



## Study of limiting factors of power performance within Li-ion batteries

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# Study of Limiting Factors of Power Performance within Li-ion Batteries

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## Context & Objectives

To meet the ever-growing requirements of the transport sector, improvement of current Li-ion systems is needed in terms of energy density, safety, cycle life, and costs. Among the different strategies to increase battery autonomy, one focuses on electrode loading to increase the ratio of active materials to inactive components. However, by doing so, the output power density becomes strongly limited by the charge transport within the composite electrodes. <sup>(1)(2)</sup>

- Objectives
  - Study of battery power performance with respect to **electrode design (loading, porosity...)** as well as **temperature and ageing**
  - Determination of the **limiting current density ( $J_{lim}$ )** and **effective limiting process** of power using an electrochemical approach
  - Optimization of the battery design to **maximize the energy/power**

## Experimental

### Positive electrode formulation

Active Materials (AM) : **LFP & NMC-811**

| Electrode        | Composition   | Loading (mg.cm <sup>-2</sup> ) | Theoretical Capacity (mAh.cm <sup>-2</sup> ) | Porosity (%) | Temperature (°C) |
|------------------|---------------|--------------------------------|--|--------------|------------------|
| LFP / C65 / PVDF | 80/ 10 / 10   | 2 - 17                         | 0.4 - 2.7                                    | 50 - 23      | 5°C - 40°C       |
|                  | 85/ 7.5 / 7.5 |                                |  |              |                  |
|                  | 90/ 5 / 5     |                                |  |              |                  |
| NMC / C65 / PVDF | 90/ 5 / 5     | 2 - 17                         | 0.5 - 3.4                                    | 50 - 20      | 5°C - 40°C       |

