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1 **X-ray structure and mechanism of ZgHAD, a L-2-haloacid**
2 **dehalogenase from the marine Flavobacterium *Zobellia***
3 ***galactanivorans***

4
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13 *running title : Structure and mechanism of ZgHAD*

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25 **ABSTRACT**

26 Haloacid dehalogenases are potentially involved in bioremediation of contaminated
27 environments and few have been biochemically characterized from marine organisms. The L-
28 2-haloacid dehalogenase (L-2-HAD) from the marine *Bacteroidetes Zobellia galactanivorans*
29 Dsij^T (ZgHAD) has been shown to catalyze the dehalogenation of C2 and C3 short-chain L-2-
30 haloalkanoic acids. To better understand its catalytic properties, its enzymatic stability, active
31 site and 3D structure were analyzed. ZgHAD demonstrates high stability to solvents and a
32 conserved catalytic activity when heated up to 60°C, its melting temperature being at 65°C. The
33 X-ray structure of the recombinant enzyme was solved by molecular replacement. The enzyme
34 folds as a homodimer and its active site is very similar to DehRhb, the other known L-2-HAD
35 from a marine *Rhodobacteraceae*. Marked differences are present in the putative substrate
36 entrance sites of the two enzymes. The H179 amino acid potentially involved in the activation
37 of a catalytic water molecule was confirmed as catalytic amino acid through the production of
38 two inactive site-directed mutants. The crystal packing of 13 dimers in the asymmetric unit of
39 an active-site mutant, ZgHAD-H179N, reveals domain movements of the monomeric subunits
40 relative to each other. The involvement of a catalytic His/Glu dyad and substrate binding amino
41 acids was further confirmed by computational docking. All together our results give new
42 insights into the catalytic mechanism of the group of marine L-2-HAD.

43

44 **Keywords:** L-2-haloacid dehalogenase, marine *Bacteroidetes, Zobellia galactanivorans*,
45 crystal structure, catalytic mechanism, computational docking, His/Glu dyad

46

47 **50-75 words statement outlining the importance and impact of the work presented**

48 The present work on the first characterized *Bacteroidetes* haloacid dehalogenase provides
49 additional knowledge on marine L-2-HAD structure/function characteristics of the

50 monophyletic group B. The 3D crystal structure reveals specificities related to unique catalytic
51 features, such as a catalytic His/Glu dyad, as compared to known L-2-HADs from soil
52 organisms of the monophyletic group A. An unprecedented helical arrangement of 13 dimers
53 observed within one crystal highlights the presence of possible movements between monomeric
54 units.

55

56 **INTRODUCTION**

57 The haloacid dehalogenases (HAD) superfamily includes dehalogenating enzymes together
58 with diverse enzymes that hydrolyze carbon-phosphorus bonds, such as epoxide hydrolases,
59 phosphatases, phosphomutases, or nucleotidases. Widely present among living organisms, they
60 are involved in a variety of cellular processes ranging from amino acid biosynthesis to
61 detoxification ([Burroughs et al., 2006](#)). Since the industrial boom, halogenated xenobiotic
62 pollutants are contaminating soils and aquatic environments. The accumulation of those toxic
63 compounds led to the research of new tools for detoxification and bioremediation.

64 The “true” dehalogenases of the HAD superfamily are classified into four types relative to the
65 substrate specificity and stereoselectivity. D-2-haloacid dehalogenases (D-2-HADs) and L-2-
66 haloacid dehalogenases (L-2-HADs) have a strict enantioselective dehalogenating activity on
67 D-2-haloacids and L-2- haloacids, respectively, to produce the corresponding alcohols with an
68 inverted chirality. The two other types are DL-2-haloacid dehalogenases (DL-2-HADs) which
69 accept both D- or L-2-haloacids as substrates: DL-2-HAD_i act with a configuration-inverting
70 mechanism whereas DL-2-HAD_r retain the configuration. The 2-haloacid dehalogenases are
71 also categorized into two groups according to their amino acid sequence homology. D-2-HADs
72 (EC 3.8.1.9) and DL-2-HADs (EC 3.8.1.10 and EC 3.8.1.11) are part of Group I and L-2-HADs
73 (E.C. 3.8.1.2) belong to Group II ([Wang et al., 2021](#); [Ang et al., 2018](#)).

74 In addition, L-2-HADs were recently classified on the basis of phylogenetic/environment
75 analyses as two monophyletic groups, where the group A contains a mix of terrestrial and
76 marine sequences and the group B includes mostly marine sequences (Grigorian *et al.*, 2021).
77 There are currently nine available crystal structures of L-2-HADs from diverse organisms,
78 mainly bacteria. Four of these 3D structures belong to the group A and were solved before 2020.
79 They correspond to L-DEX YL from *Pseudomonas* sp. YL (Hisano *et al.*, 1996), Dh1B from
80 *Xanthobacter autotrophicus* GJ10 (Ridder *et al.*, 1997), DehIVa from *Burkholderia cepacia*
81 (Schmidberger *et al.*, 2007) and DehSft from *Sulfolobus tokodaii* (Rye *et al.*, 2009). Four other
82 crystal structures were determined very recently and they led to the discovery of the first
83 defluorinating L-2-HAD enzymes for two of them, named Bpro0530 from *Polaromonas* sp.
84 JS666 and Rha0230 from *Rhodococcus* sp. RHA1 isolated from polluted environments (Chan
85 *et al.*, 2021). All these eight characterized L-2-HADs originate from terrestrial bacteria. The
86 last L-2-HAD characterized structure was DehRhb from a marine *Rhodobacteraceae* sp.
87 (Novak *et al.*, 2013) and the only one relative to the phylogenetic group B (Grigorian *et al.*,
88 2021).

89 All the biochemically and structurally described L-2-HAD enzymes to date are dimers with two
90 domains in each subunit. The L-2-HADs consist in a characteristic core domain with a
91 conserved alpha/beta hydrolase fold, similar to the “Rossmann-fold”, and a second small cap
92 domain exhibiting varying folds and functions. This cap domain is responsible for the
93 biochemical diversification within the HAD superfamily (Lahiri *et al.*, 2004). These enzymes
94 transform the substrate(s) molecule(s) according to a conserved nucleophilic substitution
95 involving a conserved aspartic acid that forms an intermediate ester bond with the substrate. In
96 the case of L-2-HADs, the enzyme-substrate ester bond is then hydrolyzed by another
97 nucleophilic attack with an activated water molecule (Liu *et al.*, 1995; Nardi-Dei *et al.*, 1997).
98 In contrast to other HAD superfamily enzymes, the cap domain is similar in all L-2-HADs and

99 is composed of a four-helix bundle where the active site is flanked by a hydrophobic cavity
100 situated in between the core and cap domains (Ridder *et al.*, 1997; Schmidberger *et al.*, 2007;
101 Rye *et al.*, 2009).

102 The complete genome sequence of the marine flavobacteria *Zobellia galactanivorans* Dsij^T
103 revealed the presence of a L-2-HAD enzyme, further on named ZgHAD. Subsequent to cloning
104 and purification, the biochemical characterization of the recombinant ZgHAD enzyme has been
105 described recently (Grigorian *et al.*, 2021). The enzyme is specific towards L-2-enantiomer
106 substrates having a short carbon chain (C2 and C3) and it can deiodinate, debrominate or
107 dechlorinate the α -carbon position. The highest activity was observed with iodoacetic and
108 bromoacetic acids, while the reactions with chloroacetic and L-2-bromopropionic acids were
109 much lower, suggesting catalytic specificities when compared to the other characterized marine
110 L-2-HAD (Grigorian *et al.*, 2021).

111 Here, we describe the 3D crystal structure of wild-type ZgHAD, as well as that of two point-
112 mutated enzymes, ZgHAD-H179A and ZgHAD-H179N, and, together with computational
113 docking we compare the observed structural details with those of related enzymes, previously
114 described. The thermal and solvent stabilities of this enzyme have also been assessed in view
115 of potential biotechnological applications.

116

117 **RESULTS**

118 **Thermostability and solvent stability**

119 A temperature gradient between 20°C and 95°C was applied to evaluate the stability zone of
120 the protein fold. The experiment showed that ZgHAD was denatured between 60 and 70°C with
121 a melting temperature of 65°C (Fig. 1A), as also confirmed by Dynamic Light Scattering (DLS)
122 measurements (Fig. S1). These results suggest that ZgHAD is thermostable up to the maximal
123 temperature of 65°C. The thermal stability of the enzymatic activity was also investigated

124 between 10°C and 90°C and after 30 minutes of enzyme incubation. The residual activity was
125 measured and plotted as a percentage of the initial activity (**Fig. 1B**). ZgHAD activity was found
126 to be stable up to 50°C but decreased rapidly between 60°C and 70°C, from 80% residual
127 activity to a complete loss of activity. The results on protein denaturation and enzymatic activity
128 are strongly correlated and show that ZgHAD turns completely unfolded and therefore inactive
129 at 70°C.

130 The solvent stability of ZgHAD was assessed by incubating the enzyme with different
131 concentrations of ethanol, methanol, acetonitrile and dimethyl sulfoxide (DMSO) for one hour
132 before evaluating the residual activity (**Fig. 1C**). The effects of ethanol and methanol treatments
133 were very similar. The activity of the enzyme was conserved up to 80%, when 10 to 20% of
134 these organic solvents were added, and decreased to 20% in presence of 20% to 40% of these
135 alcohols. Above 40% of methanol or ethanol, the activity was almost completely inhibited.
136 DMSO had the least inhibitory effect of all the solvents tested. Only 20% loss of activity was
137 observed after addition of up to 40% DMSO. The activity decreased to 60% with 50% of
138 DMSO, and higher concentrations drastically reduced the activity to 10% of the control without
139 any solvent. In acetonitrile, the activity of ZgHAD was drastically reduced to 20% in the
140 presence of 10% of this solvent. It was completely inhibited between 20 and 30% of acetonitrile,
141 and surprisingly it increased again for higher concentrations tested, with a stabilization around
142 20% of residual activity in the presence of 60 to 80% of acetonitrile.

143

144 **Overall structure of ZgHAD**

145 The crystal structure of ZgHAD was solved by the molecular replacement method at 1.6 Å,
146 using the closest structural representative, DehRhb (PDB accession: 2YML) from a marine
147 *Rhodobacteraceae*. The two proteins share 31% identity and 50% similarity of amino acid

148 sequences. ZgHAD crystallized with the space group $P2_12_12_1$ and the mutant ZgHAD_H179N
149 with space group $P2_1$; unit cell parameters are reported in **Table 1**.

150 ZgHAD crystallized as a homodimer and, likewise all reported HAD enzymes, the monomeric
151 subunit is composed of two domains, comprising a core domain, formed by the residues 18-30
152 and 110-238, and a cap domain, formed by the residues 31-109 (**Fig. 2A**). The core domain has
153 a typical “Rossmann fold” which consists of six parallel β -strands surrounded by five α -helices
154 and three 3_{10} helices. Overall, the connectivity follows the pattern β -strand - α -helix - β -strand,
155 except for β -strands 5 and 6, which are connected by a β -turn. 3_{10} helices are found before and
156 after strand $\beta 3$. The cap domain is composed of four α -helices, and a 3_{10} helix is inserted in the
157 core domain between strand $\beta 1$ and helix $\alpha 5$. The active site is located between the core and the
158 cap domains, right after strand $\beta 1$. The four helices of the cap domain shield the top of the
159 active site cavity from the solvent (**Fig. 2A**). The core and the cap domains of ZgHAD are
160 similar to those of DehRhb (2YML), DehSft (2W43), DehIVa (2NO4) and L-DEX YL (1ZRM)
161 with rmsd values between matching $C\alpha$ positions of 1.21 Å over 212, 1.65 Å over 194, 1.77 Å
162 over 213 and 1.30 Å over 203 residues, respectively.

