Channel (VIRTIS-M) instrument (Coradini et al., 2007). Poch et al. (2020) 541 propose the presence of ammonium salts (possibly including ammonium sulfate), mixed with 542 opaques, based on the well defined absoption bands at 3.1 and 3.3 µm. From the data we report in 543 the mascagnite sample, this specific spectral feature, has a great variability especially for high incidence and emergence angles which may affect its correct identification and quantification from 544 545 remote observations.

546 In the Virgil Fossae area, on Pluto, two possible ammonia bands at ~1.65 and 2.21 µm were identified (Dalle Ore et al., 2019) by the Linear Etalon Imaging Spectral Array (LEISA) onboard the 547 *New Horizons* mission (Schmitt et al., 2017). The 1.65 and 2.21 µm bands have been unambiguously 548 attributed, similarly to the findings for Ceres, to the possible presence of ammonium minerals such 549 as NH₄Cl related to recent cryovolcanism activity (Cruikshank et al., 2019). Furthermore, the surface 550 551 of Charon, the largest Pluto's moon, shows the same absorption characteristics at 2.21 µm, 552 attributed to hypothetical different ammonium minerals, in a non-univocal way given the similarity of the 2.21 µm band with that of ice. (Cook et al., 2018, 2007). The spectra here reported shown a 553 554 great variability of the 2.21 µm features, related to the different viewing geometries especially at 555 high emergence angle. The data collected in this work can help to understand if the NH₃ spectral 556 feature on these bodies is either related to ammoniated salts, in this case salammoniac or 557 mascagnite, or to other possible forms such as ice or other compounds. Recently, some features in the Charon's spectrum at 1.60 and 1.63 μm were dicovered for the first time, consistent with NH₄Cl 558 absorption bands. Our data might provide information on its abundance on this moon (Cook et al., 559 560 2023). Finally, the data presented in this paper can be usefull to interpret the spectra from the Near-Infrared Spectrograph (NIRSpec) (Rauscher et al., 2007) on board the James Webb Space Telescope 561 (JWST) of Trans Neptunian Object (TNO). 562

563 5. Conclusions

564 This study reported a systematic investigation of reflectance spectra in the near and visible infrared $(1-4.2 \mu m)$ varying the observation geometries of two natural analogues ammonium bearing 565 minerals, mascagnite [(NH₄)₂SO₄)] and salammoniac (NH₄Cl), with 32-80 μm powder granulometry 566 567 and at room temperature. These data are of potentially great interest for the detection of nitrogen 568 and ammonium compounds on the surface of icy objects. We analysed the effects of incidence and 569 emergence angles, with a series of three different angles of incidence ($i=0^\circ$, 30° and 60°) and nine emergence angles in the range $e = \pm 70^\circ$, on the spectral characteristics of NH₄⁺ absorption features 570 571 in order to provide laboratory data in relevant conditions to identify and quantify these minerals by 572 remote sensing analysis.

573 We have shown that the collected reflectance spectra are affected by the change in observation 574 geometries. We notice that divergence from nadir-like measurements becomes significant for e≥40°. Characteristic NH₄⁺ features are located at ~1.06, 1.3, 1.56, 2.02, 2.2 and 3.0 µm. Depth and 575 576 area parameters of these features display a dual behaviour: (i) for absorption bands located below 2 µm and for incidence angles close to the azimuth and low emergence angles, the parameter values 577 578 tend to increase, conversely, for high emergence angles, the values tend to decrease; (ii) spectral 579 features above 2 µm generally increase with different trends depending on the angle of incidence. 580 The greatest increase occurs for $i = 60^{\circ}$. Reflectance values generally increase with increasing angle 581 of incidence and angle of emission with maximum values for $i = 60^{\circ}$ and $e = \pm 70^{\circ}$ i.e. high phase 582 angles and very low phase angles (opposition effect). However, bands below two microns reduce 583 reflectance values increasing phase angles (g) especially for incidence angles equal to 0° and 30°.

584 The spectral features at ~3.0 μ m show the clearest variation in shape resulting in its desaturation 585 and corresponding better definition as geometry changes, important for remote sensing 586 identification of NH₄⁺ features. 587 Our results can have a direct aplication for differents objects inside the Solar System, especially 588 small icy bodies. On these bodies, large surface slopes and surfaces characterised by rough 589 topography are observed. The high emergence and incidence angles encountered during a probe 590 fly-by can produce a misinterpretation when using laboratory spectra measured with nadir viewing 591 geometries.

592 The use of laboratory data are needed to correctly deconvolve remote sensing hyperspectral data 593 of planetary surfaces. The analysis of the surfaces of icy bodies for traces of NH₄⁺ by current and 594 future missions will provide important knowledge about the abundance of nitrogen on these bodies, 595 their environment and their potential for the evolution of life. The laboratory spectra reported here 596 add information on the band dependance on viewing geometries of two selected ammonium 597 bearing minerals and should aid the remote data interpretation from *New Horizons*/LEISA and 598 *Dawn*/VIR spectrometers.

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605 On-line data

The complete sets of data (Fastelli et al. 2021b) presented in this paper are available online in the CSS database of the SSHADE database infrastructure (https://vwww.sshade.eu) [doi: 10.26302/SSHADE/EXPERIMENT BS 20231304 001]

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