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Encapsulation of cannabidiol in oil-in-water nanoemulsions and
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study
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Abstract

32 Hypothesis

The oral administration of lipophilic bioactives using conventional nanoemulsions presents certain limitations. Hence, novel hybrid vehicles could be valuable alternatives providing diverse routes of administration. To determine the suitability of a nanocarrier the elucidation of its structure is essential since it affects the release of the encapsulated compound. Specifically, concerning the transdermal delivery the evaluation of the penetration of the bioactive through the skin is also crucial.

Experiments

Oil-in-water nanoemulsions (NE) and the respective nanoemulsion-filled hydrogels (NE/HG) were formulated and structurally characterized as carriers of cannabidiol (CBD). The NE stability and size were studied and the structure of both systems was visualized with microscopic techniques to determine possible structural changes in the NE upon their incorporation into the hydrogels. Potential use of the formulated nanocarriers for topical application was evaluated through cytotoxicity tests over normal human skin fibroblasts. An *ex vivo* permeation protocol using porcine ear skin was also implemented to assess the release of CBD and its penetration through the skin.

Findings

Experimental results obtained by SAXS, EPR and confocal microscopy show that incorporation of the NE in the hydrogels does not significantly affect their structural properties. CryoEM indicated that CBD does not have observable effects on structural rigidity and ultrastructure of the NE, suggesting minimal perturbations upon drug encapsulation. The proposed systems did not exhibit any cytotoxicity and CBD was released effectively penetrating through the skin indicating their suitability for transdermal delivery.

Keywords: nanoemulsions; hydrogels; cannabinoids; DLS; EPR; cryo-EM; SAXS; confocal microscopy; cell viability; skin permeation

1. Introduction

Nowadays, the consumer interest is shifted towards more natural formulations for the treatment of various conditions. Particularly cannabidiol (CBD) has attracted considerable attention in recent years due to its multiple pharmacological actions. Cannabidiol is a 21carbon terpenophenolic, non-psychotropic compound and constitutes one of the major phytocannabinoids of the Cannabis sativa plant extract (Figure 1) [1]. Over the years, numerous research studies have demonstrated that CBD presents good safety profile, affects various physical functions and could demonstrate promising therapeutic action to medical and conditions such as seizures, epilepsy [2], psychiatric [3] and skin conditions [4], pain and inflammation [5]. In addition, due to the anti-inflammatory [6], immunomodulatory and neuroprotective [7] properties that CBD demonstrates it was proposed as a support drug during the COVID-19 pandemic [8,9]. However, due to the wide variability of the plant composition and subsequently of the CBD extracts and the broadly diversified THC/CBD ratios their use as therapeutic agents is troublesome [10,11]. Moreover, CBD is a sensitive bioactive compound that can easily undergo degradation when in solution or in contact with light and temperature [12]. For that reason, synthetically produced CBD was selected for encapsulation in the present study.

Until now oral route has been the most preferred mode of delivery for CBD. However, this method of administration presents significant limitations due to the highly lipophilic nature of this substance including gastric instability and low oral bioavailability due to rapid elimination through first-pass metabolism [13]. Another important obstacle with oral administration is that it is not feasible for people who suffer from nausea, gastrointestinal problems or seizures. To address the aforementioned limitations research groups are focused on finding more effective methods for the administration of bioactive compounds. There are indications that one of the most advantageous approaches is the transdermal delivery since it provides the possibility of self-administration, prolonged contact with the site of action offering controlled drug delivery over extended periods of time and prevention of gastrointestinal side-effects [14]. Specifically, concerning cannabinoids the transdermal administration leading to systemic delivery could provide sustained and possibly targeted delivery [15]. In fact, the beneficial effects of transdermal delivery of CBD against diverse conditions including arthritis, seizures, and anxiety and in general the overall improvement of the quality of life of the patients have been demonstrated by various *in vivo* studies [16–18].

However, the selection of the suitable delivery route is often not enough so encapsulation in various nanocarriers is frequently preferred in order to enhance the efficacy of a bioactive

 compound of interest. Commonly used nanocarriers include micelles, polymers, nanoparticles, liposomes, niosomes, nanogels, and dendrimers. The main advantages of using nanocarriers include improved solubility of highly hydrophobic compounds, protection against degradation of sensitive bioactive molecules, and efficient delivery to their exact targets [19-22]. Among different nanocarriers for the effective delivery of bioactive compounds, nanoemulsions based on biocompatible materials have been extensively studied by many research teams worldwide mainly due to their physicochemical properties [23,24]. Nanoemulsions (NE) are fine oil-in-water (O/W) or water-in-oil (W/O) dispersions stabilized by an interfacial film of surfactant molecules forming droplets of nanometric sizes. The oily phase usually consists of different types of lipids and oils. The aqueous phase consists of water or can be formulated by adding different water-soluble components. Surfactants are the molecules that adsorb into oil/water interfaces reducing the interfacial tension and stabilizing the emulsions. Nanoemulsions are thermodynamically unstable systems and can be prepared by both low and high-energy methods. Low energy methods include self-emulsification and phase inversion whereas high energy methods include high-pressure homogenization, microfluidization and ultrasonication [25,26]. The main advantages of nanoemulsions for the effective delivery and release of bioactive components are their low surfactant-to-oil ratio, the increased water-solubility, stability, and bioavailability of the encapsulated lipophilic molecules, as well as their administration through various routes [19,25].

Hydrogels (HG) are composed of natural or synthetic hydrophilic polymers that are cross-linked forming three-dimensional networks with the ability to retain large amounts of water. Due to their properties such as high water content, flexibility and biocompatibility they are popular for many biomedical applications including drug delivery [27,28]. Chitosan hydrogels is a class of hydrogels based on chitosan, a cationic natural polymer obtained by deacetylation of chitin. These hydrogels are of special interest due to their appealing physicochemical, biological, and mechanical properties [29]. Chitosan hydrogels are biocompatible, biodegradable, with low toxicity and low cost. Moreover, they are easy to handle and present mucoadhesive properties providing prolonged contact with the site of action. In addition, chitosan hydrogels are promising absorption enhancers of active compounds through constructed epithelia [30]. Despite their advantages, hydrogels are not often used as delivery vehicles of bioactive molecules due to their inability to encapsulate lipophilic substances and their possible toxicity as a result of burst release [31]. Nanoemulsions on the other hand present some limitations such as low viscosity and low spreadability making them less

 appropriate for (trans)dermal or other applications requiring adherence and prolonged contact with the site of action.

Hence, due to the limitations of the two aforementioned nanocarriers, recently there is a growing interest towards their combination resulting in a novel hybrid system the nanoemulsion-filled hydrogels or emulgels (NE/HG). Specifically, nanoemulsion-filled hydrogels consist of a polymer gel matrix in which a nanoemulsion is incorporated [32]. These systems possess different properties compared to the initial nanocarriers such as improved stability, and versatile texture [33]. In addition, they present various advantages over nanoemulsions and hydrogels since they can encapsulate both lipophilic and hydrophilic bioactive compounds, provide controlled release, increase biocompatibility and improve the release profile of the incorporated nanoemulsion [31,34,35].

The aim of the present study was to develop stable, biocompatible oil-in-water nanoemulsions and incorporate them in chitosan hydrogels to formulate a novel and possibly more effective delivery system, the respective nanoemulsion-filled hydrogels. In both systems, cannabidiol (CBD) was encapsulated as active ingredient of pharmacological interest. Since the efficacy of the carrier and its ability to release and deliver the encapsulated bioactive compound directly depends on its structure, the present study focuses on the structural study of the developed systems in order to assess their suitability as delivery vehicles. In addition, a biological evaluation was performed to assess the biocompatibility of the formulated nanocarriers both in the presence and absence of CBD. Finally, the two systems under study were compared in terms of release profiles and skin penetration potential of CBD to determine their possible use as delivery vehicles for its the transdermal delivery.

