



HAL
open science

Molecular-rotation-induced splitting of the binary ridge in the velocity map of sub-eV H⁺ ions ejected from H₂ molecules by ion impact

Z. Juhász, S T S Kovács, V. Vizcaïno, P. Herczku, Sándor Demes, R. Rácz, B. Sulik, S. Biri, N. Sens, D. V. Mifsud, et al.

► To cite this version:

Z. Juhász, S T S Kovács, V. Vizcaïno, P. Herczku, Sándor Demes, et al.. Molecular-rotation-induced splitting of the binary ridge in the velocity map of sub-eV H⁺ ions ejected from H₂ molecules by ion impact. *Physical Review A: Atomic, molecular, and optical physics* [1990-2015], 2023, 107, pp.L010801. 10.1103/PhysRevA.107.L010801 . hal-03953754

HAL Id: hal-03953754

<https://hal.science/hal-03953754v1>

Submitted on 24 Jan 2023

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution - NonCommercial 4.0 International License

Molecular-rotation-induced splitting of the binary ridge in the velocity map of sub-eV H⁺ ions ejected from H₂ molecules by ion impact

Z. Juhász^{1,†}, S.T.S. Kovács¹, V. Vizcaïno², P. Herczku¹, S. Demes¹, R. Rácz¹, B. Sulik¹, S. Biri¹, N. Sens², D.V. Mifsud^{3,1}, and J.-Y. Chesnel²

1 *Institute for Nuclear Research (Atomki), Debrecen H-4026, Hungary*

2 *Centre de Recherche sur les Ions, les Matériaux, et la Photonique (CIMAP), UMR 6252 CEA-CNRS-ENSICAEN-UNICAEN, Normandie Université, 14000 Caen, France*

3 *Centre for Astrophysics and Planetary Science, School of Physical Sciences, University of Kent, Canterbury CT2 7NH, United Kingdom*

† Corresponding author: zjuhasz@atomki.hu

Abstract

In studies of ion-induced molecular fragmentation, the challenging measurement of the velocity distribution of fragments emitted below 1-eV kinetic energy is rarely achieved, although most fragments have an energy below this value. Here, we study H⁺ fragment emission in collisions of 10-keV O⁺ ions with H₂ molecules using a field-free time-of-flight technique developed specifically to detect sub-eV fragments. We find that, in the velocity map, the binary ridge due to direct H⁺ knockout is split into two parts arising from the rotational motion of the H₂ molecule, and that this split scales with rotational velocity. The velocity distribution of the nuclei in the original molecule is determined and the thermally populated $J = 1$ rotational level is found to be the dominant contributor, although asymmetry in the split indicates projectile-induced rotational transitions between M sub-levels. These rotation effects influence fragment emission probabilities, thus carrying important consequences for the radiation-induced hydrogen loss and H₂ dissociation in the atmospheres or exospheres of planets and moons.

Letter

The simplest case of ion-induced molecular fragmentation is that of H₂. Even in this case, however, diverse processes may occur [1,2]. Following double electron removal, the residual molecule undergoes Coulomb explosion which results in a quasi-isotropic emission of H⁺ at emission energies of ~9.5 eV [1,2]. More moderate excitation or single ionisation leads to even more intense emission in the sub-eV range [2,3]. Despite its astrophysical relevance as a contributor to exospheric mass loss, this low-energy regime is scarcely studied. Complete energy and angular distributions of emitted fragments are challenging to measure, although their characterisation is of great importance since the emission may be anisotropic due to effects of charged particles on fragmentation dynamics. Furthermore, the energy spacing of the first two rotational levels of H₂ is ~15 meV, which is large compared to that of other diatomic

molecules. As such, the kinetic energy stored in the rotational degree of freedom of H₂ may also influence the fragment distribution in the sub-eV range.

The rotational excitation of molecules colliding with atoms and other molecules was previously studied using the velocity map imaging (VMI) technique for collision velocities of a few hundred m/s [4-7]. In the resultant two-dimensional velocity maps, the scattered neutral partners (after having been state-selectively ionised by a laser pulse and accelerated by the VMI electric field) show a structured distribution due to the quantised energy defects arising from transitions between rotational energy levels.

Sub-eV fragments from N₂ generated by electron impact were studied using a time-delayed spectroscopic technique [8]. In another study, fragments down to 0.5 eV emission energy near the 90° emission angle were investigated in the dissociative charge exchange and ionization of O₂ by H⁺ and O⁺ ions of 10 to 100 keV kinetic energy using a time-of-flight (TOF) technique [9]. Such work is applicable to astrophysics: the escape velocity of the Jovian moon Europa corresponds to ~0.3 eV fragment energy, meaning that dissociation products may leave the lunar exosphere only if their energy exceeds this value due to charge exchange collisions with magnetospheric ions. For H₂, thermal escape dominates, but charge exchange and knock-out processes are also relevant [10]. However, cross sections in the sub-eV emission energy range are often lacking for exospheric models.

