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### Structural, magnetic and magnetocaloric properties of (Pr,Sm)<sub>2</sub>Fe<sub>17</sub> compound at room temperature. 2

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In the present work, we report on the structural, magnetic, and magnetocaloric properties of the (Pr, Sm)<sub>2</sub>Fe<sub>17</sub> system prepared by arc melting under high pure argon and homogenized at 1073 K to minimize other possible impurity phases. X-ray diffraction (XRD) coupled with Rietveld analysis with FullProf computer code reveals the presence of a single-phase of (Pr, Sm)<sub>2</sub>Fe<sub>17</sub>. This compound crystallizes in the rhombohedral  $Th_2Zn_{17}$  type structure. The lattice parameters decrease when Sm substitutes Pr. Hence, the temperature dependence of the magnetization is determined. The Curie temperature increases from  $285 \,\mathrm{K}$  to  $299 \,\mathrm{K}$  after the Pr substitution. This increase in  $T_C$  is mainly due to the de Gennes factor. The Arrott plot around second-order magnetic transition, magnetic entropy changes  $\Delta S_{\rm M}$ , the relative cooling power, and temperature-averaged entropy change (TEC) are reported. This compound is considered a magnetic refrigerant for use at a low magnetic field based on these results.

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Keywords: Intermetallic rare-earth transition-metal compounds; Magnetic materials; Magnetocaloric Effect.

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#### INTRODUCTION I.

The R-M intermetallic compounds combining a rare 11 <sup>12</sup> earth element (R) and a transition metal (M) have interesting properties due to their extremely varied potential 13 applications. Among these applications, we may cite the 14 high-density magnetic recording  $Sm_2Co_{17}-Cu$  [1], and 15 TbFeCo [2], the hard magnetic materials SmCo, NdFeB, 16 and PrCo [3–6], and the magnetic refrigeration [7]. 17

Iron-rich  $R_2$ Fe<sub>17</sub> intermetallic compounds are studied 18 in several works for their: magnetic properties, as well 19 as their thermodynamic properties, and the absence of 20 thermal hysteresis [8–19]. Additionally, Kou et al. [20] 21 have analyzed the magnetic anisotropy in several  $R_2$ Fe<sub>17</sub> 22 systems, where R = Ce, Pr, Nd, Sm, Gd, Tb, Dy, Ho, 23 Er, Tm, Lu, Y. Moreover, absorption and storage ca-24 pacity of hydrogen were examined by Isnard et al. in 25  $R_2$ Fe<sub>17</sub> (R =Ce, Pr, Nd, Gd) [21]. Their magnetostric-26 tion and thermal expansion properties were also studied 27 by Mori *et al.* [22]. Sun *et al.* [23] have reviewed the 28 magnetic properties, Fe-Fe and R-Fe exchange interac-29 tions for  $R_2$ Fe<sub>17</sub> and their nitrides. Also, Long *et al.* [24] 30 have as well investigated neutron and Mössbauer effect 31  $_{32}$  of  $Pr_2Fe_{17}$  and  $Pr_2Fe_{17}N_{2.6}$ . Further information about 33 structural, magnetic, and microscopic physical properties of  $Sm_2Fe_{17}$  and  $Pr_2Fe_{17}$  and their nitrides have been 34 provided by Zeng et al. [25]. 35

On the other hand, for many years, human activities 36 have led to a sharp increase in greenhouse gas emis-37 sions due to the need for refrigeration, air conditioning, 38 and cryogenics. However, these systems mostly use re-39 frigerants that contain volatile organic compounds such 40 <sup>41</sup> as hydrofluorocarbons. Such compounds are made up  $_{42}$  of carbon and hydrogen, which can easily be found in  $_{77}$  to room temperature is possible to substitute the Pr

44 ages. They can also have harmful effects on animals and <sup>45</sup> plant species as well as on humans. Thus, using less <sup>46</sup> polluting refrigeration systems becomes crucial in this 47 situation.

Magnetic refrigeration is an alternative way for cool-48 <sup>49</sup> ing matter using a magnetic field. In the last years, a <sup>50</sup> reasonable interest has been giving to the study of mag-<sup>51</sup> netocaloric materials. [10, 26–32]. Many research stud-<sup>52</sup> ies have reported materials with high magnetocaloric ef-<sup>53</sup> fect (MCE) at high temperatures, but few have interest-54 ing MCE and second-order magnetic transitions around 55 300 K. Since discovering MCE, the researchers have stud-<sup>56</sup> ied several magnetocaloric materials; the most promis-57 ing materials for magnetic refrigeration at room temper-<sup>58</sup> ature are Gd<sub>5</sub>(Ge,Si)<sub>4</sub> [27, 33–35], La(Fe,Co,Si)<sub>13</sub>(H,C) <sup>59</sup> [36–41], Fe<sub>2</sub>P [42–45], manganites  $R_{1-x}A_xMnO_3$  [46–49] 60 and related alloys and compounds.

Gadolinium is the only element having a high mag-61 62 netocaloric effect near room temperature. However, its <sup>63</sup> relatively rapid oxidation corrosion, high price, and in-64 sufficient global resources do not allow it to compete <sup>65</sup> with conventional refrigeration systems. This work aims 66 to synthesize new material with an interesting magne-<sup>67</sup> tocaloric effect room temperature, chemically stable, and  $_{68}$  lower costs than Gd. We focus on iron-rich  $Pr_2Fe_{17}$ 69 based compounds, its relatively large magnetic entropy <sup>70</sup> change maximum around room temperature with a large 71 full width at half-maximum of magnetic entropy change  $_{72}$  and its reversible magnetocaloric effect make  $Pr_2Fe_{17}$  a 73 good competitor for Gd-based magnetic materials. Nev-74 ertheless, its Curie temperature (the working tempera-<sup>75</sup> ture in materials for cooling application) is lower than <sup>76</sup> room temperature. To obtain a Curie temperature equal 43 gaseous form in the atmosphere and cause climatic dam-78 atom with Sm atom because the Curie temperature of  $_{79}$  Sm<sub>2</sub>Fe<sub>17</sub> is higher than room temperature. Several sam-  $_{124}$  principle as a Faraday type balance between 300 and <sup>80</sup> ples (Pr,Sm)<sub>2</sub>Fe<sub>17</sub> have been prepared with different com-<sup>125</sup> 900 K [66]. A Physical Properties Measurement System s1 positions (x=0.24, x=0.3, x=0.36 and x=0.42), only the 126 (PPMS) magnetometer is used for magnetic measure- $_{22}$  composition with the nominal composition x=0.36 one  $_{127}$  ments at low temperatures between 200 and 340 K. Regives  $T_C = 300$  K. Here, we present the structural, mag- 128 cently, it has been reported that the demagnetizing field <sup>84</sup> netic, and magnetocaloric properties of  $(Pr, Sm)_2Fe_{17}$  <sup>129</sup> might have a notable influence on the result of magne- $^{85}$  with the nominal composition x=0.36.

#### II. EXPERIMENTAL METHODS

87 88 99.9%, praseodymium (Pr) 99.98% and samarium(Sm) 89 99.98% by arc-melting technique under a purified argon 90 atmosphere. The elements are placed in a copper crucible 91 <sup>92</sup> cooled by cold water. After ingot formation, the com- 138 <sup>93</sup> pound is wrapped in tantalum foil and introduced into a silica tube sealed under secondary vacuum  $2 \times 10^{-6}$  bar [50-52]. The ingot has been heat-treated for seven days 95 at 1073 K, and is finally water quenched [53–55]. 96

97 Phase analysis has been performed by X-ray powder diffraction (XRD), using D8 Brucker diffractometer with 98 Cu K $\alpha$  radiation  $\lambda = 1.54178$  Å. XRD data of the sam-99 ples have been collected between  $20^{\circ}$  and  $80^{\circ}$  at room 100 temperature with 0.015 step width. 101

XRD diagrams are analyzed with the Rietveld method 102 [56, 57] using the Fullprof program [58, 59]. The peak 103 <sup>104</sup> shape function has been selected as Thompson-Cox-<sup>105</sup> Hastings pseudo-Voigt type [60]. The goodness of fit 106 indicators  $R_B$  and  $\chi^2$  are calculated from the program 107 output to measure the quality of refinement, and they <sup>108</sup> are defined as:

$$R_B = 100 \frac{\sum_{K} |I_K(O) - I_K(C)|}{\sum_{K} I_K(O)}$$

109 and

86

$$\chi^2 = \frac{\sum_i w_i |y_i(O) - y_i(C)|}{N - P}$$

where  $I_K(O)$  is the observed Bragg intensity and 110  $I_{111} I_K(C)$  is the calculated one.  $y_i(O)$  is the intensity ob-<sup>112</sup> served at the  $i^{th}$  step in the step scanned powder diffraction pattern,  $y_i(C)$  is that calculated, and  $w_i$  is the 113 weight of the observation. N is the total number of points 148 114  $_{115}$  used in the refinement, and P is the number of refined  $_{149}$  Pr<sub>1.64</sub>Sm<sub>0.36</sub>Fe<sub>17</sub> compound crystallizes in the same parameters. 116

117 factors U, V, W, background points, and atomic positions 152 118 119 120 Ref. [9, 64, 65] 121

122 <sup>123</sup> neto/susceptometer Manics DSM8 operating on the same <sup>157</sup> For x = 0.36 a and c are 8.5796(4) and 12.4647(4) Å,

<sup>130</sup> tocaloric effect [67]. To get the internal field  $H_{int}=H_{ext}$ - $_{131}$  N<sub>Demag</sub>M(T, H<sub>ext</sub>), where M is the measured magneti- $_{132}$  zation, the external applied magnetic field  $H_{ext}$  has been <sup>133</sup> corrected for the demagnetization effect. Demagnetiza-

 $_{134}$  tion constant  $\mathrm{N}_{Demag}$  has been determined from M vs  $(Pr, Sm)_2Fe_{17}$  with the nominal composition x=0.36 <sup>135</sup> H<sub>ext</sub> curve in a low field region following the method has been prepared from high pure elements iron(Fe)  $_{136}$  given in Ref [68]. The corrected magnetic field  $H_i$  has <sup>137</sup> been used for the present magnetic results.