163 The homodimer of ZgHAD has the dimensions of 74 x 35 x 44 Å (**Fig. 2B**) and this oligomeric
164 state is in agreement with the estimated size of the protein in solution determined by gel
165 filtration (as referred in [Grigorian et al., 2021](#)). Up to date, all other structurally-characterized
166 L-2-HADs have been reported to occur as homodimers, except for one putative L-2-HAD
167 named PH0459 that has been crystallized in a monomeric state, but no dehalogenase activity
168 was yet reported for this enzyme ([Arai et al., 2006](#)). Similar to these other L-2-HADs, the two
169 subunits are related by a two-fold symmetry axis running nearly parallel to the $\alpha 2$ helix. On
170 dimer formation, 16% of each monomeric subunit’s accessible surface is buried at the interface
171 as identified by PDBePISA server ([Krissinel and Henrick, 2007](#)). In comparison, this value
172 ranges between 13.4% in DehSft to 19% in Dh1B. The subunit interface of ZgHAD is mainly

173 formed by helices $\alpha 2$ and $\alpha 3$. In addition, a salt bridge between glutamate E53 of helix $\alpha 2$ of
174 one monomeric subunit and arginine R188 of helix $\alpha 8$ of the other (and vice versa) reinforces
175 the interaction between the two monomeric subunits.

176

177 Active site of ZgHAD

178 While 69% of the primary amino acid sequence differ between DehRhb and ZgHAD, all the
179 amino acids of their active sites are conserved, except for one amino acid that changes from a
180 serine (S120) in ZgHAD to a threonine (T124) in DehRhb (**Fig. 3A**). This corresponds to a
181 minor difference, as both amino acids are exchangeable and conservative with respect to their
182 functional group. Previous site-directed mutagenesis performed on other L-2-HADs has shown
183 that nine conserved amino acids are essential for catalytic activity ([Kurihara *et al.*, 1995](#); [Pang
184 and Tsang, 2001](#); [Nakamura *et al.*, 2009](#); [Adamu *et al.*, 2016](#)). Among those nine essential
185 amino acids, three are different in ZgHAD compared to the terrestrial L-2-HADs, as shown by
186 the sequence alignment of structurally characterized L-2-HADs (**Fig. S2** and **Table S1**). The
187 positively charged arginine that binds and stabilizes the halide ion in the active site of Dh1B
188 (R39) and L-DEX YL (R42) ([Ridder *et al.*, 1997](#); [Kondo *et al.*, 2014](#)) is replaced by a non-polar
189 phenylalanine (F43) in ZgHAD. In L-DEX YL and DehIVa, a serine residue (S175 and S176,
190 respectively) is described to form a hydrogen bond with the catalytic aspartate (D10 and D11,
191 respectively) to maintain a suitable orientation of its carboxyl group for the nucleophilic attack
192 on the substrate ([Hisano *et al.*, 1996](#); [Schmidberger *et al.*, 2007](#)). In ZgHAD and DehRhb, this
193 serine is replaced by an alanine (A177) that cannot bind with the catalytic D14. In contrast, a
194 threonine (T18) and a lysine (K153) form hydrogen bonds with the carboxyl group of the
195 catalytic aspartate (D14), respectively. In DehRhb, a histidine (H183) was proposed to
196 participate to the activation of the catalytic water molecule instead of the conserved asparagine
197 in other characterized L-2-HADs ([Novak *et al.*, 2013](#)). Similar to DehRhb, ZgHAD possesses

198 a potential catalytic histidine at position 179 as part of the hydrophobic pocket around the active
199 site. This histidine (H179) has been changed to alanine and asparagine by site directed
200 mutagenesis to generate the ZgHAD_H179A and ZgHAD_H179N mutant enzymes leading to
201 the loss of dehalogenase activity. To analyze the potential structural rearrangements due to these
202 mutations, we have also crystallized and solved the crystal structures of both ZgHAD_H179A
203 and ZgHAD_H179N. No major structural differences were observed for the H179N and H179A
204 mutants, except for the position of glutamate E17. In both cases, the main chain of E17 moves
205 by 1 to 2 Å towards the active site and the side chain displays an alternative conformer that
206 moves the carboxyl group 6 Å closer to the catalytic residues, which may possibly interfere
207 with substrate binding or water activation (**Fig. 3B** and **3C**).

208 In the case of the mutant ZgHAD_H179N, the crystal structure revealed 13 dimers displaying
209 a helical arrangement within the asymmetric unit, as shown in **Figure 4A**. Notably, the same
210 space group and large unit cell parameters as for ZgHAD_H179N were also observed for certain
211 wild-type ZgHAD crystals, indicating that this spatial arrangement (dependent on the pH of the
212 crystallization condition) is also possible for the native protein. But due to diffraction at low
213 resolution (i.e. 3.5 - 3.2 Å), these crystals were not investigated further, since better diffracting
214 crystals were obtained for the wild-type protein. For ZgHAD_H179N, when all independent
215 dimers of the asymmetric unit are superimposed based on a single monomeric subunit, these
216 subunits match well with a rmsd between 0.288 and 0.445 (**Fig. 4B**), respectively. However,
217 the other monomeric subunits, which are not included in the superimposition calculations,
218 appear to display different relative positions with respect to the first monomeric subunit, with
219 largest main chain distances of up to 2.7 Å (**Fig. 4C**) for the outer structural elements. This
220 indicates that the dimeric arrangement has some flexibility, allowing a rotational freedom at the
221 interface of the two monomeric subunits.

222

223 Putative substrate binding residues and docking analysis

224 Molecular docking studies of ZgHAD with iodoacetic acid (IAA), bromoacetic acid (BAA),
225 chloroacetic acid (CAA) or L-2-bromopropionic acid (2-BPA) substrate analogs were
226 undertaken using AutoDock Vina (Trott and Olson, 2010). For each docking result, the top
227 ranked position based on affinity score (kcal/mol) was selected as the most likely solution. The
228 binding energy calculated for these substrates is comprised between -5.6 and -6.1 kcal/mol,
229 indicating that they can all be considered as potential substrates for ZgHAD. In these models,
230 residues V15, N16, S120, N121 and K153 are potentially interacting with each docked substrate
231 molecule through hydrogen bonds (Fig. 5).

232 According to Novak *et al.*, 2013, a ‘halogen cradle’ is formed by the side chains of residues
233 F47, I51, F66, N125 and W185 in DehRhB. Four of the corresponding amino acids in ZgHAD
234 (F43, F62, N121 and W181) are conserved and could be similarly involved in stabilizing the
235 halogen atom, as well as the fifth residue, L47, that replaces I51 in a conservative manner.

236 Active site entrances, tunnels and cavities

237 The presence of cavities and potential tunnels/channels in ZgHAD was analyzed using CAVER
238 Analyst 2.0 software (Jurcik *et al.*, 2018). Two putative entry sites were determined on each
239 monomeric subunit, a large and a smaller one, both leading to the catalytic cavity by a short
240 and tight tunnel. The two orifices are on opposite sides of the monomeric subunit but connected
241 (Fig. S3A). The largest entrance has an ellipsoid configuration and was estimated to have an
242 average surface of 74.9 Å² and the smallest an average surface of 66.5 Å² (Fig. S3B and Fig.
243 S3C). The diameters are 2.64 Å and 1.56 Å for the larger and smaller tunnel, respectively. The
244 catalytic cavities present on each monomeric subunit are shown to have a very similar volume
245 of 606.8 Å³ and 606.9 Å³. In the inactive H179A mutant, a reduction of the cavity is observed
246 with approximately 564.2 Å³, corresponding to a decrease of the volume by 7%.

247 The analysis of the amino acid composition around the cavity and along the tunnel allowed to
248 identify two patches of basic (H37, R44, H99, R198, H201 on monomeric subunit A and K76,
249 F77 on monomeric subunit B) and acidic residues (E17, D21, E26, E36, E199), complemented
250 by hydrophobic, neutral and small residues (M22, G23, N27, F39, S40, G200 on monomeric
251 subunit A and L41 on monomeric subunit B) (**Fig. S4A**). Deeper in the tunnel and near the
252 catalytic site, a patch of hydrophobic amino acids (F43, L47, W181) is present and the putative
253 substrate binding residues (V15, N16, S120, N121, K153) are more buried in the core of the
254 cavity. The electrostatic potential in proximity of the haloacid binding site was estimated to be
255 positive as calculated by APBS in PyMOL version 2.4.1 ([Jurrus *et al.*, 2018](#)). In the closest
256 homologous protein DehRhb, the amino acid composition in the same area is quite different
257 (**Fig. S4B**). Among the basic residues at the surface, two major changes are the replacement of
258 the small residues S40 and G200 in ZgHAD by cumbersome R44 and H204 in DehRhb. These
259 modifications lead to a closed conformation in DehRhb, compared to the potential entrance site
260 for the substrates in ZgHAD.

261 The second, smaller opening in the structure of ZgHAD is circular and globally more basic than
262 acidic, with 3 lysine (K67, K94 and K125) and 2 aspartic acid (D61 and D87) residues near the
263 surface (**Fig. S4C**). Neutral and hydrophobic residues (G63, T64, L90, G91, I93, N123, L126)
264 complete this patch of amino acids at the entrance site. Several hydrophobic residues (L25,
265 W42, L46, F62, S122) constitute the tunnel leading to the active site and in particular to the
266 substrate binding residues. The corresponding zone of this smaller opening is cluttered in
267 DehRhb by the presence of K94 and A70, as well as L97 although to a lesser extent.

268 Nevertheless, two channels are also present, side by side, in a nearby zone in DehRhb (**Fig.**
269 **S4D**). The equivalent to that zone is covered by K94, N123, L126, L127 and Q130, forming
270 helix α_6 in ZgHAD (**Fig. S5**). By contrast, this helix is replaced by an arch-forming structure

271 and rather divergent residues in the loop ¹²⁶SAPSPAPSP¹³⁴ in the *Rhodobacteraceae* homolog
272 (Fig. S5).

273

274 **Dimeric interface**

275 There are 12 amino acids, hydrophobic in majority, that are involved at the dimeric interface in
276 ZgHAD : L41, H48, Y49, L51, T52, E53, T56, K76, W181, R188, G200, Y204 (Fig. S6). Eight
277 of these 12 positions are also at the dimer interface in DehRhb, while only 5 equivalent
278 positions, in the alignment with DehSft, are also involved in dimerization (Fig. S2). Among
279 these corresponding residues, 3 out of 8 are identical in DehRhb (Y53, T56 and W185) and
280 only one (Y52) is conserved in DehSft. Furthermore, R188 is implicated in hydrogen bonds
281 with E53 and T56 and W181 forms a salt bridge with Y49.

