



HAL
open science

Dielectric permittivity of C-S-H

Sofiane Ait Hamadouche, Tulio Honorio, Thierry Bore, Farid Benboudjema,
Franck Daout, Eric Vourc'H

► **To cite this version:**

Sofiane Ait Hamadouche, Tulio Honorio, Thierry Bore, Farid Benboudjema, Franck Daout, et al.. Dielectric permittivity of C-S-H. Cement and Concrete Research, 2023, 169, pp.107178. <10.1016/j.cemconres.2023.107178>. <hal-04072394>

HAL Id: hal-04072394

<https://hal.science/hal-04072394v1>

Submitted on 13 Sep 2023

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire HAL, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



HAL Authorization

Dielectric permittivity of C-S-H

Sofiane Ait Hamadouche^{a,b,*}, Tulio Honorio^a, Thierry Bore^c, Farid Benboudjema^a, Franck Daout^b, Eric Vourc'h^b

^aUniversité Paris-Saclay, CentraleSupélec, ENS Paris-Saclay, CNRS, LMPS - Laboratoire de Mécanique Paris-Saclay, 91190, Gif-sur-Yvette, France.

^bSATIE, UMR CNRS 8029, ENS Paris Saclay, Université Paris Saclay, Gif-sur-Yvette, France.

^cSchool of Civil Engineering, The University of Queensland, St Lucia, QLD 4072, Australia.

Abstract

As a porous hydrophilic phase, C-S-H is expected to exhibit a frequency-dependent dielectric response. This C-S-H property remains unknown, despite the importance of dielectric assessment for non-destructive evaluation of concrete. This study adopts a multiscale approach to estimate the complex dielectric response of C-S-H over the [0; 1000 GHz] frequency range. We perform molecular dynamics simulations to compute the frequency-dependent dielectric response of water in C-S-H from the polarization correlation formula. We consider pore sizes covering the range associated with interlayer and gel pores in C-S-H. The dielectric response obtained is anisotropic and pore size dependent, as expected in layered materials. The results at the molecular scale are then used as inputs in a homogenization model to estimate the dielectric permittivity of C-S-H gel, which we compare with estimations obtained from inverse analysis based on experiments. Our results are a valuable input for multiscale modeling of non-destructive testing and evaluation in cement-based materials.

Keywords: Electromagnetic properties; Molecular Dynamics; Anisotropy; Susceptibility; Micromechanics.

1. Introduction

Dielectric permittivity measurement is one of the methods deployed for non-destructive monitoring of concrete structures [1] and to evaluate concrete property development at early ages [2, 3]. This method is sensitive to water content thanks to the dipolar character of water molecules, which feature a higher dielectric permittivity than gas and solid phases. Besides, the broadband dielectric response of porous materials also is worth measuring because it might be correlated to characteristics such as density, shrinkage, pore size distribution, and mechanical properties [4, 5, 6]. Nevertheless, assessing such characteristics requires a fundamental understanding of the dielectric response of the pore phase and microporous constituents of the material.

Several experimental studies have investigated the dielectric permittivity of cement-based materials from cement paste up to concrete scale [7, 1, 8]. Multiscale strategies have been deployed to model the complex frequency-dependent permittivity as a function of the composition [9, 10]. Micromechanics-based multiscale approaches require the knowledge of the intrinsic permittivity of the constituent phases. This property data is lacking for most of the phases in cement systems.

Calcium silicate hydrates (C-S-H) are the major hydrated phase in ordinary cement-based materials. Still, its frequency-dependent dielectric permittivity remains unknown. Guihard et al.[10] performed measurements on a mixture of C-S-H and portlandite obtained by mixing C_3S and portlandite powders

with pure water. The authors estimated the dielectric permittivity of C-S-H through an inverse analysis, assuming that it is a real value that remains constant over the studied frequency range [200 MHz; 1GHz]. This assumption can be legitimately questioned since C-S-H exhibits meso- and microporosity contributing to the dielectric response over different frequency ranges. To date, to the author's knowledge, no direct experimental measurement or indirect estimate of the imaginary part of the dielectric permittivity of C-S-H exists.

The pore phase is the main contributor to the dielectric response of porous materials, with the composition of the ionic aqueous solution playing a significant role [6]. Confinement in micro- and mesopores is known to affect the dielectric response of nanoporous materials [11, 12, 13, 14]. Molecular simulations are particularly well-suited to evaluate the behavior of confined fluids under controlled composition and thermodynamic conditions. The technique has enabled quantifying how the dielectric response of confined fluids varies with the pore size [15, 13, 16]. Recently, the pore-size dependence of the static dielectric response of water confined in C-S-H has been calculated using molecular simulations [17]. It was observed that the in-plane static dielectric constant $\epsilon_{||}$ is proportional to $\exp(-h_p/13.22)$, for a pore thickness h_p in Å for C-S-H with Ca/Si of 1.7 and 1.1. The out-of-plane dielectric response of water confined in C-S-H is still to be quantified. Following the example of the density [18], hydrodynamics [19, 20] and self-diffusion [19, 20], the dielectric response is pore-size dependent [11, 12]. The computation of the frequency-dependent complex dielectric response of C-S-H is still to be addressed and can be useful to analyze the origins of the material dielectric response. Quantifying the pore size dependence of the dielectric response

*Corresponding author

Email address: sofiane.ait_hamadouche@ens-paris-saclay.fr
(Sofiane Ait Hamadouche)

of C-S-H might be helpful in developing dielectric methods to measure interlayer and gel pore water content.

In this work, we perform molecular dynamics (MD) simulations to obtain the dielectric response of water confined in C-S-H at the molecular scale. To study the influence of the confinement, we consider various interlayer distances ranging from 11.8 to 106.3 Å (i.e., spanning the range of pore sizes associated with interlayer and gel pores in C-S-H). The Green-Kubo relation adapted to compute the dielectric response under a (slit) confined geometry is used. The results obtained at the molecular scale are then used as inputs in a homogenization model to estimate the frequency-dependent permittivity of C-S-H gel over a frequency range of [0 ; 1000 GHz].

2. Molecular models and methods

2.1. Dielectric response from molecular simulations

Molecular dynamics simulations enable computing the time evolution of the total dipole moment $\vec{M}(t)$, which is the sum of the dipole moment $\vec{\mu}_i(t)$ of each particle i with a charge q_i . For a system with N particles:

$$\vec{M}(t) = \sum_i^N \vec{\mu}_i(t) = \sum_i^N q_i \vec{r}_i(t) \quad (1)$$

where $\vec{r}_i(t)$ is the position of the center of the particle i .

The dielectric response of water results from microscopic relaxation processes described by the auto-correlation function of water polarization $\phi_{ij}(t)$ (for $i, j = x, y, z$) [21]:

$$\phi_{ij}(t) = \frac{\langle M_i(0)M_j(t) \rangle}{V_{cav}k_B T \epsilon_0} \quad (2)$$

where k_B is the Boltzmann constant, T is the temperature (in Kelvin), ϵ_0 is the vacuum permittivity, and V_{cav} is the volume of the pore occupied by confined water. Here, this volume is computed using the sum of the Voronoi volume of the oxygen atoms in water molecules. For a Debye relaxation, $\phi_{ij}(t)$ is a simple exponential decay. Fluid-wall interaction might lead to deviations of a Debye relaxation, especially in micropores.

The frequency-dependence is computed via the Fourier transform of the components of the susceptibility tensor [22]:

$$\chi_{ij}(f) = - \int_0^\infty e^{-2\pi i f t} \frac{d\phi_{ij}(t)}{dt} dt \quad (3)$$

where f is the frequency. From Eq. (2) and (3), the susceptibility can also be written as follows [21]:

$$\chi_{ij}(f) = \phi_{ij}(0) - 2\pi i f \int_0^\infty e^{-2\pi i f t} \phi_{ij}(t) dt \quad (4)$$

The dielectric permittivity tensor ϵ can be computed directly from the susceptibility tensor. It must be noted that ϵ is neither Hermitian nor symmetrical in the general case [23]. Some simplifications can be made in the absence of an external magnetic field and considering the slit pore geometry. In a slit pore with an infinite slab of fluid confined between two walls,

the dielectric tensor can be written as a function of the parallel ϵ_{\parallel} and perpendicular ϵ_{\perp} components [22]:

$$\epsilon(f) = \begin{pmatrix} \epsilon_{\parallel}(f) & 0 & 0 \\ 0 & \epsilon_{\parallel}(f) & 0 \\ 0 & 0 & \epsilon_{\perp}(f) \end{pmatrix} \quad (5)$$

in a Cartesian frame. The frequency-dependent components of the dielectric permittivity tensor are obtained as [22]:

$$\epsilon_{\parallel}(f) = 1 + \chi_{\parallel}(f) \quad (6)$$

$$\frac{1}{\epsilon_{\perp}(f)} = 1 - \chi_{\perp}(f) \quad (7)$$

In the case of C-S-H, we need to check if the in-plane response - especially in small pores - can be approximated as $\epsilon_{\parallel}(f) \approx \epsilon_{xx}(f) \approx \epsilon_{yy}(f)$. Due to translational invariance, ϵ_{\perp} is expected to be negligible. Also, due to the slit geometry, the out-of-diagonal terms are expected to be $\epsilon_{xy}(f) = \epsilon_{yx}(f) \approx 0$. In the following, we verify the validity of these conjectures.

Finally, let us remind the usual fluctuation formula used to compute the static dielectric permittivity of isotropic fluids (e.g., [11]):

$$\epsilon = 1 + \frac{\langle |\vec{M}|^2 \rangle - \langle \vec{M} \rangle^2}{3k_B T \epsilon_0 V} \quad (8)$$

where V is the volume of the (bulk or confined) fluid. This expression can be readily obtained from Eq. (6) and various values of $\phi_{ij}(t)$ evaluated at $t = 0$, considering that the average dipole moment $\langle \vec{M} \rangle$ should be zero in an isotropic fluid.

