

Convenient synthesis of substituted thiazolo [4,5-b] quinolines

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THIAZOLO[4,5-b]QUINOLINES



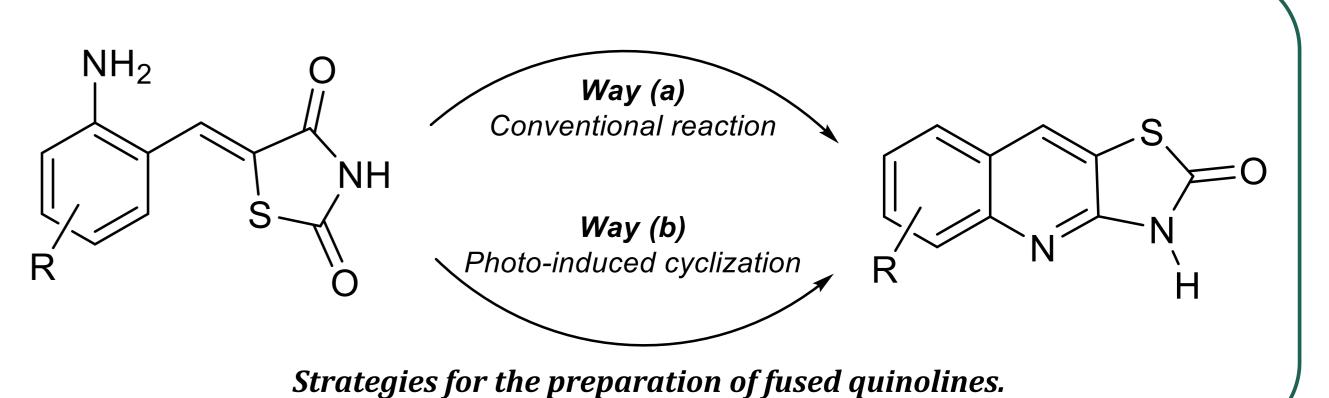
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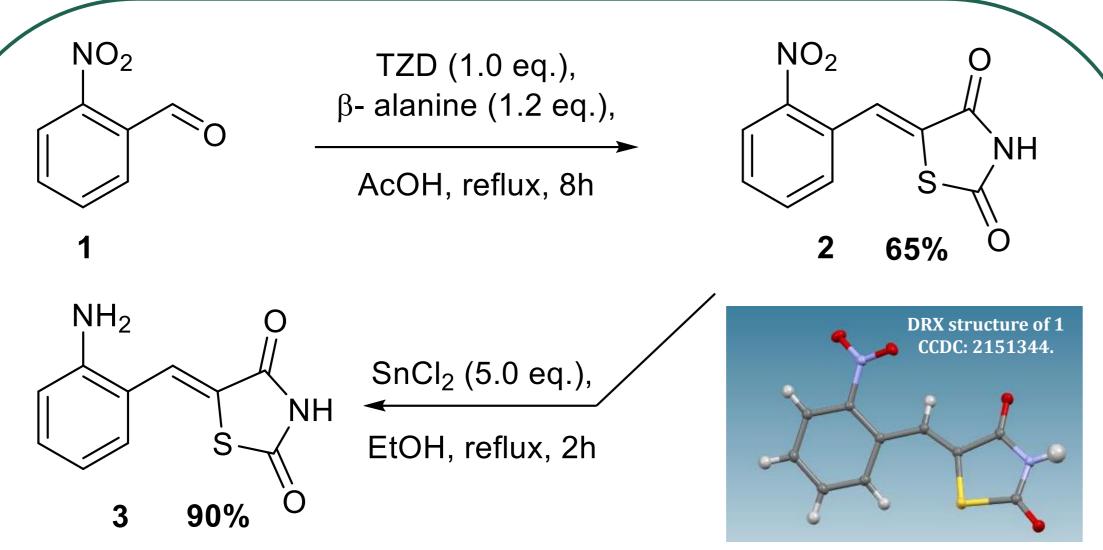
Introduction

Quinoline structure are interesting as they are present in a large variety of biologically active compounds,¹ thus their synthesis have received a large attention. Many methods are employed to prepare these moieties but they often need hard conditions.^{2,3,4} In this work, we focused our attention on a short and efficient synthesis of fused quinolines. We envisioned two strategies:

- * one with classical conditions,
- * a more original one, based on photo-induction.