282

283 **DISCUSSION**

284 Expectedly, ZgHAD displays the common fold and dimeric arrangement shared with the other
285 L-2-HADs, where the catalytic site is located in a cavity between the core and cap domain.
286 Despite this similarity, major differences regarding lengths of loops and interface interactions
287 are reflected in the low sequence identity (between 19 and 31 %) that ZgHAD shares with other
288 characterized L-2-HADs. The highest similarity is shared with DehRhb the closest homolog,
289 which is also reflected in similar catalytic properties. In all enzymes, a conserved aspartate
290 residue (D14 in ZgHAD) is the main catalytic amino acid, performing the first step of the
291 reaction. The activation of the water molecule is done by a His/Glu dyad for both ZgHAD and
292 DehRhb, whereas various other amino acid dyads, such as Asp/Asn, Asp/Lys or Lys/Tyr, have
293 been proposed to be responsible of this step in other L-2-HAD enzymes (Wang *et al.*, 2021;
294 Novak *et al.*, 2013; Schmidberger *et al.*, 2007; Hisano *et al.*, 1996; Nakamura *et al.*, 2009). A

295 similar His/Glu dyad is well-known to operate in an equivalent manner in haloalkane
296 dehalogenases that are widespread in the marine environment (Janssen, 2004).
297 Among the nine residues, depicted to be the most important for catalytic activity, five residues
298 are fully conserved, while the four other amino acids of L-DEX YL (two serines, an arginine
299 and an asparagine) are less systematically conserved, but present in most bacterial and archaeal
300 sequences (Table S1 and Fig. S2). Other conserved amino acids have been shown to belong to
301 a hydrophobic pocket surrounding the active site and appear to play an important role in
302 determining the stereo-specificity of the enzyme (Fig. S2; Novak *et al.*, 2013). F43, L46, L47,
303 F62, H179 and W181 compose this hydrophobic cluster in ZgHAD. Based on the high-
304 resolution structure of ZgHAD in the unbound state, computational analyses by docking
305 substrate molecules into the active site revealed that only two important residues for substrate
306 binding or catalysis are different between ZgHAD and DehRhb (L47 and S120 of ZgHAD vs.
307 I51 and T124 of DehRhb). Since serine and threonine are neutral polar amino acids, the
308 substitution between them is unlikely to introduce a strong significant change of the catalytic
309 activity. The same conservative replacement can be assumed for the exchange of hydrophobic
310 amino acids isoleucine to leucine in ZgHAD. Nevertheless, a mutational study of L-DEX YL
311 was shown to produce a reduction by 20% of the dehalogenation activity towards 2-
312 chloropropionic acid by a single mutant, S175T (Kurihara *et al.*, 1995), suggesting that this
313 substitution might have a similar effect in ZgHAD, as compared to DehRhb. The recombinant
314 ZgHAD is shown to have preferable activity towards short-carbon-chain substrates and most
315 specifically for C2 compared to C3 haloalkanoic acids. Iodoacetic acid and bromoacetic acid
316 are found to be the best substrates and the catalytic turnover rates are similar for both substrates
317 (Grigorian *et al.*, 2021). When compared to DehRhb, we found that ZgHAD removes bromide
318 more efficiently from bromoacetic acid, with a V_{max} value of $1.12 \mu\text{M}\cdot\text{sec}^{-1}$, in respect to 1.75
319 $\mu\text{M}\cdot\text{min}^{-1}$ for the *Rhodobacteraceae* enzyme (Grigorian *et al.*, 2021). The K_m value of ZgHAD

320 for bromoacetic acid was also found to be less than that of DehRhb (0.46 mM against 6.72 mM
321 respectively), suggesting a better affinity for this substrate. Consequently, the replacement of
322 these two amino acids might be associated to an increase of substrate reactivity and/or affinity
323 in ZgHAD, as compared to that in DehRhb. L47 is located in the cap domain, which is assumed
324 to be responsible for substrate recognition and binding. However, this hypothesis requires
325 further experimental examination, such as site-directed mutagenesis. The high structural
326 conservation of active site residues between ZgHAD and DehRhb, suggests that the equivalent
327 His/Glu dyad might be responsible for water activation. Domain movements, such as that of
328 the cap domain relative to the core domain but also of the two monomeric subunits relative to
329 each other, have been described to be important for the activity of DehIVa ([Schmidberger et al., 2007](#)).
330 Despite the difference of residues involved in the mechanism between DehIVa and
331 ZgHAD, it is interesting to note that the crystal structure of ZgHAD_H179N highlights the
332 possibility of such movements, which potentially open the access to the active site pocket,
333 allowing the entrance of substrate molecules and the outward diffusion of the reaction products.
334 The importance of these movements for activity might be the basis for the dimeric assemblage
335 of the enzymes of this class.

336 In the same line, interactions involved in dimerization of ZgHAD are most similar to those of
337 DehRhb and different from DehSft ([Rye et al., 2009](#)), L-DEX YL ([Hisano et al., 1996](#)) and
338 DhIB ([Ridder et al., 1997](#)), as described by [Novak et al., 2013](#). L-2-HADs were shown to be
339 robust enzymes as they display significant thermostability and resistance to organic solvents.
340 L-DEX YL retained 100% of its activity when incubated at 60°C for 30 minutes ([Liu et al.,](#)
341 [1994](#)) and DehRhb from *Rhodobacteraceae* retained 90% activity when incubated at 55°C
342 ([Novak et al., 2013](#)). Similarly, we also observe a high thermal stability for ZgHAD as it
343 conserved 100% of its activity after incubation at 55°C during 30 minutes. On the other hand,
344 while ZgHAD was shown to be quite stable in low concentrations of ethanol, methanol,

345 acetonitrile and DMSO, it was rapidly inactivated at high concentrations. Similar results were
346 presented for DehRhb (Novak *et al.*, 2013) and DehSft (Rye *et al.*, 2009). ZgHAD and DehRhb
347 appear to be more stable than DehSft, when incubated with the same organic solvents. The
348 optimum pH was not determined since the activity assay is pH dependent, but according to
349 previous studies, L-2-HADs enzymes are generally reported to be alkaline (Van der Ploeg *et*
350 *al.*, 1991; Liu *et al.*, 1994). In this respect, although it could only be due to crystal packing
351 artifact, notably the possible flexibility at the interface of the dimer was observed at alkaline
352 pH, as shown by the dimer positional variability of ZgHAD_H190N crystallized at pH 8.5.
353 These differences in the interactions at the dimeric interface might explain the variations
354 observed in thermal and solvent stabilities of the L-2-HADs, if the dimer formation is necessary
355 to conserve active enzymes.

356 The most important differences between DehRhb and ZgHAD are seen at the two entrances
357 leading to the active site (**Fig. S4 and S5**), possibly explaining the observed differences in
358 substrate specificity. While they both exhibit the presence of two openings connected with the
359 catalytic cavity, in DehRhb these are situated on the same side of the monomeric subunit, one
360 giving direct access to the hydrophobic pocket called the “halogen cradle”, whereas in ZgHAD
361 the two orifices are disposed on opposite sides of the monomeric subunit. This positions the
362 potential ‘halogen cradle’ differently with respect to the openings in ZgHAD. The charge
363 distribution at the openings is also different in the two enzymes, where DehRhb presents a less
364 charged environment than ZgHAD. In addition, the electrostatic properties of the catalytic
365 cavities of ZgHAD vs DehRhb, calculated the APBS software appear to be opposite, since a
366 global acidic environment is predicted for DehRhb, while it is basic in ZgHAD (data not
367 shown). This basic environment might be more attractive for small halogenated acids in the
368 enzyme from *Z. galactanivorans*, in agreement with its substrate preference (Grigorian *et al.*,

369 2021). These differences might also explain the higher catalytic efficiency of ZgHAD towards
370 bromoacetic acid than DehRhb.

371 Another interesting structural difference between ZgHAD and DehRhb that could explain
372 substrate specificity is the position of the glutamic acid at the beginning of the tunnel plunging
373 into the larger entrance of ZgHAD (E17) whereas it is located inside the core of the catalytic
374 cavity of DehRhb (E13). When H179 was mutated to alanine or asparagine in ZgHAD_H179A
375 and ZgHAD_H179N mutant enzymes, this led to the movement of E17 and an obstruction of
376 the larger entrance that potentially blocked the access to substrate binding residues (**Fig. S7**).
377 As the catalytic H179 was affected in both mutant enzymes it was not possible to observe the
378 effect of the tunnel closing by E17 on dehalogenation activity or substrate affinity but it might
379 be interesting to study in future.

380

381 **CONCLUSION**

382 In conclusion, our structural study by comparison to all available HAD structures allows
383 pinpointing the subtle differences on a same overall quaternary arrangement that lead to
384 variations of the catalytic activity and/or substrate specificity. In particular, the position and
385 charge distribution at the entrance to the active site cavity appear to vary among homologous
386 enzymes. Our data also confirm the possibility that domain movements, occurring between the
387 two monomeric subunits of the dimer, may play a key role in substrate tunneling to and from
388 the active site. Future work using site-directed mutagenesis and methods to analyze the
389 dynamics will help confirm these findings.

390

391 **MATERIALS AND METHODS**

392 **Gene cloning and site-directed mutagenesis**

393 The ZgHAD gene sequence (Zobellia_4183) was cloned from the genomic DNA of *Zobellia*
394 *galactanivorans* as described by [Barbeyron et al., 2001](#); using primers Zgal_4183fw and
395 Zgal_4183rv (**Table S2**) The PCR product was ligated into pFO4 vector using *Bam*HI and
396 *Eco*RI restriction sites and the T4 DNA ligase protocol (New England Biolabs). The
397 recombinant vector was transformed first into *Escherichia coli* DH5 α for sequence verification
398 and subsequently into *E. coli* BL21(DE3) expression strain.

399 ZgHAD mutants H179A and H179N were produced using QuickChange Lightning Site-
400 directed Mutagenesis (Agilent Technologies). Primers used are listed in **Table S2** from
401 supplementary data.

402

403 **Gene overexpression and protein purification**

404 The procedure for heterologous gene expression with subsequent production and protein
405 purification was performed as described in [Grigorian et al., 2021](#).

406

407 **Thermal unfolding experiments**

408 The proteins were diluted to a final concentration of 10 μ M. For each condition, 10 μ L of
409 sample per capillary were prepared. The samples were loaded into UV capillaries and
410 experiments were carried out using the Prometheus NT.48 (NanoTemper Technologies) that
411 can detect changes in the fluorescence of tryptophan (Trp) residues in the proteins. The
412 temperature gradient was set to an increase of 2 $^{\circ}$ C/min in a range from 20 $^{\circ}$ C to 95 $^{\circ}$ C. Protein
413 unfolding was measured by detecting the temperature-dependent change in tryptophan
414 fluorescence at 330 and 350 nm emission wavelengths. The increase of the ratio of Trp
415 fluorescence emission between 350 and 330 nm indicates the thermal unfolding transition
416 midpoint of the protein.

417

418 **Thermostability measurements**

419 The thermostability of the protein was determined by incubation in the presence of
420 monochloroacetic acid (MCA) at different temperatures between 10 and 90°C for 30 minutes.

421 The assay solution was added and then incubated on ice for 1h. The solvent stability was
422 investigated by incubating the enzyme in the presence of monochloroacetic acid and with
423 ethanol, methanol, acetonitrile and dimethylsulfoxide (DMSO) at concentrations between 10%
424 and 80% for 1h at room temperature. The assay solution was added and then incubated on ice
425 for 1h. The activity was determined by measuring the absorbance at 560 nm as previously
426 described.

