

From Quinoxaline, Pyrido[2,3-b]pyrazine and Pyrido[3,4-b]pyrazine to Pyrazino-Fused Carbazoles and Carbolines

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Article

From Quinoxaline, Pyrido[2,3-*b***]pyrazine and Pyrido[3,4-***b***]pyrazine to Pyrazino-Fused Carbazoles and Carbolines**

Frédéric Lassagne 1,*, Timothy Langlais ¹ , Elsa Caytan ¹ [,](https://orcid.org/0000-0003-0490-3074) Emmanuelle Limanton ¹ , Ludovic Paquin 1,*, Manon Boullard ¹ [,](https://orcid.org/0000-0002-5959-3029) Coline Courtel ¹ , Idriss Curbet ¹ , Clément Gédéon ¹ , Julien Lebreton ¹ , Laurent Picot 2,*, Valérie Thiéry ² [,](https://orcid.org/0000-0002-2096-0098) Mohamed Souab ³ , Blandine Baratte ³ , Sandrine Ruchaud ³ , Stéphane Bach 3,[*](https://orcid.org/0000-0002-7186-7483) , Thierry Roisnel ¹ and Florence Mongin 1,[*](https://orcid.org/0000-0003-3693-8861)

- ¹ Univ Rennes, CNRS, Institut des Sciences Chimiques de Rennes (ISCR)-UMR 6226, F-35000 Rennes, France; timothy.langlais@etudiant.univ-rennes1.fr (T.L.); elsa.caytan@univ-rennes1.fr (E.C.); emmanuelle.limanton@univ-rennes1.fr (E.L.); manon.boullard@etudiant.univ-rennes1.fr (M.B.); coline.courtel@etudiant.univ-rennes1.fr (C.C.); idriss.curbet@etudiant.univ-rennes1.fr (I.C.); clement.gedeon@etudiant.univ-rennes1.fr (C.G.); julien.lebreton@etudiant.univ-rennes1.fr (J.L.); thierry.roisnel@univ-rennes1.fr (T.R.)
- ² Laboratoire Littoral Environnement et Sociétés, UMRi CNRS 7266, Université de La Rochelle, 17042 La Rochelle, France; valerie.thiery@univ-lr.fr
- ³ Sorbonne Universités, UPMC Univ Paris 06, CNRS USR3151, "Protein Phosphorylation and Human Disease" Unit, Plateforme de criblage KISSf, Station Biologique de Roscoff, Place Georges Teissier, 29688 Roscoff, France; mo.souab@gmail.com (M.S.); baratte@sb-roscoff.fr (B.B.); sandrine.ruchaud@sb-roscoff.fr (S.R.)
- ***** Correspondence: frederic.lassagne@univ-rennes1.fr (F.L.); ludovic.paquin@univ-rennes.fr (L.P.); laurent.picot@univ-lr.fr (L.P.); bach@sb-roscoff.fr (S.B.); florence.mongin@univ-rennes1.fr (F.M.); Tel.: +33-223-236-931 (F.M.)

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Abstract: 2,3-Diphenylated quinoxaline, pyrido[2,3-*b*]pyrazine and 8-bromopyrido[3,4-*b*]pyrazine were halogenated in deprotometalation-trapping reactions using mixed 2,2,6,6-tetramethyl piperidino-based lithium-zinc combinations in tetrahydrofuran. The 2,3-diphenylated 5-iodoquinoxaline, 8-iodopyrido[2,3-*b*]pyrazine and 8-bromo-7-iodopyrido[3,4-*b*]pyrazine thus obtained were subjected to palladium-catalyzed couplings with arylboronic acids or anilines, and possible subsequent cyclizations to afford the corresponding pyrazino[2,3-a]carbazole, pyrazino[2',3':5,6] pyrido[4,3-b]indole and pyrazino[2',3':4,5]pyrido[2,3-d]indole, respectively. 8-Iodopyrido[2,3-b] pyrazine was subjected either to a copper-catalyzed C-N bond formation with azoles, or to direct substitution to introduce alkylamino, benzylamino, hydrazine and aryloxy groups at the 8 position. The 8-hydrazino product was converted into aryl hydrazones. Most of the compounds were evaluated for their biological properties (antiproliferative activity in A2058 melanoma cells and disease-relevant kinase inhibition).

Keywords: pyrazine; deprotometalation; coupling; *N*-arylation; palladium; copper

1. Introduction

Quinoxalines and pyridopyrazines are aromatic heterocycles present in compounds endowed with numerous interesting properties. Some derivatives are bioactive and are used as antimicrobial, anti-inflammatory, antimalarial, anticancer and antidepressant compounds [1,2]. Others are for example employed as organic dyes [3], electroluminescent materials [4], and organic semiconductors [5]. Quinoxaline and pyridopyrazine substrates can be readily synthesized by

condensation of 1,2-dicarbonyl compounds with 1,2-arylenediamines [6] and lend themselves to further elaboration. elaboration. For this interception of the ary little ary little with electrophic systems with electrophic systems is an architecture.

Deprotonative lithiation followed by interception of the arylmetals with electrophiles is an efficient way to functionalize aromatic compounds [7-12]. However, reactions with substrates sensitive to nucleophilic attack such as azines must be performed at very low temperatures to avoid secondary reactions between arylmetals and functions [13–15]. The use of in situ metal traps avoids the use of cryogenic conditions to achieve these reactions $[16,17]$. We have developed mixed lithium-zinc combinations based on TMP (TMP = 2,2,6,6-tetramethylpiperidino) capable of deprotonating sensitive substrates at temperatures close to rt [18-21]. In order to obtain original scaffolds such as pyrazino-fused carbazoles and carbolines, we decided to combine this deprotometalation under *in situ* trapping conditions with palladium- and copper-catalyzed coupling reactions. ritu tranni mbinations based on TMP (TMP = 2,2,6,6-tetramethylpiperidino) capable of deprotonating sensitive bstrates at temperatures close to rt [18–21]. In order to obtain original scatfolds such as razino-fused carbazoles and carbolines, we decided to combine this deprotometalation under situ trapping conditions with palladium- and copper-catalyzed coupling reactions. E μ ^t μ runctionalize aromatic compounts $[7-12]$. Trowever, reactions with substrates sensitive α efficient arrack such as azines must be performed at very low temperatures to avoid seconda lections between aryimetals and functions $[13-15]$. The use of in situ metal traps avoids the use cryogenic conditions to achieve these reactions $[16,17]$. We have developed mixed lithium-zind Consetensity lithistics followed by intercention of the exylmetels with electronides is an efficient $f(x) = \frac{f(x)}{f(x)}$ protonative ary little little protonation for the ary little ary lit efficient water such as against mast be performed at very for temperatures to avoid secondary Personalize ary littlicity interception for the ary little ary little ary little ary little ary little in the ary little ar

2. Results and Discussion Kesults anc deprotometalation under *in situ* trapping conditions with palladium- and copper-catalyzed coupling deprotometalation under *in situ* trapping conditions with palladium- and copper-catalyzed coupling

2.1. Synthesis To functionalize 2,3-diphenylquinoxaline (**1a**) and 2,3-diphenylpyrido[2,3-*b*]pyrazine (**2a**), two **2. Synthesis**

To functionalize 2,3-diphenylquinoxaline (1a) and 2,3-diphenylpyrido[2,3-b]pyrazine (2a), two deprotonation methods were tested in tetrahydrofuran (THF) (Table 1, *Method A* and *Method B*). deprotonation methods were tested in tetrahydrofuran (THF) (Table 1, *Method A* and *Method B*). To functionalize 2,3-diphenylquinoxaline (**1a**) and 2,3-diphenylpyrido[2,3-*b*]pyrazine (**2a**), two *2.1. Synthesis 2.1. Synthesis 2.1. Synthesis*

see Figure 1). 1b' was isolated in 70% yield by using ZnCl₂·TMEDA (1 equiv) and LiTMP (3 equiv). ¹ After purification (see experimental part). ² The rest is 5,8-diiodo-2,3-diphenylquinoxaline (1b'; 7% yield;

1.1 Ine infinition-zinc base of *Niethon A* is prepared from $ZnCl_2 \cdot INELJA$ (TMEDA = $N_rN_rN_rN_r$) The lithium-zinc base of *Method A* is prepared from ZnCl₂·TMEDA (TMEDA = *N,N,N',N* ramethylethylenediamine) and LiTMP in a 1:3 ratio. Previous studies have suggested that $\frac{1}{2}$ ratio. Previous studies $\frac{1}{2}$ ratio. Previous studies have suggested that $\frac{1}{2}$ ratio. Previous studies have suggested that $\frac{1}{2}$ ratio. Previous studies have suggested that $\frac{1}{2}$ ratio. Previous alitetry.componentialisme, and EFTMP depth and the substrates the subsequent and The lithium-zinc base of Method A is prepared from $ZnCl_2$ ·TMEDA (TMEDA = N,N,N',N' t_{t} test ratio. Previous studies have suggested that it is a 1:3 ratio. Previous studies have suggested that it $\frac{1}{2}$ compiled the substrates the substrate, $\frac{1}{2}$ intercepts the sub t_{t} remothylethylenediamine) and LiTMP in a 1:3 ratio. Previous studies have suggested that tetramethylethylenediamine) and LiTMP in a 1:3 ratio. Previous studies have suggested that it

is a 1:1 LiTMP-Zn(TMP)₂ combination. While LiTMP deprotonates the substrate, $Zn(TMP)_2$ intercepts the generated aryllithium [18,19,22]. A recent computer study on anisole showed that the reactive the generated aryllithium [18,19,22]. A recent computer study on anisole showed that the reactive species is solvated LiTMP. The effectiveness of the reaction derives from the stabilizing effect of the species is solvated LiTMP. The effectiveness of the reaction derives from the stabilizing effect of the transmetalation step [21]. transmetalation step [21].

It is possible to replace $Zn(TMP)_2$ by $ZnCl_2$ provided that there is no contact between LiTMP and ZnCl₂ in the absence of the aromatic compound [23,24]. Thus, *Method B* is limited to activated substrates for which deprotonation is favored over reaction between LiTMP and $ZnCl₂$.

Whereas *Method A* should provide a lithium arylzincate, *Method B* should rather generate an Whereas *Method A* should provide a lithium arylzincate, *Method B* should rather generate an arylzinc. Nevertheless, both species are known to react with iodine by aryl transfer. arylzinc. Nevertheless, both species are known to react with iodine by aryl transfer.

Thus, 2,3-diphenylquinoxaline (1a) and 2,3-diphenylpyrido[2,3-b]pyrazine (2a) were involved in Method A. After treatment at rt with the base for 2 h, addition of iodine led to iodoquinoxaline 1b and iodopyrido[2,3-*b*] pyrazine 2b-I in 74 and 70% yield, respectively (entries 1 and 3).

To evaluate *Method B*, **1a** and **2a** were mixed with ZnCl2·TMEDA before addition of LiTMP at To evaluate *Method B*, **1a** and **2a** were mixed with ZnCl2·TMEDA before addition of LiTMP at -20 °C and stirring for 0.5 h (Method B, entries 2 and 4). After subsequent interception with iodine, **1b** and **2b-I** were isolated in 70 and 62% yield, respectively (entries 2 and 4). and **2b-I** were isolated in 70 and 62% yield, respectively (entries 2 and 4).

We explored the use of other electrophiles to intercept the heteroarylzinc chloride prepared from 2a by using Method B. Conversion to the corresponding bromide 2b-Br (60% yield, entry 5) and chloride 2b-Cl (62% yield, entry 6) was performed using bromine and trichloroisocyanuric acid, respectively, as the electrophile.

Figure 1. ORTEP diagrams (30% probability) of **1b'**, **2d**, **2f**, **2i**, **2p**. **Figure 1.** ORTEP diagrams (30% probability) of **1b**⁰ , **2d**, **2f**, **2i**, **2p**.

to significant degradation before trapping (Scheme 1). While the position of the iodo group in 3b was evidenced by subsequent reaction, it was studied by advanced NMR experiments in the case of **4b**
(see Supplementary Materials) Br Br pyrido[3,4-*b*]pyrazine (**3a**) [25,26], but failed with 7-bromo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**4a**) due The deprotometalation-iodination sequence was successfully applied to 8-bromo-2,3-diphenyl (see Supplementary Materials).

