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The ammonia absorption spectrum between 4700 and 5650 cm^{-1}

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Abstract

Room temperature absorption spectra of ammonia were recorded by high-resolution Fourier transform spectroscopy in the 4700-5600 cm⁻¹ range at four pressures (13, 46, 140, and 304 Pa). Additionally, we use an existing Kitt Peak spectrum to extend the analysis up to 5650 cm^{-1} . In total, 9110 absorption lines with intensity ranging from 1×10^{-25} to 1.6×10^{-20} cm/molecule were retrieved, for the two principal isotopologues. Another Kitt Peak spectrum was used to identify 505 ¹⁵NH₃ transitions within the experimental line list. The rovibrational assignments of the ¹⁴NH₃ lines relied on the position and intensity agreement with the C2018 theoretical line list (Coles *et al.* JQSRT 2018;219 199–212) and were validated by the systematic use of Lower State Combination Difference (LSCD) relations. In the process, a number of intensity transfer between nearby transitions were identified. Finally, 6562 transitions were assigned to 61 vibrational bands of the main isotopologue, ¹⁴NH₃, representing 98.85% of the C2018 integrated absorption at room temperature in the region. The upper state empirical energy of a total of 2215 rovibrational levels of ¹⁴NH₃ were derived. Most of them are newly reported. The comparison with the current HITRAN2020 list is discussed. A recommended line list for ammonia in natural isotopic abundance is provided for the studied region.

1 1. Introduction

² Ammonia is an essential component in various astrophysical and terrestrial objects, including the Earth's atmo-

³ sphere. Therefore, understanding its infrared absorption spectra is a prerequisite for accurate monitoring of these

environments [1, 2]. This task is made difficult by numerous interactions between its six vibrational modes and the

- ⁵ internal inversion motion.
- ⁶ The current work is part of the analysis of the ammonia high resolution spectrum recorded by Fourier Transform

⁷ Spectroscopy (FTS) in the 3900-5600 cm⁻¹ range. The low energy part of the range, from 3900 to 4700 cm⁻¹, was

⁸ covered in our previous contribution [3], while the remainder of the data set, up to 5600 cm⁻¹, is analyzed in the

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⁹ current work. The present FTS data are complemented with an existing Kitt Peak FTS spectrum in the weak absorbing ¹⁰ 5600-5650 cm⁻¹ interval, in order to fill the gap up to our previous analysis of the 5650-6350 cm⁻¹ range [4].

The HITRAN line list [5] in the considered range is based on the study of a FTS spectrum recorded at Kitt Peak 11 with a 0.011 cm⁻¹ spectral resolution [6]. The spectrum is dominated by the $v_1 + v_4$ and $v_3 + v_4$ bands centered at 4956 12 cm⁻¹ and 5050 cm⁻¹, respectively. The upgrade of the HITRAN database from 2016 [7] to 2020 [5] has increased 13 the number of ¹⁴NH₃ transitions from 1875 to 2493 while the ¹⁵NH₃ dataset in the region has remained unchanged. 14 The overview of HITRAN2020 list is presented in the upper panel of Fig. 1. About 620 new transitions of $^{14}NH_3$ 15 included in HITRAN2020 were not observed but transferred from the hybrid CoYuTe list created by adjusting the 16 line positions predicted from the C2018 potential energy surface [8] according to empirical energy levels determined 17 using the MARVEL algorithm [9]. For the HITRAN2020 edition [5], it was decided to add to the HITRAN2016 18 dataset [7] only the CoYuTe transitions for which both the lower and upper levels have empirical energies and their 19 calculated line intensities are larger than 1×10^{-25} cm/molecule. In our region, the added transitions include several hot 20 band transitions for which the upper levels were determined from ground state transitions, thus at higher energy [10]. 21 Their intensities are generally less than 3×10^{-24} cm/molecule which was the intensity cut-off of the HITRAN2016 22 list. Among the ¹⁴NH₃ transitions, 1255 are given with full assignment of the lower and upper states, whereas no 23 or incomplete assignment is given for 1201 transitions. Experimental intensities have been kept for the transitions 24 transferred from HITRAN2016, whereas C2018 calculated intensity values [11] are given for the added transitions. 25 Regarding the previous experimental studies in our region, in addition to Ref. [6], it is worth mentioning the 26 recent work of Beale et al. [12]. These authors reported an analysis of ammonia FTS spectra recorded at different 27

²⁷ recent work of Beale *et al.* [12]. These authors reported an analysis of ammonia FTS spectra recorded at different ²⁸ temperatures from 295 to 973 K with a 0.01 cm⁻¹ spectral resolution. Due to its lower accuracy, this study was used ²⁹ neither for the HITRAN2020 update nor for the derivation of new MARVEL energy levels [11]. Nevertheless, it is a ³⁰ very useful source for the application of Lower State Combination Differences (LSCD) relations used for validation ³¹ of our quantum assignments [4].

As the experimental setup, conditions of the recordings, fitting process and construction of the line list were detailed in Ref. [3], they are briefly recalled in the two next sections. Section 4 describes some aspects of the assignment process before discussing the results in particular the occurrence of intensity transfers between resonant lines.

2. Experimental details

The experimental conditions of the measurements are summarized in Table 1. Four room temperature FTS absorption spectra of ammonia were recorded at ULB with a Bruker IFS 120 (upgraded to 125) spectrometer in the 38 3900-5600 cm⁻¹ range (Fig. 2).

To supplement these data, we used a couple of FTS spectra from the National Solar Observatory historical archive recorded in early nineties at Kitt Peak observatory. To fill the $5600-5650 \text{ cm}^{-1}$ interval of weak ammonia absorption we used a long-path recording (KP1, Ref. 900308R0.019), and to identify the ¹⁵NH₃ transitions, an ammonia spectrum enriched in ¹⁵NH₃ (KP2, 930826R0.016) was treated.



Figure 1: *Upper panel*: Overview of the HITRAN2020 transitions of ${}^{14}NH_3$ (green and orange) and ${}^{15}NH_3$ (cyan) in the studied region. ${}^{14}NH_3$ transitions were added using MARVEL algorithm (orange). *Lower panel*: Ammonia transitions measured in the present work: ${}^{14}NH_3$ and ${}^{15}NH_3$ (green and cyan circles, respectively), and unassigned (yellow circles).