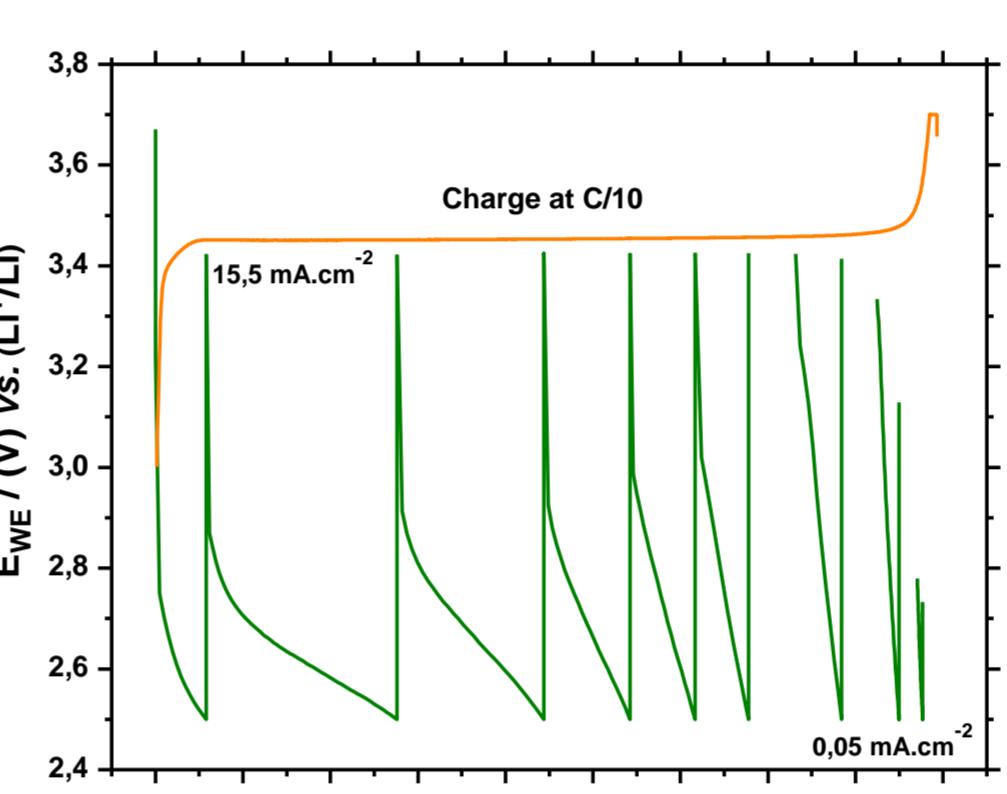
### Cell Assembly

Coin cell CR2032

Electrode //LP30//Li

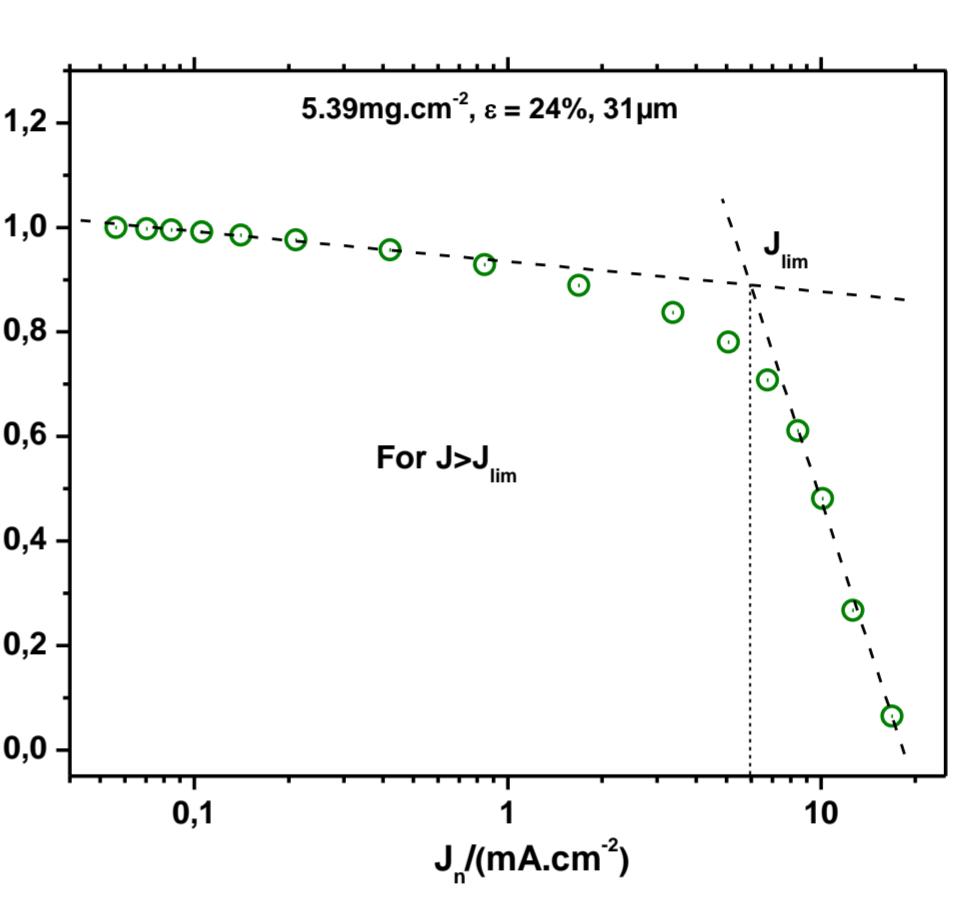


### Cycling protocol <sup>(3)(4)</sup>



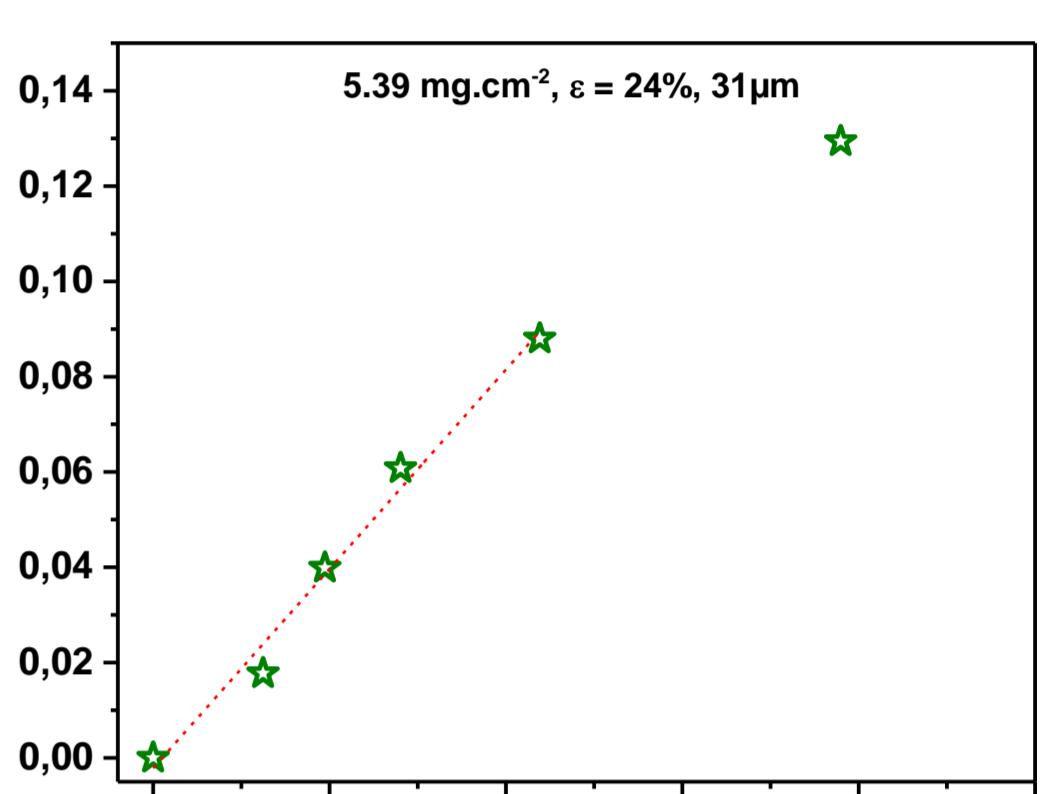
## Methodology

### Power performance



- Time-saving methodology
- One charge, followed by successive discharges at ever-decreasing C-rates

