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### Chemical profiles of venom glands in queens, foundresses, pre-wintering gynes, and workers in the hornet Vespa velutina nigrithorax

Berville L<sup>1</sup>, Lucas C<sup>1</sup>, Haouzi M<sup>1</sup>, Khalil A<sup>1,2</sup>, Gévar J<sup>3</sup>, Bagnères A-G<sup>4</sup>, Darrouzet E<sup>1\*</sup>

<sup>1</sup>IRBI, UMR CNRS 7261, University of Tours, 37200 Tours, France <sup>2</sup>Oniris, Ecole Nationale Vétérinaire, Agroalimentaire et de l'Alimentation, 44322 Nantes, France <sup>3</sup>iEES Paris, INRAE, 78026 Versailles, France <sup>4</sup>CEFE, CNRS, University of Montpellier, EPHE, IRD, Montpellier, France

16-digit ORCID: LB: 0000-0001-9540-9869 CL: 0000-0003-2839-7583 MH: 0000-0002-8851-9223 GJ: 0000-0001-9084-6445 AGB: 0000-0002-1729-379X ED: 0000-0002-7224-6937

\*Corresponding author: eric.darrouzet@univ-tours.fr

1 Abstract - Invasive species are permanently modifying the distribution and diversity of native species worldwide. 2 For nearly two decades, a hornet, Vespa velutina nigrithorax (Hymenoptera: Vespidae), has been spreading in 3 Europe. Due to its marked invasiveness, this yellow-legged hornet is of great economic and ecological concern, 4 mainly because of the damage it causes to insects in general and bees in particular. Current management methods 5 are sparse and ineffective. Naturally produced by insects, semio-chemicals have been proposed as integrated 6 management tools in this context, either for disruption or mass trapping, as an alternative tool to conventional non-7 selective traps. Here, we focused on the venom gland, which produces the alarm pheromone. In previous studies, 8 individuals showed marked diversity in their chemical profiles. However, to successfully conduct targeted 9 pheromone-based trapping, the hornet's chemical ecology must be thoroughly characterized. Therefore, it was 10 necessary to better understand the chemical composition of the alarm pheromone of not only workers but also 11 other V. v. nigrithorax females. First, we evaluated the differences in venom gland profiles between the four types 12 of females: queens, foundresses, pre-winter gynes, and workers. Next, we experimentally explored the venom 13 gland profiles of V. velutina nigrithorax workers and pre-wintering gynes by in vivo and in vitro approaches. We found 13 new compounds in the venom gland, of which 9 were identified (chain lengths:  $C_8$  to  $C_{12}$ ). Two 14 15 compounds were found exclusively in reproductive females. Profiles differed among pre-wintering gynes, 16 foundresses, and queens but not between pre-wintering gynes and workers. This result indicates that the chemical 17 signature of the female venom gland changes over the course of life history: from pre-wintering gynes to 18 foundresses to queens.

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- 20

21 Keywords - Yellow-legged hornet, Venom gland, Invasive species, Chemical communication, Vespidae

#### DECLARATIONS

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**Ethics approval:** The study does not require ethics approval since the yellow-legged hornet is classified as an invasive species; all research was carried out in compliance with the relevant national guidelines.

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Code availability: Not applicable.

Authors' contributions: The experiments and data collection were carried out by LB, JG, MH, and ED. The chemical assays were carried out by CL, AGB, LB, and AK. Formal Analysis: LB. The first draft of the manuscript was written by LB. The manuscript was revised and edited by LB, CL, AGB, and ED. Funding was obtained by ED. All the authors read the manuscript and gave their approval for submission. Conceptualization & Project Administration: ED.

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# 2

#### INTRODUCTION

3 Invasive species are increasingly affecting the Earth's ecosystems [1]. They are permanently modifying the 4 distribution and diversity of native species worldwide [2] at a variety of biological scales and with differing degrees 5 of severity [3]. In many different habitats, social insects are highly successful invaders because of their excellent 6 ability to disperse, compete for resources, mount effective defenses, reproduce at high rates, and exploit a broad 7 range of habitats and diets [4]. Native to China, the invasive yellow-legged hornet, Vespa velutina nigrithorax 8 (Lepeletier 1836) [5], was accidentally introduced into southwestern France around 2004 [6]. It is a hardy, 9 adaptable species whose flexible dietary requirements have allowed it to invade and establish itself within a variety 10 of habitats [7]. Vespa v. nigrithorax has successfully expanded its range and is now found throughout most of 11 France [8] as well as in several other European countries [9–13]. Ecosystems invaded by this hornet are impacted 12 in various ways since V. v. nigrithorax preys upon several insect and arthropod taxa [14], including the domestic 13 honey bee, Apis mellifera [15]. Finally, V. v. nigrithorax presents human health risks as hornets can aggressively 14 spray liquid in people's eyes [16]; also, their stings are painful and sometimes even fatal [17,18]. Consequently, 15 medical practitioners and researchers are increasingly seeking information on V. v. nigrithorax venom for 16 diagnostic and treatment purposes [17,19]. The hornet's introduction into Europe led to calls for control efforts, 17 which have largely focused on passively trapping adults using homemade or commercial poison baits [20]. 18 Unfortunately, this approach has done little to limit V. v. nigrithorax population sizes [20,21]. Furthermore, 19 because they lack specificity, these traps can have significant ecological impacts on a wide range of non-target 20 species [22,23]. As a result, work is currently underway [20] to develop new species-specific systems that are 21 based on pheromones [24-27].