		р	$c_{15}N$	L	R
Spectrum	(Torr)	(Pa)	(%)	(m)	(cm^{-1})
TW1	0.109 (6)	13.2(1.5)	0.78		
TW2	0.348 (8)	46.0(1.1)	0.68	13.86 (4)	0.003
TW3	1.053 (33)	140.4 (4.4)	0.62	13.80 (4)	0.003
TW4	2.28 (8)	304.0(11.0)	0.54		
KP1 (Ref. 900308R019)	0.52	69.3		193	0.01
KP2 (¹⁵ NH ₃ - Ref. 930826R016)	2.0	167.	99.64	16	0.01

Table 1: Experimental parameters of the four FTS spectra used in this work.

Notes: For the four spectra studied in this work (TW1-TW4), p is the average pressure (the numbers between parentheses

are the uncertainties of measurement, provided in the units of the last digit quoted,

L is the interaction pathlength and R is the spectral resolution used for the recording. c_{15N} is

the estimated value of the $^{15}\mathrm{NH}_3$ relative abundance (the natural value is 0.37%).



Figure 2: Overview of the Fourier transform absorption spectra of ammonia recorded in the present work with the resolution of 0.003 cm^{-1} . The arrows mark the spectral intervals of the spectra used in the analysis. In the 5600-5650 cm⁻¹ interval, the Kitt Peak spectrum (KP1) with Ref. 900308R0.019 was used.



Figure 3: Ratios of the intensities derived from the 140.4 Pa spectrum to those of the 13.2 Pa spectrum (orange dots) for isolated transitions in the 1.2×10^{-23} - 1.2×10^{-21} cm/molecule range. ¹⁵NH₃ transitions were removed from the comparison. Intensity ratios between the KP1 Kitt Peak spectrum and the 304 Pa spectrum are plotted for a few weak lines below 5600 cm⁻¹ (cyan dots). They are used for the intensity calibration of the 5600-5650 cm⁻¹ interval.

3. Line list construction

Line positions and line intensities were retrieved using a multi-line fit of the spectra. The fit was performed assuming a simple Voigt function as line profile, thus, including in an effective way the Instrument Line Shape (ILS). The Lorentzian and Gaussian components of the Voigt profile were assumed to be identical for all the lines and adjusted in the fit. The fitted value of the Gaussian width was found slightly above the calculated value of the ¹⁴NH₃ Doppler broadening (FWHM 0.015 cm⁻¹), confirming a small contribution of ILS. Constraining the line widths helped to disentangle blended lines. In Ref. [3], it was demonstrated that our data treatment neglecting the ILS leads to underestimated line intensities in case of strong absorbance, and the correction of this bias was presented. The same process was applied in the

⁵² current range. As a test of the correction and consistency of the intensity values in this new range, Fig. 3 plots the

ratio of intensities retrieved from different sources. The intensities from the 140.4 Pa spectrum are compared to those 53 retrieved from of the 13.2 Pa spectrum. For the KP1 data in the $5550-5600 \text{ cm}^{-1}$ interval, the intensity ratio between 54 the 304 Pa and the KP1 spectra is presented for a few lines. Isolated lines were preferred for the comparison (minimum 55 position separation of 0.03 cm⁻¹). The oscillator strength spans over two orders of magnitude between 1.2×10^{-23} and 56 1.2×10^{-21} cm/molecule (in order to correspond to absorbances not too small at 13.2 Pa spectrum and not too large at 57 140.4 Pa). On average, a small overestimation of 1.5% is noted between 1.2×10^{-22} and 1.2×10^{-21} cm/molecule for 58 the 140.4 Pa spectrum and the root mean square deviation less than 5% in the whole range. This confirms the efficiency 59 of the applied correction and allows for using either spectrum. To construct the global line list, we selected at least 60 two spectra for each spectral interval, according to the absorption amplitude (see Fig. 2): the 13.2 Pa spectrum was 61 used from 4850 to 5250 cm^{-1} , the 46 Pa spectrum in 4700-4850 cm⁻¹ and $5200-5300 \text{ cm}^{-1}$ regions, the 140.4 Pa in the 62 $4700-5500 \text{ cm}^{-1}$ range, the 304 Pa spectrum from 5260 to 5600 cm⁻¹ and, the KP1 spectrum in the 5550-5650 cm⁻¹ 63 range. The comparison between the 304 Pa and the KP1 spectra in the range 5550-5600 cm⁻¹ was used to calibrate in 64 intensity the latter (Fig. 3). 65

⁶⁶ Several water vapor lines (present as an impurity) were also observed in the spectra, accidentally hiding some ⁶⁷ ammonia lines above $5200 \,\mathrm{cm}^{-1}$. They were removed manually or subtracted from the spectrum by comparison with ⁶⁸ the H₂O line list from the HITRAN2020 database [5]. Finally, the resulting ammonia list includes 9110 lines with ⁶⁹ intensity ranging between 1×10^{-25} and 1.3×10^{-20} cm/molecule including 505 ¹⁵NH₃ transitions (see lower panel of ⁷⁰ Fig. 1). In case of superposition of a ¹⁴NH₃ line and a ¹⁵NH₃ line (or a water line), the intensity of the interfering line ⁷¹ was carefully estimated and subtracted in order to minimise its impact on the retrieval of the ¹⁴NH₃ line intensity.

As mentioned in Ref. [3], the FTS spectra under analysis do not correspond to the ammonia natural isotopic 72 abundance but are enriched in the ¹⁵NH₃ isotopologue due to previous recordings with ¹⁵N enriched gas samples. The 73 ¹⁵NH₃ transitions in the different spectra were identified by comparison with a Kitt Peak spectrum highly enriched 74 in ¹⁵NH₃ (Ref. 930826R0.016, p = 2.00 Torr, T = 23.8 °C, L = 2.4 m with CO for calibration). The ¹⁵NH₃ relative 75 abundance corresponding of our FTS spectra has been estimated in two ways: using the absolute intensities obtained 76 in Ref. [13] already used in Ref. [3] or by comparison with the calculated intensity values of Ref. [14]. The two results 77 agree and lead to the relative abundance values included in Table 1. The ¹⁵NH₃ abundance increases from 0.54% 78 (340 Pa) to 0.78% (13.2 Pa) to be compared to a natural value of 0.366% [5]. As the four spectra were recorded 79 successively by decreasing steps of the pressure, the isotopic ratio increased, due to the progressive desorption of 80 15 NH₃ adsorbed on the walls of the cell and exchanges with ammonia in natural isotopic abundance initially injected 81 in the cell. Note that the estimated value of the ¹⁵NH₃ abundance in the various analyzed spectra was used to scale 82 the ¹⁵NH₃ line intensities accordingly, and the reported intensity values correspond to the natural abundance. 83