3a 1) ZnCl2·TMEDA (1 equiv) 2) LiTMP (1.2 equiv) subjected to in Suzuki couplings [27,28] under standard conditions (Table 2) [29]. Phenyl- (entry 1), 3) I2, -20 °C to rt 2-thienyl- (entries 2 and 3) and 2-aminophenyl- (entries 4 and 5) boronic acids led to the 5-arylated **3b** In order to prepare original pyrazino-fused carbazoles and carbolines, iodides **1b** and **2b-I** were derivatives 1c-e and 2d,e in 42-97% yields. The more electron-rich arylboronic acids and the less **4a** electron-poor quinoxaline substrate **1b** gave the best results.

Scheme 1. Deprotonative metalation of 8-bromo-2,3-diphenylpyrido[3,4-b]pyrazine (3a) and 7-bromo-2,3-diphenylpyrido[2,3-b]pyrazine (**4a**) followed by conversion to the halogeno derivatives.

Table 2. Suzuki coupling from 5-iodo-2,3-diphenylquinoxaline (1b) and 8-iodo-2,3-diphenyl pyrido[2,3-*b*]pyrazine (**2b-I**). pp. 2. $\frac{6424M}{M} \cdot \frac{644M}{M}$.

	Ph N	$ArB(OH)_2$ $Pd(PPh_3)_4$ (5 mol%) $NaHCO3$ (4 equiv)	Ar Ph N
	Ph N	$DME-H2O$ 80 °C, 3 h	Ph
	1b: $X = CH$ $2b-1: X = N$		1c-e, $X = CH$ 2d,e, $X = N$
Entry	Substrate	$ArB(OH)_2$	Product (Ar), Yield $(\%)$ ¹
$\mathbf{1}$	1b $(X = CH)$	$PhB(OH)_{2}$	1c (Ph) , 42
$\overline{2}$	1b $(X = CH)$		1 d (2-thienyl), 97
3	2b-I $(X = N)$	\overline{B} (OH) ₂	2d (2-thienyl) 2 , 75
$\overline{4}$	1b $(X = CH)$		1e (2-aminophenyl), 92
5	2b-I $(X = N)$	H_2N $B(OH)_2$	2e (2-aminophenyl), 73
		¹ After purification (see experimental part). ² See Figure 1.	

No intramovement intramodes intramodes in the corresponding diazing diazing α is α in the corresponding diazino-fused carbazole and α is α is a corresponding diazino-fused carbazole and α is a corresponding β-carboline was obtained for the azides coming from **1e** and **2e** [29]. We thus turned to the synthesis of the original pyrazino[2,3-*a*]carbazole **1g** and the corresponding pyrazino-fused γ-carboline **2g** by combining intermolecular C-N bond formation [30–38] with intramolecular C-C bond formation No intramolecular nitrene insertion into the corresponding diazino-fused carbazole and No intramolecular nitrene insertion into the corresponding diazino-fused carbazole and β -carboline was obtained for the azides coming from 1e and 2e [29]. We thus turned to the synthesis of the original pyrazino[2,3-a]carbazole $1g$ and the corresponding pyrazino-fused γ -carboline $2g$ isomers (Scheme 2). i somers by combining intermolecular \mathcal{S}^{c} bond for \mathcal{S}^{c} with interact \mathcal{S}^{c} bond \mathcal{S}^{c} bond \mathcal{S}^{c} bond \mathcal{S}^{c} bond \mathcal{S}^{c} bond \mathcal{S}^{c} bond \mathcal{S}^{\text

formation $\frac{1}{2}$.
The first step, attempted from **1b** by using catalytic palladium(II) acetate as transition metal source, Xantphos as ligand, and sodium *tert*-butoxide as base in toluene [39], yielded only 16% of diarylamine **1f**. Applying to **1b** and **2b-I** the conditions reported by Maes and co-workers for related reactions [29], **1f** and **2f** were obtained in 92 and 67% yield, respectively (Scheme 2, left). Inspired by Pieters and co-workers, who cyclized 4-(2-chlorophenylamino)pyridine into 5*H*-pyrido[4,3-*b*]indole under these conditions [40], we successfully employed catalytic $(Pd₂(dba)₃)$ and tri-*tert*-butylphosphine as catalyst precursors, diazabicyclo[5.4.0]undec-7-ene (DBU) as base, and dioxane as solvent for the second step. After 10 min at 180 ◦C under microwave irradiation, the pyrazino-fused carbazole **1g** and γ-carboline **2g** were isolated in moderate yields (Scheme 2, right).

We decided to combine both steps in an auto-tandem process under microwave irradiation e 3). Using (Pd₂(dba)₃), we selected Xantphos for its higher efficiency in comparison w tri-tert-butylphosphine. From 2b, best results were obtained with three equivalents of DBU as base this afforded carboline 2g in 70% yield (entry 3). source, Xantphos as ligand, and sodium *tert*-butoxide as base in toluene [39], yielded only 16% of (Table 3). Using $(Pd_2(dba_3)$, we selected Xantphos for its higher efficiency in comparison with (entries 1 and 2). In addition, a longer reaction time was required to ensure complete conversion and

Scheme 2. Conversion of 5-iodo-2,3-diphenylquinoxaline (1b) and 8-iodo-2,3-diphenylpyrido[2,3-b] pyrazine (**2b-I**) into 2,3-diphenyl-11H-pyrazino[2,3-*a*]carbazole (**1g**) and 2,3-diphenyl-11H-pyrazino [2',3':5,6]pyrido[4,3-b]indole (2g), respectively. Dba = dibenzylideneacetone.

The first step, at the first step, at the step of $\frac{1}{2}$ by $\frac{1}{2}$ by $\frac{1}{2}$ and $\frac{1}{2}$ by $\frac{1}{2}$ by 2,3-diphenyl-11*H-*pyrazino[2',3':5,6]pyrido[4,3-*b*]indole (**2g**) under MW irradiation. Table 3. Study of the conversion of the 8-halogenated 2,3-diphenylpyrido[2,3-b]pyrazines 2b into 2,3-diphenyl-11*H*-pyrazino[2',3':5,6]pyrido[4,3-*b*]indole (**2g**) under MW irradiation.

х .Ph $\ddot{}$ `Ph N N 2 _b	N _{H2} СI $(1.2$ equiv)	$Pd_2(dba)_{3}$ (4 mol%) Xantphos (8 mol%) DBU (n equiv) dioxane, MW conditions	'NH CI N N N 2f	Ph $\ddot{}$ Ph	NΗ N N 2g
Entry	Substrate (X)	n Equiv	Conditions ¹	$2f$ (%) 2	$2g$ (%) ²
1	$2b-I$ (I)	1.2	180 °C, 20 min	80	20
$\overline{2}$	$2b-I$ (I)	3.0	180 °C, 20 min	41	59
3	$2b-I$ (I)	3.0	180 °C, 60 min	$<$ 5	$>95,70^3$
4	$2b-I$ (I)	3.0	180 °C, 60 min 4	75	25
5	$2b-I$ (I)	3.0	$180 °C$, 20 min ⁵	23	75
66	$2b-I$ (I)	3.0	180 °C, 20 min	95	\leq 2
77	$2b-I$ (I)	3.0	$180 °C$, 20 min	32	67
8	$2b-I$ (I)	3.0	180 \degree C, 5 min	66^{8}	θ
9	$2b-Br(Br)$	3.0	$180 °C$, 5 min	35 ⁹	θ
10	$2b-Cl$ (Cl)	3.0	$180 °C$, 5 min	22^{10}	θ

 1 Maximum microwave power applied: 150–200 W at the beginning to reach the required temperature. 2 Evaluated μ and μ and μ and μ (cluster). The analytical structure in the conduction of the conduction μ and μ and 4 min. \degree Then classical heating at 180 \degree C for 40 min. \degree Without catalyst. \degree By using 12 mol% Pd₂(dba)₃ at mol% Xantphos. 8 The rest was unreacted 2b-I (32%). 9 The rest was unreacted 2b-Br (38%) and 2a (28%). ¹⁰ The profile of irradiation: The sequence '*Maximum microwave power applied: 150*–*200 W to reach 180 °C then* from the NMR spectra of the crudes.³ Yield after purification.⁴ Microwave profile of irradiation: The sequence '*Maximum microwave power applied: 150*–*200 W to reach 180* ◦*C then 2 min at 180* ◦*C before cooling to 100* ◦*C*' was repeated every 4 min. ⁵ Then classical heating at 180 °C for 40 min. ⁶ Without catalyst. ⁷ By using 12 mol% Pd₂(dba)₃ and 30 rest was unreacted **2b-Cl** (78%).

By testing a profile to maximize the microwave power, we noticed that an increase of the applied power favored the formation of $2f$ over $2g$ (entry 4). By carrying out one third of the reaction time under **2b-Cl** (78%). microwave irradiation and the rest by classical heating at the same temperature, a small microwave effect was evidenced (entry 5). While **2g** was not formed without catalyst, C-N bond formation giving 2f could take place (entry 6; see Figure 1). However, increasing the catalyst amount had no impact on the conversion to **2g** (entry 7). Finally, we intentionally chose a short reaction time (5 min) in order to compare the palladium-catalyzed reactions under microwave irradiation from 2b-I (entry 7), 2b-Br (entry 7) and 2b-Cl (entry 10). The results clearly showed decreasing reactivity from 2b-I to 2b-Cl, and thus, we selected iodo as halogeno group to pursue our investigations.

We applied the optimized procedure to the synthesis of the pyrazino-fused α-carboline **3g** from the bromoiodo substrate 3b and aniline. No trace of the expected product 3g was detected but the formation of **3g[']** due to competitive debromination was noted, showing a less obvious intramolecular C-H arylation (Scheme 3, left). Consequently, we moved to the synthesis of the pyrazino-fused δ-carboline **3h**. Upon treatment of **3b** by 2-aminophenylboronic acid under standard conditions [29], coupling and subsequent cyclization occurred, providing **3h** in 65% yield (Scheme 3, right).

Scheme 3. Conversion of 8-bromo-7-iodo-2,3-diphenylpyrido[3,4-*b*]pyrazine (**3b**) into **Scheme 3.** Conversion of 8-bromo-7-iodo-2,3-diphenylpyrido[3,4-*b*]pyrazine (**3b**) into 2,3-diphenyl-11H-pyrazino[2',3':4,5]pyrido[2,3-d]indole (3h) and ORTEP diagram (30% probability) of 3h.

of **3h**. under copper catalysis as reported previously [41,42] (Table 4). Thus, by treating **2b-I** with pyrrole (entry 1; see Figure 1), indole (entry 2), pyrazole (entry 3), imidazole (entry 4) or 1,2,4-triazole (entry 5), in the presence of catalytic copper(I) oxide, cesium carbonate, and dimethylsulfoxide (DMSO) at 110 ℃ for 24 h, the expected *N*-arylated azoles were obtained in 51 to 79% yields. To take advantage of the iodo group on 2b-I, C-N bond formation with azoles was attempted

As previously mentioned [22], such reactions work far less efficiently when performed on $1k'$, regardless of the amount of azole employed (Scheme 4). diiodides. Indeed, reacting the diiodide 1b['] with pyrazole only gave the monofunctionalized derivative derivative **1k'**, regardless of the amount of azole employed (Scheme 4).

Scheme 4. Copper-catalyzed *N*-arylation of 5,8-diiodo-2,3-diphenylquinoxaline (1**b'**).

Different amines and hydrazine reacted with 2b-I without recourse to catalyst (Table 5), affording Distribution of the state of the the corresponding secondary amines 2n-p (entries 1-3) and arylhydrazine 2q (entry 4) in good yields. The latter was converted into the hydrazones 2r-u in the presence of aromatic aldehydes chosen for their ability to potentially interact with binding sites of biological interest [43] (Scheme 5). Finally, reaction of **2b**-**I** with a phenol also proved possible without catalyst, giving the diaryl ether **2v** in 64% yield (Scheme 6).