22 Chemical signals are an essential component of animal communication systems [28,29] and mediate a 23 variety of behaviors, including social organization. In insects, long-distance signals play an important role in intra-24 and interspecific communication [29] and elicit species-specific responses. Such communication provides fitness 25 benefits by enhancing colony cohesion, defense, and alarm systems. There is a long history of research on chemical 26 signals (e.g., pheromones) in social insects (e.g., ants, bees, hornets, and termites); these signals often trigger 27 different behaviors (recruitment, dispersal, aggressiveness, or aggregation) that can affect colony survival [30,31]. 28 Thus, a promising pest control approach is to develop species-specific attractants (e.g., trap bait) or repellents that 29 target specific life stages. Alarm pheromones are of key importance because they are deployed when colonies are 30 threatened: they prompt nestmate recruitment and defense behaviors [32,33]. They can also be used by workers to 31 signal danger in areas farther away from the nest [34]. The most effectively disseminated signals are made up of 32 highly volatile compounds that act quickly and then dissipate, limiting further responses. Most alarm pheromones 33 in social insects are multicompound blends [24,35,36] that are generally involved in both offensive and defensive 34 reactions by workers. The venom gland tends to be the primary source of alarm pheromones [33,37], although 35 interspecific variation exists. Thus, during aggressive interactions, alarm signals are released at the same time as 36 venom [38]. Such is the case in several wasp and hornet species, including Polistes dominulus [39], Vespula 37 squamosa [40], Vespa crabro [41], Vespa mandarinia [37], and Vespa simillima xanthoptera [37].

38 The first study conducted on alarm pheromones in a subspecies of the yellow-legged hornet characterized 39 the volatile compounds found in worker venom glands in a native V. velutina auraria population in China [42]. A 40 second study [43] did the same in an invasive V. v. nigrithorax population in France. In both cases, researchers 41 identified some venom gland compounds that elicited hornet attacks when colonies were threatened. When the 42 alarm pheromones were released, they attracted other workers and encouraged stinging behavior. However, 43 compound abundance differed between the two subspecies. Next, researchers explored the chemical signals 44 associated with caste and species' reproductive status using four colonies in an invasive population in Italy; the 45 study looked at compounds in volatile venom gland mixtures in gynes and workers [27]. Twelve major compounds 46 were observed, which contrasted with the 16 and 17 compounds observed previously ([42,43] respectively). Five 47 of these 12 compounds had been seen in the first two studies. However, others had never been reported before, 48 including 4 unknown acetates of aliphatic alcohols and citronellyl acetate, which had the second highest abundance 49 [27]. Finally, work on an invasive population in Spain observed 8 compounds [26], collected in vivo by SPME 50 from seven hornets, including 3 previously seen ketones [27,42,43]. However, in this most recent study, individuals 51 displayed marked diversity in their chemical profiles since 2 of the 8 compounds were found in only one hornet 52 and 3 others in fewer than 3 hornets [26]; only one compound occurred across all the profiles. There is clearly a 53 need for further research given this range of results. First, a better understanding of the chemical composition of 54 V. v. nigrithorax's alarm pheromone is needed. Second, since gynes, foundresses, and queens alike use alarm 55 pheromones, we also must clarify associations between hornet caste and alarm pheromone composition. However, 56 in the latter three cases, the volatile compounds involved remain poorly characterized. Here, we hypothesized that 57 females from different castes and/or with different tasks might differ in their venom gland profiles.

This study thus had two goals. First, we assessed differences in venom gland profiles among the four types of females: queens, foundresses, pre-wintering gynes, and workers. Second, we experimentally explored venom gland profiles of *V. velutina nigrithorax* workers and pre-wintering gynes via an *in vivo* approach (collection of volatiles, solid-phase microextraction [SPME]) and an *in vitro* approach (liquid extraction of compounds from the venom gland).

63

MATERIALS AND METHODS

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#### 67 *Study sites and hornet sampling.*

68 Hornet colonies were collected at 44 different sites in the administrative department of Indre-et-Loire, 69 which is part of the Centre-Val-de-Loire region of France (Fig. 1). The yellow-legged hornet is classified as an 70 invasive species, and all research was carried out in compliance with relevant national guidelines. To determine 71 hornet caste (worker vs. reproductive), mass and wing spacing were measured [44]. The reproductives were either 72 pre-wintering gynes (females without nest), foundresses (mated females with small nest and some eggs), or queens 73 (mated females with workers). To characterize the venom gland's compounds, hornets of each caste were stored 74 at -80°C until they could be dissected. For the second experiment, individuals from a given colony were placed in 75 a mesh breeder box and fed an *ad libitum* mixture of honey and water, which was replenished daily. The hornets 76 underwent SPME within three days of collection.

