



HAL
open science

Combining CO₂ valorization strategies to produce hydrogen and methane in a geological storage context

Emeline Vidal, Cyrielle Fauveau, Cyril Aymonier, Anaïs Cario, Samuel Marre

► To cite this version:

Emeline Vidal, Cyrielle Fauveau, Cyril Aymonier, Anaïs Cario, Samuel Marre. Combining CO₂ valorization strategies to produce hydrogen and methane in a geological storage context. 13th International Symposium on Supercritical Fluids, International Society for the Advancement of Supercritical Fluids, May 2022, Montréal, Canada. 2 p. hal-04152350

HAL Id: hal-04152350

<https://hal.science/hal-04152350>

Submitted on 5 Jul 2023

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Combining CO₂ valorization strategies to produce hydrogen and methane in a geological storage context

E. Vidal, C. Fauveau, C. Aymonier, A. Cario, S. Marre*

CNRS, Univ. Bordeaux, ICMCB, UMR 5026, F-33600 Pessac, France

*Corresponding author: samuel.marre@icmcb.cnrs.fr

1. Introduction

The dramatic increase of anthropogenic greenhouse gases concentrations in the atmosphere threatens the ecosystems equilibrium, including our way of life. The development of sustainable technologies in the frame of mitigation efforts to reduce anthropogenic CO₂ emissions remains an urgent need. A promising strategy consists in storing the CO₂ in deep geological formations, like deep saline aquifers. However, most of the considered methodology aim at forming carbonates to safely store CO₂ as a waste. Oppositely, it could be used as a raw material – and therefore valorized - for producing valuable resources such as hydrogen (through chemical reaction with Iron) or methane (through bioconversion). Far from being anecdotal, the transformation of part of the stored CO₂ into hydrogen or methane would lower the costs of capture and storage and change the paradigm of CO₂ geological storage.

Deep saline aquifers are porous environments, with temperature ranging between 40°C to 100°C and hydrostatic pressure up to 150 bar. When CO₂ is injected into these environment, several geochemical relations can proceed, in particular with iron-rich minerals, resulting in carbonation processes, which can also produce hydrogen. Besides, such reaction can be speeded up by considering the co-injection of CO₂ with iron wastes, which represent millions of tons of industrial wastes. This will results in: $\text{Fe} + \text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{FeCO}_3 + \text{H}_2$. Meanwhile, methanogens microorganisms, which are part of the microbial population living in deep saline aquifers, and more particularly hydrogenotrophic methanogens, can transform CO₂ and H₂ into methane, through methanogenesis: $\text{CO}_2 + 4 \text{H}_2 \rightarrow \text{CH}_4 + 2 \text{H}_2\text{O}$. Using their metabolisms would allow considering deep saline aquifers as macro (bio)reactors for an upgrading process of CO₂ to methane.

2. Materials and Methods

To investigate the transformation of CO₂ either in a biotic or in an abiotic way, we have developed multiscale high-pressure transparent biocompatible approaches (microfluidics and millifluidics) allowing to mimic deep environments at lab scale. The tools and approaches that we are developing and using make it possible to simulate these deep ecosystems, to estimate the yields of such approaches and the possibility of testing exploitation scenarios.

In the case of abiotic generation of hydrogen, iron powder was placed in a millifluidic reactor filled with DI or saline water and further pressurized with supercritical CO₂. The evolution of the gas phase was monitored with *ex situ* gas chromatography to access the reaction kinetics.

For the bioconversion of CO₂ to methane, microfluidics or millifluidics reactors were filled with an artificial groundwater medium, inoculated with a model microbial methanogen strain: *Methanococcus thermolithotrophicus*. This medium is then pressurized with a mixture of hydrogen and CO₂ (80/20 mol%) and the biofilm formation along with the gas production is monitored over time.

3. Results and discussion

We have determined the reaction kinetic of the abiotic production of hydrogen through iron corrosion in CO₂ / water medium, demonstrating quite fast reaction rates. Up to 80% of the iron is converted to carbonate, resulting in hydrogen production within few hours for pressure and temperature conditions ranging from $50 < p \text{ (bar)} < 200$ and $25 < T \text{ (}^\circ\text{C)} < 130$.

Similarly, microbial CO₂ conversion to methane can reached high conversion rates at similar conditions and residence times. This is also linked with the development of biofilms inside the reactor, which was monitored using microfluidics approaches.

Finally, we have combined abiotic hydrogen generation with biotic CO₂ conversion by introducing iron powder in a microbial process, without introducing hydrogen. Depending on the CO₂ partial pressure, microbes can use the hydrogen production from iron corrosion to combine it with CO₂ and to form methane.

4. Conclusions

In brief, the promotion of the (bio)chemical conversion of CO₂ to hydrogen and methane in geological formations constitutes an advantageous long-term CCUS strategy. Besides the sole storage of CO₂, deep saline aquifers may also represent an interesting option for stable long-term energy generation and storage in addition to being cost-effective (bio)reactors for CO₂ valorization at a global scale. As the global energy economy remains hydrocarbon-based, this double valorization could aid in the transition to more sustainable sources of energy and raw materials.