HO OH

Figure 1. Molecular structure of cannabidiol

2. Materials and Methods

2.1. Materials

Isopropyl tetradecanoate (IPM), 98% was purchased from Alpha Aesar, Germany. Organic extra virgin olive oil (EVOO) from the Amfissa Variety, Pelion, Magnesia, Greece, (acidity <0.4%, oleocanthal 124 mg/kg, oleacein 72 mg/kg, total hydroxytyrosol derivatives 140 mg/kg, total phenols 315 mg/kg) was a gift by "Myrolion" start-up company, Pelion, Magnesia, Greece. Polyoxyethylenesorbitan monooleate (Tween 80), pure, was from Fisher BioReagents, NH, USA. Glycerol monolinoleate (Maisine® CC) HLB 1, Caprylocaproyl Polyoxyl-8 glycerides (Labrasol®) HLB 12 and Diethylene glycol monoethyl ether (Transcutol® HP) were kindly donated by Gattefossé, France. Chitosan (200-600 mPa·s, 0.5% in 0.5% Acetic Acid at 20°C) was purchased from TCI, Belgium. CBD (99% of cannabidiol) from Cannabis sativa L. was a gift from Ekati Alchemy Labs S.L, Barcelona, Spain. 5-Doxyl stearic acid (5-DSA) ammonium salt, >99% purity, was from Avanti Polar Lipids, Alabama, USA. 12- Doxyl stearic acid (12-DSA) was from Sigma-Aldrich Chemie Gmbh Munich, Germany. Acetic Acid glacial 100% was from Merck, Darmstadt, Germany. The porcine ears used as model membranes for the ex vivo experiments conducted in the Faculty of Pharmacy, University of Belgrade, Belgrade, Serbia were kindly gifted by a local provider. Ethanol absolute, analytical grade was purchased from Fisher Scientific, Loughborough, UK. Deionized water was used.

2.2. Formulation of nanoemulsions by low-energy method

Low-energy O/W nanoemulsions (L1, L2) were prepared by the isothermal spontaneous self-emulsification procedure [23] using water, a mixture of surfactants (Labrasol, Tween 80, Maisine), and either IPM or a mixture of IPM with EVOO (1:1 mass ratio). To formulate the nanoemulsions, an oily phase was initially prepared by mixing the surfactants with the oil/oils using a magnetic stirrer. After mixing, the oily phase was slowly added into the water under constant magnetic stirring at room temperature. The prepared nanoemulsions were transferred to glass bottles and stored protected from light.

2.3. Formulation of nanoemulsions by high-energy method

High-energy O/W nanoemulsions (H1, H2) were prepared in a two-step emulsification procedure [23] using water, cosolvent (Transcutol), mixtures of surfactants (Labrasol, Tween 80, Maisine) and a mixture of IPM and EVOO (1:1 mass ratio). Initially, coarse emulsions were obtained with mechanical stirring at room temperature and then nanoemulsions were formulated by passing the emulsions through a Panda PLUS1000 (GEA, Niro Soavi, Parma,

Italy) high-pressure homogenizer at max 390 bar applying up to 10 recirculation passages.

Nanoemulsions after their exit from the homogenizer were immediately cooled using an ice

193 bath.

The composition of the nanoemulsions is shown in Table 1.

2.4. CBD-loaded nanoemulsions

Initially, CBD was dissolved in the 1:1 mixture of IPM and EVOO to result stock solutions at different concentrations. Then, CBD stock solution was mixed with the other components to result O/W nanoemulsions as mentioned above. Final concentrations of CBD in the nanoemulsions were from 1 mg/g to 10 mg/g of nanoemulsion.

2.5. Formulation of nanoemulsion-filled hydrogels

Chitosan hydrogels were formed in glass tubes by mixing chitosan flakes with 1% acetic acid solution under gentle stirring using a spatula. Nanoemulsion-filled hydrogels were formed by adding different amounts of O/W nanoemulsions to hydrogels prepared as mentioned above. Nanoemulsions were incorporated by gentle stirring into the hydrogels using a spatula.

2.6. Dynamic light scattering (DLS) measurements

Droplet size and droplet size distribution of empty and loaded O/W nanoemulsions were measured by dynamic light scattering (DLS) using a Zetasizer Nano ZS (ZEN3600) analyser (Malvern Instruments Ltd, Malvern, UK). The analyser was equipped with a He-Ne laser (633 nm) and Non-Invasive Backscatter (NIBS) optics. Measurements were carried out at 25 °C at a scattering angle of 173 ° and processed using the Malvern Zetasizer Nano software, version 6.32 (Malvern Instruments Ltd, Malvern, UK). The cumulants' analysis gives two values, a mean value for the size (Intensity mean) and a width parameter known as the polydispersity index (PdI). Samples were filtered through 0.45 µm hydrophilic PTFE filters (Millex-LCR from Millipore Merck KGaA, Darmstadt, Germany) and contained in a quartz type cuvette under dust-free conditions. Each sample was measured in triplicate. From these data were calculated the mean and standard deviations.

2.7. Stability studies

Changes in mean droplet diameter and PdI of empty and CBD-loaded O/W nanoemulsions were investigated using the Zetasizer NanoZS device (ZEN3600) (Malvern

Instruments Ltd, Malvern, UK) as mentioned above. Measurements were conducted immediately after preparation and again at given time intervals. Each sample was measured in triplicate. Mean values and standard deviations were calculated from this data.

2.8. Electron Paramagnetic Resonance (EPR) spectroscopy measurements

EPR is a spectroscopic technique that measures the absorption of microwaves by materials with unpaired electrons. Interactions of an unpaired electron with its environment influence the shape of an EPR spectral line thus giving insight into dynamic processes such as molecular motions or fluidity. In the present study, two doxyl stearic acid derivatives (5-DSA ammonium salt and 12-DSA) labelled at different positions of the fatty acid aliphatic chain (C5 and C12) were used to study the fluidity and structure of the surfactants layer at different depths.

EPR measurements were performed at room temperature with an EMX EPR spectrometer (Bruker BioSpin GmbH, Germany) operating at the X-Band (9.8 GHz). A low dielectric constant quartz flat cell for aqueous samples (Wilmad-LabGlass, Cortecnet Europe, France) was used to contain the O/W nanoemulsions and the corresponding nanoemulsion-based hydrogels. Typical EPR settings were center field of 0.348 T, scan range 0.01 T, gain of 5.64x10³, time constant of 5.12 ms, conversion time of 5 ms, modulation amplitude of 0.4 mT. Data collection and analysis were performed using the Bruker WinEPR acquisition and processing program (Bruker BioSpin GmbH, Germany).

Sample preparation for EPR measurements was as follows: Initially 2 mM stock solutions of the spin probes (5-DSA, 12-DSA) in ethanol were made. Then, appropriate amounts of these concentrated solutions were inserted in eppendorf tubes. After ethanol evaporation, 1 mL of the nanoemulsions was added. For the preparation of the corresponding hybrid systems the aforementioned nanoemulsions were incorporated to hydrogels. Final concentration of the spin probes was 0.02 mM.

2.9. Small-angle X-ray scattering (SAXS) measurements

Small angle X-ray Scattering (SAXS) experiments were carried out on a XEUSS 2.0 device (XENOCS, Grenoble, France). Coupled to a FOX 3D single reflection optical mirror centered on the Cu K α radiation (λ = 1.54 Å), the GeniX 3D source delivers an 8 keV beam which is collimated and defined by a set of 2 motorized scatterless slits. The samples were put in thin glass capillaries (optical path 2 mm) which were then airtight sealed. The data were collected by a two-dimensional PILATUS-300k detector (DECTRIS, Switzerland) placed at

different distances from the sample, giving access to the angle range $0.056^{\circ} - 2.26^{\circ}$ ($0.04 \text{ nm}^{-1} - 1.6 \text{ nm}^{-1}$). A thermostatically controlled module makes it possible to carry out studies in T from -30 °C to 150 °C. The measurements were performed at room temperature.

2.10. Confocal Fluorescence Microscopy

Confocal fluorescence microscopy was performed to detect and visualize the oil phase of the nanoemulsions and the respective hybrid systems. For that reason, the oil phase (1:1 mixture of IPM/EVOO) was stained with the lipophilic dye Nile red. Then nanoemulsions and the corresponding nanoemulsion-filled hydrogels were formulated as mentioned above (2.3 and 2.5) using the aforementioned oil phase. Following, one drop of the stained samples was placed on a slide and covered with a cover slip, ensuring that there were no air bubbles. Samples were mounted and visualized using a confocal laser scanning microscope (Leica TCS SPE, Leica Microsystems, Heidelberg, Germany), with an excitation wavelength at 488 nm and emission wavelength at 500 nm. For the images acquisition LAS AF software (Leica Microsystems) was used.