2.2. Simulation details

The atomistic C-S-H model we use is based on Kunhi et al. [24] model with molecular formula $\text{Ca}_{1.67}\text{SiO}_{3.7} \cdot n\text{H}_2\text{O}$. The Ca/Si of 1.67 is a common value observed in C-S-H [25, 26]. To describe the interatomic interactions, we use ClayFF [27] and SPC/E water [28]. These non-polarizable force fields have been successfully deployed to get the dielectric response of a vast range of nanoporous materials. The same atomistic modeling strategy has been used to compute the thermoelastic and interfacial properties of confined water in C-S-H [29, 30], and self-diffusion [20]. An analysis of the structural features of the C-S-H model combined with this force field is also provided in ref. [29] together with stability analysis under sorption.

To study the influence of confinement, we consider various interlayer pore sizes ranging from 11.8 to 106.3 Å (Fig. 1(A)), which spans the sizes associated with interlayer and gel pores in C-S-H as in [20]. Grand Canonical Monte Carlo (GCMC) simulations were performed to fill the system with the water content corresponding to ambient conditions and (liquid) saturated conditions ($T = 300$ K and $\text{RH} = 100\%$) (see ref. [29] for details). Two layers are simulated, therefore, two pore spaces are present in the simulated systems. One pore is kept at the equilibrium basal spacing of $d_{eq} = 13.7$ Å, while the other is changed (see Fig. 1(A)). The following analysis deals with water present in only this second pore.

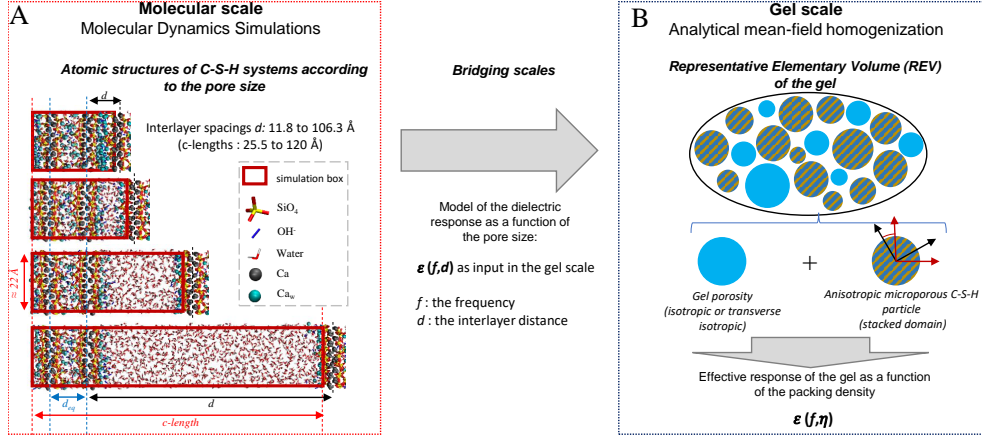


Figure 1: Multi-scale strategy adopted in this work. The dielectric response of water confined in C-S-H is computed for interlayer spacings d ranging from 11.8 to 106.3 Å. The results are then taken as inputs in a homogenization model to estimate the dielectric response at the C-S-H gel scale for different packing densities η .

We use LAMMPS [31] software for MD simulations. Simulations are performed with a timestep of $dt = 1$ fs and periodic boundary conditions. Water molecules are constrained with SHAKE algorithm [32]. For each pore size, seven configurations obtained from GCMC simulations are considered. First, a short 10 ps NVT run at $T = 300$ K with the Nosé-Hoover thermostat is performed, followed by a 20 ps NVE run. Production runs are then performed still in the NVE ensemble during 1 ns. The auto-correlations of dipole moments are computed with *fix ave/correlate* in LAMMPS using input values every 10 fs, with 10000 correlation time windows accumulated, and calculating time window averages every 100 ps. Average and standard deviations are computed according to the seven different trajectories. Independent NVT simulations are performed to compute the volume cavity V_{cav} as a sum of the Voronoï volume of water oxygens in the interlayer.

Only the water and ions in the interlayer space are time integrated during MD to gain computational efficiency and reduce fluctuations in the computation of $\phi_{ij}(t)$. Since the auto-correlation function depends on the volume cavity (Eq. (2)), time-integrating the particles in the solid layer adds more considerable variability to the values of $\phi_{ij}(t)$, especially for small interlayer spacings, in which the fluctuations due to solid layer deformability are more significant compared to the total pore volume. We have compared the results of simulations with a rigid solid layer and unconstrained atoms in the solid, and the average values are similar but with more variability associated with the second case.

For the integration of $\phi_{ij}(t)$ in Eq. (3), the upper integration limit must be chosen so that the function decay is integrated correctly by avoiding the integration of noise present in the signal from the simulation [33]. Since MD simulations are based on finite differences algorithms, very long runs are prone to accumulating numerical errors, resulting in noise present in the spectra of $\phi_{ij}(t)$. To avoid the integration of that noise and to reduce the computational time needed to compute Fourier transforms in large lists, we use two equations obtained by fitting $\phi_{ij}(t)$:

- A simple exponential decay, leading to a Debye relaxation

microscopically:

$$\phi_{ii}^{Deb}(t) = \epsilon_s \exp[-t/\tau_{Deb}] \quad (9)$$

- A polynomial exponentially decaying aiming to capture potential oscillations in the $\phi(t)$ profiles:

$$\phi_{ii}^{Poly}(t) = \epsilon_s \exp[-t/\tau_{Poly}] (a_0 + a_1 t + a_2 t^2 + a_3 t^3) \quad (10)$$

where ϵ_s is the static dielectric permittivity, τ_{Deb} and τ_{Poly} the relaxation time for each model, respectively. We provide results up to 1000 GHz which allows capturing the decrease in the imaginary response in the case of bulk water. It is noteworthy that physical phenomena involving smaller scales may arise at higher frequencies [34, 35]. The use of classical molecular simulation with non-polarizable force fields cannot of course capture the effects associated with electron polarization. The results above a few hundreds of gigahertz presented herein must be considered as the contribution of the water polarization only.

3. Upscaling the dielectric permittivity

3.1. C-S-H gel microstructure representation

We adopt a microstructure representation of C-S-H gel, assuming that it is composed of two phases: microporous solid particles and gel porosity. The former is depicted as a transverse isotropic inclusion composed of C-S-H layers stacked according to the equilibrium basal spacing $d_{eq} = 13.7$ Å, which should be the most prevalent configuration in stacked systems [29]. The gel porosity might also be considered to exhibit a transverse isotropic or a full isotropic behavior. Various conventions are used in the literature to classify the different types of porosity in C-S-H according to their sizes. Qomi et al. [36] consider the size of the gel porosity ranging from 0.5 nm to 8 nm, while Nguyen-Tuan et al. [37] define it as ranging between 1.5 nm and 10 nm. Jennings [38] proposed two characteristic

205 pore sizes for C-S-H gel porosity from small-angle scattering
 experimental data: small gel pores [1-3 nm] and larger gel pores
 [3-12 nm]. In this study, to upscale C-S-H gel properties, we
 adopt four characteristic pore diameters spanning the typical
 250 range of gel pore sizes: 2, 3, 4 and 10 nm.

210 Thus, we use as input the information obtained from molec-
 ular simulations regarding the interlayer distances of 1.37 nm
 ($=d_{eq}$) for the solid particles and 2, 3, 4, and 10 nm for the gel
 pore (Fig. 1(B)). Other gel pore sizes can easily be tested. To
 255 this aim, in the following we provide expressions that fit the
 behavior at the molecular scale, and a homogenization scheme.

3.2. Micromechanical model

Analytical micromechanics schemes are used to upscale the
 dielectric permittivity of C-S-H gel, as it was recently done for
 mass transport properties [20, 39]. By mathematical analogy,
 260 the homogenization formula derived for upscaling the diffusiv-
 ity and (thermal and electrical) conductivities can also be used
 to estimate the dielectric permittivity [40].

The following formula enables estimating the effective di-
 electric permittivity of a macroscopically isotropic heteroge-
 265 neous material composed of N (microscopically) anisotropic
 ellipsoids is (e.g. [40]):

$$\boldsymbol{\varepsilon}^{hom} = \sum_{i=1}^N f_i \boldsymbol{\varepsilon}_i \mathbf{A}_i \quad (11) \quad 270$$

where f_i , $\boldsymbol{\varepsilon}_i$ and \mathbf{A}_i are respectively the volume fraction, dielec-
 tric permittivity tensor, and the localization tensor of phase i .
 The above expression can be written as a sum, provided that
 275 the dielectric behavior per phase (or pore size) is constant.