84 4. Rovibrational assignments

⁸⁵ Following our recent analysis [1, 3, 4], the C2018 variational list of ¹⁴NH₃ [11] was used as a primary source for ⁸⁶ the rovibrational assignment of the experimental line list. The C2018 rovibrational labeling uses 13 quantum numbers:

 $v_1, v_2, v_3, v_4, L_3, L_4, \tau_{inv}, \Gamma_{vib}, J, K, \Gamma_{rot}, N_B, \Gamma_{tot}$ where v_i are the vibrational normal-mode quantum numbers, $L_i = |l_i|$ 87 are the corresponding vibrational angular momentum quantum numbers, $\tau_{inv} = a/s$ is the inversion symmetry, and Γ_{vib} is the symmetry of the vibrational motion. J is the total angular-momentum quantum number, K = |k| is the 89 projection of the total angular momentum on the molecule-fixed axis z, and Γ_{rot} is the rotational symmetry. Finally, 90 Γ_{tot} stands for the full symmetry of the eigenstate, and N_B is the block counting number used during the generation of 91 the C2018 list. Let us recall that only J and Γ_{tot} are exact quantum numbers, the others are approximate ones which 92 describe helpfully the level considering its true wavefunction and how this wavefunction is projected on the basis set. 93 In each (J, Γ_{tot}) subspace, the diagonalization process gives the eigenvalues and eigenstates which correspond to the 94 energy levels, ordered by increasing energy and with a rank number N_B . In this context, let us mention the work of 95 Huang et al. who have processed the same type of calculations, using a refined potential surface [15, 16]. The quality of the predicted energy positions is claimed to be improved compared to C2018 but the transition intensities are not 97 provided.

98

The assignment process has been described in detail in our previous contributions [1, 3]. Coincidences between 99 the experimental and C2018 lists were searched automatically with the constraints $\delta v = |v_{exp.} - v_{calc.}| \le 1 \text{ cm}^{-1}$ and 100 $0.5 \le S_{exp.}/S_{calc.} \le 2$. For a given measured line, the C2018 candidates were tested using LSCD relations combined 101 with intensity considerations. It means that the LSCD candidate partners were also subjected to the intensity screening 102 with a similar criterion as for the initial C2018 match. For the search of LSCD relations, different studies extending 103 from 3900 to 6500 cm^{-1} were considered [3, 4, 5]. The method of branches was used to complement the assignment 104 procedure [3]. The (obs.-calc.) wavenumber differences within a given vibrational branch are expected to vary slowly 105 with the rotational quantum numbers. This expected tendency was used to confirm the LSCD assignments and for 106 transitions missing LSCD partners. In some cases, the application of the LSCD relations leads to an assignment of two 107 or more C2018 predicted transitions to a single experimental line. In such a case, the original line was artificially split 108 into the corresponding number of independent transitions with positions shifted by a negligible value of 1×10^{-5} cm⁻¹. 109 The intensities of these new transitions were obtained by affecting to each transition the fraction of the measured total 110 intensity following the predictions of the C2018 line list. 111

Overall, 6562 out of 8610 transitions were assigned to 61 ¹⁴NH₃ vibrational bands listed in Table 2. In this table, 112 the vibrational bands are defined by the vibrational quantum labels: $v_{1,2,3,4}$, $L_{1,2}$ and Γ_{vib} (the τ_{inv} label has been 113 added). The line assignments are included in the list provided as a Supplementary Material. Considering the C2018 114 intensities, the assigned transitions represent 98.9% of the total integrated intensity in the region. 115



Figure 4: Comparison between the experimental lines left unassigned and the C2018 calculated lines of ¹⁴NH₃ with no experimental counterpart.

Table 2: Statistics and integrated absorption of the various vibrational bands contributing to the $^{14}NH_3$ spectrum between 4705 and 5650.3 cm⁻¹. The bands are ordered according to their C2018 integrated intensities.