The near-complete energy and angular distributions of H⁺ fragments were studied using electrostatic spectrometry [1,11,12]. Unfortunately, this technique has a practical low-energy limit of 1-2 eV. Several studies have also considered laser-induced fragmentation of H₂ and acquired complete velocity maps using TOF techniques; however, these studies focused on the Coulomb explosion process in the range of a few eV [13,14]. Other groups investigated the angular distribution of the scattered projectile in the H⁺-induced fragmentation of H₂ to demonstrate the projectile's coherence effects [2,15]. Earlier, detailed studies were performed for collisions between O⁺ and H₂, but at energies of several tens of eV and focusing on the reactive (hydrogen transfer) channel and velocity distributions of the OH⁺ product [16,17].

In this letter, an experimental observation of the effects of the rotational motion of H₂ on the velocity map of positively charged fragments is presented. A field-free TOF technique was used and allowed us to accurately measure the fragment velocity distribution down to 0.1 eV emission energy, thus providing cross sections for collision-induced H⁺ emission from H₂ in the sub-eV energy range. The rotational effects, which cause a shift in the velocity map, are observed in the low-energy part of the binary ridge due to the direct knock-out of a H⁺ fragment by the projectile ion. We show that the present findings are of both fundamental and applied interest, as they give an insight into the nuclear wavefunction of the rotating molecule and may also have astrophysical applications.

The velocity map of the emitted H⁺ fragments arising from the impact of H₂ with O⁺ ions has been studied over a 0.1-20 eV energy range. The reason behind selecting O⁺ ions as projectiles is that they are a major constituent of the Jovian magnetosphere and can be easily distinguished from the light H⁺ fragments. This letter focuses on emission energies <1 eV to demonstrate the effects of molecular rotation. Projectile ions with energies of 10 keV, which correspond to typical stellar wind and Jovian magnetospheric ion velocities [9,18], were provided by an electron cyclotron resonance ion source at Atomki [19]. The ion beam was chopped using the clock signal of the TOF apparatus, and 250 ns ion bunches with 5 kHz repetition frequency

were delivered into a vacuum chamber where they crossed with an effusive gas jet of H₂ supplied by a nozzle cooled to 125 K. The TOF of the emitted H⁺ was measured by a position-sensitive microchannel plate detector, which could be set to different observation angles with respect to the beam.

The position and TOF of the detected fragments were used to calculate their velocity vector. The gas jet velocity is estimated to be 1380 m/s, which is comparable to that of the slowest detected fragments and was therefore subtracted from the calculated fragment velocity vectors. The transversal velocity spread of the jet, estimated to be 140 m/s, induces only a small blurring of the measured velocity distributions. These velocity estimations were based on previous simulations of gas expansion through a nozzle [20]. To ensure that no magnetic or electric fields could change the velocity of the emitted charged fragments, a μ-metal double shield reduced the magnetic field to ~1 milligauss at the centre of the chamber, and metals with similar contact potentials were used throughout to avoid static electric fields. Moreover, the detector (set to high surface potential) was shielded electrically by a triple metallic grid arrangement. To minimise the detection of fragmentation products from background gases, narrow metallic tubes were placed around the ion beam with a 3 mm opening at the collision area. The experimental set-up is depicted in Fig. 1, and will be expounded upon in Ref. [21].

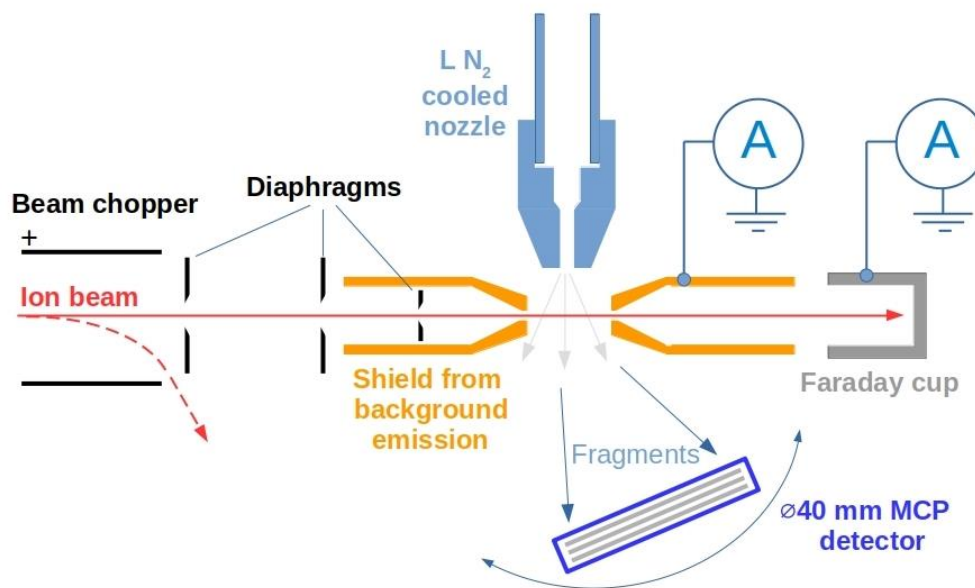
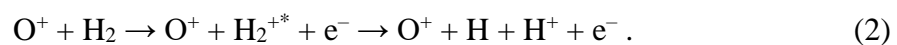
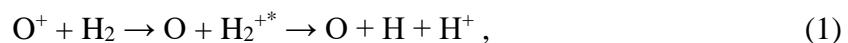