The localization tensor \mathbf{A} can be approximated using the
 strategy proposed for the thermal conductivity [41] inspired by
 the Eshelby [42] solution in elasticity. In the case of elliptical
 inclusions randomly distributed and oriented in a representa-
 280 tive elementary volume (REV), and under the assumption of the
 self-consistent (or Bruggeman) scheme (i.e., that the homoge-
 nized medium function and the reference medium mediating
 inclusion interactions), we get [20]:

$$\mathbf{A}_i^{SC} = \mathbf{A}_i^d \left(\sum_{r=1}^N f_r \mathbf{A}_r^d \right)^{-1} \approx \mathbf{A}_i \quad (12)$$

where $\mathbf{A}_i^d = (\mathbf{I} + \mathbf{S}_i^E ((\boldsymbol{\varepsilon}^{hom})^{-1} \mathbf{R} \boldsymbol{\varepsilon}_i \mathbf{R}^T - \mathbf{D}))^{-1}$ is the localiza-
 240 tion tensor in the dilute case, with \mathbf{R} being the rotation ma-
 trix, \mathbf{I} being the identity matrix, and \mathbf{S}_i^E the components of
 the diagonal (second-rank) Eshelby-like tensor [41]. Assuming
 that all orientations of the equiaxed ("spherical") inclusions are
 245 equally possible, the expression of the homogenized dielectric
 permittivity for a two-phase heterogeneous material composed
 of transverse isotropic phases is given by the implicit equation:

$$\boldsymbol{\varepsilon}^{hom} = \frac{\frac{7(1-f_g)\varepsilon_{\parallel,s}}{\varepsilon_{\parallel,s}+2\varepsilon^{hom}} + \frac{5(1-f_g)\varepsilon_{\perp,s}}{\varepsilon_{\perp,s}+\varepsilon^{hom}} - \frac{7f_g\varepsilon_{\parallel,g}}{\varepsilon_{\parallel,g}+\varepsilon^{hom}} - \frac{5f_g\varepsilon_{\perp,g}}{\varepsilon_{\perp,g}+2\varepsilon^{hom}}}{\frac{7(1-f_g)}{\varepsilon_{\parallel,s}+2\varepsilon^{hom}} + \frac{5(1-f_g)}{\varepsilon_{\perp,s}+2\varepsilon^{hom}} - \frac{7f_g}{\varepsilon_{\parallel,g}+2\varepsilon^{hom}} - \frac{5f_g}{\varepsilon_{\perp,g}+2\varepsilon^{hom}}} \quad (13) \quad 285$$

where f_g represents the gel porosity, $\varepsilon_{\parallel,p} = \varepsilon_{x,p} = \varepsilon_{y,p}$, $\varepsilon_{z,p} =$
 $\varepsilon_{\perp,p}$ the subscripts $_{,s}$ and $_{,g}$ stand, respectively, for the solid C-
 S-H particle and gel porosity. Eq. (13) can be written in an
 explicit form, but the result is too lengthy to be reported here.
 Among the different roots, we use only the one giving positive
 non-complex values for the static response. This same root is
 used in the calculation of the full complex spectra. This general
 formulation allows accounting for anisotropy on the dielectric
 response at the gel porosity. We will discuss the relevance of
 this feature in Section 4.3.

4. Results

4.1. Static dielectric response at the molecular scale

Figure 2 shows the components of the dielectric permit-
 tivity tensor in the static limit. The in-plane component ε_{\parallel}
 ($= (\varepsilon_{xx} + \varepsilon_{yy})/2$) of the permittivity of water confined in C-
 S-H is pore size dependent, with bulk-like behavior for inter-
 layer distances exceeding roughly 6 nm (approximately 5 nm if
 the thickness of the solids layer is discounted out) and a strong
 decrease in the dielectric permittivity with the confinement of
 water. This result was also recently observed by Masoumi et
 al. [17]. This decrease can be explained by water molecules
 near solid surfaces being subjected to forces that hinder their
 polarization (an effect that would be more pronounced in hy-
 drophilic systems like C-S-H). Our results corroborate that wa-
 ter confined in C-S-H follows the universal reduction in the di-
 electric response observed for other confined fluids [15, 13, 16].
 For comparison with ε_{\parallel} , we computed the dielectric constant
 of rigid bulk SPC/E water also under ambient conditions (in a
 simulation with 900 molecules). We obtained $\varepsilon = 71.2 \pm 2.11$
 (in agreement with other simulations of SPC/E in the literature
 [43, 44, 33]), which is rather close to the experimental value
 $\varepsilon = 78.02 \pm 0.20$ (interpolated value for $T = 300$ K from the
 experimental study of [45]). For comparison, the numerical and
 experimental values of bulk water are depicted in Fig. 2(A). The
 in-plane response ε_{\parallel} obtained in our simulations can be fitted
 with the following expression:

$$\varepsilon_{\parallel} = \varepsilon_{inf} \exp\left(-\frac{d-B}{h_{charac}}\right) \quad (14)$$

where ε_{inf} is the dielectric constant of bulk water computed
 using MD simulations ($= 71.2$), d the interlayer distance, $B =$
 7.1 \AA is a shift in the interlayer distance, and $h_{charac} = 20 \text{ \AA}$
 is the characteristic distance associated with the pore size de-
 pendence of the dielectric response. When $d \gg h_{charac}$, the
 static dielectric response of C-S-H approaches that of bulk wa-
 ter. A similar expression is used to fit the results of Masoumi et
 al.[17], and they obtained a h_{charac} in the same order (13.22 \AA)
 (Note that these authors used a rigid SPC water model, and this
 water model is known to yield a lower dielectric constant when
 compared to rigid SPC/E water model, which in turn underesti-
 mate the dielectric constant of a few units in relative permit-
 tivity. For the same temperature, the authors obtained a static
 response $\varepsilon_{\parallel} = 64$ for bulk water).

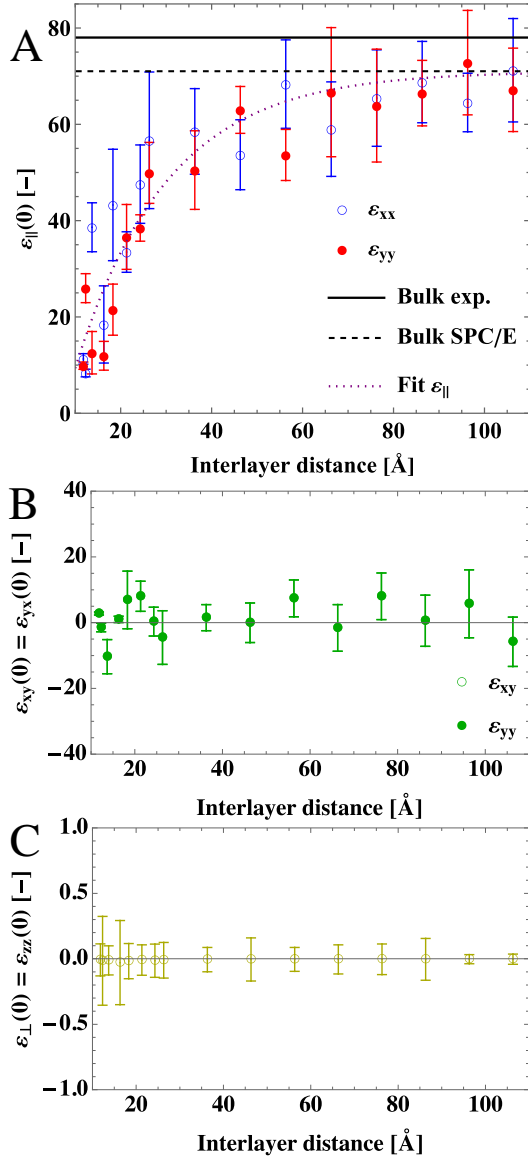


Figure 2: Components of the dielectric permittivity tensor of C-S-H at the molecular scale as a function of the interlayer distance d : (A) Parallel components ϵ_{xx} , ϵ_{yy} , and $\epsilon_{||} = (\epsilon_{xx} + \epsilon_{yy})/2$ is fitted with Eq. (14) (the associated coefficients of determination regarding the values obtained from MD are $R^2=0.845$ for ϵ_{xx} and $R^2=0.922$ for ϵ_{yy}). The values for experimental and SPC/E bulk water are provided for comparison. (B) Out-of diagonal component $\epsilon_{xy} = \epsilon_{yx}$. (C) Perpendicular component $\epsilon_{\perp} = \epsilon_{zz}$.

Our results suggest some degree of anisotropy in the in-plane response, especially for interlayer distances smaller than 25 Å. Independent fits of ϵ_{xx} and ϵ_{yy} using Eq. (14) yield similar results (with coefficients of determination of $R^2=0.845$ and $R^2=0.922$, respectively). Taking into account the variability/uncertainty associated with the MD results, the approximation $\epsilon_{||} \approx \epsilon_{xx} \approx \epsilon_{yy}$ is reasonable.

In Figure 2(B), it can be observed that the out-of-diagonal components $\epsilon_{xy} = \epsilon_{yx}$ fluctuate around 0. Figure 2(C) shows that, as expected, the values of the perpendicular component ϵ_{\perp} are negligible (with respect to the reported variability observed in the results). These results agree with the form of the dielec-

tric tensor in Eq. (5).

4.2. Frequency-dependent response at the molecular scale

Figure 3 shows the polarization correlation functions of water molecules in C-S-H pores as obtained in MD simulations. The main contribution to the dielectric response comes from the parallel components $\phi_{xx}(t)$ and $\phi_{yy}(t)$. A significant decrease in the static value (in agreement with Fig. 2) and an increase in the relaxation time (i.e., the correlation functions go to zero at larger timescales) is observed with increasing confinement. The values of the out-of-diagonal component $\phi_{xy}(t) = \phi_{yx}(t)$ and the perpendicular component $\phi_{\perp}(t)$ indicate negligible frequency-dependent responses, as expected (see Section 2.1).

Figure 4 shows the frequency-dependent permittivity of confined water in C-S-H computed by integrating the correlation functions in Eq. (4). As discussed in Section 2.2, we fit the parallel components $\phi_{xx}(t)$ and $\phi_{yy}(t)$ using Eqs. (9) and (10) to cope with the noise in MD results and gain in precision and efficiency in the computation of integrals in Eq. (4). Both permittivity profiles obtained $\epsilon^{Deb}(f)$ and $\epsilon^{Poly}(f)$ are depicted in Figure 4, along with the isotropic response of bulk water provided by experimental values according to Buchner et al. [46] and simulation values using SPC/E water according to Rinne et al. [33]. In the imaginary part, we can see that the resonance frequency decreases as we increase the interlayer distance, which shows that the relaxation frequencies of bound water are lower than that of bulk water as suggested by [47].