	C2018			This w				
Vibrational	E_0^{b}	Nb ^c	Band Int. ^d	Nb ^e	J_{min}/J_{max}^{f}	Band Int. ⁸	Int. ratio ^h	$<\delta>^{i}$
band ^a	(cm^{-1})		(cm/molecule)			(cm/molecule)	(%)	(cm^{-1})
$v_3^1 + v_4^1 (E') s$	5052.6	1508	3.15E-19	911	0/17	3.16E-19	99.8	0.03
$v_3^1 + v_4^1 (E'') a$	5053.2	1480	3.07E-19	909	0/17	3.08E-19	99.8	0.02
$v_1 + v_4^1 (E') s$	4955.8	1063	2.15E-20	564	0/14	2.18E-20	97.9	0.01
$v_1 + v_4^1 (E'') a$	4956.9	1021	2.08E-20	525	0/15	2.11E-20	97.9	0.06
$v_3^1 + v_4^1 (A_1') s$	5067.8	552	1.56E-20	337	1/13	1.60E-20	98.8	0.04
$v_3^1 + v_4^1 (A_2'') a$	5067.7	546	1.32E-20	324	0/13	1.33E-20	98.3	0.06
$v_3^1 + v_4^1 (A_2') s$	5052.0	577	6.62E-21	297	2/12	6.45E-21	96.2	0.01
$v_3^1 + v_4^1 (A_1'') a$	5052.7	585	6.21E-21	299	2/13	6.12E-21	96.3	0.00
$4v_2 + v_4^1(E') s$	5104.9	638	5.86E-21	161	4/11	5.42E-21	92.9	0.20
$2v_2 + 2v_4^0 (\mathbf{A}_2'')$ a	5093.6	402	5.03E-21	109	3/13	6.20E-21	96.0	0.17
$v_2 + v_3^1 + v_4^1 (\mathbf{E}') \mathbf{s} \leftarrow v_2$	5080.4	485	4.28E-21	257	0/14	4.20E-21	96.2	-0.05
$v_2 + v_3^1 + v_4^1 (\mathbf{E}^{\prime\prime}) \mathbf{a} \leftarrow v_2$	5104.1	467	3.25E-21	241	0/14	3.20E-21	95.6	-0.10
$2v_2 + v_3^1(E') s$	5144.9	706	2.89E-21	279	1/14	2.96E-21	86.3	0.21
$v_1 + 2v_2 (A'_1) s$	5000.2	448	2.47E-21	186	1/12	3.32E-21	92.0	0.19
$2v_2 + 2v_4^2 (E'') a$	5113.3	314	1.25E-21	58	4/15	1.06E-21	86.6	0.35
$2v_2 + v_3^1(E'') a$	5353.0	651	1.08E-21	368	0/12	9.92E-22	89.8	0.15
$3v_4^1(E')s$	4799.2	496	9.91E-22	42	4/16	4.02E-22	38.8	0.15
$3v_4^1(E'')$ a	4801.4	452	5.48E-22	13	3/16	7.24E-23	14.4	-0.09
$3v_4^3(A_2')s$	4840.9	262	4.95E-22	52	2/10	3.46E-22	60.9	0.19
$3v_4^3(A_1')s$	4841.6	245	4.94E-22	46	1/13	4.16E-22	57.9	0.12
$3v_4^3(A_2'')a$	4843.4	204	4.23E-22	30	0/12	4.08E-22	57.6	0.16
$3v_4^3(A_1'')a$	4843.0	251	3.95E-22	27	6/11	1.51E-22	54.3	0.10
$3v_2 + v_3^1(\mathbf{E}') \mathbf{s} \leftarrow v_2$	4923.6	266	3.86E-22	121	1/11	3.11E-22	74.2	0.15
$v_1 + 2v_2 (A_2'') a$	5233.8	207	3.12E-22	58	0/12	1.94E-22	73.7	0.28
$2\nu_2 + 2\nu_4^0 (A_1') s$	4756.9	163	1.80E-22	20	1/10	9.07E-23	48.0	0.16
$2\nu_2 + 2\nu_4^2$ (E') s	4773.8	294	1.71E-22	4	5/9	6.60E-24	3.0	0.16
$2v_2 + v_3^1 + v_4^1(E') s \leftarrow 2v_2$	5119.8	179	1.67E-22	23	2/8	6.51E-23	38.7	-0.20
$v_2 + v_3^1 + v_4^1 (\mathbf{A}_2'') \mathbf{a} \leftarrow v_2$	5117.2	159	1.56E-22	31	2/10	1.04E-22	59.6	-0.22
$v_2 + v_3^1 + v_4^1 (A_1') s \leftarrow v_2$	5091.8	159	1.50E-22	38	3/9	1.02E-22	63.8	-0.29
$v_2 + v_3^1 + v_4^1 (A_1'') a \leftarrow v_2$	5101.2	156	1.39E-22	33	2/9	7.75E-23	43.7	-0.20
$v_2 + v_3^1 + v_4^1 (A'_2) s \leftarrow v_2$	5075.6	113	1.20E-22	22	4/8	9.06E-23	63.6	-0.22
$v_1 + v_2 + v_4^1 (E'') a \leftarrow v_2$	4998.4	206	1.14E-22	25	1/9	4.55E-23	28.7	-0.33
$v_1 + v_3^1 (E') s \leftarrow v_4$	4982.5	267	1.09E-22	5	4/7	6.42E-24	5.8	0.02
$v_1 + v_2 + v_4^1 (\mathbf{E}') \mathbf{s} \leftarrow v_2$	4965.4	196	9.84E-23	15	2/10	2.37E-23	23.3	-0.36
$v_3^1 + 2v_4^2 (\mathbf{A}_2'') \mathbf{a} \leftarrow v_4$	5025.8	103	8.53E-23	1	6/6	1.58E-24	1.6	0.17
$v_3^1 + 2v_4^2 \left(\mathbf{A}_1^{\prime\prime} \right) \mathbf{a} \leftarrow v_4$	5025.4	81	5.67E-23	1	4/4	1.96E-24	2.5	0.10
$\nu_1 + 3\nu_2 \left(\mathbf{A}_1' \right) \mathbf{s} \leftarrow \nu_2$	4805.4	73	3.89E-23	33	1/8	2.98E-23	67.1	0.33
$2v_2 + v_3^1 + v_4^1 (E'') a \leftarrow 2v_2$	5355.5	105	3.44E-23	2	5/7	1.92E-24	3.6	-0.10