77



79 Fig. 1: Vespa velutina nigrithorax collection sites

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#### 81 *Analysis of volatile compounds.*

82 To characterize the venom gland's compounds, hornets (n = 75 workers, 40 pre-wintering gynes, 30 83 foundresses, and 30 queens) were dissected under a binocular microscope. Within each group, a set of 5 84 undamaged venom sacs (n = 15, 8, 6, and 6 replicates for the workers, pre-wintering gynes, foundresses, and queens, respectively) was placed in a vial containing heptane (1 mL). The vials were then stored at -20°C. Glands 85 86 were perforated right before performing the gas chromatography-mass spectrometry (GC-MS) analysis; the vials 87 were vortexed at 300 rpm for 1 min. Gland extract (2 µL) from each vial were analyzed using a gas chromatograph 88 (Agilent Technologies 7890B) coupled to a mass spectrometer (Agilent Technologies 7000C GC/MS Triple Quad, 89 Les Ulis, France) equipped with an HP-5 capillary column (Agilent Technology, USA; 30 m x 250 µm x 0.25 µm) 90 and using a Gerstel MPS autosampler (electron impact at 70 eV). Helium was the carrier gas (flow rate = 2.3 91 mL/min). Injectors were used in splitless mode (splitless time = 2 min) at a constant temperature of  $250^{\circ}$ C. The 92 oven temperature was set to ramp up from 50°C to 200°C at a rate of 8°C/min and then from 200°C to 315°C at a
93 rate of 5°C/min (held constant for 5 min). Masses were scanned between 15 and 550 amu at 0.1 scan/s. Immediately
94 before the analysis, 10 µL of internal standard (*n*-eicosane at 10<sup>-3</sup> g/mL diluted in heptane) was added to each
95 sample.

96 To identify and quantify the compounds released by workers or pre-wintering gynes in response to stress, 97 two experimental stimuli were assessed. The experiments were conducted with sets of four live co-workers or four 98 live pre-wintering gynes. The hornets were cold anesthetized (-18°C for 10 min) and were subsequently placed in 99 a clean 350mL glass vial at room temperature (19-21°C for 10 min). The hornets were still able to move but not 100 to fly. Each hornet was exposed once to either a stressed hornet group (SH, replicates: pre-wintering gynes n = 4; 101 workers n = 8) or a wounded hornet group (WH, replicates: pre-wintering gynes n = 4; workers n = 8). The SH 102 treatment consisted of gently shaking the hornet group inside the glass vial. The WH treatment consisted of 103 applying pressure to a hornet's thorax using metal forceps; wounded hornets were then resealed before closing the 104 glass vial. We assumed that the WH treatment induced a higher stress level than the SH treatment. To extract the 105 resulting volatile compounds, a 40-min procedure was carried out using red solid-phase microextraction (SPME) 106 fibers coated with polydimethylsiloxane (PDMS,100 µm; Supelco). Immediately after the treatments, the fibers 107 were placed within a metal wire to prevent any physical contact with the hornets. First, for each caste, two fibers 108 were desorbed, and the resulting compounds were identified using the temperature program and GC-MS procedure 109 described above. Second, individual fibers representing different castes and treatment groups were desorbed 110  $(WH_{workers} n = 8; SH_{workers} n = 8; WH_{gynes} n = 4; SH_{gynes} n = 4)$  using a GC system (CPG Agilent Technologies 111 7820A) equipped with a flame-ionization detector (FID) and a capillary column (HP-5 Agilent Technology, Santa 112 Clara, USA; 30 m  $\times$  0.32 mm  $\times$  0.25 µm); helium served as the carrier gas (1.7 ml/min). The temperature program 113 was the same as above. The relative area of each compound was calculated based on the proportions represented 114 by each peak within the chromatogram using ChemStation (v. 04.02, Agilent).

115

#### 116 *Compound identification.*

117 The data were analyzed using MassHunter (v. B.07.00; Agilent) and AMDIS (v. 2.0g; NIST, 2011). Compounds were identified based on their mass spectra, which were interpreted via fragmentation analyses [45]. 118 119 The results were then compared to published spectra and/or confirmed using commercial standards (Table 1) 120 purchased from different suppliers (Sigma-Aldrich, Saint-Louis, Missouri, USA; GreenPharma, Orléans, France; 121 and ChemSpace, Monmouth Junction, New Jersey, USA). The molecules 4,8-dimethyl-7-nonen-2-one and 4,8-122 dimethyl-1,7-nonadiene were synthesized by Synthenova (Hérouville-Saint-Clair, France). Kovats retention index 123 (KRI) values were calculated based on the retention times of C8-C20 n-alkane standards (Fluka, 94234) that were 124 analyzed under the same set of GC and GC-MS conditions.