Preparation of o-aminobenzylidene scaffold

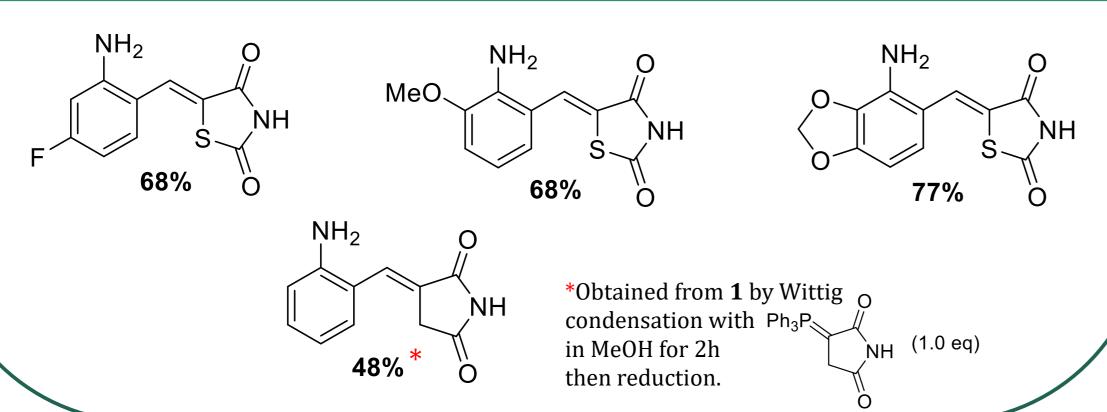


- Easy and fast synthesis.
- Good overall yield (59%).

 NH_2

Only pure isomer (Z) is obtained.

Extension of the sequence to various aminobenzylidenes



Classical way

The preparation of quinoline 4 depends on the isomerization of the double bond.

NH₂ O Reagents, Solvent,
$$\begin{array}{c} NH_2 \\ \hline NH \\ \hline \end{array}$$
 $\begin{array}{c} NH_2 \\ \hline \end{array}$ $\begin{array}{c} NH_2 \\$

Conditions for preparation of quinoline 4.

Starting material	Solvent	T (°C)	time	Reagents	Results
NH ₂ O NH S NH (Z)-3	Acetone	r.t.	2 h to 48 h	none	Traces of 4
	Acetone	r.t.	2 months	none	80% of 4
	Acetone	reflux	2 h to 24 h	none	degradation
	Toluene	reflux	3 h to 24 h	APTS (0.1 eq)	s.m. ^c
	EtOH	180°Ca	3 h	none	degradation
	EtOH	150°Ca	3 h	none	degradation
	EtOH	120°Ca	3 h	none	degradation
NO ₂ O NH S O	EtOH	reflux ^b	8 h	SnCl ₂ (5.0 eq)	3 (90%), 4 (10%) ^d
	EtOH	reflux	8 h	SnCl ₂ (20.0 eq)	3 (90%), 4 (10%) ^d
	EtOH	120°Ca	3 h	SnCl ₂ (5.0 eq)	degradation

^a Thermal reactions were run in sealed vessel. ^b Under microwaves. ^c s.m. Starting material. ^d Yield estimated by ¹H NMR.

