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High accuracy spectroscopy of H_2 rovibrational transitions in the (2-0) band near 1.2 μ m

H. Fleurbaey¹, A.O. Koroleva^{1,2}, S. Kassi¹, A. Campargue^{1,*}

¹ Univ. Grenoble Alpes, CNRS, LIPhy, Grenoble, France ² Institute of Applied Physics of RAS, Nizhniy Novgorod, Russia

Accurate transition frequencies of six lines of the (2-0) vibrational band of H₂ are measured near 13 14 1.2 µm, namely Q1-Q4, S0 and S1. These weak electric-quadrupole transitions were measured at room 15 temperature by comb referenced cavity ring-down spectroscopy. Accurate transition frequencies were 16 determined applying a multi-spectrum fit procedure with various profile models including speed 17 dependent collisional broadening and shifting phenomena. Although none of the considered profiles 18 allows reproducing the shape of the strongest lines at the noise level, the zero-pressure line centers are 19 found mostly independent on the used profile. These values are the first H_2 (2-0) transition frequencies referenced to an absolute frequency standard. As a result, a 1σ -accuracy better than 100 kHz is 20 21 achieved for the O1, S0 and S1 transition frequencies, improving by three orders of magnitude the 22 accuracy of previous measurements. For the six measured transitions, the most recent calculated 23 frequencies are found systematically underestimated by about 2.58 MHz, about twice their claimed 24 uncertainties. The energy separation between the J=2 and J=0 rotational levels of the vibrational 25 ground state is derived from the Q2 and S0 transition frequencies and found within the 110 kHz 26 uncertainty of its theoretical value. The same level of agreement is achieved for the energy separation 27 between the J=3 and J=1 rotational levels obtained by difference of the Q3 and S1 transition 28 frequencies. The *ab initio* values of the intensity of the six transitions are validated within a few 29 thousandths.

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33 Introduction

34 Due to its simplicity, dihydrogen (H_2) and its isotopologues (HD and D_2) are quantum physics test systems for advanced theoretical calculations including nonadiabatic, relativistic, quantum 35 electrodynamics (QED) effects [1], [2], [3], and hyperfine structure calculations [4], [5], [6]. For 36 37 instance, the measurement of highly accurate rotation-vibration (RV) transition frequencies provide stringent tests of the level energies recently obtained by fully variational calculations including high 38 39 level QED corrections [1], [2], [3]. In recent years, significant progress has resulted from the fruitful 40 emulation between theory and experiment. Due to the weakness of the considered RV transitions, precision spectroscopy is experimentally challenging in H₂. As a homonuclear diatomic molecule, H₂ 41 42 has no electric dipole spectrum and its RV spectrum is formed by very weak electric-quadrupole vibrational bands, (V-0), of decreasing intensity (see Fig. 6 of Ref. [7]). For instance, the (1-0) 43 fundamental and (2-0) first overtone bands centered at about 4160 and 8090 cm⁻¹, have maximum line 44 intensity of 3.2×10^{-26} and 5.2×10^{-27} cm/molecule at 296 K, respectively. In the recent years, hydrogen 45 deuteride (HD) has been preferred for metrological measurements because HD shows stronger 46 47 absorption bands than H₂. This is due to the charge asymmetry and resulting weak dipole moment existing in this heteronuclear isotopologue giving rise to a small electric dipole moment. As a result, in 48 49 the (2-0) band, the electric dipole transitions in HD are about two orders of magnitude stronger than 50 electric quadrupole transitions in H₂ (and HD).

Accuracies better than 100 kHz level have been recently achieved by several groups for a few 51 52 (2-0) transition frequencies in HD [8], [9], [10], [11], [12], [13], [14], [15], [16], [17]. This level of 53 accuracy represents about one thousandth of the Doppler width of the (2-0) HD transitions at room temperature. Part of the experimental studies were performed in saturation regime in order to 54 55 circumvent this large Doppler broadening but underlying hyperfine components were found to lead to 56 dispersive-like spectral structure which limited the final accuracy on the line center determination [8], 57 [10], [11], [12], [14]. As a result, accuracies on HD transition frequencies achieved in saturation 58 regime are similar to those obtained in Doppler regime [15], [16], in particular at low temperature 59 [17].