Figure 4(C) and (F) show the difference $\epsilon_{yy} - \epsilon_{xx}$ in the real and imaginary parts quantifying the possible anisotropy in the in-plane response. To quantify the anisotropy we compute the error $e_{ani} = \int_0^{\infty} |\epsilon_{yy} - \epsilon_{xx}| df / \int_0^{\infty} \epsilon_{bulk} df$ using as reference the response of bulk water. With this definition, the anisotropy is in the order of 34 % for $d = 13.7$ Å, and ranges between 3 % and 22 % for the other interlayer distances, being more pronounced in the 0-10 GHz range for interlayer distances above 25 Å, and in the low frequency range for smaller pores. The anisotropy can also be evaluated in Fig. 3, where we see that for interlayer distances > 13.7 , the spectra of ϕ_{yy} is inside the limits $\phi_{xx} \pm$ standard deviation, except for $d = 56.3$ Å.

Regarding the choice of fitting functions for the microscopic process, while $\epsilon^{Deb}(f)$ shows a Debye-type relaxation profile as expected, $\epsilon^{Poly}(f)$ exhibits oscillations that may not necessarily have a physical meaning. The real part of $\epsilon^{Poly}(f)$ have negative values around 1 GHz for $d = 13.7$ Å in the xx direction, whereas the dielectric permittivity of common materials (including dielectrics) is generally non-negative [48].

The results obtained so far can now be used to get an expression of the pore size dependence of the complex frequency-dependent dielectric response in C-S-H. Cole-Cole model [49], which is one of the most used relaxation models, shows a good performance in fitting $\epsilon^{Poly}(f)$. With this model, the frequency dependence in the response reads:

$$\epsilon^*(f) = \epsilon_{\infty} + \frac{\epsilon_{cc} - \epsilon_{\infty}}{1 + (i2\pi f\tau_{cc})^{1-\alpha_{cc}}} \quad (15)$$

where ϵ_{cc} and ϵ_{∞} are respectively the static and infinite frequency dielectric permittivity, τ_{cc} the relaxation time and α_{cc}

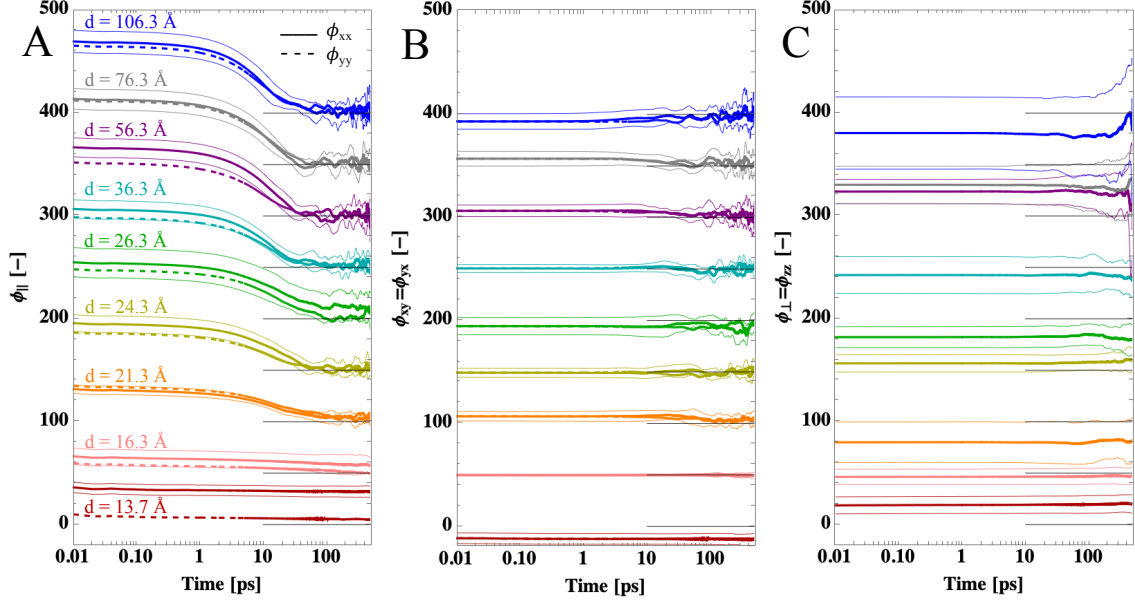


Figure 3: Polarization correlation functions of water in C-S-H for various interlayer distances: A) Parallel components ϕ_{xx} (full line), ϕ_{yy} (dashed line), (B) Out-of-diagonal component $\phi_{xy} = \phi_{yx}$, and (C) Perpendicular component $\phi_{\perp} = \phi_{zz}$. The functions regarding each interlayer distance are shifted 50 units upwards for readability (the horizontal gray line marks the zero for each curve). Thin lines represent \pm one standard deviation computed over the seven trajectories considered.

the exponent parameter which has a value between -1 and 1 (with $\alpha = 0$ corresponding to a Debye model). For ε_{cc} , we fixed the values of the static response obtained in Section 4.1 using Eq.(14) and ε_{∞} is taken equal to 1, since the force fields used in MD simulations do not include atomic polarization [6]. The parameters α_{cc} and τ_{cc} were determined by fitting (using least squares methods) $\varepsilon^{Poly}(f)$ with the Cole-Cole model. The Cole-Cole model does not fit the negative values, nor the bumps observed in the $\varepsilon^{Poly}(f)$ dielectric spectra. There is a good agreement between $\varepsilon^{Poly}(f)$ and the Cole-Cole fit in the real part as well as in the imaginary part (see Appendix 5).

Figure 5 depicts how Cole-Cole parameters τ_{cc} and α_{cc} vary as a function of the interlayer distance. These parameters can be fitted with the expressions below:

$$\tau_{cc} = A \exp\left(-\frac{d - A_0}{h_{charac}^P}\right) + \tau_{inf} \quad (16)$$

$$\alpha_{cc} = -A_1\{1 - \exp[-A_2(d - A_3)]\}^2 + A_4 \quad (17)$$

where d is the interlayer distance, $h_{charac}^P = 2.90 \text{ \AA}$ is the characteristic distance, $\tau_{inf} = 14 \text{ ps}$ is the value towards which the data converge; and A , A_0 , A_1 , A_2 , A_3 , and A_4 are other fitting coefficients whose values are reported in Table 1

The relaxation time τ_{cc} exponentially decays towards bulk water values, as expected. The relaxation frequency is the inverse of the relaxation time $\tau = 1/(2\pi f_{relaxation})$, and as discussed above, relaxation frequencies of bulk water are higher than that of bound water. Together with Eq. (14), the expressions above can be used as input in the micromechanical model aiming to estimate the dielectric response at the C-S-H gel scale.

Table 1: Coefficients of the fitting equations of α_{cc} and τ_{cc} .

Coefficients	Values
A [ps]	12655
A_0 [Å]	3.33
A_1 [-]	0.589
A_2 [1/Å]	0.307
A_3 [Å]	16.046
A_4 [-]	0.673

4.3. Dielectric response of C-S-H gel

We estimate the static and frequency-dependant permittivity of C-S-H gel using Eq. (13) in accordance with the representation of the C-S-H gel structure detailed in Section 3.1. In all cases, the microporous solid particles are represented as transverse isotropic inclusions with layers stacked according to the equilibrium basal spacing $d_{eq} = 13.7 \text{ \AA}$. In the case of gel porosity, two cases are considered:

- A 2D response (as in Eq. 5) with $\varepsilon_{x,g}(f) = \varepsilon_{y,g}(f) = \varepsilon_{\parallel,g}(f)$, and $\varepsilon_{z,g}(f) = 0$
- An isotropic response with $\varepsilon_{x,g}(f) = \varepsilon_{y,g}(f) = \varepsilon_{z,g}(f)$

Figure 6 shows the static response of C-S-H gel $\varepsilon_{gel}(0)$ as a function of the packing density $\eta = 1 - f_g$. The filled and hollow circles at $\eta = 0.64$ and $\eta = 0.74$ refer to the static permittivity of low density (LD) and high density (HD) C-S-H, respectively, the two main packing densities of C-S-H [50]. The values of the static response of LD and HD C-S-H for the 2D and isotropic cases are summarized in Table 2. Our values are in agreement with Guihard et al. [10] reporting a static dielectric permittivity of C-S-H in the range 2-20. In this reference, these values are

