



HAL
open science

Multiscale processes controlling niobium mobility during supergene weathering

Quentin Bollaert, Mathieu Chassé, Thierry Allard, Alexandra Courtin,
Laurence Galois, Gautier Landrot, Cécile Quantin, Delphine Vantelon,
Georges Calas

► To cite this version:

Quentin Bollaert, Mathieu Chassé, Thierry Allard, Alexandra Courtin, Laurence Galois, et al.. Multi-scale processes controlling niobium mobility during supergene weathering. *Geochimica et Cosmochimica Acta*, 2023, 353, pp.142-157. 10.1016/j.gca.2023.05.023 . hal-04134293

HAL Id: hal-04134293

<https://hal.science/hal-04134293>

Submitted on 4 Jul 2023

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Journal Pre-proofs

Multiscale processes controlling niobium mobility during supergene weathering

Bollaert Quentin, Chassé Mathieu, Allard Thierry, Courtin Alexandra, Galois Laurence, Landrot Gautier, Quantin Cécile, Vantelon Delphine, Calas Georges

PII: S0016-7037(23)00242-9
DOI: <https://doi.org/10.1016/j.gca.2023.05.023>
Reference: GCA 13066

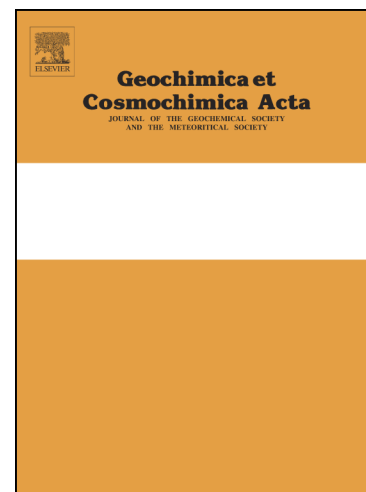
To appear in: *Geochimica et Cosmochimica Acta*

Received Date: 30 November 2022
Revised Date: 20 February 2023
Accepted Date: 30 May 2023

Please cite this article as: Quentin, B., Mathieu, C., Thierry, A., Alexandra, C., Laurence, G., Gautier, L., Cécile, Q., Delphine, V., Georges, C., Multiscale processes controlling niobium mobility during supergene weathering, *Geochimica et Cosmochimica Acta* (2023), doi: <https://doi.org/10.1016/j.gca.2023.05.023>

This is a PDF file of an article that has undergone enhancements after acceptance, such as the addition of a cover page and metadata, and formatting for readability, but it is not yet the definitive version of record. This version will undergo additional copyediting, typesetting and review before it is published in its final form, but we are providing this version to give early visibility of the article. Please note that, during the production process, errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

© 2023 Published by Elsevier Ltd.



1 **Multiscale processes controlling niobium mobility during supergene**
2 **weathering**

3 **Authors :** Bollaert Quentin^{1,*}, Chassé Mathieu¹, Allard Thierry¹, Courtin Alexandra², Galoisy
4 Laurence¹, Landrot Gautier³, Quantin Cécile², Vantelon Delphine³, Calas Georges¹.

5 *corresponding author

6 **Affiliations :**

7 ¹Sorbonne Université, Institut de Minéralogie, de Physique des Matériaux et de Cosmochimie, 4
8 place Jussieu, Paris, 75005, France

9 ²Université Paris-Saclay, UMR 8148 GEOPS, 91405 Orsay Cedex, France

10 ³Synchrotron SOLEIL, L'Orme des Merisiers, Saint-Aubin, BP 48 91192 Gif-sur-Yvette Cedex,
11 France

12

13

14

15

16

17

18

19

20

21

ABSTRACT

22 Niobium (Nb) is one of the most immobile elements during supergene weathering, widely used
23 for mass-balance calculations, despite elusive information on the mechanisms controlling its
24 dynamics in the critical zone. Here, a multiscale approach, from weathering profile to atomic-scale,
25 is developed to monitor Nb speciation along a thick (about 50 m) lateritic regolith formed over the
26 Pitinga pluton (Amazonas, Brazil). In the A-type parent granite, Nb is mainly hosted in ilmenite
27 (1.5–2 wt% Nb₂O₅) and, to a lesser extent, in rutile (2–3 wt% Nb₂O₅). A quantitative assessment of
28 the average Nb speciation from the parent rock to the upper horizons has been carried out by
29 combining spectroscopic and spatially-resolved chemical techniques. The contribution of Nb-
30 bearing Ti oxides (2–6 wt% Nb₂O₅) and Fe oxides (0.1–0.3 wt% Nb₂O₅) in the average Nb
31 speciation increases with the degree of weathering and reaches 80 % in the most altered horizon.
32 This unusual Nb speciation results from the nature of the primary Nb carriers, less resistant to
33 weathering than common Nb ore minerals such as pyrochlore. X-ray absorption spectroscopy
34 demonstrates that Nb released from the weathering of ilmenite substitutes for Fe and Ti in goethite
35 and Ti oxides, respectively, providing atomic-scale evidence of the high affinity of Nb for these
36 phases. Elemental mobility of Nb, Ta and Sn followed by geochemical mass-transfer calculations
37 evidences lateral transport from Nb-Sn-enriched laterites developed over the surrounding Nb-
38 enriched facies of the Pitinga granite to the studied profile. Elements such as Ti, Zr and Hf, often
39 considered immobile, are leached or redistributed in the studied profile. Our results demonstrate Nb
40 remobilization during intense weathering. Although Nb scavenging by secondary Fe and Ti oxides
41 limits Nb mobility at the mineral scale, this work questions the unrestricted use of Nb as a chemical
42 invariant during surficial alteration processes.

43 **Keywords** : Niobium, supergene weathering, laterite, speciation, XANES spectroscopy, mass-
44 balance calculations, elemental mobility

45

1. INTRODUCTION

46 Niobium (Nb) is among the most immobile elements during weathering processes (MacLean
47 and Barrett, 1993) due to corrosion resistance of Nb-bearing minerals (McMaster et al., 2018) and
48 low solubility in fluids (Peiffert et al., 2010). Such conservative behavior is essential to document
49 geological processes (Hickmott and Spear, 1992; Ballouard et al., 2016) including mass-balance
50 calculations of weathering budget (Nahon and Merino, 1996; Kurtz et al., 2000). Niobium
51 immobility is at the origin of world-class deposits such as those found in Brazilian laterites, i.e.
52 Araxá and Catalão Nb deposits, which represent about 90 % of the world production of this critical
53 metal (U.S.G.S, 2022). Unparalleled Nb enrichment in laterites results from the tropical alteration of
54 Nb-enriched carbonatite hosting weathering-resistant Nb phases from the pyrochlore supergroup
55 minerals $[A_{2-x}B_2(O,OH)_6(OH,F,H_2O)_{1-y}]$ (Mitchell, 2015). However, these Nb-enriched carbonatites
56 are formed during complex multi-stage processes and, consequently, they are neither representative
57 of typical carbonatite nor of the continental crust (Chakhmouradian, 2006).

58 In magmatic rocks, multicomponent Nb oxides are not the only Nb-bearing minerals.
59 Widespread Ti oxides such as rutile, brookite and ilmenite are also typical hosts of Nb (Černý and
60 Ercit, 1989). Niobium substitution for titanium (Ti) can reach high levels (30 wt% Nb₂O₅ in brookite
61 and rutile and 15 wt% Nb₂O₅ in ilmenite) due to the vicinity of Nb and Ti ionic radii
62 (Chakhmouradian and Mitchell, 1999; Ballouard et al., 2020). As titanium oxides are expected to be
63 resistant to weathering (Milnes and Fitzpatrick, 1989; Meinhold, 2010), Nb should be preserved in
64 these primary carriers and will remain immobile. However, the alteration of Ti oxides and ilmenite
65 during intense chemical weathering in tropical regions has long been evidenced (Anand and Gilkes,
66 1984; Nahon and Merino, 1996; Du et al., 2012), questioning the fate of Nb after the dissolution of
67 such minerals. Our understanding of Nb dynamics in supergene context is limited by the lack of data
68 on Nb solubility and mobility in fluids (Lukyanova et al., 2017; Filella and May, 2020) and by the
69 scarcity of studies identifying the evolution of Nb carriers during alteration of lithotypes distinct

70 from pyrochlore-bearing carbonatite. The study of Nb speciation in regoliths that formed at the
71 expense of rocks containing Nb-bearing Ti oxides- and ilmenite allows to discuss Nb mobility in the
72 critical zone.

73 In this study, we investigate Nb speciation in regolith samples from the region of the world-
74 class tin (Sn)-Nb-tantalum (Ta) Pitinga deposit (Bastos Neto et al., 2009). Lateritic profiles formed
75 from the weathering of the albite-enriched granite (Horbe and da Costa, 1999; Alves et al., 2018)
76 exhibit exceptional Nb concentration (*ca* 1500 ppm Nb) due to the combination of singular
77 magmatic and hydrothermal processes (Bastos Neto et al., 2009). Pyrochlore and columbite
78 [(Mn,Fe)Nb₂O₆] from the parent rock have been preserved in the altered horizons despite the
79 fragmentation of columbite into thin particles and the minor release of Nb from pyrochlore and
80 columbite which was later incorporated into Fe oxides (Alves et al. 2018). Here, we complement the
81 knowledge of the Pitinga province by studying Nb speciation in a laterite developed over the barren
82 biotite alkali-feldspar granite (Lenharo et al., 2003). The investigation of this profile will uncover
83 the processes of Nb concentration during supergene weathering that apply to widespread continental
84 rocks and help discuss the role of the parent rock mineralogy.

85 2. MATERIALS AND METHODS

86 2.1 Geological context and sampling

87 The Pitinga province is located within an A-type granitoid belt typical of extensional settings
88 in the southern part of the Guiana shield, dated at 1.82 Ga (Bastos Neto et al., 2009; Bettencourt et
89 al., 2016). As the Pitinga alluvial ore deposit is now exhausted, the Sn-Nb-Ta-enriched Pitinga
90 granite is now mined for Sn, Nb and Ta. The studied lateritic profile (S0°44'52",W60°6'28") was
91 sampled over the biotite–alkali feldspar Pitinga granite (1.822 ± 0.002 Ga) (Costi et al., 2000)
92 exhibiting a moderate Nb enrichment (Fig. 1), more representative of granitic rocks than

93 surrounding highly-enriched facies characterized by a unique association of cryolite (Na_3AlF_6), Sn,
94 Nb and several other rare metals (Bastos Neto et al., 2009).