60 To the best of our knowledge, in the case of the H₂ major isotopologue, no RV transition 61 frequencies have been reported with sub-MHz accuracy, so far. This matter of fact results both from 62 the above-mentioned weakness of the H_2 RV transitions and from the lack of recent measurements of H₂ spectra referenced to frequency standards. In this context, let us mention the S3 electric-quadrupole 63 64 transition of the (3-0) band near 795 nm reported with a 1.6 MHz accuracy by measuring the 65 frequency shift to a Rb line [18], [19]. Regarding the (2-0) band presented in Fig. 1, the most accurate measurements were performed in our group by high sensitivity cavity ring down spectroscopy (CRDS) 66 but the frequency calibration of these spectra were based on nearby reference lines of water vapor 67 (present as an impurity in the H₂ cell) [7], [20]. As a result, the claimed uncertainty on the reported 68 transition frequencies was no better than 10^{-3} cm⁻¹ (30 MHz). 69



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Figure 1. Calculated stick spectrum of the (2-0) band of H_2 [21]. Absorption lines previously measured by CRDS [7], [20] are highlighted. The rotational labeling is indicated for the six transitions presently measured.

73 In the present work, we report a gain of up to three orders of magnitude on high accuracy 74 frequency measurements of the Q1-Q4transitions and complement the dataset with S0 and S1, in the 75 (2-0) vibrational band of H₂. The measurements were performed in Doppler regime at room 76 temperature, by comb referenced cavity ring-down spectroscopy. As just mentioned, the Q1-Q4 77 transitions have been measured in [20] using standard CRDS while the S0 and S1 transitions were 78 apparently not revisited since the recordings of Bragg et al. in 1982 [22]. These authors used the Kitt 79 Peak Solar Fourier transform spectrometer with an absorption pathlength of 434 m and pressure values 80 of 1.4 and 2.8 atm.

81 Let us recall that the determination of the transition frequencies in Doppler regime requires to 82 correct the pressure-induced line shift in order to extrapolate the line center at zero pressure limit. This 83 necessitates for a careful line profile analysis of series of recordings at different pressures to determine the pressure dependence of each spectroscopic parameter. As non-Voigt line-shape effects (e.g. Dicke 84 85 narrowing and speed-dependent effects) are exceptionally pronounced in the case of H_2 (see [23] and 86 [24]), sophisticated line shapes such as the speed-dependent Nelkin-Ghatak profile will be used in the forthcoming analysis. Note that the high sensitivity provided by cavity-enhanced techniques like 87 CRDS is a crucial advantage as high quality recordings at low pressure (e.g. a few Torr in the present 88 89 case of the Q1, S0 and S1 lines) limit the uncertainty on the line center extrapolation at zero-pressure. 90 The experimental setup and the recording procedure are described in the following section. In

91 the next one, we present the frequency determination of the different transitions from a multi-spectrum

92 fit of the series of spectra recorded at different pressures. The obtained transition frequencies are 93 compared with theory in the penultimate section, before the concluding remarks.

Experiment

The room temperature absorption spectrum of natural dihydrogen (Alphagaz2, 99.9999 % chemical purity) was recorded in flow regime by high sensitivity frequency comb referenced cavity ring-down spectroscopy. The frequency CR-CRDS method and setup used for the recordings have been described in details in Refs. 25], [26]. The accurate frequency values associated "on the fly" to each ring-down event allow not only for an absolute calibration of the frequency axis but also for a reduction of the noise amplitude in particular on the sharp slopes of the line profiles [27], [28].