### Sand Plot



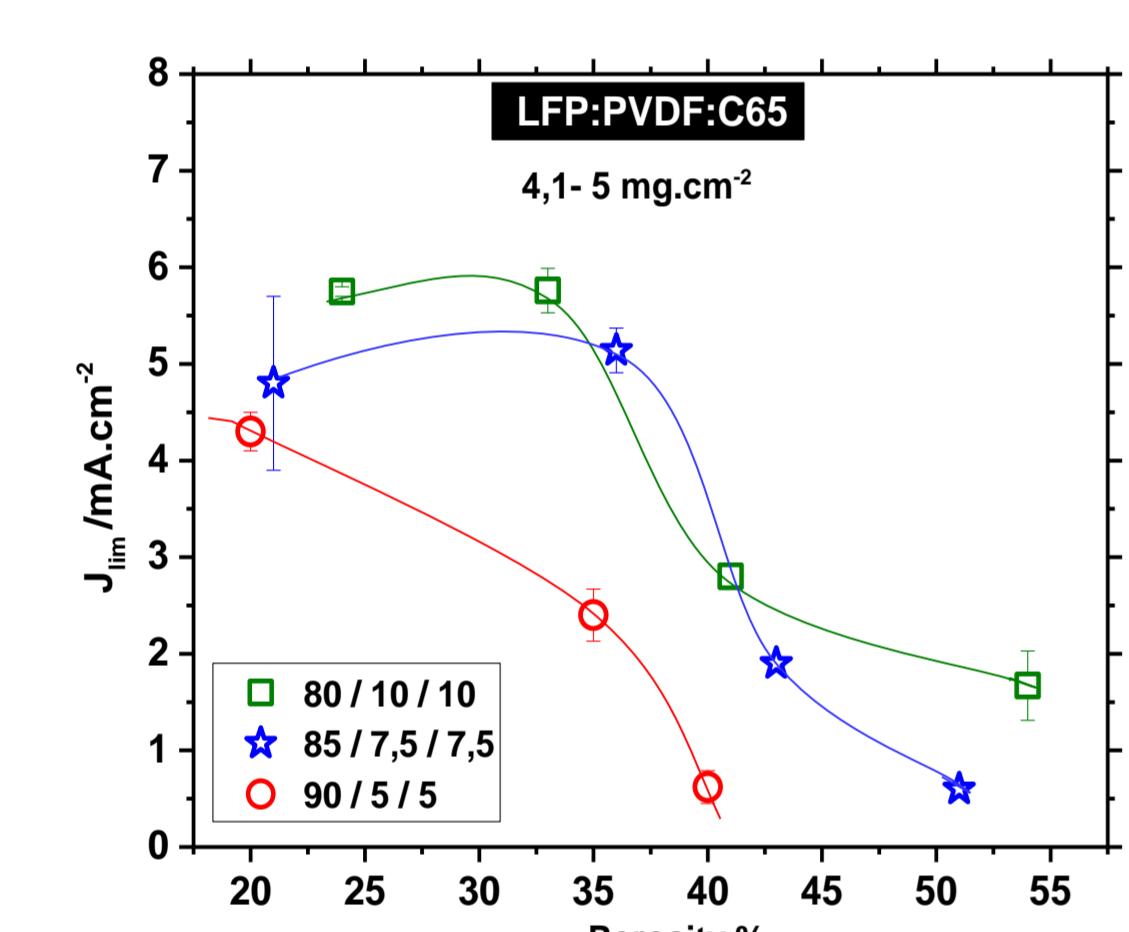
### Sand Equation <sup>(4)(5)</sup>

$$t_{\text{Sand}} = \pi \cdot D_{\text{eff}} \cdot \frac{n \cdot S \cdot F \cdot C_{\text{max}}}{(2 \cdot * J)^2}$$

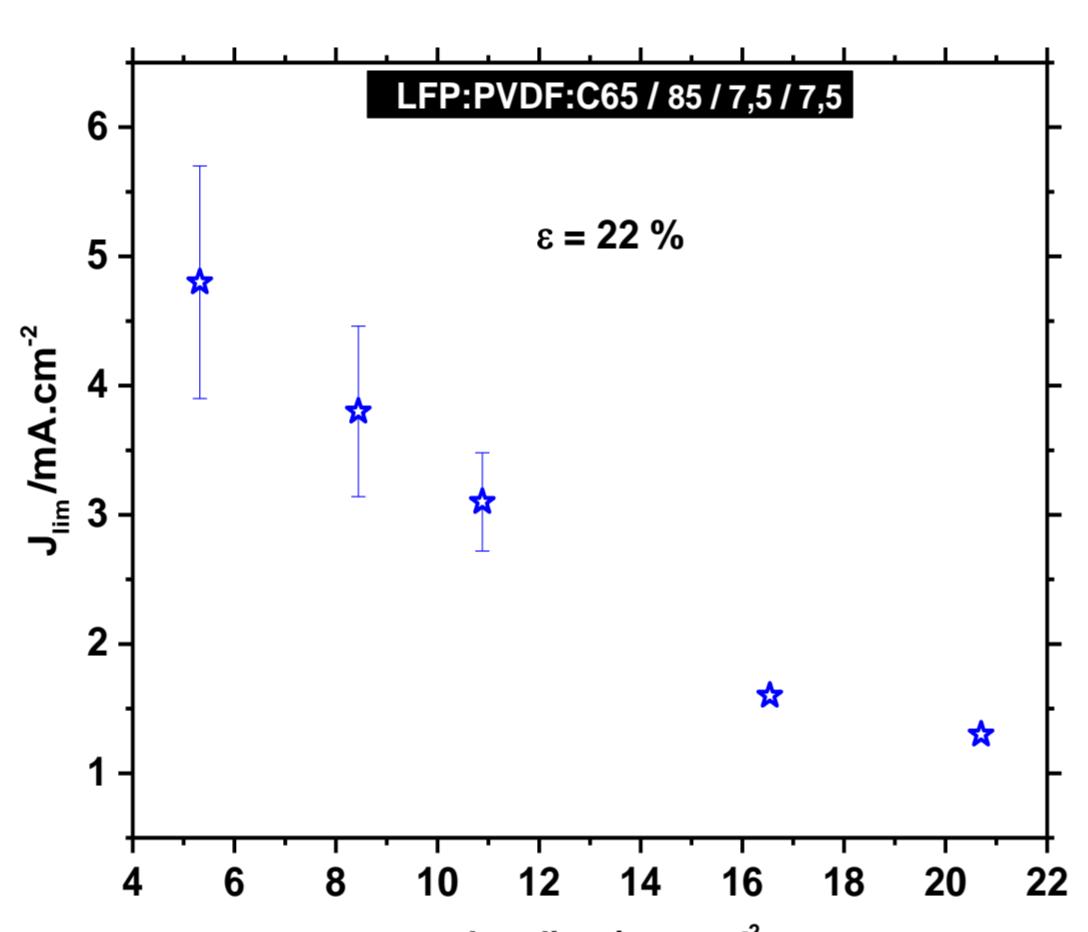
Quick access to the effective diffusion coefficient