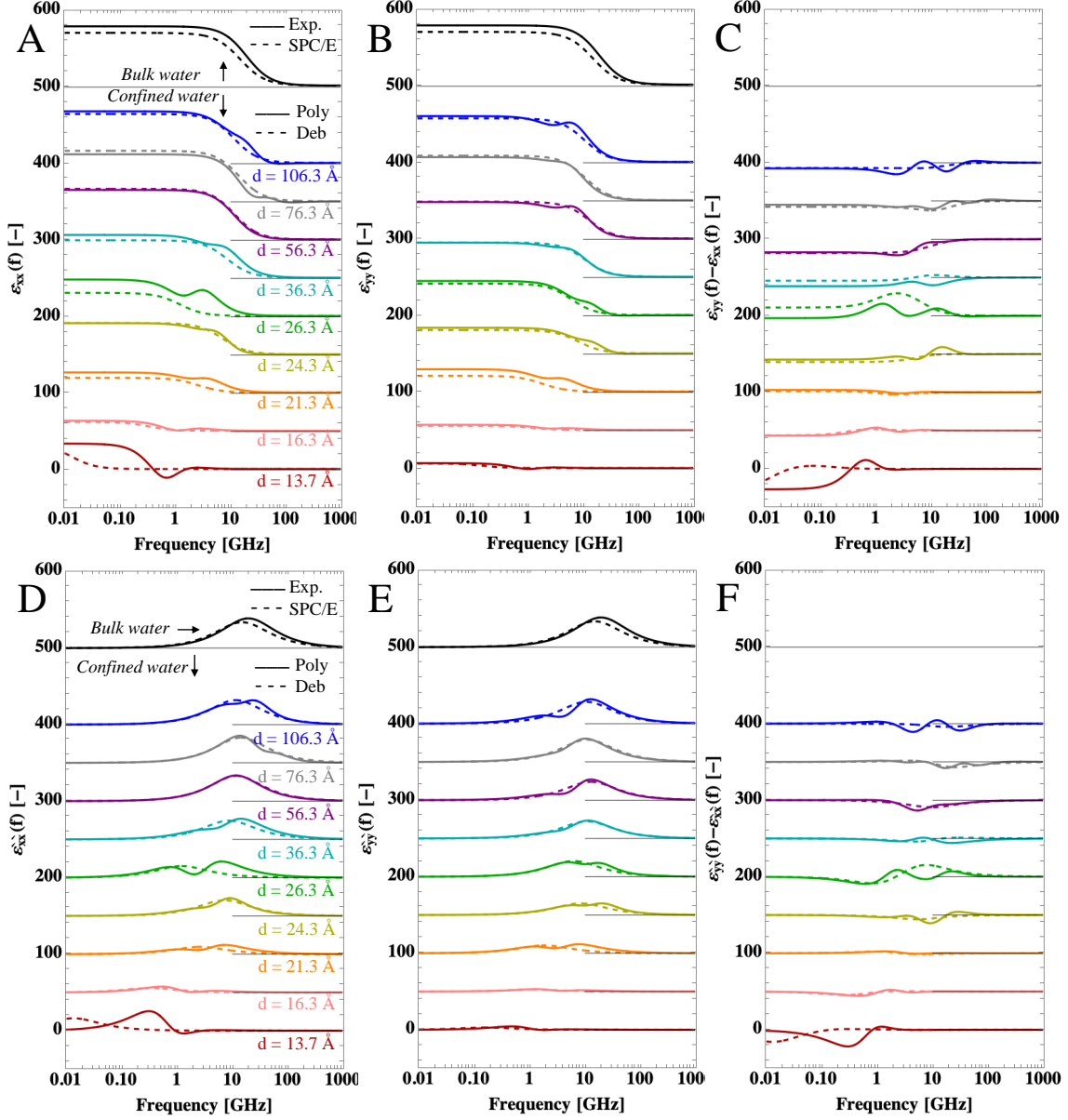


Figure 4: Dielectric permittivity of confined water in C-S-H: (A),(B) real and (D),(E) imaginary parts. The (isotropic) response of bulk water is shown for comparison (experimental values according to Buchner et al. [46] and simulation values for SPC/E water according to Rinne et al. [33]). The functions regarding each interlayer distance are shifted 50 units upwards for readability (the horizontal gray line marks the zero for each curve); bulk water values are shifted 500 units upwards. To quantify the (an)isotropy in the parallel components of the dielectric tensor, the difference $\epsilon_{yy} - \epsilon_{xx}$ is calculated for the (C) real and (D) imaginary parts of the dielectric response for selected slit pore sizes.

Table 2: Static permittivity of LD and HD C-S-H when a 2D or an isotropic behaviour is assumed for the gel porosity configurations.

	case	$d_g = 2$ nm	$d_g = 3$ nm	$d_g = 4$ nm	$d_g = 10$ nm
$\epsilon_{gel}(0)$ (LD)	2D	8.75	9.86	10.40	11.08
$\epsilon_{gel}(0)$ (HD)	2D	8.30	9.03	9.36	9.77
$\epsilon_{gel}(0)$ (LD)	Iso	15.14	18.20	19.87	22.14
$\epsilon_{gel}(0)$ (HD)	Iso	12.72	15.58	15.53	16.77

obtained through inverse analysis of experimental data in the frequency range [200 MHz; 1GHz]. Our results however show that some frequency dependency in C-S-H would arise, even at this relatively low frequency range, due to the contribution of

interlayer pores as discussed below.

Figure 7 shows the frequency dependence of the effective dielectric response of C-S-H gel. It depicts the real and imaginary parts of $\epsilon_{gel}(f)$ for both 2D and isotropic cases for gel

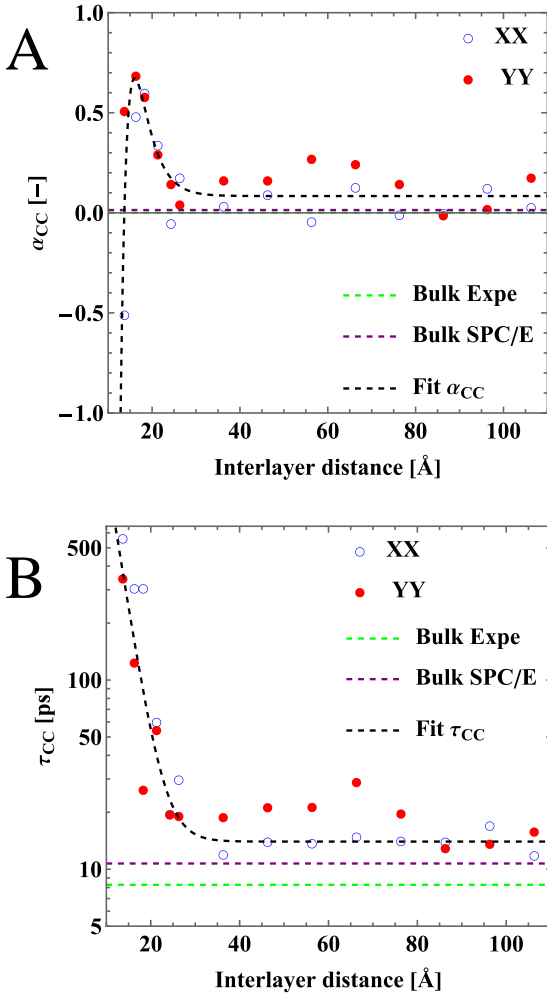


Figure 5: Variation of Cole-Cole parameters of water confined in C-S-H at the molecular scale (A) α_{cc} and (B) τ_{cc} as a function of the interlayer distance. Fits using Eqs. 16 and 17 are provided. The purple and green dashed lines represent, respectively, the Cole-Cole parameters from experiments [46] and MD simulations using the SPC/E water model [33].

porosity according to various packing densities (ranging from 0.1 to 1) and for the four characteristic gel pore sizes considered here (2, 3, 4, and 10 nm). We observe that the dielectric response at the gel scale depends the gel pore size. The dielectric response of C-S-H gel is fitted using a Cole-Cole model. As at the molecular scale, we fix the values of the static response (Fig. 6) for ε_{cc} . The variation of the two remaining Cole-Cole parameters (α_{cc} and τ_{cc}) as a function of the packing density is illustrated in Fig. 8. The behaviour of system with pores > 2 nm features the same tendency. Hence, α_{cc} and τ_{cc} presented for $d_g=4$ nm are also valid for mid to large pore gels. Nevertheless, the static response at $d_g = 10$ nm has higher values than at $d_g = 4$ nm, which induces differences at the frequency-dependent dielectric response (Fig. 7). Table 3 outlines α_{cc} and τ_{cc} of LD and HD C-S-H.

According to [51], confined water shows an anisotropic response ($\varepsilon_{||} \neq \varepsilon_{\perp}$) in pore as large as 100 nm. The authors stud-

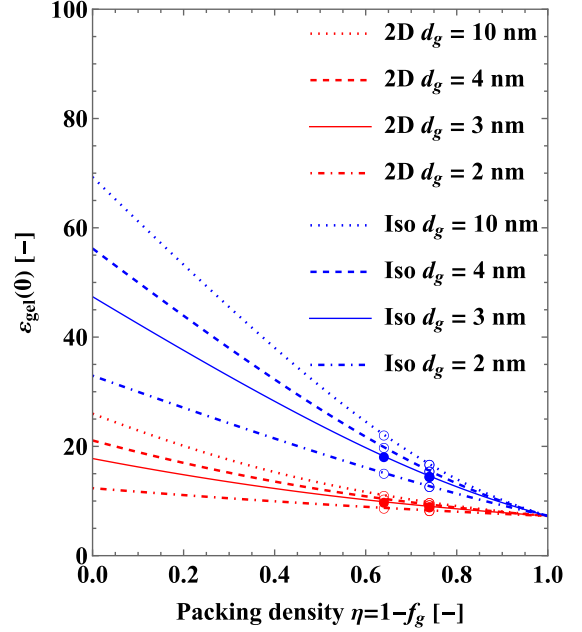


Figure 6: Static response of C-S-H gel as a function of the packing density.

ied the dielectric response of water molecules confined between hydrophobic surfaces, under the influence of external electric fields. Since the nature of the surfaces between which water molecules are confined is deemed to have little influence on the dielectric response [13], we can assume that $\varepsilon_{||} \neq \varepsilon_{\perp}$ in the gel porosity of our hydrophilic system, and therefore we consider the 2D case as the more relevant.