	Molecular Major fragments Kovats Retention Index								Compound identifications				
ID	name	formula	weight	CAS	(m/z)			Vespa velutina	in this study		previous studies		
			(a/mol)			from literature	measured	sub-species	methods id				
1	Heptan-2-one	C <sub>7</sub> H <sub>14</sub> O	114	110-43-0	43,58,71,27	888	-	a & n	-	-	0		
2	Heptan-2-ol	C7H16O	116	543-49-7	45,27,43	[877-924]	-	a & n	-	-	0		
3	3 2,6-Heptanedione		128	13505-34-5	43,39,82,58	-	-	a & n	-	-	0		
4	Acetophenone	C <sub>8</sub> H <sub>8</sub> O	120 140 142 142 144	98-86-2 5009-32-5 821-55-6 124-19-6 628-99-9	105, 77, 43 67, 43, 82 58, 43, 71 57,41,43,56 69, 55, 56, 57	[1026-1091] [1042-1055] [1046-1094] 1099 [1072-1110]	1032 1045 1052 - 1062	a & n	<b>+</b> Δε•	а	0		
5	8-Nonen-2-one	$C_9H_{16}O$						a&n a&n n a&n	•3Δ+ •Δε• - •Δε•	b c - d	00		
6	Nonan-2-one	C <sub>9</sub> H <sub>18</sub> O									000		
7	Nonanal	C <sub>9</sub> H <sub>18</sub> O									6		
8	Nonan-2-ol	C <sub>9</sub> H <sub>20</sub> O									0		
9	9 Isoamyl isovalerate		172	659-70-1	70, 85, 43, 57	[1080-1110]	1065	n	<b>+</b> Δε•	e	-		
10	Unknown 1 (cyclohexanol)	C <sub>10</sub> H <sub>20</sub> O	156	-	43, 71, 95, 138		1117	a & n	•	f	1?2?		
11	Decanal	C10H20O	156	112-31-2	43,41,57,55	1200		n	-	-	8		
12	Isoamyl senecioate	C10H18O2	170	56922-73-7	83, 100, 55	1146	1145	п	<b>+</b> ε•	g			
(!)	4.8-dimethyl-1.7-nonadiene	C11H20	152	62108-28-5	41.69.55.68	1343 🚯	1149	n	Δ	-	6		
13	X-C12:1	C12H24	-	-	55, 83, 70, 56		1152	n	•	h			
14	Unknown 2	-	-	-	44, 71, 95, 91	-	1181	n	•	i			
15	4.8-dimethyl-7-nonen-2-one	C11H200	168	3664-64-0	95, 110, 85, 69	-	1188	a & n	<b>+</b> Δε•	i	008		
16	2-Nonanyl acetate	C11H22O2	186	14936-66-4	43.87.55	[1186-1221]	1195	n	+ = •	, k	0:0?		
17	β-Citronellol, methyl ether	C11H2202	170	55915-70-3	82.85.81.69	1220	1204	n	٨.	ï	0.07		
18	2-Phenethyl acetate	C10H12O2	164	103-45-7	82.85.81.109	[1221-1279]	1219	n	+ = •	m			
19	X-Undecen-2-one	C11H200		-	43.81.118.68	-	1234	a & n	• x	 n	0		
20	X-Lindecen-6-one	C., H., O	-	-	43 71 99 58	-	1242	a & n	• ×		Ă		
21	X'-Lindecen-2-one	C11H220	-	-	44 69 97	-	1242	a & n	• ×	n	Ä		
22	Lindecan-2-one	C11H200	170	112-12-9	58 43 71 59	[1247-1342]	1245	a & n	<b>+</b> Λε•	р 0	คคล		
22	Undecan-2-ol	C1111220	172	1652.20.1	15 83 55 69	1302	1245	a & n	Λ	ч			
23	Citronellyl acetate	C1111240	198	150.94.5	81 95 69 123	[1314-1360]	1205	n	+	-	ă		
25	Unknown 3 *	C. Has	-	100 04 0	96 109 82 81	-	1210	n	•	•			
26	Geranyl acetate	C111.120	196	105 97 2	69 68 41 43	[1340-1409]	1226	n	+	+	0		
26'	Geranyl acetone	C12H2002	19/	2706.70.1	A3 69 A1 151	1/137	1330	n n			ě		
20	Decane-2 9-dione	C13H220	170	16529.01.2	43,05,41,151	1452	1220	n n	+				
20	Linknown 4	01011802	1/0	10330-31-3	110 81 68 42		1335	n n					
20	Isopontyl bonzoato	с н о	102	94.46.2	105 70 122 55	[1402-1426]	1370	n n	+				
2.5	2 Mathul 2 hutanul hanzasta	C H O	100	54-40-2	105,70,123,33	[1402-1430]	1394	"		w			
30	3-ivietnyi-3-butenyi benzoate	C <sub>12</sub> H <sub>14</sub> O <sub>2</sub>	190	5205-12-9	105, 68, 77	[1412-1421]	1403	n	Δ•	x			
31	X-Undecene-2,10-dione	C <sub>11</sub> H <sub>18</sub> O <sub>2</sub>	-	-	43, 81, 124	-	1416	a & n	ו	У			
32	Undecane-2,10-dione	C <sub>11</sub> H <sub>20</sub> O <sub>2</sub>	184	1800/-24-8	43, 71, 109, 58	1485	1442	a & n	<b>τ</b> Δε•	z			
33	α-Farnesene	C <sub>15</sub> H <sub>24</sub>	204	502-61-4	41,93,69,55	. 1494	- Molecule	n	-	-	Ð		
The major <i>m</i> /z fragments are sorted based on their occurrence frequency (highest to			<ul> <li>Literature and/or published spectra (PubChem)</li> <li>Δ Standard</li> </ul>				a to z	a to z in this study, see table 2					
X: corresponds to the putative position of the			ε Mass spectral data bases (NIST)				0	in Cheng et al. 20	016 and Thier	ry et al.	2018		
	double bond for each compound		Fragmentation analyses				0	2 in Cappa et al. 2019					
(!) standard of the molecule suggested by			Tragmenta don anal joco				ด้	in Rodriguez-Flores et al. 2021					
* Molec	cule mass spectra of the unknown		a: Vespa velutina auraria				0:0?	? not named in articles, postulated based on profile sighting					
someound /Fig 3A			n: Vesna veluting narithoray										