427

428 **Crystallization, data collection, structure determination and refinement**

429 The purified L-haloacid dehalogenase (ZgHAD) and its variants (ZgHAD_H179A and
430 ZgHAD_H179N) were concentrated using a 10 kDa membrane Amicon Ultra-15 centrifugals
431 Filters (Mercks Millipore) at 3600 g and 4°C until a final concentration of 15 mg/mL was
432 reached. Hanging drops were prepared by mixing 2µL of ZgHAD (15 mg/mL) and 1 µL of
433 reservoir, and were equilibrated by vapor diffusion at 20°C. Diffraction-quality crystals
434 appeared after approximately three days in a condition containing 0.33 M potassium
435 thiocyanate and 31% (w/v) PEG 3350 for ZgHAD and ZgHAD_H1179A; and containing 25%
436 (w/v) PEG3350, 0.1 M Tris-HCl pH 8.5 and 0.2 M NaCl for ZgHAD_H179N. Crystals were
437 soaked in their reservoir solutions supplemented with 10% (v/v) glycerol before flash-freezing
438 in liquid nitrogen. Diffraction data were collected at 100 K at microfocus beamline Proxima 2-
439 A (Soleil, France). The data were processed using XDS ([Kabsch, 2010](#)) and scaled with Aimless
440 from the CCP4 program package ([Winn et al., 2011](#)). The structure of ZgHAD was solved by
441 molecular replacement with the CCP4 suite program MolRep ([Vagin & Teplyakov, 2010](#)) using
442 the marine *Rhodobacteraceae* L-Haloacid Dehalogenase as the starting model (PDB code:

443 2YML). Iterative rounds of model building and refinement were carried out using Coot ([Emsley](#)
444 [et al., 2010](#)) and the Phenix.Refine module of PHENIX ([Adams et al., 2010](#)). The validation of
445 the crystal structures was performed with MolProbity ([Chen et al., 2010](#)).

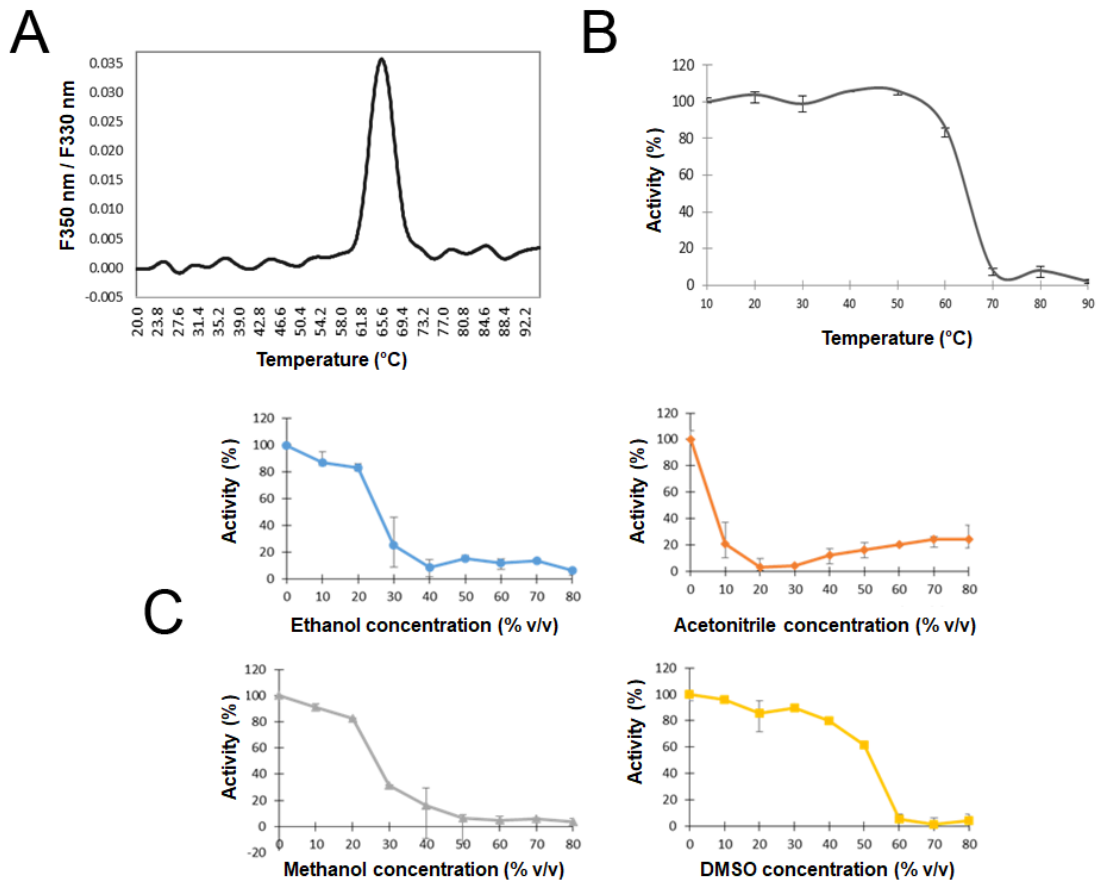
446

447 **Computational docking**

448 Computational docking of haloacetic acids (Cl-, Br- and I-) and of the 2-bromopropionic acid
449 to ZgHAD X-ray structure was performed using AutoDock Vina ([Trott and Olson, 2010](#)). The
450 initial coordinates of these molecules were generated from the SMILES string using
451 PHENIX.eLBOW ([Liebschner et al., 2019](#)). The ZgHAD protein was kept rigid during
452 docking. A docking grid with dimensions 25 Å × 25 Å × 25 Å, encompassing the entire active
453 site, was used. The calculation yielded 9 possible models, of which the one with the highest
454 affinity in kcal/mol was selected as the most likely. Then the complexes were energy minimized
455 using the Yasara energy minimization server ([Krieger et al., 2009](#)).

456

457 **FIGURES AND LEGENDS**



458

459 Figure 1: Thermal and solvent stability of the recombinant L-2-HAD from *Z. galactanivorans*.

460 (A) Thermal break point of ZgHAD protein as determined by the Prometheus NT.48. (B)

461 Thermal stability of ZgHAD activity as determined after the pre-incubation of the enzyme at

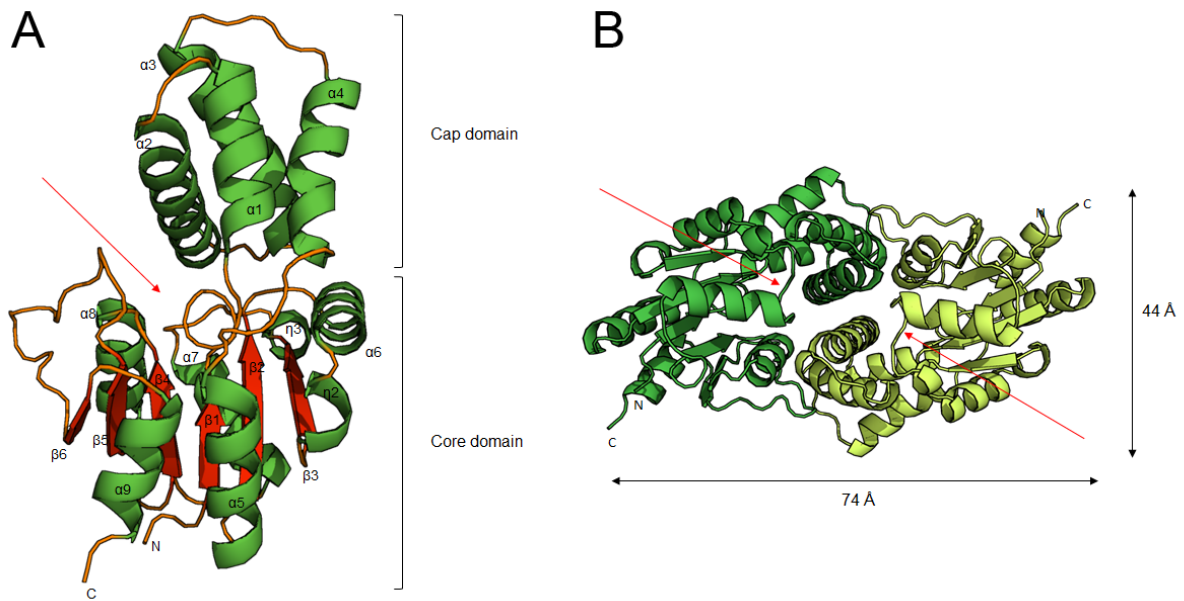
462 varying temperatures for 30 minutes before measuring the residual activity at 20°C for all the

463 points. (C) Solvent stability of ZgHAD activity as determined after pre-exposure to different

464 concentrations of ethanol, methanol, acetonitrile and DMSO for 1 hour before measuring the

465 residual activity in a standard solvent concentration.

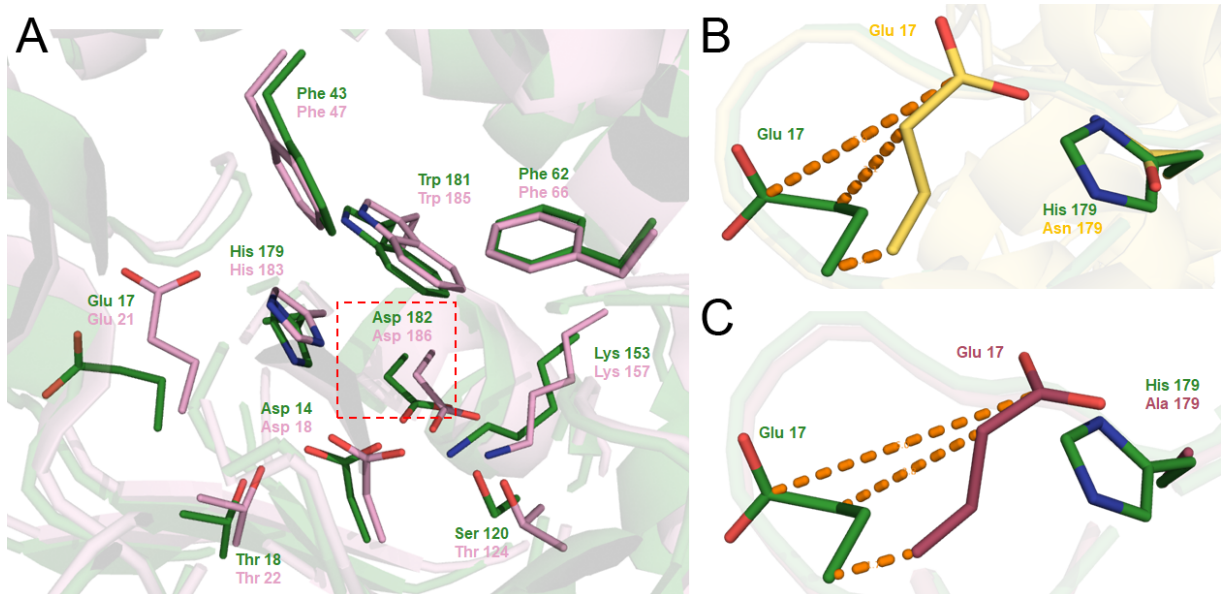
466



467

468 Figure 2: (A) Overall fold of the ZgHAD monomeric subunit presented as a ribbon diagram
 469 colored by secondary structure elements. The α helices, β strands and loops are colored in green,
 470 red and orange respectively. (B) Ribbon diagram of the ZgHAD dimer viewed along the two-
 471 fold horizontal axis. Each subunit is shown in a different color. The red arrows indicate the
 472 position of the catalytic site.