5. Conclusion

In this work, we computed the dielectric response of water confined in C-S-H at the molecular scale using MD simulations. These results were used as input in a micromechanical model in order to get the response at the gel scale. Our results provide valuable input for the multiscale modeling of dielectric permittivity of cement-based materials, estimating the intrinsic frequency-dependent permittivity of C-S-H gel for the first time. The main conclusions are:

- *Dielectric response of confined water in C-S-H at the molecular scale.* The static dielectric response of water confined in C-S-H decreases with confinement, confirming that the universal reduction in the dielectric response of confined fluids [15, 13, 16, 17] is also valid for C-S-H. For interlayer distances above roughly 6 nm, the dielectric response retrieves that of bulk water. The parallel component of frequency-dependent dielectric permittivity of confined water in C-S-H is adequately fitted by a Cole-Cole model with parameters depending on the pore size.
- *Anisotropy in the dielectric response at the molecular scale.* Both pore in-plane directions (x and y) show a

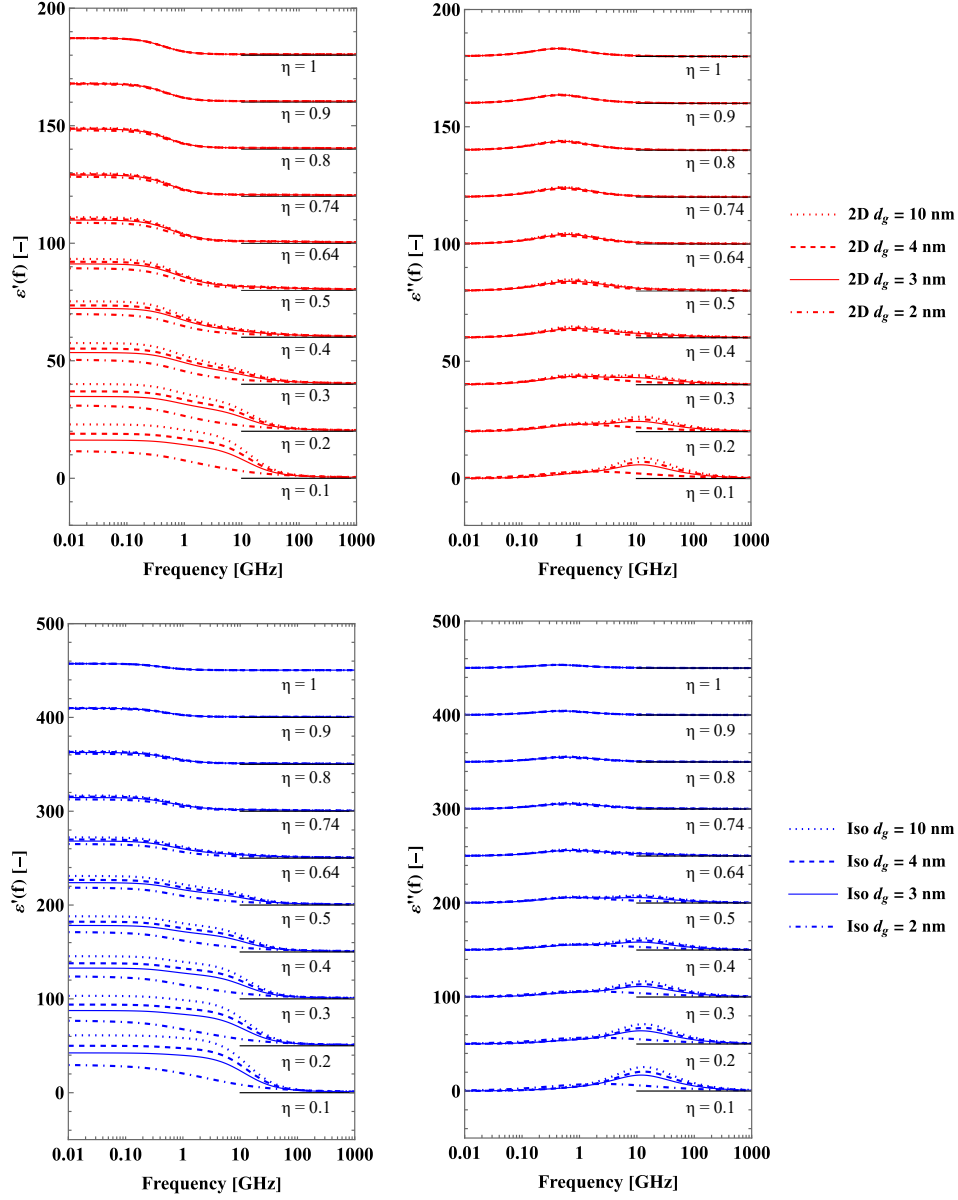


Figure 7: Frequency dependent dielectric response of C-S-H gel considering 2 cases for the response of the gel porosity ε_g : (A) a 2D response i.e. $\varepsilon_{x,g} = \varepsilon_{y,g} = \varepsilon_{\parallel,g}$ while $\varepsilon_{z,g} = \varepsilon_{\perp,g} = 0$ and (B) considering an isotropic response i.e. $\varepsilon_{\parallel,g} = \varepsilon_{\perp,g}$. The graphics for each packing density are shifted upwards 15 units in (A) and 40 units in (B) for readability..

Table 3: Cole-Cole parameters of LD and HD C-S-H.

	case	$d_g = 2$ nm	$d_g = 3$ nm	$d_g = 4$ nm	$d_g = 10$ nm
α_{cc} [-] (LD)	2D	0.024	0.041	0.048	0.051
τ_{cc} [ps] (LD)	2D	281	243	245	251
α_{cc} [-] (HD)	2D	0.018	0.017	0.016	0.016
τ_{cc} [ps] (HD)	2D	312	289	293	299
α_{cc} [-] (LD)	Iso	0.212	0.279	0.297	0.306
τ_{cc} [ps] (LD)	Iso	211	127	121	116
α_{cc} [-] (HD)	Iso	0.118	0.145	0.152	0.153
τ_{cc} [ps] (HD)	Iso	259	207	206	208

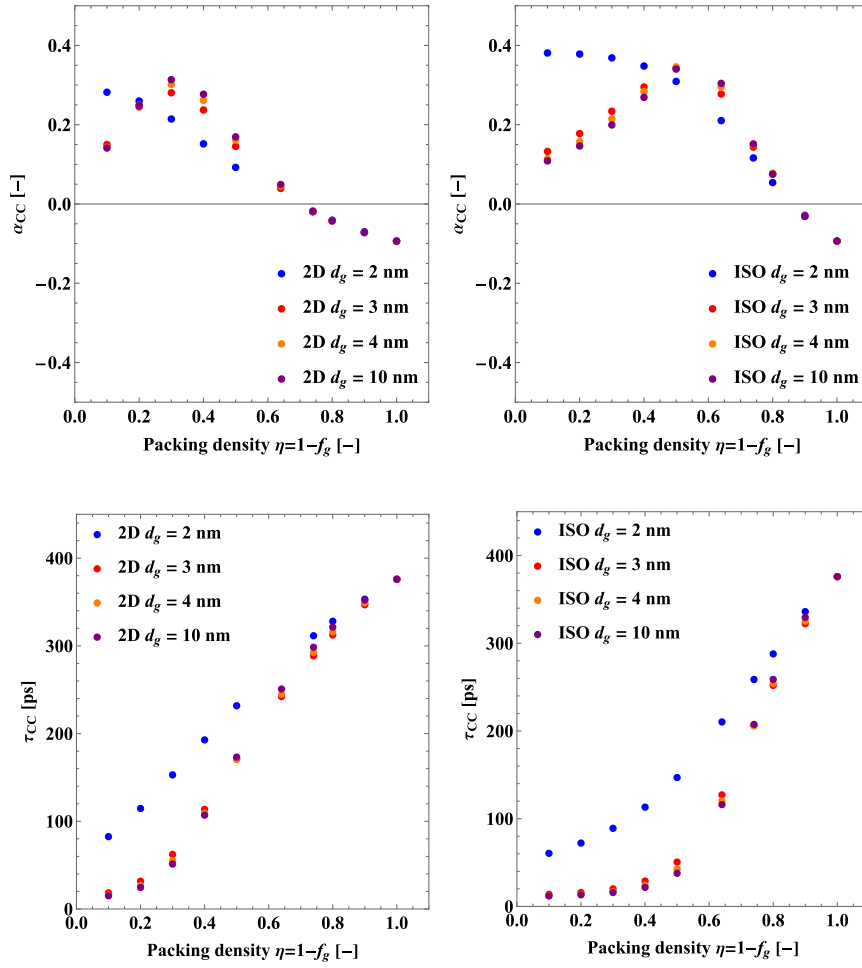


Figure 8: Variation of Cole-Cole parameters obtained from fitting the effective dielectric response of C-S-H gel as a function of the packing density η .

similar response, which justifies attributing a transverse isotropic behavior to C-S-H at the molecular scale, and that even for small interlayer pores. The perpendicular component of the dielectric tensor vanishes, which indicates a reduction in the dimensionality at the molecular level. Evidence from the study of water confined in large mesopores suggests that the anisotropic character of the response persists up to pore sizes as large as 100 nm [51], which justifies adopting a transverse isotropic behavior even for gel pores. The out-of-diagonal components of the dielectric tensor are shown to fluctuate around 0. Accounting for the anisotropy at the molecular level is critical in determining the response of C-S-H.

- *Dielectric response of C-S-H gel.* We quantify the complex dielectric response of C-S-H as a function of the packing density, which is input for multiscale approaches coping with C-S-H densification [52]. The values of the static response of LD and HD C-S-H are in the range of 2-20 identified in experimental studies [10]. The Cole-Cole model also fits pretty well the results at the gel scale. In our analysis, we considered various gel pore sizes. Our

results provide, therefore, a full characterization of the variability of the complex dielectric response of C-S-H gel, accounting for the entire range of packing density and gel pore sizes. In future micromechanical modeling of the dielectric properties of cement-based materials, C-S-H must be considered as a phase showing a complex frequency-dependent behavior.

The perspective of this work includes considering the effect of unsaturated pores and also applying the multiscale strategy to other hydrates/ microporous phases in cement systems (e.g., AF-phases).

Acknowledgement

This work has benefited from the financial support of the LabEx LaSIPS (ANR-10-LABX-0032-LaSIPS) managed by the French National Research Agency under the "Investissements d'avenir" program (ANR-11-IDEX-0003).

Appendix. Correlation between ε^{Poly} and the Cole-Cole fit

Figure 9 shows fittings of autocorrelation functions ϕ_{xx} and ϕ_{yy} using simple exponential decay and polynomial exponentially decaying for selected interlayer distances. The associated quality of fittings is assessed according to the determination coefficient (R^2) and overall standard deviation (in relative dielectric permittivity units). Overall, polynomial fits yield a better R^2 and lower standard deviation than the exponential decay.