## LFP based electrode

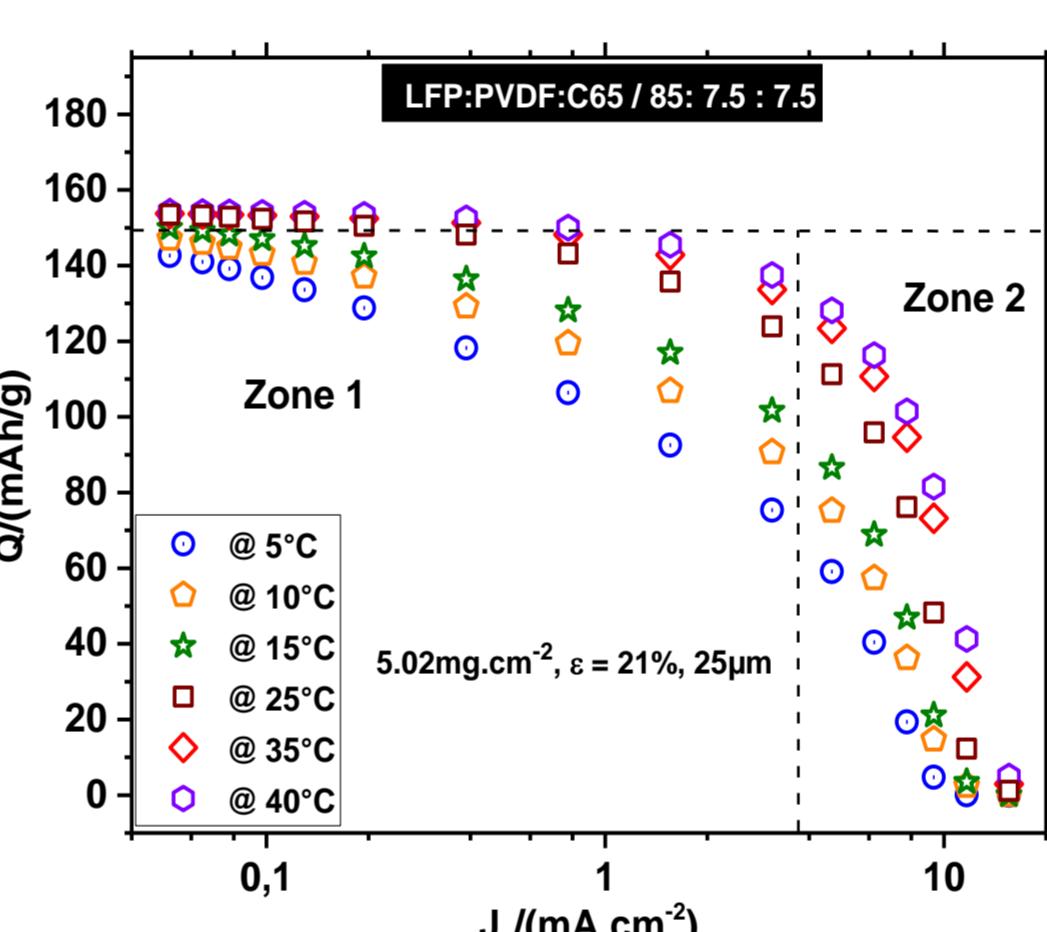
### 1) Porosity effect on $J_{lim}$



### 2) Loading effect



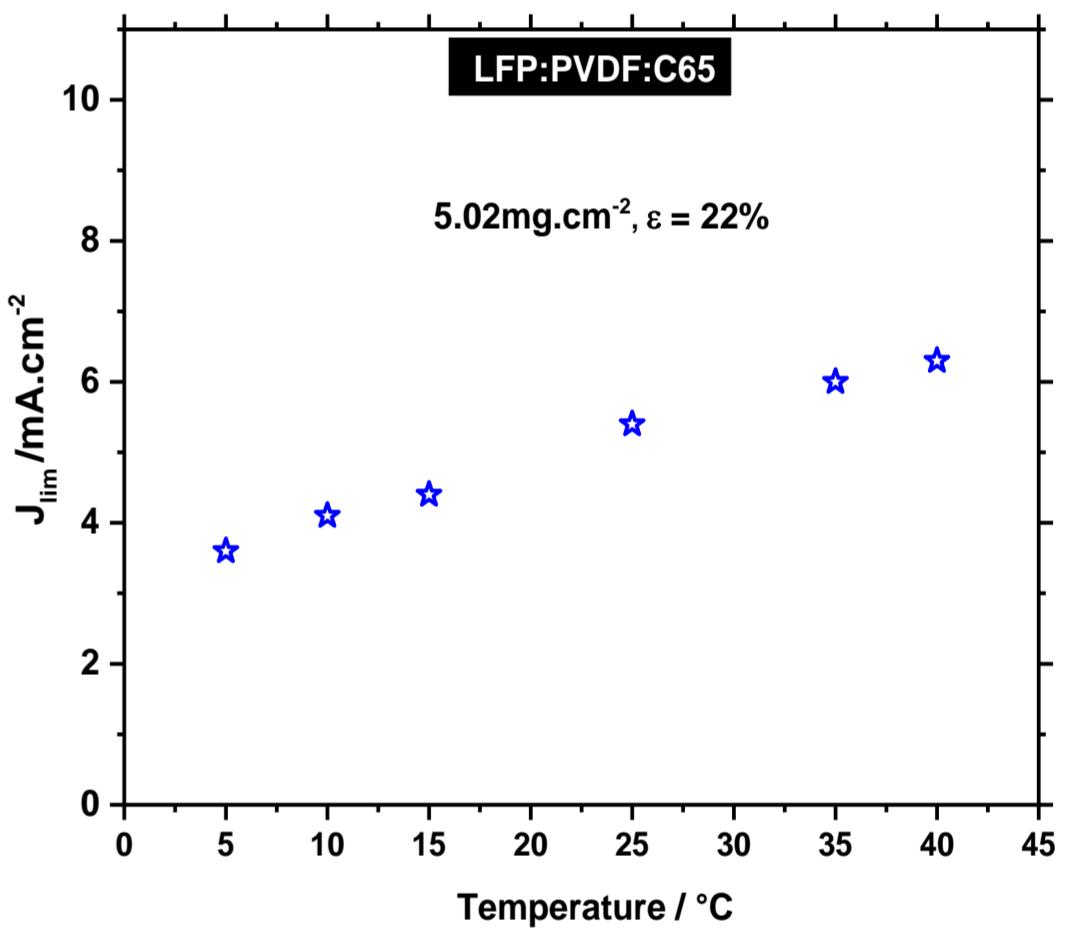
### 3) Temperature effect



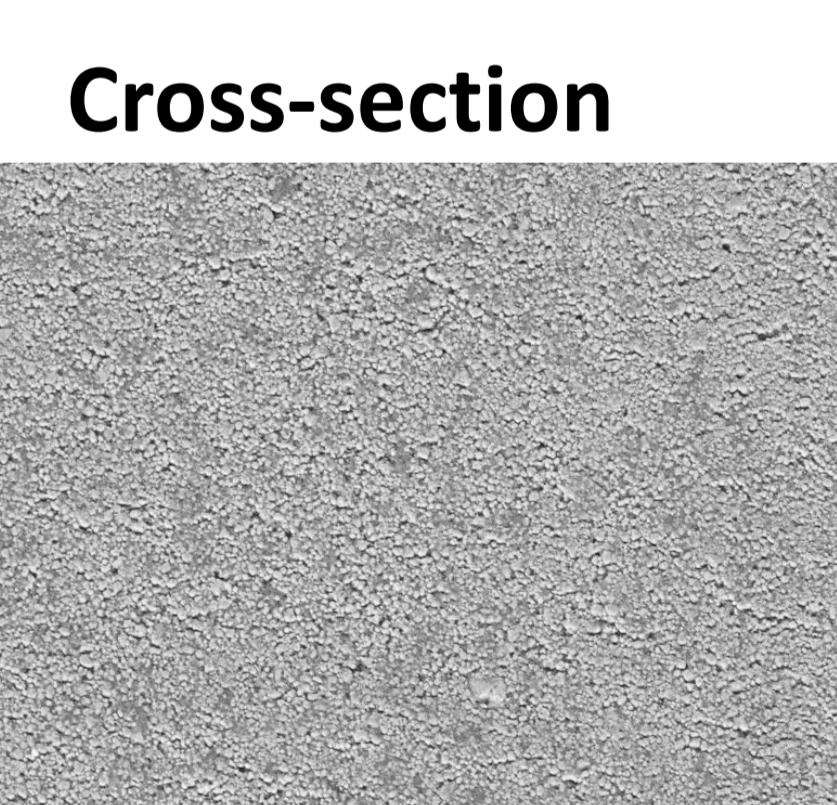