$2b-1$.Ph N. Azole $\ddot{}$ Ph (2 equiv)	Cu ₂ O (10 mol%) Cs ₂ CO ₃ (2 equiv) DMSO 110 °C, 24 h	ŅRR' .Ph Ph $2i-m$
Entry	Azole	Product, Yield $(\%)$ 1	
$\,1\,$	Pyrrole	\mathcal{P} h N. ้N ² Ph	2i, 67
$\sqrt{2}$	$\operatorname*{Indole}% \left(X\right) \equiv\operatorname*{Ind}% \left(X$,Ph N Ñ Ph	2j, 51
$\ensuremath{\mathfrak{Z}}$	Pyrazole	.Ph N. N^2 Ph	2k, 71
$\bf 4$	Imidazole	.Ph N. Ph N	21,69
$\mathbf 5$	1,2,4-Triazole	N $N^{\overline{N}}$.Ph N, Ph Ν	2m, 79

Table 4. Copper-catalyzed *N*-arylation of 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**) using azoles. azoles.

 1 After purification (see experimental part). The rest is starting material and the corresponding deiodinated compound.

beneme b **Scheme 5.** Conversion of 8-hydrazino-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2q**) into aryl hydrazo Scheme 5. Conversion of 8-hydrazino-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2q**) into aryl hydrazones.

Scheme 5. Conversion of 8-hydrazino-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2q**) into aryl hydrazones.

	$2b-1$	(n equiv)	$2n-q$	
Entry	$R-NH2$	$\bf Conditions$	Product, Yield (%) ¹	
$\mathbf 1$	iPrNH ₂ (1.2)	EtOH, 150 °C, 18 h	HN .Ph N. Ph	2n, 69
\overline{c}	$4-MeOC_6H_4CH_2NH_2(1.2)$	EtOH, 150 °C, 24 h	NΗ Ph Ν MeO Ph	20, 71
$\ensuremath{\mathfrak{Z}}$	PhCH ₂ NH ₂ (1.2)	EtOH, 150 °C, 24 h	NH Ph N Ph N	$2p^2$, 79
$\bf 4$	$NH2NH2·H2O(10)$	iPrOH, reflux, 4 h	$HN^{\sim NH_2}$.Ph N. Ph	2q, 92

Table 5. Conversion of 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**) into corresponding amines and hydrazine. **Table 5.** Conversion of 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**) into corresponding amines on converse. \mathbf{e}

Scheme 6. Conversion of 8-iodo-2,3-diphenylpyrido[2,3-b]pyrazine (2b-I) into ether 2v.

2.2. Biological Activity

2.2. Biological Activity melanoma cells and proved to exert a modest to good activity (Figure 2). The best results were obtained with the 4-(trifluoromethyl)benzaldehyde hydrazone 2u and the 8-benzylamino pyrido[2,3-b]pyrazine Some of the synthesized compounds were tested [44] for their antiproliferative activity in A2058 **2o** which induced ~64% growth inhibition at 10−⁵ M.

 δ/ε or ε), and mitotic kinase Haspin). Table S1 in Supplementary Materials shows the results obtained. ⁴ NH₂NH₂-H₂(10) iPrOH, reflux, 4 h

¹ After purification (see the Materials and Methods section).³ See Figure 1.

¹ After purification (see the Materials and Methods section).³ See Figure 1.
 1 Age of th Compounds **1c**–**e**, **1g**, **2d**–**g**, **2i**–**v** and **3h** were evaluated [44] against a short panel of disease-relevant protein kinases. Protein kinases are drug targets often deregulated in diseases such as cancers and neurodegenerative disorders [45]. No significant inhibition of the following kinases was observed: Cyclin-dependent kinases 2 (CDK2/Cyclin A), 5 (CDK5/p25) and 9 (CDK9/Cyclin T), proto-oncogene kinase PIM1, CDC2-like kinase 1 (CLK1), dual specificity tyrosine phosphorylation regulated kinase 1A (DYRK1A), glycogen-synthase kinase 3 (GSK3; α/β or β), casein kinase 1 (CK1;

Figure 2. Antiproliferative activity of some of the synthesized compounds at 10⁻⁵ M after 72 h in A2058 human melanoma cells. A2058 human melanoma cells.

3. Materials and Methods

3.1. General Information 3.1. General Information

All the reactions were performed under a dry argon atmosphere. THF was distilled over All the reactions were performed under a dry argon atmosphere. THF was distilled over sodium/benzophenone. Column chromatography separations were achieved on silica gel (40–63 µm). μm). Melting points were measured on a Kofler apparatus. IR spectra were taken on an ATR Melting points were measured on a Kofler apparatus. IR spectra were taken on an ATR Spectrum 100 spectrometer (Perkin-Elmer). ¹H- and ¹³C-Nuclear Magnetic Resonance (NMR) spectra were recorded either on an Avance III spectrometer (291 K) at 300 MHz and 75 MHz, respectively, or on an Avance III on an Avance III HD spectrometer (298 K) at 500 MHz and 126 MHz, respectively (Bruker, Billevica, HD spectrometer (298 K) at 500 MHz and 126 MHz, respectively (Bruker, Billevica, Massachussets, USA). ¹H chemical shifts (δ) are given in ppm relative to the solvent residual peak and ¹³C chemical shifts are relative to the central peak of the solvent signal [46]. 2,3-Diphenylpyrido[2,3-*b*]pyrazine $(2a)$ [6], 8-bromo-2,3-diphenylpyrido[3,4- b]pyrazine $(3a)$ [25,26] and 7-bromo-2,3-diphenylpyrido[2,3- b] pyrazine (4a) [6] were prepared as reported previously. The biological activity assays were performed as reported previously $[44]$.

3.2. Crystallography. 3.2. Crystallography

(graphite monochromatized Mo-K α radiation ($\lambda = 0.71073$ Å)) for the compounds **1b⁰** and **2i**, or using a (graphite monochromatized Mo-Kα radiation (λ = 0.71073 Å)) for the compounds **1b'** and **2i**, or D8 VENTURE Bruker AXS diffractometer (multilayer monochromatized Mo-Kα radiation (λ = 0.71073 using a D8 VENTURE Bruker AXS diffractometer (multilayer monochromatized Mo-Kα radiation (λ Å)) equipped with a (CMOS) PHOTON 100 detector for **2f**, **2p**, **3h** and **2d**, at the temperature given in the crystal data. For 1b' and 2i, the structure was solved by direct methods using *SIR97* [47]. For 2f, temperature given in the crystal data. For **1b** and **2i**, the structure was solved by direct methods on $\frac{1}{2}$ and $\frac{1}{2}$. **2p, 3h** and **2d**, they were solved by dual-space algorithm using the *SHELXT* program [48]. Structural program [48]. Structural refinements were performed with full-matrix least-square methods based case of **2f** and **3h**, the contribution of the disordered solvents to the calculated structure factors was estimated following the *BYPASS* algorithm [50], implemented as the *SQUEEZE* option in *PLATON* [51]; commuted following the *BTTTIBS* algorithm [50], implemented as the *BRASSIS* option in FEIT BTV [51], a new data set, free of solvent contribution, was then used in the final refinement. All non-hydrogen a field data set, free of solvent contribution, was then used in the final refinement. This horning drogen atoms were refined with anisotropic atomic displacement parameters. Except nitrogen linked hydrogen final refinement. All non-hydrogen atoms were refined with anisotropic atomic displacement atom that was introduced in the structural model through Fourier difference maps analysis (**2f**, **2p**, **3h**), H atoms were finally included in their calculated positions and treated as riding on their parent **3h**), H atoms were finally included in their calculated positions and treated as riding on their parent on), I dome were many included in their calculated positions and treated as Hang on their parent atom with constrained thermal parameters. The molecular diagrams were generated by ORTEP-3 The X-ray diffraction data were collected either using an APEXII Bruker-AXS diffractometer refinements were performed with full-matrix least-square methods based on *F* 2 (*SHELXL*) [49]. In the (version 2.02) [52].

3.3. Deprotometalation Followed by Trapping with Electrophiles

3.3.1. General Procedure 1

To a solution of 2,2,6,6-tetramethylpiperidine (0.51 mL, 3.0 mmol) in THF (3 mL) at 0 °C were successively added BuLi (about 1.6 M hexanes solution, 3.0 mmol) and, 15 min later, $ZnCl_2$ ·TMEDA [53] (0.25 g, 1.0 mmol). After 15 min at 0 \degree C, the pyrazine (2.0 mmol) was introduced, and the mixture was stirred for 2 h at rt before addition of I₂ (0.76 g, 3.0 mmol) in THF (3 mL) at 0 °C. The mixture was

stirred at this temperature for 1 h before addition of an aqueous saturated solution of $\text{Na}_2\text{S}_2\text{O}_3$ (10 mL) and extraction with EtOAc (3×20 mL). The combined organic layers were dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by chromatography over silica gel (the eluent is given in the product description).

3.3.2. 5-Iodo-2,3-diphenylquinoxaline (**1b**)

The general procedure 1 using 2,3-diphenylquinoxaline (**1a**, 0.56 g) gave **1b** (eluent: heptane-CH₂Cl₂ 60:40; R_f = 0.55) in 74% yield as a pale yellow powder. Mp: 148 °C. IR: 486, 529, 551, 602, 689, $695, 701, 763, 776, 796, 892, 978, 1023, 1068, 1079, 1184, 1281, 1336, 1384, 1497, 1534, 3051 cm⁻¹. ¹H-NMR$ (CDCl3): 7.31–7.41 (m, 6H), 7.48 (dd, 1H, *J* = 8.3 and 7.4 Hz), 7.54–7.57 (m, 2H), 7.64–7.67 (m, 2H), 8.15 (dd, 1H, *J* = 8.4 and 1.3 Hz), 8.36 (dd, 1H, *J* = 7.4 and 1.3 Hz). ¹³C-NMR (CDCl₃): 102.8 (C), 128.3 (2CH), 128.5 (2CH), 129.2 (CH), 129.3 (CH), 129.9 (2CH), 130.0 (CH), 130.5 (2CH), 131.0 (CH), 138.2 (C), 138.7 (C), 140.1 (CH), 140.9 (C), 141.3 (C), 153.9 (C), 154.1 (C). Anal. Calc. for C₂₀H₁₃IN₂ (408.24): C 58.84, H 3.21, N, 6.86. Found: C 59.05, H 3.27, N, 6.70. 5,8-Diiodo-2,3-diphenylquinoxaline (1b') was similarly isolated (eluent: heptane-CH₂Cl₂ 60:40; R_f = 0.69) in 7% yield as a yellow powder. Mp: 222 ◦C. IR: 533, 572, 613, 649, 692, 771, 824, 893, 978, 1025, 1055, 1077, 1169, 1209, 1325, 1383, 1447, 2930, 3059 cm^{−1}. ¹H-NMR (CDCl₃): 7.34–7.45 (m, 6H), 7.64–7.76 (m, 4H), 8.02 (s, 2H). ¹³C-NMR (CDCl₃): 103.5 (2C), 128.4 (4CH), 129.7 (2CH), 130.4 (4CH), 137.7 (2C), 140.6 (2CH), 140.8 (2C), 154.5 (2C). *Crystal* data for **1b'**. C₂₀H₁₂I₂N₂, *M* = 534.12, *T* = 150(2) K, monoclinic, *P* 2₁, *a* = 10.1153(9), *b* = 5.8725(5), *c* = 14.9603(14) Å, β = 98.489(4) °, *V* = 878.94(14) Å³, *Z* = 2, *d* = 2.018 g cm⁻³, μ = 3.581 mm⁻¹. A final refinement on F^2 with 3888 unique intensities and 217 parameters converged at $\omega R(F^2)$ = 0.0701 (*R*(*F*) = 0.0343) for 3602 observed reflections with *I* > 2σ(*I*). CCDC 1858478.

3.3.3. 8-Iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**)

The general procedure 1 using 2,3-diphenylpyrido[2,3-*b*]pyrazine (**2a**, 0.57 g) gave **2b-I** (eluent: CH₂Cl₂; R_f = 0.34) in 70% yield as a whitish powder. Mp: 220 °C. IR: 534, 562, 613, 624, 637, 699, 980, 1023, 1336, 1416, 1519, 1570, 3068 cm^{−1}. ¹H-NMR (CDCl₃): 7.32–7.44 (m, 6H), 7.64–7.69 (m, 4H), 8.28 (d, 1H, *J* = 4.5 Hz), 8.70 (d, 1H, *J* = 4.6 Hz). ¹³C-NMR (CDCl₃): 116.1 (C), 128.3 (2CH), 128.4 (2CH), 129.7 (CH), 129.8 (CH), 130.3 (2CH), 130.3 (2CH), 135.6 (CH), 136.6 (C), 137.6 (C), 137.6 (C), 149.1 (C), 153.6 (CH), 155.0 (C), 157.1 (C). Anal. Calc. for C₁₉H₁₂IN₃ (409.23): C 55.77, H 2.96, N, 10.27. Found: C 55.91, H 3.06, N, 10.03.

