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# Hydrogen retention in ITER's Diagnostic First Wall submitted to cyclic thermomechanical loadings

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#### Abstract

A simplified ITER's Diagnostic First Wall is modelled in Abaqus Finite Element software to evaluate hydrogen retention after cyclic plasma exposure, and especially, the impact of mechanical fields. Using several User Subroutines, a strongly coupled chemothermo-mechanical problem is set and solved. Hydrogen field results with or without mechanical fields are compared, during the different loading phases, leading to the conclusion that, for this simplified problem, mechanical fields have no important impact on hydrogen retention.

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Keywords: Hydrogen; Diffusion; Trapping; Thermomechanic; Finite Element; Abaqus

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#### 1. Introduction

ITER (International Thermonuclear Experimental Reactor) is an international nuclear fusion research and engineering project aimed at replicating the fusion processes of the Sun to produce energy on Earth. The largest nuclear fusion reactor in the world, the ITER tokamak, is under construction.

ITER's Diagnostic First Wall (DFW) structure (Loesser et al., 2017) is composed of structural elements intended to support plasma diagnostic devices and protective panels of the vacuum chamber. These are made of 316L steel, and are subjected to cycles of both heat and particle fluxes (including tritium, a radioactive isotope of hydrogen). These particles may diffuse in the structure and get trapped by material defects (vacancies, dislocations, ...), assisted by thermo-mechanical fields.

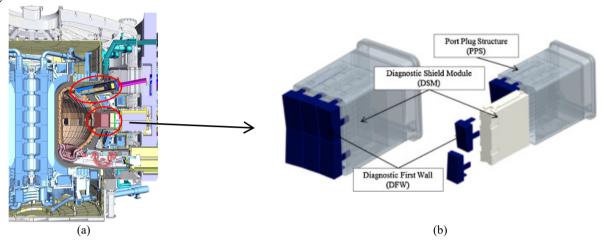
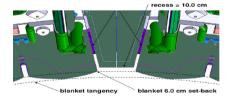


Fig. 1. (a) ITER DFW location (reproduced from (Loesser et al., 2017)); (b) DFW (reproduced from (Giacomin et al., 2015))

Estimating the amount of tritium contained in DFW is mandatory for safety issues and component reliability (e.g., to avoid any Hydrogen Embrittlement-related phenomenon), especially to assess tritium permeation fluxes towards the DFW's cooling pipes, linked to a pollution risk.



e been performed on this component (Khodak et al., 2017; Smith et al., sis has yet been provided related to tritium diffusion and trapping in the al and mechanical fields (Benannoune et al., 2020).

agus Finite Element software, tritium transport through a very simplified

re equations are presented, as well as the modelling strategy. Results are

#### MUNICHCIALUIC diffusive hydrogen concentration Α elastic limit $C_T$ trapped hydrogen concentration В hardening constant interstitial sites density Cstrain rate constant trap density hardening exponent ninterstitial site occupancy thermal softening exponent mtrap site occupancy equivalent plastic strain $\varepsilon_n$ density yield stress specific heat $T_{room}$ room temperature temperature T<sub>melt</sub> melting temperature

| λ                           | thermal conductivity          | $R_p$           | particle implantation depth    |
|-----------------------------|-------------------------------|-----------------|--------------------------------|
| $oldsymbol{arepsilon}_{th}$ | thermal strain tensor         | $\varphi_{imp}$ | implantation flux of hydrogen  |
| α                           | thermal expansion coefficient | $D_L$           | hydrogen diffusion coefficient |
| I                           | identity tensor               | $C_{L0}$        | hydrogen boundary condition    |
|                             |                               |                 |                                |

### 2. Constitutive equations

Several physical phenomena are involved in the problem resolution: hydrogen transport and trapping, mechanical behavior, and heat transfer, all being coupled. In the following, hydrogen is considered instead of tritium.