473

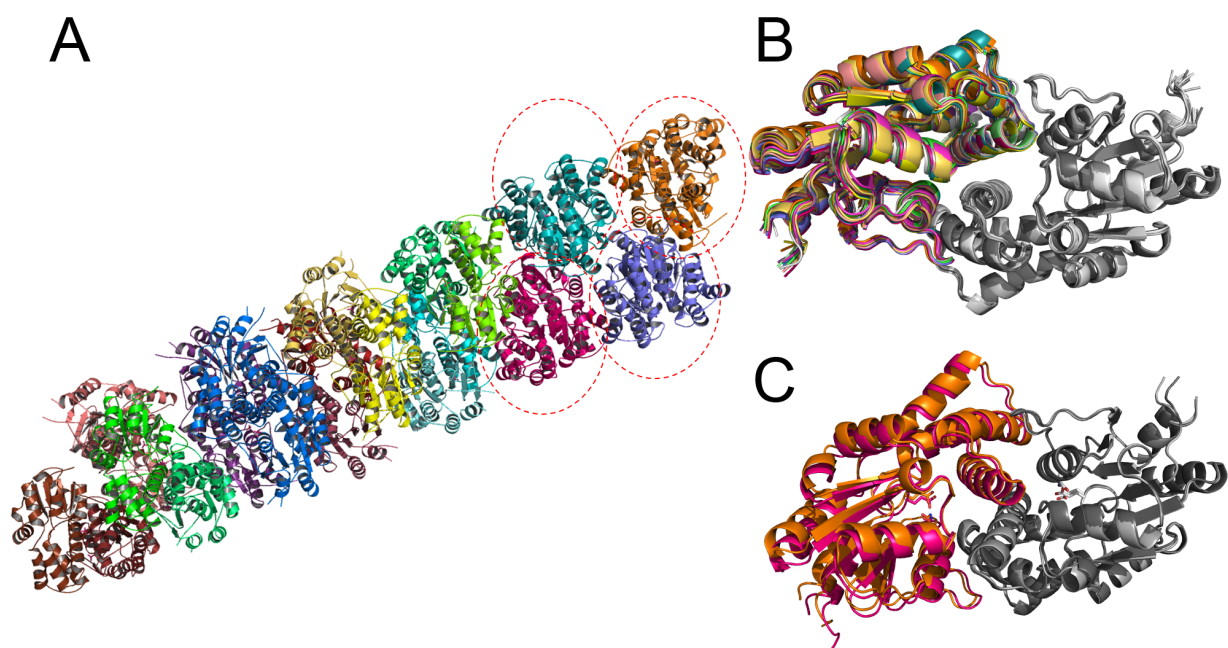


474

475 Figure 3: Ribbon representation showing the active site residues in the structures of ZgHAD
 476 and DehRhb. The side chains of selected residues are shown as sticks (carbons are colored in

477 green, pink, yellow or purple, oxygen red and nitrogen blue). (A) Superimposition of the
478 structures of ZgHAD (green) and DehRhb (pink). The red square shows the fixation site of the
479 substrate. (B) Superimposition of ZgHAD (green) and mutant H179N (yellow). (C)
480 Superimposition of ZgHAD (green) and mutant H179A (purple). Orange dotted lines show
481 difference of positions of side chain carbons of the movement of Glu17 between wild-type and
482 mutant proteins.

483

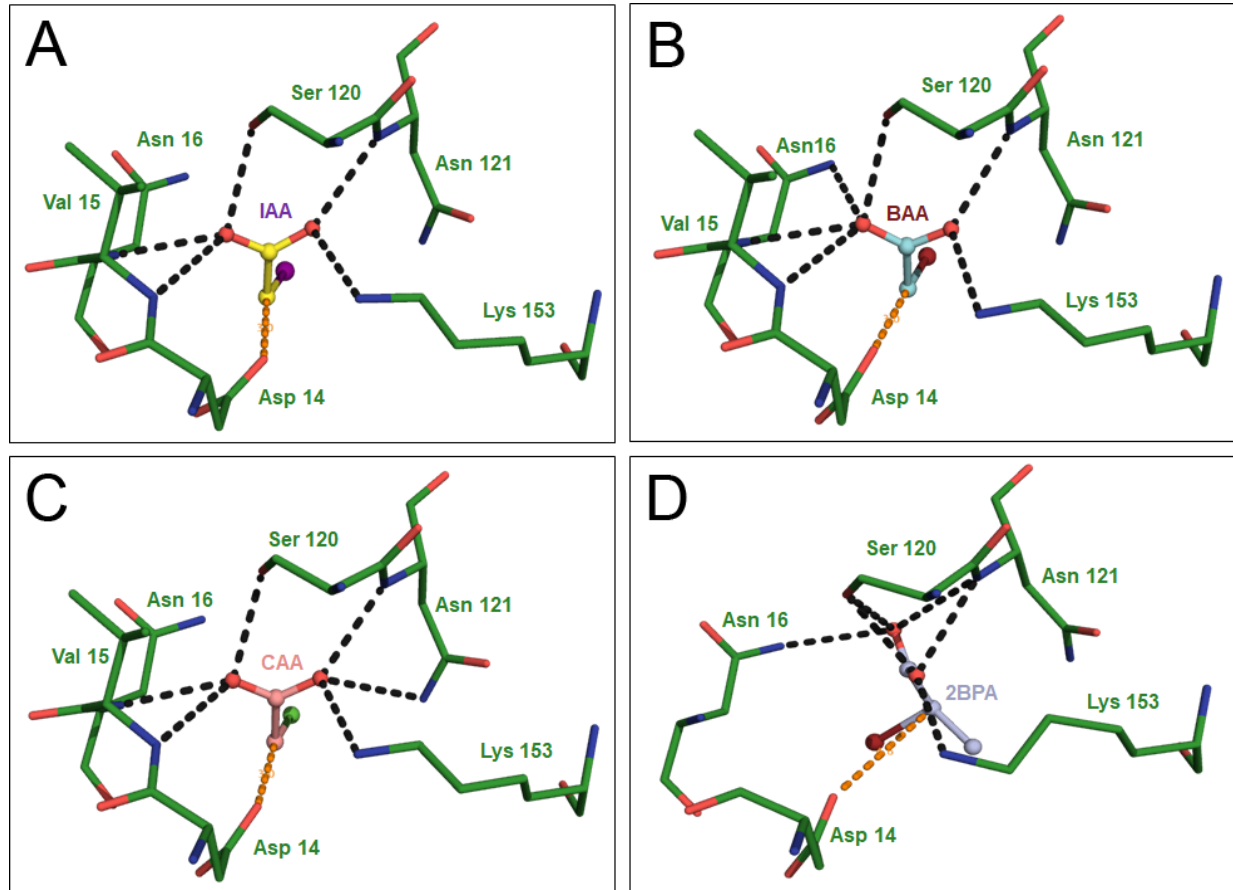


484

485 Figure 4: Crystal structure of the mutant ZgHAD_H179N shows an unexpected helical
486 arrangement of 13 dimers within the asymmetric unit. (A) Ribbon representation of the thirteen
487 ZgHAD_H179N dimers present in the asymmetric unit. The individual monomeric subunits of
488 each dimer are colored with similar colors, highlighting the helical arrangement of the dimers.
489 At one end, each circle of dotted lines surrounds one dimer. (B) Superimposition of all dimers
490 based on the calculation of a single monomeric subunit of each (grey). The colored monomeric
491 subunits highlight the variability of relative positions to the grey monomeric subunits, which in
492 contrast are almost identical. (C) Two of the dimers from (B), displaying the most distant

493 relative orientations, are presented as ribbons and the amino acids E17 and N179, represented
494 as sticks, highlight the position of the respective catalytic active sites.

495



496

497 Figure 5: Illustration of the molecular docking calculations of various substrate molecules in
498 the active site of ZgHAD. Selected residues are shown as sticks. (A) Docking of IAA (yellow)
499 in ZgHAD (green) structures. (B) Docking of BAA (cyan) in ZgHAD (green) structure. (C)
500 Docking of CAA (pink) in ZgHAD (green) structure. (D) Docking of 2BPA (purple) in ZgHAD
501 (green) structure. Orange dotted lines represent the distance between Asp14 and the carbon 2 of
502 the substrate. Black dotted lines represent hydrogen bonds of the substrates with surrounding
503 amino acids.

504

505

506 Table 1: Data collection and refinement statistics for ZgHAD. Values in parentheses refer to
 507 the outer resolution shell.

	ZgHAD wt	ZgHAD H179A	ZgHAD H179N
Beam line	PROXIMA-2A	PROXIMA-2A	PROXIMA-2A
Space group	P 2 ₁ 2 ₁ 2 ₁	P 2 ₁ 2 ₁ 2 ₁	P 2 ₁
Average unit cell (Å)	59.53 68.83 103.74	59.716 71.655 116.009	76.17 132.79 275.70
Wavelength (Å)	0.97986	0.980116	0.980114
Resolution (Å)	45.03 to 1.60	45.874 to 1.716	49.46 to 2.75
<i>R</i> _{merge}	0.098 (0.745)	0.080 (0.872)	0.132 (1.471)
<i>R</i> _{meas}	0.102 (0.774)	0.083 (0.908)	0.143 (1.612)
<i>R</i> _{pim}	0.028 (0.208)	0.023 (0.252)	0.054 (0.643)
No. unique reflections	41657 (2364)	41318 (2628)	144,483 (4,343)
Mean I/σI	15.2 (3.1)	21.6 (3.5)	9.6 (0.9)
CC _{1/2}	0.998 (0.896)	0.999 (0.925)	0.998 (0.447)
Completeness (%)	100 (100)	100 (100)	98.0 (60.1)
Average redundancy	13.5 (13.7)	13.0 (13.1)	6.9 (5.6)
Wilson B-factor (Å ²)	21.70	25.06	83.7
<i>Refinement</i>			
Resolution (Å)	1.78	1.88	2.75
<i>R</i> _{free} / <i>R</i> _{work}	0.1774/0.1552	0.1991/0.1680	0.2543/0.2065
Total number of atoms	7481	7415	44949
Water	473	413	56
Average B factor (Å ²)	24.7	29.6	73.31
Ligands	PO4 ; SCN	PO4 ; SCN	PO4
<i>R.m.s deviations</i>			
Bonds	0.010	0.017	0.009
Angles	1.0	1.4	1.5
<i>MolProbity analysis</i>			
Clashscore, all atoms	3.14	2.14	4
MolProbity score	1.21	1.02	1.21
Ramachandran outliers	0.0%	0.0%	0.12% (7)
Ramachandran allowed	0.46%	0.68%	1.96%
Ramachandran favored	99.54%	99.32%	97.91%
PDB entry	7ARP	7ASZ	7QNM

508

509

510 **SUPPLEMENTARY MATERIAL DESCRIPTION**

511 Figure S1: Thermal stability of the ZgHAD enzyme by Dynamic Light Scattering (DLS)
512 measurement. The peak of denaturation corresponding to a Z-average diameter increase
513 appears at 65°C.

514 Figure S2: Amino acid sequence alignment of ZgHAD and its homologues DehRhB, DehIVa,
515 L-DEX, DehIB and DehSft, using the programs MultAlin (Corpet, 1988) and ESPript (Robert
516 and Gouet, 2014).