3.3.4. 8-Bromo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-Br**)

To a stirred mixture of 2,3-diphenyl pyrido[2,3-b]pyrazine ($2a$, 0.28 g, 1.0 mmol) and $ZnCl_2$ · TMEDA [53] (0.26 g, 1.0 mmol) in THF (1 mL) at -20 °C was added dropwise a solution of LiTMP (prepared by adding BuLi (about 1.6 M hexanes solution, 1.2 mmol) to a stirred, cooled (−20 ◦C) solution of 2,2,6,6-tetramethylpiperidine (0.24 mL, 1.2 mmol) in THF (2 mL) and stirring for 15 min) cooled at −20 °C. After 30 min at −20 °C, Br₂ (97 µL, 2.0 mmol) was introduced, and the mixture was stirred for 1 h before addition of an aqueous saturated solution of $Na_2S_2O_3$ (5 mL) and extraction with EtOAc (3×20 mL). The combined organic layers were dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by chromatography over silica gel (eluent: CH₂Cl₂-EtOAc 90:10; R_f = 0.50) to give 2**b-Br** in 60% yield as a whitish powder. Mp: 183 °C. IR: 491, 538, 563, 615, 625, 649, 698, 767, 839, 985, 1021, 1049, 1090, 1179, 1241, 1336, 1387, 1421, 1460, 1524, 1584, 3067 cm−¹ . ¹H-NMR (CDCl3): 7.32–7.42 (m, 6H), 7.63–7.66 (m, 4H), 8.00 (d, 1H, *J* = 4.7 Hz), 8.91 (d, 1H, *J* = 4.7 Hz). ¹³C-NMR (CDCl₃): 128.3 (CH), 128.4 (CH), 128.7 (CH), 129.7 (CH), 129.8 (CH), 130.3 (CH), 130.3 (CH), 134.7 (C), 136.3 (C), 137.7 (C), 137.9 (C), 150.1 (C), 153.4 (CH), 154.9 (C), 157.0 (C). Anal. Calc. for C₁₉H₁₂BrN₃ (362.23): C 63.00, H 3.34, N, 11.60. Found: C 63.24, H 3.58, N, 11.43.

3.3.5. 8-Chloro-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-Cl**)

To a stirred mixture of 2,3-diphenyl pyrido[2,3-b]pyrazine (2a, 0.28 g, 1.0 mmol) and ZnCl₂· TMEDA [53] (0.26 g, 1.0 mmol) in THF (1 mL) at -20 °C was added dropwise a solution of LiTMP (prepared by adding BuLi (about 1.6 M hexanes solution, 1.2 mmol) to a stirred, cooled (−20 ◦C) solution of 2,2,6,6-tetramethylpiperidine (0.24 mL, 1.2 mmol) in THF (2 mL) and stirring for 15 min) cooled at −20 ◦C. After 30 min at −20 ◦C, trichloroisocyanuric acid (0.30 g, 1.3 mmol) was introduced (CAUTION: dissolution of trichloroisocyanuric acid in THF at a temperature above $0 °C$ produces intense heat), and the mixture was stirred at this temperature for 1 h before addition of water (5 mL) and extraction with EtOAc (3×20 mL). The combined organic layers were dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by chromatography over silica gel (eluent: CH2Cl2-EtOAc 90:10; R^f = 0.60) to give **2b-Cl** in 62% yield as a whitish powder. Mp: 180 ◦C. IR: 534, 544, 617, 658, 699, 770, 851, 991, 1025, 1055, 1193, 1242, 1341, 1388, 1422, 1442, 1452, 1532, 1583, 3034, 3051 cm−¹ . ¹H-NMR (CDCl3): 7.31–7.44 (m, 6H), 7.62–7.66 (m, 4H), 7.79 (d, 1H, *J* = 4.7 Hz), 9.02 (d, 1H, *J* = 4.7 Hz). ¹³C-NMR (CDCl₃): 125.1 (CH), 128.3 (CH), 128.5 (CH), 129.7 (CH), 129.8 (CH), 130.2 (CH), 130.3 (CH), 133.7 (C), 137.8 (C), 138.1 (C), 144.5 (C), 150.5 (C), 153.3 (CH), 154.8 (C), 157.1 (C). Anal. Calc. for C₁₉H₁₂ClN₃ (317.78): C 71.81, H 3.81, N, 13.22. Found: C 71.77, H 3.85, N, 13.14.

3.3.6. General Procedure 2

To a stirred mixture of the pyrazine (1.0 mmol) and ZnCl2·TMEDA [53] (0.26 g, 1.0 mmol) in THF (1 mL) at -20 °C was added dropwise a solution of LiTMP (prepared by adding BuLi (about 1.6 M hexanes solution, 1.2 mmol) to a stirred, cooled (−20 °C) solution of 2,2,6,6-tetramethylpiperidine (0.24 mL, 1.2 mmol) in THF (2 mL) and stirring for 15 min) cooled at -20 °C. After 30 min at -20 °C, I² (0.37 g, 1.5 mmol) in THF (2 mL) was introduced, and the mixture was stirred at this temperature for 1 h before addition of an aqueous saturated solution of $Na₂SO₃$ (5 mL) and extraction with EtOAc $(3 \times 20 \text{ mL})$. The combined organic layers were dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by chromatography over silica gel (the eluent is given in the product description).

3.3.7. 8-Bromo-7-iodo-2,3-diphenylpyrido[3,4-*b*]pyrazine (**3b**)

The general procedure 2 using 8-bromo-2,3-diphenylpyrido[3,4-*b*]pyrazine (**3a** [25,26], 0.36 g) gave **3b** (eluent: CH₂Cl₂-petroleum ether 80:20; R_f = 0.44) in 67% yield as a red powder. Mp: 186–188 ◦C. IR: 493, 529, 559, 600, 658, 695, 765, 984, 1025, 1055, 1117, 1238, 1315, 1373, 1399, 1446, 1493, 1551, 3034, 3060 cm−¹ . ¹H-NMR (CDCl3): 7.34–7.47 (m, 6H), 7.54–7.57 (m, 2H), 7.62–7.65 (m, 2H), 9.27 (s, 1H). ¹³C-NMR (CDCl3): 121.7 (C), 128.6 (2CH), 128.7 (2CH), 129.7 (C), 129.8 (2CH), 130.1 (CH), 130.5 (2CH), 130.5 (CH), 136.0 (C), 137.4 (C), 137.7 (C), 142.3 (C), 152.5 (CH), 156.2 (C), 158.6 (C). Anal. Calc. for C₁₉H₁₁BrIN₃ (488.13): C 46.75, H 2.27, N, 8.61. Found: C 46.89, H 2.49, N, 8.55. 8-Bromo-5,7-diiodo-2,3-diphenyl pyrido[3,4-*b*]pyrazine, also formed in <5% yield, was identified by its ¹H-NMR (CDCl₃): 7.36–7.47 (m, 6H), 7.54–7.57 (m, 2H), 7.65–7.68 (m, 4H).

3.3.8. 7-Bromo-6-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**4b**)

The general procedure 2 using 7-bromo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**4a** [54], prepared in 90% yield [6], 0.36 g) gave **4b** (eluent: CH₂Cl₂-heptane 70:30; R_f (heptane-CH₂Cl₂ 80:20) = 0.80) in 5% yield as a yellow powder. Mp: 150–152 ◦C. IR: 495, 547, 596, 615, 697, 731, 770, 778, 903, 937, 1025, 1060, 1107, 1178, 1274, 1332, 1390, 1448, 1502, 1562, 1603, 1699, 1768, 2734, 2940, 3064 cm^{−1}. ¹H-NMR (CDCl3): 7.30–7.43 (m, 6H), 7.52–7.55 (m, 2H), 7.59–7.62 (m, 2H), 8.62 (s, 1H). ¹³C-NMR (CDCl3): 128.3 (2CH), 128.6 (2CH), 128.7 (C), 129.9 (CH), 129.9 (2CH), 130.0 (CH), 130.0 (C), 130.3 (2CH), 135.5 (C), 137.6 (C), 138.0 (C), 139.5 (CH), 148.0 (C), 155.9 (C), 157.1 (C). Anal. Calc. for C₁₉H₁₁BrIN₃ (488.13): C 46.75, H 2.27, N, 8.61. Found: C 46.93, H 2.38, N, 8.49.

3.4. Suzuki Coupling Reactions

3.4.1. General Procedure 3

To a stirred mixture of the iodide (0.50 mmol) and $Pd(PPh₃)₄$ (29 mg, 25 µmol) in degassed 1,2-dimethoxyethane (5 mL) was added the boronic acid (0.60 mmol) and NaHCO₃ (2.0 mmol) in degassed water (1.6 mL). The resulting mixture was heated at 80 °C for 3 h and cooled to rt before addition of water (5 mL) and extraction with EtOAc (3×10 mL). The combined organic layers were washed with brine (10 mL), dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by chromatography over silica gel (the eluent is given in the product description).

3.4.2. 2,3,5-Triphenylquinoxaline (**1c**)

The general procedure 3 using 5-iodo-2,3-diphenyl quinoxaline **(1b**, 0.20 g) and phenylboronic acid (73 mg) gave **1c** (eluent: CH₂Cl₂-heptane 60:40; $R_f = 0.35$) in 42% yield as a white powder. Mp: 150 ◦C. IR: 763, 804, 841, 927, 984, 1023, 1081, 1128, 1233, 1336, 1388, 1433, 1444, 1491, 1566, 2858, 2927, 2965, 3064 cm−¹ . ¹H-NMR (CDCl3): 7.26–7.34 (m, 3H), 7.36–7.48 (m, 4H), 7.52–7.64 (m, 6H), 7.81–7.89 (m, 4H), 8.21 (dd, 1H, *J* = 7.3 and 2.5 Hz). ¹³C-NMR (CDCl₃): 127.7 (CH), 128.0 (2CH), 128.1 (2CH), 128.5 (2CH), 128.7 (CH), 128.8 (CH), 129.0 (CH), 129.8 (CH), 129.9 (2CH), 130.3 (2CH), 130.4 (CH), 131.1 (2CH), 138.4 (C), 139.0 (C), 139.1 (C), 139.4 (C), 140.6 (C), 141.3 (C), 152.4 (C), 152.9 (C). Anal. Calc. for $C_{26}H_{18}N_2$ (358.44): C 87.12, H 5.06, N, 7.82. Found: C 87.25, H 5.22, N, 7.70.

3.4.3. 2,3-Diphenyl-5-(2-thienyl)quinoxaline (**1d**)

The general procedure 3 using 5-iodo-2,3-diphenylquinoxaline (**1b**, 0.20 g) and 2-thienylboronic acid (77 mg) gave **1d** (eluent: CH₂Cl₂-heptane 60:40; R_f = 0.20) in 97% yield as a yellow powder. Mp: 210 °C. IR: 738, 766, 796, 828, 854, 916, 933, 969, 1025, 1053, 1083, 1163, 1238, 1336, 1390, 1442, 1495, 1562, 1592, 3064 cm−¹ . ¹H-NMR (CDCl3): 7.18 (dd, 1H, *J* = 5.1 and 3.7 Hz), 7.32–7.40 (m, 6H), 7.51 (dd, 1H, *J* = 5.1 and 1.2 Hz), 7.58–7.61 (m, 2H), 7.67–7.70 (m, 2H), 7.76 (dd, 1H, *J* = 8.3 and 7.4 Hz), 7.88 (dd, 1H, *J* = 3.7 and 1.2 Hz), 8.08 (dd, 1H, *J* = 8.3 and 1.3 Hz), 8.13 (dd, 1H, *J* = 7.4 and 1.3 Hz). ¹³C-NMR (CDCl3): 126.7 (CH), 126.9 (CH), 127.5 (CH), 128.1 (CH), 128.2 (2CH), 128.5 (2CH), 128.8 (CH), 129.0 (CH), 129.0 (CH), 129.9 (2CH), 129.9 (CH), 130.6 (2CH), 133.0 (C), 137.6 (C), 138.8 (C), 138.9 (C), 139.2 (C), 141.4 (C), 152.3 (C), 153.2 (C). Anal. Calc. for C₂₄H₁₆N₂S (364.47): C 79.09, H 4.43, N, 7.69. Found: C 79.11, H 4.48, N, 7.72.