95 The Pitinga region in the Amazon rainforest is characterized by a tropical-humid climate
96 with intense rainfall ($2000 \text{ mm} \cdot \text{yr}^{-1}$) year-round, warm temperatures (*ca* $26 \text{ }^\circ\text{C}$ on average) and a
97 long wet season (5–6 months). The lateritic cover is located on the summit of regional plateaus at *ca*
98 250 m elevation. In the region, they attain 50 m thick (Horbe and Da Costa, 1990; Alves et al.,
99 2018). The minimum age assessed for such thickness is 1 Ma, considering an estimate of the
100 progression of the weathering front at a rate of $0.05 \text{ mm} \cdot \text{yr}^{-1}$ in this region (Mathieu et al., 1995).
101 Such long-term development of lateritic regolith is made possible by the elevation of the alteration
102 profile providing protection from marine transgression events and by the stability of the latitude of
103 the Amazon craton since 160 Ma, thus maintaining a tropical climate over the Cenozoic and part of
104 the Mesozoic (Monteiro et al., 2018).

105 From the bottom to the top horizons, the studied regolith is composed of five units
106 distinguished from their macroscopic properties: the granitic parent rock, the clayey mottled zone,
107 the lateritic horizon, the bauxitic horizon and the ferruginous duricrust. Ten regolith samples,
108 representative of these different horizons were collected from 3 to 50 m depth. The saprock and
109 saprolite were not accessible at the outcrop and could not be sampled.

110 2.2 Bulk mineralogical characterization

111 X-ray diffraction was performed on powdered samples with a PANALYTICAL X'pert Pro
112 MPD diffractometer. Measurements were carried out in Bragg-Brentano geometry using a $\text{Co K}\alpha$
113 anode in order to minimize the X-ray absorption of Fe, enriched in the uppermost lateritic horizons.
114 Data were recorded with an X'Celerator detector between $3^\circ 2\theta$ and $90^\circ 2\theta$ with 0.017° steps.
115 Incident beam mask was fixed at 20 mm and Soller slits at 0.04 rad. Total counting time of
116 measurement was 6 hours per sample. Identification and semi-quantification of the major mineral

117 phases were performed with X'Pert HighScore software. The accuracy of the mineralogical
118 quantification is discussed in Supplementary Material (Table S1).

119 2.3 Bulk geochemical characterization

120 Two grams of powdered samples, representative of each horizon, were sent to the SARM
121 (Service d'Analyses des Roches et des Minéraux, CRPG, Nancy) for chemical analysis (Table S2).
122 Bulk concentrations of major and trace elements were analyzed by ICP–OES (iCap6500
123 ThermoFisher) and quadrupole ICP–MS (iVapQ ThermoFisher), respectively, after a lithium
124 tetraborate alkali fusion (Carignan et al., 2001). Loss on ignition (L.O.I.) was obtained by
125 gravimetric analysis at 1020 °C.

126 2.4 Mass-balance calculations

127 Quantification of the net chemical gains and losses was carried out using the following mass
128 conservation equation (Brimhall et al., 1991).

$$129 \frac{V_p \rho_p C_{j,p}}{100} + m_{j,flux} = \frac{V_w \rho_w C_{j,w}}{100} (1)$$

130 where V is the volume in cm^3 , ρ is the dry bulk density (g/cm^3) and C is the concentration (wt%) of
131 an element j in the parent material $C_{j,p}$ and in the weathered product $C_{j,w}$. The mass of an element j is
132 denoted by the $m_{j,flux}$ term which is positive when an element j accumulates in the sample and
133 negative when it is leached from the sample. In this work, we use the biotite granite from Pitinga as
134 the parent rock in order to capture the chemical transfers due to lateritization processes. The use of
135 thorium (Th) as the conservative element is further discussed.

136 Volumetric strain index ($\epsilon_{i,w}$) corresponding to the volume change between the weathered
137 material and the parent rock is estimated by :

$$\varepsilon_{i,w} = \left(\frac{\rho_p C_{i,p}}{\rho_w C_{i,w}} \right) - 1 \quad (2)$$

where ρ is the bulk density (g/cm^3) obtained by the paraffin method (Blake, 1965), $C_{i,p}$ and $C_{i,w}$ are the concentration of a reference element (considered as immobile during weathering) in the parent rock and weathered material, respectively. Volumetric strain around zero indicates isovolumetric weathering. Positive values indicate expansion and negative ones indicate collapse (Table S3).

Quantification of chemical mass gain or loss through a profile uses the open-system mass-transport function $\tau_{j,w}$ (Table S4), defined by :

$$\tau_{j,w} = \left(\frac{\rho_w C_{j,w}}{\rho_p C_{j,p}} \right) (\varepsilon_{i,w} + 1) - 1 \quad (3)$$

If $\tau_{j,w}$ is equal to 0, the element is immobile during weathering. A positive value reflects an absolute mass gain in element j of the weathered rock compared to the mass present in the parent rock whereas negative value indicates a mass loss. For instance, a positive value of 0.5 indicates that the mass of the element j results from an addition of 50 % relative to the mass of j present in the fresh parent rock.

2.5 Spatially-resolved chemical analyses

Polished sections of selected samples were made by impregnating chips in epoxy resin and polishing on cloth with diamond pastes.

Scanning electron microscopes (SEM) equipped with focused electron gun (FEG) (Zeiss Ultra55, Sorbonne Université and Zeiss Sigma, École Normale Supérieure, Paris) were used to map the polished sections with an AsB detector collecting backscattered electrons. The optimal working distance was chosen according to the geometry of the sample chamber and the beam energy fixed to 15.0 kV. Energy dispersive X-ray spectroscopy (EDS) chemical mapping was performed with a Bruker Quantax EDS system using an Oxford 50 mm^2 silicon drift detector (SDD).

160 Electron probe microanalysis (EPMA) was performed on a CAMECA SXFive EPMA
161 equipped with five wavelength-dispersive spectrometers (WDS) at the Centre d'Analyse des
162 Minéraux de Paris (CAMPARIS). An accelerating potential of 15 keV and a sample current of
163 40 nA have been used except for Fe oxides. Due to their small size and to protect them from beam
164 radiation damage, the current was decreased to 15 nA. As a consequence, counting time was
165 increased to analyze minor and trace elements such as Nb. Wavelength dispersive spectrometer
166 analyses were performed using the following standards: Fe₂O₃ for iron (Fe), almandine for
167 aluminum (Al) and silicon (Si), MnTiO₃ for manganese (Mn) and Ti, NaNbO₃ for Nb, metallic Ta
168 for Ta, monazite for Th.

169 2.6 Reference compounds for X-ray absorption analyses

170 A large set of natural and synthetic Nb-bearing reference materials were considered as
171 potential compounds for X-ray absorption near-edge structure (XANES) analyses (Bollaert et al.,
172 2023). Apart from columbite-(Mn) and Nb-bearing ilmenite which were provided by École
173 Nationale Supérieure des Mines de Paris (MINES ParisTech), Nb reference materials were
174 synthesized in order to incorporate Nb concentrations representative of natural enrichment.

175 Niobium-substituted goethite was prepared by adapting a hydrothermal method used for Sc-
176 bearing Fe oxides (Chassé et al., 2017). Niobium-substituted goethite was converted into Nb-
177 substituted hematite (α -Fe₂O₃) by dehydroxylation. The synthesis of Nb-substituted anatase and
178 rutile has been carried out using a sol-gel method at moderate temperature. Detailed mineral
179 synthesis procedures are described in Bollaert et al. (2023). The synthesis of Nb-sorbed phases at
180 concentrations representative of natural lateritic media could not be carried out because Nb oxides
181 reprecipitation instantly before adsorption.

182 2.7 Procedures for synchrotron-based analyses

183 XANES data at Nb L₃- and K-edges have been collected at SOLEIL synchrotron facility
184 (Saint-Aubin, France) operating with a storage ring current of 450 mA and energy of 2.75 GeV on
185 the LUCIA (Vantelon et al., 2016) and SAMBA (Fonda et al., 2012) beamlines. The acquisition of
186 spectra at the K and L₃-edges is complementary, since two similar spectra at the L₃-edge may be
187 different at the K-edge, or inversely, since the information about the atomic-scale environment of
188 Nb is different (Fig. S1).

189 A Si(111) double-crystal monochromator crystal was used on the LUCIA beamline with an
190 energy resolution of 0.25 eV at 2400 eV (Schaeffers et al., 2007). The monochromator was calibrated
191 at the energy of the Nb L₃-edge using a Nb₂O₅ powder pellet. Due to high Nb contents in Nb
192 mineralogical references, the spectra were recorded on cellulose-diluted pellets before being
193 mounted on a sample-holder. The low penetration depth of X-rays at this energy or the low Nb
194 contents required a data acquisition in the fluorescence mode. Energy steps were (2, 0.2, 1) eV for
195 energy ranges of (2300–2350), (2350–2400) and (2400–2455) eV, respectively, with a 1 s
196 integration time. Two measurements per sample at room temperature and under vacuum were
197 sufficient to get adequate signal-to-noise ratio and check for reproducibility of the edge features.

198 Spatially resolved μ -XRF and μ -XANES analyses were conducted by focusing the beam with
199 four total reflection mirrors in Kirkpatrick-Baez geometry. Micro-XRF element maps were carried
200 out on polished sections using an energy dispersive–silicon drift diode (SDD). For each map, from 5
201 to 10 μ -XANES spectra of pre-identified minerals were recorded with an excitation energy of
202 2450 eV and a micro-beam focus of 2.8 x 3.8 μ m.

203 Bulk Nb K-edge scans were collected from 18,700 to 20,000 eV using a continuous scan
204 acquisition mode on the SAMBA beamline. The XANES region corresponds to the first 50 eV
205 above the energy of the edge (*ca* 18,990–19,040 eV). The energy range for the EXAFS region is
206 between the end of the XANES region up to 20,000 eV. The beamline was equipped with a Si(220)

207 monochromator providing a 2 x 0.5 mm (H x V) monochromatic beam. The samples were analyzed
208 at 20 K using a He cryostat. Scans were obtained in 170 s and featured 1625 data points with a *ca*
209 0.8 eV step. Multiple scans were collected for each sample until no significant improvement in the
210 signal-to-noise ratio of their merged spectra was observed. Up to 30 scans were collected for the
211 bulk rock samples.

212 Normalization and least-squares linear combination fitting of X-ray absorption spectra were
213 performed using Larch software (Newville, 2013).

214 3. RESULTS

215 3.1 Mineralogy and geochemistry

216 The biotite-alkali feldspar granite is a medium- to fine-grained leucocratic granite with a
217 texture ranging from equigranular to porphyritic. It is composed of mesoperthitic alkali feldspar
218 (Fig. 2a,b) (*ca* 53 %), albite (*ca* 33 %), quartz (*ca* 11 %) and biotite (*ca* 4 %) with chlorite, fluorite,
219 zircon, monazite, ilmenite, rutile, apatite, xenotime, thorite, columbite, galena, cassiterite and barite,
220 in order of abundance as characterized using SEM. Zircon, monazite, ilmenite and rutile are 10-
221 20 μm large and found embedded in large biotite flakes (Fig. 2c), partially replaced by chlorite (Fig.
222 2d). Rutile also occurs as submicrometric grains within biotite. Thorite, apatite, monazite and
223 cassiterite are found as isolated submicrometric crystal within feldspar. The alkali concentration of
224 the granite in comparison to alumina concentration classifies it as a peraluminous granite. The
225 concentration in high-field-strength elements (HFSE: i.e., zirconium (Zr), hafnium (Hf), Th, Nb, Ta,
226 uranium (U) and rare-earth elements - REE) is *ca* 1200 ppm including *ca* 700 ppm of REE (Table
227 S2). Niobium concentration is 46 ppm.