517 Figure S3: Opening zones, tunnels and determination of ZgHAD dimer obtained by CAVER
518 Analyst program.

519 Figure S4: Visualization of the possible entrance sites leading to the active site cavity of
520 ZgHAD as compared to DehRhB. The surface view is modelled on the left and the
521 corresponding residues are represented as sticks on the right view.

522 Figure S5: Visualization of the possible entrance sites leading to the active site cavity in
523 DehRhB. (A) representation of the arch-like structure formed by a loop rich in proline residues
524 that liberates two orifices connected with the active site.

525 Figure S6: Representations of the dimeric interface of ZgHAD and DehRhB, analyzed using
526 PDBe PISA v1.52. (A) ZgHAD dimer and (B) dimer of DehRhB.

527 Figure S7: Surface view of the position of Glu17 at the putative substrate entrance site in (A)
528 ZgHAD wild type enzyme, (B) ZgHAD mutant H190A and (C) ZgHAD mutant H190N.

529 Table S1: Residues found to be essential for catalysis in L-DEX YL (Kurihara *et al.*, 1995),
530 and the corresponding amino acids in DhIB, DehIVa, DehSft, DehRhB and ZgHAD.

531 Table S2: sequences of primers used for PCR-based cloning for ZgHAD and and site-directed
532 mutants H179A and H179N.

533

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545

546 **AUTHOR CONTRIBUTIONS**

547 EG and LD conceived the experiments. EG contributed to the majority of the experiments and
548 analyses. MC contributed to crystal preparation, data collection and refinement and TR
549 contributed to structure resolution, refinement and docking. EG wrote the manuscript with the
550 support of LD, CL and MC. All authors corrected and approved the final manuscript.

551

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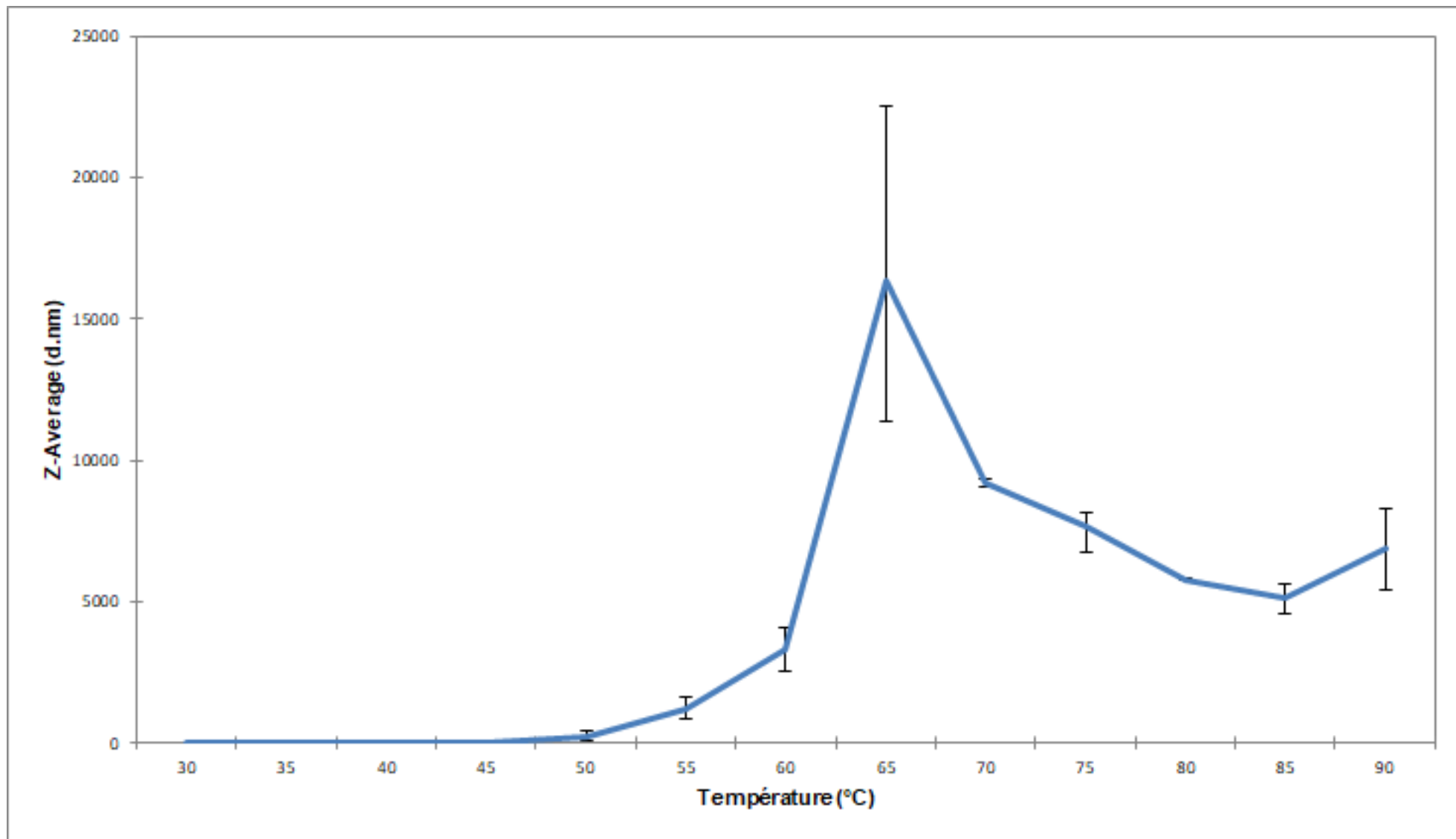


Figure S1: Thermal stability of the ZgHAD enzyme by Dynamic Light Scattering (DLS) measurement.

The peak of denaturation corresponding to a Z-average diameter increase appears at 65°C.

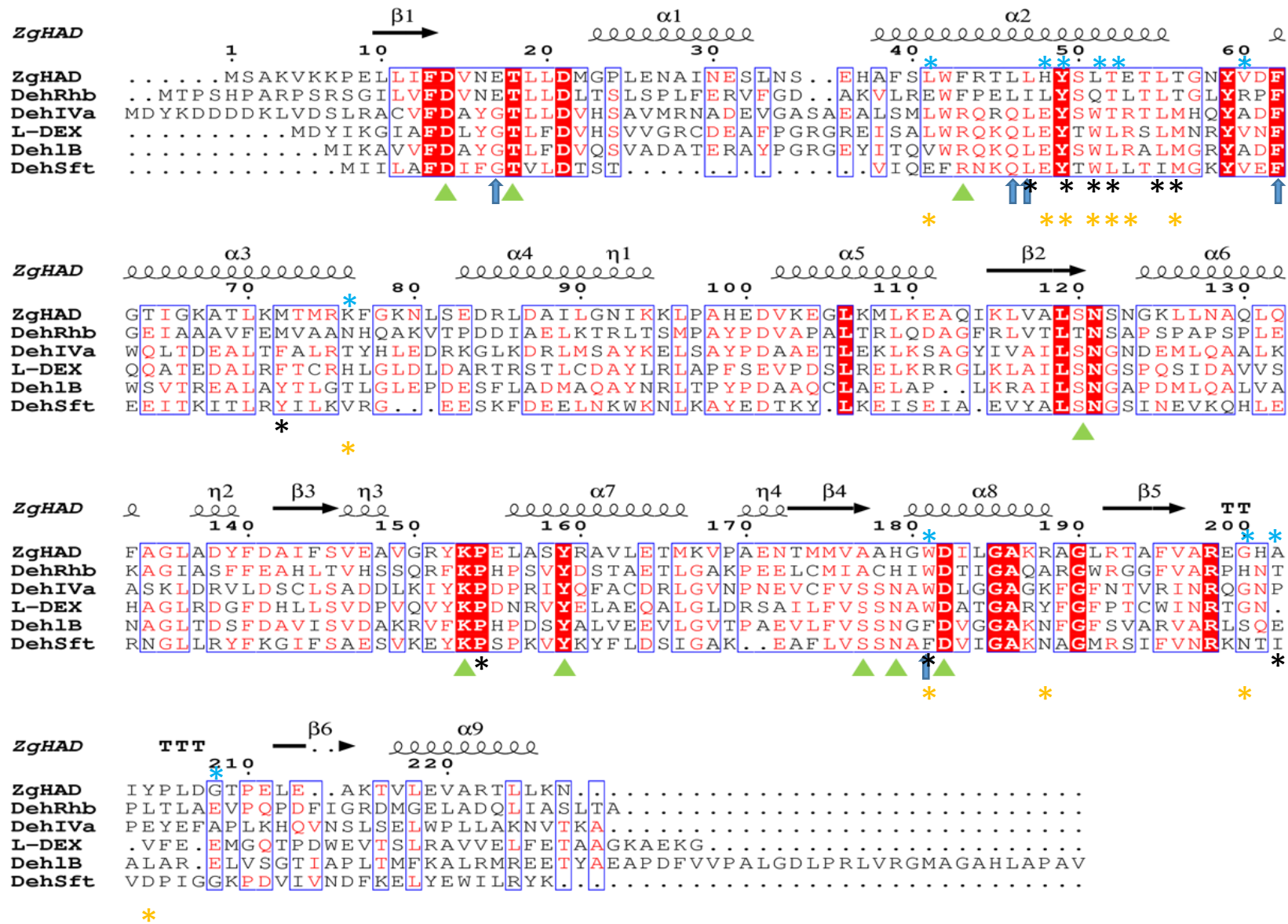


Figure S2: Amino acid sequence alignment of ZgHAD and its homologues DehRhb, DehIVa, L-DEX, Deh1B and DehSft, using the programs MultAlin (Corpet, 1988) and ESPript (Robert & Gouet 2014). Identical residues are shown in red boxes and residues with similar properties are in blue boxes. The green triangles highlight the residues that have been shown to be important for catalytic activity and substrate recognition. The blue arrows highlight additional residues belonging to the hydrophobic cluster around the active site in comparison with DehRhb. Black, blue and orange stars indicate the conserved residues respectively involved in dimerization of DehSft according to Rye *et al.*, 2009 and DehRhb according to Novak *et al.*, 2013 and in ZgHAD.