100 The recording procedure is similar to that adopted to measure D_2 transitions in the (2-0) band 101 [26], [29], [30] except that an external cavity diode laser (ECDL) is used as light source instead of distributed-feedback (DFB) laser diodes. The ECDL (Toptica fiber-connected DL pro, 1200 nm) was 102 tuned to record small spectral intervals around the Q1-Q4, S0 and S1 transitions spanning the 7970 -103 8605 cm⁻¹ region. For the evacuated cavity, the ring-down time τ_0 varied from about 180 µs to 360 µs 104 depending on wavenumber. For each frequency point, about 50 to 120 ring-downs were averaged 105 leading to a minimum detectable absorption coefficient between 2×10^{-12} and 6×10^{-11} cm⁻¹ for a single 106 scan. Following [27], [29], a self-referenced frequency comb (Model FC 1500-250 WG from Menlo 107 Systems) was used for the frequency calibration of the spectra. The frequency attached to each ring 108 109 down event is determined from (i) the frequency measurement of the beat note between a fraction of 110 the ECDL light and a tooth of the frequency comb and (ii) the tooth number deduced from the 111 frequency value provided by a commercial Fizeau type wavemeter (HighFinesse WSU7-IR, 5 MHz 112 resolution, 20 MHz accuracy over 10 hours). For each frequency step, the average central emission 113 frequency of the ECDL was actively stabilized using a software based Proportional-Integral loop, with 114 100 Hz bandpass, acting on the laser current.

A drawback of Doppler limited absorption spectroscopy compared to Doppler-free saturation spectroscopy is that the retrieved line parameters can be affected by interferences with lines due to gas impurities present in the cell. Water vapor desorbing from the CRDS cell or from the injection tubes is the most problematic species. In order to minimize the concentration of gas impurities, the spectra were recorded in flow regime. As explained in the next section, water lines were apparent in some of the spectra and allowed to estimate the water concentration in the cell to be typically a few tens of ppm.

The hydrogen pressure in the CRDS cell was actively regulated to values between 1.0 and 750 Torr through a needle valve connecting the cell to a turbo pump group, using a computer based Proportional/Integral controller. The gas pressure in the cell was continuously monitored by two capacitance gauges (MKS Baratron, 10 Torr and 1000 mbar full range). For each line, the recordings were performed for a series of pressure values listed in **Table 1**. The range of the pressure values depends on the line intensities. In the case of the Q4 line which has the smallest intensity (about 10⁻²⁹)

cm/molecule), a minimum pressure value of 90 Torr was used as the quality of the spectra was 128 129 insufficient at lower pressure. In the case of the strongest lines (Q1, S0 and S1) which have an intensity larger than 10⁻²⁷ cm/molecule range, recordings could be performed down to 1 Torr. The 130 strong pressure impact on the line profiles is illustrated in Fig. 2. For each line, the absorption 131 132 coefficients obtained for the lowest and highest pressure values have been divided by the pressure 133 values (in Torr) and superimposed. The self-induced pressure shift of the line centre and the narrowing 134 of the profile at higher pressure are clearly observed. From the relative area of water lines visible in S0 and S1 spectra one concludes that water is essentially desorbed from the tubing and/or cell rather than 135 present in the sample bottle, as its relative amount clearly varies with pressure. 136



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Fig. 2 Comparison of the profile of the six (2-0) H_2 lines observed in this work, for the lowest and highest pressure value of each dataset. For the clarity of the figure, the absorption coefficients have been corrected from the baseline and divided by the pressure values in Torr. Note the pressure shift of the line centre and the narrowing of the profile at higher pressure.

142 In order to improve the line parameter determinations, for each pressure value, a number of 143 spectra were recorded successively (from 5 to 72 – see **Table 1**). The duration of a single scan was about 20 minutes. Special efforts were made to accumulate spectra for the lowest pressure values which are crucial to minimize the uncertainty on the zero-pressure line position. In the case of the Q1 line, in addition to the recordings listed in **Table 1**, series of 30 to 100 spectra were recorded between 2 and 10 Torr, with a pressure step of 1 Torr (525 spectra in total). Those spectra were analyzed as an independent dataset because that measurement campaign was separated from the others by several years. The temperature varied between 294.0 and 295.4 K according to the recordings.