Continued on next page

		C20	18		This we	ork		
Vibrational	E_0^{b}	Nb ^c	Band Int. ^d	Nb ^e	J_{min}/J_{max}^{f}	Band Int. ^g	Int. ratio ^h	$<\delta>^i$
band ^a	(cm^{-1})		(cm/molecule)			(cm/molecule)	(%)	(cm^{-1})
$v_1 + 2v_4^0 (\mathbf{A}_2'') \mathbf{a} \leftarrow v_4$	4978.4	74	2.73E-23	1	3/3	1.15E-24	3.5	0.09
$v_2 + v_3^1 (E'') a$	4435.4	33	2.72E-23	13	11/14	2.05E-23	79.6	0.04
$v_1 + 3v_2 (A'_1) s$	5737.9	35	2.59E-23	21	4/8	2.79E-23	89.2	0.42
$6v_2(A'_1) s \leftarrow v_2$	5110.9	22	2.48E-23	4	8/9	1.87E-23	63.0	0.00
$3v_2 + 2v_4^2(E')s$	5622.6	61	2.30E-23	3	1/5	3.16E-24	11.9	0.87
$4v_2 + v_4^1(E'')$ a	5708.3	51	1.81E-23	1	7/7	2.52E-25	1.3	0.12
$2v_2 + v_3^1 + v_4^1 (\mathbf{E}') \mathbf{s} \leftarrow v_4$	5091.0	84	1.77E-23	1	6/6	6.06E-25	0.6	-0.35
$3v_2 + v_3^1(\mathbf{E}'') \mathbf{a} \leftarrow v_2$	5398.9	79	1.72E-23	9	2/7	4.00E-24	21.6	0.11
$v_2 + v_3^1(E') s$	4416.9	25	1.54E-23	16	12/15	1.39E-23	90.9	0.12
$3v_2 + v_3^1(E') s$	5856.1	35	1.21E-23	5	7/9	5.37E-24	37.6	0.22
$v_3^1 + 2v_4^2 (A_2'') a \leftarrow 2v_2$	5054.6	43	9.89E-24	1	6/6	8.00E-25	7.8	0.18
$3v_2 + 2v_4^0(\mathbf{A}_2'')\mathbf{a} \leftarrow v_2$	5201.1	17	9.56E-24	2	9/9	6.11E-24	40.2	0.27
$v_1 + 3v_2 (\mathbf{A}_2'') \mathbf{a} \leftarrow v_2$	5297.9	35	6.86E-24	2	4/6	8.34E-25	8.7	0.57
$v_1 + 2v_2 + v_4^1(\mathbf{E}') \mathbf{s} \leftarrow v_4$	4955.9	32	5.71E-24	1	6/6	5.03E-25	9.3	-0.30
$2v_2 + v_3^1 + v_4^1 (E') s \leftarrow v_2$	5784.8	19	4.87E-24	1	6/6	4.80E-25	4.2	-0.35
$v_1 + v_2 (A'_1) s$	4294.5	5	2.41E-24	3	12/13	2.10E-24	85.8	0.10
$v_2 + 3v_4^3 (A_1') s$	5718.7	3	2.26E-24	1	4/4	1.35E-24	43.0	0.50
$v_1 + v_2 (A_2'') a$	4320.0	8	2.02E-24	4	12/14	1.40E-24	64.5	0.02
$v_2 + 2v_4^2 (E') s$	4135.8	5	1.89E-24	1	13/13	8.83E-25	40.5	0.42
$v_2 + 2v_4^2 (E'') a$	4192.9	4	1.62E-24	2	11/13	1.10E-24	78.9	0.29
$v_2 + 3v_4^3 (A_1') s \leftarrow v_2$	4786.3	4	1.54E-24	1	4/4	5.74E-25	32.2	0.49
$v_1 + 3v_2 (\mathbf{A}_2'') \mathbf{a} \leftarrow 2v_2$	4632.8	4	4.66E-25	1	6/6	1.15E-25	26.8	0.57
$v_2 + 2v_4^0 (A_2'') a$	4173.1	1	4.13E-25	1	12/12	2.88E-25	100.0	-0.11
Sum		16759	7.384E-19	6562		7.389E-19	•	
Others ^j		1182	0.005E-19	2048		0.03E-19		
Total ^k		17941	7.389E-19	8610		7.43E-19		

Table 2 – Continued from previous page

Notes:

^{*a*} Vibrational band labeling: v_i (i = 1 - 4) are the vibrational normal modes corresponding to the v_1 symmetric stretch, v_2 symmetric bend, v_3 the antisymmetric stretch and v_4 antisymmetric bend, respectively. $L_3 = |l_3|$ and $L_4 = |l_4|$ are the vibrational angular momentum quantum numbers of the v_3 and v_4 modes, respectively.

^b Energy of the lowest level of the upper vibrational state (corresponding generally to J = K = 0 as calculated in the C2018 list. In the case of the hot bands, the J = K = 0 value of the lowest state was subtracted.

^c Number of lines of the corresponding band with C2018 intensity larger than 1×10^{-25} cm/molecule.

^d Sum of the C2018 intensities of given band, larger than 1×10^{-25} cm/molecule. Note that the C2018 intensities are calculated for pure ¹⁴NH₃.

^e Number of lines of the corresponding band assigned in the studied spectrum.

^f Minimum and maximum values of the rotational quantum numbers of the assigned lines.

^g Sum of the experimental intensities derived from the optical depth, temperature and total pressure of the spectrum for the considered band.

^{*h*} Ratio of the sum of the C2018 intensities of the transitions identified in the studied spectrum by the total C2018 intensity of the considered band.

^{*i*} Average value of the (exp.- C2018) position differences of the considered band.

Continued on next page

Table 2 – Continued from previous page										
Vibrational $E_0^{\ b}$ Nb^c Band Int. ^d Nb^e J_{min}/J_{max}^f Band Int. ^g Int. ratio ^h <										
band ^a	(cm^{-1})		(cm/molecule)			(cm/molecule)	(%)	(cm^{-1})		
^{<i>j</i>} For the C2018 line list, "others" correspond to the intensity sum of the predicted vibrational bands absent in the list,										
while for the experimental	list of trans	itions. it	corresponds to the	intensity	sum of unassi	gned transitions.	Removing			

while for the experimental list of transitions, it corresponds to the intensity sum of unassigned transitions. Removing the $^{15}NH_3$ components, the measured intensities of the assigned transitions represent 99.5% of the total measured intensities.

^k Note that the C2018 intensities of assigned transitions represent 98.85% of the total C2018 intensities in the region.

Figure 4 compares the set of unassigned lines to that of the C2018 lines with no experimental counterpart. All unassigned lines have an intensity smaller than 2.5×10^{-23} cm/molecule. Except for one transition, all the C2018 transitions with intensity above 2.5×10^{-23} cm/molecule have been found. Overall, the appearance of the two data sets is similar. We nevertheless note that the unpaired C2018 transitions have a larger intensity than the unassigned lines at the high energy edge near 5600 cm⁻¹ while the situation is opposite near the low energy edge around 4800 cm^{-1} . Extra experimental absorption could be partly due to NH₂D which has a natural abundance of about 3.4×10^{-4} which leads to transitions with rough estimated intensity of a few 10^{-24} cm/molecule for the stronger lines. Let us recall that the identified ¹⁵NH₃ lines have been removed from the experimental panel.

¹²³ 5. Upper state energies of ¹⁴NH₃

¹²⁴ By adding the lower state energies [9] to the measured experimental line positions, the upper state empirical energies of 2215 ¹²⁵ ¹⁴NH₃ rovibrational levels were derived from the assigned transitions. The determined upper levels were reached by up to 11 ¹²⁶ transitions. 560 levels were derived from a single transition, 496 from two transitions and the remaining (1151) from more than ¹²⁷ two transitions. In the last case, the average value and standard deviation were calculated in two ways, for all the transitions and ¹²⁸ for a 30% trimmed selection of them. In the last case, the *rms* of the standard deviation over the determined energies was reduced ¹²⁹ from (0.9 ± 1.8) × 10⁻³ cm⁻¹ to (0.3 ± 0.3) × 10⁻³ cm⁻¹. The list of derived energy levels is given in Supplementary Materials.