3.4.4. 2,3-Diphenyl-8-(2-thienyl)pyrido[2,3-*b*]pyrazine (**2d**)

The general procedure 3 using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.20 g) and 2-thienylboronic acid (77 mg) gave **2d** (eluent: CH₂Cl₂-EtOAc 95:5; R_f = 0.50) in 75% yield as a pale yellow powder. Mp: 215 ◦C. IR: 540, 695, 744, 1025, 1096, 1120, 1188, 1238, 1336, 1384, 1435, 1480, 551, 1568, 2927, 2965, 3060 cm−¹ . ¹H-NMR (CDCl3): 7.23 (dd, 1H, *J* = 5.1 and 3.8 Hz), 7.32–7.45 (m, 6H), 7.66–7.72 (m, 5H), 7.98 (d, 1H, *J* = 4.8 Hz), 8.10 (dd, 1H, *J* = 3.8 and 1.2 Hz), 9.10 (d, 1H, *J* = 4.8 Hz). ¹³C-NMR (CDCl₃): 120.3 (CH), 127.2 (CH), 128.3 (2CH), 128.4 (2CH), 129.2 (CH), 129.4 (CH), 129.5 (CH), 130.3 (2CH), 130.5 (2CH), 132.4 (CH), 132.4 (C), 136.0 (C), 138.2 (C), 138.3 (C), 140.9 (C), 150.2 (C), 153.1 (C), 153.9 (CH), 155.8 (C). *Crystal data for* **2d.** C23H15N3S, *M* = 365.44, *T* = 150(2) K, triclinic, *P* 1, *a* = 6.6311(18), *b* = 9.939(3), *c* = 13.655(4) Å, α = 81.914(12), β = 80.405(11), γ = 89.955(10) ◦ , *V* = 878.3(4) Å³, *Z* = 2, *d* = 1.382 g cm⁻³, μ = 0.197 mm⁻¹. A final refinement on F^2 with 7113 unique intensities and 236 parameters converged at $\omega R(F^2)$ = 0.3351 ($R(F)$ = 0.1327) for 6147 observed reflections with $I > 2σ(I)$. CCDC 1858479.

3.4.5. 5-(2-Aminophenyl)-2,3-diphenylquinoxaline (**1e**)

The general procedure 3 using 5-iodo-2,3-diphenylquinoxaline **(1b**, 0.20 g) and 2-aminophenylboronic acid (82 mg) gave **1e** (eluent: heptane-CH₂Cl₂ 70:30; R_f = 0.31) in 92% yield as a yellow powder. Mp: 178 ◦C. IR: 689, 702, 740, 771, 977, 1307, 1342, 1492, 1626, 3025, 3060, 3212, 3328, 3468 cm−¹ . ¹H-NMR (CDCl3): 3.87 (br s, 2H, NH2), 6.85 (dd, 1H, *J* = 7.9 and 1.1 Hz), 6.92 (td, 1H, *J* = 7.4 and 1.2 Hz), 7.21–7.30 (m, 5H), 7.35–7.40 (m, 3H), 7.47–7.50 (m, 2H), 7.55–7.58 (m, 2H), 7.78–7.86 (m, 2H), 8.20 (dd, 1H, $J = 7.8$ and 2.1 Hz). ¹³C-NMR (CDCl₃): 116.5 (CH), 118.8 (CH), 125.7 (C), 128.1 (2CH), 128.5 (2CH), 128.9 (CH), 129.0 (CH), 129.0 (CH), 129.0 (CH), 129.9 (2CH), 130.2 (CH), 130.3 (2CH), 132.0 (CH), 132.3 (CH), 138.8 (C), 139.2 (C), 139.3 (C), 139.6 (C), 141.3 (C), 145.0 (C), 152.5 (C), 153.2 (C). Anal. Calc. for C₂₆H₁₉N₃ (373.46): C 83.62, H 5.13, N, 11.25. Found: C 83.81, H 5.26, N, 11.17.

3.4.6. 8-(2-Aminophenyl)-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2e**)

The general procedure 3 using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine **(2b-I**, 0.20 g) and 2-aminophenylboronic acid (82 mg) gave **2e** (eluent: CH₂Cl₂-EtOAc 70:30; R_f = 0.50) in 73% yield as a yellow powder. Mp: 205 ◦C. IR: 687, 742, 766, 854, 981, 1015, 1047, 1237, 1307, 1382, 1489, 1623, 3024, 3055, 3345 cm⁻¹. ¹H-NMR (CDCl₃): 3.99 (br s, 2H, NH₂), 6.87 (dd, 1H, *J* = 8.4 and 1.2 Hz), 6.94 (td, 1H, *J* = 7.4 and 1.2 Hz), 7.25–7.40 (m, 8H), 7.51–7.54 (m, 2H), 7.65–7.68 (m, 2H), 7.73 (d, 1H, *J* = 4.5 Hz), 9.19 (d, 1H, *J* = 4.4 Hz). ¹³C-NMR (CDCl₃): 116.9 (CH), 118.7 (CH), 122.6 (C), 126.3 (CH), 128.2 (2CH), 128.2 (2CH), 129.3 (CH), 129.5 (CH), 130.1 (CH), 130.1 (2CH), 130.1 (2CH), 132.0 (CH), 134.3 (C), 138.1 (C), 138.2 (C), 144.9 (C), 149.0 (C), 149.8 (C), 153.4 (C), 154.1 (CH), 155.7 (C). Anal. Calc. for C25H18N⁴ (374.45): C 80.19, H 4.85, N, 14.96. Found: C 80.07, H 4.87, N, 14.85.

3.4.7. 2,3-Diphenyl-11*H*-pyrazino[2⁰ ,30 :4,5]pyrido[2,3-*d*]indole (**3h**)

In a tube containing a stirred mixture of 8-bromo-7-iodo-2,3-diphenylpyrido[3,4-*b*]pyrazine (**3b**, 0.24 g, 0.50 mmol) and Pd(PPh₃)₄ (29 mg, 25 µmol) in degassed 1,2-dimethoxyethane (5 mL) was introduced 2-aminophenylboronic acid (82 mg, 0.60 mmol) and Na_2CO_3 (2.0 mmol) in degassed water (1.6 mL). The sealed tube was heated overnight at 140 °C and cooled to rt before addition of saturated aqueous NaHCO₃ (5 mL) and extraction with EtOAc (3 \times 10 mL). The combined organic layers were washed with brine (10 mL), dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by chromatography over silica gel (eluent: CH_2Cl_2 -EtOAc 90:10; R^f = 0.28) to give **3h** in 65% yield as a yellow powder. Mp: 284–286 ◦C. IR: 695, 748, 763, 1025, 1092, 1190, 1236, 1315, 1328, 1336, 1376, 1446, 1495, 1540, 1624, 3034, 3064, 3420 cm⁻¹. ¹H-NMR (CDCl3): 7.30–7.42 (m, 7H), 7.50–7.60 (m, 6H), 8.45 (d, 1H, *J* = 7.9 Hz), 9.47 (s, 1H), 9.78 (br s, 1H). ¹³C-NMR (CDCl₃): 111.9 (CH), 120.6 (CH), 121.3 (CH), 123.1 (C), 126.7 (C), 127.2 (CH), 128.4 (2CH), 128.5 (2CH), 129.2 (CH), 129.6 (CH), 130.0 (2CH), 130.1 (2CH), 132.4 (C), 134.9 (C), 138.4 (C), 138.6 (C), 138.8 (C), 139.5 (C), 146.5 (CH), 153.6 (C), 155.9 (C). *Crystal data for* **3h.** C25H16N4, *M* = 372.42, *T* = 150(2) K, orthorhombic, *Pbca*, *a* = 7.1524(9), *b* = 16.3313(17), *c* = 33.798(4) Å, *V* = 3947.9(8) Å³ , *Z* = 8, $d = 1.253$ g cm⁻³, $\mu = 0.076$ mm⁻¹. A final refinement on F^2 with 4429 unique intensities and 265 parameters converged at $ωR(F^2) = 0.1564 (R(F) = 0.0739)$ for 3511 observed reflections with $I > 2σ(I)$. CCDC 1858477. This compound was also obtained in 64% yield under microwave irradiation (300 W; Monowave 300, Anton Paar, Graz, Austria) for 30 min at 150 ◦C.

3.5. 8-(2-Azidophenyl)-2,3-diphenylpyrido[2,3-b]pyrazine

To a stirred solution of 8-(2-aminophenyl)-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2e**, 94 mg, 0.25 mmol) in acetic acid (1.5 mL) at 0 °C was added 1M aqueous NaNO₂ (0.35 mL, 0.35 mmol). After stirring for 1 h at rt, the solution was cooled to 0 °C before addition of 1M aqueous NaN₃ (0.35 mL, 0.35 mmol). After stirring overnight at rt, 3 mL of saturated aqueous NaHCO₃ were added. Extraction with EtOAc $(3 \times 10 \text{ mL})$, washing of the combined organic layers with brine (10 mL),

drying over MgSO4, filtration and concentration under reduced pressure afforded a brown powder which was purified by chromatography over silica gel (eluent: CH_2Cl_2 -EtOAc 95:5; R_f = 0.50) to afford the azide in 64% yield. IR: 685, 745, 1288, 1440, 1577, 2088, 2124, 3064 cm $^{-1}$. 1 H-NMR (CDCl $_3$): 7.23–7.40 (m, 8H), 7.45–7.57 (m, 4H), 7.66–7.69 (m, 3H), 9.18 (d, 1H, *J* = 4.4 Hz). ¹³C-NMR (CDCl₃): 118.8 (CH), 124.7 (CH), 126.0 (CH), 127.9 (C), 128.2 (2CH), 128.2 (2CH), 129.2 (CH), 129.5 (CH), 130.1 (2CH), 130.3 (2CH), 130.4 (CH), 132.5 (CH), 134.5 (C), 138.3 (C), 138.5 (C), 138.7 (C), 146.7 (C), 149.8 (C), 153.5 (CH), 153.7 (C), 155.8 (C).

3.6. Palladium-Catalyzed N-arylation

3.6.1. General Procedure 4

To a stirred mixture of the halide (0.50 mmol) and Cs_2CO_3 (0.48 g, 1.5 mmol) in 2-chloroaniline (63) μ L, 0.60 mmol) was added a solution of the catalyst prepared by stirring Pd₂(dba)₃ (11 mg, 12.5 μ mol) and Xantphos (16 mg, 27.5 µmol) in degassed dioxane (2 mL) for 10 min at rt. The resulting mixture was heated at 110 °C for 24 h and cooled to rt before addition of water (5 mL) and extraction with EtOAc (3×10 mL). The combined organic layers were washed with brine (10 mL), dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by chromatography over silica gel (the eluent is given in the product description).

3.6.2. 5-(2-Chlorophenylamino)-2,3-diphenylquinoxaline (**1f**)

The general procedure 4 using 5-iodo-2,3-diphenylquinoxaline (**1b**, 0.20 g) gave **1f** (eluent: heptane-CH₂Cl₂ 60:40; R_f = 0.42) in 92% yield as a yellow powder. Mp: 182 °C. IR: 695, 729, 748, 959, 1021, 1055, 1072, 1098, 1182, 1218, 1317, 1343, 1356, 1394, 1442, 1454, 1497, 1534, 1562, 1579, 1594, 1613, 3060, 3347 cm−¹ . ¹H-NMR (CDCl3): 6.95 (td, 1H, *J* = 7.7 and 1.4 Hz), 7.27–7.39 (m, 7H), 7.46–7.52 (m, 2H), 7.55–7.62 (m, 4H), 7.64–7.66 (m, 2H), 7.71 (dd, 1H, *J* = 8.2 and 1.4 Hz), 8.56 (br s, 1H). ¹³C-NMR (CDCl3): 109.1 (CH), 118.9 (CH), 118.9 (CH), 122.6 (CH), 125.1 (C), 127.6 (CH), 128.2 (2CH), 128.4 (2CH), 128.9 (CH), 128.9 (CH), 129.9 (2CH), 130.1 (2CH), 130.2 (CH), 130.9 (CH), 132.1 (C), 138.4 (C), 138.9 (C), 139.2 (C), 139.3 (C), 141.9 (C), 150.4 (C), 153.9 (C). Anal. Calc. for C₂₆H₁₈ClN₃ (407.90): C 76.56, H 4.45, N, 10.30. Found: C 76.89, H 4.58, N, 10.13.