	Q4	Q3	Q2	Q1	S0	S1
Position (cm^{-1}) [21]	7971.1001	8017.1831	8051.9877	8075.3074	8406.3609	8604.2151
Intensity (cm/molecule) [21]	1.04×10 ⁻²⁹	2.36×10 ⁻²⁸	3.29×10 ⁻²⁸	2.60×10 ⁻²⁷	1.56×10 ⁻²⁷	5.24×10 ⁻²⁷
Pressure (Torr)	Number of recordings					
1				а	63	31
5				а	50	40
10		34	63	$10^{\ a}$	19	11
45		5	7	8	51	40
90	5	6	5	10	16	
210	6	5	7	10	57	
500	7	6	6	8	72	
750	7	7	5	6	36	

Table 1. Summary of the recordings of the Q1-Q4, S0 and S1 (2-0) transitions of H_2

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Note

^a Additional series of 30 to 100 spectra were recorded for pressures between 2 and 10 Torr, with a pressure step of 1 Torr. These spectra (525 in total) were analyzed as an independent dataset.

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We recall that the studied H₂ lines are observed superimposed to the broad H₂ collision induced 155 absorption (CIA) band which extends from 7800 to 9200 cm⁻¹ [31], [32]. In the present recordings 156 limited to narrow spectral intervals around the H₂ lines, the CIA shows up as an increase of the 157 baseline level of the CRDS spectra. Being due to dipole moments induced by interactions between 158 colliding H₂ molecules, the CIA has a pressure squared dependence. The sensitivity of the CRDS 159 technique and the stability of the setup allow for the measurements of such weak CIA at sub-160 atmospheric pressures. In [20], the P^2 dependence of the baseline level around the Q1 line was 161 checked up to 1 atm and the corresponding CIA binary cross-section coefficient was determined. A 162 systematic measurements of the H₂ CIA first overtone band in the region accessible with the current 163 ECDL (8000-8650 cm⁻¹) will be undertaken in the future for validation tests of the semi-empirical CIA 164 values [31], [32] largely used for planetary applications. 165

166 Line parameter retrieval

As mentioned earlier, the standard Voigt profile is not sufficient to adequately model the data 167 168 due to the prevalence of narrowing effects in H_2 , justifying the use of more sophisticated line shapes. 169 The Nelkin-Ghatak profile (NGP) includes the Dicke effect, characterized by the frequency of the velocity-changing collisions parameter, v_{VC} , in the hard-collision model. The quadratic speed-170 dependent Nelkin-Ghatak profile (qSDNGP) further takes into account the speed dependence of the 171 172 broadening and shift (γ_2 and δ_2 , respectively), modeled with a quadratic law, leading to a narrowing 173 and asymmetry of the line profile. As for the self-induced pressure shift and broadening (δ_0 , and γ_0 , respectively), the unit of the v_{VC} , γ_2 and δ_2 parameters is cm⁻¹atm⁻¹. The fitting program MATS [33] 174 175 developed at NIST, based on the HAPI Python library and the Levenberg-Marquardt minimizing 176 algorithm, was used to fit the parameters of the qSDNG profile to the spectra.

The recorded spectra were typically 0.5 cm⁻¹ wide. Some of the lines (S1, S0 and Q1) were 177 affected by interference from strong water lines in the same spectral region present due to outgassing 178 179 (see Fig. 2). These water lines were identified on the basis of the recently published [34] line list. In the case of the S1 line, a $H_2^{17}O$ line was situated less than 0.05 cm⁻¹ from the H_2 line, strongly 180 affecting the observed position if it was not included in the fit. The observed isotopic abundance of 181 H₂¹⁷O and HDO were very different from natural abundance due to previous measurements of 182 isotopically enriched water in the same CRDS cell, and were determined experimentally by recording 183 spectra while evacuating the cell. In the analysis, water lines were simulated as Voigt profiles using 184 line parameters from [34]. The water mole fraction was floated during the analysis of the S1, S0 and 185 186 Q1 spectra, and varied between approximately 10 and 200 ppm, well beyond the sub-ppm amount stated in the gas sample. No water interference was observed for the other Q lines. 187