#### **RESULTS AND DISCUSSION** III.

#### Structure analysis Α.

XRD patterns of the  $Pr_2Fe_{17}$  and the  $Pr_{1.64}Sm_{0.36}Fe_{17}$ 140 141 compounds are presented with their Rietveld refinement  $_{142}$  in Fig. 1(a) and Fig. 1(b), respectively. The pure  $Pr_2Fe_{17}$ <sup>143</sup> compound crystallizes in a rhombohedral structure of the <sup>144</sup> Th<sub>2</sub>Zn<sub>17</sub> type in the space group  $R\bar{3}m$ . In this structure,  $_{145}$  Pr atoms occupy 6c sites, while Fe atoms occupy four different crystallographic sites: 18f, 18h, 6c and 9d (in 146 <sup>147</sup> Wyckoff notation).

Table I. a and c unit cell parameters,  $R_B$ ,  $\chi^2$  factors, and atomic positions from Rietveld refinement of  $Pr_{2-x}Sm_xFe_{17}$ .

	x = 0	x = 0.36
a (Å)	8.5848 (3)	8.5796(4)
c (Å)	12.4659(1)	12.4647(4)
c/a	1.4521	1.4526
V (Å <sup>3</sup> )	795.67	794.63
$\chi^2$	1.45	2.08
$R_B$	2.98	4.103
$x\{18f\}(Fe)$	0.284	0.283
$x\{18h\}(Fe)$	0.506	0.503
$z\{6c\}(\Pr)$	0.345	0.356
$x\{6c\}(Fe)$	0.093	0.092
$z\{18h\}$ (Fe)	0.157	0.166

The Rietveld refinement that the shows <sup>150</sup> space group as  $Pr_2Fe_{17}$   $(R\bar{3}m)$ . The atomic, structural The refined parameters are unit cell parameters, scale <sup>151</sup> parameters,  $R_B$  and  $\chi^2$  factors are listed in Table I.

Fig. 2 shows that after the partial substitution of Pr by [5, 61–63]. The atomic input parameters that have been 153 Sm, we notice a small decrease in the lattice parameters used to determine the structure of our samples are from  $_{154} a, c$  and V. For x = 0, a and c are respectively 8.5848(3)<sup>155</sup> and 12.4659(1) Å. Similar lattice parameter values and Magnetic properties are measured using a mag- 156 atom positions were found in the previous work [9, 64].



Figure 1. The Rietveld refinement pattern of XRD data for (a)  $Pr_2Fe_{17}$  and (b)  $Pr_{1.64}Sm_{0.36}Fe_{17}$  compounds. The black and red line presents respectively the calculated intensities and the experimental intensities. The green vertical bars correspond to (hkl) line positions (Positions of Bragg peaks). The blue line shows the difference between the calculated and experimental intensities.

159 160 161 the praseodymium one. 162

For Sm concentration 0 and 0.36, we note that all 165 <sup>166</sup> angular positions shift to the right, in Fig. 3 we show angular position shift of the peak that corresponds to 167  $(hkl) \equiv (303)$ , a decrease of lattice parameters causes 168 the observed shift, several works reported the same be-169 170 171  $_{176}$  as follows [71]:



Figure 2. Cell parameters versus Sm content x, Inset shows cell volume and c/a vs Sm content.



Figure 3. (303) Bragg position for  $Pr_2Fe_{17}$ ,  $Pr_{1.64}Sm_{0.36}Fe_{17}$ 

$$a_{A-B} = a_A(1-x) + a_B$$

<sup>177</sup> Where  $a_{A-B}$ ,  $a_A$ ,  $a_B$  are lattice parameters of binary <sup>158</sup> respectively, while slight decrease in cell volume from <sup>178</sup> compound, pure A compound and pure B compound, re-795.67 Å<sup>3</sup> to 794.63 Å<sup>3</sup> and c/a ratio remains almost con- <sup>179</sup> spectively. From a value of  $Pr_2Fe_{17}$ ,  $Pr_{2-x}$  Sm<sub>x</sub>Fe<sub>17</sub> and stant. This behaviour can be explained by the size of the 100 Sm<sub>2</sub>Fe<sub>17</sub> the preceding equation gave  $x \sim 0.35$  which is Sm atom, which is slightly small compared to the size of 181 very close to the nominal composition within the exper-182 imental uncertainties.

#### В. Magnetic properties

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Here, we present the magnetic properties of haviour, we can cite Ref. [69]. In order to estimate the  $_{185}$  Pr<sub>1.64</sub>Sm<sub>0.36</sub>Fe<sub>17</sub> compound. It is worthy to note that concentration of the prepared sample from the outputs 186 this composition has been chosen for the following rea- $_{172}$  of Rietveld refinement, we use Vegard's law; this law has  $_{187}$  son: It is well known that  $T_C$  of  $Pr_2Fe_{17}$  and  $Sm_2Fe_{17}$ <sup>173</sup> mainly been used in the calculation of solid solution com- <sup>188</sup> are equals to 285 and 386 K, respectively. To obtain a <sup>174</sup> positions and even recently [69, 70], the simplest expres-<sup>189</sup> compound with a Curie temperature around the room  $_{175}$  sion of this law for a binary (A-B) compound is defined  $_{190}$  temperature, we started by preparing Pr<sub>2</sub>Fe<sub>17</sub> as the ini-<sup>191</sup> tial mixture, and we used the nearly linear variation of  $_{192}$  Pr<sub>2-x</sub>Sm<sub>x</sub>Fe<sub>17</sub> Curie temperature with Sm content to  $_{210}$  ferromagnetic state to a paramagnetic one. The value of 193 is achieved. 194

195 196 197 <sup>198</sup> culating the first derivative of M(T). The value of the <sup>216</sup> Vegard's law mentioned above, similar behavior has been Curie temperature corresponds to the minimum of the 217 reported in [65, 72]. 199  $_{200}$  magnetization derivative curve. We notice that  $T_C$  in-<sup>201</sup> creases after the substitution of the Pr atom by the Sm 202 atom.



Figure 4. Temperature dependence of magnetization of  $Pr_{1.64}Sm_{0.36}Fe_{17}$  under an applied magnetic field  $\mu_0 H =$ 0.05 T. 203

204



Figure 5.  $T_C$  versus Sm content (x) of  $Pr_{2-x}Sm_xFe_{17}$  com- <sup>230</sup> pounds 205

206

Fig. 4 shows the variation of the magnetization as a  $_{231}$  where G is the de Gennes factor: 207 <sup>208</sup> function of the temperature. It shows a sudden drop in <sup>209</sup> magnetization, the signature of a phase transition from a

reach  $T_C \sim 300$  K. Therefore, for x = 0.36, this goal <sup>211</sup>  $T_C$  for the compound Pr<sub>2</sub>Fe<sub>17</sub> is equal to 285 K, this value  $_{212}$  is consistent with the one found by Gorria *et al.* [13]. Magnetization variation as a function of the temper-  $_{213}$  After the partial substitution of Pr by Sm  $T_C$  increases, ature is measured under a weak magnetic field equal to 214 it equals 299 K. Fig. 5 shows a quasi-linear variation of 0.05 T. The Curie temperature  $T_C$  is determined by cal-  $_{215}$   $T_C$  with Sm content which is in a good agreement with

Table II. Interatomic distances  $(d_{\text{Fe-Fe}}(A))$  and coupling type for the rhombohedral  $R\bar{3}m$  structure type for  $Pr_{2-x}Sm_xFe_{17}$ , x = 0 and x = 0.36. NN is the number of near neighbor atom.