# 2.1. Hydrogen transport

It is assumed that the total hydrogen concentration can be split into a diffusive part,  $C_L$ , and a trapped one,  $C_T$ . If several kinds of traps are involved,  $C_T = \sum_i C_{T,i}$ , i being the trap kind number. These concentrations are related to trap densities  $N_{T,i}$  (or interstitial site density  $N_L$ ) by the occupancy  $\theta_{T,i} \in [0,1]$  (denoted  $\theta_L$  for diffusive hydrogen):

$$\begin{cases}
C_L = N_L \theta_L \\
C_T = \sum_i C_{T,i} = \sum_i N_{T,i} \theta_{T,i}
\end{cases}$$
(1)

Coupling mechanically-assisted hydrogen flux  $\varphi_L$  (based on Fick's law) (Bockris et al., 1971; Li et al., 1966) and mass conservation yields the diffusion and trapping equation (Sofronis and McMeeking, 1989)

$$\frac{\partial C_L}{\partial t} + \sum_i \frac{\partial C_{T,i}}{\partial t} = \nabla \cdot \left( D_L \nabla C_L + D_L C_L \frac{V_H}{RT} \nabla P_H \right)$$
 (2)

where R is the ideal gas constant, T the absolute temperature,  $D_L$  the hydrogen diffusion coefficient, and  $V_H$  the partial molar volume of hydrogen.  $P_H = -1/3 \text{ tr}(\sigma)$  is the hydrostatic stress, where  $\sigma$  is the stress tensor. The temporal evolution of the trapped hydrogen concentration (for each trap) can be computed using the so-called McNabb and Foster equation(McNabb and Foster, 1963), for  $C_L \ll N_L$  (or  $\theta_L \ll 1$ )

$$\frac{\partial C_{T,i}}{\partial t} = \frac{k_i}{N_L} C_L \left( N_{T,i} - C_{T,i} \right) - p_i C_{T,i} \tag{3}$$

where  $p_i$  and  $k_i$  are the detrapping and trapping reaction rates.

## 2.2. Heat transfer

The heat transfer equation is

$$\rho C_p \frac{\partial T}{\partial t} = \nabla \cdot (\lambda \nabla T) \tag{4}$$

where  $\rho$  is the density,  $C_p$  the specific heat and  $\lambda$  the thermal conductivity. Any temperature variation induces a thermal expansion tensor such as

$$\boldsymbol{\varepsilon}_{th} = \alpha (T - T_{init}) \boldsymbol{I}$$

$$\boldsymbol{\varepsilon}_{th} = \alpha (T - T_{init}) \boldsymbol{I}$$
(5)

where  $T_{init}$  is the initial temperature, and I is the identity tensor.

# 2.3. Mechanical behavior

Isotropic thermo-elasticity is considered, based on Young modulus E and Poisson ratio  $\nu$ . Isotropic hardening is assumed, based on a Johnson-Cook law (G. R. Johnson and Cook, 1983)

$$\sigma_{Y} = \left( A + B \varepsilon_{p}^{n} \right) \left( 1 - \left( \frac{T - T_{room}}{T - T_{room}} \right)^{n} \right)$$

$$\left( 1 - \left( \frac{T - T_{room}}{T_{melt} - T_{room}} \right)^{n} \right)$$

$$(6)$$

where  $\sigma_Y$  represents the yield stress and  $\varepsilon_p$  the equivalent plastic strain.  $T_{room}$  and  $T_{melt}$  represent respectively the room and melting temperatures (assumed to be equal to 300 and 1356 K respectively). Last, A, B, n and m are material parameters.

#### 2.4. Material parameters

All parameters are extracted from literature.  $N_L$  and  $D_L$  can be found in (Penzhorn et al., 2012). One trap has been considered, with a trapping energy equal to 0.7 eV (Guillermain, 2016), and a density such that  $\log N_T = 25.26 - 2.33e^{-5.5\epsilon_p}$  site/m³ (adapted from (Kumnick and H. H. Johnson, 1980) to account for an initial dislocation density equal to  $10^{12}$ - $10^{14}$  m/m³). Heat transfer parameters have been taken from (Umbrello et al., 2007), with an temperature-independent expansion coefficient (Sayman et al., 2009). Last, thermomechanical parameters have been extracted from (Chandrasekaran et al., 2005; Umbrello et al., 2007).

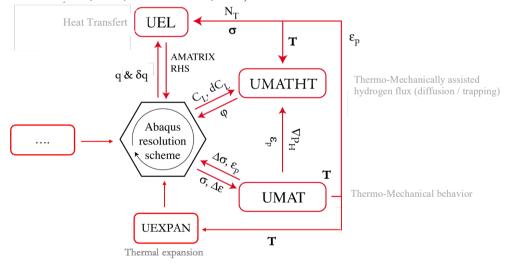


Fig. 2 Implementation scheme of a strongly coupled thermo-chemo-mechanical problem in Abaqus software.