228 Above the parent granite, the lateritic profile consists of four distinct horizons (Fig. 3a). The
229 clayey mottled zone exhibits pinkish (10R 8/4, from Munsell color chart, Soil Survey Staff, 2017)

230 and light red (2.5YR 7/6) colors with white clayey areas. The friable lateritic horizon is
231 characterized by the predominance of reddish yellow matrix (5YR 7/8) composed of clays. A dark
232 indurated bauxitic horizon (2.5YR 6/8) with white mottles is identified. The ferruginous duricrust
233 (5YR 6/8) corresponds to a Fe-enriched crust containing consolidated nodules of Fe oxides (Fig. 3b,
234 c).

235 The dominant mineral throughout the profile is kaolinite in variable proportions depending on
236 the horizon (Fig. 3b). Gibbsite is found from the lateritic horizon up to the duricrust forming the top
237 horizon and reaches a maximum in the bauxitic horizon (*ca* 45 wt%). Other minerals include quartz
238 (5–13 wt%), Fe oxides (i.e., hematite and goethite, 2–11 wt%), anatase (2–3 wt%). Small amounts
239 of rutile (< 1 wt%) are detected by XRD in the bauxitic horizon and the ferruginous duricrust. The
240 proportion of Fe oxides increases from the bottom to the top of the profile (Fig. 3b). In the bauxitic
241 horizon with the highest Zr content, zircon was also identified as a trace mineral on the XRD pattern
242 (Fig. 3d).

243 The clayey mottled horizon is texturally similar to the lateritic horizon (Fig. 4a, b).
244 Submillimetric grains of quartz are found along the profile (Fig. 4c). In the bauxitic horizon,
245 kaolinite can be found as microcrystalline, forming stacks of lamellae discordant to the surrounding
246 matrix (Fig. 4d). Iron oxides occur both as agglomerates of submicron size crystals (Fig. 4d) and as
247 crystals of hundreds of microns isolated in the matrix in the lateritic horizon (Fig. 4b). Accessory
248 minerals identified are zircon, Ti oxides, monazite, xenotime, thorite, columbite and cassiterite.

249 The bulk concentrations of major and minor elements reflect the mineralogical evolution. The
250 Al₂O₃ and TiO₂ contents increase toward the surface with a maximum in the bauxitic horizon (41.5
251 and 1.0 wt%, respectively) and then slightly decrease in the ferruginous duricrust (Fig. 3c). The
252 concentration of Fe₂O₃ increases up to 25 wt% in the duricrust while SiO₂ is depleted. The
253 concentrations of Zr, Sn, Nb and Th increase from the bottom to the top of the laterite (Fig. 3d). The

254 concentration of REE in the laterite is below that of the granite and displays a complex evolution
255 throughout the profile (Fig. 3d). The index of lateritization ($IOL = 100 \times [(Al_2O_3 + Fe_2O_{3(T)}) / (SiO_2 + Al_2O_3 + Fe_2O_{3(T)})]$) used to quantify the stage of chemical weathering of a sample (Babechuk
256 et al., 2014) varies between 47 and 71 % from the bottom to the top of the profile. Using this index,
257 the investigated profile is considered as kaolinitized to weakly laterititized (Fig. 5a). The
258 concentrations of HFSE such as Nb, Th, Zr are positively correlated ($r = 0.81-0.99$) with the IOL
259 index (Fig. 5b).

261 The mobility of some HFSE and immobile elements were investigated in the profile
262 considering Th as immobile (Fig. 6). Globally, the REE are lost throughout the profile (Fig. 6). The
263 mass loss of most light REE (LREE) is strong in the entire profile (τ from -0.8 to -1.0) whereas the
264 mass loss of heavy REE (HREE) and Y is weak to moderate in the bottom profile (τ from -0.3 to -
265 0.6) and then reaches 0.8–0.9 of mass loss in the uppermost part of the profile (Table S4). The
266 evolution of the mass fraction transport function τ for Zr and Ti is similar with a mass gain in the
267 lower laterite followed by a continuous loss from -0.2 up to -0.7. Although the evolution of Hf and
268 Zr is similar along the profile, the loss of Hf is lesser compared to Zr. (Fig. 6). On the scale of the
269 profile, we note a moderate mass gain of Nb and Ta through the profile ($\tau = 0.1-0.7$). Finally, we
270 note a net gain of Sn of 1.0–2.0 relative to the parent rock with a maximum occurring in the
271 ferruginous duricrust ($\tau = 2.8$).

272 3.2 Identification and characterization of the niobium carriers

273 In the granitic parent rock, ilmenite is the main Nb-bearing phase. Ilmenite forms euhedral
274 crystals of tens of micron isolated in millimetric biotite crystals (Fig. 7a, b). It is chemically
275 characterized by high Mn and Nb concentrations (up to 10 wt % MnO and 2 wt % Nb₂O₅, Table 1).
276 Niobium-bearing ilmenite (Fig. 7c) is replaced by another Nb-bearing Ti-Fe oxide (Fig. 7d). This
277 mineral is presumably pseudorutile (Table 1) according to its Ti/(Fe+Ti) ratio, between 0.6 and 0.7,

278 and low totals obtained by microprobe analysis expected from its hydration (Grey et al., 1994).
279 Alternatively, this phase could be a mixture between ilmenite and rutile. Micron-size crystals of Nb-
280 bearing rutile also occur within the biotite matrix. Two types of Nb-bearing oxides occurring as
281 trace minerals are also found. Micrometric Nb-bearing inclusions in ilmenite exhibiting preferential
282 orientation (Fig. 7a) may be ascribed to fergusonite-(Y) or to Y-enriched columbite. Ferrocolumbite
283 with slight enrichment in Sc is also identified (Fig. 7b).

284 In the altered horizons, based on chemical and textural differences, two types of Nb-bearing
285 ilmenite (i.e. (i) preserved and (ii) altered ilmenite) and of Nb-bearing Ti oxides (i.e. (i) type-1 and
286 (ii) type-2) are identifiable (Fig. 8). We found preserved Nb-bearing ilmenite in all the horizons. In
287 the bauxitic horizon, Nb-bearing ilmenite are found within lamellae of kaolinite (Fig. 8a). Compared
288 to the parent rock, Nb-bearing ilmenite is also manganoan but with a little less Mn (typically *ca*
289 5.0 wt% MnO) and with lower Nb concentrations (*ca* 1.0 wt % Nb₂O₅, Table 2). The replacement of
290 Nb-bearing ilmenite by pseudorutile as observed in the granite (Fig. 7d) is not found in the laterite.
291 Instead, some crystals can be described as pseudorutile surrounding an ilmenite core (Fig. 8b). Other
292 ilmenite crystals display signs of alteration such as corrosion features and cracks (Fig. 8c,d).
293 Although altered in all the horizons, the most evident signs of alteration of Nb-bearing ilmenite are
294 observed in the bauxitic horizon.

295 Type-1 Nb-bearing Ti oxide with low amounts of Fe (*ca* 5 wt% Fe₂O₃) and without Mn is
296 only observed in the bauxitic horizon. Significant amounts of Th observed on the EDS spectra of
297 type-1 Ti oxide (*ca* 0.5 wt% ThO₂) may result from a substitution of Th for Ti (Gamaletsos et al.,
298 2011). The presence of nanoscale Th-bearing inclusions in Ti oxide is also expected, as this Th for
299 Ti substitution is limited by the large difference in ionic radii (1.0 and 0.61 Å). Thorium adsorption
300 is excluded considering the low solubility of Th. The composition of these oxides could not be
301 analyzed using EPMA due to their sub-micrometric size. Type-1 Ti oxide occurs as elongated

302 lamellae crystallizing between kaolinite sheets (Fig. 8e) and as poorly crystalline minerals
303 surrounding kaolinite (Fig. 8f). Pure Nb-bearing Ti oxide is also observed (type-2 Ti oxide, Fig. 8 g,
304 h). The crystals are larger than for type-1 Ti oxide ($> 20 \mu\text{m}$) and they contain more Nb than
305 ilmenite (Table 2).

306 Micrometric size Fe oxides with low Nb contents (0.1–0.3 Nb_2O_5 wt%, Table 3) are also
307 evidenced (Fig. 9a, b). The last Nb-bearing phase identified is columbite-(Mn) and -(Fe) (Table 4),
308 found as fragmented grains (10–20 μm) within the kaolinite matrix (Fig. 9c). Columbite-(Fe) with
309 low amounts of Sc is also found as inclusion in Nb-bearing ilmenite (Fig. 9d). The detection of Ti on
310 the EDS spectrum may result from the interaction volume of the electron beam, larger than the
311 columbite size (Fig. 9d). As for type-1 Ti oxide, the size of the inclusions of columbite prohibits an
312 accurate determination of their composition, an EDS spectrum is shown instead (Fig. 9d). The
313 qualitative evolution of Nb speciation along the regolith obtained by SEM observations is
314 summarized in Table 5.

315 3.3 Characteristics of the X-ray absorption spectra of niobium carriers

316 The atomic environment of Nb in type-1 and type-2 Ti oxides, Fe oxides, ilmenite and
317 columbite has been characterized using Nb L_3 -edge micro-XANES spectroscopy and compared with
318 spectra of mineral references (Bollaert et al., 2023).

319 The Nb XANES spectra of columbite (Fig. 10a) and Fe oxide (Fig. 10b) from Pitinga match
320 those of the mineral references. We find two distinct spectra between type-1 and type-2 Ti oxides.
321 The position and intensity of the features of type-1 Ti oxide match the reference spectrum of Nb-
322 bearing anatase considering the resolution of the spectra (Fig. 10c). The relative intensity of the
323 features of type-2 Ti oxide spectrum better matches that of Nb-bearing rutile (Fig. 10d). As the
324 spectra of mineral references match those of the Nb carriers from Pitinga, the reference spectra can

325 be used as components in the linear combination fitting procedure (Fig. 11,12). In absence of a
326 reference for Nb-bearing ilmenite spectrum, we compare the spectrum of Pitinga ilmenite with Nb-
327 bearing hematite and anatase in order to highlight the specificities of this spectrum (Fig. 10e). The
328 micro-XANES spectrum of ilmenite being distinct from the other spectra, it will be further used as a
329 fitting component.

330 3.4 Quantitative evolution of the average niobium speciation

331 A least-squares linear combination fitting procedure of Nb L₃- and K-edges XANES spectra
332 on bulk samples was performed using spectra of relevant mineral references (Fig. 10, S1) to
333 quantitatively follow the average Nb speciation along the profile (Fig. 11, 12).