2.11. Cryo-Electron Microscopy

The morphology of the CBD-loaded and empty nanoemulsion samples were studied using cryogenic transmission electron microscopy method. For that purpose, quantifoil® R2/1 grids were utilized as a substrate. The grids were treated and prepared according to a well-established protocol which is fully described elsewhere [36]. The grids were then loaded in a ThermoFisher Scientific Vitrobot Mark IV grid plunger. The chamber of the grid plunger was adjusted at 4 °C and >95% relative humidity. A 3.5 μ L aliquot from each sample were applied on the prepared grids and blotted using type 595 filter paper for 6 seconds. The blotted grids were plunged in liquified ethane were further assembled in autogrids and loaded inside the ThermoFisher Scientific Glacios 200 kV CryoEM under cryogenic conditions. The micrographs were acquired at the pixel size of 3.17 Å/pixel and 30 e/Å2 electron dose using ThermoFisher Scientific EPU V 2.11 software.

2.12 Cell Culture and Cell Proliferation Assay

WS1 human normal skin fibroblasts were purchased from the American Type Culture Collection (ATCC, Manassas, VA, USA). The cell line was grown in complete EMEM culture medium (containing 2 mM L-glutamine, 1 mM sodium pyruvate, and 1500 mg/L sodium bicarbonate), supplemented with 10% FBS and 1% penicillin/streptomycin (Gibco-Life

 Technologies), at 37 $^{\circ}$ C in a humidified incubator with 5% CO2. Cells were cultivated in 96 well tissue culture plates (1.6 \times 10⁴ per well) and treated with different concentrations (0.5–1.75 mg/mL) of the formulated nanoemulsion and the respective nanoemulsion-filled hydrogel in absence and presence of the bioactive compound for 72 h. Non-treated cells were used as control.

Inhibition of cell proliferation was assessed 72 h after treatment by the MTT (3-(4,5-Dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide) assay according to the manufacturer's standard protocol. To each culture being assayed MTT stock solution (5 mg/mL) was added to equal one tenth of the original culture volume and then incubated for 3 h. At the end of the incubation period the solution was discarded, and the converted dye was solubilized in 1:1 isopropanol/DMSO solution to dilute the insoluble purple formazan crystals. Absorbance of the converted dye was measured at 570 nm. Cell viability was calculated using the equation below:

Cell viability (%) = (OD of treated cells)/ (OD of control) \times 100 where, OD is the optical density.

2.13. *In vitro* permeation study

In order to evaluate the skin permeation of CBD from the nanoemulsions and the respective nanoemulsion-filled hydrogels an in vitro permeation protocol was implemented using full-thickness porcine ear skin as the model membrane. The porcine ears were obtained from a local slaughter house in Serbia and were processed appropriately. The porcine ears were washed with cold tap water immediately after slaughter, dried carefully with soft tissue, wrapped with aluminum foil and stored at $-20~^{\circ}\text{C}$ to be preserved as intact as possible until use. Subsequently, for the implementation of the in vitro protocol, the porcine ears were defrosted at room temperature and the hair were shortened carefully with electric trimmer to obtain a sufficiently clear surface without destroying hair follicles or damage the stratum corneum (SC). Then the full-thickness skin was detached from the cartilage using a scalpel and punched with a custom-made spherical cutter of a 25 mm diameter to achieve the desired discs used as membranes for the experiment. To ensure the integrity of the skin barrier transepidermal water loss (TEWL) (Tewameter® TM 210; Courage+Khazaka, Köln, Germany) was measured.

The full-thickness skin discs were placed on modified Franz diffusion cells with a surface of 2.01 cm² and a receptor chamber of 12 mL volume (Gauer Glas, Germany) and the waterbath temperature was set at a constant of 32 °C in agreement with the mean skin surface temperature. To conduct the experiment the receptor compartments were filled with a preheated mixture of PBS 1X solution and ethanol 96% V/V (1:1) which was under constant stirring using a magnet at 500 rpm. The porcine full-thickness skin was placed between the donor and receptor compartments with the SC towards the donor chamber. After 30 min of equilibration, the donor compartments were uniformly filled with 0.5 mL of the formulations (CBD-loaded nanoemulsions or nanoemulsion filled hydrogels) and covered with Parafilm® to prevent evaporation. Samples of 500 μL volume were collected every 2 hours for a total of 30 hours of experiment starting from the formulation application. The receptor compartments were filled again each time after sampling with an equal volume of fresh pre-warmed medium. To determine the amount of CBD in the samples a LC-MS/MS method was used as described below (paragraph 2.15). From the obtained results were determined the release profiles from the two proposed nanocarriers and were calculated the steady-state flux (Jss), cumulative amount of CBD permeated through the skin at the end of the 30 h of the experiment (Q_{30h}), and permeability coefficient (K_p) as described elsewhere [37].

2.14. Tape stripping experiments

The technique of tape stripping using porcine ear was implemented to evaluate the two systems under study in terms of CBD penetration through the stratum corneum (SC) [38]. For this experiment a porcine ear preserved as described above was initially thawed, and after hairs were removed with a hair trimmer, it was fixed on Styrofoam plates without detaching the skin from the cartilage. In order to determine the appropriate time point to apply the formulation the TEWL was measured periodically using a Tewameter® TM 210 and when it reached a value of approximately 15 g.m-2.h-2 [39] the CBD-loaded nanoemulsion and the respective hybrid were cautiously applied on the marked skin sites. The study was conducted under infinite dose regimen, using the same dosing conditions as for in vitro skin permeation study (1 mL of each of the formulations were left on the skin for 2 h (effective diffusion area: 4.02 cm²)). To determine the amount of SC removed, each adhesive tape was weighted accurately before starting the tape stripping. At the end of the 2 h incubation time, the excess formulations were carefully removed using dry cotton swabs and the process of gradual removal of the SC began, for which 30 adhesive D-squame® discs (CuDerm Corporation, USA) were used. For this procedure the adhesive tapes were pressed onto the skin with a roller to

avoid any wrinkling and then were removed. After repeated weighing, each tape was inserted into a centrifuge tube to extract the CBD. To achieve that, 4 ml of ethanol (70%, v/v) was added in each tube and followed sonication (Sonorex RK 120H, Bandelin, Berlin, Germany) for 15 min and the centrifugation at 4000 rpm for 5 min (Centrifuge MPW-56; MPW Med. Instruments, Warszawa, Poland). The final supernatant was collected and analysed for CBD content using LC-MS/MS method [38].

2.15. LC-MS/MS method

Determination of CBD in aliquots obtained during in vitro penetration and permeation studies was performed by using LC-MS/MS method. A liquid chromatographic system Accela 1000 (Thermo Fisher Scientific, San Jose, California, USA) consisting of an auto sampler and quaternary pump was used. Chromatographic separation was achieved using Infinity Lab Poroshell 120 EC-C18 (100 mm x 4.6 mm, 2.7 μm; Agilent Technologies, USA) at 40 °C. Mixture of acetonitrile and 0.1% formic acid (80:20, v/v) was used for isocratic elution at flow rate of 500 μL/min. Total analysis time was 6 min. A TSQ Quantum Access MAX triple quadripole spectrometer was used mass analyses. The spectrometer was equipped with electrospray ionization source and worked with nitrogen as nebulizing gas. ESI source parameters were optimized by syringe infusion (20 μL/min) of CBD standard solution. For data acquisition selected reaction monitoring (SRM) positive scan mode was used. ESI source and mass spectrometry parameters were as follows: spray voltage 3500 V; vaporizer temperature 400 °C; sheath gas pressure 50 units; ion sweep gas 0 units; auxiliary gas 45 units; ion transfer capillary temperature 250 °C; capillary offset 35 units; tube lens offset 50 units; skimmer offset 0 units; peak width relating to resolution 0.7 for Q1; scan with (m/z) 0.02; scan time 200 ms. Xcalibur software v. 2.1.0.1139 was used for data acquisition and processing (Thermo Fisher Scientific, San Jose, CA, USA). Mass spectrometry was used for detecting specific ions for analyte identification. MS full-scan mass spectra indicated that in ESI source [M+H]+ was the dominant ion for CBD (m/z 315.3). The setting of SRM transition channel for CBD monitoring was selected product ions at m/z 193.2 (collision energy was 19 V).