#### 3. Implementation

The chemo-thermo-mechanical system described in the three paragraphs is resolved in a fully coupled way (strong coupling) using the Abaqus Finite Element software (Simulia, 2011), based on the 'coupled temp-displacement' procedure and user subroutines (see (Charles et al., 2017; Vasikaran et al., 2020) for details):

- UMAT subroutine: thermomechanical behavior (section 2.3);
- UMATHT subroutine: hydrogen transport and trapping (section 2.1);
- UEL subroutine: heat transfer (equation (4));
- UEXPAN subroutine: thermal expansion (equation (5)).

The interactions of the different subroutines are depicted on Fig. 2.

# 4. Model definition

The geometry of the DFW is presented in Fig. 3(a). For the sake of simplicity, only a part of this geometry will be modelled (indicated by a red rectangle). Each cylindrical conduit represents a tube for the cooling fluid.

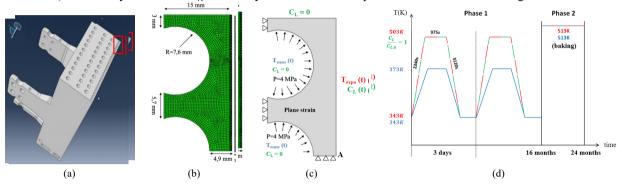


Fig. 3. (a) DFW and the part modelled (in a red rectangle), (b) mesh and dimension, (c) applied boundary conditions. (d) Loading scenario (temperature & hydrogen)

The part of interest is modelled in 2D and meshed with just under 4000 fully integrated linear elements (Fig. 3b). Symmetry boundary conditions are imposed on the lower and left edges (Fig. 3c), and plane strain is assumed. A water flow circulating in the cooling tubes induces a constant pressure on the DFW of 4 MPa. On the plasma-exposed face, heat and hydrogen loading cycles are imposed, while on the upper surface and on the cooling tubes, an instantaneous hydrogen draining is assumed ( $C_L = 0$ ).

The scenario chosen is illustrated in Fig. 3d. It consists of 160 loading cycles of 3 days each (Phase 1), and an annealing cycle of 8 months (Phase 2), at constant temperature and without plasma exposure. Each 3-day cycle corresponds to the temporal connectantian even 2 days of 12 daily places of 400 seconds. This cycle is repeated Phase 1 Phase 2 for 16 months (160 cycles). Hy n  $C_{L0}$  on the exposure surface, such as in (Hodille et al 503K (baking)  $C_{L0} = \frac{R_p \varphi_{imp}}{D_L} = 1.46 \times$ (7) 2340s 373K where  $\varphi_{imp}$  represents the h plantation depth (set to 13 nm (Benannoune et al., 202 343K 5. Results 343K Three calculation T<sub>water</sub>

erted by the water Hydrogen → time with only the pressure 3 days 16 months exerted by the water -

The diffusion fields at the end of Phase 1 are given in Fig. 4 for the three cases. It can be observed that, while the cooling pipes-induced pressure increases the level of  $C_L$ , at least near the exposed surface, the depth of  $C_L$ 's penetration seems not affected by mechanical fields. This impact of the stress field on  $C_L$  levels is linked to the hydrostatic pressure

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effect on hydrogen flux, which affects the maximal  $C_L$  level. To detail that effect, in Fig. 5 are plotted the  $P_H$  distributions when only the cooling pipes pressure is considered (Fig. 5a) and with all the mechanical fields, at the

higher and lower exposition temperatures (Fig. 5b and Fig. 5c respectively).

5. Results

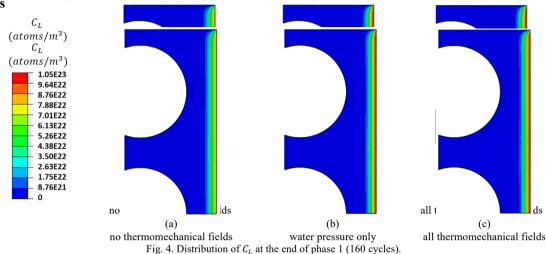
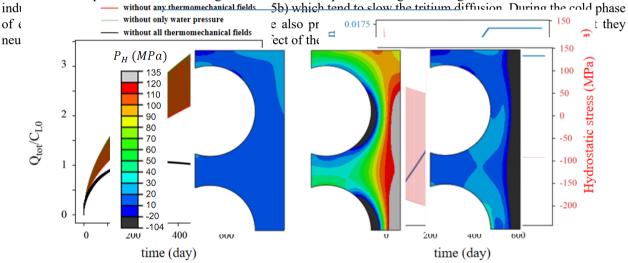