334 The Nb L₃-edge spectrum of the clayey mottled horizon can be reproduced with the spectra
335 of Nb-bearing ilmenite (60 %) and anatase (40 %), identified by SEM as the major Nb-bearing
336 phases (Fig. 11). For the lateritic horizon, the combination of Nb-bearing ilmenite and Nb-bearing
337 anatase spectra does not reproduce the L₃-edge spectrum (Fig. S2) and the addition of a 50%
338 contribution of the spectrum of Nb-bearing Fe is required (Fig. 11). The result of the fit does not
339 change depending on whether Nb-bearing hematite or goethite spectra is used due to their
340 resemblance (Fig. 10b). In the bauxitic horizon, the contribution of Nb-bearing Fe oxides reaches
341 60 % (Fig. 11). Average Nb speciation derived from Nb L₃-edge XANES is summarized in Table 5.

342 The Nb XANES spectrum of the granite is only collected at the Nb K-edge due to low Nb
343 concentration (45 ppm) preventing the acquisition of a spectrum with a good signal-to-noise ratio at
344 the L-edge. The fit is not satisfactory with ilmenite and rutile reference spectra, which are the main
345 Nb carriers in the granite according to SEM-EDS observations (Fig. S3). This discrepancy stems
346 from major differences between the spectra of reference ilmenite and Pitinga ilmenite which was
347 characterized by microanalysis (Fig. S4). The Nb K-edge XANES spectrum of Pitinga ilmenite has
348 matching features in energy and relative intensity with the granite spectrum thus suggesting it is the

349 main Nb carrier (Fig. S4). The low signal-to-noise ratio of the μ -XANES spectrum of Pitinga
350 ilmenite prevents the quantitative analysis of the Nb speciation of the granite (Fig. 12). In the
351 following fits, the spectra of Nb-bearing rutile and anatase will be used to account for the
352 proportions of Nb-bearing Ti oxides and ilmenite in absence of relevant Nb-bearing ilmenite
353 spectrum. As a complement to the L₃-edge which allows determining the nature of the Ti oxides, the
354 fit at the K-edge (Fig. 12) is thus used to distinguish the nature of the Nb-bearing Fe oxides, which
355 have distinct spectra at the K-edge (Fig. S1). We obtain a satisfactory fit of the clayey mottled
356 horizon spectrum only with Nb-bearing Ti oxides (60 % anatase and 40 % rutile). Like at the L₃-
357 edge, the combination of Nb-bearing ilmenite and Ti oxides spectra does not reproduce the K-edge
358 spectra of the lateritic and bauxitic horizons (Fig. S5). The best fit for the lateritic horizon spectrum
359 is found when Nb-bearing goethite spectrum (50 %) is added to Nb-bearing Ti oxides (Fig. 12). In
360 the bauxitic horizon, the fit with only Nb-bearing goethite and Ti oxides components fails to
361 reproduce the features of the edge (Fig. S6). As ilmenite contributes less to the average Nb
362 speciation in this horizon, this difference is more likely due to a missing component. The addition of
363 15 % of columbite, for which SEM-EDS analyses indicate increasing abundance in the top horizons
364 (Fig. 9c), to 60 % goethite and 25 % anatase significantly improves the quality of the fit (Fig. 12).
365 We also collected the Nb K-edge spectrum of the duricrust horizon which stands out by an intense
366 pre-edge at ca. 18, 980 eV and a low intensity feature between the two main features (Fig. 12). This
367 spectrum can only be fitted with a high proportion of columbite (50 %) and with Nb-bearing
368 goethite (50 %). On average, Nb speciation obtained at the K-edge is consistent with that provided at
369 the L₃-edge (Table 5).

370

4. DISCUSSION

371 4.1 Origin of the parent niobium mineralization

372 The Nb mineralization of the biotite granite from Pitinga is dominated by ilmenite and rutile
373 with minor columbite (Fig. 7). The equilibrium temperature and pressure of formation of the granite
374 were estimated to be 800–860 °C and 7.5 kbar (Lenharo et al., 2003). The Nb/Y value of the parent
375 rock is typical of a A2-type granite indicating emplacement in post-collisional environment (Ferron
376 et al., 2010). The low Nb/Ta ratio of the granite (*ca* 7) arises from the early crystallization of micas
377 ($D_{\text{Nb}}/D_{\text{Ta}} = 2.00$) (Stepanov et al., 2014; Ballouard et al., 2016). The presence of ilmenite rather than
378 magnetite indicates that the melt first evolved in reducing conditions ($< \text{Ni-NiO}$ buffer, Ishihara,
379 1977). The formation of pseudorutile (Fig. 7d) is controlled by an increase of the oxygen fugacity of
380 the magma body during its ascent, which increases the $\text{Fe}^{3+}/\text{Fe}^{2+}$ ratio (Anand, 1984). This supports
381 the hypothesis of Lenharo et al., (2003) suggesting that the biotite granite underwent partial
382 crystallization followed by emplacement at shallow levels. Reduced Fe and Mn from the ilmenite
383 structure are oxidized and released explaining the lower amount of Mn in pseudorutile (Schroeder et
384 al., 2002).

385 At the late magmatic stage, the exsolution of a Y-rich Nb oxide from ilmenite, either
386 fergusonite-(Y) or yttracolumbite (Fig. 7d), is controlled by substitution of heterovalent cations
387 other than Fe^{2+} , namely Mn^{2+} , Nb^{5+} and Ta^{5+} and by the oxidation state of Fe. The manganoan nature
388 of ilmenite may have prevented greater incorporation of Nb due to the limited solubility of Nb in
389 manganoan ilmenite at low temperature (Capitani, 2017). Despite its rarity, the presence of
390 yttracolumbite is likely as columbite has been reported in the studied biotite granite (Lenharo et al.,
391 2003) and as inclusions in ilmenite (Beurlen and Thomas, 2006; Capitani, 2017). Similar orientation
392 of exsolved phases may result from the crystallization process of columbite during exsolution which
393 follows the crystallographic orientation of the octahedral chains of ilmenite (Capitani, 2017). The
394 exsolution process may have concentrated Y and HREE explaining the formation of a Y-rich
395 columbite.

396 4.2 Crystal-chemical and geochemical controls on niobium mobility in the critical zone

397 The combination of SEM (Fig. 8e, f) and XANES data (Fig. 11,12) demonstrates the
398 alteration of ilmenite due to chemical weathering. Ilmenite can be resistant to supergene alteration
399 (Berger et al., 2014; Janots et al., 2015) or prone to weathering (Mordberg et al., 2001; Du et al.,
400 2012) depending on the geochemical conditions or on its crystal-chemistry. In oxidic weathering
401 environment, Mn-rich ilmenite would be destabilized due to the possible formation of soluble Mn^{3+}
402 and Mn^{4+} easing its weathering. In the studied profile, Nb speciation is therefore controlled by the
403 capacity of alteration of primary ilmenite, releasing Nb in supergene fluids and leading to
404 scavenging by secondary Fe and Ti oxides (Fig. 11,12).

405 Goethite is the main Nb-bearing Fe oxide as shown by the fit of K-edge XANES spectra of
406 lateritic samples (Fig. 12, Fig. S7) and by crystal-chemistry of the Fe oxides (Table 3). The
407 scavenging of Nb by Fe oxides is favored by the formation of goethite at the expense of hematite
408 under cool and humid tropical climate with near-neutral pH fluids (Schwertmann, 1983; Kämpf and
409 Schwertmann, 1983), in agreement with the expected conditions of alteration in the studied profile.

410 According to XRD results, anatase is the main Ti oxide in the profile but rutile is also
411 detected in the bauxitic horizon and in the duricrust. The fingerprint analysis of type-1 and type-2 Ti
412 oxides XANES spectra indicates they may be anatase and rutile, respectively (Fig. 10c, d). Type-2
413 Ti oxide is probably primary, inherited from the biotite granite. As Nb-bearing rutile component is
414 not required in the fitting procedure (Fig. 11), the proportion of Nb hosted in rutile is likely minor at
415 the scale of the profile.

416 4.3 Atomic-scale processes controlling niobium mobility during supergene weathering

417 Atomic-scale mechanisms of Nb incorporation into supergene phases will control the affinity
418 of Nb for Ti and Fe oxides and Nb mobility after its release from ilmenite. They can be deduced

419 from the XANES signature of supergene Nb carriers (Fig. 10).

420 Goethite is a major host of Nb in the weathering profile (Table 5). Due to similar ionic radius
421 ($[^6\text{Fe}^{3+}] = 0.645 \text{ \AA}$ and $[^6\text{Nb}^{5+}] = 0.64 \text{ \AA}$, - Shannon, 1976) substitution between Nb^{5+} and Fe^{3+} is made
422 possible by the formation of Fe vacancies (5Fe^{3+} for 3Nb^{5+}) which maintain charge balance (Bollaert
423 et al., 2023). In goethite, the $\text{Nb}^{5+}-\text{Fe}^{3+}$ substitution process could be accompanied by the loss of two
424 H^+ occupying the tunnels between the double chains which would lead to the formation of
425 $\text{NbO}_5(\text{OH})$ octahedra. Although goethite from Pitinga incorporates only 0.1–0.3 % Nb_2O_5 , goethite
426 incorporating more than 1 % Nb_2O_5 has been reported in supergene environments (Wall et al.,
427 1996 ; Melgarejo et al., 2012). Experimental studies have shown that goethite can accommodate up
428 to 0.3 apfu (atom per formula unit) of Nb (Oliveira et al., 2008, 2009) but increasing Nb content
429 leads to a decrease of the crystallinity due to the transformation of double octahedral chains into
430 single ones as a result of the withdrawal of H atoms (Bolanz et al., 2013). The absence of Nb in
431 hematite could be due to the limited affinity of Nb for this phase. Niobium ions would occupy face-
432 sharing octahedra of the hematite structure leading to a strong electrostatic repulsion between
433 cations (Bolanz et al., 2013).

434 The occurrence of Nb-adsorbed Fe oxides is unlikely at the concentrations found in goethite
435 (*ca* 0.20 wt % Nb_2O_5) considering the low solubility of Nb ($< 1.10^{-7} \text{ mol.kg}^{-1}$, Peiffert et al. 2010).
436 The similarity of Nb-substituted goethite reference XANES spectrum with that of the Pitinga
437 goethite also excludes the possibility of Nb adsorption onto Fe oxides. Such chemisorption process
438 occurring onto hematite and goethite is nonetheless possible in near-neutral pH with 80–90 % of Nb
439 being adsorbed (Ghosh et al., 2017) but only at extremely low Nb concentrations (*ca* $2.10^{-9} \text{ mol.L}^{-1}$).