3. Results and discussion

- 3.1. O/W nanoemulsions
- 391 3.1.1. Formulation of O/W nanoemulsions

O/W nanoemulsions are thermodynamically metastable but kinetically stable nanosized emulsions, composed of oil droplets dispersed in an aqueous medium and stabilized by

 surfactant molecules. Usually, for the formulation of nanoemulsions either high-energy or low-energy methods are applied. As can be seen in Table 1, low-energy O/W nanoemulsions were prepared with water as continuous phase, a 2:2:1 (mass ratio) mixture of Tween 80/Labrasol/Maisine as surfactants and IPM as dispersed oil phase (L1). Following, low-energy nanoemulsions which contained water as continuous phase, the same blend of surfactants and a 1:1 mixture of IPM/EVOO as oil phase were formulated (L2). Both low energy nanoemulsions were opaque and remained visually stable for at least 80 days. When nanoemulsions having the same composition as L2 but homogenized by the two-step high-energy emulsification procedure were developed (H1), visual destabilization occurred 50 days after their formulation. High-energy nanoemulsions containing 1% Transcutol, a colorless liquid commonly used as cosolvent in many surfactant formulations, were formulated and remained visually stable for 60 days (H2).

In the present study, all nanoemulsions contained a 2:2:1 mixture of Tween 80, Labrasol and Maisine. These surfactants are biocompatible and non-ionic having hydrophilic lipophilic balance (HLB) values ranging from 1 to 15. As reported in a previous study the HLB value of the mixture of Tween 80/Labrasol/Maisine was calculated by adding values of the individual surfactants and was found 11 [40]. It is well established in the literature that mutual solubilisation of two immiscible liquids such as oil and water drastically increase when blends of surfactants with different HLB values are used [41,42].

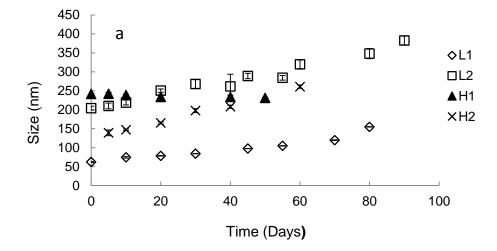
Following their formulation, both low- and high-energy O/W nanoemulsions were studied using analytical techniques based on scattering, spectroscopy, and microscopy to investigate their structural characteristics in relation to their composition and preparation method.

Table 1. Composition of the nanoemulsions.

Ingredients (% w/w)		System		
	L1	L2	H1	H2
Water	91	91	91	90
Tween 80	2	2	2	2
Labrasol	2	2	2	2
Maisine	1	1	1	1
Transcutol	-	-	-	1
IPM	4	2	2	2
EVOO	-	2	2	2

3.1.2. Droplet size, polydispersity index and stability

Dynamic light scattering (DLS) was applied to measure the hydrodynamic diameter of the oil droplets and their size distribution. Measurements were taken immediately after preparation and at given time intervals to evaluate storage stability of the samples. Figure 2 shows the variation of mean droplet diameter and PdI with time for all nanoemulsions tested (L1, L2, H1, H2). As can be seen in Fig 2a mean droplet diameter of L1 nanoemulsions increased from 62 nm to 155 nm after a total of 80 days before phase separation. In the case of L2 nanoemulsions larger droplets were formulated probably due to the presence of EVOO in the oil phase. Immediately after preparation mean diameter of the dispersed oil droplets was 204 nm. The size of the droplets was gradually increased up to 383 nm after 90 days of storage. In the case of H1 nanoemulsions a small droplet size decrease from 242 nm to 231 nm within 50 days of storage was observed. Finally, when H2 nanoemulsions were examined, oil droplets of 139 nm that increased up to 261 nm after 60 days of storage were obtained.



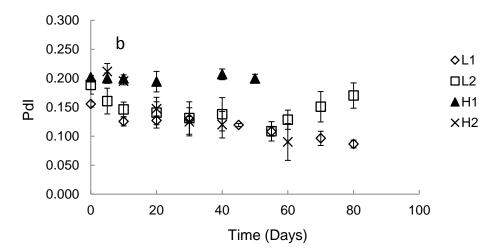


Figure 2. DLS study of O/W nanoemulsions: a) Droplet size and b) polydispersity index (PdI) as a function of time. (◊): L1; (□): L2; (▲): H1; (×): H2.

 In Figure 2b, the polydispersity index is shown as a function of storage time. Both lowand high energy nanoemulsions were monodispersed with PdI <0.2. Similar findings have been reported by our group in a previous study where O/W nanoemulsions containing 2% w/w of the same oils but at different proportions and surfactant's mixture of HLB 11 were evaluated [40].

In Figure 2 we can clearly observe that nanoemulsions containing EVOO in the oil phase (L2, H1, H2) had larger oil droplets and higher PdI as compared to the one composed of IPM only (L1). EVOO is a mixture of triglycerides with a variety of minor components (monoand diacylglycerols, free fatty acids, phospholipids, sterols, tocopherols, polyphenols etc.) and seems to affect both the size of the oil droplets and the physical stability of the final formulation [43]. In general, the choice of the oil in terms of structure, polarity, surface tension and viscosity has an important impact on the formulation of nanoemulsions. IPM is an oil of lower viscosity as compared to EVOO generating nanoemulsions with smaller average droplet size. On the other hand, oils of lower polarity such as vegetable oils are generally more suitable regarding the physical stability of nanoemulsions, being less susceptible to Ostwald ripening [44,45]. When the composition of the oil phase is complex, relative contribution of each component to the final characteristics of the formulation must be considered.

To summarize, droplet size, size distribution, and stability of the O/W nanoemulsions were affected by their composition and the emulsification procedure. Low energy nanoemulsion in the absence of EVOO and Transcutol (L1) showed the lowest values of mean droplet diameter and PdI and remained stable for 80 days. When EVOO was added in the oil phase mean droplet diameter was increased by several nanometres (L2, H1, H2). Besides, in the presence of Transcutol (H2), smaller oil droplets as compared to L2, H1 were obtained although these systems were destabilized 60 days after preparation. More specifically, nanoemulsions with same composition but produced with different emulsification procedures (L2, H1) exhibited differences in the stability measured in terms of hydrodynamic diameter and polydispersity index (PdI).

Considering the results of the DLS study regarding droplet size and stability, nanoemulsions consisted of 91% w/w water, 2% w/w IPM, 2% w/w EVOO and 5% w/w blend of surfactants (L2) were proved as more appropriate for the encapsulation of CBD. These nanoemulsions are more advantageous than the other systems as encapsulation vehicles of CBD since they contain EVOO, a natural vegetable oil with health benefits [46]. What is more, L2 nanoemulsions are stabilized in the absence of cosolvent applying the low-energy

emulsification technique making them more promising as drug carriers in potential industrial application.

3.2. Encapsulation of CBD

3.2.1 Formulation of CBD-loaded nanoemulsions

Cannabidiol (CBD), a naturally occurring cannabinoid, is a highly lipophilic terpenophenolic compound. For its encapsulation in the oil droplets of O/W nanoemulsions a procedure based on emulsification was applied as described in Methods (2.4. CBD-loaded nanoemulsions). The effect of CBD encapsulation on droplet size, polydispersity, stability, and membrane dynamics was evaluated using DLS, EPR and SAXS. For the DLS experiments CBD concentration in the final nanoemulsions was 1 mg/g and 5 mg/g.

3.2.2. Droplet size, polydispersity index and stability of CBD-loaded nanoemulsions

As can be seen in Table 2, immediately after preparation, smaller oil droplets as compared to empty L2 nanoemulsions were observed. The dispersion index PdI which reflects the distribution of particle size, was below 0.25 showing uniform dispersions. In CBD loaded nanoemulsions, oil droplets of 126±3 nm diameter and 0.18±0.01 PdI were formulated in the presence of 1 mg/g CBD. When the concentration of the lipophilic active compound was increased up to 5 mg/g, the size of the droplets was further decreased and the PdI increased to 0.24±0.01. In both cases mean oil droplet diameter was increased with time while the PdI was initially decreased and then remained practically unchanged and lower than 0.2. These findings indicate that the presence of CBD in the oily droplets of the nanoemulsions influences certain physicochemical characteristics and therefore further studies should be carried out to elucidate the interplay between all components. In previous studies when encapsulation of curcumin, vitamin D3, and ibuprofen in O/W nanoemulsions was considered, different effects were observed depending on the nature of the added compound. More specifically, encapsulation of vitamin D3 or curcumin in the dispersed oil phase of edible O/W nanoemulsions caused an increase in the size of the oil droplets. This was related to the interfacial location of both bioactive compounds and their effect on membrane dynamics in terms of rigidity and order [47,48]. In another study, ibuprofen, a lipophilic drug, was encapsulated in O/W nanoemulsions based on limonene and surfactants' mixture without affecting droplet size or stability. In that case localization of the drug exclusively in the oil cores was indicated and related to the strong lipophilic character of the added compound [49]. In a recent study of Zhao et al., encapsulation of lycopene within O/W nanoemulsions using

 lactoferrin and sesame oil resulted in droplet size increase and was related to the limited solubility of lycopene in oil [50].