Fig. 1: The field-free TOF experimental set-up.

In this study, H⁺ fragments may arise from dissociative charge transfer (which is dominant at the present projectile velocity), or dissociative ionisation after the formation of H₂⁺:



H⁺ emission from the 1σ_g ground state occurs mostly at emission energies below 0.1 eV, while emission from the 2sσ_g and 2pσ_u excited states of H₂⁺* occurs over the 4-12 eV range [2,3]. The corresponding cross sections integrated over the solid angle and energies as obtained from

the velocity maps produced in this work are respectively 7 and 1×10^{-16} cm² over these energy ranges, indicating that excitation is not a strong channel in this collision system. However, we note that the cross sections are subject to a systematic uncertainty of at least 15% due to limitations in the accuracy of the pressure gauge. In particular, the cross section for the <0.1 eV energy range should be regarded with caution, since this range is below the limit of accurate measurement. In this range, the emission appears as being anisotropic (see Fig. 2(a)), while one would expect an isotropic distribution for dissociative charge transfer processes in large impact parameter collisions.

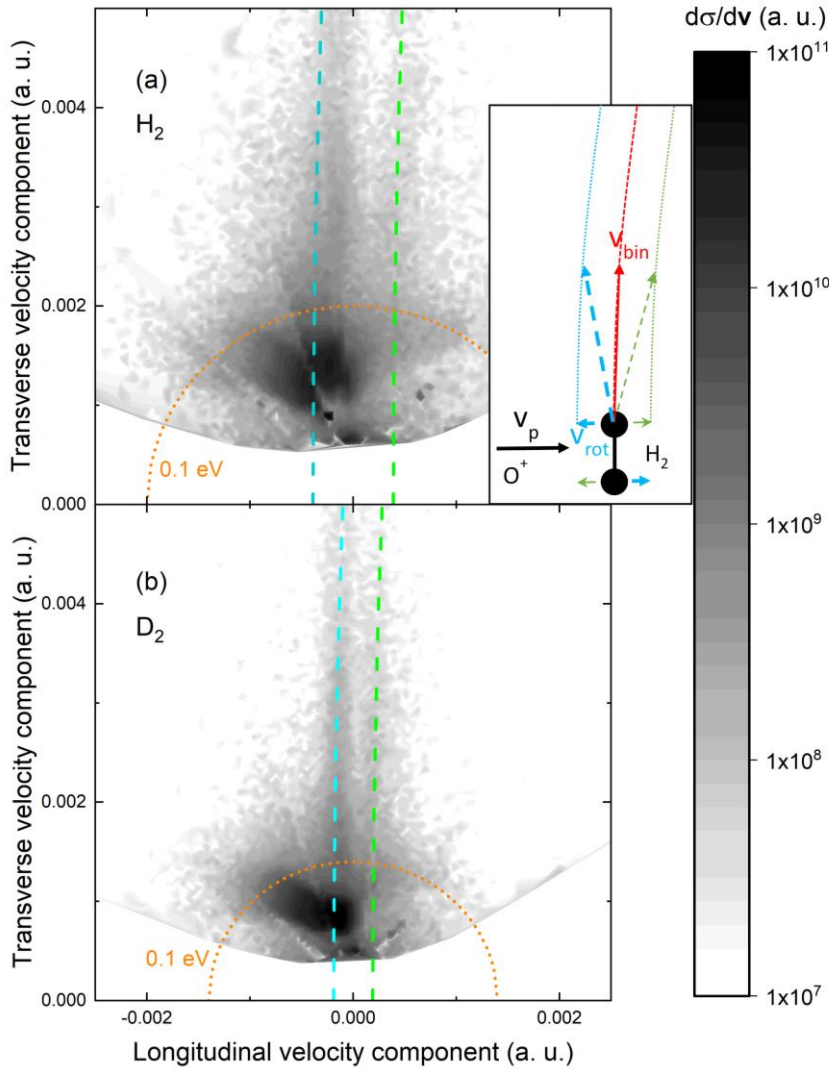


Fig. 2: Velocity distributions of (a) H⁺ and (b) D⁺ ions respectively arising from H₂ or D₂ molecules in collision with 10 keV O⁺ ions. The atomic velocity unit is used. The cross section $\frac{d\sigma}{dv}$ for H⁺ (or D⁺) emission (differential in emission velocity \mathbf{v}) is also given in atomic units. The ridges for the direct H⁺ (or D⁺) binary knock-out process are marked by segmented lines. The nozzle was placed 5 mm away from the ion beam axis. Velocities corresponding to a kinetic energy of 0.1 eV are marked by dotted lines for reference. The intense spots below this line at a backward emission angle of $\sim 135^\circ$ are mostly due to dissociative charge transfer processes (Eq. 1). The expected isotropic distribution at such low emission velocities is distorted by the spurious field of surface charges. Inset: the expected binary ridge for a target at rest (red, dashed line) and the shifted binary ridges (blue and green dotted lines) caused by molecular rotation. The $\frac{d\sigma}{dv}$ values over the binary ridges are given in tabular form in the Supplemental Material Tables S1 and S2[22].

Actually, the low-energy fragment distribution (<0.1 eV) is disturbed by the field of unavoidable surface charges. Moreover, the possible contribution of the H_2^+ molecular ion should be considered: some of the ground-state H_2^+ ions that are produced may remain intact after the collisions, the process of which does not involve kinetic energy release, thus rendering the expected emission energy close to zero. However, by repeating the experiment and biasing one of the grids placed in front of the detector, the yield of H_2^+ was found to be negligible compared to that of H^+ . The obtained H^+ fragment velocity distribution is shown in the low-velocity range (<0.5 eV emission energy) in Fig. 2(a).