Sites		x = 0	x = 0.36	NN
6c - 6c	AFM	2.31	2.28	1
6c - 9d	$\mathbf{FM}$	2.64	2.637	3
6c - 18f	$\mathbf{FM}$	2.69	2.677	6
6c-18h	$\mathbf{FM}$	2.68	2.6	3
9d - 18f	AFM	2.43	2.42	2
9d-18h	$\operatorname{AFM}$	2.43	2.4	4
9d-6c	$\mathbf{FM}$	2.63	2.63	4
18f - 18f	AFM	2.43	2.42	2
18f - 18h	AFM	2.55	2.4	2
18f - 6c	$\mathbf{FM}$	2.69	2.68	1
18h - 9d	$\operatorname{AFM}$	2.43	2.4	3
18h - 18h	$\mathbf{FM}$	2.48	2.5	1
18h-6c	$\mathbf{F}\mathbf{M}$	2.67	2.6	2
18h - 9d	AFM	2.44	2.4	2
18h - 18f	AFM	2.56	2.4	2

 $T_C$  in rare-earth transition-metal intermetallic com-<sup>219</sup> pounds are governed by three kinds of exchange coupling  $_{220}$   $J_{RR}$ ,  $J_{TT}$  and  $J_{RT}$ . Therefore,  $T_C$  can be given by the 221 following relation [73]: which allows us to discuss the  $_{222}$  different parameters which influence the value of  $T_C$ :

$$3k_BT_C = a_{\rm FeFe} + a_{\rm RR} + [(a_{\rm FeFe} - a_{\rm RR})^2 + 4a_{\rm RFe}a_{\rm FeR}]^{1/2}$$

<sup>223</sup> Where  $a_{\alpha-\beta}$  is the magnetic interaction energy between  $\alpha$  and  $\beta$  spin. 224

Neglecting the exchange interaction between two rare-225 226 earth atoms  $a_{RR}$  is relatively weak compared to the other 227 terms of energy of magnetic interaction. So, it can be written as follows: 228

$$3k_B T_C = a_{\rm FeFe} + [a_{\rm FeFe}^2 + 4a_{\rm RFe}a_{\rm FeR}]^{1/2}$$

229 Where

$$a_{\rm FeFe} = Z_{\rm FeFe} J_{\rm FeFe} S_{\rm Fe} (S_{\rm Fe} + 1)$$

$$a_{\rm RFe}a_{\rm FeR} = Z_1 Z_2 S_{\rm Fe} (S_{\rm Fe} + 1) G J_{\rm RFe}^2$$

$$G = (g_{\rm R} - 1)^2 J_{\rm R} (J_{\rm R} + 1)^2$$

 $_{232}$   $J_{\rm FeFe}$  and  $J_{\rm RFe}$  are the exchange coupling constant be-  $_{268}$  iron pairs become ferromagnetically coupled. The presearth atom, respectively. 234

 $g_{\rm R}$  is the Landé factor of rare-earth atom. 235

236 the total moment of rare-earth atom, respectively. 237

238  $_{239}$  erage of  $Z_1 = 19$  iron near-neighbours; each an atom of  $_{275}$  Pr<sub>1.64</sub>Sm<sub>0.36</sub>Fe<sub>17</sub>. After the partial substitution of Pr by  $_{240}$  iron has an average of  $Z_2 = 2$  atoms of Pr or Sm and a  $_{276}$  Sm in  $Pr_2Fe_{17}$  the coupling nature of the different pairs  $_{241}$  number of  $Z_{\text{FeFe}} = 10$  iron neighbors. This relation shows  $_{277}$  of Fe is not affected, but we notice a slight decrease in 242 243 244  $_{245}$  sition metal. Consequently,  $T_C$  is mainly controlled by  $_{281}$  distances. Ultimately, there is a competition between the  $_{246}$  the exchange interaction between the atoms of iron  $J_{\text{FeFe}}$   $_{282}$  two contributions to  $T_C$ : de Gennes factor and iron-iron  $_{247}$  and between the rare-earth atom and the iron atom  $J_{\rm RFe}$   $_{283}$  distance. In Sm<sub>2</sub>Fe<sub>17</sub> compound  $T_C$  is equal to 400 K but  $_{248}$  [52, 74]. Many parameters can have an influence on the  $_{284}$  in  $Pr_2Fe_{17}$  T<sub>C</sub> is equal to 285 K, although the unit cell <sup>249</sup> value of  $T_C$  in  $R_2$ Fe<sub>17</sub> intermetallic materials. Among <sup>285</sup> volume and iron pairs distance for these two compounds 250 these, we can mention de Gennes's factor. Indeed,  $T_C$  is 286 is very close  $(V_{Pr_2Fe_{17}} - V_{Sm_2Fe_{17}} = 3\mathring{A}^3)$  there is a rela-251 governed by the magnetic energies of exchange  $a_{Fe-Fe}$  287 tively large difference in the value of  $T_C$ . This difference  $a_{R-Fe}$ , the strength of  $a_{R-Fe}$  is proportional to de 288 shows that rare-earth-transition metal magnetic energy <sup>253</sup> Gennes factor of rare-earth [75–77]. Ref [23] shows the <sup>289</sup> exchange is much stronger in Sm<sub>2</sub>Fe<sub>17</sub> than in Pr<sub>2</sub>Fe<sub>17</sub>.  $_{254}$  dependence of  $T_C$  on the de Gennes factor in  $R_2$ Fe<sub>17</sub> com-  $_{290}$  In conclusion, the contribution of Sm de Gennes factor 255  $_{256}$  A maximum of  $T_C$  is shown for Gd with the highest de  $_{292}$   $T_C$  after the partial substitution of Pr by Sm. Similar <sup>258</sup> Gennes factor increases with different slopes for light and <sup>294</sup>  $Pr_{2-x}Dy_xFe_{17}$  [65]. <sup>259</sup> heavy rare-earth atom. Table III shows de Gennes factor <sup>260</sup> values tabulated by Kirchmayr and Poldy (1978)[78]

Figure 6. Curie temperature of  $R_2$ Fe<sub>17</sub> for different rare-earth atom

Table III. de Gennes factor value of different rare earth

$4f^n$	$3^+ion$	G
1	Ce	0.18
2	$\mathbf{Pr}$	0.80
3	Nd	1.84
5	$\operatorname{Sm}$	4.46
7	$\operatorname{Gd}$	15.75
8	$\mathrm{Tb}$	10.5
9	Dy	7.08
10	Ho	4.5
11	$\mathbf{Er}$	2.55
12	Tm	1.17

261 263 fact, the sign of the exchange integrals is determined by 299 ferromagnetic state, the magnetization increases notably  $_{264}$  the distances  $d_{\rm FeFe}$ , when  $d_{\rm FeFe}$  is less than 2.45 Å  $J_{\rm FeFe}$  is  $_{300}$  according to the magnetic field. However, for tempera- $_{265}$  negative and the coupling between the pair of iron atoms  $_{301}$  tures higher than  $T_C$ , the material is paramagnetic, the  $_{266}$  is antiferromagnetic [79]. Otherwise,  $d_{\rm FeFe}$  is greater than  $_{302}$  magnetization increases slowly with the applied magnetic

tween tow iron atoms and between iron atom and rare- 269 ence of Fe-Fe distances less than and greater than 2.45 Å  $_{270}$  in  $Pr_2Fe_{17}$  and  $Pr_{1.64}Sm_{0.36}Fe_{17}$  compounds results in a 271 competition between two types of ferromagnetic and an- $S_{\rm Fe}$  and  $J_{\rm R}$  are the spin moment of the iron atom and  $_{272}$  tiferromagnetic couplings (Table II). The coexistence of 273 a negative and positive coupling is the origin of the rel-In Pr<sub>1.64</sub>Sm<sub>0.36</sub>Fe<sub>17</sub>, each atom of Pr or Sm has an av- <sup>274</sup> atively low temperature in the compounds Pr<sub>2</sub>Fe<sub>17</sub> and that  $T_C$  is governed by the 3d - 3d exchange interactions 278 the Fe-Fe distances in the 6c - 6c site, which can debetween the transition metal atoms and the 3d - 4f ex-  $^{279}$  crease the value of Curie temperature. So the increase change interactions between the rare-earth and the tran-  $^{280}$  of  $T_C$  cannot be explained by the variation of the Fe-Fe pounds (R =Ce, Pr, Nd, Sm, Gd, Tb, Dy, Ho, Er, Tm).<sup>291</sup> in magnetic energy exchange can explain the increase of Gennes factor,  $T_C$  increases almost linearly when the de 293 behavior has been noted for other compounds such as



Figure 7. magnetization  $M(\mu_0 H, T)$ curves of  $Pr_{1.64}Sm_{0.36}Fe_{17}$ .

Fig. 7 presents isothermal magnetization  $M(\mu_0 H, T)$ 295 <sup>296</sup> measured at different magnetic fields between 0 and 3 T Another parameter is affecting  $T_C$  in  $R_2$ Fe<sub>17</sub> com- <sup>297</sup> and temperatures from 200 to 340 K. It shows that for pounds which is the distances between the iron atoms. In  $_{298}$  temperatures lower than the  $T_C$  which corresponds to a 267 2.45 Å the exchange integral becomes positive, and the 303 field. We also note that for a given magnetic field, the <sup>304</sup> magnetization decreases when the temperature increases. Fig. 8 shows the magnetization as a function of the 305 magnetic field applied at the temperature 200 K. To de-306 termine the value of the magnetization at saturation and 307 the anisotropy constant we used the following saturation 308 309 law:

$$M = M_s \left( 1 - \frac{a}{H^2} \right)$$

310 with

$$a = \left(\frac{8}{105}\right) \left(\frac{K}{M_s}\right)^2$$

 $_{311}$  Where K and  $M_s$  are the anisotropy constant and the 312 saturation magnetization, respectively. The anisotropy 313 field can be easily deduced from  $H_a = 2K/M_s$ . The  $_{314}$  values of  $M_s$ , K and  $H_a$  are listed in Table IV.