188 The analysis was performed in two steps, separately for each line. Firstly, a subset of spectra 189 (one for each pressure value) was fit in a multi-spectrum configuration to determine the line shape 190 parameters (broadening, speed dependent parameters for more advanced profiles). The multi-spectrum 191 fitting procedure constrained the pressure broadening coefficient (γ_0) and the v_{VC} , γ_2 , δ_2 parameters to be identical for the different pressure values but the positions and line areas were fitted independently 192 for each spectrum, δ_0 being fixed to zero. The baseline of each spectrum was adjusted as a linear 193 194 function of frequency. Figure 3 presents such a subset of spectra for the Q1 and Q2 lines, along with 195 the (meas. - calc.) residuals from the multi-spectrum fits (panel (b)). Note the increase of the baseline level with increasing pressure due to the CIA contribution. The line shape parameters (γ_0 , v_{VC} , γ_2 , δ_2) 196 197 obtained for each line are given in **Table 2**. We have also tried fitting the line shape parameters separately for each pressure value, fitting all spectra recorded at a given pressure value together. As 198 199 can be seen in the lower panel (c) of Figure 3, the fit quality is better. Nevertheless, the line 200 parameters appear to vary in a somewhat non-linear manner with pressure and the speed-dependent 201 shift parameter δ_2 could not be determined with good precision at lower pressure values leading to 202 much uncertainty on the apparent line position. For this reason, we chose to use the line shape

203 parameters from the constrained fits in the rest of the analysis.

204 Table 2: Line shape parameters obtained from multi-spectrum treatment of the CRDS spectra recorded for a series of pressure values using a qSDNGP profile. The δ_0 , γ_0 , v_{VC} , γ_2 , and δ_2 values are given in 10^{-3} cm⁻¹ atm⁻¹ 205 206 unit. For a pressure series, the γ_0 , v_{VC} , γ_2 , and δ_2 values were constrained to be identical but the line centers and 207 line areas were fitted independently for each spectrum, and used afterwards to derive the zero-pressure position 208 (v_{0b}) , line intensity (S) and self-pressure shift (δ_0) . The line intensity was directly converted to the reference 209 temperature 296 K in the MATS program. For each line, the line position and shift obtained using the simpler 210 *NGP* profile (with no speed dependence, i.e. γ_2 and δ_2 fixed to zero) are given in italics in the row below. The 211 uncertainties given within parenthesis in the unit of the last quoted digit correspond to the (1σ) statistical values provided by the fit. The NGP uncertainty values are identical to the qSDNGP values and not repeated. 212 213

	$v_0 ({\rm cm}^{-1})$		S (296 K) ^c	2				s
	Ab initio ^a	Measured	$(10^{-28} \text{ cm/molecule})$	O_0	<i></i>	V_{VC}	<i>Y</i> 2	o_2
Q4 7971.10017(5)	7971.1003373(893)	0.10534(7)	-2.43(14)	4.98	22.38	7.01).77	
	9/1.10017(3)	7971.1003193(893)		-2.95(14)	3.12	35.54		
Q3 3017.18316(5)	8017.1832480(238)	2.3759(6)	-2.93(24)	5.85	21.34	7.23	0.68	
	8017.1832481(238)		-3.45(24)	3.87	34.89			
02	02 8051 08772(5)	8051.9878210(114)	3.2982(4)	-3.10(13)	5.80	21.25	7.07	0.62
Q2 5031.98775(3)	8051.9878211(114)		-3.58(13)	3.45	33.26			
01	01 075 20747(5)	8075.3075828(23)	26.021(3)	-3.25(2)	5.18	20.85	7.18	0.61
QI 075.30747(5)	8075.3075825(23)		-3.72(2)	3.21	34.25			
$\mathbf{O1}^{b}$		8075.3075822(10)	25.915(1)	-3.05(4)	6.66	12.24	Э.04	1.59
ŲI		8075.3075822(10)		-4.25(4)	2.94	25.27		
S0 3406.36086(5)	8406.3609484(20)	15.554(1)	-3.08(1)	4.90	21.72	5.74	0.75	
	400.30080(3)	8406.3609487(20)		-3.64(1)	3.24	34.51		
G1 C04 21510(5)	8604 21510(5)	8604.2152759(10)	52.380(3)	-2.53(3)	6.16	10.77	12.04	1.44
51	5004.21519(5)	8604.2152760(10)		-3.75(3)	2.82	26.01		