In the region above the 0.1 eV energy level indicated in Fig. 2, emitted H^+ fragments exhibit an anisotropic distribution as a result of their direct knock-out by the projectile. The corresponding tabular-form cross-sectional data are given in the Supplemental Material Table S1 [22]. H^+ fragments produced in this way from a target at rest are expected to be distributed along a binary ridge that appears as an arc in the two-dimensional velocity map, starting at the origin and spreading along the transversal direction. This arc has a radius of:

$$v_r = v_p m_O / (m_H + m_O) = 0.139 \text{ a.u.} \quad (3)$$

where $v_p = 0.148$ a.u. is the projectile velocity and m_H and m_O are the masses of the target H nucleus and O^+ projectile, respectively. This radius is large when compared to the dimensions of Fig. 2. In the velocity map, however, two arcs are visible which are shifted from the origin in the longitudinal direction.

The inset of Fig. 2 provides a graphical explanation of the shifted binary ridges. In collisions involving relatively small impact parameters ($b < 1.4$ a.u.), in which only one target atom is approached, that atom and the projectile jointly determine the collision dynamics. The velocity \mathbf{v}_{rot} of the atom in the rotating molecule adds to the velocity \mathbf{v}_{bin} gained by binary momentum transfer, the latter being almost transversal. Depending on its sign, the longitudinal component of \mathbf{v}_{rot} shifts the binary ridge either backwards or forwards along the projectile axis. Hence, this component splits the expected ridge for a non-rotating target into two ridges. Conversely, the transversal component of \mathbf{v}_{rot} has a negligible influence on the ridge arc, which is nearly a vertical line.

The initial population of the rotational levels of the H_2 gas is determined by the room-temperature Boltzmann distribution. The cooling of the nozzle and the rapid jet expansion cannot change the population [20], which requires several hours to equilibrate. Calculations show that the $J = 1$ rotational level is the most important (66%), and its contribution is thus expected to dominate the H^+ fragment velocity distribution. At the $J = 1$ level, the average H atom velocity \mathbf{v}_{rot} is 3.9×10^{-4} a.u., based on the classical formula for angular momentum

$$J = m_H r_0 v_{\text{rot}} \quad (4)$$

where $r_0 = 1.4$ a.u. is the internuclear distance in H_2 at equilibrium. Nearly the same shift sizes are observed in both the positive and negative directions, which may stem from the two rotational directions or from the emission of two counter-moving atoms at $J = 1$. To confirm the rotational origin of the binary ridge split, the experiment was repeated using a D_2 target (Fig. 2(b)). Since D has twice the mass of H, the rotational velocity and, by extension, the induced shifts, are expected to be half of those observed for H_2 assuming that the contribution of the $J = 1$ level remains the most dominant. The velocity map depicted in Fig. 2b confirms our expectations (tabular-form data are given in the Supplemental Material 1 Table S2 [22]).