Figure 8. Magnetization M(H) vs magnetic field of  $\mathrm{Pr}_{1.64}\mathrm{Sm}_{0.36}\mathrm{Fe}_{17}$  measured at  $T=200\,\mathrm{K}$  .

We plotted the Arrott plots  $M^2 = f(H/M)$  deduced 315  $_{316}$  from isothermal magnetization M(H) curves (Fig. 9). From the shape and the slope of the curves, we can de-317 termine the type of the phase transition: If the slope is 318 <sup>319</sup> negative and the Arrott plots close to the Curie temper-320 ature have the S shape, the transition from the ferromagnetic state to the paramagnetic state is, therefore, a 321 first-order transition. If the curves have a single inflexion 336 322 <sup>323</sup> point and a positive slope, the transition from the ferro-<sup>337</sup> rameters determined from the equilibrium condition by <sup>324</sup> magnetic state to the paramagnetic one is a second-order <sup>338</sup> plotting  $\mu_0 H$  as a function of the magnetization M. a(T)325 transition.

#### Study of the phase transition according to the С. 326 Landau model 327

328 329  $_{330}$  transition phase is determined by developing free mag-  $_{347}$  phase. For  $Pr_{1.64}Sm_{0.34}Fe_{17}$   $b(T_C) \ge 0$  which confirms  $_{331}$  netic energy as a function of temperature around the  $T_C$ ,  $_{348}$  that the magnetic transition is a second-order type.



Figure 9. The Arrott plots of Pr<sub>1.64</sub>Sm<sub>0.36</sub>Fe<sub>17</sub> compound.

Table IV. magnetic characterization of  $Pr_{2-x}Sm_xFe_{17}$  and  $Pr_2Fe_{17}$ 

	$M_s(\mathrm{Am^2kg^{-1}})$	$T_c(\mathbf{K})$	$K(\mathrm{MJ.m^{-3}})$	$\mu_0 H_a(\mathbf{T})$
$Pr_2Fe_{17}$	165	285	5.09	7.33
$\rm{Pr}_{1.64}Sm_{0.36}Fe_{17}$	146	299	2.49	4.07

<sup>332</sup> neglecting the very high power terms of magnetization. 333 Free energy F as a function of total magnetization can 334 be developed in the form:

$$F = \frac{1}{2}a(T)M^2 + \frac{1}{4}b(T)M^4 + \frac{1}{6}c(T)M^6 - \mu_0 MH$$

335 Applying the equilibrium condition for free energy

$$\frac{dF}{dM} = 0$$

we obtain:

$$u_0 H = a(T)M + b(T)M^3 + c(T)M^5$$

The coefficients a(T), b(T) and c(T) are the Landau pa- $_{339}$  and c(T) are always positive (Fig 10 and Fig. 11), from  $_{340} a(T)$  we can determine the value of the  $T_C$  which corre-<sup>341</sup> sponds to its minimum. However, b(T) can be positive,  $_{342}$  zero, or negative, the sign of b(T) can indicate if the mag-343 netic transition is first-order phase transition or second-<sup>344</sup> order phase one (Fig. 12). Indeed, if  $b(T_C) > 0$  the mag-We develop in this part the Landau model to confirm <sup>345</sup> netic transition is a second-order transition phase, oththe nature of the phase transition. The order of the 346 erwise the magnetic transition is a first-order transition



Figure 10. Landau parameter a(T) vs temperature around the  $T_C$  for  $Pr_{1.64}Sm_{0.36}Fe_{17}$  sample.



Figure 11. Landau parameter c(T) vs temperature around the T<sub>C</sub> for Pr<sub>1.64</sub>Sm<sub>0.36</sub>Fe<sub>17</sub> sample.

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## D. Magnetocaloric effect

To calculate the variation of the magnetic entropy the application of a magnetic field, it is necessary to integrate the following Maxwell relation:

$$\Delta S_{\rm M} = \mu_0 \, \int_0^H \, \left(\frac{\partial M}{\partial T}\right)_H \, dH$$

Where M is the magnetization, T is the temperature, and H is the applied magnetic field. Since, the isothermal M(H) curves are measured by discrete field changes, the following expression might be used [80]:

$$|\Delta S_{\mathrm{M}}| = \mu_0 \sum_{i} \frac{M_i - M_{i+1}}{T_{i+1} - T_i} \,\Delta H_i$$



Figure 12. Landau parameter b(T) vs temperature around the T<sub>C</sub> for Pr<sub>1.64</sub>Sm<sub>0.36</sub>Fe<sub>17</sub> sample.

<sup>357</sup> Where  $M_i$  and  $M_{i+1}$  are the initial magnetization at <sup>358</sup>  $T_i$  and  $T_{i+1}$  temperature respectively when the magnetic <sup>359</sup> field increases by  $\Delta H_i$ . In table V, we have presents <sup>360</sup>  $-\Delta S_{\rm M}$  values for some  $R_2$ Fe<sub>17</sub> compounds. It clearly <sup>361</sup> shows that (Pr,Sm)<sub>2</sub>Fe<sub>17</sub> has a good magnetocaloric ef-<sup>362</sup> fect at a relatively low field.

Table V.  $T_C$ , magnetic entropy change and magnetic field change of some  $R_2$ Fe<sub>17</sub> compounds.

Compound	$T_C$	$-\Delta S_{\rm M}$	$\mu_0 \Delta H$	Ref
	(K)	(J/kgK)	(T)	
$Y_2 Fe_{17}$	303	3.2	5	[8]
$Lu_2Fe_{17}$	264	1.5	2	[10]
$\mathrm{Nd}_{2}\mathrm{Fe}_{17}$	339	2.5	1.5	[81]
$\mathrm{Nd}_{2}\mathrm{Fe}_{17}$	331	3.4	1.5	[18]
$\mathrm{Gd}_{2}\mathrm{Fe}_{17}$	475	1.2	1.5	[16]
$Pr_2Fe_{17}$	286	2	2	[13]
$\rm Pr_{1.64}Sm_{0.36}Fe_{17}$	299	2.5	2	This work
$\operatorname{Gd}$	293	4.8	2	[82]
GdNi	69	9	2	[83]
HoNi	37	6.8	2	[83]
${ m ErNi}$	10	15	2	[83]
$\mathrm{Gd}_3\mathrm{Co}$	128	5	2	[84]
$\mathrm{Tb}\mathrm{Co}_2$	230	4.61	3	[85]
$\mathrm{Gd}_5\mathrm{Ge}_2\mathrm{Se}_2$	275	14	2	[82]

Fig. 13 illustrates the variation of the magnetic entropy around Curie temperature of  $Pr_{1.64}Sm_{0.36}Fe_{17}$ for external variation of the magnetic field between of 0 and 3 T. For a magnetic field change  $\mu_0\Delta H=3$  T  $\lambda_{57} |\Delta S_{\rm M}| \sim 3.5 \,\text{J/(kg.K)}$  at T= 305 K. The partial substitution of Pr By Sm increase the operating temperature that corresponds to the maximum of  $|\Delta S_{\rm M}|$ , is higher that corresponds to the maximum of  $|\Delta S_{\rm M}|$ , is higher in  $Pr_{1.64}Sm_{0.36}Fe_{17}$  compared to the parent compound



Magnetic entropy change  $\Delta S(T,\mu_0 H)_{\rm M}$  for  $^{\rm 400}$ Figure 13.  $Pr_{1.64}Sm_{0.36}Fe_{17}$  sample.

Table VI. Magnetocaloric effect of Pr<sub>1.64</sub>Sm<sub>0.36</sub>Fe<sub>17</sub>.

$\mu_0 \Delta H(\mathrm{T})$	1	1.6	2	2.6	3
$ \Delta S_{\rm M} ({\rm J}/({\rm kg.K}))$	1.22	1.87	2.5	2.98	3.24
$\delta T^{\rm FWHM}({\rm K})$	51	58	63.76	69.4	78
RCP(J/kg)	62.22	108.46	159.4	202.65	247.26
$RC_{Area}(J/kg)$	48.46	86.81	121.1	164	195.1

 $_{371}$  Pr<sub>2</sub>Fe<sub>17</sub>.  $|\Delta S_{\rm M}|$  remains almost constant after the substi-<sup>372</sup> tution with Sm. Even though many research studies have reported materials with high ECM at low temperatures, 373 there are few materials with high MCE and second-order 374 transition phase from ferromagnetic state to paramag-375 netic state around to 300 K. 376

The relative cooling power (RCP) is an essential pa-377 rameter in the magnetocaloric application. It is related 378 with the maximum of entropy change by the following  $_{380}$  relation [37]:

## $RCP = -\Delta S_{\rm M} \ \delta T^{\rm FWHM}$

381 382  $_{333}$  change, the maximum of  $-\Delta S_{\rm M}$ , the variation of the tem-  $_{428}$  magnetic transition from the ferromagnetic state to the  $_{324}$  perature  $\Delta T$  and relative cooling power (RCP) are given  $_{429}$  paramagnetic state is a first order transition phase show a 385 in Table VI.