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215 216 Notes

^{*a}</sup> Ab initio values provided by the H2spectre software [35],*</sup>

^b Parameters obtained from the separate series of recordings between 2 and 10 Torr (see Table 1 and experiment section).

^c The given intensity values include the H_2 isotopic abundance factor (0.9997)

In a second step, all spectra were fit separately, fixing the line shape parameters to the values 221 222 determined in the first step, and floating only the position and intensity. Figure 4 presents the apparent 223 line positions as a function of pressure, for lines Q1 and S0. The zero-pressure line position and the 224 pressure shift coefficient were obtained from a linear regression of the apparent line centers vs. 225 pressure. As a clear departure from the linearity is observed at high pressure (see Fig. 4), the linear 226 regression was performed only below 210 Torr. In the case of the Q1 line, we have included an inset 227 corresponding to the separate series of recordings between 2 and 10 Torr. An excellent agreement 228 (better than 20 kHz) is achieved between the two independent determinations of the zero-pressure Q1 229 line positions.



Fig. 3 Sample of spectra of the Q2 (left) and Q1 (right) lines, recorded at various pressures and used to determine the line shape parameters. The panels present the experimental spectra (top), and the residuals of a multi-spectrum fit with qSDNGP profile and three different configurations: with all parameters constrained and

234 (a) δ_2 fixed to zero, or (b) δ_2 floated; (c) with all parameters determined separately for each pressure. Note the

235 different vertical scale for the two lines, and the increase in the baseline at higher pressure due to CIA.



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Fig. 4 Determination of the zero-pressure position from a linear extrapolation of the apparent qSDNGP and NGP line positions, for the lines Q1 and S0. The linear regression was performed below 210 Torr due to the clear non-linearity of the observed shift at higher pressures. This plot also illustrates the difference in apparent position due to the inclusion of the speed-dependence of the pressure shift (δ_2) which takes into account the asymmetry of the line profile (see the residuals in Fig. 3). The insert on the Q1 panel corresponds to the separate series of recordings between 2 and 10 Torr (see Text).

In order to evaluate the impact of the choice of the profile to the zero-pressure positions, we 244 also applied the procedure described above using simpler line profiles: NGP which does not include 245 the speed dependence of the broadening and shift ($\gamma_2 = 0$ and $\delta_2 = 0$), and qSDNGP with δ_2 fixed to 246 247 zero. The positions obtained with these two simpler profiles are very close but differ importantly from 248 the qSDNGP positions (see Fig. 4). This observation evidences the importance of the speeddependence of the collisional shift (δ_2) on the apparent line position at a given pressure value. This is 249 reflected by the larger residuals obtained with $\delta_2 = 0$ (Fig. 3a). Nevertheless, as illustrated in Fig. 4, the 250 zero-pressure position is found mostly identical for the different profiles (within a few kHz). The 251 position and line shift values obtained using a NGP profile are included in Table 2. Let us mention 252 253 that the NGP pressure shift parameters of the O1-O4 lines agree with the values retrieved in [17] using 254 a Galatry profile (both NGP and Galatry profiles neglect speed dependent effects).