The determined ¹⁴NH₃ energy levels have been highlighted on Figure 5 which presents the overall set of C2018 energy values for upper states of transitions predicted in the 4705-5650 cm⁻¹ interval, with intensity larger than 1×10^{-25} cm/molecule. The average rotational term $[9.7J(J + 1) - 3.6K^2]$ have been subtracted for the clarity of the plot.

Three groups of energy levels are observed as upper levels of transitions located in our region of interest. Their reduced energies belong to the 4500-5500, 5500-6100 and 6100-7000 cm⁻¹ ranges depending if they were determined from cold transitions, hot transitions starting from v_2 , hot transitions from $2v_2$ and v_4 , respectively.

Among the 2215 upper state energy levels determined in this work, 1939 are new compared to the MARVEL set of levels of Ref. [9]. Figure 6 presents the differences between upper state energies and the calculated C2018 values, *versus* the quantity [J + K/(J + 1)]. Except for a few outliers, the agreement with MARVEL values [9] is very good (average and *rms* values of 2×10^{-4} and 7×10^{-4} cm⁻¹ for the (MARVEL-exp.) differences, respectively). A small overestimation of the MARVEL values of 2×10^{-4} cm⁻¹ is noted (right panel). As concerns the C2018 values [11], the deviations remain in a ±0.6 cm⁻¹ range.



Figure 5: Upper state energies of the ¹⁴NH₃ transitions of the C2018 list in the 4705-5650 cm⁻¹ interval predicted with intensity larger than 1×10^{-25} cm/molecule (grey dots). The average rotational term $[9.7J(J + 1) - 3.6K^2]$ has been subtracted for clarity. The experimental values determined in this work are superimposed (orange dots) together with the existing empirical values provided by the MARVEL procedure (green squares)[9]. Some vibrational hot and cold bands from which the energies have been derived are indicated. Note that in some cases, the same upper levels are reached through both hot and cold bands.



Figure 6: Differences between the C2018 [11] (grey) and MARVEL (green) [9] energies of the upper levels of $^{14}NH_3$ and the corresponding values derived in the present work. The values corresponding to upper levels derived from four vibrational bands are highlighted.

141 6. Comparison with the HITRAN line list

The overview comparison of our experimental list to the HITRAN2020 list (Fig. 1) illustrates the significant amount of new information provided by the analysis. Unassigned and assigned 14 NH₃ lines, as well as minor 15 NH₃ isotopologue lines can be distinguished on the plot. Our 14 NH₃ line list includes about four times more lines than the HITRAN2020 list. As for the 3900-4700 cm⁻¹ range [3], the 15 NH₃ transitions contribute to about 0.4% of the absorption in the studied range which is consistent with the natural relative abundance of 0.366% [5]. Note that both our work and HITRAN database do not consider the possible

¹⁴⁷ contribution of weak NH₂D lines.

Table 3: Statistics and integrated absorption in various databases for ammonia in the 4703-5650.3 cm⁻¹ spectral range (intensity cut-off of 1×10^{-25} cm/molecule at 296 K).

$ \begin{array}{c c c c c c c c c c c c c c c c c c c $		HITRAN2008 ^a		HITRAN2016		HITRA	N2020	This work		C2018
Transitions20011291875126249312686324651794Assignments7711296391261255 ^b 126656201794Intensity sum7.660.01467.650.01467.660.01467.420.0347.39		¹⁴ NH ₃	$^{15}\mathrm{NH}_3$	$^{14}NH_3$	$^{15}\mathrm{NH}_3$	¹⁴ NH ₃	$^{15}\mathrm{NH}_3$	$^{14}\mathrm{NH}_3$	¹⁵ NH ₃	¹⁴ NH ₃
Assignments 771 129 639 126 1255 ^b 126 6562 0 1794 Intensity sum 7.66 0.0146 7.65 0.0146 7.66 0.0146 7.42 0.034 7.39	Transitions	2001	129	1875	126	2493	126	8632	465	17941
Intensity sum $7.66 0.0146 7.65 0.0146 7.66 0.0146 7.42 0.034 7.39$	Assignments	771	129	639	126	1255 ^b	126	6562	0	17941
	Intensity sum (×10 ⁻¹⁹ cm/molecule)	7.66	0.0146	7.65	0.0146	7.66	0.0146	7.42	0.034	7.39

Notes:

 a 140 assignments were lost from the 2008 to 2016 editions of the HITRAN database with no explanation.

^b This number is the sum of the HITRAN2016 transitions (639) and those added using the MARVEL process (616) [9].

In Fig. 7, a part of the 140.4 Pa spectrum is compared to a simulation based on the HITRAN2020 list around 5288 cm⁻¹. Transitions with experimental line parameters transferred from HITRAN2016 are distinguished from the transitions newly added through the MARVEL procedure. In general, the latters show a nice agreement in position and intensity. A number of additional lines are observed in our FTS spectrum illustrating the achieved sensitivity.

We have performed a systematic comparison of our assignments to those attached to the HITRAN2020 transitions. Among the 152 1275 transitions present in HITRAN2020 in our region, only 756 can be found within a 0.005 cm⁻¹ interval around the measured 153 position value. Indeed, a significant number of HITRAN transitions missing in our list corresponds to newly added transitions with 154 MARVEL line positions and C2018 intensities which are too weak to be detected in our spectra (see Fig. 1). 70 of the transitions 155 with coincident positions show a large intensity disagreement. Part of these intensity mismatches might be due to accidental 156 position coincidence between a weak HITRAN transitions with MARVEL origin and a strong measured line. Among the 686 157 remaining transitions in common, the assignment fully coincides for 362, is incompatible for 53 as J' value differs and, for the 158 remaining part (271), some differences are observed. For example, J and K are correct but the vibrational assignment is different, 159 mainly interchanged between $v_1 + v_4$ and $v_3 + v_4$ above J = 7. 160

161 7. Intensity considerations

As the assignment process is supported by the intensity matching between the experimental spectrum and the C2018 calculations, it is important to discuss their level of agreement. The corresponding intensity ratios are plotted in Fig. 8 where the values relative to the main vibrational bands - $v_3 + v_4$ (E), $v_1 + v_4$ (E) and $v_3 + v_4$ (A)) - are highlighted. These bands represent 63.5 % of the total number of assigned transitions but bring 95.4 % of the total intensity. Overall, the agreement between the experimental



Figure 7: FTS spectrum at 140.4 Pa in the $5285-5290 \text{ cm}^{-1}$ range (blue line) compared to HITRAN2020 simulation (red): four lines came from experimentally observed data and were present in HITRAN2016 issue, and three lines indicated with a "M" corresponds to transitions added in HITRAN2020 from MARVEL process [8]. All the other lines are not present in HITRAN2020.