Another important parameter to evaluate the magne-386 tocaloric effect is the refrigerant capacity  $(RC)_{Area}$ , it is 387 the amount of heat that can be transferred in one thermo-388 dynamic cycle. The RC value represents the area under 389  $-\Delta S_{\rm M}$  curve using the temperature of FWHM. It can be 390 <sup>391</sup> calculated by integrating the following expression:

$$RC_{\text{Area}}(H) = \int_{T_{min}}^{T_{max}} \Delta S_M(T, H) dT;$$
$$T_{max} - T_{min} = \delta T^{FWHM}$$

In addition to RCP and  $RC_{area}$ , temperature-averaged entropy change (TEC) is another parameter that has 394 been used to check the applicability of materials for mag-395 netic refrigeration technologies. TEC was introduced by L. D. Griffith et al. [90] as another figure of merit. TEC <sup>398</sup> can be calculated using magnetic entropy change data as 399 follows:

$$TEC = \frac{1}{\Delta T_{H-C}} \max \left\{ \int_{T_{mid} - \frac{\Delta T_{H-C}}{2}}^{T_{mid} + \frac{\Delta T_{H-C}}{2}} |\Delta S_M(T)| dT \right\}$$

Where  $\Delta T_{H-C}$  is defined as the temperature range for 401 the measuring device and represents the difference be- $_{402}$  tween hot and cold heat exchangers.  $T_{mid}$  represents <sup>403</sup> the temperature value in the middle of  $\Delta T_{H-C}$  that 404 maximises TEC. Fig. 14 shows the variation of TEC 405 with  $\Delta T_{H-C}$  under several magnetic field changes be- $_{406}$  tween 1 T and 3 T. One can see that TEC decreases 407 slightly and monotonously with  $\Delta T_{H-C}$ , it is very close 408 to the maximum of entropy change value for low values 409 of  $\Delta T_{H-C}$ . It is easy to see that TEC increases mono-410 tonically with the applied magnetic field for a given tem-<sup>411</sup> perature. The obtained TEC( $\Delta T_{H-C}=5$  K,  $\mu_0$ H=3 T) <sup>412</sup> and TEC( $\Delta T_{H-C}=5$  K,  $\mu_0$ H=1 T) for Pr<sub>1.64</sub>Sm<sub>0.36</sub>Fe<sub>17</sub> <sup>413</sup> are 3.71  $Jkg^{-1}.K^{-1}$  and 1.4  $Jkg^{-1}.K^{-1}$ , respectively. <sup>414</sup> Table VII shows TEC value of our sample and TEC val- $^{_{415}}$  ues of other materials for comparison.  $\mathrm{Pr}_{1.64}\mathrm{Sm}_{0.36}\mathrm{Fe}_{17}$  $^{_{416}}\frac{TEC}{\Delta\mu_0H}$  is comparable to  $\frac{TEC}{\Delta\mu_0H}$  of  $\mathrm{Pr}_{0.5}\mathrm{Er}_{0.1}\mathrm{Sr}_{0.4}\mathrm{MnO}_3$  $_{417}$  [86] and TmFe<sub>0.7</sub>Mn<sub>0.3</sub>O<sub>3</sub> [87], slightly higher than  $\frac{TEC}{\Delta \mu_0 H}$ <sup>418</sup> of  $Pr_{0.5}Eu_{0.1}Sr_{0.4}MnO_3$  [86],  $\frac{TEC}{\Delta\mu_0 H}$  of  $TmFe_{0.8}Mn_{0.2}O_3$ <sup>419</sup> [87],  $\frac{TEC}{\Delta\mu_0 H}$  of La<sub>0.8</sub>Na<sub>0.2</sub>MnO<sub>3</sub> [88], and  $\frac{TEC}{\Delta\mu_0 H}$  of  $_{420}$  Ho<sub>2</sub>CoMn $O_6$  [89].

RCP for  $Pr_{1.64}Sm_{0.36}Fe_{17}$  is around ~ 250 J/kg for a 421 <sup>422</sup> magnetic field change equal to 3 T.

There are several magnetocaloric materials that have 423 <sup>424</sup> been studied for magnetic refrigeration at room tempera-<sup>425</sup> ture: Gd<sub>5</sub>(Ge,Si)<sub>4</sub> [27, 33–35], La(Fe,Co,Si)<sub>13</sub>(H,C) [36– Where  $\delta T^{\text{FWHM}}$  is the full width at half-maximum  $_{426}$  41], Fe<sub>2</sub>P [42–45], manganites R<sub>1-x</sub>A<sub>x</sub>MnO<sub>3</sub> [46–49] and in the temperature dependence of the magnetic entropy 427 related alloys and compounds. In these materials the  $_{430}$  large  $|\Delta S_{\rm M}|$  at operating temperature, but possess a low

Material	$T_{mid}$	$\Delta \mu_0 H$	$\Delta T_{H-C}$	TEC	$\frac{TEC}{\Delta \mu_0 H}$	Ref
	Κ	Т	Κ	$J(Kkg)^{-1}$	$J(KkgT)^{-1}$	
$\rm Pr_{1.64}Sm_{0.36}Fe_{17}$	300	3	5	3.71	1.24	This work
$\mathrm{Pr}_{0.5}\mathrm{Eu}_{0.1}\mathrm{Sr}_{0.4}\mathrm{MnO}_3$	279.9	5	5	4.45	0.89	[86]
$\mathrm{Pr}_{0.5}\mathrm{Eu}_{0.1}\mathrm{Sr}_{0.4}\mathrm{MnO}_3$	188.5	5	5	4.9	0.98	[86]
$\mathrm{Tm}\mathrm{Fe}_{0.7}\mathrm{Mn}_{0.3}\mathrm{O}_3$	11	7	5	7	1	[87]
$\mathrm{Tm}\mathrm{Fe}_{0.8}\mathrm{Mn}_{0.2}\mathrm{O}_3$	12	7	5	6.4	0.914	[87]
$\rm La_{0.8}Na_{0.2}MnO_3$	330	5	5	4.6	0.92	[88]
$\mathrm{La_{0.8}Ca_{0.05}Na_{0.15}MnO_{3}}$	315	5	5	4.55	0.91	[88]
${\rm Ho_2CoMnO_6}$	77	7	5	6.5	0.928	[89]

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Table VII. TEC value of our sample and TEC values of other materials for comparison.



Figure 14. TEC vs  $\Delta T_{H-C}$  at several magnetic fields for  $Pr_{1.64}Sm_{0.36}Fe_{17}$ .

<sup>431</sup> RCP because of their low working temperature range  $_{\rm 432}~\delta T^{\rm FWHM} \sim 10\,{\rm K}$  as opposed to materials with second or-<sup>433</sup> der transition phase like the parent compound  $Pr_2Fe_{17}$ <sup>434</sup> with  $\delta T^{FWHM} \sim 100 \text{ K}$  [72].

For  $Pr_{1.64}Sm_{0.36}Fe_{17}$  RCP is found to be around 435  $159.4 \,\mathrm{J.kg^{-1}}$  under a magnetic field change of  $2 \,\mathrm{T}$  which 436 is around 75% of that observed in pure Gd and makes 437 our compound a potential candidate for magnetic refrig-438 eration around the room temperature [82]. 439

440 441 442 443 ample, the materials mentioned above like (Gd<sub>5</sub>(Ge,Si)<sub>4</sub>, 466 of rescaled temperature [94]. 444 La(Fe,Co,Si)<sub>13</sub>(H,C),Fe<sub>2</sub>P) exhibit a first-order transi-445 tion phase, show a large  $|\Delta S_{\rm M}|$ . However, with a rela-446 tively large thermal hysteresis contrary to materials that 447 <sup>448</sup> display a second-order magnetic transition. Compared to <sup>449</sup> these materials,  $Pr_{1.64}Sm_{0.36}Fe_{17}$  exhibits a second-order  $_{450}$  magnetic transition at a T<sub>C</sub> very close to room temper-  $_{468}$  Where  $\Theta$  is the rescaled temperature, Tr1 and Tr2 are <sup>451</sup> ature with a moderate  $|\Delta S_{\rm M}|$  and high *RCP*.

