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Analysis of Limiting Diffusion Processes within Li-ion Batteries





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Context & Objectives

Objectives

To meet the ever-growing requirements of the transport sector, improvement of current Li-ion systems is needed in terms of energy density, safety, cycle life, and costs. Among the different strategies to increase battery autonomy, one focuses on electrode loading to increase the ratio of active materials to inactive components. However, by doing so, the output power density becomes strongly limited by the charge transport within the composite electrodes. ⁽¹⁾⁽²⁾

Study of battery power performance with respect to electrode design (loading, porosity...) as well as temperature and ageing

Determination of the limiting current density (J_{lim}) and effective limiting process of power using an electrochemical approach

Optimization of the battery design to maximize the energy/power



LFP based electrode : 1) Porosity effect



- Lower porosity results in higher J_{lim}, and higher operating C-rates
- ➔ Calendering enhances contact particleparticle and particle-current collector

Porosity (%)	53	22
J _{lim} (mA.cm⁻²)	0,7	5
D _{eff} (cm ² .s ⁻¹)	2,2.10 ⁻¹⁵	1,7.10 ⁻¹⁴

2) Loading effect



- Thicker is the electrode; lower is J_{lim}
- →At high loading, the limiting process is probably the diffusion within the porous electrode

Loading (mg.cm ²)	5	19
J _{lim} (mA.cm⁻²)	5	2,5
D _{eff} AM limitation (cm ² .s ⁻¹)	1,7.10 ⁻¹⁴	2,5.10 ⁻¹⁵
D _{eff} electrolyte limitation (cm ² .s ⁻¹)	5,2.10 ⁻⁷	1,1.10-6

3) Temperature effect



NMC based electrode : 1) Loading effect

2) SEM images

Conclusion