440 The comparison between type-1 and type-2 Ti oxide XANES spectra with their respective
441 reference counterparts (Fig. 10) gives access to the mechanisms of incorporation of Nb in Ti oxides.
442 The significant shift of 0.5 eV of the second feature on type-1 Ti oxide XANES spectrum relative to

443 Nb-bearing anatase of reference mirrors distinct Nb environments (Fig. 10c). The presence of Fe in
444 type-1 Ti oxide may ease Nb incorporation by a co-substitution of Nb^{5+} and Fe^{3+} for 2Ti^{4+} ($2\text{Ti}^{4+} =$
445 $\text{Fe}^{3+} + \text{Nb}^{5+}$) and may prevent significant distortion of the local environment of Nb in type-1 Ti
446 oxide. Such substitution, expected to occur in Nb-bearing Ti oxide from other laterites (Giovannini
447 et al., 2020) is common in presence of high amounts of Fe such as it is the case in the studied
448 profile. By contrast, the similarity between the spectra of type-2 Ti oxide and Nb-bearing rutile
449 reference indicates similar Nb environment and therefore similar mechanisms of substitution. In
450 absence of Fe, the incorporation of Nb results in the increased distortion of the Nb site and Nb–O
451 distances (Bollaert et al., 2023). In this case, the incorporation of Nb is made possible by the
452 formation of one Ti^{4+} vacancy for four Nb^{5+} which leads to the increase of Nb–O distances (Bollaert
453 et al., 2023).

454 4.4 Selection of an immobile element for mass-balance calculations in the profile

455 The choice of the immobile element is a challenging task in mass-balance calculations as it is
456 critical in determining element leaching or redistribution within the regolith and constraining
457 biogeochemical cycles (Oh and Richter, 2005). In addition to Nb, typical elements chosen as
458 immobile include Al, Ti, Zr, Hf, Ta (Grant, 2005; Szilas et al., 2016). However, the mineral carriers
459 of these elements display signs of alteration in the studied profile. Ilmenite (Fig. 8c) and zircon (Fig.
460 S6) are altered in the lateritic units making Ti, Zr and Hf unsuitable for such purpose. Tantalum,
461 being carried by ilmenite, it cannot be used either. Given the textural characteristics of weathering-
462 resistant cassiterite (SnO_2) found in the lateritic units (Fig. S6), Sn may be the least mobile element
463 of the alteration profile. However, the use of Sn as a conservative element in mass-balance
464 calculations leads to extreme mass loss of other relatively immobile elements (up to 90 % for Ti, Nb
465 and Zr) which seems inconsistent with the relative abundance of Ti oxides and zircon. Such low
466 values could be explained by an addition of Sn in the profile. The Sn concentration is 250 times

467 higher in the albite-enriched granite facies nearby than in the studied parent granite (Table S5). Even
468 limited lateral transport of material from the laterite developed over the Sn-enriched granite would
469 lead to a significant Sn mass gain in the studied laterite through a nugget effect. The use of Sn thus
470 skews the results of mass-balance calculations leading to overestimation of elemental mass loss.
471 Such lateral transport is evidenced mineralogically by the rare presence of Zr-enriched thorite (Fig.
472 9b) and columbite-(Mn) (Fig. 9c), absent in the parent rock but present in the surrounding enriched
473 albite-enriched granite facies (Alves et al., 2018). As the profile is located on a slope downstream of
474 the mineralized laterite (Fig. 1c), these minerals may have been brought in the form of solid particles
475 to the studied profile by circulation of sub-surficial fluids or by mechanical sliding. Their occurrence
476 is unlikely related to heterogeneities between the two granitic facies given the continuity of the
477 geochemical data along the profile (Fig. 3c, d) and the absence of minerals typical of the albite-
478 enriched granite (such as pyrochlore, columbite, cryolite) in the biotite granite.

479 Despite signs of monazite weathering mirrored by the mass loss of LREE (Table S4), the
480 weak signs of alteration of thorite (Fig. S8) and the smaller difference in Th concentration between
481 the studied parent rock and the neighboring albite-enriched granite (6-fold enrichment, Table S5),
482 Th was chosen as the best conservative element for mass-balance calculations.

483 4.5 Assessment of Nb mobility at the scale of the profile

484 As evidenced by the weathering of ilmenite and further incorporation of Nb into Ti oxides
485 and goethite, the remobilization of Nb at the micrometric scale implies potential mass loss of Nb at
486 the scale of the profile which would result in τ_{Nb} close to 0 or slightly negative. Mass-balance
487 calculations nonetheless indicate a moderate mass gain of Nb relative to the parent rock all along the
488 profile (Fig. 6). Together with the contribution of columbite to Nb speciation in the bauxitic horizon
489 and duricrust (Fig. 12), this supports the hypothesis of lateral transport. The importance of such
490 lateral transport has been shown in laterites, where it affects directly the vertical evolution of trace

491 elements such as Ni and Sc along alteration profiles (Quesnel et al., 2017; Chassé et al., 2019). The
492 very high contribution of columbite in the fit of the duricrust XANES spectrum could be due to a
493 higher proportion of columbite in the analyzed powdered sample than in the rest of the sample. This
494 nugget effect is corroborated by the τ_{Nb} value in the duricrust (0.24), lower than that of the bauxitic
495 horizon (0.58).

496 Due to the *ca* 30-fold difference in the concentrations of Nb between the albite-enriched and
497 biotite granite facies (Table S5), lateral transport masks potential losses of Nb due to supergene
498 weathering and prevents to conclude on the mobilization of Nb at the profile scale. This may explain
499 the absence of correlation between the index of lateritization (IOL) and the mass fraction transport
500 function τ (Fig. 13a). Given the poor solubility of oxide species that Nb could form under the
501 slightly acidic to near-neutral pH prevailing in laterites (Peiffert et al., 2010) and the affinity of Nb
502 for supergene Fe and Ti oxides demonstrated by this study, Nb mobility is likely low and
503 constrained to the scale of mineralogical assemblages. Along with this significant influence of pH
504 on Nb speciation and therefore on Nb mobility, the effect of the presence of inorganic (sulphate,
505 chloride, phosphate) and organic ligands remains to be explored (Friis and Casey, 2018)

506 4.6 Comparison with other 'immobile' elements

507 The behavior of other elements considered as immobile during supergene weathering was
508 also studied. The study of Ta and Sn dynamics is hindered by mass gain resulting from lateral
509 transport. The weak correlation obtained between the IOL index and τ for Ta (Fig. 13b) can be
510 explained in the same way as for Nb because Ta, which is 30 times more enriched in the albite
511 granite, substitutes for Nb in columbite which is transported to the studied profile. For Sn, the
512 magnitude of τ is explained by lateral transport due the extreme differences in Sn concentration
513 between the two source granites (by a factor of 250). The positive correlation showing that τ
514 increases with the IOL index (Fig. 13c) and thus with depth (Fig. 5a), suggests that lateral transport

515 occurs in the entire profile and is more important in the top horizons. The negative correlations
516 between the mass loss experienced by Ti (Fig. 13d), Zr (Fig. 13e) and Hf (Fig. 13f) and the IOL
517 index evidence the increasing mobility of these elements with increasing degree of weathering. For
518 these elements, lateral transport does not compensate the net mass loss induced by supergene
519 weathering because the difference of concentrations between the biotite and albite facies is less
520 significant (Table S5).

521 Several 'immobile' elements proved to be mobile in specific conditions of alteration (Braun
522 et al., 2005). Titanium is mobile at the macroscopic scale due to intense weathering and
523 complexation with organic matter (Cornu et al., 1999). Zirconium was found to be mobile in
524 weathering profiles formed over rocks with alterable Zr-bearing silicates (Duvallet et al., 1999) but
525 also with zircon which experienced metamictization (Balan et al., 2001). Intense and long-term
526 alteration processes are responsible for a series of precipitation-dissolution reactions which induce
527 Sc remobilization at least at the local scale (Chassé et al. 2019). Along with this study, possible
528 mobility of 'immobile' elements shows that the choice of the best chemical invariant for mass-
529 balance calculations must be a thought-out choice according to the conditions of formation of the
530 studied profile and to the mineralogy of the source rock to accurately quantify element fluxes in the
531 critical zone.

532 **4.7 Influence of primary mineralization on Nb mobility**

533 The role of the parent rock on the fate of Nb in the inherited lateritic profile is illustrated by
534 the mineralogy of lateritic profiles resulting from the weathering of different granitic facies from
535 Pitinga. These profiles formed within the same geodynamic context and under similar climatic
536 conditions but over different granitic facies and distinct primary Nb-bearing phases. In the laterites
537 formed from the albite-enriched facies, pyrochlore and columbite have been preserved despite the
538 fragmentation of columbite into thin particles and the Nb released from pyrochlore and columbite

539 later incorporated secondary Fe oxides (Horbe and da Costa, 1999; Alves et al., 2018). The
540 differences of Nb speciation in laterites from the Pitinga region are related to the diversity of
541 primary Nb minerals in the parent rock. Lateritization processes will be more amenable to mobilize
542 Nb from rocks containing Nb-bearing ilmenite and rutile than from rarer lithotypes bearing
543 pyrochlore or columbite phases which are more resistant to supergene alteration.

544 CONCLUSION

545 The speciation of Nb along an entire lateritic regolith has been determined and quantitatively
546 monitored with a multiscale approach combining SEM and EPMA microanalyses and XANESS
547 data. The Nb mineralogy of the parent biotite granite, characterized by Ti-bearing oxides, i.e.
548 ilmenite and rutile, is distinct from that encountered in the surrounding Nb-enriched Pitinga granites
549 with complex multicomponent Nb oxides. Lateritization processes result in the formation of Nb-
550 bearing Ti oxides (2–6 wt% Nb₂O₅) and goethite (0.1–0.3 wt% Nb₂O₅) at the expense of primary
551 carriers in the altered horizons.

552 The mobility of Nb in the studied profile results from the alteration of primary Nb-bearing
553 minerals which releases Nb in supergene fluids. However, the high affinity of Nb for Ti oxides and
554 goethite, evidenced at the atomic-scale using XANES spectroscopy, limits the mobility of Nb to the
555 scale of the mineralogical assemblage. Mass-balance calculations evidence a mass gain of Nb in the
556 upper laterite which is explained by lateral transport of surrounding Sn-Nb-enriched laterite formed
557 from the albite-enriched granite. XANES data show that this mass gain of Nb comes from a lateral
558 transfer of columbite towards the studied profile, in particular in the upper horizons. An enrichment
559 through lateral transport is also observed for Sn and Ta. By contrast, Zr, Hf and Ti, usually
560 immobile, are leached in the studied profile.

561 The difference of Nb speciation in laterites from the Pitinga region is related to the diversity of
562 primary Nb minerals with different degrees of resistance to weathering. The use of Nb as a chemical

563 invariant must be carefully considered as the mobility of Nb is dependent on the processes
564 controlling primary Nb mineralization.