3.6.3. 8-(2-Chlorophenylamino)-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2f**)

The general procedure 4 using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.20 g) gave **2f** (eluent: CH₂Cl₂-EtOAc 90:10; R_f = 0.32) in 67% yield as a yellow powder. Mp: 202 °C. IR: 542, 699, 755, 1021, 1102, 1242, 1313, 1336, 1356, 1437, 1452, 1534, 1558, 1583, 1646, 3060, 3322, 3631 cm^{−1}. ¹H-NMR (CDCl3): 7.13–7.19 (m, 2H), 7.30–7.41 (m, 7H), 7.53 (dd, 1H, *J* = 8.0 and 1.5 Hz), 7.57–7.65 (m, 4H), 7.68 (dd, 1H, *J* = 8.1 and 1.5 Hz), 8.77 (br s, 1H, NH), 8.81 (d, 1H, *J* = 5.4 Hz, H₆). ¹³C-NMR (CDCl₃): 102.8 (CH), 122.2 (CH), 125.5 (CH), 127.2 (C), 127.5 (C), 127.8 (CH), 128.2 (2CH), 128.4 (2CH), 129.2 (CH), 129.4 (CH), 130.0 (2CH), 130.3 (2CH), 130.5 (CH), 136.2 (C), 138.4 (C), 138.5 (C), 147.0 (C), 150.3 (C), 151.1 (C), 155.0 (CH), 156.5 (C). *Crystal data for* **2f.** C₂₅H₁₇ClN₄, *M* = 408.88, *T* = 150(2) K, orthorhombic, *P c a* 2₁, *a* = 15.3485(15), *b* = 18.8937(16), *c* = 6.9936(7) Å, *V* = 2028.1(3) Å³, Z = 4, *d* = 1.339 g cm^{−3}, μ = 0.208 mm $^{-1}$. A final refinement on F^2 with 4578 unique intensities and 274 parameters converged at ω *R*(*F*²) = 0.1478 (*R*(*F*) = 0.0583) for 4133 observed reflections with *I* > 2σ(*I*). CCDC 1858474.

3.7. Palladium-Catalyzed N-arylation

2,3-Diphenyl-11*H*-pyrazino[2,3-*a*]carbazole (**1g**) was prepared by adapting a reported procedure [40]. To a stirred mixture of 5-(2-chlorophenylamino)-2,3-diphenylquinoxaline (**1f**, 0.24 g, 0.60 mmol) and 1,8-diazabicyclo[5.4.0]undec-7-ene (0.13 mL, 0.90 mmol), was added a solution of the catalyst prepared by stirring $Pd_2(dba)$ ₃ (14 mg, 15 µmol) and $P(fBu)$ ₃ (12 mg, 60 µmol) in degassed dioxane (1 mL) for 10 min at rt. The resulting mixture was heated by microwave irradiation (300 W; Monowave 300, Anton Paar, Graz, Austria) for 10 min at 180 °C before addition of water (5 mL)

and extraction with EtOAc (3×10 mL). The combined organic layers were washed with brine (10 mL), dried over MgSO4, filtered and concentrated under reduced pressure. The crude product was purified by chromatography over silica gel (eluent: heptane-CH₂Cl₂ 60:40; R_f = 0.48) to give **1g** in 62% yield as a yellow powder. Mp: 260 ◦C. IR: 1025, 1087, 1102, 1175, 1242, 1326, 1347, 1362, 1384, 1444, 1459, 1624, 1731, 2854, 2922, 3420 cm−¹ . ¹H-NMR ((CD3)2SO): 6.86 (ddd, 1H, *J* = 8.0, 7.1 and 1.0 Hz), 6.91–6.97 (m, 6H), 7.01–7.10 (m, 3H), 7.14–7.17 (m, 2H), 7.28 (d, 1H, *J* = 8.3 Hz), 7.39 (d, 1H, *J* = 8.7 Hz), 7.84 (d, 1H, *J* = 7.8 Hz), 8.14 (d, 1H, *J* = 8.7 Hz), 12.13 (br s, 1H). ¹³C-NMR ((CD₃)₂SO): 112.2 (CH), 118.8 (CH), 119.7 (CH), 120.3 (CH), 120.7 (C), 122.7 (C), 124.2 (CH), 125.6 (CH), 128.0 (2CH), 128.0 (2CH), 128.5 (CH), 128.6 (CH), 129.7 (2CH), 129.9 (2CH), 130.0 (C), 134.3 (C), 139.0 (C), 139.2 (C), 139.8 (C), 139.9 (C), 150.7 (C), 151.4 (C). Anal. Calc. for C₂₆H₁₇N₃ (371.44): C 84.07, H 4.61, N, 11.31. Found: C 84.19, H 4.52, N, 11.12.

3.8. One-Pot Palladium-Catalyzed N-arylation/C-H Arylation

3.8.1. General Procedure 5

To a mixture of the halide (0.25 mmol), 1,8-diazabicyclo[5.4.0]undec-7-ene (118 µL, 0.75 mmol), 2-chloroaniline (38 mg, 0.30 mmol), $Pd_2(dba)$ ₃ (9.2 mg, 10 µmol) and Xantphos (13 mg, 22 µmol), was added degassed 1,4-dioxane (1 mL). The mixture was heated by microwave irradiation (150 W; Monowave 300, Anton Paar, Graz, Austria) under the conditions given in the product description. The cooled residue was taken up with EtOAc (20 mL). The organic layer was washed with brine (10 mL) , dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by chromatography over silica gel (the eluent is given in the product description).

3.8.2. 2,3-Diphenyl-11*H*-pyrazino[2⁰ ,30 :5,6]pyrido[4,3-*b*]indole (**2g**)

The general procedure 5 (1 h at 180 ◦C) using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.10 g) gave **2g** (eluent: CH₂Cl₂-EtOAc 90:10; R_f = 0.43) in 70% yield as a white powder. Mp > 260 °C. IR: 525, 542, 551, 626, 699, 750, 768, 1025, 1045, 1075, 1100, 1236, 1339, 1373, 1444, 1555, 1736, 2665, 3056 cm−¹ . ¹H-NMR ((CD3)2SO): 7.40–7.45 (m, 7H), 7.55–7.63 (m, 5H), 7.77 (dd, 1H, *J* = 8.2 and 0.9 Hz), 8.42 (dt, 1H, *J* = 7.8 and 1.0 Hz), 9.87 (s, 1H), 13.19 (s, 1H). ¹³C-NMR ((CD₃)₂SO): 112.6 (CH), 118.2 (C), 120.7 (CH), 121.3 (CH), 121.4 (C), 126.4 (C), 126.5 (CH), 128.0 (CH), 128.0 (2CH), 128.1 (2CH), 128.8 (CH), 129.8 (2CH), 129.9 (2CH), 138.6 (C), 138.7 (C), 139.4 (C), 140.0 (C), 147.4 (C), 148.5 (CH), 151.4 (C), 153.4 (C). Anal. Calc. for C₂₅H₁₆N₄ (372.43): C 80.63, H 4.33, N, 15.04. Found: C 80.54, H 4.28, N, 14.89.

3.8.3. 7-(Phenylamino)-2,3-diphenylpyrido[3,4-b]pyrazine (3g')

The general procedure 5 (40 min at 180 ◦C) using 8-bromo-7-iodo-2,3-diphenylpyrido[3,4-*b*] pyrazine (**3b**, 0.12 g) gave **3g'** (eluent: CH₂Cl₂-MeOH 99:1; R_f = 0.27) in 32% yield as a yellow powder. Mp: 224–226 ◦C. IR: 699, 750, 770, 978, 1025, 1057, 1077, 1169, 1197, 1261, 1336, 1349, 1435, 1450, 1527, 1555, 1588, 1613, 2854, 2927, 2961, 3025, 3232 cm−¹ . ¹H-NMR (CDCl3): 7.14 (p, 1H, *J* = 4.4 Hz), 7.23 (br s, 1H), 7.29–7.49 (m, 15H), 9.26 (s, 1H). ¹³C-NMR (CDCl₃): 158.3 (C), 155.8 (C), 154.0 (CH), 151.4 (C), 146.3 (C), 139.8 (C), 138.8 (C), 138.7 (C), 132.4 (C), 129.8 (2CH), 129.7 (2CH), 129.7 (2CH), 129.5 (CH), 128.9 (CH), 128.4 (2CH), 128.4 (2CH), 124.1 (CH), 121.4 (2CH), 98.3 (CH). Anal. Calc. for C₂₅H₁₈N₄ (374.45): C 80.19, H 4.85, N, 14.96. Found: C 80.17, H 4.99, N, 14.84.

3.9. Copper-Catalyzed N-arylation

3.9.1. General Procedure 6

A mixture containing the iodide (0.50 mmol) and azole (1.0 mmol), $Cu₂O$ (6.0 mg, 0.10 mmol), Cs₂CO₃ (0.33 g, 1.0 mmol) and DMSO (0.5 mL) was stirred at 110 °C for 24 h. The cooled residue was taken up with EtOAc (20 mL) and filtered through a Celite pad. The organic layer was washed with water (10 mL) and brine (10 mL), dried over $MgSO₄$, filtered and concentrated under reduced pressure. The crude product was purified by chromatography over silica gel (the eluent is given in the product description).

3.9.2. 2,3-Diphenyl-8-(*N*-pyrrolyl)pyrido[2,3-*b*]pyrazine (**2i**)

The general procedure 6 using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.20 g) and pyrrole (67 mg) gave **2i** (eluent: CH₂Cl₂-EtOAc 90:10; R_f = 0.47) in 67% yield as a yellow powder. Mp: 210 °C. IR: 946, 1025, 1072, 1096, 1107, 1173, 1238, 1289, 1328, 1362, 1388, 1433, 1454, 1482, 1549, 1588, 3025, 3060, 3111, 3141, 3180 cm $^{-1}$. 1 H-NMR (CDCl₃): 6.39–6.40 (m, 2H), 7.24–7.35 (m, 6H), 7.49–7.53 (m, 3H), 7.59–7.62 (m, 2H), 7.65–7.66 (m, 2H), 9.01 (d, 1H, $J = 5.0$ Hz). ¹³C-NMR (CDCl₃): 112.0 (2CH), 115.1 (CH), 122.7 (2CH), 128.3 (2CH), 128.4 (2CH), 129.4 (C), 129.5 (CH), 129.8 (CH), 130.0 (2CH), 130.2 (2CH), 137.8 (C), 138.1 (C), 144.6 (C), 150.7 (C), 153.2 (C), 153.9 (CH), 156.0 (C). *Crystal data for* **2i.** C23H16N4, *M* = 348.40, *T* = 150(2) K, orthorhombic, *P 2¹ 2¹ 21*, *a* = 6.3672(5), *b* = 13.0997(10), $c = 21.5377(18)$ Å, $V = 1796.4(2)$ Å³, $Z = 4$, $d = 1.288$ g cm^{−3}, $\mu = 0.079$ mm^{−1}. A final refinement on F^2 with 2367 unique intensities and 245 parameters converged at $\omega R(F^2) = 0.1207$ ($R(F) = 0.0498$) for 1679 observed reflections with $I > 2\sigma(I)$. CCDC 1858475.

3.9.3. 8-(*N*-indolyl)-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2j**)

The general procedure 6 using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.20 g) and indole (0.12 g) gave **2j** (eluent: CH2Cl2; R^f = 0.36) in 51% yield as a red powder. Mp: 136 ◦C. IR: 1023, 1154, 1208, 1236, 1324, 1356, 1379, 1442, 1454, 1478, 1519, 1555, 1577, 1592, 3240, 3339, 3639 cm^{−1}. ¹H-NMR (CDCl3): 6.82 (d, 1H, *J* = 3.4 Hz), 7.22–7.44 (m, 8H), 7.53–7.56 (m, 2H), 7.67 (d, 1H, *J* = 8.3 Hz), 7.70–7.72 (m, 3H), 7.86 (dd, 1H, *J* = 4.9 and 1.2 Hz), 7.94 (d, 1H, *J* = 3.4 Hz), 9.17 (d, 1H, *J* = 4.9 Hz). ¹³C-NMR (CDCl3): 106.1 (CH), 111.4 (CH), 118.0 (CH), 121.5 (CH), 122.1 (CH), 123.2 (CH), 128.4 (2CH), 128.4 (2CH), 128.5 (C), 129.7 (CH), 129.9 (CH), 130.1 (2CH), 130.3 (C), 130.3 (2CH), 130.5 (C), 130.8 (CH), 136.2 (C), 137.8 (C), 138.0 (C), 144.8 (C), 150.8 (C), 153.6 (CH), 156.4 (C). Anal. Calc. for C₂₇H₁₈N₄ (398.47): C 81.39, H 4.55, N, 14.06. Found: C 81.26, H 4.67, N, 13.84.