Figure 10 shows the coefficient of determination R^2 between the Cole-Cole fit of the dielectric response at the molecular scale for both real and imaginary parts. The R^2 values are all above 0.9 for interlayer distances above 3 nm. For smaller pores, the R^2 are lower but remain in the domain of good correlation.

References

- [1] C. Andrade, V. M. Blanco, A. Collazo, M. Keddad, X. R. Nóvoa, H. Takenouti, Cement paste hardening process studied by impedance spectroscopy, *Electrochimica Acta* 44 (24) (1999) 4313–4318.
- [2] Van Beek, A., Milhorst, M.A., Dielectric measurements to characterize the microstructural changes of young concrete, *HERON*, vol. 44 (1), 2000 (Jan. 2000).
- [3] X. Z. Ding, X. Zhang, C. K. Ong, B. T. G. Tan, J. Yang, Study of dielectric and electrical properties of mortar in the early hydration period at microwave frequencies, *Journal of Materials Science* 31 (20) (1996) 5339–5345.
- [4] S. Tang, X. Cai, Z. He, W. Zhou, H. Shao, Z. Li, T. Wu, E. Chen, The review of early hydration of cement-based materials by electrical methods, *Construction and Building Materials* 146 (2017) 15–29. doi:10.1016/j.conbuildmat.2017.04.073. URL <https://linkinghub.elsevier.com/retrieve/pii/S0950061817307262>
- [5] V. P. Drnevich, A. K. Ashmawy, X. Yu, A. M. Sallam, Time domain reflectometry for water content and density of soils: study of soil-dependent calibration constants, *Canadian Geotechnical Journal* 42 (4) (2005) 1053–1065. doi:10.1139/t05-047. URL <https://www.nrcresearchpress.com/doi/abs/10.1139/t05-047>
- [6] T. Honorio, T. Bore, F. Benboudjema, E. Vourc'h, M. Ferhat, Dielectric properties of the pore solution in cement-based materials, *Journal of Molecular Liquids* 302 (2020) 112548.
- [7] V. Guihard, F. Taillade, J.-P. Balayssac, B. Steck, J. Sanahuja, F. Deby, Permittivity measurement of cementitious materials with an open-ended coaxial probe, *Construction and Building Materials* 230 (2020) 116946. doi:10.1016/j.conbuildmat.2019.116946. URL <https://www.sciencedirect.com/science/article/pii/S0950061819323888>
- [8] P. R. Camp, S. Bilotta, Dielectric properties of portland cement paste as a function of time since mixing, *Journal of Applied Physics* 66 (12) (1989) 6007–6013, publisher: American Institute of Physics. doi:10.1063/1.343577. URL <https://aip.scitation.org/doi/citedby/10.1063/1.343577>
- [9] V. Guihard, F. Taillade, J. P. Balayssac, B. Steck, J. Sanahuja, F. Deby, Prediction of Cement-Based Materials' Water Content with the Use of Electromagnetic Homogenisation Schemes, in: 2018 Progress in Electromagnetics Research Symposium (PIERS-Toyama), 2018, pp. 164–168, iSSN: 1559-9450. doi:10.23919/PIERS.2018.8597698.
- [10] V. Guihard, C. Patapy, J. Sanahuja, J.-P. Balayssac, F. Taillade, B. Steck, Effective medium theories in electromagnetism for the prediction of water content in cement pastes, *International Journal of Engineering Science* 150 (2020) 103273. doi:10.1016/j.ijengsci.2020.103273. URL <https://www.sciencedirect.com/science/article/pii/S0020722520300616>
- [11] V. Ballenegger, J.-P. Hansen, Dielectric permittivity profiles of confined polar fluids, *The Journal of Chemical Physics* 122 (11) (2005) 114711.
- [12] D. J. Bonhuis, S. Gekle, R. R. Netz, Dielectric Profile of Interfacial Water and its Effect on Double-Layer Capacitance, *Physical Review Letters* 107 (16) (Oct. 2011).
- [13] C. Zhang, F. Gygi, G. Galli, Strongly Anisotropic Dielectric Relaxation of Water at the Nanoscale, *The Journal of Physical Chemistry Letters* 4 (15) (2013) 2477–2481.
- [14] L. Fumagalli, A. Esfandiar, R. Fabregas, S. Hu, P. Ares, A. Janardan, Q. Yang, B. Radha, T. Taniguchi, K. Watanabe, G. Gomila, K. S. Novoselov, A. K. Geim, Anomalously low dielectric constant of confined water, *Science* 360 (6395) (2018) 1339–1342.
- [15] S. Senapati, A. Chandra, Dielectric Constant of Water Confined in a Nanocavity, *The Journal of Physical Chemistry B* 105 (22) (2001) 5106–5109, publisher: American Chemical Society. doi:10.1021/jp011058i. URL <https://doi.org/10.1021/jp011058i>
- [16] M. H. Motevaselian, N. R. Aluru, Universal Reduction in Dielectric Response of Confined Fluids, *ACS Nano* 14 (10) (2020) 12761–12770. doi:10.1021/acsnano.0c03173. URL <https://pubs.acs.org/doi/10.1021/acsnano.0c03173>
- [17] S. Masoumi, S. Zare, H. Valipour, M. J. Abdolhosseini Qomi, Effective Interactions between Calcium-Silicate-Hydrate Nanolayers, *The Journal of Physical Chemistry C* 123 (8) (2019) 4755–4766.
- [18] T. Honorio, Monte Carlo Molecular Modeling of Temperature and Pressure Effects on the Interactions between Crystalline Calcium Silicate Hydrate Layers, *Langmuir* 35 (11) (2019) 3907–3916.
- [19] A. Botan, B. Rotenberg, V. Marry, P. Turq, B. Noetinger, Hydrodynamics in Clay Nanopores, *The Journal of Physical Chemistry C* 115 (32) (2011) 16109–16115.
- [20] T. Honorio, H. Carasek, O. Cascudo, Water self-diffusion in C-S-H: Effect of confinement and temperature studied by molecular dynamics, *Cement and Concrete Research* 155 (2022) 106775. doi:10.1016/j.cemconres.2022.106775. URL <https://www.sciencedirect.com/science/article/pii/S0008884622000667>
- [21] R. Kf, G. S, N. Rr, Ion-specific solvation water dynamics: single water versus collective water effects, *The journal of physical chemistry. A* 118 (50), publisher: J Phys Chem A (Dec. 2014). doi:10.1021/jp5066874. URL <https://pubmed.ncbi.nlm.nih.gov/25474321/>
- [22] V. A. Froltsov, S. H. L. Klapp, Dielectric response of polar liquids in narrow slit pores, *The Journal of Chemical Physics* 126 (11) (2007) 114703. doi:10.1063/1.2566913. URL <https://aip.scitation.org/doi/full/10.1063/1.2566913>
- [23] V. M. Agranovich, V. Ginzburg, *Crystal Optics with Spatial Dispersion, and Excitons*, Vol. 42 of Springer Series in Solid-State Sciences, Springer Berlin Heidelberg, Berlin, Heidelberg, 1984. doi:10.1007/978-3-662-02406-5. URL <http://link.springer.com/10.1007/978-3-662-02406-5>
- [24] A. Kunhi Mohamed, S. C. Parker, P. Bowen, S. Galmarini, An atomistic building block description of C-S-H - Towards a realistic C-S-H model, *Cement and Concrete Research* 107 (2018) 221–235.
- [25] I. G. Richardson, Tobermorite/jennite- and tobermorite/calcium hydroxide-based models for the structure of C-S-H: applicability to hardened pastes of tricalcium silicate, β -dicalcium silicate, Portland cement, and blends of Portland cement with blast-furnace slag, metakaolin, or silica fume, *Cement and Concrete Research* 34 (9) (2004) 1733–1777.
- [26] A. J. Allen, J. J. Thomas, H. M. Jennings, Composition and density of nanoscale calcium-silicate-hydrate in cement, *Nature Materials* 6 (4) (2007) 311–316.
- [27] R. T. Cygan, J.-J. Liang, A. G. Kalinichev, Molecular Models of Hydroxide, Oxyhydroxide, and Clay Phases and the Development of a General Force Field, *The Journal of Physical Chemistry B* 108 (4) (2004) 1255–1266.
- [28] H. J. C. Berendsen, J. R. Grigera, T. P. Straatsma, The missing term in effective pair potentials, *The Journal of Physical Chemistry* 91 (24) (1987) 6269–6271.
- [29] T. Honorio, F. Masara, F. Benboudjema, Heat capacity, isothermal compressibility, isosteric heat of adsorption and thermal expansion of water

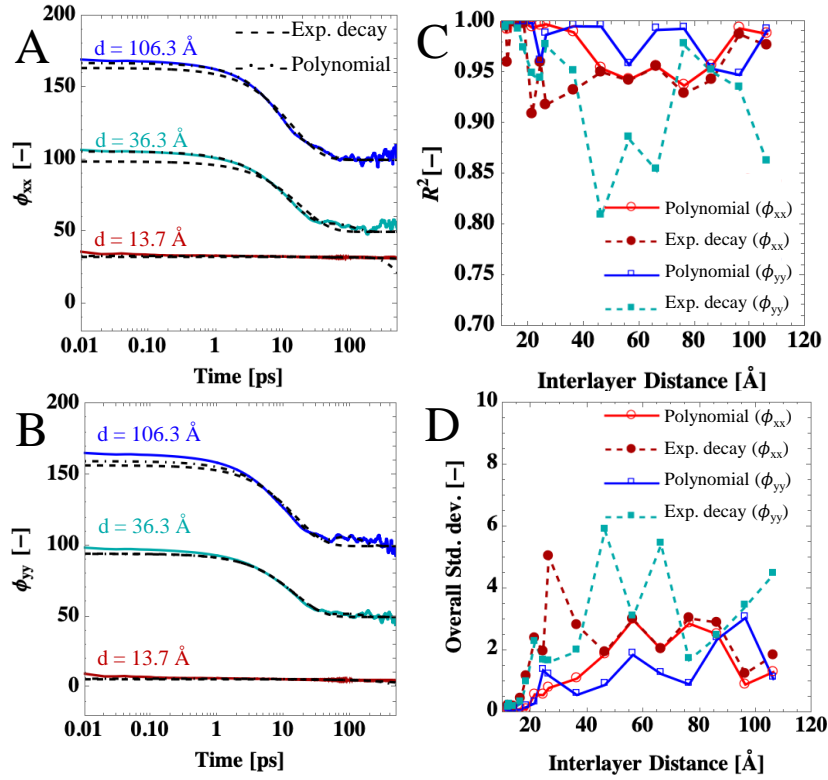


Figure 9: Fitting of autocorrelation functions (A) ϕ_{xx} and (B) ϕ_{yy} using simple exponential decay (dashed lines) and polynomial exponentially decaying (dot-dashed lines) for selected interlayer distances. The associated quality of fittings is assessed according to (C) determination coefficient (R^2) and (D) overall standard deviation (in relative dielectric permittivity units).