Fig 4. Distribution of  $C_L/C_{L0}$  at the end of phase 1

When only the water pressure is accounted for, it can be seen that the DFW section is slightly under dilatation (Fingetival pressure), the distribution considering and the pressure as a testing the highest the pressure as a testing to the highest the pressure as a testing to the highest the pressure in the pressure of the highest the pressure in the pressure in the pressure of the highest the pressure in the pr



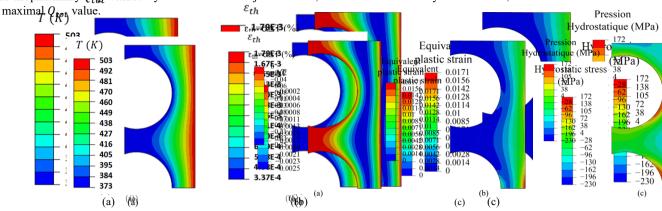
(a) (b) all thermomechanical fields Fig. 5. Distribution of  $P_H$  at the end of phase 1's last cycle (160 cycles) considering (a) water pressure only and (b-c) all thermomechanical fields at higher and lower exposition temperature  $T_{expo}$  respectively.

The impact of mechanical fields on hydrogen retention can be investigated by introducing its total amount per unit thickness (for a 2D configuration)  $Q_{tot} = \int (C_L + C_T) dV$  in the DFW section;  $Q_{tot}$  evolution with time is plotted in Fig. 7a for the three cases.

The impact of the cooling pipes pressure is here very clear: at each cycle,  $Q_{tot}$  increases continuously, exhibiting cycles corresponding to the plasma loading ones. When the DFW section is exposed to both hydrogen and temperature,  $Q_{tot}$  increases. Hydrogen amount decreases as soon as these boundary conditions are modified, during phase 2 (no

The impact of mechanical fields on hydrogen retention can be investigated by introducing its total amount  $Q_{tot} = \int (C_L + C_T) dV$  in the DFW section;  $Q_{tot}^{inhae} \theta'_{tot}^{inhae} \theta'_$ 

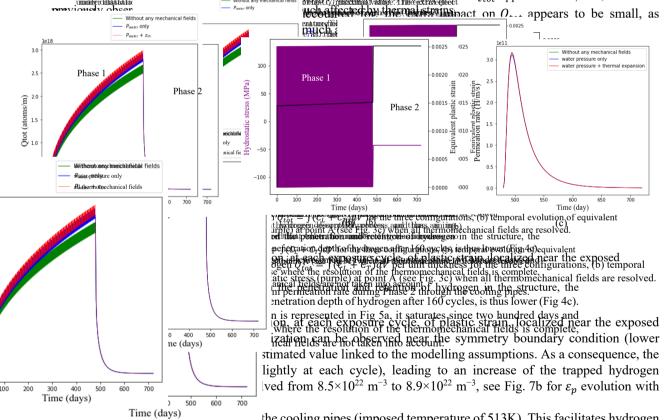
The impact of the cooling pipes pressure is here very clear: at each cycle,  $Q_{tot}$  increases continuously, exhibiting cycles corresponding the loading ones. When the DFW section is exposed to both hydrogen and temperature,  $Q_{tot}$  more leading ones. When the DFW section is exposed to both hydrogen and temperature,  $Q_{tot}$  more leading ones. When the DFW section is exposed to both hydrogen and temperature,  $Q_{tot}$  more leading ones. When the behavior indicates are important hydrogen and temperature process, and ydrogen comportant hydrogen and distributed and the roots of the following action of the cooling process of the process of the



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Fig. 6. (AFTERMINITERMI

Fig. 6. (a) Temperature and (b) thermal strain fields and (c) equivalent plastic strain at the end of Phase 1's last cycle, at the higher  $T_{expo}$  value. When the condition is the condition is the condition of the condition is the condition of the condition is the condition of the condition



the cooling pipes (imposed temperature of 513K). This facilitates hydrogen diffusion and detrapping. In a stress-free configuration, H transport is only driven by traps and  $C_L$  gradients, while mechanical fields (and especially compression) hardly affect the global desorption process (Fig. 7c).

#### 6. Conclusion

It has been shown that the mechanical fields, induced by both plane strain, symmetry assumptions, and thermal expansion, have a slight effect on hydrogen fields during the different loading phases of the DFW section, and thus, on hydrogen retention. It has also been seen that boundary conditions induce plastic strain localization in their vicinity, leading to a needed interrogation on their relevance; this localization, however, has no influence on the tritium retention or the desorption flux in the DFW section.

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