Acid activation X High pressure X High temperature X Microwaves X

Long time agitation <



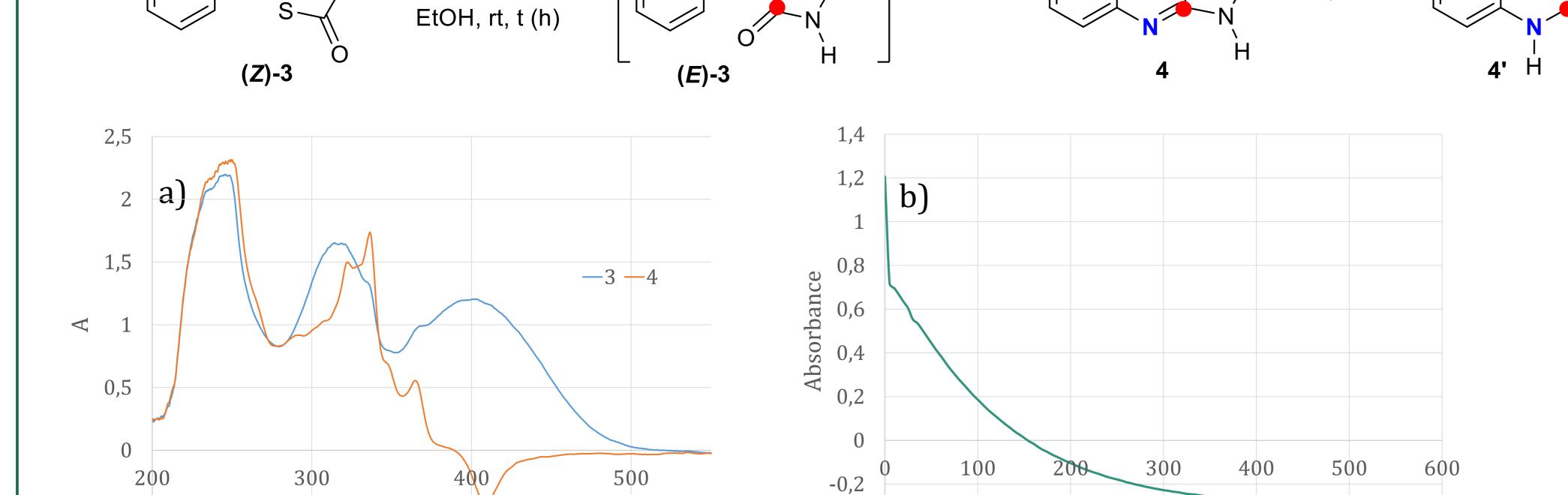
Light

Photochemistry way

hv (405 nm),

Wavelength (nm)

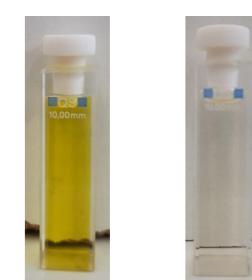
The nuclophilic attack of the amine moiety on a carbonyl group depends on the double bond configuration: an inversion is necessary to obtain the fused quinoline. As a consequence of the latter results (acetone, r.t., 2 months), we investigated the photo-induced isomerization of the double bond with light.^{5,6,7}



 NH_2

a) UV-visible spectra of amine 3 and quinoline 4 in EtOH; b) kinetic study of amine 3 consumption under irradiation at 405 nm.

- → Product 4 was prepared after a 48h irradiation at 405 nm in an excellent 84% yield (overall yield 49 %).
- > Extension of the photoinduced cyclization sequence afforded various fused azaheterocycles in 48 to 77% yields.



DRX structure of quinoline 4' Solutions of amine 3 and CCDC: 2151345. quinoline 4 in EtOH

Examplification

Various fused azaheterocycles were obtained by photoinduction.

68%
$$\lambda = 405 \text{ nm}$$
 $t = 48 \text{ h}$
Overall yield : 33%*

77% $\lambda = 430 \text{ nm}$
 $t = 48 \text{ h}$

Overall yield: 18%*

 $\lambda = 405 \text{ nm}$ t = 48 hOverall yield: 39%*

 $\lambda = 375 \text{ nm}$ t = 48 h

*Overall yield calculated for 3 steps.

Overall yield: 30%*

Conclusion

Time (min)

We succedeed in preparing various fused quinolines with a 3-steps synthesis in 18 to 49 % overall yields. The first two steps were fast and effective as we obtained amines 3 as pure products in very good yields (23 to 62 %). The third step consisted in a photo-induced isomerization followed by a subsequent cyclization. This method is convenient, clean and really effective compared to classical ways.⁸

References

-0,5