<sup>452</sup> It should be mentioned that detailed researches show that <sup>453</sup> the Maxwell relation can not be utilized close to the Curie <sup>454</sup> temperature because of the coexistence of paramagnetic 455 and ferromagnetic phases; the sudden drop in the mag-<sup>456</sup> netization causes a discontinuity in the entropy change <sup>457</sup> at Curie temperature. Consequently, the large entropy 458 peak is a fallacious result because of the inappropriate <sup>459</sup> use of the Maxwell relation [92, 93]. For materials that



Figure 15. Universal curve of entropy change VS reduced temperature  $\Theta$  for  $Pr_{1.64}Sm_{0.36}Fe_{17}$  sample

461 From the point of view of magnetic refrigeration, the 462 exhibit a second-order magnetic phase transition from irreversibility of the magnetocaloric effect is a major dis- 463 the ferromagnetic to the paramagnetic state, it is possiadvantage. Indeed, a minimal thermal hysteresis is a 464 ble to obtain a universal phenomenological curve using basic constraint for a practical refrigerant [91]. For ex- 465 normalized entropy change  $-\Delta S/-\Delta S_{\text{max}}$  as a function

$$\Theta = -(T - T_{\rm C})/(T_{\rm r1} - T_{\rm C}), T \le T_{\rm C}$$
  
 $\Theta = (T - T_{\rm C})/(T_{\rm r2} - T_{\rm C}), T \ge T_{\rm C}$ 

469 the temperatures of the two reference points that have

470 been selected as those corresponding to  $-\Delta S_{\mathrm{M}} = -a \times$  $_{471} \Delta S_{\text{max}}$  (where  $0 \le a \le 1$ ) [95]. Some theoretical studies 472 have proved that it is not necessary to use two reference  $_{473}$  point temperatures lower and higher than  $T_C$  Tr1 and Tr2, but only one Tr can be used for a material with a 475 single magnetic phase. Nevertheless, in the case of an im-476 portant demagnetization factor or the existence of multiple magnetic phases, it is required to use two reference 477 temperature points [96]. 478

479 netic homogeneity. This approach should remove the de-480 pendency of the set of curves  $\Delta S_{\rm M}({\rm H, T})$  on both tem-481 perature and field so that all curves treated using the 482 same scaling procedure will collapse on a common uni-483 versal curve. A lack of success in achieving this universal 484 collapse can be explained by a magnetic inhomogeneity 485 in the material [96]. As described above, we use detailed <sup>487</sup> statistical analysis to determine the magnetic homogeneity in the material and the order of magnetic transition. 488 After rescaling temperature and choosing a=0.5, 489 <sup>490</sup> Fig. 13 shows that the normalized entropy change for <sup>491</sup> all applied magnetic fields collapses in a single universal  $_{492}$  curve. This suggests that  $Pr_{1.64}Sm_{0.36}Fe_{17}$  is magnet-<sup>493</sup> ically homogeneous undergoes a second-order magnetic <sup>494</sup> transition phase around  $T_C = 299 \text{ K}$ .



Figure 16. Field dependence of magnetic entropy change at  $T_C$  for  $Pr_{1.64}Sm_{0.36}Fe_{17}$  sample.

Fig. 13 clearly shows that magnetic entropy change 495  $_{496}$  of  $\mathrm{Pr}_{1.64}\mathrm{Sm}_{0.36}\mathrm{Fe}_{17}$  sample reach its maximum around <sup>497</sup> Curie temperature. The maximum of magnetic entropy <sup>498</sup> change depends strongly on the applied magnetic field. A 499 nonlinear relation can be expected between  $\Delta S_M^{max}$  and 500  $\mu_0 H$  as follows:

$$\Delta S_M^{max} \sim \mu_0 H^n$$

501 <sup>502</sup> real ferromagnetic material the previous relation was rec-<sup>530</sup> magnetic field; it is known that for materials with a <sup>503</sup> tified by Lyubina *et al* [98]. It was modified as follow:

$$\Delta S_M^{max} = A(\mu_0 H + H_0)^n - AH_0^{2/3} + B(\mu_0 H)^{4/3}$$

Where A and B are intrinsic parameters of the material  $_{\rm 505}$  and  $H_0$  is a measure of homogeneity of the compound. 506 Fig. 16 shows  $\Delta S_M^{max}$  vs  $\mu_0 H$  and fit parameters. One 507 can easily notice that the fit of the experimental data 508 with  $\Delta S_M^{max} = a(\mu_0 H)^{2/3}$  equation is failed, on the other The analysis of the universal scale demonstrates mag-tic homogeneity. This approach should remove the de-stic homogeneity been in good agreement with the  $_{511}$  MCE behaviour of  $Pr_{1.64}Sm_{0.36}Fe_{17}$  sample, similar be-<sup>512</sup> havior was observed in Ref. [86, 99]. It confirms the 513 second-order nature of his magnetic transition, as demon-<sup>514</sup> strated above by Both the Landau model and Arrott plot. 515 The experimental data follow  $\Delta S_M^{max} = a(\mu_0 H)^n$  equa-<sup>516</sup> tion with the local exponent n equals to 0.581. The ob-517 tained values of n are slightly deviated from the typical value of n = 2/3 within the mean-field model [100]. This <sup>519</sup> deviation justifies the invalidity of the mean-field model 520 for our sample. For materials that exhibit a second-order <sup>521</sup> magnetic transition phase, it has been demonstrated in <sup>522</sup> previous work that magnetic entropy change can be ex-523 pressed as  $\Delta S_M^{max} \propto H^{n(T,H)}$  [101]. The exponent n can <sup>524</sup> be locally determined using the following relation:

$$n(T,H) = \frac{dln\Delta_M}{dlnH}$$



Figure 17. Temperature dependence of the local exponent n at different magnetic fields from 1 to 3 T for  $Pr_{1.64}Sm_{0.36}Fe_{17}$ sample.

In the paramagnetic state  $n \rightarrow 2$  when  $T \gg T_C$ , in the 525 526 ferromagnetic state n $\longrightarrow$ 1 when T  $\ll T_C$  and around <sub>527</sub> magnetic phase transition  $(T \rightarrow T_C) n(T_C) = 1 + (\beta - \beta)$  $_{528}$  1)/( $\beta + \gamma$ ). Previous works have reported that the exwhere n=2/3 in the case of mean field model [97]. For 529 ponent n depends on the temperature and the applied <sup>531</sup> single magnetic phase, n is field independent at Curie

<sup>532</sup> temperature. Nevertheless, in the case of materials that <sup>553</sup> <sup>533</sup> present multi magnetic phase, n is field-dependent at any <sup>534</sup> temperature [102]. Fig. 17 shows the variation of the  $_{535}$  local exponent n with the temperature at several mag-536 netic fields calculated using the logarithmic derivative of  $_{537}$   $\Delta S_M$  with an applied magnetic field. Fig. 17 shows that 538 the exponent n is field-dependent below and above  $T_C$ , 539 in contrast, at  $T=T_C$  is almost constant, unlike the be-540 haviour observed for multiphase systems where n is field-<sup>541</sup> dependent in Ref. [103, 104]. The Rietveld refinement of  $_{542}$  Pr<sub>1.64</sub>Sm<sub>0.36</sub>Fe<sub>17</sub> sample, a thermal derivative of magne-543 tization, the universal curve of magnetic entropy change <sup>544</sup> and the behaviour of n exponent at Curie temperature indicates that our sample presents a single-phase and a 545  $_{\rm 546}$  single magnetic phase transition. For different magnetic  $_{\rm 547}$  field  $n\sim 0.58,$  is deviated from the n value of the mean-548 field model but is consistent with the value calculated <sup>549</sup> from  $n(T_C)$  expression using the values of  $\beta$  and  $\gamma$  of 550 the Ising model ( $\beta = 0.325, \gamma = 1.241$ ), this suggests <sup>551</sup> that the mean-field model is not the appropriate model <sup>552</sup> to describe the magnetic phase transition in our sample. <sup>570</sup>

#### IV. CONCLUSION

554 The structural, magnetic, and magnetocaloric proper- $_{555}$  ties have been studied for the intermetallic (Pr, Sm)<sub>2</sub>Fe<sub>17</sub> <sup>556</sup> compound. X-ray diffraction at room temperature shows 557 that this compound crystallizes in the rhombohedral  $_{558}$  Th<sub>2</sub>Zn<sub>17</sub> type structure. Rietveld refinement shows that  $_{\rm 559}$  the cell parameters decrease when the Sm replaces the Pr. 560 After the substitution of Pr by the Sm  $T_C$  increases from  $_{561}$  285 to 299 K. This increase of  $T_C$  is mainly due to the de  $_{\rm 562}$  Gennes factor. The study of the phase transition accord- $_{563}$  ing to the Landau model proves that  $Pr_{1.64}Sm_{0.36}Fe_{17}$ <sup>564</sup> exhibits a second-order transition phase. The magnetic 565 entropy change, relative cooling power and TEC, were 566 calculated. The value of  $T_C$ , which is close to the room <sup>567</sup> temperature and a large temperature range with a quasi 568 constant  $\Delta S_{\rm M}$  value, consequently the studied compound <sup>569</sup> exhibits a good magnetocaloric effect.