565 **ACKNOWLEDGMENTS**

566 The ANR RECA (ANR-17-CE01-0012) project and Adriana Horbe are acknowledged for
567 providing samples for this study. Benoit Baptiste, Damien Deldicque, Ludovic Delbes and Imène
568 Estève, as well as Michel Fialin and Nicolas Rividi from Sorbonne Université–CAMPARIS
569 microprobe facility, are acknowledged for their help in sample preparation and analysis. We
570 acknowledge SOLEIL for provision of synchrotron radiation beamtime and thank the staff of the
571 LUCIA and SAMBA beamlines for their assistance in the measurement of Nb L_{2,3}-edges (Proposal
572 No. 20191239) and K-edge spectra (Proposal No. 20200363). This work has been conducted in the
573 framework of the PhD thesis of Q. Bollaert at ED398 GRNE, Sorbonne University.

574 **RESEARCH DATA**

575 Research Data associated with this article can be accessed at
576 <https://doi.org/10.5281/zenodo.7376789>.

577 **APPENDIX A. SUPPLEMENTARY MATERIAL**

578 Supplementary material includes additional information on semi-quantitative analysis (Table
579 S1), geochemical data (Table S2), mass-transfer calculations (Tables S3 and S4) and a comparison
580 between the concentration of the biotite granite and the mineralized albite-enriched facies (Table
581 S5). Supplementary figures show the Nb K-edge spectra of references (Fig. S1), comparison of
582 spectra and tests of fits (Figs. S2, S3, S4, S5, S6, S7) and backscattered images of trace minerals
583 more or less resistant to weathering (Fig. S8).

- Alves M. A. da S., Pereira V. P., Bastos Neto A. C. and Menegotto E. (2018) Weathering of the Madeira world-class Sn-Nb-Ta (Cryolite, REE, U, Th) deposit, Pitinga Mine (Amazon, Brazil). *J. Geochem. Explor.* 186, 61–76.
- Anand R. R. (1984) Weathering of Ilmenite in a Lateritic Pallid Zone. *Clays Clay Miner.* 32, 363–374.
- Anand R. R. and Gilkes R. J. (1984) Mineralogical and chemical properties of weathered magnetite grains from lateritic saprolite. *J. Soil Sci.* 35, 559–567.
- Babechuk M. G., Widdowson M. and Kamber B. S. (2014) Quantifying chemical weathering intensity and trace element release from two contrasting basalt profiles, Deccan Traps, India. *Chem. Geol.* 363, 56–75.
- Balan E., Neuville D. R., Trocellier P., Fritsch E., Muller J.-P. and Calas G. (2001) Metamictization and chemical durability of detrital zircon. *Am. Mineral.* 86, 1025–1033.
- Ballouard C., Massuyeau M., Elburg M. A., Tappe S., Viljoen F. and Brandenburg J.-T. (2020) The magmatic and magmatic-hydrothermal evolution of felsic igneous rocks as seen through Nb-Ta geochemical fractionation, with implications for the origins of rare-metal mineralizations. *Earth-Sci. Rev.* 203, 103–115.
- Ballouard C., Poujol M., Boulvais P., Branquet Y., Tartèse R. and Vignerresse J.-L. (2016) Nb-Ta fractionation in peraluminous granites: A marker of the magmatic-hydrothermal transition. *Geology* 44, 231–234.
- Bastos Neto A. C., Pereira V. P., Ronchi L. H., de Lima E. F. and Frantz J. C. (2009) The world-class Sn, Nb, Ta, F (Y, REE, Li) deposit and the massive cryolite associated with the albite-enriched facies of the Madeira a-type granite, Pitinga Mining District, Amazonas state, Brazil. *Can. Mineral.* 47, 1329–1357.
- Berger A., Janots E., Gnos E., Frei R. and Bernier F. (2014) Rare earth element mineralogy and geochemistry in a laterite profile from Madagascar. *Appl. Geochem.* 41, 218–228.
- Bettencourt J. S., Juliani C., Xavier R. P., Monteiro L. V. S., Bastos Neto A. C., Klein E. L., Assis R. R., Leite W. B., Moreto C. P. N., Fernandes C. M. D. and Pereira V. P. (2016) Metallogenic systems associated with granitoid magmatism in the Amazonian Craton: An overview of the present level of understanding and exploration significance. *J. South Am. Earth Sci.* 68, 22–49.
- Beurlen H. and Thomas R. (2006) Manganocolumbite and cassiterite exsolution lamellae in ilmenite from the Pitombeiras pegmatite (Acari - Rio Grande do Norte) in the Borborema pegmatitic province, NE-Brazil. 16, 3–15.
- Blake G. R. (1965) Bulk Density. In *Methods of Soil Analysis* (ed. C. A. Black).

- Bolanz R. M., Bläss U., Ackermann S., Ciobotă V., Rösch P., Tarcea N., Popp J. and Majzlan J. (2013) The Effect of Antimonate, Arsenate, and Phosphate on the Transformation of Ferrihydrite to Goethite, Hematite, Feroxyhyte, and Tripuhyite. *Clays Clay Miner.* 61, 11–25.
- Bollaert Q., Chassé M., Elnaggar H., Juhin A., Courtin A., Galois L., Quantin C., Retegan M., Vantelon D. and Calas G. (2023) Niobium speciation in minerals revealed by L_{2,3}-edges XANES spectroscopy. *Am. Mineral.* doi: 10.2138/am-2022-8293.
- Braun J.-J., Ngoupayou J. R. N., Viers J., Dupre B., Bedimo Bedimo J.-P., Boeglin J.-L., Robain H., Nyeck B., Freydier R., Nkamdjou L. S., Rouiller J. and Muller J.-P. (2005) Present weathering rates in a humid tropical watershed: Nsimi, South Cameroon. *Geochim. Cosmochim. Acta* 69, 357–387.
- Brimhall G. H., Christopher J. L., Ford C., Bratt J., Taylor G. and Warin O. (1991) Quantitative geochemical approach to pedogenesis: importance of parent material reduction, volumetric expansion, and eolian influx in lateritization. *Geoderma* 51, 51–91.
- Capitani G. C. (2017) Complex exsolution microstructures in ilmenite–pyrophanite from the Garnet Codera dyke pegmatite (Central Italian Alps): an electron microscopy investigation. *Mineral. Mag.* 81, 1087–1104.
- Carignan J., Hild P., Mevelle G., Morel J. and Yeghicheyan D. (2001) Routine Analyses of Trace Elements in Geological Samples using Flow Injection and Low Pressure On-Line Liquid Chromatography Coupled to ICP-MS: A Study of Geochemical Reference Materials BR, DR-N, UB-N, AN-G and GH. *Geostand. Geoanalytical Res.* 25, 187–198.
- Černý P. and Ercit T. S. (1989) Mineralogy of Niobium and Tantalum: Crystal Chemical Relationships, Paragenetic Aspects and Their Economic Implications. In *Lanthanides, Tantalum and Niobium* (eds. P. Möller, Petr Černý, and F. Saupé). Springer Berlin Heidelberg, Berlin, Heidelberg. pp. 27–79.
- Chakhmouradian A. R. (2006) High-field-strength elements in carbonatitic rocks: Geochemistry, crystal chemistry and significance for constraining the sources of carbonatites. *Chem. Geol.* 235, 138–160.
- Chakhmouradian A. R. and Mitchell R. H. (1999) Niobian ilmenite, hydroxylapatite and sulfatian monazite: alternative hosts for incompatible elements in calcite kimberlite from Internatsional'naya, Yakutia. *Can. Mineral.* 37, 1177–1189.
- Chassé M., Griffin W. L., O'Reilly S. Y. and Calas G. (2019) Australian laterites reveal mechanisms governing scandium dynamics in the critical zone. *Geochim. Cosmochim. Acta* 260, 292–310.
- Chassé M., Griffin W. L., O'Reilly S. Y. and Calas G. (2017) Scandium speciation in a world-class lateritic deposit. *Geochem. Perspect. Lett.*, 105–114.
- Cornu S., Lucas Y., Lebon E., Ambrosi J. P., Luizão F., Rouiller J., Bonnay M. and Neal C. (1999) Evidence of titanium mobility in soil profiles, Manaus, central Amazonia. *Geoderma* 91, 281–295.

- Costi H. T., Dall'agnol R. and Moura C. A. V. (2000) Geology and Pb-Pb Geochronology of Paleoproterozoic Volcanic and Granitic Rocks of Pitinga Province, Amazonian Craton, Northern Brazil. *Int. Geol. Rev.* 42, 832–849.
- Du X., Rate A. W. and Gee M. A. M. (2012) Redistribution and mobilization of titanium, zirconium and thorium in an intensely weathered lateritic profile in Western Australia. *Chem. Geol.* 330–331, 101–115.
- Duvallet L., Sabatier U. P. and Jules-Guesde A. (1999) The mobility of zirconium and identification of secondary Zr-bearing phases in bauxite from Poços de Caldas, Minas Gerais, Brazil: A mass-balance and X-ray absorption spectroscopic study. *Can. Mineral.* 37, 635–651.
- Ferron J. M. T. M., Bastos Neto A. C., Lima E. F., Nardi L. V. S., Costi H. T., Pierosan R. and Prado M. (2010) Petrology, geochemistry, and geochronology of Paleoproterozoic volcanic and granitic rocks (1.89–1.88 Ga) of the Pitinga Province, Amazonian Craton, Brazil. *J. South Am. Earth Sci.* 29, 483–497.
- Filella M. and May P. M. (2020) The aqueous solution thermodynamics of niobium under conditions of environmental and biological interest. *Appl. Geochem.* 122, 104729.
- Fonda E., Rochet A., Ribbens M., Barthe L., Belin S. and Briois V. (2012) The SAMBA quick-EXAFS monochromator: XAS with edge jumping. *J. Synchrotron Radiat.* 19, 417–424.
- Friis H. and Casey W. H. (2018) Niobium Is Highly Mobile As a Polyoxometalate Ion During Natural Weathering. *Can. Mineral.* 56, 905–912.
- Gamaletsos P., Godelitsas A., Mertzimekis T. J., Göttlicher J., Steininger R., Xanthos S., Berndt J., Klemme S., Kuzmin A. and Bárdossy G. (2011) Thorium partitioning in Greek industrial bauxite investigated by synchrotron radiation and laser-ablation techniques. *Nucl. Instrum. Methods Phys. Res. Sect. B Beam Interact. Mater. At.* 269, 3067–3073.
- Ghosh M., Swain K. K. and Verma R. (2017) Interaction of niobium with iron-oxide colloids and the role of humic acid. *J. Environ. Radioact.* 178–179, 101–109.
- Giovannini A. L., Mitchell R. H., Bastos Neto A. C., Moura C. A. V., Pereira V. P. and Porto C. G. (2020) Mineralogy and geochemistry of the Morro dos Seis Lagos siderite carbonatite, Amazonas, Brazil. *Lithos* 360–361, 105433.
- Grant J. A. (2005) Isocon analysis: A brief review of the method and applications. *Phys. Chem. Earth Parts ABC* 30, 997–1004.
- Grey I. E., Watts J. A. and Bayliss P. (1994) Mineralogical nomenclature: pseudorutile revalidated and neotype given. *Mineral. Mag.* 58, 597–600.
- Hickmott D. and Spear F. S. (1992) Major-and Trace-Element Zoning in Garnets from Calcareous Pelites in the NW Shelburne Falls Quadrangle, Massachusetts: Garnet Growth Histories in Retrograded Rocks. *J. Petrol.* 33, 965–1005.