128 Table 1: Compounds found in venom glands of V. velutina subspecies. D Peaks present in both V. v. auraria [42] 129 and V. v. nigrithorax [43]. Peaks present in V. v. nigrithorax populations in 2 Italy [27] and 3 Spain [26]. All 130 the compounds observed in this study (a to z) underwent MS analysis (•) and were identified by comparing their 131 mass spectra with those in the NIST 2017 database ( $\varepsilon$ ) and/or in the literature (+); final validation was performed 132 using available standards ( $\Delta$ ). The major *m/z* fragments are sorted based on their occurrence frequency (highest to 133 lowest). The Kovats retention index values were calculated and obtained from PubChem and the literature. 134 Chemical Abstracts Service (CAS) of the American Chemical Society. (!) compound (Fig. 3B) identified by 135 Rodríguez-Flores (2021) on the basis of the HS-SPME chromatogram and mentioned as eluting after undecan-2-136 one. However, its MS spectra, retention time, and KRI value differed from those of the 4,8-dimethyl-1,7-nonadiene 137 standard. \* Mass spectra of unknown compound s (Fig. 3A) isolated from the venom gland (Fig. 3). n: Vespa 138 velutina nigrithorax. a: Vespa velutina auraria.

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141 Statistical analyses.

142 Venom gland profile composition was analyzed using multivariate principal component analysis (PCA) 143 performed in Statistica (v. 10; Statsoft). The analysis used the percentage representation of identified compounds 144 that occurred above a minimum threshold of 0.5%. The worker and gyne profiles obtained in the stress experiment 145 were explored using another PCA, which employed the percentage representation of all the compounds collected

146	via PDMS-SPME. To assess profile-based grouping patterns, we conducted a K-means cluster analysis [46].					
147	Because the number of groups is unknown ahead of time, the choice of initial group number is somewhat arbitrary.					
148	To optimize this number, we used the Caliñski-Harabasz index [47].					
149						
150						
151	RESULTS					
152						
153	Identification of venom gland compounds					
154	The venom gland profiles contained 38 compounds with chain lengths ranging from 8 to 12 carbons (Fig.					
155	2, Tables 1 & 2). Of these, 26 occurred at levels higher than the 0.5% threshold (Table 2). This group included 11					
156	ketones, 8 esters, 4 alcohols, and 3 unknown molecules. The MS spectrum were investigated, and the identities of					
157	the compounds were confirmed using chemical standards (Table 1). The unknown compounds were subject to MS					
158	interpretation. The third unknown compound (s, Table 2, Fig. 3A) was thought to be 4,8-dimethyl-1,7-nonadiene					
159	(Fig. 3B, Table 1). Indeed, this compound was identified [26] based on the HS-SPME chromatogram. However,					
160	its MS spectra, retention time, and KRI value differed from those of the 4,8-dimethyl-1,7-nonadiene standard (Fig.					
161	3). The MS spectra for compound $s$ (Fig. 3A) and citronellyl acetate (compound $r$ ) shared some similarities.					
162	However, they differed in the presence of a fragment at $m/z$ 152 fragment and a trace of $m/z$ 170. Furthermore,					
163	compared to the citronellyl acetate, compound s had higher peaks at $m/z$ 137 (instead of at $m/z$ 138), $m/z$ 96, and					
164	m/z 109. Unfortunately, spectra for the citronellyl family remain quite rare, limiting more detailed comparisons.					
165	Additionally, we found no other commercial compounds with more similar spectra, thus we were not able to					
166	confirm our hypothesis.					
167						



- 169 Fig. 2: Chemical profiles of hornet workers, gynes, foundresses, and queens (unit of analysis = pools of five venom
- 170 glands). The letters refer to the different compounds observed (Table 2).
- 171