From Table 2, it appears that all the values of the profile parameters of the different lines are 255 256 consistent except those relative to the separate series of Q1 recordings with maximum pressure of 10 257 Torr and those of the S1 recordings with maximum pressure of 45 Torr. This observation indicates that 258 the line parameters depend not only on the chosen profile but also on the maximum pressure of the recordings used for their determination. The reader is referred to [25] and [26] for a detailed 259 260 discussion of the correlation between the profile parameters and an improved treatment of the Q1 (2-0) 261 and Q1 (3-0) profiles using a corrected Hartmann-Tran profile adjusted to particular model of the 262 velocity-changing collisions.

263 Line intensities were floated for each spectrum and obtained as the average of the line intensity 264 values weighted by their fit uncertainty. The agreement between the reported intensities (Table 2) and HITRAN values (Table 1) is excellent: the four determinations of the Q1, S0 and S1 line intensities 265 lead to an average intensity ratio of 1.0013 with a standard deviation less than 0.0020. The largest 266 relative deviation for the very weak Q4 line with intensity of 1.04×10^{-29} cm/molecule is less than 1%. 267 This level of agreement validates with an unprecedented accuracy the *ab initio* intensity values of [37] 268 269 reproduced in the HITRAN database [21]. We note that the uncertainty of 10% attached to the 270 HITRAN intensity value is largely overestimated.

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Comparison with *ab initio* calculations

In the recent years, significant progress has been achieved in the *ab initio* calculations of the H₂, 272 HD and D₂ energy levels [1], [2], [3], [36]. In these works, the authors performed 4-particle fully 273 variational calculations including relativistic and quantum electrodynamics (QED) effects in the 274 275 nonadiabatic treatment of the nuclear motion. The resulting calculated transition frequencies of the six H₂ transitions presently studied, as provided by the H2spectre software [35], are included in Table 2. 276 The corresponding deviations from the experimental values are displayed on the lower panel of Fig. 5. 277 278 The first observation is that the deviations obtained for Q3, Q2, S0 and S1 are very close, with 279 an average value of 2.58 MHz and a standard deviation of only 63 kHz. The Q1 transition frequency 280 determined shows a higher deviation of 3.23 MHz and appears to be an outlier. For comparison purpose, we present on the upper panel of Fig. 5, a similar plot for the six (electric dipole) transitions 281 282 of HD in the (2-0) band whose transition frequencies have been recently measured at high accuracy. 283 The experimental values obtained in saturation regime [8], [10], [11] and in Doppler regime [16], [17] have a similar accuracy of a few tens of kHz. Very similar (meas.-calc.) frequency differences are 284 285 obtained for the six HD transitions which are on average 1.91 MHz larger than predicted by theory 286 with a standard deviation of only 49 kHz. Let us note that for both H₂ and HD, the calculated frequencies are underestimated and deviate on average by about 1.7 times their claimed uncertainties 287 (1.6 and 1.1 MHz, respectively). The absence of rotational dependence of the deviations leads to the 288 289 conclusion that the V=2 vibrational term is underestimated by theory, both for H₂ and HD.



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Fig. 5 Differences between experiment and theory for the most accurate experimental frequency determinations in the first overtone band of HD and H_2 (upper and lower panels, respectively). The theoretical values [Puchalski2019] were obtained from the H2spectre software [35]. The grey arrows represent the uncertainty on the calculated values (1.1 and 1.6 MHz, respectively). The H_2 experimental values are obtained in the present work while the HD experimental values were reported in [8], [10], [11], [16], [17]. The black arrows on the lower panel connect pairs of transitions involved in ground state combination relations.

While the uncertainties on the *ab initio* transition frequencies are much larger than the corresponding experimental error bars (about one MHz compared to few tens of kHz), the energy separations between the ground state rotational levels are predicted with a much better accuracy, allowing for more stringent validation tests. The lower uncertainty on the energy separation isprobably related to error cancellations in the computation of the energy levels.