Table 2. Droplet size (d, nm) and polydispersity index (PdI) of CBD-loaded nanoemulsions. The final concentration of CBD was 1 mg/g and 5 mg/g of nanoemulsion. Temperature was 25 °C. Data expressed as the mean \pm SD.

[CBD]				Tin	ne (days)			
		0	10	20	30	45	80	95
1 mg/g	Size (nm)	126±3	178±3	204±2	223±4	236±4	277±6	296±8
	PdI	0.18±0.01	0.11±0.01	0.09±0.01	0.10±0.21	0.09±0.03	0.11±0.02	0.11±0.02
5 mg/g	Size (nm)	102±3	185±10	178±6	182±9	214±6	342±10	389±8
3, 0	PdI	0.24±0.01	0.23±0.01	0.22±0.01	0.20±0.02	0.15±0.02	0.15±0.02	0.16±0.02

3.3. Nanoemulsion-filled hydrogels

Chitosan hydrogels are three dimensional networks of a natural polycationic copolymer that can absorb and retain large amounts of water. In general, hybrid hydrogels are networks of natural or synthetic polymers that incorporate proteins, peptides, or nanoand microstructures, interconnected via physical or chemical means, to yield materials with improved mechanical strength and expand their range of applications [51].

In recent years, different approaches were applied to develop nanoparticle hybrid hydrogels with varying nanoparticle types and varying hydrogel frameworks [52]. As reported by our group in an earlier study, hybrid hydrogel can be developed adding 1.25 % w/v chitosan in a 1:1 mixture of acetic acid solution/nanoemulsions under mild conditions [40]. This method of preparation could be more advantageous compared to other ones previously mentioned in the literature since it's simple and rapid [53]. These hybrids were used as carriers for the encapsulation and effective delivery/release of vitamin D3 and curcumin. In the present study, the nanoemulsions used for the development of chitosan hybrid hydrogels were formulated with the low-energy homogenization procedure and consisted of 91% w/w water, 2% w/w IPM, 2% EVOO and 5% w/w surfactants' mixture (L2). For the encapsulation of CBD in hydrogels, nanoemulsions loaded with CBD were initially formulated as described in Methods (2.2.1. Preparation of nanoemulsions). These loaded nanoemulsions were then mixed with acetic acid 1% and chitosan as mentioned above to result the nanoemulsions-

based hydrogels. The concentration of CBD in the nanoemulsion-based hydrogels was 2.5 mg/mL.

3.4. Dynamics of the surfactant layer

3.4.1. O/W nanoemulsions

EPR experiments using 5-DSA ammonium salt and 12-DSA as amphiphilic spin probes were conducted to investigate the effect of the presence of CBD on local viscosity, order and local polarity of the surfactants' layer in the O/W nanoemulsions. These probes are spin-labelled fatty acid analogues consisting of stearic acid and an N–O· moiety attached to the C-5 and C-12 position of the hydrocarbon chain, respectively. In the case of 5-DSA the paramagnetic moiety of the probe is localized closer to the polar head groups of the surfactants while in the case of 12-DSA it is localized closer to the surfactant tails. In other words, 5-DSA senses variations closer to the aqueous phase whereas 12-DSA senses changes of the oil phase that penetrates the lipophilic surfactant chains [54].

Figure 3 shows the experimental EPR spectra of 5-DSA ammonium salt in empty and CBD-loaded O/W nanoemulsions and nanoemulsions-filled hybrid hydrogels. Three-line EPR spectra of unequal heights and widths characteristic of doxyl derivatives when located in membranes, were obtained in all cases.

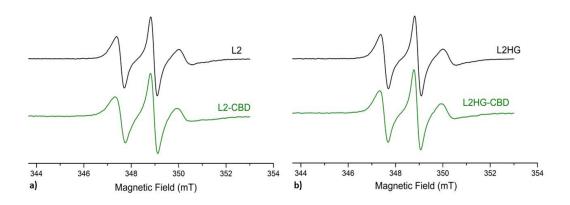


Figure 3. EPR spectra of 5-DSA ammonium salt in (a) empty (L2) and full, CBD-loaded (L2-CBD) O/W nanoemulsions and (b) empty (L2HG) and full, CBD-loaded (L2-CBD).

In Table 3, three parameters (τ_R , S and α_N) obtained from the EPR spectral characteristics in the absence and the presence or CBD are presented. The detailed equations and the physical meaning of those parameters have been reported in previous studies [55,56]. As can be observed, in the case of 5-DSA, both τ_R and S values were increased upon CBD encapsulation in the oil phase of the O/W nanoemulsions. More specifically, τ_R was increased from 2.5 ± 0.1 ns to 3.2 ± 0.1 ns and S from 0.13 ± 0.01 to 0.19 ± 0.01 (Table 3). The increase of rotational correlation time (τ_R) in the presence of the bioactive compound is indicative of a more hindered motion of the spin probe in the area close to the surfactants' polar head groups. In addition, order parameter of the surfactants' layer was also increased showing a more rigid environment in the same area. These changes indicate penetration and interaction of CBD with the surfactant molecules rendering the surfactants' monolayer more viscus and less flexible. The nitrogen hyperfine splitting, α_N , of the N–O· moiety of 5-DSA is a local polarity parameter which remained practically the same in the presence of CBD. Similar findings were recorded when the other spin probe, 12-DSA, was considered. In that case, τ_R and S values were lower as compared to those of 5-DSA which could be explained by the different location of 12-DSA in the membrane. The nitrogen hyperfine splitting, α_N , of the N–O· moiety of 12-DSA was higher both in the presence and absence of CBD, indicating differences in the micro polarity at different depths of the surfactants' monolayer. Similar findings have been reported in previous studies of our group when encapsulation of lipophilic compounds such as Vitamin D3 in food grade O/W nanoemulsions [47] or extra virgin olive oil-in-water nanoemulsions [48] were considered. Encapsulation of curcumin in O/W nanoemulsions containing eucalyptol (EUC) or pinene in the dispersed oil phase also affected the surfactants layer increasing local viscosity [57].

3.4.2. Nanoemulsion-filled hydrogels

Nanoemulsion-filled hybrid hydrogels were investigated by EPR spin probing spectroscopy applying 5-DSA ammonium salt and 12-DSA. These amphiphilic spin probes were used to monitor possible changes in the surfactants' layer after incorporation of the O/W nanoemulsions into the chitosan hydrogels. As can be observed, membrane dynamics in nanoemulsion-filled hydrogels as expressed by τ_R , S and α_N where similar to the corresponding nanoemulsions showing that chitosan used as gelling agent at a concentration of 1.25 % w/v has no effect on local viscosity, order and polarity. On the other hand, hydrogels were less affected by the presence of CBD. More specifically, rotational correlation time (τ_R) was slightly increased but order parameter (S) and local polarity remained the same.

Table 3. Rotational correlation time (τ_R), order parameter (S) and isotropic hyperfine splitting constant (α_N) of 5-DSA and 12-DSA in empty and loaded nanoemulsions (L1, L1-CBD) and hybrid hydrogels (L1HG, L1HG-CBD).

		5-DSA			12-DSA	
System	τ _R (ns)	S	$lpha_{ m N}$ (x10 ⁻⁴ T)	τ _R (ns)	S	$lpha_{ extsf{N}}$ (x10 ⁻⁴ T)
L1	2.5±0.1	0.13±0.01	13.8±0.0	1.4±0.0	0.08±0.00	14.4±0.0
L1-CBD	3.2±0.1	0.19±0.01	13.5±0.0	2.3±0.1	0.14±0.01	14.0±0.1
L1HG	2.4±0.2	0.15±0.02	13.5±0.1	1.5±0.1	0.09±0.01	14.4±0.1
L1HG-CBD	2.7±0.1	0.15±0.02	13.6±0.1	1.6±0.0	0.09±0.00	14.3±0.1

3.5. Confocal Microscopy

Confocal microscopy was implemented to determine the structure of the formulated systems and identify possible changes in the shape of the oil droplets after incorporation in the chitosan hydrogels. The fluorescent confocal microscopy images indicated that the oil droplets of the O/W nanoemulsions were spherical in shape and about 120 nm in diameter (Figure 4). Moreover, confocal micrographs obtained for the nanoemulsion-filled hydrogels (hybrids) show no significant changes in the shape or size of the incorporated nanoemulsions (Figure 4). Similar microscopy studies of both initial O/W nanoemulsions and the corresponding hydrogels have not been so far reported in the literature. In a study of McClements and co-workers, confocal microscopy images of O/W nanoemulsions containing Vitamin D3 and prepared using natural surfactants and different carrier oils indicated the presence of spherical nanodroplets [58]. In another study, confocal fluorescence microscopy was used to study the morphology of soluble hydrogels formulated with flavour oil nanoemulsions and biopolymers. Confocal micrographs of the systems under different conditions were obtained showing visible red structures of oil stained with Nile red [59].