			VENOM G	LAND 's profile	2	SPME analysis					
						SH	WH	SH	WH		
Name:		Queens	Foundresses	Workers	Prewintering gynes	Worke	rs	Pre-wintering gynes			
Acetophenone	a	1.7 ± 1	0.4 ± 0.3	0.2 ± 0.2	8.10-3 ± 1.10-3						
8-Nonen-2-one	b	0.4 ± 0.3	0.8 ± 0.2	3.7 ± 0.9	3.2 ± 0.5	1.1 ± 0.9	$1.4 \pm 0.4$	3.1 ± 2.5	1.1 ± 1.05		
Nonan-2-one	с	3.4 ± 0.9	16.4 ± 2.8	18.2 ± 3.2	14.2 ± 3.7	7.7 ± 3.8	10.9 ± 3.8	3.3 ± 2.2	6.3 ± 2.6		
Nonan-2-ol	d	0.3 ± 0.3	0.4 ± 0.2	1.6 ± 0.8	0.8 ± 0.9	$1.2 \pm 0.7$	$1.5 \pm 0.5$	1.8 ± 1.1	0.6 ± 0.3		
Isoamyl isovalerate	е	0.1 ± 0.1	0.8 ± 0.2	2,0 ± 0.9	2.5 ± 0.8	1.2 ± 1.3	0.8 ± 0.5	0.4 ± 0.5	1.2 ± 0.9		
Unknown 1 (cyclohexanol)	f	3 ± 0.4	0.7 ± 0.1	1.3 ± 0.9	0.8 ± 0.7						
Isoamyl senecioate	g	0.6 ± 0.5	0.2 ± 0.1	0 ± 0	$0.1 \pm 0.1$						
X-C12:1	h	0.4 ± 0.4	0.4 ± 0.1	0.7 ± 0.9	0.8 ± 1.1						
Unknown 2	i	$1.4 \pm 0.3$	0.2 ± 0.1	0.8 ± 0.6	0.5 ± 0.5						
4,8-dimethyl-7-nonen-2-one	j	28.3 ± 5.7	21.8 ± 2.9	23.3 ± 5.3	24.6 ± 3.1	34.4 ± 4.8	38.2 ± 3.1	37.7 ± 7.6	32.5 ± 5.5		
2-Nonanyl acetate	k	1 ± 0.6	3.7 ± 0.5	3.4 ± 1.1	4 ± 0.8	5.6 ± 1.4	6.7 ± 1.6	6.7 ± 1.2	5.2 ± 1.1		
β-Citronellol, methyl ether	1	4.5 ± 2.1	1,0 ± 0.4	1.2 ± 0.8	1.7 ± 0.8	4.1 ± 0.6	3.5 ± 1.1	5.7 ± 2.1	3.8 ± 0.8		
2-Phenethyl acetate	m	0 ± 0	0.4 ± 0.2	0 ± 0	0.2 ± 0.3						
X-Undecen-2-one	n	2.6 ± 0.2	5.5 ± 0.4	9.3 ± 2.9	5.8 ± 1.4	7.1 ± 2.4	7.5 ± 2.8	4.9 ± 1.2	5.7 ± 2.4		
X'-Undecen-6-one	0	6.7 ± 1.8	2.3 ± 0.7	8.5 ± 2	11.5 ± 2,0	$1.3 \pm 0.8$	0.9 ± 1.3	1.5 ± 1.3	3.5 ± 0.7		
X"-Undecen-2-one	р	0.2 ± 0.3	0.2 ± 0.1	1.2 ± 0.9	1.5 ± 0.5	$1.5 \pm 0.5$	$1.3 \pm 0.3$	0.3 ± 0.2	$1.5 \pm 0.4$		
Undecan-2-one	q	2.4 ± 1	6.1 ± 1	7.7 ± 1.6	7.7 ± 2,0	6.9 ± 2.9	8.1 ± 3.1	2.7 ± 0.4	5 ± 1.4		
Citronellyl acetate	r	1.6 ± 0.8	5.9 ± 2	0.5 ± 0.5	0.7 ± 0.6	5.8 ± 3.8	1.8 ± 1.1	7.9 ± 3.3	11.3 ± 4.9		
Unknown 3	5	23 ± 5	15.8 ± 2.2	5.2 ± 1.6	7.2 ± 2.6	21.9 ± 6.1	17.3 ± 5.6	23.7 ± 10	22.4 ± 5.9		
Geranyl acetate	t	1.2 ± 0.5	1.7 ± 0.4	0.1 ± 0.1	0.1 ± 0.2						
Decane-2,9-dione	u	0.2 ± 0.2	0.1 ± 0.2	0.8 ± 0.6	0.6 ± 0.4						
Unknown 4	v	0.2 ± 0.2	0.5 ± 0.1	1.2 ± 0.7	$1 \pm 0.3$						
Isopentyl benzoate	w	$1.3 \pm 0.1$	2.4 ± 1	0,0 ± 0	0.3 ± 0.4						
3-Methyl-3-butenyl benzoate	x	1.2 ± 0.3	2.0 ± 0.8	0.2 ± 0.2	0.4 ± 0.6						
X-Undecene-2,10-dione	y	2 ± 0.6	0.8 ± 0.4	4.5 ± 2.1	4.7 ± 1.8						
Undecane-2,10-dione	z	$1.6 \pm 0.5$	0.5 ± 0.3	1.3 ± 0.8	0.4 ± 0.5						

173 Table 2: Chemical composition of the venom glands of V. v. nigrithorax queens, foundresses, workers, and pre-

174 wintering gynes (percentages  $\pm$  standard error), which were obtained either via solvent- or fiber-based extraction.

175 Experimental treatments: SH, stressed hornet; WH, wounded hornet.

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Fig 3: (A) Mass spectra for unknown compound s (Kovats retention index = 1310), isolated from the venom gland
of *V. v. nigrithorax*; (B) Mass spectra for the 4,8-dimethyl-1,7-nonadiene standard (Kovats retention index = 1149); (C) combined GC-MS profiles for C10-C18 and 4,8-dimethyl-1,7-nonadiene. The blue arrow represents
the molecule s location (RT), in the gland profile.

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While the 26 compounds were all present in the pre-wintering gynes and foundresses, 3 were absent from the workers (**g**, **m**, and **w**; Table 2), and 1 was absent from the queens (**m**; Table 2). In the queens and foundresses, two compounds, 4,8-dimethyl-7-nonen-2-one (**j**) and compound **s**, accounted for more than 58% and 42% of the overall profile, respectively. In the workers and pre-wintering gynes, the two most common compounds were 4,8dimethyl-7-nonen-2-one (**j**) and nonan-2-one (**c**); they accounted for 43% and 42% of the overall profile, respectively.