The fragment distribution along the longitudinal velocity axis provides an insight into the rotational wavefunction of the original target molecule in the longitudinal velocity space. In a similar collision process between H_2^+ and He, it was shown that the spatial imaging of the vibrational wavefunction of H_2^+ is possible by Coulomb explosion imaging, where the only limiting factor stems from the Heisenberg uncertainty relations [23]. Although pump-probe laser experiments have been extensively exploited to study rotational wave packets providing spatial information on the molecular axis distribution [24], the imaging of low- J rotational eigenstates remains scarce [25]. In our case, the insight into the velocity space is possible because the original target atom velocity can be traced back from the two-body kinematics of the binary collision, since the effect of the third atomic center is negligible. In Fig. 3, a double peaked structure is observed in the longitudinal velocity distributions for both H_2 and D_2 and stems from the ridge split caused by the rotational motion of the molecules. The distance between peaks for D_2 (Fig. 3(b)) is about half of that for H_2 (Fig. 3(a)) due to the factor-of-two difference between the rotational velocities of these targets.

These curves are compared to the velocity distribution of the nuclei in the molecules calculated using the approximate nuclear wavefunction of a rotating molecule in its vibrational ground state:

$$\Psi \approx \exp(-m \omega (r - r_0)^2 / 2) Y_{JM}(\theta, \varphi) \quad (5)$$

wherein polar coordinates (r, θ, φ) are defined with a polar axis perpendicular to the scattering plane; Y_{JM} is the spherical harmonic function for the $J = 1$ rotational and $M = \pm 1$ angular momentum projection quantum numbers, r_0 is the internuclear distance in H_2 (or D_2) at equilibrium, m is the atomic mass of H (or D), and ω is the molecular vibration angular frequency.

The blue dotted lines in Fig. 3 correspond to the application of a Fourier transformation in the longitudinal direction at each axis parallel to the initial projectile velocity so as to obtain the longitudinal velocity distributions with results being coherently added. The probability distribution in the velocity space is derived from the square of the absolute value of the results. The resultant peak positions match those obtained when using the classical formula (Eq. 4) for angular momentum (3.9×10^{-4} and 1.9×10^{-4} a.u. for H_2 and D_2 , respectively), but the experimental peak positions are marginally closer to the origin.

When incoherent superposition was applied (red dashed lines in Fig. 3), the resultant peak positions matched the experimental ones. However, the dip in the origin was not reproduced and, thus, coherent superposition was a better approach in this respect. These observations suggest that the wavefunction imaging by the projectile appears to be in line with a partial coherent superposition. This is also expected from the coherence properties of the ion beam given the slit geometry [2,15]: the transversal coherence length of the projectile at the collision centre is in the range of 1 a.u., which is comparable to the size of the nuclear wavefunction.

It should be noted that the absolute intensities of the binary ridges are significantly smaller for D_2 than for H_2 . The total cross sections for the binary ridges in the energy range 0.1-4 eV are 2.5 and $4.9 \times 10^{-16} \text{ cm}^2$, respectively. Similar isotopic effects were observed in Refs. [26] and [27]. For our total cross section for dissociation *via* excited molecular ions over the 4-12 eV range, the value of $0.45 \times 10^{-16} \text{ cm}^2$ for D_2 may be compared to that of $1 \times 10^{-16} \text{ cm}^2$ for H_2 . Kusakabe et al. [26] attributed this difference to the more compact vibrational states of the D_2

isotopologue and the corresponding smaller Franck-Condon factors. Interestingly, in the low-energy region (<0.1 eV), the opposite trend is observed in our results: 20×10^{-16} cm² for the D₂ isotopologue compared to 7×10^{-16} cm² for H₂. However, these values should be regarded with caution due to uncertainties induced by possible instrumental surface charge effects.