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- [1] J. X. Zhang, L. Bessais, C. Djega Mariadassou, 606 575 E. Leroy, and A. Percheron Guegan. Appl. Phys. Lett., 576 607 80(11):1960-1962, 2002. 608 577
- M. Murakami. J. Appl. Phys., 101(9):09C522, 2007. 578
- [3] J. F. Liu, P. Vora, M. H. Walmer, E. Kottcamp, 610 579 S. A. Bauser, A. Higgins, and S. Liu. J. Appl. Phys., 580 97(10):10H101, 2005. 581
- [4] K. J. Strnat. Cobalt, 36 (1967) 133-143, 1967. 582
- R. Fersi, N. Mliki, L. Bessais, R. Guetari, V. Russier, [5] 583
- and M. Cabie. J. Alloys Compd., 522:14-18, 2012. 584 W. Bouzidi, N. Mliki, and L. Bessais. J. Magn. Magn. 585 [6]
- Mater., 441:566-571, 2017. 586
- A. M. Tishin and Y. I. Spichkin. Institute of Physics 587 |7|618 Publishing, Bristol and Philadelphia, 2003. 588
- S. Yu Dankov, A. M. Tishin V. V. Ivtchenko, K. A. Jr. 8 589 Gschneidner, and V. K. Pecharsky. Adv. Cryog. Eng., 590 46:397, 2000. 591
- G. Calestani, N. Magnani, A. Paoluzi, L. Pareti, and [9] 592 C. Rizzoli. Phys. Rev. B, 68(5):054424, 2003. 593
- A. M. Tishin and Y. I. Spichkin. Institute of Physics [10]625 594 Publishing, Dirac House Temple Back, Bristol UK, 595 626 2003.596
- K. Mandal, A. Yan, P. Kershl, A. Handstein, O. Gut-[11] 597 fleisch, and K. H. Muller. J. Phys. D, 37:2628, 2004. 598
- H. Chen, Y. Zhang, J. Han, H. Du, Ch. Wang, and 599 630 Y. Yang. J. Magn. Magn. Mater., 320:1382–1384, 2008. 631 600
- P. Gorria, P. Alvarez, J. Sanchez Marcos, J. Sanchez-601 [13]632 Llamazares, M. J. Perez, and J. A. Blanco. Acta Mater., 602 633 57:1724–1733, 2009. 634 603
- S. Dan, S. Mukherjee, C. Mazumdar, and R. Ran-[14] 604 635 ganathan. RSC Adv, 6(97):94809-94814, 2016. 605

- [15] S. Dan, S. Mukherjee, C. Mazumdar, and R. Ranganathan. J. Phys. Chem. Solids, 115:92–96, 2018.
- [16]M. Saidi, K. Nouri, S. Walha, E. Dhahri, A. Kabadou, M. Jemmali, and L. Bessais. J. Electron. Mater., 48:2242-2253, 2019.

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620

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623

624

627

628

629

- [17] S. Dan, S. Mukherjee, C. Mazumdar, and R. Ranganathan. J. Phys. Chem. Solids, 21(5):2628-2638, 2019.
- W. Bouzidi, K. Nouri, T. Bartoli, R. Sedek, H. Lassri, [18]J. Moscovici, and L. Bessais. J. Magn. Magn. Mater., 497:166018, 2020.
- Y. Cao, K. Lin, Z. Liu, J. Hu, C-W. Wang, E. Tereshina-[19]Chitrova, K. Kato, Q. Li, J. Deng, J. Chen, et al. Inorg. Chem., 59(16):11228-11232, 2020.
- [20]X. C. Kou, F. R. de Boer, R. Grössinger, G. Wiesinger, H. Suzuki, H. Kitazawa, T. Takamasu, and G. Kido. J. Magn. Magn. Mater., 177-181:1002-1007, 1998.
- [21]O. Isnard, S. Miraglia, D. Fruchart, E. Akiba, and K. Nomura. J. Alloys Compd., 257:150-155, 1997.
- K. Mori, A. E. Clark, and O. D. McMasters. J. Magn. [22]Magn. Mater., 31-34:855-856, 1983.
- H. Sun, J. M. D Coey, Y. Otani, and D. P. F. Hurley. [23]J. Phys. Condens. Matter, 2:6465-6470, 1990.
- [24]G. J. Long, O. A. Pringle, F. Grandjean, W. B. Yelon, and K. H. J. Buschow. J. Appl. Phys., 73:6050, 1993.
- Y. Zeng, Z. H. Lu, N. Tang, X. W. Li, R. W. Zhao, and [25]F. M. Yang. J. Magn. Magn. Mater., 139:11-18, 1995.
- [26]S. A. Nikitin, G. Myalikgulyev, K. A. Asatryan A. M. Tishin, M. P. Annaorazov, and A. L. Tyurin. Phys. Lett. A, 148:363–366, 1990.

- [27] V. K. Pecharsky and K. A. Gschneidner. Phys. Rev. 700 636 637 Lett., 78:4494, 1997. 701
- [28] H. Wada, S. Tomekawa, and M. Shiga. Cryogenics, 638 702 39:915-919, 1999. 639
- [29]H. Wada and Y. Tanabe. Appl. Phys. Lett., 79:3302-640 704 3304, 2001. 641 705
- [30] K. A. Gschneidner, V. K. Pecharsky, and A. O. Tsokol. 642 Rep. Prog. Phys., 68:1479, 2005. 643
- [31 A. Fujita. J. Appl. Phys., 127:123902, 2020. 644
- [32]W. J. Lai, X. M. You, I Dugulan, B. W. Huang, J. Liu, 709 645 M. Maschek, L. van Eijck, N. Dijk, and E. Bruck. J. 710 646 Alloys Compd., 821:153451, 2020. 647
- [33] V. K. Pecharsky and K. A. Gschneidner. J. Magn. 648 Magn. Mater., 167:L179, 1997. 649
- [34] K. A. Gschneidner, V. K. Pecharsky, E. Brück, H. G. M. 714 650 Duijin, and E. M. Levin. Phys. Rev. Lett., 85:4190, 2000. 715 651
- [35] H. Tang, V. K. Pecharsky, G. D. Samolyuk, M. Zou, 716 652 K. A. Gschneidner, V. P. Antropov, D. L. Schlagel, and 717 653 T. A. Lograsso. Phys. Rev. Lett., 93:237203, 2004. 654 718
- [36] S. Fujieda, A. Fujita, and K. Fukamichi. Appl. Phys. 655 719 Lett., 81:811276-8, 2002. 656
- [37] A. Fujita, S. Fujieda, and K. Fukamichi. J. Magn. Magn. 657 Mater., 310:e1006-e1007, 2007. 658
- [38] A. Fujita, S. Fujieda, and K. Fukamichi. J. Magn. Magn. 659 Mater., 321:3553-3558, 2009. 660
- [39] M. Phejar, V. Paul-Boncour, and L. Bessais. Inter-661 metallics, 18:2301, 2010. 662
- [40] A. Boutahar, M. Phejar, V. Paul-Boncour, L. Bessais, 663 727 and H. Lassri. J. Supercond. Nov. Magn., 27:1795-1800, 728 664 2014.665
- [41] M. Phejar, V. Paul-Boncour, and L. Bessais. J. Solid 730 666 State Chem., 223:95-102, 2016. 667 731
- [42] O. Tegus, B. Fuquan, W. Dagula, L. Zhang, E. Brück, 732 668 P. Z. Si, F. R. de Boer, and K. H. J. Buschow. J. Alloys 733 669 Compd., 396:6-9, 2005. 670
- [43] A. Bartok, M. Kustov, L. F. Cohen, A. Pasko, K. Ze-671 735 hani, L. Bessais, F. Mazaleyrat, and M. LoBue. J. 736 672 Magn. Magn. Mater., 400:333-338, 2016. 673
- [44] A. Bartok, M. Kuepferling, C. Curcio, V. Basso, 738 674 A. Pasko, K. Zehani, L. Bessais, F. Mazaleyrat, and 739 675 M. Lobue. Refrigeration Science and Technology, pages 740 676 119-122. 2016. 677
- [45] A. Pasko, A. Bartok, K. Zehani, L. Bessais, F. Maza-742 678 leyrat, and M. LoBue. AIP Advances, 6:056204, 2016. 743 679
- 680 [46] A. R. Dinesen, S. Linderoth, and S. Morup. J. Phys. 744 Condens. Matter, 17:6257, 2005. 681 745
- [47] Z. Wei, A. Chak-Tong, and D. You-Wei. Chin. Phys. B, 682 22:057501, 2013. 683
- [48] H. Felhi, M. Smari, A. Bojorek, K. Nouri, E. Dhahri, 748 684 and L. Bessais. Prog. Nat. Sci., 29:198-209, 2019. 685
- [49] N. Ameur, F. Elleuch, M. Triki, E. Dhahri, L. Bessais, 686 and E. K. Hlil. Solid State Comm., 289:30-37, 2019. 687
- [50] L. Bessais, E. Dorolti, and C. Djega-Mariadassou. Appl. 688 Phys. Lett., 87, 2005. 689
- [51] N. Hosni, K. Zehani, T. Bartoli, L. Bessais, and 754 690 H. Maghraoui-Meherzi. J. Alloys Compd., 694:1295-691 1301, 2017. 692
- S. Khazzan, N. Mliki, L. Bessais, and C. Djega-[52]693 757 Mariadassou. J. Magn. Magn. Mater., 322(2):224-229, 758 694 2010.695 759
- R. Bensalem, W. Tebib, S. Alleg, J. J. Sunol, L. Bessais, [53]760 696 and J. M. Greneche. J. Alloys Compd., 471:24-27, 2009. 761 697
- K. Zehani, R. Bez, A. Boutahar, EK Hlil, H. Lassri, 762 54698 J. Moscovici, N. Mliki, and L. Bessais. J. Alloys Compd., 763 690

591:58-64, 2014.