- 194 PCAs were carried out on these 26 compounds. Two rounds of analysis were performed. The first
- explored differences among pre-wintering gynes, foundresses, and queens (n = 7, 6, and 6 pools, respectively, Fig.
- 1964). The first and second axes accounted for 40% and 21% of the total variance, respectively, and revealed a marked
- 197 separation among the three groups. This clustering was confirmed by the K-means analysis, in which the
- foundresses and queens were assigned to distinct groups. Only one pre-wintering gyne (G2) was incorrectly placedin the queens' group. The queens stood apart from the other two groups based on the presence of three compounds:
- Acetophenone (a) and two unknown compounds (f and s). The foundresses formed a distinct group based on the
- 201 presence of four compounds: citronellyl acetate (**r**), geranyl acetate (**t**), isopentyl benzoate (**w**), and isoprenyl
- benzoate (x, *i.e.*, 3-methyl-3-butenyl benzoate). The second round of analysis added profile information for the
- 203 workers (n = 15 pools) (Fig. 5). No differences were seen between workers and pre-wintering gynes, a result
- 204 confirmed by the K-means analysis resulting in no group separation.



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Fig. 4: (A) Principal component analysis results and (B) correlation circle plot for the venom gland profiles of queens, foundresses, and pre-wintering gynes. Q: queens (n = 6), F: foundresses (n = 6), and G: pre-wintering gynes (n = 7). The ellipses correspond to the 95% confidence interval of the data. Compounds found closer to the circumference of the circle convey greater amounts of information.



Fig. 5: (A) Principal component analysis and (B) correlation circle plot for the venom gland profiles of
queens, foundresses, workers, and pre-wintering gynes. Q: queens (n = 6), F: foundresses (n = 6), W: workers (n
= 15), and G: gynes (n = 7). The ellipses indicate the 95% confidence interval of the data.

The fiber-based extractions revealed that the workers and pre-wintering gynes released 13 compounds under stressful conditions (SH and WH). All the compounds were present in the solvent extract obtained from the venom glands (Table 2). Two compounds accounted for more than 50% of the total profile: 4,8-dimethyl-7-nonen-

- 223 2-one (j) and the unknown compound s. A PCA was carried out on the profiles of the workers (n = 16 pools) and
- 224 pre-wintering gynes (n = 8 pools) subject to the SH and WH treatments. The first and second axes accounted for
- 42% and 22% of the total variance, respectively. The analysis did not show any clear differences based on caste
- or treatment (Fig. 6).
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Fig. 6: Principal component analysis of the 13 compounds identified via PDMS-SPME after workers and prewintering gynes were experimentally stressed (SH treatment, G: n = 4 and W: n = 8) or wounded (WH treatment, G: n = 4 and W: n = 8). The ellipses indicate the 95% confidence interval of the data.

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#### DISCUSSION

240 As insects evolved to form organized societies, it became essential for them to develop communication 241 systems for sharing diverse messages [48]. Eusocial insects most commonly employ chemical communication 242 [49], and it can be challenging to study the cues and signal patterns involved in different behaviors (e.g., 243 recruitment, dispersal, aggressiveness, or aggregation) [30,35,50]. Chemical signals, like pheromones, can rapidly 244 and efficiently disseminate information and are used by all eusocial insects [51,52]. Volatile pheromones are 245 common and may convey information across great distances [33,52,53]. Previous studies identified 19 compounds 246 with chain lengths ranging between 7 and 12 carbons in workers of both V. v. auraria and V. v. nigrithorax 247 [26,27,42,43]. The majority were ketones, the most phylogenetically widespread class of alarm pheromones [52]. 248 In the study presented here, 9 new compounds were identified among the 13 compounds found, which displayed 249 the same range of chain lengths as in previous research (Table 1). To our knowledge, this is the first time that three 250 of them ( $\beta$ -citronellol methyl ether [1], 2-nonanyl acetate [k], and 2,9-decanedione [u]; Table 2) could been 251 identified as potential semiochemicals or as molecules involved in toxic actions. Other compounds that we 252 observed are known to have communication functions in Hymenoptera. Geranyl acetate (t) is used by three Bombus 253 species [54], and citronelly acetate ( $\mathbf{r}$ ), isoamyl isovalerate ( $\mathbf{e}$ ), and isoamyl senecioate ( $\mathbf{g}$ ) occur in the European 254 hornet Vespa crabro [55]. Undecan-2-one (q) occurs in Dolichovespula maculate [56] and Polistes species [39]. 255 Nonan-2-one (c), 4,8-dimethyl-7-nonen-2-one (j), and undecan-2-one (q) are known to elicit different levels of 256 alarm in Vespa orientalis [57]. Thus, these compounds may be common in Vespidae. Some compounds have also 257 been found in other insects. For example, phenethyl acetate (m) is an attractant in Diachasmimorpha longicaudata [58] and is part of a pheromone released by the milkweed bug Neacoryphus bicrucis [59]. Others, such as isopentyl 258 259 benzoate (w) and isoprenyl benzoate (x), are volatiles in plants [60,61]. The presence of undecan-2-ol, in contrast 260 to previous research in V. v. auraria [34] and V. v. nigrithorax [57], was not observed. Since a reference sample 261 of undecan-2-ol was also injected, using the same method, and present in the profile, we are confident that the 262 absence of this compound does not reflect a dehydration or thermolysis artifact in our GC-MS injector.