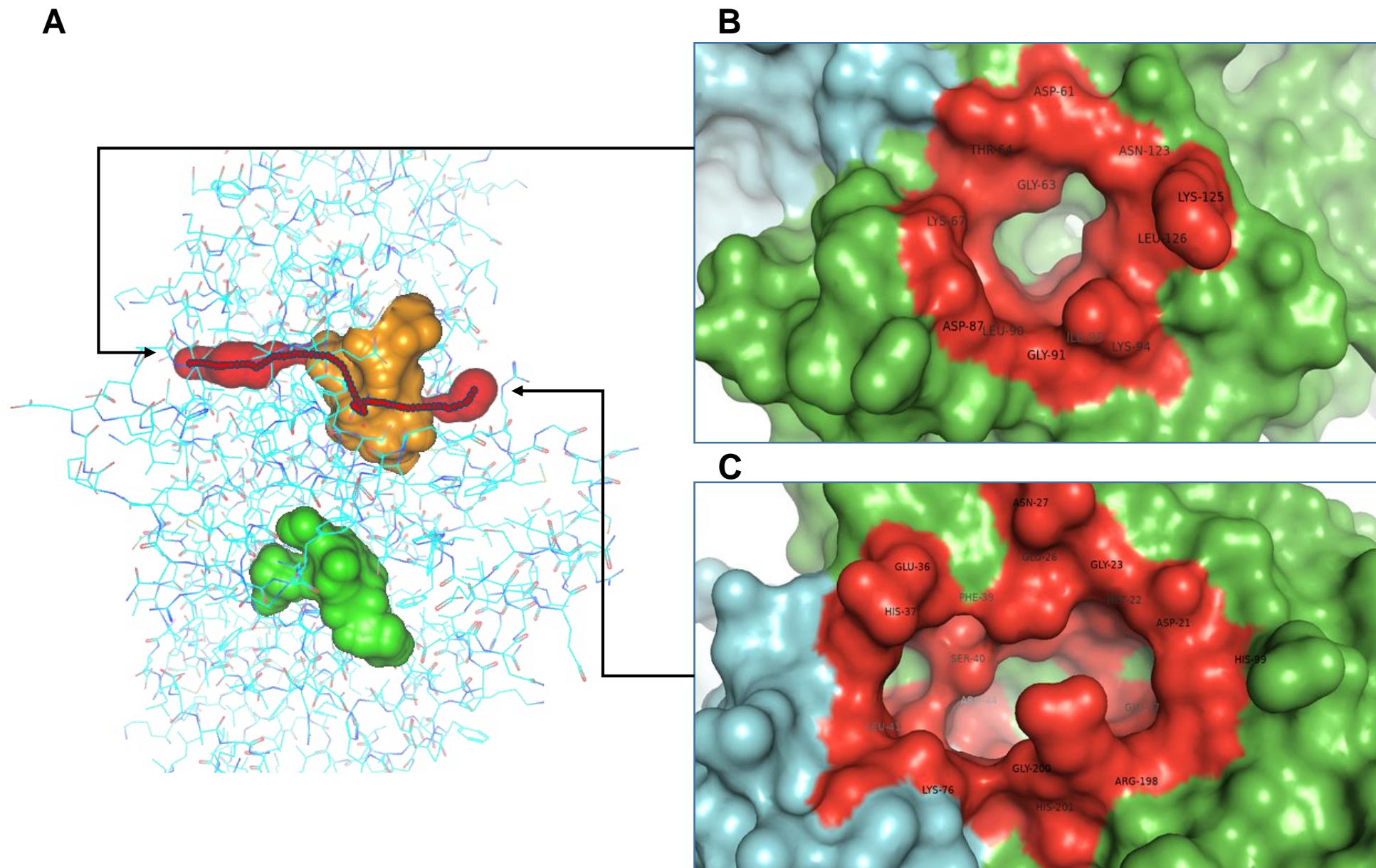


Figure S3: Opening zones, tunnels and determination of ZgHAD dimer obtained by CAVER Analyst program.

(A) Global view of the cavities and tunnels present in the dimer. The 2 catalytic cavities of each subunit are orange and green coloured. The tunnels of one monomer are shown in red and the pathway from one extremity to the other is marked with red dots. (B) Zoom of the smaller entrance site. (C) Zoom on the larger entrance site.

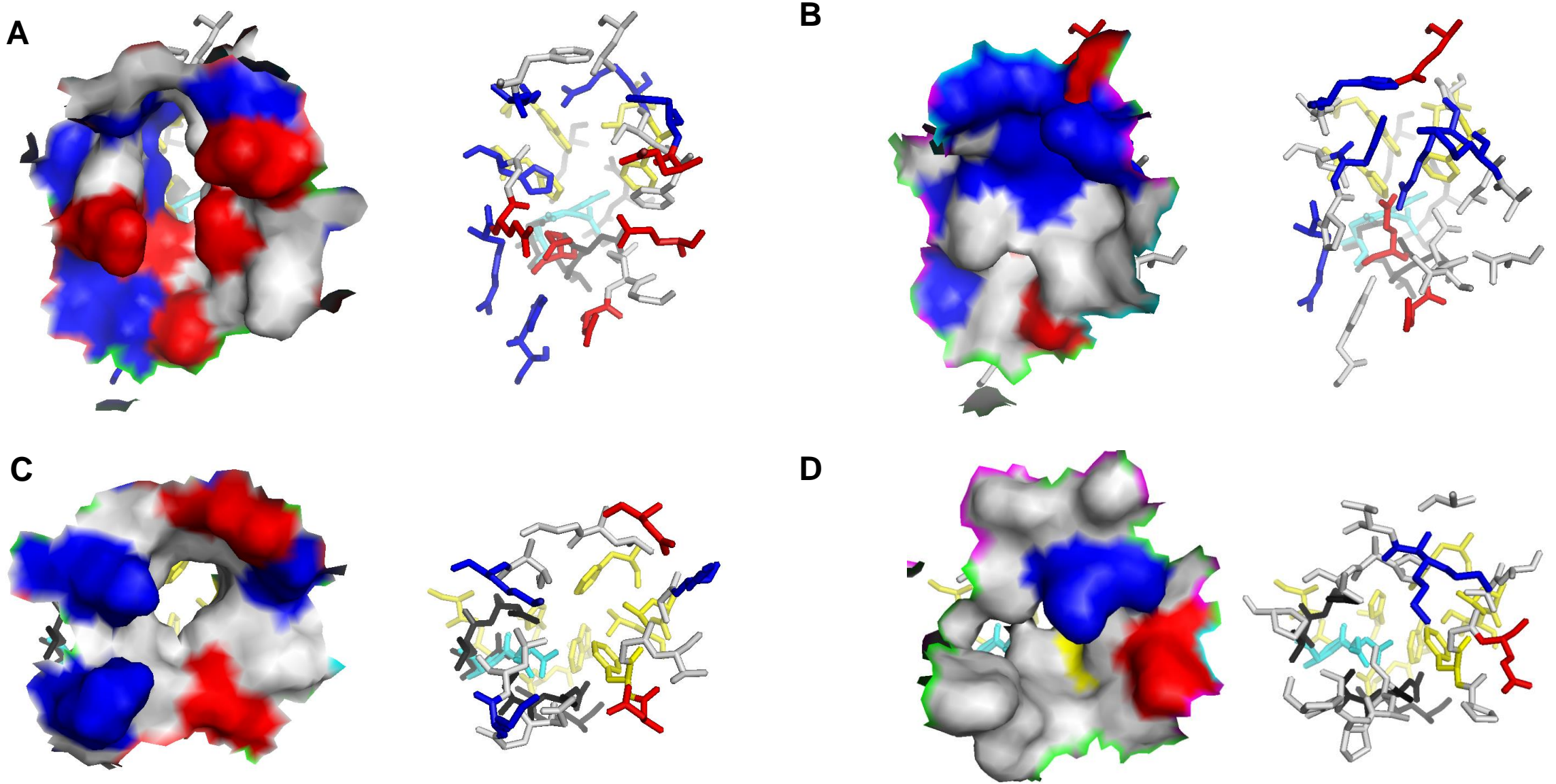


Figure S4: Visualization of the possible entrance sites leading to the active site cavity of ZgHAD compared to DehRhb.

The surface view is modelled on the left and the corresponding residues are represented on the right via stick view. (A) Large entrance accession zone in ZgHAD and (B) corresponding obstructed zone in DehRhb. (C) Small entrance accession zone in ZgHAD and (D) corresponding obstructed zone in DehRhb. Acidic negatively charged, basic positively charged and neutral amino acids are respectively in red, blue and light grey. Residues implicated in substrate binding and in the hydrophobic pocket are in black and in yellow. Aspartic acid-4-carboxymethyl ester representing the intermediate of chloroacetic acid linked to D14 in DehRhb is shown in cyan. This intermediate has been artificially introduced in the catalytic site of ZgHAD by overimposition of the DehRhb structure (PDB entry: 2YMP).

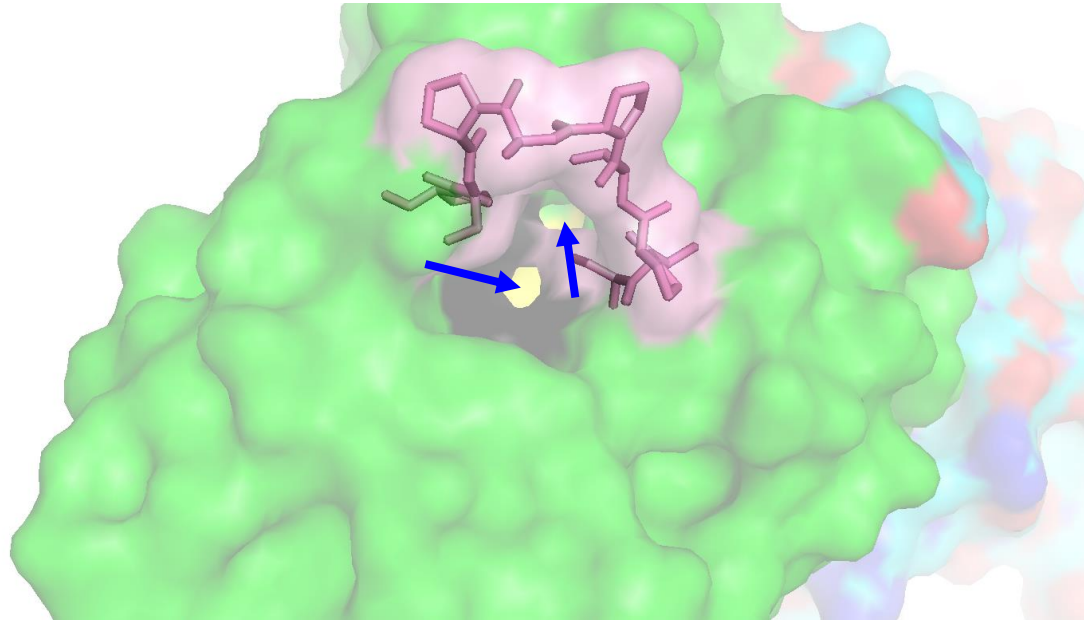
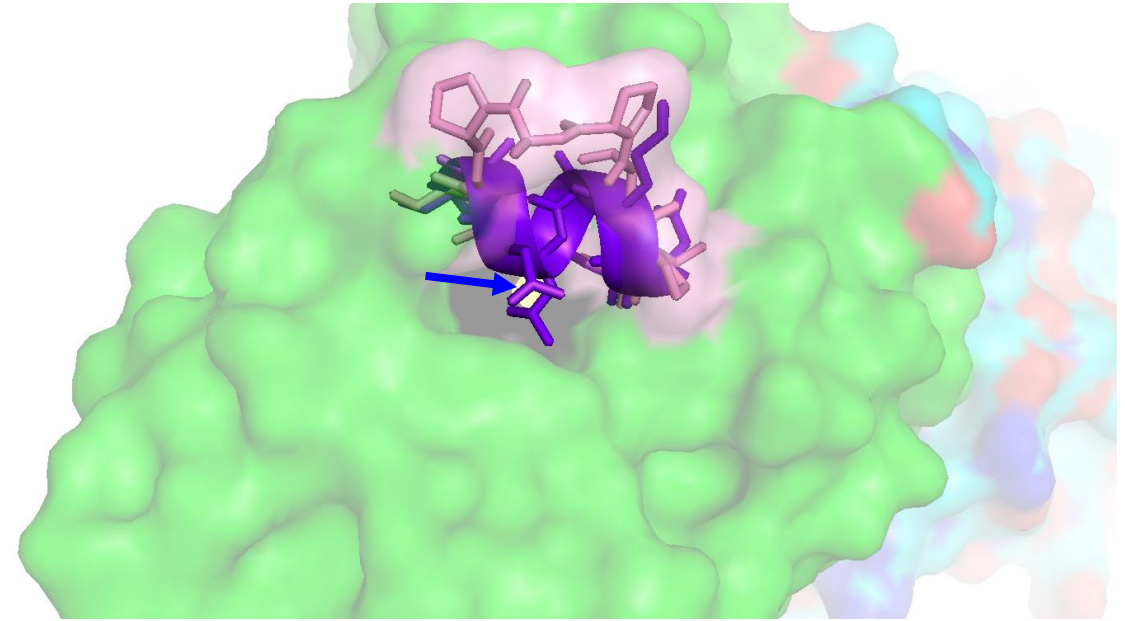
A**B**

Figure S5: Visualization of the possible entrance sites leading to active site cavity in DehRhb.