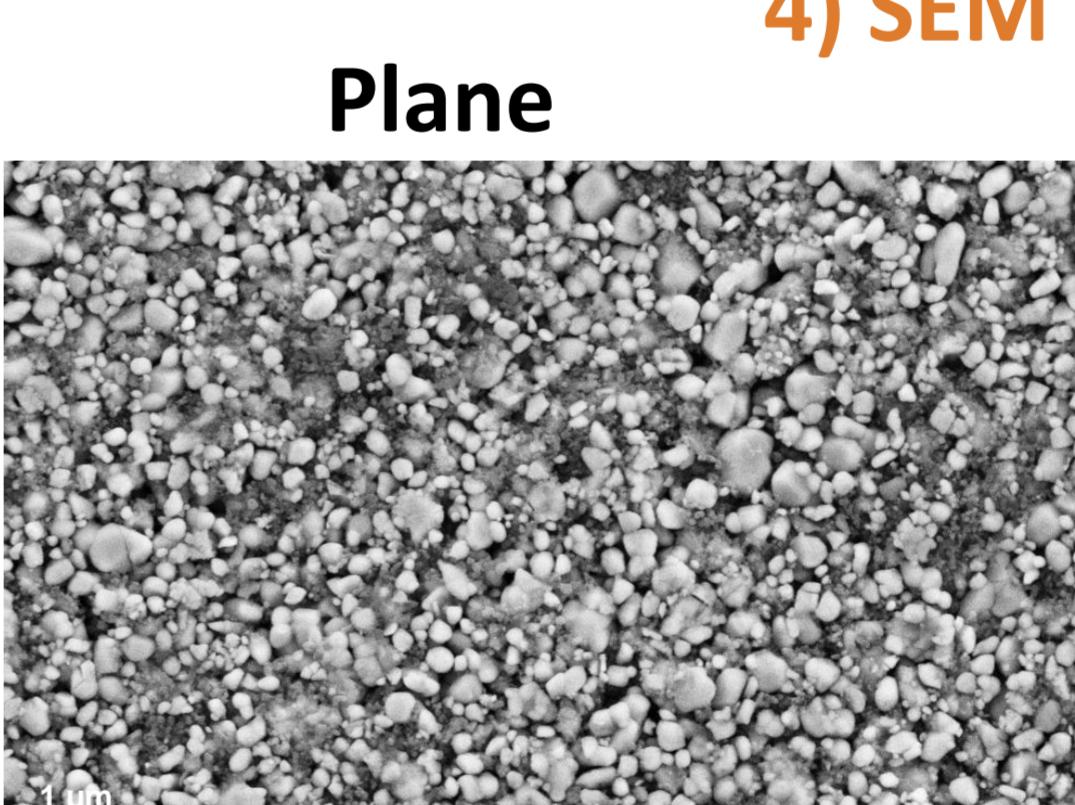
- For  $J < J_{lim}$  (zone1), the slope  $Q$  vs  $J$  increases when  $T$  decreases
- The variation of  $J_{lim}$  with  $T$  shows **linear behaviour**