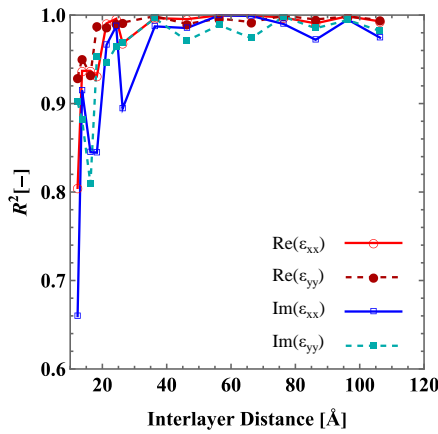


Figure 10: The coefficient of determination R^2 between the Cole-Cole fit and the real and imaginary parts of $\epsilon_{xx}^{Poly}(f)$ and $\epsilon_{yy}^{Poly}(f)$ as a function of the interlayer distances.

confined in C-S-H, Cement 6 (2021) 100015. doi:10.1016/j.cement.2021.100015.

URL <https://www.sciencedirect.com/science/article/pii/S2666549221000128>

[30] F. Masara, T. Honorio, F. Benboudjema, Sorption in C-S-H at the molecular level: Disjoining pressures, effective interactions, hysteresis, and cavitation, Cement and Concrete Research 164 (2023) 107047. doi:10.1016/j.cemconres.2022.107047.

URL <https://www.sciencedirect.com/science/article/pii/>

S0008884622003398

[31] S. Plimpton, Fast Parallel Algorithms for Short-Range Molecular Dynamics, Journal of Computational Physics 117 (1) (1995) 1–19. doi:10.1006/jcph.1995.1039.

URL <http://www.sciencedirect.com/science/article/pii/S002199918571039X>

[32] J.-P. Ryckaert, G. Ciccotti, H. J. Berendsen, Numerical integration of the cartesian equations of motion of a system with constraints: molecular dynamics of n-alkanes, Journal of Computational Physics 23 (3) (1977) 327–341. doi:10.1016/0021-9991(77)90098-5.

URL <https://linkinghub.elsevier.com/retrieve/pii/S0021999177900985>

[33] K. F. Rinne, S. Gekle, R. R. Netz, Dissecting ion-specific dielectric spectra of sodium-halide solutions into solvation water and ionic contributions, The Journal of Chemical Physics 141 (21) (2014) 214502. doi:10.1063/1.4901927.

URL <http://aip.scitation.org/doi/10.1063/1.4901927>

[34] H. Yada, M. Nagai, K. Tanaka, Origin of the fast relaxation component of water and heavy water revealed by terahertz time-domain attenuated total reflection spectroscopy, Chemical Physics Letters 464 (4) (2008) 166–170. doi:10.1016/j.cpllett.2008.09.015.

URL <https://www.sciencedirect.com/science/article/pii/S0009261408012487>

[35] G. Brodie, Modeling the very broad band dielectric properties of water, Journal of Microwave Power and Electromagnetic Energy 55 (1) (2021) 80–89, publisher: Taylor & Francis eprint: <https://doi.org/10.1080/08327823.2021.1877243>. doi:10.1080/08327823.2021.1877243.

URL <https://doi.org/10.1080/08327823.2021.1877243>

[36] M. J. Abdolhosseini Qomi, L. Brochard, T. Honorio, I. Maruyama, M. Vandamme, Advances in atomistic modeling and understanding of drying shrinkage in cementitious materials, Cement and Concrete Research 148 (2021) 106536. doi:10.1016/j.cemconres.2021.106536.

- 675 URL <https://www.sciencedirect.com/science/article/pii/S000888462100185X>
- [37] L. Nguyen-Tuan, M. A. Etzold, C. Rößler, H.-M. Ludwig, Growth and porosity of C-S-H phases using the sheet growth model, *Cement and Concrete Research* 129 (2020) 105960. doi:10.1016/j.cemconres.2019.105960.
- 680 URL <https://www.sciencedirect.com/science/article/pii/S0008884619306428>
- [38] H. M. Jennings, Refinements to colloid model of C-S-H in cement: CM-II, *Cement and Concrete Research* 38 (3) (2008) 275–289. doi:10.1016/j.cemconres.2007.10.006.
- 685 URL <https://www.sciencedirect.com/science/article/pii/S0008884607002761>
- [39] T. Honorio, Permeability of C-S-H (Jun. 2022). doi:10.2139/ssrn.4132286.
- 690 URL <https://papers.ssrn.com/abstract=4132286>
- [40] S. Torquato, *Random Heterogeneous Materials: Microstructure and Macroscopic Properties*, Springer Science & Business Media, 2002.
- [41] H. Hatta, M. Taya, Effective thermal conductivity of a misoriented short fiber composite, *Journal of Applied Physics* 58 (7) (1985) 2478–2486. doi:10.1063/1.335924.
- 695 URL <http://aip.scitation.org/doi/10.1063/1.335924>
- [42] J. Eshelby, *Progress in solid mechanics: Vol. 1*. Edited by IN Sneddon and R. Hill, North-Holland Publishing Company, Amsterdam, 1960, 448 pp., 100s (1961).
- 700 [43] M. Rami Reddy, M. Berkowitz, The dielectric constant of SPC/E water, *Chemical Physics Letters* 155 (2) (1989) 173–176. doi:10.1016/0009-2614(89)85344-8.
- URL <https://www.sciencedirect.com/science/article/pii/S0009261489853448>
- 705 [44] R. Fuentes-Azcatl, N. Mendoza, J. Alejandro, Improved SPC force field of water based on the dielectric constant: SPC/ε, *Physica A: Statistical Mechanics and its Applications* 420 (2015) 116–123. doi:10.1016/j.physa.2014.10.072.
- 710 URL <https://www.sciencedirect.com/science/article/pii/S0378437114009108>
- [45] A. P. Gregory, R. N. Clarke, Traceable measurements of the static permittivity of dielectric reference liquids over the temperature range 5–50 °C, *Measurement Science and Technology* 16 (7) (2005) 1506. doi:10.1088/0957-0233/16/7/013.
- 715 URL <https://dx.doi.org/10.1088/0957-0233/16/7/013>
- [46] R. Buchner, G. T. Hefter, P. M. May, Dielectric Relaxation of Aqueous NaCl Solutions, *The Journal of Physical Chemistry A* 103 (1) (1999) 1–9. doi:10.1021/jp982977k.
- 720 URL <http://pubs.acs.org/doi/abs/10.1021/jp982977k>
- [47] M. C. Dobson, F. T. Ulaby, M. T. Hallikainen, M. A. El-rayes, Microwave Dielectric Behavior of Wet Soil-Part II: Dielectric Mixing Models, *IEEE Transactions on Geoscience and Remote Sensing* GE-23 (1) (1985) 35–46, conference Name: IEEE Transactions on Geoscience and Remote Sensing. doi:10.1109/TGRS.1985.289498.
- 725 [48] G. Fan, K. Sun, Q. Hou, Z. Wang, Y. Liu, R. Fan, Epsilon-negative media from the viewpoint of materials science, *EPJ Applied Metamaterials* 8 (2021) 11, publisher: EDP Sciences. doi:10.1051/epjam/2021005.
- URL <https://epjam.edp-open.org/articles/epjam/abs/2021/01/epjam200025/epjam200025.html>
- 730 [49] K. S. Cole, R. H. Cole, Dispersion and Absorption in Dielectrics I. Alternating Current Characteristics, *The Journal of Chemical Physics* 9 (4) (1941) 341–351, publisher: American Institute of Physics. doi:10.1063/1.1750906.
- 735 URL <https://aip.scitation.org/doi/10.1063/1.1750906>
- [50] G. Constantinides, F.-J. Ulm, The effect of two types of C-S-H on the elasticity of cement-based materials: Results from nanoindentation and micromechanical modeling, *Cement and Concrete Research* 34 (1) (2004) 67–80. doi:10.1016/S0008-8846(03)00230-8.
- 740 URL <https://www.sciencedirect.com/science/article/pii/S0008884603002308>
- [51] S. De Luca, S. K. Kannam, B. D. Todd, F. Frascoli, J. S. Hansen, P. J. Daivis, Effects of Confinement on the Dielectric Response of Water Extends up to Mesoscale Dimensions, *Langmuir* 32 (19) (2016) 4765–4773, publisher: American Chemical Society. doi:10.1021/acs.langmuir.6b00791.
- 745 URL <https://doi.org/10.1021/acs.langmuir.6b00791>
- [52] A. C. A. Muller, K. L. Scrivener, A. M. Gajewicz, P. J. McDonald, Densification of C-S-H Measured by 1H NMR Relaxometry, *The Journal of Physical Chemistry C* 117 (1) (2012) 403–412.