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706

707

708

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726

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750

751

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753

755

756

- [55] A. Hamrita, Y. Slimani, M. K. Ben Salem, E. Hannachi, L. Bessais, F. Ben Azzouz, and M. Ben Salem. Ceram. int., 40:1461–1470, 2014.
- [56] H. Rietveld. Acta Crystallogr., 22:151, 1967.
- [57]H. Rietveld. J. Appl. Crystallogr., 2:65, 1969.
- [58]J. Rodriguez-Carvajal. Physica B, 192:55, 1993.
- [59]J. Rodriguez-Carvajal, M. T. Fernandez-Diaz, and J. L. Martinez. J. Phys. , 81:210, 2000.
- [60]L.Bessais, S. Sab, C. Djega-Mariadassou, NH. Dan, and NX. Phuc. Phys Rev B, 70(13):134401, 2004.
- [61]C. Djega-Mariadassou, L. Bessais, A. Nandra, and E. Burzo. Phys. Rev. B, 68:24406, 2003.
- [62] L. Bessais, S. Sab, C. Djega-Mariadassou, N. H. Dan, and N. X. Phuc. Phys. Rev. B, 70:134401, 2004.
- [63] K. Younsi, V. Russier, and L. Bessais. J. Appl. Phys., 107:083916, 2010.
- Qu. Johnson, D. H. Wood, G. S. Smith, and A. E. [64]Ray. Acta Crystallogr. B Struct. Cryst. Cryst. Chem, 24(2):274-276, 1968.
- [65] R. Guetari, R. Bez, C. B. Cizmas, N. Mliki, and L. Bessais. J. Alloys Compd., 579:156-159, 2013.
- S. Khazzan, N. Mliki, and L. Bessais. J. Appl. Phys., [66]105:103904, 2009.
- [67]C. Romero-Muniz, J. J. Ipus, J.S. Blázquez, V. Franco, and A. Conde. Appl. Phys. Lett., 104(25):252405, 2014.
- A. K. Pramanik and A. Banerjee. J. Phys: Condens. [68]Matter, 20(27):275207, 2008.
- [69] M. Vatanparast, Y.T. Shao, M Rajpalke, B. Fimland, T. Reenaas, E. Holmestad, P. Vullum, and J.M. Zuo. Ultramicroscopy, page 113299, 2021.
- [70] I. V. Bodnar, I.A. Victorov, O.V Alita, V.V. Khoroshko, and E. Arushanov. Solid State Sci., 113:106550, 2021.
- [71] L. Vegard. Zeitschrift für Physik, 5(1):17-26, 1921.
- [72] R. Guetari, R. Bez, A. Belhadi, K. Zehani, A. Bezergheanu, N. Mliki, L. Bessais, and C. B. Cizmas. J. Alloys Compd., 588:64-69, 2014.
- [73] K. H. J. Buschow. volume 4, page 44. North-Holland, Amsterdam, north-holland, amsterdam edition, 1988.
- [74] L. Bessais, C. Djega-Mariadassou, A. Nandra, M. D. Appay, and E. Burzo. Phys. Rev. B, 69:64402, 2004.
- [75] K. Inoue, Y. Nakamura, A. V. Tsvyashchenko, and L. Fomichiva. J. Phys. Soc. Jpn., 64(6):2175–2182, 1995.
- [76] M. R. Ibarra J. L. Wang, C. Marquina and G. H. Wu. Phys. Rev. B, 73(9):094436, 2006.
- R. M. Bozorth. J. Appl. Phys., 38(3):1366-1371, 1967. [77]
- [78]H. R. Kirchmayr and C. A. Poldy. J. Magn. Magn. Mater., 8(1):1-42, 1978.
- Z. W. Li and X. A. H. Morrish. Phys. Rev. B, 55:3670-[79]3676, 1997.
- M. Foldeaki, R. Chahine, and T. K. Bose. J. Appl. [80]Phys., 77:3528, 1995.
- [81] P. Alvarez, P. Gorria, V. Franco, J. Sanchez Marcos, M. J. Perez, J. Sanchez-Llamazares, I. Puente-Orench, and J. Blanco. J. Phys. Condens. Matter, 22:216005, 2010.
- [82]V. K. Pecharsky and K. A. Gschneidner Jr. Phys. Rev. Lett., 78(23):4494, 1997.
- P. Kumar, KG. Suresh, AK. Nigam, and O. Gutfleisch. [83]J. Phys. D J PHYS D APPL PHYS, 41(24):245006, 2008
- [84]S K. Tripathya, K G. Suresha, and A K. Nigam. J. Magn. Magn. Mater. 306 24, 2006.

- [85] Z. Jun-Ding, S. Bao-Gen, , and S. Ji-Rong. Chin. Phys. 788 764 Soc., 16(12):3843, 2007. 765
- [86] A. Sakka, R. M'nassri, MM. Nofal, S. Mahjoub, 790 766 W. Cheikhrouhou-Koubaa, N. Chniba-Boudjada, 791 767
- M. Oumezzine, and A. Cheikhrouhou. J. Magn. Magn. 792 768 Mater., 514:167158, 2020. 769 793
- [87] L. Su, X.Q. Zhang, Q.Y. Dong, H. T. Yangand S. H. 770 794 Li, and Z. H. Cheng. Ceram. Int, 47(13):18286-18294, 771 795 2021. 772 796
- [88] S. Choura-Maatar, M.M. Nofal, R. Mnassri, 797 773 W. Cheikhrouhou Koubaa, N. Chniba Boudjada, 798 774 and A. Cheikhrouhou. J. Mater. Sci.: Mater. Electron., 799 775 31:1634-1645, 2020. 776
- [89] D. Mazumdar and I. Das. J. Appl. Phys., 129(6):063901, 777 2021.778
- [90] L. D. Griffith, Y. Mudryk, J. Slaughter, and V. K. 803 779 Pecharsky. Journal of Applied Physics, 123(3):034902, 780 804 2018.805 781
- [91] R. C. Zhang, M. F. Qian, X. X. Zhang, F. X. Qin, L. S. 782 Wei, D. W. Xing, X. P. Cui, J. F. Sun, L. Geng, and 783
- 784 H. X. Peng. J. Magn. Magn. Mater., 428:464–468, 2017. [92]809
- G. J. Liu, J. R. Sun, J. Shen, B. Gao, H. W. Zhang, F. X. 785 Hu, and B. G. Shen. Appl. Phys. Lett., 90(3):032507, 786 2007.787

[93] M. Balli, D. Fruchart, D. Gignoux, and R. Zach. Appl. Phys. Lett., 95(7):072509, 2009.

789

800

801

802

807

- [94] A. Das and A. K. Majumdar. J. Magn. Magn. Mater., 128(1-2):47-57, 1993.
- [95] V. Franco, J. S. Blazquez, and A. Conde. Appl. Phys. Lett, 89:222512, 2006.
- [96] V. Franco and A. Conde. Int. J. Refrig., 33(3):465-473, 2010.
- [97] M. Seeger, S.N. Kaul, K. Kronmüller, and R. Reisser. Phys. Rev. B, 51(18):12585, 1995.
- J. Lyubina, M.D. Kuzmin, K. Nenkov, O. Gutfleisch, [98]M. Richter, D.L. Schlagel, T.A. Lograsso, Jr. Gschneidner, and A. Karl. Phys Rev B, 83(1):012403, 2011.
- [99]S. Chandra, A. Biswas, S. Datta, B. Ghosh, V. Siruguri, AK. Raychaudhuri, MH. Phan, and H. Srikanth. J. Condens. Matter Phys., 24(36):366004, 2012.
- [100]H. Oesterreicher and F.T. Parker. J. Appl. Phys., 55(12):4334-4338, 1984.
- [101] J. S. Blazquez V. Franco and A. Conde. Appl. Phys. 806 Lett, 89:222512, 2006.
- [102]V. Franco, J.S. Blázquez, B. Ingale, and A. Conde. 808 Annu. Rev. Mater. Res, 42, 2012.
- [103] J.J. Ipus, J.S. Blázquez, V. Franco, A. Conde, and L.F. 810 Kiss. Int. J. Appl. Phys, 105(12):123922, 2009. 811
- [104]H. Ben Khlifa, R. M'nassri, W. Cheikhrouhou-Koubaa, 812 813 G. Schmerber, and A. Cheikhrouhou. J. Magn. Magn. Mater., 466:7-16, 2018. 814