- Horbe A. M. C. and da Costa M. L. (1999) Geochemical evolution of a lateritic Sn–Zr–Th–Nb–Y–REE-bearing ore body derived from apogranite: the case of Pitinga, Amazonas — Brazil. *J. Geochem. Explor.* 66, 339–351.
- Ishihara S. (1977) The magnetite-series and ilmenite-series granitic rocks. *Min. Geol.* 27, 293–305.
- Janots E., Bernier F., Brunet F., Muñoz M., Trcera N., Berger A. and Lanson M. (2015) Ce(III) and Ce(IV) (re)distribution and fractionation in a laterite profile from Madagascar: Insights from in situ XANES spectroscopy at the Ce L_{III}-edge. *Geochim. Cosmochim. Acta* 153, 134–148.
- Kämpf N. and Schwertmann U. (1983) Goethite and hematite in a climosequence in southern Brazil and their application in classification of kaolinitic soils. *Geoderma* 29, 27–39.
- Kurtz A. C., Derry L. A., Chadwick O. A. and Alfano M. J. (2000) Refractory element mobility in volcanic soils. *Geology* 28, 683–686.
- Lenharo S. L. R., Pollard P. J. and Born H. (2003) Petrology and textural evolution of granites associated with tin and rare-metals mineralization at the Pitinga mine, Amazonas, Brazil. *Lithos* 66, 37–61.
- Lukyanova E. V., Akinfiev N. N., Zotov A. V., Rass I. T., Kotova N. P. and Korzhinskaya V. S. (2017) Niobium in hydrothermal systems related to alkali granites: Thermodynamic description of hydroxo and hydroxofluoride complexes. *Geol. Ore Depos.* 59, 305–314.
- MacLean W. H. and Barrett T. J. (1993) Lithochemical techniques using immobile elements. *J. Geochem. Explor.* 48, 109–133.
- Mathieu D., Bernat M. and Nahon D. (1995) Short-lived U and Th isotope distribution in a tropical laterite derived from granite (Pitinga river basin, Amazonia, Brazil): Application to assessment of weathering rate. *Earth Planet. Sci. Lett.* 136, 703–714.
- McMaster S. A., Ram R., Faris N. and Pownceby M. I. (2018) Radionuclide disposal using the pyrochlore supergroup of minerals as a host matrix—A review. *J. Hazard. Mater.* 360, 257–269.
- Meinhold G. (2010) Rutile and its applications in earth sciences. *Earth-Sci. Rev.* 102, 1–28.
- Melgarejo J. C., Costanzo A., Bambi A. C. J. M., Gonçalves A. O. and Neto A. B. (2012) Subsolidus processes as a key factor on the distribution of Nb species in plutonic carbonatites: The Tchivira case, Angola. *Lithos* 152, 187–201.
- Milnes A. R. and Fitzpatrick R. W. (1989) Titanium and Zirconium Minerals. In *Minerals in Soils Environments (2nd edition)* (eds. J. B. Dixon and S. B. Weed). pp. 1031–1206.
- Mitchell R. H. (2015) Primary and secondary niobium mineral deposits associated with carbonatites. *Ore Geol. Rev.* 64, 626–641.
- Monteiro H. S., Vasconcelos P. M. P., Farley K. A. and Lopes C. A. M. (2018) Age and evolution of diachronous erosion surfaces in the Amazon: Combining (U-Th)/He and cosmogenic ³He records. *Geochim. Cosmochim. Acta* 229, 162–183.

- Mordberg L. E., Stanley C. J. and Germann K. (2001) Mineralogy and geochemistry of trace elements in bauxites: the Devonian Schugorsk deposit, Russia. *Mineral. Mag.* 65, 81–101.
- Nahon D. and Merino E. (1996) Pseudomorphic replacement versus dilation in laterites: petrographic evidence, mechanisms, and consequences for modelling. *J. Geochem. Explor.* 57, 217–225.
- Newville M. (2013) Larch: An Analysis Package for XAFS and Related Spectroscopies. *J. Phys. Conf. Ser.* 430, 012007.
- Oh N.-H. and Richter D. D. (2005) Elemental translocation and loss from three highly weathered soil–bedrock profiles in the southeastern United States. *Geoderma* 126, 5–25.
- Oliveira L. C. A., Ramalho T. C., Souza E. F., Gonçalves M., Oliveira D. Q. L., Pereira M. C. and Fabris J. D. (2008) Catalytic properties of goethite prepared in the presence of Nb on oxidation reactions in water: Computational and experimental studies. *Appl. Catal. B Environ.* 83, 169–176.
- Oliveira L. C. A., Zaera F., Lee I., Lima D. Q., Ramalho T. C., Silva A. C. and Fonseca E. M. B. (2009) Nb-doped hematites for decomposition of isopropanol: Evidence of surface reactivity by in situ CO adsorption. *Appl. Catal. Gen.* 368, 17–21.
- Peiffert C., Nguyen-Trung C., Palmer D. A., Laval J. P. and Giffaut E. (2010) Solubility of B-Nb₂O₅ and the Hydrolysis of Niobium(V) in Aqueous Solution as a Function of Temperature and Ionic Strength. *J. Solut. Chem.* 39, 197–218.
- Quesnel B., de Veslud C. L. C., Boulvais P., Gautier P., Cathelineau M. and Drouillet M. (2017) 3D modeling of the laterites on top of the Koniombo Massif, New Caledonia: refinement of the per descensum lateritic model for nickel mineralization. *Miner. Deposita* 52, 961–978.
- Schaefers F., Mertin M. and Gorgoi M. (2007) KMC-1: A high resolution and high flux soft x-ray beamline at BESSY. *Rev. Sci. Instrum.* 78, 123102.
- Schroeder P. A., Le Golvan J. J. and Roden M. F. (2002) Weathering of ilmenite from granite and chlorite schist in the Georgia Piedmont. *Am. Mineral.* 87, 1616–1625.
- Schwertmann U. (1983) Effect of pH on the Formation of Goethite and Hematite from Ferrihydrite. *Clays Clay Miner.* 31, 277–284.
- Shannon R. D. (1976) Revised effective ionic radii and systematic studies of interatomic distances in halides and chalcogenides. *Acta Crystallogr.* A32, 751–767.
- Soil Survey Staff (2017) Keys to soil taxonomy. *U. S. Dep. Agric. Wash. DC USA*, 639.
- Stepanov A., Mavrogenes J., Meffre S. and Davidson P. (2014) The key role of mica during igneous concentration of tantalum. *Contrib. Mineral. Petrol.* 167, 1009.
- Szilas K., Maher K. and Bird D. K. (2016) Aluminous gneiss derived by weathering of basaltic source rocks in the Neoproterozoic Storø Supracrustal Belt, southern West Greenland. *Chem. Geol.* 441, 63–80.

Vantelon D., Trcera N., Roy D., Moreno T., Mailly D., Guilet S., Metchalkov E., Delmotte F., Lassalle B., Lagarde P. and Flank A.-M. (2016) The LUCIA beamline at SOLEIL. *J. Synchrotron Radiat.* 23, 635–640.

Wall F., Williams C. T., Woolley A. R. and Nasraoui M. (1996) Pyrochlore from Weathered Carbonatite at Lueshe, Zaire. *Mineral. Mag.* 60, 731–750.

585 **Figure 1.** Geological context of the sampled profile: (a) location and geological map of the Pitinga
586 region (modified after Costi et al. 2000); (b) close-up of the sampling region indicating Nb
587 concentration in the different facies of the Madeira pluton. (c) Geological cross-section close to the
588 studied profile (modified after Bastos Neto et al. 2009).

589 **Figure 2.** Backscattered electron images of representative mineral assemblages in the parent granite:
590 (a) perthitic feldspars (Kfs) and local alteration of biotite (Bt); (b) EDS map of perthitic feldspars;
591 (c) zircon (Zrn), monazite (Mzn), ilmenite (Ilm) and xenotime (Xtm) around biotite; (d) ilmenite,
592 monazite and zircon within biotite.

593 **Figure 3.** Mineralogical and geochemical evolution of the studied profile: (a) log diagram of the
594 profile; (b) bulk mineralogy obtained by semi-quantitative analyses of X-ray diffraction patterns
595 (Ab: albite, Kfs: alkaline feldspar, Qz: quartz, Bt: biotite, Hem: hematite, Gt: goethite, Kln:
596 kaolinite, Ant: anatase, Gbs: gibbsite); (c) evolution of the concentration of major and (d) some
597 selected trace elements including HFSE with depth.

598 **Figure 4.** Backscattered electron images representative of the different horizons: (a) clayey mottled
599 horizon with quartz (Qz) and zircon (Zrn) grains dispersed in the kaolinite (Kln) matrix; (b) lateritic
600 horizon with scattered ilmenite (Ilm), zircon and Fe oxides in a gibbsite matrix; (c) bauxitic horizon
601 exhibiting large crystals of quartz distributed in a clay and gibbsite matrix on the scale of the thin
602 section; (d) stacking of kaolinite lamellae are associated to partially altered ilmenite. Numerous
603 micrometric crystals of Fe oxides are embedded in gibbsite (Gbs).

604 **Figure 5.** Evolution of the degree of weathering of the profile: (a) mass $\text{SiO}_2\text{--Al}_2\text{O}_3\text{--Fe}_2\text{O}_{3(\text{T})}$ (SAF)
605 ternary plot showing the degree of weathering experienced and the sampling depth; (b) correlations
606 between the index of lateritization (IOL) and Nb, Sn, Th, Zr. Pearson correlation coefficient r is
607 calculated for each element.

608 **Figure 6.** Mass fraction transport function τ obtained by mass-balance calculations as a function of
609 depth for Nb, Ta, Zr, Hf, Sn, Y and Ti using Th as the immobile element of reference. Uncertainties
610 on τ are between ± 1 and 20%.

611 **Figure 7.** Backscattered electron images of the Nb carriers of the parent granite: (a) inclusion of Nb-
612 Y-bearing oxide (Nb-Y ox.) in Nb-bearing ilmenite (Nb ilm) that crystallized within biotite (Bt); (b)
613 partial replacement of Nb-bearing ilmenite by Sc-rich ferrocolumbite (Sc-Clb); (c) mineralogical
614 relationship between Nb-bearing ilmenite, monazite (Mnz), biotite and zircon (Zrn); (d) close-up on
615 (c) showing the exsolution of Nb-Y-bearing oxide in Nb-bearing ilmenite. Niobium-bearing ilmenite
616 is replaced by Nb-bearing pseudorutile (Nb pdrt).