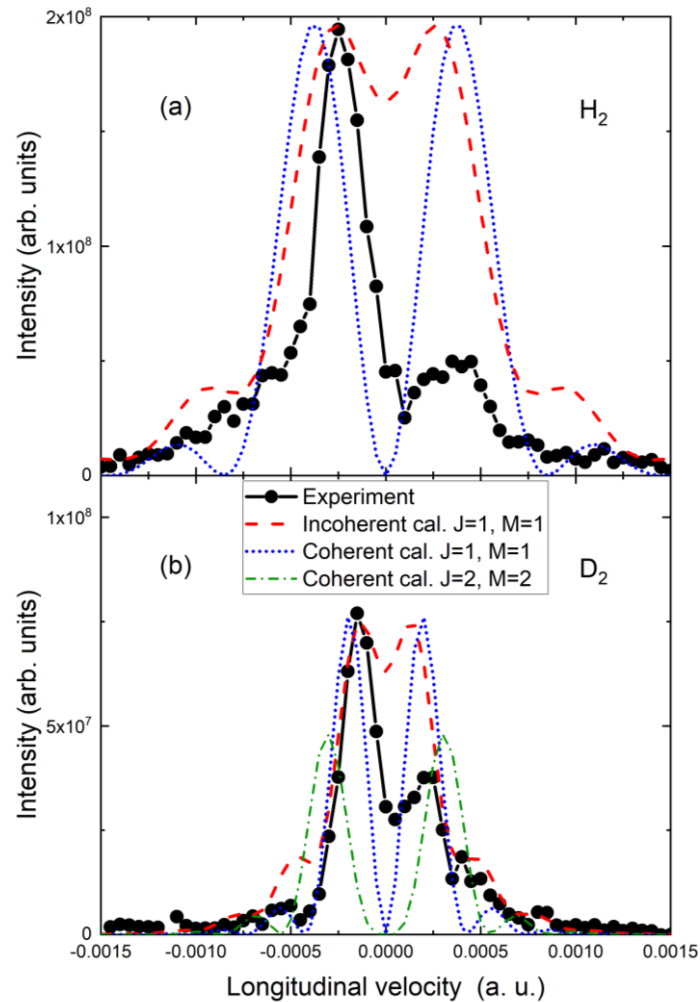


Fig. 3: Measured nuclear velocity distributions in the longitudinal direction (joined circles) with (a) H₂ and (b) D₂ targets. The results are averaged over 0.003-0.004 a.u. transversal velocities for good statistics and \mathbf{v}_{bin} was subtracted. The calculated velocity distributions from the nuclear wavefunction for the rotational quantum numbers $J = 1, M = \pm 1$ are also shown when coherent superposition (blue dotted line) and incoherent superposition (red dashed line) were applied in the transversal direction. These theoretical curves are normalised to the experiments at the maximum. In panel (b), the result for the coherent calculation for $J = 2, M = \pm 2$ is shown by the green dash-dotted line with the same normalisation as that of the coherent calculation for $J = 1, M = \pm 1$, thus permitting the comparison of their intensities

A striking difference between the experiment and theory is the strong experimental asymmetry compared to the theoretical symmetry of the distributions. This may be explained by collision-induced rotational excitation of the molecules. The attractive force between the incoming O⁺ ion and the approached H atom exerts a torque on the molecule causing the H atom to more likely move backwards rather than forwards. This is reflected in the more intense peak on the

negative side of Fig. 3(a). For D_2 , the asymmetry is smaller (Fig. 3(b)) as a result of its larger moment of inertia.

For H_2 at 300 K, the population of the $J = 0$ level (13%) is small compared to the $J = 1$ level (66%). This explains why a significant contribution from $J = 0$, which would lead to a peak at zero-velocity in Fig. 3(a), is not observed. However, precollisional rotational excitations induced by the torque exerted by the projectile on the molecule can affect the population of the observed rotational levels. Although transitions between odd and even rotational levels in H_2 are nuclear spin forbidden, transitions between states characterised by different $M = -1, 0, 1$ angular momentum projection quantum numbers are likely given that there is no energy change involved during transition. The excitations result in a non-equilibrium M -distribution, which is dependent on the impact parameter of the collision with respect to the molecular centre. This is the quantum mechanical analogue of the explanation of the observed asymmetric velocity distributions in Fig. 3, since the resulting M -distribution is such that the projectile will collide with the counter-moving atom in the molecule with a higher probability.

Since the energy spacing of the D_2 rotational levels is smaller than that of H_2 , the equilibrium population of the first few rotational levels of D_2 at room temperature is more evenly distributed. From $J = 0$ to $J = 4$, they are respectively 19%, 21%, 39%, 11%, and 9%, according to Boltzmann population calculations. It is perhaps intuitive to expect that the contribution from the $J = 2$ level is dominant in the longitudinal velocity distribution in Fig. 3(b), but our results demonstrate the dominance of the $J = 1$ level, as in the case of H_2 . This may be explained by comparing the intensities resulting from the coherent calculation for the different rotational states shown in Fig. 3(b): higher rotational levels having more nodes possess more evenly balanced positive and negative parts, which results in lower relative contributions due to destructive interference. As shown in Fig. 3(b), the resultant intensity for the $J = 1, M = \pm 1$ rotational state is twice that of the $J = 2, M = \pm 2$ state. Conversely, when the wavefunction is anti-symmetric to the polar axis, as in the case of the $J = 2, M = \pm 1$ and the $J = 1, M = 0$ states, interference is fully destructive and results in no intensity being observed from these states. Despite this, however, the contribution of high- J states is indeed noticeable for D_2 (Fig. 3(b)) from the larger peak width compared to that obtained by coherent calculation. This apparent broadening is the result of the merging of peaks for multiple- J values.