Figure 8: Ratios of the ¹⁴NH₃ line intensities measured in this work in the 4705-5650 cm⁻¹ spectral region to the corresponding C2018 values [11]. The intensity ratios corresponding to the $v_3 + v_4$ (E), $v_1 + v_4$ (E) and $v_3 + v_4$ (A) strong bands are distinguished (red, blue and green, respectively). The highlighted outlier corresponding to a strongly underestimated C2018 intensity is discussed in the text as an example of intensity transfer.

and calculated values is excellent. The increase of the dispersion for small line intensities is a usual phenomenon partly due to the larger relative uncertainty of the small experimental intensity values. However, there are some cases of strong transitions showing a large disagreement as for example, the transition measured at 4967.3479 cm⁻¹ with intensity of 7.09×10^{-22} cm/molecule, (orange square on Fig. 8) which is predicted to have a ten times smaller C2018 intensity. This particular example is analyzed in details just below.

Remember that the first step of the assignment process is to associate C2018 and experimental lines with similar intensities and close line centers. Obviously, in the considered example, the intensity mismatch prevented the association to the C2018 transition with closest position. Nevertheless, loosening the intensity criterion, the LSCD search gives a reasonable agreement for the C2018 transition at 4967.2327 cm⁻¹ with a smaller intensity of 7.3×10^{-23} cm/molecule. This transition belonging to the $\nu_1 + 2\nu_2$ vibrational band starts from J = 7, K = 6 s level and reaches J = 6, K = 3 s upper level with an energy of 5389.6889 cm⁻¹. The systematic search for the LSCD partners revealed three hot band transitions assigned in our previous study at lower energy [3] and four matches in the interval presently studied, thus confirming the assignment in spite of a strong underestimation of the



Figure 9: Calculated [11] and experimental transitions reaching the $N_B = 80$ upper level of the block (J = 6, $\Gamma_{tot} = A_2^{\prime\prime}$ at 5389.6889 cm⁻¹). The experimental transitions of the $v_1 + 2v_2 - v_2$ band were reported in Ref. [1] while the $v_1 + 2v_2$ cold band transitions are observed in the present work. The transition under study is the one at 4967.3479 cm⁻¹ with a C2018 intensity underestimated by a factor of 9.68. The C2018 values are plotted with different symbols (open squares and open triangles) according to the inversion symmetry (a or s, respectively) of the lower state. The corresponding exp./C2018 intensity ratios are indicated for each transition. The experimental counterparts (orange circles) were found in this work and in Ref. [3].

- C2018 intensity. While the hot transitions show an almost perfect intensity agreement, the FTS/C2018 intensity ratio for the five 178
- $v_1 + 2v_2$ band transitions vary from 1.09 up to 9.7 (Fig.9). 179

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- Let us recall that the only two good quantum numbers for ${}^{14}NH_3$ are J and Γ_{tot} . The other labels given in C2018 list are 180 only approximate. In fact, each wavefunction is found through the diagonalization of the Hamiltonian inside J, Γ_{tot} blocks and is 181
- expressed as a linear combination of basis wavefunctions characterized by (J, Γ_{tot}) and completed by quantum labels v_1, v_2, v_3, v_4 , 182
- L_3 , L_4 , τ_{inv} , K, Γ_{vib} , Γ_{rot} . The labeling is completed by the rank in the block, N_B , which unambiguously designates the N_B^{th} level 183
- of the (J, Γ_{tot}) block ordered by increasing value of the energy. The upper state involved in the discussed example corresponds to 184
- $N_B = 80$ in the (6, A') block and has an experimental energy of 5389.80603 cm⁻¹. The experimental value of the energy separation 185
- from the next $N_B = 81$ level is only 0.225 cm⁻¹. The $N_B = 81$ upper state is the upper state of strong transitions and we interpret 186 the large intensity of the $v_1 + 2v_2$ transitions reaching the $N_B = 80$ level as due to an intensity borrowing from those involving the
- $N_B = 81$ levels. It is interesting to note that the energy separation of the $N_B = 80$ and $N_B = 81$ levels is significantly smaller than its 188

 189 C2018 value of 0.2945 cm⁻¹. This closer energetic resonance might explain the fact that the resulting intensity transfer is much larger than predicted by theory.

Table 4 presents the experimental and calculated energies and intensities of the five LSCD pairs of the $v_1 + 2v_2$ band together with the corresponding intensity ratios. If we sum the experimental intensities of the doublets of transitions reaching the $N_B = 80$ and 81 levels (last set of columns in Tab. 4), and compare them to the corresponding C2018 sums, an agreement within the experimental error bar (5%) is obtained.

Table 4: An example of C018 intensity transfer between the transitions arriving into two close upper states of the same J', Γ'_{tot} subspace.