617 **Figure 8.** Backscattered electron images of the different types of Nb-bearing ilmenite and Ti oxides
618 found in the laterite: (a) EDS-SEM map of a heterogeneous crystal of Nb-bearing ilmenite (Nb ilm)
619 within kaolinite (Kln) in the bauxitic horizon; (b) Nb-bearing pseudorutile (Nb pdrt) crystal in the
620 clayey mottled horizon displaying relics of Nb-bearing ilmenite and inclusions of columbite (Clb);
621 (c) and (d) typical alteration features of Nb-bearing ilmenite in a kaolinitic matrix (Kln) in the
622 bauxitic horizon; (e) type-1 Nb-bearing Ti oxide crystallized at the margins and between the
623 kaolinite layers (Kln); (f) porous and fibrous type-1 Ti oxide crystallizing at the margins of
624 kaolinite; (g) large crystal of pure Nb-bearing Ti oxide associated with monazite (Mnz) in the
625 lateritic horizon; (h) euhedral type-2 Ti oxide intergrown with kaolinite.

626 **Figure 9.** Backscattered electron SEM images of Nb-bearing Fe oxides and columbite: (a) micron-
627 size Nb-bearing Fe oxides in the clay matrix; (b) Zr-rich thorite probably mixed to Fe oxides
628 overgrown by Nb-bearing Fe oxides; (c) euhedral crystal of columbite-(Mn) (Clb-Mn) in a kaolinite
629 matrix (sample Col.1 in Table 4); (d) inclusion of ferrocolumbite (Clb-Fe) in ilmenite. The region of
630 analysis of associated SEM-EDS spectra is indicated by the star symbols.

631 **Figure 10.** Fingerprint analysis of Nb L₃-edge XANES spectra with reference spectra: (a)
632 columbite; (b) Nb-bearing Fe oxides (c) type-1 Nb-bearing Ti oxide; (d) type-2 Nb-bearing Ti oxide;
633 (e) Nb-bearing ilmenite.

634 **Figure 11.** Evolution of the average Nb L₃-edge XANES spectra along the profile. Least-squares
635 linear combination fitting of the altered horizons performed at the L₃-edge using Nb-bearing Fe
636 oxides (Nb-Fe), Nb-bearing anatase (Nb-ant) and Nb-bearing ilmenite (Nb-ilm).

637 **Figure 12.** Evolution of the average Nb K-edge XANES spectra along the profile. Least-squares
638 linear combination fitting of the lateritic horizons performed at the K-edge using Nb-bearing
639 goethite (Nb-gt), Nb-bearing anatase (Nb-ant), Nb-bearing rutile (Nb-rt) and columbite (Clb). The
640 spectrum of the granite cannot be reproduced in absence of relevant ilmenite spectrum, this phase
641 being the main Nb host (Fig. S3).

642 **Figure 13.** Binary plots of the index of lateritization (IOL) versus mass fraction transport function τ .
643 The increase of IOL is correlated with the decrease of τ for Ti, Zr and Hf (Pearson correlation
644 coefficient: $r > 0.8$). Uncertainties on τ are between ± 1 and 20%.

645 **Table 1.** Representative compositions of Nb carriers from the parent granite (ILM: ilmenite, PDRT:
646 pseudorutile, RT : rutile). Ti/(Ti+Fe) at% ratios between 0.6 and 0.7 are indicative of pseudorutile.

647 The sum of cations above 1 in rutile may arise from the interaction volume of the electron beam,
 648 larger than the rutile size, which may also analyze biotite and ilmenite.
 649

wt%	ILM.1	ILM.2	ILM.3	ILM.4	PDRT.1	PDRT.2	PDRT.3	RT.1	RT.2	RT.3
TiO ₂	52.66	50.17	48.76	50.69	61.93	67.06	65.62	78.84	73.19	70.00
MnO	10.09	7.60	7.49	9.55	0.78	0.47	0.68	0.38	0.42	0.40
FeO	35.76	39.20	40.29	37.38	28.76	25.73	27.12	18.04	20.15	20.00
Nb ₂ O ₅	1.60	1.74	2.11	1.87	1.17	2.25	2.32	2.74	2.57	2.50
Ta ₂ O ₅	0.11	0.38	0.22	0.31	0.08	0.08	0.18	0.22	0.34	0.30
Al ₂ O ₃	/	0.22	/	/	/	/	/	/	/	/
SiO ₂	0.08	1.30	0.03	0.11	0.70	0.91	1.36	1.25	1.75	2.00
Total	100.30	100.60	98.90	99.91	93.42	96.50	97.28	101.47	98.42	100.00
Ti/(Ti+Fe) at%	0.53	0.50	0.48	0.51	0.66	0.70	0.68	0.77	0.74	0.70
<i>Structural formulae</i>										
Ti	0.99	0.95	0.95	0.97	3.48	3.570	3.731	0.85	0.823	0.800
Mn	0.21	0.16	0.16	0.206	0.049	0.028	0.042	0.005	0.005	0.000
Fe	0.75	0.82	0.87	0.795	1.795	1.523	1.603	0.216	0.252	0.200
Nb	0.018	0.02	0.025	0.022	0.039	0.072	0.074	0.018	0.017	0.010
Ta	0.001	0.003	0.002	0.002	0.002	0.002	0.002	0.001	0.001	0.000
Al	/	0.007	/	/	/	/	/	/	/	/
Si	0.002	0.033	0.001	0.001	0.052	0.064	0.097	0.018	0.026	0.030
∑ Cations	1.971	1.993	2.008	1.996	5.417	5.259	5.549	1.108	1.124	1.040

650

651 **Table 2.** Representative compositions of Nb-bearing ilmenite (Ilm) and type-2 Ti oxide from

652 lateritic horizons. Some analyses are far from 100 % due to contamination with neighboring phases.

653 Type-1 Ti oxide could not be analyzed using EPMA.

wt%	ILM.1	ILM.2	ILM.3	ILM.4	Type-2 Ti ox.1	Type-2 Ti ox.2	Type-2 Ti ox.3
TiO ₂	52.92	53.08	53.93	55.47	98.60	92.39	96.45
MnO	4.86	6.33	4.89	4.05	/	/	0.04
Fe ₂ O ₃	42.36	42.71	35.00	36.08	0.15	0.53	0.43
Nb ₂ O ₅	0.79	0.91	1.27	1.01	1.42	3.69	2.13
Ta ₂ O ₅	0.57	0.17	0.20	0.16	0.23	0.15	0.40
Al ₂ O ₃	/	/	/	/	/	0.33	/
SiO ₂	/	/	0.01	/	0.03	0.60	/
Total	101.5	103.20	95.30	96.78	100.43	97.69	99.45

Structural formulae

Ti	0.995	0.982	1.049	1.059	0.988	0.956	0.981
Mn	0.103	0.132	0.107	0.087	/	/	0.000
Fe	0.885	0.879	0.757	0.766	0.002	0.006	0.005
Nb	0.009	0.010	0.015	0.012	0.009	0.023	0.013

wt%	ILM.1	ILM.2	ILM.3	ILM.4	Type-2 Ti ox.1	Type-2 Ti ox.2	Type-2 Ti ox.3
Ta	0.004	0.001	0.001	0.001	0.001	0.001	0.001
Al	/	/	/	/	/	0.005	/
Si	/	/	0.001	/	0.000	0.008	/
Σ Cations	1.996	2.004	1.930	1.925	1.000	0.999	1.000

654

655 **Table 3.** Representative compositions of Nb-rich goethite (Gt) from the lateritic horizons. The
656 presence of Si may result from the simultaneous analysis of kaolinite due to the interaction volume
657 of the electron, similar to the size of the analyzed Fe oxides crystals.

wt%	Gt.1	Gt.2	Gt.3	Gt.4	Gt.5	Gt.6	Gt.7
Fe ₂ O ₃	78.03	73.72	78.34	85.60	86.92	75.43	81.10
TiO ₂	1.75	1.31	0.16	2.01	0.76	0.99	1.10
Nb ₂ O ₅	0.12	0.20	0.01	0.04	0.18	0.21	0.20
Al ₂ O ₃	5.92	7.08	8.55	4.74	3.53	10.93	8.28
SiO ₂	1.74	5.59	2.09	1.05	2.27	8.98	6.07
Total	87.56	87.89	89.15	93.45	93.67	96.53	96.76

658

659 **Table 4.** Representative compositions of columbite-(Mn) (Col) from the Pitinga weathering profile.
660 The composition of Sc-rich columbite could not be determined by EPMA due to their sub-
661 micrometric size (Fig. 9d).

wt%	Col.1	Col.2	Col.3
FeO	5.76	9.10	17.92
MnO	13.68	10.40	1.97
Al ₂ O ₃	0.12	0.05	/
Nb ₂ O ₅	70.82	71.27	73.27
TiO ₂	0.22	0.87	4.39
Ta ₂ O ₅	7.50	6.95	2.88

wt%	Col.1	Col.2	Col.3
Total	98.12	98.63	100.42
<i>Structural formulae</i>			
Fe	0.298	0.464	0.848
Mn	0.717	0.537	0.094
Al	0.009	0.003	0.000
Nb	1.980	1.965	1.874
Ti	0.010	0.040	0.187
Ta	0.126	0.115	0.044
A site	2.117	2.121	2.105
B site	1.024	1.005	0.942

662

663 **Table 5.** Comparison of the average Nb speciation between the L₃- and K-edges obtained by linear
664 combination fitting procedures. Qualitative Nb speciation as observed using SEM is also indicated.
665 Relative abundances range from trace (o) to dominant (+++) phases.

Technique	Sample	Nb speciation				
		Ilmenite	Rutile	Anatase	Goethite	Columbite
SEM	Bauxitic horizon	+	/	++	+	+
	Lateritic horizon	+	/	++	+	o
	Clayey mottled horizon	++	/	++	o	o
	Biotite granite	+++	+	/	/	o
Nb L ₃ -edge		Ilmenite	Rutile	Anatase	Fe oxide	Columbite
	Bauxitic horizon	20 %	/	20 %	60 %	/

	Lateritic horizon	20 %	/	30 %	50 %	/
	Clayey mottled horizon	60 %	/	40 %	/	/
Nb K-edge		Ilmenite/Ti oxide		Goethite	Columbite	
	Duricrust horizon	/		50 %	50 %	
	Bauxitic horizon	25 %		60 %	15 %	
	Lateritic horizon	50 %		50 %	/	
	Clayey mottled horizon	100 %		/	/	

666

667

668 **Declaration of interests**

669

670 x The authors declare that they have no known competing financial interests or personal
671 relationships that could have appeared to influence the work reported in this paper.

672

673 The authors declare the following financial interests/personal relationships which may be considered as
674 potential competing interests:

675

676

677