3.9.4. 2,3-Diphenyl-8-(*N*-pyrazolyl)pyrido[2,3-*b*]pyrazine (**2k**)

The general procedure 6 using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.20 g) and pyrazole (68 mg) gave 2k (eluent: CH_2Cl_2 -EtOAc 80:20; $R_f = 0.47$) in 71% yield as a pale yellow powder. Mp: 200 ◦C. IR: 1027, 1032, 1092, 1164, 1229, 1324, 1356, 1388, 1532, 1549, 1592, 3034, 3060, 3159 cm−¹ . ¹H-NMR (CDCl3): 6.55 (d, 1H, *J* = 2.2 Hz), 7.30–7.41 (m, 6H), 7.55–7.58 (m, 2H), 7.64–7.66 (m, 2H), 7.82 (s, 1H), 8.34 (dd, 1H, *J* = 5.3 and 2.4 Hz), 9.12 (dd, 1H, *J* = 5.2 and 2.2 Hz), 9.46 (t, 1H, *J* = 2.5 Hz). ¹³C-NMR (CDCl₃): 109.2 (CH), 115.0 (CH), 127.9 (C), 128.3 (2CH), 128.6 (2CH), 129.6 (CH), 129.8 (CH), 129.9 (2CH), 130.3 (2CH), 134.4 (CH), 137.6 (C), 138.3 (C), 142.5 (CH), 143.2 (C), 150.5 (C), 153.4 (C), 154.3 (CH), 156.1 (C). Anal. Calc. for C₂₂H₁₅N₅ (349.40): C 75.63, H 4.33, N, 20.04. Found: C 75.71, H 4.42, N, 19.86.

3.9.5. 8-(*N*-imidazolyl)-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2l**)

The general procedure 6 using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.20 g) and imidazole (68 mg) gave **2l** (eluent: EtOAc-MeOH 95:5; $R_f = 0.48$) in 69% yield as a yellow powder. Mp: 209 ◦C. IR: 1019, 1053, 1075, 1105, 1115, 1169, 1236, 1319, 1334, 1379, 1429, 1446, 1459, 1482, 1549, 1594, 3064, 3124, 3639 cm−¹ . ¹H-NMR (CDCl3): 7.32–7.45 (m, 7H), 7.56 (d, 2H, *J* = 6.6 Hz), 7.65–7.71 (m, 3H), 7.80 (br s, 1H), 8.82 (br s, 1H), 9.20 (d, 1H, *J* = 4.8 Hz). ¹³C-NMR (CDCl3): 115.6 (CH), 119.5 (CH), 128.3 (2CH), 128.4 (2CH), 128.9 (C), 129.7 (CH), 129.9 (CH), 129.9 (2CH), 130.1 (2CH), 130.3 (CH), 137.5 (C), 137.6 (C), 138.8 (CH), 141.4 (C), 150.7 (C), 154.1 (C), 154.2 (CH), 156.6 (C). Anal. Calc. for C22H15N⁵ (349.40): C 75.63, H 4.33, N, 20.04. Found: C 75.74, H 4.37, N, 19.92.

3.9.6. 2,3-Diphenyl-8-[1-(1,2,4-triazolyl)]pyrido[2,3-*b*]pyrazine (**2m**)

The general procedure 6 using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.20 g) and 1,2,4-triazole (69 mg) gave $2m$ (eluent: CH₂Cl₂-EtOAc 80:20; R_f = 0.35) in 79% yield as an orange powder. Mp: 205 ◦C. IR: 708, 995, 1025, 1049, 1079, 1124, 1158, 1223, 1242, 1276, 1332, 1386, 1403, 1459, 1508, 1551, 1590, 3064, 3146 cm−¹ . ¹H-NMR (CDCl3): 7.35–7.49 (m, 6H), 7.59 (d, 2H, *J* = 7.0 Hz), 7.68 (d, 2H, *J* = 7.2 Hz), 8.22 (s, 1H), 8.37 (br s, 1H), 9.27 (br s, 1H), 10.16 (br s, 1H). ¹³C-NMR (CDCl3): 115.1 (CH), 127.3 (C), 128.2 (2CH), 128.5 (2CH), 129.8 (CH), 129.8 (2CH), 129.9 (CH), 130.1 (2CH), 137.3 (C), 137.7 (C), 140.4 (C), 147.2 (CH), 150.4 (C), 152.0 (CH), 154.1 (C), 154.5 (CH), 156.7 (C). Anal. Calc. for $C_{21}H_{14}N_6$ (350.39): C 71.99, H 4.03, N, 23.99. Found: C 72.19, H 4.15, N, 23.81.

3.9.7. 5-Iodo-2,3-diphenyl-8-(*N*-pyrazolyl)quinoxaline (**1k**⁰)

The general procedure 6 using 5-iodo-2,3-diphenylquinoxaline (1b', 0.27 g) and pyrazole (68 mg) gave $1k'$ (eluent: CH₂Cl₂-heptane 80:20; R_f = 0.45) in 50% yield as an pale yellow powder. Mp: 200–202 ◦C. IR: 536, 585, 602, 692, 696, 755, 843, 894, 946, 1040, 1092, 1182, 1193, 1221, 1336, 1397, 1465, 1519, 1543, 1592, 3060, 3159 cm−¹ . ¹H-NMR (CDCl3): 6.56 (dd, 1H, *J* = 2.6, 1.8 Hz), 7.35–7.45 (m, 6H), 7.57–7.60 (m, 2H), 7.70–7.73 (m, 2H), 7.82 (d, 1H, *J* = 1.8 Hz), 8.12 (d, 1H, *J* = 8.2 Hz), 8.43 (d, 1H, *J* = 8.3 Hz), 8.97–8.98 (m, 1H). ¹³C-NMR (CDCl3): 99.3 (C), 107.6 (CH), 123.7 (CH), 128.4 (2CH), 128.5 (2CH), 129.5 (CH), 129.6 (CH), 130.0 (2CH), 130.4 (2CH), 133.2 (C), 133.5 (CH), 137.1 (C), 137.8 (C), 138.2 (C), 139.7 (CH), 140.7 (C), 141.1 (CH), 153.0 (C), 153.6 (C). Anal. Calc. for C₂₃H₁₅IN₄ (474.31): C 58.24, H 3.19, N, 11.81. Found: C 58.33, H 3.26, N, 11.68.

3.10. Nucleophilic Substitution Using Amines

3.10.1. General Procedure 7

A sealed tube containing the iodide (0.50 mmol) and amine (amount given in the product description) in ethanol (2 mL) was heated (conditions given in the product description). The cooled residue was concentrated before chromatography over silica gel (eluent given in the product description).

3.10.2. 8-(Isopropylamino)-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2n**)

The general procedure 7 (150 ◦C, 18 h) using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.20 g) and isopropylamine (51 μ L, 0.60 mmol) gave **2n** (eluent: CH₂Cl₂-EtOAc 50:50; R_f = 0.20) in 69% yield as a beige powder. Mp: 179 ◦C. IR: 699, 703, 772, 804, 1156, 1178, 1236, 1313, 1336, 1538, 1564, 1592, 2965, 3038, 3064, 3390 cm−¹ . ¹H-NMR (CDCl3): 1.27 (d, 6H, *J* = 6.4 Hz, Me), 3.77 (dp, 1H, *J* = 7.9 and 6.4 Hz, C*H*Me2), 6.40 (br d, 1H, *J* = 8.0 Hz, NH), 6.45 (d, 1H, *J* = 5.5 Hz), 7.15–7.28 (m, 6H), 7.39–7.42 (m, 2H), 7.46–7.49 (m, 2H), 8.59 (dd, 1H, *J* = 5.4, 0.6 Hz). ¹³C-NMR (CDCl₃): 22.3 (2CH₃), 44.1 (CH), 100.7 (CH), 127.1 (C), 128.0 (2CH), 128.2 (2CH), 128.7 (CH), 129.0 (CH), 129.9 (2CH), 130.2 (2CH), 138.4 (C), 138.9 (C), 149.8 (C), 150.1 (C), 150.1 (C), 154.8 (CH), 155.8 (C). Anal. Calc. for $C_{22}H_{20}N_4$ (340.43): C 77.62, H 5.92, N, 16.46. Found: C 77.72, H 6.14, N, 16.19.

3.10.3. 8-(4-Methoxybenzylamino)-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2o**)

The general procedure 7 (150 ◦C, 24 h) using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.20 g) and 4-methoxybenzylamine (78 μ L, 0.60 mmol) gave **2o** (eluent: CH₂Cl₂-EtOAc 50:50; R_f = 0.48) in 71% yield as a yellow powder. Mp: 190 ◦C. IR: 697, 832, 1175, 1236, 1302, 1341, 1437, 1459, 1510, 1585, 2828, 2910, 3064, 3232 cm−¹ . ¹H-NMR (CDCl3): 3.81 (s, 3H, OMe), 4.55 (d, 2H, *J* = 5.9 Hz), 6.58 (d, 1H, *J* = 5.3 Hz), 6.90 (d, 2H, *J* = 8.7 Hz), 7.02 (t, 1H, *J* = 5.7 Hz), 7.27–7.35 (m, 8H), 7.49–7.51 (m, 2H), 7.58–7.61 (m, 2H), 8.69 (d, *J* = 5.3 Hz, 1H). ¹³C-NMR (CDCl3): 46.5 (CH2), 55.3 (CH3), 101.1 (CH), 114.3 (2CH), 127.2 (C), 128.0 (2CH), 128.2 (2CH), 128.6 (2CH), 128.8 (CH), 129.1 (CH), 129.1 (C), 129.9 (2CH), 130.3 (2CH), 138.5 (C), 138.8 (C), 150.0 (C), 150.2 (C), 150.9 (C), 154.9 (CH), 155.9 (C), 159.2 (C). Anal. Calc. for C₂₇H₂₂N₄O (418.50): C 77.49, H 5.30, N, 13.39. Found: C 77.58, H 5.44, N, 13.20.

3.10.4. 8-(Benzylamino)-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2p**)

The general procedure 7 (150 ◦C, 24 h) using 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.20 g) and benzylamine (66 μ L, 0.60 mmol) gave 2p (eluent: CH₂Cl₂-EtOAc 50:50; R_f = 0.50) in 79% yield as a yellow powder. Mp: 238 ◦C. IR: 697, 768, 873, 1150, 1238, 1300, 1324, 1339, 1439, 1538, 1590, 2910, 3064, 3201 cm−¹ . ¹H-NMR (CDCl3): 4.63 (d, 2H, *J* = 6.0 Hz), 6.56 (d, 1H, *J* = 5.4 Hz), 7.11 (t, 1H, *J* = 6.0 Hz), 7.27–7.39 (m, 11H), 7.49–7.52 (m, 2H), 7.58–7.61 (m, 2H), 8.68 (d, 1H, *J* = 5.4 Hz). ¹³C-NMR (CDCl3): 47.0 (CH2), 101.2 (CH), 127.2 (2CH), 127.2 (C), 127.8 (CH), 128.1 (2CH), 128.3 (2CH), 128.9 (CH), 128.9 (2CH), 129.1 (CH), 129.9 (2CH), 130.3 (2CH), 137.2 (C), 138.5 (C), 138.8 (C), 150.0 (C), 150.3 (C), 151.0 (C), 154.9 (CH), 156.0 (C). *Crystal data for* 2p. $C_{26}H_{20}N_4$, $M = 388.46$, $T = 150(2)$ K, monoclinic, *P* 2₁/*n*, *a* = 6.0721(6), *b* = 12.8640(10), *c* = 25.460(2) Å, β = 91.436(4) °, *V* = 1988.1(3) Å³, $Z = 4$, $d = 1.298$ g cm⁻³, $\mu = 0.078$ mm⁻¹. A final refinement on F^2 with 4438 unique intensities and 274 parameters converged at $ωR(F^2) = 0.1432$ ($R(F) = 0.0626$) for 3710 observed reflections with $I > 2σ(I)$. CCDC 1858476.