(A) Ark structure forming by a loop rich in proline residues liberates 2 orifices connected to the active site. (B) Corresponding ark zone in ZgHAD forms a alpha helix masking the accession to the active site.

Arch structure is represented in pink over the 2 orifices composed respectively of the substrate binding residues in black and the hydrophobic pocket residues in yellow indicated by blue arrows. The overimposed alpha helix structure of ZgHAD is shown in violet.

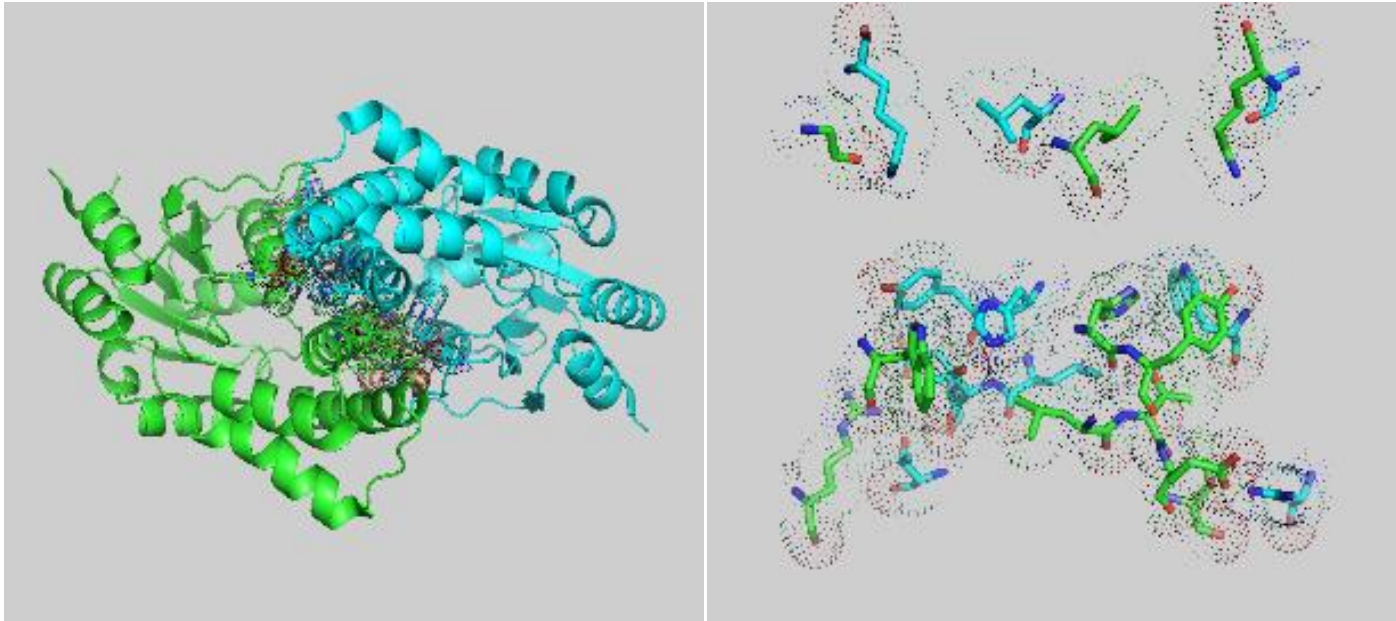
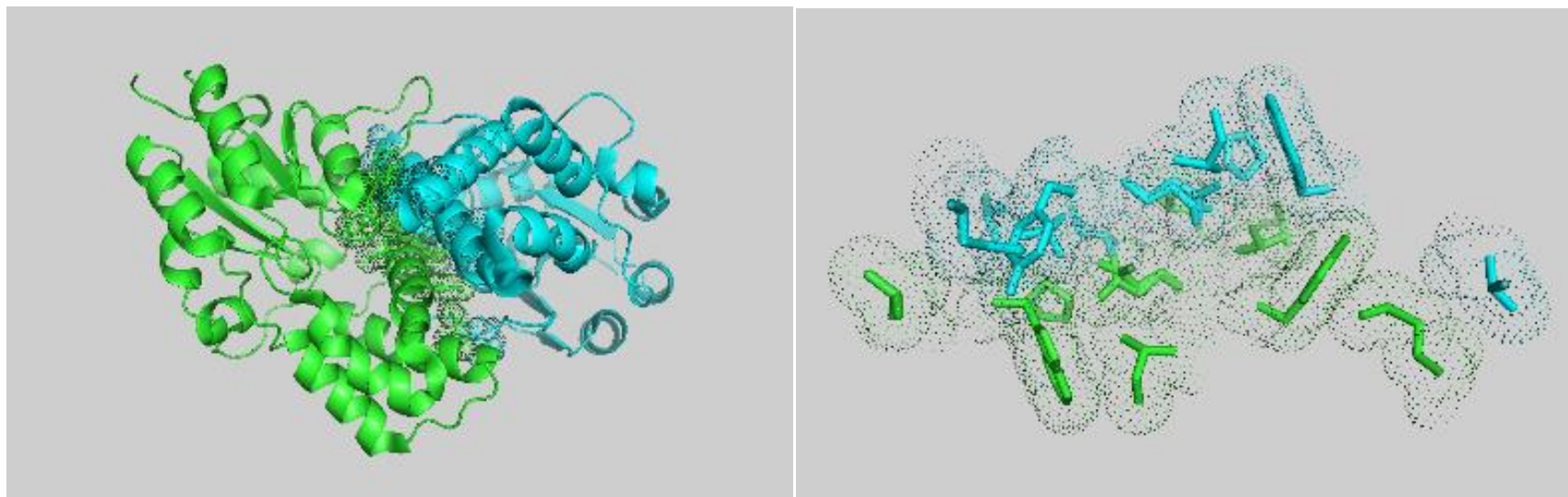
A**B**

Figure S6: Dimeric interface representation of ZgHAD compared to DehRhb analysed using PDBe PISA v1.52.

Every left panel represents the overall view and right panel shows the zoom on dimeric interface.

(A) Pictures of the ZgHAD dimer and (B) Pictures of DehRhb. In each enzyme, one monomer is coloured in green and the other coloured in cyan. Amino acids are surrounded by dots edited in pymol to visualize the interaction zone between the residues.

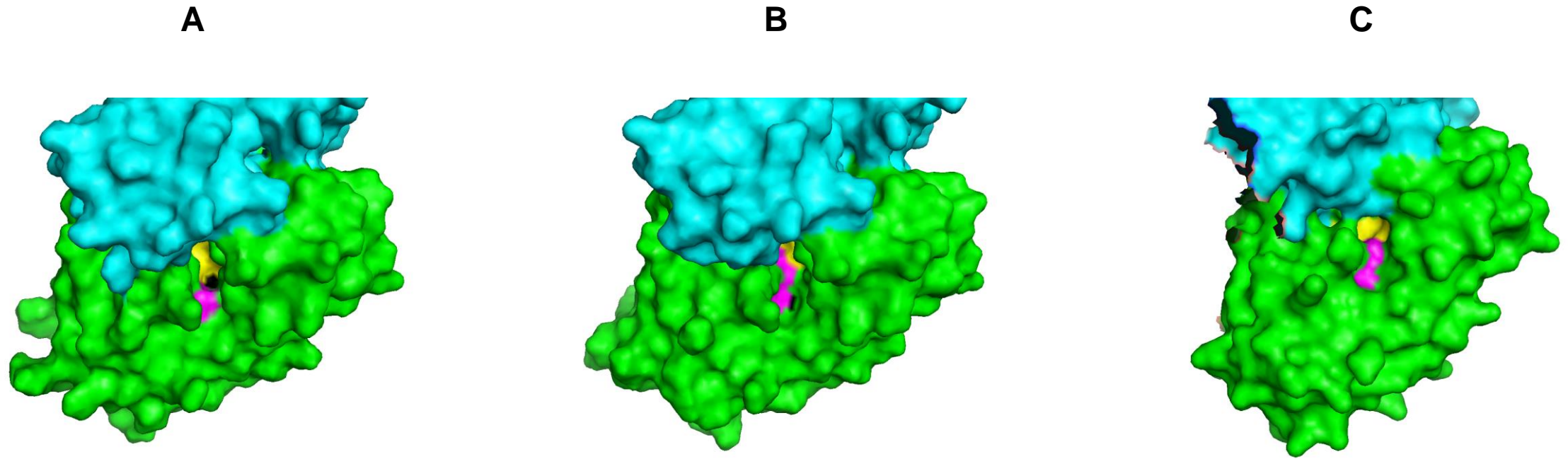


Figure S7: Surface view of E17 amino acid position at the putative substrate entrance site in (A) ZgHAD wild type enzyme, (B) ZgHAD mutant H190A and (C) ZgHAD mutant H190N.

Each monomer is represented in green and cyan. Putative substrate binding residues are in black and hydrophobic pocket residues of the active site are in yellow. E17 amino acid is shown in magenta.

L-DEX YL	DhlB	DehIVa	DehSft	DehRhb	ZgHAD
D10	D8	D11	D7	D18	D14
T14	T12	T15	T11	T22	T18
R41	R39	R42	R23	F47	F43
S118	S114	S119	S95	T124	S120
K151	K147	K152	K128	K157	K153
Y157	Y135	Y158	Y134	Y163	Y159
S175	S171	S176	S150	A181	A177
N177	N173	N178	N152	H183	H179
D180	D176	D181	D155	D186	D182

Table S1: Comparison of residues found to be essential for catalysis in L-DEX YL (Kurihara *et al.*, 1995), and the corresponding amino acids in DhIB, DehIVa, DehSft, DehRhb and ZgHAD.

Differences between enzymes are highlighted in red.

H179N_Fwd	CCATGATGGTCGCTGCCAACGGATGGGATATTTTAGG
H179N_Rv	CCTAAAATATCCCATCCGTTGGCAGCGACCATCATGG
H179A_Fwd	CCATGATGGTCGCTGCCGCCGGATGGGATATTTTAGG
H179A_Rv	CCTAAAATATCCCATCCGGCGGCAGCGACCATCATGG
ZgHAD_Fwd	GGATCCTCGGCAAAAAGTAAAGAAACCCGAG
ZgHAD_Rv	GAATCCTTAGTTCTTCAATAAGGTTCTGGGCTAC

Table S2: sequences of primers used for PCR-based cloning for ZgHAD and site-directed mutants H179A and H179N.