263 This same previous research [42,43] discovered that four compounds were common in the venom glands 264 of V. velutina workers: nonan-2-one (c), 4,8-dimethyl-7-nonen-2-one (j), X-undecen-2-one (n), and undecan-2-265 one (q). In our study, however, the main compound observed in all castes was 4,8-dimethyl-7-nonen-2-one (21-266 28 %, liquid extraction), not nonan-2-one. Nonan-2-one was the second most common compound in workers, pre-267 wintering gynes, and foundresses but not in queens (relative representation: 4%). Instead, the second and third 268 most common compounds in queens were compound s (23%, Fig 3) and undecen-6-one (o), respectively. In Cheng 269 et al. (2017), Vespa v. auraria collected did not display the peak for compound s. Nevertheless, compound s might 270 correspond to the unidentified peak placed after the citronellyl acetate in both Vespa v. auraria (seen by [62]) and 271 V. v. nigrithorax (seen by [43]). Additionally, compound quantity differed between our work here and that of 272 Cheng et al. (2017). For example, in V. v. auraria, Cheng et al. (2017) found that X-undecen-2-one (n) and 273 undecan-2-one (q) were the second and third most common compounds, respectively; in our study, however, they 274 were present at far lower levels (2.6-9.3%, liquid extraction). In short, our results differ quantitatively and 275 qualitatively from those of both Cheng et al. (2017) and Thiéry et al. (2018).

Considering that undecan-2-ol was absent from our results and that we observed a different set of common compounds (i.e., like citronellyl acetate in the French and Spanish hornet populations), it could be that the hornet's chemical blends are changing at the population scale, as has been seen in some ants [63–65]. Whether such shifts between and within subspecies could arise from isolation by distance or difference in evolutionary lineages is a question that merits further exploration.

281 In the venom gland profiles, there were clear differences among the gynes, queens, and foundresses. This 282 result indicates that the chemical signature of the female venom gland changes over the course of life history: from 283 pre-wintering gynes (females without nest) to foundresses (mated females with small nest and some eggs) to 284 queens (mated females with workers). However, no differences between the pre-wintering gynes and workers 285 profiles were observed. This result shows that pre-wintering gynes start off with profiles similar to those of workers 286 but that those profiles then change over time. The most obvious change could be seen in compound s: its mean 287 relative levels were 5% in workers versus 7, 15, and 23% in pre-wintering gynes, foundresses, and queens, 288 respectively. Unfortunately, no match was found for compound  $\mathbf{s}$  in any of the available databases. However, this 289 compound did have a fragmentation spectrum highly similar to that of citronellyl acetate (r): fragments 137 and 290 152 were present, and fragment 138 was almost completely absent.

291 The differences among the females raise questions about the hornet's chemical ecology. The blend 292 observed could act as an alarm pheromone for pre-wintering gynes, which occur in their colony of origin and can 293 play a defensive role alongside workers. However, the queens displayed a different compound ratio, which could 294 mean that the blend had a more queen-specific function. Indeed, this compound ratio could serve as a specific 295 signal from the queen, potentially informing the colony's workers that their queen has a problem or is in danger. 296 Alternatively, it could act as a signal of queen presence and health. Indeed, research on wasps, bees, and ants has 297 shown that reproductives may chemically communicate their fertility [66]. Consequently, it could be that 298 compound s is a signal generated by mature egg-laying queens. Further study is needed to explore this hypothesis 299 in V. v. nigrithorax.

300 Previous studies identified 12 compounds in the crushed venom glands of workers (V. v. nigrithorax; 301 [43]) and in living hornets (V. v. auraria; [42]) using blue PDMS fibers (Supleco). Here, we used SPME and liquid extracts to characterize the compounds released by the venom glands of V. velutina hornets as a result of 302 303 experimental stress or wounding. 13 out of 26 compounds in this study were identified in living hornets using red 304 PDMS fibers (Supleco). We found 4,8-dimethyl-7-nonen-2-one (32–38%; j) to be the most common compound, 305 followed by compound s. The results were unaffected by either caste (worker vs. pre-wintering gyne) or treatment. 306 When stressed, both groups of hornets emitted an alarm signal that was the same as the one emitted when facing 307 a life-threatening event. Eight of the compounds associated with this signal had been previously observed in both 308 V. v. auraria and V. v. nigrithorax [42,43]. Recent work by Rodriguez-Flores et al. (2021) identified 8 compounds 309 in queens (n = 4) and workers (n = 3) using HS-SPME/GC-MS [26]. Among them were nonane-2-one, undecan-310 2-one, and 4,8-dimethyl-7-nonen-2-one, which have been seen in all studies on the hornet's chemical ecology; in 311 contrast, nonanal, decanal, geranyl acetone, and 4,8-dimethyl-1,7-nonadiene were not seen [26]. The differences 312 between our results and previous research could be explained by SPME fiber type, collection materials, and/or 313 chromatography methods and machine systems.

This study represents an important step towards developing pheromone-based traps. Indeed, the next step is to identify all the components of *V. v. nigrithorax* pheromones, including compound **s**, which might be important in queen signaling. A useful approach could be to assess the effects of each major compound, first individually

317	and then in proportionally accurate blends. Indeed, certain compounds may only elicit a strong response when they
318	occur in blends, as is the case with alarm pheromones in honey bees [67] or sex pheromones in moths [68].
319	Moreover, the precise compound ratio of blends could be of major importance. It is also essential to test the effects
320	of any target compounds on bees. Given that alarm pheromones are produced during both defensive and offensive
321	situations, they could also serve as a kairomone for the hornet's prey. Indeed, bees may detect certain components
322	of hornet alarm pheromones and modify their behavior accordingly.
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