### Activation Energy

Ea (5-40°C) 0.2 eV

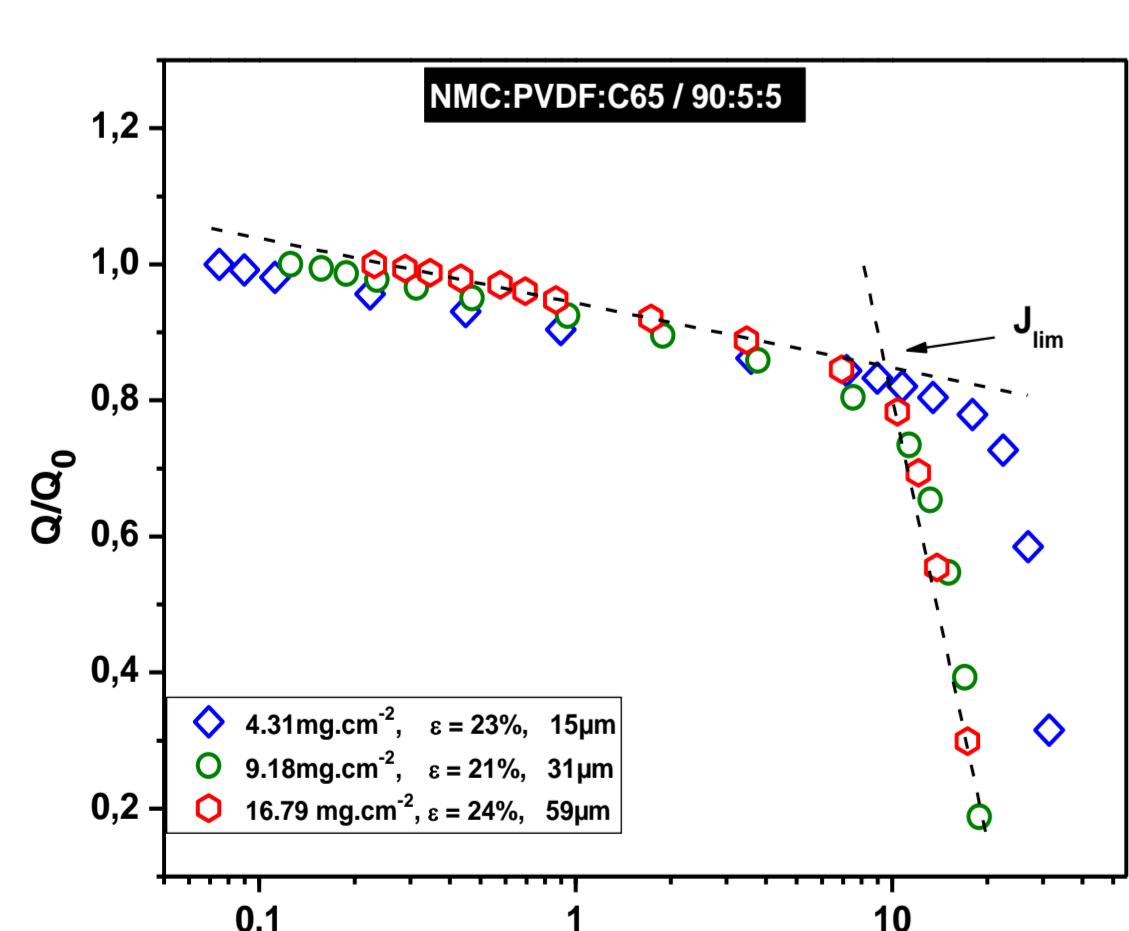