3.11. Nucleophilic Substitution using Hydrazine Hydrate: 8-Hydrazino-2,3-diphenylpyrido[2,3-b]pyrazine (2q)

A solution of 8-iodo-2,3-diphenyl pyrido [2,3-*b*]pyrazine (**2b-I**, 0.20 g, 0.50 mmol) and hydrazine hydrate (0.25 mL, 5.0 mmol) in isopropanol (2 mL) was heated under reflux for 4 h. The cooled residue was concentrated and taken up with EtOAc (20 mL). The organic layer was washed with water (10 mL), dried over MgSO4, filtered and concentrated under reduced pressure to give the title compound **2q** in 92% yield as a red powder. Mp > 250 ◦C. ¹H-NMR (CDCl3): 4.22 (br s, 2H, NH), 6.91 (d, 1H, *J* = 5.6 Hz), 7.22–7.36 (m, 7H), 7.40–7.43 (m, 2H), 7.48–7.51 (m, 2H), 8.60 (d, 1H, *J* = 5.6 Hz). ¹³C-NMR (CDCl3): 101.2 (CH), 126.2 (C), 128.2 (2CH), 128.3 (2CH), 129.0 (CH), 129.2 (CH), 129.9 (2CH), 130.3 (2CH), 138.4 (C), 138.7 (C), 149.7 (C), 150.3 (C), 153.0 (C), 155.0 (CH), 156.1 (C). Anal. Calc. for C₁₉H₁₅N₅ (313.36): C 72.83, H 4.83, N, 22.35. Found: C 72.96, H 4.89, N, 22.31.

3.12. Condensation Reactions from the Hydrazine 2q

3.12.1. General Procedure 8

A sealed tube containing 8-hydrazino-2,3-diphenylpyrido[2,3-*b*] pyrazine (**2q**, 0.16 g, 0.50 mmol) and the aldehyde (0.55 mmol) in ethanol (2 mL) was heated at 110 ◦C overnight. The cooled residue was concentrated under vacuum, washed with methanol and isolated by filtration.

3.12.2. 2-Hydroxybenzaldehyde 2-[8-(2,3-diphenylpyrido[2,3-*b*]pyrazinyl)]hydrazone (**2r**)

General Procedure 8 using 2-hydroxybenzaldehyde (67 mg) gave **2r** (R^f (CH2Cl2-EtOAc 80:20) = 0.44) in 60% yield as a yellow powder. Mp > 260 ◦C. IR: 952, 1019, 1096, 1163, 1233, 1270, 1309, 1328, 1422, 1540, 1562, 1594, 1618, 3064, 3317 cm^{−1}. ¹H-NMR (CDCl₃): 6.96 (td, 1H, *J* = 7.5 and 1.1 Hz), 7.07 (d, 1H, *J* = 8.2 Hz), 7.23–7.44 (m, 9H), 7.51–7.54 (m, 2H), 7.58–7.62 (m, 2H), 8.26 (s, 1H), 8.91 (d, 1H, *J* = 5.3 Hz), 9.71 (br s, 1H), 10.60 (br s, 1H). The ¹³C spectra could not be recorded due to low solubility in CDCl₃ and DMSO. Anal. Calc. for C₂₆H₁₉N₅O (417.47): C 74.80, H 4.59, N, 16.78. Found: C 74.72, H 4.39, N, 16.67.

3.12.3. Piperonal 2-[8-(2,3-diphenylpyrido[2,3-*b*]pyrazinyl)]hydrazone (**2s**)

General Procedure 8 using piperonal (83 mg) gave **2s** (R_f (CH₂Cl₂-EtOAc 80:20) = 0.37) in 70% yield as a yellow powder. Mp: 254 ◦C. IR: 933, 1038, 1150, 1255, 1339, 1450, 1489, 1501, 1545, 1568, 1590, 2901, 3060, 3322, 3648 cm^{−1}. ¹H-NMR (CDCl₃): 6.03 (s, 2H), 6.84 (d, 1H, *J* = 8.0 Hz), 7.07 (dd, 1H, *J* = 8.1 and 1.6 Hz), 7.28–7.42 (m, 7H), 7.50 (t, 3H, *J* = 6.6 Hz), 7.59 (d, 2H, *J* = 6.8 Hz), 7.97 (s, 1H), 8.85 (d, 1H, *J* = 5.3 Hz), 9.66 (s, 1H). ¹³C-NMR ((CD₃)₂SO): 101.5 (CH₂), 103.5 (CH), 104.9 (CH), 108.5 (CH), 123.0 (CH), 125.6 (C), 128.1 (2CH), 128.2 (2CH), 128.8 (CH), 129.1 (CH), 129.2 (C), 129.7 (2CH), 130.1 (2CH),

138.3 (C), 138.6 (C), 145.1 (CH), 147.7 (C), 148.1 (C), 148.8 (C), 149.6 (C), 150.0 (C), 154.5 (CH), 155.5 (C). Anal. Calc. for C₂₇H₁₉N₅O₂ (445.48): C 72.80, H 4.30, N, 15.72. Found: C 72.95, H 4.44, N, 15.83.

3.12.4. 2-Hydroxy-4-methoxybenzaldehyde 2-[8-(2,3-diphenylpyrido[2,3-*b*]pyrazinyl)]hydrazone (**2t**)

General Procedure 8 using 2-hydroxy-4-methoxybenzaldehyde (84 mg) gave **2t** (R_f (CH₂Cl₂-EtOAc 80:20) = 0.58) in 80% yield as a yellow powder. Mp > 260 ◦C. IR: 1032, 1135, 1163, 1238, 1291, 1339, 1431, 1439, 1461, 1510, 1543, 1566, 1631, 2845, 2931, 3004, 3056, 3176, 3317 cm $^{-1}$. 1 H-NMR (CDCl3): 3.85 (s, 3H), 6.52 (dd, 1H, *J* = 8.5 and 2.5 Hz), 6.58 (d, 1H, *J* = 2.5 Hz), 7.15 (d, 1H, *J* = 8.6 Hz), 7.19 (d, 1H, *J* = 5.2 Hz), 7.28–7.43 (m, 6H), 7.50–7.54 (m, 2H), 7.58–7.61 (m, 2H), 8.19 (s, 1H), 8.88 (br s, 1H), 9.59 (br s, 1H), 10.81 (s, 1H). The ¹³C spectra could not be recorded due to low solubility in CDCl₃ and DMSO. Anal. Calc. for C₂₇H₂₁N₅O₂ (447.50): C 72.47, H 4.73, N, 15.65. Found: C 72.53, H 4.89, N, 15.60.

3.12.5. 4-(Trifluoromethyl)benzaldehyde 2-[8-(2,3-diphenylpyrido[2,3-*b*]pyrazinyl)]hydrazone (**2u**)

General Procedure 8 using 4-(trifluoromethyl)benzaldehyde (87 mg) gave **2u** (R_f (CH₂Cl₂- EtOAc 80:20) = 0.51) in 73% yield as a yellow powder. Mp: 258–260 ◦C. IR: 1017, 1066, 1109, 1124, 1145, 1236, 1300, 1321, 1512, 1545, 1562, 1588, 3060, 3184, 3317, 3652 cm^{−1}. ¹H-NMR (CDCl₃): 7.28–7.43 (m, 6H), 7.50–7.53 (m, 2H), 7.56–7.61 (m, 3H), 7.68 (d, 2H, *J* = 8.2 Hz), 7.87 (d, 2H, *J* = 7.8 Hz), 8.11 (s, 1H), 8.91 (d, 1H, *J* = 5.2 Hz), 9.90 (br s, 1H). ¹³C-NMR ((CD₃)₂SO, 333 K): 103.8 (CH), 124.0 (q, CF₃, *J* = 272 Hz), 125.4 (C), 125.5 (q, 2CH, *J* = 3.7 Hz), 127.1 (2CH), 127.8 (2CH), 127.9 (2CH), 128.7 (CH), 128.8 (CH), 129.2 (q, *C*-CF3, *J* = 31.7 Hz), 129.5 (2CH), 129.8 (2CH), 138.1 (C), 138.4 (C), 138.5 (C), 143.2 (CH), 147.4 (C), 149.4 (C), 150.3 (C), 154.3 (CH), 155.5 (C). Anal. Calc. for $C_{27}H_{18}F_3N_5$ (469.47): C 69.08, H 3.86, N, 14.92. Found: C 69.25, H 3.97, N, 14.78.

3.13. Nucleophilic Substitution Using a Phenolate: Methyl 2-[8-(2,3-diphenylpyrido[2,3-b]pyrazinyl)]oxy-5-methoxybenzoate (2v)

A mixture of 8-iodo-2,3-diphenylpyrido[2,3-*b*]pyrazine (**2b-I**, 0.20 g, 0.50 mmol), methyl 2-hydroxy-5-methoxy-benzoate (0.10 g, 0.55 mmol), K_2CO_3 (77 mg, 0.55 mmol) and DMSO (1 mL) was heated at 110 °C for 2 h. The cooled residue was treated by an aqueous solution of $Na₂CO₃$ (10 mL) before extraction with Et₂O (3 \times 10 mL). The organic layer was dried over MgSO₄, filtered and concentrated under reduced pressure, and the residue was chromatographed over silica gel (eluent: CH₂Cl₂-MeOH 95:5; Rf (CH₂Cl₂-EtOAc 95:5) = 0.50) to give the title compound 2v in 64% yield as a beige powder. Mp: 206 ◦C. IR: 542, 698, 773, 856, 1021, 1072, 1109, 1205, 1235, 1263, 1333, 1350, 1434, 1468, 1496, 1554, 1594, 1719, 2845, 2956, 3041 cm $^{-1}$. 1 H-NMR (CDCl₃): 3.65 (s, 3H), 3.90 (s, 3H), 6.64 (d, *J* = 5.2 Hz, 1H), 7.19 (dd, 1H, *J* = 8.9, 3.0 Hz), 7.24 (d, 1H, *J* = 9.5 Hz), 7.30–7.40 (m, 6H), 7.55–7.58 (m, 3H), 7.61–7.64 (m, 2H), 8.86 (d, 1H, *J* = 5.2 Hz). ¹³C-NMR (CDCl₃): 52.4 (CH₃), 56.0 (CH₃), 107.6 (CH), 116.3 (CH), 120.6 (CH), 124.6 (C), 125.0 (CH), 128.2 (2CH), 128.4 (2CH), 129.0 (C), 129.1 (CH), 129.4 (CH), 130.2 (2CH), 130.3 (2CH), 138.2 (C), 138.7 (C), 147.0 (C), 151.1 (C), 153.6 (C), 154.4 (CH), 156.6 (C), 157.4 (C), 163.0 (C), 164.7 (C). Anal. Calc. for C₂₈H₂₁N₃O₄ (463.49): C 72.56, H 4.57, N, 9.07. Found: C 72.49, H 4.65, N, 9.01.

4. Conclusions

Original pyrazino-fused polycyclic scaffolds were synthesized by combining deprotometalation-iodolysis with palladium- or copper-catalyzed couplings or direct substitution reactions. This study highlights the interest in preparing iodo derivatives of sensitive aromatic heterocycles by using lithium-zinc basic combinations to access scaffolds of potential biological interest. Interestingly, bromine and trichloroisocyanuric acid were successfully employed as electrophiles to intercept the intermediate heteroarylzinc halides.

Supplementary Materials: Supplementary materials are available online.

Author Contributions: F.L., T.L., M.B., C.C., I.C., C.G. and J.L. synthesized and analyzed all compounds presented in this article; E.C. contributed to the identification of some synthesized compounds by NMR; E.L. and L.P. contributed with the experiments performed under microwave irradiation; L.P., M.S., B.B. and S.B. performed the bioassays; T.R. collected the X-ray diffraction data and solved the structures. F.M. wrote the paper with the help of E.C., L.P., V.T., S.R., S.B. and T.R.; F.L. started the project, designed the molecules and revised the paper.

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Sample Availability: Samples of the synthesized compounds are available from the corresponding authors.

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