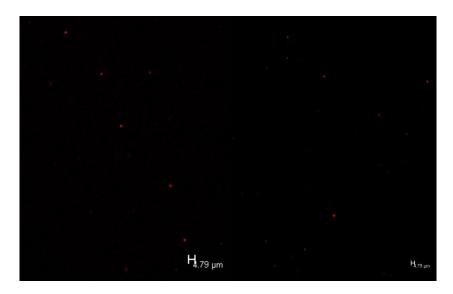


Figure 4. Images of the Nile red-stained oil mixture of the formulated nanoemulsion (left) and nanoemulsion-based hydrogels (right) obtained by confocal fluorescence microscopy (X 40 magnification).

3.6. Cryo-Electron Microscopy

To directly observe the architecture of nanoemulsions and their CBD-loaded conformation utilizing cryogenic electron microscopy (cryo-EM), all specimens were vitrified by plunge-freezing (Materials and Methods). Nanoemulsions are cooled rapidly (10⁵ °C s⁻¹) to avoid ice crystal formation that perturbs their structure, producing, therefore, vitreous ice that is amorphous similar to a liquid with high density. Due to nanoemulsions' high viscosity, formation of vitreous ice was challenging as it is majorly affected by the intrinsic sample viscosity that must be kept to a bare minimum [60]. Despite above-mentioned challenges, we systematically probed vitrification conditions for all specimens and retrieved cryo-EM data for the L2 nanoemulsions (Table 1) that eventually allowed direct observation of the content in both unloaded and CBD-loaded states (Figure 5). This result shows that nanoemulsion samples are in principle amenable to cryo-EM analysis.

Empty O/W nanoemulsions localized in thick vitreous ice regions, systematically forming spherical signatures (Figure 5A). L2 nanoemulsions had preferential distribution in thicker vitreous ice like polymersomes previously observed with cryo-EM [61–63] and contrasts with the intrinsic surface properties of (proteo)liposomes that are frequently observed on the amorphous holey carbon grid edge [64,65]. Indeed, as DLS data suggest (Figure 2) their diameters slightly vary; due to their high sphericity, cryo-EM and DLS data regarding diameters agree well (Figure 5B, Figure 2). We also characterized via cryo-EM rare nanoemulsions with substantially larger diameters (Figure 5B) that may well serve for delivery

 of more complex material in terms of size. CBD-loaded nanoemulsions exhibited similar architectural properties (Figure 5C) and statistical analysis of their particle diameters did not show any substantial change in their observed shapes (Figure 5D). This result suggests that CBD, which is encapsulated in those nanoemulsions (Figure 3) does not have observable effects on their structural rigidity and ultrastructure, suggesting minimal perturbations upon drug encapsulation.

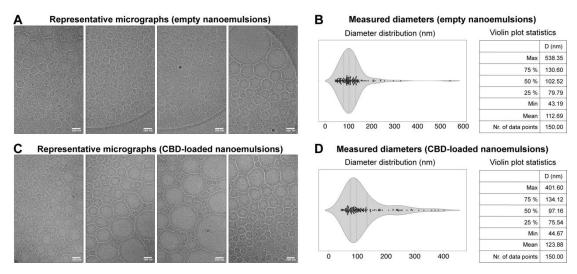


Figure 5. Cryo-EM study of L2 nanoemulsions. (A) Representative micrographs of empty nanoemulsions showing their distribution, which is regulated by ice thickness (the thicker the ice, the highest the odds to identify the particles). Background black dots identified in the micrographs represent possible surfactants. (B) Violin plot illustrating the measurements of diameters for 150 single-particle nanoemulsions along with the underlying statistics in the accompanying table. Note the dots in the violin plots represents each single data point. (C) Same as (A), but for drug-loaded full nanoemulsions. (D) is same as (C), but for drug-loaded nanoemulsions.

3.7 Small angle X-ray Scattering (SAXS)

SAXS is an analytical and non-destructive method implemented for the investigation of nanostructures in liquids and solids. SAXS is able to probe the colloidal length scales of 10–1000 Å thus being a suitable tool for the determination of the size and the structure of colloidal systems such as microemulsions, nanoemulsions and hydrogels [66]. Actually, reaching the upper limit requires specific equipment and favourable systems. In the present study this method was initially implemented to determine the effect on the size when different concentrations of CBD were encapsulated in the nanoemulsions. Specifically, Figure 6a, represents the intensity profile of the formulated nanoemulsions in absence and in

 presence of various concentrations of CBD in the so-called Porod representation. Since characteristic sizes are quite large, the scattering intensity I is dominated by the asymptotic power-law decay $1/q^4$, also known as Porod's law, in the investigated scattering wave vector q range. The Porod representation, namely I X q^4 vs. q, magnifies deviations from Porod's law, unravelling the structure [67].

Another practical limitation nevertheless comes here from the low contrast of organic matter with respect to water and the ensuing low signal-to-noise ratio. Some firm structural information still emerges from the processed data, at least in a semi-quantitative sense. The depicted spectra in combination with the simulation (red line) provide an indication of the structure of the nanoemulsions. First of all, the below spectra illustrate an oscillating red curve simulating the scattering by an assembly of non-interacting, identical spheres with radius about 49.3 nm, suggesting that the spherical model invoked in DLS and confocal microscopy is structurally relevant also in SAXS for all the samples.

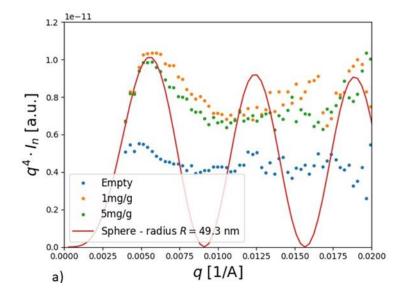
Then, the encapsulation of two different CBD concentrations, namely 1 mg/g and 5 mg/g, was investigated and compared with the respective nanoemulsions in absence of the bioactive compound. It was observed that that the (truncated) «peak» around 0.0045 A⁻¹ for the CBD-free system moves to a slightly larger value, ca. 0.005 A⁻¹ with CBD concentration which should be associated with the decrease of the droplet size of the nanoemulsions with the encapsulation of CBD. However, size is not significantly affected by concentration for the two CBD-loaded systems. In contrast with the light scattering data, size remains unchanged for both the CBD-loaded nanoemulsions indicating that the concentration of the encapsulated compound does not affect the size of the formulated nanoemulsions.

On the other hand, as may be observed from the SAXS curves (Figure 6a) the diameter of the oil nanodroplets calculated by fitting SAXS data were always smaller than the values measured by DLS (Figure 2 and Table 2). This was expected since DLS measures the hydration layer that moves with the nanodroplets and not only droplet's size [68].

In addition, the structure of the chitosan hydrogel and the size of the nanoemulsions after their incorporation in the hydrogels were studied to determine possible changes. First, the chitosan hydrogel was examined and as can be seen from the spectrum presented in Figure 6c two-dimensional (thin, with large lateral dimensions) objects like platelets may describe the structure of the chitosan hydrogel. This hypothesis arises by the fact from the $1/q^2$ intensity decay at large wave vectors. Consequently, nanoemulsions both in absence and in presence of CBD were incorporated in the chitosan hydrogel resulting in the hybrid systems. In Figure 6b are depicted the hybrid systems in absence of CBD and the CBD loaded hybrid

systems with CBD concentration 0.5 mg/g and 2.5 mg/g. These systems were also studied in terms of structure and size. The spectrum illustrated in Figure 6b demonstrates that the spherical structure and the size of the nanoemulsions were not affected by their incorporation in the hydrogel, with the CBD-free system again exhibiting a larger characteristic size than both CBD-loaded systems.

To summarize, the results obtained by the SAXS complete those acquired by EPR and confocal microscopy suggesting that indeed the incorporation of the nanoemulsions in the chitosan-based hydrogel could provide protection to the nanoemulsions without affecting significantly their structural properties.