Higher charge state projectiles presumably initiate more efficient rotational excitations since the interaction with the molecule is stronger. To demonstrate this, experiments were also performed with 20 keV O^{2+} projectiles; the results of which are described in Ref. [21]. Briefly, the main finding is that the appearance of the higher-level rotational excitations suggests that transitions between different J states are open for the doubly-charged projectile, in contrast to the singly-charged projectile. The observed $J = 3$ state is mainly due to the projectile-induced transition from the thermally populated $J = 1$ state, while the $J = 2$ state is sourced from the $J = 0 \rightarrow 2$ transition.

In conclusion, we have measured the velocity distribution of sub-eV fragments from H_2 (D_2) arising from collisions with 10 keV O^+ ions. The binary ridge of the knocked-out H^+ (D^+) ions is found to be split into two parts. By comparing the isotopologues, we have shown that this phenomenon results from molecular rotational motion, and the split binary ridge provides an insight into the nuclear wavefunction of the rotating molecule in the momentum (velocity) space. We have observed strong isotopic effects in terms of cross sections. Asymmetry in the

left- and right-shifted binary ridges indicates rotational transitions prior to the dissociation of the molecule. Using singly-charged projectile ions, these transitions are between states associated with different M angular momentum projection quantum numbers, while when using doubly-charged projectiles, transitions between different J rotational states may occur.

The molecular rotation effects described in this letter may greatly influence fragment emission processes, as evidenced by our results. This may have important consequences in astrophysics, where rates of H_2 destruction by stellar winds or magnetospheric ions may significantly differ in the atmospheres of warm planets compared to the rotationally relaxed cooler atmospheres (or exospheres) of comets and moons. This may be particularly important on Earth where reactive collisions between O^+ ions and H_2 play an important role in atmospheric chemistry, since small changes in the concentration of H_2 may result in large perturbations in the ionosphere [28]. Small rotational effects have already been documented in the production of OH^+ [29]. The present study indicates that the influence of rotational motion on the fragmentation process is important to consider.

Acknowledgements

This work was supported by the Hungarian Scientific Research Fund of the National Research, Development and Innovation Office financed under the K18 funding scheme through Project No. K128621, and by the Centre National de la Recherche Scientifique (CNRS, France) through International Emerging Action (IEA-PICS-CNRS No. 7739 & Project NKM-115/2017). This article is based upon work from COST Action CA18212–Molecular Dynamics in the GAS phase (MD-GAS), supported by COST (European Cooperation in Science and Technology). Z.J. is grateful for the support of the Hungarian Academy of Sciences through the János Bolyai Research Scholarship. D.V.M. is the grateful recipient of a University of Kent Vice-Chancellor’s Research Scholarship.

References

- [1] P. Sobocinski, J. Rangama, G. Laurent, L. Adoui, A. Cassimi, J.-Y. Chesnel, A. Dubois, D. Hennecart, X. Husson, and F. Frémont, *J. Phys. B: At. Mol. Opt. Phys.* **35**, 1353 (2002).
- [2] B. R. Lamichhane, A. Hasan, T. Arthanayaka, M. Dhital, K. Koirala, T. Voss, R. A. Lomsadze, and M. Schulz, *Phys. Rev. A* **96**, 042708 (2017).
- [3] M. Waitz, R. Y. Bello, D. Metz, J. Lower, F. Trinter, C. Schober, M. Keiling, U. Lenz, M. Pitzer, K. Mertens, M. Martins, J. Viehhaus, S. Klumpp, T. Weber, L. Ph. H. Schmidt, J. B. Williams, M. S. Schöffler, V. V. Serov, A. S. Kheifets, L. Argenti et al., *Nat. Commun.* **8**, 2266 (2017).
- [4] G. Sarma, S. Marinakis, J. J. ter Meulen, D. H. Parker, and K. G. McKendrick, *Nat. Chem.* **4**, 985 (2012).
- [5] J. Onvlee, S. N. Vogels, A. von Zastrow, D. H. Parker, and S. Y. T. van de Meerakker, *Phys. Chem. Chem. Phys.* **16**, 15768 (2014).
- [6] Z. Gao, T. Karman, S. N. Vogels, M. Besemer, A. van der Avoird, G. C. Groenenboom, and S. Y. T. van de Meerakker, *Nat. Chem.* **10**, 469 (2018).
- [7] Z. Gao, J. Loreau, A. van der Avoird, and S. Y. T. van de Meerakker, *Phys. Chem. Chem. Phys.* **21**, 14033 (2019).
- [8] N. Ferreira, L. Sigaud, V. L. B. de Jesus, A. B. Rocha, L. H. Coutinho, and E. C. Montenegro, *Phys. Rev. A* **86**, 012702 (2012).
- [9] H. Luna, C. McGrath, M. B. Shah, R. E. Johnson, M. Liu, C. J. Latimer, and E. C. Montenegro, *Astrophys. J.* **628**, 1086 (2005).
- [10] R. E. Johnson, M. H. Burger, T. A. Cassidy, F. Leblanc, M. Marconi, W. H. Smyth, and R. Dotson, Composition and detection of Europa’s sputter-induced atmosphere, in *Europa*, edited by R. T. Pappalardo, W. B. McKinnon, and K. K. Khurana (University of Arizona Press, Tucson, 2009), pp. 507–528.
- [11] F. Frémont, C. Bedouet, M. Tarisien, L. Adoui, A. Cassimi, A. Dubois, J.-Y. Chesnel, and X. Husson, *J. Phys. B: At. Mol. Opt. Phys.* **33**, L249 (2000).
- [12] Z. Juhász, B. Sulik, E. Lattouf, E. Bene, B. A. Huber, P. Herczku, S. T. S. Kovács, A. Méry, J.-C. Pouilly, J. Rangama, J. A. Tanis, V. Vizzaino, and J.-Y. Chesnel, *Phys. Rev. A* **100**, 032713 (2019).