### 4) SEM images

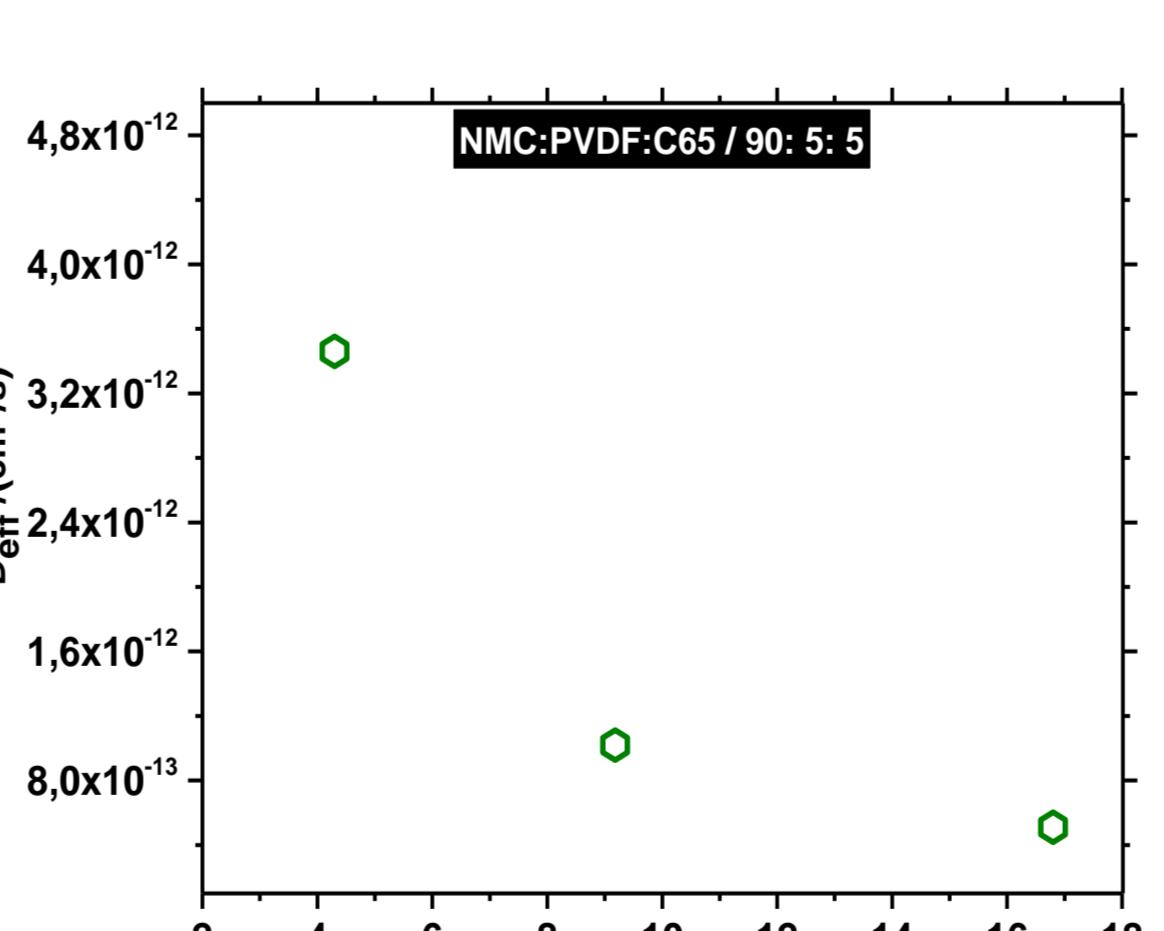
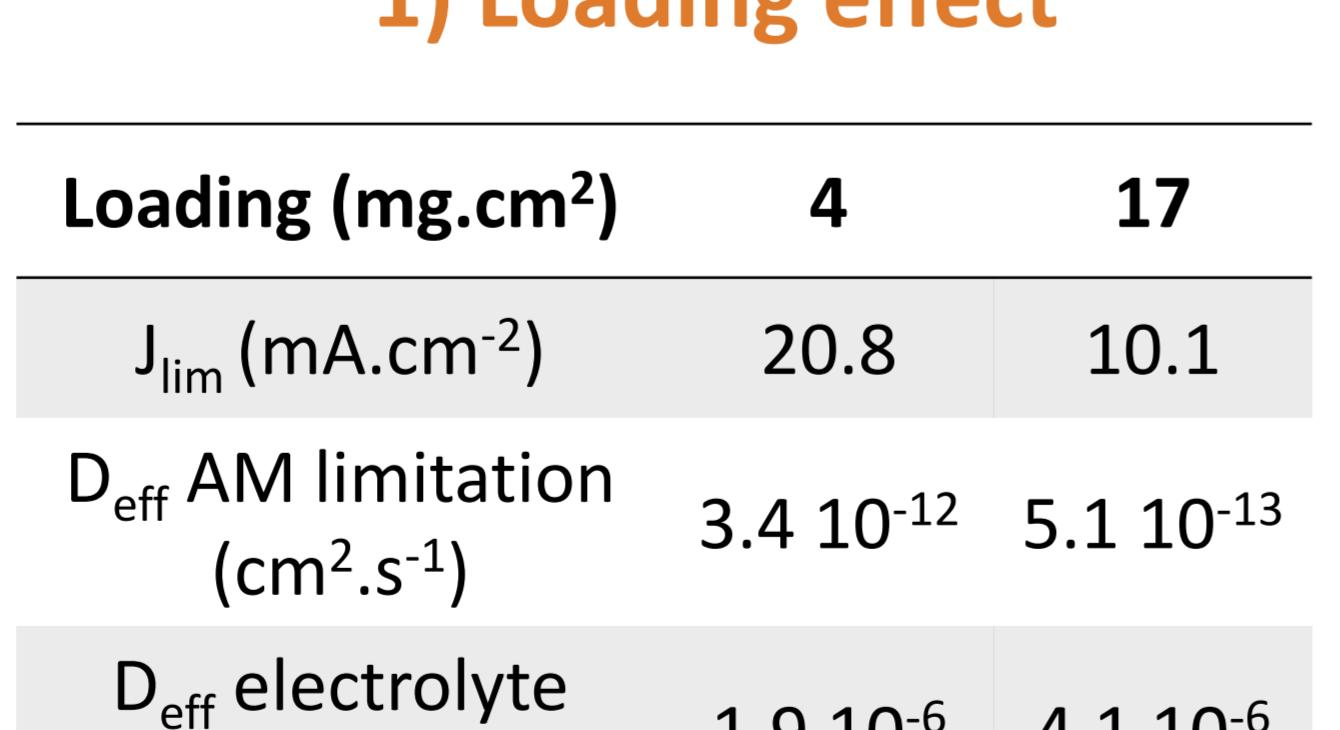