303 Combination differences of measured transition frequencies can be used to derive accurate 304 values of the energy level separation in the ground vibrational state. From the six presently measured 305 H₂ (2-0) transition frequencies, the energy separation between the J=2 and J=0 levels, $\Delta_{(J=2)-(J=0)}$ is obtained by difference of S0 and Q2 transition frequencies sharing the same J=2 (V=2) upper level. 306 307 Similarly, the separation between the J=3 and J=1 levels, $\Delta_{(J=3)-(J=1)}$, is obtained by difference of S1 and Q3 transition frequencies. The comparison summarized in Table 3 shows a convincing agreement 308 between theory and experiment, within 100 kHz for both $\Delta_{(J=2)-(J=0)}$ and $\Delta_{(J=3)-(J=1)}$. This level of 309 310 agreement is surprisingly good as it is better than the uncertainties on the *ab initio* values (110 and 180 kHz, respectively) which are themselves significantly better than the uncertainties on the experimental 311

312 values (390 and 750 kHz, respectively).

In summary, the experimental data at disposal confirm the sub-MHz accuracy of the energy separation of the *ab initio* ground state energy levels and seem to indicate that the 1.6 MHz claimed uncertainty on the *ab initio* frequencies of the studied (2-0) transitions is underestimated.

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Table 3. Energy separations between ground state rotational levels obtained by combination differences of the measured and calculated transitions. The difference between the J=2 and J=0 energy levels is obtained from the S0 and Q2 transition frequencies. That between the J=3 and J=1 energy levels is obtained from the S1 and Q3 transition frequencies. The (1σ) uncertainties are given within parenthesis in the unit of the last quoted digit.

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	Exp. (MHz)	Calc. (MHz)	Exp. (cm^{-1})	Calc. (cm^{-1})
$\Delta_{(J=2)-(J=0)}$	10 623 839.09(39)	10 623 839.17(11)	354.373 127(13)	354.373 130(4)
$\Delta_{(J=3)-(J=1)}$	17 598 777.46(75)	17 598 777.37(18)	587.032 028(25)	587.032 025(6)

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324 Compared to the average value for the Q3, Q2, S0 and S1 positions, the 650 kHz offset of the Q1 frequency difference to theory remains to be explained. We considered a possible experimental 325 bias due to an underlying impurity line located in the range of the Q1 line profile but the typical 326 concentration of water vapor in our spectra leads to a negligible contribution of this species. A 327 328 systematic search of the possible absorbers on the basis of the HITRAN database indicates that a weak 329 ammonia line could be present in the range of the Q1 profile but the spectra do not show stronger ammonia lines located in the vicinity. Let us recall, that the Q1 transition frequency reported in this 330 331 work was retrieved from spectra recorded in two measurement campaigns, three years apart and an 332 agreement within 18 kHz is obtained between the two determinations. Thus the origin of the 650 kHz offset of the Q1 frequency is unexplained so far. As the O3 (2-0) transition at 7488.27 cm⁻¹ is sharing 333 334 the same J=1 (V=2) upper level than the Q1 transition, the (planned) accurate measurement of the O3 transition frequency should bring a decisive confirmation on the existence and on the amplitude of thisoffset.

337 In summary, the transition frequencies reported in the present work are the most accurate reported so far for H₂, the first ones in the (2-0) band referenced to an absolute frequency standard. 338 Their relative uncertainty is at the 10^{-10} level in the case of the Q1, S0 and S1 transitions. The 339 comparison to the most recent theoretical values shows a systematic underestimation of 2.58 MHz of 340 the calculated transition frequencies, a situation similar to that found for (2-0) transitions in HD. This 341 observation provides a challenge for future refinement of the *ab initio* calculations of the energy levels 342 of the hydrogen molecule. Finally, we note that ab initio values of the intensity of the six studied 343 344 transitions are validated within a few thousandths.

345 *Conflicts of interest*

346 There are no conflicts to declare.

347348 Acknowledgements

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