- [13] A. S. Alnaser, X. M. Tong, T. Osipov, S. Voss, C. M. Maharjan, P. Ranitovic, B. Ulrich, B. Shan, Z. Chang, C. D. Lin, and C. L. Cocke, *Phys. Rev. Lett.* **93**, 183202 (2004).
- [14] B. Manschwetus, T. Nubbemeyer, K. Gorling, G. Steinmeyer, U. Eichmann, H. Rottke, and W. Sandner, *Phys. Rev. Lett.* **102**, 113002 (2009).
- [15] B. R. Lamichhane, T. Arthanayaka, J. Remolina, A. Hasan, M. F. Ciappina, F. Navarrete, R. O. Barrachina, R. A. Lomsadze, and M. Schulz, *Phys. Rev. Lett.* **119**, 083402 (2017).
- [16] K. T. Gillen, B. H. Mahan, and J. S. Winn, *J. Chem. Phys.* **58**, 5373 (1973).
- [17] K. T. Gillen, B. H. Mahan, and J. S. Winn, *J. Chem. Phys.* **59**, 6380 (1973).
- [18] S. J. Bolton, F. Bagenal, M. Blanc, T. Cassidy, E. Chané, C. Jackman, X. Jia, A. Kotova, N. Krupp, A. Milillo, C. Plainaki, H. T. Smith, and H. Waite, *Space Sci. Rev.* **192**, 209 (2015).
- [19] S. Biri, I. K. Vajda, P. Hajdu, R. Rácz, A. Csík, Z. Kormány, Z. Perduk, F. Kocsis, and I. Rajta, *Eur. Phys. J. Plus* **136**, 247 (2021).
- [20] A. Naß and E. Steffens, *Nucl. Instr. Meth. Phys. Res. A* **598**, 653 (2009).
- [21] Z. Juhász, S. T. S. Kovács, V. Vizcaíno, P. Herczku, S. Demes, R. Rácz, B. Sulik, S. Biri, N. Sens, D.V. Mifsud, and J.-Y. Chesnel (unpublished).
- [22] See Supplemental Material at <http://link.aps.org/supplemental/10.1103/PhysRevA.107.L010801> for the measured double-differential $H^+(D^+)$ fragment emission cross section.
- [23] L. Ph. H. Schmidt, T. Jahnke, A. Czasch, M. Schöffler, H. Schmidt-Böcking, and R. Dörner, *Phys. Rev. Lett.* **108**, 073202 (2012).
- [24] K. Mizuse, N. Chizuwa, D. Ikeda, T. Imajo, and Y. Ohshima, *Phys. Chem. Chem. Phys.* **20**, 3303 (2018).
- [25] Y. Mi, P. Peng, N. Camus, X. Sun, P. Fross, D. Martinez, Z. Dube, P. B. Corkum, D. M. Villeneuve, A. Staudte, R. Moshhammer, and T. Pfeifer, *Phys. Rev. Lett.* **125**, 173201 (2020).
- [26] T. Kusakabe, L. Pichl, R. J. Buenker, M. Kimura, and H. Tawara, *Phys. Rev. A* **70**, 052710 (2004).
- [27] S. Rai, K. I. Bijlsma, I. Rabadán, L. Méndez, P. A. J. Wolff, M. Salverda, O. O. Versolato, and R. Hoekstra, *Phys. Rev. A* **106**, 012804 (2022).
- [28] M. Yasar and M. Canyılmaz, *Therm. Sci.* **22**, S47 (2018).
- [29] X. Li, Y.-L. Huang, G. D. Flesch, and C. Y. Ng, *J. Chem. Phys.* **106**, 564 (1997).