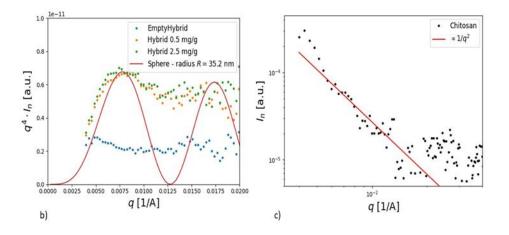


Figure 6. a) Data of SAXS in the Porod representation for empty and loaded nanoemulsions Empty nanoemulsion (blue ●) and loaded systems with 1 mg/g CBD (orange ●) and 5 mg/g

CBD (green ●), simulation (red line). b) Empty nanoemulsion-filled hydrogel (blue ●) and loaded systems with 0.5 mg/g CBD (orange ●) and 2.5 mg/g CBD (green ●), simulation (red line). c) Chitosan hydrogel (black ●), simulation (red line).

3.8 Cell viability

The suitability of a nanodispersion to be used as an effective carrier of bioactive compounds depends on various factors. Biocompatibility of the nanocarrier itself, in absence of the bioactive compound, is one of the most crucial factors. For that reason, the potential skin irritation and cytotoxic effect of the formulated nanoemulsions and the respective nanoemulsion-filled hydrogels were investigated in vitro. Following, the CBD-loaded systems were also studied to evaluate the effects of the bioactive compound on the cell viability. Cytotoxicity was assessed through the colorimetric MTT assay. This tetrazolium-based assay measures the reduction of tetrazolium salt by intracellular dehydrogenases of viable living cells. Since the formulated systems are intended for topical use and possibly for wound healing, human normal fibroblasts (WS1) were used in order to compare the biocompatibility of the two nanodispersions. Thus, WS1 cells were treated with the nanoemulsions and the corresponding hydrogels in the absence and presence of CBD at a concentration range of 0.5 to 1.5 mg/mL in the culture medium. As can be seen in Figure 7, both the NEs and the NE/HGs did not exhibit any cytotoxic effect up to the concentration threshold of 1.5 mg/mL in the cell culture medium. The high biocompatibility of these systems could be attributed to the low concentration of surfactants. In fact, there are indications that the administration of the empty systems causes a dose-dependent induction of cell proliferation. This result could be attributed to the use of olive oil in the oil phase [69,70] and the chitosan. As has been reported before by Howling et al. [71], highly deacetylated chitosan, like the one used in our case, strongly stimulated fibroblast proliferation. The exact mechanism of action has not been elucidated however it is suggested that chitosan binds to serum factors, such as growth factors [72]. Furthermore, at higher concentrations the CBD-loaded systems result in an increase of cell viability indicating that CBD could also promote the proliferation of fibroblasts. Since until recently, CBD was most commonly administered orally, there is no previous literature evaluating the effect of CBD on normal human fibroblasts. However, there are indications that the endocannabinoid system is involved in the wound healing response in fibroblast cells through the upregulation of epidermal growth factors based on studies carried out using murine models [4]. Furthermore, a recent in vivo evaluation of the CBD effect on the wound healing process using rats demonstrated a noticeable regeneration of epidermis and dermis

 [73]. Hence, the observed fibroblast proliferation when treated with the CBD-loaded systems could be interpreted by the aforementioned studies. However, at the minimum concentration administered (0.5 mg/mL) this effect is not evident and the systems under study did not demonstrate significant different cell viability values. Recently, a similar cytotoxicity study was described by Nigro et al. concerning the evaluation of fibroblasts cell viability after the administration of nanoemulsion-filled hydrogels intended for topical administration of coenzyme Q10 (coQ10) [74]. The specific research group demonstrated cell viability below 50% when the system in the absence of coQ10 was administered in concentration 1.25 mg/mL.

To conclude, the cell viability evaluation indicates that the nanodispersions under study present high biocompatibility and could stimulate skin fibroblast proliferation. Hence, these systems could be ideal candidates to be used as carriers for bioactive compounds with antimicrobial and anti-inflammatory properties for the development of a novel wound healing formulation.

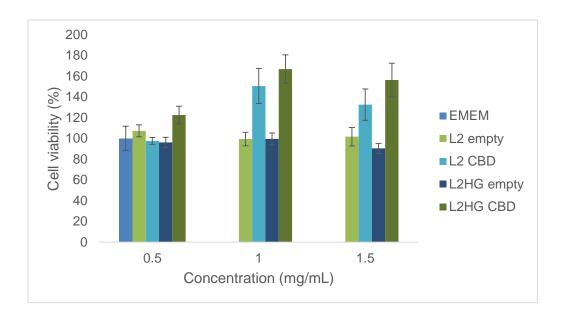


Figure 7. Cell viability testing results in WS1 human fibroblasts cell line after 72 h of treatment with L2 nanoemulsion and the respective L2/HG nanoemulsion-filled hydrogel in the absence and presence of CBD using the MTT cell proliferation assay. Each column represents the mean of five replicates ± SD.

3.9. In vitro permeation study

The evaluation of in vitro drug permeation from a carrier is a valuable tool since it provides an indication concerning the bioavailability of the compound of interest and the suitability of the formulated nanocarrier as a delivery vehicle. The effective release of a bioactive compound as well as its diffusion through the skin are essential in order to obtain the desired biological effects. In the present study the aim was to compare two different nanocarriers namely the nanoemulsions (NE) and the respective nanoemulsion-filled hydrogels (NE/HG) as delivery vehicles of cannabidiol (CBD) since the structural difference of these two formulations could be expected to have impact on their ability to release and deliver the bioactive substance into/through the skin [75]. In spite the fact that in vitro release studies using synthetic membranes are very popular due to the ease of application, more than often the obtained results cannot correlate with the in vivo performance or cannot detect possible differences in release due to different structure [76]. For that reason, in the present study the described in vitro permeation protocol was implemented using modified Franz diffusion cells and full-thickness porcine ear skin as the model membrane. The experiment was conducted under infinite dose regiment despite the fact that the amount of applied formulation is significantly higher than the actual amount applied by patients. The infinite dose experiments are selected since they allow to determine the maximum flux of the CBD from a formulation and therefore enable to parametrically compare investigated formulations.

The amount of CBD permeated through the full-thickness porcine ear skin in function of time is depicted in figure 8 Interestingly, no CBD has permeated through the porcine skin at the beginning of the test as can be observed by the diagram in figure 8. Specifically, CBD was initially detected in the acceptor compartment after the 24th and 22nd hours of administration of the NE and the NE/HG, respectively. Before these time points, CBD concentration was below the LC/MS-MS quantification limit. In the later sampling time points, a higher permeation of CBD was observed from NE/HG compared to NE. In this regard, it is important to emphasize that donor compartments were covered with Parafilm® in order to prevent an extensive evaporation of water from the formulations during 30 h.

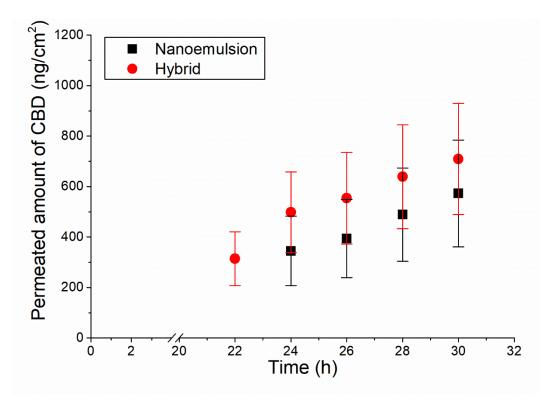


Figure 8. Ex vivo permeation profiles of cannabidiol (CBD) determined across the full-thickness pig ear skin (mean \pm SD, n=6) reflecting the influence of differences in nanocarrier type on the ex vivo skin absorption of CBD.

As presented in table 4. the hybrid systems demonstrated higher permeation rate and permeation coefficient which provides an indication of enhanced transdermal absorption. By the end of the 30 h incubation time, the cumulative amount of CBD that was detected in the acceptor compartment of the cells where hybrids were applied was also higher compared to the NE ones.