			$J' = 6, \Gamma'_{tot} =$	$A_2'', N_B' = 80$			$J' = 6, \Gamma'_{tot} =$	$A_2'', N_B' = 81$					
Lower	State		$E'_{C2018} = 538$	9.6889 cm ⁻¹			$E'_{C2018} = 538$		Sums and ratio				
			$E'_{emp} = 5389$	$.80603 \text{ cm}^{-1}$			$E'_{emp} = 5390$						
$J^{\prime\prime} \tau^{\prime\prime}_{inv} N^{\prime\prime}_B$	E_{C2018}''	v_{TW}	S _{TW}	S_{2018}	r _{TW/C2018}	v_{TW}	S _{TW}	S C2018	r _{TW/C2018}	Σ_{C2018}	Σ_{TW}	r_{Σ}	
	(cm^{-1})	(cm ⁻¹)	(cm/mol.)	(cm/mol.)		(cm ⁻¹)	(cm/mol.)	(cm/mol.)		(cm/mol.)	(cm/mol.)		
7 a 2	522.2222	4867.5832	2.00E-23	1.85E-23	1.09			1.07E-24		1.95E-23	2.00E-23	1.03	
7 s 1	422.4562	4967.3479	7.09E-22	7.32E-23	9.68	4967.5729	3.01E-21	3.62E-21	0.83	3.69E-21	3.72E-21	1.01	
6 a 2	383.9772	5005.8282	1.55E-23	6.21E-24	2.49	5006.0535	1.25E-23	2.27E-23	0.55	2.89E-23	2.80E-23	0.97	
6 s 1	283.5719	5106.2317	2.22E-22	2.29E-23	9.70	5106.4570	9.59E-22	1.20E-21	0.80	1.23E-21	1.18E-21	0.96	
5 a 1	265.2263	5124.5797	3.75E-23	1.87E-23	2.00	5124.8042	3.12E-23	5.42E-23	0.58	7.29E-23	6.86E-23	0.94	

The situation observed for the $N_B = 80$ and 81 pair of levels is not isolated and many other cases of intensity transfer were found during the assignment process. This is also the case in other spectral regions. For instance, some examples can be found in our preceding study in the 5650-6300 cm⁻¹ range [1] and, Huang *et al.* [17] pointed out two similar occurrences in their re-analysis of the ammonia spectrum in this region.

Let us now try to select systematically the states with possible interaction with their nearby states in our range of interest. In the 4705-5650 cm⁻¹ interval, 1061 doublets of transitions with intensity above 1×10^{-25} cm/molecule have been found in the C2018 list with upper energy levels separated by less than 2 cm^{-1} (in identical blocks). In addition, 115 triplets, 16 quadruplets and 2 quintuplets fulfill this criterion. Loosening the transition cut-off, the number of possibly interacting transitions groups rises up to 1724 from previously mentioned 1194 ones. Comparing this list to our dataset we have found all experimental counterparts for 266 cases (250 doublets, 13 triplets and 3 quadruplets).

To test possible intensity transfer within the identified multiplets, we present in the upper panel of Fig. 10 the exp./C2018 inten-205 sity ratios with different symbols for the transitions reaching the resonant multiplets and the isolated levels. The first observation is 206 that most of the large deviations correspond to resonant levels. In the lower panel of Fig. 10, we plotted the ratios for the intensity 207 sum of the resonant transitions. The obtained result is impressive: while a significant fraction of the individual components show 208 important disagreement between the experimental and C2018 intensities, their intensity sums coincide very well. We believe that 209 the sensitivity of the wavefunctions to small inaccuracies of the potential energy surface could explain the phenomenon: a small 210 variation of the energy separation of the interacting upper levels leads to a large intensity change for the weak component of the 211 resonant doublets but the resulting intensity transfer does not affect the intensity sum of the involved transitions. Note that at higher 212 energy the number of resonances within blocks increases sharply leading to more frequent intensity transfers and might degrade the 213 agreement with C2018 intensity calculations. The results presently obtained can be used to improve future theoretical predictions 214 of line intensities at higher energy. 215



Figure 10: Upper panel: S_{exp}/S_{C2018} intensity ratios for the ¹⁴NH₃ transitions assigned in the 4705-5650 cm⁻¹ region. According to the energy separation of the upper energy level in a given (J', Γ'_{tot}) subspace, a different symbol is used: black circles and cyan points when the energy separation from the neighbour level is more or less than 2 cm^{-1} , respectively. Lower panel: Intensity ratios for cases when all the possibly interacting components in a given group (belonging to the same J', Γ'_{tot} with $\Delta E < 2 \text{ cm}^{-1}$ and sharing the same lower state) were identified experimentally, individual ratio (cyan dots) and ratio of corresponding intensity sums (red /yellow dots).

216 8. Conclusion

New high resolution and low-pressure FTS recordings have allowed an in depth analysis of the room temperature ammonia 217 spectrum from 4703 to 5650.3 cm⁻¹ range. Among the 9110 measured lines, 505 are identified as due to ¹⁵NH₃ minor isotopologue, 218 and a rovibrational assignment is proposed for 6562 ¹⁴NH₃ lines. This is a significant improvement compared to the HITRAN2020 219 list [5], which provides 2493 ¹⁴NH₃ transitions (with 1255 assignments), 626 of them predicted by theory and not experimentally 220 observed. Overall, our assigned transitions belong to 35 cold and 26 hot bands and bring over 98.8% of the integrated absorption 221 in the region. 2215 upper energies were derived with accuracy better than 0.001 cm^{-1} . Among them 1939 are new, compared to 222 those derived in Ref. [9] using the MARVEL procedure. A recommended list of 9110 transitions for ammonia in natural isotopic 223 abundance is provided as Supplementary Material. The ¹⁴NH₃ line positions were calculated using our best empirical determination 224 of the upper energy levels (estimated accuracy on the order of 5×10^{-4} cm⁻¹). Despite the quality of *ab initio* calculations to 225 accurately predict position and intensity of the ¹⁴NH₃ transitions [11] and the power of the Lower State Combination Difference 226 technique to assign them, a number of important disagreements between the measured and C2018 intensities was evidenced. Most 227 of the intensity mismatches concern weak transitions reaching an upper level with a close lying neighbour level reached by a 22 stronger transition. As the intensity sum of the two interacting transitions coincides with the C2018 predictions, we conclude that 229 the problematic situations result from intensity transfer between interacting multiplets. In those situations, line intensities provide 230 a sensitive tool to reveal small inaccuracies of the potential energy surface used to generate the C2018 list [11]. As a result, the 231 intensity agreement with C2018 calculations which is in general highly valuable for rovibrational assignment, should be used with 232 caution, in particular at higher energy where the level density increases rapidly making resonance interactions more and more 233 probable. In particular, the rovibrational assignments of the ammonia spectrum in the 1.5 μ m range (6300-7000 cm⁻¹) already 234 studied in Refs. [18, 19] might be extended by a careful consideration of possible intensity transfers. 235

236 Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

239 CRediT authorship contribution statement

P. Cacciani: Investigation. P. Čermák: Investigation. J. Vander Auwera: Investigation. A. Campargue: Investigation.

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