LFP electrode at 20% porosity

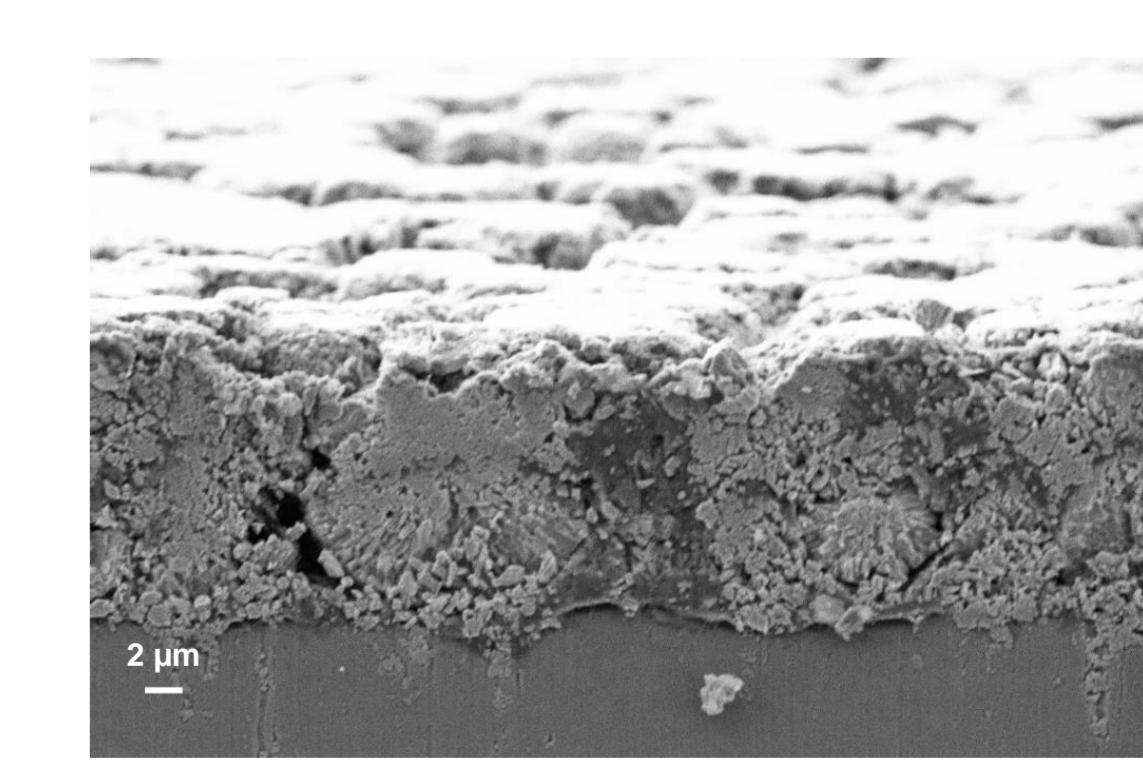
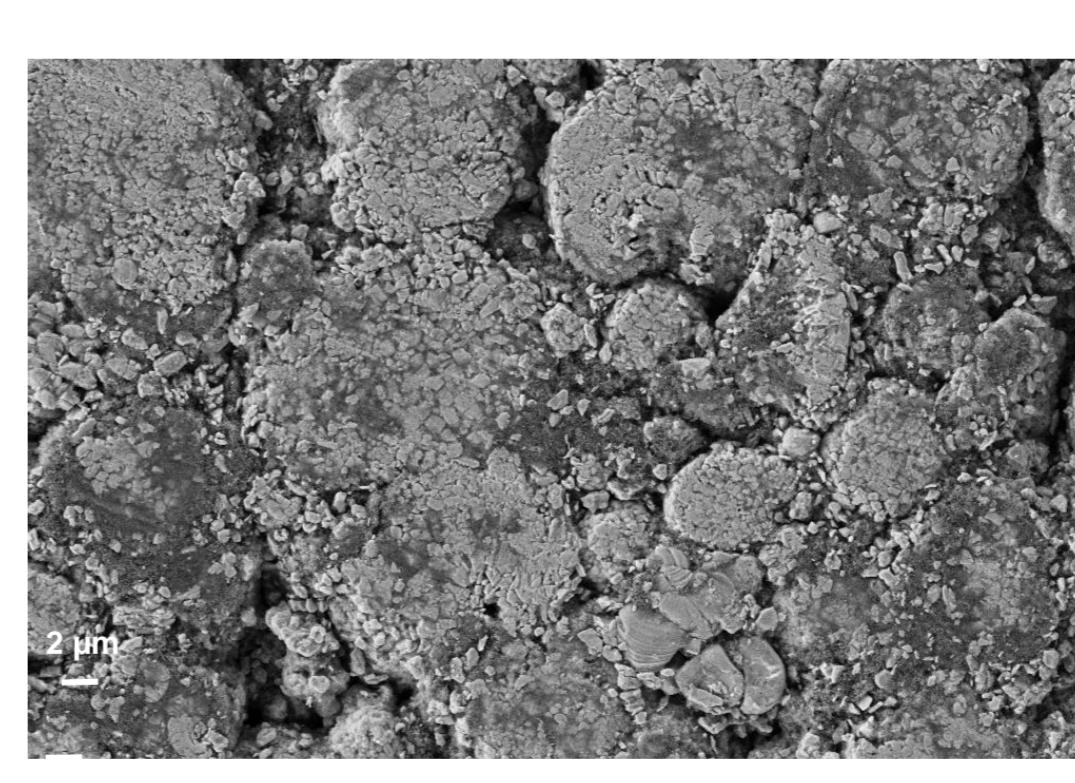
## NMC based electrode



### 1) Loading effect



### 2) SEM images



NMC electrode at 20% porosity

## Conclusion

- Quick determination of limiting current density as well as limiting process of power.
- Electrode design (Porosity, loading, microstructure) has a substantial impact on the power performance of the battery
- Figure out a physical interpretation of the impact of design parameters on power performance
- An optimum of all parameters has to be found to maximize power performance

## References

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