Table 4. Ex vivo skin penetration data for the different investigated colloidal systems after 30 h of application (mean \pm SD, n=6).

	steady-state flux (μg/cm²h ⁻¹)	Q _{30h} Cumulative amount of CBD permeated through the skin (μg/cm)	Permeation coefficient (mg/cm²h)
Nanoemulsions	403.37 ± 130.09	572.56 ± 211.11	80.67 ± 211.11
Hybrids	475.68 ± 162.57	709.22 ± 220.62	95.14 ± 32.51

These results are expected since it is well-known that chitosan acts as penetration enhancer due to its interaction with the SC in various ways including the modification of the

protein structure of the SC and acting as a moisturizing agent increasing the SCs' water content [77–79]. In addition, skin occlusion also enhances the percutaneous absorption, especially for very lipophilic substances such as CBD [80]. This effect is proportional to the duration of occlusion hence the penetration of the bioactive compounds is enhanced with increasing duration of occlusion [81]. Skin occlusion leads to increased skin hydration, while extended skin hydration (> 8h) causes a swelling of corneocytes, creating inter-corneocyte ruptures, and causes microstructural changes in lipid self-assembly, leading to changes in skin barrier properties [82–84].

Therefore, the obtained results indicated that the NE/HG itself contributed to the improved penetration of CBD. However, it is important to emphasize that no statistically significant difference was found regarding monitored permeation parameters, probably due to high variability of obtained results. The high variation is highly expected and is frequently reported in the literature, since porcine ear skin was used as the biological membrane. It is well known that it is difficult to control the breeding and diet conditions of pigs, which can affect, inter alia, the skin properties (the ears are obtained from local abattoir immediately after slaughter).

3.10. Tape stripping experiments

As aforementioned the presence of chitosan in the proposed NE/HG used in permeation studies can promote a reversible disruption of the skin layers and long-time used for in vitro skin permeation studies is able to promote higher degree of skin hydration and possible changes of this biological material reducing its barrier function. Therefore, preliminary tape stripping experiments were performed on porcine ear (skin remained on the cartilage during the experiment) to investigate penetration of CBD into the skin after shorter exposure time, utilizing the same infinite dose setting. The samples were obtained after 2 h of application of the two examined formulations and the CBD amount was quantified using LC-MS/MS technique. In order to achieve more accurate results five independent experiments were performed starting with a preliminary study which was necessary for the determination of the number of strips required for the complete removal of CBD localized in the SC. It is important to mention that the first two tapes removed had to be discarded due to the very high amount of CBD detected in these samples. This phenomenon was especially observed on the area where the hybrids were applied which could be attributed to the bioadhesive properties of chitosan which result in the system remaining on the skin surface even after its removal with the cotton swabs [85].

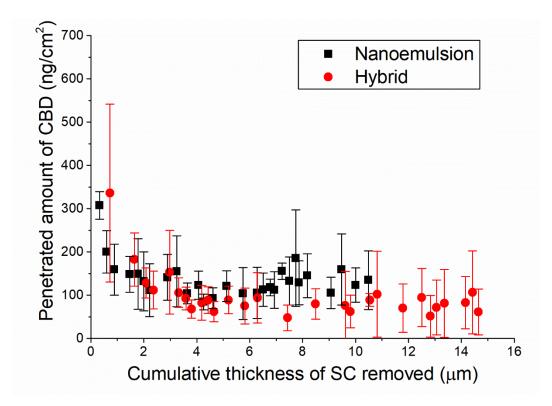


Figure 9. Diagram representing the stratum corneum concentration of CBD in relation to the cumulative thickness of stratum corneum removed profiles following a 2h application of tested samples (mean \pm SE, n=4).

The total amount of CBD recovered in the SC for NE (3853.20 \pm 543.06 ng/cm²) was slightly higher than for NE/HG (2745.41 \pm 1022.44 ng/cm²), but no statistically significant difference was found, despite the remarkable differences in viscosity between the tested formulations. However, as the graph demonstrates (figure 9), there is obvious difference regarding the achieved penetration depth of CBD between the tested formulations. In other words, significantly higher amount of SC was removed with 28 tapes after application of the NE/HG (14.644 \pm 8.338 μ m) compared to the NE (10.493 \pm 3.958 μ m). It may be assumed that the cohesion of the corneocytes might have been decreased after application of the NE/HG as aforementioned in the previous paragraph due to the chitosan interaction with the SC. It has been mentioned in previous studies that chitosan demonstrates permeation-promoting properties through diverse mechanisms including the alteration of the conformation of keratins reducing the cohesion of the SC, the action on the tight junctions and the increase of water content [81,86]. Additionally, comparing this finding with in vitro permeation results, it appears that skin occlusion potentiated the observed effect, leading to improved CBD delivery during the prolonged test duration. These results indicate a rapid and high permeation

through the SC suggesting that the proposed nanocarriers are suitable for the transdermal delivery of CBD with the hybrid system demonstrated enhanced properties as a delivery system for percutaneous delivery. In order to confirm the relevance of the obtained data, additional experiments utilizing finite dose setting, representative for common in-use conditions, are needed.

4. Conclusion

The goal of the present research was to develop a novel delivery system for the effective transdermal administration of lipophilic bioactive compounds that would avoid topical irritation caused by burst release. For that reason, O/W nanoemulsions and the corresponding nanoemulsion-filled chitosan hydrogels were successfully developed as encapsulation media of CBD. The proposed formulation method could be widely used since it is simple and rapid in contrast to other previously reported methods [53]. Initially a series of O/W nanoemulsions were formulated and studied using DLS to evaluate their droplet size, size distribution, and stability in relation to their composition and preparation method. Monodispersed and stable for three months, nanoemulsions containing EVOO in the oil phase were selected as more suitable for the encapsulation of CBD. These nanoemulsions were prepared with low energy input in the absence of co-solvent or any permeation enhancers. Then, CBD-loaded nanoemulsions were added in chitosan hydrogels and the resulting hybrid formulations were structurally characterized by means of EPR, SAXS, confocal microscopy and cryo-EM. A crucial part of the present study was the elucidation of the structure of the formulated systems since it has great impact on the release of the encapsulated bioactive compound [87-89]. Specifically, this is the first study having in the centre of attention the investigation of possible structural changes after the incorporation of the nanoemulsion into the hydrogel that could in turn lead to differences in the release of the encapsulated CBD. Experimental results from EPR spin probing showed that membrane dynamics in terms of local viscosity, order and polarity in nanoemulsion-filled hydrogels where similar to the corresponding nanoemulsions in the absence of CBD. However, the observed changes in rotational correlation time values between the CBD-loaded nanoemulsions and nanoemulsion-filled hydrogels provide an indication that the hybrid systems could release more effectively the encapsulated compound. Confocal micrographs showed the existence of spherical oil droplets of comparable sizes in both formulations. Empty and CBD-loaded nanoemulsions characterized via cryo-EM showed spherical structures and droplet diameters

 that agreed well with the DLS results. SAXS results also indicated that the spherical structure and the size of the nanoemulsions were not affected by their incorporation in the hydrogel.

Subsequently, cell viability was assessed in the absence and presence of CBD indicating that both the nanoemulsions and the hybrid hydrogels present high biocompatibility and possibly dose-dependent induction of cell proliferation. It is worth noting that the developed nanocarriers were administered at very high concentrations compared to previous similar systems that have been reported in the literature without showing the corresponding cytotoxicity [53,90]. These results are expected due to the low surfactant concentration used for the formulation of the systems since it is widely known that various synthetic surfactants could present dose-dependent toxicity [91]. Furthermore, the addition of extra virgin olive oil as component of the oil phase also enhances the biocompatibility of the systems since its health promoting properties are established [70,92]. Finally, to evaluate the suitability of the nanoemulsions and the hybrid systems as vehicles for the transdermal delivery of CBD ex vivo evaluation was performed using porcine ear skin as the model membrane. The bioactive compound was detected in the acceptor medium of Franz diffusion cells first from the hybrid systems, followed by the nanoemulsions and sustained release was observed from both the nanocarriers under study during the 30 h of the experiment. Additionally, differential tape stripping technique demonstrated high permeability of CBD through the stratum corneum with the hybrid system exhibiting enhanced penetration possibly due to change of the structure of the stratum corneum in the presence of chitosan. Hence, the present study could be a new approach for the development of effective and at the same time non-toxic transdermal delivery vehicles since it suggests that the presence of penetration enhancers, alcohols or high concentrations of co-surfactants or other possibly toxic components is not necessary to obtain effective transdermal delivery of a strongly lipophilic substance such as CBD [93–96].

Therefore, these systems could be ideal candidates for use as carriers for (trans)dermal delivery of highly lipophilic bioactive compounds. In the future, their ability to simultaneously encapsulate lipophilic and hydrophilic bioactive substances in order to obtain synergistic action and improved biological effects remains to be studied.

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